

Characterization of Motor Oils and Other Lubricants by High Performance Liquid
Chromatography, Three-Dimensional Excitation Emission Matrices and Two-
Dimensional Low Temperature Fluorescence Spectroscopy

by

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Abstract

Characterization of Motor Oils and Other Lubricants by High Performance Liquid Chromatography, Three-Dimensional Excitation Emission Matrices and Two-Dimensional Low Temperature Fluorescence Spectroscopy

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Criminals often use automobiles during the commission of crimes. Criminalists routinely analyze automotive transfer evidence, including paint, rubber and glass in order to establish an association between a suspect car and a person or object that may have come in contact with that car. Currently, there is no routine forensic method to analyze transferred automotive undercarriage residue. This feasibility study will use fluorescence spectroscopy, both room-temperature and low-temperature (77K), and HPLC, to analyze this material and will investigate the use of compiled spectral libraries to assist in determining the evidentiary value of this process for this sample set.

Automobile undercarriage residues are a complex mixture of compounds from many sources including automotive fluids, asphalt residues, exhaust particulates and tire particulates. Each of these sources may contain compounds that fluoresce when

excited with light in the ultraviolet or visible region of the electromagnetic spectrum. Given the many sources of fluorescing compounds and the effect of environmental conditions on these compounds, it is hypothesized that the two-dimensional and three-dimensional fluorescence spectrum of automobile undercarriage residues along with HPLC analysis can be used to differentiate among automobiles.

Five undercarriage locations were swabbed from ten motor vehicles. Extraction via sonication in different solvents separated each residue into fractions. These fractions were analyzed by fluorescence spectroscopy and HPLC with UV detection. The types of fluorescence spectroscopy used were emission scans, synchronous scans, excitation-emission matrices, and low-temperature fluorescence spectroscopy. Previous studies with fluorescence characterization of lubricants relied on visual examination of the spectral data to determine differences or similarities (Purcell 2002). This study will use visual examination, peak number and position, chromatographic data and spectral libraries to show similarities and differences among the samples.

The research contributes to the body of knowledge about analytical methods that may characterize a long-observed transferable material, automotive undercarriage residue, to create associations between automobiles and persons or objects. These associations can be used to help the trier-of-fact decide upon the likelihood that an automobile came into contact with a person or object. Additionally, this research makes contributions to the field of environmental science which seeks to characterize petroleum-derived organic contaminants.

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Table of Contents

Abstract	iv
Introduction	1
Instrumental Theory	5
Literature Review	12
Methods	23
Data Collection	47
Results	72
Discussion	92
Appendices	105
Bibliography	446

List of Figures and Tables

Figure 1. A Joblonski diagram of the processes that result in luminescence.	6
Figure 2. Flow Chart of Sampling and Extraction Scheme for a Single Motor Vehicle	27
Figure 3. Pentane Extraction Blank	31
Figure 4. Toluene Extraction Blank	31
Figure 5. Pyridine Extraction Blank	32
Figure 6. Acetonitrile Blank	33
Figure 7. Methylcyclohexane Blank	33
Figure 8. Chromatograms from triplicate swabs of Vehicle 1 Location 3 (13)	40
Figure 9. Anthracene/Naphthalene Polymer Block Standard – Single EEM.	44
Figure 10. Anthracene/Naphthalene Polymer Standard – Excitation 290 nm. All scans.	45
Figure 11. Anthracene/Naphthalene Polymer Standard – Excitation 320 nm. All scans.	45
Figure 12. Anthracene/Naphthalene Polymer Standard – Excitation 360 nm. All scans.	46
Figure 13. Known Sample 41A at Scan rate 1200 nm/min	50
Figure 14. Known Sample 41 A at Scan rate 600 nm/min	50
Figure 15. 83BT Before Rayleigh-Tyndell Scatter removal.	54
Figure 16. 83BT After Rayleigh-Tyndell Scatter Removal	55

List of Figures and Tables (continued)

Figure 17. Farand Dewar cap with quartz-tipped glass sample tube.	57
Figure 18. Farand Dewar cap top view showing vent holes.	57
Figure 19. Flow Manifold	60
Figure 20. Quartz Tipped Dewar	61
Figure 21. Cary Eclipse sample compartment with Dewar housing.	61
Figure 22. Modified interior for LT Fluorescence (Dewar housing not pictured).	62
Figure 23. Operational View of Low Temperature Set-Up	63
Figure 24. Clear freeze of methylcyclohexane	64
Figure 25. Low Temperature Data Collection	67
Figure 26. Low Temperature Scans – Full Data Set	68
Figure 27. 51CT in Dewar at Room Temperature	70
Figure 28. 51 CT in Dewar at 77 K	70
Figure 29. Region A, B, and C Excitation – Emission Regions. (Item 22A, pentane extract)	73
Figure 30. Summary of hits obtained using spectral libraries. (Yellow = hit to correct vehicle, pink = hit to correct vehicle and specific location).	87
Figure 31. Percentage of EEMs collected and the distribution of Fluorescent intensities.	94

List of Figures and Tables (continued)

Table 1. Vehicle Sample Set and Specific Locations	23
Table 2. Extraction Method	24
Table 3. HPLC Method	37
Table 4. Standard PAH Solutions	38
Table 5. HPLC Detector Linearity - Phenanthrene	38
Table 6. HPLC Detector Linearity - Naphthalene	39
Table 7. Anthracene/Naphthalene Polymer Block Standard – Intensity and Emission values.	44
Table 8. Cary Eclipse Setting for Room Temperature Data Collection	47
Table 9. Cary Eclipse Settings for Low Temperature Data Collection	65
Table 10. Classification with Pentane Extract EEM Spectra.	74
Table 11. Classification with Toluene Extract EEM Spectra	76
Table 12. Coded samples using EEM peak number and position.	78
Table 13. Classification with Pyridine Extract EEM Spectra.	83
Table 14. Determination of Unknown Residue Origin Using HPLC Data.	90

List of Appendices	105
A. Unknown Labeling Scheme	106
B. Cary Eclipse Specifications	107
C. Pentane Extract Chromatograms	108
D. Pentane Extract Contour Plots	267
E. Toluene Extract Contour Plots	327
F. Pyridine Extract Contour Plots	386

Introduction

The criminalist's analysis of transfer evidence can be a powerful tool to create associations. Incidents involving automobiles often yield several different types of transfer evidence. Transferred paint is primarily analyzed by light microscopy, infrared microspectrophotometry, pyrolysis gas chromatography-mass spectroscopy and scanning electron microscopy with energy dispersive X-ray spectrometry (Caddy 2001). The density, refractive index and elemental composition of transferred glass can be established. Transferred rubber is identified by attenuated total reflectance microspectrophotometry and pyrolysis gas chromatography. The results of these multi-instrumental analyses are compared to find a likelihood of association between an automobile and a person or object. Historically, this likelihood has been unquantifiable but forensic science researchers and practitioners are beginning to adopt a Bayesian approach to generating likelihood ratios (Taroni *et al*, 2010).

Missing from the list of regularly analyzed transfer evidence is automotive undercarriage residue. This material, mainly consisting of petroleum-derived products, is a complex, dynamic and easily transferable material. This study investigated the usefulness of liquid chromatography to adjust the concentration of fluorescing compounds, fluorescence spectroscopy to characterize automotive undercarriage residue and visual examination coupled with spectral library searching techniques to gauge the discriminatory power of the latter technique.

Contribution to the Field of Forensic Science and Criminal Justice

The contribution of this research to forensic science is the ability to use a long-observed transferable material, automotive undercarriage residue, to create associations between automobiles and persons or objects. These associations can be used to help the trier of fact determine the likelihood that a specific automobile came into contact with a person or object. This project illustrates the discriminatory power of transferred petroleum materials and the usefulness of fluorescence spectroscopy and high performance liquid chromatography with UV detection. This type of evidence is found in “hit and run” type vehicular crimes and in environmental crimes.

1. Vehicular Crimes

Incidents where a vehicle comes in contact with a person or object result in a several different types of transfer evidence. Glass, paint, and rubber transferred from a vehicle to person or object, and fibers and biological evidence transferred from person to vehicle are routinely analyzed to determine if contact took place. At present, the transfer of petroleum products from the vehicle to a person or object are not routinely analyzed. I suggest that automotive undercarriage residue analysis could be valuable in creating an association between a suspect car and person or object that came in contact with the underside of that vehicle. This association becomes valuable in the case of a hit-and-run incident where investigators may need to connect a suspect vehicle, not found at the scene but located at a later time, to a victim. If the victim made contact with the underside of the offending vehicle it is very likely that there was a transfer of automotive

undercarriage residue. Characterizing the fluorescing components of this residue by conducting a spectral comparison of the transferred and native materials could prove especially valuable in the efforts to associate a vehicle to a victim. This is especially true when there are no classic types of transfer evidence (e.g paint, glass) recovered or it is of little evidentiary value. In cases where the identity of the vehicle is known but the details of the collision are contested, automotive undercarriage residue may be helpful for reconstruction purposes (Purcell, 2002).

2. Environmental crimes

Oil spills on both ground and water are a much too frequent occurrence. These events have environmental, legal and financial consequences to the municipalities that detect and oversee the clean-up of the spilled petroleum products. For the fiscal year 2007 there were 527 oil spill incidents, both water and ground, in New York City (NYC DEP, 2007). Clean-up costs can vary wildly depending on such factors as type and amount of oil spilled, rate of spillage, spill location, weather, season and competence of clean-up personnel. It is not always clear, especially in heavily industrialized areas, where the spill originated and therefore what organization or company is the responsible party. Careful chemical analyses of the spilled material using chromatographic and spectroscopic methods have been successful in establishing the source of spilled petroleum materials (Zieba, 1985). This current research, despite its focus on automotive sources of transferred petroleum products, builds on the body of work regarding environmental crimes. Similar laboratory techniques can be used for environmental soil

or water samples. If this method proves to be highly discriminatory it would be extremely helpful in cases where there are hundreds of potential polluters. If culpability can be established in a greater percentage of oil spill events, then more of the clean-up costs can be shifted away from the taxpayer and onto the responsible party.

Instrumental Theory

1. Fluorescence and Phosphorescence

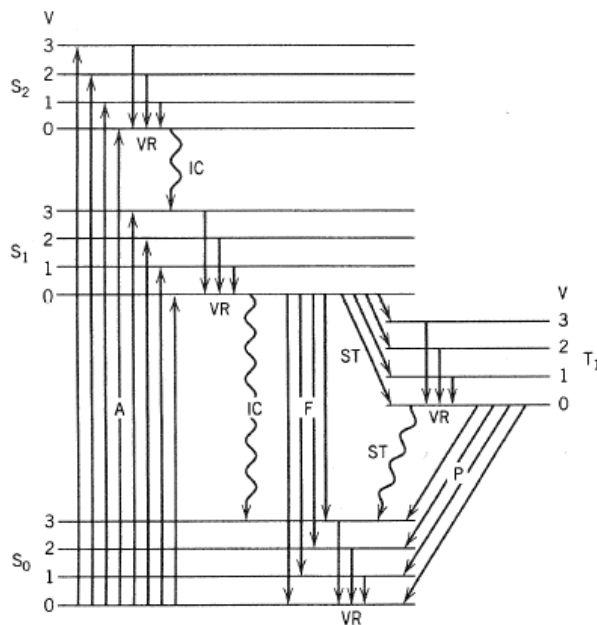
All electrons in an atom are represented by four quantum numbers; n , l , m_l , m_s . The principle quantum number, n , represents the energy of the orbital. The azimuthal quantum number, l , determines the shape of the orbital and the angular momentum of the electron. The magnetic quantum number, m_l , determines the orientation of the orbital in space. The spin quantum number, m_s , describes the spin of the electron and can only be represented by the values, $+1/2$ or $-1/2$ (Sharma, A., Schulman, S. G., 1999).

Upon excitation with a photon, an electron goes to a higher energy state and maintains its original m_s value, the spin of that electron is still paired with the one left in the ground state and the total angular momentum is zero, $S=0$ therefore the multiplicity $(2S+1)$ equals one. When multiplicity equals 1 the excited electron is in the singlet state. Fluorescence occurs when an electron is excited to a singlet state and radiatively relaxes down to the ground state. This process is short-lived, usually 10^{-9} to 10^{-7} seconds (Sharma, A., Schulman, S. G., 1999).

If an electron is excited and has the same spin quantum number as the lone electron left in the ground state then the spin direction is parallel and the total angular momentum is equal to one and the multiplicity $(2S+1)$ equals three. When the multiplicity equals three, the excited electron is in the triplet state. Phosphorescence occurs

when an electron enters an excited triplet state and radiatively relaxes down to the ground state. This process is much longer comparatively ranging from 10^{-5} to several seconds. Any condition that inhibits radiationless relaxation, lower temperature or a rigid medium, increases the amount of observable phosphorescence. Phosphorescence in solutions is observed less often than fluorescence mainly because of the non-radiative competing processes of collisional deactivation, quenching, photochemical reactions and other energy transfer process. Competing with both fluorescence and phosphorescence are several non-radiative processes including internal conversion and vibrational relaxation (Sharma, A., Schulman, S. G., 1999).

Figure 1. A Joblonski diagram of the processes that result in luminescence.



Source: Sharma and Schulman (1999)

When a molecule absorbs a photon of ultraviolet or visible wavelength, an electron is promoted to a higher energy state. This excited electron can return to the ground state through several processes. Luminescence occurs when this relaxation process results in the emission of a photon. If the electron radiatively transitions from the first singlet state back to the ground state then the emission of this photon is called fluorescence. Most observable luminescence from organic compounds comes from the excitation of an electron in a π orbital. If the electron radiatively transitions from the first triplet state to the ground state then the emission of this photon is called phosphorescence. An electron enters the first triplet state after an internal conversion from the excited singlet state. Because the transition can end on any one of the vibrational levels of the ground state, the phosphorescence band itself reflects the ground state of vibrational levels (Ingle, J. D., Crouch, S. R., 1988) (Schenk, 1973).

The excited electron may return to the ground state via non-radiative processes of internal conversion or vibrational relaxation. Whenever a non-radiative relaxation is energetically preferred over a radiative process the total amount of photons emitted by the sample will be reduced. The relationship between exciting photons and emitted photons is described as the quantum yield (Φ).

Fluorescent intensity at a specific wavelength is directly proportional to the number of photons absorbed. The non-radiative processes of intersystem crossing and

internal conversion compete with fluorescence for transition of the lowest excited singlet state electron and therefore not all potentially fluorescent molecules will return to the ground state by fluorescence. The amount of the excited molecules that fluoresce is called the quantum yield (Ingle, J. D., Crouch, S. R., 1988). Quantum yield represents the efficiency of the fluorescence emitted. The following equation states the relationship between fluorescence intensity and quantum yield (Guilbault, G. G., 1990).

$$F = [P_0(1 - e^{-\epsilon bc})] \Phi \quad \text{Equation 1.}$$

Where F is the fluorescence intensity, P_0 is the source intensity, ϵ is the molar absorptivity, b is the path length, c is the concentration and Φ is the quantum yield. Guilbault states that when Eq. 1 is subjected to a McLaren (sic) expansion, the relationship between fluorescence intensity and concentration is linear when fluorophore concentrations is low and when $\epsilon bc < 0.05$. This linearity is represented by Eq. 2.

$$F = [K\Phi P_0] \epsilon bc \quad \text{Equation 2.}$$

At higher concentrations, quenching causes deviations from linearity. Therefore, fluorescence spectrophotometry is a useful quantitative technique for concentrations within the linear range. Rigid structures have a much higher quantum yield than

compounds that are not rigid due to the inhibition of collisions that result in non-radiative relaxation and therefore will have a larger range of fluorophore concentrations that are linearly related to fluorescence intensity (Sharma, A., Schulman, S. G., 1999).

Fluorescence spectrometry is much more sensitive and selective than UV-Vis spectrophotometry. This makes it a very valuable technique for the analysis of very small concentrations (Schenk, 1973). Fluorescence intensity and concentration have a linear relationship only when the total absorbance of the system is below 0.05 absorbance units as measured on a UV-Vis spectrophotometer. Above this absorbance the relationship begins to plateau and could actual decrease due to the increased likelihood of solute-solute interactions.

Polycyclic aromatic hydrocarbons are highly fluorescing compounds. Polycyclic aromatic hydrocarbons (PAHs) contain the rigid aromatic structure necessary to inhibit radiationless relaxation that competes with fluorescence (Sharma, A., Schulman, S. G., 1999).

2. Effect of substituents

Any compound that allows the correct electronic transitions may fluoresce. Compounds with rigid aromatic rings are strong absorbers and are much less likely to

return to the ground state via non-radiative means, however not all aromatic systems behave in the same manner. Electron-donating groups (e.g. NH_2) bound to aromatic rings can increase fluorescent efficiency while electron-withdrawing groups (e.g. Cl^-) may quench fluorescence (Sharma, A., Schulman, S. G., 1999).

3. Effect of concentration

The same compound or mixture of compounds may exhibit differently shaped spectra at different concentrations. Shifts of the peak wavelength maxima to longer wavelengths are referred to as “red shifts.” Red shifts occur at higher concentrations and not only alter peak maxima, but may result in changes in peak shape. As solute-solute interactions increase it is possible for excited molecules to aggregate with ground state molecules producing an excited dimer. This excited dimer could relax to the ground state non-radiatively, essentially quenching the fluorescence from the excited single molecule. If the excited dimer relaxes radiatively, the luminescence will be at longer wavelengths than that from the single excited molecule. This aggregation affects the shape, position and intensity of the resultant fluorescence (Sharma, A., Schulman, S. G., 1999).

Changes in the position and shape with concentration of the fluorophore are due in part to self quenching and self absorption. Self quenching occurs when energy is transferred via the collision between excited molecules. This process does not result in radiation. Because of the increased likelihood of collision, self quenching increases with

concentration and contributes to deviations from the linearity of Beer's law. At higher concentrations there is an increase in solute-solute interactions. As it does in absorbance spectroscopy, an increase in these interactions causes deviations in the linearity between concentration and intensity of fluorescent light (Sharma, A., Schulman, S. G., 1999).

This study evaluated the use of chromatographic and gravimetric analysis to establish concentration. Proper concentration determination would ensure that all spectral differences and similarities seen in a comparison will be attributed to different sample composition and not different concentrations of the same composition.

Literature review

1. Early Work

Early work by the U. S. Coast Guard in the 1970's was successful in identifying the source of spilled oil using the complementary methods of fluorescence spectroscopy, IR spectroscopy, thin layer chromatography and gas chromatography. Each of these techniques is sensitive to different components in the sample. A multi-instrument approach is thorough but time consuming (Curtis, 1977).

2. SSE and VSSE

The first fluorescence spectra were made by either by keeping a constant excitation wavelength and scanning the emission wavelength range, or by keeping a constant emission wavelength and scanning the excitation wavelength range. Synchronous scan excitation (SSE) was a significant advancement. Both the excitation wavelength and the emission wavelength are scanned synchronously at a fixed wavelength interval ($\Delta \lambda$). The SSE technique usually results in more detailed spectra and therefore increased the discriminating power of fluorescence spectroscopy.

J.B.F. Lloyd addressed the evidential value of luminescence of engine oils and similar materials using synchronous scan excitation and emission fluorescence (SSE).

Lloyd was able to differentiate among samples with high molecular weight with the SSE technique (Lloyd, 1971, Part I, II, III).

SSE has been a very popular technique in the forensic analysis of transfer evidence. Rubber bumper guards and petroleum lubricants from sexual assaults were characterized by Blackledge (Blackledge 1980, 1983).

Pharr, McKinzie and Hickman used SSE to characterize gasoline, kerosene, diesel oil, asphalt and fuel oil in organic and aqueous extracts of ground water. They claimed this characterization is a “fingerprinting” technique. I find this terminology ill-fitting to describe the categorization they were successful in achieving. Any technique made analogous to fingerprint analysis should be an individualizing technique (Pharr et. al. 1992).

Patra and Mishra used SSE to classify diesel fuel, gasoline and kerosene. They actually induced the red shift seen with concentration dependent samples to characterize these materials and increase the discrimination of the technique. The wavelength shifting that the researchers used to characterize their samples is precisely why it is so important to establish a concentration for a proper comparison of samples in this study. If identical materials of different concentrations generate spectra of different shape and intensity false exclusions are probable (Patra et al, 2001).

Variable separation synchronous excitation (VSSE) was developed by Kubic, Lasher, Dwyer and Sheehan in 1983. This technique differs from SSE in that the $\Delta \lambda$ is not fixed. This advancement enabled the user to produce any cross-section of a three-dimensional total fluorescence plot between 45° and 90°. With multiple scans at different angles the three dimensional plot could be constructed. Using the VSSE technique, the researchers were able to differentiate between 61 new motor oils and 45 used motor oils, their results were presented in a two-part publication (Kubic et. al . 1983).

3. EEM and contour plots

When computers became the rule rather than the exception in data collection, three-dimensional fluorescence data representation was simplified. The three dimensions of excitation wavelength, emission wavelength and fluorescence intensity are referred to as an excitation-emission matrix (EEM). The EEM can be represented in two dimensions as a contour plot.

Early work with EEMs was done by Freearde, Hatchard and Parker. They investigated the use of contour plots from fluorescence spectral data to differentiate oils and were successful in distinguishing among nine samples (Freearde et. al. 1971).

In 1989 Siegel and Cheng were able to differentiate between samples of mid-range petroleum distillates, which appeared to be identical with gas chromatography, using 3D fluorescence plots (Siegel, et. al. 1989).

In 2006, Holbrook et al. at NIST investigated the effect of different corrections made to EEM spectra. The corrections significantly improve intra-laboratory data comparisons. Instrumental bias varies among instruments. This type of bias may result in “systematic errors in measured fluorescence intensity caused by the wavelength-dependent output or response of individual instrument components, such as excitation lamps, monochromator gratings, and detectors” (Holbrook, 2006). In order for laboratories to compare data collected on different instruments this instrument dependent bias needs to be corrected. Other researchers have addressed these corrections on the Varian Cary Eclipse (Holbrook, 2006) (Hall, Clow and Kenny, 2005). Since all fluorescence spectra were collected on the same instrument these corrections are not necessary for this feasibility study. The light source in the Varian Cary Eclipse is a xenon flash lamp. Since a flash lamp is only illuminated in very short period of time it has a much longer useable lifetime and is not prone to source decay. If future work aims to create a comparable data set, the 2006 Holbrook work should be consulted to inform data collection and processing.

4. Low Temperature Fluorescence Spectroscopy

Spectra collected at liquid nitrogen temperatures may show phosphorescence as well as fluorescence therefore it is most accurate to describe all low temperature work as luminescence spectrometry (Fortier, 1978).

Fortier and Eastwood, pioneers in low temperature luminescence, used this technique to differentiate among different classes of fuel oil. They noticed an increase in spectral structure and observable phosphorescence at 77K. Subtle spectral details were enhanced at liquid nitrogen temperatures. This increase in detail arises from the lack of collisional broadening and other non-radiative relaxation mechanisms that are hindered in the rigid, frozen glass. They determined that the optimum concentration for analysis is 10 to 20 ppm. This concentration gave sharp spectral structure with a good signal-to-noise ratio. These researchers, along with Hendrick, in a later article recommend the combined use of low temperature luminescence and synchronous scanning to differentiate among very similar oil samples (Fortier, 1978) (Eastwood, 1978).

In 1983, Kubic, Lasher and Dwyer, used low temperature luminescence and with variable synchronous excitation fluoremetry (VSSE) to distinguish among 61 automobile motor oils. They coupled the low temperature spectral data with room temperature spectral data to obtain complete differentiation among engine oils (Kubic, et al 1983).

In a 1998 Analytical Chemistry review article, Soper, Warner and McGown note that the publications regarding low-temperature luminescence have slowed but progress has been made in the applications of online detection in capillary electrophoresis to distinguish structurally similar analytes (Soper, et. al. 1998).

A 2006 review article compiled the literature from January 2004 to December 2005. The authors site advances using LEDs in time-resolved fluorescence of samples contaminated with crude oils, novel fluorescence sensors have been developed for the analysis of biological and chemical warfare agents, and numerous multivariate statistical analysis techniques are being applied to luminescence data. Much of the current research in the field of fluorescence spectroscopy is striving to make this technology more portable so measurements can be taken in the field. Portability is especially important to environmental chemists and hazardous material responders who work “in the field” and need fast but accurate results (Fletcher, et. al. 2006).

5. Other instrumental techniques

The only study to focus specifically on automotive chassis residue was conducted by Zieba. Zieba used IR spectrophotometry to identify the specific lubricating oils in the samples. She was most successful with fresh oil samples and less so with the undercarriage residues. She observed spectral difference among the undercarriage residue samples but could not determine if these differences were due to sample

composition or random errors or poor reproducibility of the technique. Her results show that significant changes to the class characteristics of the unused oils occur during use. Changes in lubricant chemistry brought on by mechanical and thermal stress caused observable differences in the IR spectra of the samples (Zieba, 1985).

In 2007 Reardon and his research team used high temperature gas chromatography-mass spectrometry to characterize new and used motor oils. PAH presence was investigated using extracted ion profiling and not chemical extraction. The percentage of aromatic compounds increased with the degree of oil wear. Of the different brands and grades, 32 of 34 samples were distinguished. Of the dipstick swabs 26 of the 30 samples were distinguished. All differentiation was done with visual examination of the smoothed chromatograms. The Reardon study shows that there are measurable differences in used motor oils (Reardon et. al. 2007).

To aid analysis of a more common forensic evidence type, Petraco, Gill, Pizzola and Kubic investigated the usefulness of multivariate pattern recognition procedures to distinguish among 20 liquid gasoline samples collected during casework. Their work showed that these samples could be differentiated using canonical variate analysis, orthogonal canonical variate analysis and principal component analysis with varying dimensionality requirements (Petraco et. al. 2008).

6. Sample Preparation

Kershaw and Fetzer developed an extraction procedure to analyze coal liquids by fluorescence spectroscopy. They fractionated coal liquids into oils, asphaltenes and pre-asphaltenes. Oils are soluble in n-pentane, asphaltenes are soluble in toluene and pre-asphaltenes are soluble in pyridine. Pinero-Iglesias *et. al.* compared Soxhlet, ultrasonic and microwave assisted extraction techniques for the extraction of PAHs from diesel particulate matter using a standard reference material from NIST (SRM 1650). Extraction of PAHs from these types of particulates can be difficult because they form at the same time as the particles and can be adsorbed into their internal structure. The researchers found that the ultrasonic, microwave and Soxhlet extractions all had comparable recoveries of the individual PAH compounds (Kershaw *et. al.* 1995) (Pineiro-Iglesias *et. al.* 2002).

8. The Material - Automotive undercarriage residue

Automotive undercarriage residue is a complex mixture that has multiple contributing sources including motor oils and other lubricants, asphalt particles, tire particles and exhaust particles. All of these sources contain fluorescing compounds, particularly polycyclic aromatic hydrocarbons (PAHs) (Rogge, *et. al.* 1993) (Pineiro-Iglesias *et. al.* 2002). Logic dictates that the residues on the lowest parts of the automotive undercarriage are most likely to be transferred when the underside of an automobile comes in contact with a person or object.

Motor oil is used to inhibit metal-on-metal contact on engine parts. Motor oil is refined from crude oil and comes from the high boiling point fraction of crude oil. The base stock of motor oil comes from the lubricating portion of crude oil. This refined product contains additives to improve its performance as a lubricating agent. These additives include thickening agents, pressure additives, antioxidants, detergents, alkaline components, rust inhibitors, pour-point depressants and antifoaming agents (Motor Oil Guide, 1971) (Wilkinson, 1992). The Society of Automotive Engineers (SAE) assigns grades to motor oils based on viscosity. Some oils have two grades that reflect a changing viscosity with temperature. When stressed during use by high temperatures and mechanical movements, new compounds are formed through oxidative and other processes. These new compounds change the fluorescing characteristics of a motor oil, making the fluorescence spectrum on an oil a dynamic pattern (Kubic et al. 1983).

Road dust particulates come from many sources including vehicle exhaust, brake lining particles, road top particles, tire wear debris and settled aerosols from local combustion of fossil fuels. Rogge *et. al.* studied the organic contribution of all but the last contributor in this list and found polycyclic aromatic hydrocarbons present in all material types (Rogge et. al. 1993).

Regardless of the contributing source to the automotive undercarriage residue, only those compounds that contain fluorophores will contribute to the total fluorescence spectrum. A fluorophore is a functional group in a molecule that after photon excitation allows for the proper electronic relaxation to the ground state. The main fluorophore that will contribute the fluorescence spectrum of each sample is the aromatic ring.

Automotive undercarriage residue has multiple contributors. Of particular importance are those contributors which are dynamic in nature; the more dynamic the source, the greater the chance of individualization.

9. Importance of Concentration

It is important that a concentration of fluorescing compounds is established before final spectral characterization. Both intensity of luminescence and shape of the spectrum will change with concentration. It is imperative, for two reasons, that the concentration of the samples be established. First, comparisons of two samples with the same fluorescing compounds but different concentrations may result in a false discrimination if smaller secondary and tertiary peaks are not observed in the less concentrated sample. Second, the fluorescence exhibited by the detected compounds may shift to longer wavelengths at higher concentration. This shift may also result in false discrimination when comparing two samples of very different concentration but of common origin. False discrimination

in the method could result in false positives and false negatives if the method is applied to a questioned sample and a sample of known origin.

Some researchers have monitored and used the red shift as a way to characterize samples. Patra and Mishra investigated this concept to differentiate among different petroleum products. In their study they wanted to observe the red shifting to characterize the compounds (Patra et al 2001). Since individual compounds may not be identified, the total concentration of the compounds will be determined by measuring the total integrated area of the absorbing compounds resolved by liquid chromatography or by the mass of the extracted residues.

Methods

1. Sample Collection

Ten motor vehicles were included in this study. All ten were passenger vehicles, either sedans or sport utility vehicles, of various years, makes and models. Five undercarriage locations on each motor vehicle were sampled, including the right and left front ball joints on all vehicles along with three additional locations as detailed in Table 1.

Table .1 Vehicle Sample Set and Specific Locations

01 GMC Denali	01	Right ball joint
	02	Left ball joint
	03	Engine oil pan
	04	In front of right ball joint
	05	In front of left ball joint
02 Ford Bronco II XLT	01	Right ball joint
	02	Left ball joint
	03	Cross member
	04	Oil pan
	05	Under radiator in front of engine
03 BMW 330xi	01	Right ball joint
	02	Left ball joint
	03	Inner right joint – left of ball joint
	04	Inner left joint – right of ball joint
	05	Right joint in the back
04 Ford Taurus	01	Right ball joint
	02	Left ball joint

Table .1 Vehicle Sample Set and Specific Locations (continued)

	03	Engine oil pan
	04	Under radiator
	05	Lower part of engine block
05 Honda Accord	01	Right ball joint
	02	Left ball joint
	03	Middle strut
	04	Undercarriage Position a
	05	Undercarriage Position b
06 Mercury Marquis	01	Right ball joint
	02	Left ball joint
	03	Oil Pan
	04	Undercarriage Position a
	05	Undercarriage Position b
07 Ford Escort	01	Right ball joint
	02	Left ball joint
	03	Oil Pan
	04	Undercarriage Position a
	05	Undercarriage Position b
08 BMW 740 iL	01	Right ball joint
	02	Left ball joint
	03	Front skid guard
	04	Left frame
	05	Right bushing
09 Chevrolet Cavalier RS	01	Right ball joint
	02	Left ball joint
	03	Oil pan center
	04	In front of oil pan

Table .1 Vehicle Sample Set and Specific Locations (continued)

	05	To the right of the oil pan
10 Dodge Caravan	01	Left ball joint
	02	Right ball joint
	03	Behind radiator
	04	Oil pan
	05	Left front of cross member

Known samples have a naming scheme that reflects their original source. The generic name is “##X” where #1 equals the origin vehicle, #2 equals the origin vehicle location and X equals the known set it came from (A, B, C). For example, sample name 21A indicates that this swab came from car 2 (Ford Bronco II XLT), location 1(right ball joint) and is from known set “A.”

Samples were taken with sterile cotton swabs and stored in labeled screw-capped test tubes. Each location was swabbed five times creating 5 sets of 50 samples. Additional samples of used motor oils were collected from the dipstick of each vehicle in the study. These liquid samples were used for instrumental method development due to the ease of making solutions of known w/v concentrations. These samples were not used in the known sets because they would not be representative of transferred materials as they are found on the underside of a motor vehicle. Instead, swabbed samples were taken from the bottom of vehicle oil reservoirs. Residues from the exterior of the reservoir, while they may contain used motor oils that have seeped out of the interior of the oil reservoir, they also contain road particulates and other debris. It is this exterior-pan

material that is more likely to be transferred when a person or item comes into contact with the underside of a motor vehicle.

From the five swabs taken at each location, three sets were used as the identified known samples, one set was used for extraction and instrumental method development and one set was re-labeled to produce samples of unknown origin. The unknown sample set was re-labeled by Kate Grimley, who recorded the origin of each unknown and sealed these results in a signed and dated envelope. Additionally, Ms. Grimley chose a numbering scheme that was random with respect to the origin naming scheme. Each unknown name begins with the letter “M” and is followed by a two digit number. I arbitrarily selected, without replacement, ten of these relabeled vials to be the unknowns in this study.

There were observable differences between the residue colors, textures and ease of transfer from vehicle to vehicle. In general, older vehicles yielded greater quantities of transferable undercarriage residues which were more greasy than the residues from younger vehicles. Newer vehicles, especially the BMW 330xi, yielded less material per swab and this material was grittier and less oily than that found older cars. This implies that the tendency to transfer oils and other lubricants may be a function of age, use and vehicle design.

2. Extraction Procedure

The Kershaw and Fetzer three-solvent extraction scheme was used for this study. Three fractions, oils, asphaltenes and pre-asphaltenes, were extracted from each swab using ultrasonication solvent extraction with pentane, toluene and pyridine respectively. The extraction took place inside the same screw-capped test tube used for swab storage.

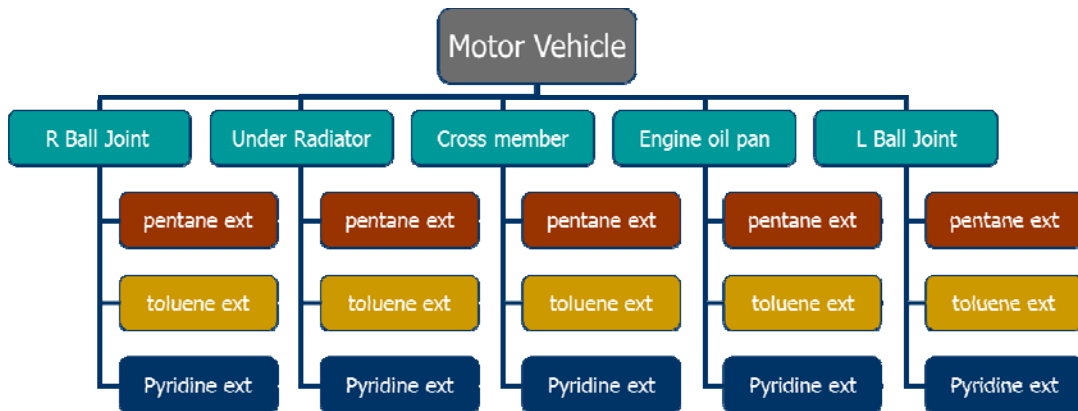
The extraction technique is summarized in Table 2 below:

Table 2. Extraction Method

Extraction volume	2 aliquots of 2mL each
Temperature	Pentane – 30 °C Toluene – 60 °C Pyridine – 60 °C
Sonication time	30 minutes each aliquot (1 hr total)
Centrifugation time	20 minutes for the combined aliquots.

The total sampling scheme is summarized in the Figure 2.

Figure 2. Total Sampling Scheme - Flow Chart of Sampling and Extraction Scheme for a Single Motor Vehicle



After extraction, the two aliquots were combined and filtered with successive 0.42 micron and 0.22 micron nylon syringe filters to reduce turbidity. A turbid solution can contribute to undesirable Rayleigh-Tyndell (RT) scattering. RT scattering inhibited spectral details in previous work. (Purcell 2002) After filtration, extracts were centrifuged for 20 minutes and then were gently evaporated at room temperature for at least eight hours in preweighed aluminum weighing dishes. Each dish mass was taken in triplicate and averaged. The average pre-residue mass was subtracted from the average post-residue mass to determine the actual residue mass. Residues were then quantitatively transferred to a 5 mL volumetric flask using successive washes of the dilution solvent. The concentration in w/w ppm for each undiluted solution of pentane extract was calculated using the equation below:

$$C = (g_{\text{residue}} * 1000) / (\rho_{\text{solvent}} * V_{\text{flask}} / 1000) \quad \text{Eq. 2}$$

Where C equals the concentration in ppm of the undiluted solution, g_{residue} is the average mass of the pentane extract residue, ρ_{solvent} is the density of the solvent¹ and V_{flask} is the volume of the volumetric flask.

The undiluted solution concentrations of known samples ranged from 9600 ppm to 22 ppm. The triplicate knowns from 22A were run at no dilution, 200 ppm, 100 ppm and 50 ppm at three excitation wavelengths, 220 nm, 230, nm and 290 nm. The spectra from the samples at 100 ppm concentration produced the best results with the relative

¹ Acetonitrile for the pentane extracts and methylcyclohexane for the toluene and pyridine extracts.

intensities of each lambda max at each excitation wavelength showing agreement among the triplicate samples from the same motor vehicle location. The undiluted solutions from the triplicate samples did not exhibit the same relative ratio of fluorescence intensities for the lambda max peaks at each excitation wavelength. These variations in spectral shape and relative intensity could be due to red shift observed at higher concentrations. This result favors dilution to a common concentration over no dilution of the extract solution. This result is in agreement with previous work by Eastwood and Kubic, who in separate studies recommended bringing all solutions to a common concentration prior to spectral data collection.

Previous researchers have used a variety of sample concentrations for the fluorescence analysis of oils. Eastwood , Siegel and Kubic independently recommended 20 ppm for direct analysis of oils. Purcell, whose work was also on motor vehicle undercarriage residues used a range of 10-100 ppm. The trials done using known 22A also recommend using a concentration at or close to 100 ppm. This may differ from previous work on oils since the automotive residue may contain many more material types than straight motor oil (e.g. road dust, diesel particulates). The 100 ppm concentration is based on total residue weight and a smaller portion of that residue contributes to the overall fluorescence of the sample. Any extract solution that was at 100 ppm or lower was left undiluted. The same sample naming scheme was expanded to

represent the extract type. The letter P, T, or Y was added to the end of each sample name to denote the pentane, toluene or pyridine extract respectively.²

Three extraction system blanks were run. Clean swabs were subjected to the entire extraction procedure including sonication, centrifugation, filtering, evaporation and reconstitution in a solvent. The blank extracts were run on the Cary Eclipse using the room temperature fluorescence excitation emission matrix parameters and low temperature emission scan fluorescence parameters. The resulting fluorescence spectra show that none of the materials used for sample preparation (i.e. swabs, glass test tube, filters or solvents) contributed any fluorescence to the extracted sample. This illustrated in figures 3-5. Note that there is no 2nd order Rayleigh scatter exhibited in figures 3-5 because of monochromator filter settings.

² Sample 22AT is the reconstituted toluene extract residue from vehicle 2 (Ford Bronco), position 2 (left ball joint), known set A.

Figure 3. Pentane Extraction Blank

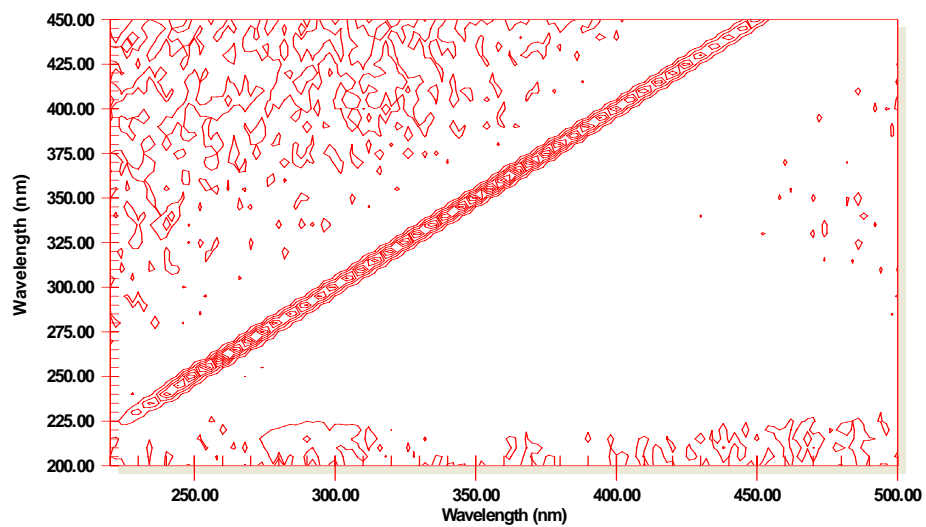


Figure 4. Toluene Extraction Blank

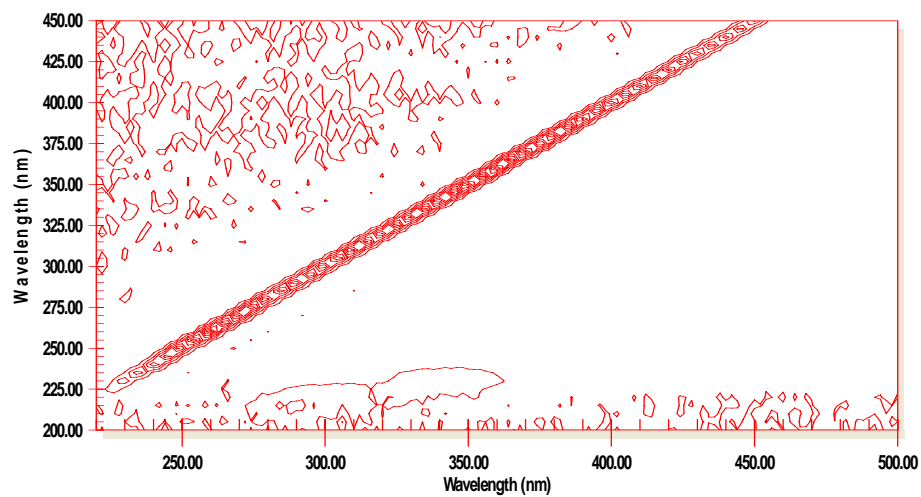
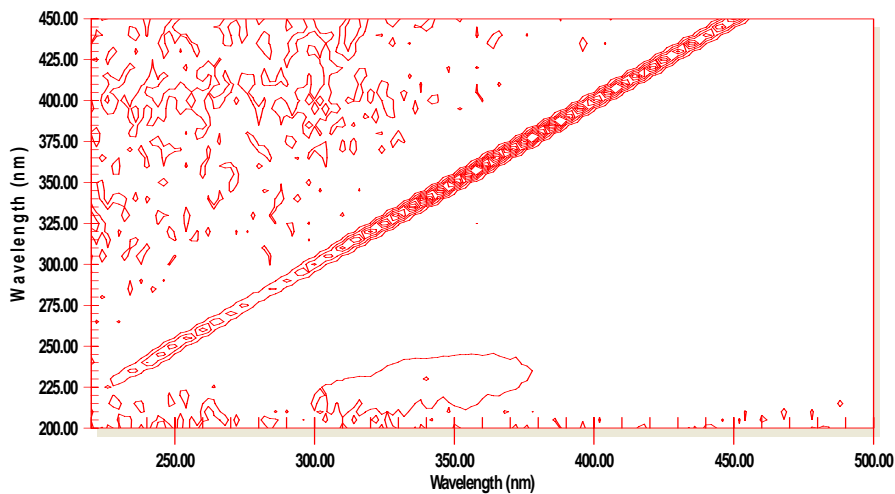


Figure 5. Pyridine Extraction Blank



Acetonitrile and methylcyclohexane, the two solvents used for residue reconstitution and fluorescence analysis were run as blanks. There was no contribution to the observed fluorescence spectra from these two solvents. This is illustrated in figures 6 and 7. Please note that figures 6 and 7 were collected at a different emission filter setting than figures 3-5. The open setting resulted in the collection of 2nd order Rayleigh scatter.

Figure 6. Acetonitrile Blank

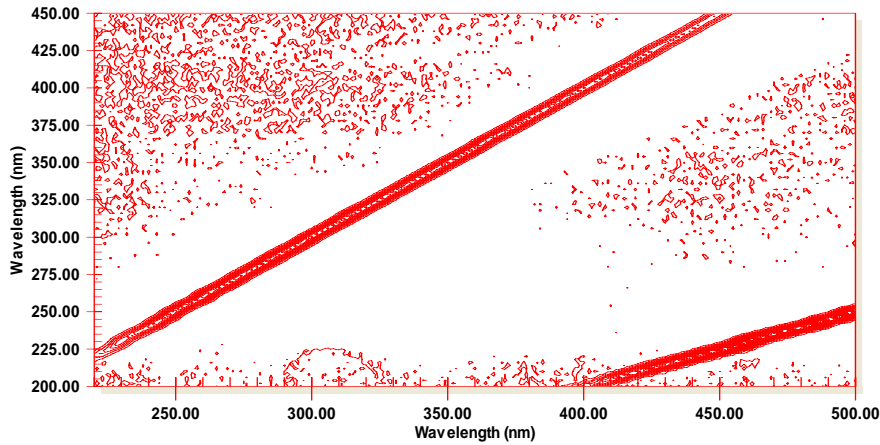
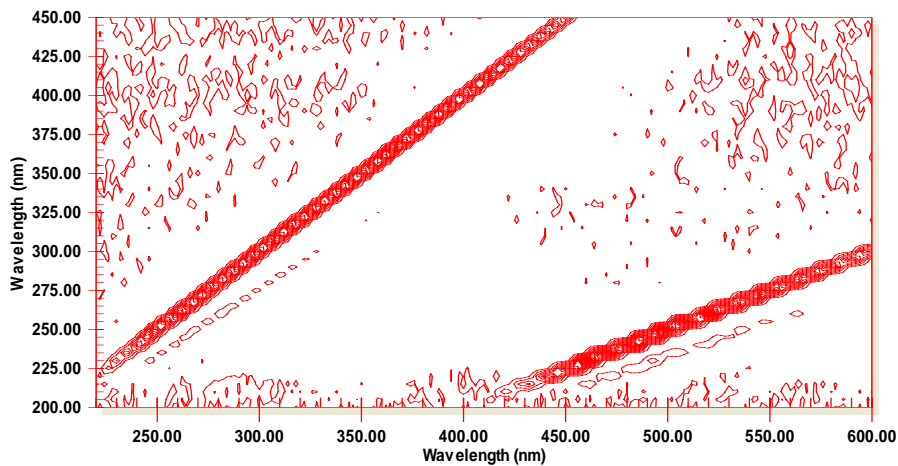


Figure 7. Methylcyclohexane Blank



2. HPLC Method Development

Since the concentration of the fluorescing compounds in each sample will effect the peak intensity and number. It was important to determine a suitable method to create extracts of similar concentrations. Purcell's work used calculated residue weights diluted to a constant volume to create extract solutions which were then diluted to common concentrations. While he was successful at creating solutions with a common ppm

concentration, this was based on weight of all compounds extracted, and the true concentration of fluorescing compounds remained unknown. One of his recommendations for future work included using chromatographic data, specifically peak areas of a compound common to all extracts in that class to base concentration calculations on.

It was determined that HPLC was the preferred chromatographic technique. There were established methods, solvent systems and column types for the separation of polyaromatic hydrocarbons and if asphaltenes, much larger molecules with thermally labile peripheral alkyl groups, were extracted in high concentrations, this method would be preferred to gas chromatography. Additionally, HPLC has a wider variety of detector types available. It was determined that a fluorescence detector would be the best suited for this task.

Using a fluorescence detector would ensure that the compounds used to adjust the concentration exhibit fluorescence and directly influence the fluorescence excitation-emission spectra. A review of the several different commercially available fluorescence detectors was done. Funds were available through a grant from the Borough of Manhattan President's Office. I reviewed the commercially available units and obtained quotes from vendors. All necessary paperwork was submitted to the city representatives to complete the purchase. After multiple communications with representatives of the City of New York purchasing department, it became clear that the procurement process

and the hurdles encountered would delay the project timeline so much so that it became prohibitive to proceed with procurement for this study. This was indeed the case as the equipment was actually delivered in November 2010 and installation has just been completed as of this writing; almost two years after the purchasing process began. Without the ability to procure a fluorescence detector for the HPLC instrument in-house, I decided it was best to proceed with UV detector already available and installed on the HPLC. The instrument used was a Shimadzu HPLC with a SCL-10A system controller module, SIL -10A auto injector, SPD-10A UV absorbance detector. All data was collected on a HP3396A integrator.

The first of three swab extracts (pentane) was evaporated to dryness, reconstituted and brought to 5 mL total volume in acetonitrile. These undiluted solutions were analyzed via HPLC. Peaks common to all three swab extracts from a single vehicle location were identified and their areas added to create a total peak area for each known. This method is similar to that used for compounds that were historically difficult to resolve such as PCBs (Butcher *et al*, 1997).

Historically, this quantification method has been used to quantify polychlorinated biphenyl compounds analyzed by gas chromatography. Most commercial PCB production was conducted by the Monsanto Corporation and marketed under the brand name Aroclor. Packed GC columns lack the resolving ability to separate the compounds in the PCB mixture. These products were quantified using the total peak areas of selected

Aroclor peaks resolved in the chromatogram. In the method employed for this project, the peaks identified as common to each triplicate sample are analogous to the Aroclor peaks selected to represent concentration of the entire PCB sample.

3. Solvent Incompatibility

The literature most strongly supported the use of a reverse phase system for polycyclic aromatic hydrocarbons detection by HPLC. However, the aqueous solvent system and the methylcyclohexane necessary for the low temperature spectral analysis are not miscible. Since both the investigation of using chromatographic techniques and the characterization of sample fluorescence at liquid nitrogen temperatures were important parts of this project it became necessary to find a compromise. Any extract residue that was reconstituted in acetonitrile would not be able to be analyzed at low temperatures and any reconstituted in methylcyclohexane would not be analyzed by HPLC. Of the three swab extracts, pentane, toluene, and pyridine, it was expected that the toluene and pyridine extracts, if they contain asphaltenes and pre-asphaltenes as recorded by Kershaw and Fetzer, would contribute the most fluorescing components due the preponderance of aromaticity, while the oils would contribute the least due to the preponderance of unconjugated compounds. I decided to analyze the pentane extracts with HPLC and UV detections to adjust concentrations with the oils extract while reserving the other two extracts for low temperature analysis.

Table 3. HPLC Method

Isocratic Elution ³	10:90 H ₂ O: ACN
Flow Rate	1ml/min
Injection Volume	3 microliter
Detector λ	254 nm

The HPLC column used was a Supelco brand Supelcosil LC-PAH, 5 cm x 4.6 mm ID with 3 micrometer particles. The LC-PAH stationary phase is C-18. A guard column with the same packing material was used to prolong the life of the analytical column. Column performance was tested with three standard polyaromatic hydrocarbons listed in Table 4. Detector response linearity was tested with dilutions of phenanthrene and naphthalene in concentrations ranging from 0.019 $\mu\text{g}/\text{mL}$ to 4.00 $\mu\text{g}/\text{mL}$ and 0.23 $\mu\text{g}/\text{mL}$ to 50 $\mu\text{g}/\text{mL}$ respectively. The results are shown in Tables 6 and 7.

³ Gradient elution was tested on dilute solutions of used waste oils. The instrument had difficulty returning to a stable baseline after the end of the gradient program. The isocratic program eliminated the baseline drift problem by maintaining a constant solvent system and ensuring complete elution of all injected compounds.

Table 4. Standard PAH Solutions

Fluorene	10 µg /mL
Naphthalene	50 µg /mL
Phenanthrene	4 µg /mL

	Inj 1	Inj 2	Inj 3	%RSD
Naphthalene				
3 µ/L inj	38257	37947	44751	9.5290
Phenanthrene				
6/µL inj	75879	75810	75483	0.27936
Fluorene				
6/µL inj	75377	75901	75568	0.35071

Table 5. HPLC Detector Linearity - Phenanthrene

Compound	µg/mL	average peak area
Phenanthrene	4.00000	640978
standard dilution 1	0.66700	111696
standard dilution 2	0.11100	22414
standard dilution 3	0.01850	6415
standard dilution 4	0.00309	ND

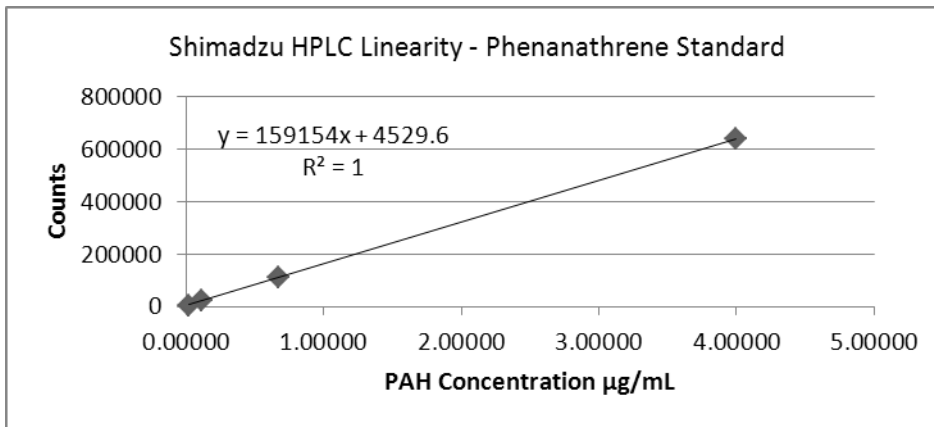
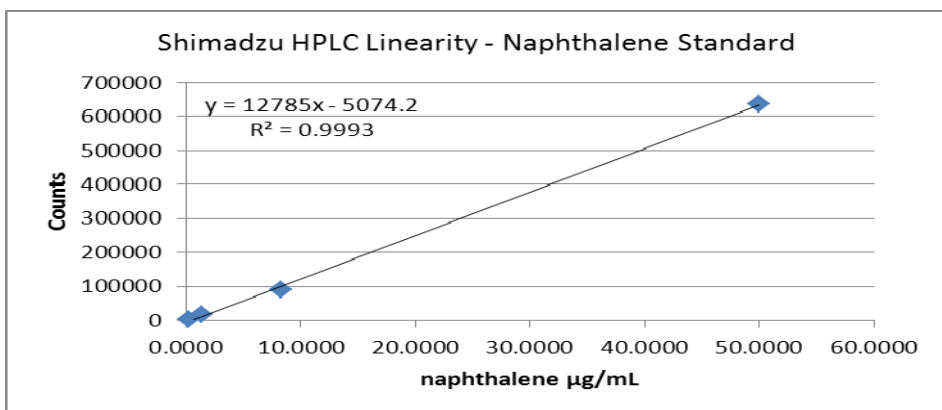


Table 6. HPLC Detector Linearity - Naphthalene

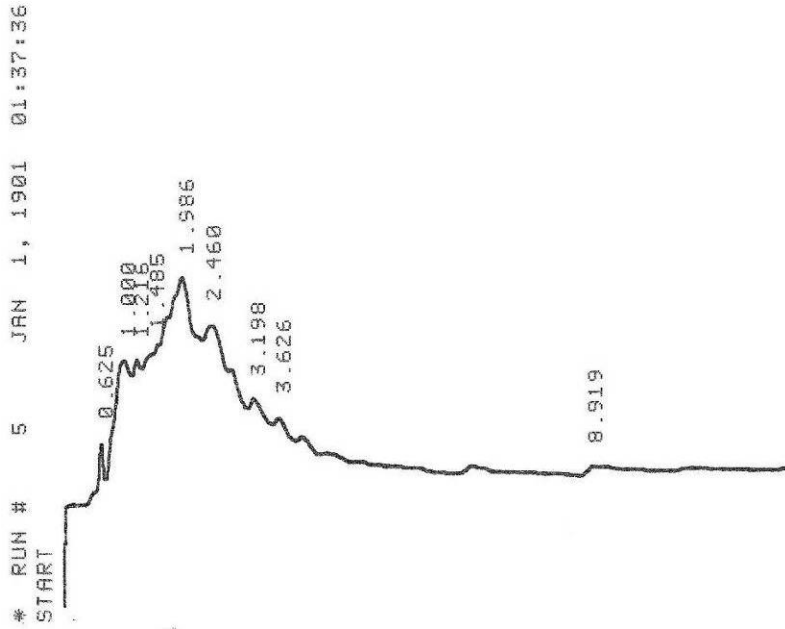
Compound	µg/mL	average peak area
Naphthalene	50.0000	635958
standard dilution 1	8.3330	89891
standard dilution 2	1.3890	16974
standard dilution 3	0.2315	3383
standard dilution 4	0.0386	ND



Chromatograms for all triplicate samples from the fifty motor vehicle undercarriage locations were collected. Figure 8 shows the reproducibility of the chromatograms among samples of common motor vehicle origin.

Figure 8. Chromatograms from triplicate swabs of Vehicle 1 Location 3 (13).

13A



13B

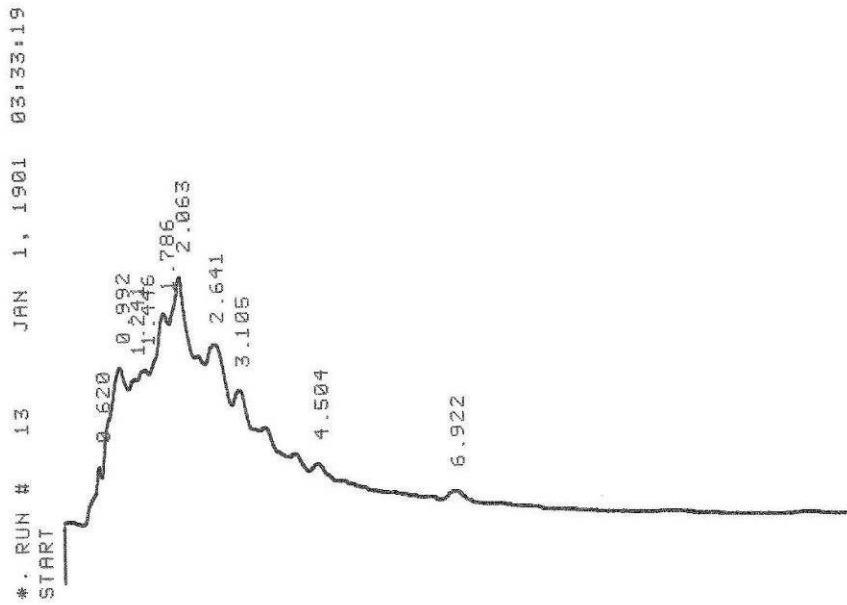
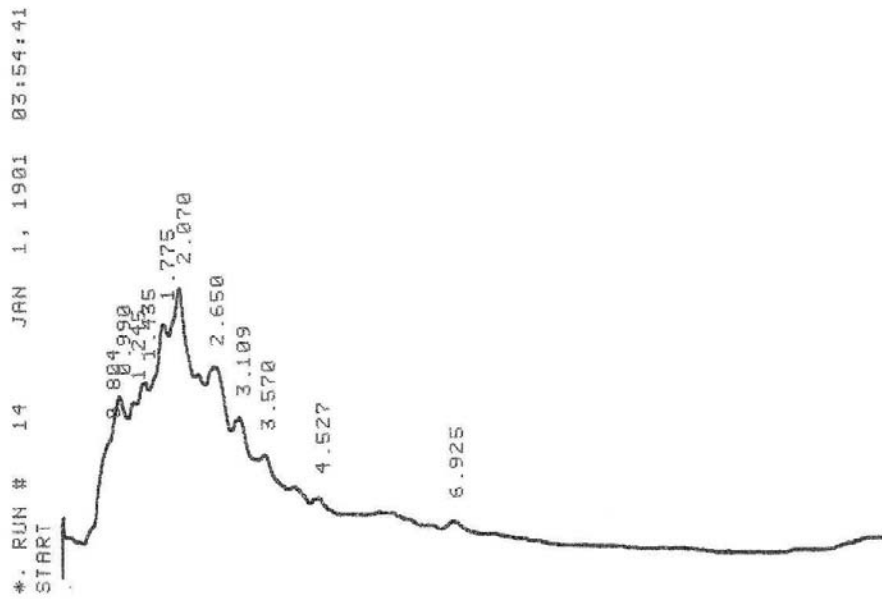


Figure 8. Chromatograms from triplicate swabs of Vehicle 1 Location 3 (13) continued
13C



4. Cary Eclipse Instrumentation

All fluorescence spectral measurements were made on a Varian Cary Eclipse Fluorescence Spectrometer. It has a xenon flash lamp source and an extended range (200 nm – 900 nm) photomultiplier tubes, both sample and reference. Wavelength selection is accomplished with two grating monochrometers. The xenon flash lamp and signal processing technology makes the Cary Eclipse nearly immune to room light contamination. The Cary Eclipse was designed with the detector located at 90 degrees from the source. This geometry limits detection of unscattered photons from the source lamp. (Ingle, Crouch)

The Cary Eclipse software contains a validation function that tests the wavelength accuracy of the excitation and emission monochromator and the spectral bandwidth accuracy of the excitation and emission slits.

Stray light is monitored by taking the ratio of the intensity of scattered light detected by the sample PMT at a wavelength which is at the very edges of the excitation wavelength (i.e. in the excitation wavelength is set to 350 nm this test will measure at 371 nm) The ratio must be between -0.0005 and 0.0005 a.u. to pass the test. The Raman band of water is measured at 350 nm and 300 nm and was used to test the sensitivity of the instrument. Wavelength reproducibility is measured by repeatedly measuring the wavelength at the 541.92 nm Xe lamp peak for 10 scans and calculating the difference between the extremes of the wavelengths measured. The difference must be less than 0.2 nm to pass. Wavelength accuracy is also tested using lines from the Xe source. Peaks must be within 0.5 nm of the line at 541.92 nm and 1.0 nm of the 260.54 nm line in order to pass. Initial execution of these validation tests show that the Cary Eclipse was performing within the desired parameters as set by the manufacturer.

Table 7. Instrument Validation Tests preset by Varian. Results from a 16 day period.

<i>Varian Validation Tests</i>		<i>1</i>	<i>2</i>	<i>3</i>	<i>4</i>	<i>5</i>	<i>RSD</i>
	<i>Expected values</i>	nm	nm	nm	nm	nm	
<i>WL Accuracy Ex. Mono (Xe) (nm)</i>	<i>541.92</i>	542.01	542.08	541.93	541.93	541.92	0.012823
	<i>260.54</i>	260.92	260.92	260.84	260.84	260.83	0.017567
<i>WL Accuracy Em. Mono (Xe)</i>	<i>541.92</i>	542.15	542.02	541.86	541.04	541.86	0.080174
	<i>260.61</i>	260.61	260.54	260.38	259.53	260.38	0.16732
<i>Spectral BW Acc. Ex. Slit (nm)</i>	<i>1.5</i>	1.651	1.667	1.664	1.751	1.766	3.188
	<i>2.5</i>	2.478	2.473	2.539	2.51	2.546	1.341
	<i>5</i>	5.336	5.345	5.386	5.321	5.339	0.456
<i>Spectral BW Acc. Em. Slit (nm)</i>	<i>1.5</i>	1.077	1.024	1.121	1.159	1.151	5.077
	<i>2.5</i>	2.423	2.394	2.454	2.428	2.377	1.249
	<i>5</i>	5.154	5.168	5.231	5.316	5.249	1.254
	<i>10</i>	10.434	10.375	10.273	10.29	10.37	0.64094
	<i>20</i>	20.697	20.521	20.555	20.539	20.62	0.35088
		a.u.	a.u.	a.u.	a.u.	a.u.	
<i>Stray Light (a.u.)</i>		0.00034	0.00034	0.00031	0.0003	0.00031	5.8
<i>Raman H2O Sensitivity Ex.350 nm (#:1)</i>		1325	1536	1405	1336	1474	6.375
<i>Raman H2O Sensitivity Ex.500 nm (#:1)</i>		863	892	889	900	828	3.36

During the course of data collection these quality assurance tests and spectral analysis of naphthalene and anthracene in fluorescent polymer blocks were run trimonthly in order to ensure that the instrument was functioning correctly.

Figure 9. Anthracene/Naphthalene Polymer Block Standard – Single EEM.

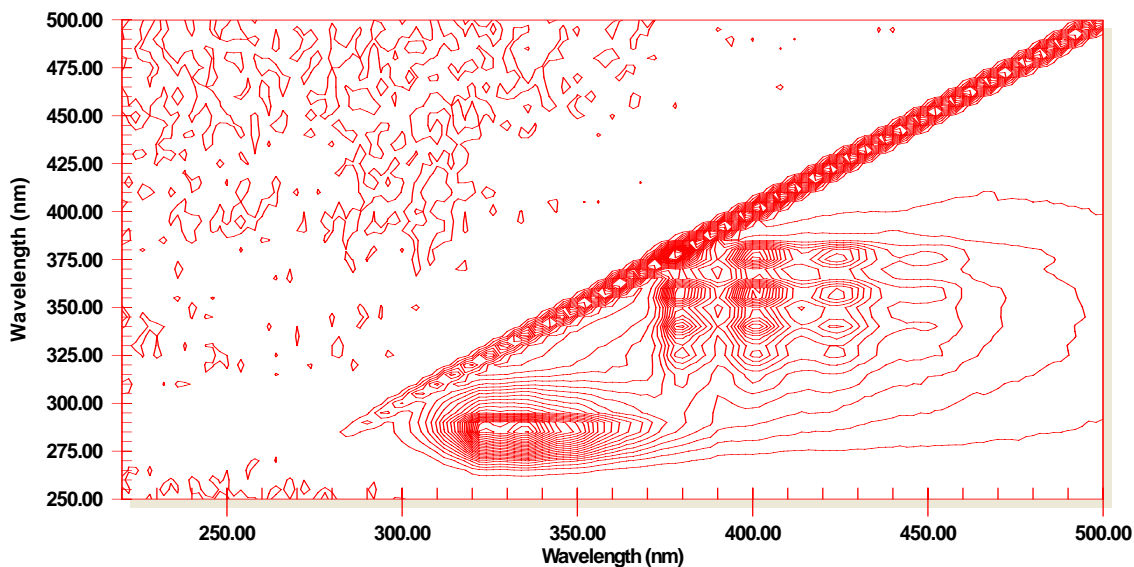


Table 7. Anthracene/Naphthalene Polymer Block Standard – Intensity and Emission values. I=Intensity

	Ex 290				Ex 320		Ex 360	
Date	Peak λ 1	I	Peak λ 2	I	Peak λ	I	Peak λ	I
5-Feb-09	324.00	249.61	336.00	248.70	401.07	97.916	402.00	250.53
20-Feb-09	324.00	243.30	336.00	241.60	402.00	93.765	402.00	238.46
27-Feb-09	324.00	245.27	336.00	244.00	402.00	84.058	402.00	204.29
2-Mar-09	324.00	248.41	336.00	249.90	402.00	86.710	402.00	207.45
16-Mar-09	324.00	242.02	336.00	283.25	402.00	85.040	402.00	199.78
31-Mar-09	324.00	238.62	336.00	239.42	402.00	85.148	402.00	196.99
9-Apr-09	324.00	239.05	335.07	244.82	402.00	85.421	401.07	183.19
15-Apr-09	324.00	239.25	335.07	238.47	401.07	84.830	401.07	191.79
28-Apr-09	324.00	242.48	335.07	241.79	401.07	85.385	402.00	194.26
% RSD	0	1.6461	0.13852	5.5522	0.11576	5.5297	0.10207	10.786

Figure 10. Anthracene/Naphthalene Polymer Standard – Excitation 290 nm. All scans
Each color represents a single scan taken on one of the dates listed in Table 7. Scans from
all dates are represented.

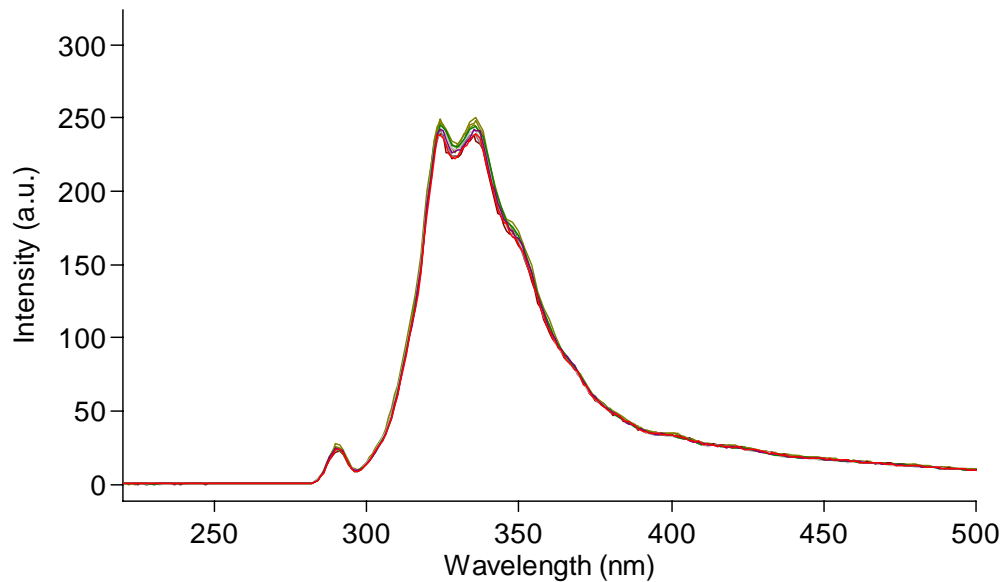


Figure 11. Anthracene/Naphthalene Polymer Standard – Excitation 320 nm. All scans.
Each color represents a single scan taken on one of the dates listed in Table 7. Scans from
all dates are represented

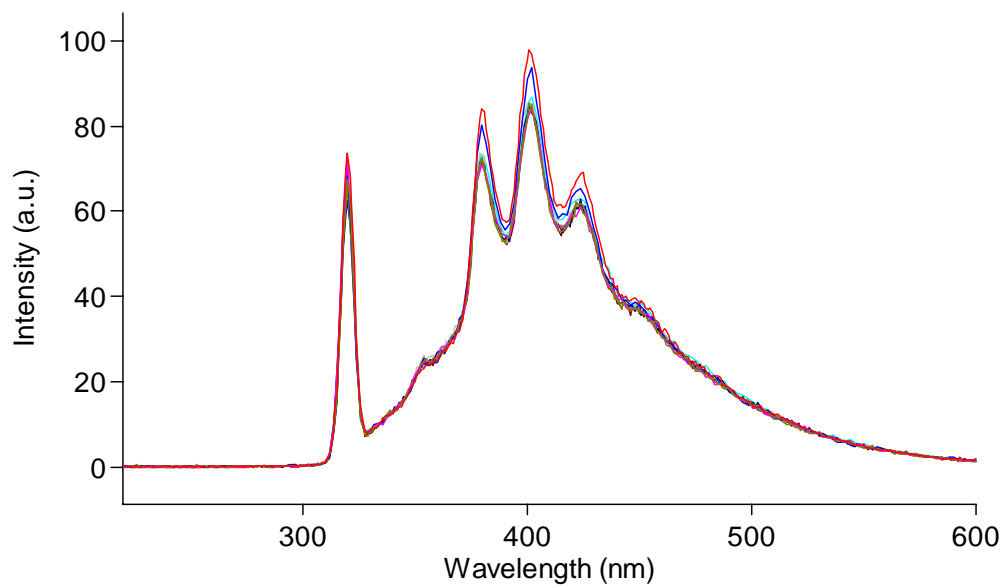
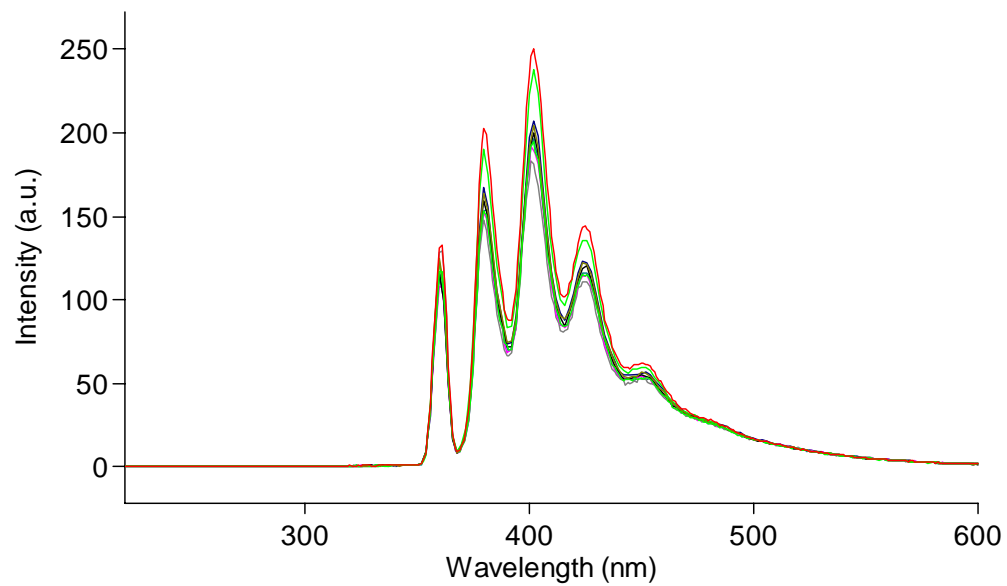


Figure 12. Anthracene/Naphthalene Polymer Standard – Excitation 360 nm. All scans. Each color represents a single scan taken on one of the dates listed in Table 7. Scans from all dates are represented.



Data Collection

Excitation Emission Matrix Room Temperature Fluorescence

One sample from each vehicle undercarriage location was selected for room temperature excitation emission matrix fluorescence spectral data collection. Table 8 shows the instrument settings for this data collection. The EEM spectra from known samples collected are listed shown in the Appendices C-E.

Table 8. Cary Eclipse Setting for Room Temperature Excitation Emission Matrix Data Collection

Excitation Start	200 nm
Excitation Stop	500 nm
Emission Range	220-600nm ⁴
Data Step ⁵	5 nm
Ex Slit	5 nm
Em Slit	5 nm
Scan Rate	1200 nm/min
Data Interval ⁶	2.0 nm
Averging Time	0.1 sec
Ex Filter	Auto
Em Filter	Auto ⁷
PMT	Medium (600 V)

⁴ The emission range for EEM data went from 220-600. No peaks were observed past 400 nm. The scale was reset to 220-400 nm during data processing to improve the presentation of EEM spectral detail.

⁵ In an excitation emission scan, the *Data Step* is the amount that the excitation monochromator moves prior to the next scan. It is the distance between the slices that form the EEM.

⁶ The *Data Interval* determines the size of the steps between each data point in the emission range.

⁷ For the pentane and some of the toluene extracts, this filter was set to "Open." This setting resulted in the collection of secondary RT scatter at wavelengths from 440 nm and above. Data processing with the Excel macro removed these peaks. The emission filter setting "Auto" is recommended for future work.

1. Parameter Selection

The excitation wavelength range was chosen based on previous work with this sample type and with direct analysis of motor oils (Purcell 2002) (Fortier and Eastwood 1978). Most organic molecules fluorescence at excitations below 500 nm. (Schenk 1973) And since the majority of the fluorescence observed is expected to be from polycyclic aromatic hydrocarbons, this excitation range is sufficient to capture these data.

Slit widths for the excitation and emission monochrometers were chosen based on previous work (Purcell 2002). The spectral resolution would have been improved with smaller slits however, there was concern about losing information from more weakly fluorescing samples. Since this resolution was sufficient for successful sample discrimination and known-unknown association in a previous study and in this work, produced EEM spectra with sufficient detail to collect the main points of comparison (i.e., the excitation and emission wavelengths of the peak fluorescence) the same slit widths were adopted for room temperature data collection for this work.

Varian claims the Cary Eclipse is capable of scanning at speeds up to 24,000 nm/min without exhibiting peak shifts. However at such a high scanning rate finer secondary spectral features may be lost. Trials with sample 41A established that a scan rate of 1200 nm/min, when compared to a scan rate of 600 nm/min, would produce spectra with no peak shift from the slower scan rate and sufficient secondary spectral detail in an EEM. This is illustrated in the Figures 13 and 14 below. The EEMs of 41A at both 600 nm/min and 1200 nm/min exhibit the same number of peaks at the same positions and all secondary spectral detail is the same. Scan rate, averaging time and data

interval are all inter-related on the Cary Eclipse. When set to the 1200 nm/min the Cary automatically sets the data interval to 2.0 nm. At this interval the noise observed at excitation wavelengths below 225 nm was reduced.

A PMT settings of medium (600 V) and high (800 V) were tested on 17 samples of known origin from 6 different motor vehicle locations. For the room temperature data collection, the PMT setting at 800 V produced many fluorescence spectra with intensities that exceed the range of the PMT. This problem was not observed with at the 600 V setting.

The use of spectral smoothing, specifically Savitzky-Golay, was considered. Setting the instrument to smooth during collection would not have allowed the collection of the raw, unsmoothed data. Attempts to smooth after collection would be extremely time consuming. Each EEM is constructed with 60 emission scans and each would have to be separately opened, smoothed and then all 60 would be reassembled in a matrix. Considering that the main data points for the comparison of spectra, the emission and excitation wavelengths of the peaks, would not be improved with smoothing, it was decided that the collection of raw but noisier data was preferable to smoothed.

Figure 13. Known 41A at Scan rate 1200 nm/min

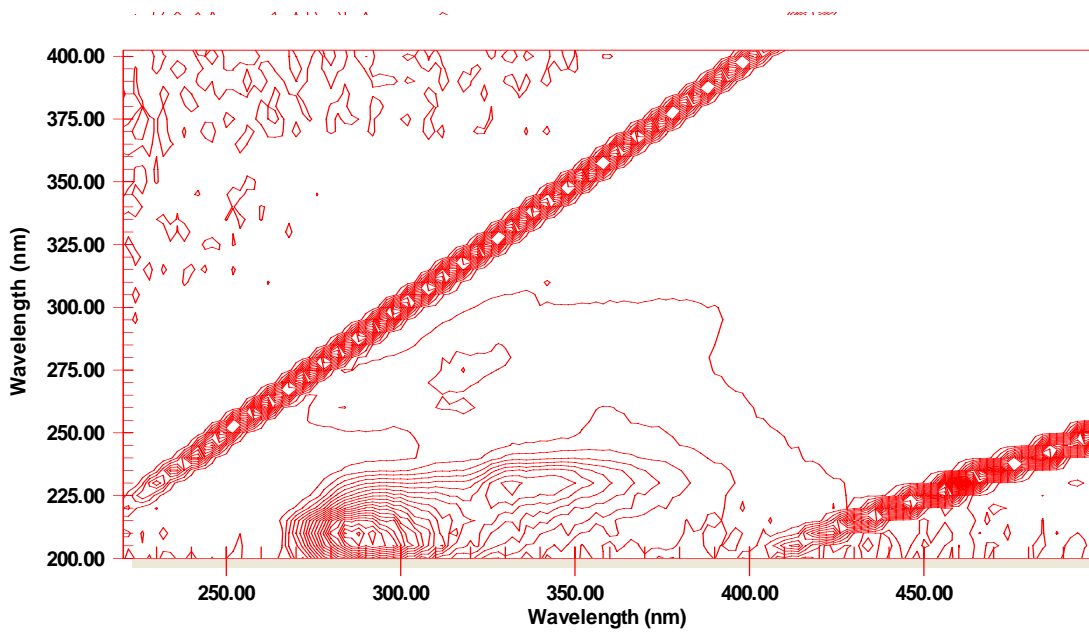
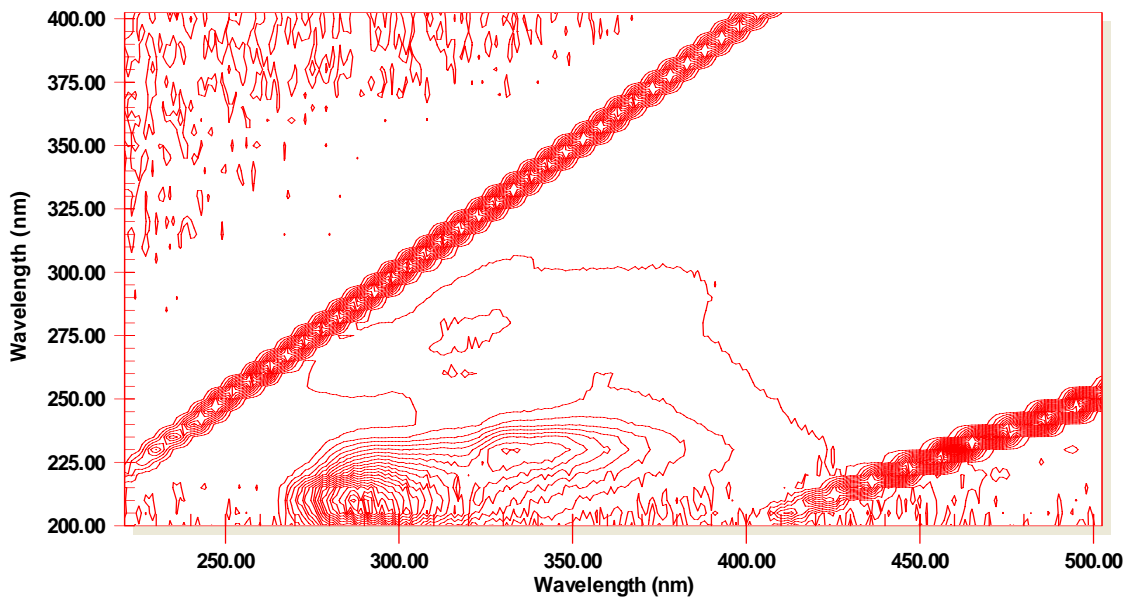


Figure 14. Known 41 A at Scan rate 600 nm/min



At the slower scan rate, and smaller data interval (1nm), figure 14 does exhibit smoother contour lines, especially in the excitation wavelength regions above 250 nm.

Below an excitation wavelength of 250 nm, the smaller data interval also captures more of the noise at this wavelength region where a xenon source is less intense. Since most of the peaks from these undercarriage residues samples occur at these low excitation wavelengths, the increased resolution provided by the smaller data interval did not outweigh the increased difficulty in finding peak maxima below excitation 250nm. Therefore, the faster scan at 1200nm with the larger data interval was preferred for the EEM data collection.

2. EEM Spectral Processing

The primary and secondary Rayleigh Tyndall (RT) scattering observed and in previous work by Purcell was thought to be caused by turbid solutions. Turbidity can cause this type of elastic scattering because light is physically deflected from the small particles suspended in solution. However, the intensity of the RT scattering was relatively constant. Samples from the previous work were filtered once through 0.44 micron syringe filters and the results were normalized to the largest peak intensity. As a result the RT scattering looked very large on spectra that had very weak fluorescence intensity. Filter settings can eliminate collection of second order RT scattering. The emissions filter was set to "Auto". These settings were used on most of the toluene extract solutions and all of the pyridine extract solutions. To generate accurate contour plots the RT scatter must be removed so that the Grams software does not include these peaks when scaling the contour slices. Contour plots are made by taking the total fluorescence intensity range and dividing it into a pre-determined number of sections.

The more sections the software is asked to create the more contour lines will appear on the plot. If the RT scatter is the most intense peaks in the data set, the actual fluorescence emission will be divided into less contour slices than the analyst may intend. This effect is most damaging for weakly fluorescing spectra that are more ‘dwarfed’ by the intense RT scattering.

A Varian representative sent an application development language (ADL)⁸ program that was to stagger the starting wavelength on the emission monochrometers to avoid collecting any RT scatter. Instead of all emission scans beginning at the same wavelength this program purported to stagger the start wavelength of the emission monochrometer. For example, the first excitation wavelength 200 nm with emission scanning from 220 nm to 600 nm would collect no RT scatter because there was never a time when excitation wavelength = emission wavelength but when the excitation wavelength moved above 220 nm the emission scan would start at a longer wavelength (e. g. 240 nm) thereby cutting out the RT scatter. This program never performed as advertised. The file as-sent would not upload because of syntax problems. I edited the format into one readable by the ADL shell. After several more edits to the ADL code a single EEM spectra was collected successfully. However attempts to display this file as a contour plot resulted in repeated crashes of the Cary Eclipse program. These experiences

⁸ The application development language (ADL) is a programming language, similar to BASIC, that is used by the Cary Eclipse software.

led to the decision to collect the EEM files without real-time efforts to avoid collecting RT scatter.

This scatter was removed post data collection by a customized Excel macro program. The excel macro is designed to replace intensity values with a zero value in key places where first and second order RT scatter is found. The macro deletes every intensity value where the emission wavelength is less than or equal to the excitation wavelength minus 5nm. Since fluorescence emission does not usually occur at wavelengths higher in energy than the excitation wavelength, no important spectral data is lost in this deletion. Any fluorescence due to the anti-stokes shift, where the emission is at shorter wavelengths than the excited radiation, will not be removed by this macro if it occurs at wavelengths 5 nm shorter than the excitation radiation. Peaks from anti-stokes shifting are usually much less intense than RT scatter peaks and are not necessary to remove from the data set because they do not overwhelm the spectral data.

This is repeated for intensity values at positions 2x the previous wave to delete 2nd order RT scatter. This had the largest impact on contours created from samples with weak fluorescence intensity where the RT scattering has the most obscuring power. The effect of this program is illustrated in Figures 15 and 16. Figure 16 shows that after removal of the RT scatter, three additional low intensity peaks are displayed by the contour plotting function.

Figure 15. 83BT Before Rayleigh-Tyndell scatter removal. Each color represents an excitation scan at a difference excitation wavelength.

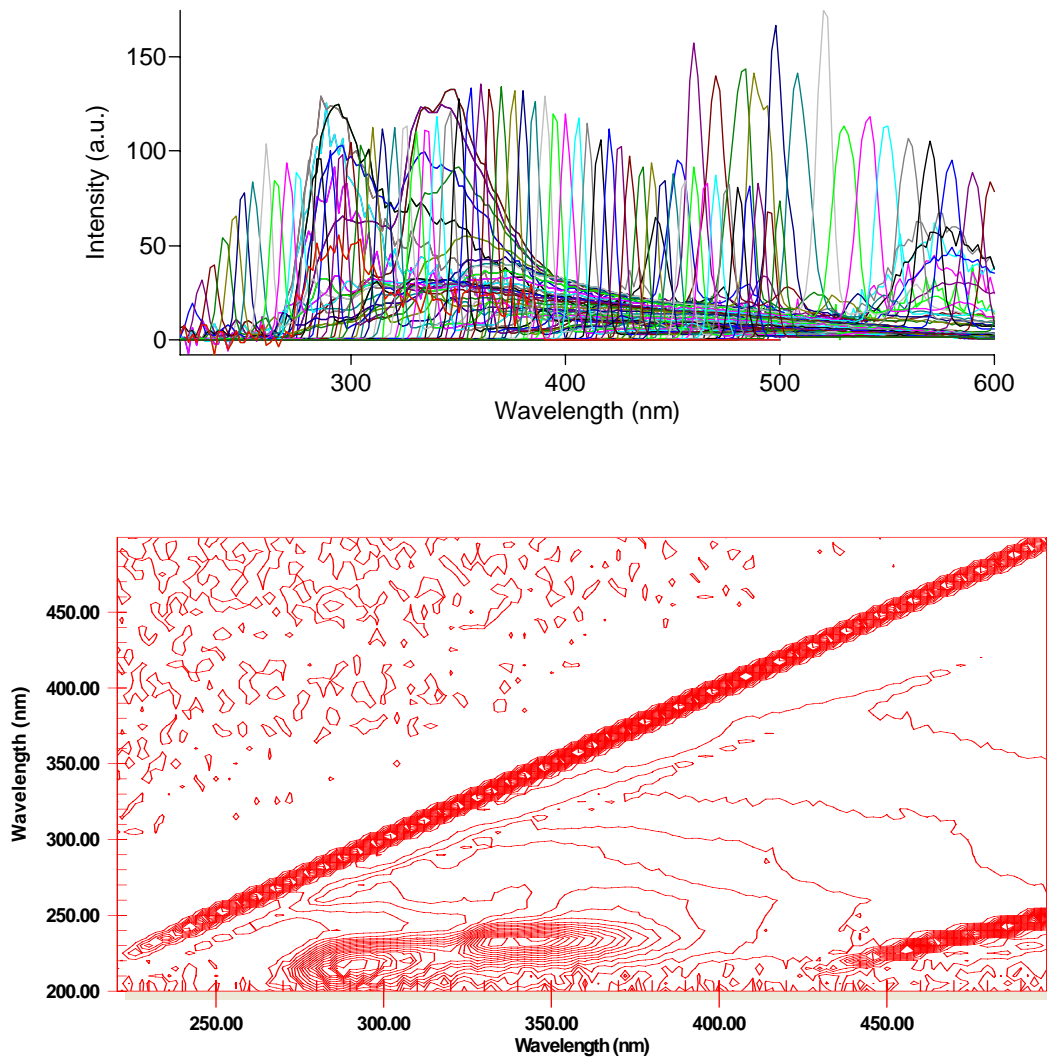
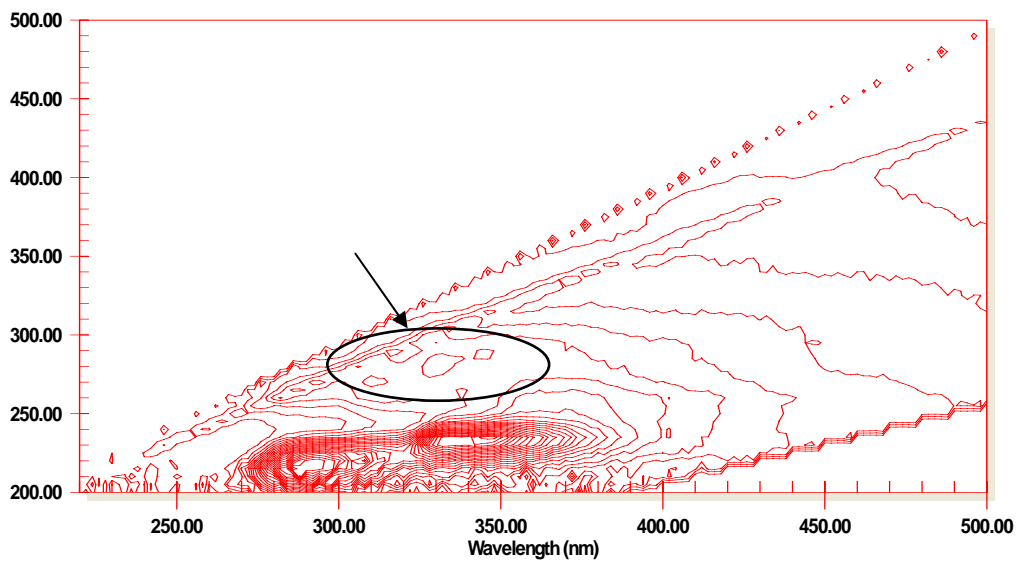
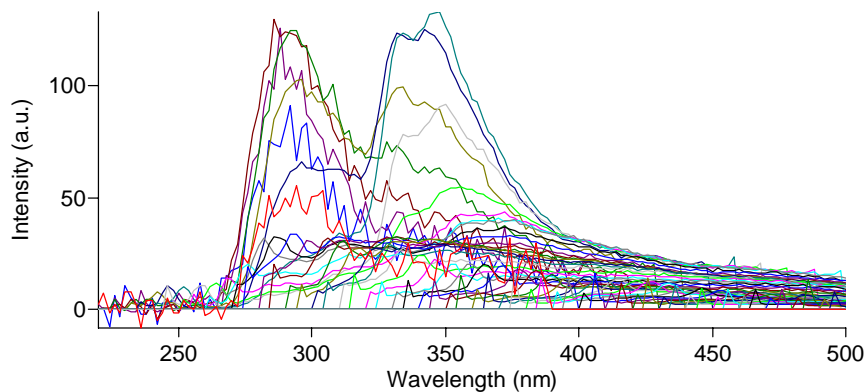


Figure 16. 83BT After Rayleigh-Tyndell Scatter Removal. Each color represents an excitation scan at a difference excitation wavelength.



3. Low Temperature Fluorescence

Development of the Low Temperature Fluorescence Apparatus

In order to achieve the decreased temperatures needed to generate more detailed spectra I used an apparatus designed to freeze the solutions in liquid nitrogen temperature, 77K. There are several challenges associated with this process and that

must be overcome in order to generate useful spectra, most notably, maintaining frost-free sample windows and producing clear and crack-free frozen samples.

4. Apparatus

The original design supplied by Varian was inoperable. The Dewar itself had no cap and was to remain open to the atmosphere when filled with liquid nitrogen. This would not work as the freezing agent would evaporate before the completion of a single EEM, run which can take upwards of 20 minutes. Without a cap to support it, the sample tube was supposed to sit in the bottom of the Dewar and rest on the side of the reservoir resulting in a tube that was at an angle and other than normal to ground. Test scans showed that this angled tube resulted in increased Raleigh-Tyndell scatter that was most likely due to the excitation beam meeting the sample tube surface at an angle other than perpendicular.

5. Design Modifications

A cap similar to the Farand design was necessary to slow the evaporation of the liquid nitrogen and to position the tube normal to the excitation beam.

Figure 17. Farand Dewar cap with quartz-tipped glass sample tube.

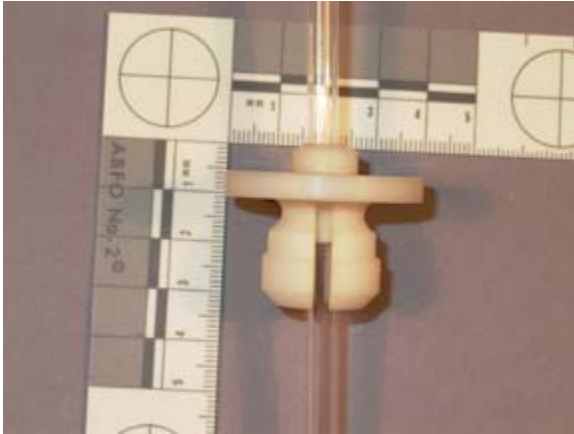


Figure 18. Farand Dewar cap top view showing vent holes.



The Varian Dewar was taller than the top of the sample compartment and wider than the original Farand design⁹. To make room for this a new top was supplied with a hole cut in

⁹ The Farand Dewar owned by Dr. Thomas A. Kubic is 4.3 cm OD while the Dewar supplied by Varian is 7.7 cm OD.

the top. This top was a solid piece and therefore the Dewar and the support apparatus had to be inserted into the instrument through the hole itself.

This was extremely awkward and eliminated any possibility of ensuring that the Dewar support and the quartz bottom aligned with the source. The sample tubes supplied were quartz NMR tubes with an extremely small inner diameter (3mm) and a height less than that of the Dewar itself. This limited the amount of liquid nitrogen allowable in the reservoir because it could not be higher than the top of the tube. In order to solve these issues a memorandum was drafted and sent to the vendor requesting the lid be split in two to and a cap for the Dewar. Photographs and dimensions of the original Farand cap and Dewar, measurements of the new Dewar and desired tube specifications and material preferences were sent to guide the production. Work proceeded once these modified products were delivered.

After adoption of the split lid, longer tubes and sample cap work continued on apparatus. One of the main differences between the original Farand design and the Varian design was that the Farand Dewar sat in an enclosed metal collar with inlet and exit ports for purging gas. Clearance between the collar and Dewar was not more than 5 mm. This reduced the amount of empty space around the container thereby requiring a slower stream of purge gas to ensure a dry environment. A dry environment is necessary because any moisture will freeze on the surface of the cold Dewar as white snowy ice; a

highly scattering beam-blocking coating. The better the vacuum in the Dewar walls the less likely this was to occur but it must still be avoided.

The Cary Eclipse has a large open sample compartment. While this is good for fitting in a variety of accessories it is not good for low temperature analysis. Extra space was reduced by introducing black paper-covered Styrofoam blocks to the sides of the container.

4. Final Design of the Low Temperature Apparatus.

A series of modifications to the interior of the sample compartment were tested to allow for the driest environment with the slowest flow of dry N₂ gas. Determination of the necessary modifications was an iterative process. As each design element was adopted or altered, the relative humidity (RH) of the sample compartment was monitored using a relative humidity meter. Monitoring would begin with the sample compartment at equilibrium with the laboratory RH and continue as the flow of dry nitrogen gas was delivered from the flow meter manifold. RH data was collected once every minute until the sample compartment interior reached a stable baseline. These data sets were used to identify the sample compartment design that would allow for the fastest drop in RH and the lowest sustainable RH at equilibrium.

The final design of the low temperature apparatus include a quartz-tipped liquid nitrogen Dewar held in a fixed position by an adjustable frame. Two nozzles attached to the frame delivered a steady, but adjustable stream of dry nitrogen gas to the tip of the Dewar to prevent frost formation. The nitrogen gas was supplied by the headspace of the liquid nitrogen tank and the flow was regulated by a custom built manifold of flow meters.

Figure 19. Flow Manifold¹⁰

Front View



Back View



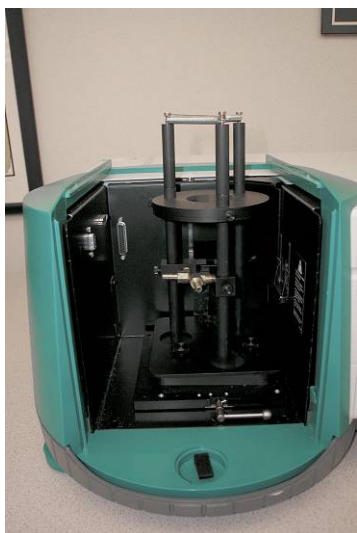
¹⁰ As designed by Dr. Thomas A. Kubic and assembled by the author.

Styrofoam blocks, covered with black paper, were cut to fit the chamber shapes were used to limit the empty space in the sample chamber. All exterior seams on the instrument sample compartment were covered with masking tape to reduce the exposure to humid air. The Dewar was accessed through a split top with a black cap. This set-up allowed access to the Dewar bell to refill with liquid nitrogen and to remove and insert sample tubes without exposing the quartz sample tip to humid air.

Figure 20. Quartz Tipped Dewar



Figure 21. Cary Eclipse sample compartment with Dewar housing.



Despite all efforts to limit chamber humidity it was still necessary to keep the flow of dry nitrogen high (1L/min) and to monitor the sample chamber humidity with a relative humidity sensor with digital display. This final design would keep the quartz tip of the Dewar frost free for up to 10 hours and allow the user to do multiple quartz tube sample changes and liquid nitrogen refills. The illustration of this final design are shown in the figures below.

Figure 22. Modified interior for LT Fluorescence (Shown with cuvette holder, Dewar housing not pictured).

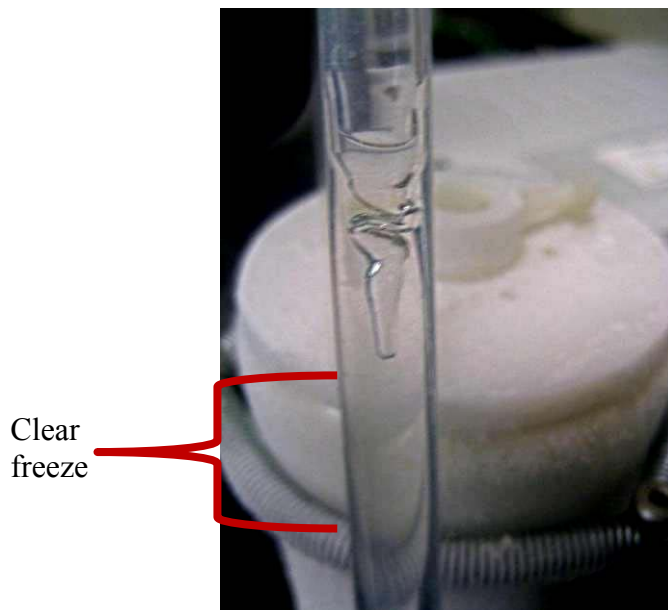


Figure 23. Operational View of Low Temperature Set-Up



Careful selection of sample solvent can protect against the effects of a highly scattering solid. Previous work by Kubic and Sheehan found that methylcyclohexane will freeze to a clear solid. When cooled slowly through gradual immersion, a clear frozen glass is obtained. This is shown in Figure 24.

Figure 24. Clear freeze of methylcyclohexane



The total data set for low temperature scans of the samples from known vehicle locations equaled 5,400 data files. The ensemble averaged scans (16 single scans of the same sample at the same excitation wavelength averaged automatically) at excitations 235 nm and synchronous scan at Δ 70 nm exhibited the most consistently intense spectra of all the low temperature data collected. Therefore, these two data sets were chosen to examine the usefulness of using spectral library searching for making correct origin associations among transferred vehicle undercarriage residues. Instrumental settings for low temperature fluorescence data collection are shown below in Table 9.

Table 9. Cary Eclipse Settings for Low Temperature Data Collection

Excitation Wavelengths for Toluene extracts (nm)	220, 235, 275
Excitation Wavelengths for Pyridine extracts (nm)	220, 275, 290
Delta setting for Synchronous Scanning	70, 105
Ex Slit	10nm Round
Em Slit	1.5 nm
Scan Rate	600 nm/min
Data Interval	1.00 nm
Averging Time	0.1 sec
Ex Filter	Auto
Em Filter	Auto
PMT	High (800 V)

Because a smaller emission slit was chosen for the low temperature scans, the PMT voltage was changed to High (800 V) to compensate for the decreased spectral bandwidth and total signal. The scan rate and data interval were decreased in order to enhance the resolution gains expected from the lower temperature analysis.

Several deltas were considered for the scans at synchronous settings. Previous work with oils recommended synchronous scanning be performed between Δ 20nm and 30 nm (Lloyd 1971) (VoDinh 1978). Initial runs of oils at Δ 23nm, Δ 70 nm, Δ 90 nm, Δ 105 nm at room temperature showed inconsistent results at the four deltas, with the greatest intensity observed at Δ 23 nm and Δ 70 nm. However, when the extracts from known motor vehicle locations were scanned at liquid nitrogen temperatures there was an overall drop in fluorescence intensity. A review of the room temperature EEMs showed that each known sample exhibited the most intense peaks at an emission wavelength

range between 280 nm to 290 nm (Ex. 220 nm) and a second group of peaks in an emission wavelength range of 330 nm to 340nm (Ex225 nm). Concerned that any Δ setting that did not intersect these peaks would show little-to-no fluorescence emission, I chose to proceed with the settings $\Delta 70$ nm and $\Delta 105$ nm. Ultimately, this may have been a redundant since these spectral characteristics were already being captured by individual excitation scans.

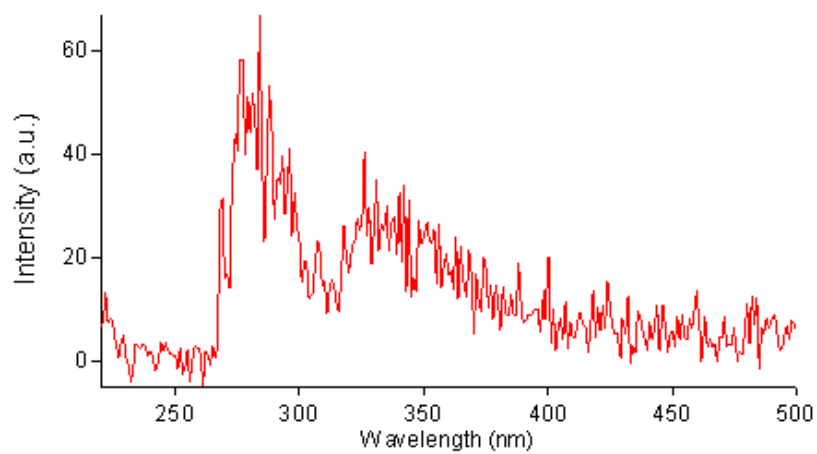
6. Fluorescence procedure

Low Temperature data collection occurred as a series of scans at three predetermined excitation wavelengths and two $\Delta \lambda$ in synchronous scanning mode. The wavelengths were chosen after review of the EEM matrices determined the areas of most intense fluorescence. Single scans and the average of 16 scans were captured for each excitation wavelength. Figure 25 below shows the increase in signal to noise that results from scan averaging. All analysis was done using the ensemble averaged scan produced from 16 successive excitation scans averaged together. Although spectral details varied among samples, Figure 26 below shows the typical result for the low temperature data collection. Low temperature fluorescence data was collected on all triplicate samples for each motor vehicle undercarriage location.

Figure 25. Low Temperature Data Collection

Single Scan

22BT



16 scans averaged - 22BT

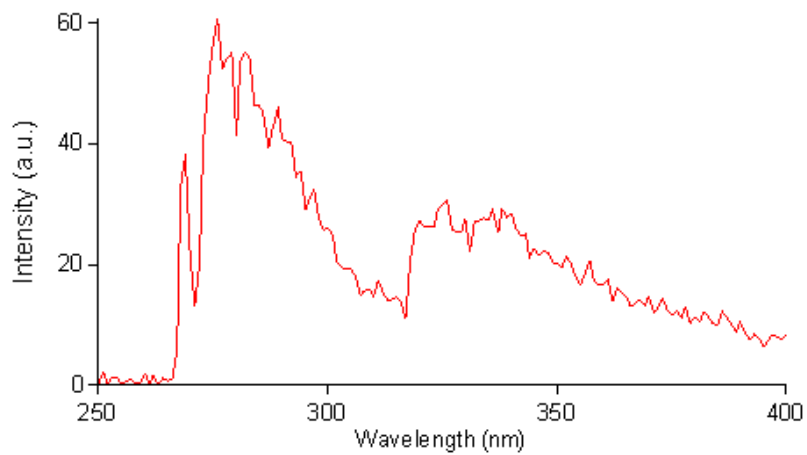
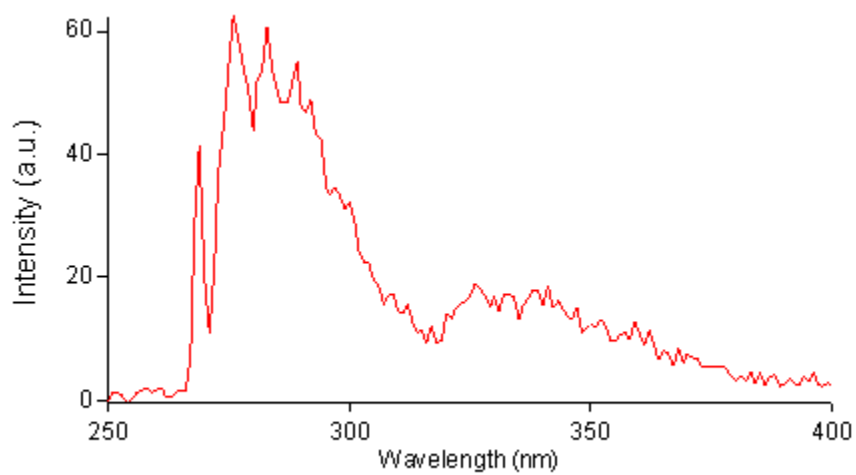


Figure 26. Low Temperature Scans – Full Data Set

24AT

Ex. 220 nm



Ex 235 nm

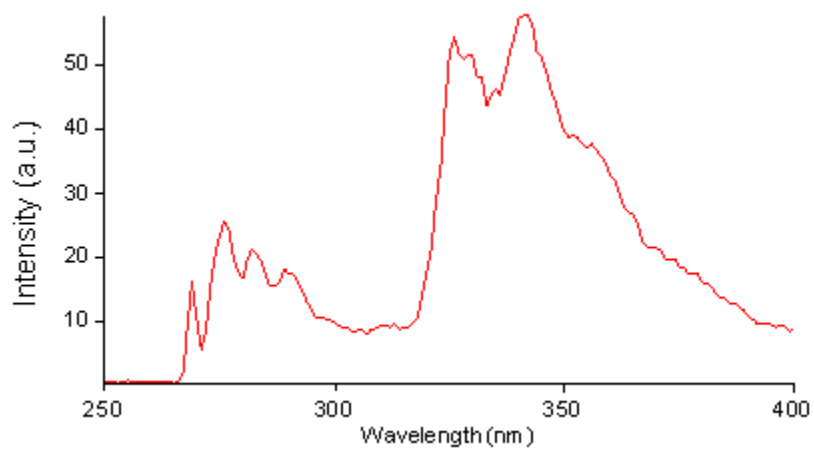
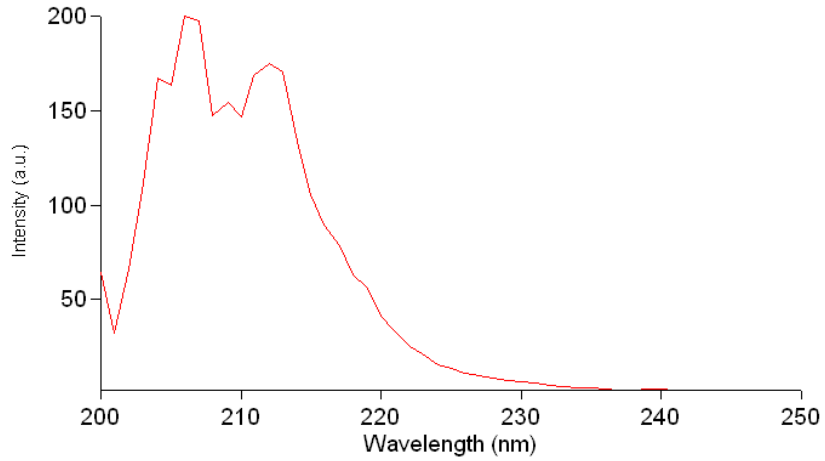
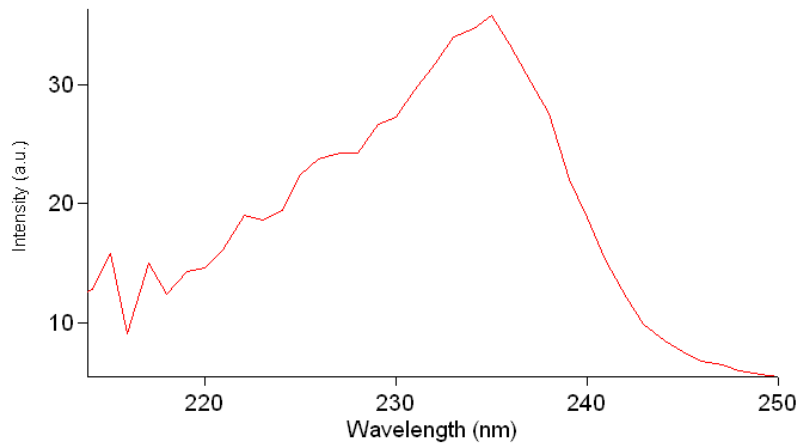


Figure 26. Low Temperature Scans – Full Data Set (continued)

Synch Δ 70



Synch Δ 105



The emission wavelength range for the synchronous scans is 50 nm (200nm – 250 nm).

At this scale and data interval (1nm) it creates a jagged spectral appearance.

Figure 27. 51CT in Dewar at Room Temperature Ex. at 220 nm.

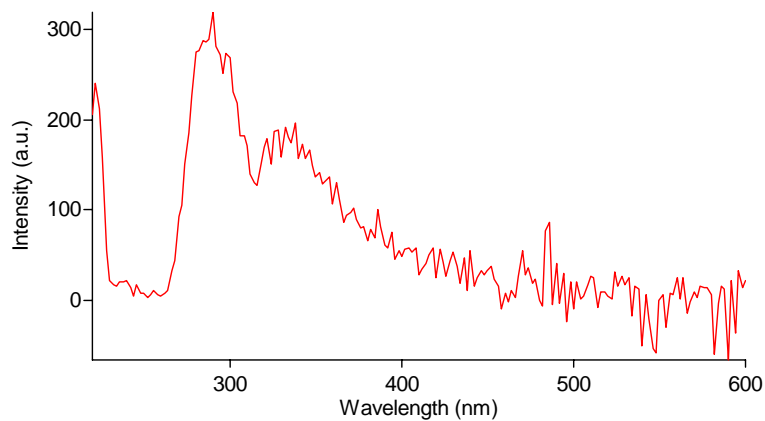
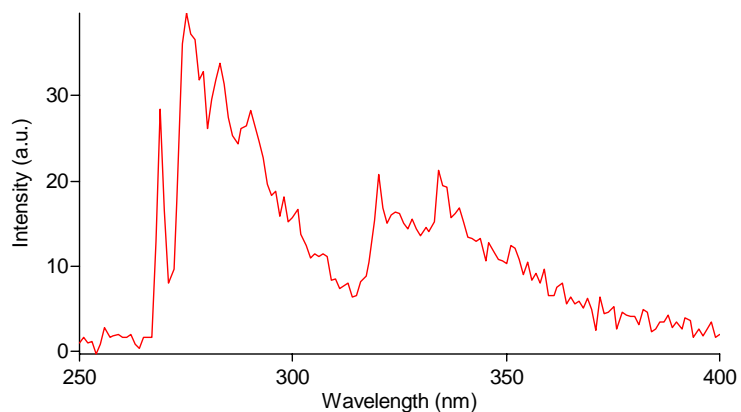


Figure 28. 51 CT in Dewar at 77 K Ex. at 220 nm.



Spectral libraries were constructed using Grams Spectral ID software. All spectra collected at low temperature were converted individually into Grams files with the file extension “.spc”. Three libraries were constructed using the emission scans of the toluene

extracts collected at excitation wavelengths 220 nm and 235 nm and with synchronous scan taken at Δ 70 nm. This extract type and these parameters yielded the most consistently intense spectra in the data set. All samples of unknown origin at these excitation and synchronous settings were searched against the libraries.

The Grams Spectral ID software allows for several types of library searches. All searches conducted for this study were correlation searches. The correlation setting uses an algorithm that centers the unknown and the library data about their means before a vector dot product is calculated (Grams/AI Help Resource). This means that the library and unknown spectra do not have to be normalized prior to searching. Upon searching, the unknown and the library entries are ranked by their hit quality index (H. Q. I.) this value represents the numeric rank of each comparison. The H.Q.I. values lie from zero for the best possible match to 1.000 to the worst. The closer the H.Q.I. is to zero, the better the hit.

Results

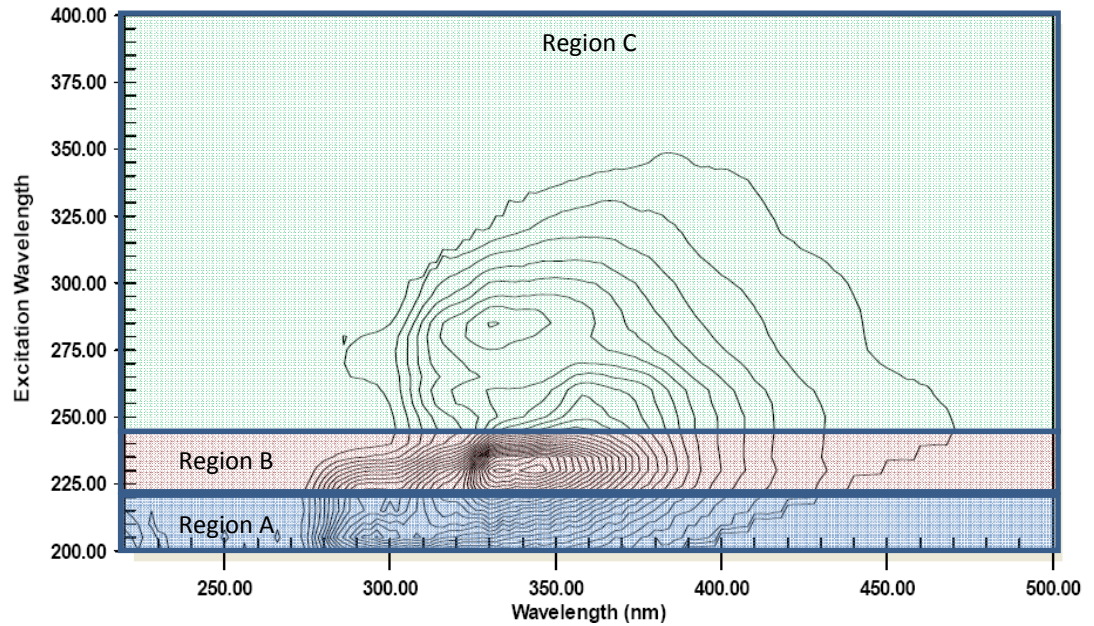
1. General Material Characteristics

There were observable differences between the residue colors, textures and ease of transfer from vehicle to vehicle. In general, older vehicles yielded greater quantities of transferable undercarriage residues which were more greasy than the residues from younger vehicles. Newer vehicles, especially the BMW 330xi, yielded less material per swab and this material was grittier and less oily than that found on older cars. This implies that the tendency to transfer oils and other lubricants may be a function of age, use and vehicle design.

2. General Fluorescence Characteristics

In all fractions the most intense fluorescence was observed at excitations within the wavelength range of 200nm – 250nm. In the excitation wavelength range of 200nm-225nm it was common to find 1 or 2 peaks at emission wavelengths of 285-295 nm. Peaks found in this region of the EEM were labeled “Group A” peaks. In the excitation wavelength range of 225nm-250nm, and 2 additional peaks at emission wavelengths of 334nm and 345nm were observed. Peaks in this region of the EEM were labeled “Group B” peaks. The most intense fluorescence was usually observed in either the Group A or Group B region. All peaks found at excitations above 250nm are described as Region C peaks. While these were usually less intense than the peaks in Group A and B, it was in the Group C region where most of the variation from among samples was found. See Figure 29 below for a graphical representation of this peak grouping.

Figure 29. Region A, B, and C Excitation – Emission Regions. (Item 22A, pentane extract)



EEM spectra from the pentane and toluene extracts were grouped according to the number of peaks in each group and the excitation/emission λ s all peaks present in Group C. After sorting with pentane extracts only, there were 22 unique groups, the largest with 11 vehicle locations and the smallest with 1. Results of this pentane extract peak classification are shown in Table 10.

Table 10. Classification with Pentane Extract EEM Spectra.

ND=None Detected.

Pentane Extract Groups	Sample	No. of Peaks in Group A	No. of Peaks in Group B	No. of Peaks in Group C	Group C Peak 1 (Ex., Em.) nm	Group C Peak 2 (Ex., Em.) nm	Group C Peak 3 Ex., Em.) nm
Group 1	84	1	0	1	258, 283	ND, ND	ND, ND
	72	1	0	1	264, 290	ND, ND	ND, ND
	32	1	0	1	265, 293	ND, ND	ND, ND
	33	1	0	1	269, 312	ND, ND	ND, ND
Group 2	55	1	0	3	266, 304	295, 380	295, 390
Group 3	52	1	1	1	260, 286	ND, ND	ND, ND
	53	1	1	1	266, 300	ND, ND	ND, ND
	81	1	1	1	260, 283	ND, ND	ND, ND
	M19	1	1	1	260, 282	ND, ND	ND, ND
	93	1	1	1	269, 309	ND, ND	ND, ND
	63	1	1	1	291, 383	ND, ND	ND, ND
	54	1	1	1	296, 387	ND, ND	ND, ND
Group 4	M6	1	1	2	260, 283	282, 330	ND, ND
Group 5	101	1	2	0	ND, ND	ND, ND	ND, ND
Group 6	71	1	2	1	268, 312	ND, ND	ND, ND
	64	1	2	1	273, 309	ND, ND	ND, ND
	M46	1	2	1	276, 326	ND, ND	ND, ND
	82	1	2	1	277, 325	ND, ND	ND, ND
	62	1	2	1	280, 324	ND, ND	ND, ND
	65	1	2	1	280, 328	ND, ND	ND, ND
	92	1	2	1	280, 330	ND, ND	ND, ND
	102	1	2	1	280, 329	ND, ND	ND, ND
	M28	1	2	1	280, 328	ND, ND	ND, ND
	21	1	2	1	282, 332	ND, ND	ND, ND
	22	1	2	1	284, 330	ND, ND	ND, ND
Group 7	M36	1	2	2	272, 305	281, 331	ND, ND
Group 8	74	1	3	1	275, 314	ND, ND	ND, ND
Group 9	91	2	0	2	261, 288	278, ND	308, ND
	34	2	0	2	270, 295	270, ND	309, ND
	104	2	0	2	280, 305	286, 356	ND, ND

Table 10. Classification with Pentane Extract EEM Spectra. (continued)

Group 10	85	2	1	1	260, 385	ND, ND	ND, ND
	41	2	1	2	262, 288	270, 321	ND, ND
	94	2	1	1	265, 337	ND, ND	ND, ND
	105	2	1	1	268, 301	ND, ND	ND, ND
	31	2	1	1	269, 312	ND, ND	ND, ND
	42	2	1	1	273, 323	ND, ND	ND, ND
	M29	2	1	2	279, 293	279, 310	ND, ND
	23	2	1	0	ND, ND	ND, ND	ND, ND
Group 11	75	2	2	0	ND, ND	ND, ND	ND, ND
	51	2	2	0	ND, ND	ND, ND	ND, ND
Group 12	83	2	2	1	271, 319	ND, ND	ND, ND
	24	2	2	1	275, 321	ND, ND	ND, ND
	13	2	2	1	276, 334	ND, ND	ND, ND
	103	2	2	1	276, 308	ND, ND	ND, ND
	12	2	2	1	278, 323	ND, ND	ND, ND
	43	2	2	1	280, 330	ND, ND	ND, ND
	M12	2	2	1	280, 328	ND, ND	ND, ND
	M24	2	2	1	280, 330	ND, ND	ND, ND
Group 13	11	2	2	2	260, 326	280, 328	ND, ND
	14	2	2	2	260, 296	282, 330	ND, ND
	73	2	2	2	274, 315	295, 377	ND, ND
Group 14	44	2	3	2	280, 308	281, 354	ND, ND
Group 15	45	2	3	0	ND, ND	ND, ND	ND, ND
Group 16	15	3	2	1	279, 323	ND, ND	ND, ND
Group 17	95	3	1	2	274, 313	260, 371	ND, ND
Group 18	61	3	1	1	281, 328	ND, ND	ND, ND
Group 19	25	4	1	2	254, 332	282, 328	ND, ND
Group 20	35	4	0	1	275, 310	ND, ND	ND, ND
Group 21	M10	4	2	2	256, 322	277, 322	ND, ND
Group 22	M40	4	2	1	275, 317	ND, ND	ND, ND

The same process was used with the EEM spectra obtained from the toluene extracts.

After sorting with the toluene extracts only, there are 25 unique groups, the largest with 5 vehicle locations and the smallest with 1. Results of this classification are shown in Table 12 .

Table 11. Classification with Toluene Extract EEM Spectra

Toluene Extract Groups	Sample	No. of Peaks in Group A	No. of Peaks in Group B	No. of Peaks in Group C	Group C Peak 1 (Ex., Em.) nm	Group C Peak 2 (Ex., Em.) nm	Group C Peak 3 (Ex., Em.) nm
Group 1	105	1	1	1	260, 281	ND, ND	ND, ND
	M46	1	1	1	262, 308	ND, ND	ND, ND
	43	1	1	1	281, 333	ND, ND	ND, ND
	M36	1	1	1	281, 333	ND, ND	ND, ND
	22	1	1	1	287, 350	ND, ND	ND, ND
Group 2	M29	1	1	0	ND, ND	ND, ND	ND, ND
Group 3	33	1	1	2	260, 285	283, 336	ND, ND
	44	1	1	2	261, 284	282, 338	ND, ND
	15	1	1	2	262, 319	284, 332	ND, ND
	55	1	1	2	264, 288	285, 338	ND, ND
	21	1	1	2	286, 330	287, 350	ND, ND
Group 4	83	1	1	3	367, 311	280, 332	286, 346
Group 5	M10	1	2	0	ND, ND	ND, ND	ND, ND
Group 6	104	1	2	1	258, 281	ND, ND	ND, ND
	M40	1	2	1	262, 285	ND, ND	ND, ND
	101	1	2	1	275, 233	ND, ND	ND, ND
	13	1	2	1	282, 337	ND, ND	ND, ND
	75	1	2	1	285, 338	ND, ND	ND, ND
Group 7	M19	1	3	2	260, 287	280, 379	ND, ND
	54	1	3	2	262, 282	289, 380	ND, ND
Group 8	65	1	2	2	264, 340	282, 334	ND, ND
	102	1	2	2	268, 341	284, 331	ND, ND
	74	1	2	2	270, 314	277, 330	ND, ND
Group 9	63	1	2	3	259, 280	286, 331	290, 347
	41	1	2	3	263, 287	263, 338	285, 341

Table 11. Classification with Toluene Extract EEM Spectra (continued)

Group 10	11	2	1	1	267, 292	ND, ND	ND, ND
	14	2	1	1	267, 392	ND, ND	ND, ND
Group 11	35	2	1	2	260, 287	284, 338	ND, ND
	42	2	1	2	262, 288	284, 338	ND, ND
	71	2	1	2	259, 380	285, 340	ND, ND
	72	2	1	2	257, 282	285, 340	ND, ND
Group 13	82	2	1	4	256, 318	261, 329	261, 349
Group 14	84	2	1	1	284, 337	ND, ND	ND, ND
	85	2	1	1	282, 323	ND, ND	ND, ND
	103	2	1	1	258, 280	ND, ND	ND, ND
Group 15	92	2	1	2	262, 285	284, 335	ND, ND
	M28	2	1	2	270, 346	276, 352	ND, ND
Group 16	M12	2	1	0	ND, ND		
Group 17	23	2	2	1	295, 427	ND, ND	ND, ND
	24	2	2	1	280, 331	ND, ND	ND, ND
	34	2	2	1	270, 292	ND, ND	ND, ND
	45	2	2	1	266, 288	ND, ND	ND, ND
	95	2	2	1	262, 375	ND, ND	ND, ND
Group 18	31	2	2	2	265, 288	275, 330	ND, ND
	51	2	2	2	266, 288	288, 347	ND, ND
	53	2	2	2	262, 285	286, 340	ND, ND
	62	2	2	2	264, 288	285, 340	ND, ND
Group 19	M24	2	3	0	ND, ND	ND, ND	ND, ND
Group 20	73	3	1	0	ND, ND	ND, ND	ND, ND
Group 21	52	3	1	1	262, 288	ND, ND	ND, ND
	81	3	1	1	282, 338	ND, ND	ND, ND
Group 22	32	3	1	3	269, 285	281, 315	281, 335
	64	3	1	3	265, 293	285, 336	287, 348
Group 23	61	3	2	3	264, 288	285, 337	287, 348
	91	3	2	3	259, 315	259, 323	279, 331
Group 24	94	3	2	*6	261, 276	332, 276	435, 298

Table 11. Classification with Toluene Extract EEM Spectra (continued)

Group 25	25	4	2	0	ND, ND	ND, ND	ND, ND
	93	4	2	1	261, 332	ND, ND	ND, ND
Group 26	M6	4	1	0	ND, ND	ND, ND	ND, ND

The classifications performed using the pentane and toluene extracts were combined to achieve maximum discrimination. Each sample was labeled using a two value code; the first number is the pentane group and the second is the toluene group. These two numbers are shown for each sample below in Table 12. The undistinguished pairs are shown in gray.

Table 12. Coded samples using EEM peak number and position.

Sample ID #	Pentane Group	Toluene Group
33	1	3
72	1	11
84	1	13
32	1	21
55	2	3
54	3	7
M19	3	7
63	3	9
53	3	17
52	3	20
81	3	20

Table 12. Coded samples using EEM peak number and position (continued).

93	3	24
M6	4	25
101	5	6
22	6	1
M46	6	1
21	6	3
65	6	8
102	6	8
71	6	11
82	6	12
92	6	14
M28	6	14
62	6	17
64	6	21
M36	7	1
74	8	8
104	9	6
34	9	16
91	9	22

Table 12. Coded samples using EEM peak number and position (continued).

105	10	1
M29	10	2
41	10	9
11	10	10
42	10	11
85	10	13
23	10	16
31	10	17
94	10	23
13	11	6
75	11	6
24	11	16
51	11	17
12	12	1
43	12	1
83	12	4
103	12	13
M12	12	15
M24	12	18
14	13	10
73	13	19

Table 12. Coded samples using EEM peak number and position (continued).

44	14	3
45	15	16
15	16	3
95	17	16
61	17	22
25	18	24
35	19	11
M10	20	5
M40	20	6

Five pairs, shaded gray in Table 12, were indistinguishable using this method.

Two of these pairs contain a known sample with sample of unknown origin. When the origin key was consulted it is found that one pair (M46 and 22) are incorrectly grouped together. Unknown M46 is not from vehicle 2 (Ford Bronco XLT) A second pair M19 and 54 were correctly grouped. They share a common origin. Unknown M19 is from vehicle 5 (Honda Accord). None of the other samples of unknown origin were correctly

identified using this classification method. Nine of the ten samples of unknown origin were not grouped with the known sample that came from the same vehicle location.

Since much of the variation between samples occurs in the Group C region of the EEM spectra, a second grouping scheme was used that relies solely on that peak number and position information was performed. This scheme did not result in improved associations.

Table 13 shows the results from the pyridine extract sample group. These data were not used in the classification scheme since so many of the EEMs exhibited very weak intensities. This made the identification of peaks in the Group A region (200 nm – 225 nm) impossible to determine from 19 of the knowns and 5 of the unknowns.

Table 13. Classification with Pyridine Extract EEM Spectra.

Pyridine Extract Groups	Sample	No. of Peaks in Group A	No. of Peaks in Group B	No. of Peaks in Group C	Group C Peak 1 (Ex., Em.) nm	Group C Peak 2 (Ex., Em.) nm	Group C Peak 3 (Ex., Em.) nm
Group 1	13	0	1	1	285, 342	ND, ND	ND, ND
Group 2	M10	0	1	2	260, 320	280, 327	ND, ND
Group 3	15	0	2	2	304, 380	304, 415	ND, ND
Group 4	22	0	2	1	390, 450	ND, ND	ND, ND
Group 5	81	1	1	3	260, 326	280, 331	203, 317
Group 6	102	1	2	1	280, 340	ND, ND	ND, ND
	14	1	2	1	280, 451	ND, ND	ND, ND
Group 7	M40	1	2	2	270, 310	273, 313	ND, ND
Group 8	73	2	1	1	265, 328	ND, ND	ND, ND
	23	2	1	1	280, 408	ND, ND	ND, ND
	65	2	1	1	280, 330	ND, ND	ND, ND
	82	2	1	1	280, 306	ND, ND	ND, ND
	M36	2	1	1	280, 328	ND, ND	ND, ND
	54	2	1	1	290, 320	ND, ND	ND, ND
Group 9	21	2	1	2	286, 340	303, 458	ND, ND
	64	2	1	2	260, 323	282, 343	ND, ND
	M28	2	1	2	259, 228	280, 230	ND, ND
Group 10	11	2	2	0	ND, ND	ND, ND	ND, ND
Group 11	42	2	2	1	284, 312	ND, ND	ND, ND
	55	2	2	1	279, 327	ND, ND	ND, ND
	M12	2	2	1	259, 332	ND, ND	ND, ND
Group 12	53	2	3	1	262, 318	ND, ND	ND, ND
Group 13	95	3	1	1	256, 370	ND, ND	ND, ND
	52	3	1	1	280, 310	ND, ND	ND, ND
Group 14	24	3	2	2	280, 312	280, 328	ND, ND
Group 15	74	3	2	1	275, 328	ND, ND	ND, ND
	75	3	2	1	287, 342	ND, ND	ND, ND

Table 13. Classification with Pyridine Extract EEM Spectra. (continued)

Group 16	63	4	1	1	282, 339	ND, ND	ND, ND
Group 17	92	4	1	2	258, 320	283, 332	ND, ND
	62	4	1	2	270, 320	279, 330	ND, ND
Group 18	104	4	2	2	265, 318	294, 351	ND, ND
Group 19	12	4	3	2	269, 345	280, 450	ND, ND
Group 20	83	4	3	3	260, 315	260, 323	280, 324
Group 21	61	6	1	2	259, 329	279, 331	ND, ND
Group 22	M24	ND	1	0	ND, ND	ND, ND	ND, ND
Group 23	M29	ND	1	1	258, 320	ND, ND	ND, ND
	45	ND	1	1	274, 322	ND, ND	ND, ND
	31	ND	1	1	280, 310	ND, ND	ND, ND
	32	ND	1	1	280, 310	ND, ND	ND, ND
	35	ND	1	1	280, 310	ND, ND	ND, ND
	25	ND	1	1	283, 330	ND, ND	ND, ND
Group 24	44	ND	1	2	260, 325	282, 347	ND, ND
	M19	ND	1	2	262, 318	290, 311	ND, ND
Group 25	M6	ND	2	0	ND, ND	ND, ND	ND, ND
Group 26	101	ND	2	1	258, 320	ND, ND	ND, ND
	72	ND	2	1	260, 323	ND, ND	ND, ND
	33	ND	2	1	280, 310	ND, ND	ND, ND
	34	ND	2	1	280, 310	ND, ND	ND, ND
	51	ND	2	1	385, 360	ND, ND	ND, ND
Group 27	41	ND	2	2	259, 324	260, 310	ND, ND
	94	ND	2	2	259, 325	278, 312	ND, ND
	71	ND	2	2	260, 323	286, 314	ND, ND
Group 28	93	ND	3	3	260, 317	279, 317	283, 311
Group 29	103	ND	4	3	260, 321	273, 336	285, 402
Group 30	105	ND	ND	1	257, 320	ND, ND	ND, ND
	84	ND	ND	1	259, 323	ND, ND	ND, ND
	85	ND	ND	1	260, 317	ND, ND	ND, ND
	91	ND	ND	1	260, 323	ND, ND	ND, ND
	M46	ND	ND	1	260, 313	ND, ND	ND, ND

The spectral classification scheme based on peak number and position proved to be highly distinguishing but not reproducible in this case. Combination of data from pentane and toluene extracts uniquely identified all but 5 pairs of samples. One of these pairs shared a common origin. However, this classification scheme was not successful at correctly associating the knowns with the samples of unknown origin. This could be due to the fact that known and unknown sample concentrations were adjusted using residue weight rather than HPLC data. Differences in concentration of fluorescing components are known to cause wavelength shifting. Samples at low concentrations could have experienced loss of smaller secondary peaks. If the unknown and the known exhibit different concentrations of fluorescing compounds they would be incorrectly grouped using a system that relies on peak number as well as peak position to classify spectra. This could also be due to the sample origins having a high degree of heterogeneity at the sample source itself. Every care was taken to ensure that the multiple swabs from the same vehicle location were applied evenly to the same areas. If the areas touched by the swabs varied in their composition of fluorescing compounds in that area then two swabs from the same vehicle location may contain very different compounds. This could occur if the original motor vehicle undercarriage residues exist as a layer or heterogeneous mixture where the sample composition could vary by grease-layer depth. A variation in the spectral classification scheme which relied only on Group C peaks (i.e. those at excitation λ s above 250) did not yield more accurate results.

3. Low Temperature Fluorescence

The discriminating power of the fluorescence spectra from the low temperature data collection were determined using spectral libraries constructed using Grams Spectra ID software. Three libraries were constructed from the data generated from the toluene extracts; one at excitation wavelength 220 nm, a second at excitation wavelength 235 nm and a third at synchronous scanning $\Delta 70$ nm.¹¹

All 10 unknowns at each excitation (or synchronous) setting were searched against the libraries. Using the library of spectra collected at excitation 220 nm, 2 of the 10 unknowns were identified to the correct motor vehicle (M19 associated to 51A and 54A, M10 associated to 84C) in the top three library search results. And 1 of those associations was to the correct motor vehicle location (M19 and known 54A). Using the library of spectra collected at excitation 235 nm, 3 of the 10 unknowns were associated to the correct motor vehicle (M29 associated to 41A, M06 associated to 42A, M24 associated to 42A) in the top three results and 1 of those associations was to the correct motor vehicle specific location (M29 and known 41A). Using the library of spectra collected at synchronous $\Delta 70$ nm, 3 of the 10 unknowns were associated with the correct motor vehicle (M46 associated to 81A, M19 associated to 55A, and M24 associated to 41A) in the top three library search results. One of those associations was to correct motor vehicle location (M29 and known 41A). No other unknown was associated with

¹¹ The spectra generated at low temperature from the pyridine extracts were not used due their weak fluorescence intensity and their similarity to the peaks seen on the spectra from the toluene extracts.

their correct source motor vehicle. These results, along with the H.Q.I. values are summarized in the table below. The closer the H.Q.I. value. is to zero, the better the library match.

Figure 30. Summary of hits obtained using spectral libraries. (Yellow = library search result to correct vehicle, pink = library search result to correct vehicle and specific location).

	Ex. 220nm		Ex. 235 nm		$\Delta 70$	
	MV Location	H. Q. I.	MV Location	H. Q. I.	MV Location	H. Q. I.
M19	51A	0.037595	33A	0.034737	105C	0.193519
	14C	0.04079	41A	0.039597	41A	0.193519
	54A	0.043603	43C	0.042814	55A	0.194749
M40	43C	0.010982	33A	0.015491	81A	0.211121
	42C	0.016132	41A	0.017078	33B	0.212594
	54C	0.016327	74C	0.02427	53A	0.213347
M29	103B	0.010294	33A	0.012905	41A	0.131694
	11B	0.010406	81A	0.018088	105C	0.132507
	14A	0.011237	41A	0.025181	55A	0.136528
M10	52C	0.077357	41C	0.357834	55A	0.585036
	84C	0.086463	32C	0.039889	91A	0.585536
	53B	0.093352	33B	0.040026	54A	0.588557
M06	53C	0.029646	52A	0.033286	81A	0.36348
	23C	0.02987	42A	0.036255	105C	0.3657
	82B	0.029872	33B	0.036751	94B	0.3676
M24	61B	0.010009	52A	0.037216	105C	0.386102
	64A	0.014316	34C	0.037803	94B	0.387406
	63A	0.015576	42A	0.03991	41A	0.387497

Interestingly, the one unknown that was correctly grouped with its exact motor vehicle source location with the comparison of EEM fluorescence spectra, M19, was also associated to the correct location (car 5 location 4) using the excitation = 220 nm spectral library. This indicates that the two methods are comparably ineffective at making correct associations between known and unknown. It should be noted that the top three library search results for each unknown as shown in the table above, rarely show matches to multiple samples from the same location. One would expect the triplicate swabs (i.e. the three swabs from the same vehicle and location) would be grouped together in the library search result list regardless of their overall position on that list. Since, this did not occur, it indicates that there is a fair amount of heterogeneity among the triplicate swabs taken from the same location. This could be due to actual compositional differences if the motor vehicle lubricants components vary with lubricants layer depth and are not a homogeneous mixture. Future studies should investigate alternative methods of residue collection designed to reduce this variability and collect more representative and homogeneous samples.

4. Characterization Using the HPLC Results

As previously stated, the chromatograms produced from extract solutions from triplicate swabs (i.e. three swabs from the same vehicle location) demonstrated excellent reproducibility and inter-vehicle-location variability in peak size, retention time and size and shape of the unresolved envelope. However, the technique was not used to determine concentration prior to fluorescence due to solvent incompatibility issues. The

chromatograms produced from the swabs of unknown origin were compared to these known sets on four criteria and the outcomes reported in the results section.

Comparisons of the knowns to the samples of unknown origin were based on both numeric (peak retention time) and spatial (unresolved envelope shape) characteristics.

The table below shows each unknown and the predicted vehicle origins from the HPLC data and the actual origins as documented in the unknown key created by Ms. Grimly.

The results show 7 of 10 correct associations made to vehicles with 6 of 10 to exact to the specific location on that vehicle. This data exhibits far better associative results than those achieved from the examination of peak number and position of EEM plots. This is especially remarkable considering that these data were collected in a solvent system designed for complete elution and maintenance of a stable baseline rather than optimum peak resolution and qualitative analysis.

Table 14. Determination of Unknown Residue Origin Using HPLC Data.

Unknown	Actual Origin		Predicted with HPLC data	
	Origin Motor Vehicle (MV)	Origin MV location	Predicted Motor Vehicle (MV)	Predicted MV location
M46	07 Ford Escort	071 Right Ball Joint	02 Ford Bronco II XLT	024 Engine Oil Pan
			09 Chevrolet Cavalier RS	091,092 R & L Ball Joints
M40	08 BMW 740 iL	083 Front Skid Guard	08 BMW 740 iL	083 Front Skid Guard
M36	06 Mercury Marquis	064 Engine Oil Pan	06 Mercury Marquis	064 Engine Oil Pan
				063 Undercarriage Position 1
M29	04 Ford Taurus	041 Right Ball Joint	04 Ford Taurus	041 Right Ball Joint
M12	01 GMC Denali	015 Undercarriage Position 2	01 GMC Denali	011 Left Ball Joint
				012 Right Ball Joint
				014 Undercarriage Position 1
				015 Undercarriage Position 2
M28	06 Mercury Marquis	061 Right Ball Joint	02 Ford Bronco II XLT	025 Under Radiator
M19	05 Honda Accord	054 Undercarriage Position 1	05 Honda Accord	055 Undercarriage Position 2
M10	08 BMW 740 iL	081 Right Ball Joint	08 BMW 740 iL	081 Right Ball Joint
M6	04 Ford Taurus	044 Under Radiator	05 Honda Accord	051 Right Ball Joint
M24	04 Ford Taurus	043 Engine Oil Pan	04 Ford Taurus	043 Engine Oil Pan

Number of correct vehicle associations = 7 of 10.

Number of correct vehicle location associations = 6 of 10.

Discussion

In this study, three analytical data sets were collected: room temperature EEM fluorescence, low temperature excitation and synchronous averaged scans at selected wavelengths, and chromatograms from the pentane extracts. Of those three, the chromatographic data exhibited the most agreement among triplicate swabs from the same motor vehicle location and more importantly, the best ability to associate motor vehicles of an unknown origin to the known collected from the same motor vehicle location.

Conclusion 1

Three-dimensional fluorescence spectroscopy, as performed in this experiment, and collected at room temperature, of motor vehicle undercarriage residues is highly distinguishing but unreliable for the successful association of residues of unknown origin to their source motor vehicles by the methods employed in this study.

Excitation-emission matrices are excellent representations of the overall fluorescence emission characteristics of a material. The ability to display data at multiple excitation wavelengths as a contour plot is a simple way to represent a large amount of spectral data. This type of data presentation is best suited to observe gross characteristics of fluorescence spectra (e.g. the number and position of multiple peaks at multiple wavelengths). It is not suitable for the representation of fine spectral detail. In this study, fluorescence EEMs were not a good tool to distinguish among motor vehicle residues with similar compositions from different sources.

Regarding the usefulness of this technique to distinguish among motor vehicle residues, an analogy can be made between the motor vehicle grease residues and the highly complex mixtures encountered during the analysis of ignitable liquid residues (ILRs) from fire debris. Ignitable liquid residues are multicomponent mixtures produced mostly from the refining of petroleum, that through use and exposure are subject to degradation and weathering. Gas chromatography – mass spectrometry (GC/MS) is the preferred analytical instrumentation for ILRs because it performs the separation of complex mixtures and supplies the mass spectral data necessary to identify compounds. If GC/MS is not available, the GC with a flame ionization detector will be able to identify many of the known ILRs through comparison of chromatographic patterns. Complex mixtures are not suitable for analysis by mass spectrometry alone. As chromatographic separation is more useful than the spectral data for the identification of ILRs of unknown origin, the same appears to be true of motor vehicle undercarriage residues.¹²

Effectiveness of the Extraction Method

The three-solvent extraction system was adopted in order to separate the motor vehicle residues into three distinct compound classes. In all extracts, the peaks in the Group A and B regions were the most intense and the variations peak λ maxima were not consistent among extracts from different solvents. This indicates that the three solvent extraction system did not result in three fractions with different groups of fluorophores.

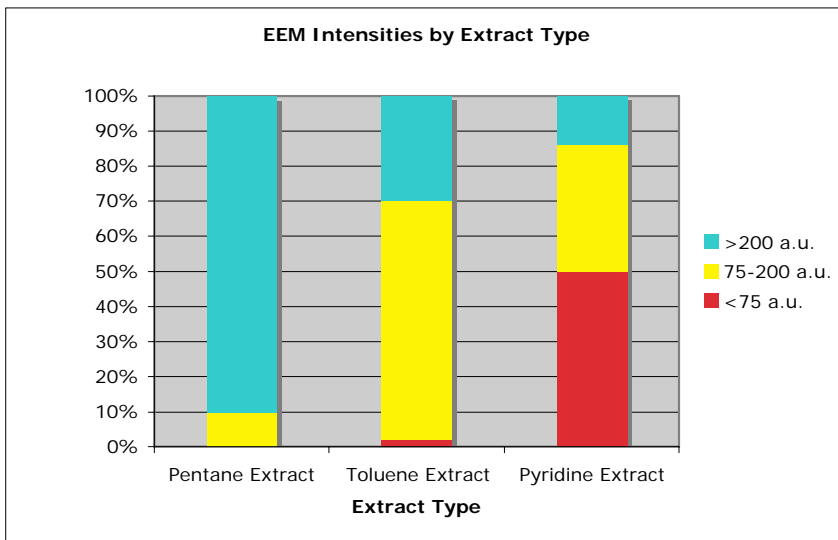
¹² This analogy is not meant to equate mass and optical spectral data, just their usefulness in analyzing complex mixtures.

The Group C region of the EEM (excitations above 250nm) exhibited some spectral differences from in each extraction type. Ultimately, the fluorescence spectra data set from the successive three-solvent extraction system did not successfully associate the ten samples of unknown origin with the motor vehicles of origin.

Since the overall fluorescence intensity decreased with each successive extraction type (pentane → toluene → pyridine) along with the number of peaks observed, I advise that future research proceed with a single extraction procedure designed to maximize extractions of fluorescing compound types. Figure 31 shows how the fluorescence intensity of the most intense peak observed in the excitation emission matrix changed with each extraction.

Figure 31. Percentage of EEMs collected and the distribution of Fluorescent intensities.

N=50



This decrease in intensity seen from the first extraction (pentane) to the third (pyridine) could be due to two factors. First, the three solvent extraction system successfully separated the motor vehicle residues in the oil, asphaltene and pre-asphaltene fractions and fluorescence component of the total extract residue decreased with each class of compounds. Or, second, the extraction system was not successful at the separation by compound class and the total fluorescing component of the motor vehicle undercarriage residue was unevenly distributed among the three fractions. Since no qualitative analysis of individual compounds in each fraction was performed it is not possible to opine on the actual components of the extractions. The fact that all three extract types exhibited their major peaks at the same excitation ranges (200 nm – 250 nm) that all three contained similar compound types, however, the less intense peaks present in the group C region (>250 nm) indicate compositional differences between the pyridine extracts and the other two solvent extracts.

Conclusion 2

In this study, adjusting concentration of extract residue solutions based on residue mass (e.g. w/w ppm) did not adjust all the fluorescing components of each triplicate swab to the same concentration.

Using total residue weights to bring motor vehicle undercarriage residues from the sample source location to a common concentration did not result in common fluorescent intensities among samples from the same source. It was thought that if the ratio of fluorescence compounds was the same in each swab from the same motor vehicle

location, then adjusting concentration based on to the total extract mass would also bring the fluorescing compounds to a common concentration. Since the fluorescent intensities of the major peaks in spectra from each triplicate sample were not the same, this was not the case.

Without adjusting the concentration of the fluorescing compounds alone it is not possible to know if weak fluorescence emission observed in one sample, but not in another, is due to the compositional or concentration differences. This could lead to false exclusions in very dilute samples.

Conclusion 3

In the conditions used in this study, low temperature fluorescence emission scans, collected at 77K, of motor vehicle undercarriage residues exhibit increased resolution but are not reliable for the successful association of residues of unknown origin to their source motor vehicles.

The spectra collected at 77K exhibited increased resolution. This technique has some specific challenges. The sample compartment must be kept at extremely low humidity and dry nitrogen gas must flow at all times onto the tip of the Dewar in order to keep it frost free. The long sample tubes require an extensive cleaning procedure to ensure no sample carryover. After each cleaning, the tubes were scanned with the sample solvent to ensure that there was no carry-over from one sample to the next. The tubes

themselves are subject to breakage if the frozen solution is warmed too quickly. Clear freezes are only accomplished if the tube itself is lowered very slowly into the liquid nitrogen. Despite these disadvantages, the apparatus itself is very useful for generating spectra of increased resolution. However the observed increase in resolution did not result in an increased ability to distinguish among motor vehicle undercarriage residues, nor did it allow for a more successful association of the unknown residues with their respective source vehicles.

Conclusion 4

In this study, chromatography of motor vehicle undercarriage residues was better than room temperature or low temperature fluorescence spectroscopy at both discriminating among vehicle locations and making correct associations between samples of known origin and samples of unknown origin.

The successful associations made using the chromatographic data (7/10 unknown origin vehicles and 6/10 unknown vehicle locations) have important implications for future research. These data were produced primarily to determine concentration. They were not collected under conditions meant to optimize resolution and specific peak identification. The primary goal was to accomplish complete elution of all compounds injected to avoid carryover and then use total peak areas to represent overall concentration. Solvent incompatibility between the low temperature apparatus and the

HPLC system resulted in the a design shift to determine concentration by residue weights instead. The chromatographic data set exhibited poor peak resolution but excellent reproducibility among triplicate knowns and variation in chromatographic patterns between knowns. As a result the chromatographic data was more useful as a means to associate samples of unknown origin to their source vehicles rather than as a means to adjust concentration.

Conclusion 5

The low temperature apparatus, as designed, can be used to successfully generate fluorescence spectra that exhibit greater resolution than room temperature measurement.

The low temperature apparatus, after extensive modification and testing, can be used to produce fluorescence spectra with better resolution than room temperature data collection. Once frozen, the rigid matrix successfully inhibits band broadening due to non-radiative relaxation of excited fluorophores. This increase in resolution was accompanied by a decrease in fluorescence intensity.

The low temperature apparatus was successfully used to generate spectra with better resolution. This technique would be very useful to research projects that aim to better understand the specific electronic transitions that occur during fluorescence

emission. Future work with low temperature fluorescence for characterization of motor vehicle residues should proceed once a Dewar is obtained that can maintain a single freeze, without refilling the Dewar, for at least 30 minutes. This will provide enough time to obtain an excitation emission matrix data set for each sample. Low temperature work in this study was limited to the collection of scans at several excitation wavelengths. This did not provide the spectral detail necessary to successfully distinguish between motor vehicles nor did it provide enough information to make successful associations among all the samples of unknown origin and known samples using spectral library searching software. Additionally, future work should investigate alternative fluorescence instrumentation available to collect fluorescence spectra at low temperature.

Recommendation 1

I recommend that future research not use fluorescence excitation emission matrices, collected under the parameters used in this study, to distinguish among motor vehicles undercarriage residues until a suitable method is developed to establish the concentration of fluorescent materials in the sample. Basing sample concentration on the mass of all compounds extracted, as opposed to, relative amount of fluorescence compounds does not result in spectra of similar fluorescent intensity for samples from the same source motor vehicle.

Recommendation 2

Based on the results of these experiments, I do not recommend that low temperature fluorescence be used to distinguish among different sources of automotive undercarriage residue until a more reliable method of determining sample concentration is developed, most likely through a chromatographic process, and until an instrument design is identified that better suits the needs of the quartz-tipped Dewar.

This work confirms the recommendation made by Purcell about the concentration adjustments necessary to sample prior to collection of fluorescence spectral data. Residue weights are not ideal as they do not represent the overall concentration of the fluorescing compounds in the sample. Chromatographic methods that are more compatible with the solvent needs of low temperature analysis are necessary.

Recommendation 3

Future work should use HPLC with fluorescence detection to characterize motor vehicle undercarriage residues.

If successful associations can be made using a simple isocratic elution method with single wavelength UV absorbance detection methods, then it stands to reason that improvements in individualization will be obtained at conditions of higher chromatographic resolution and sensitivity. I recommend that future work be done using

HPLC with fluorescence detection. A fluorescence detection system would provide a better sensitivity than UV absorbance detection.

Impact on Criminal Justice and Forensic Science Policy and Practice

The contribution of this research to the field of forensic science is the advancement of knowledge about the analytical techniques applied to a long-observed transferable material, motor vehicle undercarriage residue, to attempt to create associations between automobiles and persons or objects. When the best analytical technique is identified, these associations can be used to help the trier of fact determine the likelihood that a specific automobile came into contact with a person or object. This project illustrates the inability of room temperature three dimensional fluorescence spectroscopy and low temperature two dimensional fluorescence spectroscopy to discriminate among transferred petroleum materials. This type of evidence can be found in “hit and run” type vehicular crimes.

Analysis of transfer evidence is one of the oldest and most challenging types of forensic analysis due to the limitless number of material types that may transfer between persons and/or objects during the commission of a criminal act. (Houck, 2001) The vast majority of the resources provided to forensic science laboratories and forensic science researchers through federal funding are put towards the forensic disciplines that are able to identify suspects through direct associations (i.e. DNA and fingerprints) (NAS, 2009).

While there is no doubt that both are very powerful investigative tools, the narrowing field-of-view by both funders and criminal justice researchers could have the unforeseen consequence of the underdevelopment of new and novel transfer evidence analysis techniques.

Transfer evidence is highly variable in composition and mode of transfer. The analyst must master a myriad of identification and quantification methods in order to proficiently analyze physical evidence. This research sought to add tools to the forensic chemist's arsenal in order to expand the impact of this forensic discipline. The conclusions of this project and the recommendations made for future research expand the body of knowledge about this form of transfer evidence and about the lessons one can apply from traditional oil and grease analysis.

This project is one of many that have investigated the discriminating power of an instrumental technique on a specific class of materials. Motor vehicle undercarriage residues have been observed as transferred materials in casework but are not routinely reported as a subset of forensic chemistry (Durose, 2008). The conclusions of this research project may be used to guide future research that continues to seek the optimal analytical tools needed to distinguish among motor vehicle undercarriage residues. Once future research identifies that method, more forensic laboratories may include oil and grease analysis in their forensic chemistry sections, thereby increasing the evidentiary

value of this physical evidence type as well as the impact of transfer evidence analysis on criminal investigations and prosecutions.

Summary

Overall, the fluorescence and chromatographic techniques used to differentiate transferred vehicle undercarriage residue from different vehicle locations and to associate samples of unknown origin with the correct source vehicle have been produced results of varying reliability. Chromatographic data proved to be more reliable than spectral data for this purpose. In this study and with the instrumentation used, the EEM contour plots collected at room temperature with overlapping excitation scans produced sufficient variation to differentiate most motor vehicle samples but did not successfully associate most of the unknown samples. The single scans taken at low temperature exhibited increased peak resolution than room temperature scans but the spectral detail in the five wavelengths collected (three excitation and two synchronous) were not sufficient to differentiate most samples through visual comparison alone. In this study, searching with spectral libraries produced increased differentiation but were also not more successful than visual examination of room temperature EEMs at the correct association of unknowns with known vehicle locations. It would be advantageous to combine the EEM data collection with the increased resolution of the low temperature environments. Any future work using fluorescence EEM to distinguish among motor vehicle undercarriage residues should utilize alternative fluorescence spectrometers that allow the user to have greater control over instrument parameters. Most surprising was the ability of

chromatographic data, collected under isocratic conditions, to correctly associate most of the unknown motor vehicle residues while exhibiting excellent reproducibility among the triplicate samples. Future work in this area should focus on the use of HPLC coupled with fluorescence detection to combine the best of each of these methods to differentiate among transferred vehicle petroleum products.

Appendices

- A. Unknown Labeling Scheme
- B. Cary Eclipse Specifications
- C. Pentane Extract Chromatograms
- D. Pentane Extract Contour Plots
- E. Toluene Extract Contour Plots
- F. Pyridine Extract Contour Plots

Original Set D	Unknown Code
11	M25
12	M37
13	M22
14	M32
15	M12
21	M01
22	M30
23	M17
24	M23
25	M39
31	M14
32	M38
33	M03
34	M08
35	M26
41	M29
42	M34
43	M24
44	M6
45	M41
51	M27
52	M45
53	M49
54	M19
55	M13
61	M28

62	M11
63	M20
64	M36
65	M50
71	M46
72	M35
73	M18
74	M44
75	M48
81	M10
82	M15
83	M40
84	M31
85	M42
91	M2
92	M43
93	M47
94	M04
95	M33
101	M21
102	M9
103	M7
104	M5
105	M16

Appendix B. Cary Eclipse Specifications

Cary Eclipse Specifications:

Monochromators

Excitation: Czerny-Turner, f3.6, 0.125 m focal length,

Emission: Czerny-Turner, f3.6, 0.125 m focal length,

Gratings

Excitation: 30 x 35 mm, 1200 l/mm, blaze at 370 nm,

Emission: 30 x 35 mm, 1200 l/mm, blaze at 440 nm

- All reflective optical system with quartz over-coated optics.
- Schwarzschild source optics for increased energy throughput and precise imaging and focusing.
- 80 data points per second maximum measurement rate in fluorescence mode.
- non- measurement phase stepping wavelength drive.
- room light immunity in fluorescence mode,
- Source Xenon pulse lamp with exceptionally long lifetime, pulsed at 80 Hz. Pulse width at half peak height ~ 2 μ s, peak power equivalent to 75 kW.

Wavelength range (nm)

Mechanical

Excitation: 190 to 1100 nm. Zero order selectable.

Emission: 190 to 1100 nm. Zero order selectable.

Operational

Excitation: 200 to 900 nm with standard PM tube. Zero order selectable.

Emission: 200 to 900 nm with standard PM tube. Zero order selectable.

Wavelength accuracy (nm) \pm 1.5 nm

Wavelength reproducibility (nm) \pm 0.2 nm

Detectors

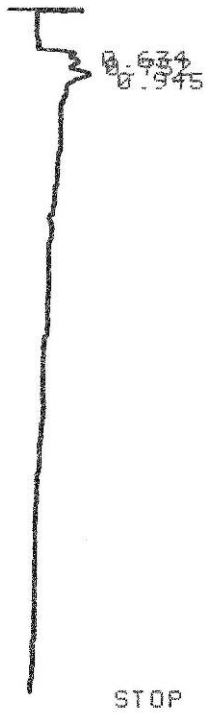
High performance R928 photomultiplier detector and separate R928 PMT

for reference signal.

Sensitivity (Raman Band of Water)

>500:1 RMS, 500 nm excitation, excitation and emission slits 10 nm,
1 sec Signal Averaging time. >750:1 RMS, 350 nm excitation, excitation
and emission slits 10 nm, 1 sec Signal Averaging time.

* RUN # 24 JAN 1, 1901 08:43:19
START



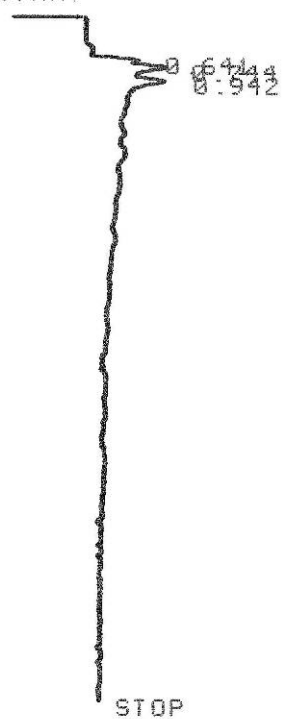
101A

RUN# 24 JAN 1, 1901 08:43:19

AREA#	RT	AREA	TYPE	WIDTH	AREA%
	.634	2228	PU	.073	16.00000
	.752	4539	UU	.144	32.59605
	.945	7158	UU	.192	51.40397

* RUN # 25 JAN 1, 1901 08:54:37
START

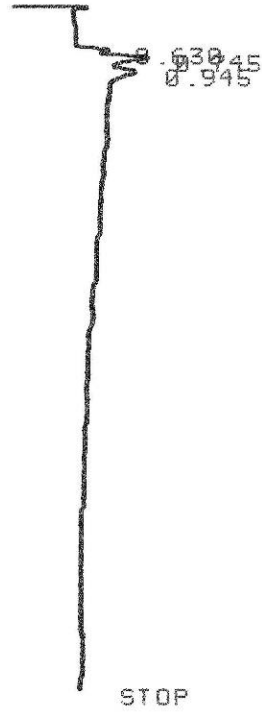
101B



RUN# 25 JAN 1, 1901 08:54:37

RT	AREA	TYPE	WIDTH	AREA%
.641	2489	PU	.067	13.61449
.744	7742	UU	.132	42.34766
.942	8051	UU	.156	44.03786

* RUN # 26 JAN 1, 1901 09:05:55
START

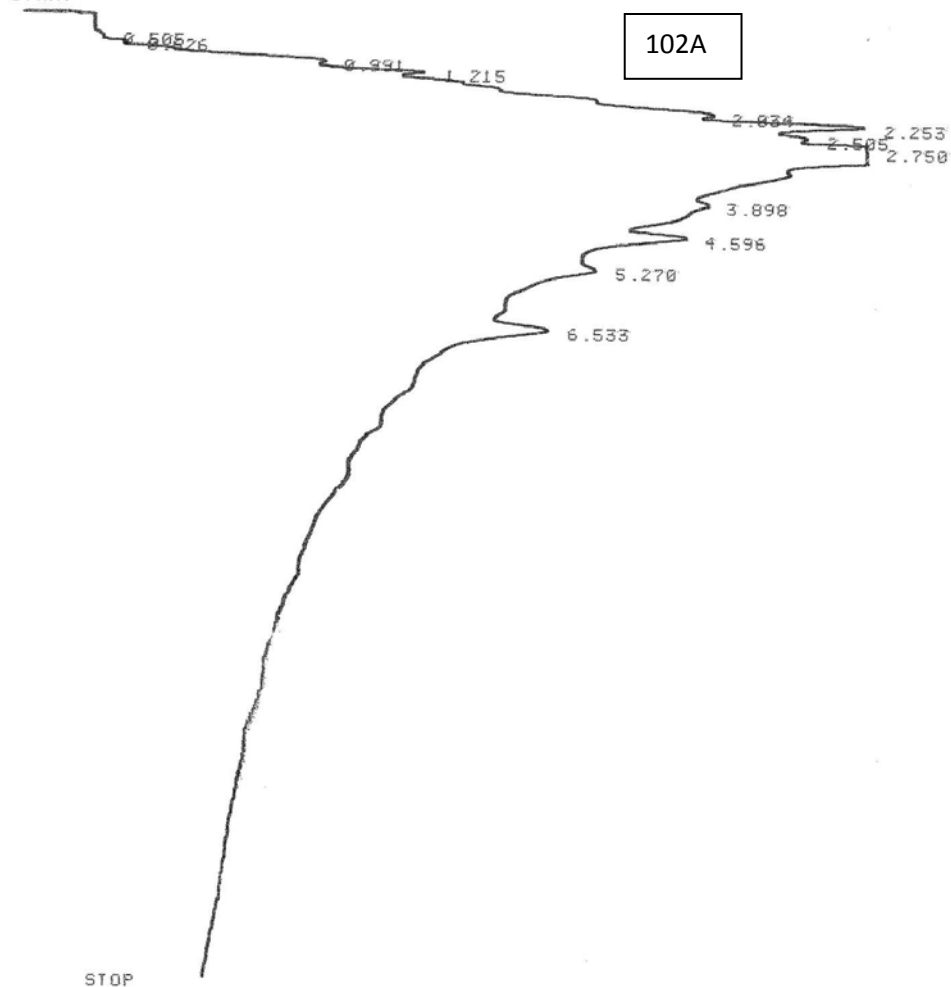


101C

RUN# 26 JAN 1, 1901 09:05:55

AREA#	RT	AREA	TYPE	WIDTH	AREA%
	.630	1574	PU	.064	12.08167
	.745	6089	UU	.117	46.73779
	.945	5365	UU	.155	41.18053

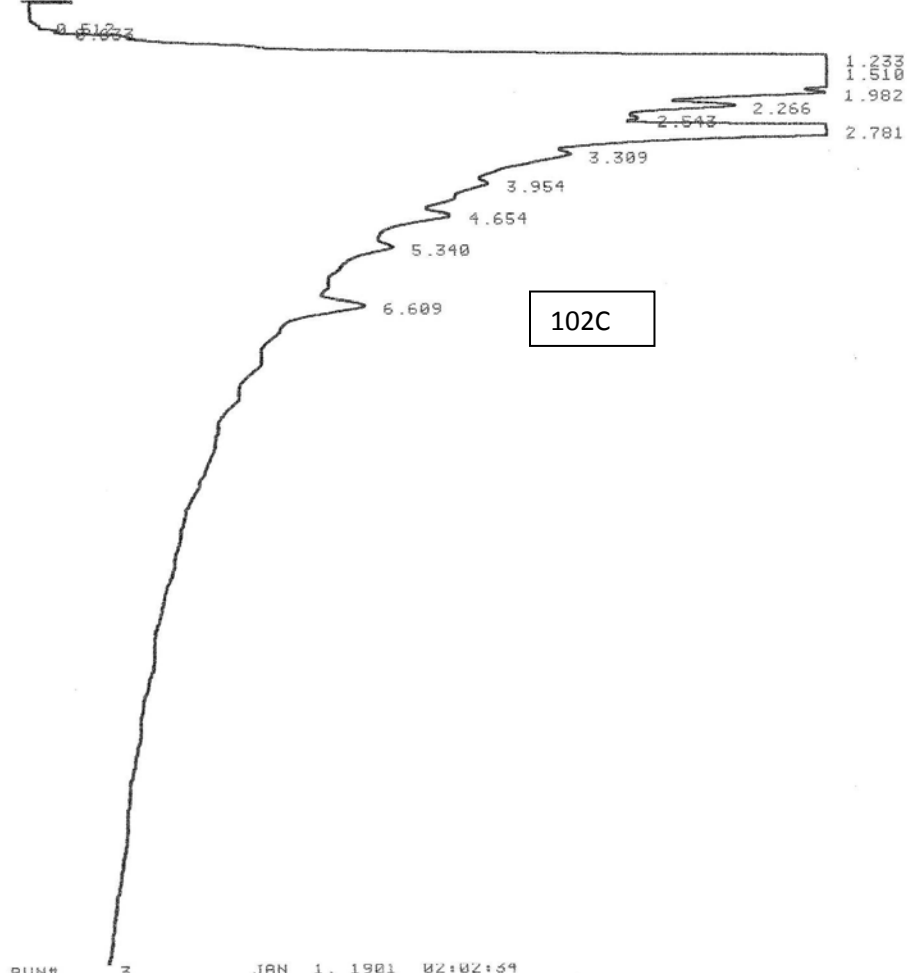
* RUN # 10 JAN 1, 1901 04:00:03
 START



RUN# 10 JAN 1, 1901 04:00:03

RT	AREA	TYPE	WIDTH	AREA%
.505	389	PP	.098	.01889
.626	1467	PU	.062	.07125
.991	54305	UU	.232	2.63745
1.215	60914	UU	.182	2.95843
2.034	385839	UU	.625	18.73919
2.253	199357	UU	.255	9.68225
2.505	124023	UU	.179	6.02347
2.750	657457	UU	.302	31.93097
3.898	260195	UU	.569	12.63699
4.596	168049	UU	.450	8.16170
5.270	131018	UU	.628	6.36320
6.533	15982	BP	.190	.77620

* RUN # 3 JAN 1, 1901 02:02:34
 START



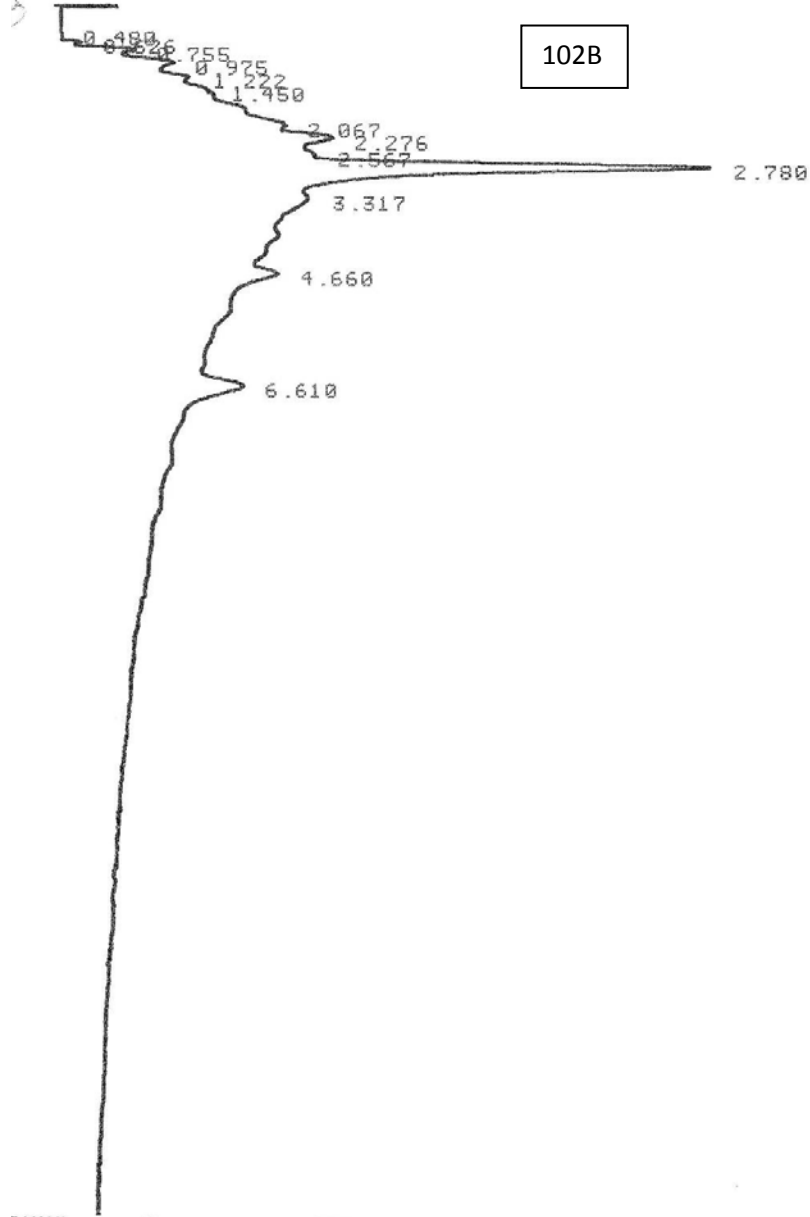
RUN# 3 JAN 1, 1901 02:02:34

AREA#	RT	AREA	TYPE	WIDTH	AREA%
	.512	616	BU	.109	.01688
	.633	1687	UU	.065	.04623
	1.233	460292	UU	.226	12.83314
	1.510	820615	UU	.431	22.70750
	1.982	216106	UU	.225	5.92220
	2.266	233179	UU	.280	6.39000
	2.543	100160	UU	.154	2.96403
	2.781	580332	UU	.328	15.90351
	3.309	332261	UU	.561	9.10534
	3.954	281794	UU	.604	7.72233
	4.654	220710	UU	.558	6.04837
	5.340	217691	UU	.723	5.96564
	6.609	159638	UU	.711	4.37475

RUN # 2 JAN 1, 1901 01:21:15

START

102B

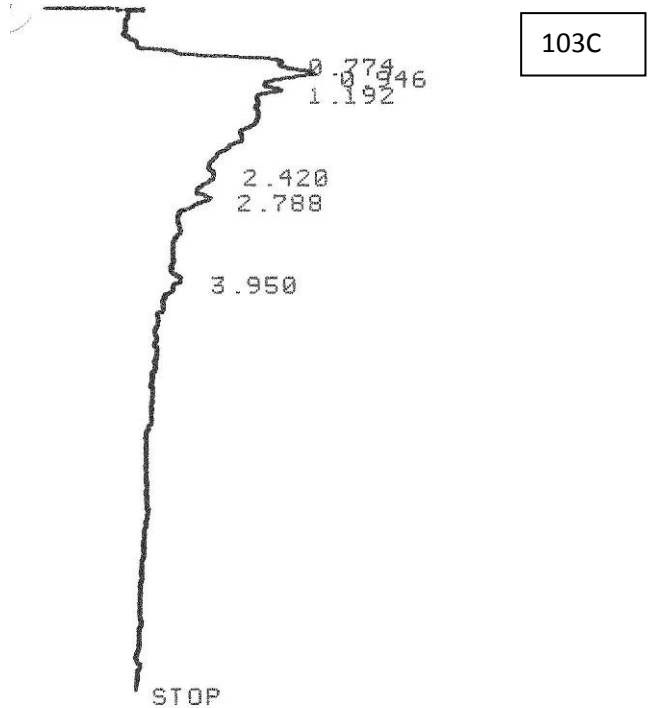


RUN# 2 JAN 1, 1901 01:21:15

AREA#

RT	AREA	TYPE	WIDTH	AREA#
.480	136	PB	.046	.02479
.626	1239	BU	.060	.22585
.755	8081	UU	.113	1.47304
.975	22141	UU	.216	4.03595
1.222	17764	UU	.162	3.23809
1.450	22591	UU	.187	4.11797
*2.067	84648	UU	.480	15.42996
*2.276	57161	UU	.263	10.41952
*2.567	32105	UU	.171	5.85222
*2.780	179637	UU	.308	32.74491
3.317	80915	UU	.523	14.74950
4.660	26415	UU	.361	4.81503
*6.610	15762	UP	.304	2.87316

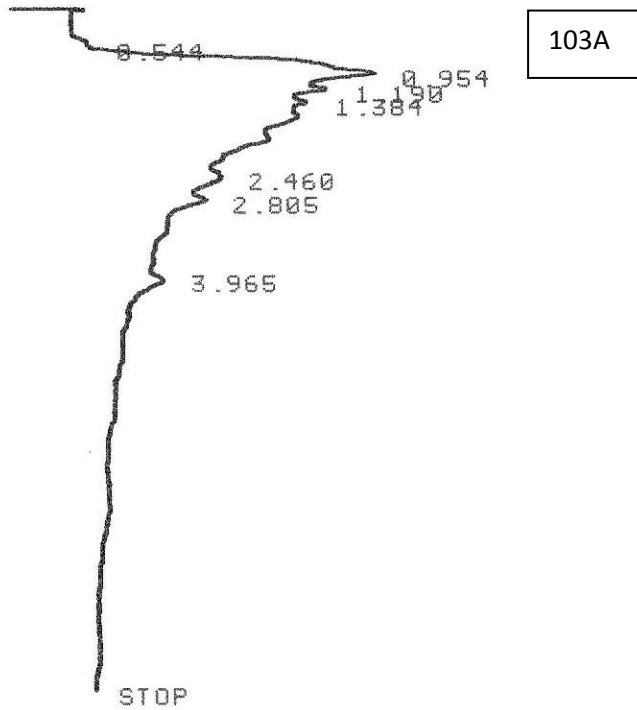
* RUN # 29 JAN 1, 1901 09:39:49
START



RUN# 29 JAN 1, 1901 09:39:49

RT	AREA	TYPE	WIDTH	AREA%
.774	19286	UU	.148	15.55234
.946	40770	UU	.260	32.87718
* 1.192	22249	UU	.175	17.94173
* 2.420	21047	UU	.328	16.97242
* 2.788	15679	UU	.276	12.64364
* 3.950	4976	UU	.188	4.01268

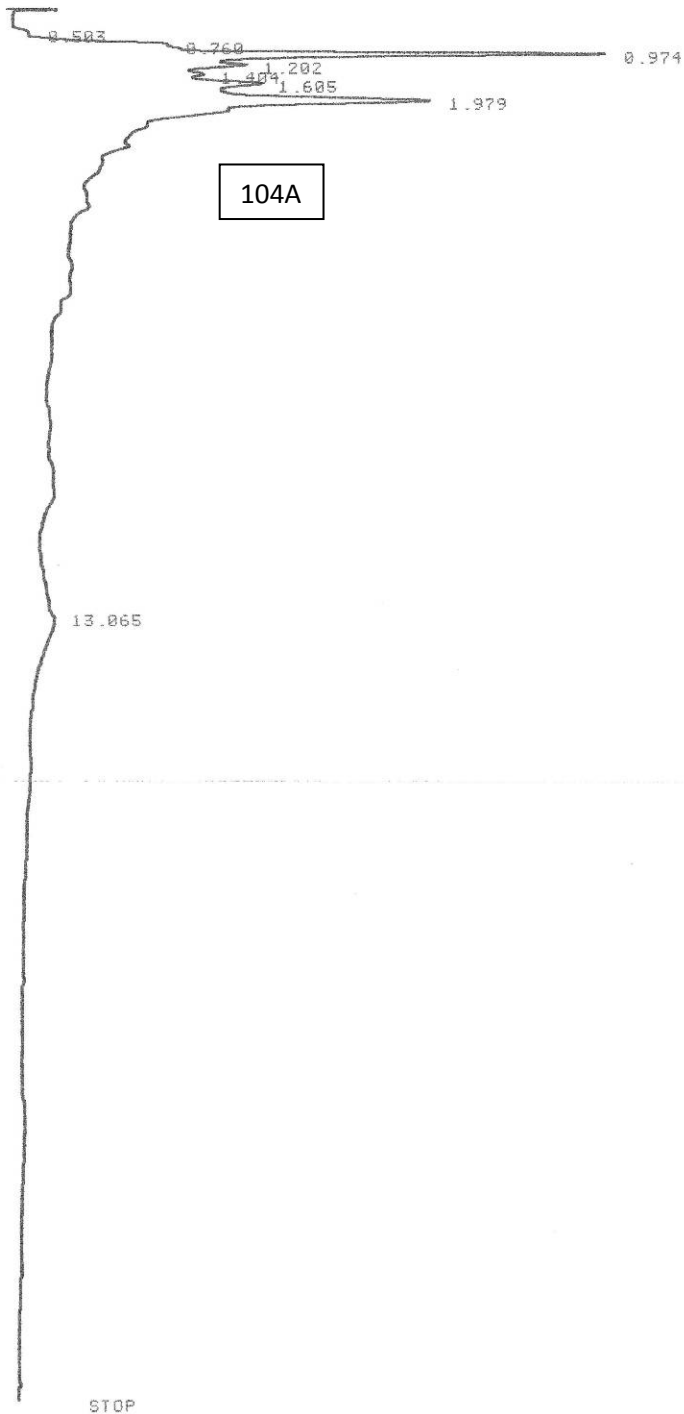
* RUN # 27 JAN 1, 1901 09:17:14
 START



RUN# 27 JAN 1, 1901 09:17:14

AREA#	RT	AREA	TYPE	WIDTH	AREA#
	.544	1757	BU	.123	.72380
*	.954	95531	UU	.376	39.35430
*	1.190	34451	UU	.162	14.19220
	1.384	30965	UU	.161	12.75614
*	2.460	36324	UU	.335	14.96379
*	2.805	27977	UU	.305	11.52522
*	3.965	15741	UU	.352	6.48456

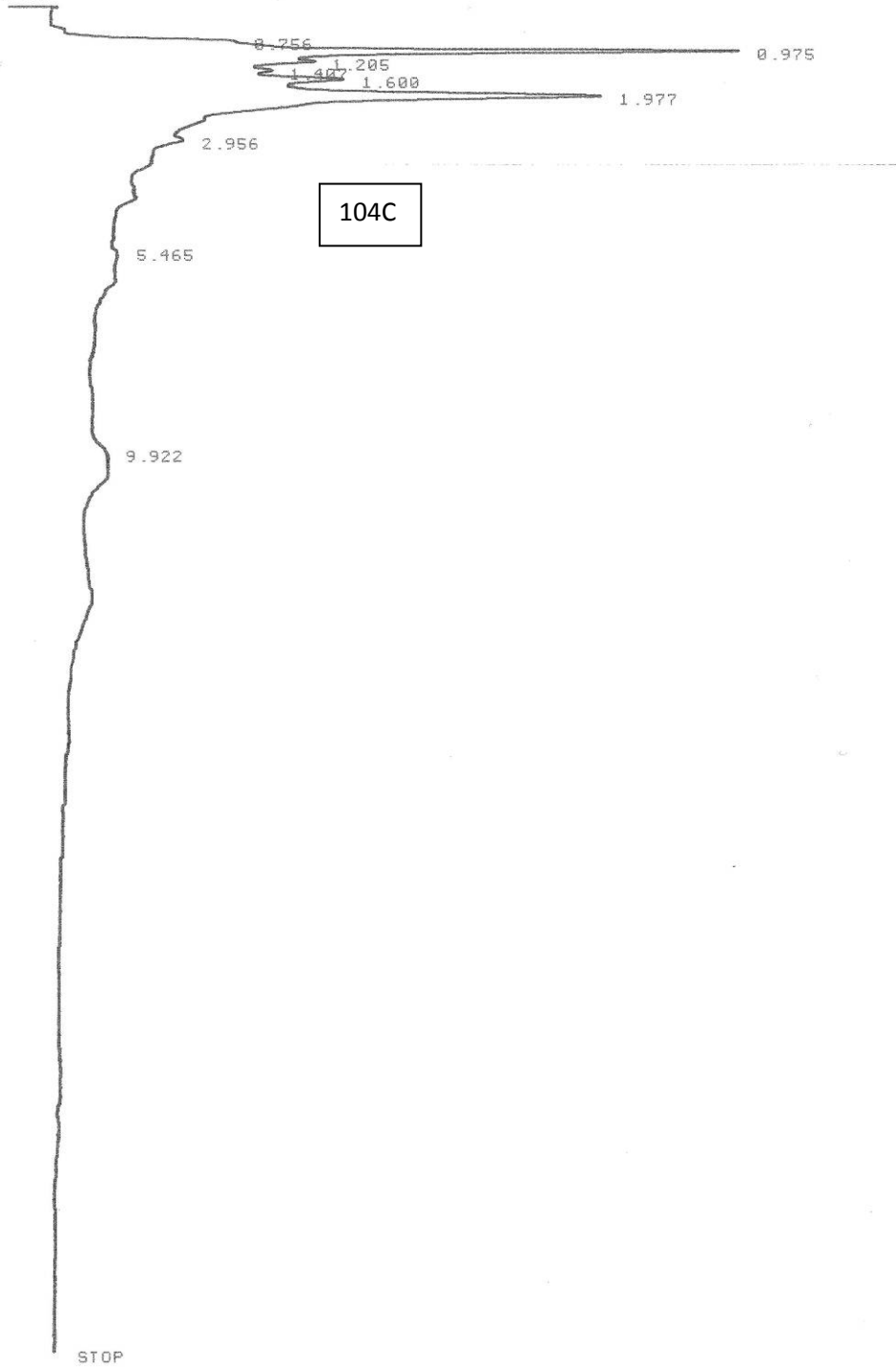
* RUN # 4 JAN 1, 1901 03:06:53
 START



RUN# 4 JAN 1, 1901 03:06:53

AREA#	RT	AREA	TYPE	WIDTH	AREA%
	.503	160	PB	.029	.04054
	.760	18389	PU	.105	4.65875
	.974	121475	UU	.166	30.77498
	1.202	42470	UU	.163	10.75952
	1.404	28735	UU	.144	7.27985
	1.605	57020	UU	.214	14.44569
	1.979	120140	UU	.262	31.85502

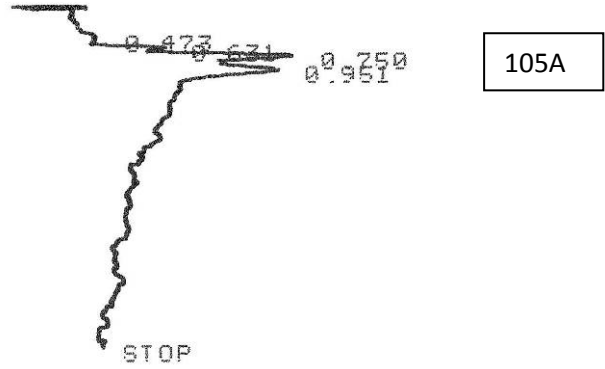
* RUN # 5 JAN 1, 1901 03:38:12
 START



RUN# 5 JAN 1, 1901 03:38:12

RT	AREA	TYPE	WIDTH	AREA%
.756	23377	PU	.103	4.04090
* .975	157724	UU	.181	27.26384
1.205	49990	UU	.162	8.64116
* 1.407	35541	UU	.145	6.14354
1.600	70457	UU	.214	12.17905
* 1.977	223390	UU	.344	38.61472
2.956	16147	UU	.247	2.79114
5.465	71	PB	.028	.01227
9.922	1813	UU	.232	.31339

* RUN # 18 JAN 1, 1901 01:58:38
 START

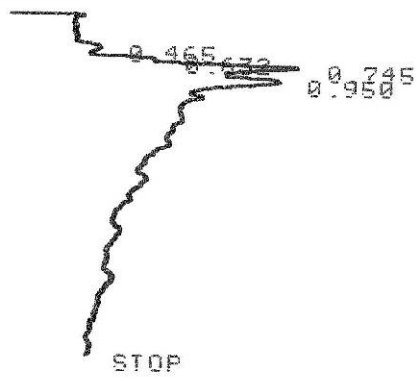


RUN# 18 JAN 1, 1901 01:58:38

AREAX

RT	AREA	TYPE	WIDTH	AREAX
0.473	1331	UU	.163	4.13444
.631	2501	UU	.071	7.76877
0.750	11050	UU	.125	34.32424
0.951	17311	UU	.219	53.77256

* RUN # 5 JAN 1, 1901 00:36:35
 START



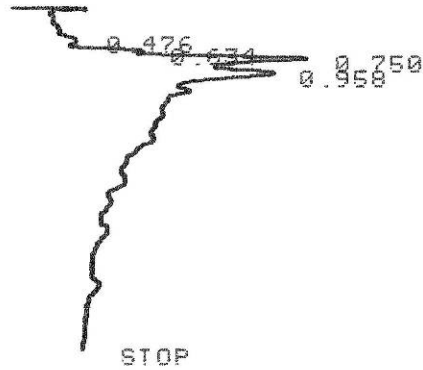
105B

RUN# 5 JAN 1, 1901 00:36:35

AREA#	RT	AREA	TYPE	WIDTH	AREAX
	0.465	1137	PU	.116	3.51849
	.632	1731	UU	.057	5.35665
	*.745	13176	UU	.146	40.77363
	*.950	16271	UU	.205	50.35123

TOTAL AREA= 32215

* RUN # 19 JAN 1, 1901 02:04:56
 START

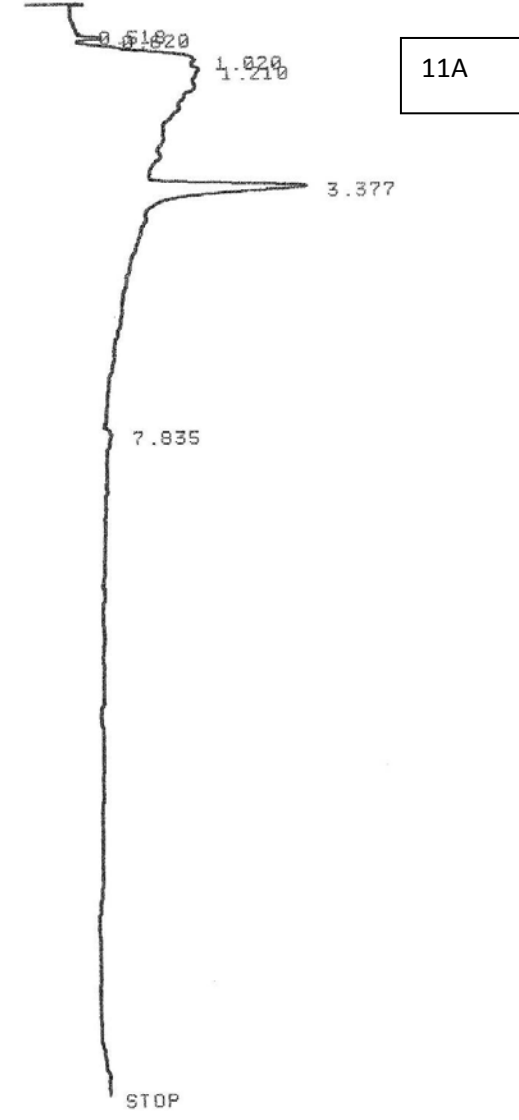


105C

RUN# 19 JAN 1, 1901 02:04:56

AREA#	RT	AREA	TYPE	WIDTH	AREAX
1	0.476	1207	UU	.130	3.17966
2	.634	2332	UU	.065	6.14331
3	.750	14714	UU	.139	38.76186
4	.958	19707	UU	.219	51.91518
TOTAL		37960			

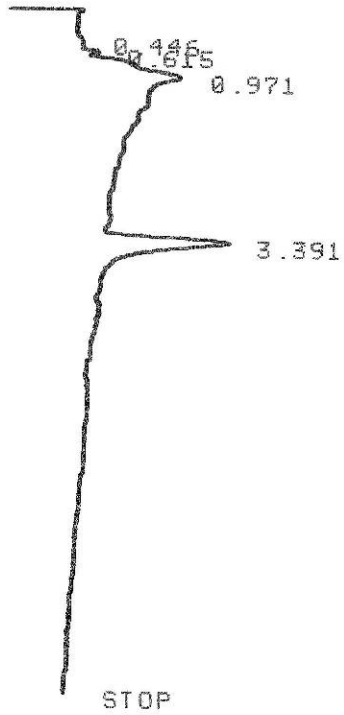
* RUN # 3 JAN 1, 1901 00:54:59
 START



RUN# 3 JAN 1, 1901 00:54:59

AREA#	RT	AREA	TYPE	WIDTH	AREA%
	.518	985	PU	.219	.81147
	.620	1859	UP	.064	1.53149
	1.020	35151	PU	.283	28.95826
	1.210	32075	UU	.253	26.42419
	3.377	50754	UU	.251	41.81242
	7.835	551	PR	.111	.46217

* RUN # 6 JAN 1, 1901 02:05:32
START

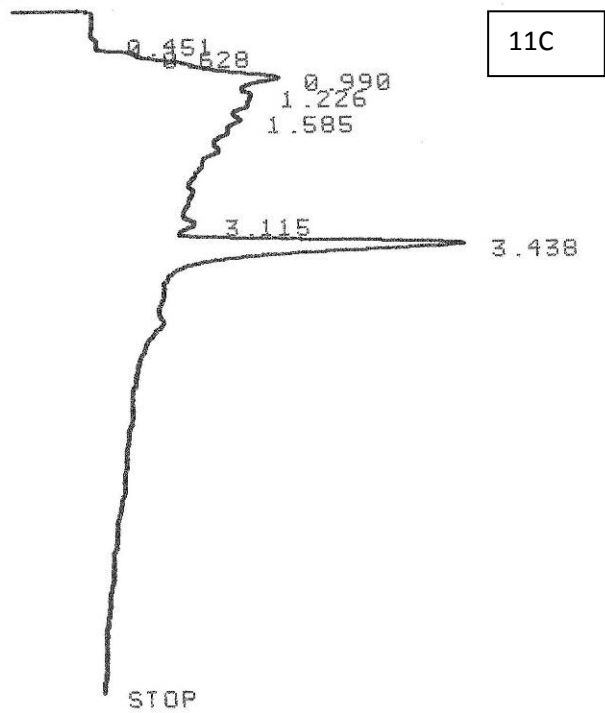


11B

RUN# 6 JAN 1, 1901 02:05:32

AREA%	RT	AREA	TYPE	WIDTH	AREA%
	.446	491	PU	.105	1.01965
	.615	783	UU	.056	1.62603
	0.971	24596	UU	.329	51.07779
	0.3.391	22284	PP	.207	46.27653

* . RUN # 10 JAN 1, 1901 02:50:46
 START



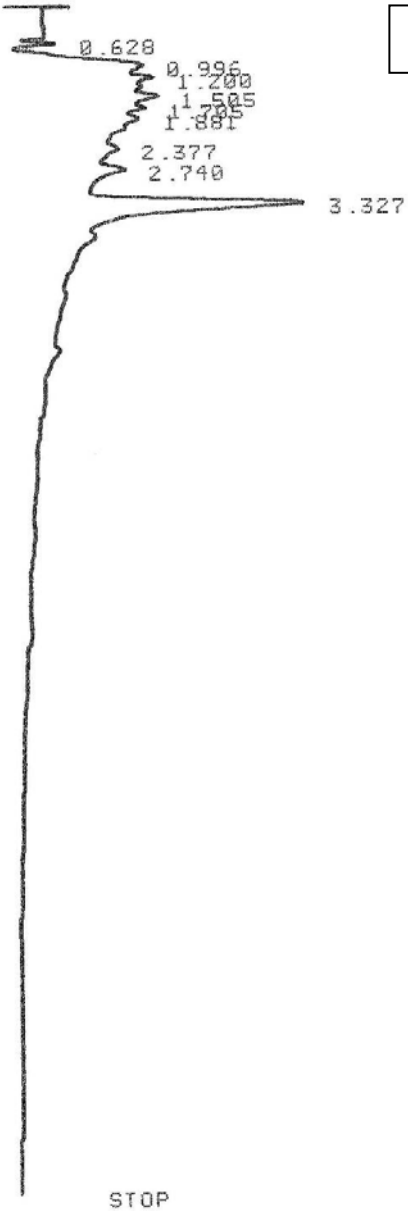
RUN# 10 JAN 1, 1901 02:50:46

AREA#	RT	AREA	TYPE	WIDTH	AREA%
	.451	945	PU	.144	.45774
	.628	2459	UU	.065	1.19110
	.990	52568	UU	.336	25.46307
	1.226	25279	UU	.191	12.24473
	1.585	23138	UU	.193	11.20766
	3.115	17982	UU	.255	8.66175
	3.438	84177	UU	.285	40.77394

TOTAL AREA= 206448
 MUL FACTOR=1.0000E+00

* RUN # 4 JAN 1, 1901 01:16:17
 START

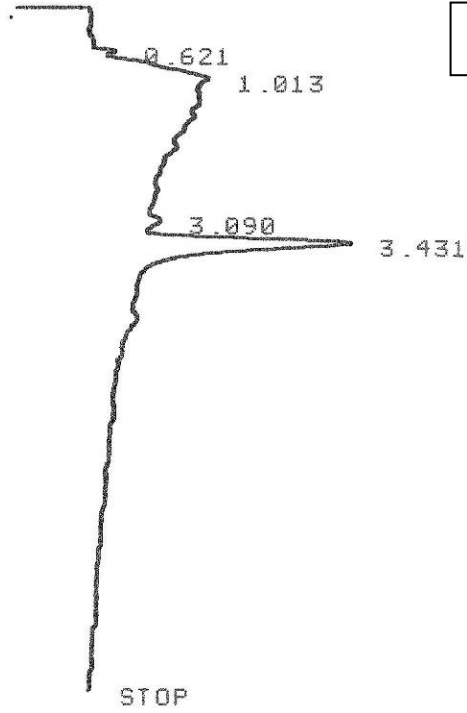
12A



RUN# 4 JAN 1, 1901 01:16:17

RT	AREA	TYPE	WIDTH	AREA%
.628	2377	UP	.062	.90440
.996	31780	PU	.254	12.09170
1.200	20332	UU	.152	7.73595
1.505	28266	UU	.200	10.75468
1.705	17959	UU	.148	6.83307
1.881	21357	UU	.187	8.12594
2.377	26719	UU	.299	10.16608
2.740	35817	UU	.305	13.62770
3.327	78218	UU	.295	29.76048

* RUN # 7 JAN 1, 1901 02:16:51
START

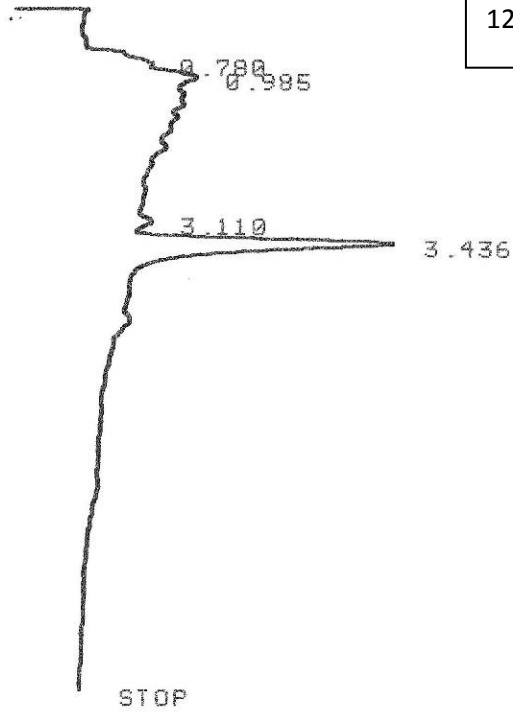


RUN# 7 JAN 1, 1901 02:16:51

AREA%	RT	AREA	TYPE	WIDTH	AREA%
	.621	1238	PU	.065	1.72495
*	1.013	30666	UU	.344	42.72816
	3.090	1723	BU	.135	2.40072
°	3.431	38143	UU	.212	53.14614

* RUN # 11 JAN 1, 1981 03:02:05
START

12C

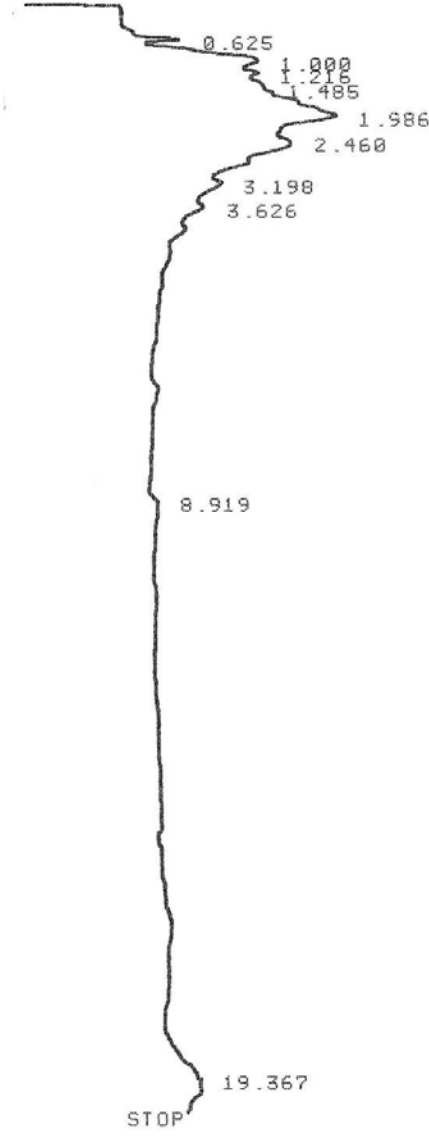


RUN# 11 JAN 1, 1981 03:02:05

RT	AREA	TYPE	WIDTH	AREA%
.780	9039	PU	.181	11.01887
.985	22620	UU	.265	27.57460
3.110	2115	BU	.147	2.57826
3.436	48258	UU	.214	58.82827

* RUN # 5 JAN 1, 1901 01:37:36
 START

13A

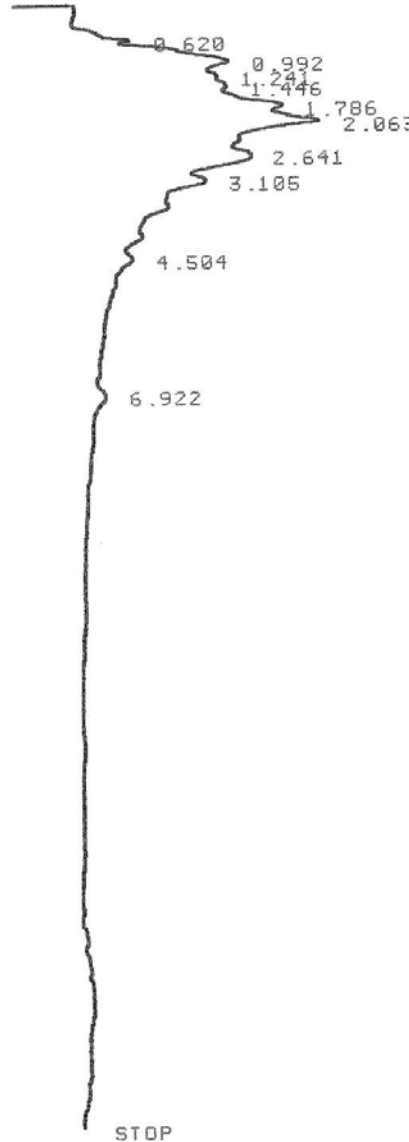


RUN# 5 JAN 1, 1901 01:37:36

AREA#	RT	AREA	TYPE	WIDTH	AREA%
	.625	6339	PU	.109	2.01454
*	1.000	43277	UU	.315	13.75344
	1.216	21907	UU	.161	6.96205
	1.485	28941	UU	.204	9.19746
	1.986	127216	UU	.600	40.42930
	2.460	21265	UU	.132	6.75802
	3.198	28489	UU	.349	9.05381
	3.626	18188	UU	.306	5.78015
	8.919	873	PU	.122	.27744
	19.367	18168	UU	.472	5.77380

TOTAL AREA= 314663

*. RUN # 13 JAN 1, 1901 03:33:19
 START



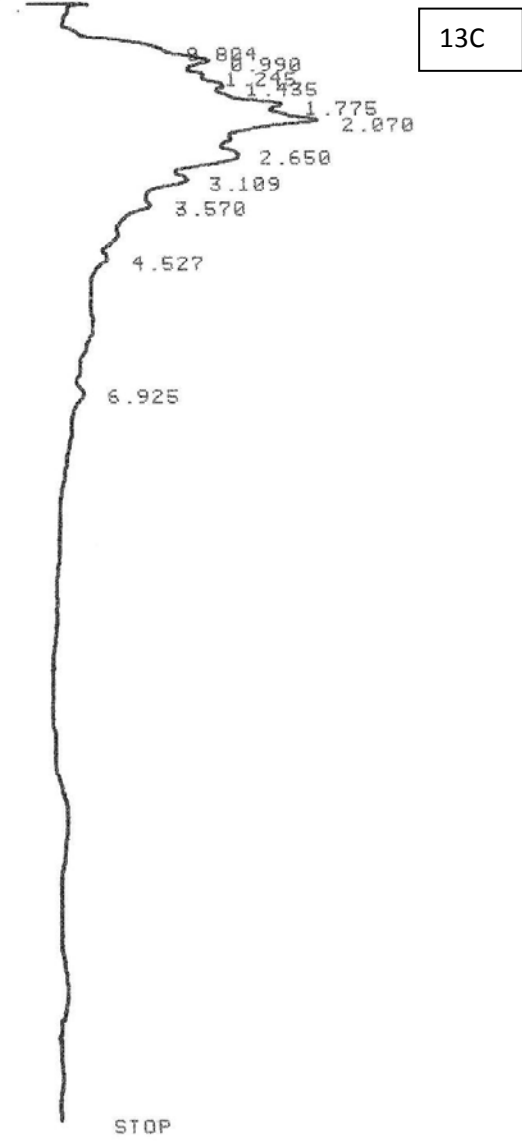
13B

STOP

RUN# 13 JAN 1, 1901 03:33:19

AREA#	RT	AREA	TYPE	WIDTH	AREA%
	.620	5020	PV	.129	3.28293
*	.992	39948	UU	.352	26.12483
.	1.241	9964	UU	.121	6.51617
.	1.446	17949	UU	.231	11.73812
.	1.786	32153	UU	.294	21.02713
.	2.063	36252	UU	.286	23.70776
.	2.641	372	PB	.067	.24328
.	3.105	4754	BP	.195	3.10890
.	4.504	2872	PP	.219	1.87820
.	6.922	3628	BU	.289	2.37261

*. RUN # 14 JAN 1, 1901 03:54:41
 START

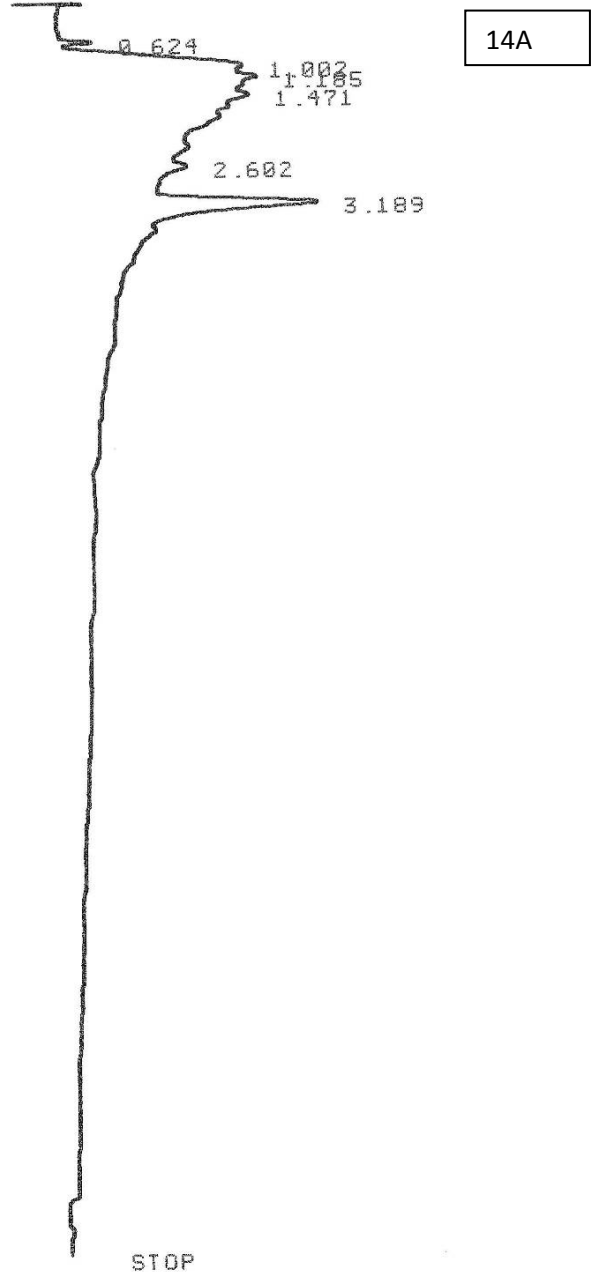


RUN# 14 JAN 1, 1901 03:54:41

AREA#	RT	AREA	TYPE	WIDTH	AREA#
	.804	23068	BU	.222	5.88103
	.990	40849	UU	.273	10.41418
	1.245	19355	UU	.136	4.93443
	1.435	35428	UU	.218	9.03213
	1.775	68951	UU	.311	17.57860
	2.070	94751	UU	.368	24.15614
	2.650	27912	UU	.164	7.11598
	3.109	43050	UU	.376	10.97531
	3.570	28336	UU	.383	7.22408
	4.527	7393	UU	.307	1.88480
	6.925	3151	UP	.281	.80333

TOTAL AREA= 392244

* RUN # 9 JAN 1, 1901 02:43:40
 START

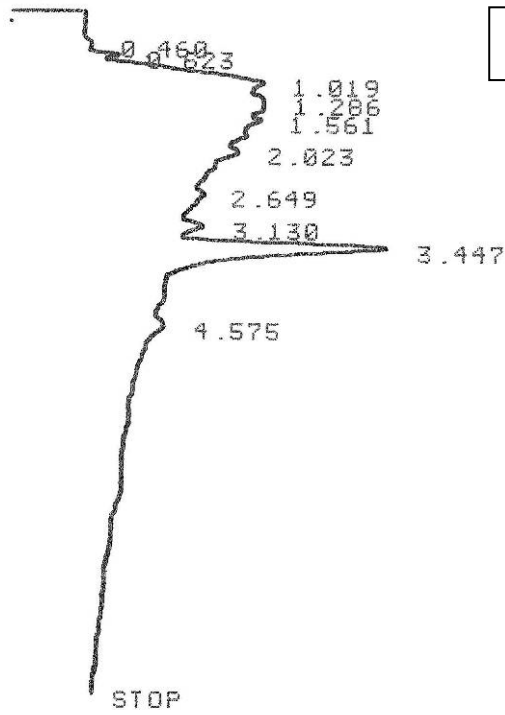


RUN# 9 JAN 1, 1901 02:43:40

RT	AREA	TYPE	WIDTH	AREA%
.624	1727	BU	.058	.85737
1.002	40595	UU	.245	20.15350
1.185	31023	UU	.175	15.40146
1.471	34143	UU	.205	16.95038
2.602	28474	UU	.295	14.13600
3.189	65467	UU	.307	32.50128

* RUN # 8 JAN 1, 1901 02:28:09
 START

14B

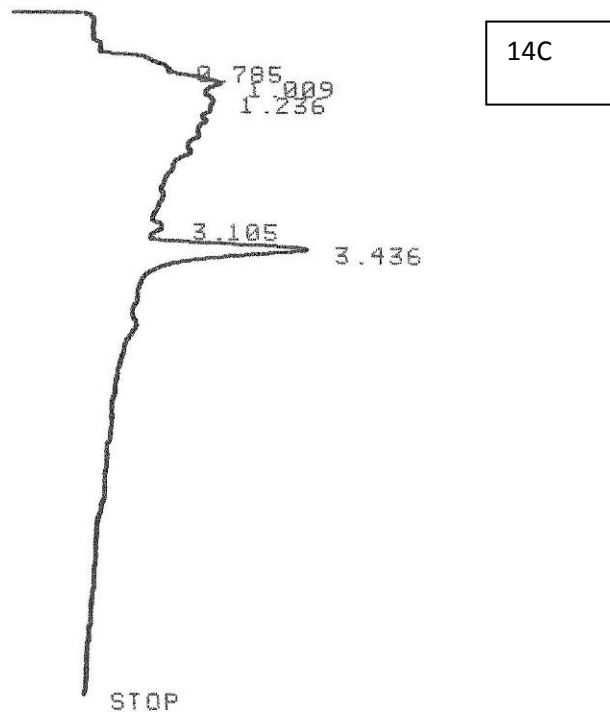


RUN# 8 JAN 1, 1901 02:28:09

AREA#

RT	AREA	TYPE	WIDTH	AREA%
.460	1007	BU	.127	.36712
.623	2374	UU	.084	.86549
*1.019	49370	UU	.331	17.99887
*1.286	27448	UU	.186	10.00674
1.561	29265	UU	.206	10.66917
2.023	29533	UU	.248	10.76688
. 2.649	32330	UU	.389	11.78658
3.130	19696	UU	.250	7.18059
*3.447	70962	UU	.305	25.87068
4.575	12310	UU	.377	4.48787

* RUN # 12 JAN 1, 1901 03:13:23
 START



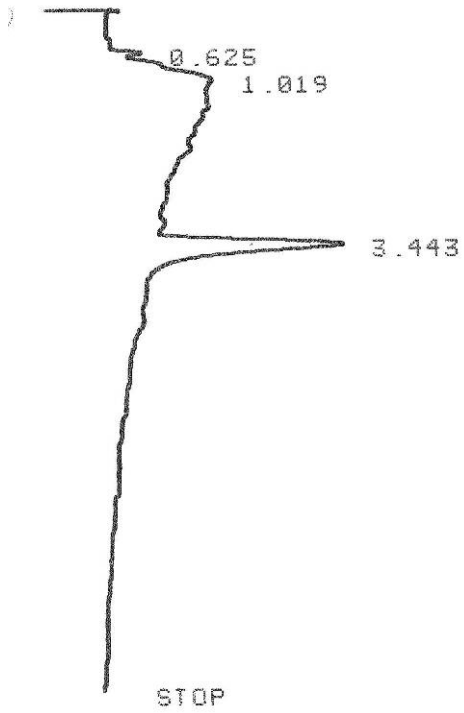
RUN# 12 JAN 1, 1901 03:13:23

AREA#	RT	AREA	TYPE	WIDTH	AREA%
	.785	9865	PU	.193	12.55728
*	1.009	22673	UU	.260	28.86075
*	1.236	13919	UU	.192	17.71766
.	3.105	1881	PU	.148	2.39435
*	3.436	30222	UU	.216	38.46995

TOTAL AREA= 78560
 MUL FACTOR=1.0000E+00

* RUN # 9 JAN 1, 1901 02:39:28
START

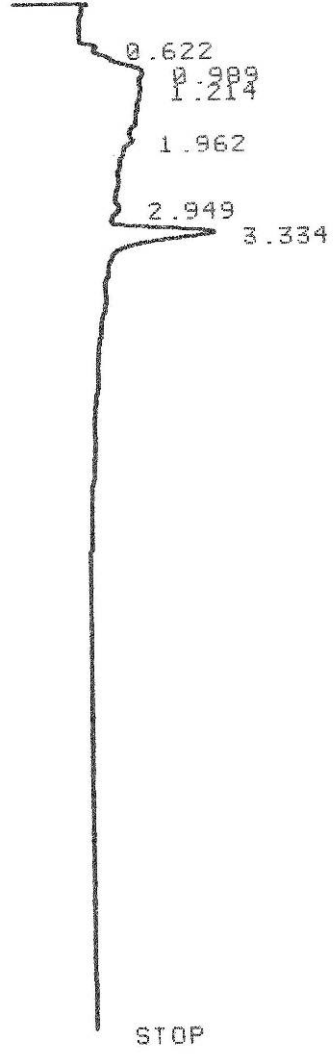
15b



RUN# 9 JAN 1, 1901 02:39:28

AREA%	RT	AREA	TYPE	WIDTH	AREA%
	.625	2176	UU	.073	2.95905
	1.019	30559	UU	.352	41.55595
	3.443	40802	UU	.231	55.48499

* RUN # 21 JAN 1, 1901 07:19:39
 START

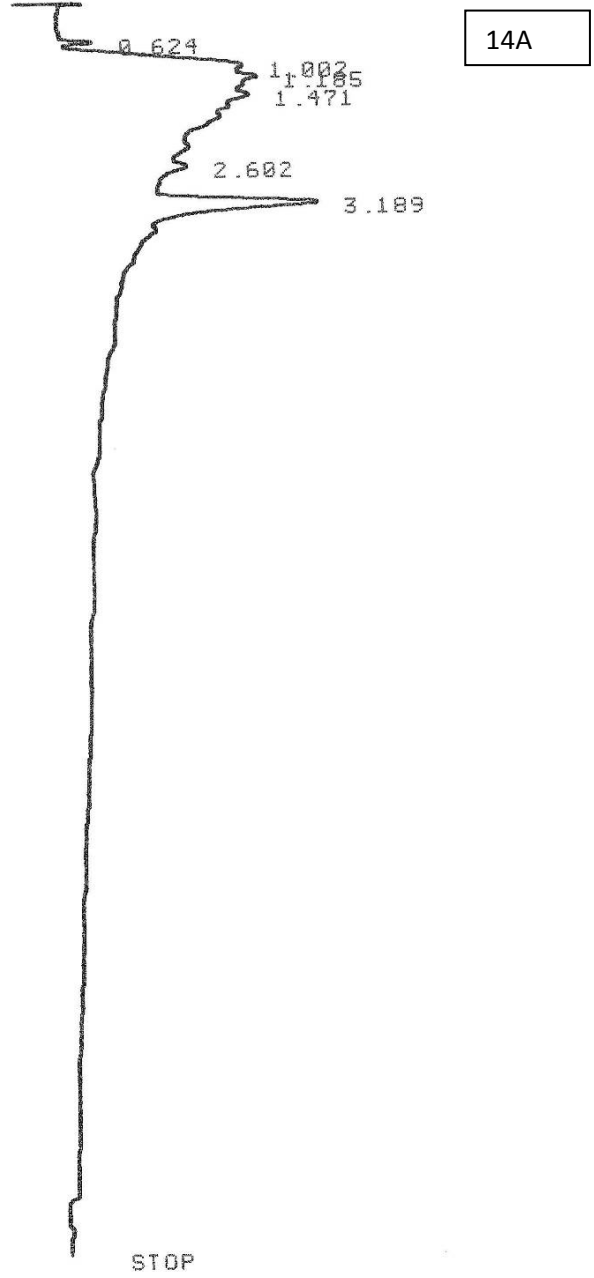


150

RUN# 21 JAN 1, 1901 07:19:39

RT	AREA	TYPE	WIDTH	AREA%
.622	2158	PU	.071	1.68904
.989	33760	UU	.325	26.42350
1.214	18343	UU	.187	14.35683
1.962	23921	UU	.343	18.72266
2.949	9016	UU	.254	7.05670
3.334	40567	UU	.214	31.75126

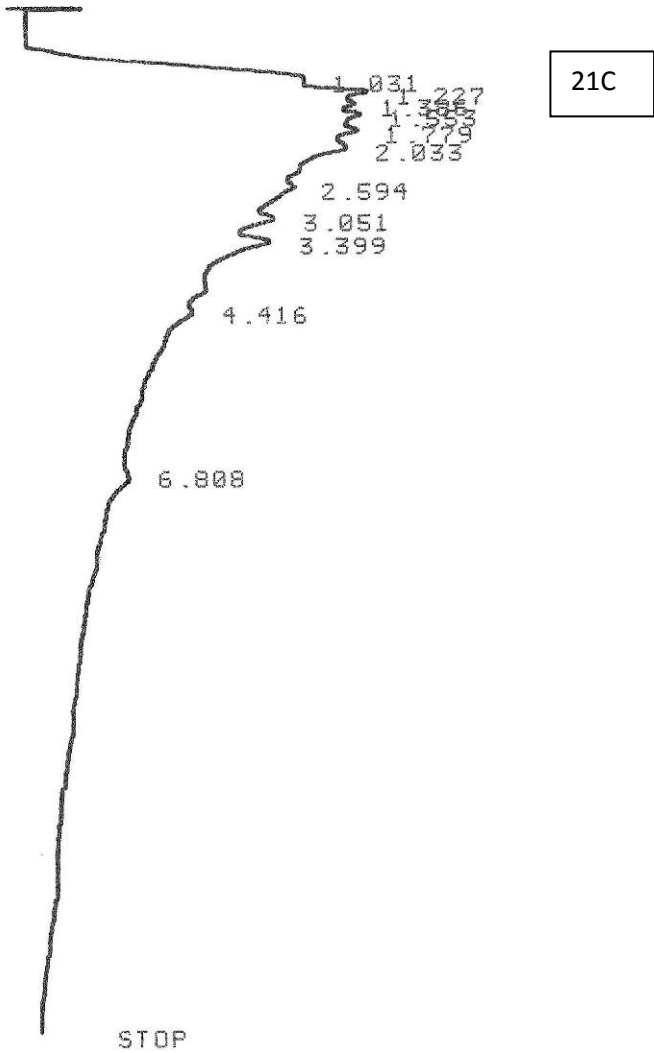
* RUN # 9 JAN 1, 1901 02:43:40
 START



RUN# 9 JAN 1, 1901 02:43:40

RT	AREA	TYPE	WIDTH	AREA%
.624	1727	BU	.058	.85737
1.002	40595	UU	.245	20.15350
1.185	31023	UU	.175	15.40146
1.471	34143	UU	.205	16.95038
2.602	28474	UU	.295	14.13600
3.189	65467	UU	.307	32.50128

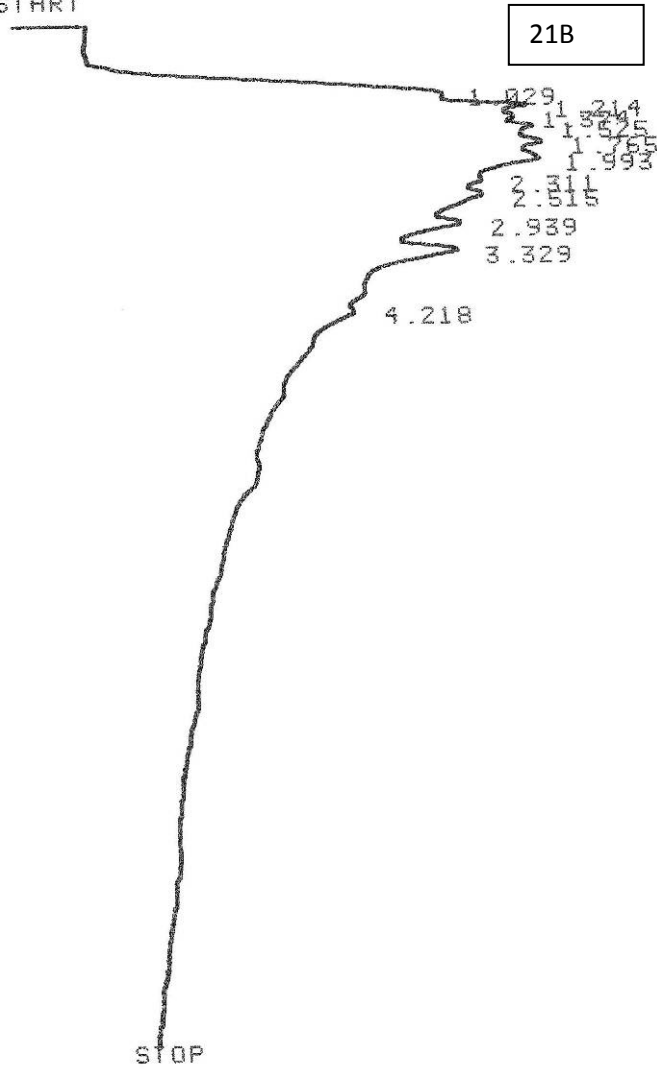
* RUN # 23 JAN 1, 1901 07:52:15
 START



RUN# 23 JAN 1, 1901 07:52:15

AREA#	RT	AREA	TYPE	WIDTH	AREA#
*	1.031	126954	BU	.279	10.95871
*	1.227	115740	UU	.209	9.99071
*	1.386	80560	UU	.154	6.95396
*	1.553	102196	UU	.194	8.82159
*	1.779	106877	UU	.210	9.22566
*	2.033	184464	UU	.382	15.92299
	2.594	145713	UU	.396	12.57799
	3.051	91757	UU	.291	7.92049
*	3.399	142735	UU	.488	12.32093
	4.416	56027	UB	.482	4.83627
	6.808	5453	UB	.268	.47070

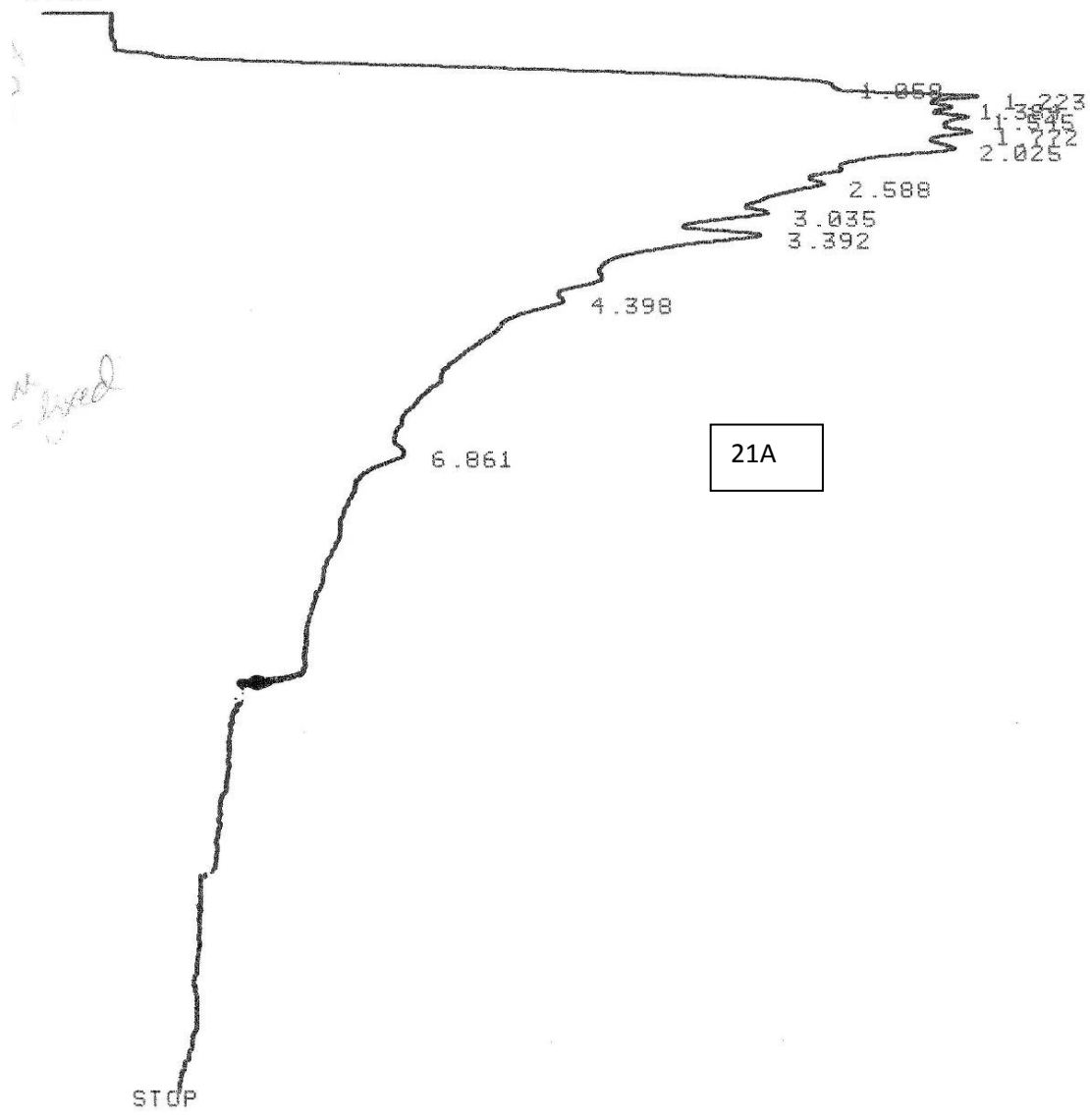
* RUN # 22 JAN 1, 1901 07:35:57
 START



RUN# 22 JAN 1, 1901 07:35:57

AREA#	RT	AREA	TYPE	WIDTH	AREA#
1	1.029	134305	PU	.253	11.55763
2	1.214	127522	UU	.196	10.97392
3	1.374	93751	UU	.154	8.06775
4	1.525	113761	UU	.184	9.78972
5	1.765	126774	UU	.212	10.90955
6	1.993	193309	UU	.345	16.63523
7	2.311	73363	UU	.175	6.31326
8	2.515	136293	UU	.348	11.72871
9	2.939	86699	UU	.295	7.46089
10	3.329	70255	UU	.304	6.04580
11	4.218	6014	UP	.203	.51754

* RUN # 15 JAN 1, 1901 04:19:24
 START

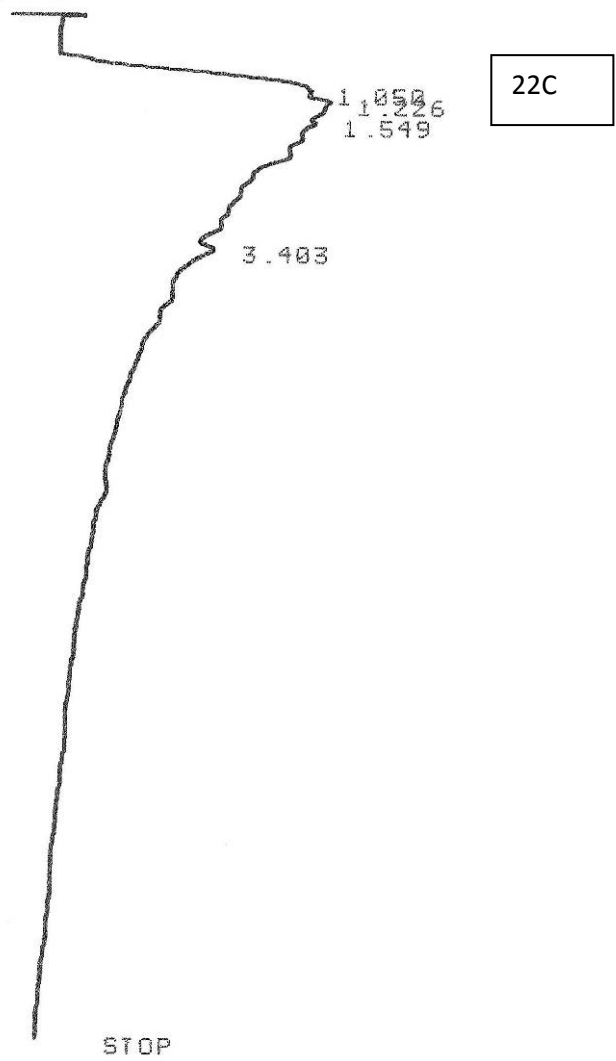


RUN# 15 JAN 1, 1901 04:19:24

AREA#

RT	AREA	TYPE	WIDTH	AREA#
1.058	160232	PU	.261	10.23710
1.223	161934	UU	.221	10.34583
1.384	110563	UU	.158	7.06378
1.545	133537	UU	.190	8.53157
1.772	151660	UU	.220	9.68943
2.025	240229	UU	.365	15.34804
2.588	211513	UU	.421	13.51340
3.035	120441	UU	.285	7.69488
3.392	199276	UU	.507	12.73158
4.398	70410	UB	.475	4.49844
6.861	5415	BP	.256	.34596

* RUN # 25 JAN 1, 1901 08:24:52
START

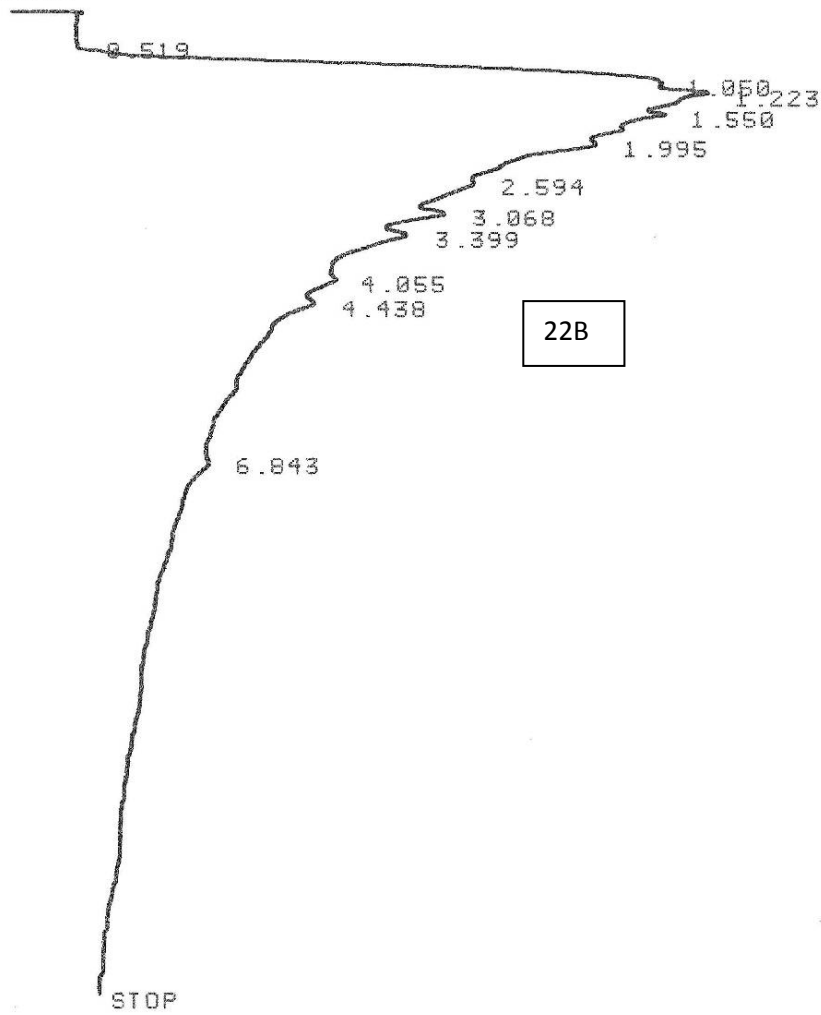


RUN# 25 JAN 1, 1901 08:24:52

AREA%

RT	AREA	TYPE	WIDTH	AREA%
1.050	144923	PU	.348	33.11466
1.226	88882	UU	.198	20.30938
1.549	92257	UU	.221	21.08057
3.403	111578	UU	.540	25.49539

* RUN # 24 JAN 1, 1901 08:08:34
 START



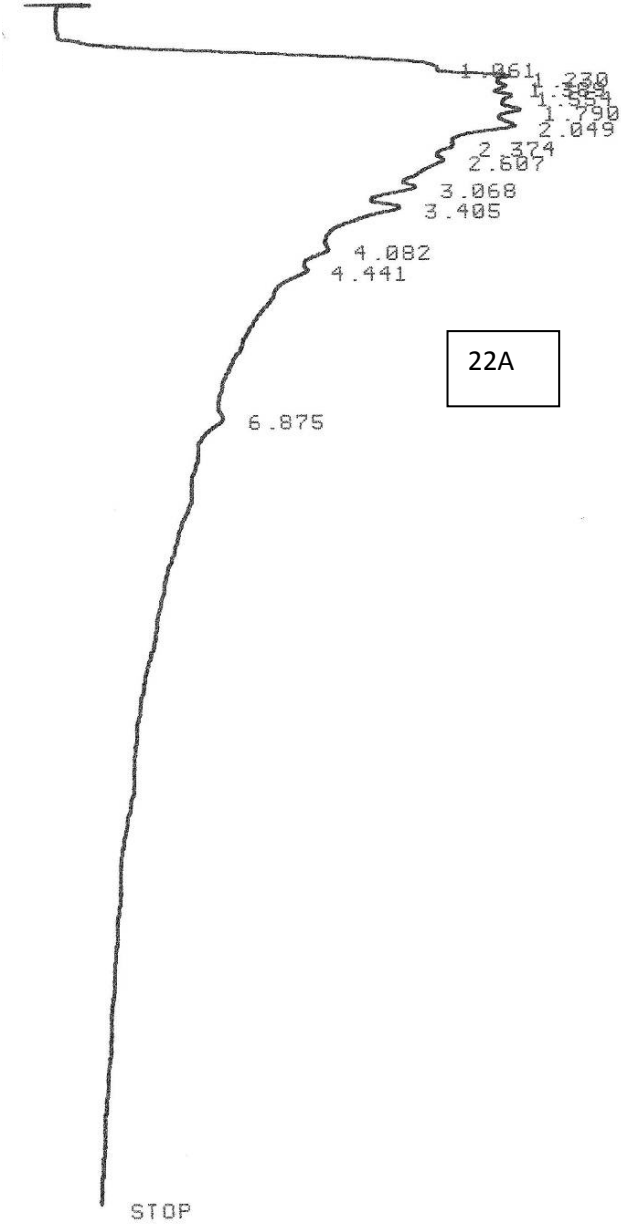
RUN# 24 JAN 1, 1901 08:08:34

AREAX

RT	AREA	TYPE	WIDTH	AREAX
.519	141	PP	.094	.00691
* 1.050	330862	PU	.330	16.20513
* 1.223	364063	UU	.338	17.83127
* 1.550	223593	UU	.229	10.95126
1.995	423725	UU	.514	20.75342
2.594	219352	UU	.386	10.74354
3.068	139986	UU	.288	6.85631
* 3.399	191656	UU	.481	9.38703
4.055	73766	UU	.324	3.61295
4.438	71697	UB	.432	3.51161
6.843	2870	BB	.159	.14057

TOTAL AREA=2041711
 MUL FACTOR=1.0000E+00

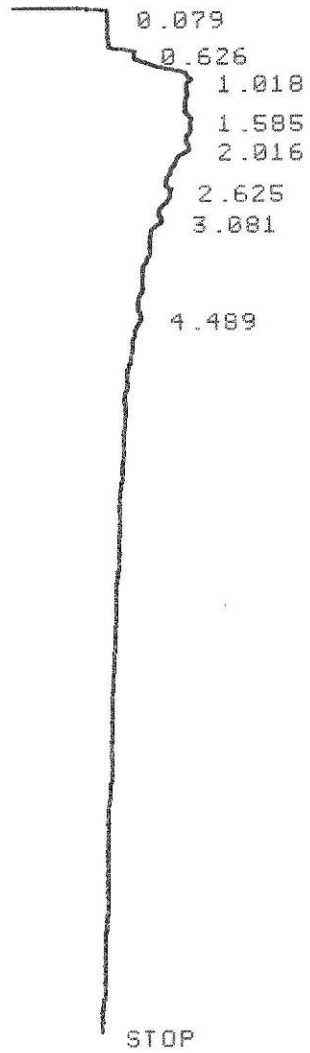
*. ATT 2^ 2 @
 *. RUN # 16 JAN 1, 1901 04:40:43
 START



RUN# 16 JAN 1, 1901 04:40:43

AREA#	RT	AREA	TYPE	WIDTH	AREA#
o	1.061	178981	PV	.258	8.74037
o	1.230	160996	UU	.196	7.86209
	1.389	153309	UU	.191	7.48671
o	1.554	148213	UU	.184	7.23785
	1.790	177930	UU	.222	8.68905
	2.049	290497	UU	.375	14.18616
	2.374	132178	UU	.209	6.45479
	2.607	244888	UU	.411	11.95888
	3.068	145879	UU	.288	7.12387
o	3.405	241156	UU	.533	11.77664
	4.082	88515	UU	.333	4.32255
	4.441	78958	UU	.401	3.85584
	6.875	6250	BP	.256	.30521

* RUN # 27 JAN 1, 1901 00:57:29
 START



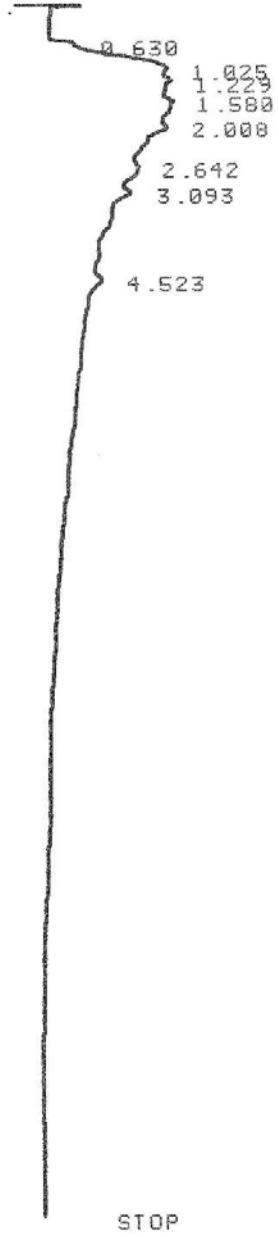
23C

RUN# 27 JAN 1, 1901 00:57:29

AREAX	RT	AREA	TYPE	WIDTH	AREAX
	.079	4805	BU	.163	2.98648
	.626	3169	PU	.072	1.96964
	1.018	41177	UU	.303	25.59294
	1.585	19134	UU	.145	11.89245
	2.016	51933	UU	.408	32.27816
	2.625	13046	UU	.149	8.10855
	3.081	20724	UU	.290	12.88069
	4.489	6904	UU	.291	4.29108

* RUN # 17 JAN 1, 1901 05:02:02
 START

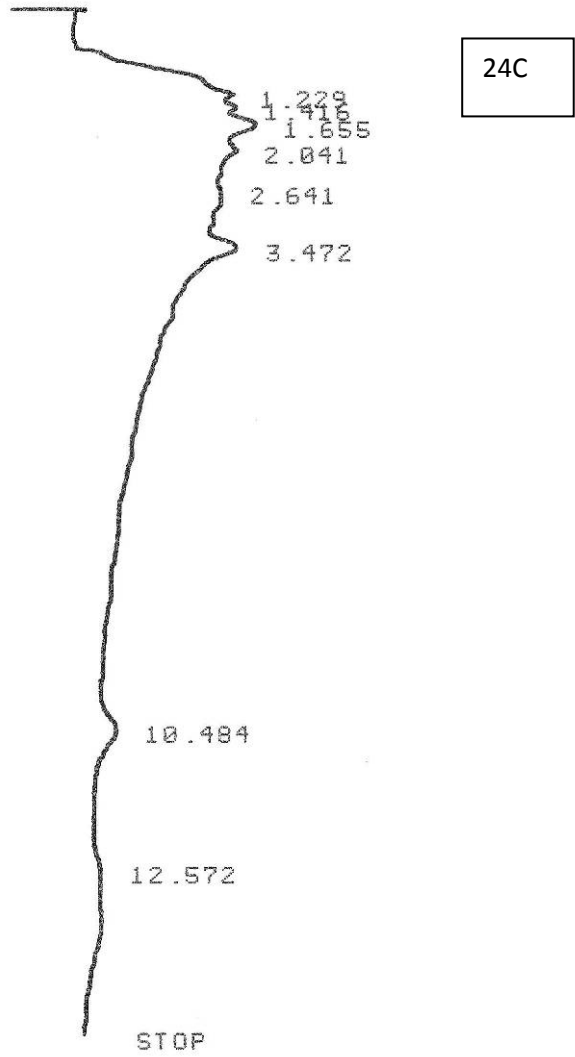
23A



RUN# 17 JAN 1, 1901 05:02:02

AREA#	RT	AREA	TYPE	WIDTH	AREA#
	.630	3148	PU	.064	.79489
*	1.025	73907	UU	.334	18.66192
	1.229	46642	UU	.205	11.77736
*	1.580	34958	UU	.154	8.82708
	2.000	109850	UU	.514	27.73773
	2.642	60477	UU	.400	15.27078
*	3.093	41728	UU	.311	10.53655
	4.523	25321	UU	.380	6.39369

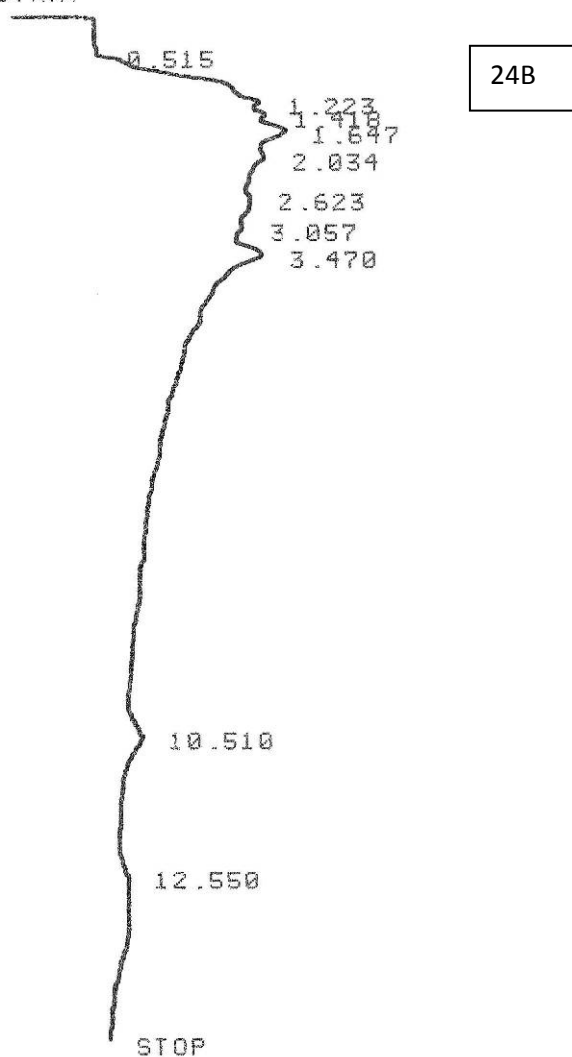
* RUN # 29 JAN 1, 1901 09:30:10
 START



RUN# 29 JAN 1, 1901 09:30:10

AREA%	RT	AREA	TYPE	WIDTH	AREA%
	1.229	118571	PU	.470	19.99160
	1.416	44302	UU	.174	7.46952
	1.655	103643	UU	.366	17.47468
	2.041	98046	UU	.410	16.53100
	2.641	50345	UU	.252	8.48840
	3.472	157719	UU	.759	26.59214
	10.484	17872	BU	.527	3.01330
	12.572	2606	BU	.359	.43938

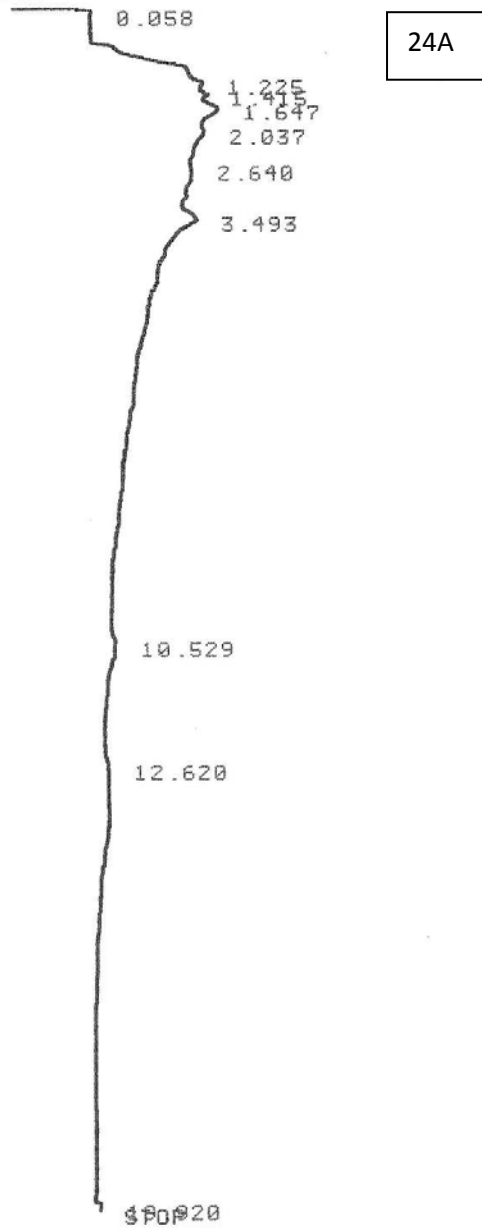
* RUN # 28 JAN 1, 1901 09:13:48
 START



RUN# 28 JAN 1, 1901 09:13:48

AREA#	RT	AREA	TYPE	WIDTH	AREA%
	.515	402	PU	.140	.05491
*	1.223	122363	UU	.467	16.71498
*	1.418	45085	UU	.169	6.15868
*	1.647	106456	UU	.361	14.54206
	2.034	138511	UU	.552	18.92082
	2.623	96383	UU	.457	13.16607
	3.057	45274	UU	.241	6.18450
*	3.470	159111	UU	.753	21.73481
	10.510	16834	UP	.547	2.29955
	12.550	1637	PU	.284	.22362

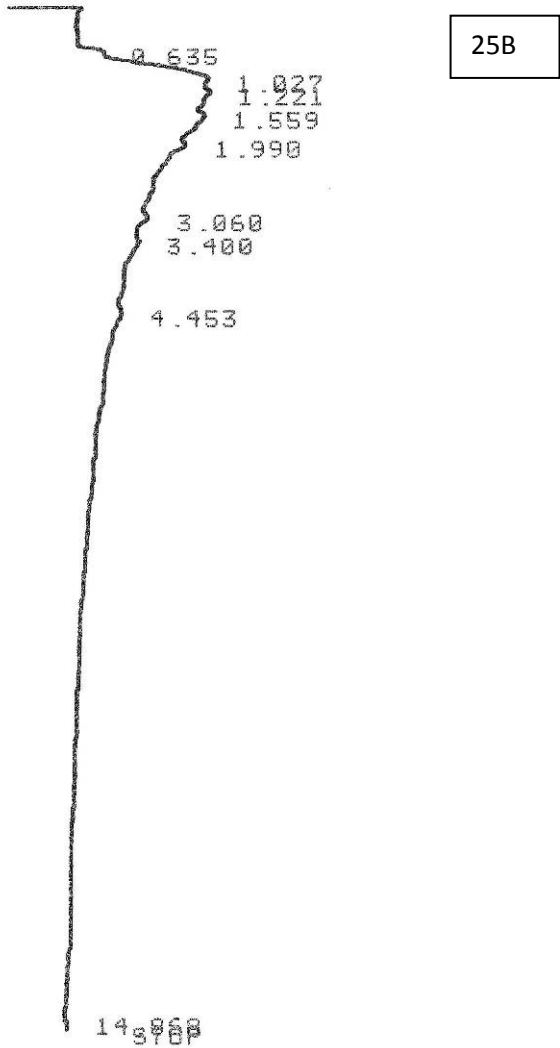
* RUN # 18 JAN 1, 1901 05:23:20
 START



RUN# 18 JAN 1, 1901 05:23:20

AREA%	RT	AREA	TYPE	WIDTH	AREA%
	.058	1471	BB	.080	.38039
*	1.225	95671	PU	.493	24.74011
*	1.415	32049	UU	.165	8.20774
*	1.647	79147	UU	.377	20.46700
	2.037	92304	UU	.541	23.86942
	2.640	19115	UU	.150	4.94306
*	3.493	60106	UU	.555	15.56304
	10.529	5217	UU	.303	1.34909
	12.620	1282	PU	.227	.33152
	19.920	262	PB	.067	.06775

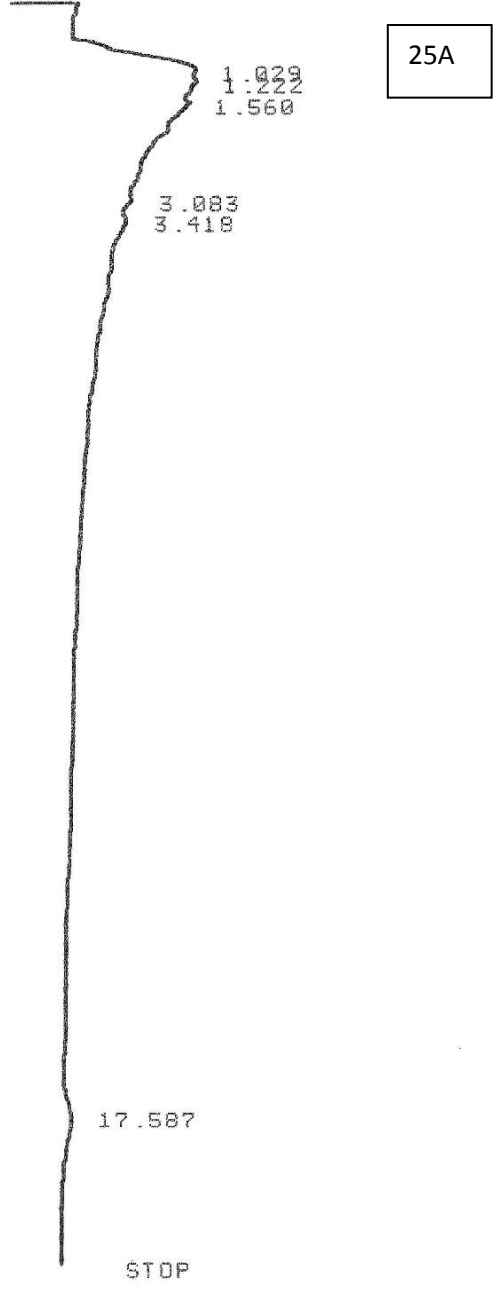
* RUN # 30 JAN 1, 1981 09:46:28
 START



RUN# 30 JAN 1, 1981 09:46:28

AREA#	RT	AREA	TYPE	WIDTH	AREA%
	.635	2671	PU	.065	.91866
*	1.027	70194	UU	.329	24.14230
	1.221	39846	UU	.187	13.70451
*	1.559	72487	UU	.371	24.93095
	1.990	70288	UU	.460	24.17464
*	3.060	16984	UU	.256	5.84142
	3.400	14954	UU	.358	5.14323
	4.453	3112	BU	.225	1.07033
	14.868	215	PU	.030	.07395

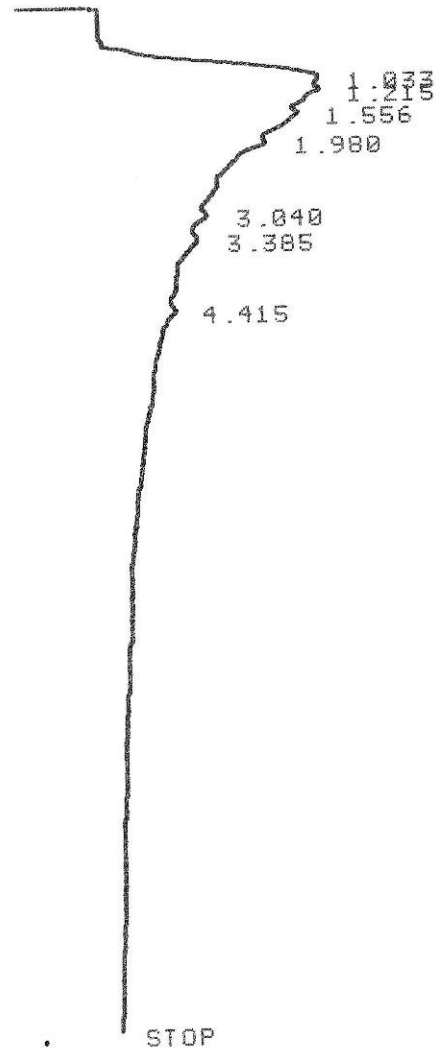
* RUN # 19 JAN 1, 1901 05:44:39
 START



RUN# 19 JAN 1, 1901 05:44:39

AREA#	RT	AREA	TYPE	WIDTH	AREAX
*	1.029	79222	PU	.359	33.75357
	1.222	39757	UU	.182	16.93899
*	1.560	75363	UU	.386	32.10939
*	3.083	15833	UU	.260	6.74586
	3.418	15330	UU	.354	6.53155
	17.587	9202	BU	.579	3.92063

* RUN # 31 JAN 1, 1901 10:02:47
 START



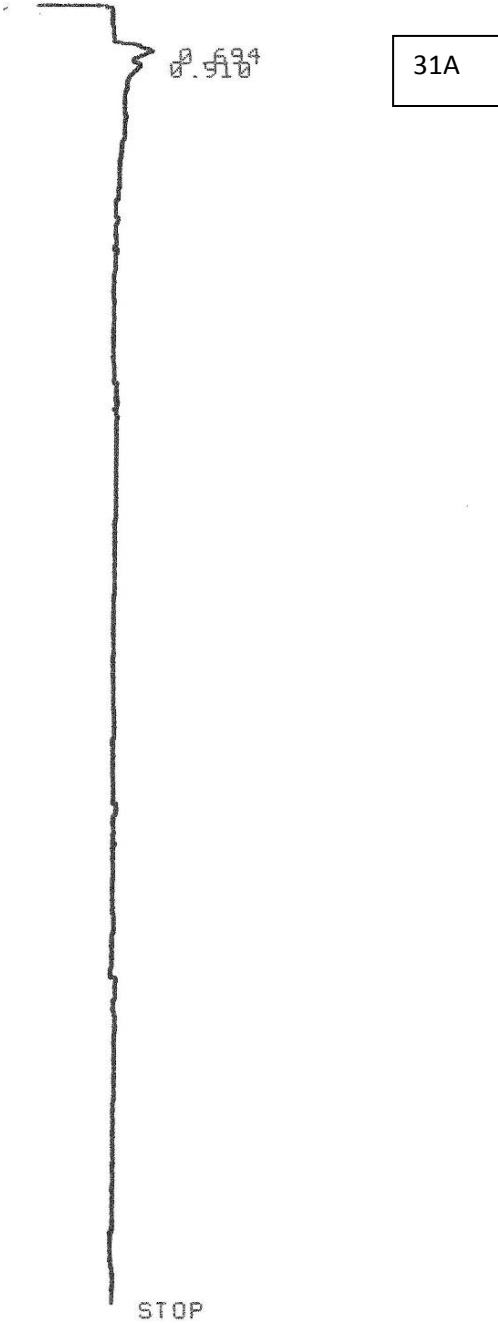
25C

RUN# 31 JAN 1, 1901 10:02:47

AREA#	RT	AREA	TYPE	WIDTH	AREA%
1	*1.033	129238	PU	.354	20.83724
2	1.215	115966	UU	.318	18.69738
3	*1.556	125930	UU	.392	20.30389
4	1.980	120459	UU	.468	19.42179
5	*3.040	39071	UU	.282	6.29948
6	3.385	61290	UU	.529	9.88188
7	4.415	28272	UB	.497	4.55834

TOTAL AREA= 620276

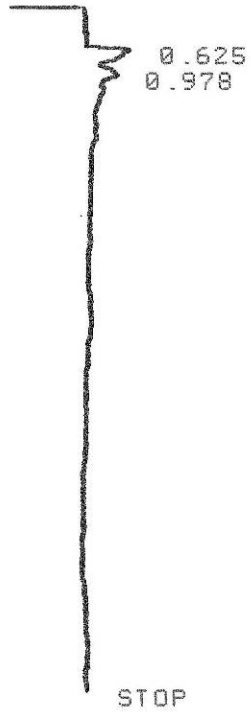
* RUN # 6 JAN 1, 1901 01:13:08
START



AREA%	RT	AREA	TYPE	WIDTH	AREA%
	.694	5668	BU	.183	64.10314
	.910	3174	UU	.165	35.89685

* RUN # 19 JAN 1, 1901 06:51:17
START

31B

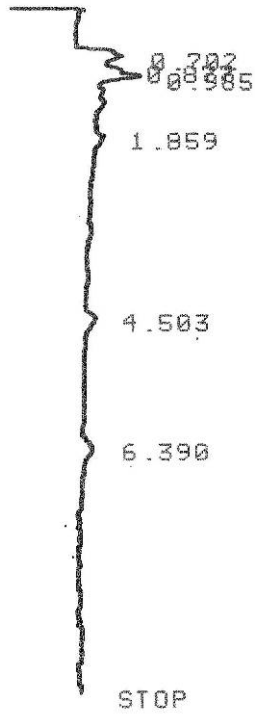


RUN# 19 JAN 1, 1901 06:51:17

AREA%	RT	AREA	TYPE	WIDTH	AREA%
.625		2696	PU	.070	33.07570
.978		6455	UU	.206	66.92429

* RUN # 20 JAN 1, 1901 07:02:35
 START

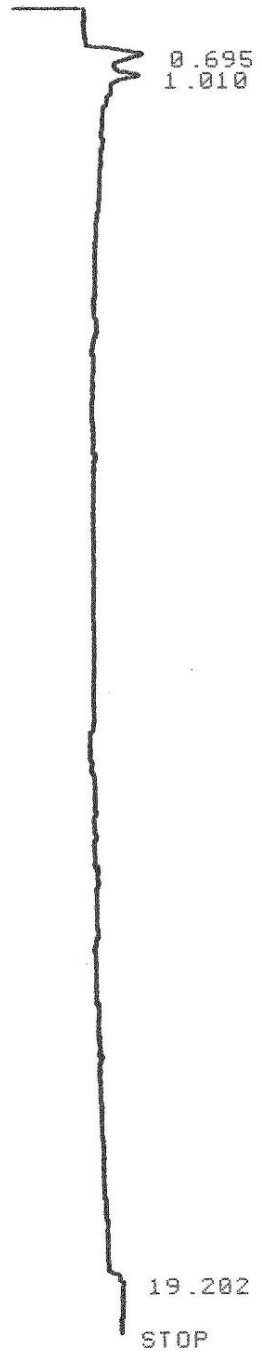
31C



RUN# 20 JAN 1, 1901 07:02:35

AREA#	RT	AREA	TYPE	WIDTH	AREA%
	.702	6061	PU	.168	33.59384
	.877	2403	UU	.080	13.31892
	.985	5820	UU	.129	32.25806
	1.859	538	UU	.065	2.98193
	4.503	2235	UP	.223	12.38776
	6.390	985	BU	.126	5.45948

* RUN # 7 JAN 1, 1901 01:34:26
START



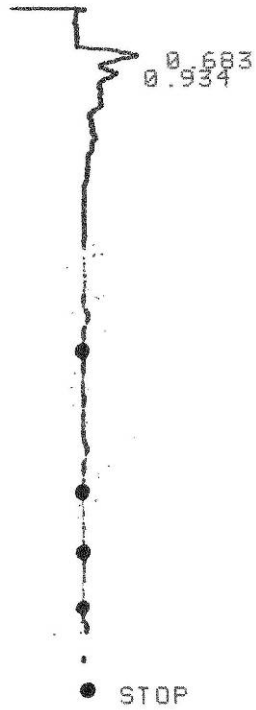
32A

RUN# 7 JAN 1, 1901 01:34:26

AREA#	RT	AREA	TYPE	WIDTH	AREA%
	.695	8503	PU	.175	48.61357
	1.010	7526	UU	.192	43.02784
	19.202	1462	PU	.200	8.35859

* RUN # 21 JAN 1, 1901 07:13:54
START

32B



RUN# 21 JAN 1, 1901 07:13:54

AREA%	RT	AREA	TYPE	WIDTH	AREA%
.683		8949	PU	.169	57.55725
.934		6599	UU	.202	42.44277

* RUN # 22 JAN 1, 1901 07:25:12
START



0.627
0.956

32C

RUN# 22 JAN 1, 1901 07:25:12

AREA%	RT	AREA	TYPE	WIDTH	AREA%
	.627	1863	BU	.063	32.92101
	.956	3796	UU	.183	67.07901

* RUN # 9 JAN 1, 1901 02:01:03
START

0.031
0.690
0.995

33A



STOP

RUN# 9 JAN 1, 1901 02:01:03

AREA#	RT	AREA	TYPE	WIDTH	AREA%
	.031	462	BU	.031	2.47854
	.690	9475	PU	.195	50.83154
	.995	8703	UU	.194	46.68990

* RUN # 23
START

JAN 1, 1901 07:36:31

33B

0.691
0.906

STOP

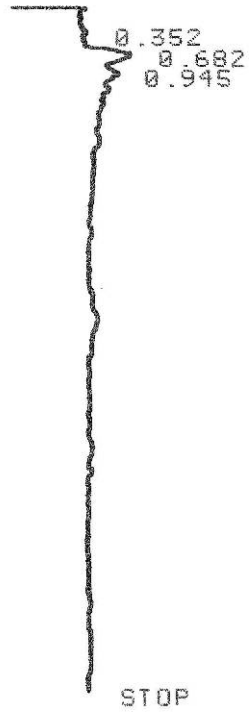
RUN# 23

JAN 1, 1901 07:36:31

AREA#

RT	AREA	TYPE	WIDTH	AREA#
.691	6625	PU	.167	56.02536
.906	5200	UU	.216	43.97462

* RUN # 24 JAN 1, 1901 07:47:49
START



33C

RUN# 24 JAN 1, 1901 07:47:49

AREA%	RT	AREA	TYPE	WIDTH	AREA%
	.352	454	PU	.097	3.31580
	.682	7246	PU	.104	52.92142
	.945	5992	UU	.219	43.76278

* RUN # 10 JAN 1, 1901 02:22:21
START

0.127
0.692
0.910
1.000

34A

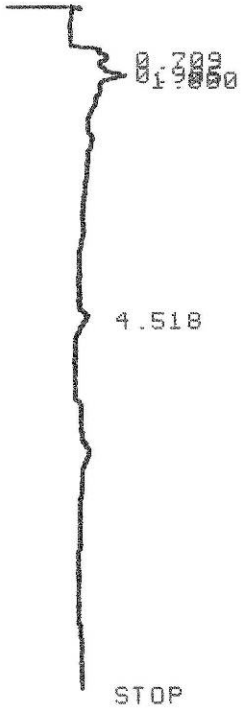
STOP

RUN# 10 JAN 1, 1901 02:22:21

AREA#	RT	AREA	TYPE	WIDTH	AREA%
	.127	1274	BU	.173	6.12412
	.692	8518	PU	.193	40.94602
	.910	3458	UU	.094	16.62261
	1.000	7553	UU	.144	36.30726

* RUN # 25 JAN 1, 1901 07:59:08
START

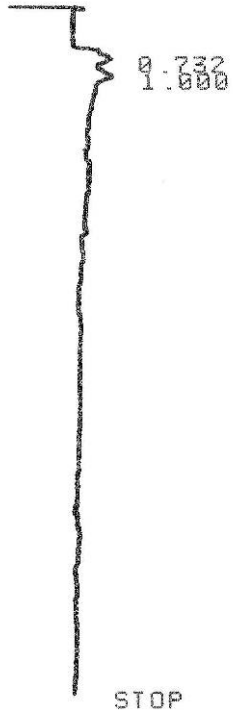
34B



RUN# 25 JAN 1, 1901 07:59:08

AREA#	RT	AREA	TYPE	WIDTH	AREA#
	.709	5765	PU	.194	39.18037
	.905	2453	UU	.094	16.67119
	1.000	5764	UU	.149	39.17357
	4.518	732	PU	.094	4.97485

* RUN # 26 JAN 1, 1901 08:10:26
START



34C

RUN# 26 JAN 1, 1901 08:10:26

AREA#	RT	AREA	TYPE	WIDTH	AREA%
	.732	6236	PU	.206	50.92691
	1.000	6009	UU	.204	49.07309

* RUN # 11 JAN 1, 1901 02:43:40
START

35A

0.686
1.002

2.278

STOP

RUN# 11 JAN 1, 1901 02:43:40

AREA%	RT	AREA	TYPE	WIDTH	AREA%
	.686	8991	PU	.170	34.12274
	1.002	16058	UU	.191	60.94349
	2.278	1300	UU	.179	4.93377

* RUN # 27 JAN 1, 1901 08:21:48
START

35B

0.696
1.000



RUN# 27 JAN 1, 1901 08:21:48

AREA#	RT	AREA	TYPE	WIDTH	AREA%
	.696	6146	PV	.201	42.48290
	1.000	8321	UV	.156	57.51712

* RUN # 28 JAN 1, 1901 08:33:07
START

35C

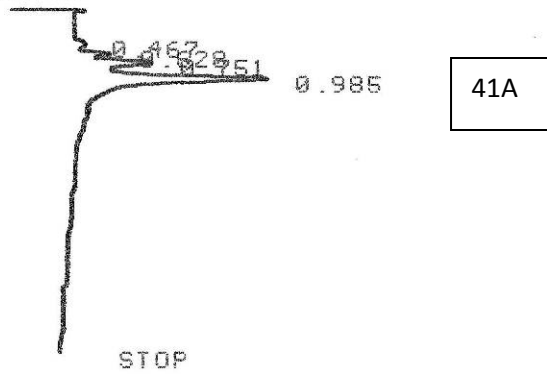
0.686
0.994

STOP

RUN# 28 JAN 1, 1901 08:33:07

AREA%	RT	AREA	TYPE	WIDTH	AREA%
	.686	8055	UU	.169	50.75702
	.994	5654	UU	.199	41.24298

* RUN # 13 JAN 1, 1901 07:59:23
START



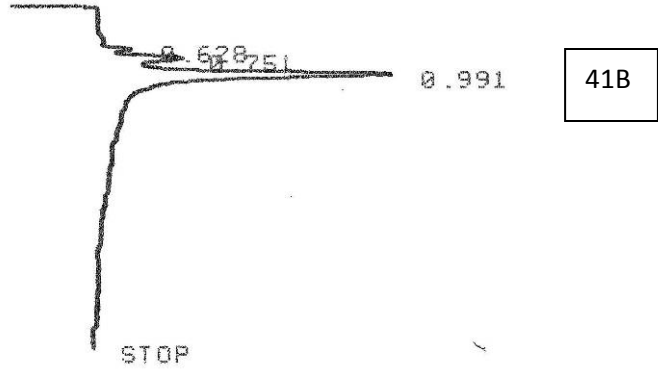
RUN# 13 JAN 1, 1901 07:59:23

AREAX

RT	AREA	TYPE	WIDTH	AREAX
.467	1297	UU	.120	3.24998
.628	2364	UU	.072	5.92363
*.751	8071	UU	.119	20.22402
*.985	28176	UU	.169	70.60240

TOTAL AREA= 39908
MUL FACTOR=1.0000E+00

* RUN # 14 JAN 1, 1901 08:05:42
START



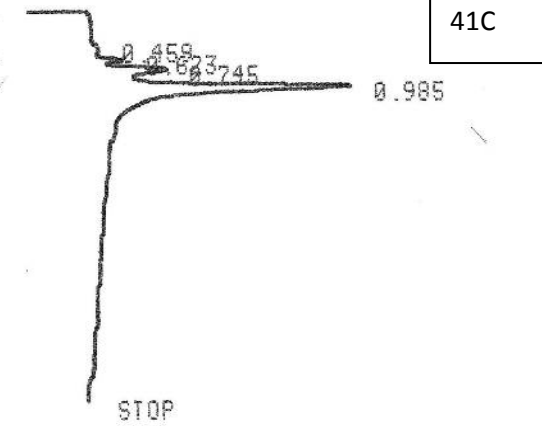
RUN# 14 JAN 1, 1901 08:05:42

AREA%

RT	AREA	TYPE	WIDTH	AREA%
.628	1519	FU	.061	3.35705
* .751	7087	UU	.100	15.66250
* .991	36642	UU	.151	80.98038

TOTAL AREA= 45248
MUL FACTOR=1.0000E+00

* RUN # 15 JAN 1, 1901 08:12:00
START



RUN# 15 JAN 1, 1901 08:12:00

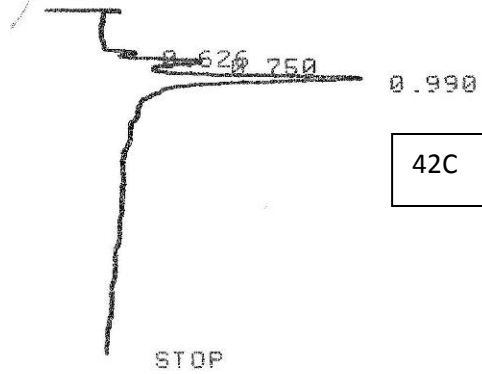
AREA%

RT	AREA	TYPE	WIDTH	AREA%
.459	862	PU	.117	1.55120
.623	1972	UU	.064	3.54868
* .745	8077	UU	.108	14.53482
* .985	44659	UU	.167	80.36528

TOTAL AREA= 55570
MUL FACTOR=1.0000E+00

* RUN # 18 JAN 1, 1901 00:30:54

START



RUN# 18 JAN 1, 1901 00:30:54

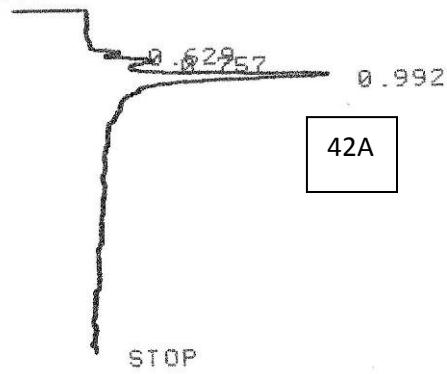
AREA%

RT	AREA	TYPE	WIDTH	AREA%
.626	1791	UU	.064	3.58609
.750	9161	UU	.111	18.34291
.990	38991	UU	.180	78.07098

TOTAL AREA= 49943

MUL FACTOR=1.0000E+00

* RUN # 16 JAN 1, 1901 00:18:18
START

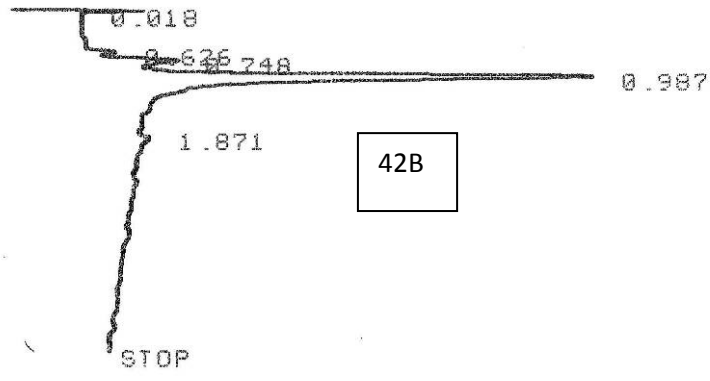


RUN# 16 JAN 1, 1901 00:18:18

AREA%	RT	AREA	TYPE	WIDTH	AREA%
	.629	1915	UU	.070	4.46762
	.757	6429	UU	.121	14.99860
	.992	34520	UU	.170	80.53379

TOTAL AREA= 42864
MUL FACTOR=1.0000E+00

* RUN # 17 JAN 1, 1901 08:24:36
 START

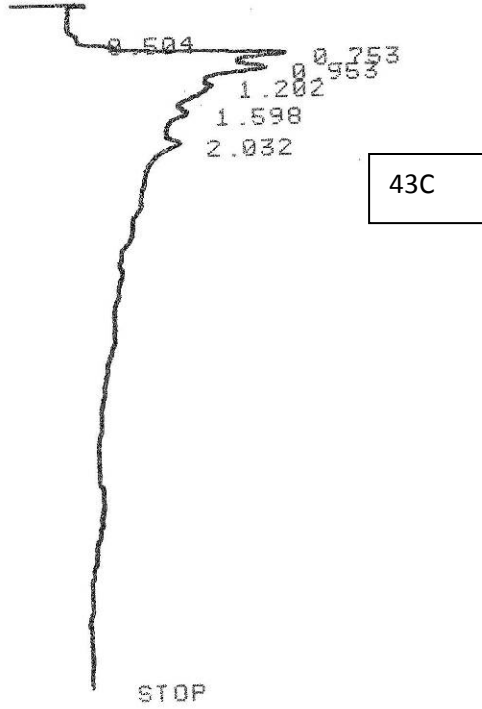


RUN# 17 JAN 1, 1901 08:24:36

RT	AREA	TYPE	WIDTH	AREA%
.018	83	BB	.002	.12786
.626	1244	BU	.058	1.91641
.748	6836	UU	.102	10.53102
.987	55641	UB	.137	85.71629
1.871	1109	PU	.106	1.70844

TOTAL AREA= 64913
 MUL FACTOR=1.0000E+00

* RUN # 9 JAN 1, 1901 07:12:10
 START

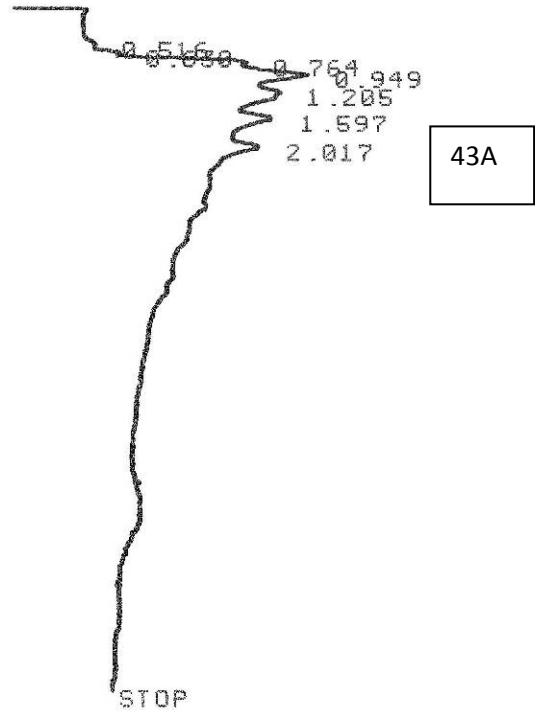


RUN# 9 JAN 1, 1901 07:12:10

AREA#	RT	AREA	TYPE	WIDTH	AREA#
	.504	736	PV	.119	.58571
	.753	25508	UU	.146	20.36305
*	.953	34938	UU	.232	27.80382
*	1.202	29163	UU	.300	23.20805
*	1.598	20290	UU	.310	16.14687
*	2.032	14944	UU	.325	11.89250

TOTAL AREA= 125659
 MUL FACTOR=1.0000E+00

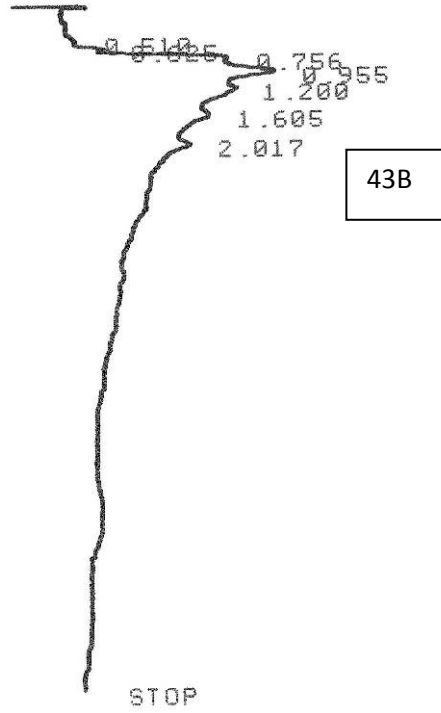
* RUN # 7 JAN 1, 1901 06:49:32
 START



RUN# 7 JAN 1, 1901 06:49:32

AREA#	RT	AREA	TYPE	WIDTH	AREAX
	.516	401	PU	.090	.25648
	.630	1269	UU	.059	.81165
	.764	13024	UU	.104	8.33008
*	.949	41567	UU	.241	26.58605
*	1.205	21678	UU	.153	13.86514
*	1.597	40188	UU	.325	25.70404
↓	2.017	38222	UU	.384	24.44659

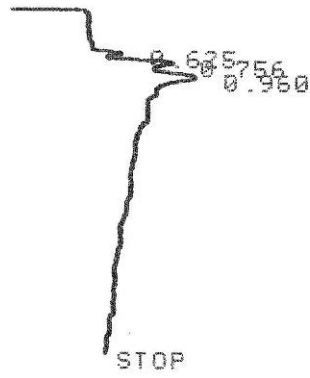
* RUN # 8 JAN 1, 1901 07:00:51
 START



RUN# 8 JAN 1, 1901 07:00:51

AREA#	RT	AREA	TYPE	WIDTH	AREA%
	.510	295	PU	.059	.20997
	.625	1436	UU	.059	1.02207
	.756	12864	UU	.101	9.15594
	.955	43763	UU	.272	31.14827
	1.200	36846	UU	.308	26.22510
	1.605	27840	UU	.329	19.81509
	2.017	17455	UU	.320	12.42358

* RUN # 19 JAN 1, 1901 08:52:45
START



44A

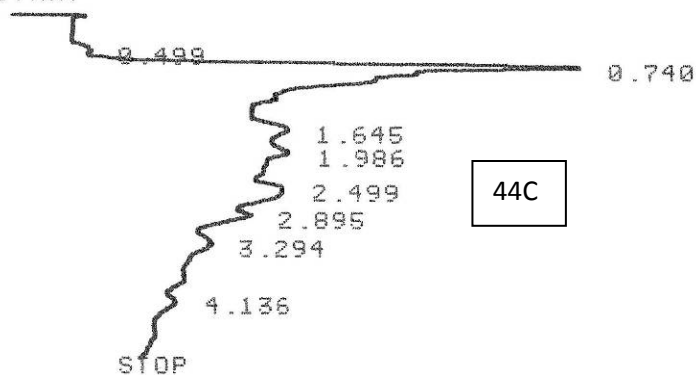
RUN# 19 JAN 1, 1901 08:52:45

AREAX

RT	AREA	TYPE	WIDTH	AREAX
.625	874	PU	.047	5.46421
.756	5236	UU	.103	32.73523
.960	9885	UU	.181	61.80056

TOTAL AREA= 15995
MUL FACTOR=1.0000E+00

* RUN # 21 JAN 1, 1901 09:05:22
 START

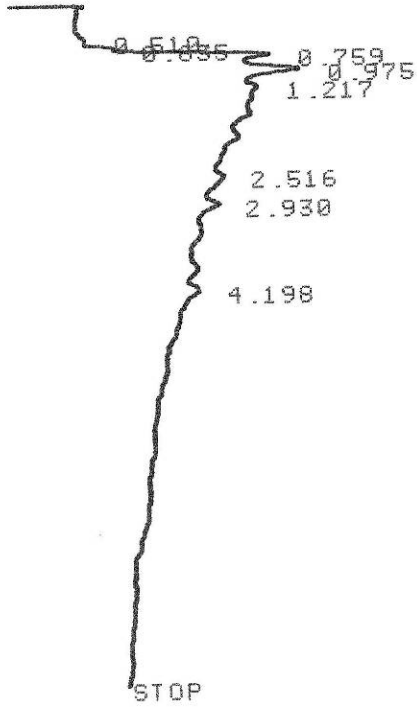


RUN# 21 JAN 1, 1901 09:05:22

AREA%

RT	AREA	TYPE	WIDTH	AREA%
.499	717	PP	.091	.74652
0.740	43276	PU	.110	45.05757
1.645	7235	BU	.212	7.53285
1.986	11735	UU	.260	12.21810
2.499	18961	UU	.339	19.74158
2.895	7194	UP	.181	7.49016
3.294	4459	PU	.250	4.64257
4.136	2469	PP	.210	2.57064

* RUN # 10 JAN 1, 1901 07:23:28
 START

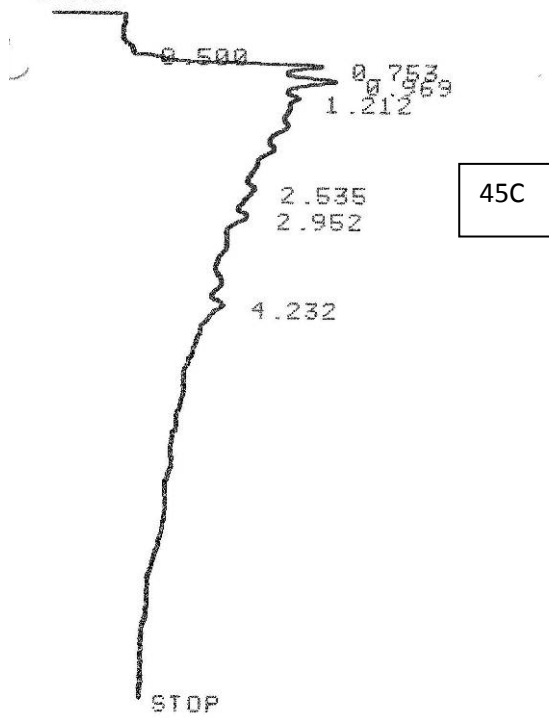


45A

RUN# 10 JAN 1, 1901 07:23:28

AREA%	RT	AREA	TYPE	WIDTH	AREA*
	.510	496	FU	.118	.38927
	.635	1495	UU	.058	1.17331
	*.759	18998	UU	.123	14.91010
	0.975	43344	UU	.249	34.01744
	1.217	25675	UU	.196	20.15038
	2.516	21771	UU	.349	17.08642
	2.930	12145	UU	.260	9.53170
	4.198	3493	UU	.227	2.74139

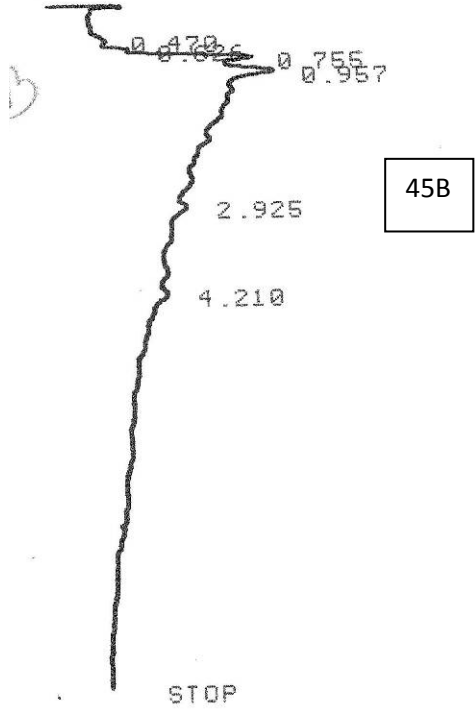
* RUN # 12 JAN 1, 1901 07:46:03
 START



RUN# 12 JAN 1, 1901 07:46:03

AREA#	RT	AREA	TYPE	WIDTH	AREA%
	.500	435	PP	.104	.81744
*	.753	16976	PU	.129	31.90078
*	.969	22203	UU	.206	41.72318
	1.212	4497	UU	.131	8.45062
	2.535	2300	PU	.218	4.32209
	2.952	2240	UP	.156	4.20934
	4.232	4564	UU	.270	8.57653

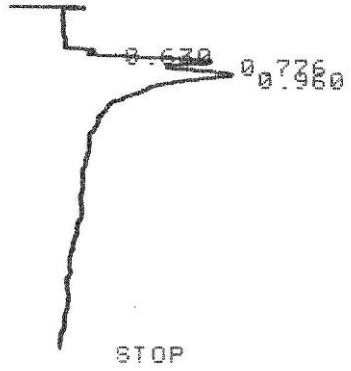
* RUN # 11 JAN 1, 1901 07:34:45
 START



RUN# 11 JAN 1, 1901 07:34:45

AREA#	RT	AREA	TYPE	WIDTH	AREA#
	.470	1609	UU	.137	1.79546
	.626	2113	UU	.066	2.35786
*	.755	17217	UU	.127	19.21219
*	.957	39755	UU	.260	44.36198
	2.925	17833	UU	.287	19.89958
	4.210	11088	UU	.315	12.37293

* RUN # 22 JAN 1, 1901 09:13:52
START

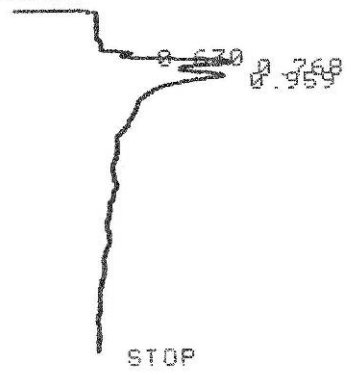


RUN# 22 JAN 1, 1901 09:13:52

AREA#	RT	AREA	TYPE	WIDTH	AREA#
	.630	1370	PU	.056	2.35016
	.776	14709	UU	.122	25.23246
	.960	42215	UU	.315	72.41738

TOTAL AREA= 58294
MUL FACTOR=1.0000E+00

* RUN # 23 JAN 1, 1901 09:20:11
START

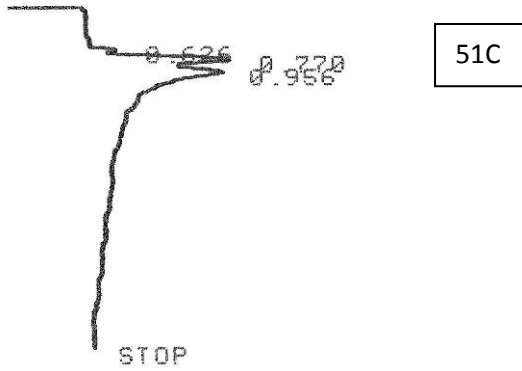


51B

RUN# 23 JAN 1, 1901 09:20:11

AREA%	RT	AREA	TYPE	WIDTH	AREA%
	.630	1604	UU	.063	3.62633
	.768	13884	UU	.127	31.38904
	.959	28744	UU	.283	64.98461

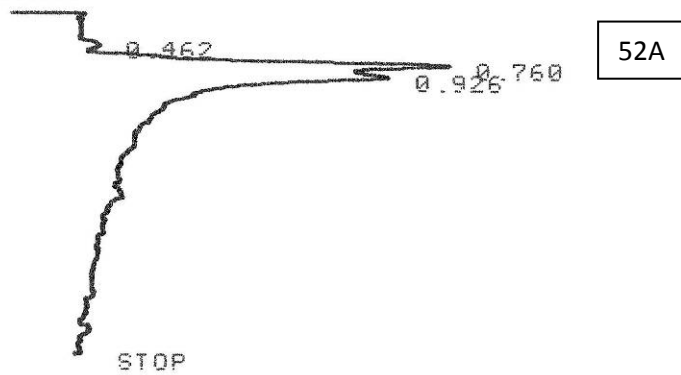
* RUN # 24 JAN 1, 1901 09:26:29
START



RUN# 24 JAN 1, 1901 09:26:29

AREA#	RT	AREA	TYPE	WIDTH	AREA%
	.626	1426	PU	.061	3.01251
*	.770	15050	UU	.127	31.79399
*	.956	30860	UU	.282	65.19350

* RUN # 6 JAN 1, 1901 00:42:54
START



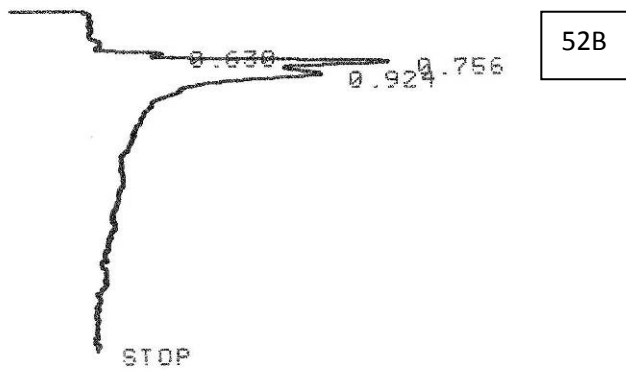
RUN# 6 JAN 1, 1901 00:42:54

AREA%

RT	AREA	TYPE	WIDTH	AREA%
.462	860	BU	.117	1.57003
.760	21578	UU	.142	39.39315
.926	32338	UU	.260	59.03680

TOTAL AREA= 54776
MUL FACTOR=1.0000E+00

* RUN # 7 JAN 1, 1901 00:49:12
START

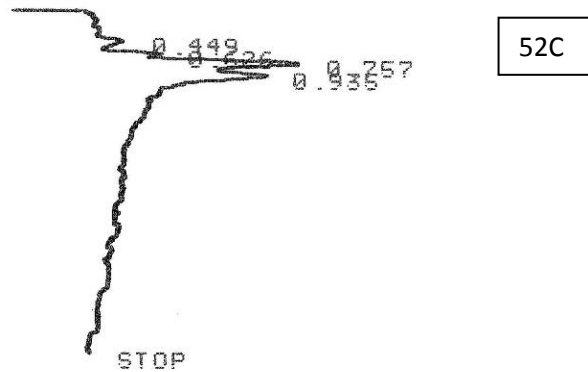


RUN# 7 JAN 1, 1901 00:49:12

AREA#	RT	AREA	TYPE	WIDTH	AREA#
	.630	1668	PU	.063	4.78211
*	.756	13660	UU	.114	39.16285
*	.924	19552	UU	.225	56.05506

TOTAL AREA= 34880
MUL FACTOR=1.0000E+00

* RUN # 8 JAN 1, 1901 00:55:31
START



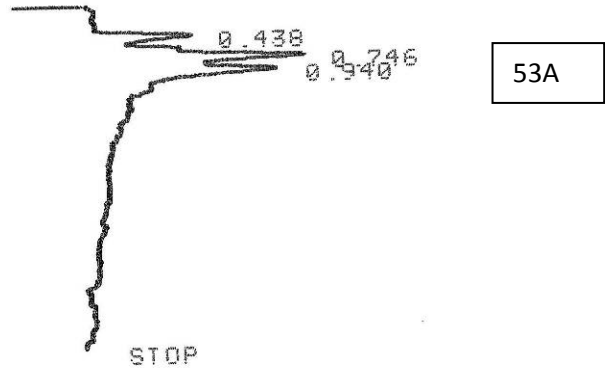
RUN# 8 JAN 1, 1901 00:55:31

AREA%

RT	AREA	TYPE	WIDTH	AREA%
.449	1007	PP	.094	4.77116
.626	1390	PU	.061	6.50500
.757	9455	UU	.122	44.79768
.935	9254	UU	.164	43.84536

TOTAL AREA= 21106
MUL FACTOR=1.0000E+00

* RUN # 9 JAN 1, 1901 01:01:49
 START

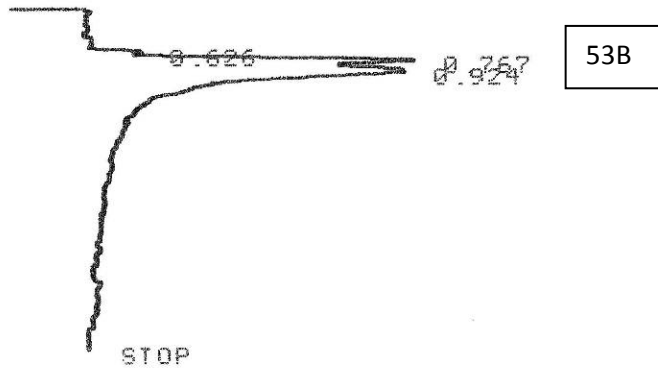


RUN# 9 JAN 1, 1901 01:01:49

AREA%	RT	AREA	TYPE	WIDTH	AREA%
	.438	4503	PU	.112	15.21849
*	.746	13200	UU	.157	44.61117
*	.940	11886	UU	.170	40.17034

TOTAL AREA= 29589
 MUL FACTOR=1.0000E+00

* RUN # 10 JAN 1, 1901 01:08:08
START

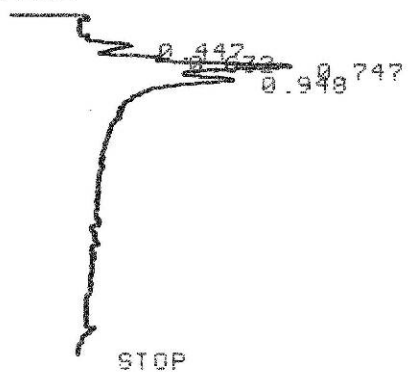


RUN# 10 JAN 1, 1901 01:08:08

AREAX	RT	AREA	TYPE	WIDTH	AREAX
	.626	1339	PV	.060	2.58954
	.767	14858	UU	.109	28.73444
	.924	35511	UB	.273	68.67603

TOTAL AREA= 51708
MUL FACTOR=1.0000E+00

* RUN # 11 JAN 1, 1901 01:14:26
 START



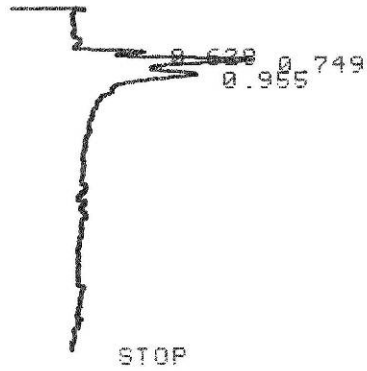
53C

RUN# 11 JAN 1, 1901 01:14:26

RT	AREA	TYPE	WIDTH	AREA%
.447	2464	UU	.127	9.61336
.632	1985	UU	.066	7.74453
.747	10843	UU	.128	42.30424
.948	10339	UU	.181	40.33787

TOTAL AREA 25631

* RUN # 15 JAN 1, 1901 01:39:43
START.



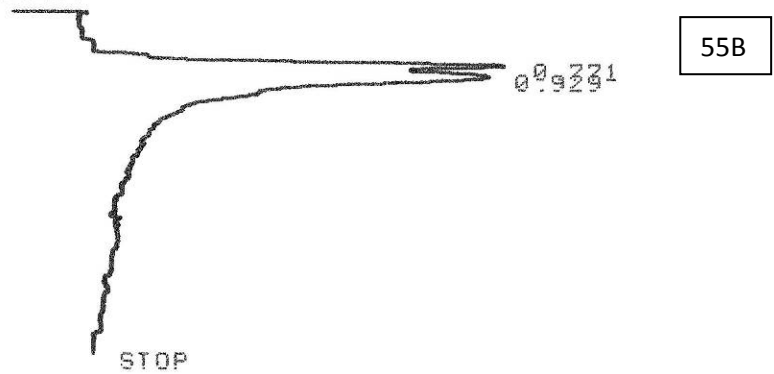
55A

RUN# 15 JAN 1, 1901 01:39:43

AREAX	RT	AREA	TYPE	WIDTH	AREAX
	.628	2063	PU	.068	10.44346
	^ .749	8926	UU	.119	45.18579
	^ .955	8765	UU	.174	44.37075

TOTAL AREA= 19754
MUL FACTOR=1.0000E+00

* RUN # 16 JAN 1, 1901 01:46:02
START



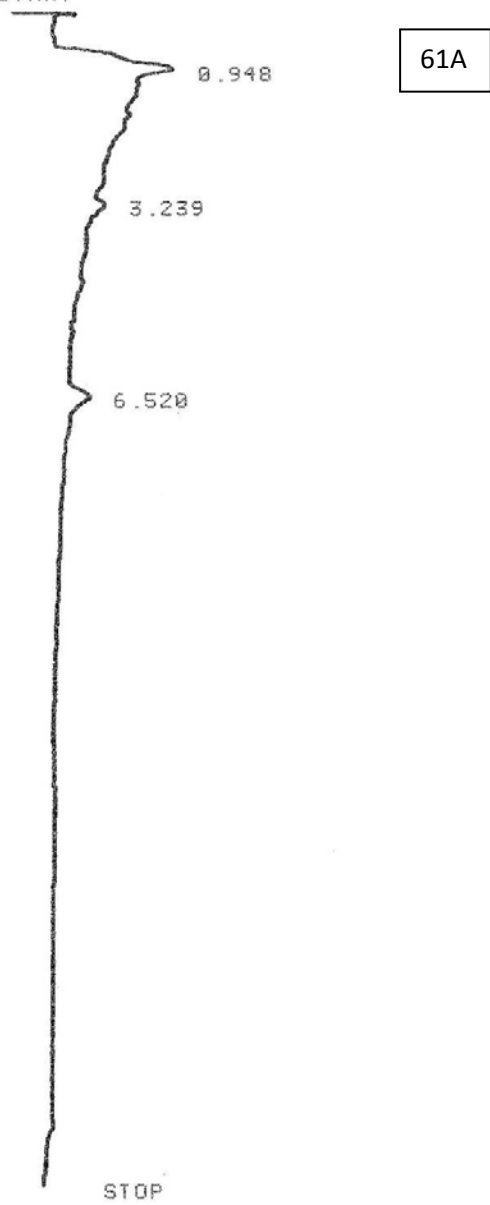
RUN# 16 JAN 1, 1901 01:46:02

AREA#

RT	AREA	TYPE	WIDTH	AREA#
.771	22652	UU	.127	29.17982
.929	54977	UU	.322	70.82019

TOTAL AREA= 77629
MUL FACTOR=1.0000E+00

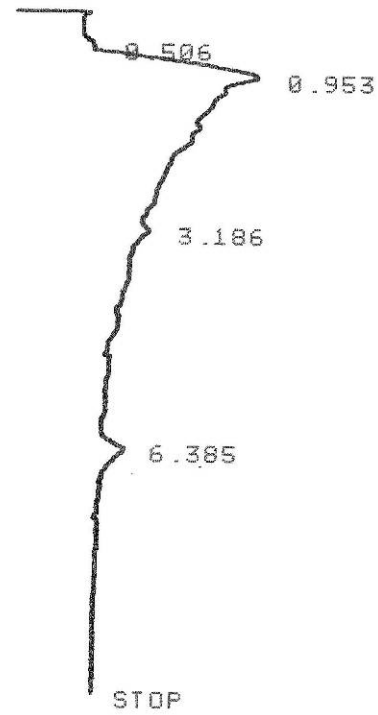
* RUN # 12 JAN 1, 1901 03:04:58
START



RUN# 12 JAN 1, 1901 03:04:58

AREA#	RT	AREA	TYPE	WIDTH	AREA#
*	.948	38091	PU	.362	84.71632
*	3.239	2824	PU	.210	6.28072
*	6.520	4048	UU	.192	9.00296

* RUN # 29 JAN 1, 1901 08:45:55
 START

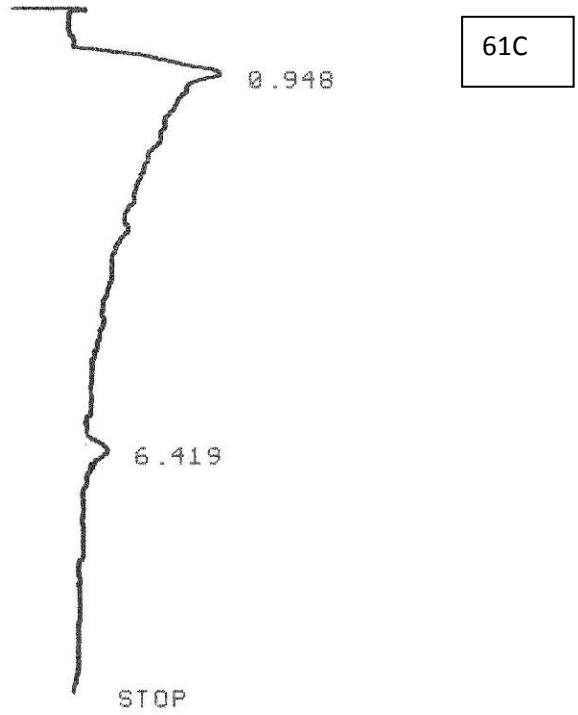


61B

RUN# 29 JAN 1, 1901 08:45:55

AREA#	RT	AREA	TYPE	WIDTH	AREA#
	.506	555	PU	.117	1.00981
*	.953	49789	UU	.380	90.58970
	3.186	1908	BP	.186	3.47155
*	6.385	2709	UU	.136	4.92895

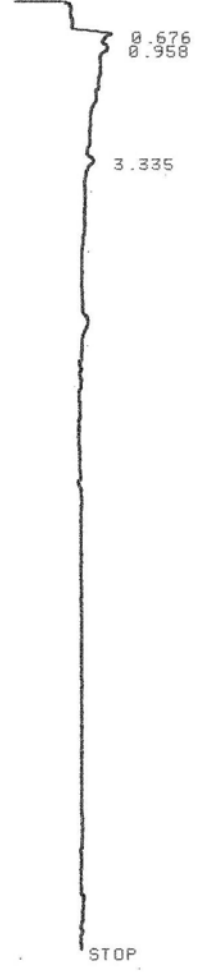
* RUN # 30 JAN 1, 1901 08:57:14
START



RUN# 30 JAN 1, 1901 08:57:14

AREA#	RT	AREA	TYPE	WIDTH	AREA%
*	.948	48567	PU	.390	90.26317
o	6.419	5239	BV	.280	9.73683

* RUN # 13 JAN 1, 1901 03:26:17
START

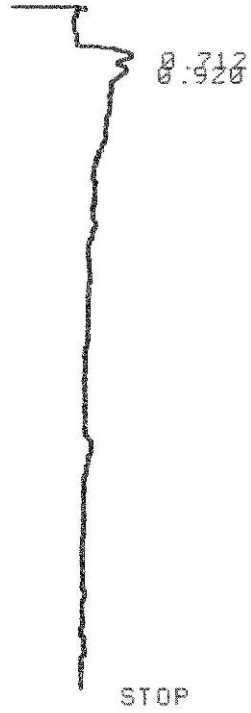


62A

RUN# 13 JAN 1, 1901 03:26:17

AREA%	RT	AREA	TYPE	WIDTH	AREA%
	.676	5012	PU	.120	44.72205
	.958	4144	UU	.171	36.97690
	3.335	2051	BP	.181	18.30106

* RUN # 31 JAN 1, 1901 09:08:32
START

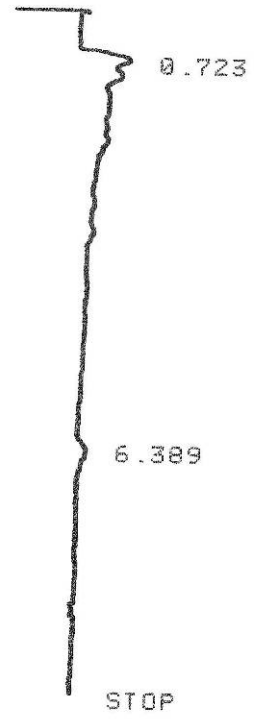


62B

RUN#	31	JAN	1,	1901	09:08:32
AREA#					
RT	AREA	TYPE	WIDTH	AREA#	
.712	9010	PU	.192	47.38862	
• .920	10003	UU	.244	52.61139	

* RUN # 32 JAN 1, 1901 09:19:51
START

62C



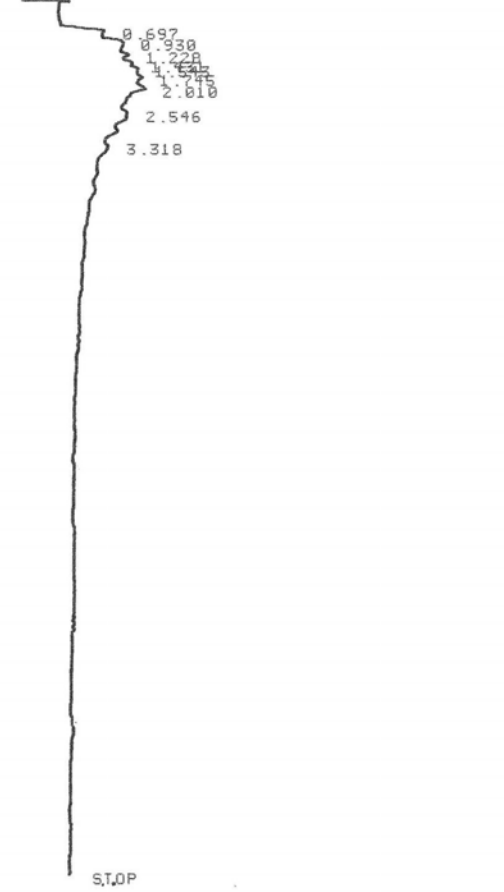
RUN# 32 JAN 1, 1901 09:19:51

AREA#	RT	AREA	TYPE	WIDTH	AREA%
	.723	7634	PU	.190	87.59610
	6.389	1081	PU	.118	12.40390

TOTAL: -----

* RUN # 14 JAN 1, 1901 03:47:35
 START

63A

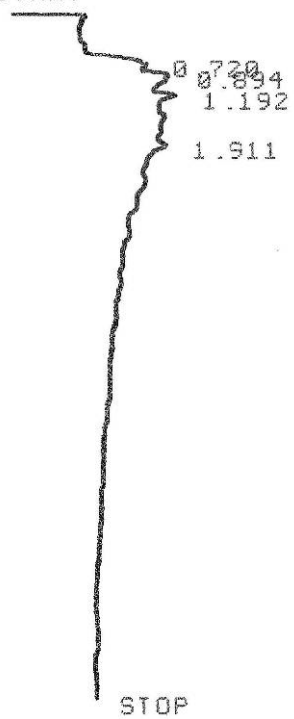


RUN# 14 JAN 1, 1901 03:47:35

AREA#	RT	AREA	TYPE	WIDTH	AREA%
	.697	6970	PU	.121	4.41504
→	.930	10984	UU	.138	6.95890
→	1.220	13570	UU	.161	8.59726
	1.431	13454	UU	.152	8.52377
	1.543	10818	UU	.115	6.85373
	1.745	26745	UU	.274	16.94426
	2.010	43053	UU	.439	27.78302
	2.546	15139	UU	.211	9.59130
	3.318	16308	UU	.419	10.33192

* RUN # 33 JAN 1, 1901 09:31:09
START

63B

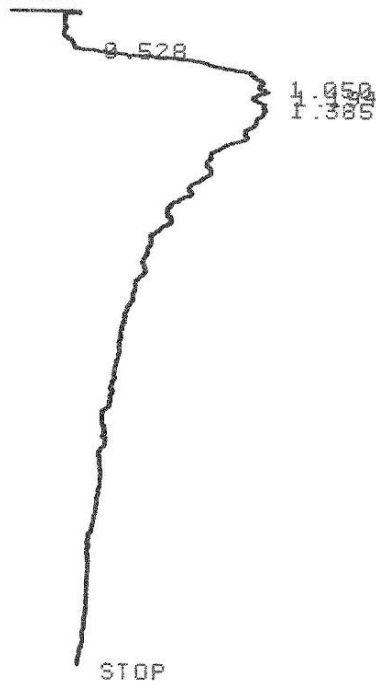


RUN# 33 JAN 1, 1901 09:31:09

AREA#	RT	AREA	TYPE	WIDTH	AREA#
	.720	10323	UU	.190	16.24569
→	.894	19611	UU	.265	30.86257
→	1.192	14520	UU	.190	22.86326
	1.911	19081	UU	.298	30.02849

* RUN # 34 JAN 1, 1901 09:42:27
 START

63C



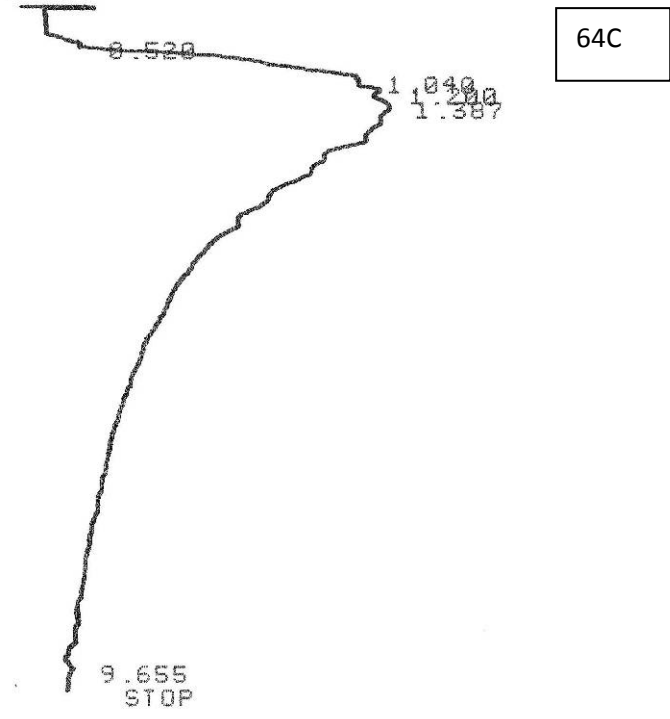
RUN# 34 JAN 1, 1901 09:42:27

AREA%

RT	AREA	TYPE	WIDTH	AREA%
• .528	1156	PV	.139	1.02850
↘ 1.050	65636	UU	.402	58.39658
→ 1.194	26554	UU	.162	23.62518
1.385	19051	UU	.122	16.94974

TOTAL AREA= 112397
 MUL FACTOR=1.0000E+00

* RUN # 36 JAN 1, 1901 10:05:04
 START



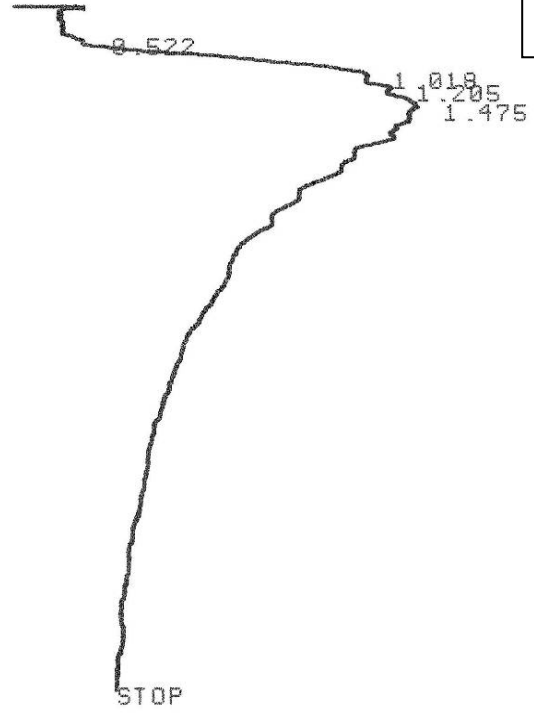
RUN# 36 JAN 1, 1901 10:05:04

AREA#	RT	AREA	TYPE	WIDTH	AREA#
	.520	804	PP	.100	1.43145
*	1.040	40277	PU	.422	71.70938
*	1.200	10657	UU	.165	18.97377
	1.387	3484	UB	.257	6.20293
	9.655	945	PU	.127	1.68248

TOTAL AREA= 56167

* RUN # 35 JAN 1, 1901 09:53:45
START

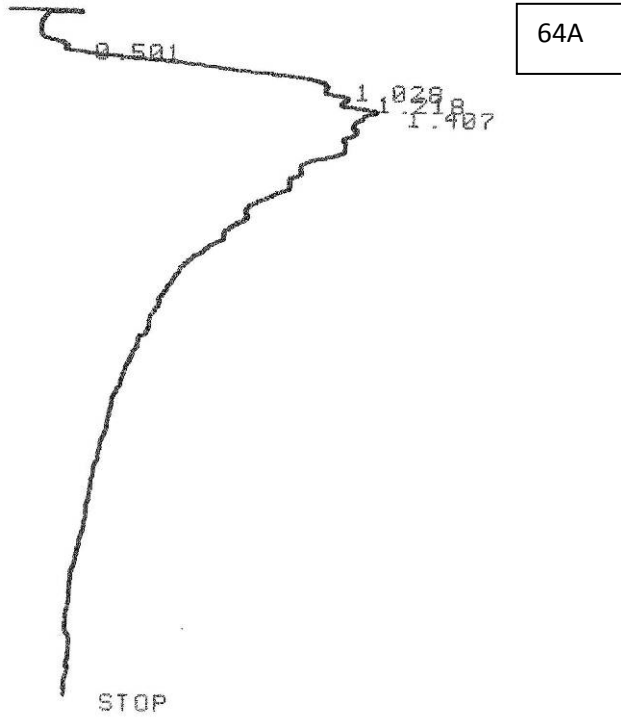
64B



RUN# 35 JAN 1, 1901 09:53:45

RT	AREA	TYPE	WIDTH	AREA*
.522	1326	PU	.089	.63236
1.018	84447	UU	.346	40.27192
1.205	43131	UU	.168	20.56874
1.475	80788	UU	.303	38.52698

* RUN # 17 JAN 1, 1901 05:58:36
 START

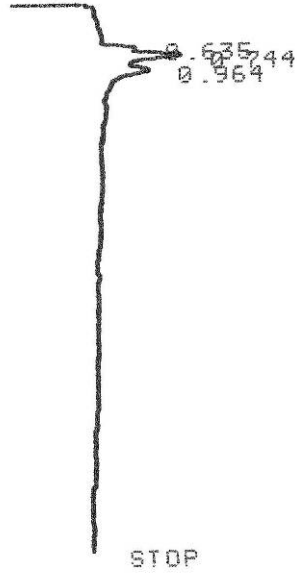


RUN# 17 JAN 1, 1901 05:58:36

AREA#	RT	AREA	TYPE	WIDTH	AREA#
	.501	1619	PU	.092	.80422
-	1.028	76694	UU	.344	38.09669
	1.218	46928	UU	.199	23.31086
	1.407	76073	UU	.298	37.78822

TOTAL AREA = 100.00000

* RUN # 25 JAN 1, 1901 09:35:07
START



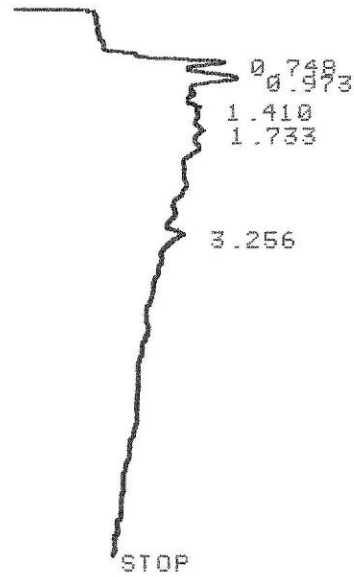
65A

RUN# 25 JAN 1, 1901 09:35:07

AREA#	RT	AREA	TYPE	WIDTH	AREA%
	.635	2238	UU	.070	12.25697
	.744	9125	UU	.130	49.97536
	.964	6896	UU	.166	37.76768

TOTAL AREA= 18259
MUL FACTOR=1.0000E+00

* RUN # 26 JAN 1, 1901 09:44:26
START

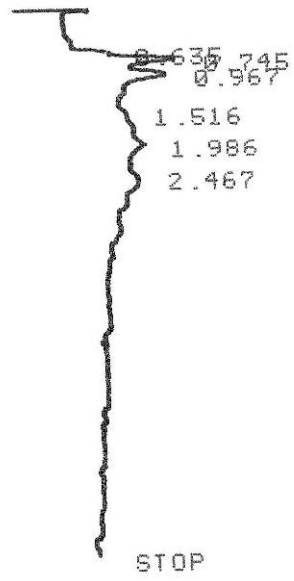


65B

RUN# 26 JAN 1, 1901 09:44:26

AREA#	RT	AREA	TYPE	WIDTH	AREA%
	.748	14213	PU	.145	24.13278
	.973	23124	UU	.222	39.26309
	1.410	7835	UU	.132	13.30334
	1.733	11653	UU	.214	19.78606
	3.256	2070	PU	.129	3.51473

* RUN # 27 JAN 1, 1901 09:53:44
 / START

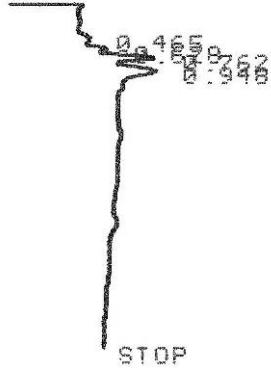


65C

RUN# 27 JAN 1, 1901 09:53:44

AREAX	RT	AREA	TYPE	WIDTH	AREAX
	.635	1834	UU	.055	3.05860
	.745	12507	UU	.140	20.85821
	.967	17406	UU	.219	29.02838
	1.516	6460	UU	.167	10.77349
	1.986	16452	UU	.348	27.43738
	2.467	5303	UU	.144	8.84394

* RUN # 5 JAN 1, 1901 00:24:15
START

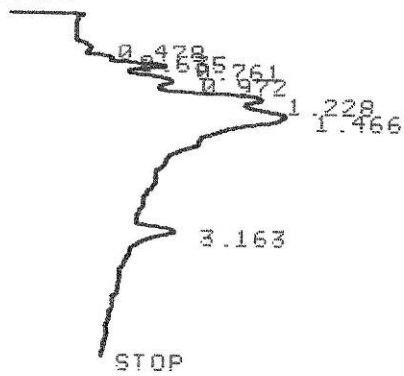


RUN# 5 JAN 1, 1901 00:24:15

AREA%	RT	AREA	TYPE	WIDTH	AREA%
	.465	248	PB	.067	1.67432
	.628	908	PV	.058	6.13017
	.762	4977	UV	.095	33.60114
	.948	8679	UV	.190	58.59438

TOTAL AREA= 14812
MUL FACTOR=1.0000E+00

* RUN # 6 JAN 1, 1901 00:36:13
 START



71B

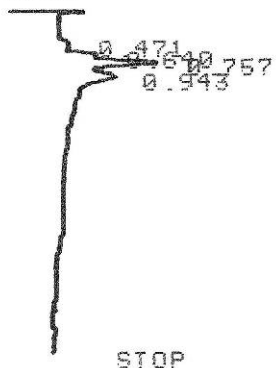
RUN# 6 JAN 1, 1901 00:36:13

AREA#

RT	AREA	TYPE	WIDTH	AREA#
.478	1298	PU	.126	1.10037
.635	1516	UU	.063	1.28518
.761	7756	UU	.114	6.57511
.972	12505	UU	.182	10.60105
1.228	25723	UU	.185	21.80654
1.466	62263	UU	.401	52.78314
3.163	6899	BP	.191	5.84859

TOTAL AREA= 117960
 MUL FACTOR=1.0000E+00

* RUN # 7 JAN 1, 1901 00:42:31
START



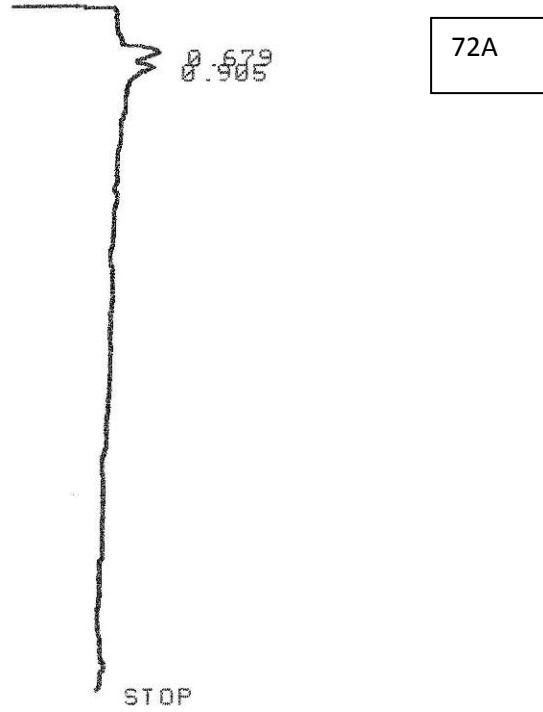
71C

RUN# 7 JAN 1, 1901 00:42:31

AREA#	RT	AREA	TYPE	WIDTH	AREA%
	.471	1042	PU	.119	5.07426
	.640	1935	UU	.063	9.42294
	.757	9094	UU	.112	44.28538
	.943	8464	UU	.201	41.21744

TOTAL AREA= 20535
MUL FACTOR=1.0000E+00

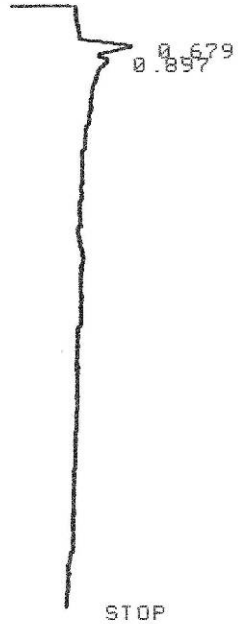
* RUN # 4 JAN 1, 1901 01:45:18
START



RUN# 4 JAN 1, 1901 01:45:18

AREA%	RT	AREA	TYPE	WIDTH	AREA%
	.679	5175	BU	.165	54.13749
	.905	4384	UU	.172	45.86253

* RUN # 10 JAN 1, 1901 04:15:51
START

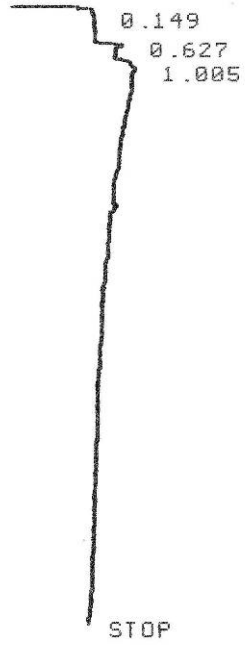


72B

RUN# 10 JAN 1, 1901 04:15:51

AREA#	RT	AREA	TYPE	WIDTH	AREA%
	.679	6829	PU	.146	68.63315
	0.897	3121	UU	.163	31.36684

* RUN # 11 JAN 1, 1981 04:27:09
START

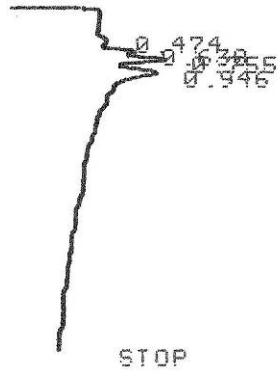


72C

RUN# 11 JAN 1, 1981 04:27:09

AREA%	RT	AREA	TYPE	WIDTH	AREA%
	.149	2960	BU	.249	33.34835
	.627	1714	PU	.070	19.31050
	1.005	4202	UU	.232	47.34114

* RUN # 10 JAN 1, 1901 01:01:27
START



73A

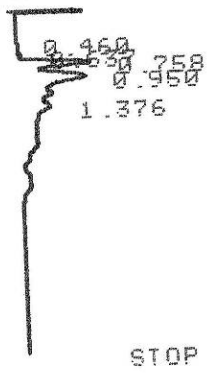
RUN# 10 JAN 1, 1901 01:01:27

AREA%

RT	AREA	TYPE	WIDTH	AREA%
.474	1268	PU	.123	6.06815
.638	2520	UU	.073	12.05972
.755	6549	UU	.109	31.34093
.946	10559	UU	.196	50.53122

TOTAL AREA= 20896
MUL FACTOR=1.0000E+00

* RUN # 9 JAN 1, 1901 00:55:09
 START



73B

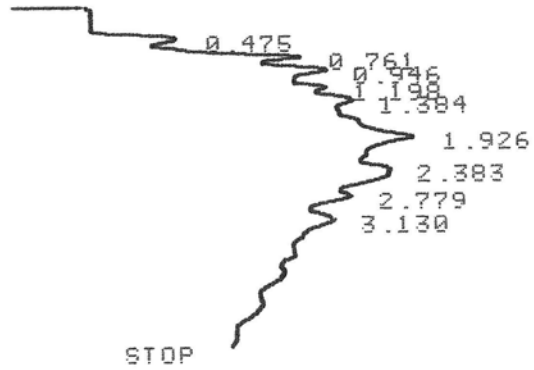
RUN# 9 JAN 1, 1901 00:55:09

AREA%	RT	AREA	TYPE	WIDTH	AREA%
	.460	4614	BU	.133	9.05665
	.637	3542	UU	.077	6.95246
	.758	12469	UU	.124	24.47494
	.950	19996	UU	.222	39.24941
	1.376	10325	UU	.212	20.26655

TOTAL AREA= 50946
 MUL FACTOR=1.0000E+00

* RUN # 8 JAN 1, 1901 00:48:50
 START

73C

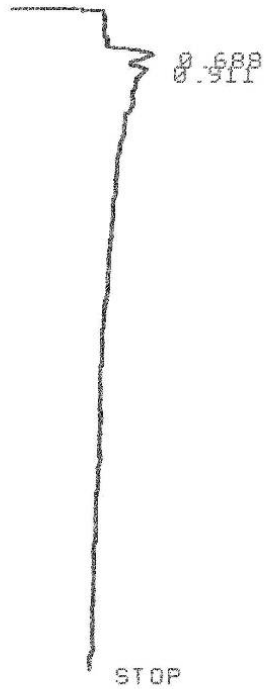


RUN# 8 JAN 1, 1901 00:48:50

RT	AREA	TYPE	WIDTH	AREA%
.475	9653	PU	.141	2.49854
.761	28136	UU	.169	7.28259
.946	41145	UU	.224	10.64978
1.190	27382	UU	.156	7.08743
1.384	41153	UU	.212	10.65185
1.926	110631	UU	.480	28.63522
2.383	31192	UU	.158	8.07359
2.779	38978	UU	.253	10.08888
3.130	58076	UU	.447	15.03212

TOTAL AREA= 386346
 MUL FACTOR=1.0000E+00

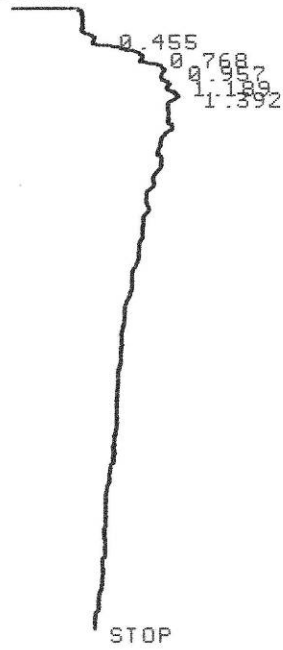
* RUN # 5 JAN 1, 1901 01:56:36
START



73A

AREA%	RT	AREA	TYPE	WIDTH	AREA%
	.688	5714	BU	.160	63.61611
	.911	3268	UU	.160	36.38389

* RUN # 12 JAN 1, 1901 04:38:27
 START



73B

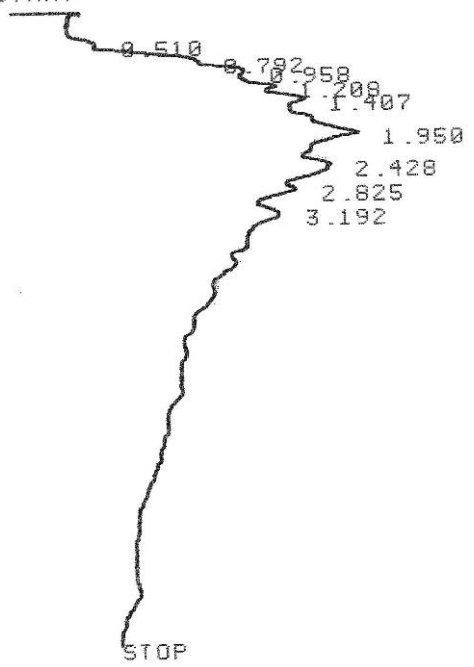
RUN# 12 JAN 1, 1901 04:38:27

AREA%	RT	AREA	TYPE	WIDTH	AREAX
.455	336	PB	.073	1.26587	
.768	6322	PV	.181	23.81795	
.957	9181	UU	.225	34.58915	
1.189	5487	UU	.177	20.67212	
1.392	5217	UU	.197	19.65490	

* RUN # 13 JAN 1, 1901 04:49:45

START

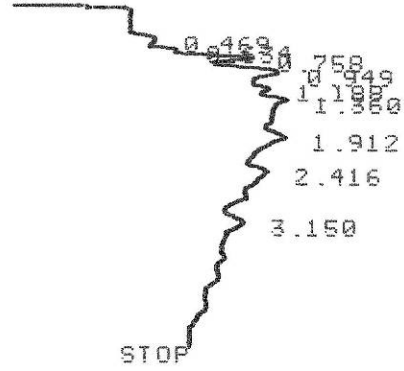
73C



RUN# 13 JAN 1, 1901 04:49:45

AREA#	RT	AREA	TYPE	WIDTH	AREA#
	.510	1637	BU	.111	.62929
	.792	15991	UU	.169	6.14722
	.958	25172	UU	.194	9.67655
	1.208	31787	UU	.219	12.21947
	1.407	35896	UU	.222	13.79904
	1.950	91292	UU	.496	35.09422
	2.428	23257	UU	.173	8.94039
	2.825	19524	UU	.231	7.50536
	3.192	15578	UU	.305	5.98845

* RUN # 11 JAN 1, 1901 01:07:46
 START



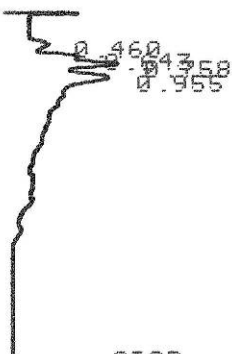
74A

RUN# 11 JAN 1, 1901 01:07:46

RT	AREA	TYPE	WIDTH	AREAX
.469	1547	PP	.107	3.11200
.634	1045	PU	.055	2.10270
.758	7228	UU	.103	14.54385
.949	13658	UU	.178	27.48199
1.188	5627	UU	.121	11.32239
1.360	6559	UU	.147	13.19772
1.912	4507	BU	.197	9.06877
2.416	5897	UU	.268	11.86567
3.150	3630	PU	.187	7.30412

TOTAL AREA= 49698
 MUL FACTOR=1.0000E+00

* RUN # 12 JAN 1, 1901 01:14:04
START



74B

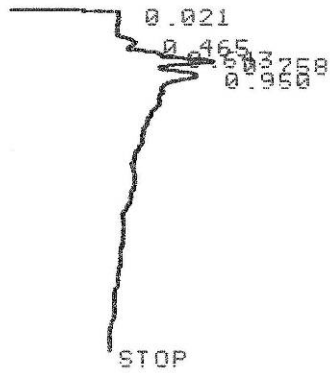
RUN# 12 JAN 1, 1901 01:14:04

AREA%

RT	AREA	TYPE	WIDTH	AREA%
.460	1217	PP	.093	11.76982
.643	720	PB	.054	6.96325
.758	3092	BP	.080	29.90329
.955	5311	PU	.141	51.36365

TOTAL AREA= 10340

* RUN # 13 JAN 1, 1901 01:20:26
 START



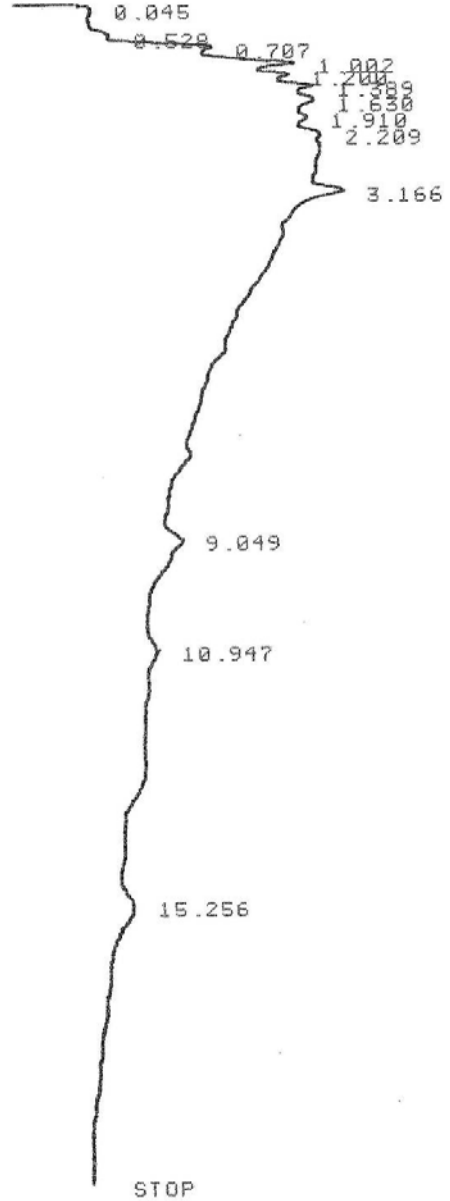
74C

RUN# 13 JAN 1, 1901 01:20:26

AREA#	RT	AREA	TYPE	WIDTH	AREA#
	.021	562	PB	.025	2.13234
	.465	1709	PU	.125	6.48429
	.643	2042	UU	.063	7.74776
	.758	9610	UU	.124	36.46229
	.950	12433	UU	.206	47.17333

TOTAL AREA= 26356
 MUL FACTOR=1.0000E+00

* RUN # 19 JAN 1, 1901 06:55:57
 START

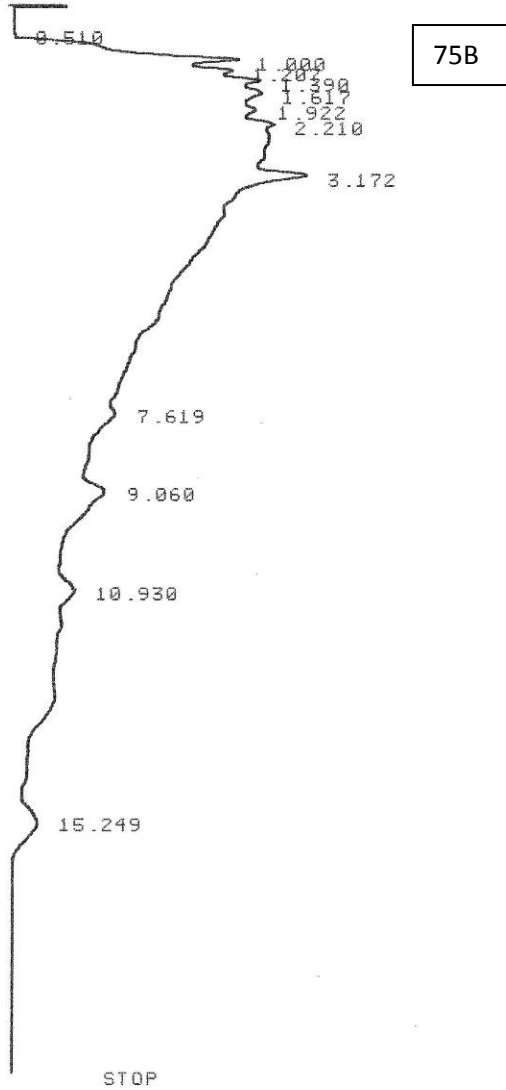


75C

RUN# 19 JAN 1, 1901 06:55:57

AREA#	RT	AREA	TYPE	WIDTH	AREA#
	.045	336	BB	.054	.21494
	.707	13165	PU	.150	8.42177
	1.002	29205	UU	.211	18.68271
	1.200	17200	UU	.155	11.00300
	1.389	20866	UU	.180	13.34818
	1.630	23060	UU	.252	14.75169
	1.910	14436	UU	.261	9.23484
	2.209	7018	UU	.215	5.00125
	3.166	10211	UP	.236	6.53207
*	9.049	12425	UP	.583	7.94839
	10.947	3292	BU	.312	2.10592
	15.256	4307	UU	.271	2.75523

* RUN # 18 JAN 1, 1901 06:34:39
 START

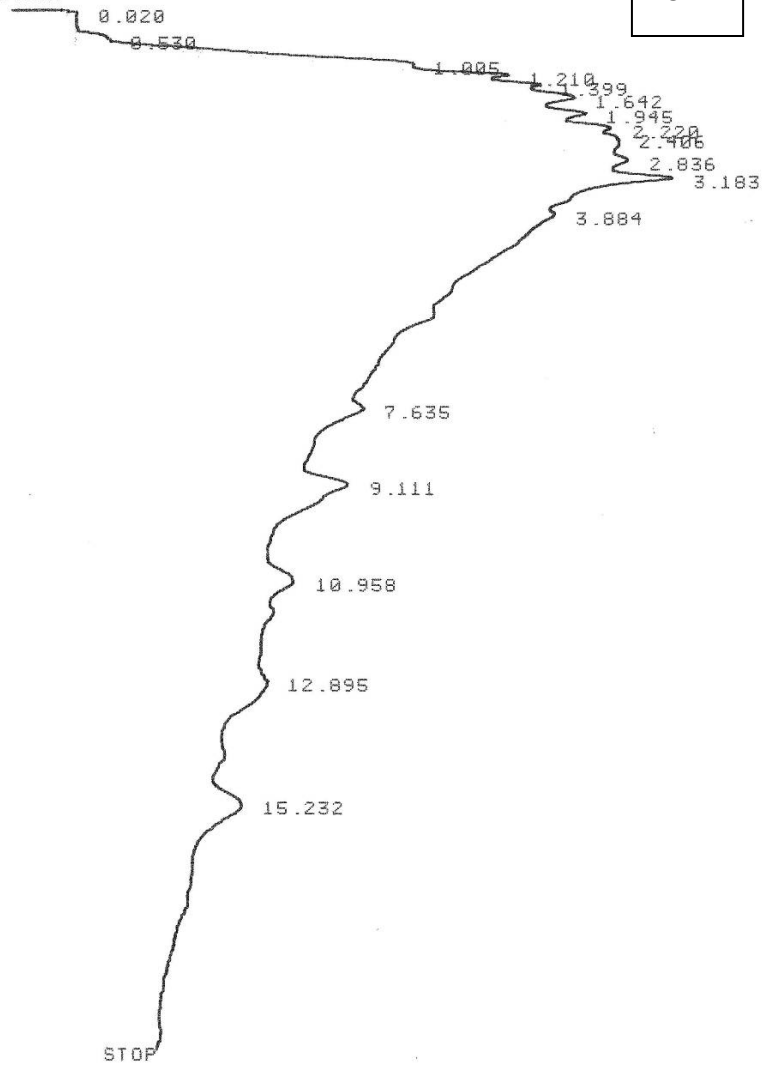


RUN# 18 JAN 1, 1901 06:34:39

AREA#	RT	AREA	TYPE	WIDTH	AREA%
	.510	238	PU	.035	.05477
	1.000	66625	UU	.283	15.33181
	1.207	35121	UU	.161	8.08208
	1.390	45466	UU	.191	10.46268
	1.617	65199	UU	.285	15.00366
	1.922	56401	UU	.270	12.97906
	2.210	45908	UU	.215	10.56440
	3.172	85046	UU	.419	19.57087
	7.619	4744	UP	.335	1.09169
	9.060	11462	PU	.384	2.63765
	10.930	5970	PP	.364	1.37382
	15.249	12374	I BB	.560	2.84752

* RUN # 17 JAN 1, 1901 06:13:21

75A



RUN# 17 JAN 1, 1901 06:13:21

AREA#	RT	AREA	TYPE	WIDTH	AREA%
	.020	648	BU	.061	.16733
	.530	942	PP	.151	.24324
	1.005	52447	PU	.245	13.54292
	1.210	51074	UU	.196	13.18038
	1.399	41128	UU	.170	10.62012
	1.642	62168	UU	.299	16.05310
	1.945	33653	UU	.256	8.68991
	2.220	15788	UU	.198	4.07680
	2.406	4723	UU	.148	1.21958
	2.836	5715	BU	.208	1.47573
	3.183	26548	UU	.265	6.85525
	3.884	10518	UB	.600	2.71597
	7.635	6291	BP	.268	1.62447
	9.111	33818	PP	.545	8.73252
	10.958	12027	PU	.407	3.10563
	12.895	6255	UU	.254	1.61517
	15.232	23522	BU	.588	6.07388

TOTAL AREA= 387265

* RUN # 5 JAN 1, 1901 01:54:35
START

.444
.625
1.008

81A



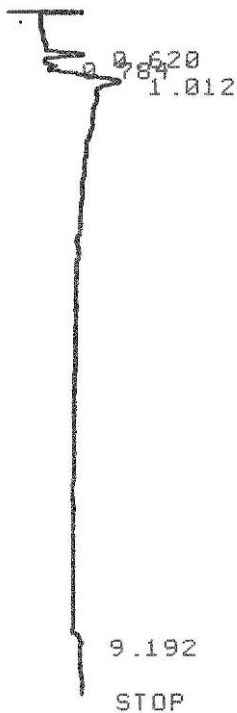
19.179

STOP

RUN# 5 JAN 1, 1901 01:54:35

AREA#	RT	AREA	TYPE	WIDTH	AREA%
	.444	788	PV	.151	5.44876
	.625	3242	UU	.081	22.41737
	1.008	9274	PV	.192	64.12669
	19.179	1158	PV	.166	8.00719

* RUN # 10 JAN 1, 1901 03:51:51
 START



81B

RUN# 10 JAN 1, 1901 03:51:51

AREA#	RT	AREA	TYPE	WIDTH	AREA#
	.620	2629	UP	.071	19.02591
	.784	442	PP	.057	3.19873
	1.012	9843	PU	.189	71.23318
	.9.192	904	PU	.146	6.54219

*. RUN # 11
START

JAN 1, 1981 04:03:10

81C

0.625
1.007



STOP

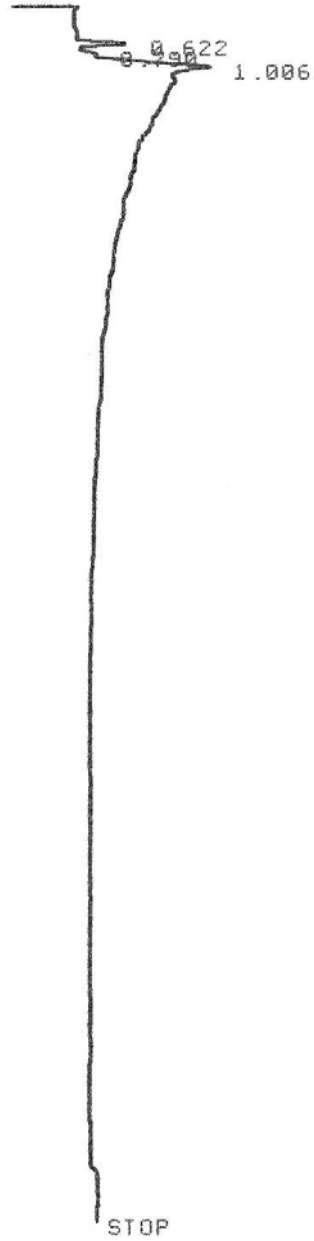
RUN# 11

JAN 1, 1981 04:03:10

AREA%

RT	AREA	TYPE	WIDTH	AREA%
.625	3479	UU	.089	25.37194
1.007	10233	PU	.206	74.62810

*. RUN # 6 JAN 1, 1901 02:15:53
START



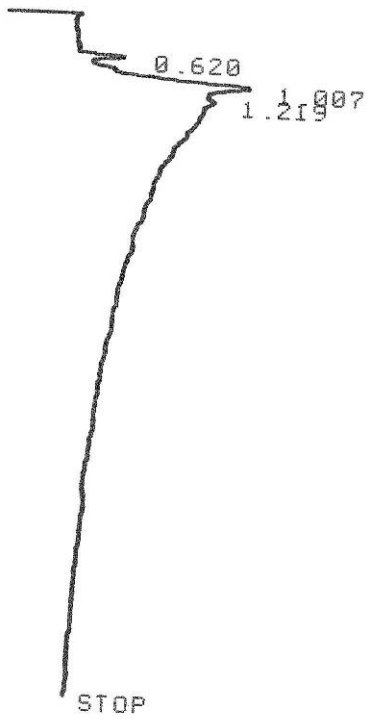
82A

RUN# 6 JAN 1, 1901 02:15:53

AREA%	RT	AREA	TYPE	WIDTH	AREA%
.622		3092	UP	.067	10.43537
.790		1046	PU	.068	3.53021
1.006		25492	UU	.215	86.03443
TOTAL	0050-	29630			

*. RUN # 12 JAN 1, 1901 04:14:29
 START

5
 0



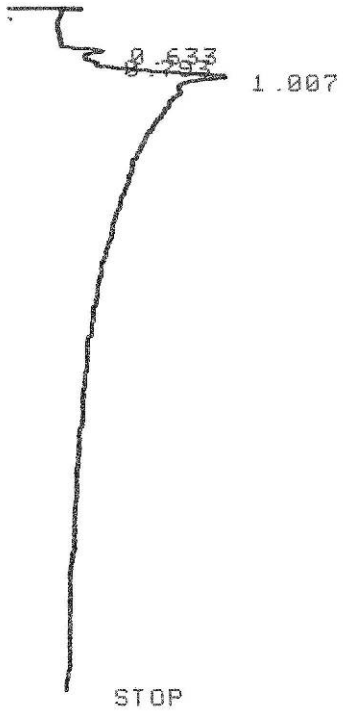
82B

RUN# 12 JAN 1, 1901 04:14:29

AREA#	RT	AREA	TYPE	WIDTH	AREA%
	.620	2045	PU	.060	5.23339
	1.007	24988	UU	.209	63.94717
	1.219	12043	UU	.158	30.81942
TOTAL AREA-					30.81942

* RUN # 13 JAN 1, 1901 04:25:50
START

82C



RUN# 13 JAN 1, 1901 04:25:50

AREA#	RT	AREA	TYPE	WIDTH	AREA%
	.633	1656	PU	.062	6.33730
	.793	1409	UU	.094	5.39206
	1.007	23066	UU	.197	88.27062

TOTAL AREA 26171

*. RUN # 7 JAN 1, 1901 02:37:12
 START

83A



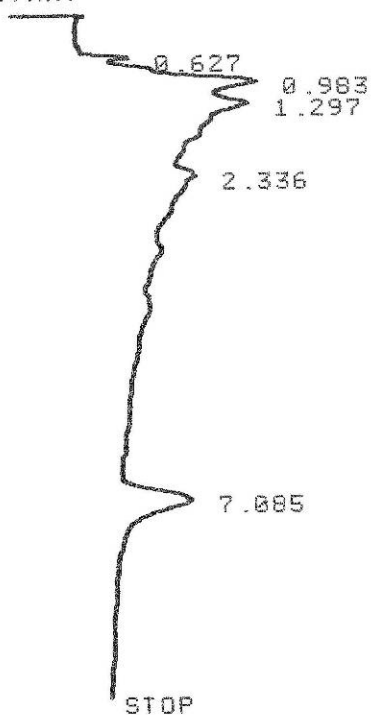
RUN# 7 JAN 1, 1901 02:37:12

AREA%	RT	AREA	TYPE	WIDTH	AREA%
	.496	692	PU	.154	.90680
	.634	7626	UU	.136	9.99318
	.980	20100	UU	.246	26.34972
	1.294	20533	UU	.280	26.90665
	2.325	5778	UU	.226	7.57155
	3.452	1058	UU	.122	1.38641
	7.160	20517	PU	.374	26.88568

*. RUN # 14 JAN 1, 1901 04:37:09

START

83B



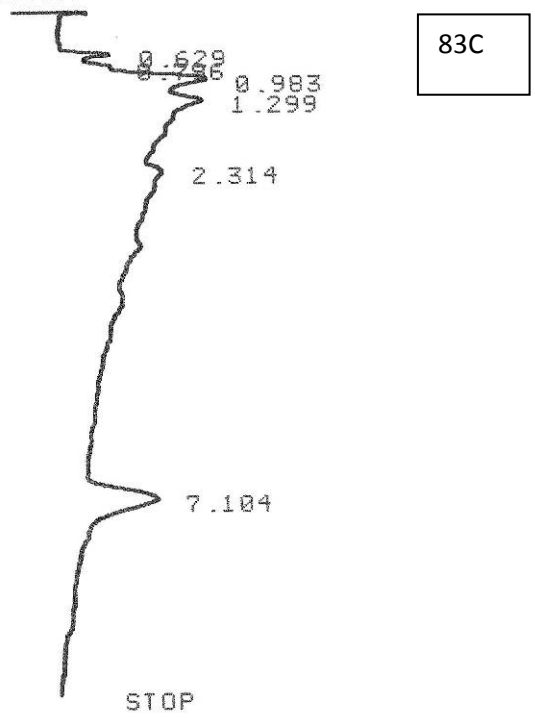
RUN# 14 JAN 1, 1901 04:37:09

AREA%

RT	AREA	TYPE	WIDTH	AREA%
.627	3790	PV	.088	2.50788
.983	43138	UU	.297	29.45538
1.297	40797	UU	.303	27.85690
2.336	37450	UU	.494	25.57152
7.085	21277	UU	.357	14.52831

TOTAL AREA = 146450

*. RUN # 15 JAN 1, 1901 04:48:28
START

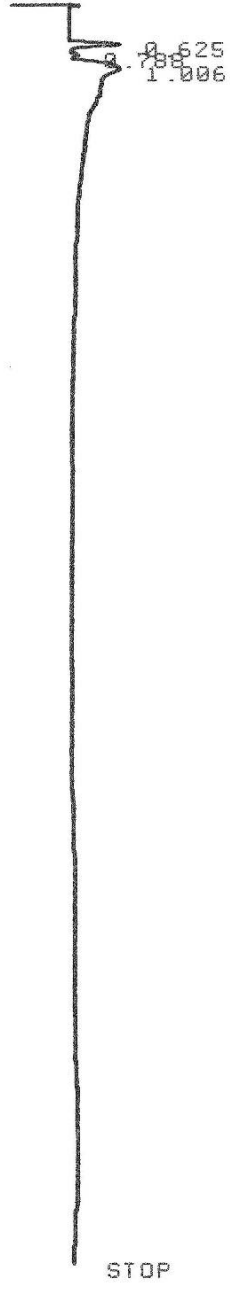


RUN# 15 JAN 1, 1901 04:48:28

AREA%	RT	AREA	TYPE	WIDTH	AREA%
	.629	2964	PU	.076	3.16045
	.796	2930	UU	.083	3.12420
	.983	26453	UU	.238	28.20631
	1.299	28773	UU	.293	30.68007
	2.314	8224	UU	.240	8.76909
	7.104	24440	BU	.382	26.05989

*. RUN # 8 JAN 1, 1901 02:58:30
START

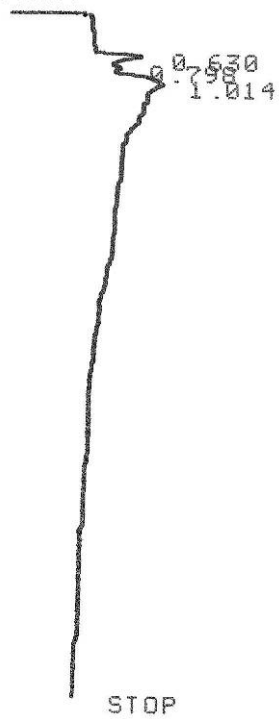
84A



AREA#	RT	AREA	TYPE	WIDTH	AREA#
	.625	3388	UP	.071	23.15790
	.788	470	PU	.057	3.21258
	.1.006	10772	UU	.237	73.62954

*. RUN # 16 JAN 1, 1901 04:59:46
START

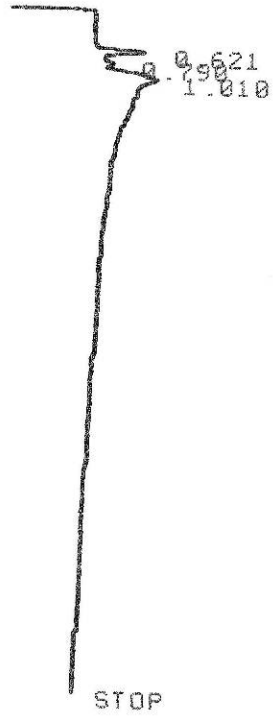
84B



RUN# 16 JAN 1, 1901 04:59:46

AREA%	RT	AREA	TYPE	WIDTH	AREA%
	.630	3809	PU	.098	21.44949
	.798	1449	UU	.081	8.15970
	1.014	12500	UU	.253	70.39082

* RUN # 17 JAN 1, 1901 05:11:05
START



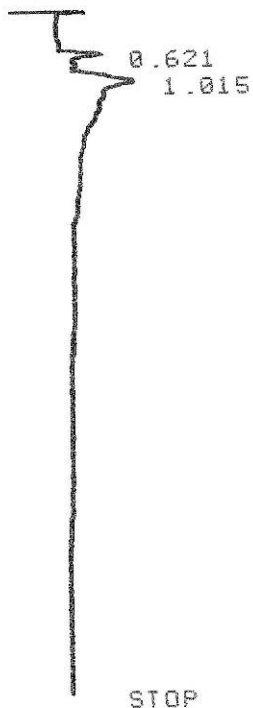
84C

UN# 17 JAN 1, 1901 05:11:05

REAR	RT	AREA	TYPE	WIDTH	AREAX
	.621	2649	PU	.068	20.69531
	.790	878	UU	.082	6.85938
	1.010	9273	UU	.215	72.44531

TOTAL AREA= 12800
JL FACTOR=1.0000E+00

* RUN # 19 JAN 1, 1901 05:33:41
START



85C

RUN# 19 JAN 1, 1901 05:33:41

AREA#	RT	AREA	TYPE	WIDTH	AREA%
	.621	2889	PU	.079	22.40926
	1.015	9726	UU	.190	77.59075

*. RUN # 9 JAN 1, 1901 03:19:49
START

0.625
1.025

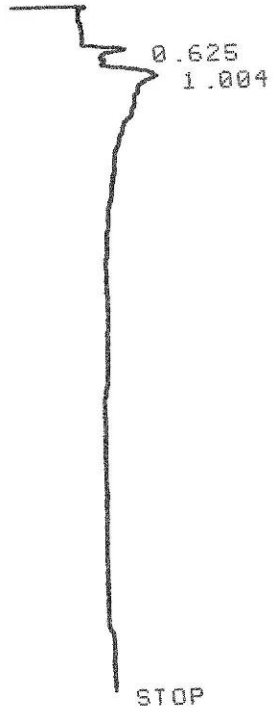
85A

STOP

RUN# 9 JAN 1, 1901 03:19:49

AREA#	RT	AREA	TYPE	WIDTH	AREA#
	.625	3793	UU	.086	30.23515
	1.025	8752	PU	.227	69.76483

* RUN # 18 JAN 1, 1901 05:22:23
START

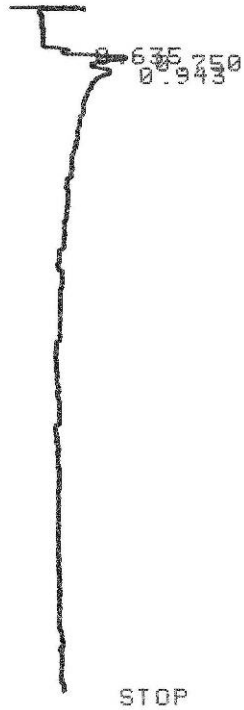


85B

RUN# 18 JAN 1, 1901 05:22:23

AREA%	RT	AREA	TYPE	WIDTH	AREA%
.625		3384	BU	.098	19.79990
1.004		13707	UU	.247	80.20010

* RUN # 14 JAN 1, 1901 05:41:20
START



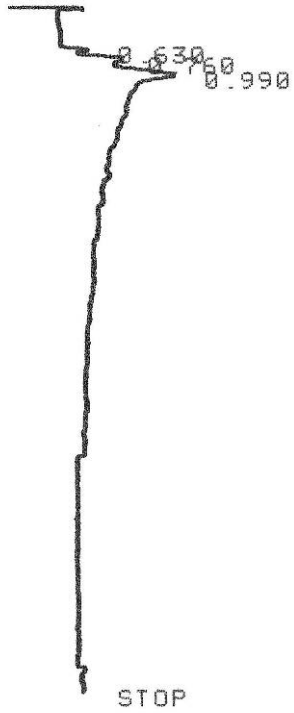
91B

RUN# 14 JAN 1, 1901 05:41:20

RT	AREA	TYPE	WIDTH	AREA%
.635	1430	PU	.068	5.28905
.750	8145	UU	.118	30.12538
.943	17462	UU	.339	64.58557

* RUN # 15 JAN 1, 1901 05:52:39
START

91C



RUN# 15 JAN 1, 1901 05:52:39

AREA%	RT	AREA	TYPE	WIDTH	AREA%
	.630	1299	PU	.060	5.43902
	.760	4550	UU	.109	19.05121
	.990	18034	UU	.255	75.50976

* RUN # 4 JAN 1, 1901 02:00:12
 START

0.627
 0.753
 0.934

91A

1.740

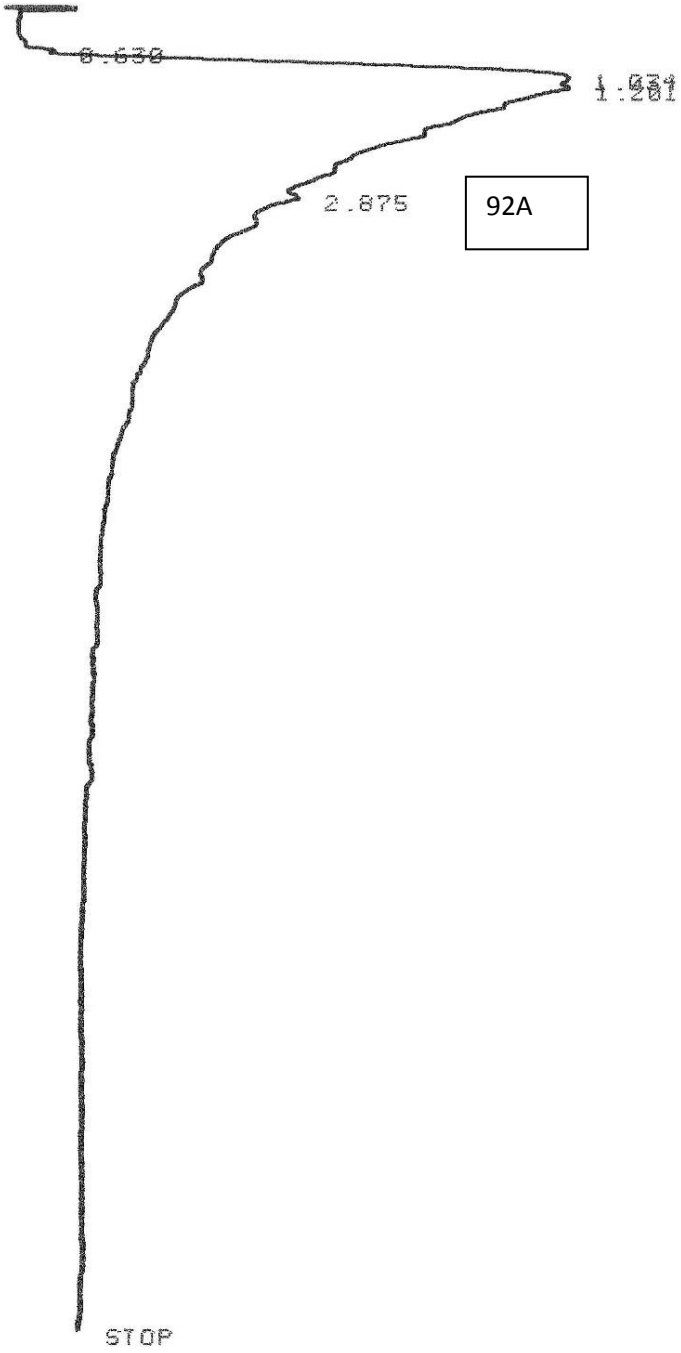
17.476

STOP

RUN# 4 JAN 1, 1901 02:00:12

AREA#	RT	AREA	TYPE	WIDTH	AREA#
	.627	1608	BU	.063	10.68154
	.753	3615	UU	.109	24.01355
	.934	7973	UU	.190	52.96267
	1.740	1061	UU	.088	7.04796
	17.476	797	FU	.125	5.29427

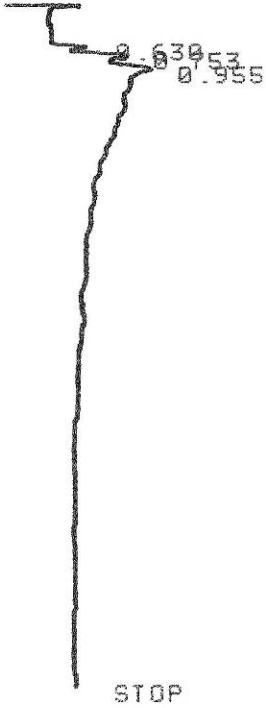
* RUN # 5 JAN 1, 1901 02:21:31
 START



RUN# 5 JAN 1, 1901 02:21:31

AREA#	RT	AREA	TYPE	WIDTH	AREA#
	.630	1331	PV	.058	.38160
	1.034	149635	UU	.321	42.90043
	1.201	138641	UU	.299	39.74845
	2.875	59189	UU	.326	16.96952

* RUN # 16 JAN 1, 1901 06:03:57
START

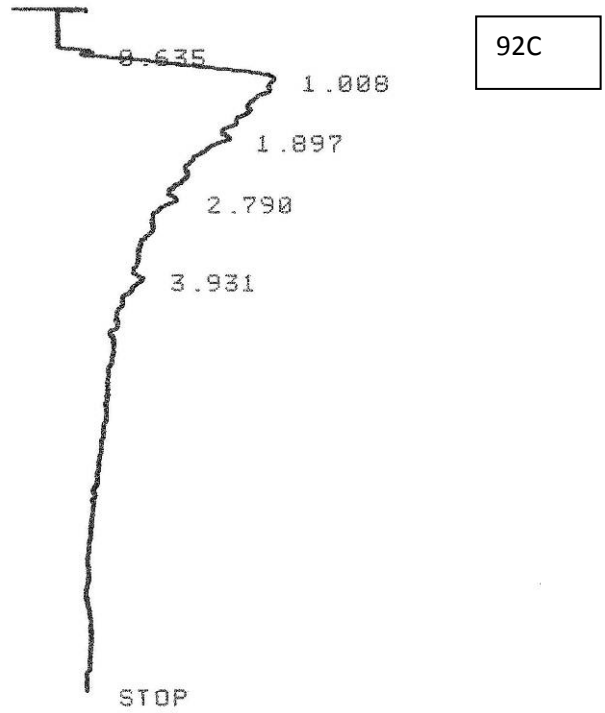


92B

RUN# 16 JAN 1, 1901 06:03:57

RT	AREA	TYPE	WIDTH	AREA%
.630	1382	PU	.056	7.62946
.753	4952	UU	.101	27.33797
.955	11780	UU	.212	65.03261

* RUN # 17 JAN 1, 1901 06:15:15
 START



RUN# 17 JAN 1, 1901 06:15:15

AREA%

RT	AREA	TYPE	WIDTH	AREA%
.635	1563	PU	.061	1.82538
1.000	52777	UU	.321	61.63666
1.897	27409	UU	.327	32.01013
2.790	2124	BP	.145	2.48056
3.931	1753	BU	.164	2.04728

TOTAL AREA= 85626
 MUL FACTOR=1.0000E+00

* RUN # 6 JAN 1, 1901 02:42:49
START

0.632
0.779
0.950
1.620

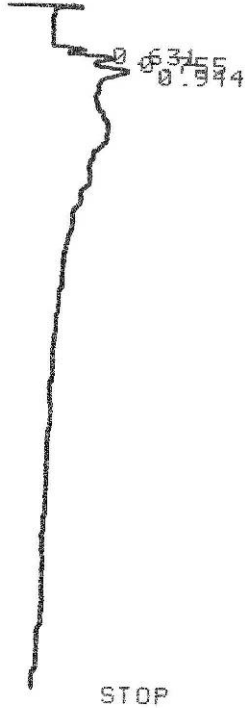
93A

STOP

RUN# 6 JAN 1, 1901 02:42:49

AREA#	RT	AREA	TYPE	WIDTH	AREA%
	.632	1739	PU	.066	6.90792
	.779	3922	UU	.121	15.57957
	.950	12354	UU	.245	49.07445
	1.620	7159	UU	.265	28.43808

* RUN # 18 JAN 1, 1901 06:26:33
START

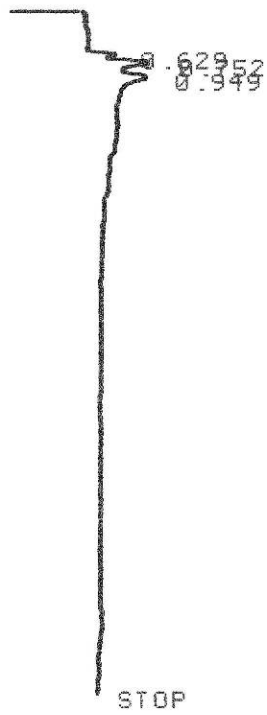


93B

RUN# 18 JAN 1, 1901 06:26:33

RT	AREA	TYPE	WIDTH	AREA%
.631	1535	PU	.061	8.90423
.755	4963	UU	.110	28.78938
.944	10741	UU	.202	62.30640

* RUN # 19 JAN 1, 1901 06:37:51
START

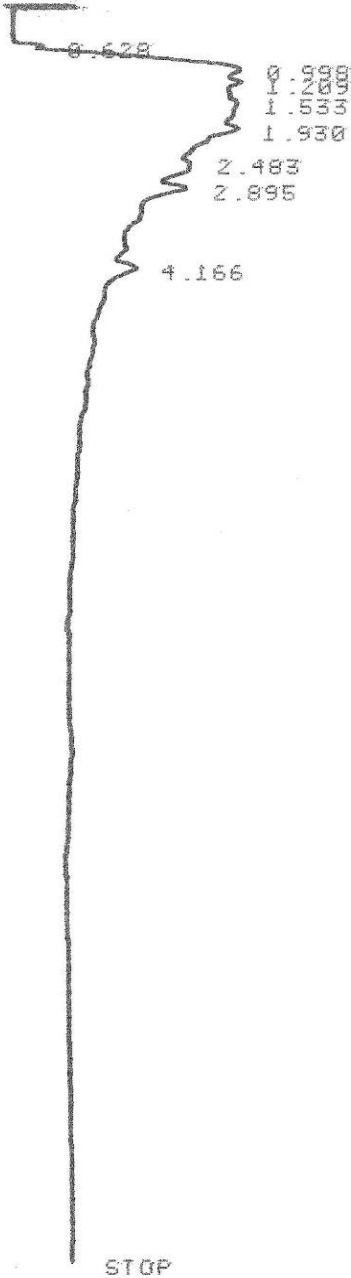


93C

RUN# 19 JAN 1, 1901 06:37:51

AREA%	RT	AREA	TYPE	WIDTH	AREA%
	.629	1488	BU	.067	10.33549
	.752	5473	UU	.115	38.01486
	.949	7436	UU	.190	51.64966

* RUN# 7 JAN 1, 1901 03:04:08
 START



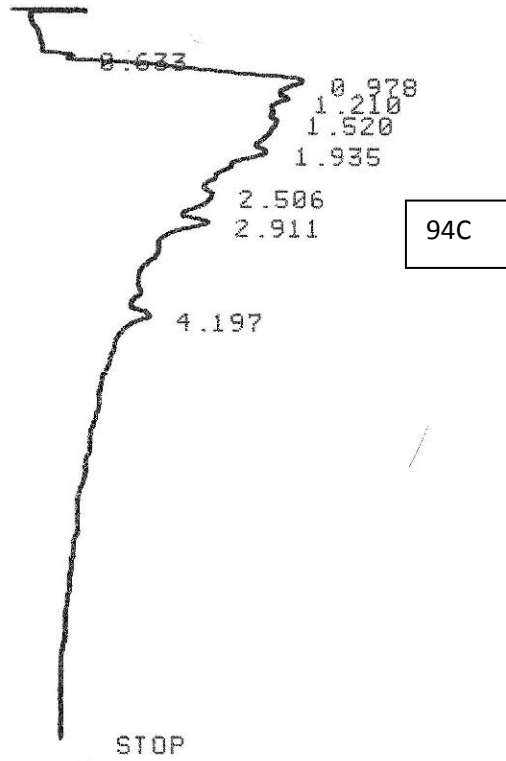
94A

RUN# 7 JAN 1, 1901 03:04:08

AREA#	RT	AREA	TYPE	WIDTH	AREA#
	.628	1302	PU	.059	.63439
*	.998	58463	UU	.320	28.48561
*	1.209	32940	UU	.189	16.04974
	1.533	35791	UU	.226	17.43886
o	1.930	37261	UU	.263	18.15511
*	2.483	22799	UU	.299	11.10862
-	2.895	12094	UU	.210	5.89270
	4.166	4587	UP	.202	2.23498

TOTAL AREA= 205237
 MUL FACTOR=1.0000E+00

* RUN # 21 JAN 1, 1901 07:00:27
 START

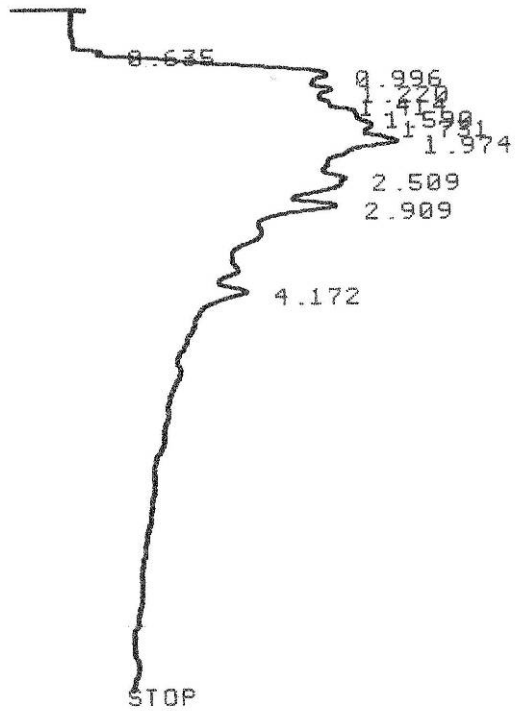


RUN# 21 JAN 1, 1901 07:00:27

AREA#	RT	AREA	TYPE	WIDTH	AREA#
	.633	2014	UU	.067	.55717
*	.978	73994	UU	.331	20.47047
*	1.210	40053	UU	.193	11.08068
	1.520	74525	UU	.381	20.61737
*	1.935	64629	UU	.355	17.87964
*	2.506	46961	UU	.367	12.99178
*	2.911	38915	UU	.325	10.76585
	4.197	20376	UU	.360	5.63703

TOTAL AREA= 361467
 MUL FACTOR=1.0000E+00

* RUN # 20 JAN 1, 1981 06:49:09
 START

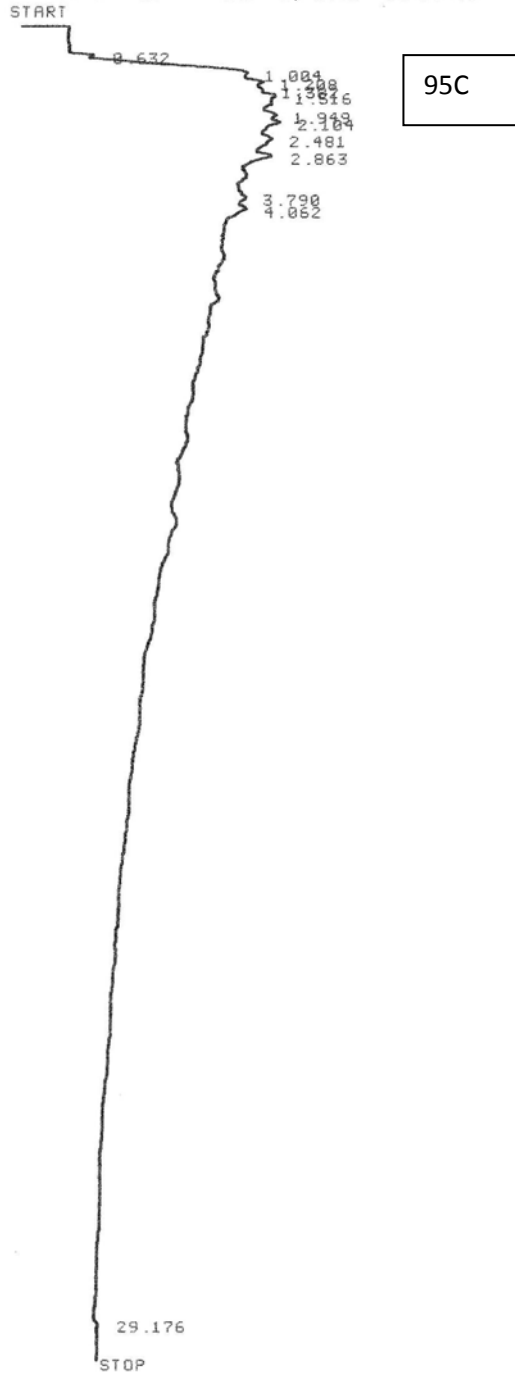


94B

RUN# 20 JAN 1, 1981 06:49:09

AREA#	RT	AREA	TYPE	WIDTH	AREA%
	.635	1395	PU	.059	.30878
*	.996	69447	UU	.327	15.37186
*	1.220	38572	UU	.181	8.53779
	1.414	26445	UU	.127	5.85351
	1.590	35579	UU	.155	7.87529
	1.731	53929	UU	.224	11.93700
*	1.974	93547	UU	.363	20.70631
*	2.509	35908	UU	.174	7.94812
*	2.909	56414	UU	.291	12.48705
	4.172	40544	UU	.409	8.97428

TOTAL AREA= 451780
 MUL FACTOR=1.0000E+00



RUN# 23 JAN 1, 1901 08:09:48

RT	AREA	TYPE	WIDTH	AREA%
.632	1376	PU	.058	.50078
*1.004	51772	UU	.268	18.84194
*1.208	38807	UU	.198	14.12345
*1.382	25646	UU	.141	9.33363
*1.516	32428	UU	.174	11.80197
1.949	31282	UU	.209	11.38480
2.104	37671	UU	.267	13.71002
2.481	33952	UU	.344	12.35652
2.863	15166	UU	.229	5.51953
3.790	1321	PU	.098	.48077
4.062	4495	UP	.211	1.63591
29.176	854	PU	.115	.31081

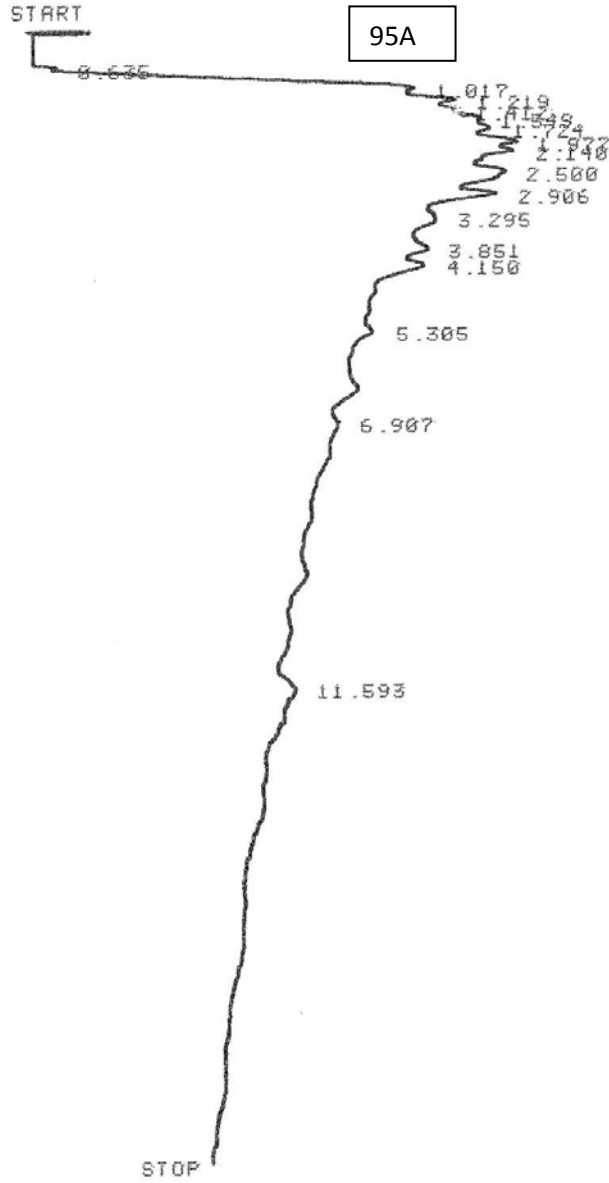
START

95B



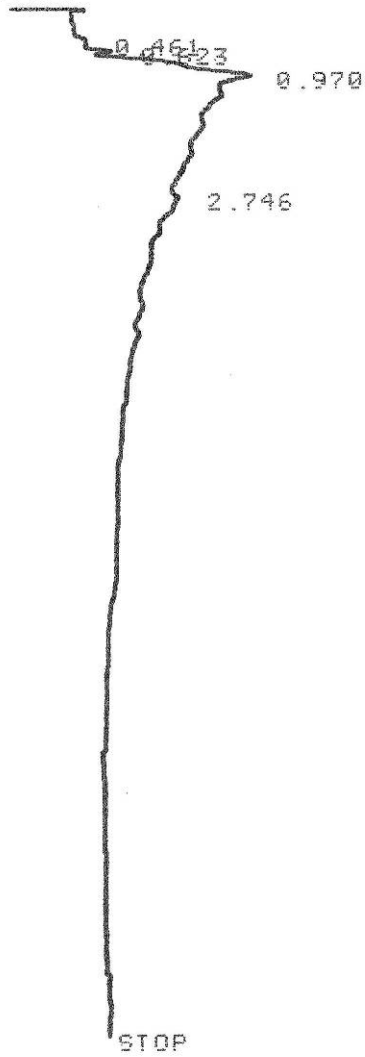
RUN# 22 JAN 1, 1901 07:38:30

RT	AREA	TYPE	WIDTH	AREAX
* .986	98965	UU	.204	12.28132
* 1.217	61815	UU	.183	7.67110
* 1.410	45216	UU	.140	5.61120
* 1.535	49300	UU	.148	6.11802
1.735	73506	UU	.220	9.12192
1.991	78255	UU	.241	9.71126
2.165	89706	UU	.282	11.13230
2.515	110274	UU	.385	13.68474
2.939	77313	UU	.314	9.59436
3.347	33907	UU	.208	4.20778
3.896	50301	UU	.438	6.24224
4.202	23494	UU	.291	2.91555
5.388	3312	UP	.211	.41101
6.164	2318	BU	.164	.28766
7.036	2550	PU	.276	.31645
11.799	5585	UU	.355	.69309



RUN#	0	JAN	1,	1901	03:25:26
AREA#	RT	AREA	TYPE	WIDTH	AREA%
	.635	1115	PU	.055	.12468
*	1.017	90476	UU	.256	10.11672
*	1.219	80793	UU	.214	9.03399
*	1.412	49872	UU	.139	5.57652
*	1.549	51867	UU	.138	5.79959
	1.724	70189	UU	.190	7.84829
	1.977	87812	UU	.235	9.81884
	2.140	97895	UU	.274	10.94628
	2.500	125366	UU	.396	14.01799
	2.906	79023	UU	.290	8.83608
	3.295	75921	UU	.433	8.48922
	3.851	46349	UU	.389	5.18259
	4.150	25742	UU	.290	2.87838
	5.305	3569	UP	.238	.39907
	6.907	3773	PU	.306	.42188
	11.593	4560	BU	.250	.50988

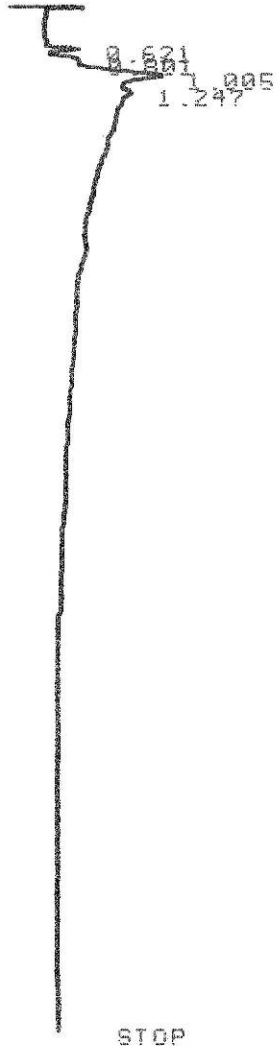
* RUN # 3 JAN 1, 1901 00:21:20
 START



RUN# 3 JAN 1, 1901 00:21:20

AREA#	RT	AREA	TYPE	WIDTH	AREA#
	.461	1279	PU	.134	2.18349
	.623	1801	UU	.065	3.07464
	.970	41595	UU	.301	71.01030
	2.746	13901	UU	.347	23.73156

* RUN # 4 JAN 1, 1901 00:37:38
 START



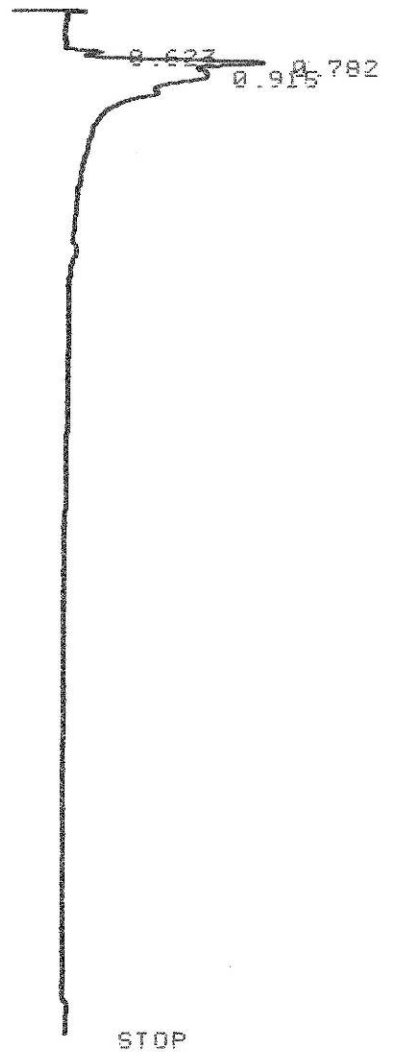
M10

RUN# 4 JAN 1, 1901 00:37:38

AREA%

RT	AREA	TYPE	WIDTH	AREA%
.621	1460	PP	.054	4.21576
.001	2628	PU	.109	7.58836
1.005	18794	UU	.210	54.26773
1.247	11750	UU	.209	33.92816
TOTAL	34632			

* RUN # 6 JAN 1, 1901 01:10:15
START

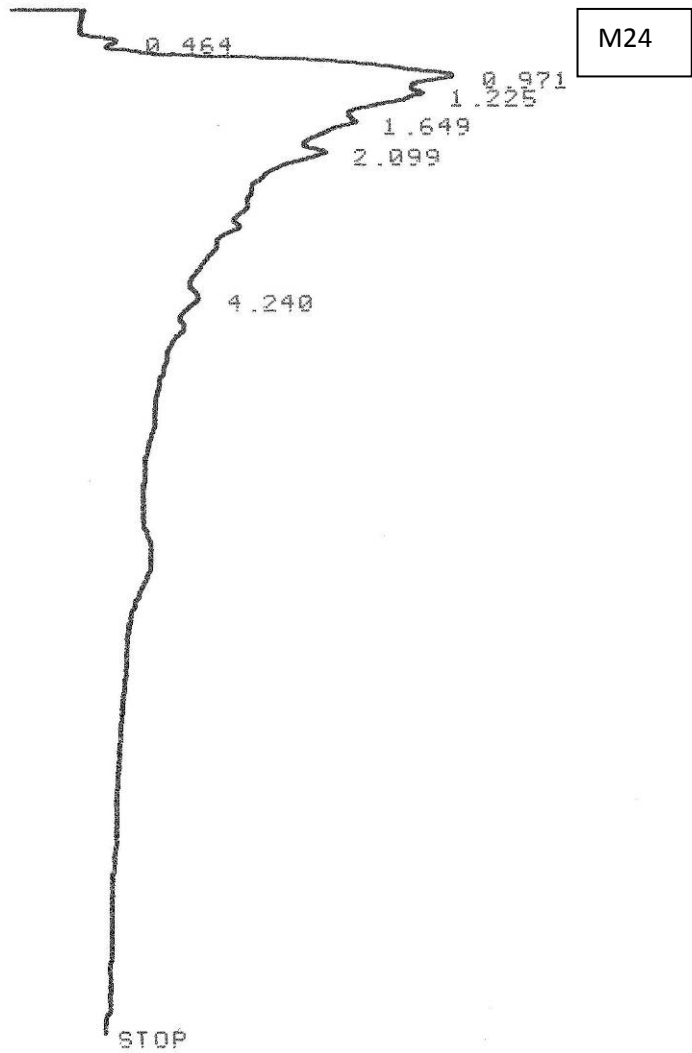


M19

RUN# 6 JAN 1, 1901 01:10:15

AREA#	RT	AREA	TYPE	WIDTH	AREA#
	.623	912	PP	.046	6.64965
	.782	10885	PU	.095	79.36566
	.915	1918	UB	.092	13.98469

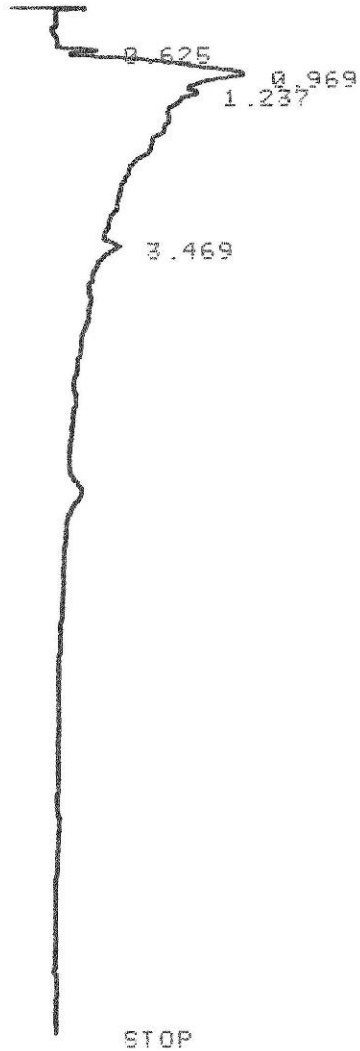
* RUN # 7 JAN 1, 1901 01:26:34
 START



RUN# 7 JAN 1, 1901 01:26:34

RT	AREA	TYPE	WIDTH	AREA%
.464	2716	PU	.111	.90998
.971	105911	UU	.375	35.48501
1.225	84371	UU	.348	28.26811
1.649	60388	UU	.373	20.23272
2.099	42017	UU	.377	14.07761
4.240	3064	PU	.235	1.02658

* RUN # 8 JAN 1, 1901 01:42:52
 START

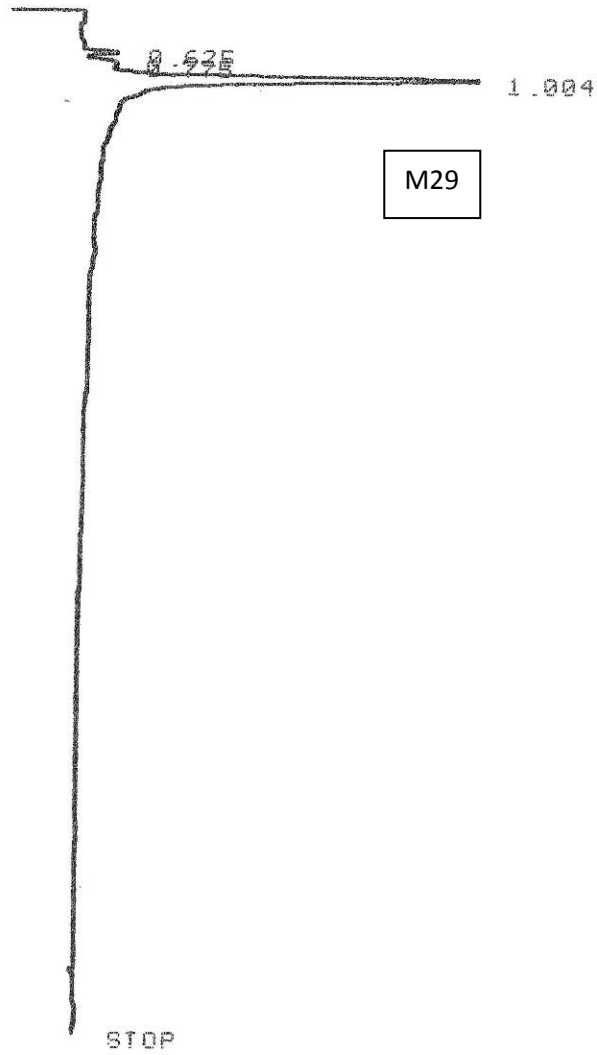


M28

RUN# 8 JAN 1, 1901 01:42:52

RT	AREA	TYPE	WIDTH	AREA%
.625	2134	UU	.061	2.01338
.969	55726	UU	.351	52.57618
1.237	34433	UU	.294	32.48672
3.469	13698	UU	.382	12.92374

* RUN # 9 JAN 1, 1901 01:59:11
START

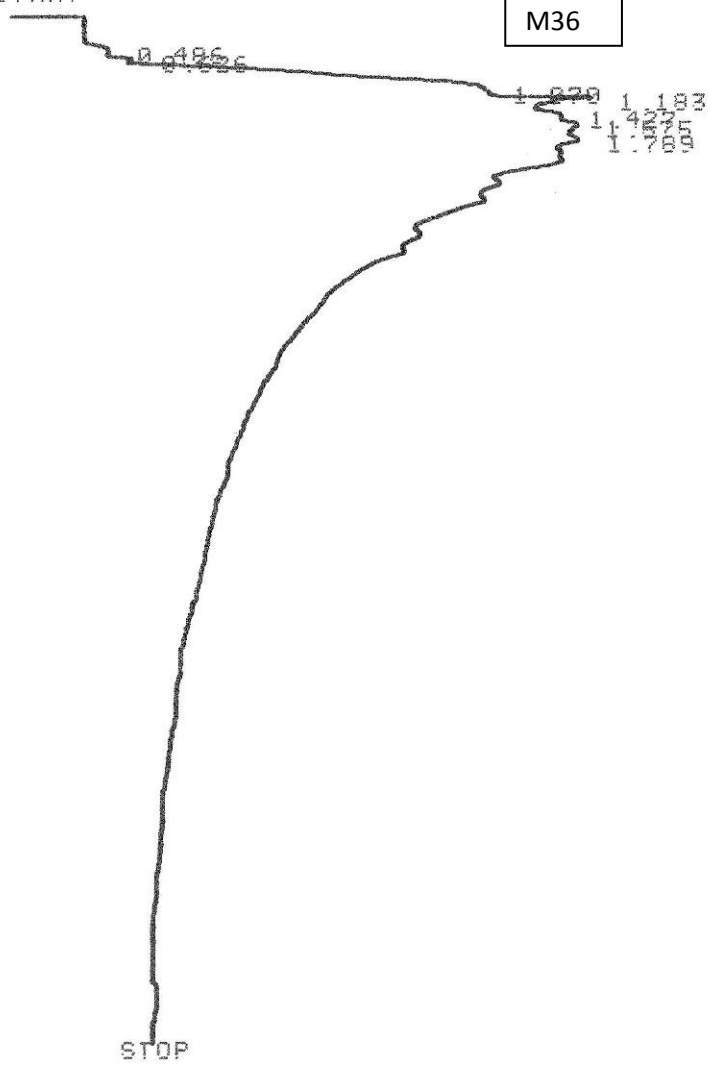


RUN# 9 JAN 1, 1901 01:59:11

AREA#	RT	AREA	TYPE	WIDTH	AREA#
	.625	1647	PP	.056	3.40300
	.775	2579	PU	.112	5.45520
	1.004	43050	UB	.133	91.06099

* RUN # 10 JAN 1, 1901 02:15:30
 START

M36

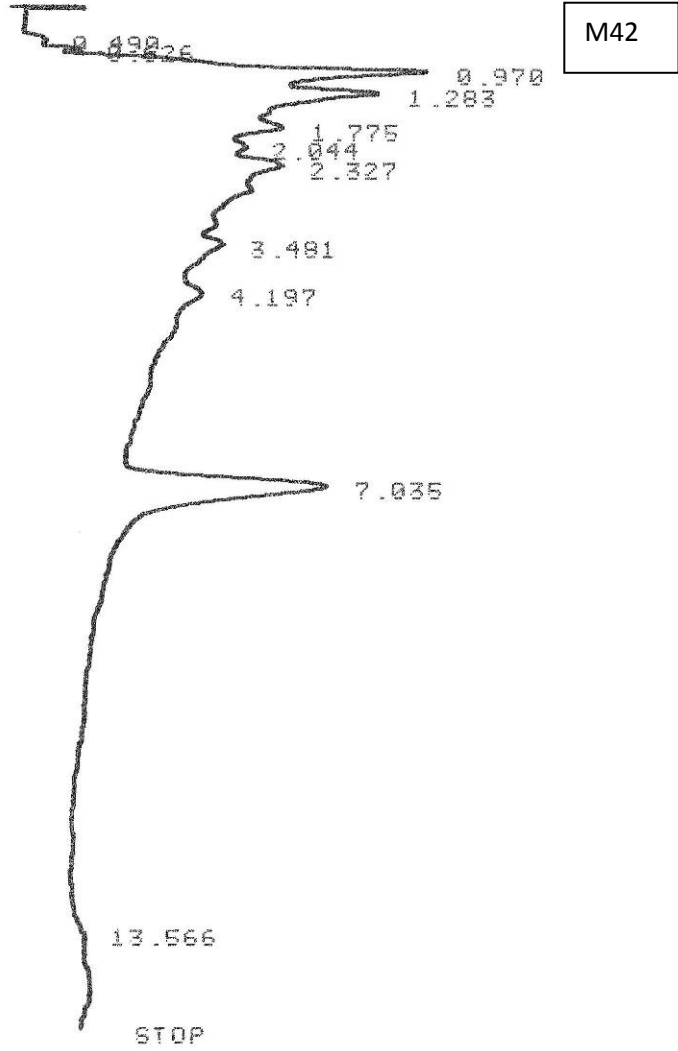


RUN# 10 JAN 1, 1901 02:15:30

RT	AREA	TYPE	WIDTH	AREA%
.486	1939	PU	.127	.53046
.626	1987	UU	.069	.54359
1.070	85965	UU	.275	23.51765
1.183	85004	UU	.214	23.25474
1.427	42033	UU	.116	11.49906
1.575	69639	UU	.189	19.05130
1.789	78967	UU	.218	21.60318

TOTAL AREA= 365534

* RUN # 11 JAN 1, 1901 02:31:51
 START

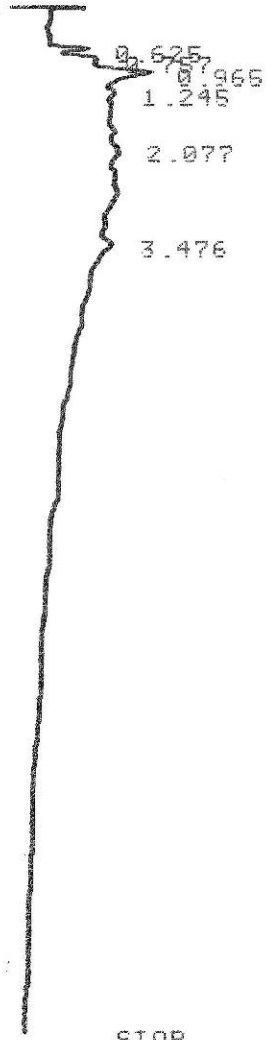


RUN# 11 JAN 1, 1901 02:31:51

AREA%

RT	AREA	TYPE	WIDTH	AREA%
.490	2356	BU	.133	.43952
.626	3317	UU	.074	.61879
.970	97445	UU	.290	18.17854
1.283	88092	UU	.303	16.43373
1.775	55699	UU	.277	10.39075
2.044	37226	UU	.226	6.94450
2.327	72377	UU	.378	13.50207
3.481	59477	UU	.482	11.09554
4.197	52236	UU	.550	9.74472
7.035	67407	BU	.384	12.57490
13.566	412	PU	.202	.07686

* RUN # 12 JAN 1, 1901 02:48:10
 START



M46

STOP

RUN# 12 JAN 1, 1901 02:48:10

RT	AREA	TYPE	WIDTH	AREA%
.625	1737	PU	.059	5.53572
.757	2927	UU	.095	9.32819
.965	13514	UU	.207	43.06838
1.245	3632	UU	.153	11.57499
2.077	3894	UU	.291	12.40997
3.476	5674	UU	.300	10.08274

```

* LIST: ATT 2^ = 0
* ATT 2^ 1 @
* RUN # 2 JAN 1, 1901 00:05:01
START

```

```

0.620
0.778

```

Blank

STOP

```

RUN# 2 JAN 1, 1901 00:05:01

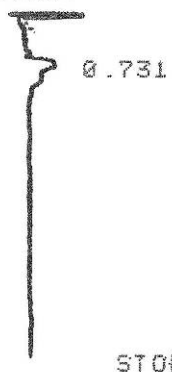
```

AREA%

RT	AREA	TYPE	WIDTH	AREA%
.620	2004	UP	.066	73.51430
.778	722	PV	.096	26.48569

TOTAL AREA= 7726

* RUN # 3 JAN 1, 1901 01:51:57
START

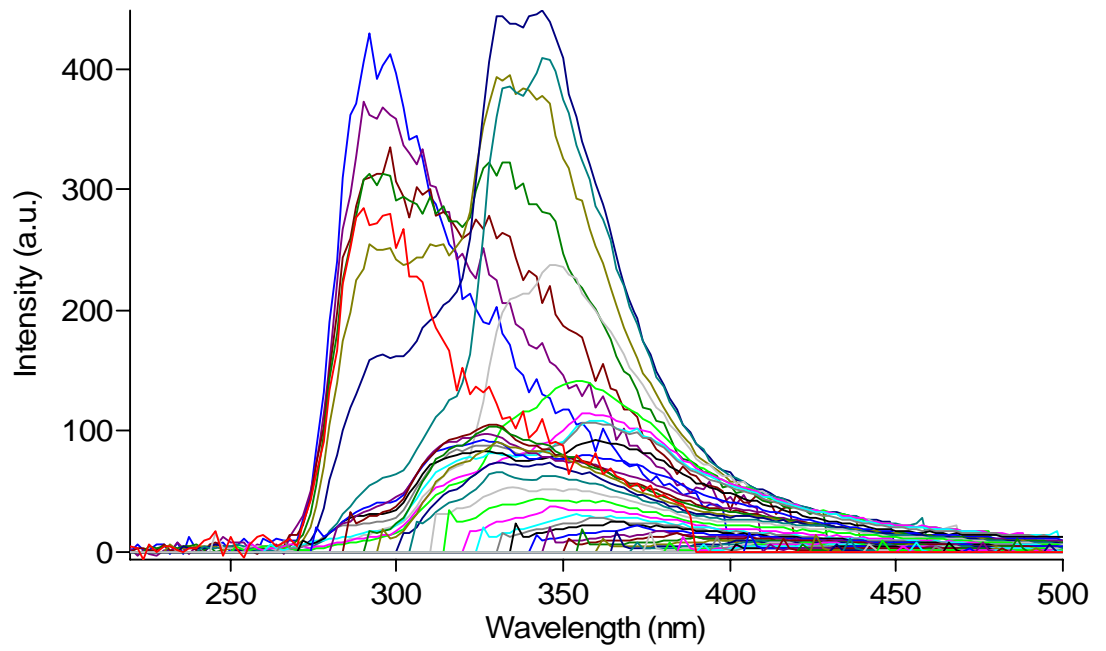
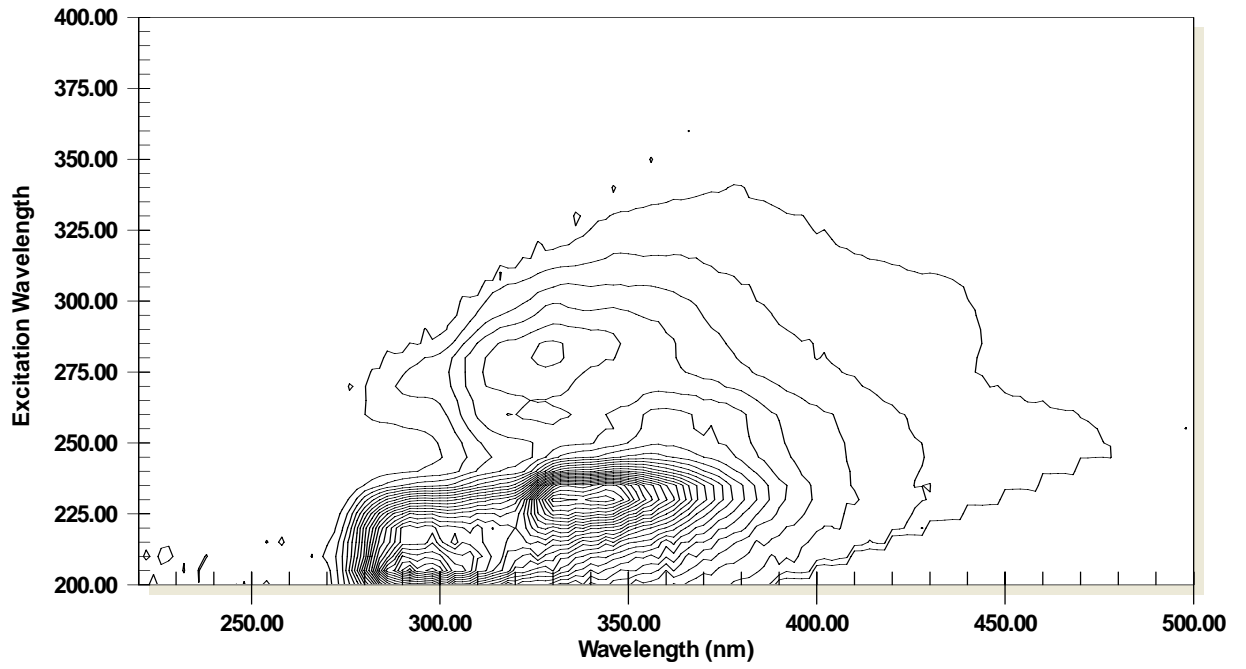


BLANK

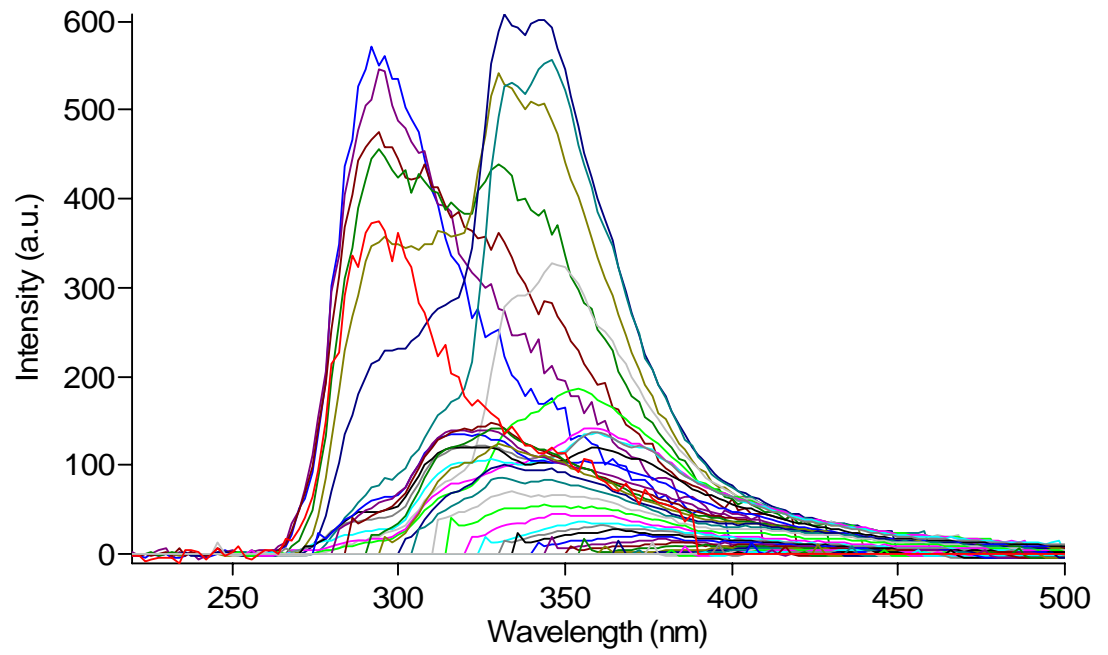
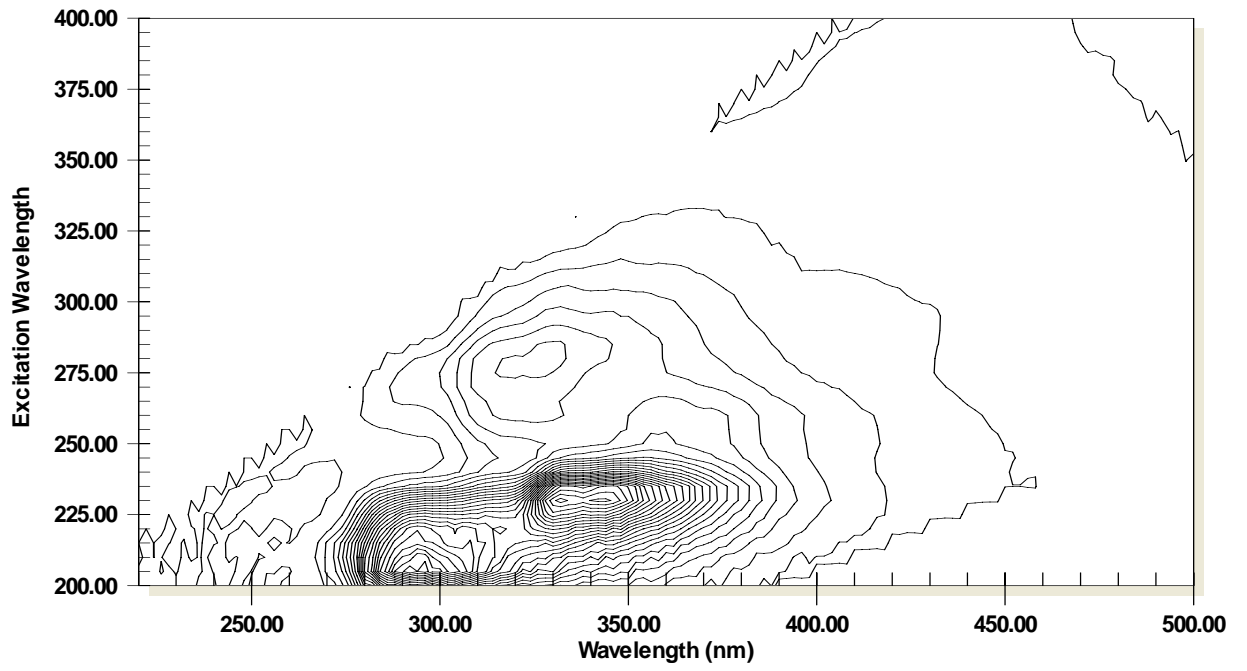
RUN# 3 JAN 1, 1901 01:51:57

AREA#	RT	AREA	TYPE	WIDTH	AREA%
	.731	5322	UU	.191	100.00000

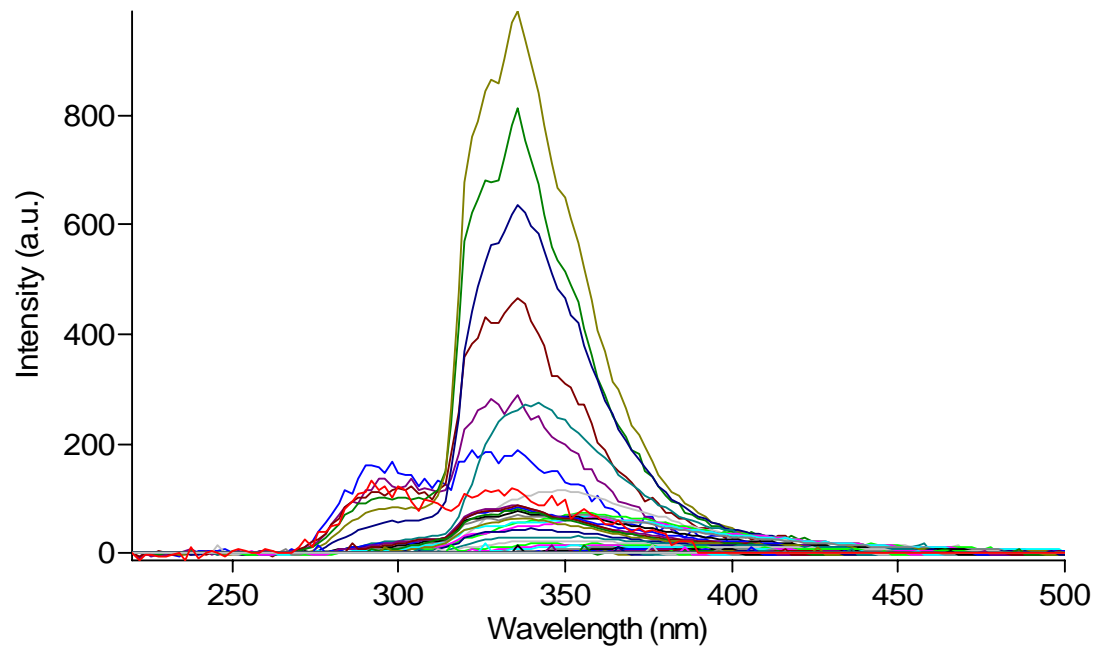
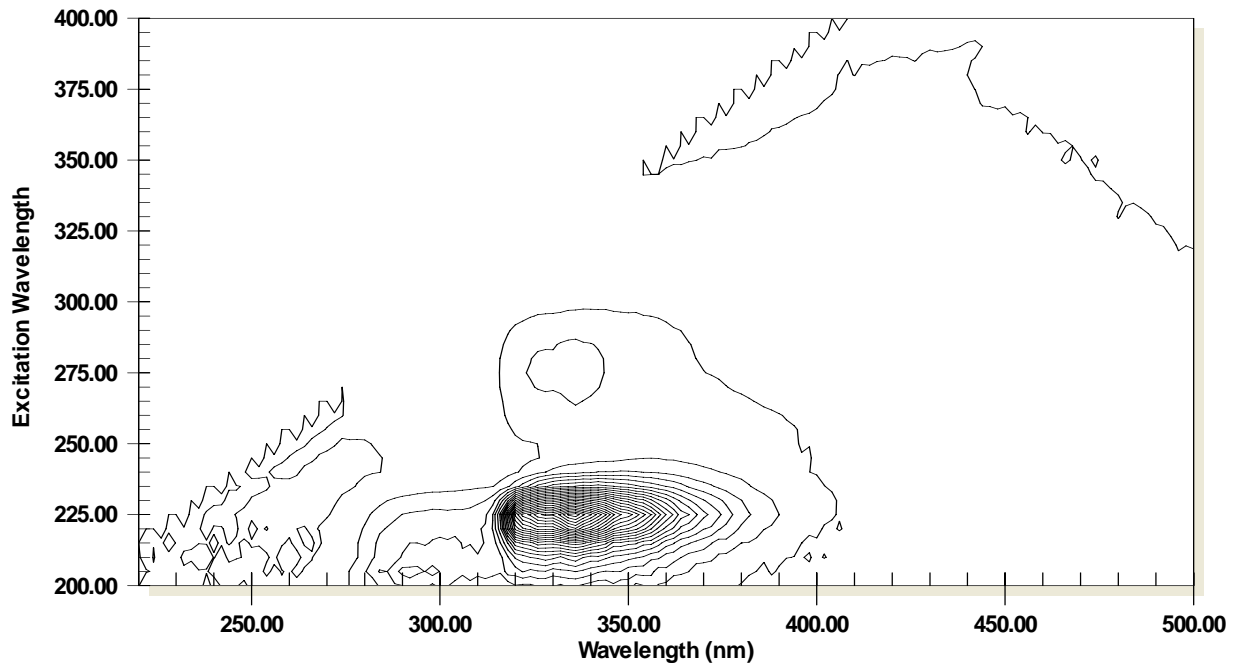
Pentane Contour Plots and Excitation Scan Overlays
11CP



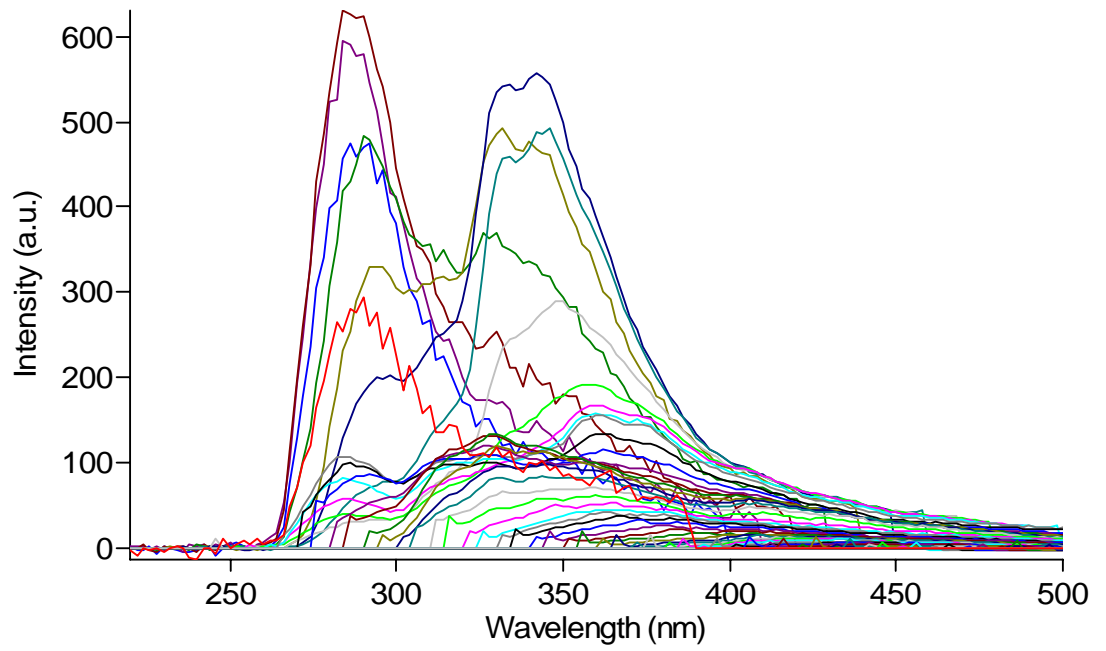
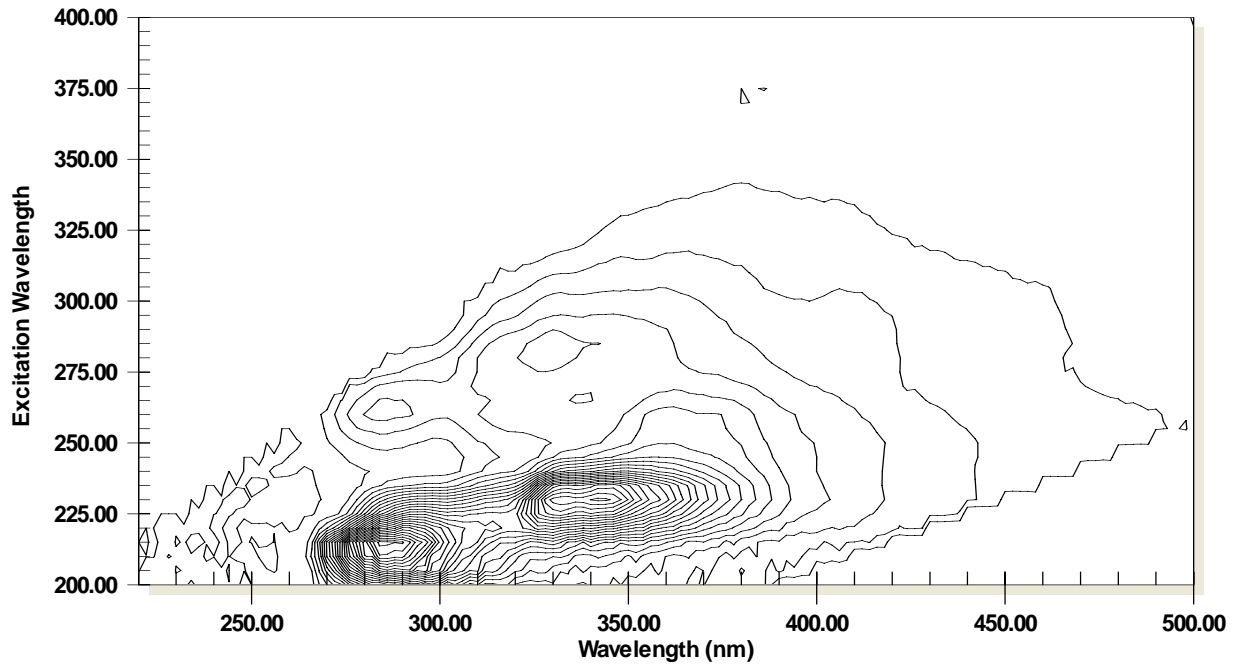
12CP



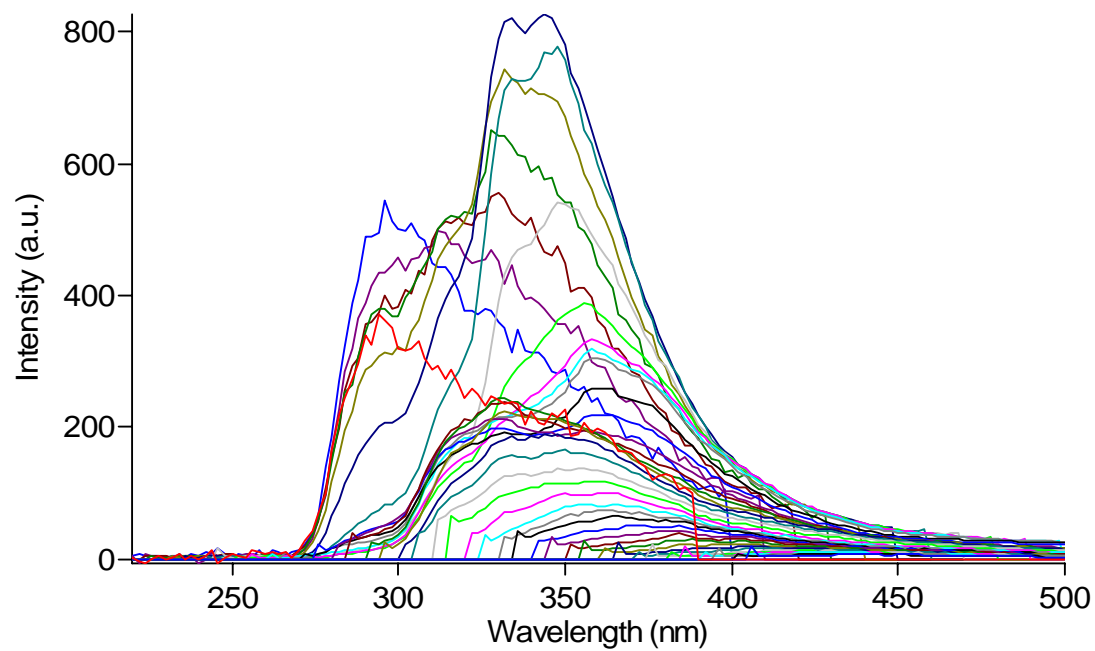
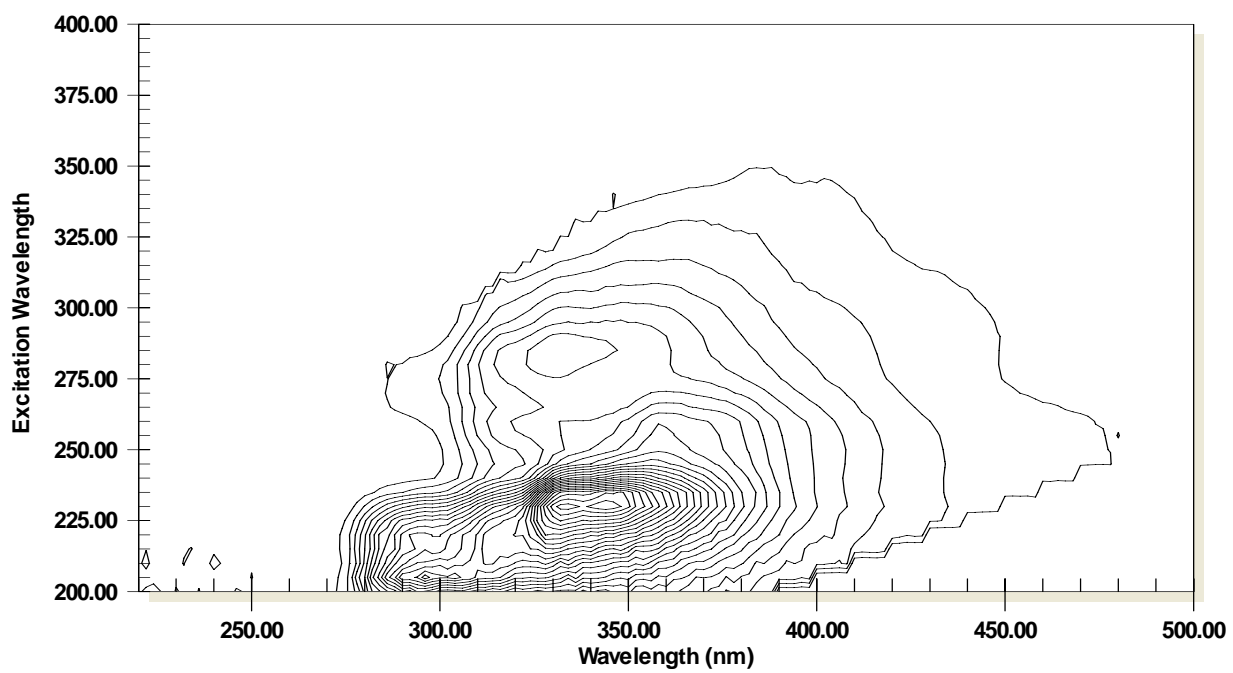
13BP



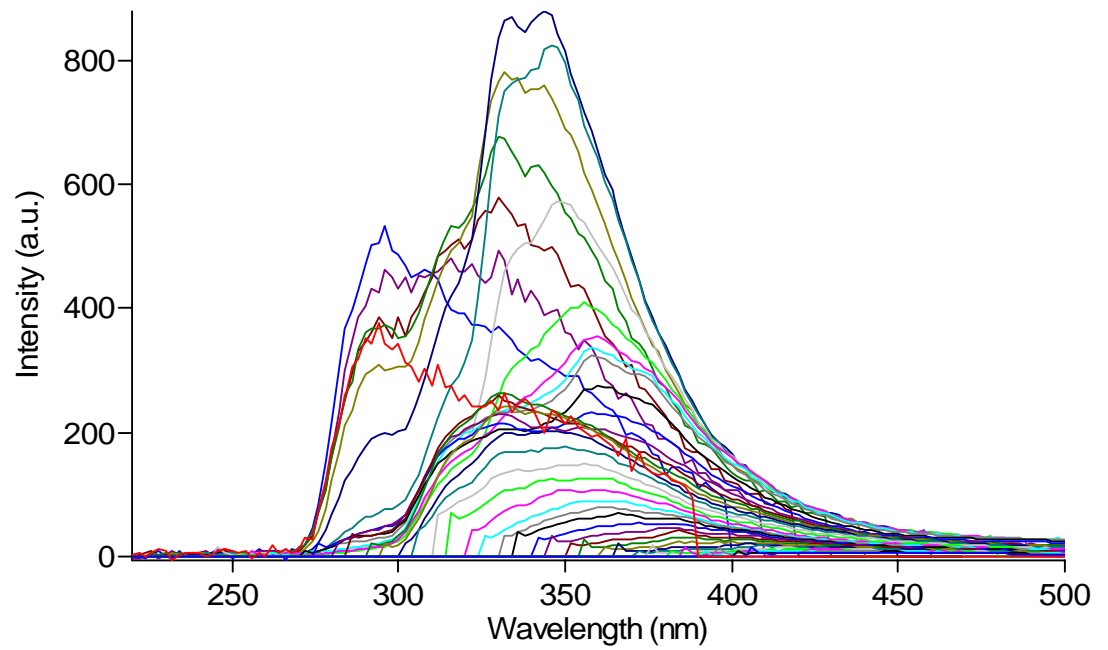
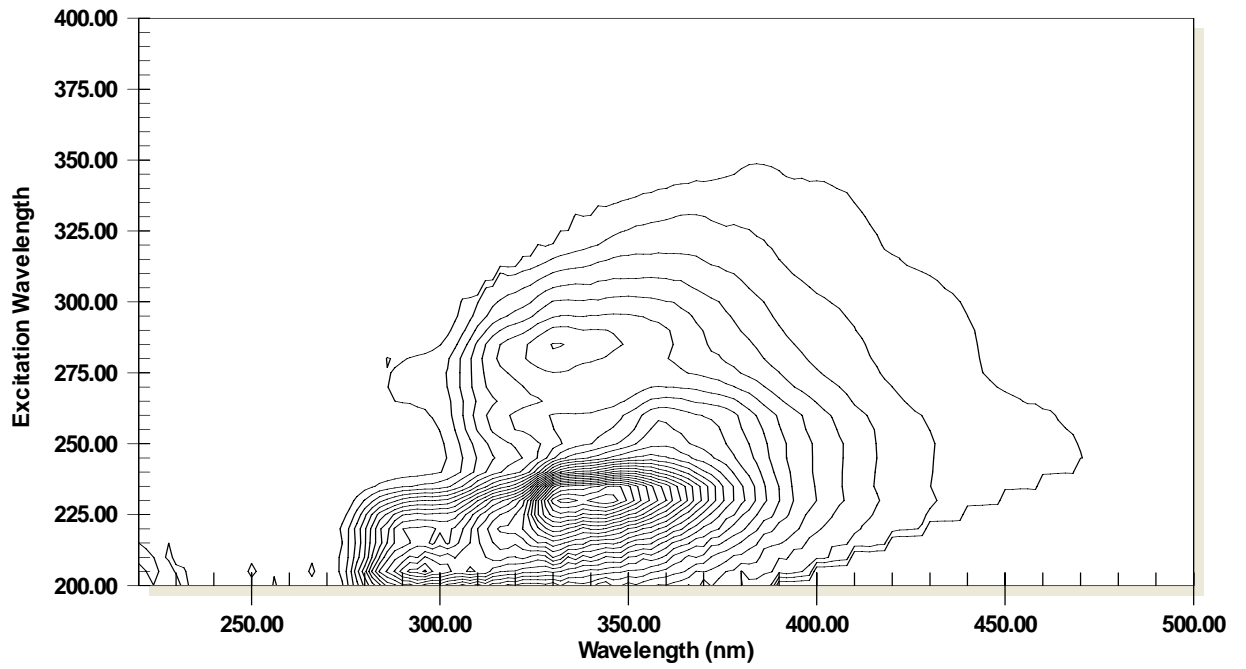
14BP



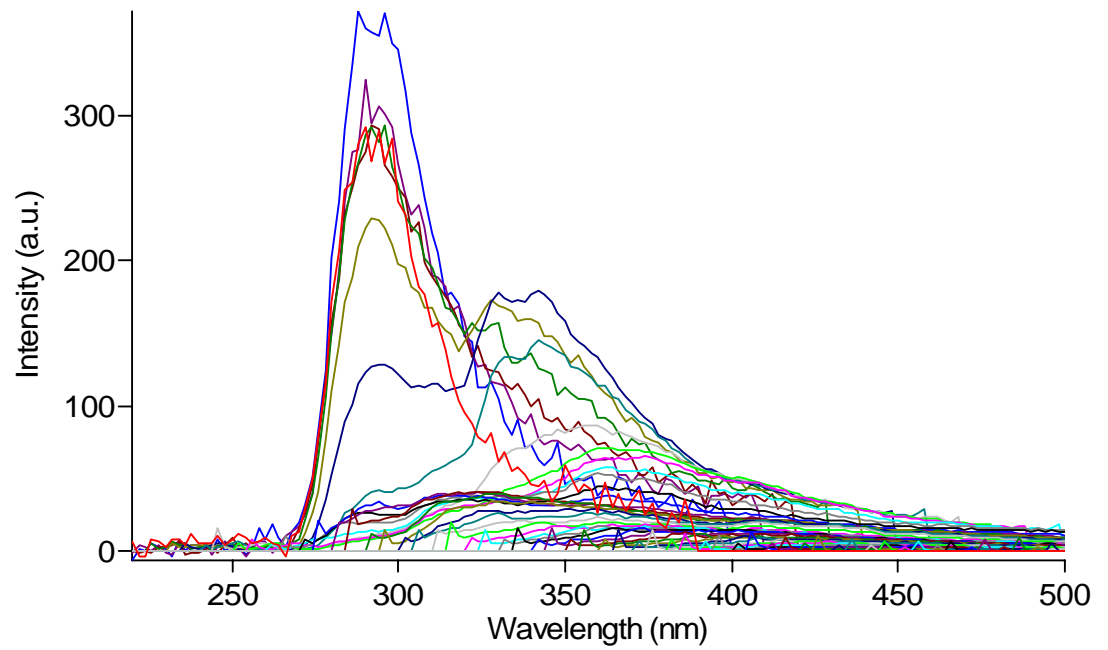
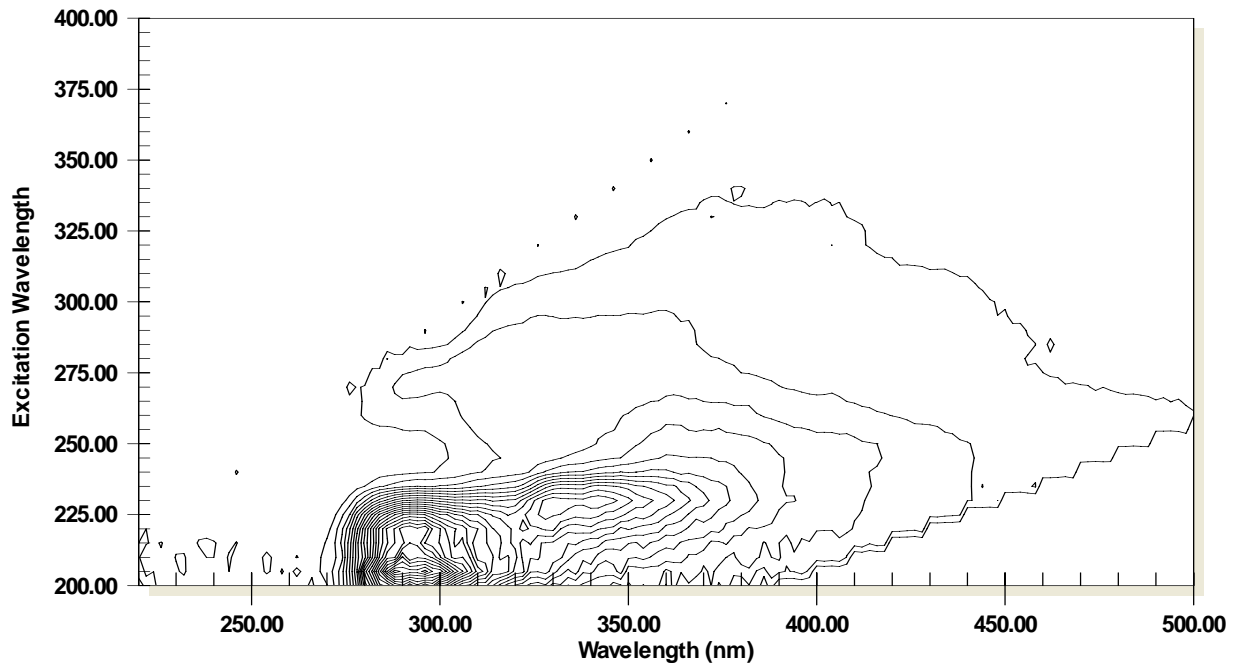
21BP



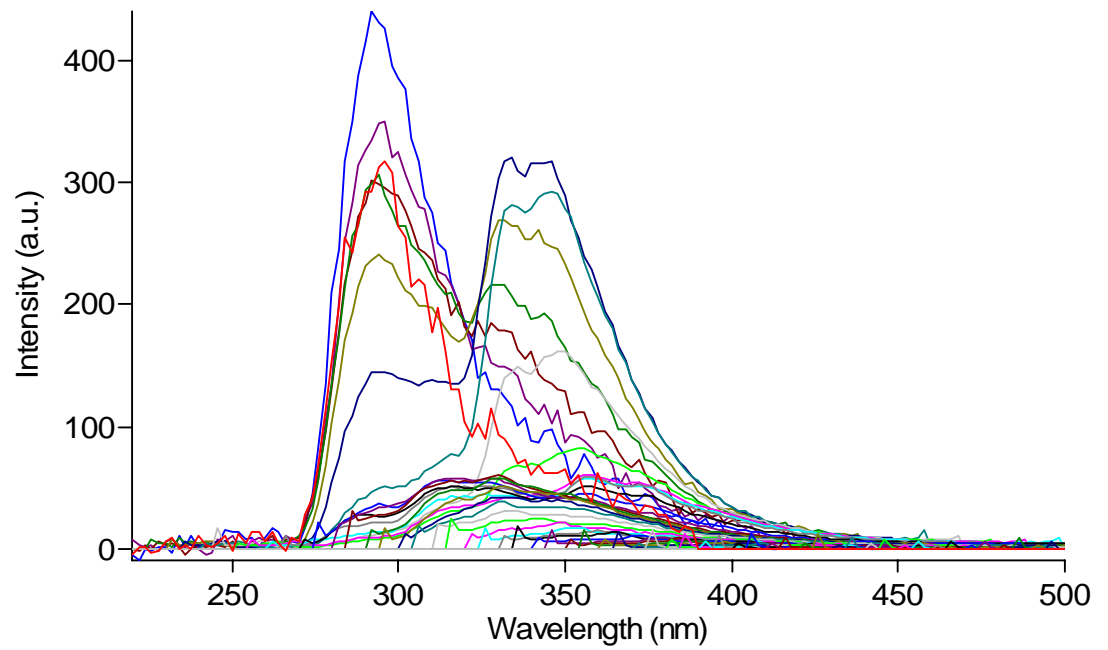
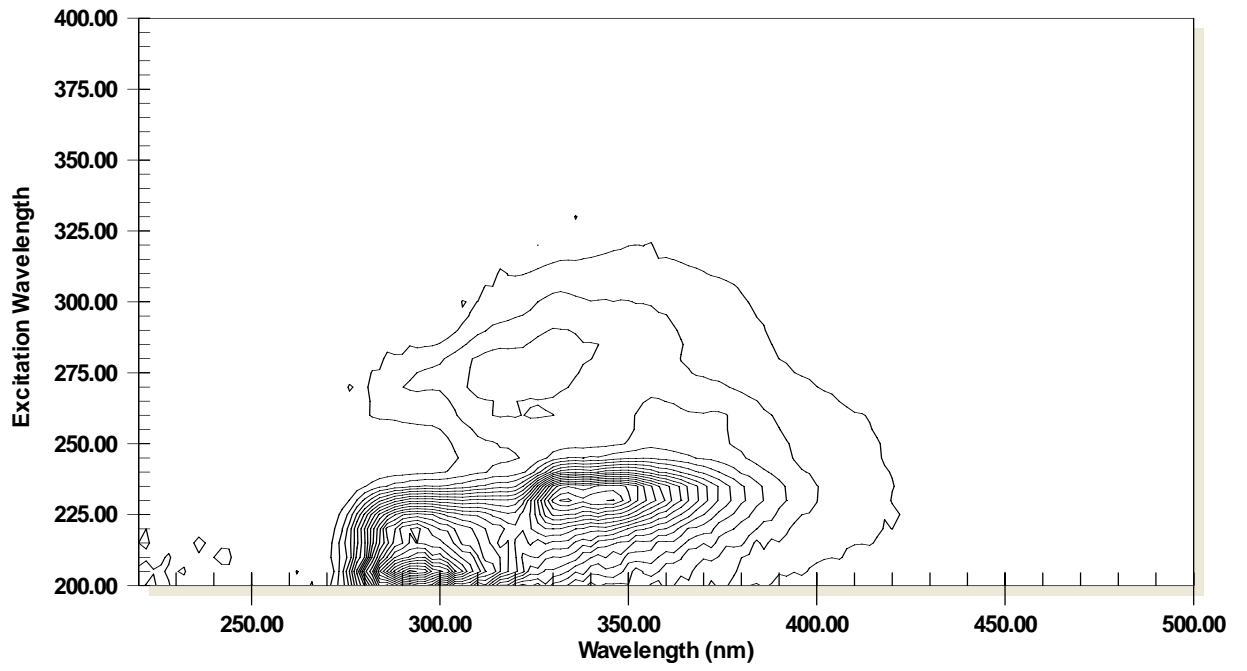
22AP



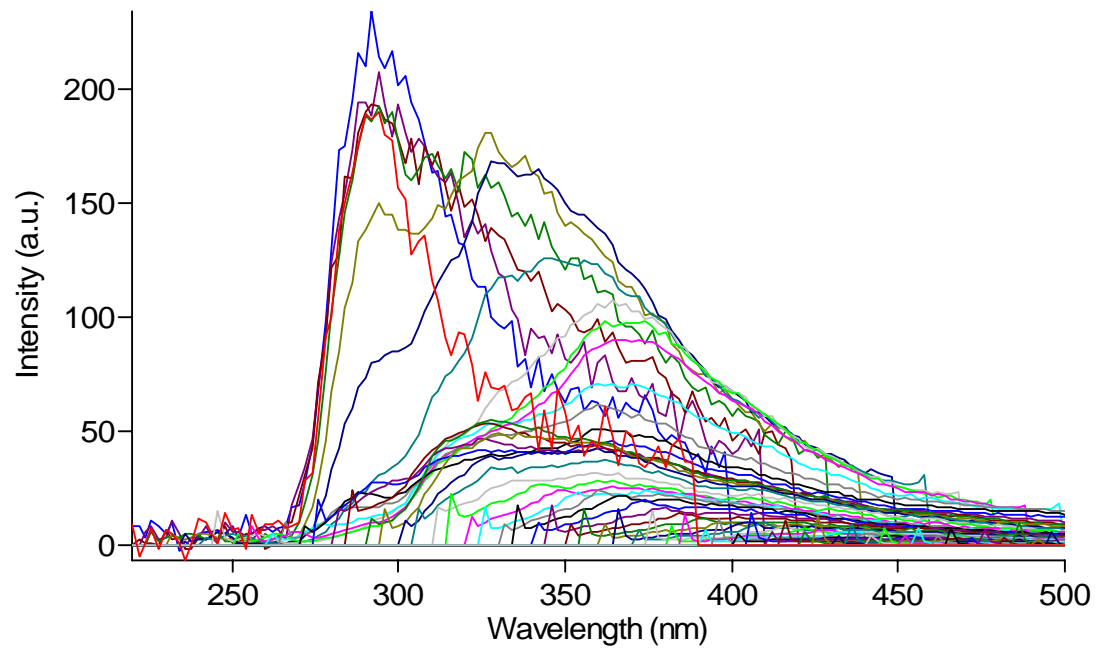
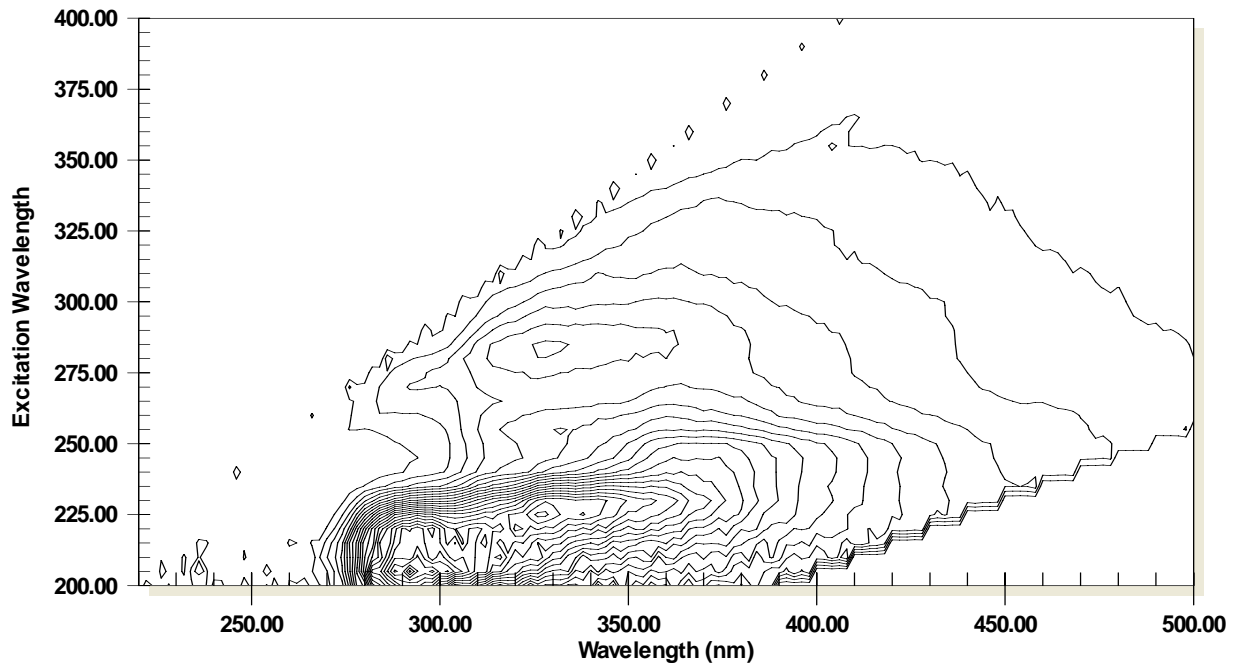
23AP



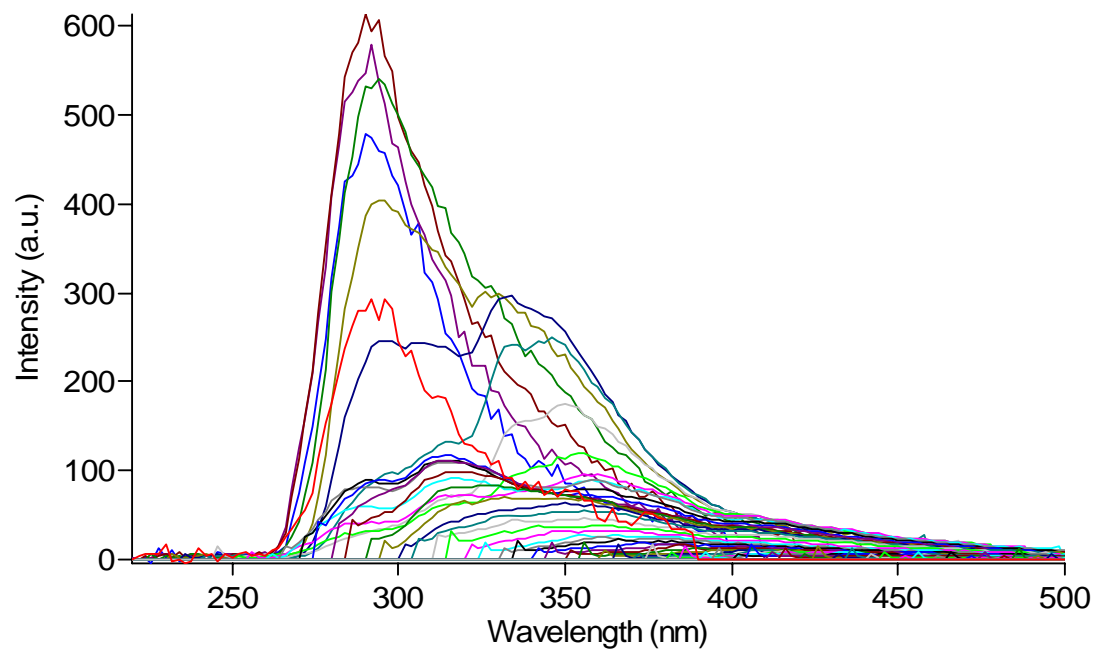
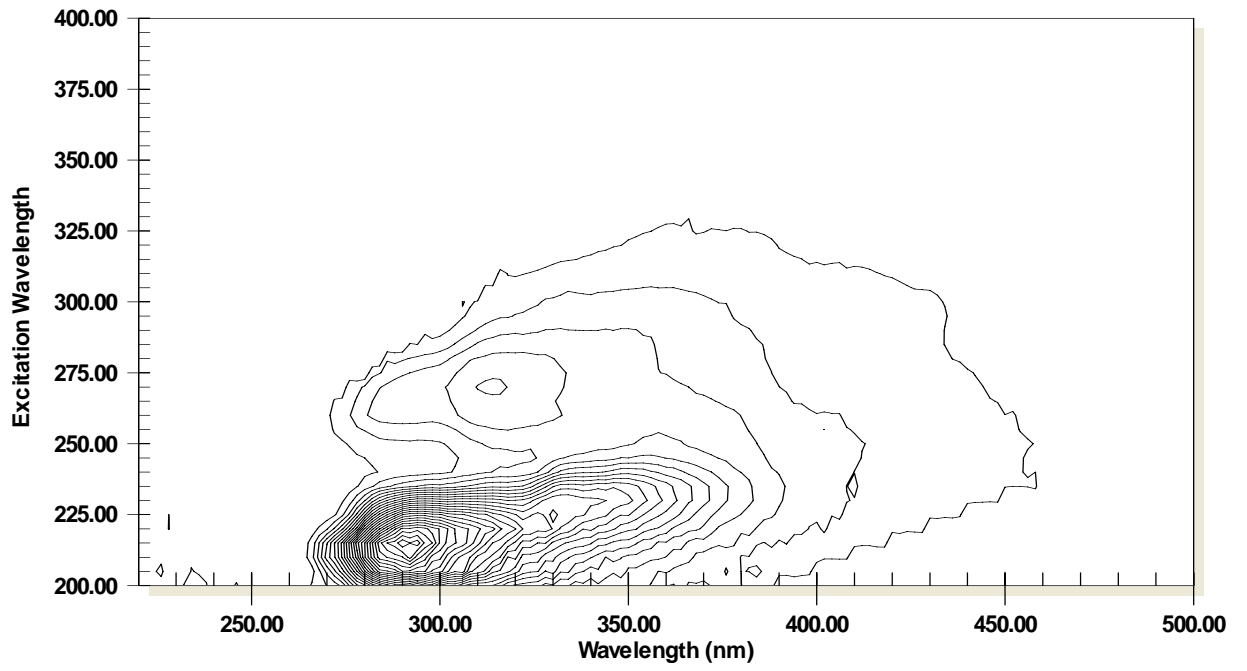
24CP

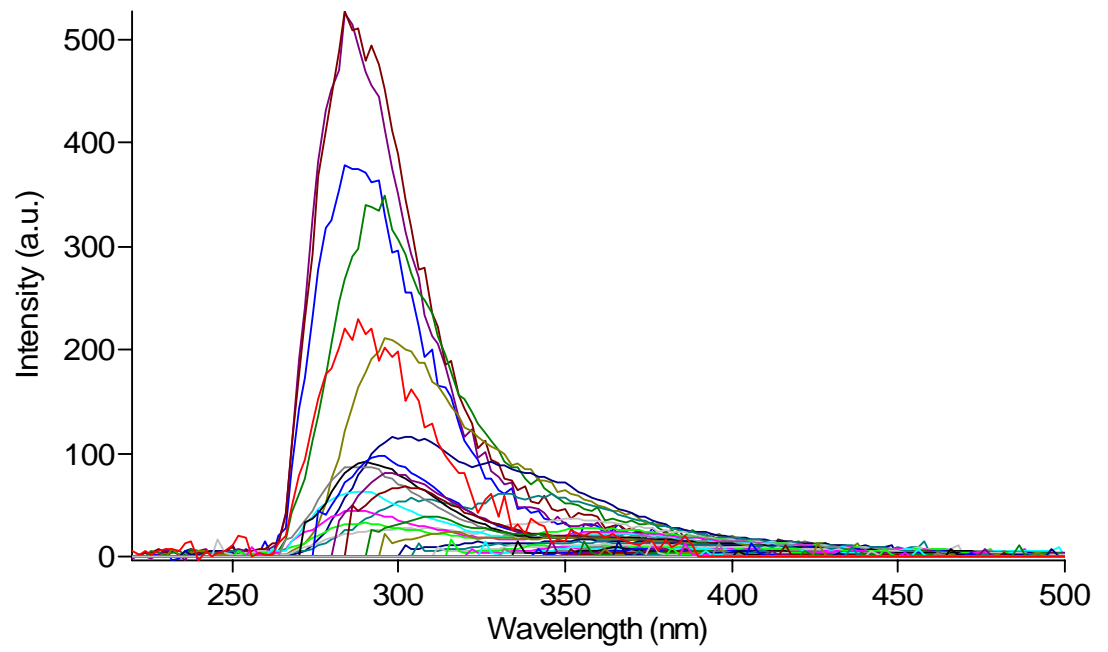
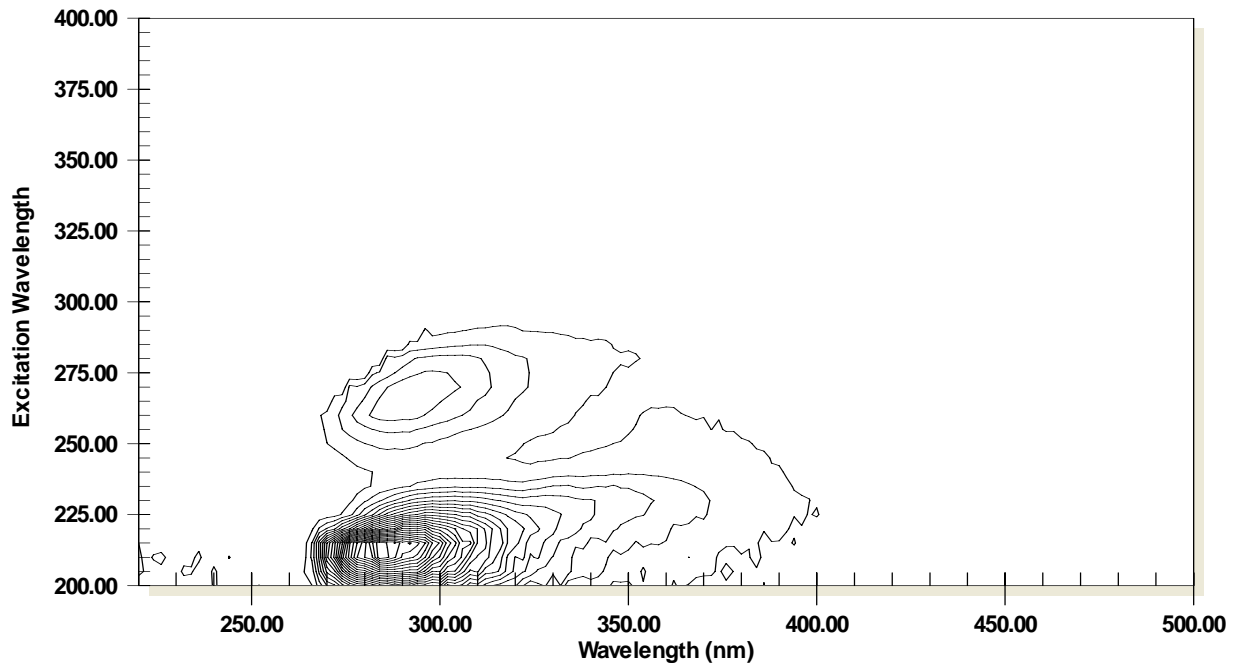


25CP

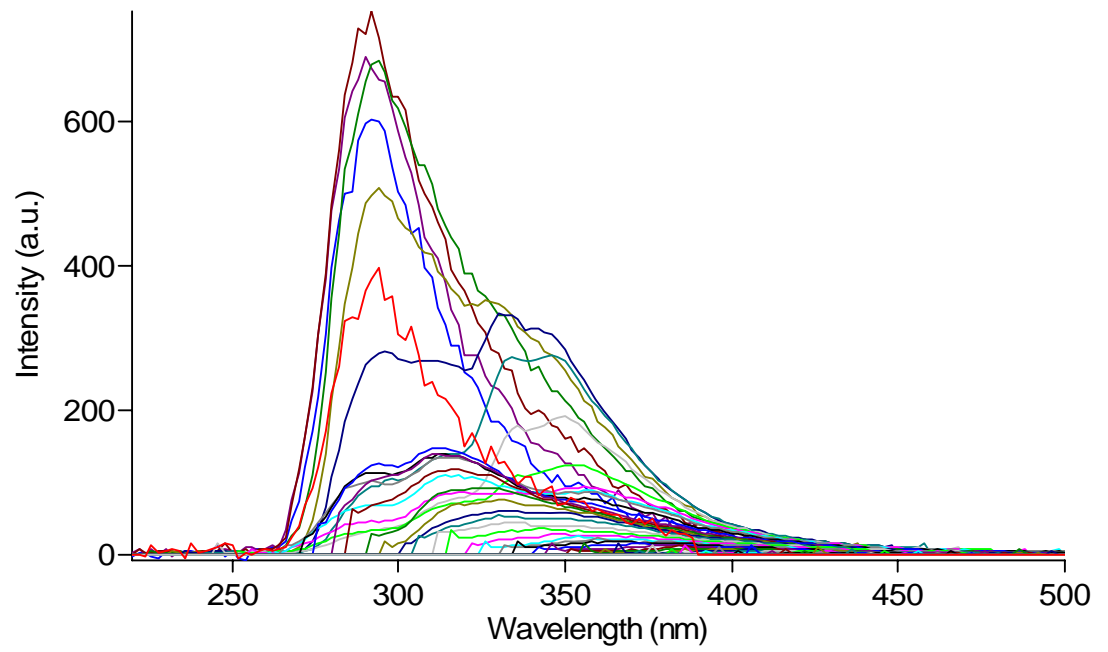
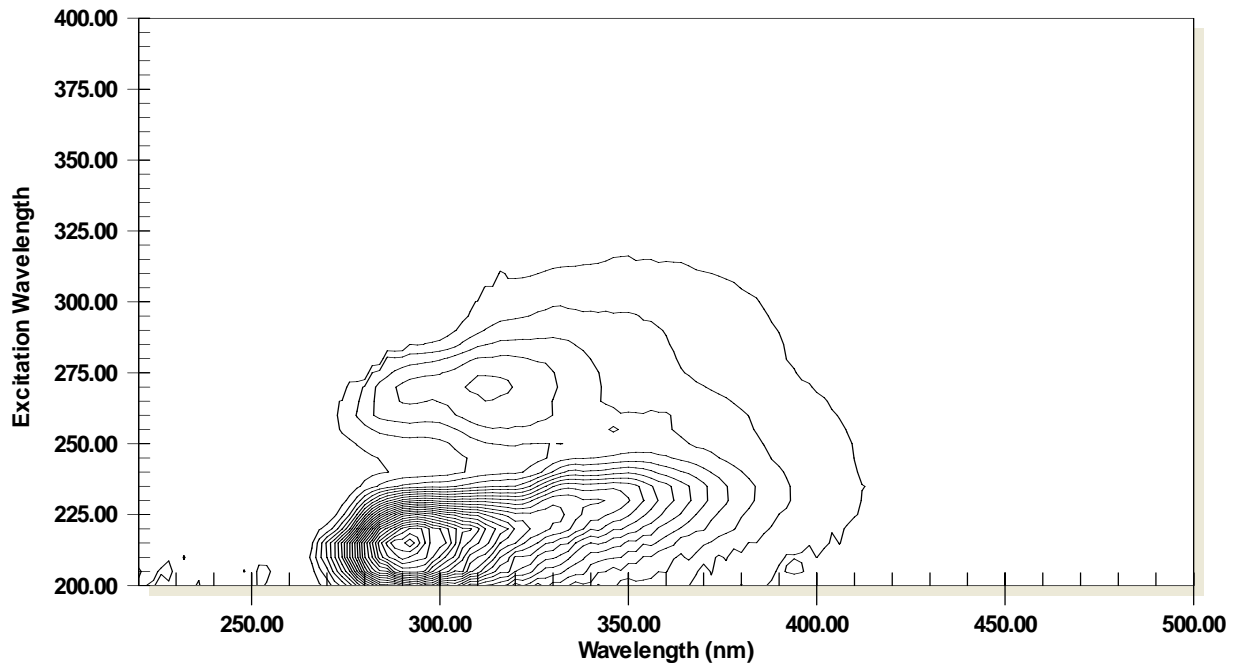


31CP

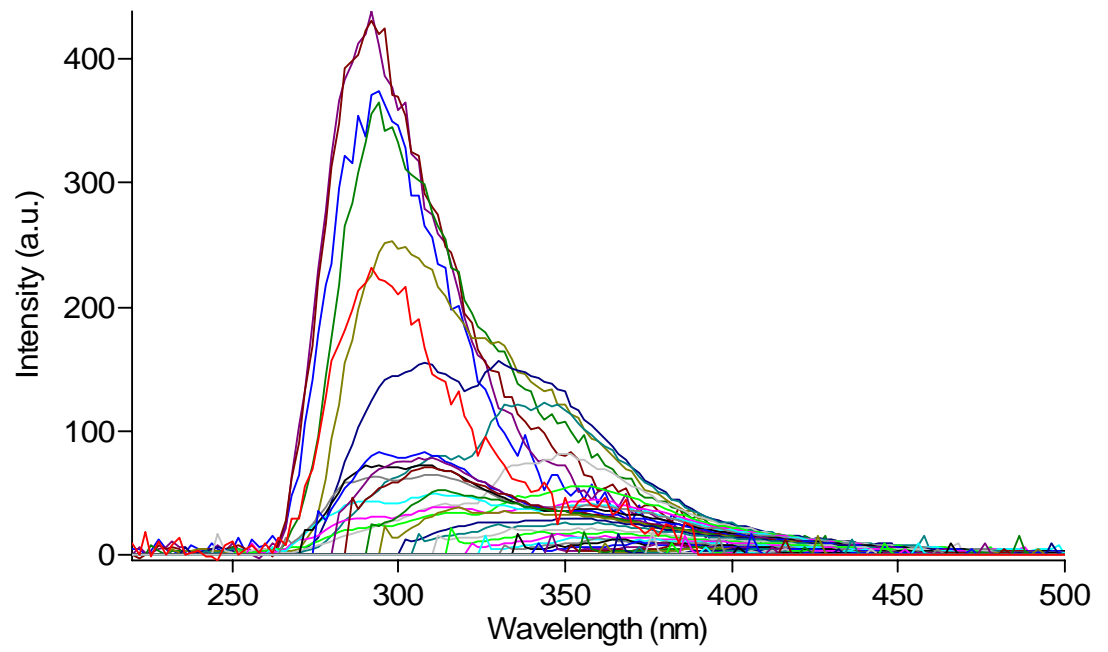
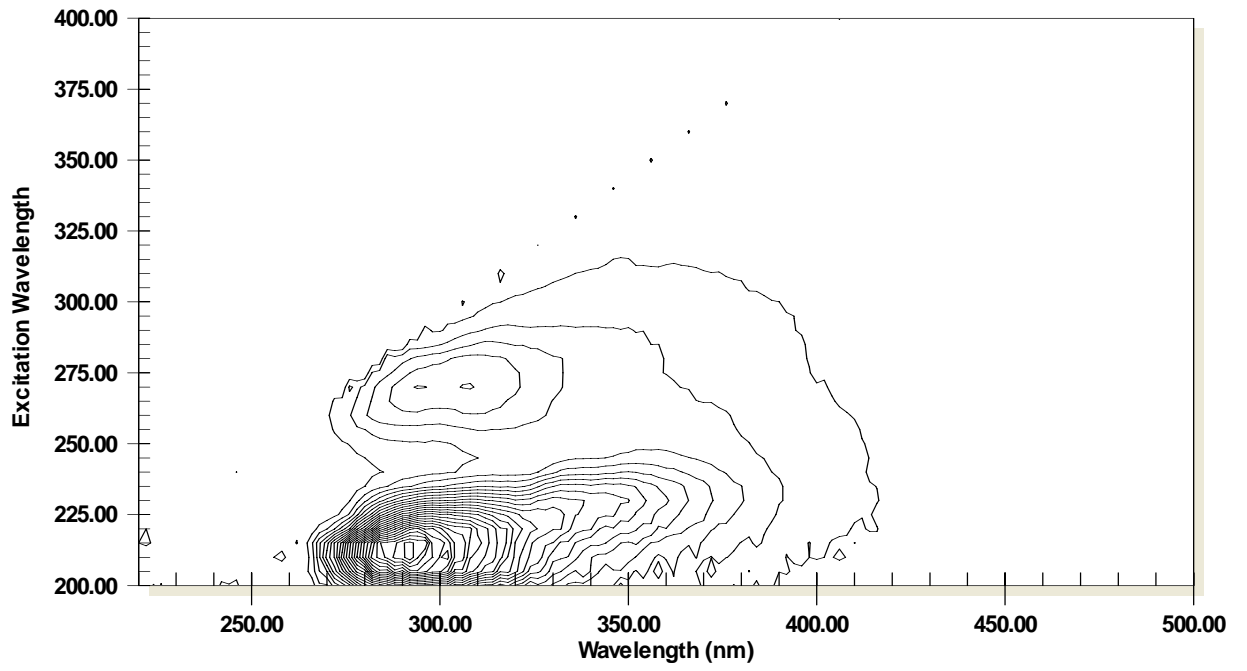


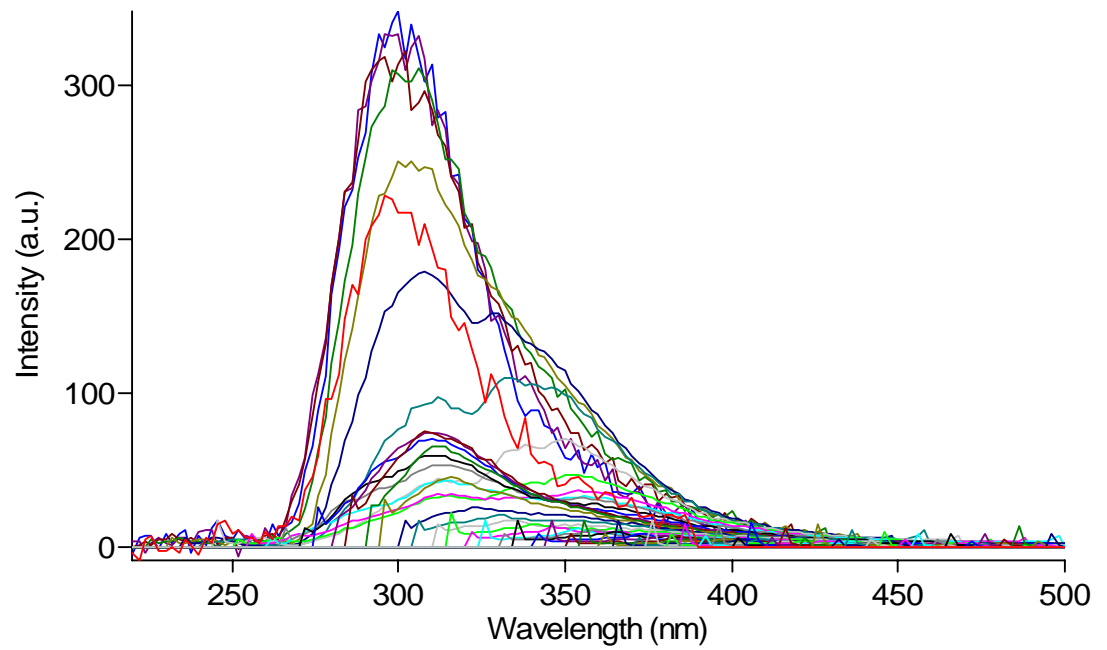
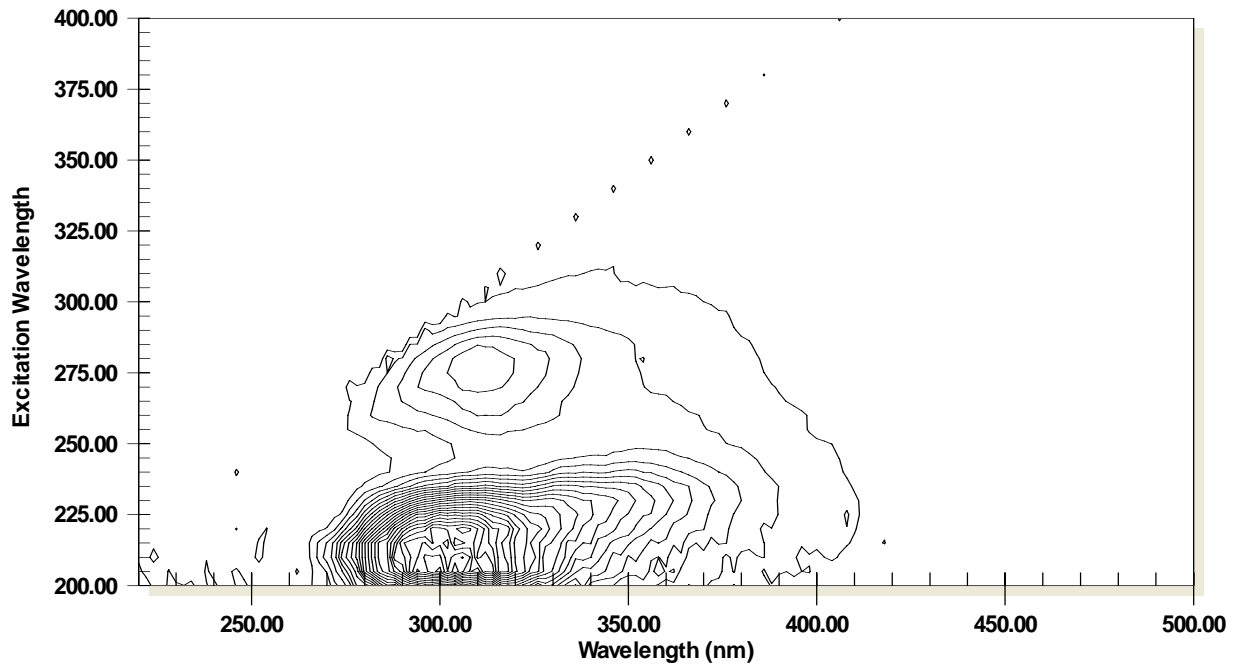


33AP

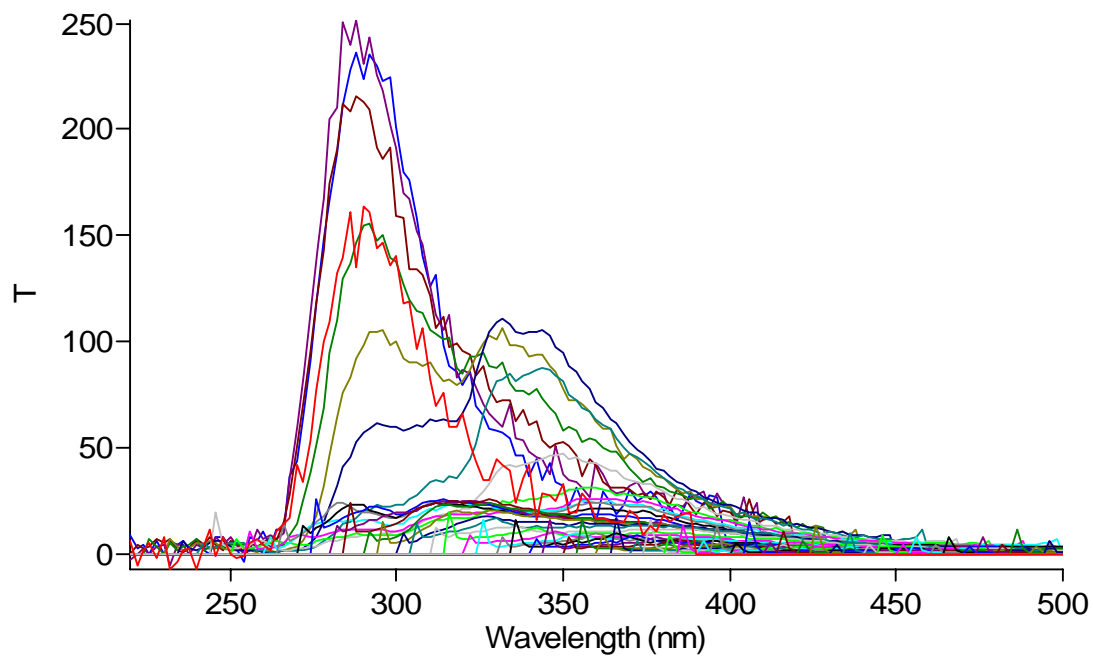
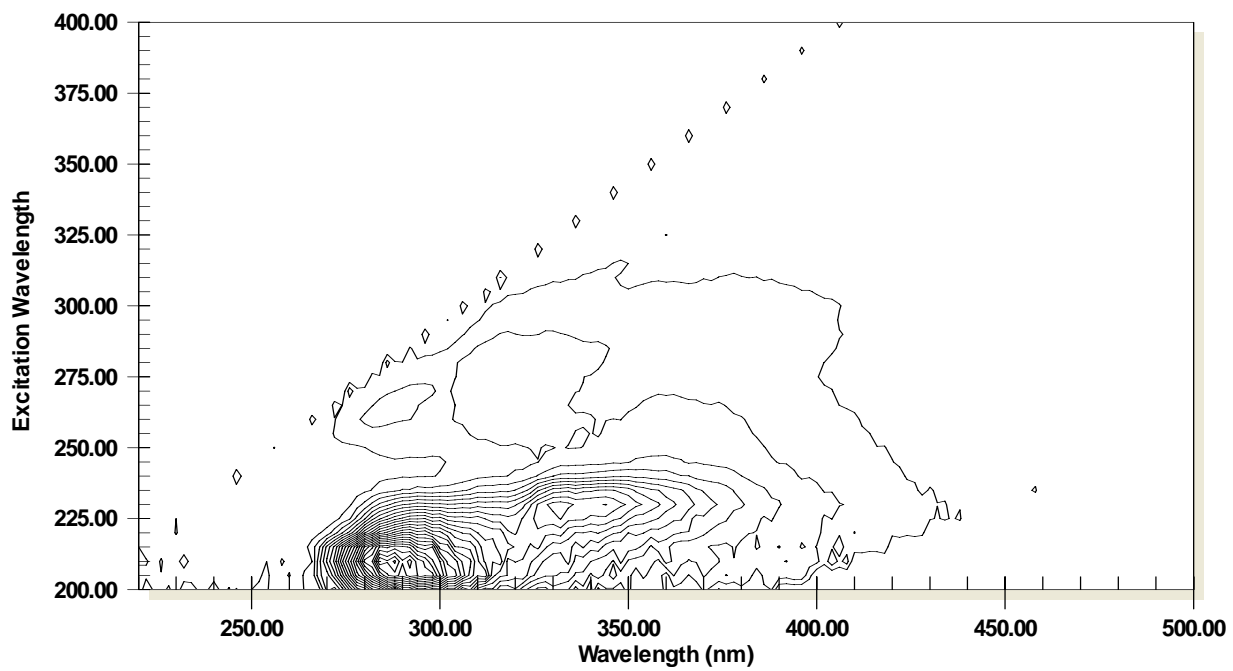


34BP

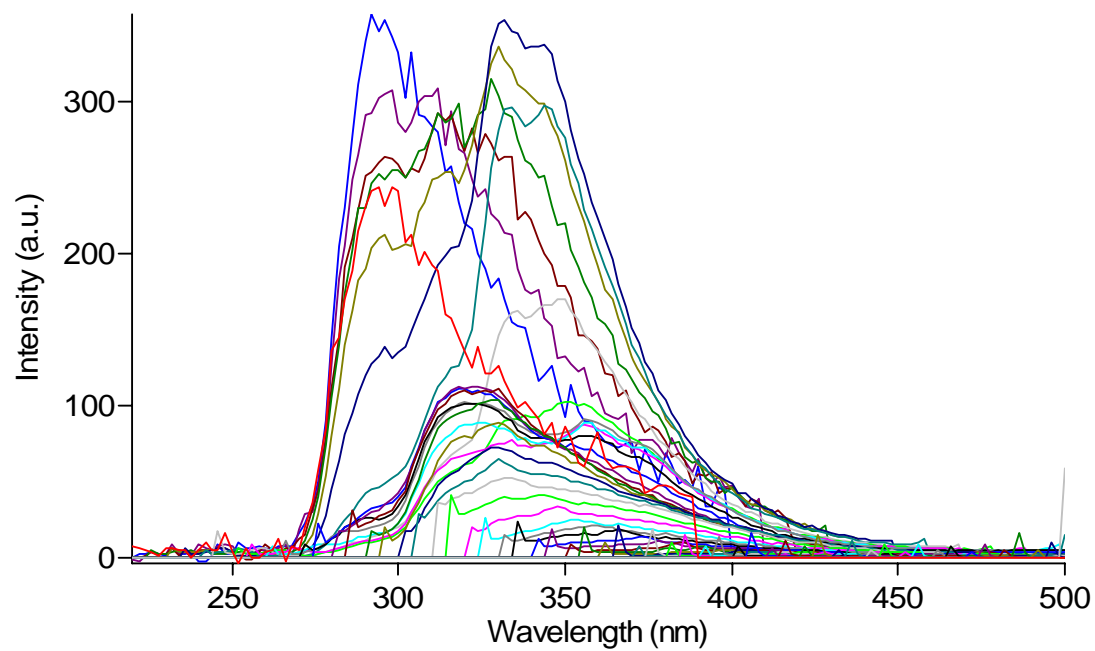
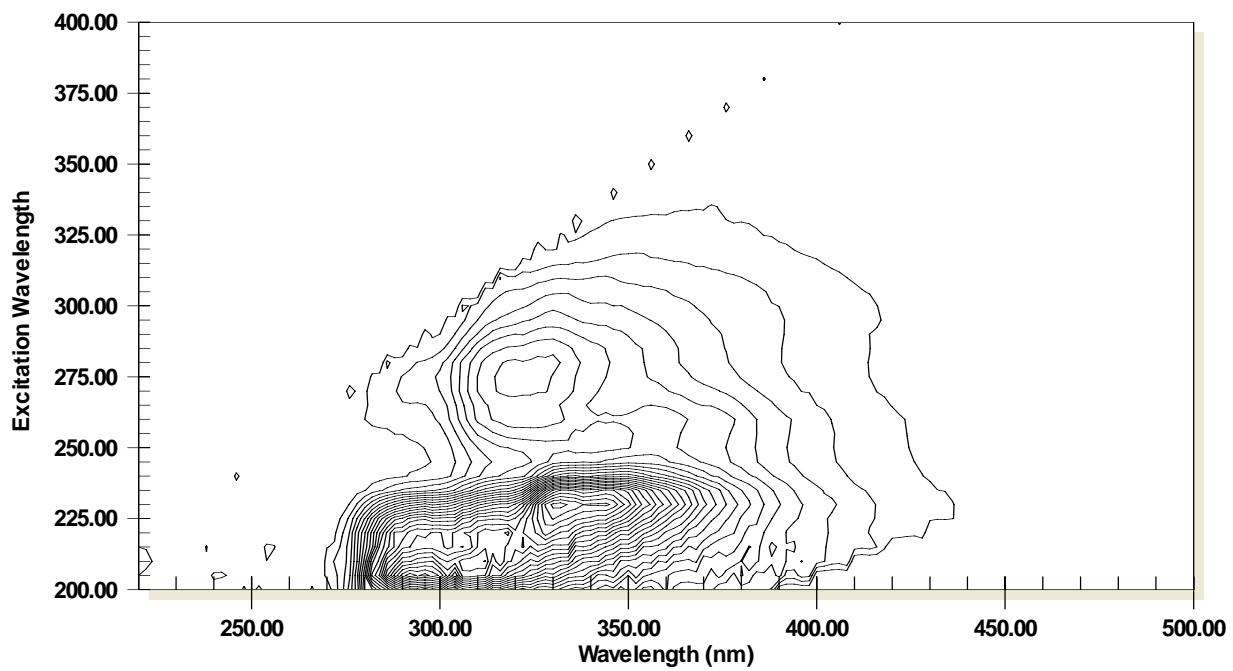




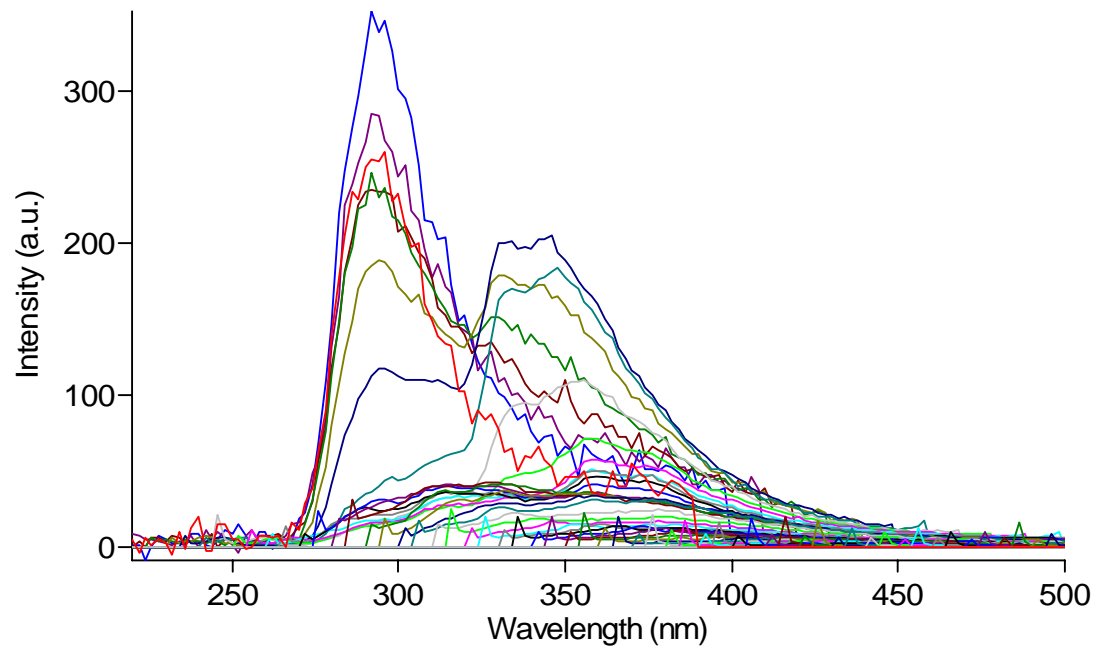
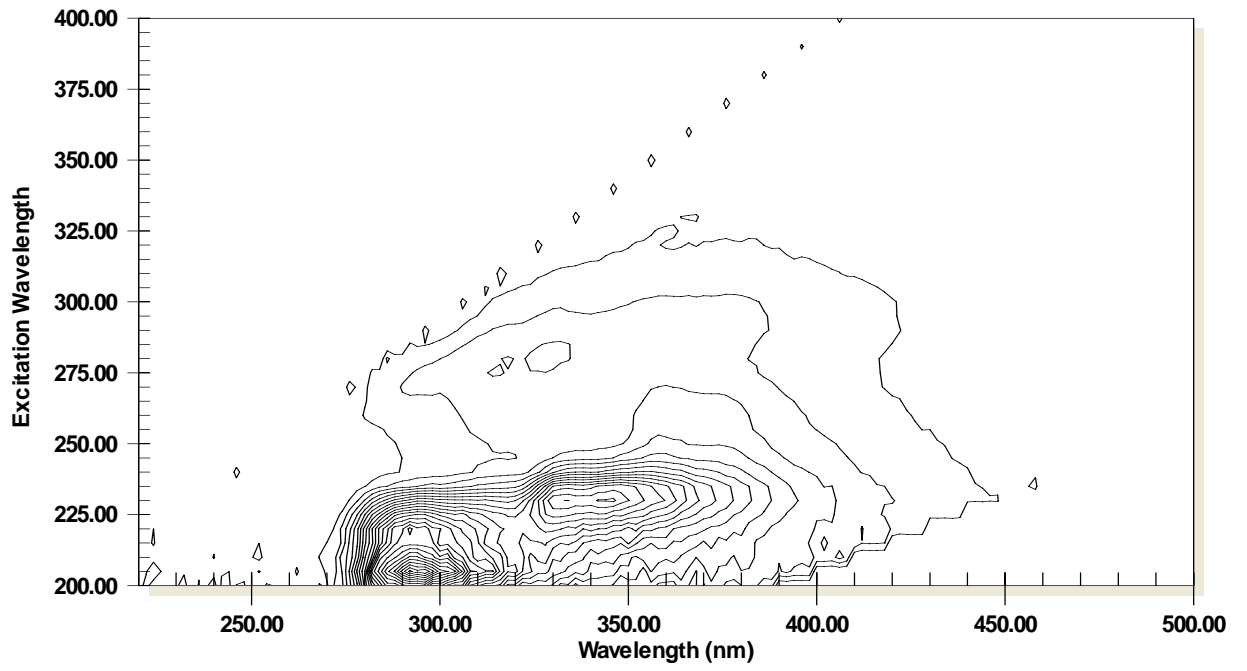
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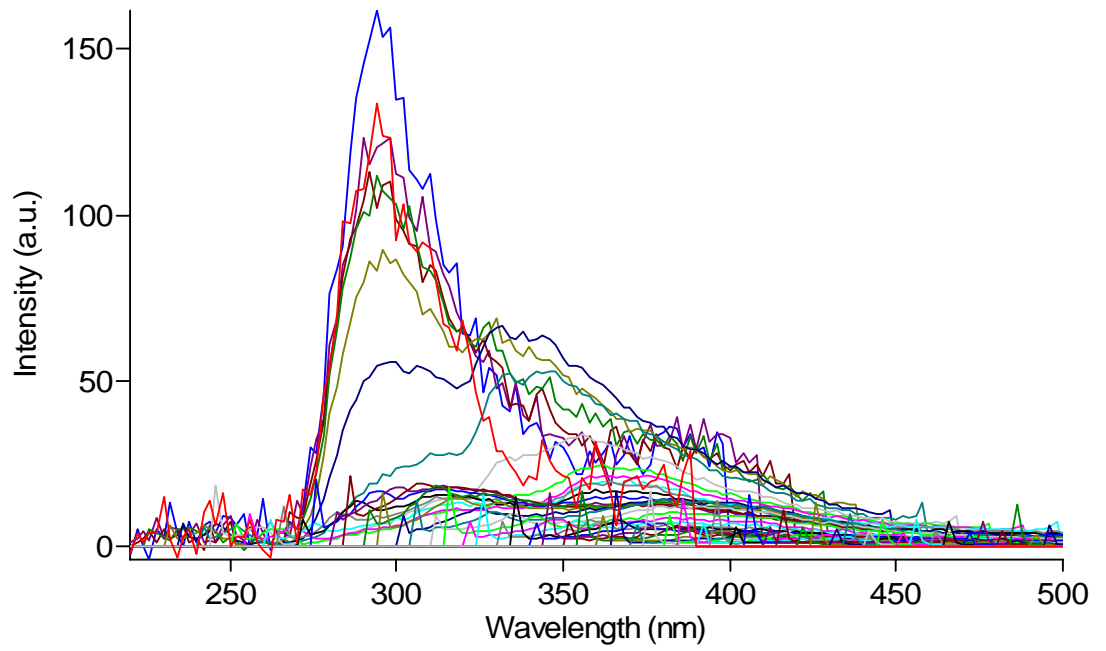
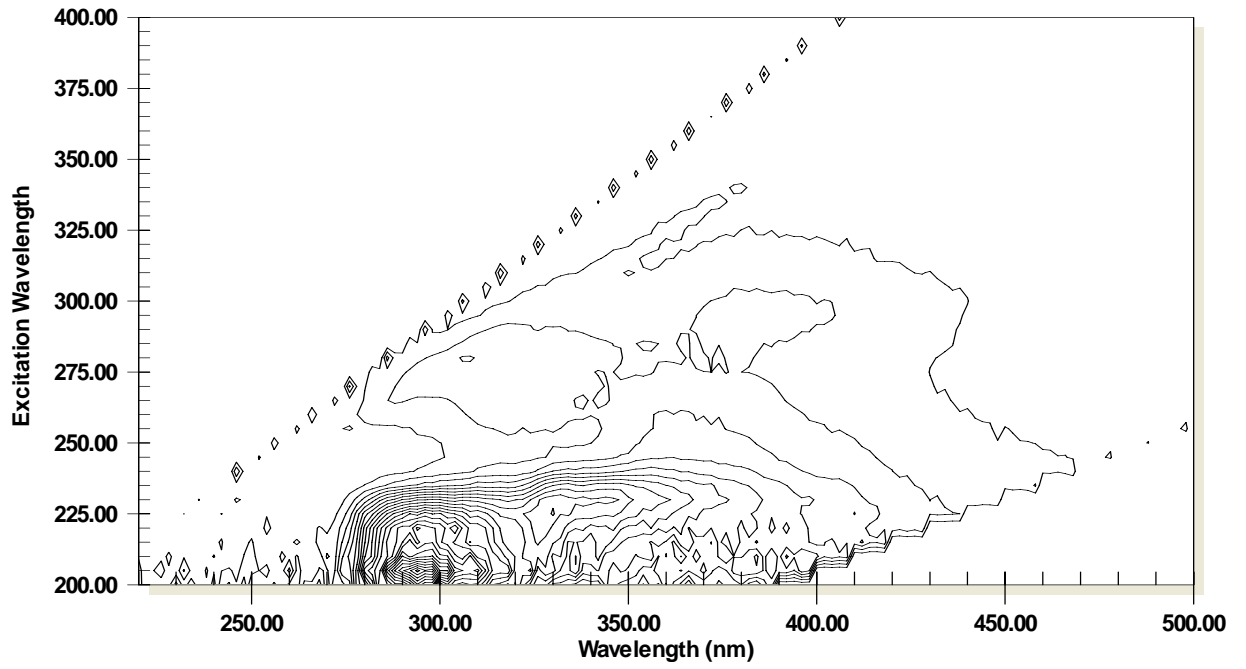
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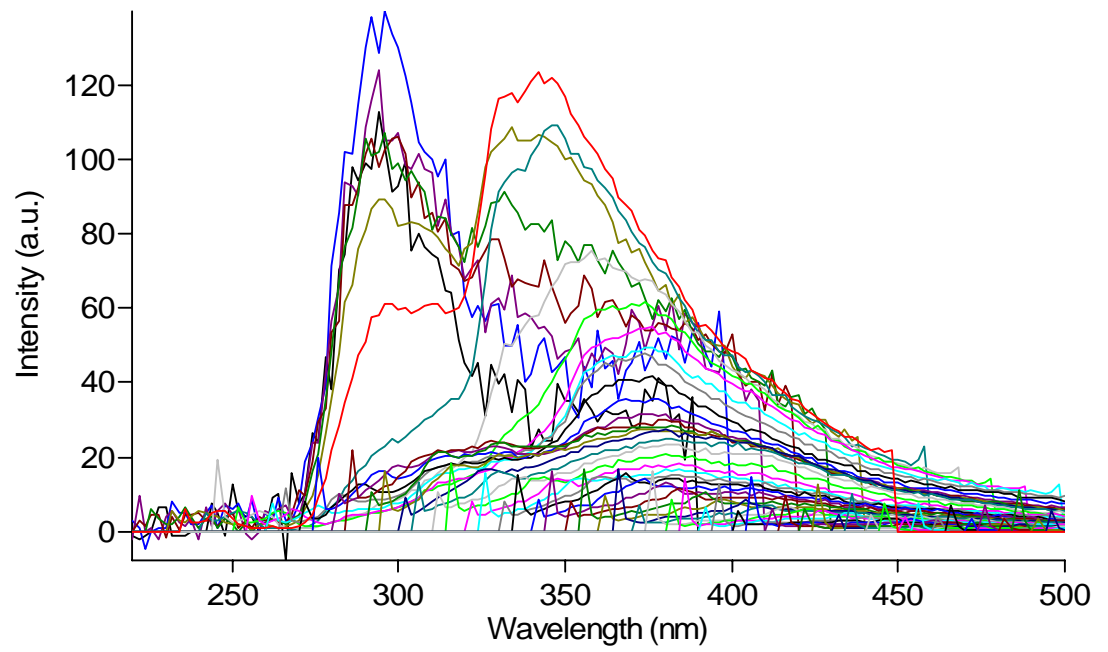
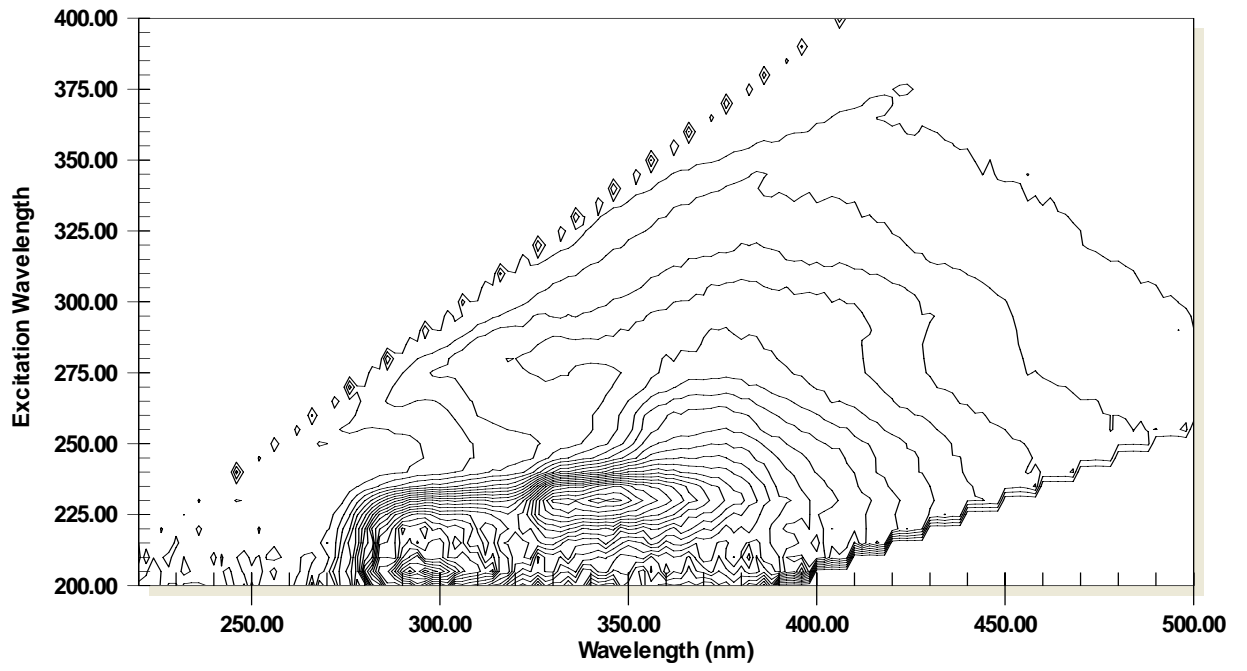
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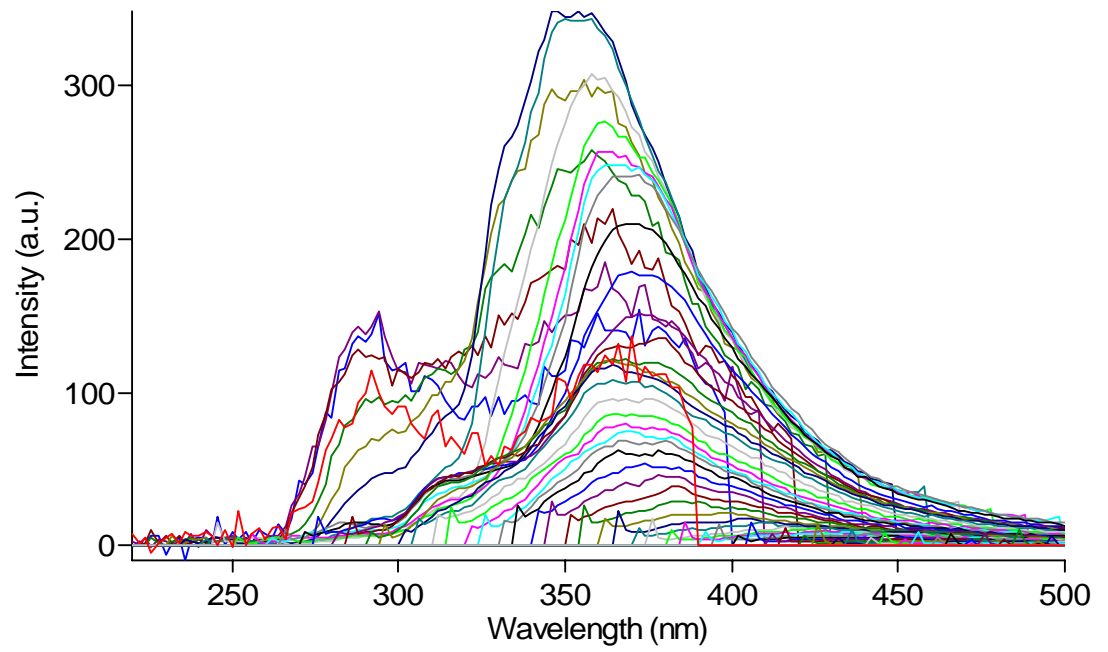
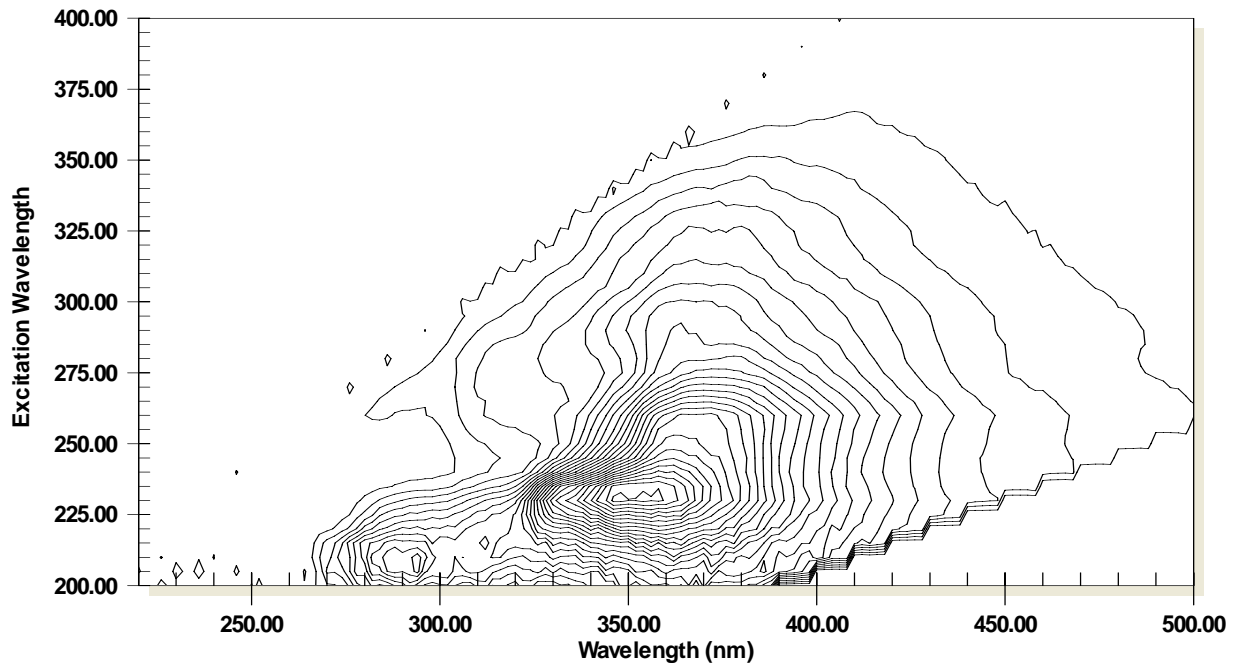
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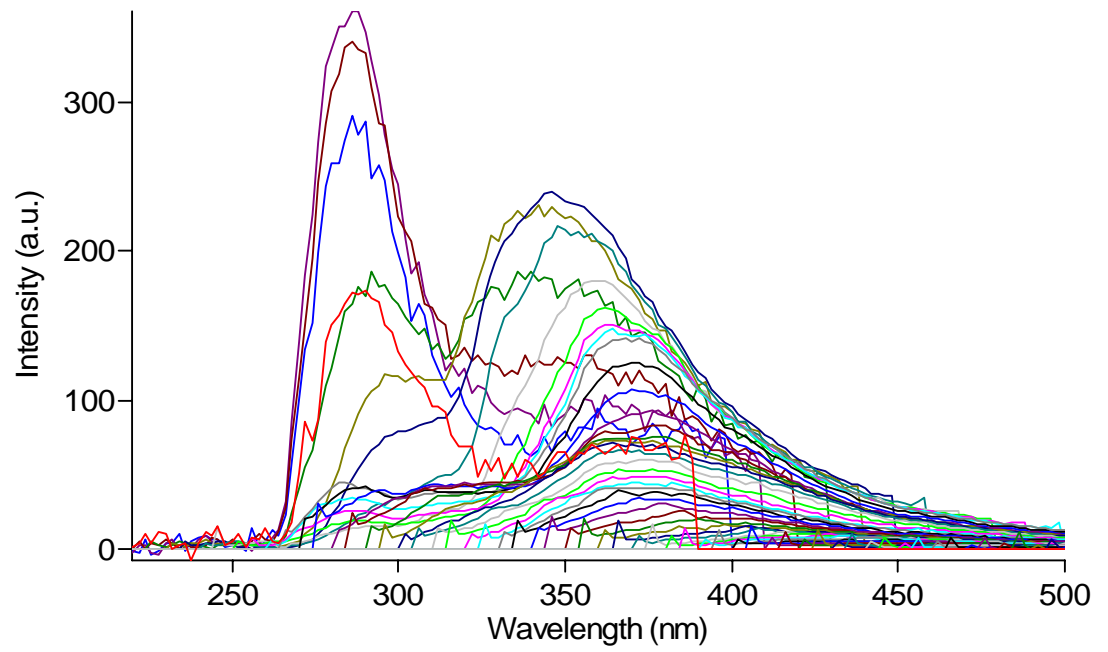
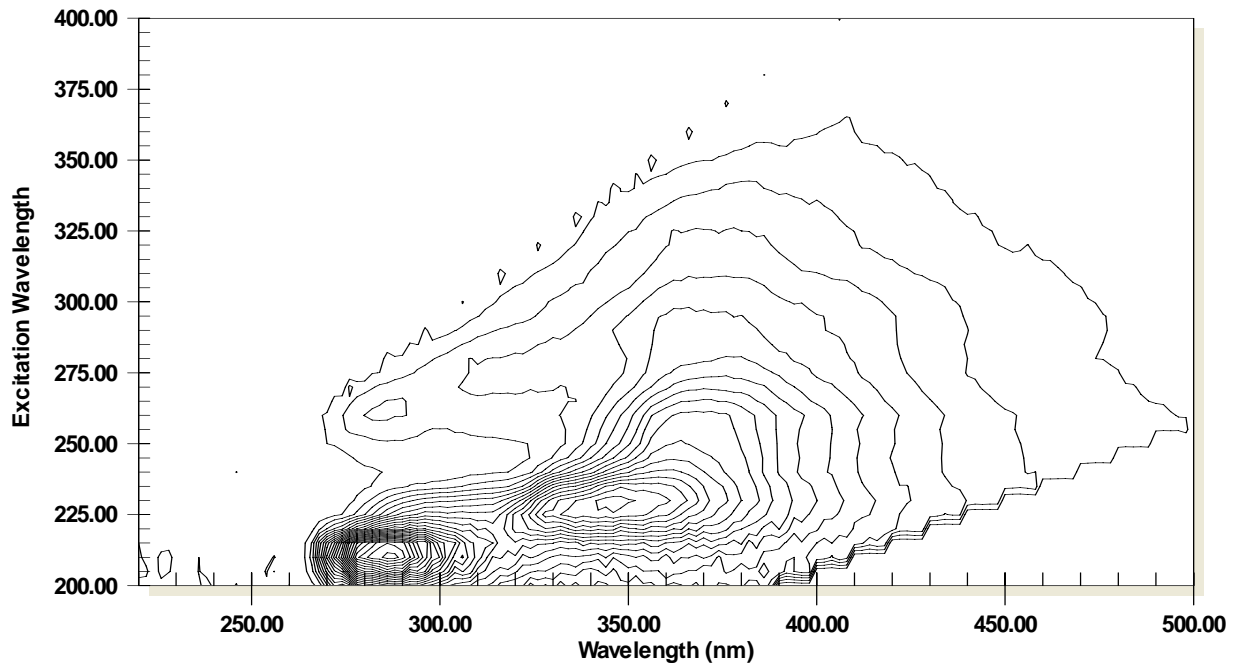
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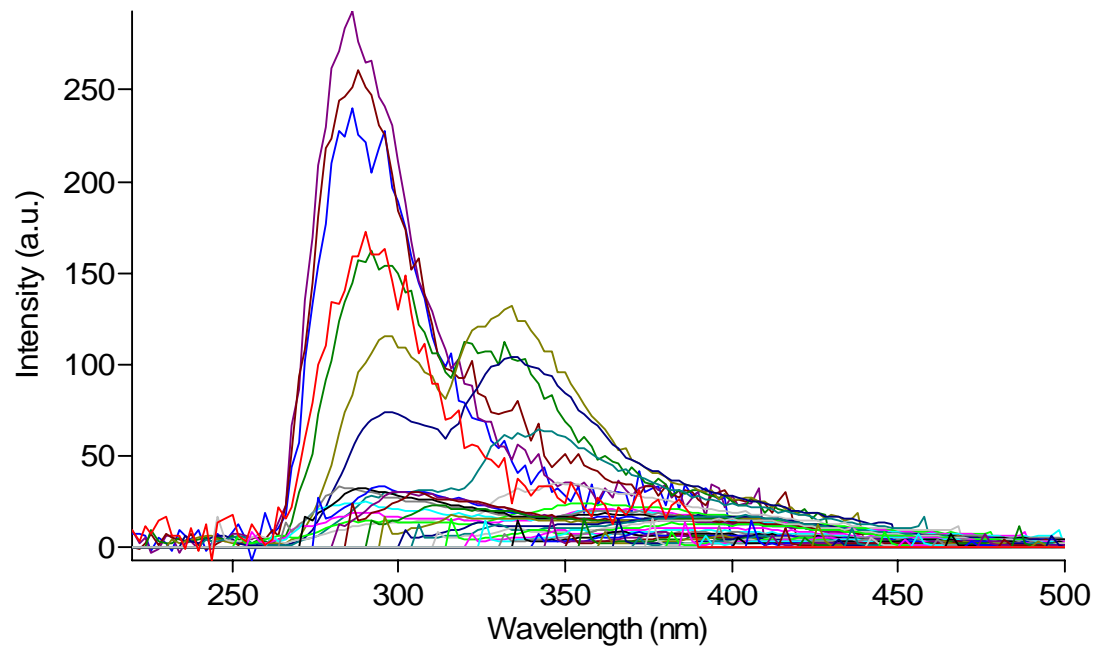
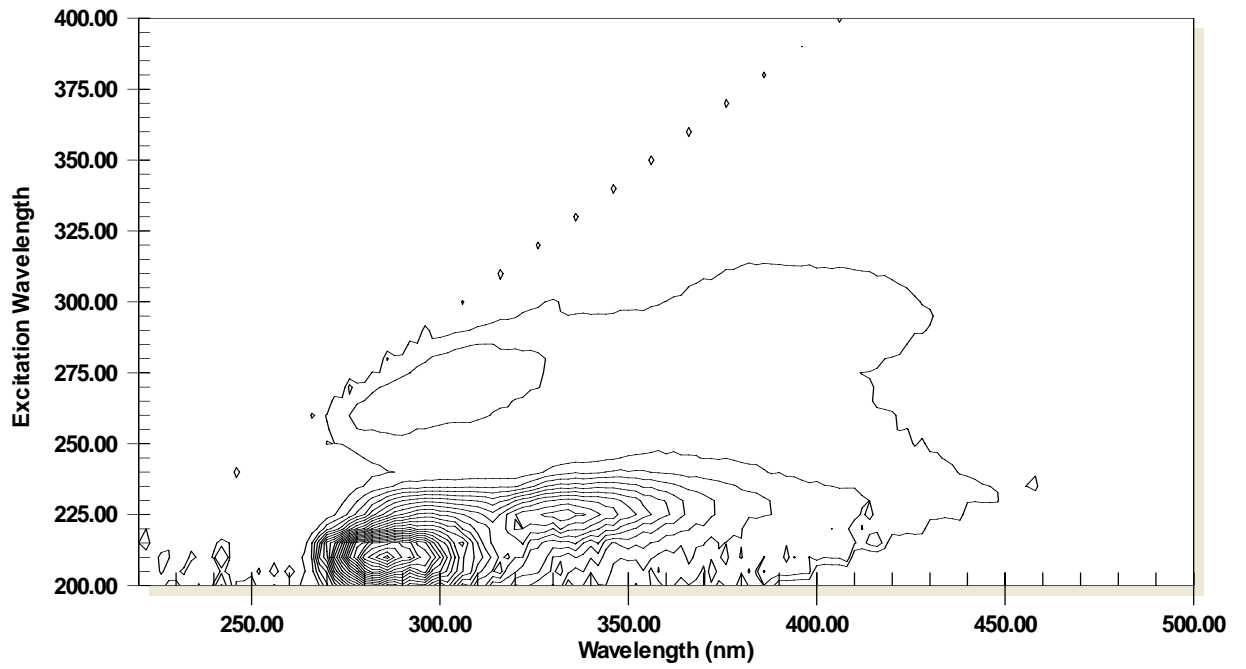
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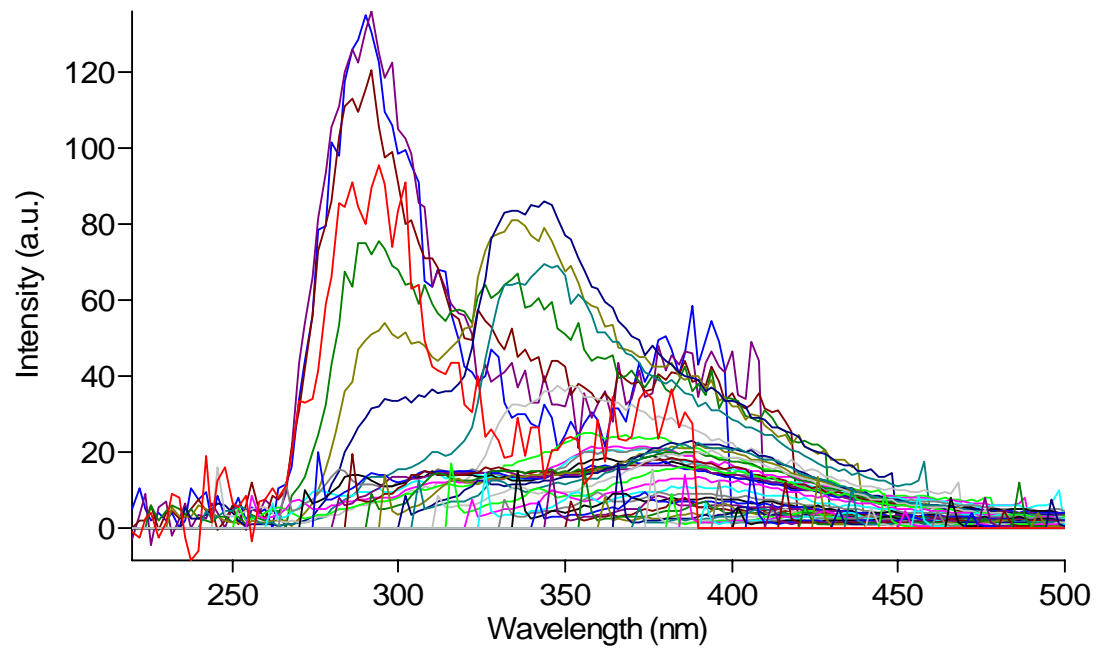
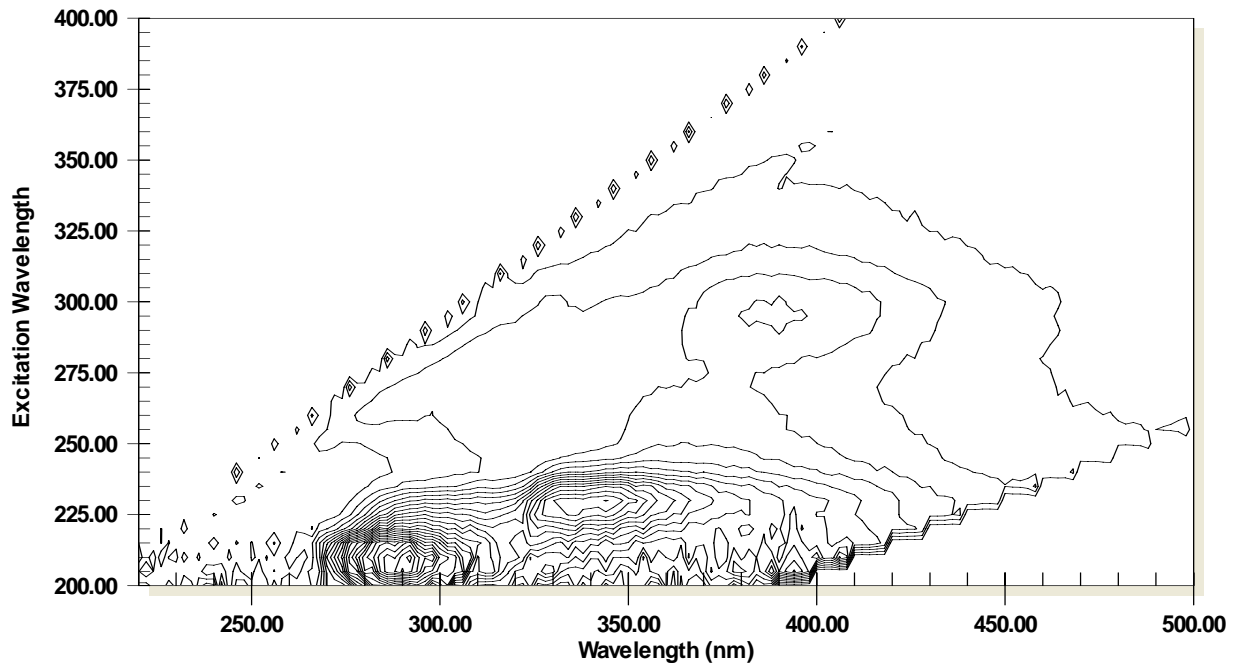
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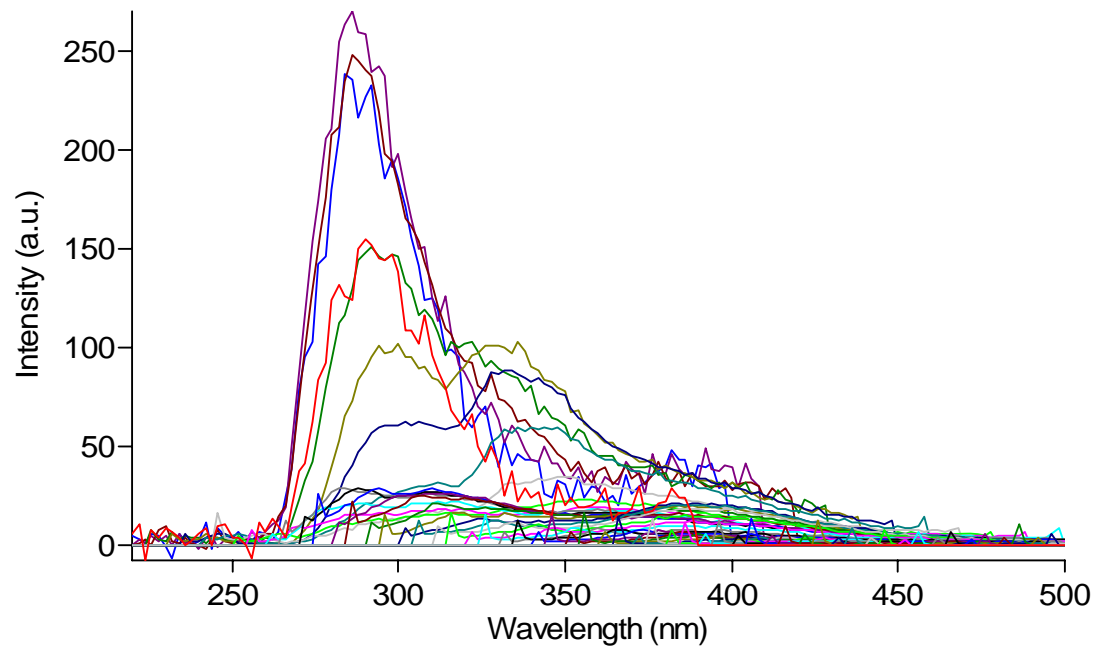
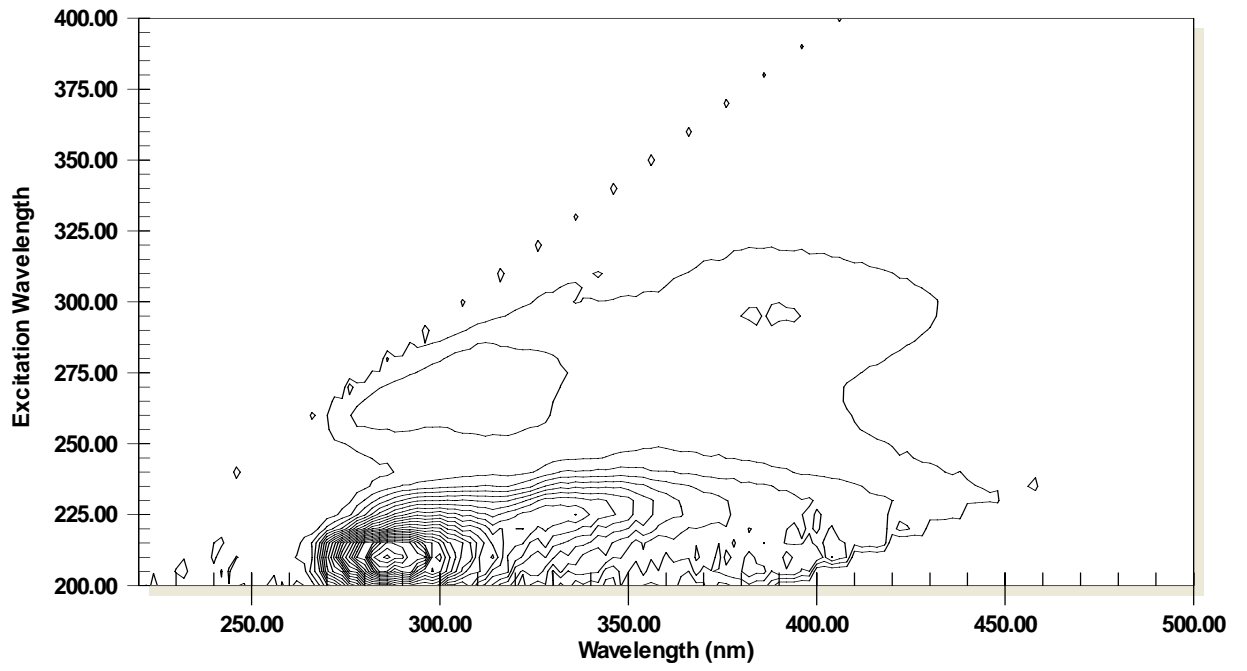


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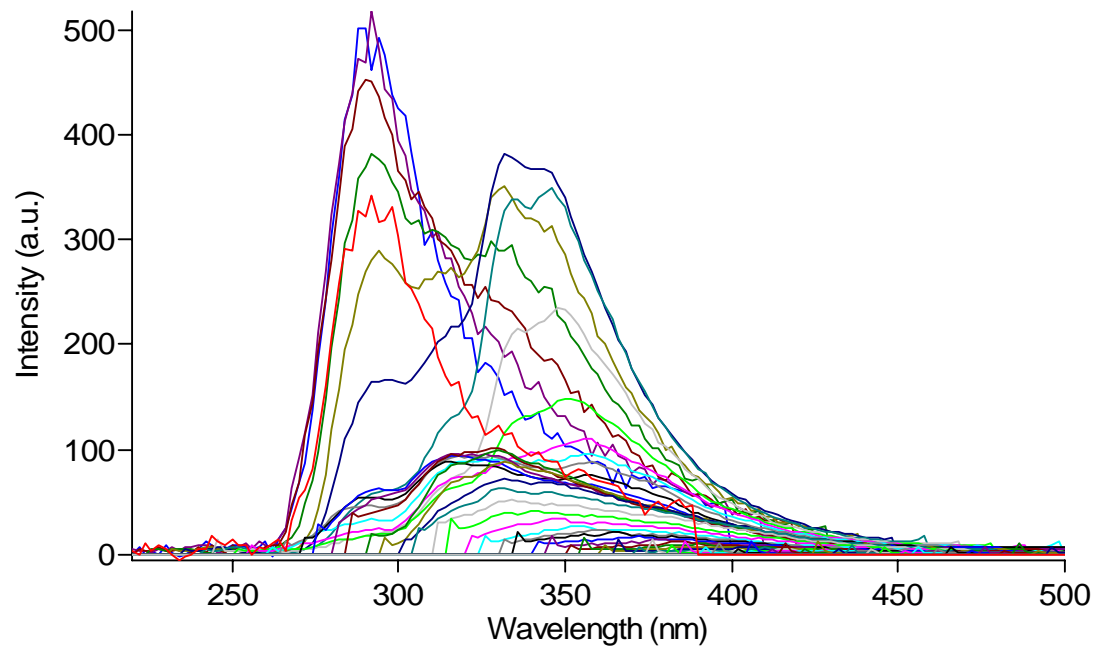
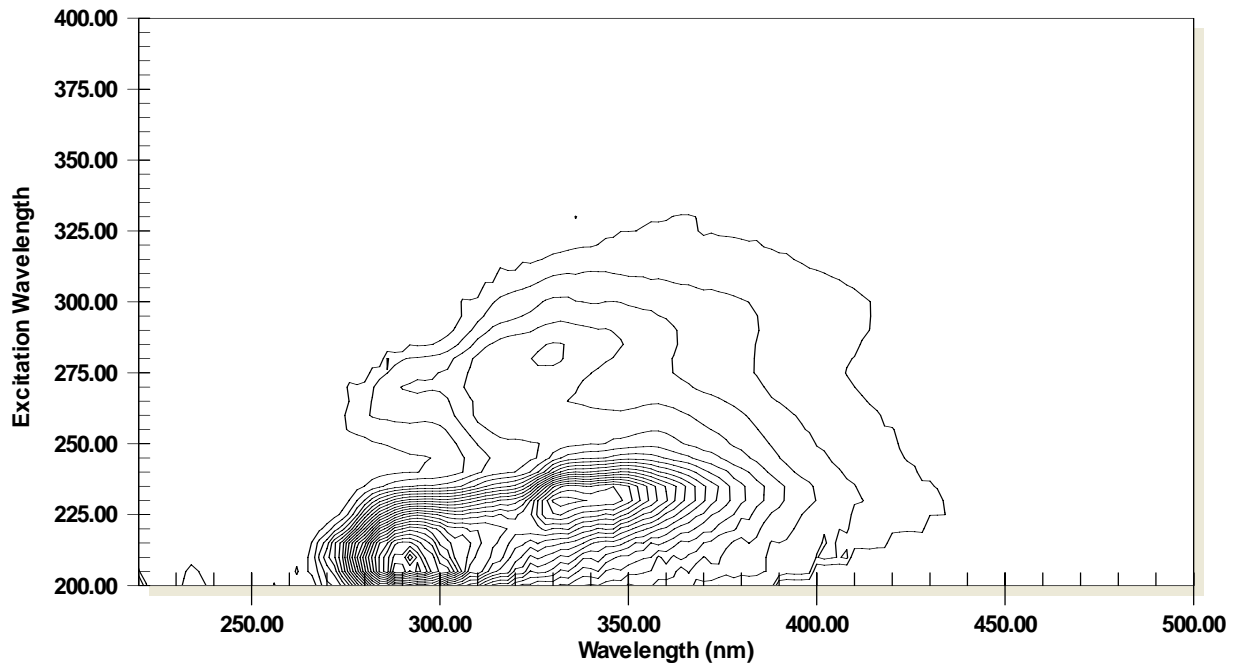


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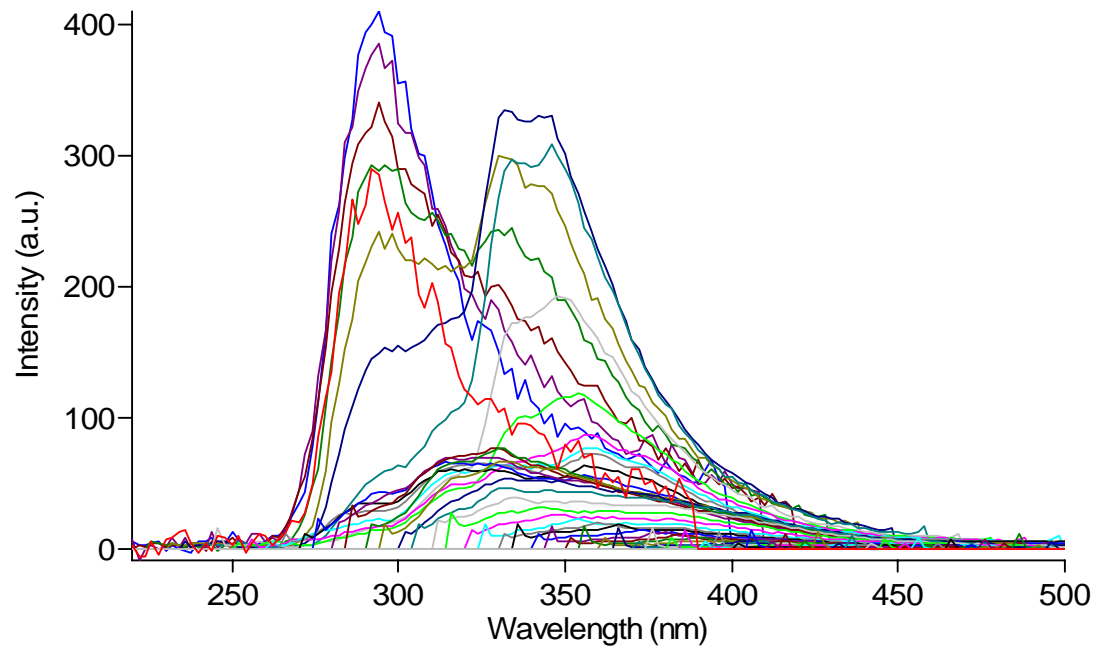
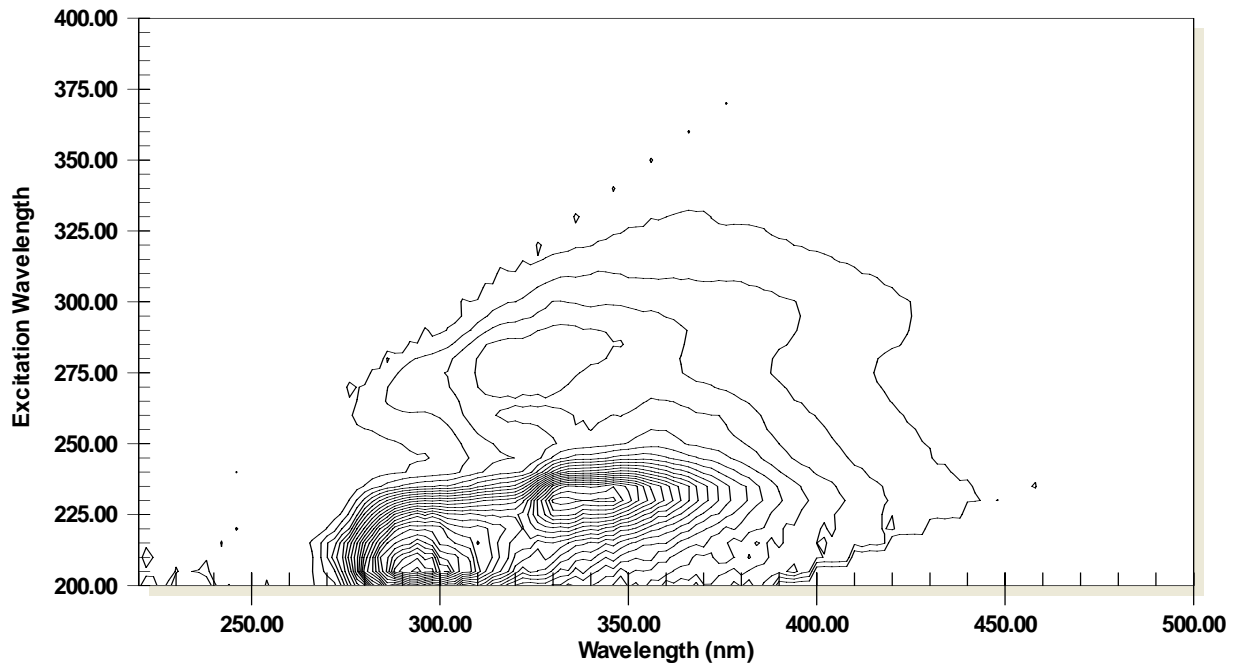




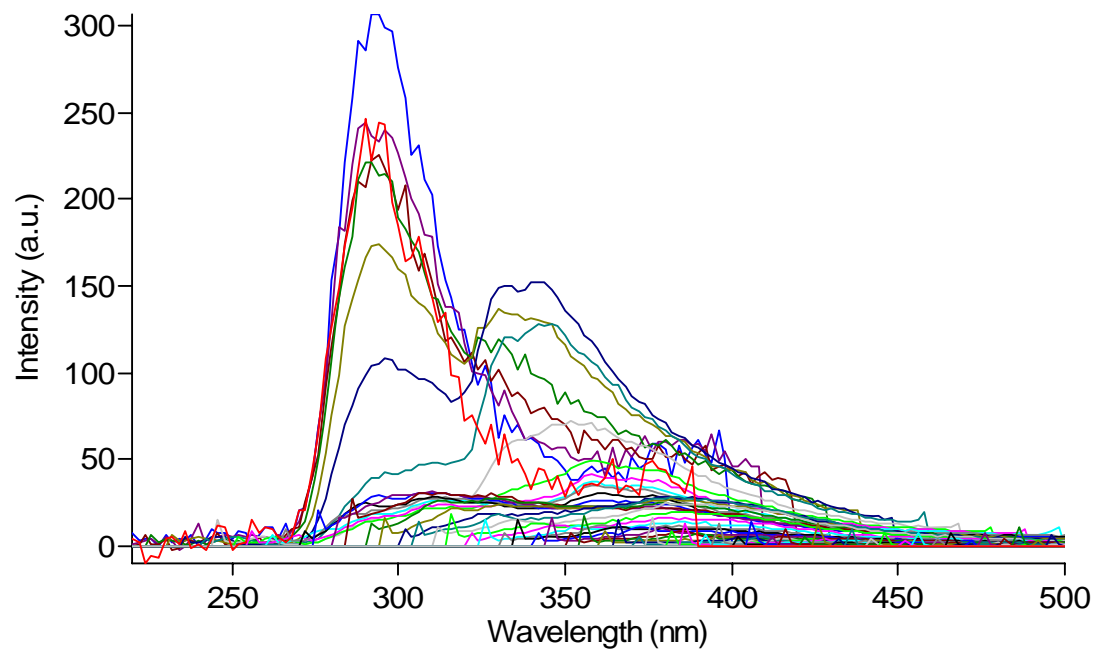
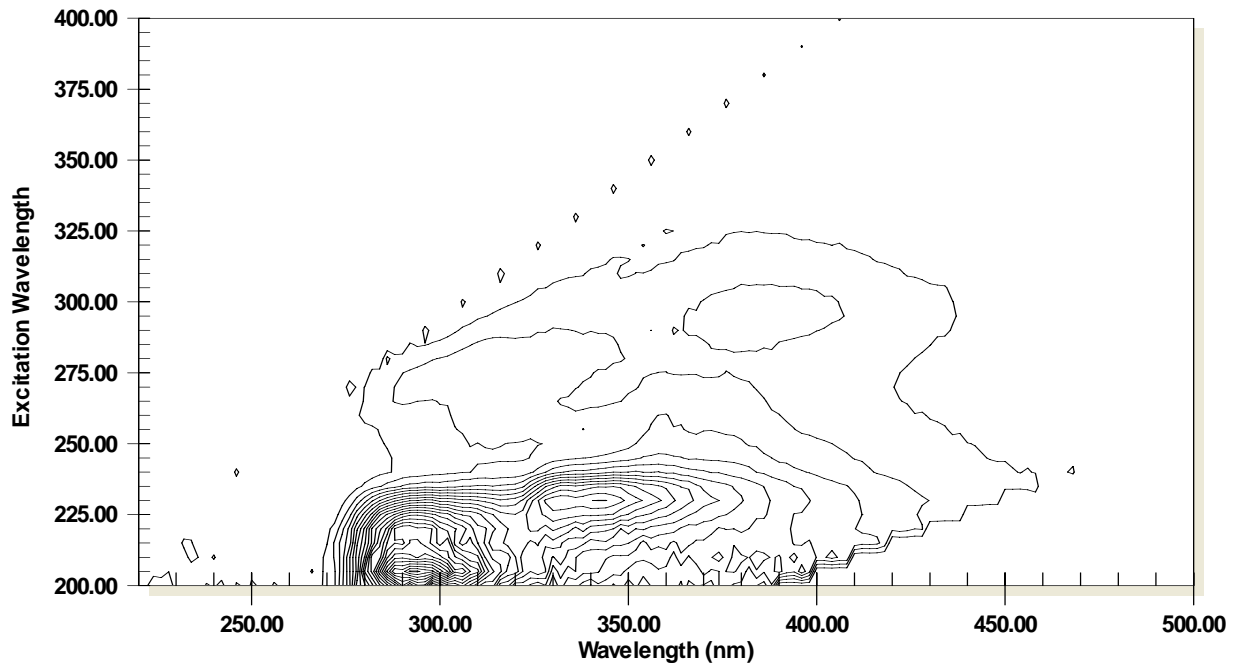
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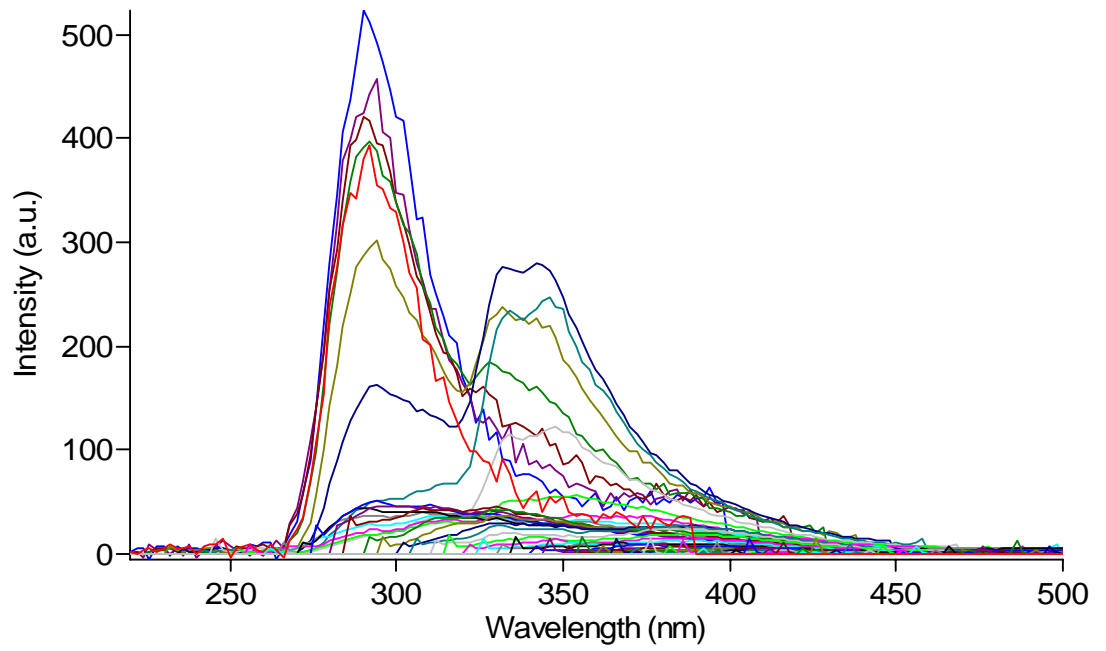
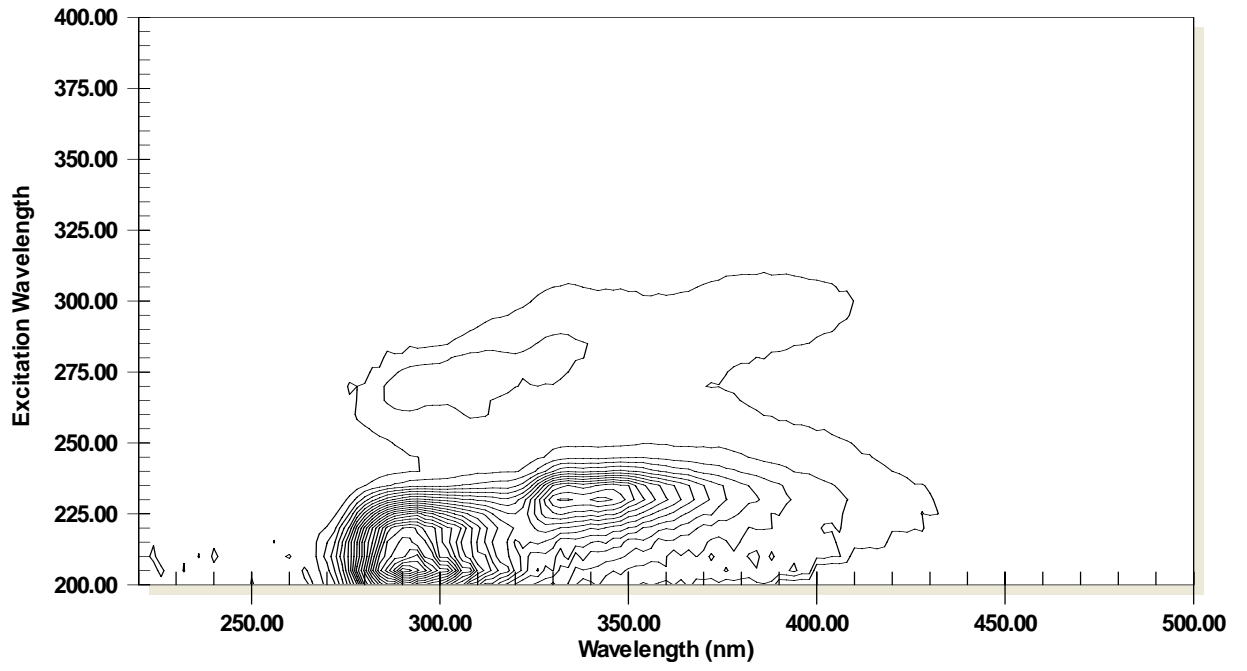
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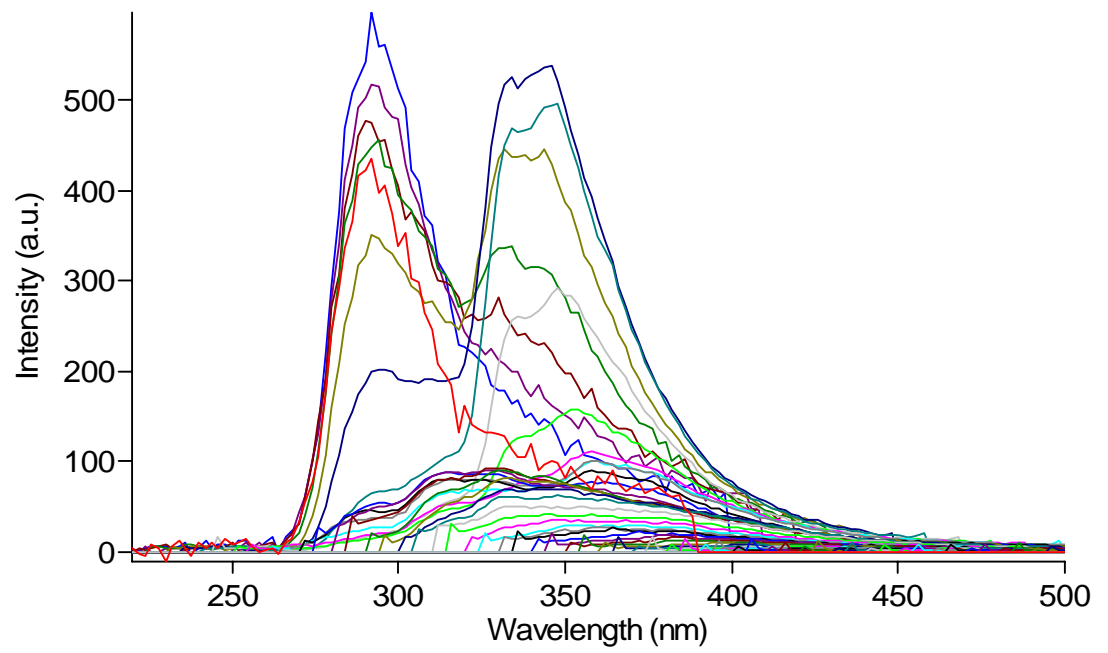
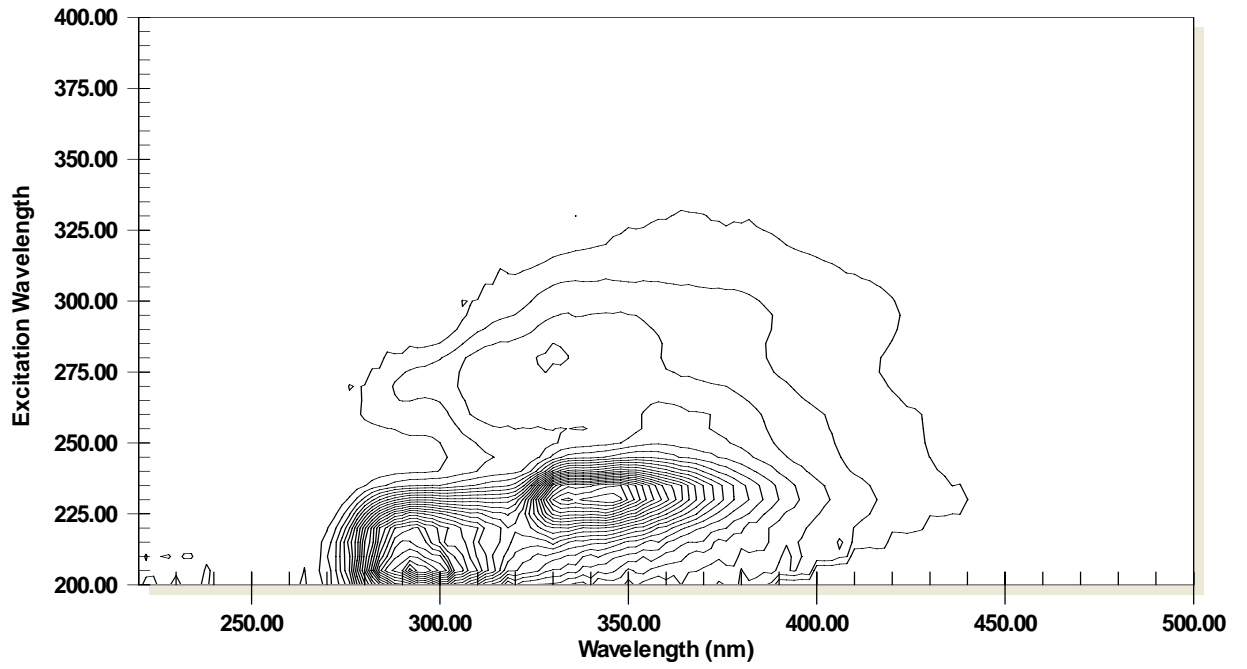
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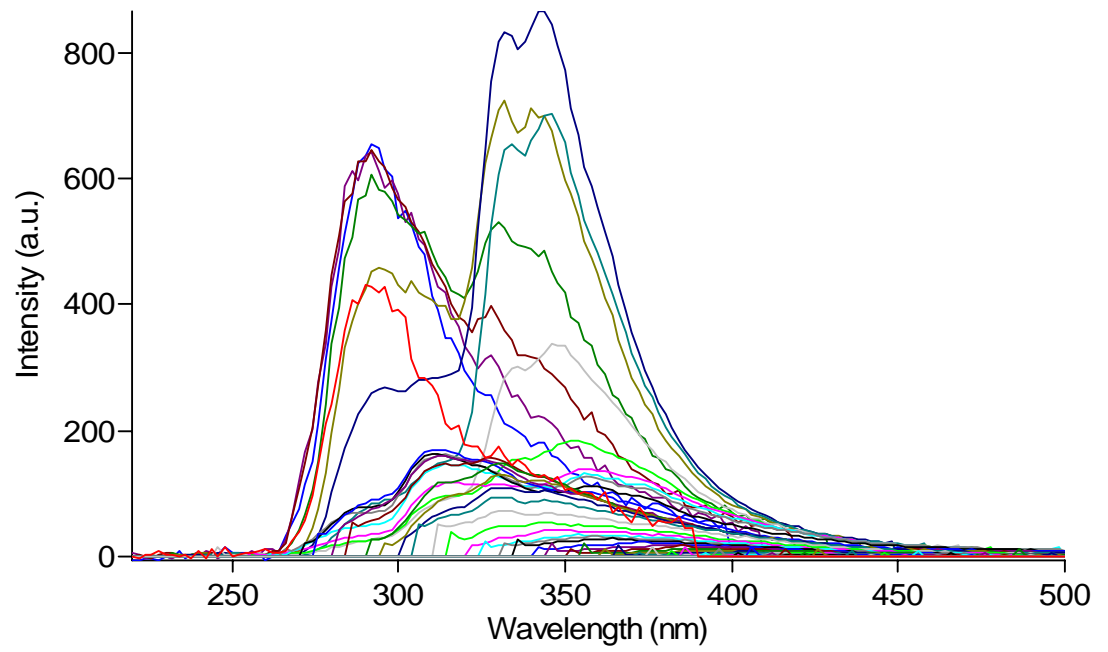
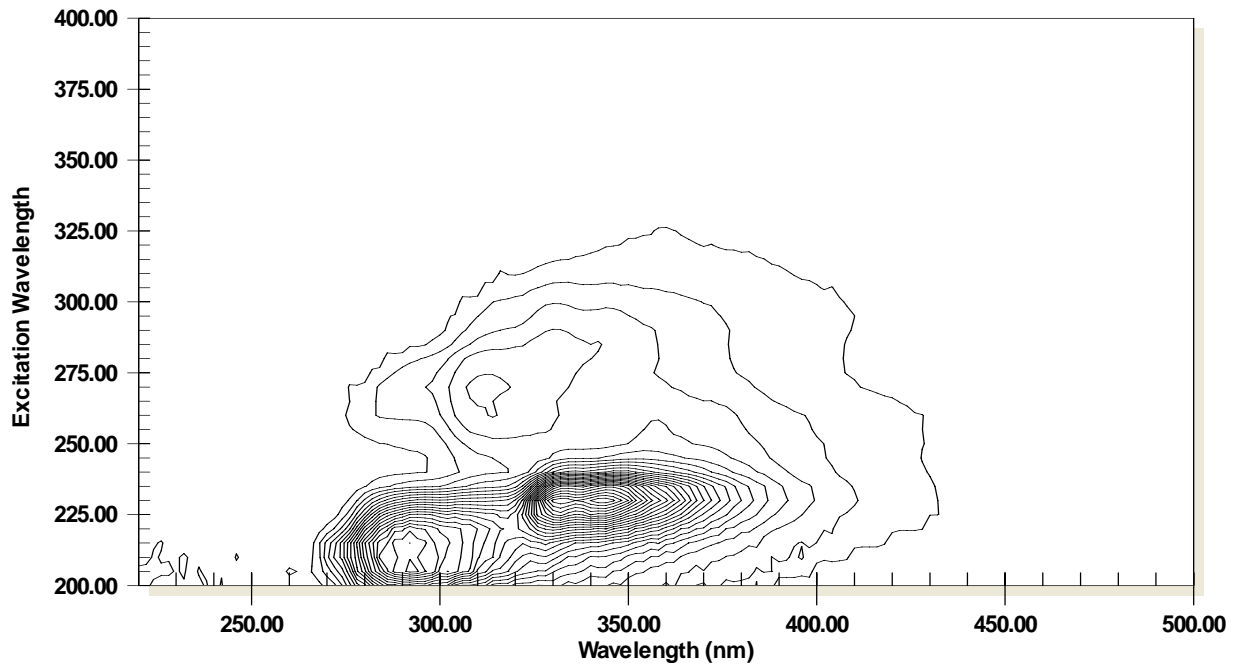
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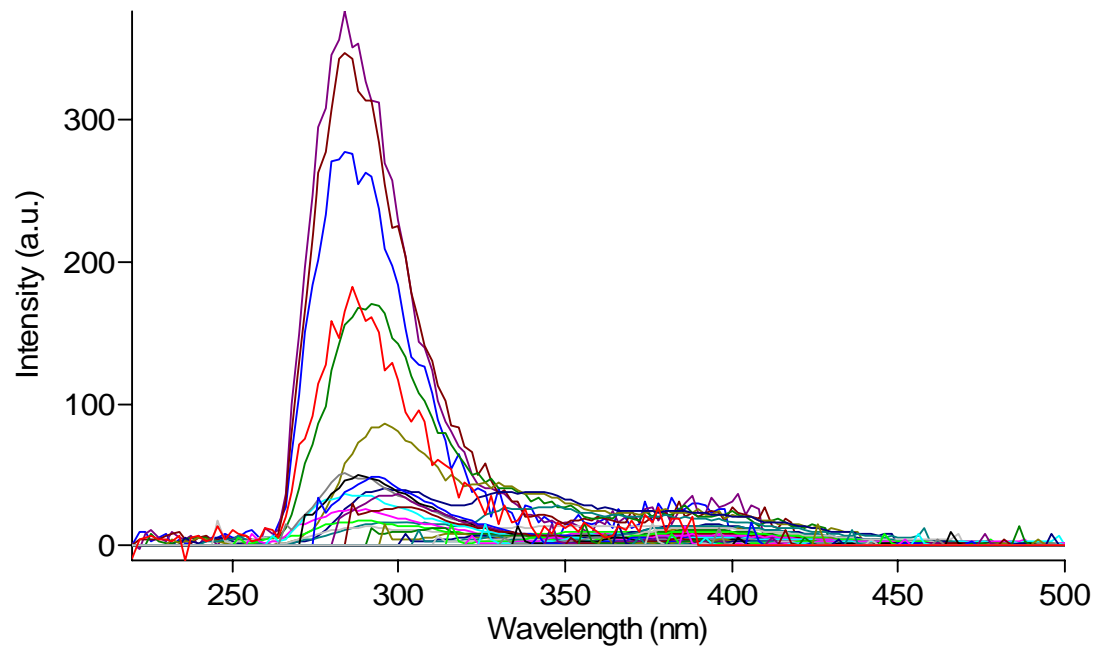
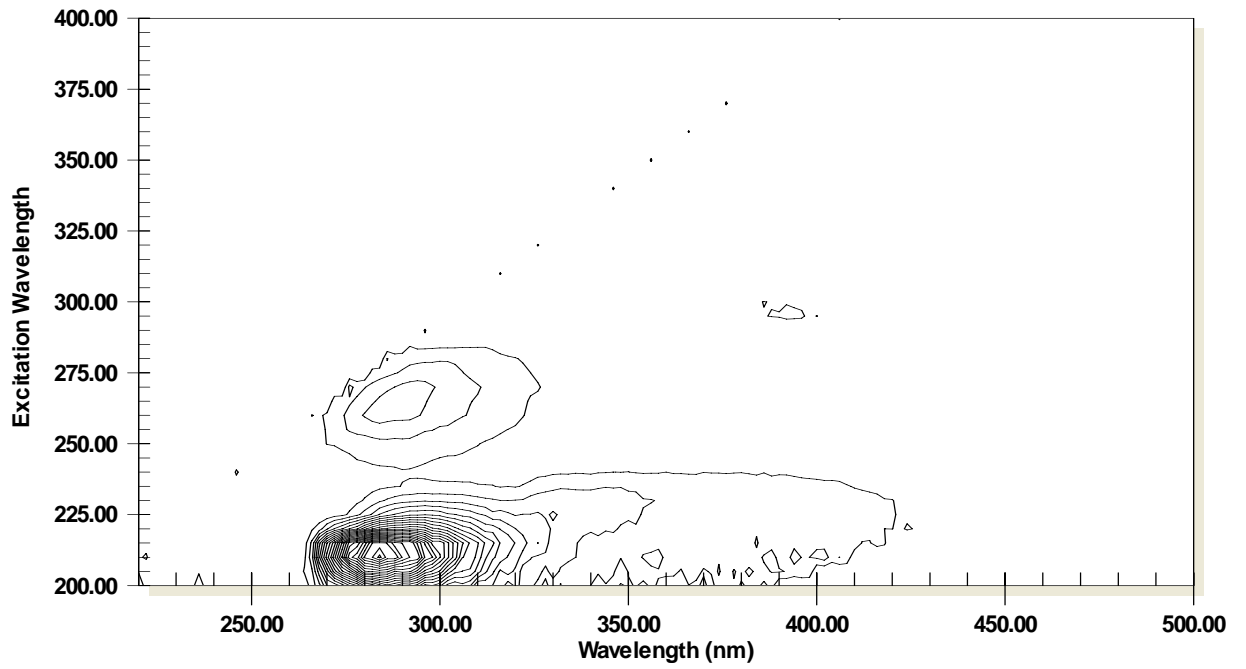
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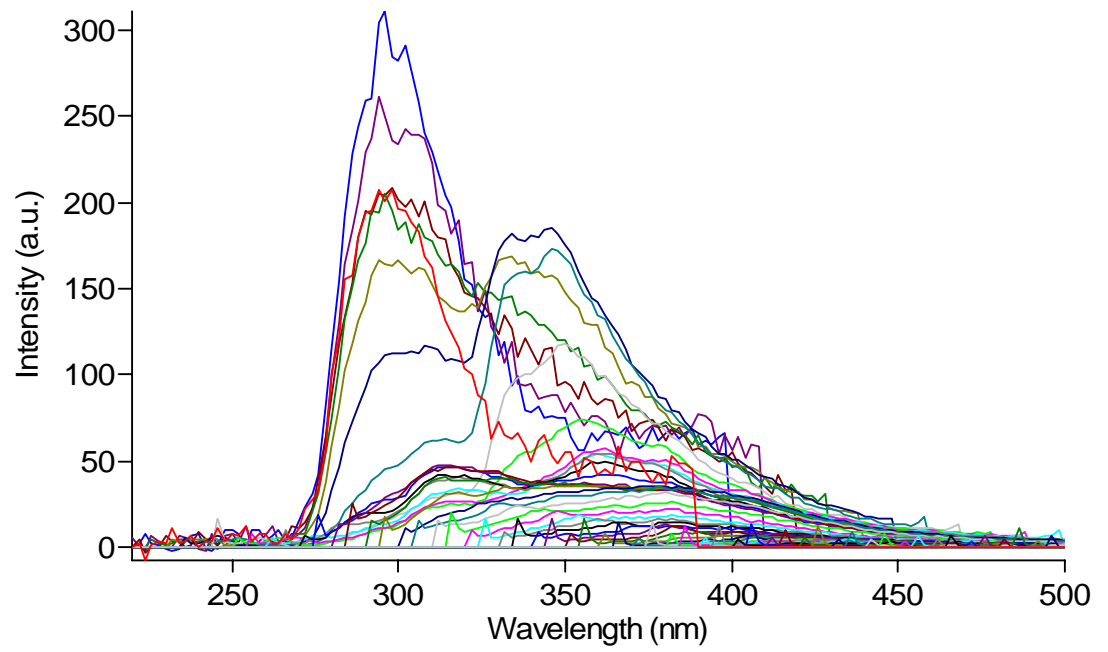
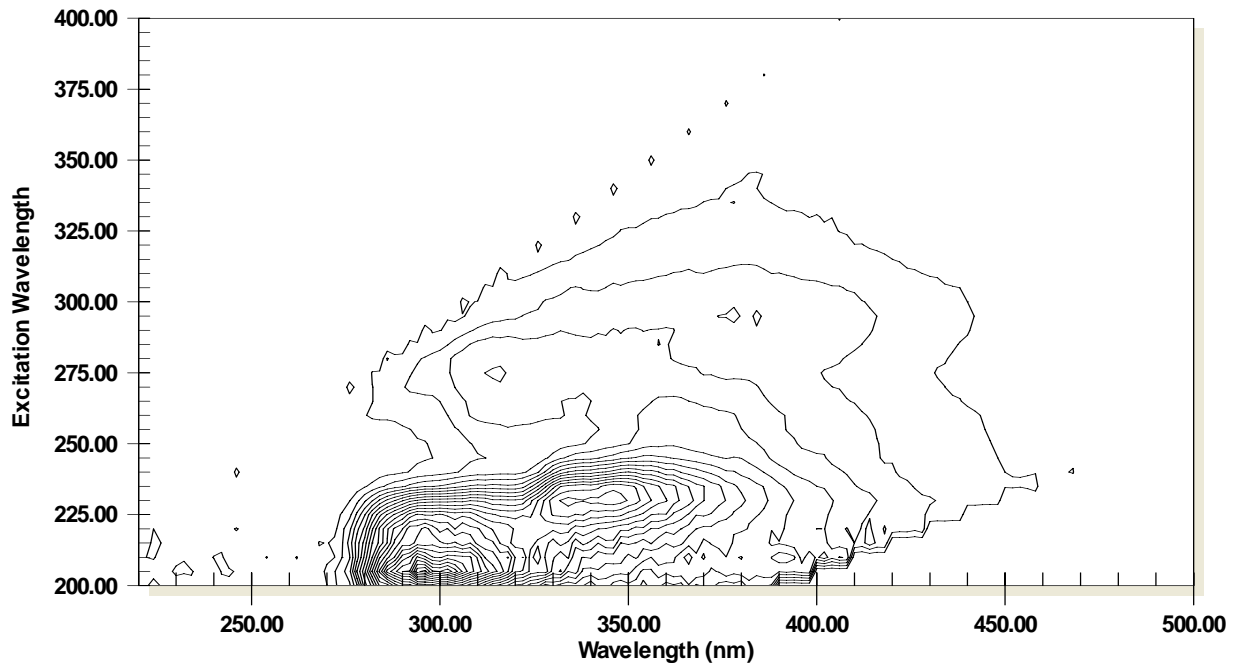


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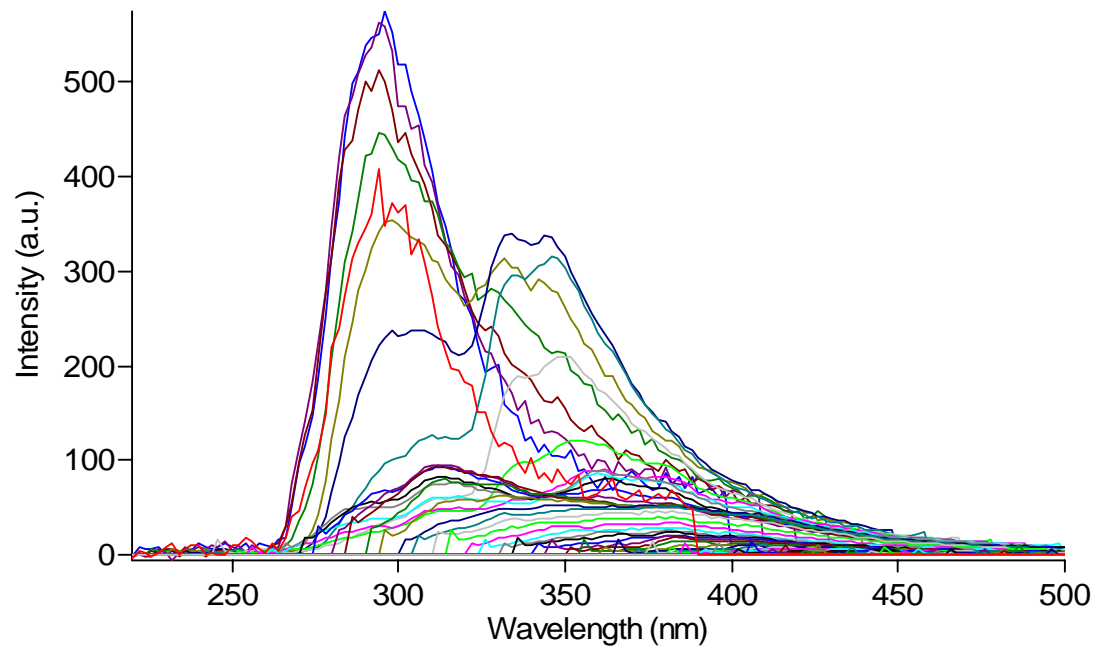
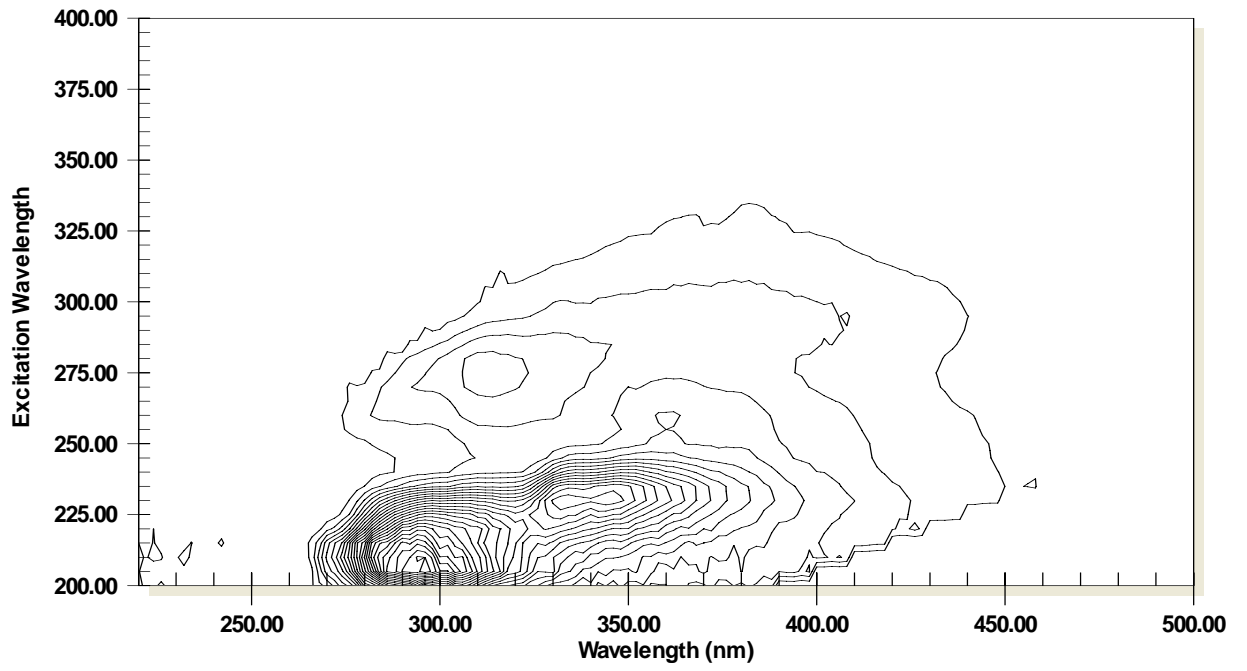


72AP

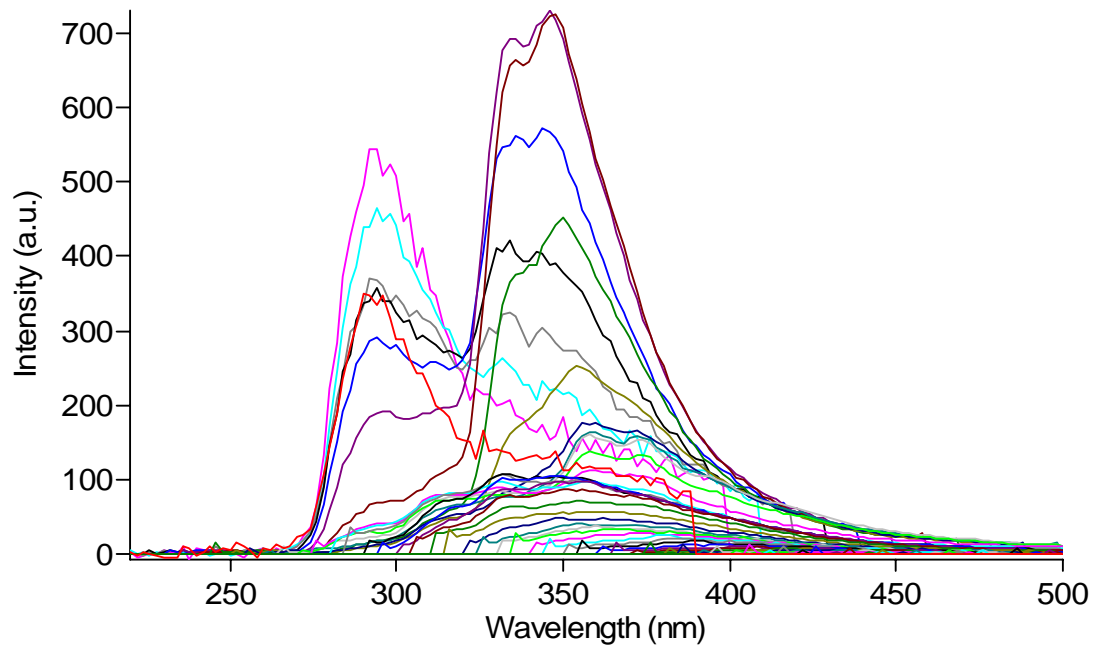
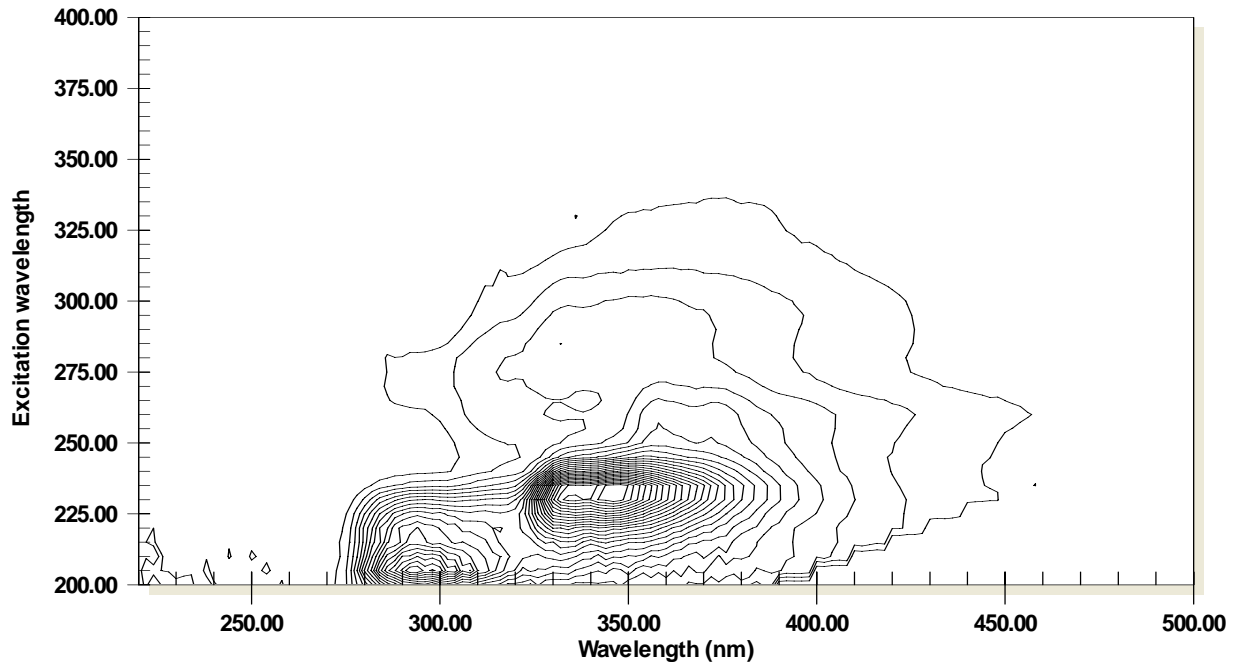




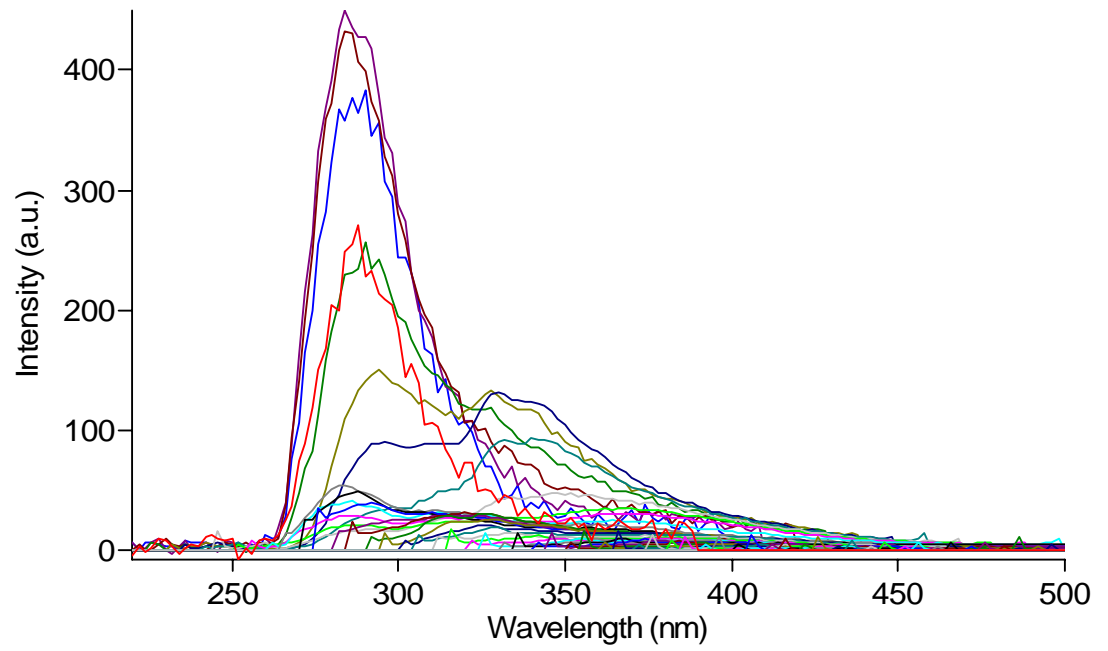
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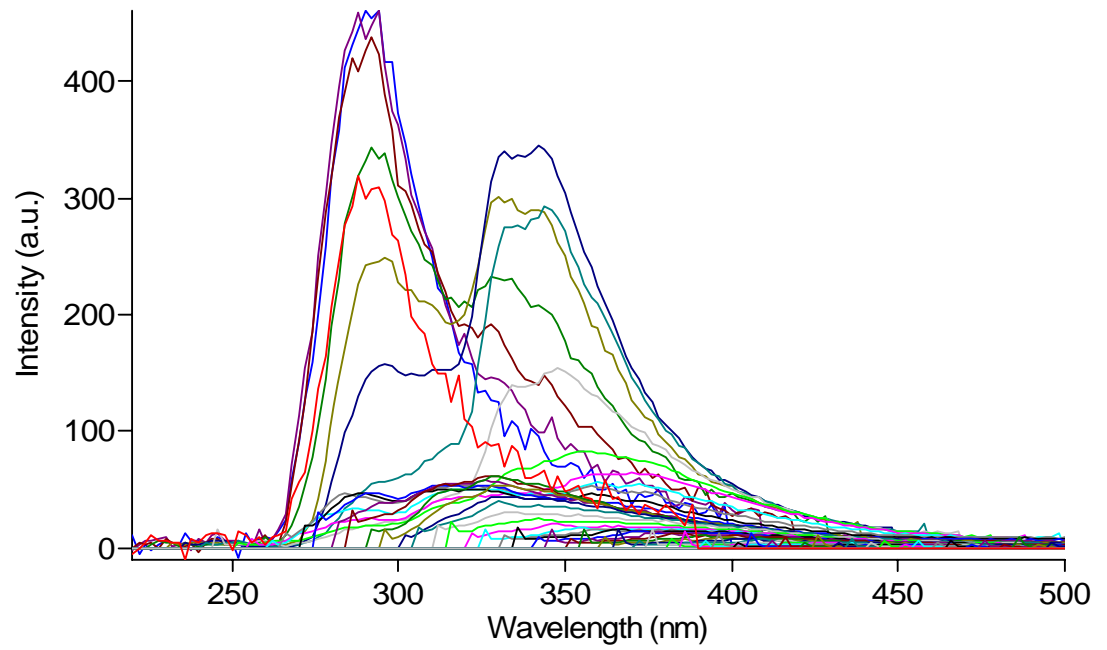
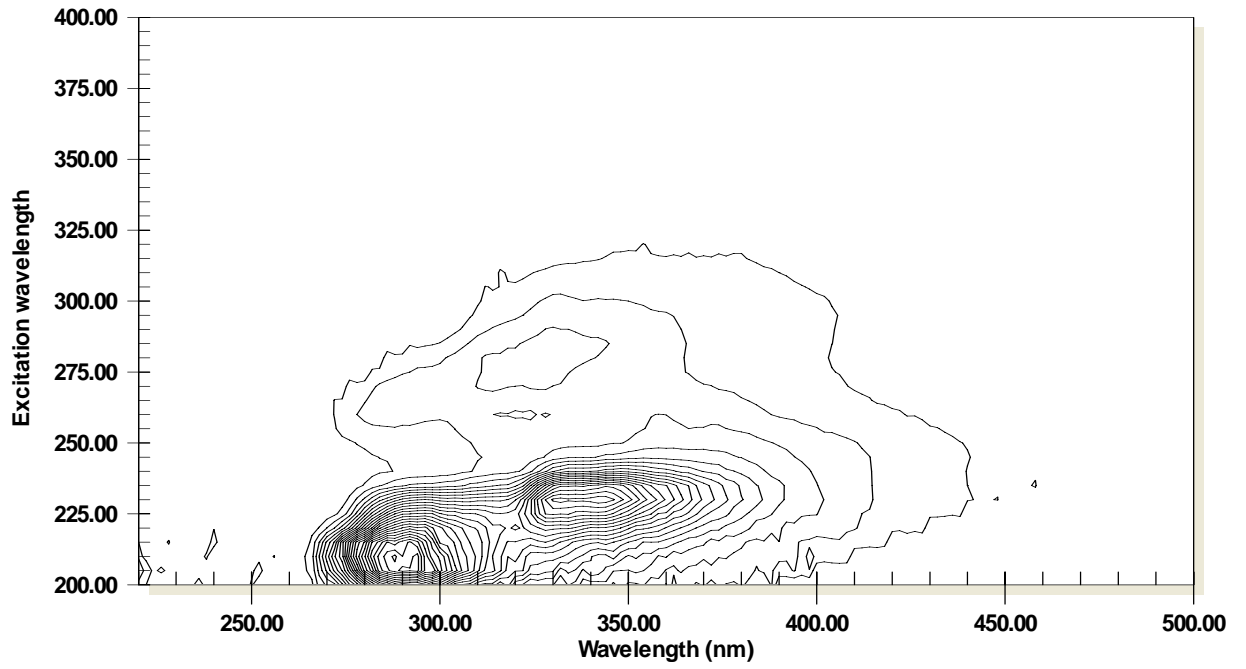
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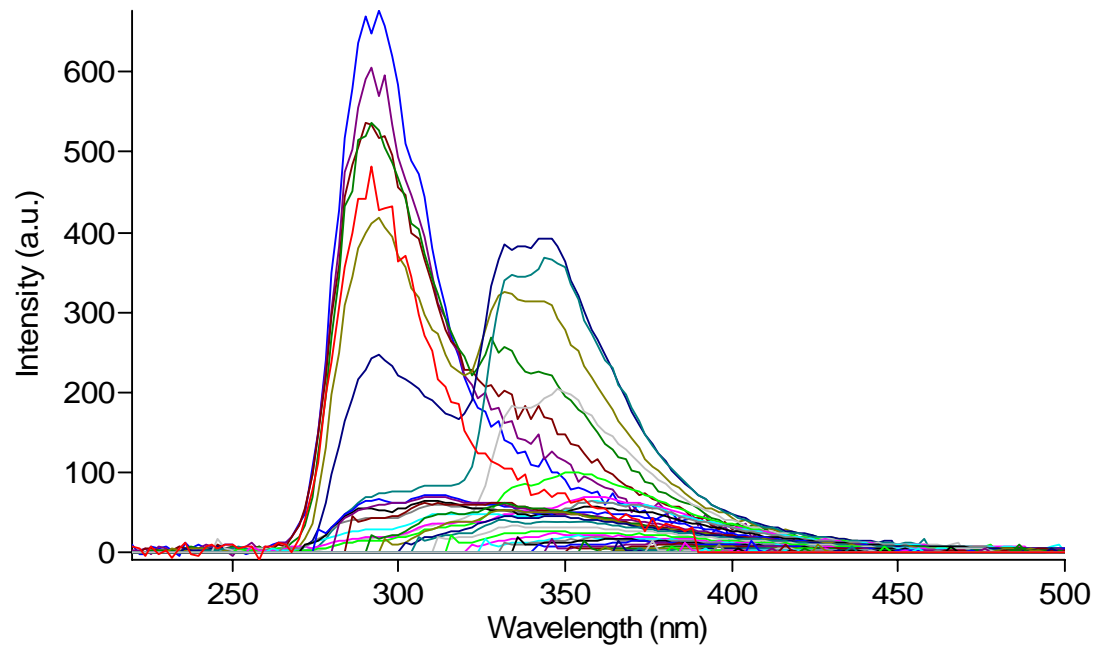
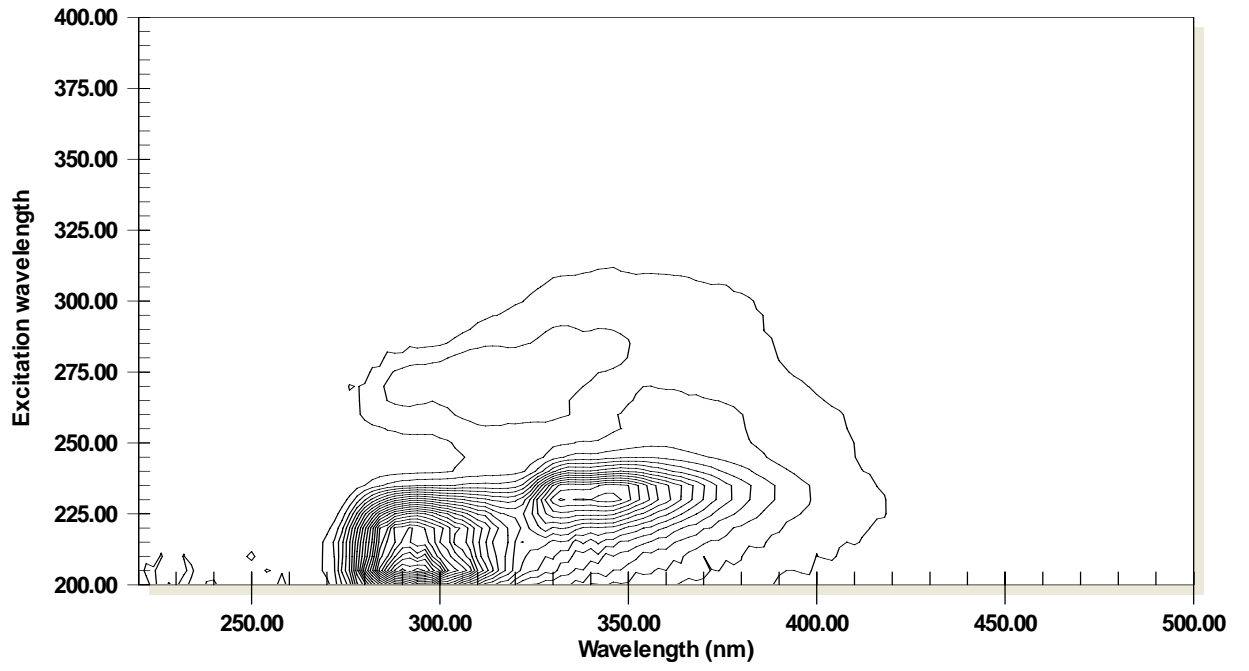
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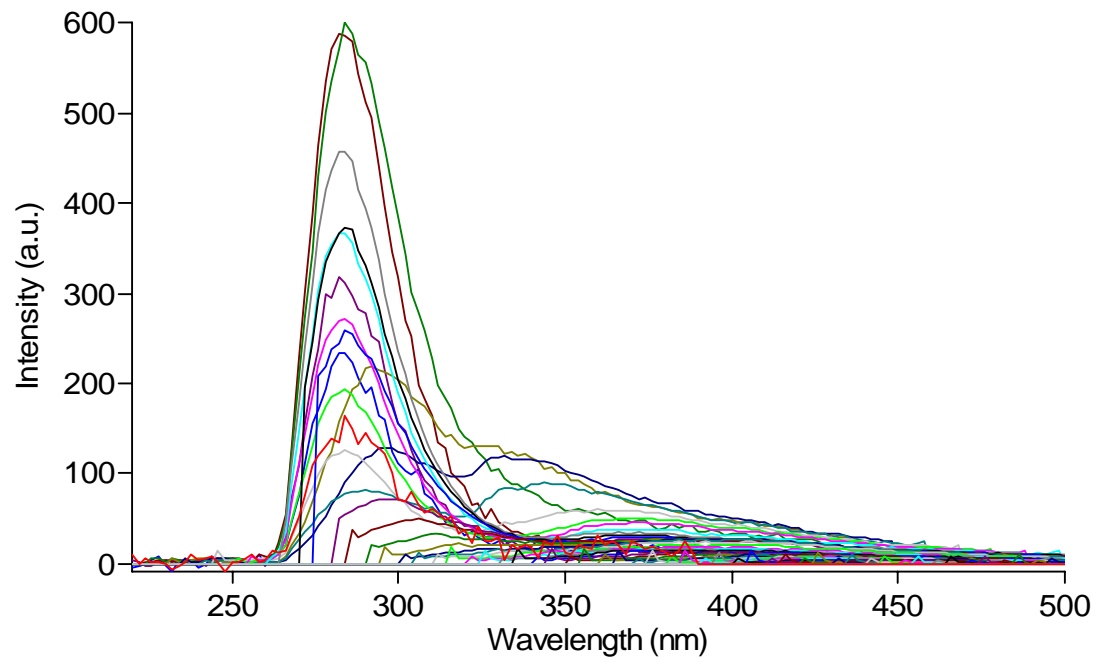
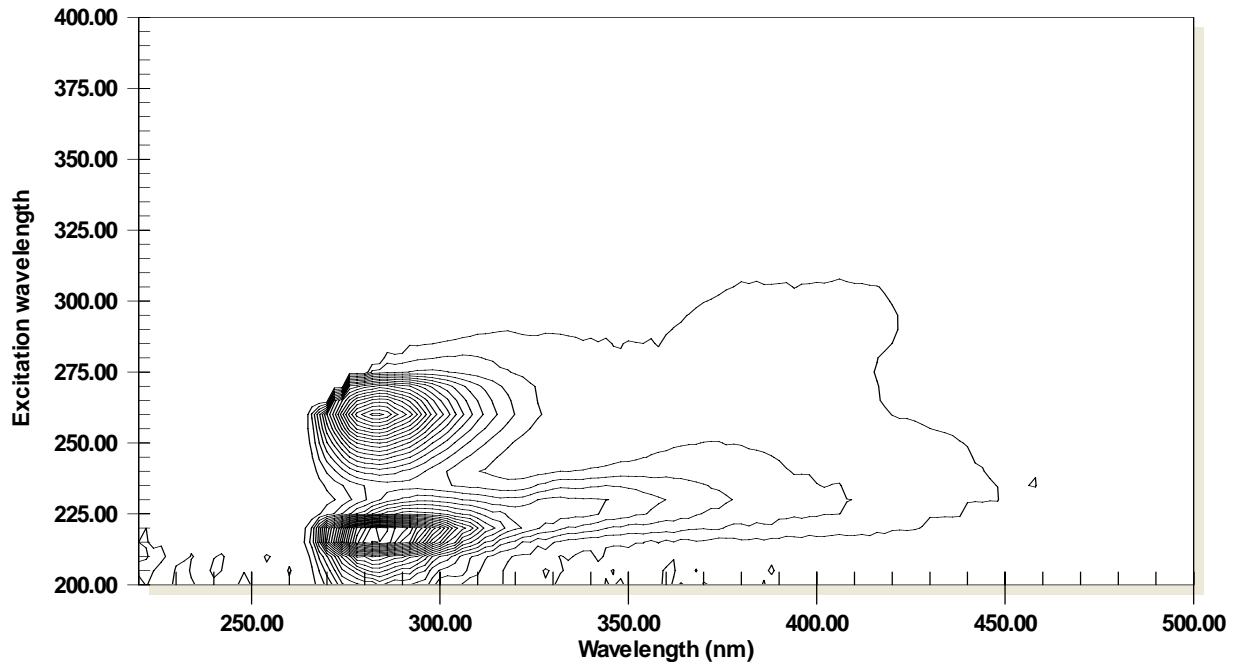
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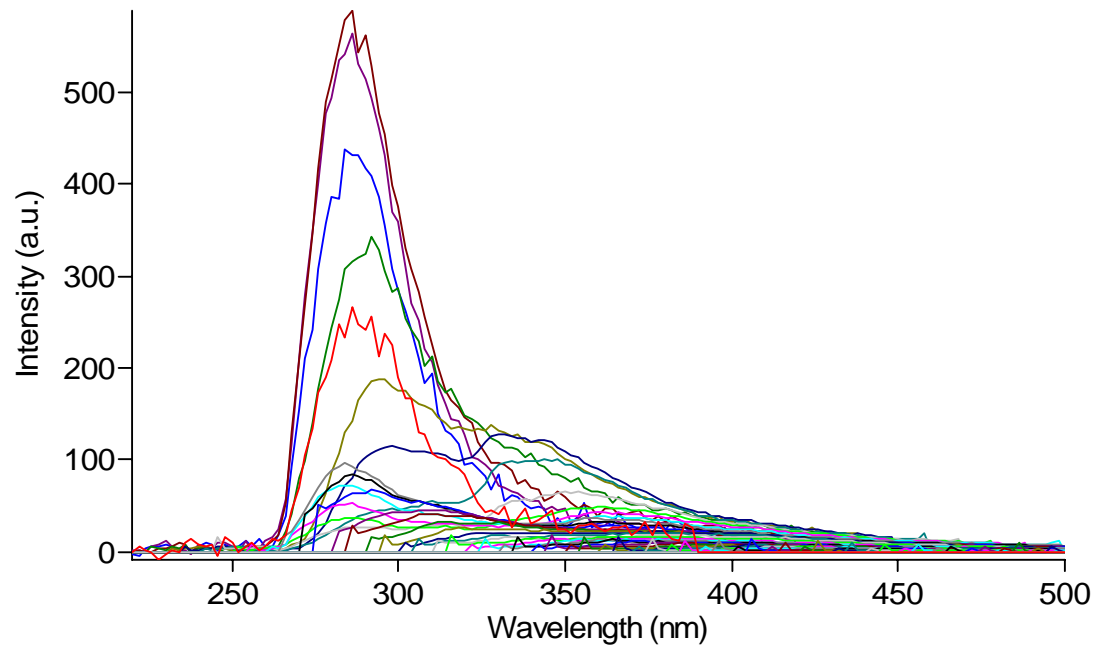
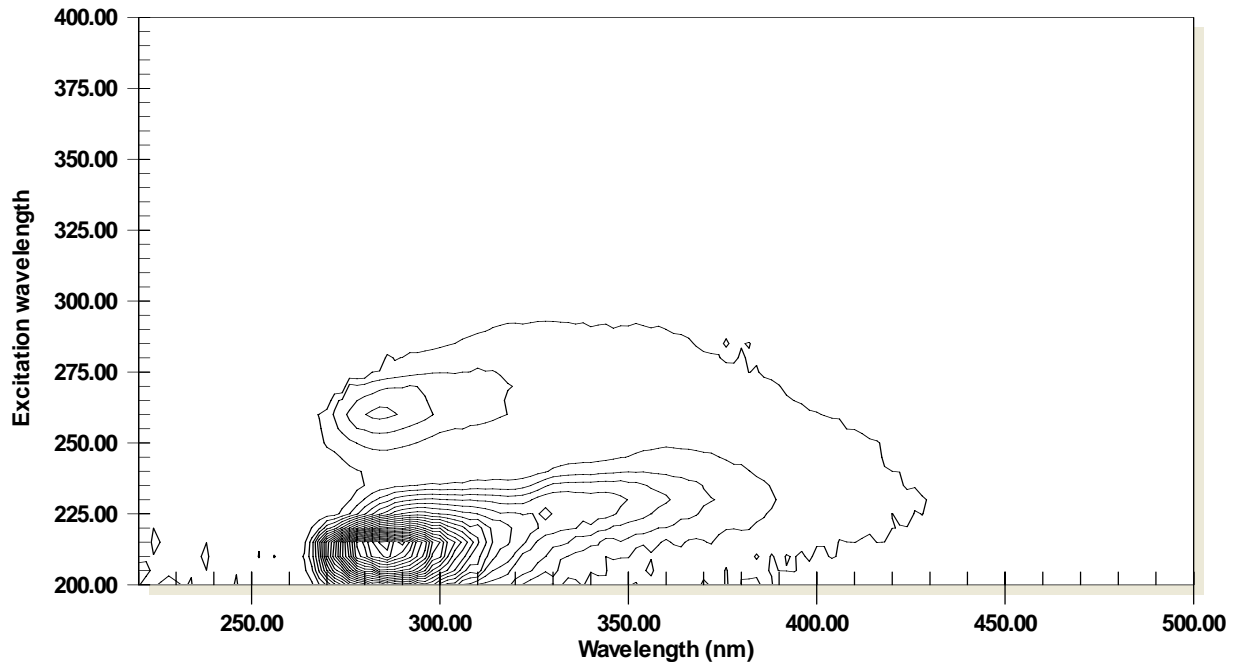
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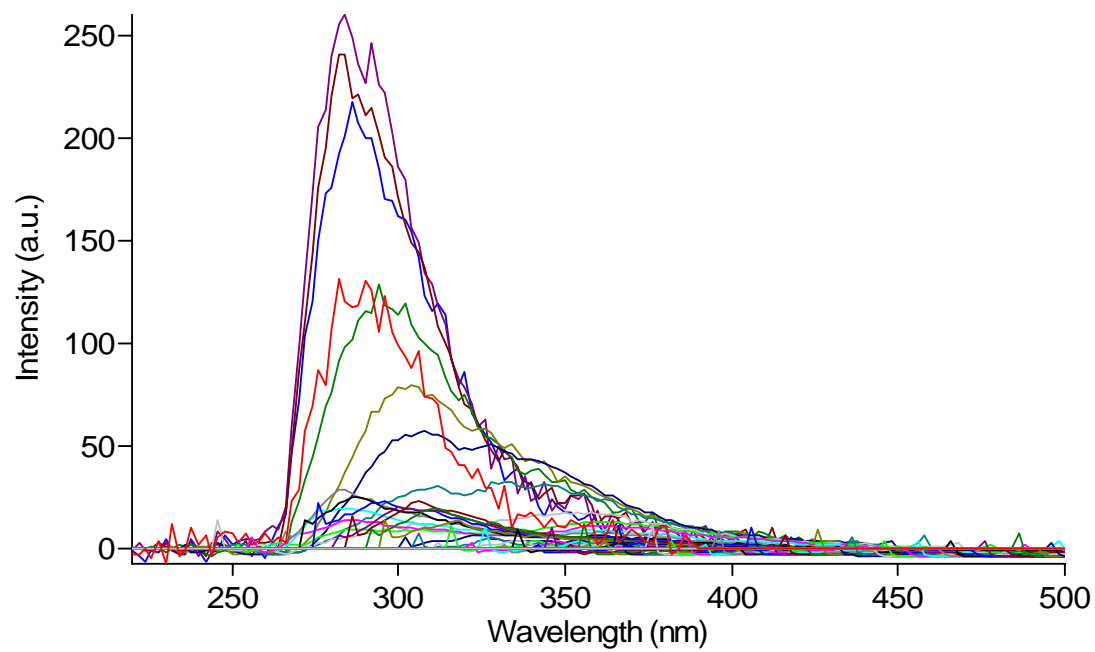
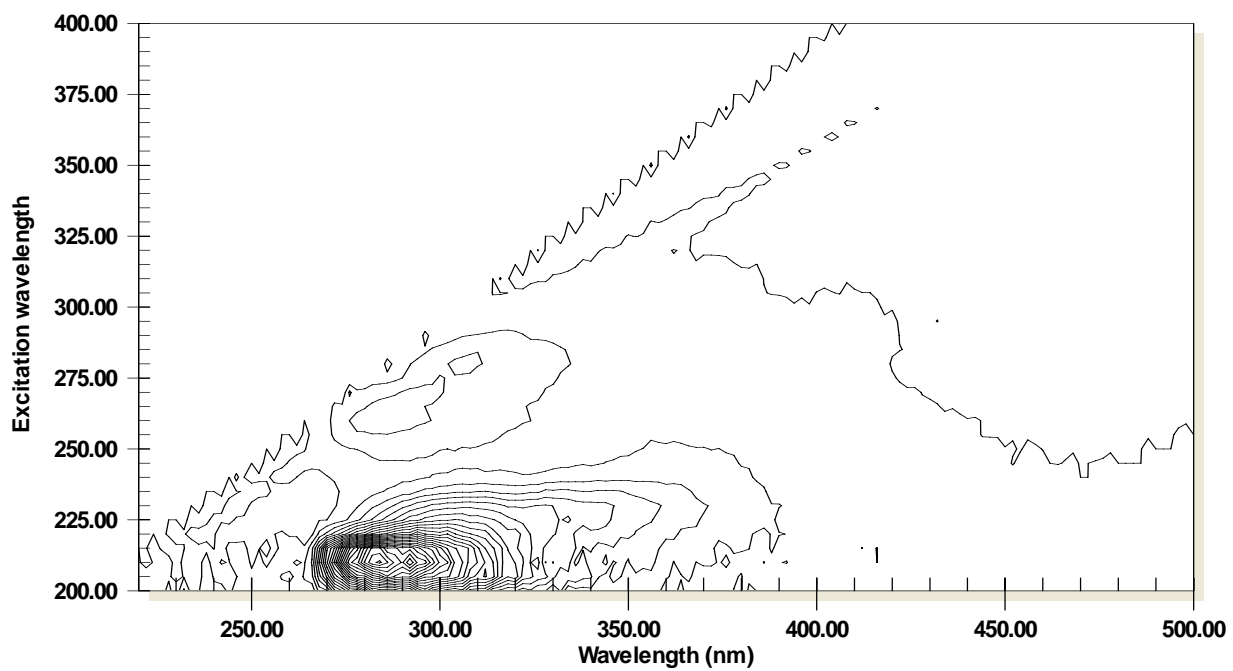
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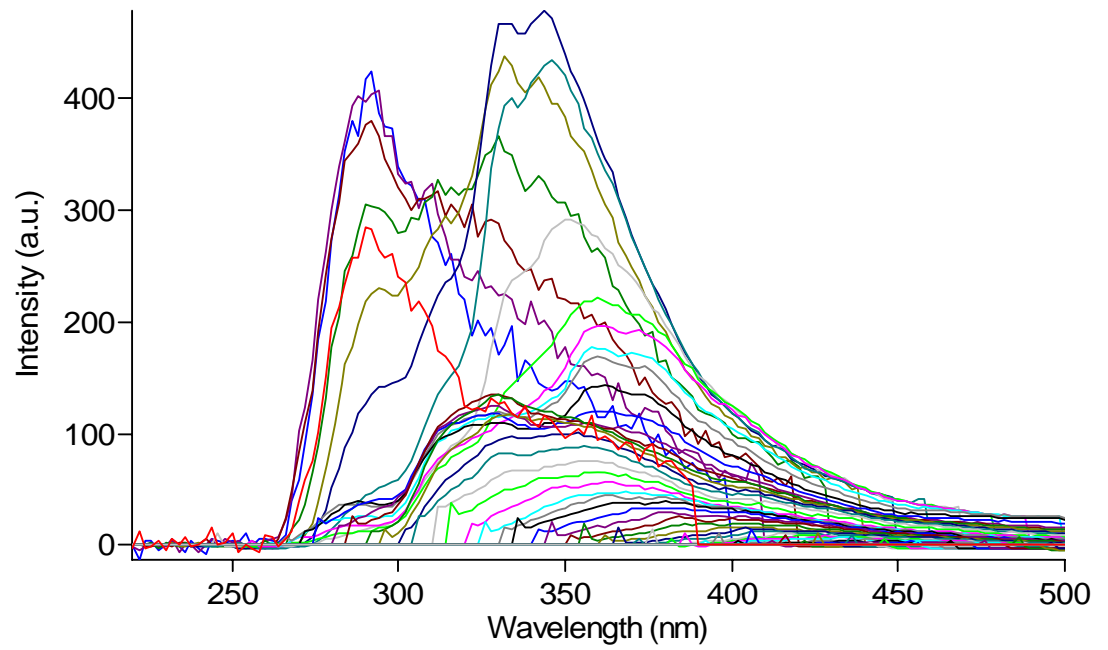
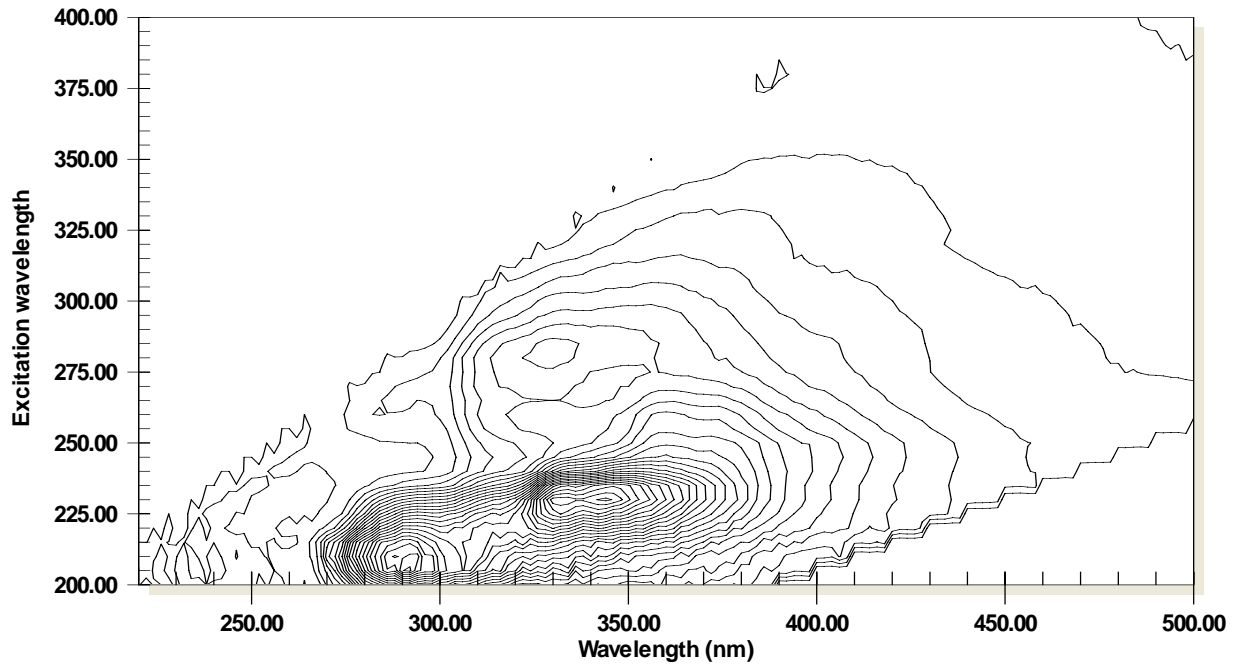
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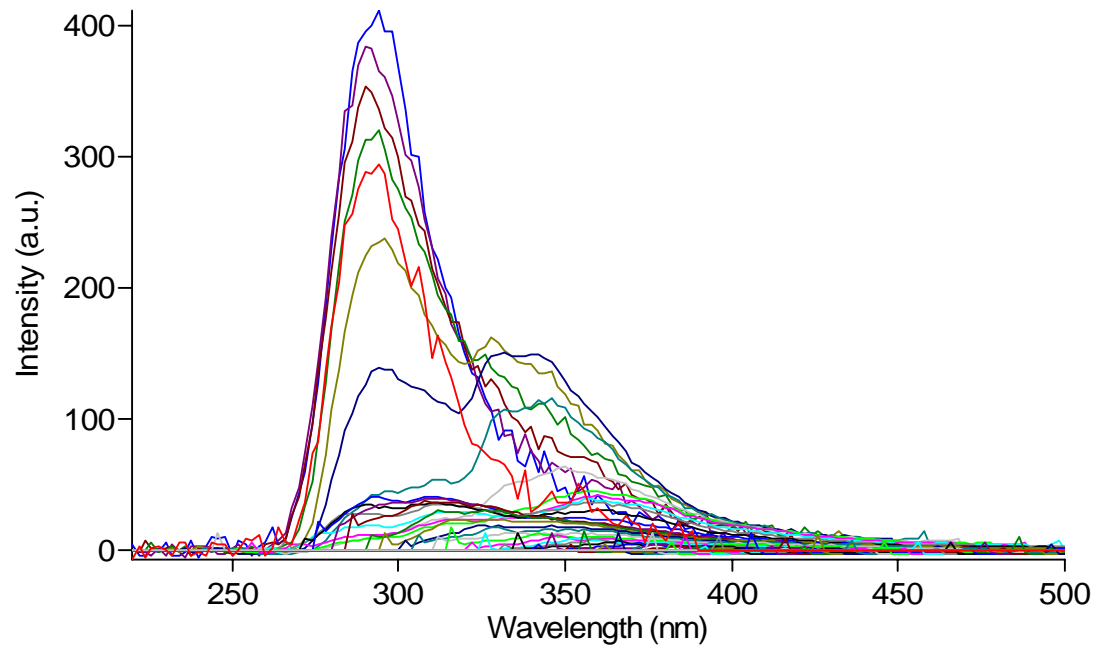
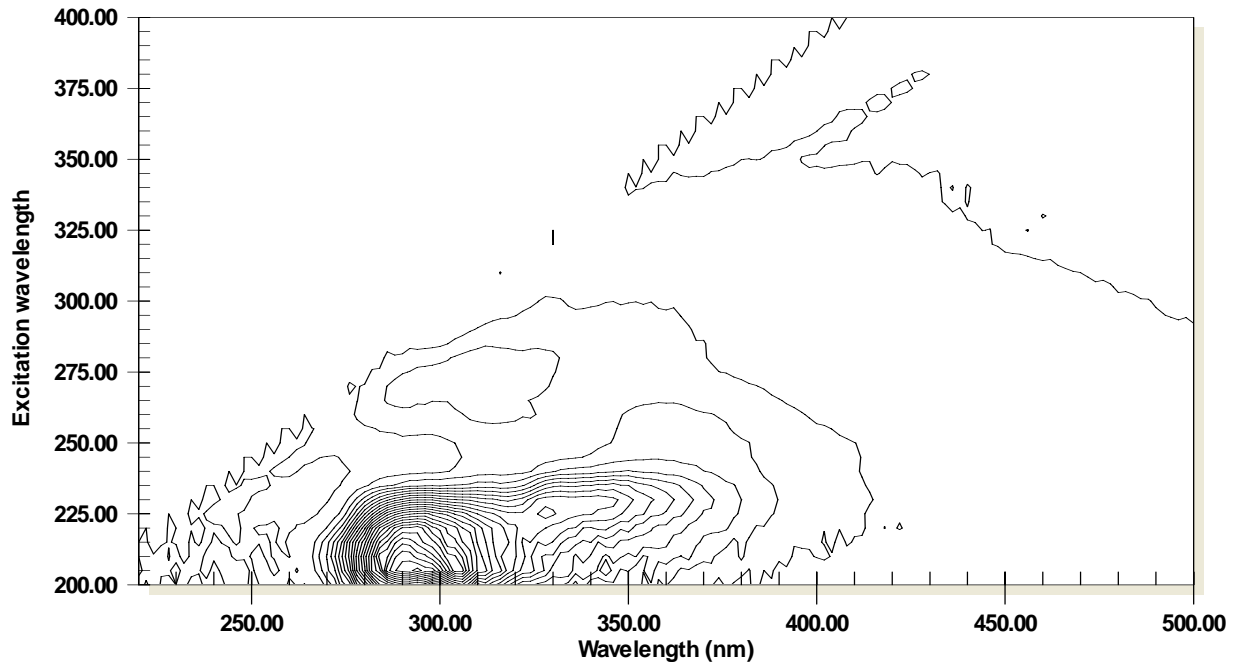
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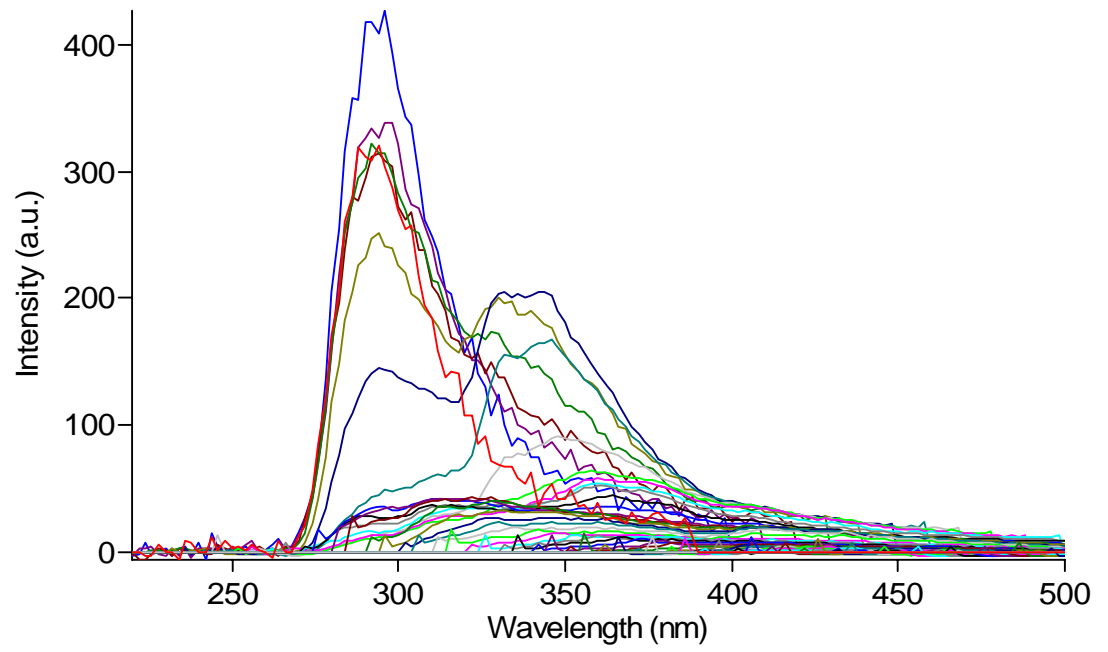
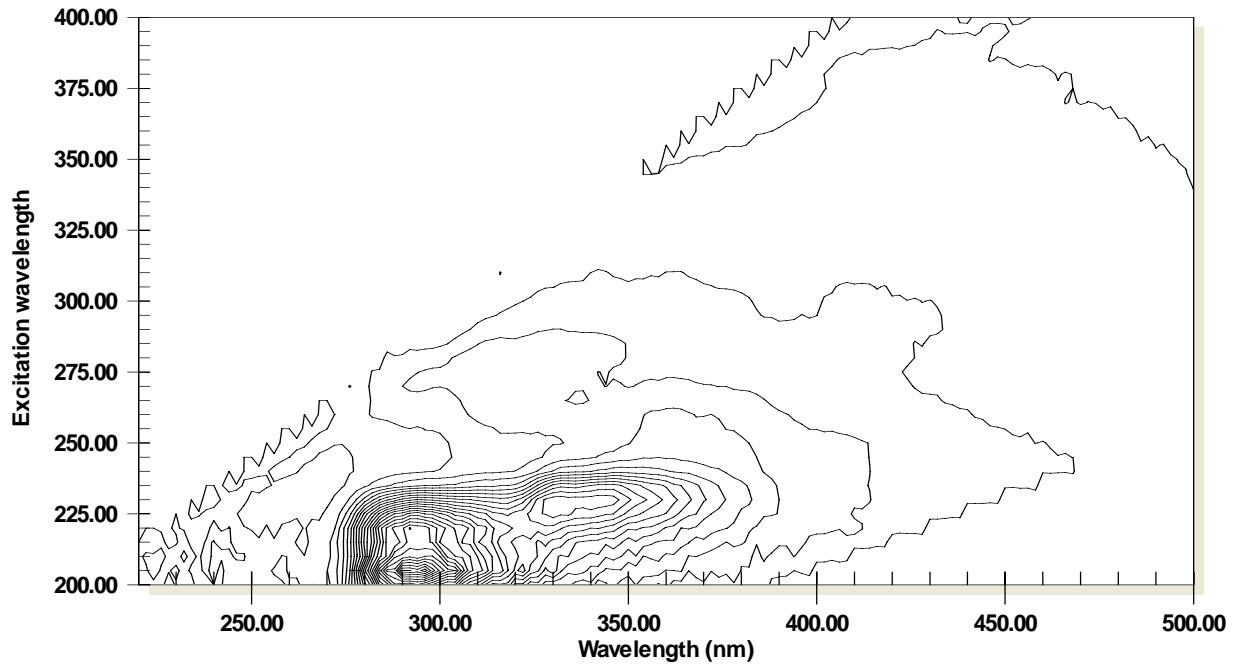
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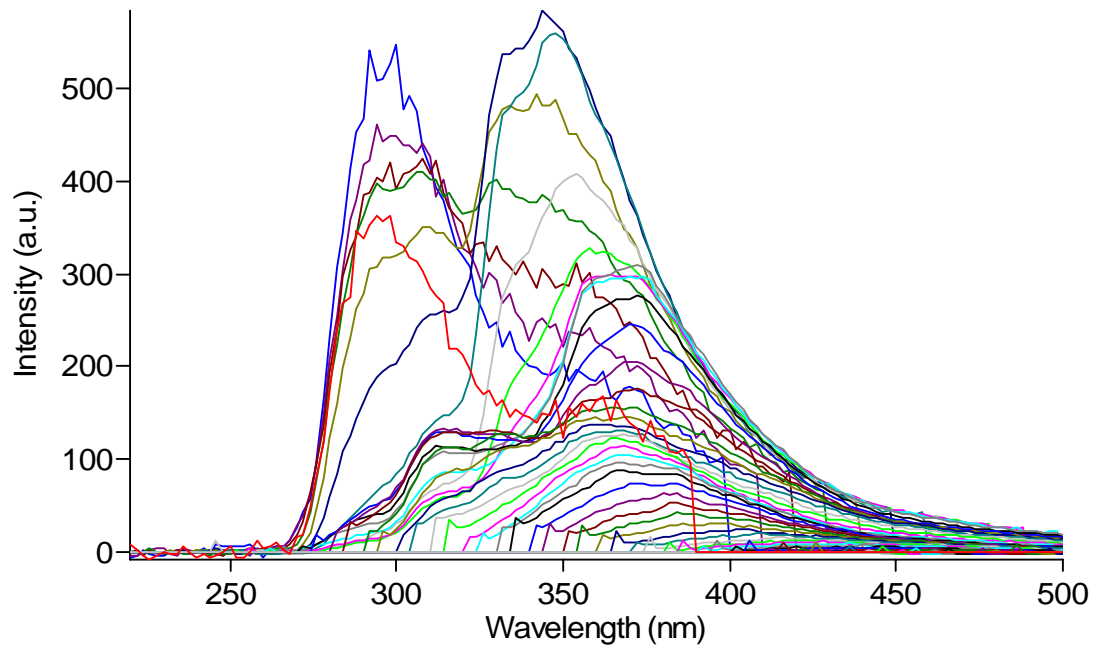
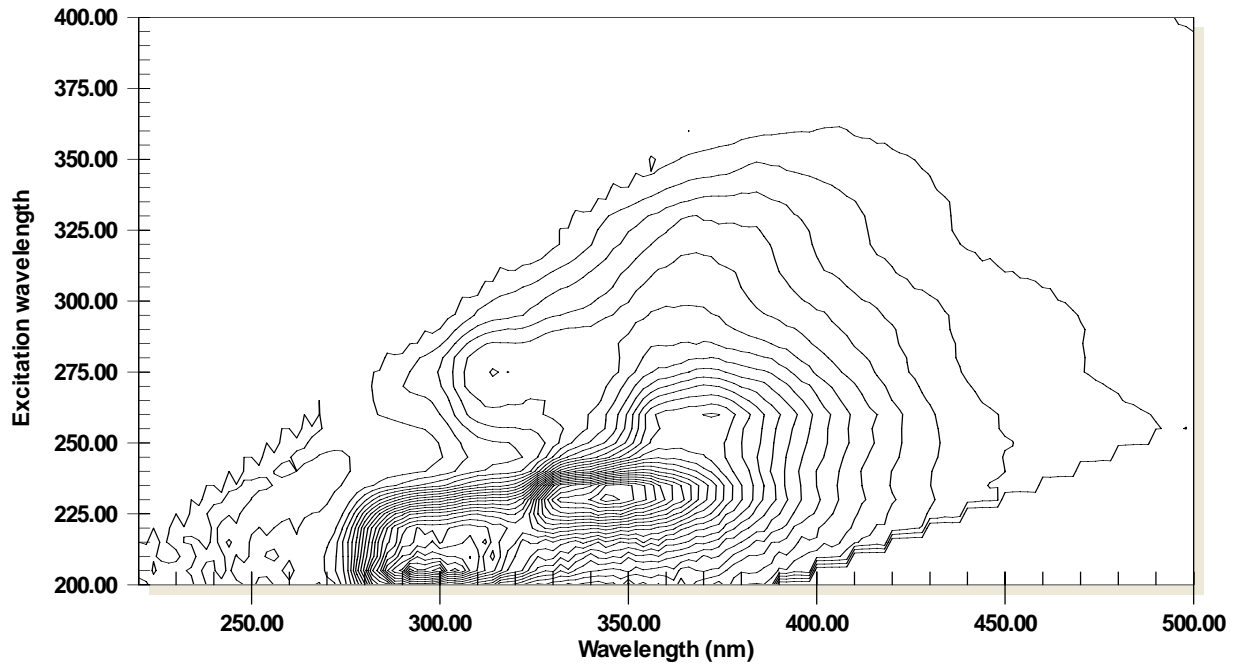
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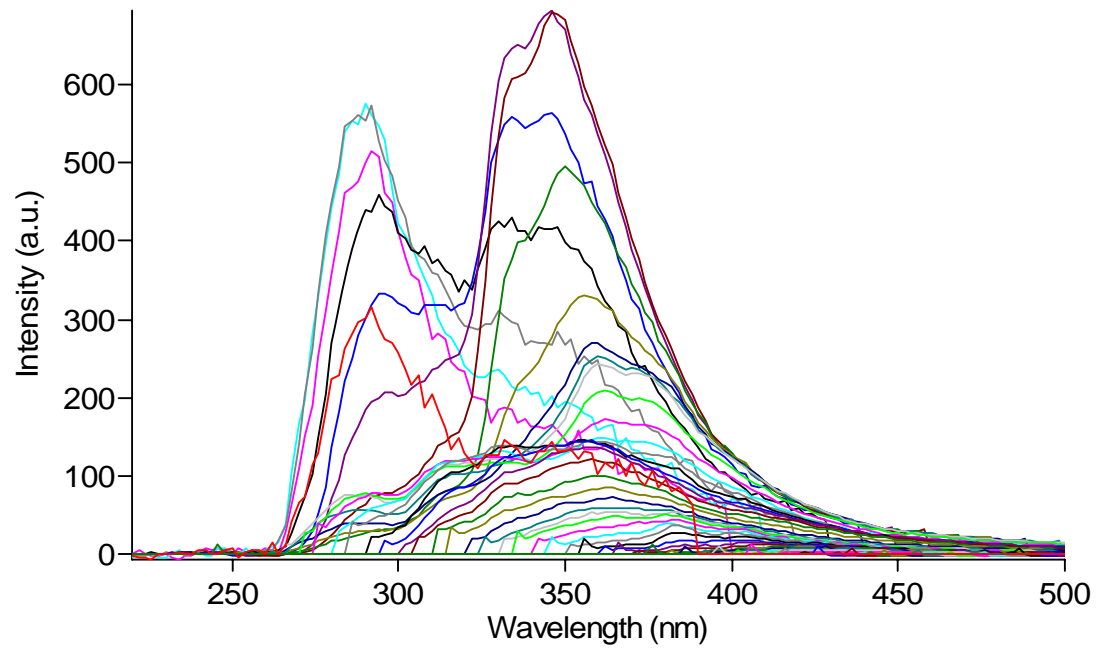
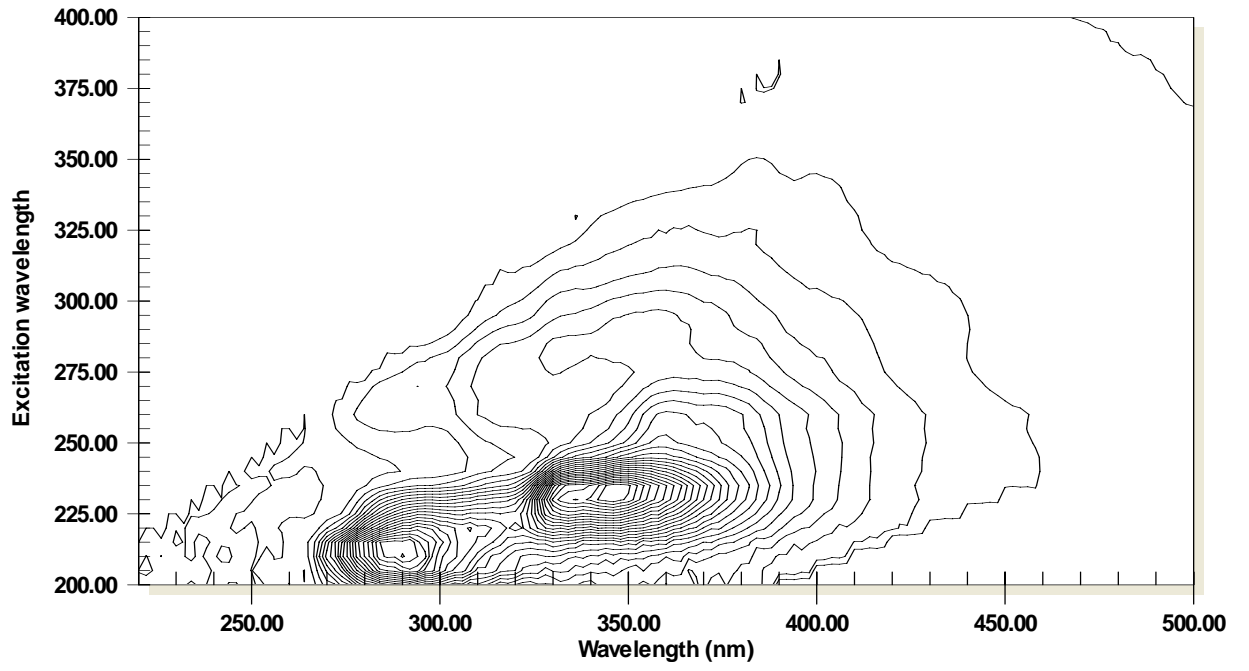
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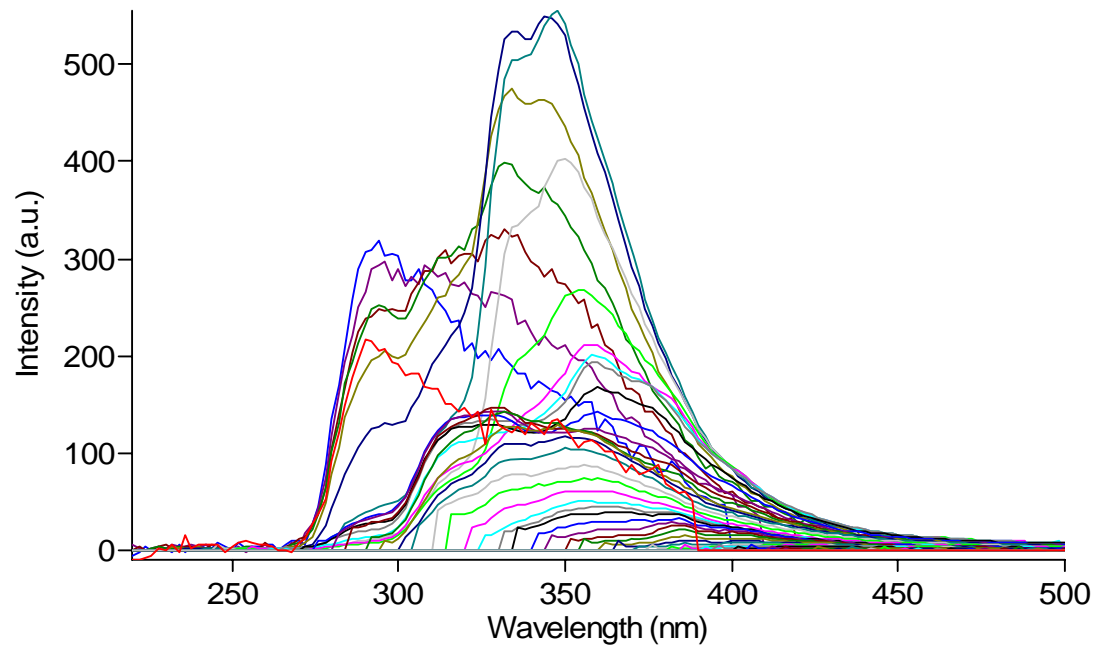
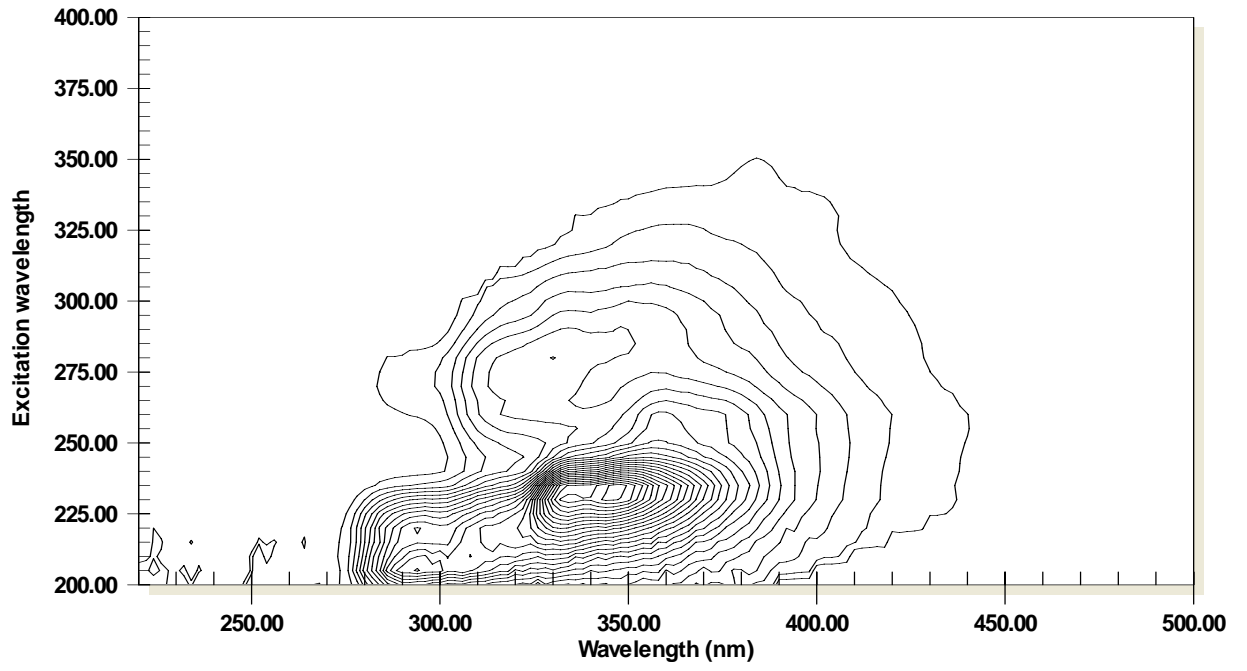
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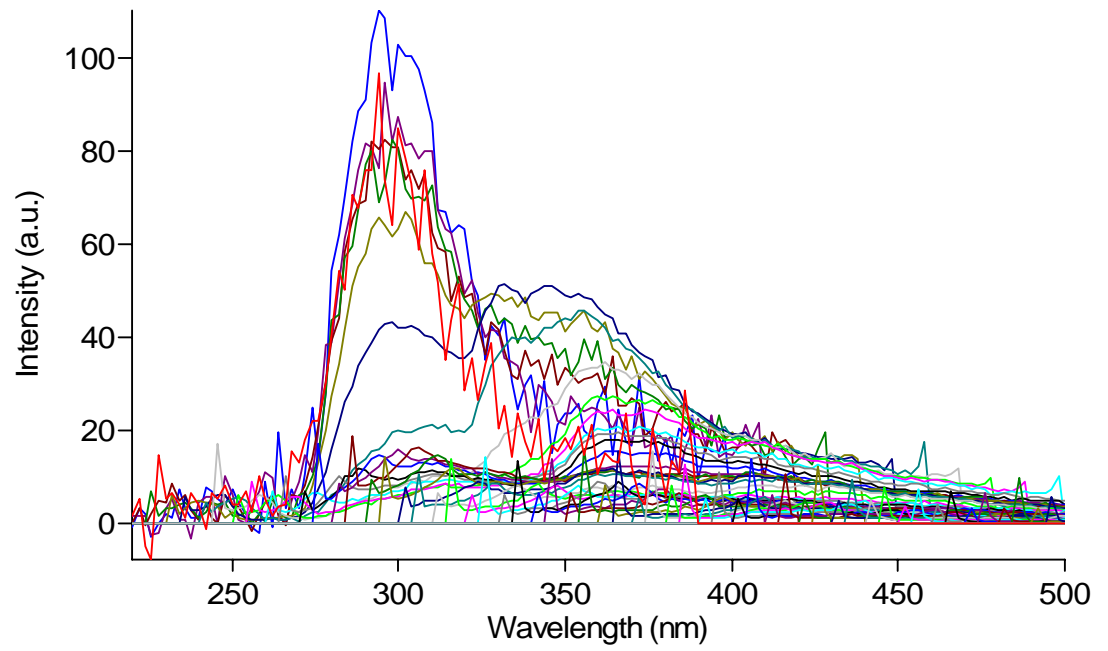
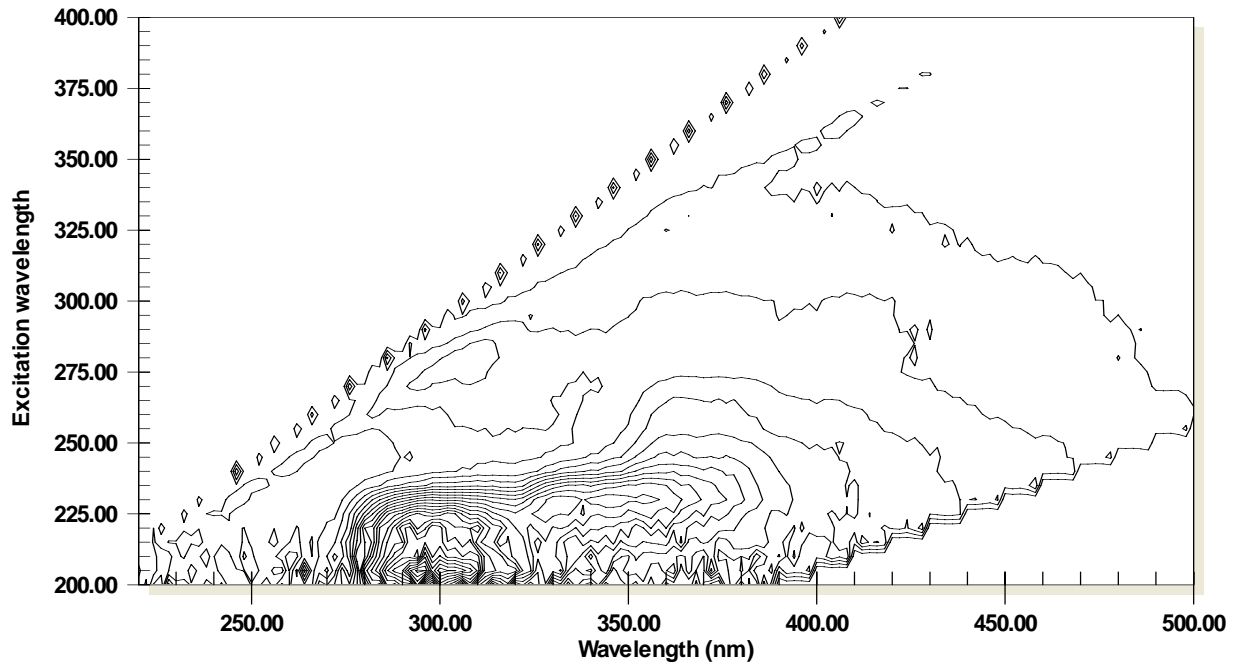
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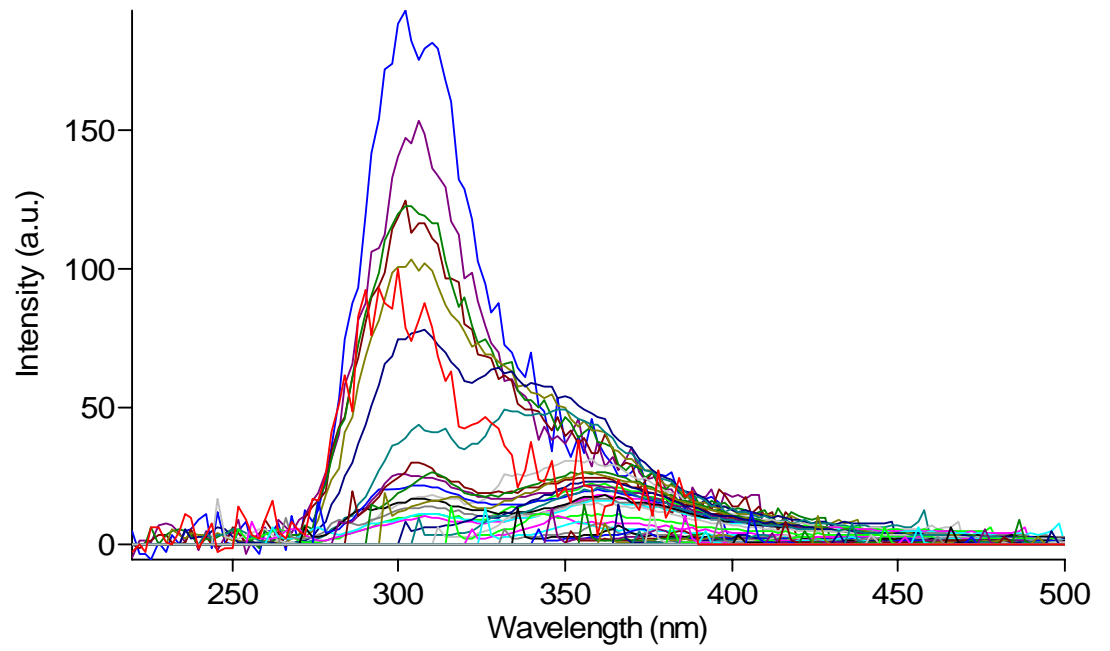
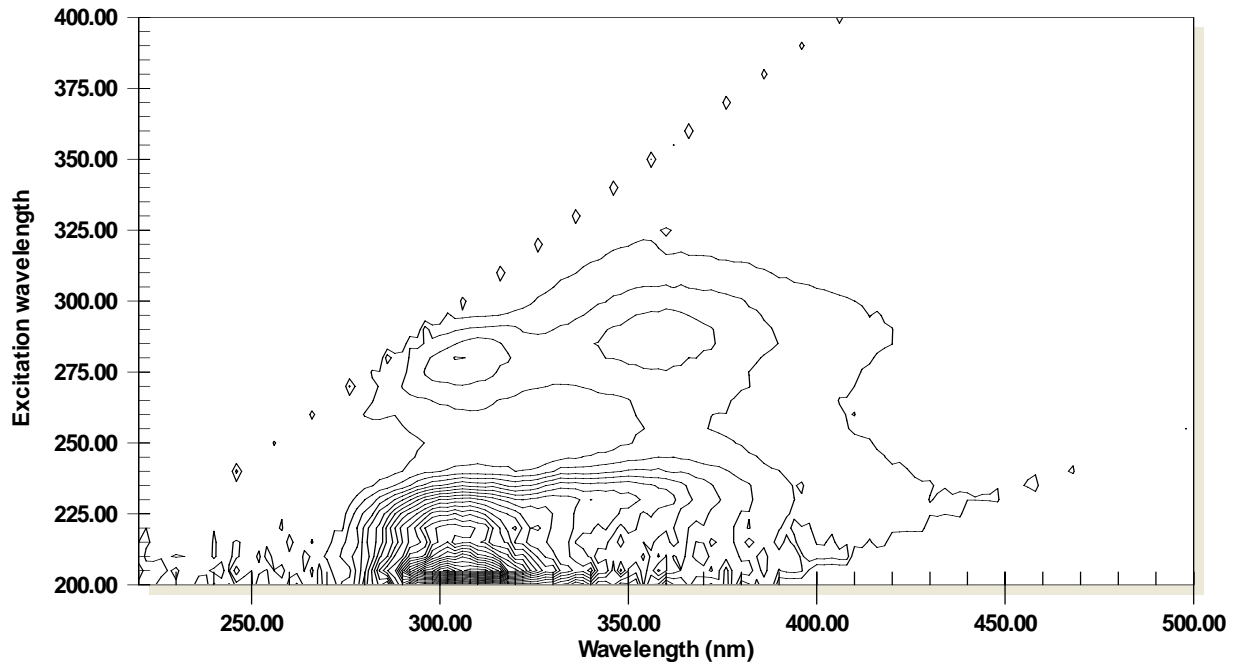


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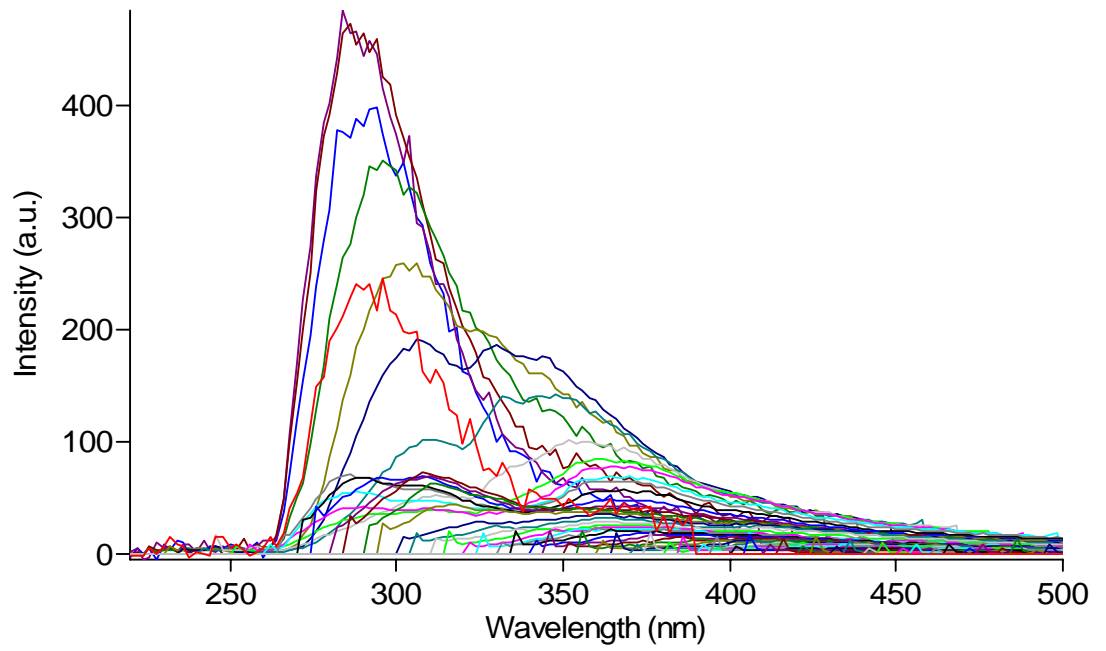
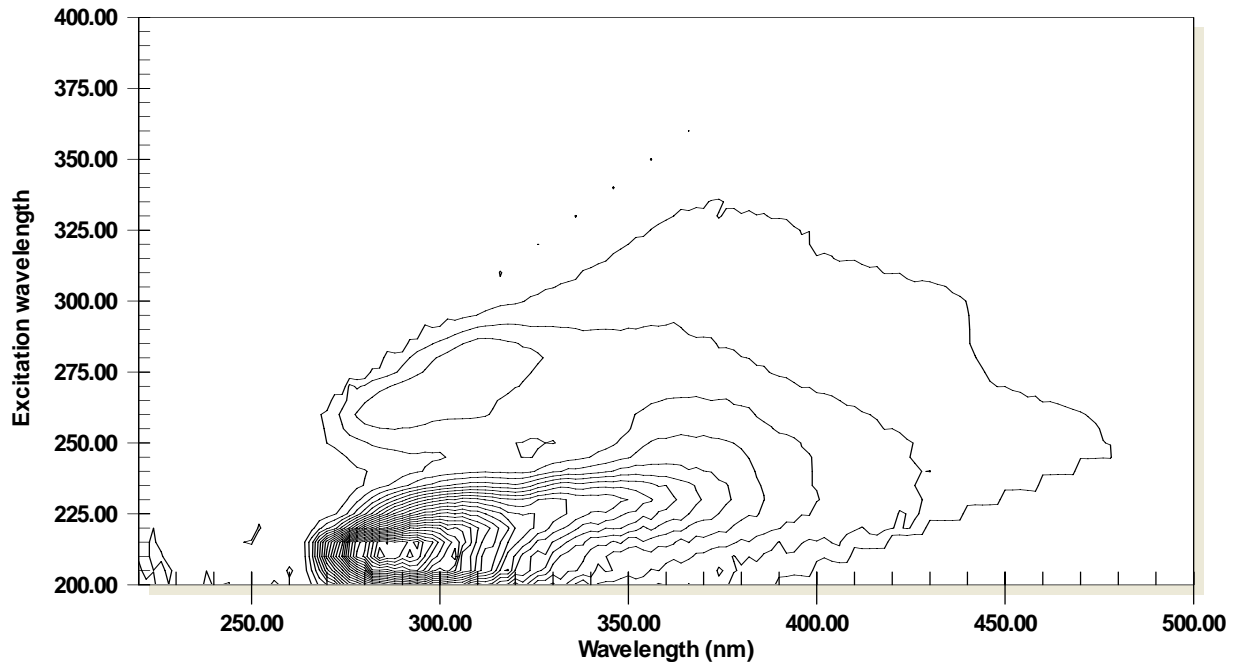


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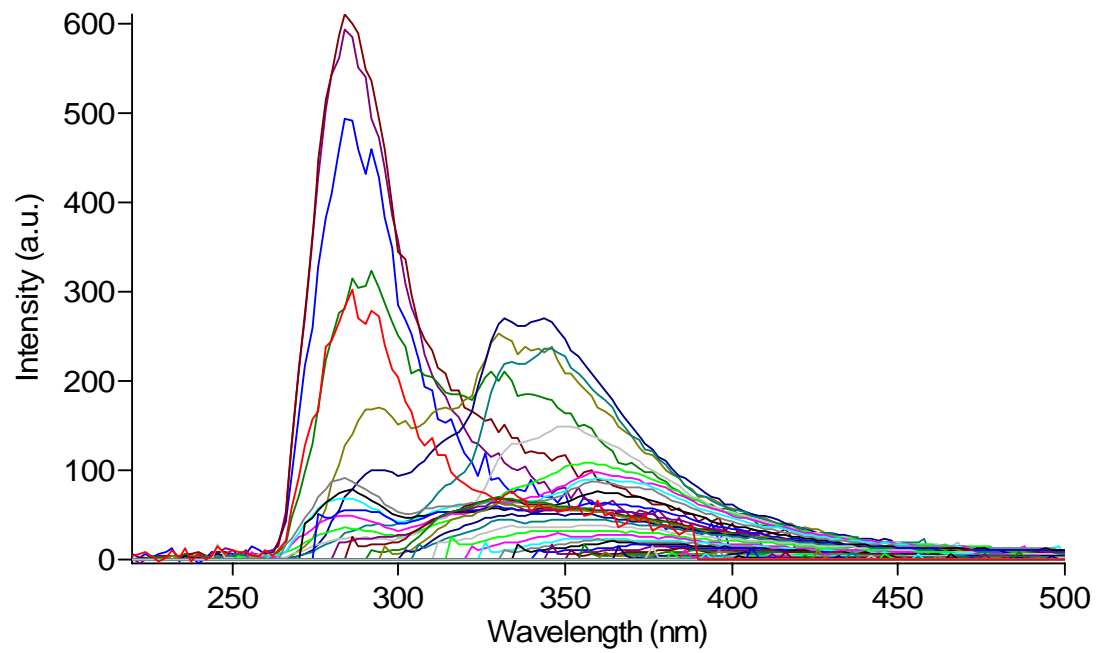
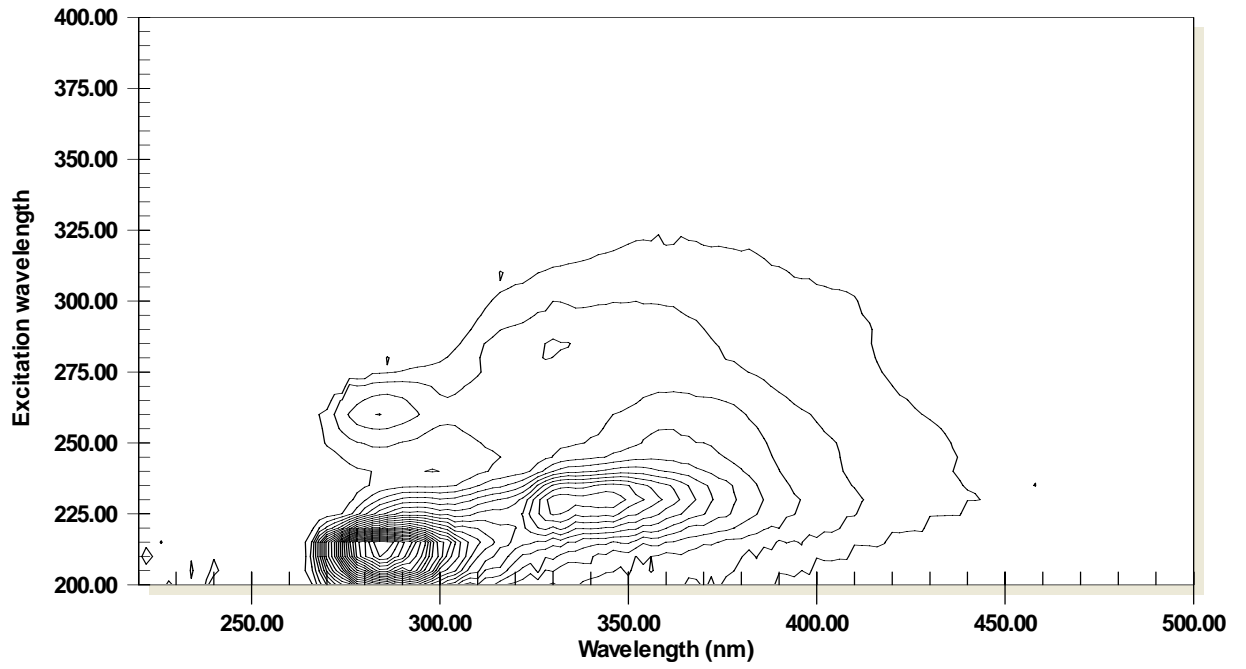




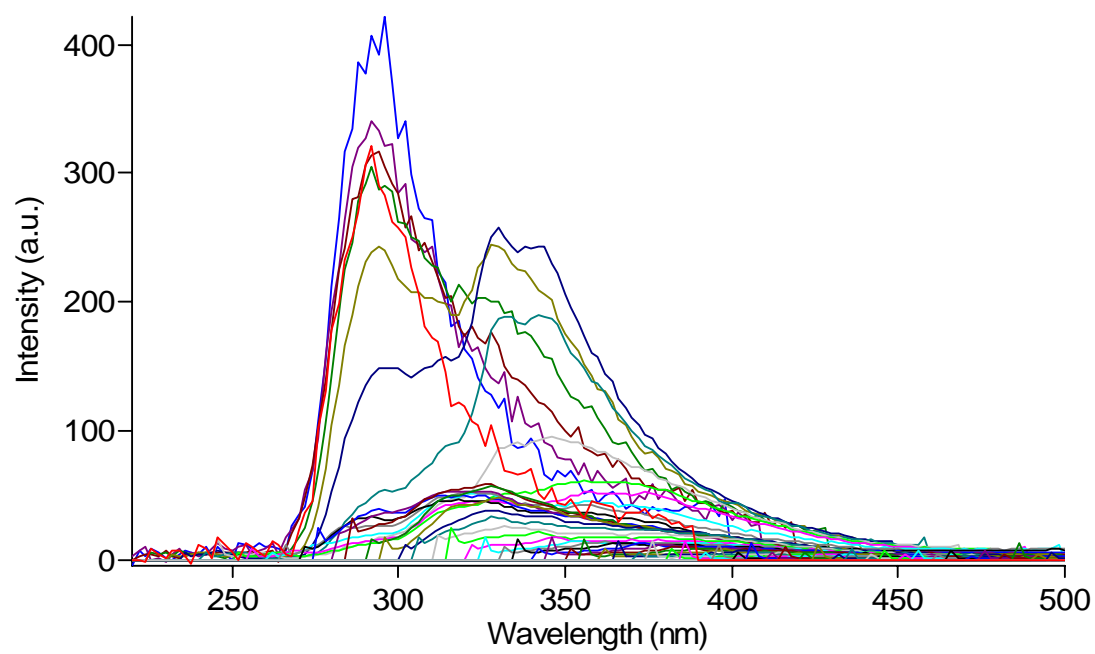
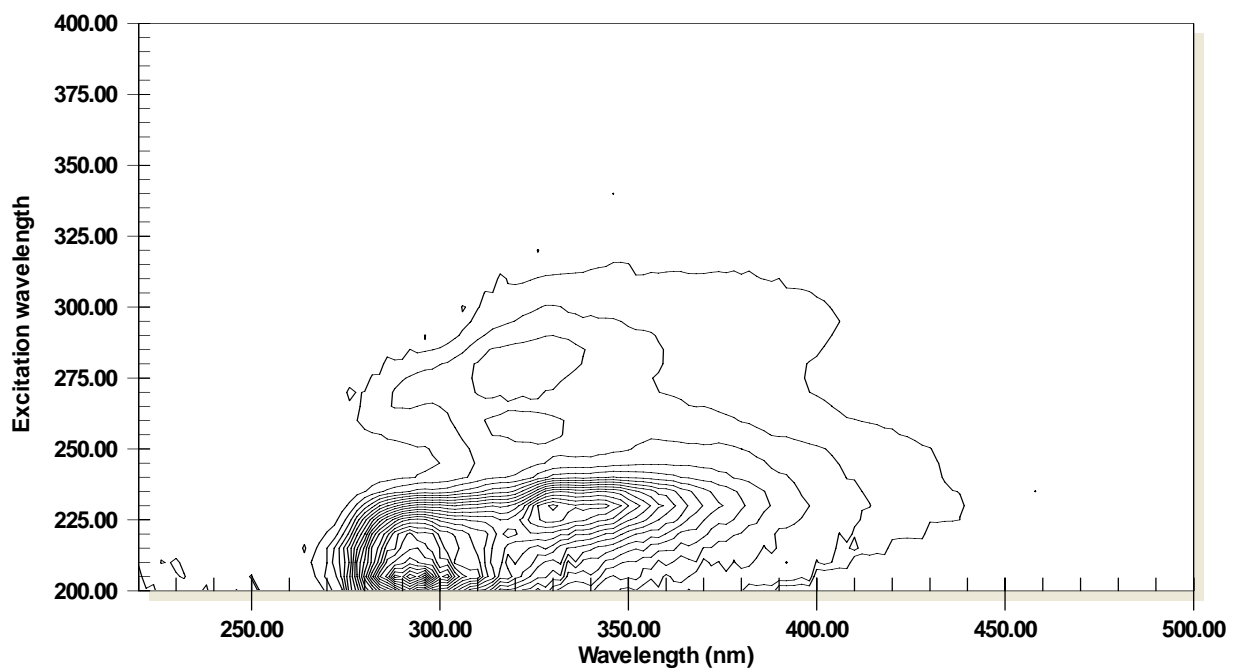
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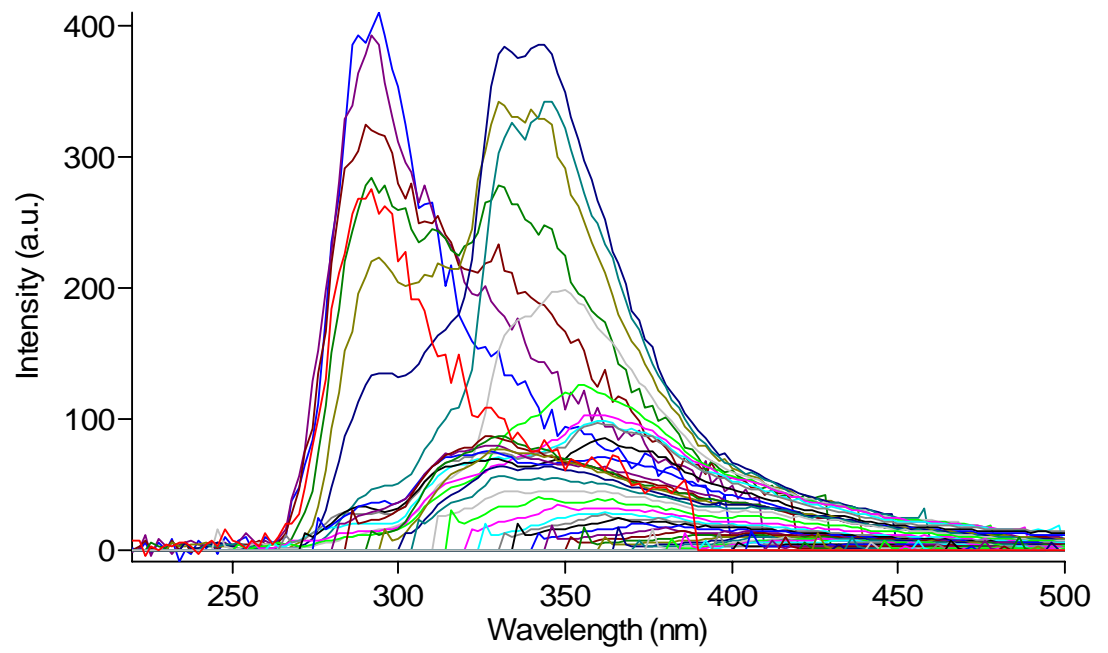
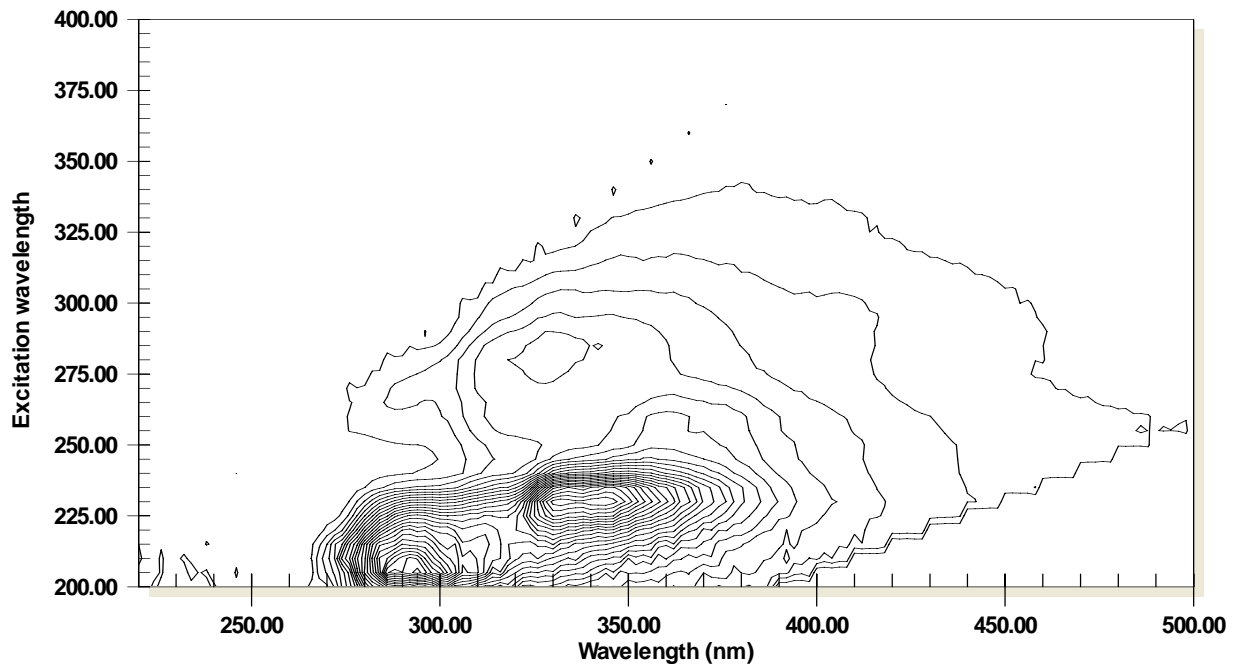
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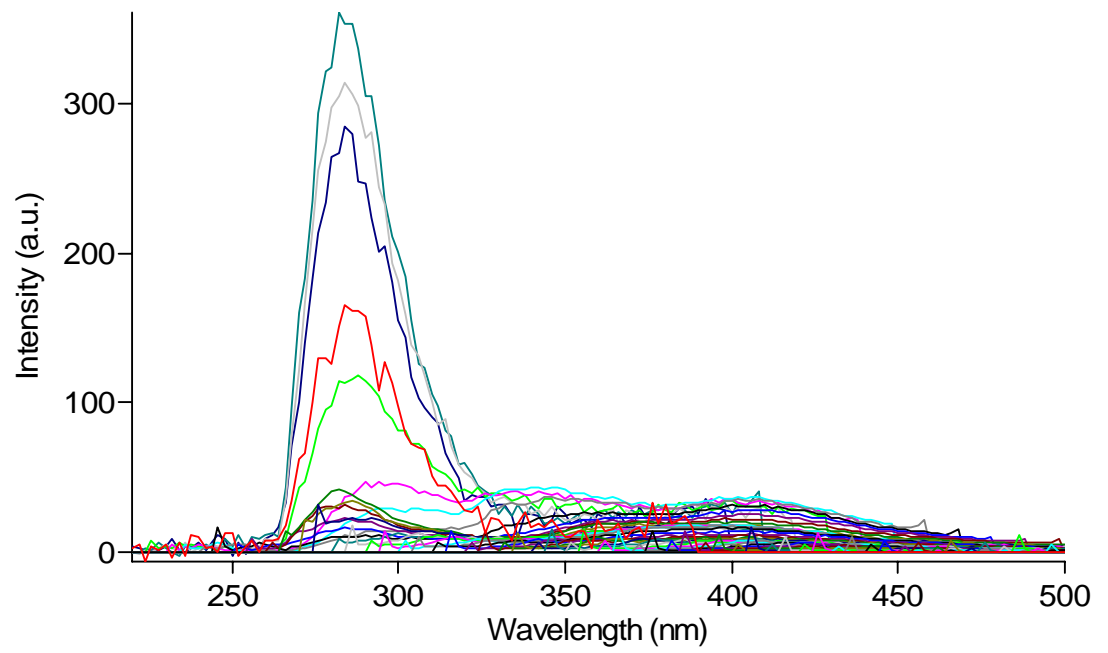
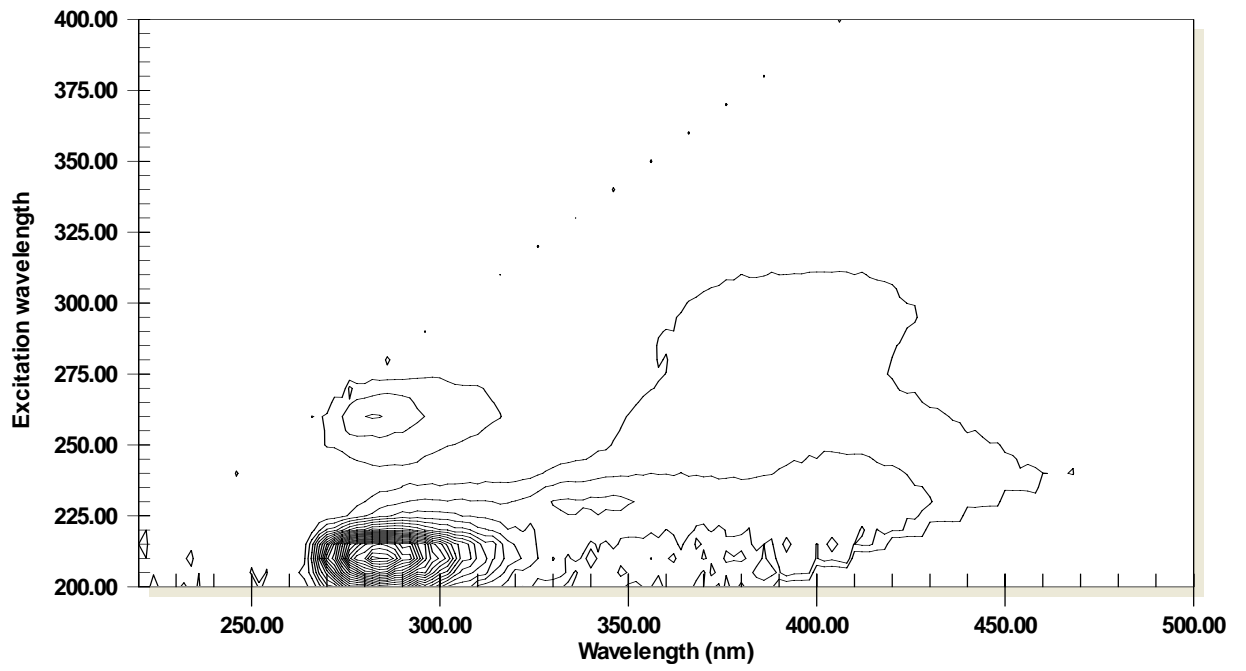
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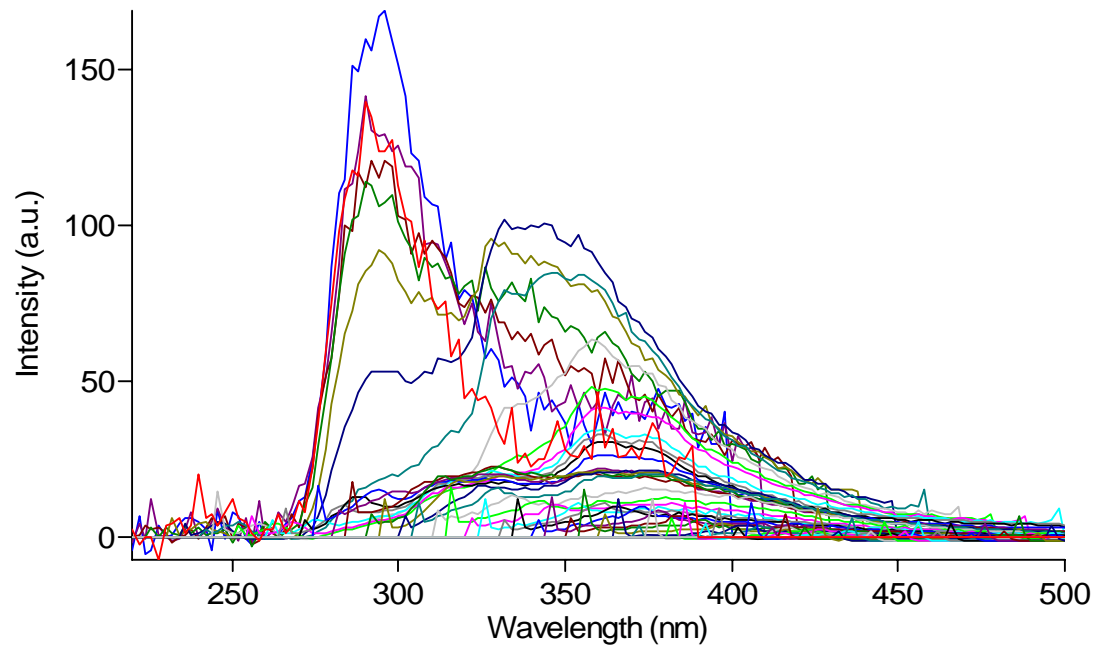
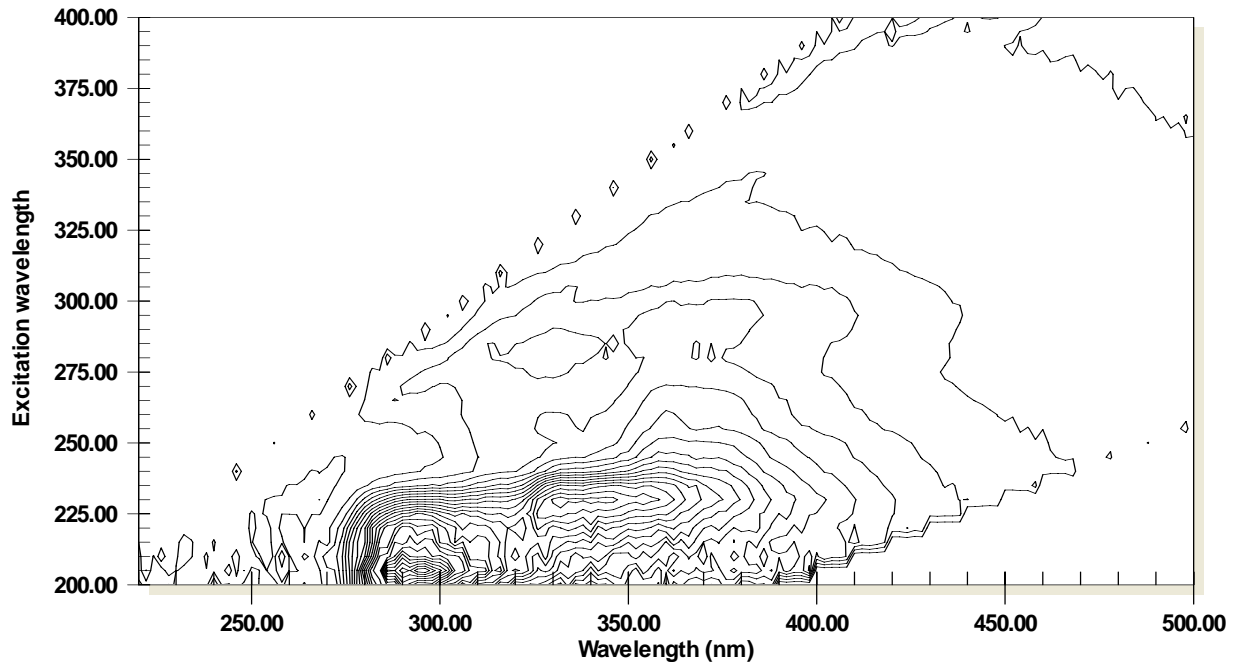
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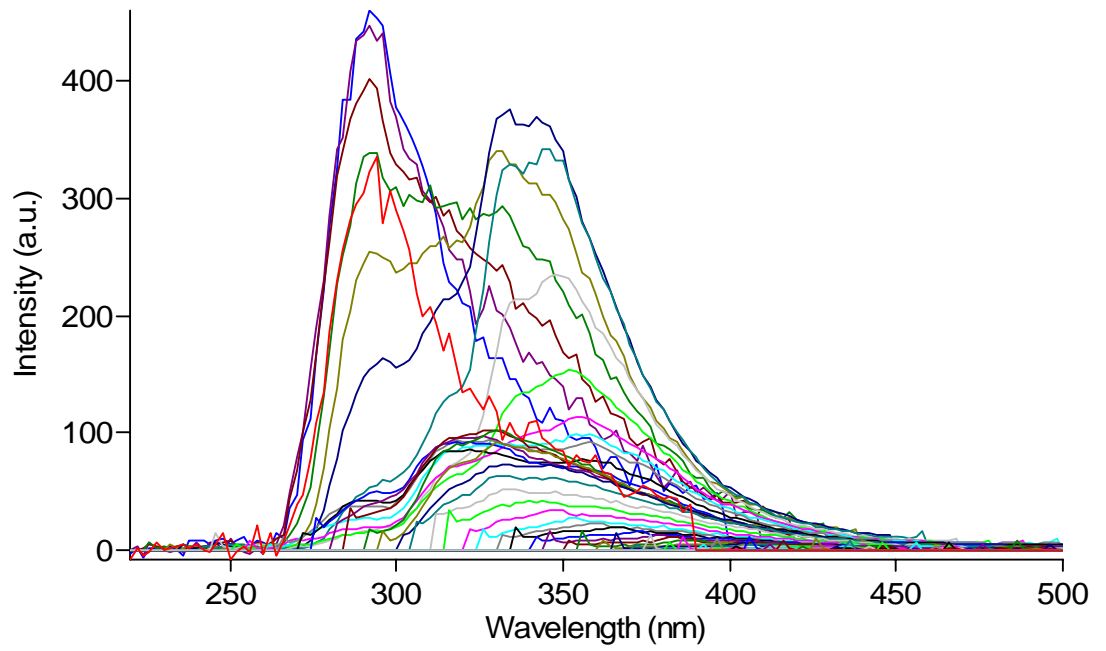
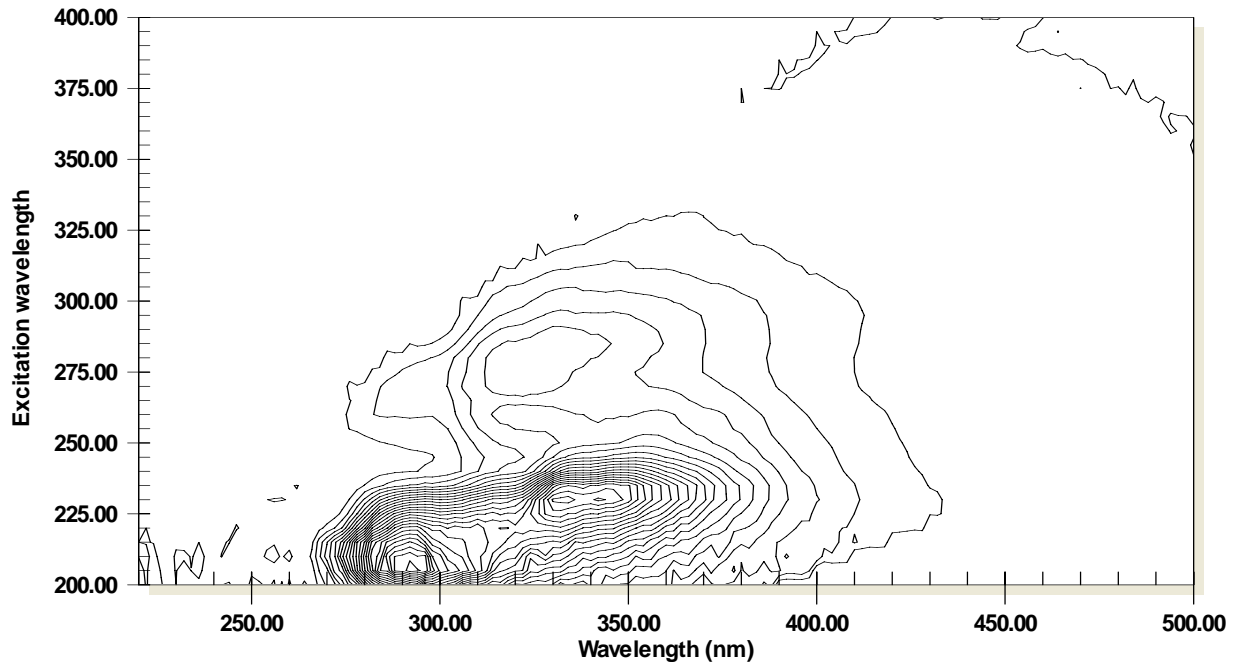
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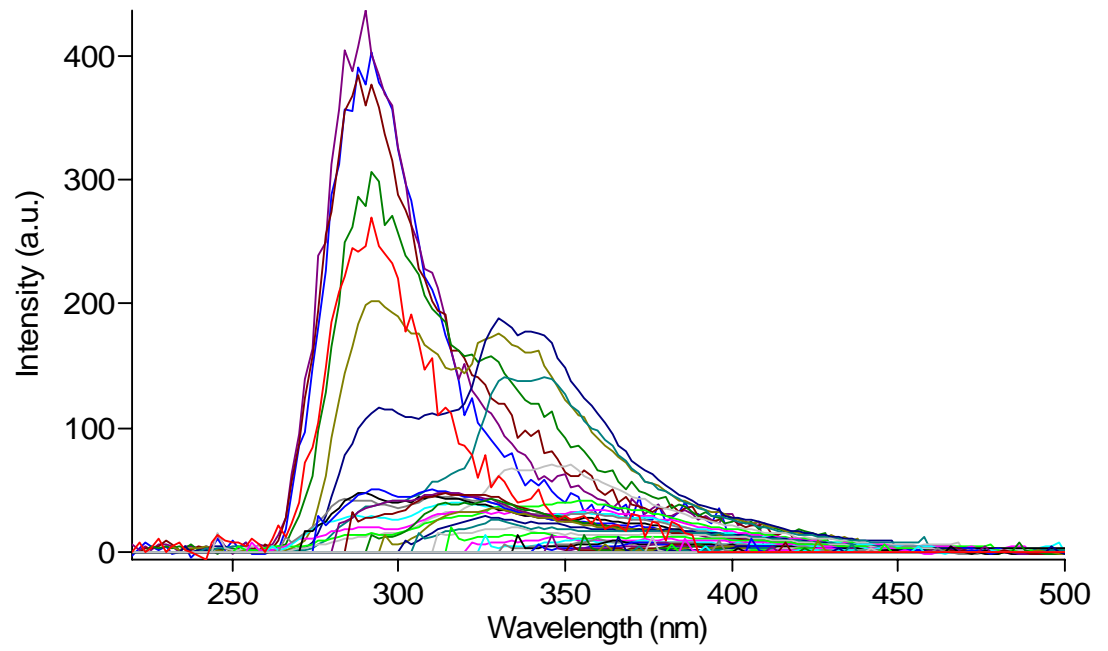
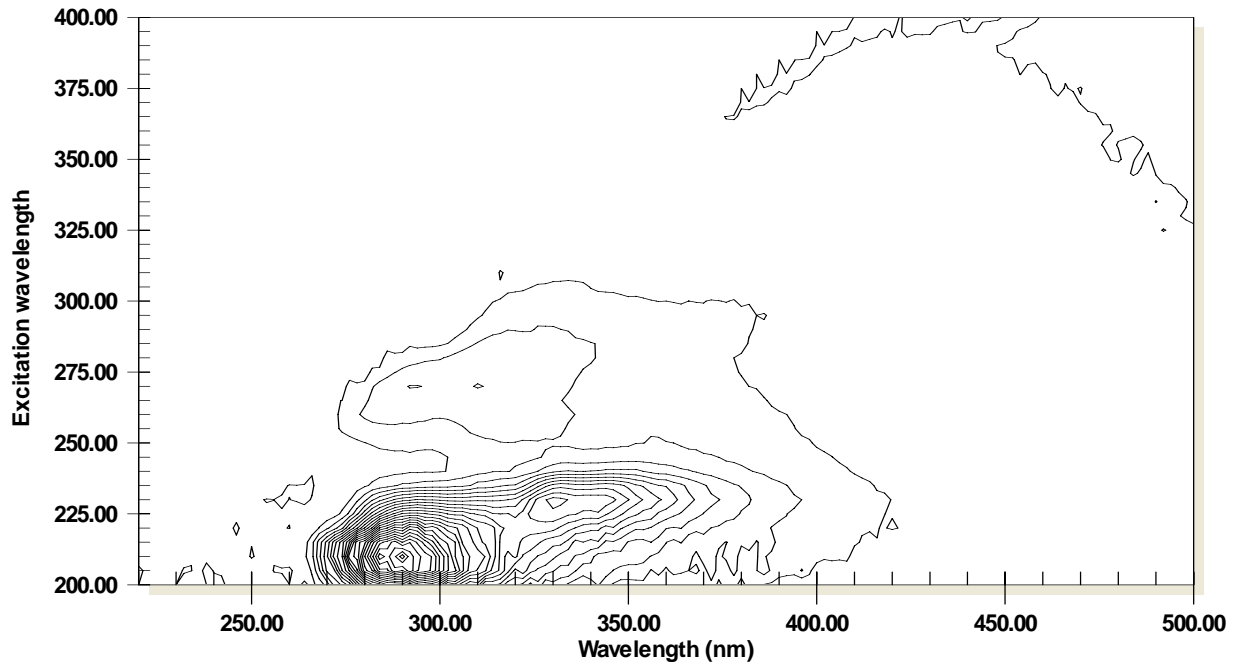
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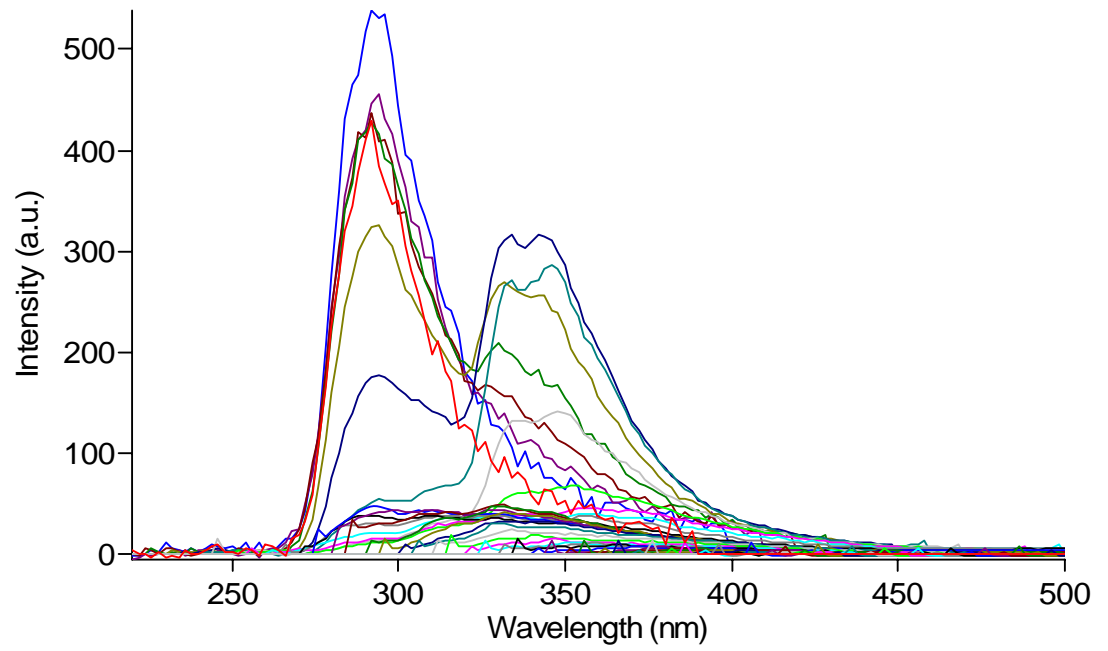
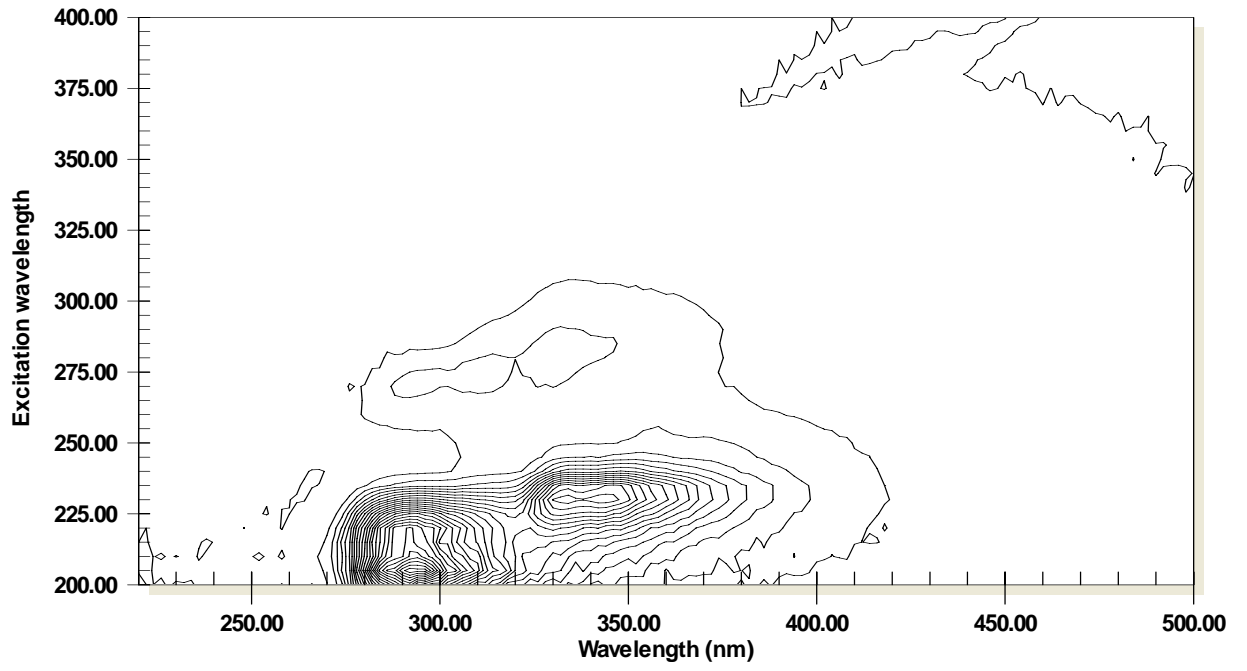
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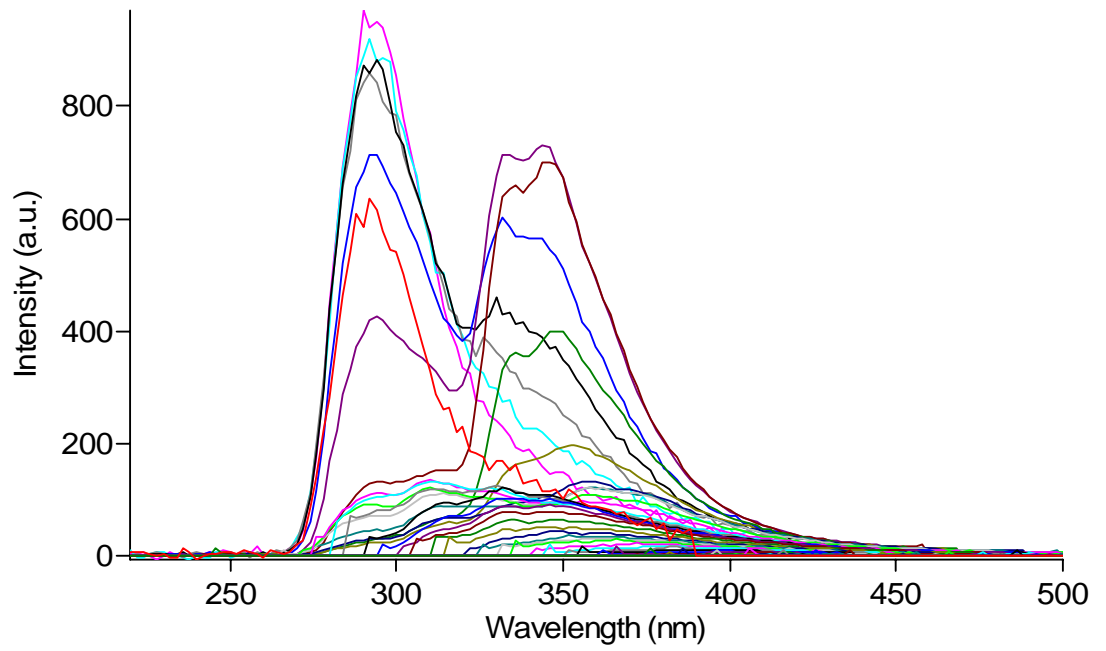
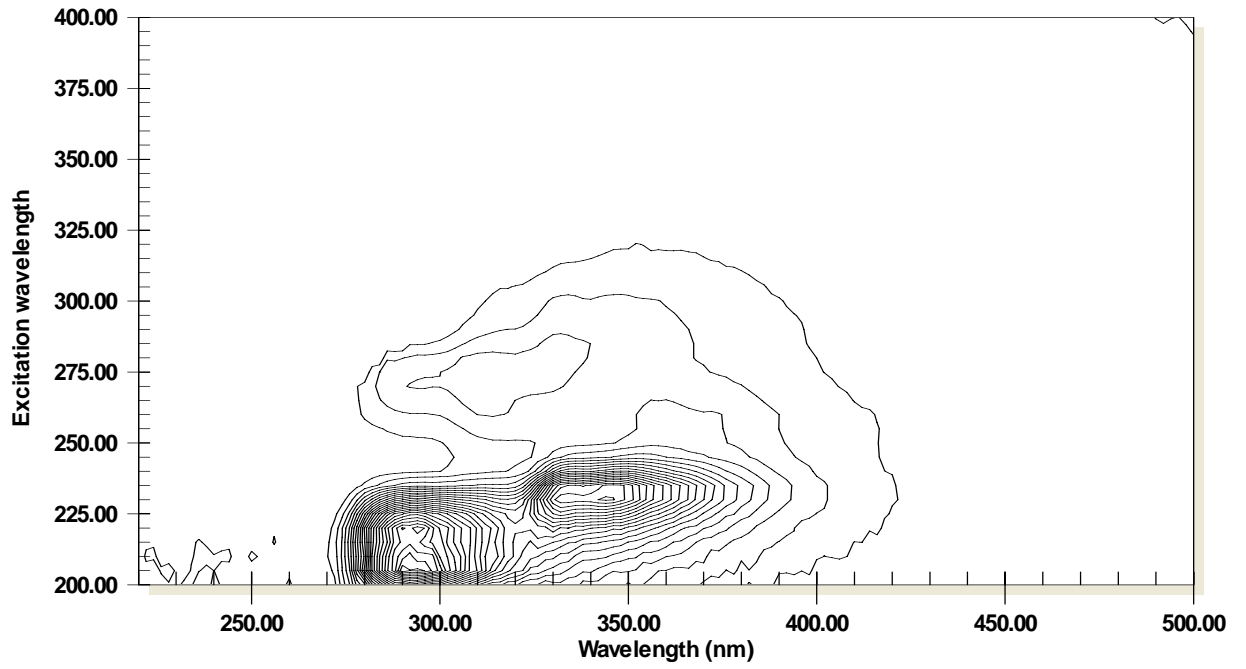
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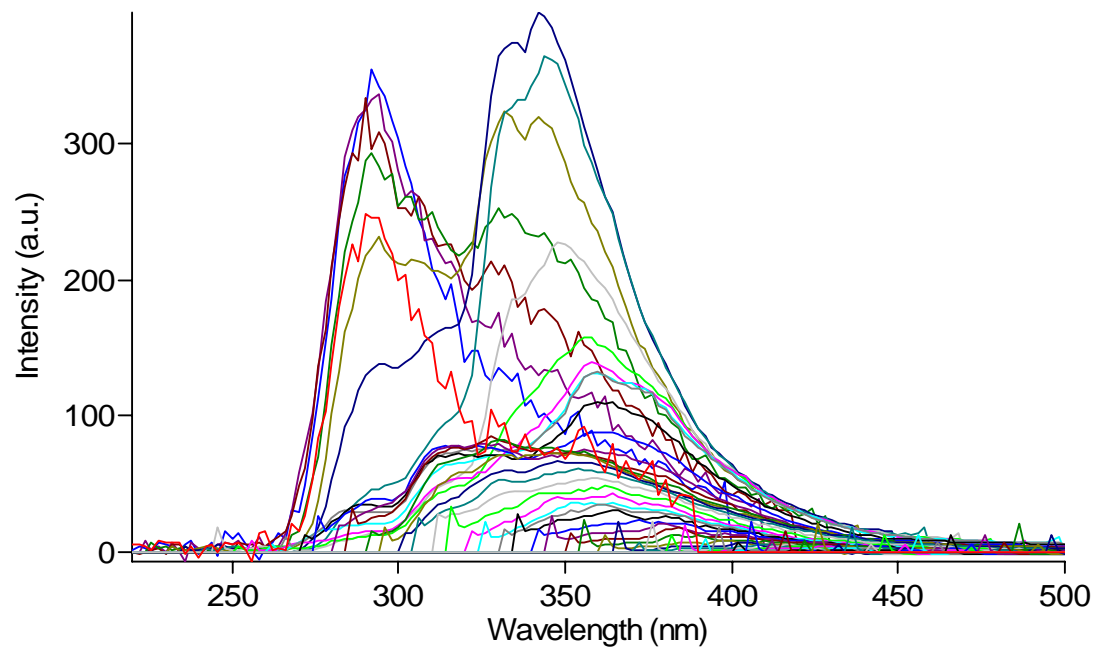
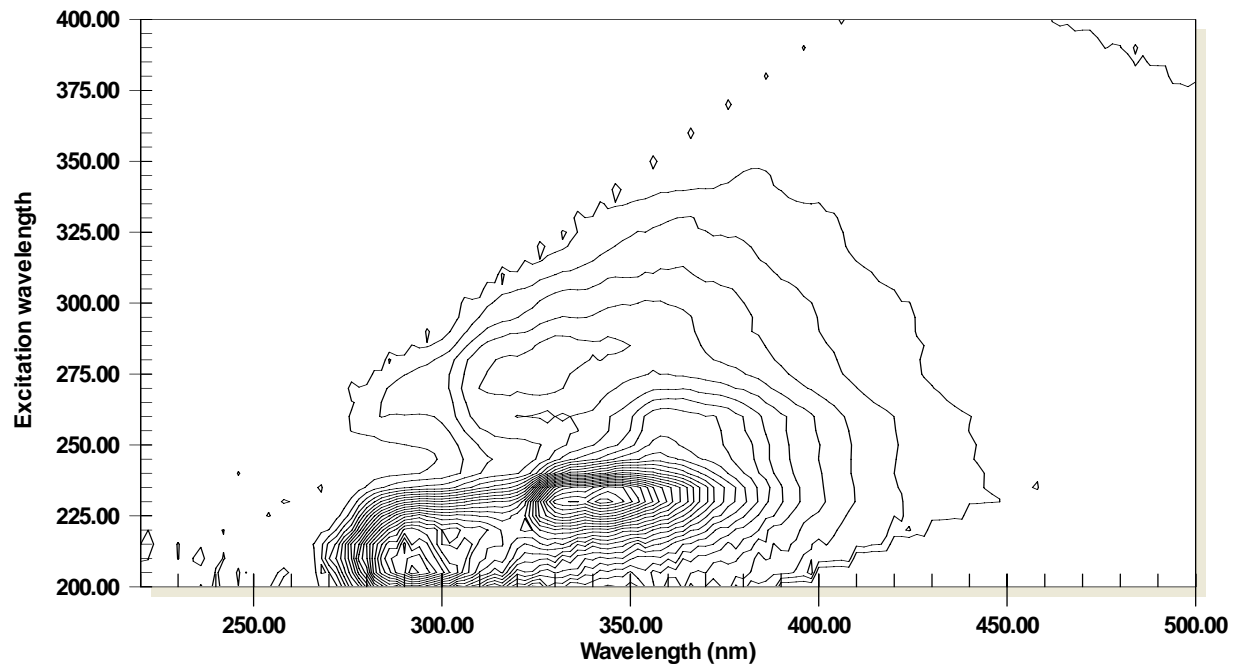
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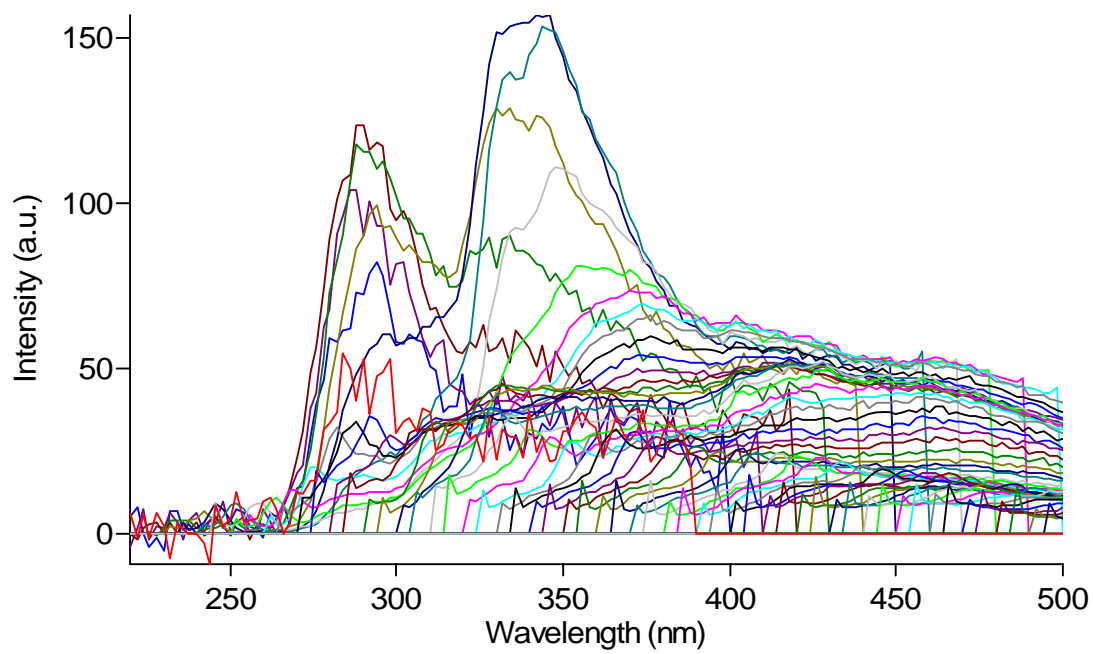
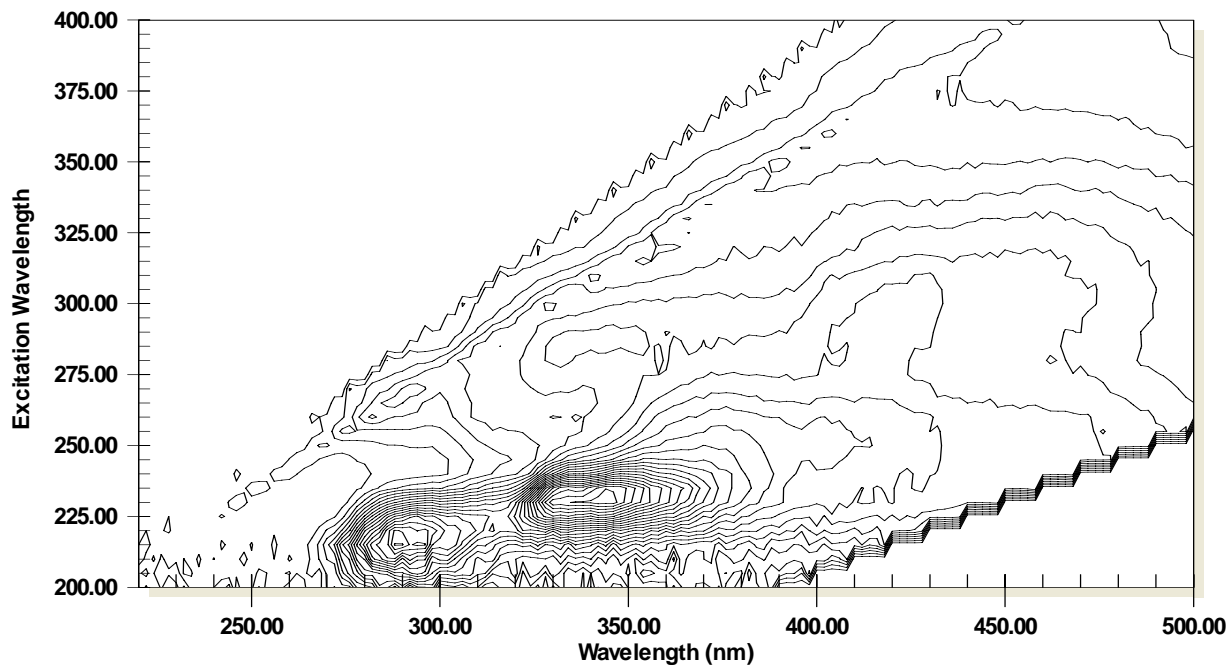
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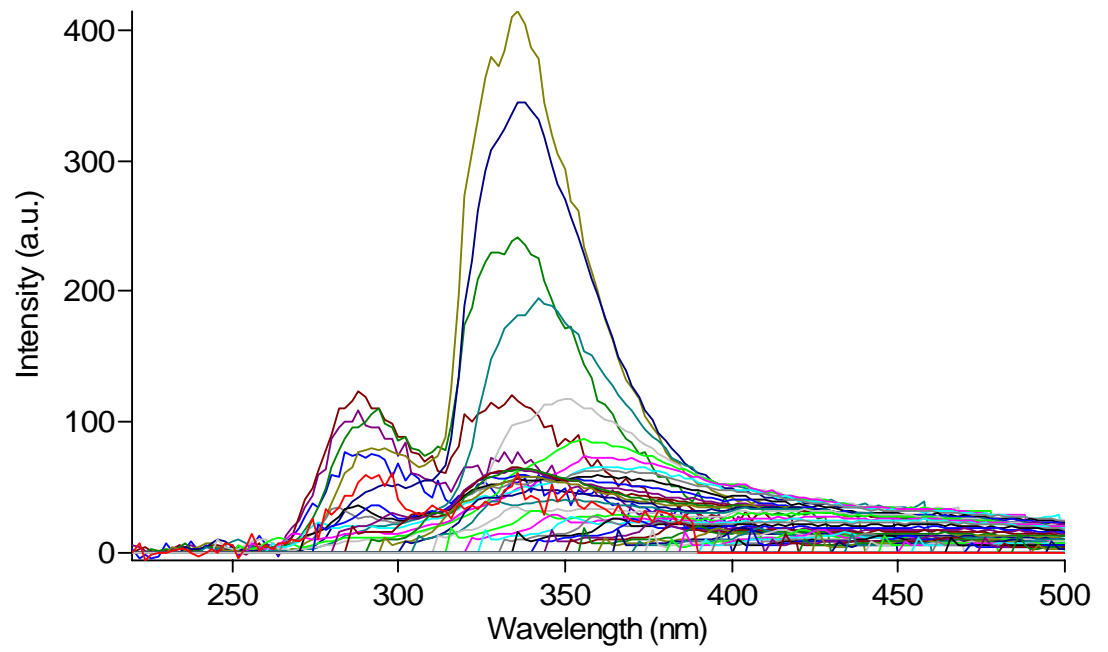
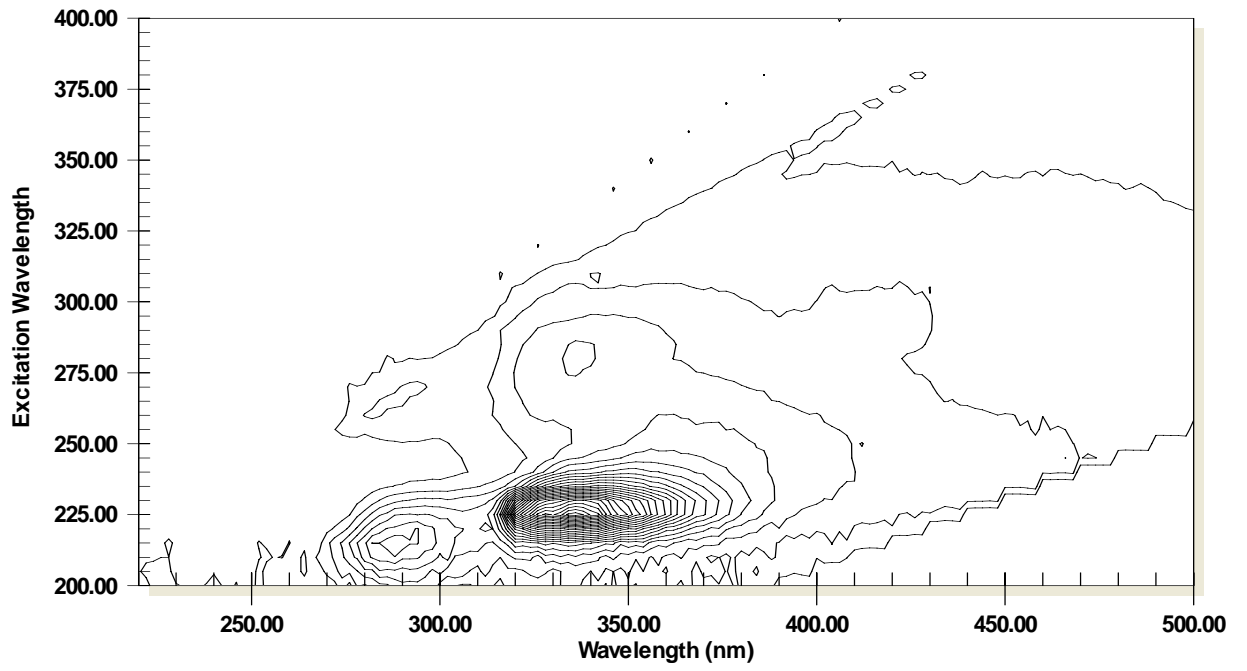
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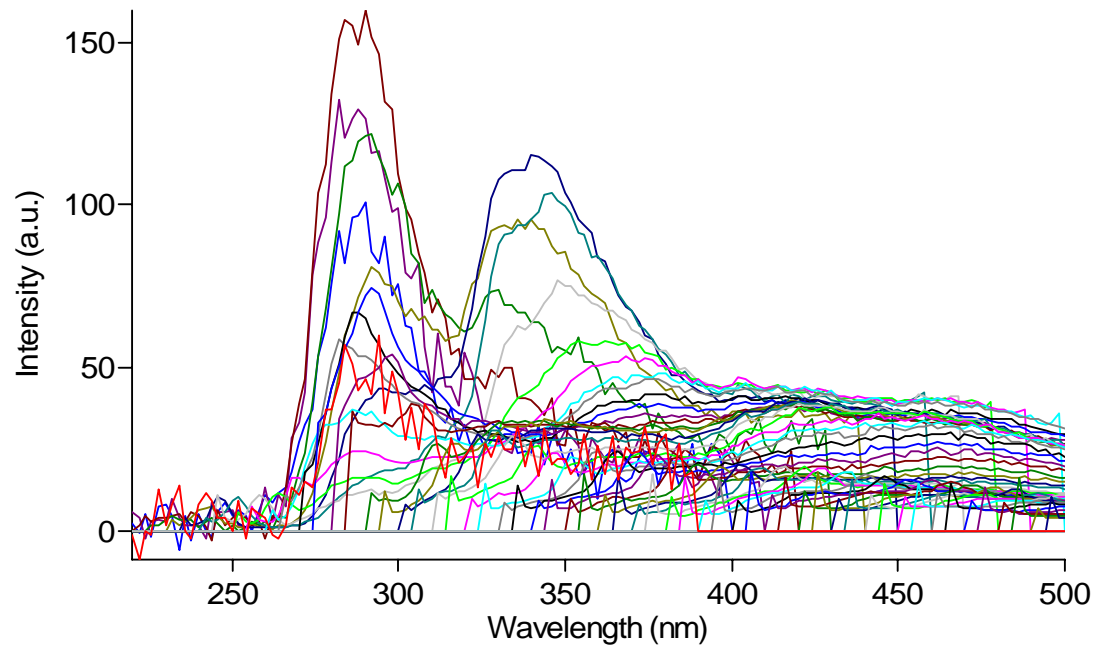
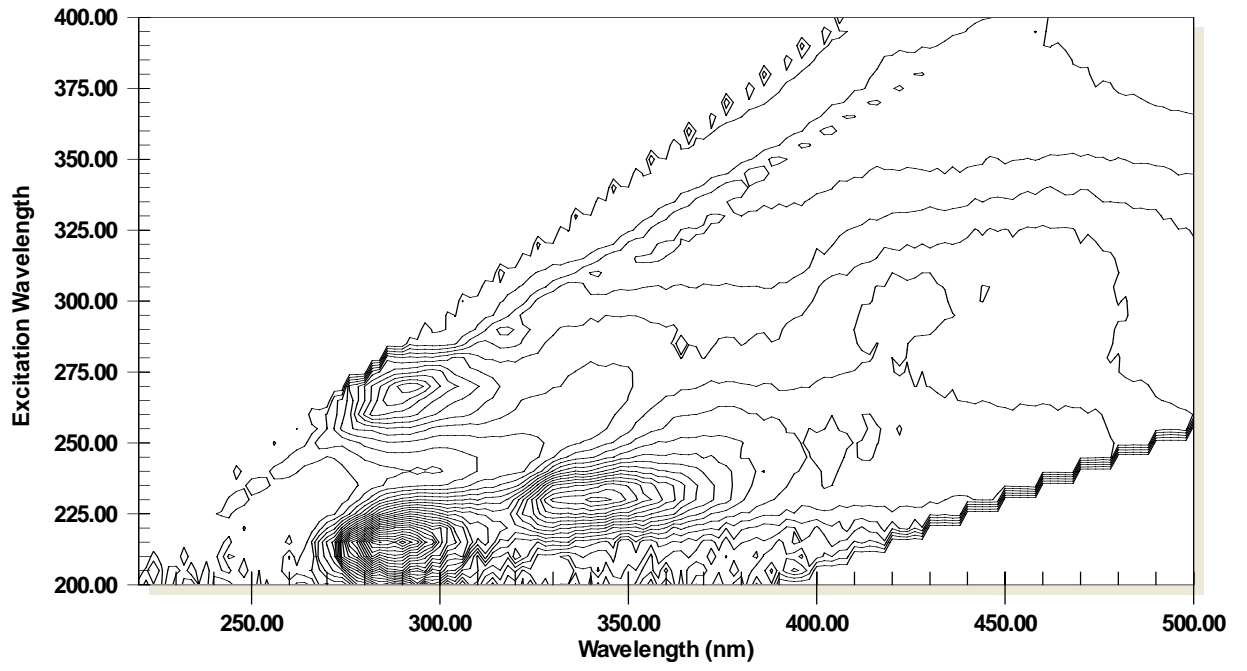
Toluene Contour Plots and Excitation Scan Overlays
11CT



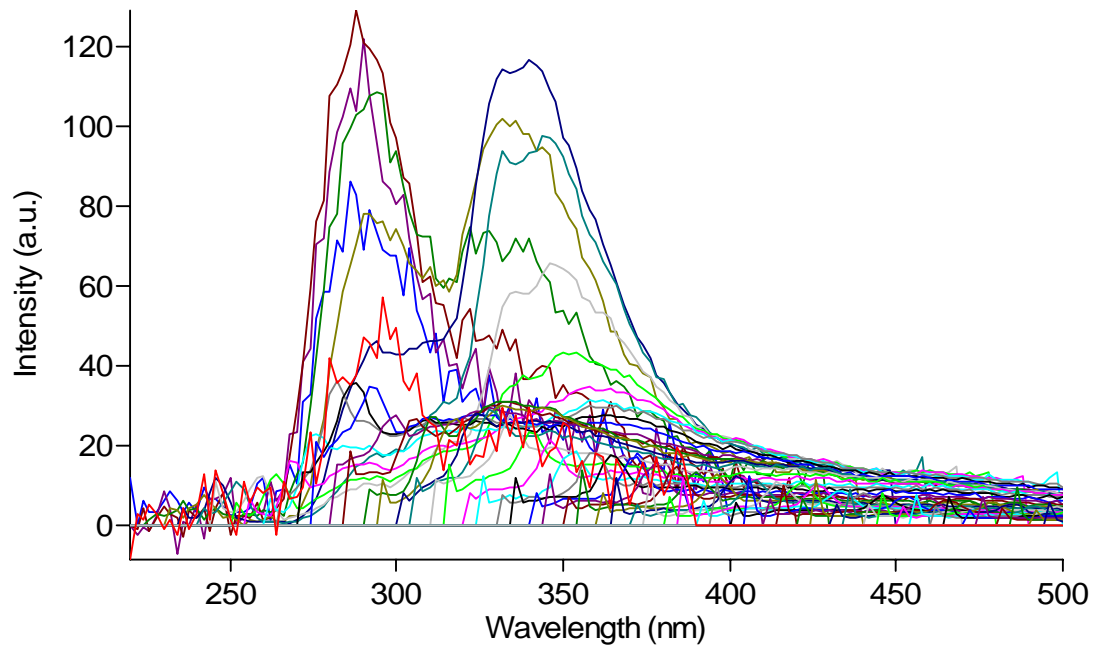
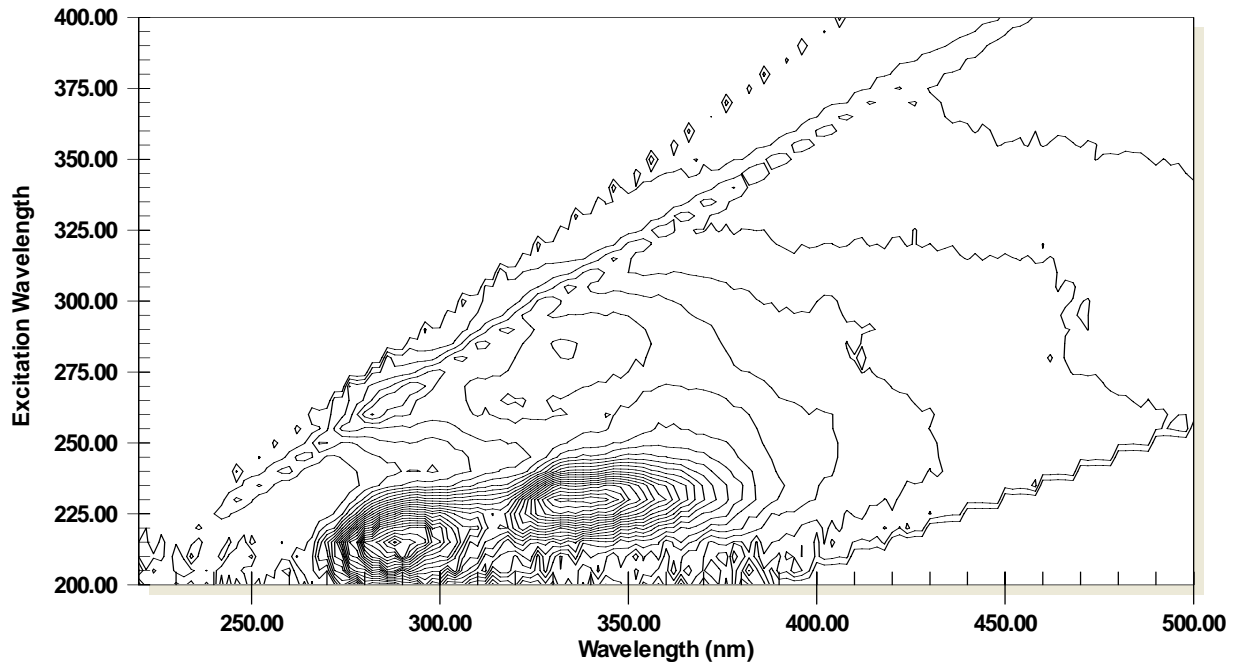
13BT



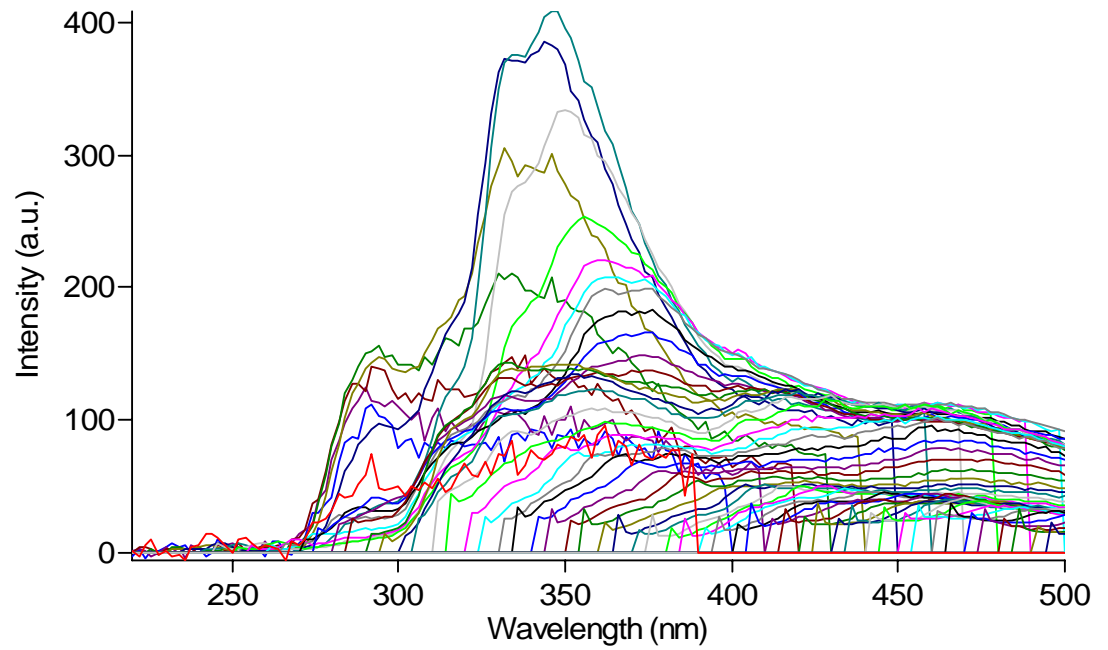
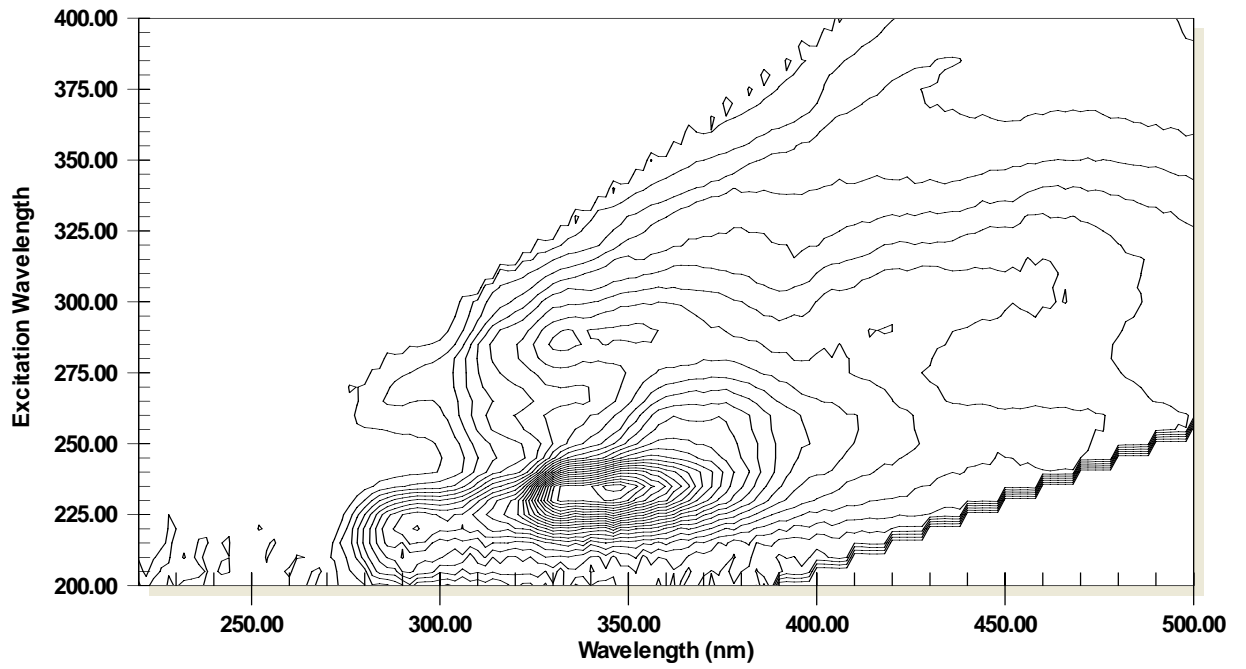
14BT



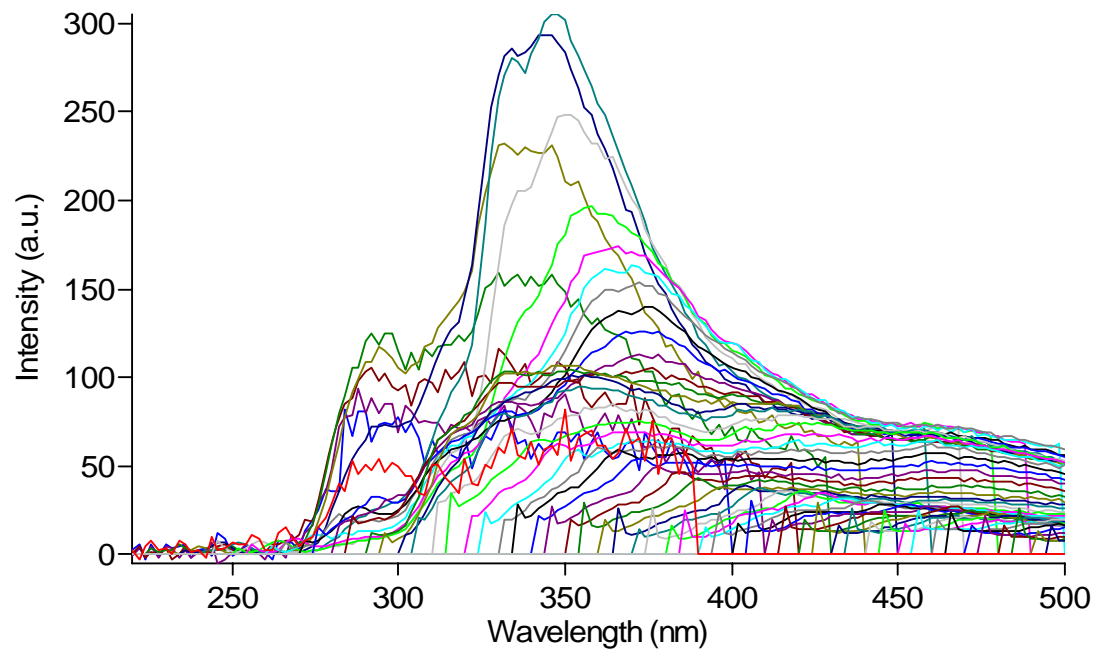
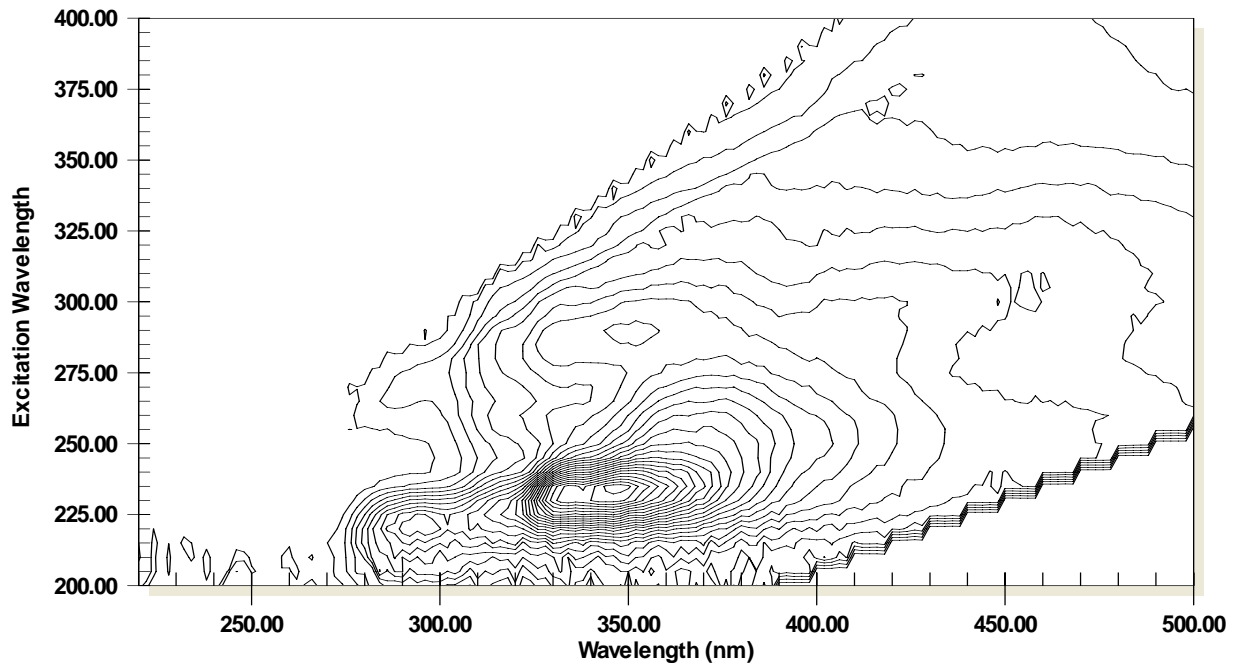
15BT



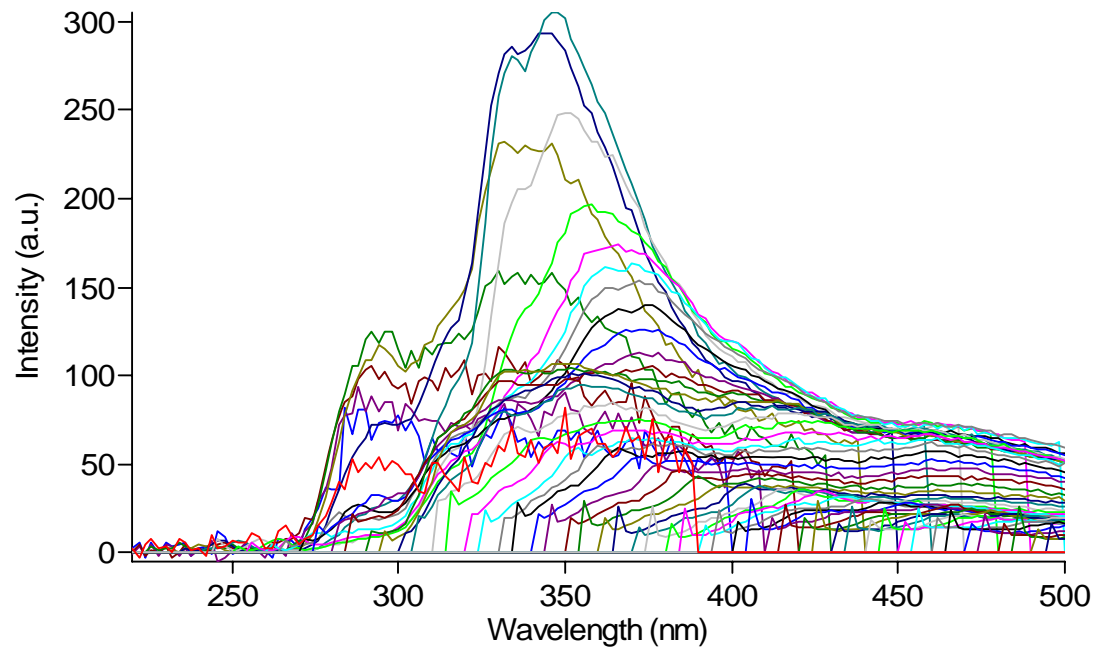
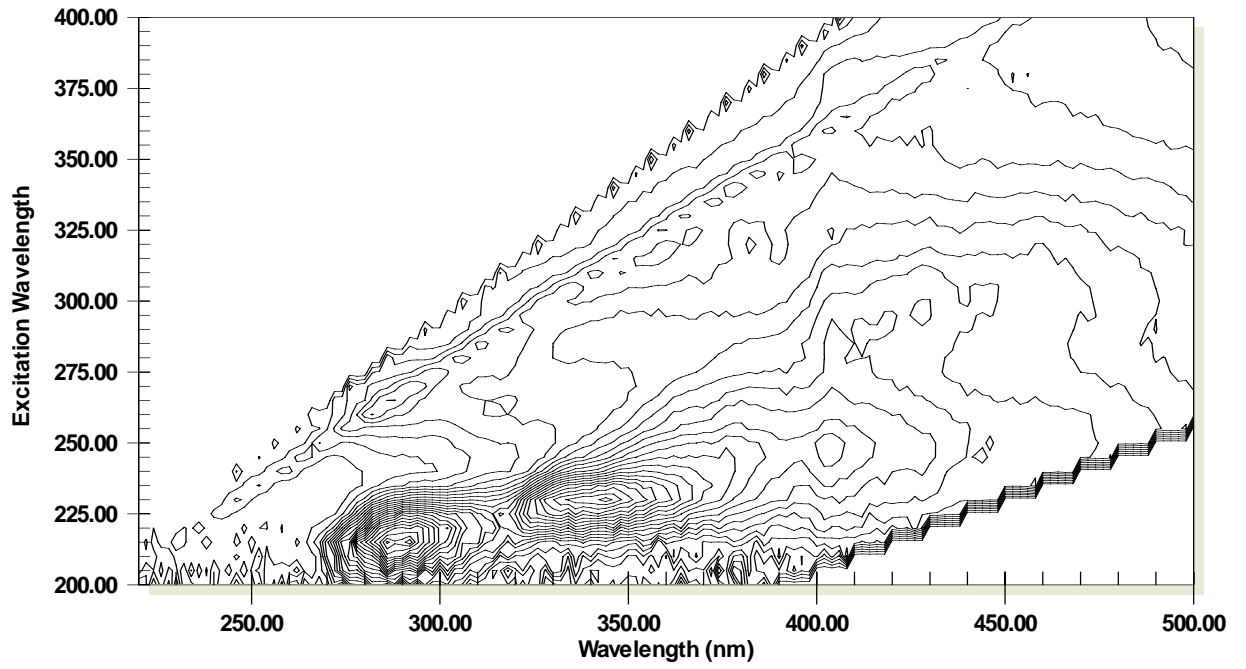
21BT



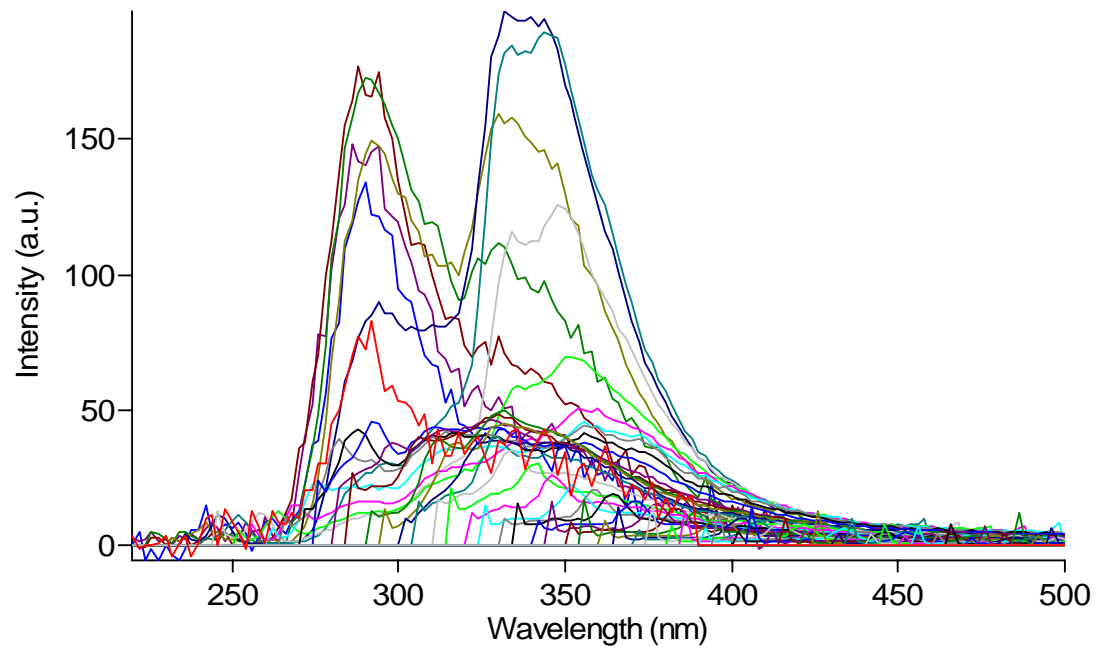
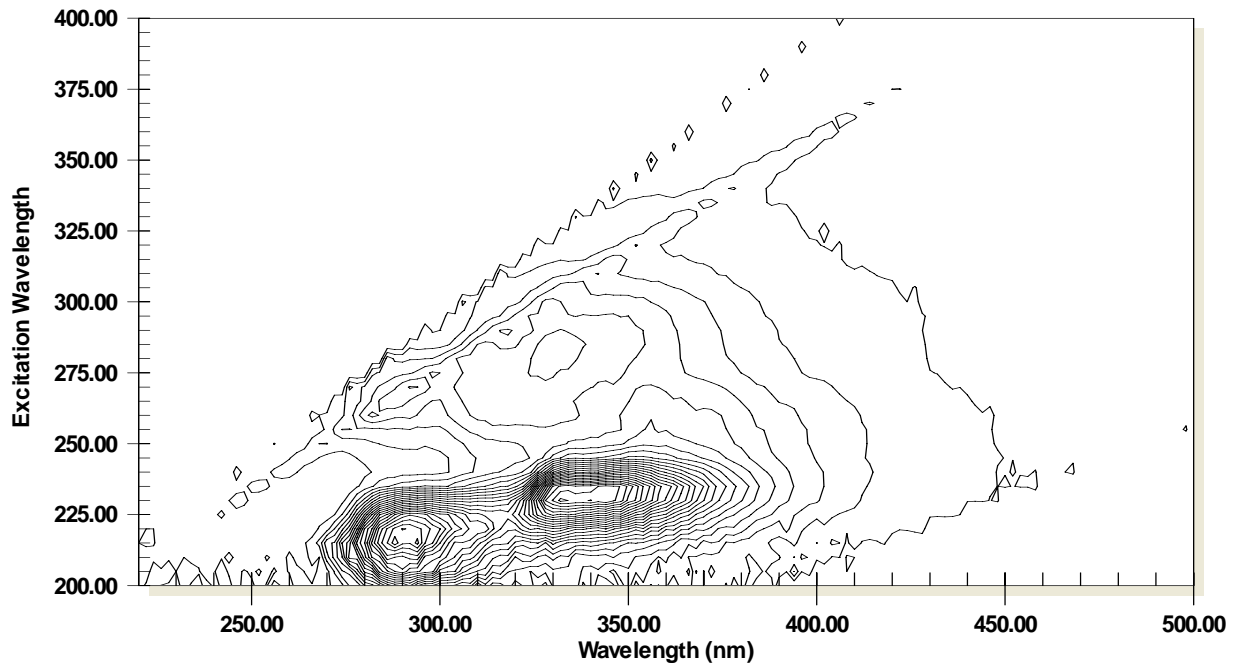
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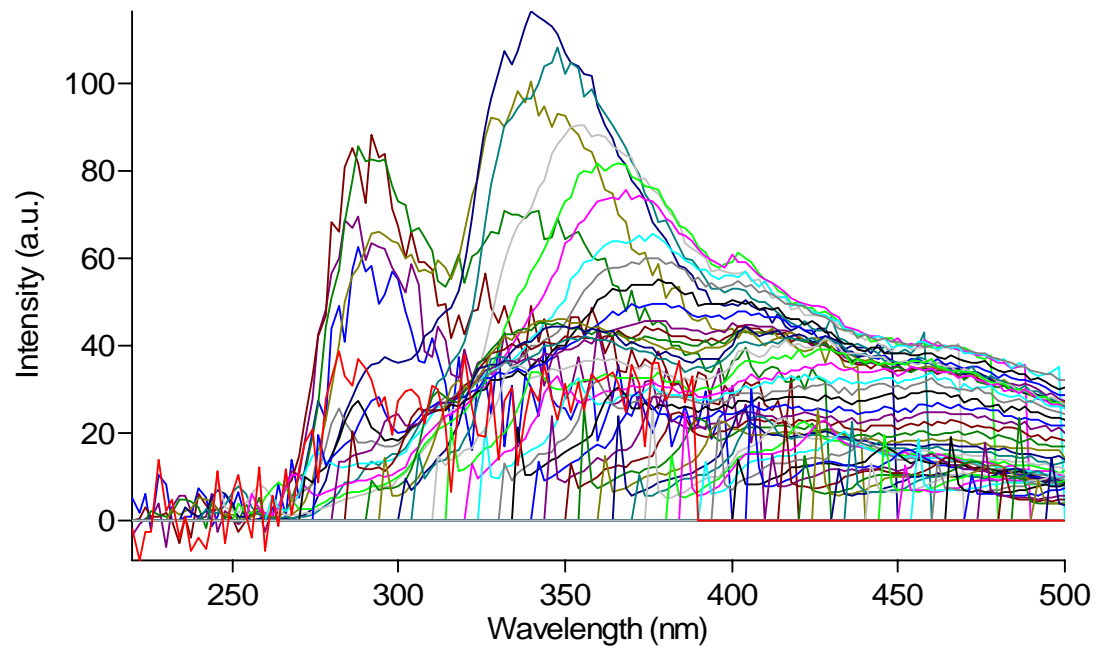
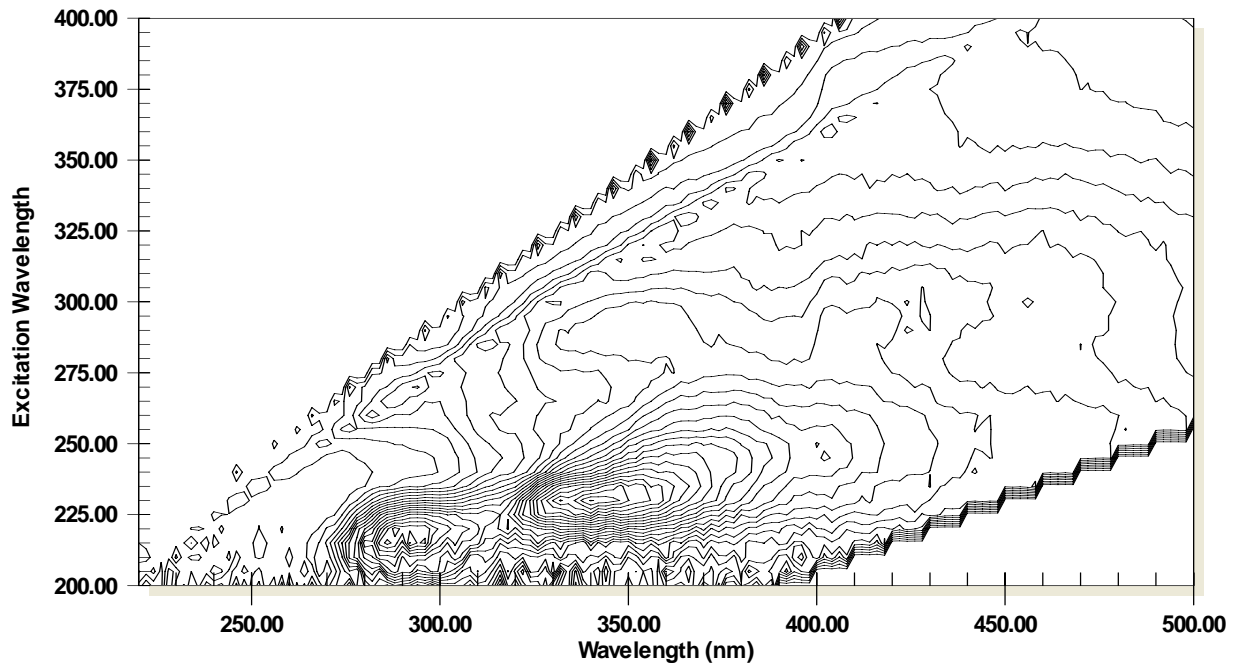
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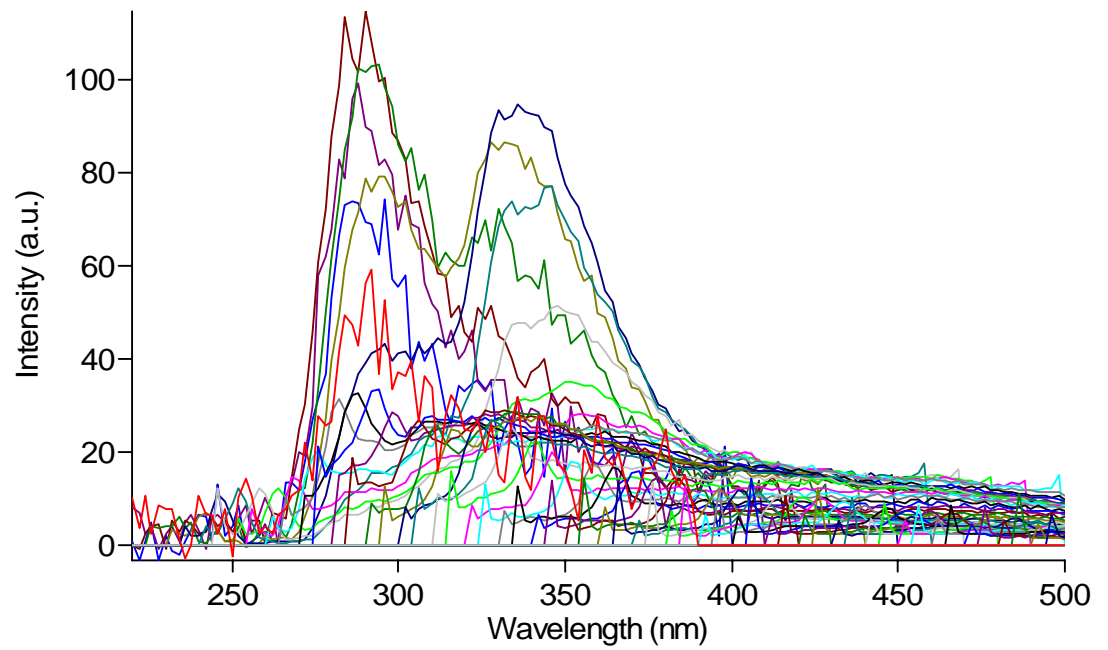
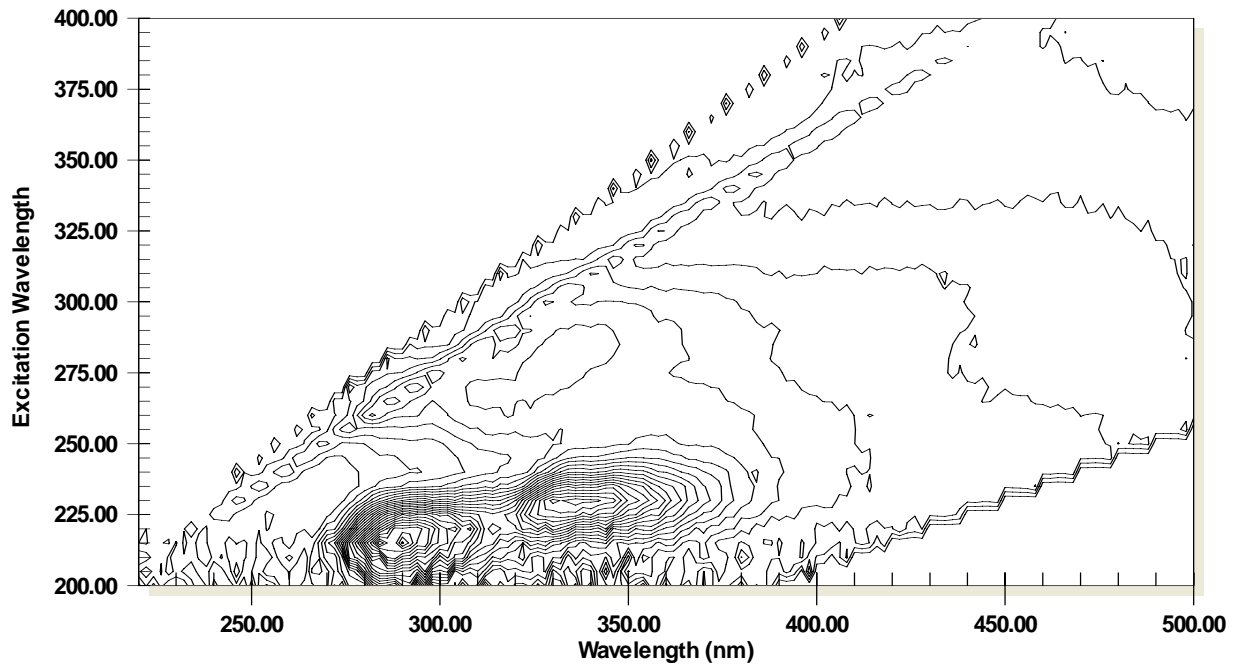
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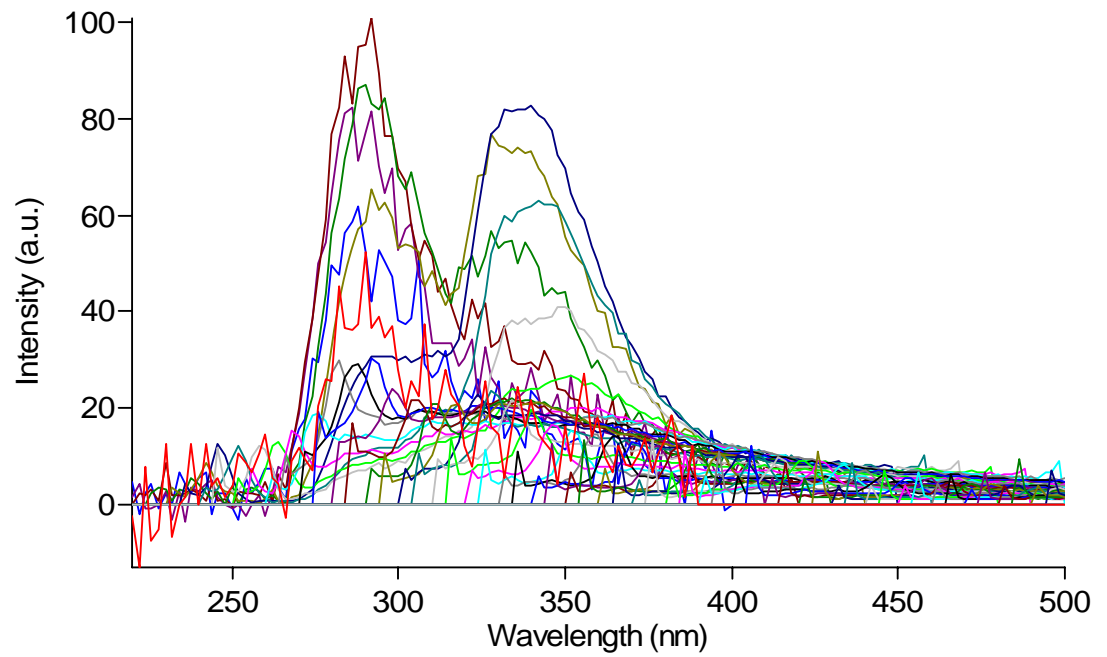
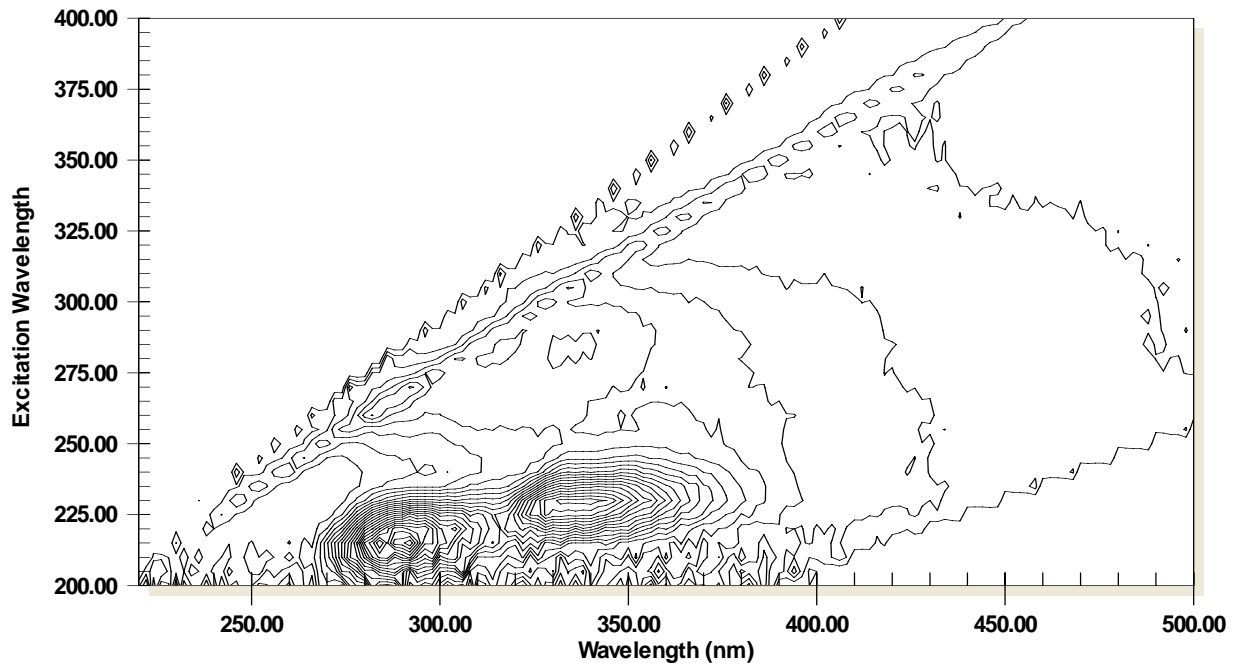
25CT



31CT

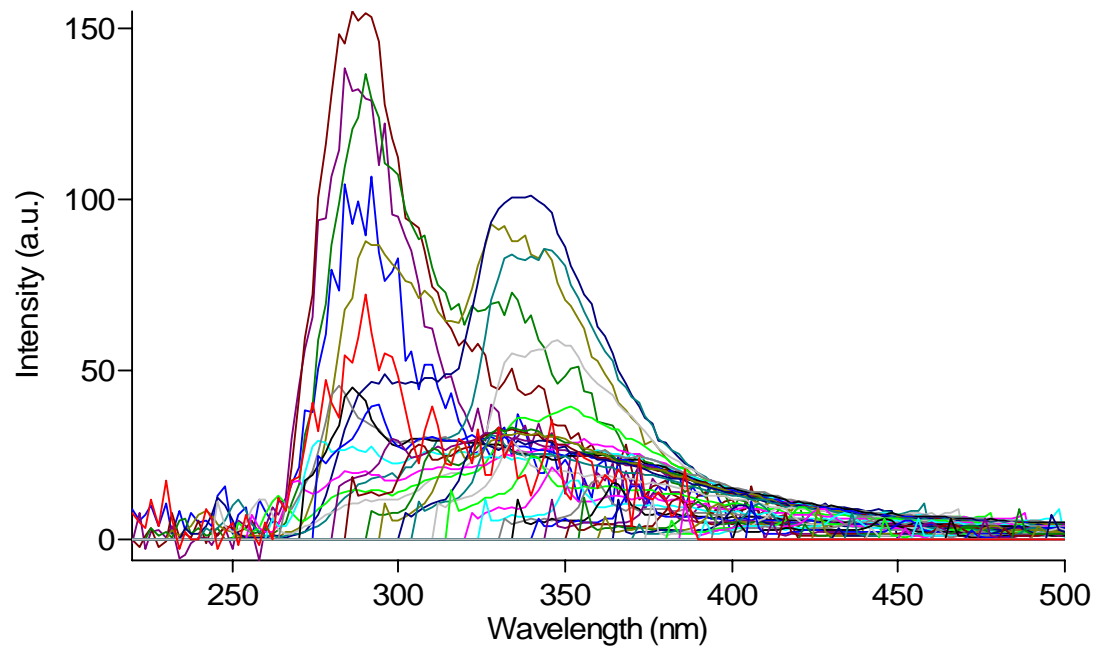
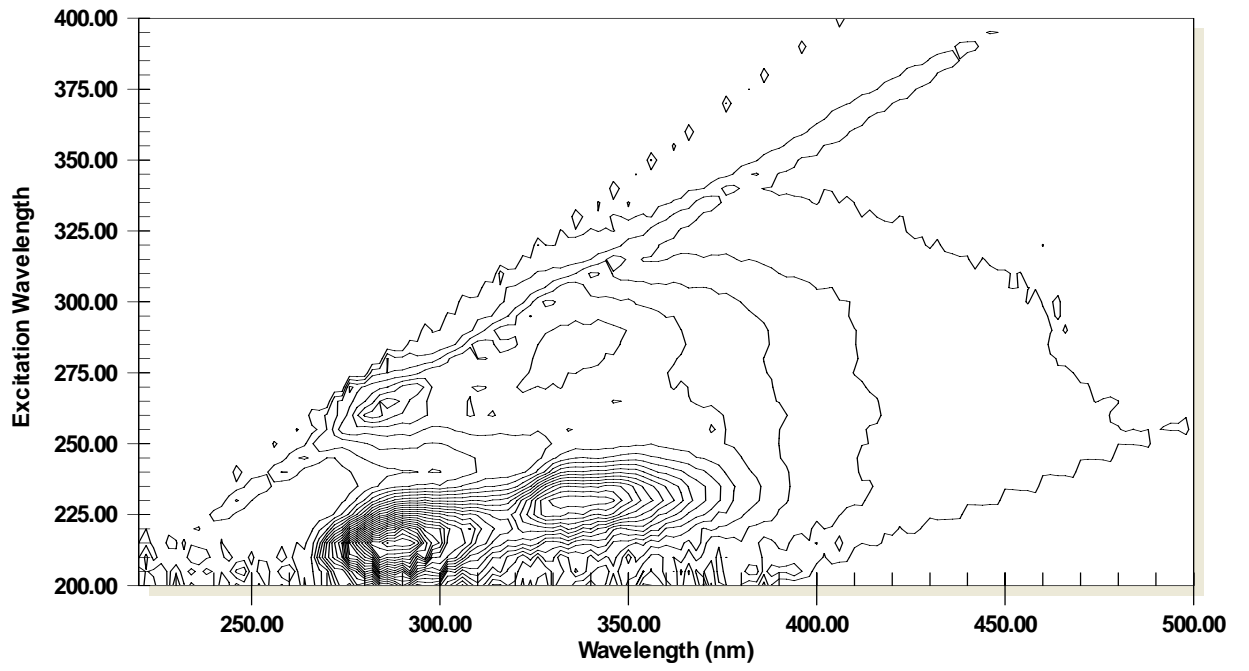


32CT

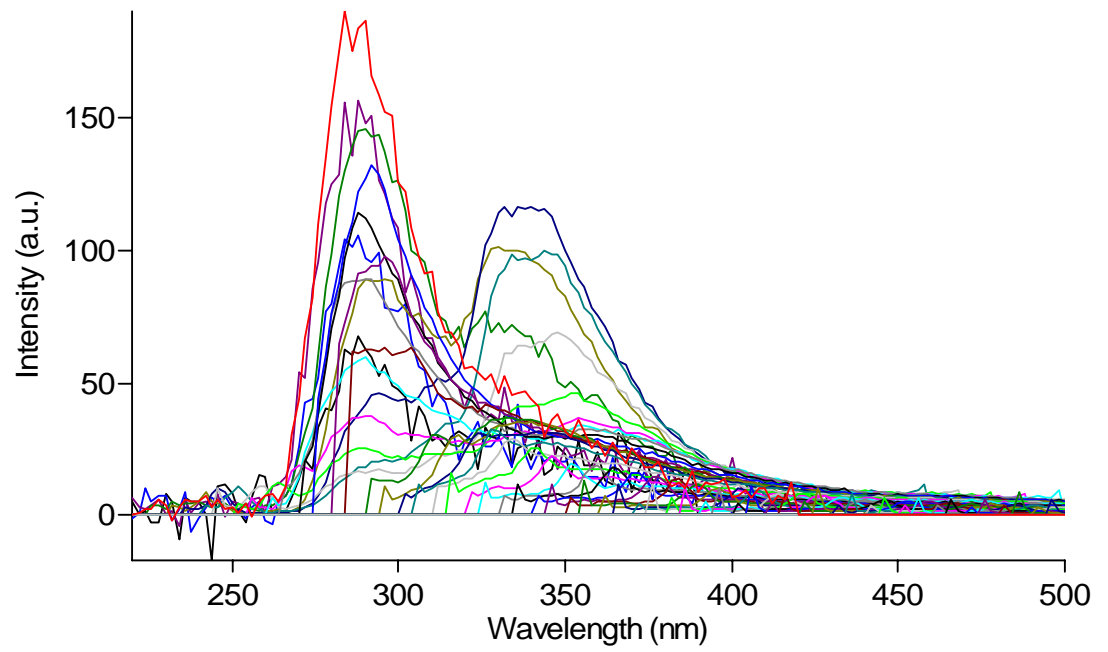
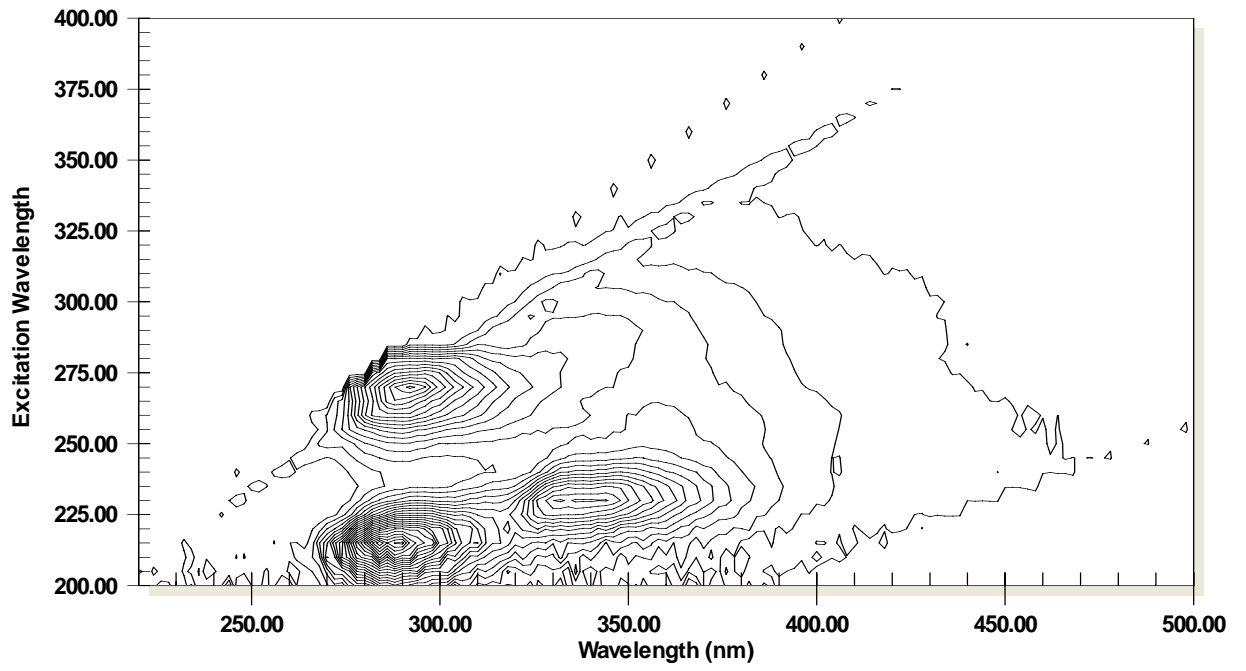


337

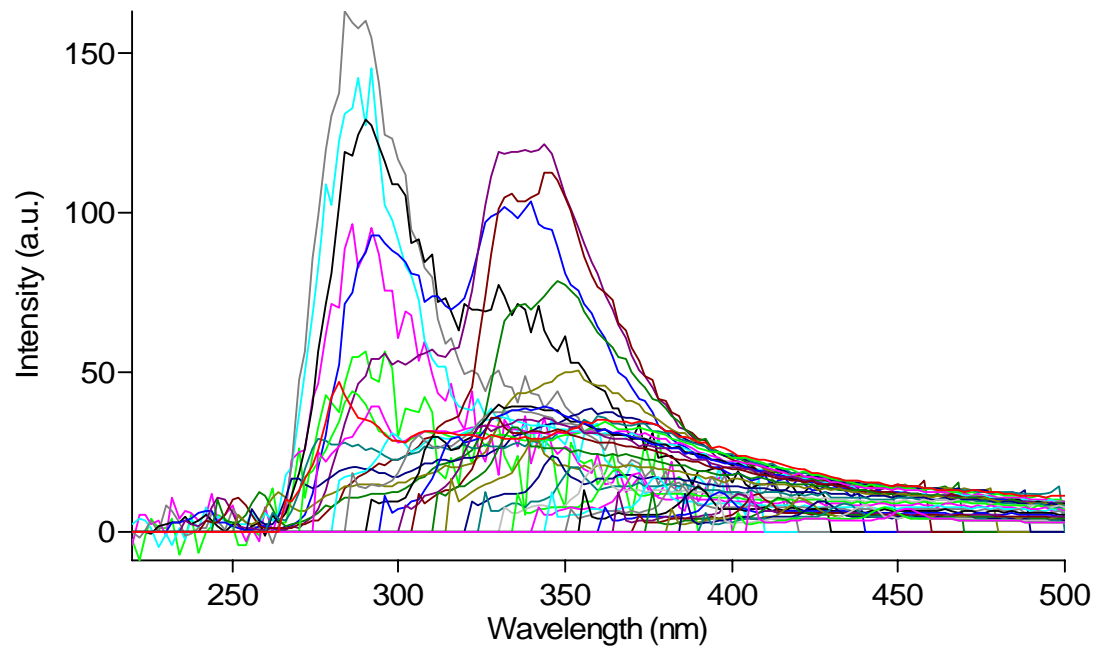
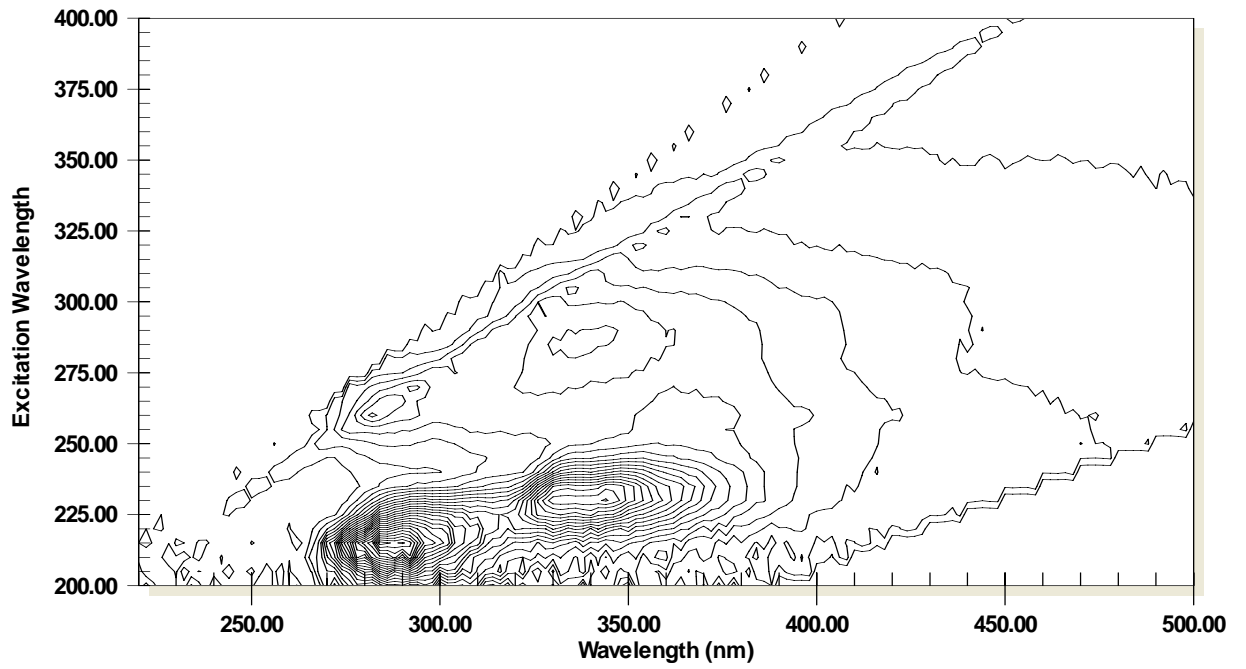
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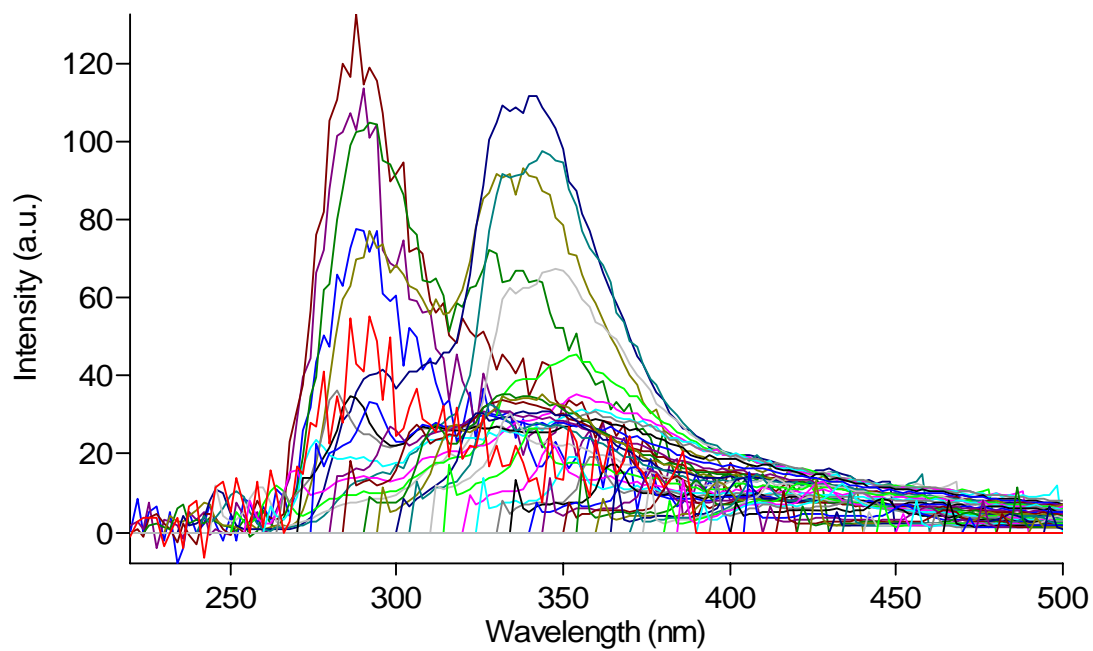
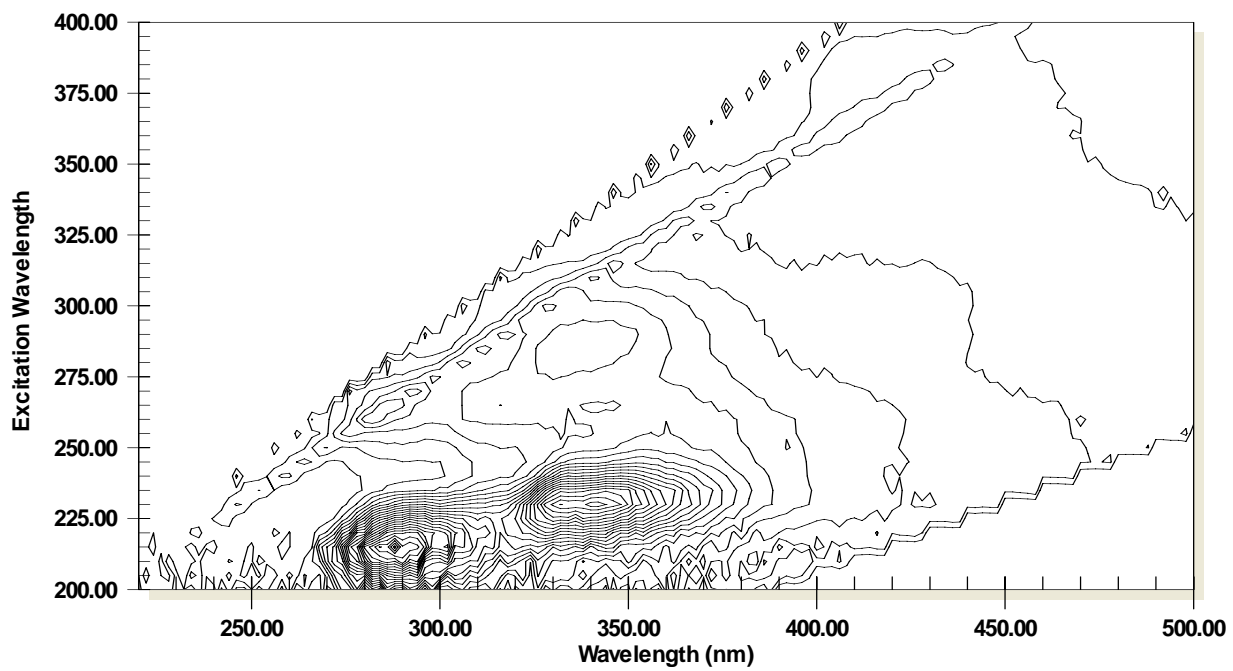
34BT



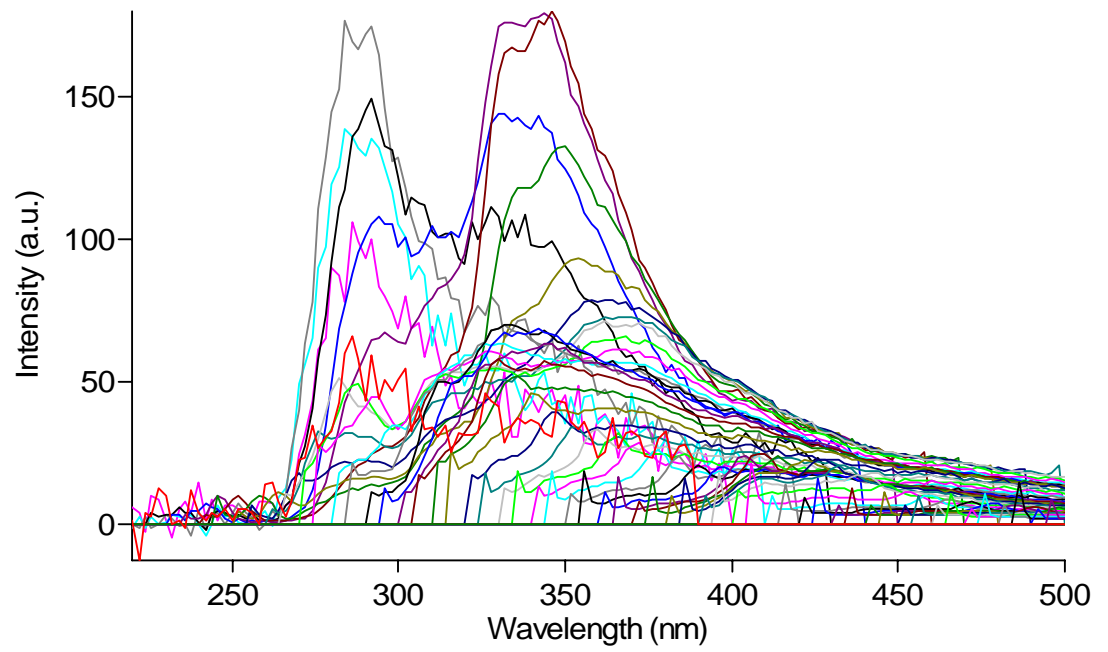
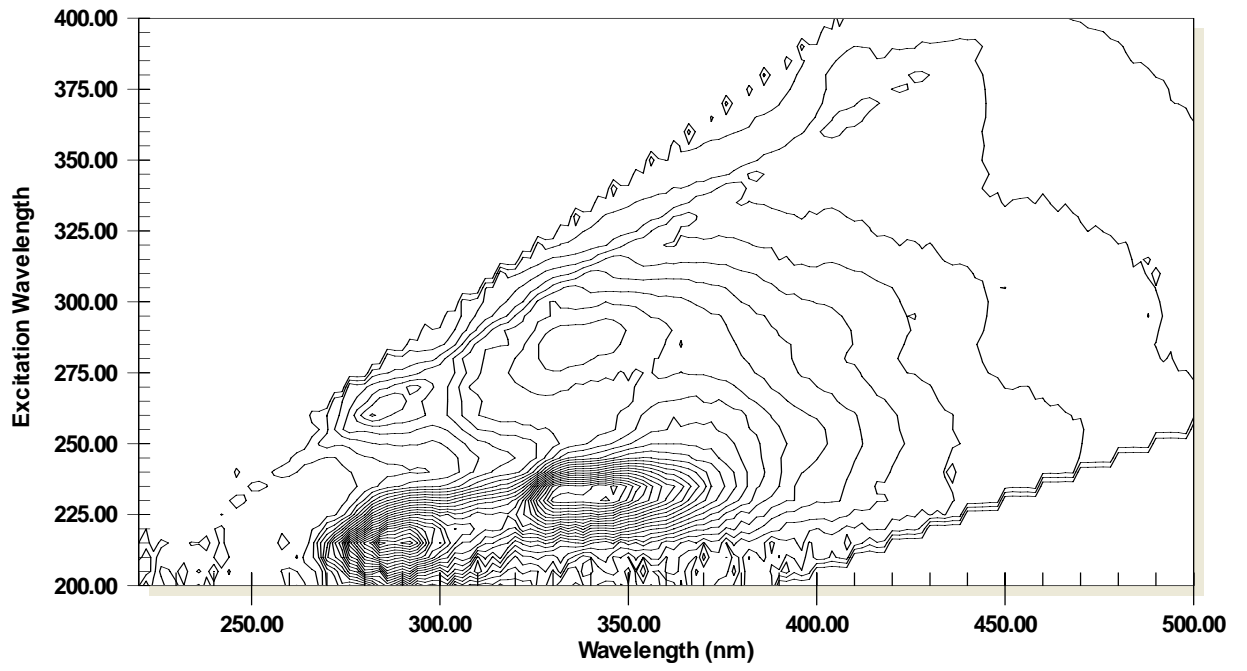
35CT



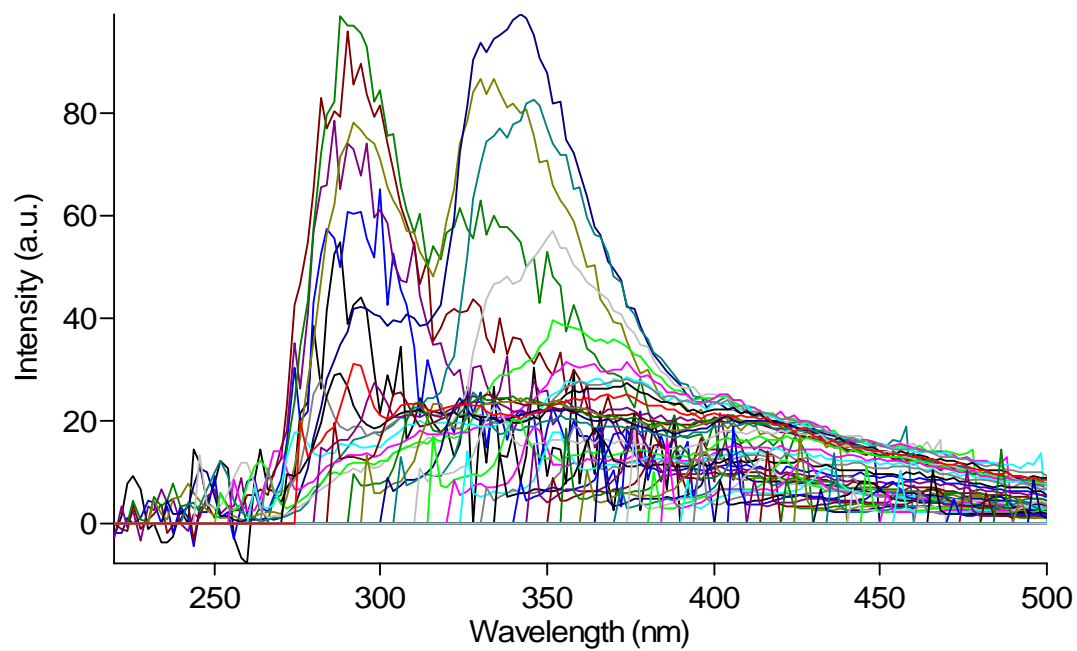
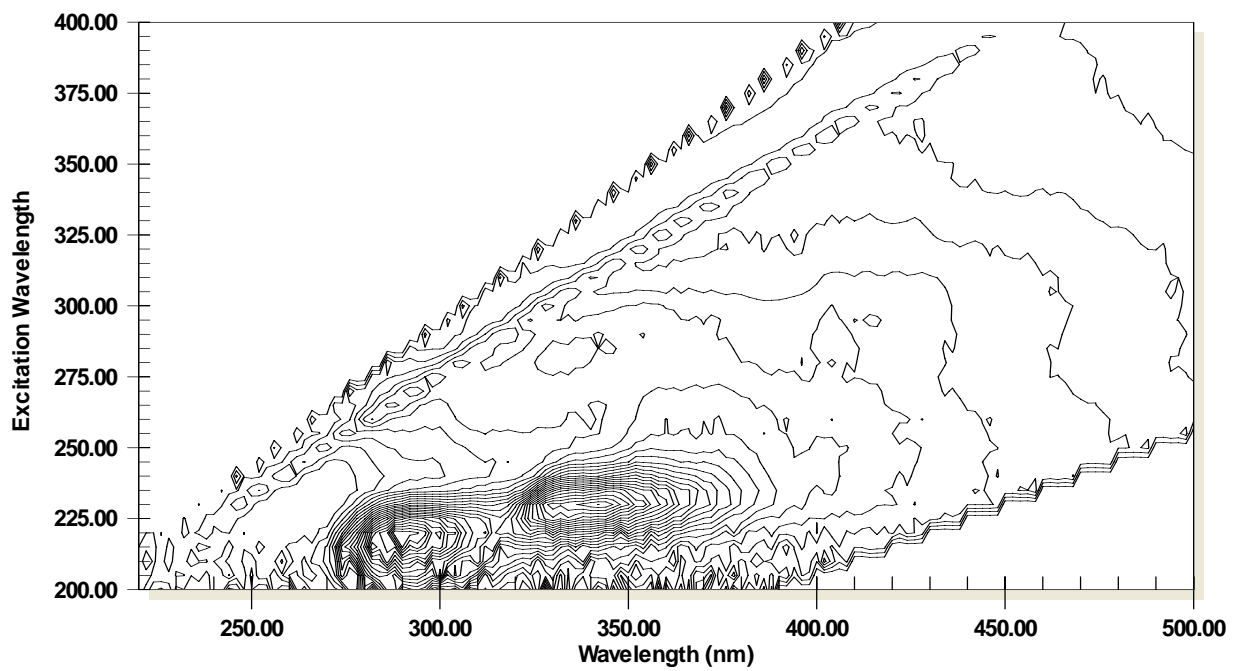
41CT



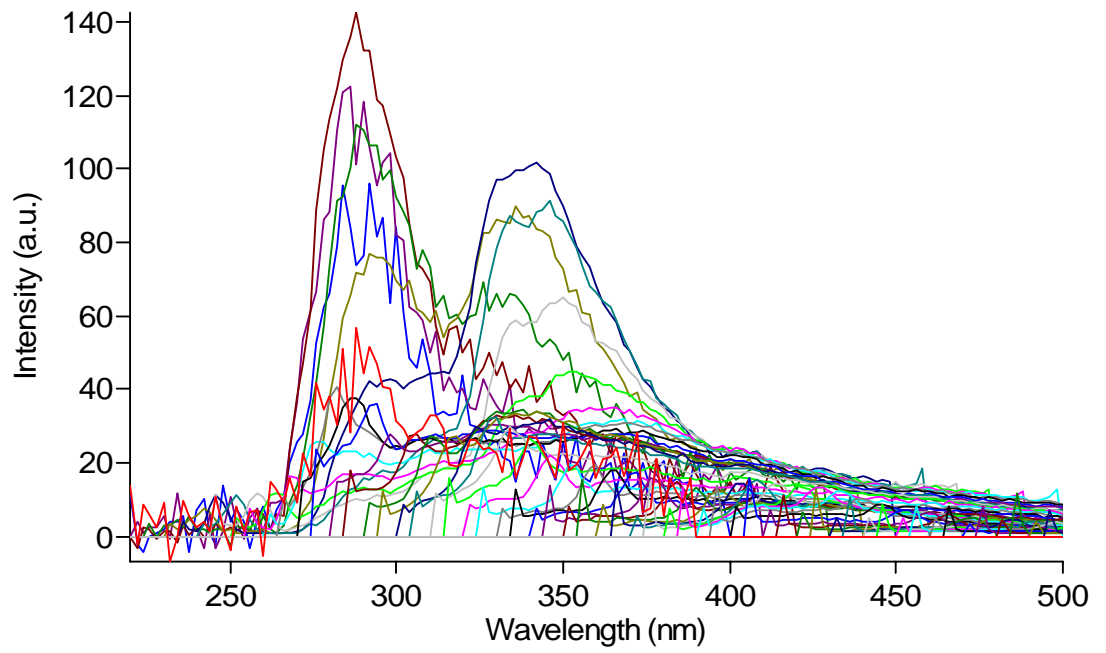
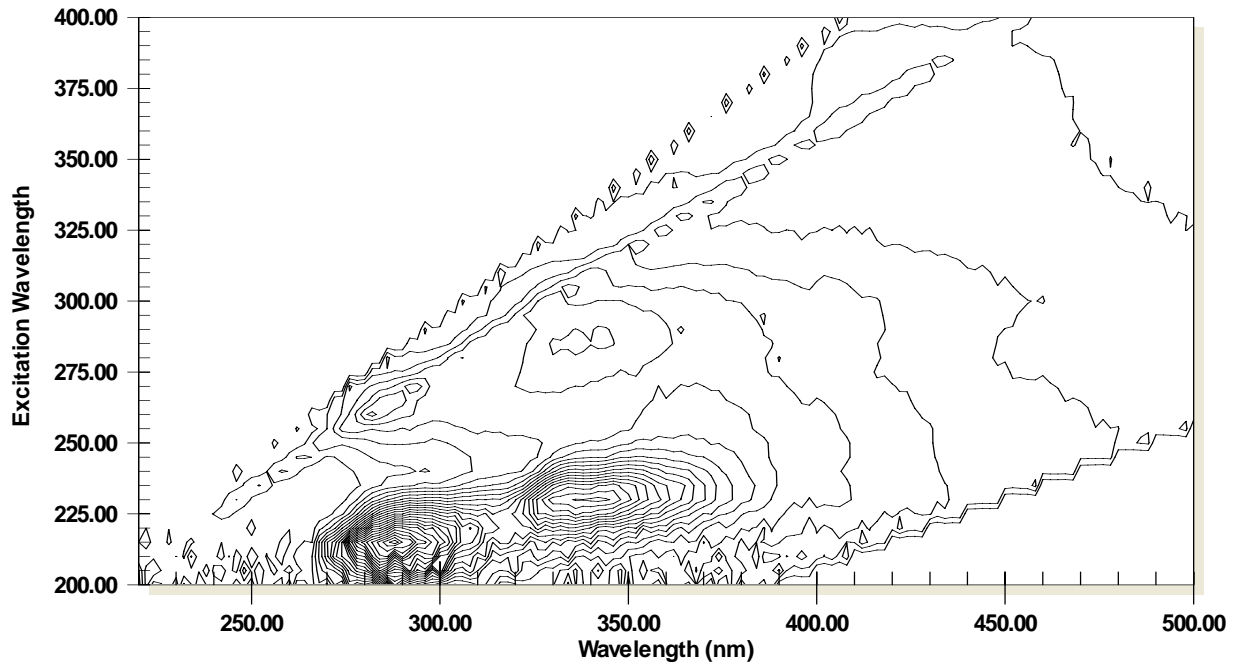
42BT



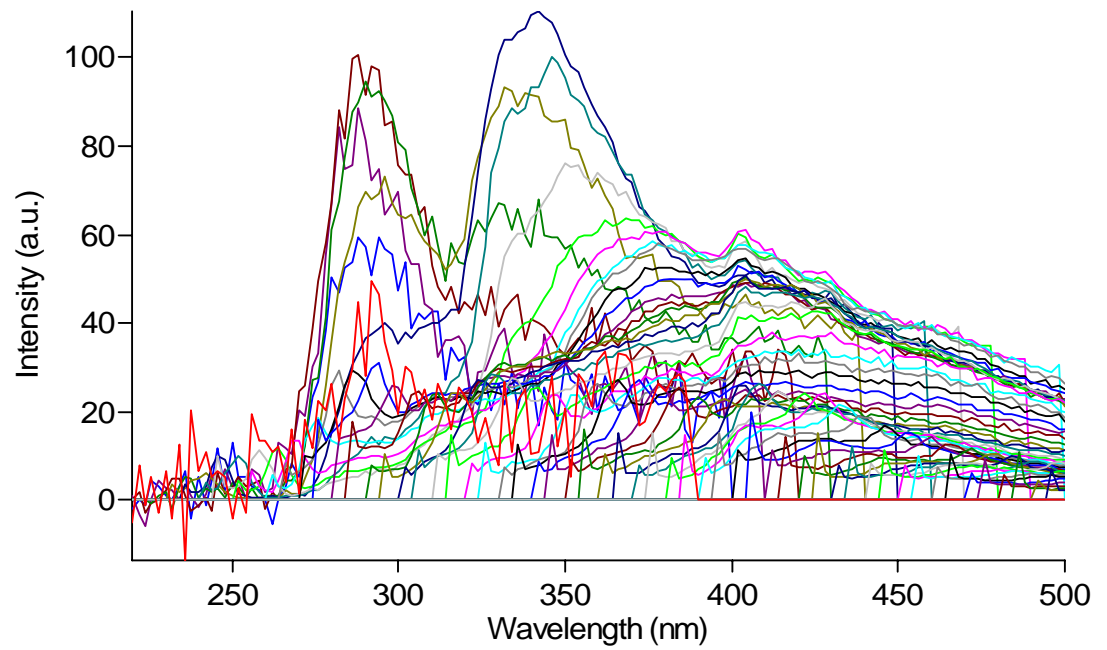
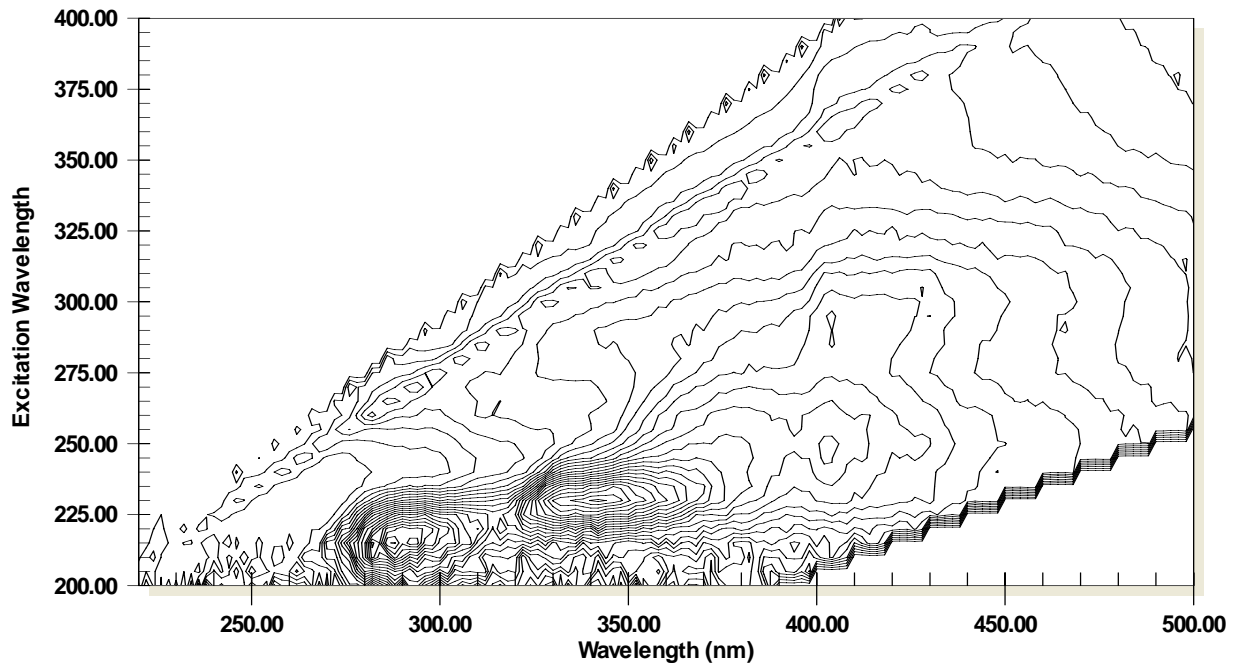
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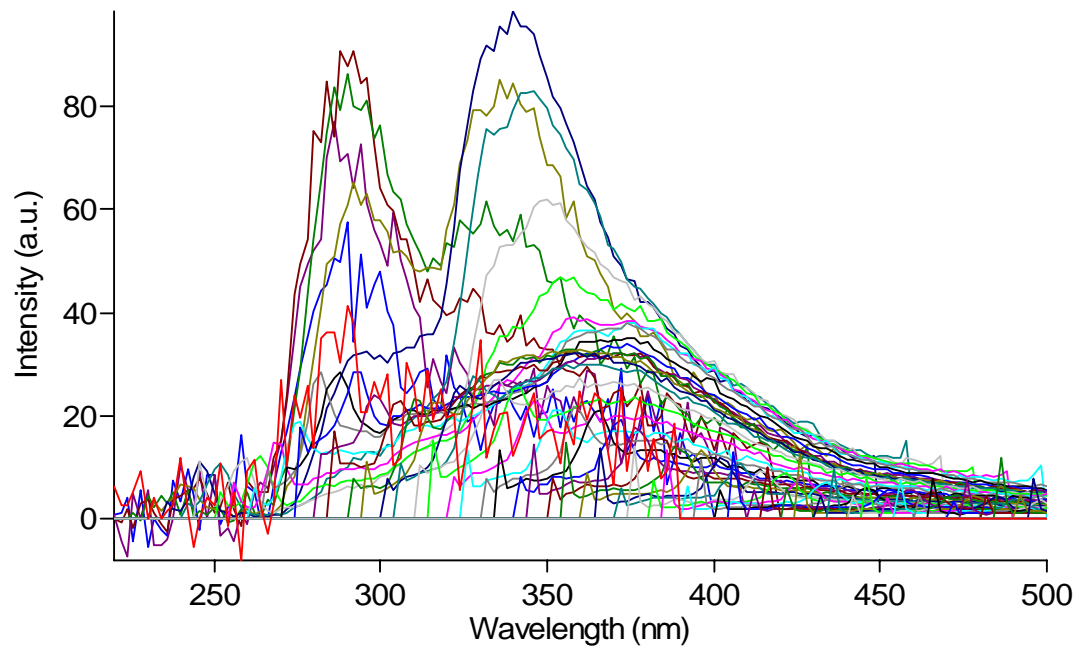
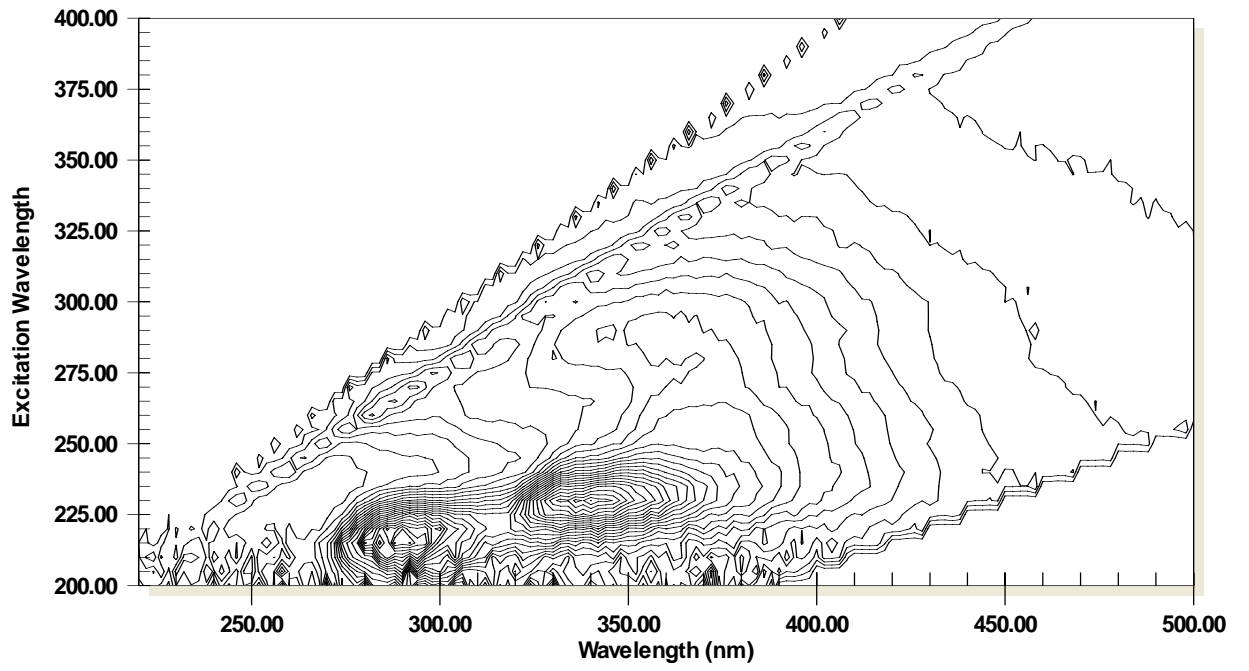
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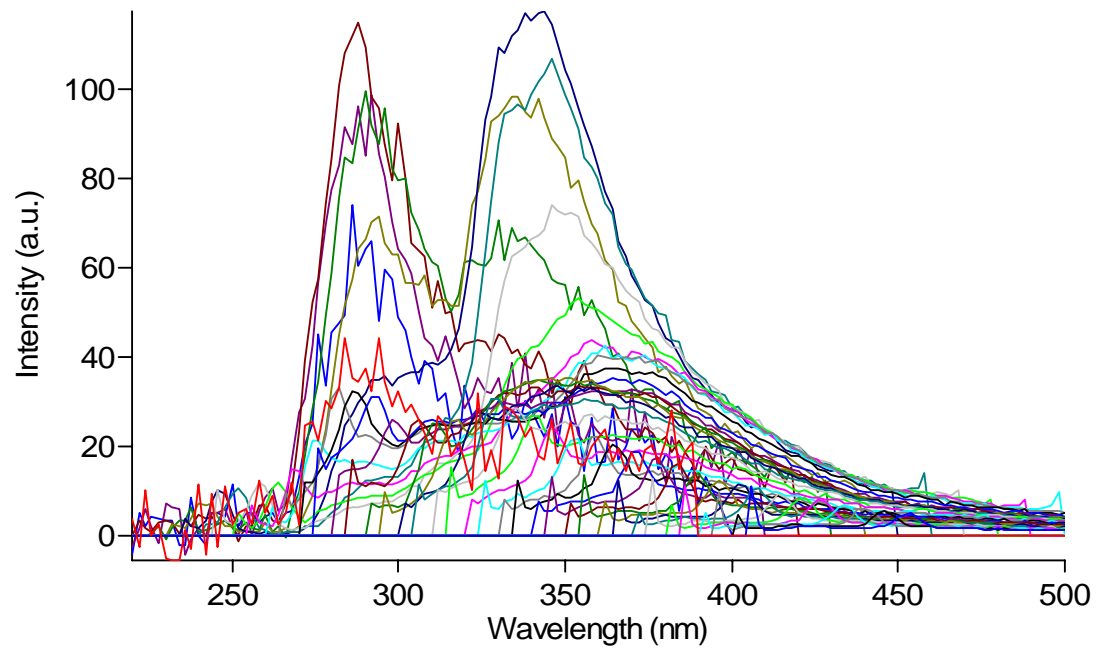
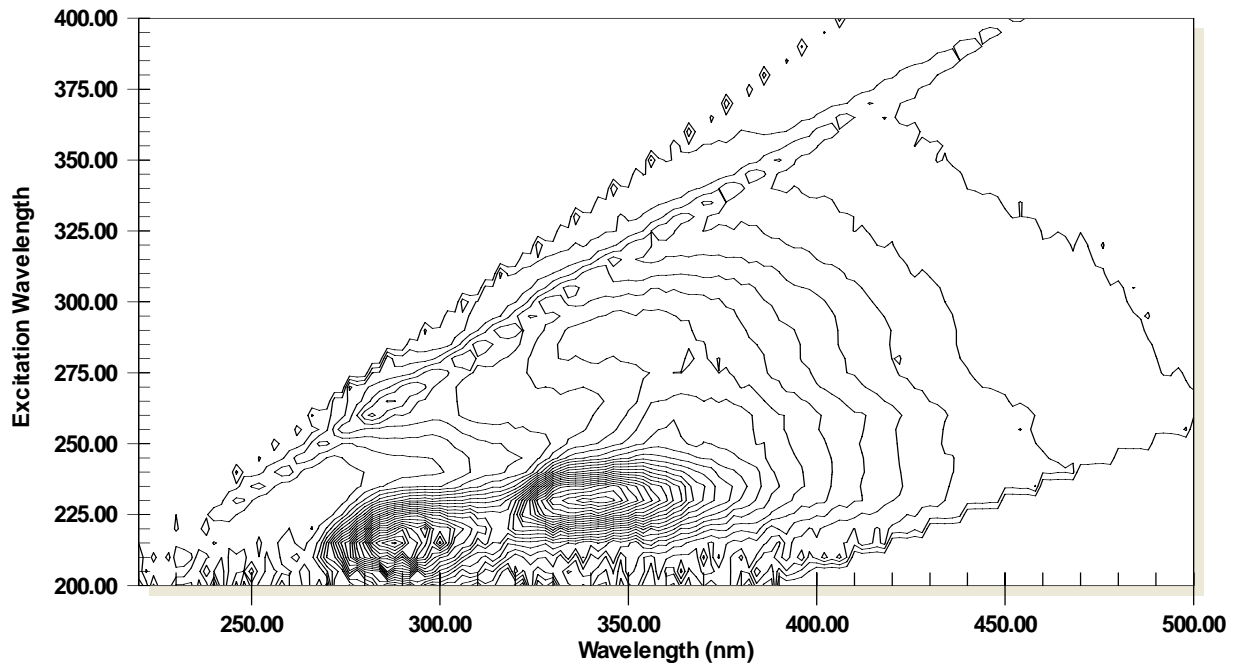
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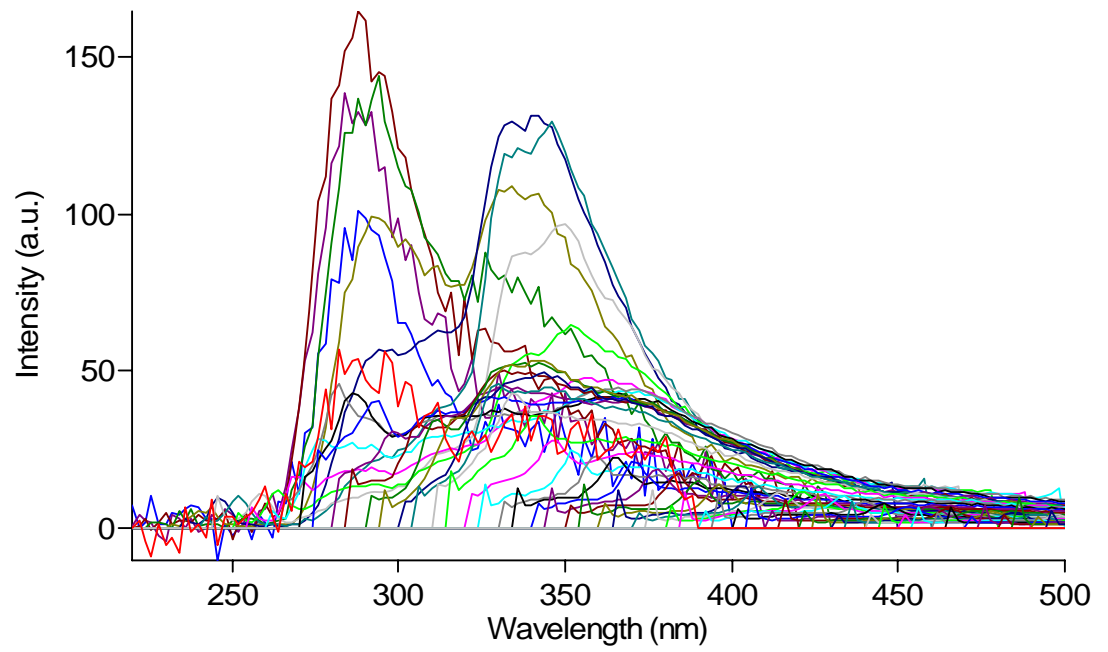
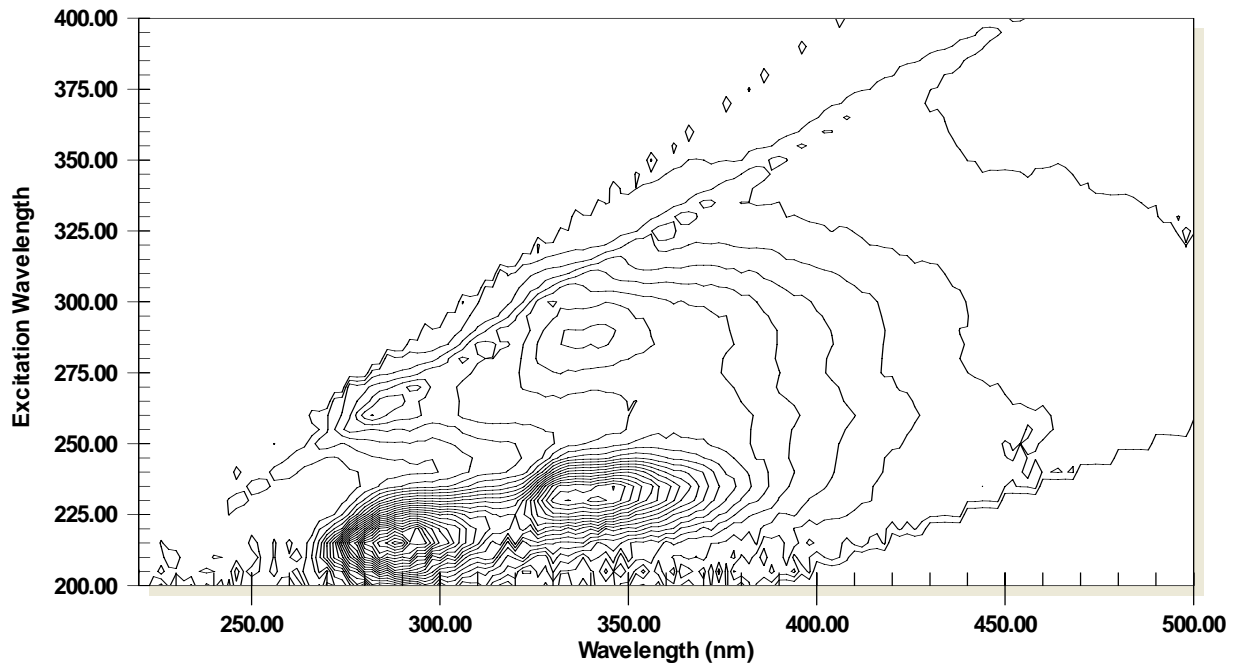
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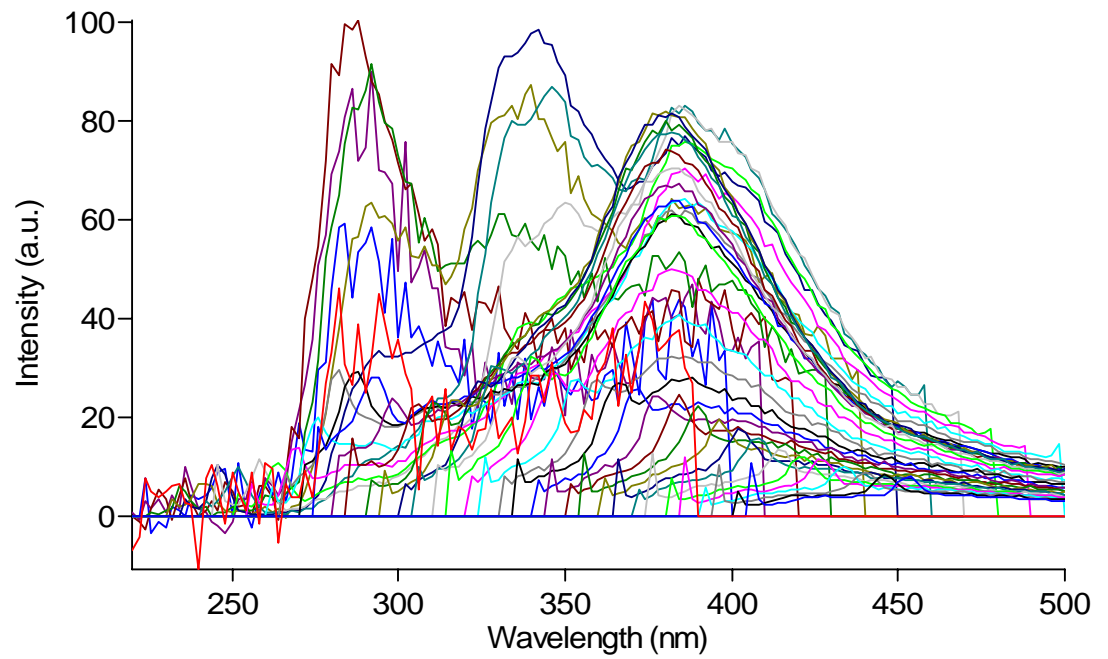
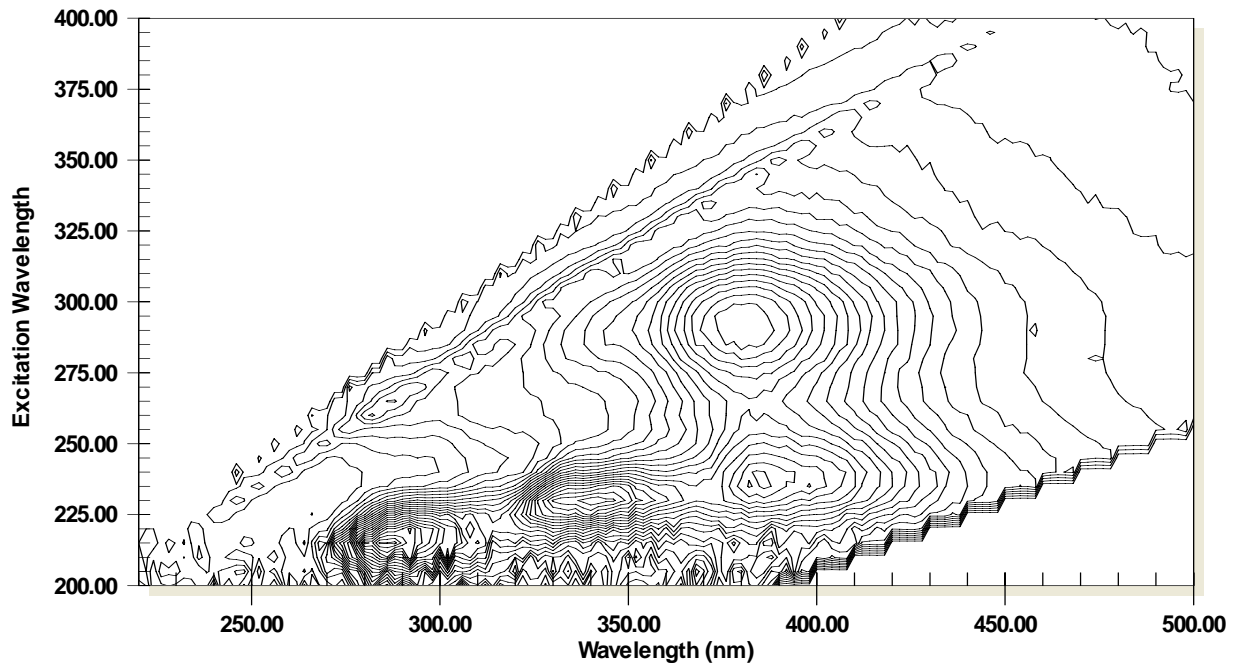
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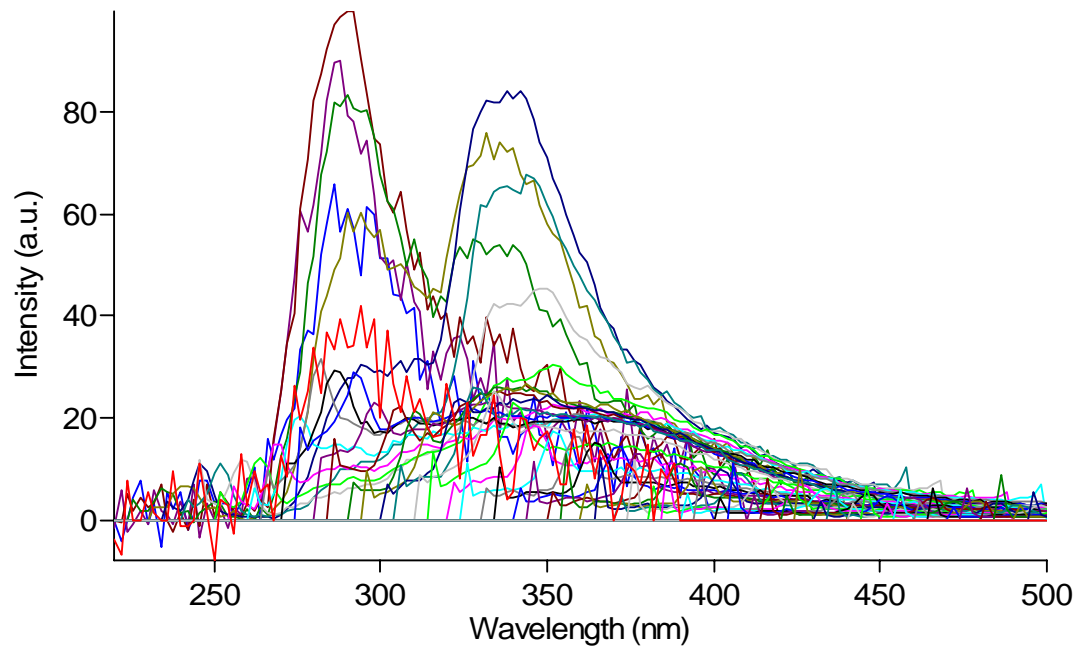
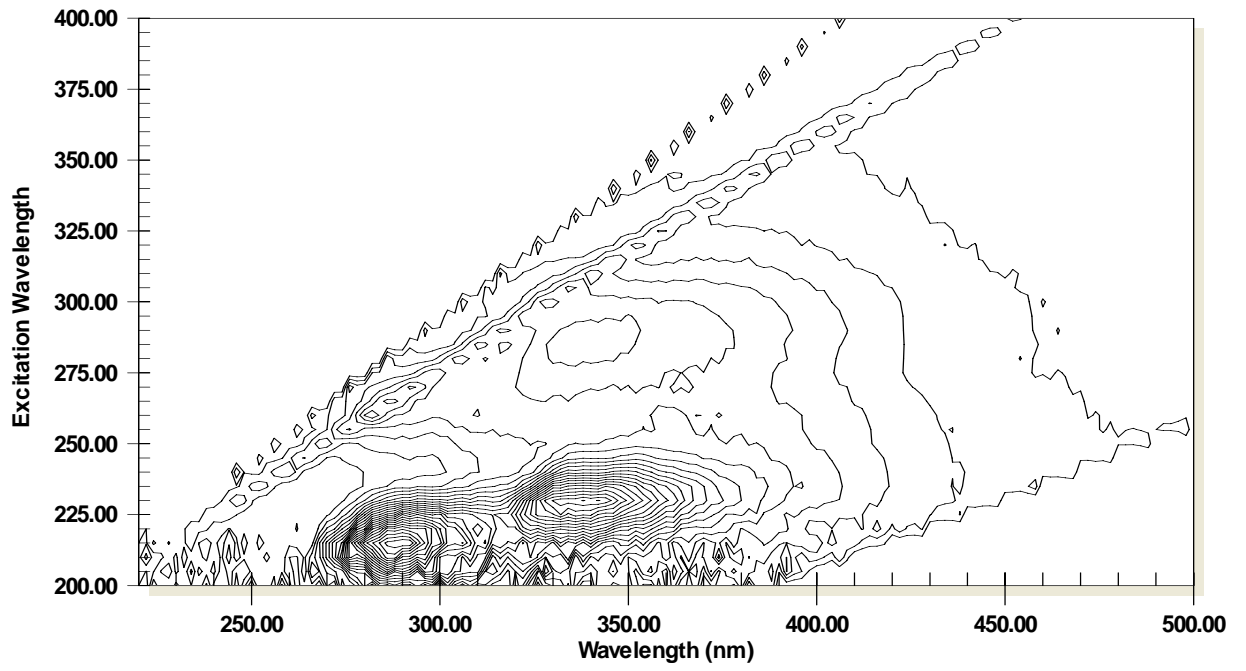
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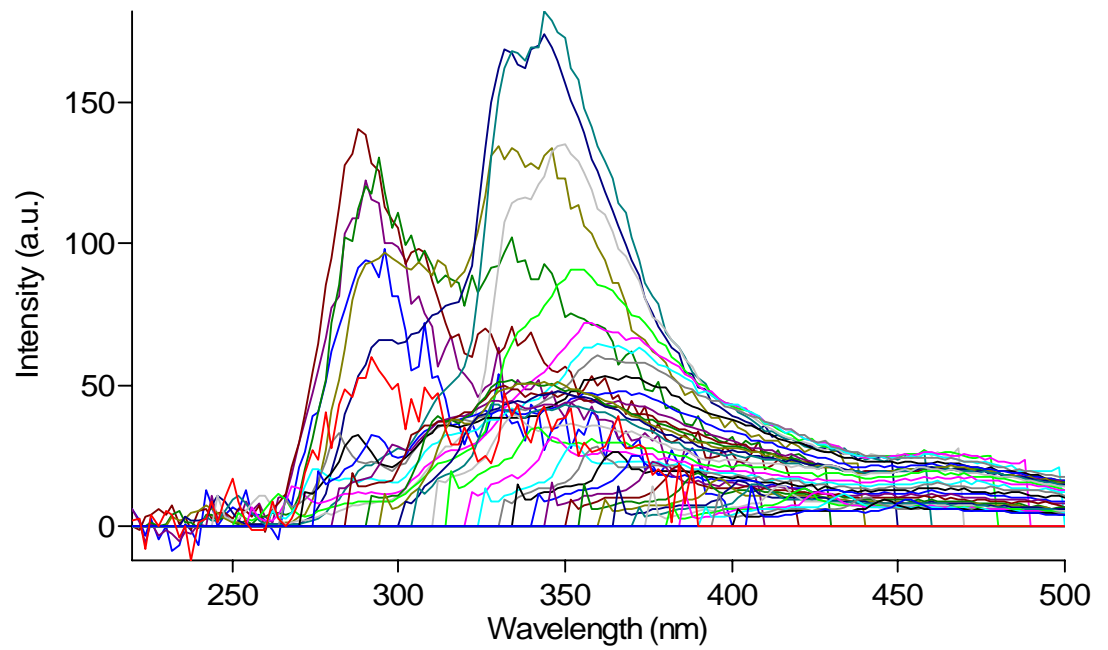
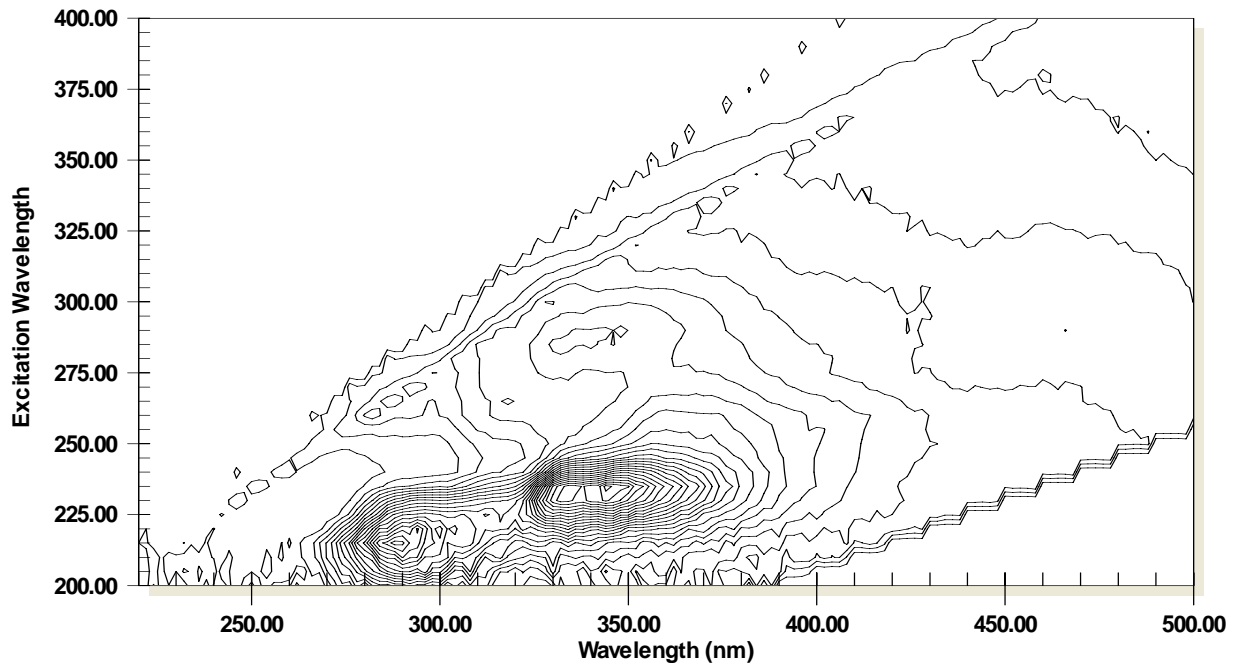
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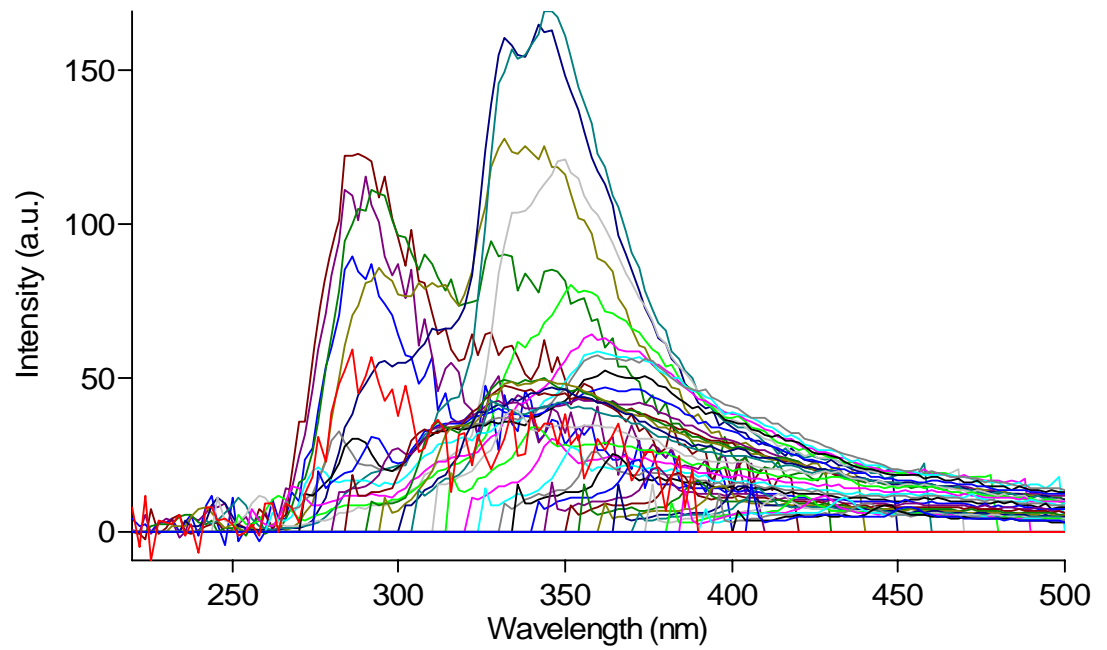
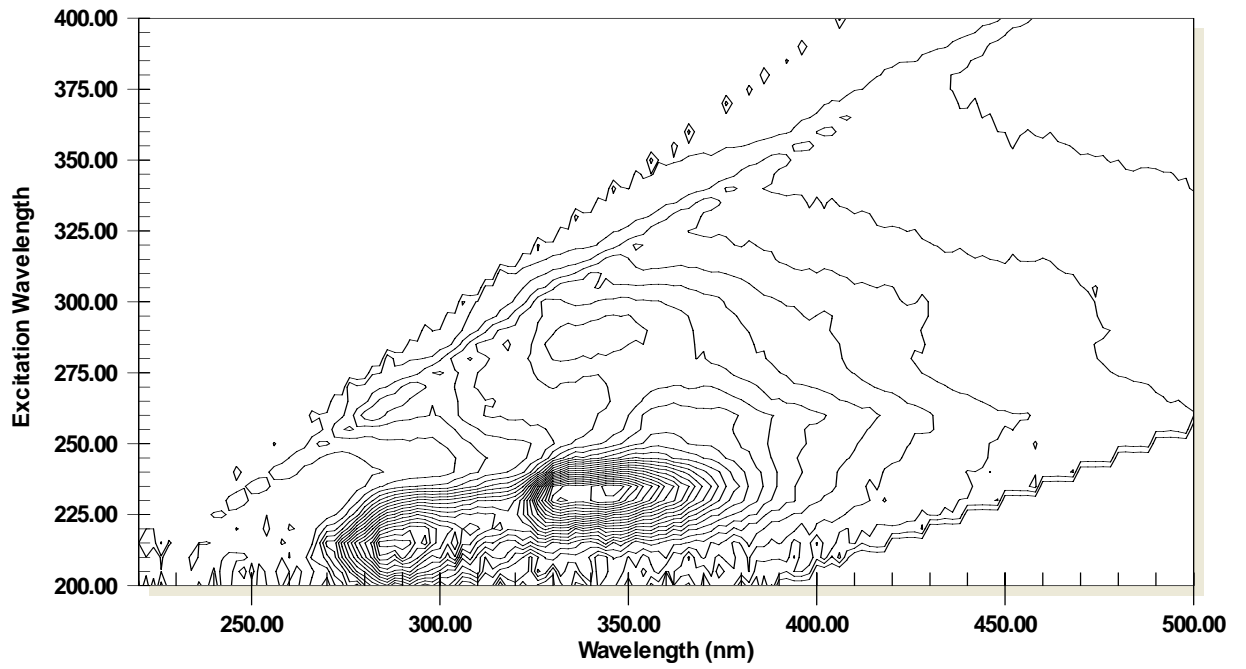
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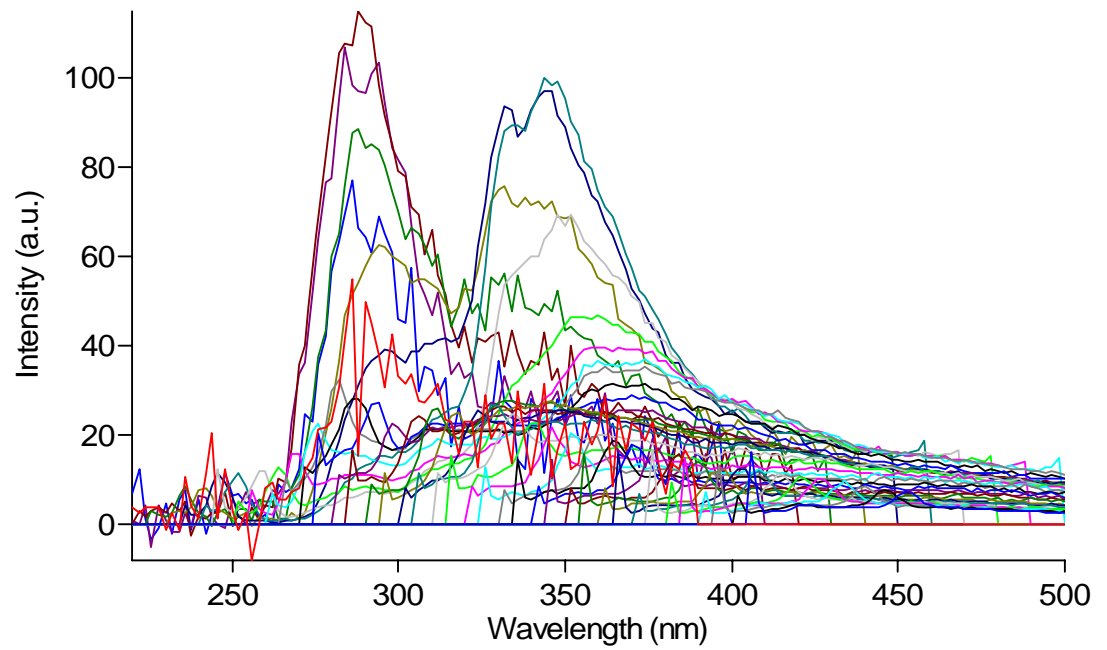
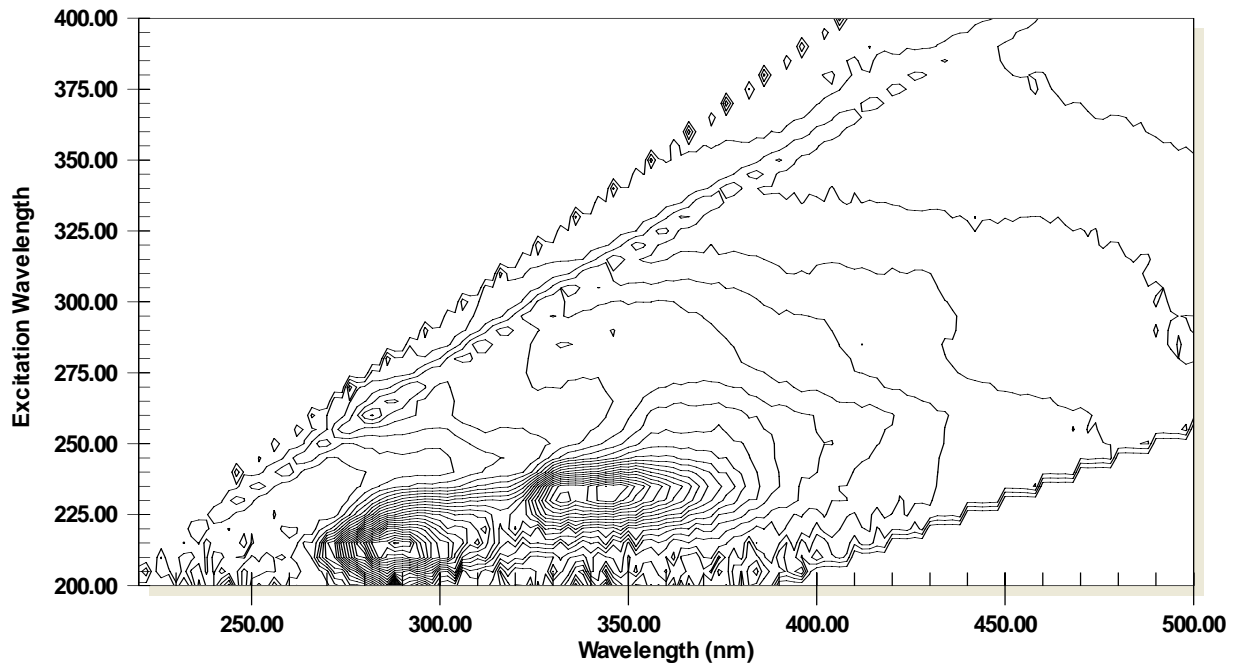
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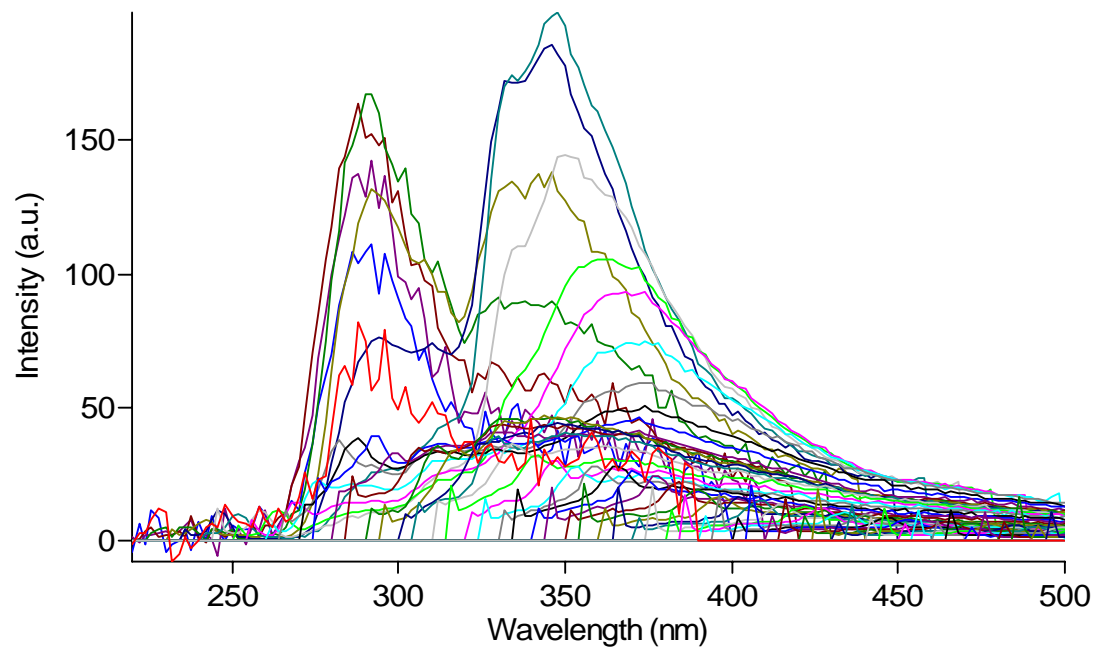
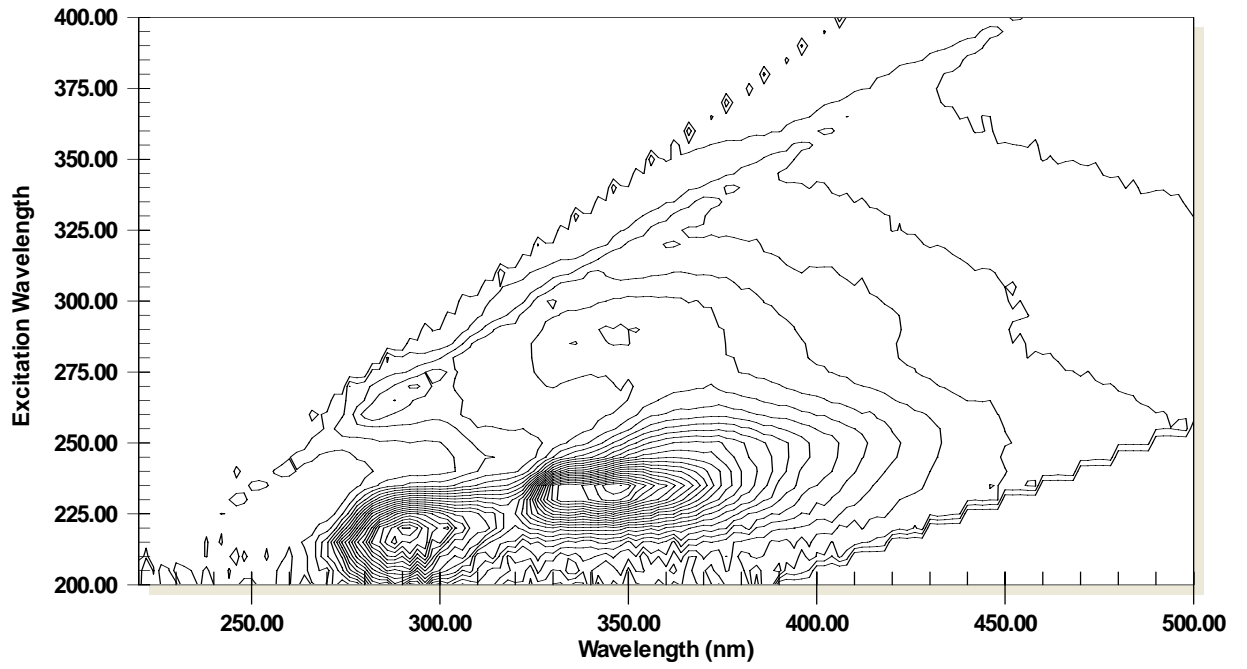
62CT



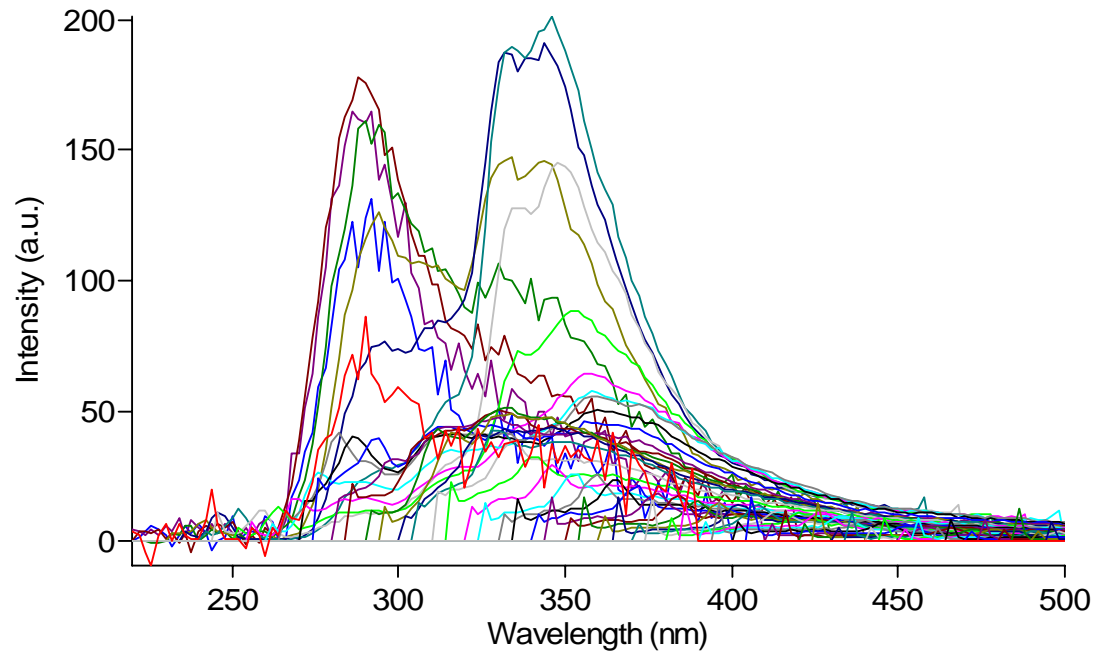
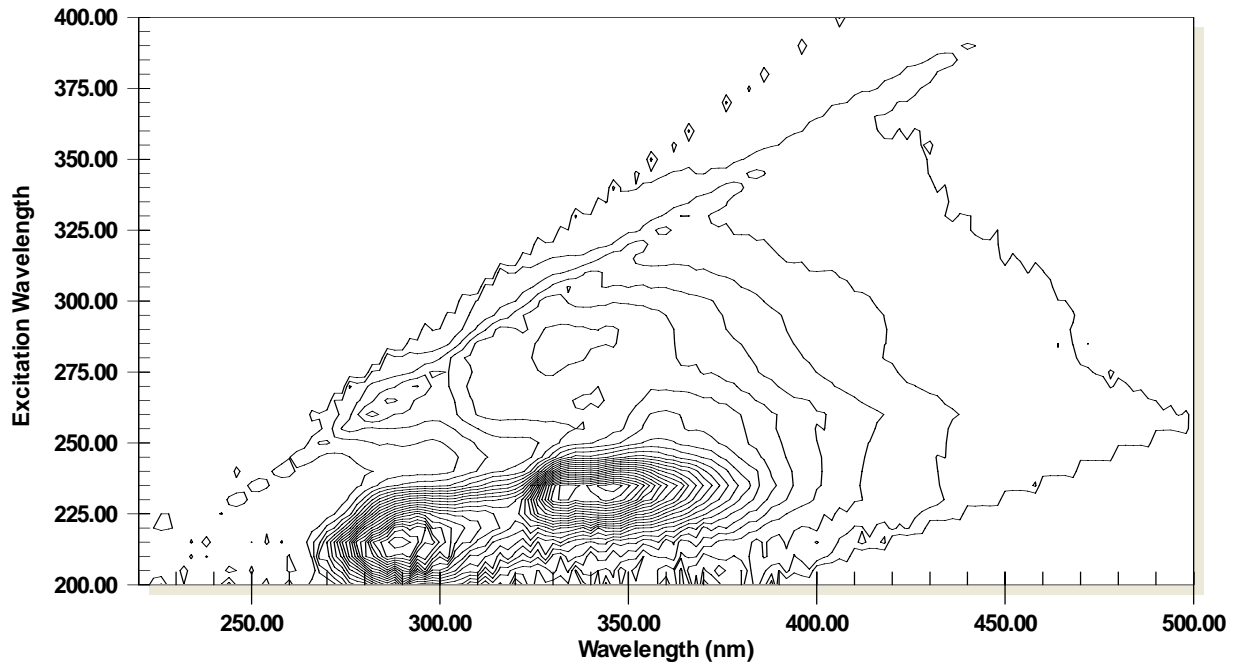
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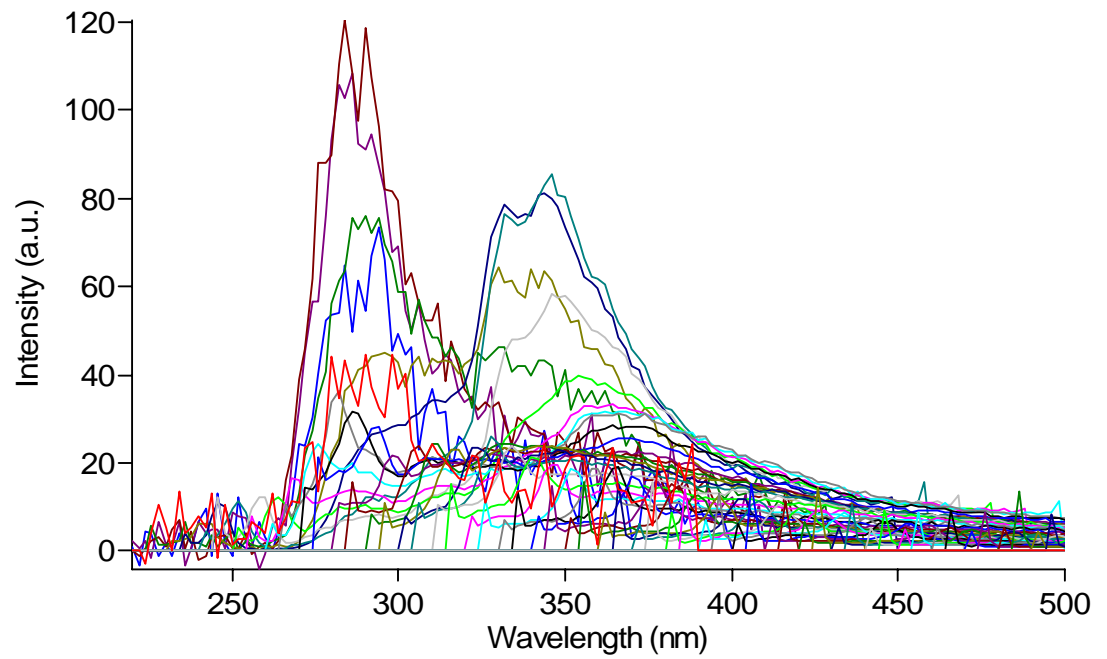
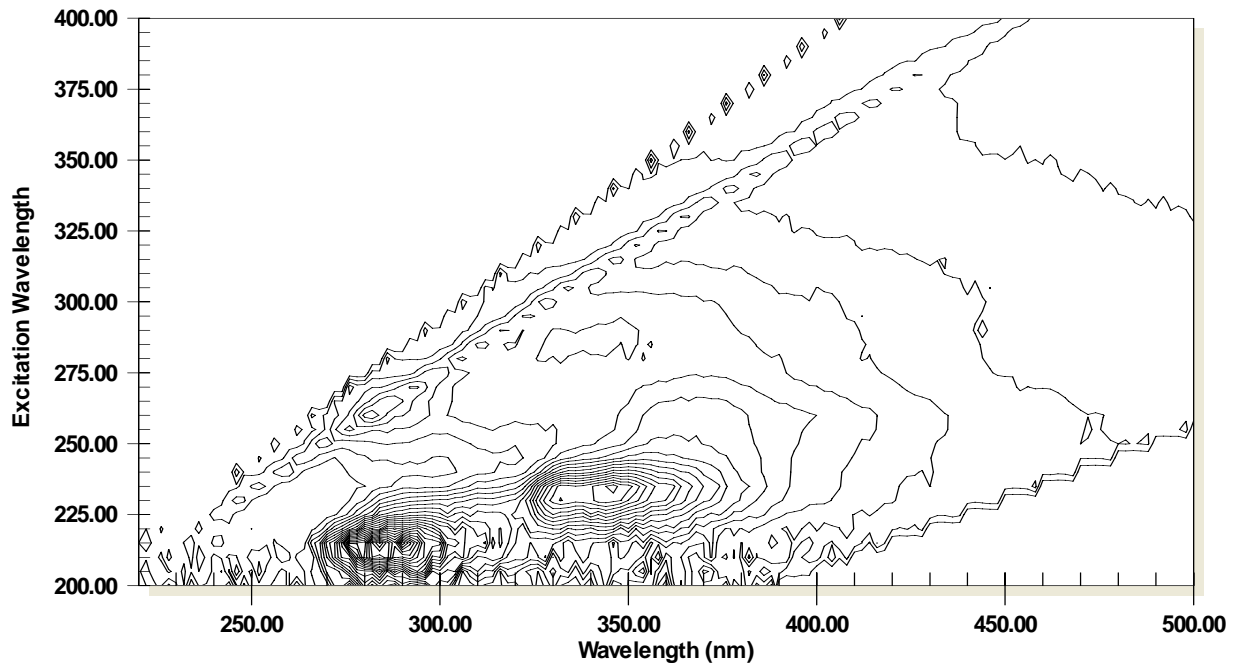
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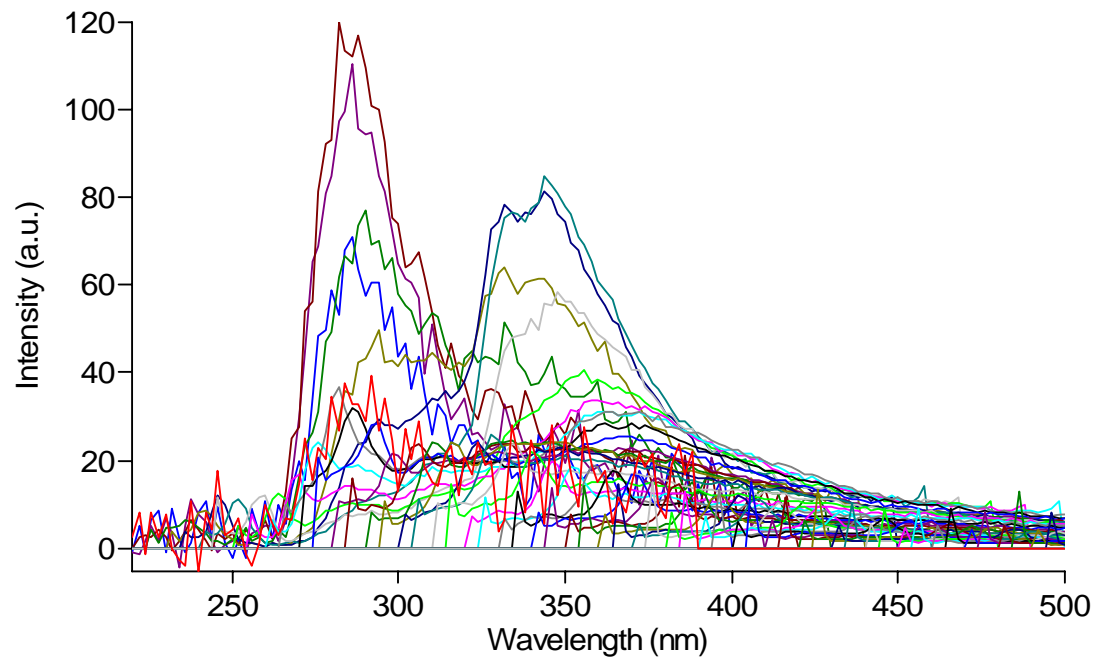
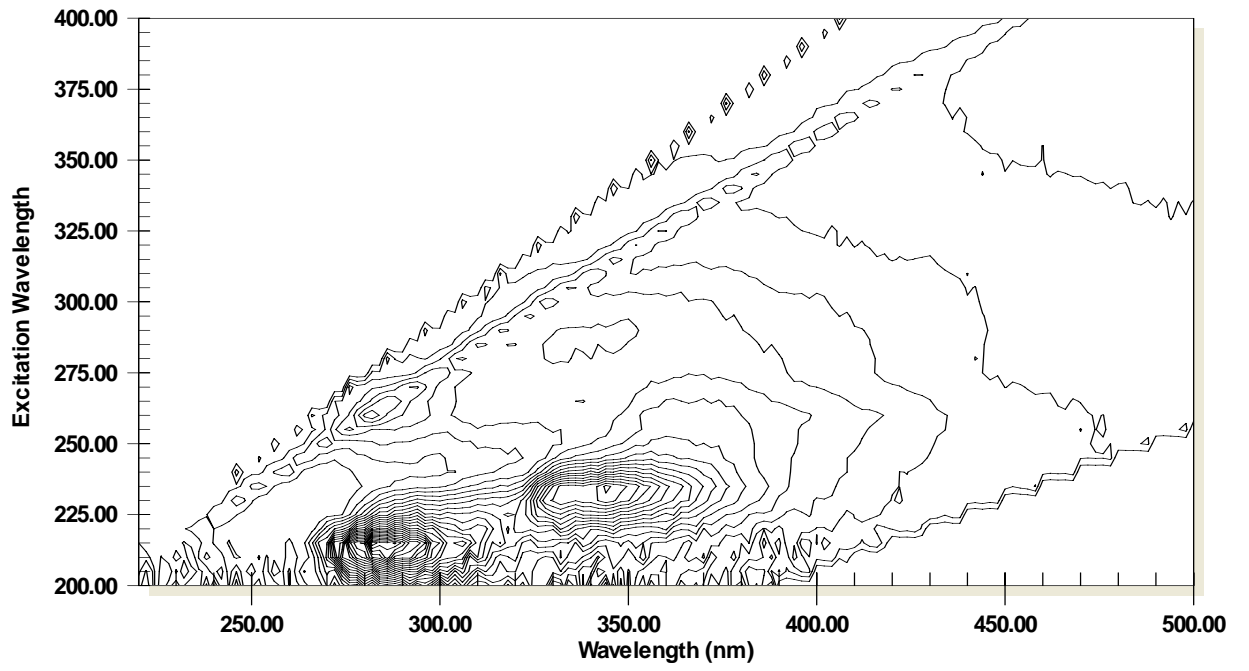
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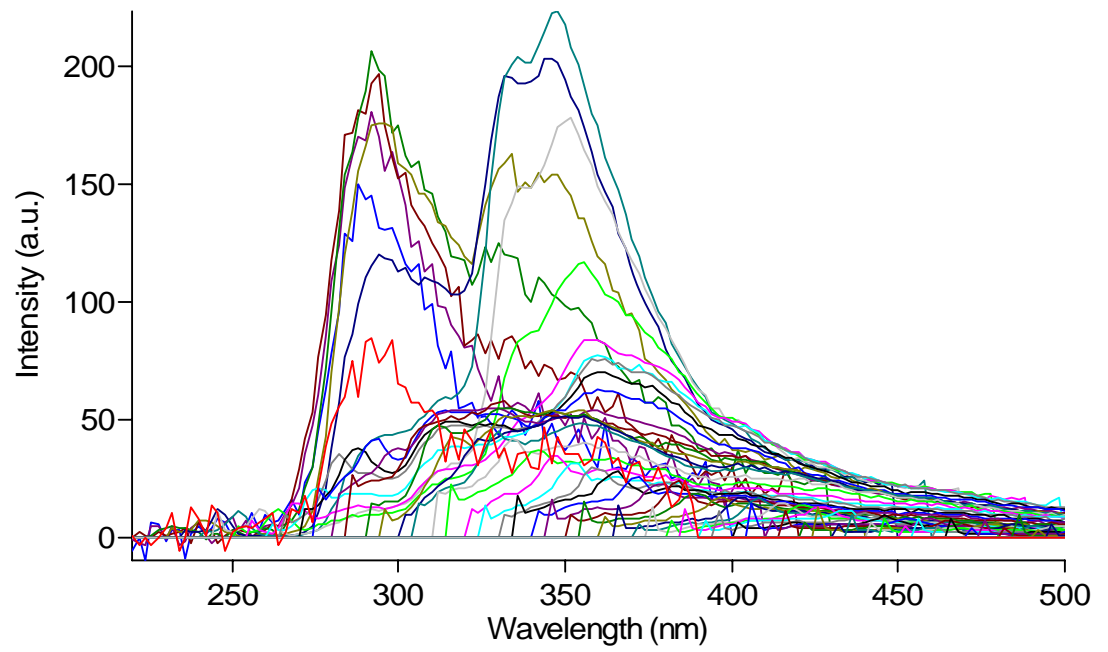
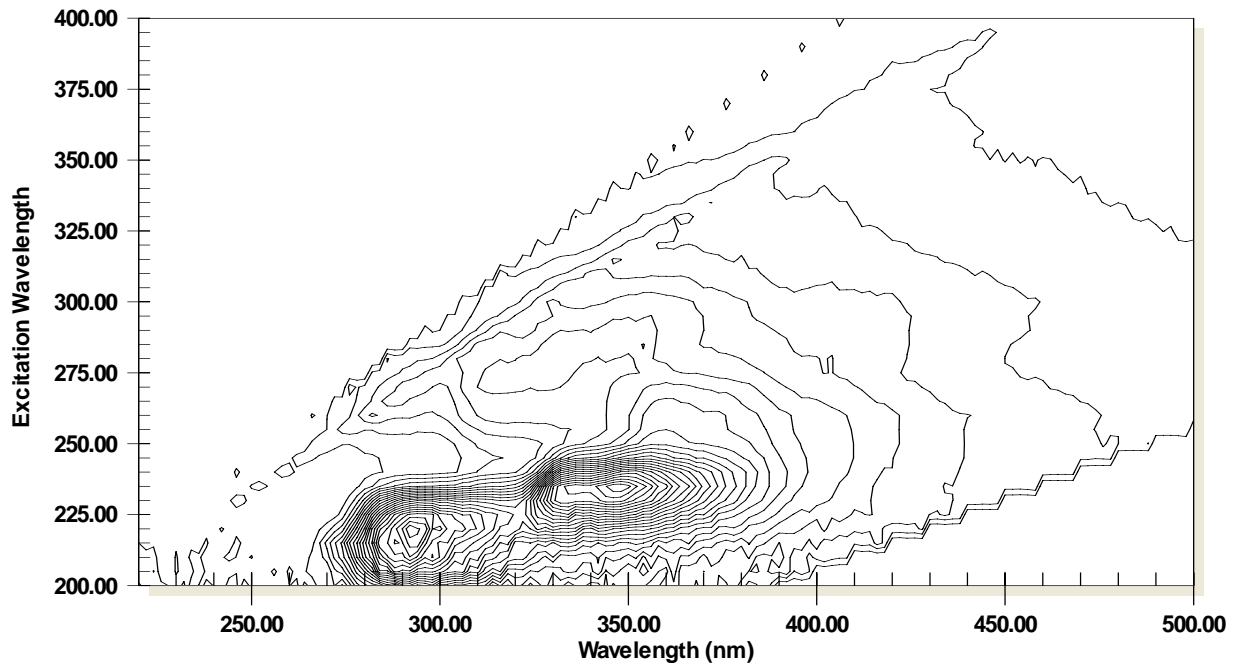
71BT



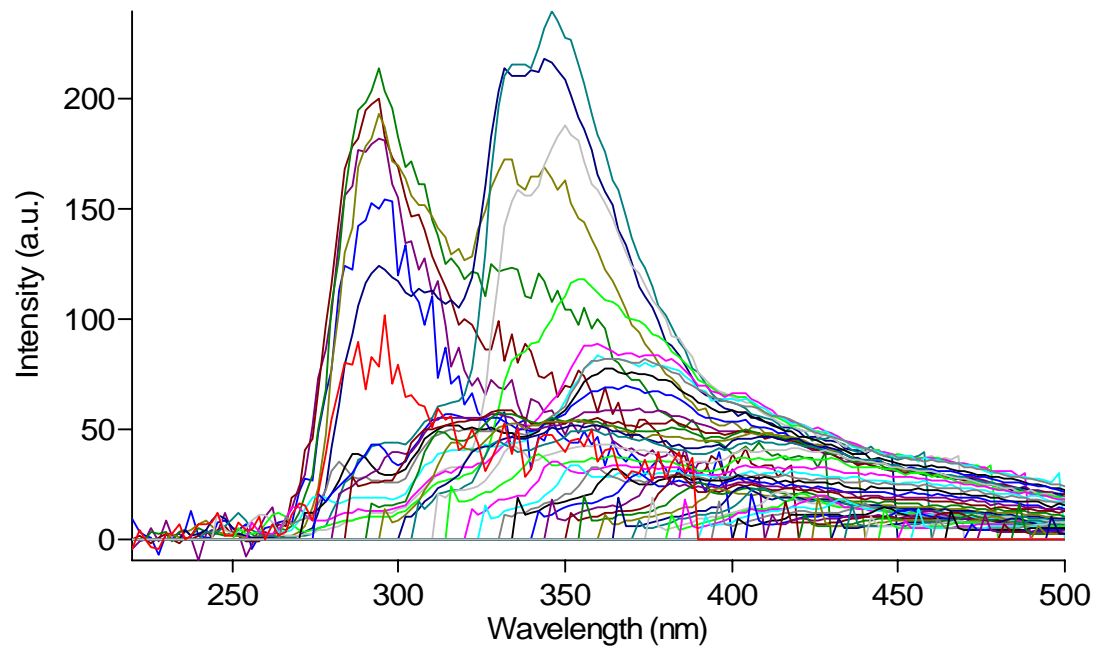
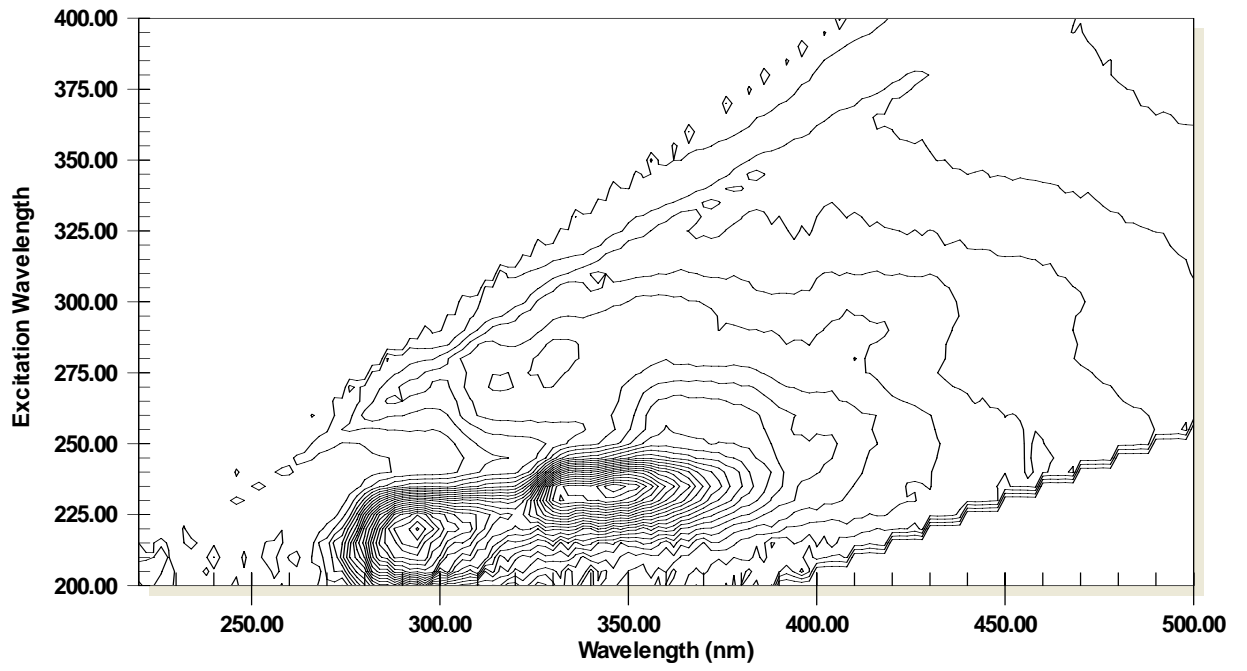
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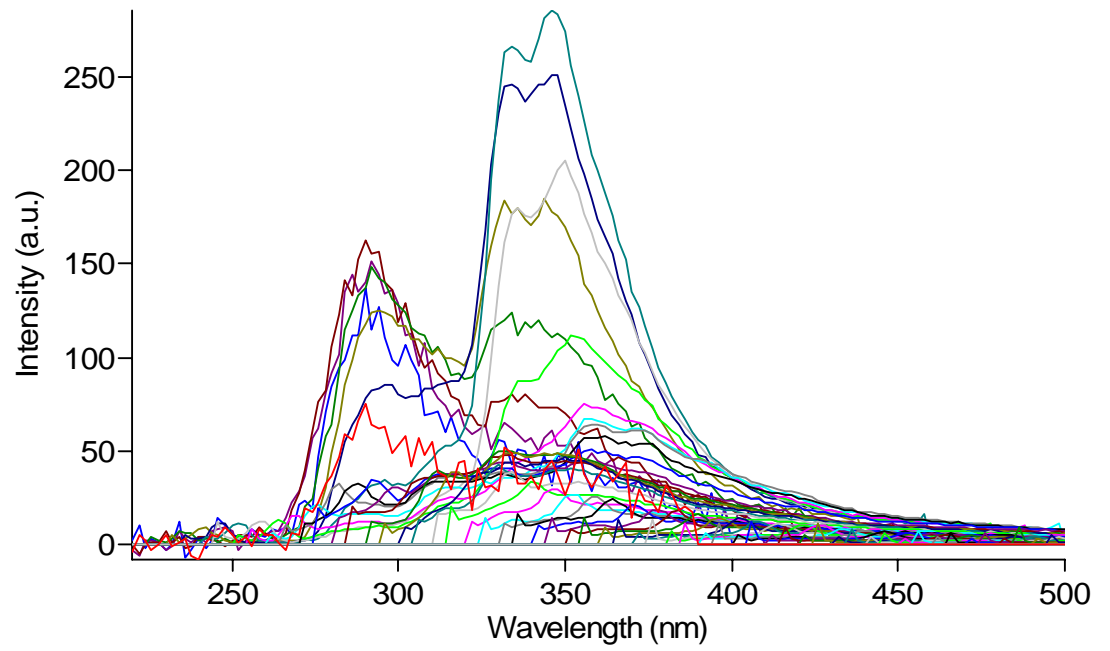
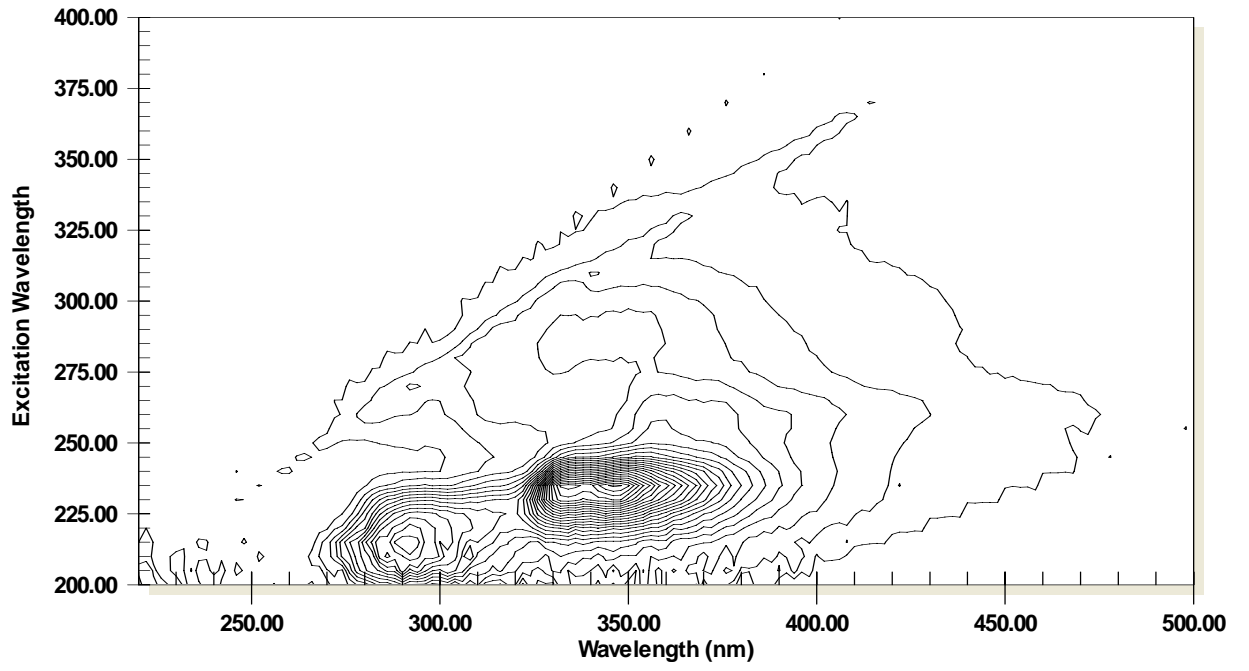
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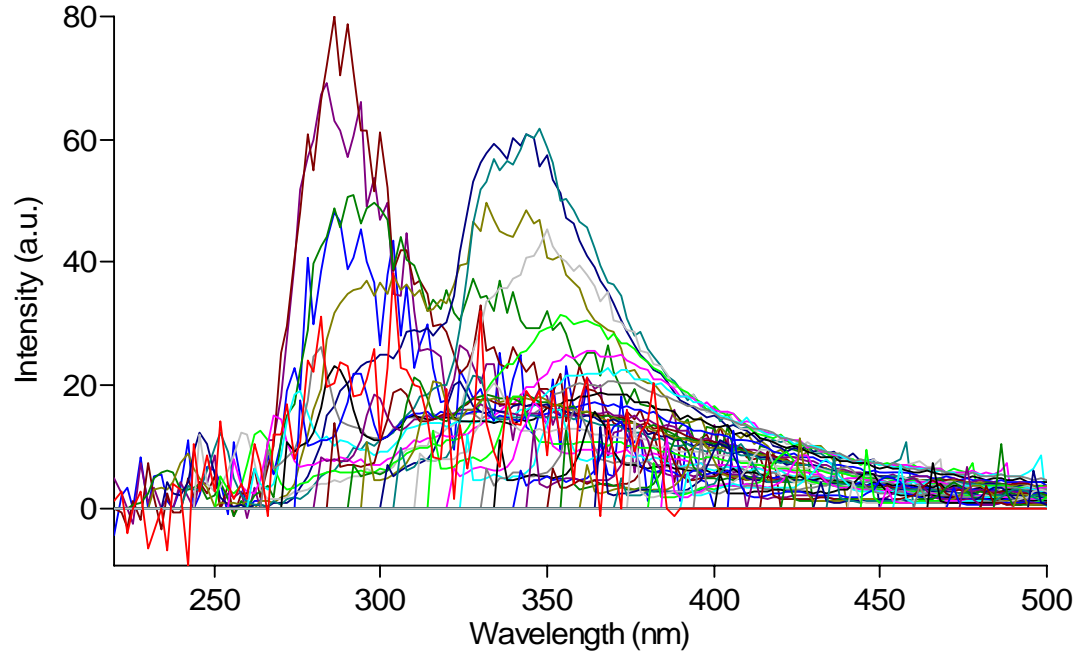
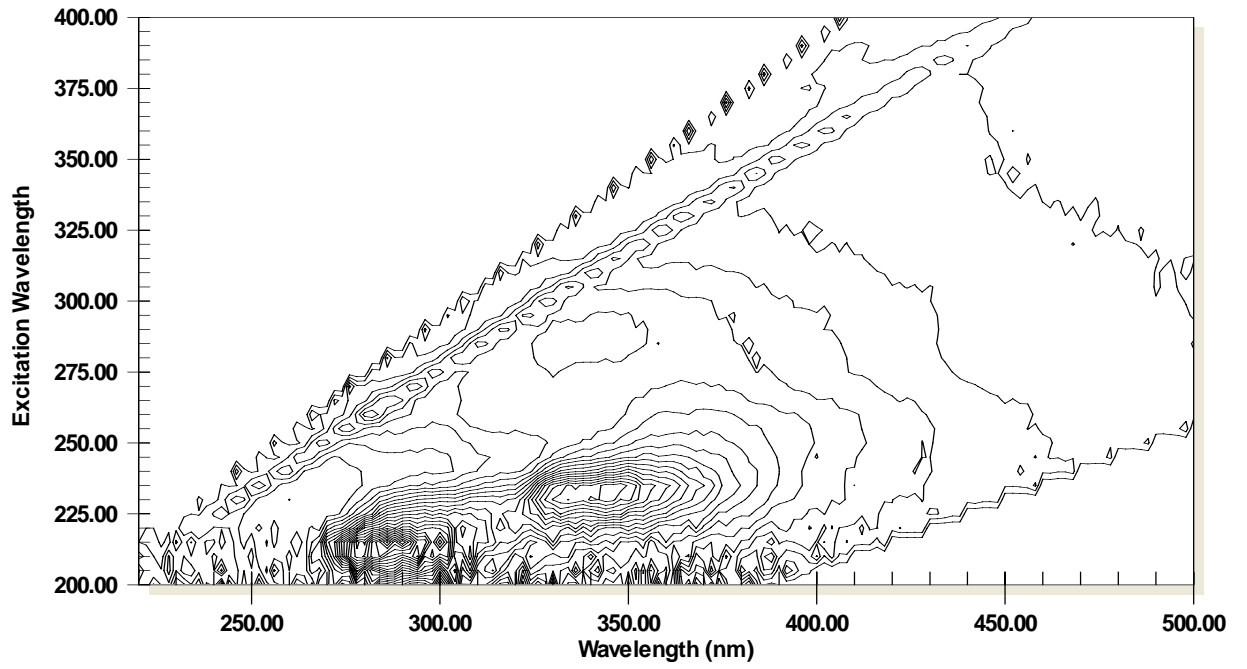
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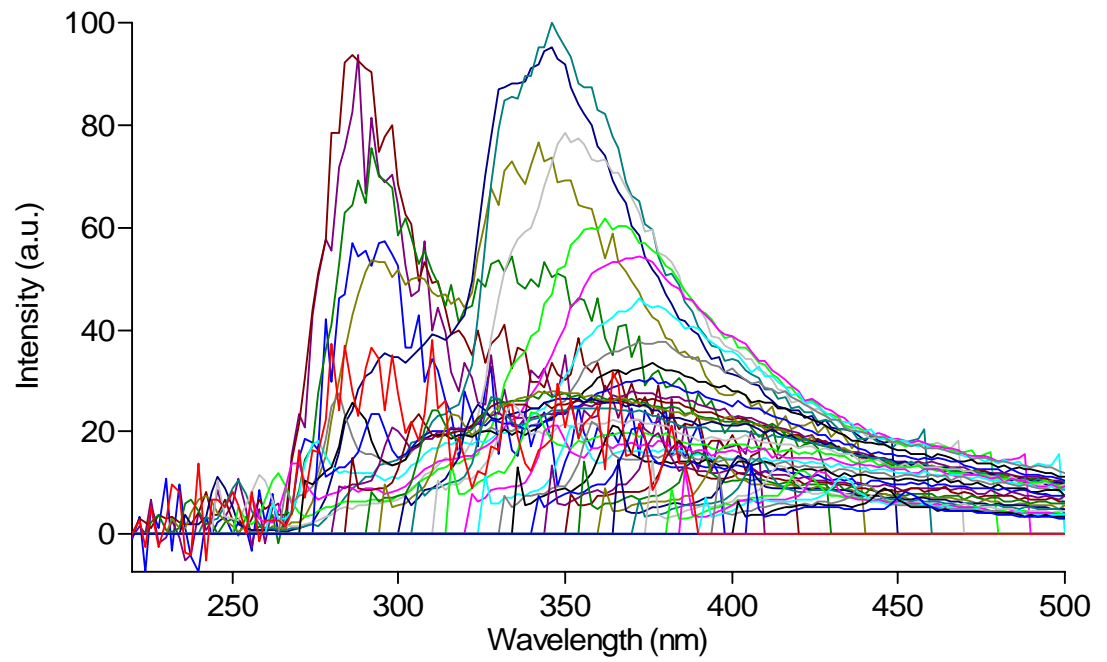
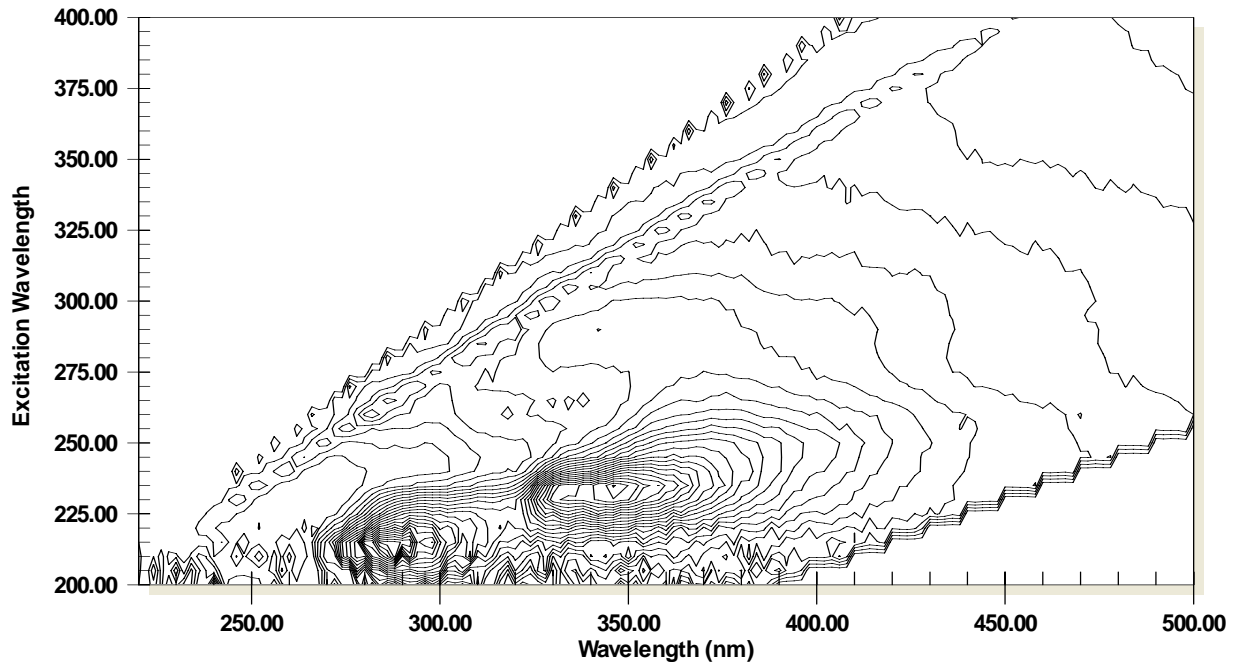
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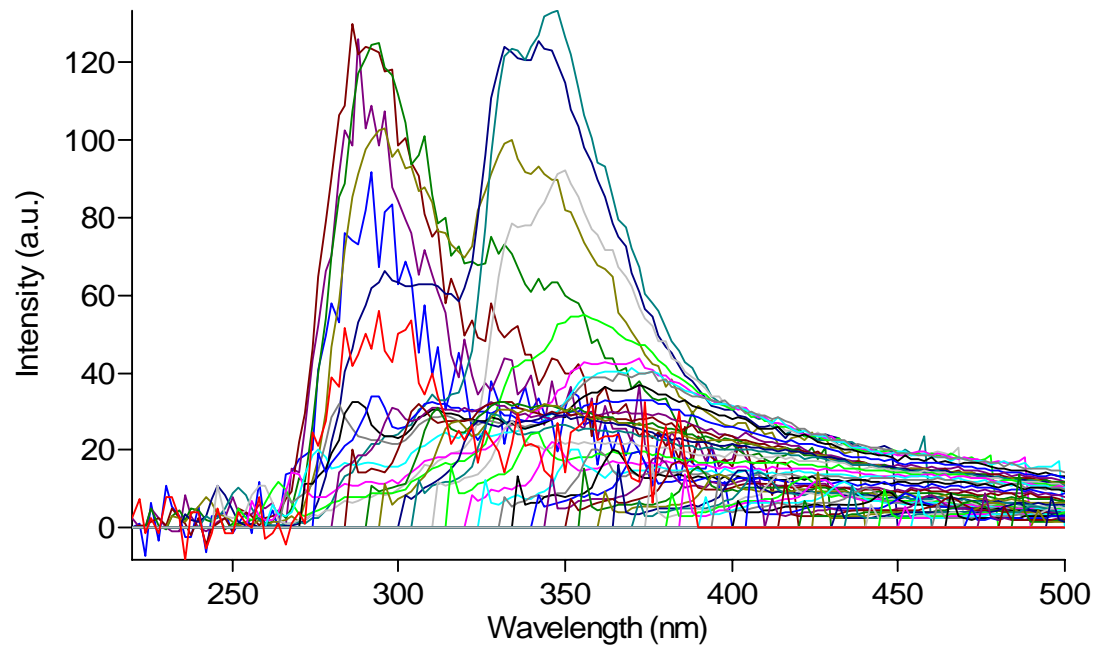
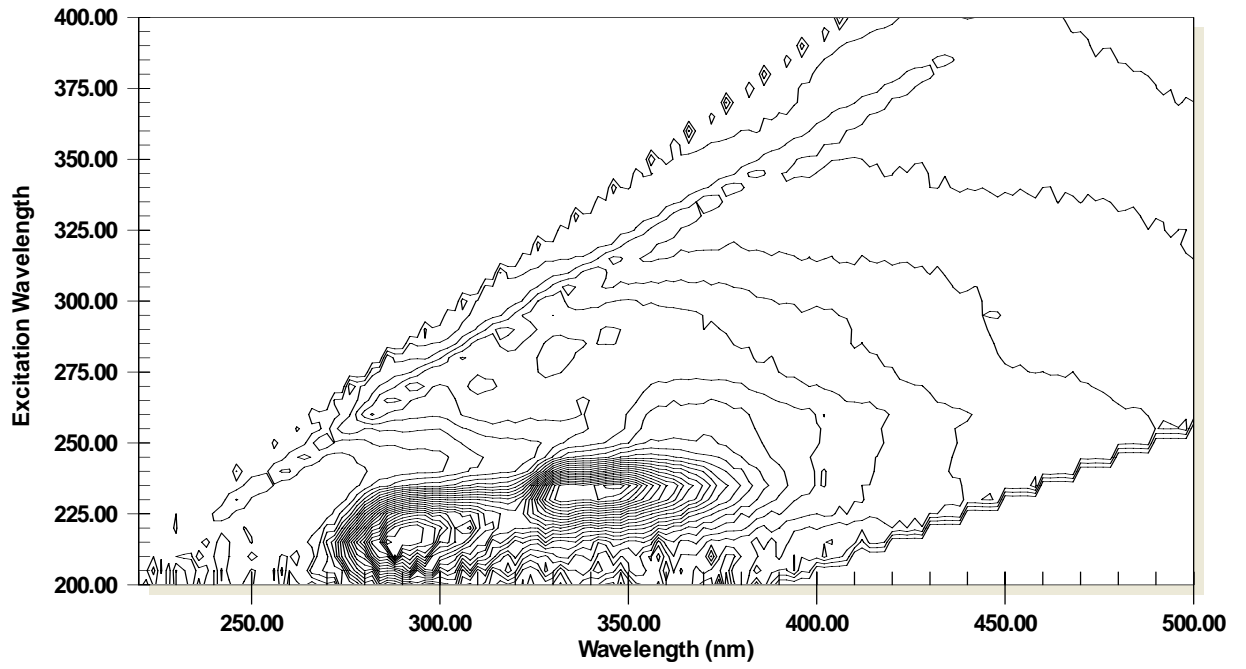
81AT



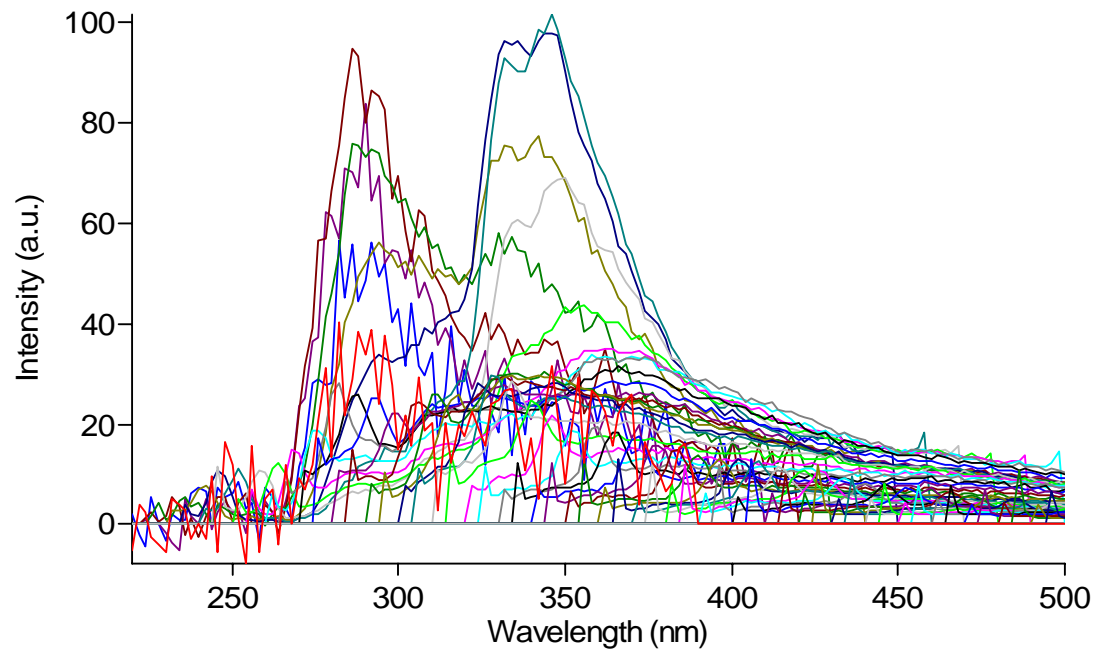
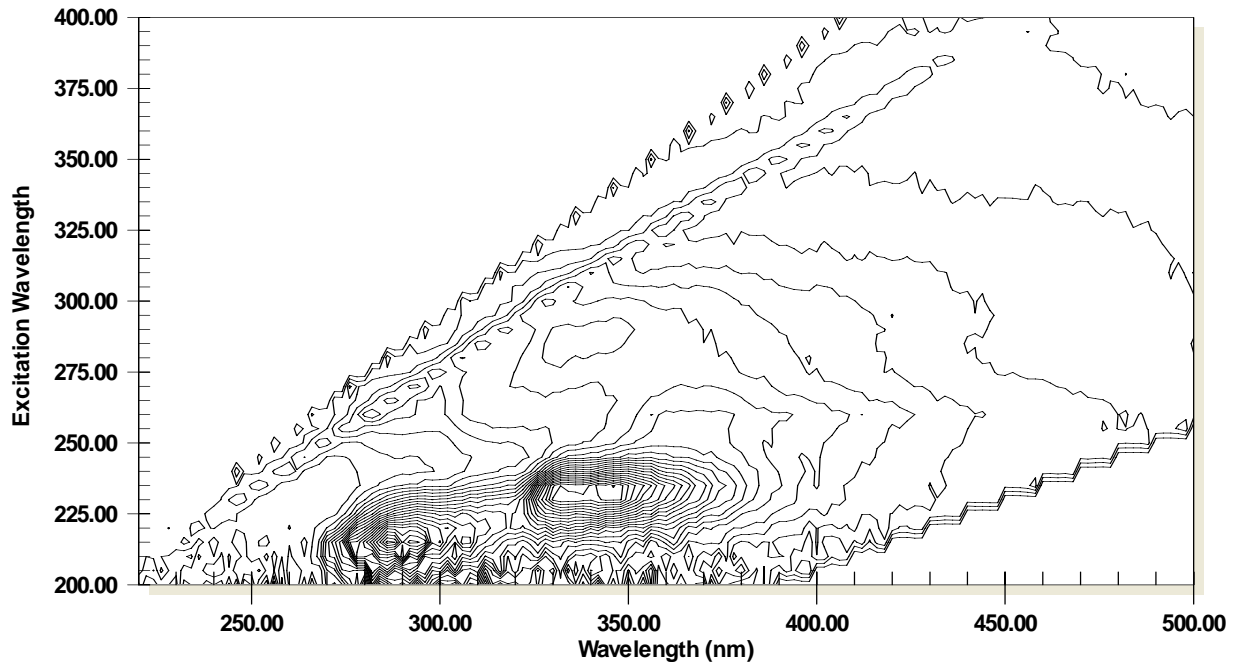
82BT



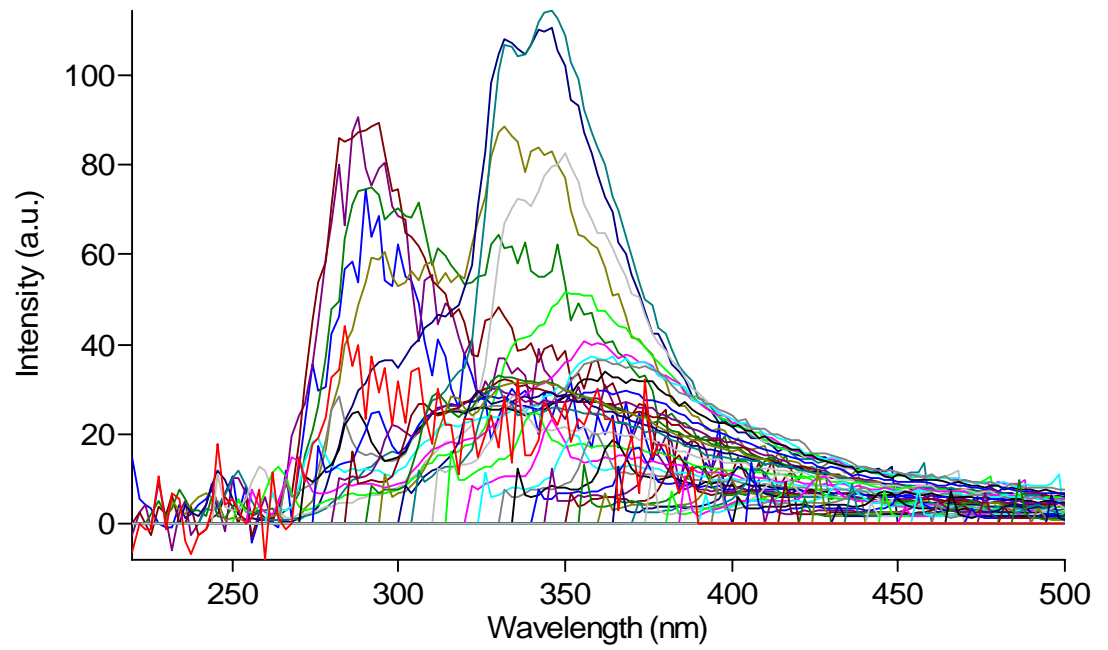
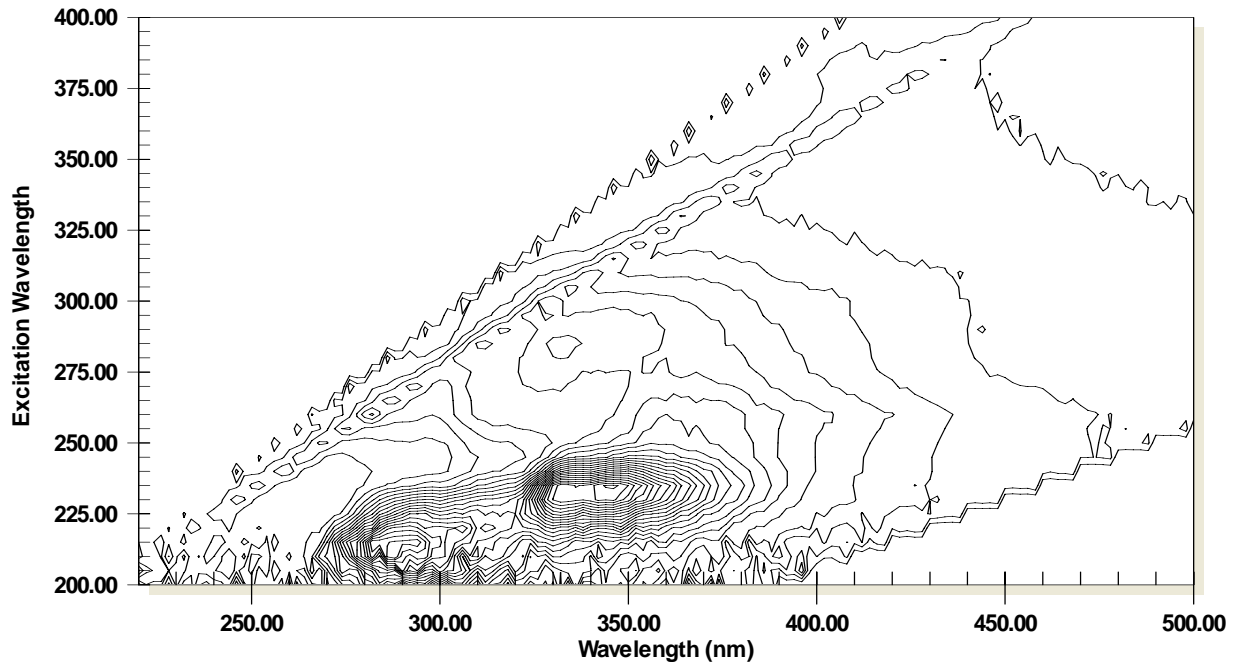
83BT



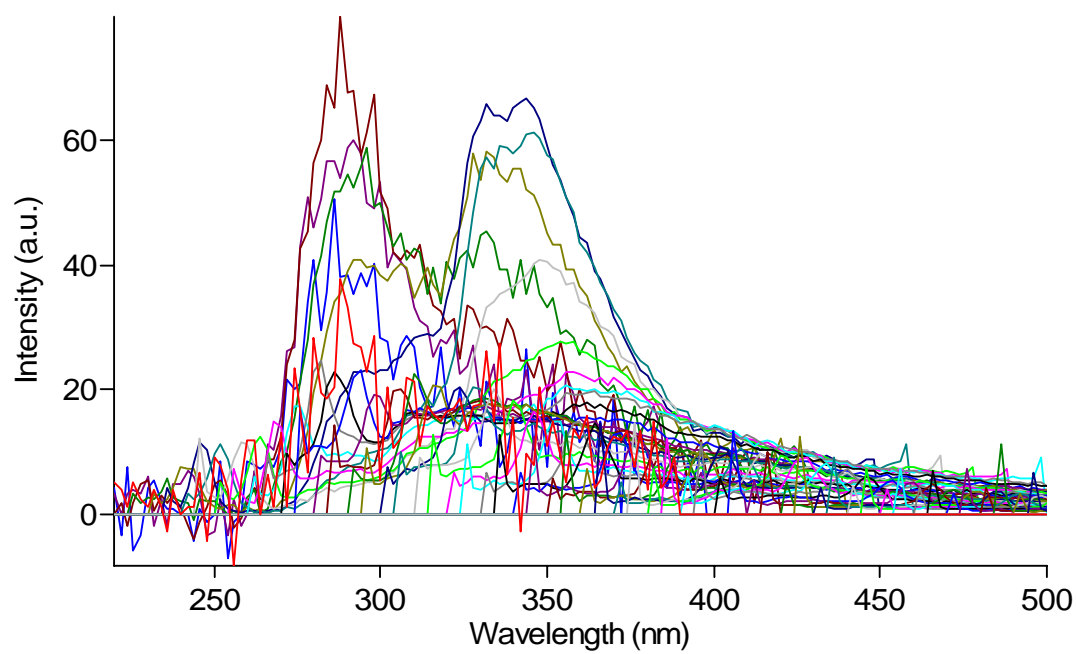
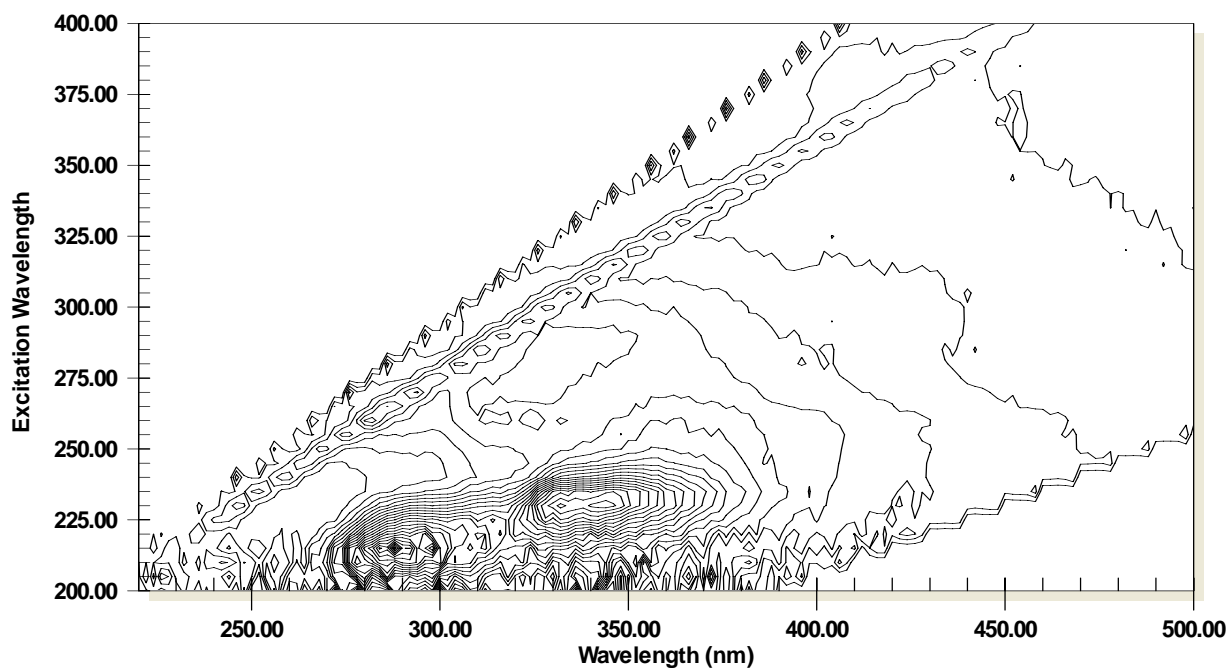
84BT



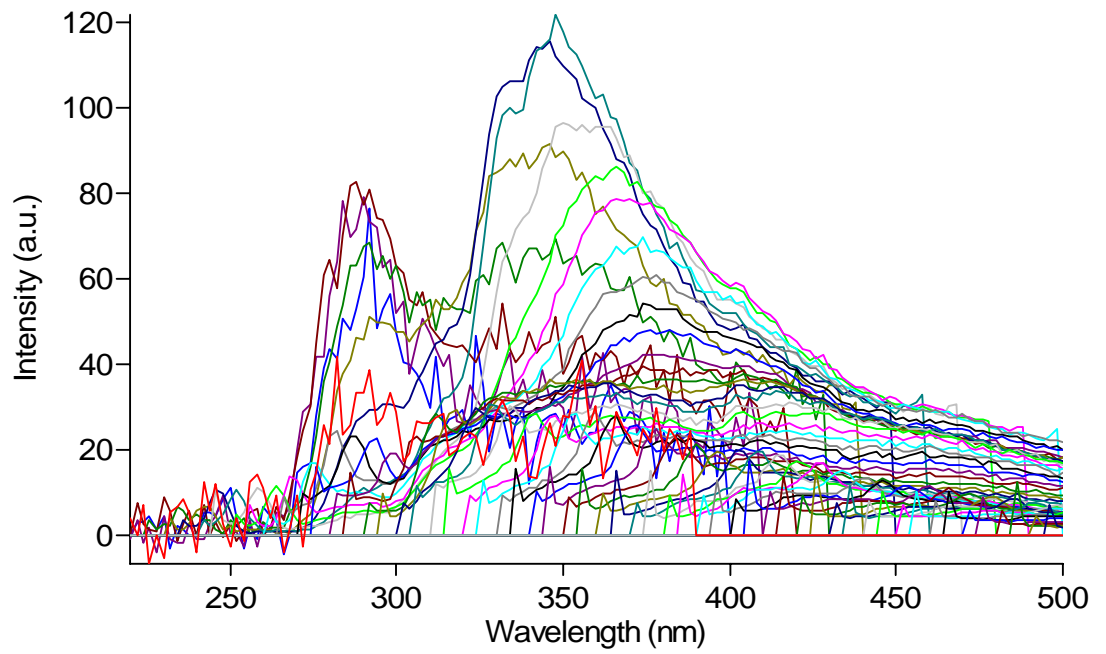
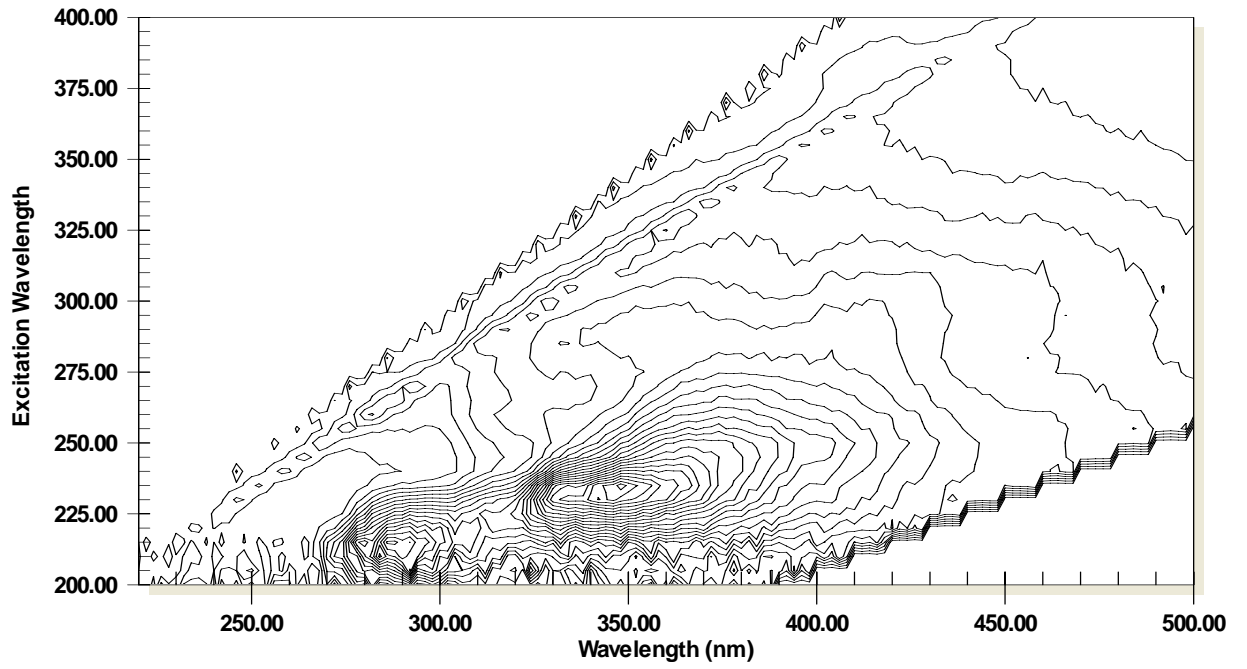
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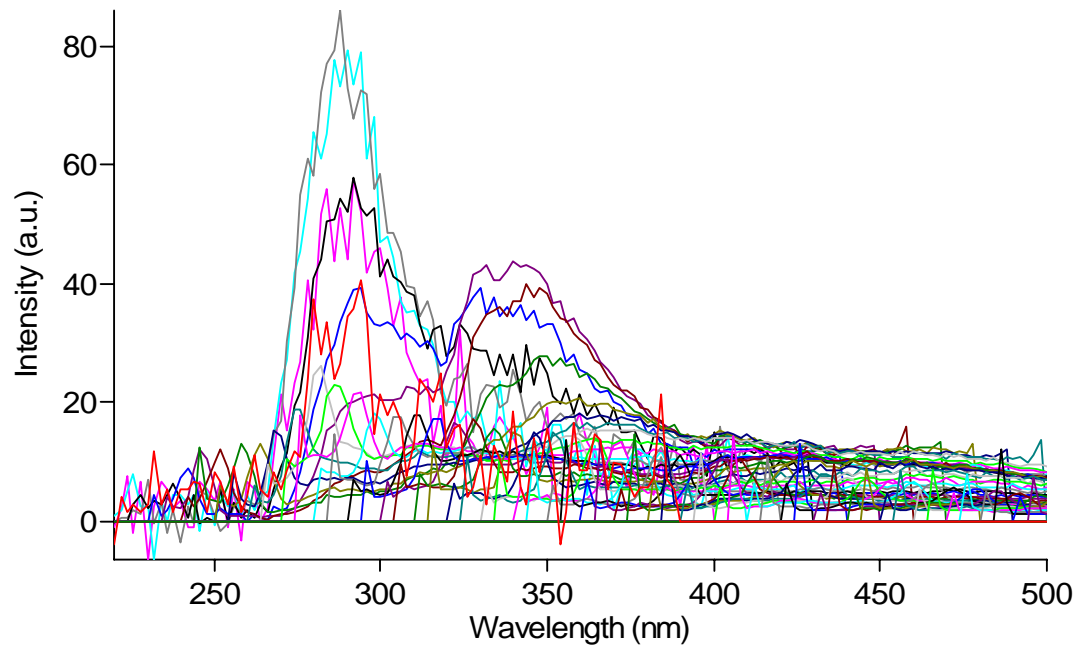
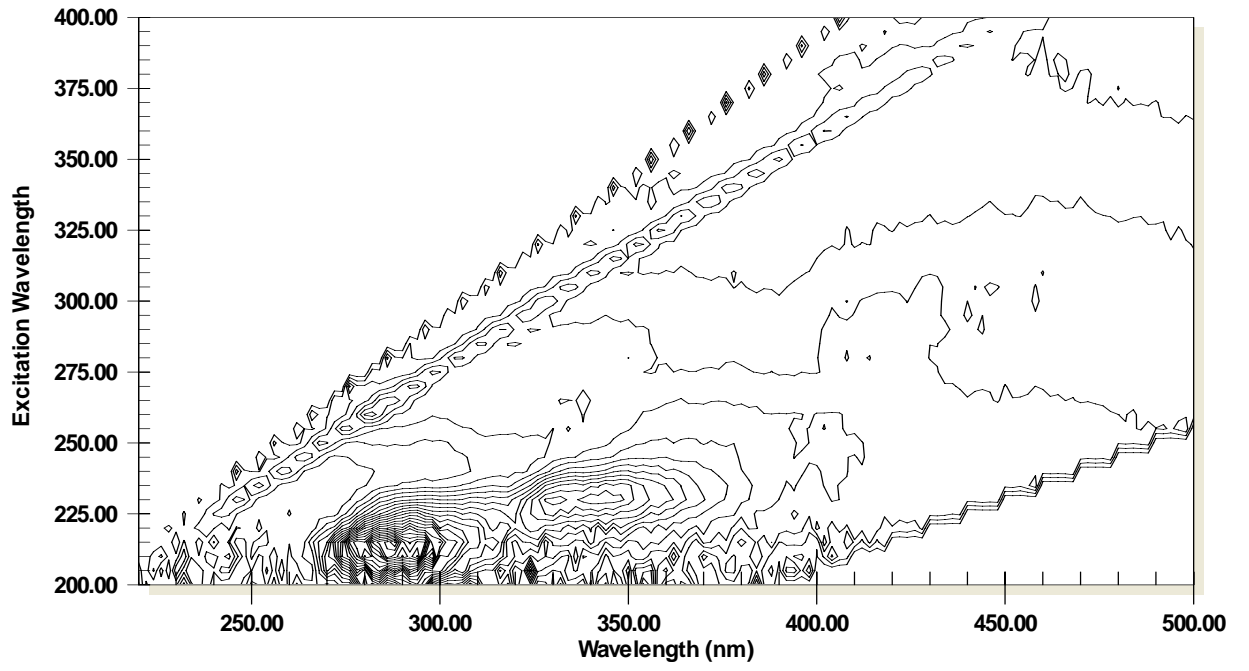
91AT



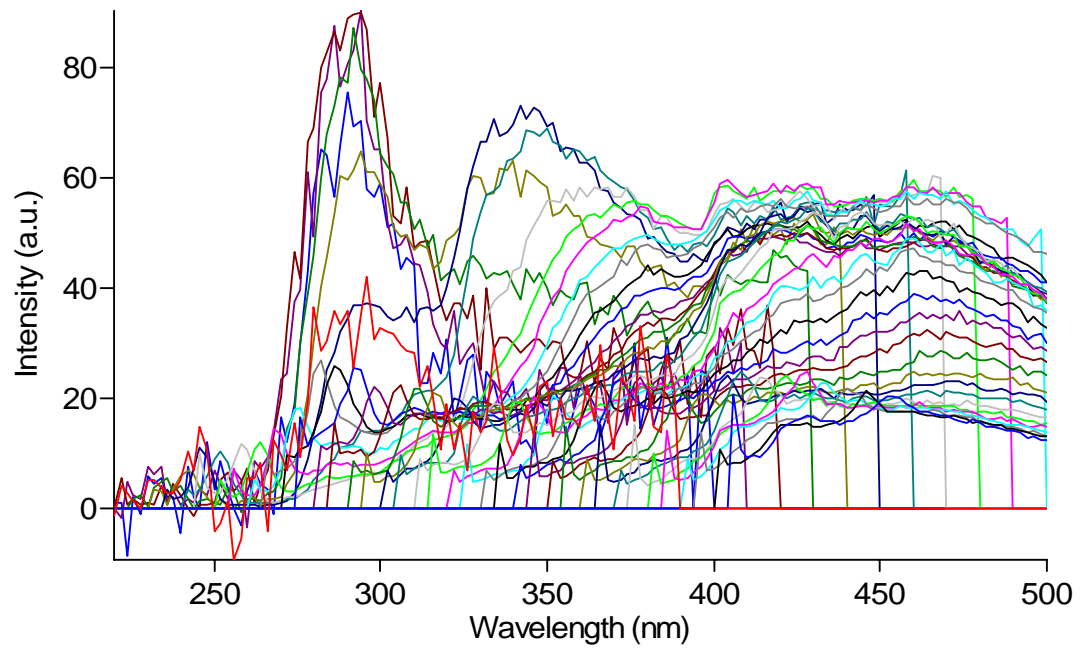
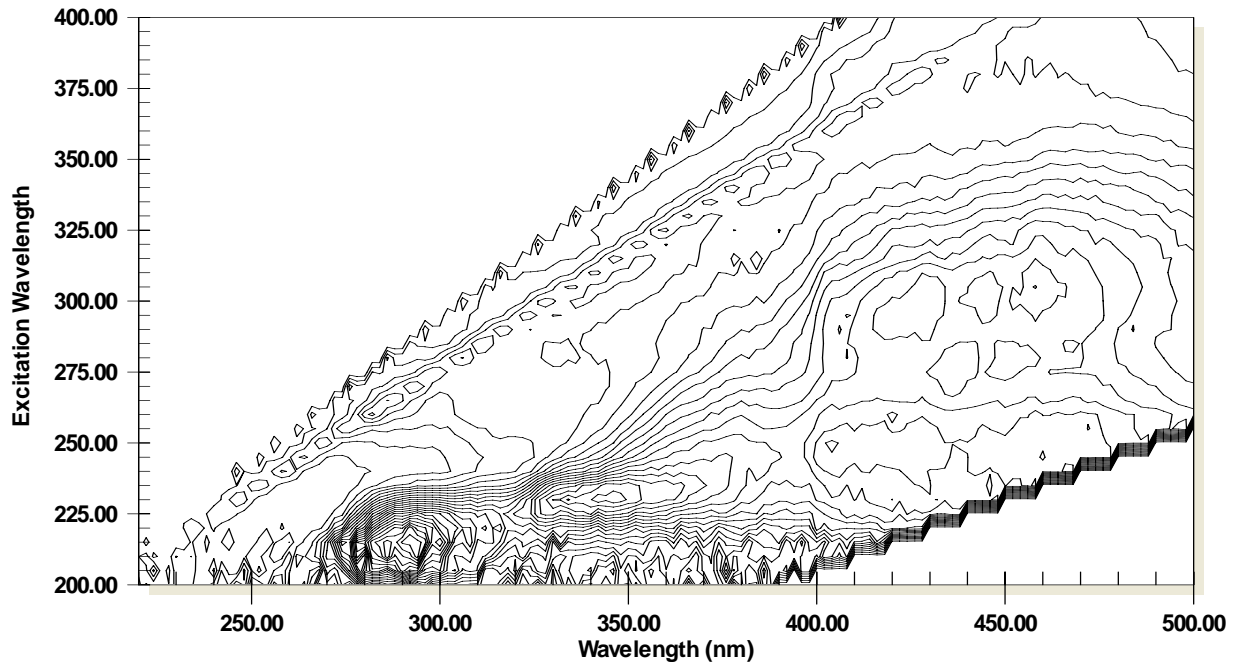
92AT



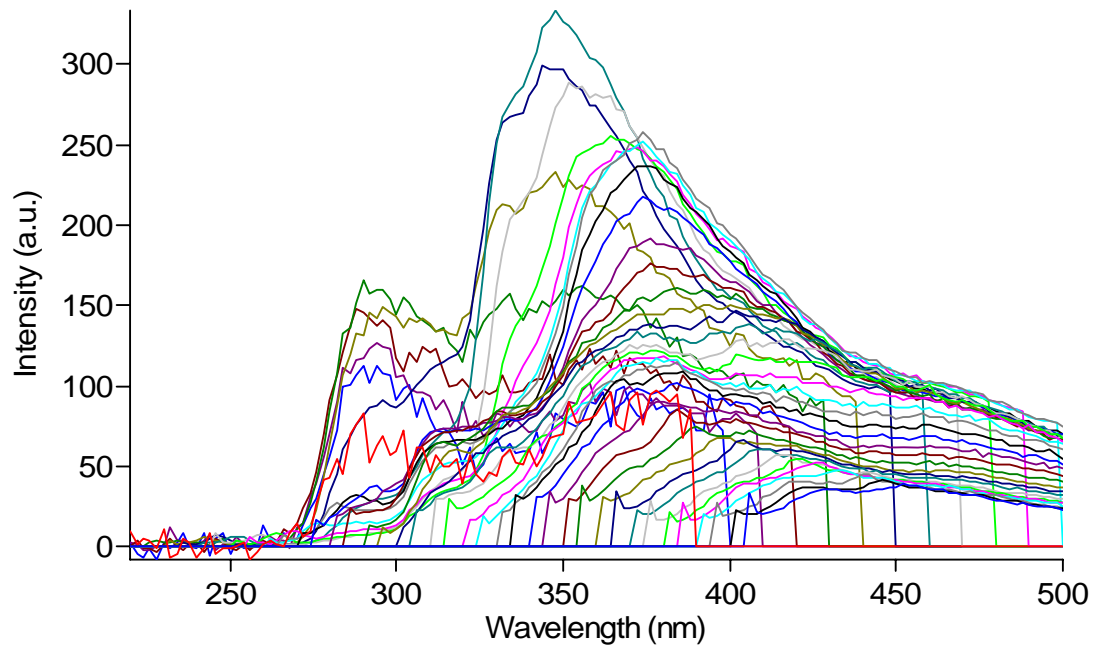
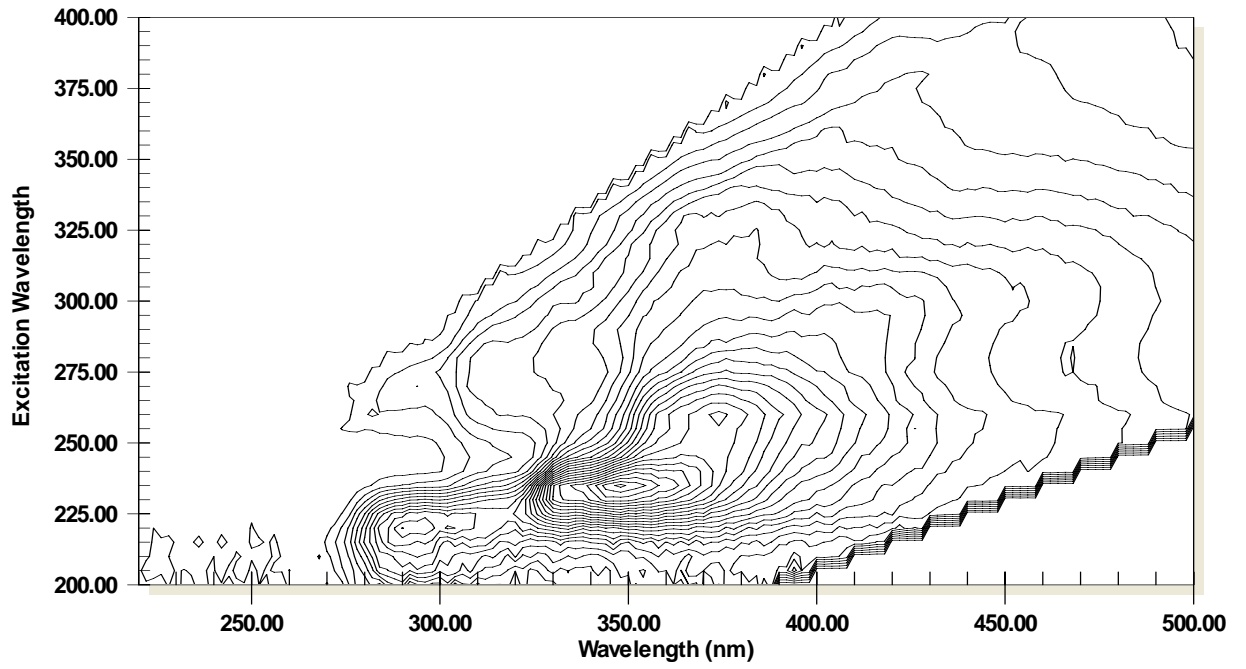
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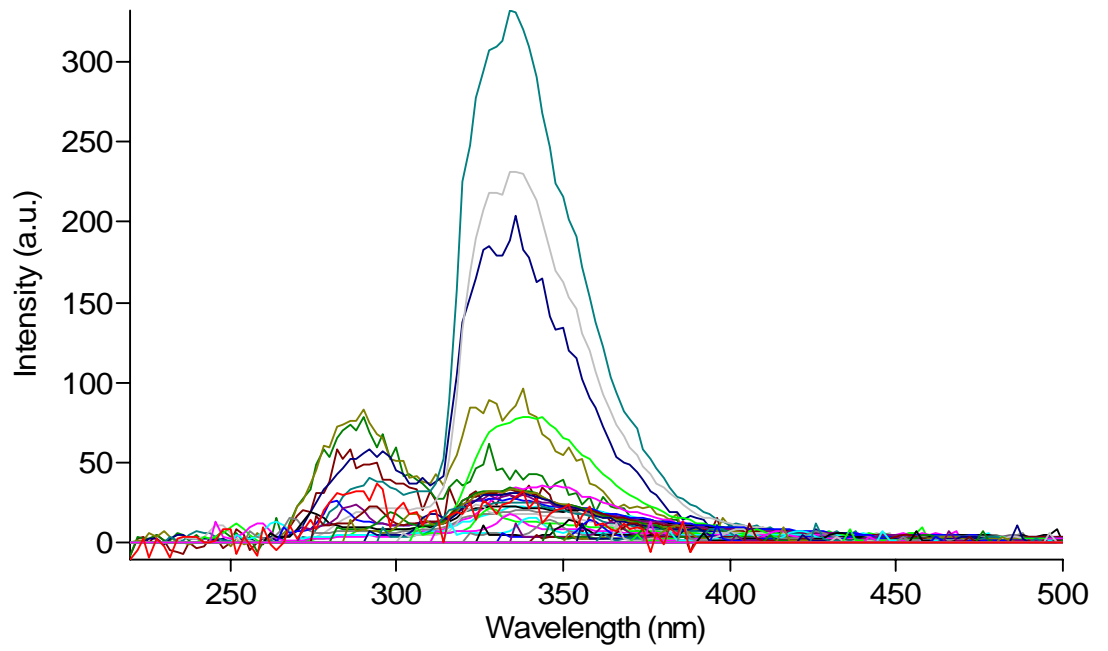
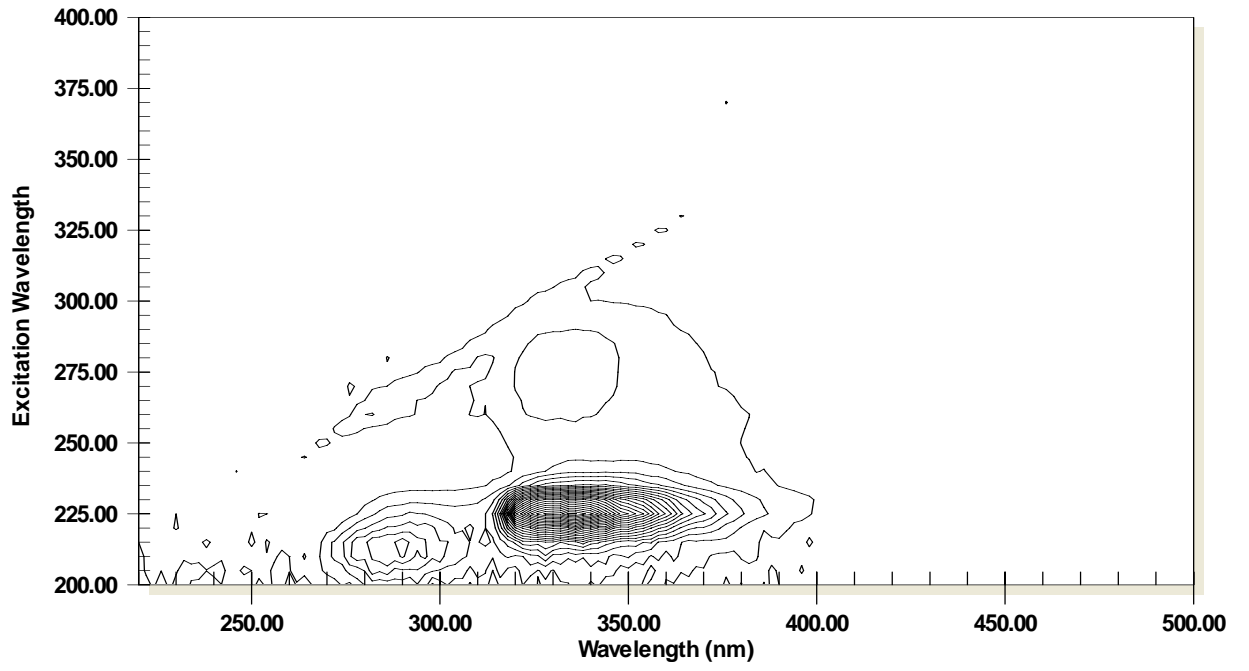
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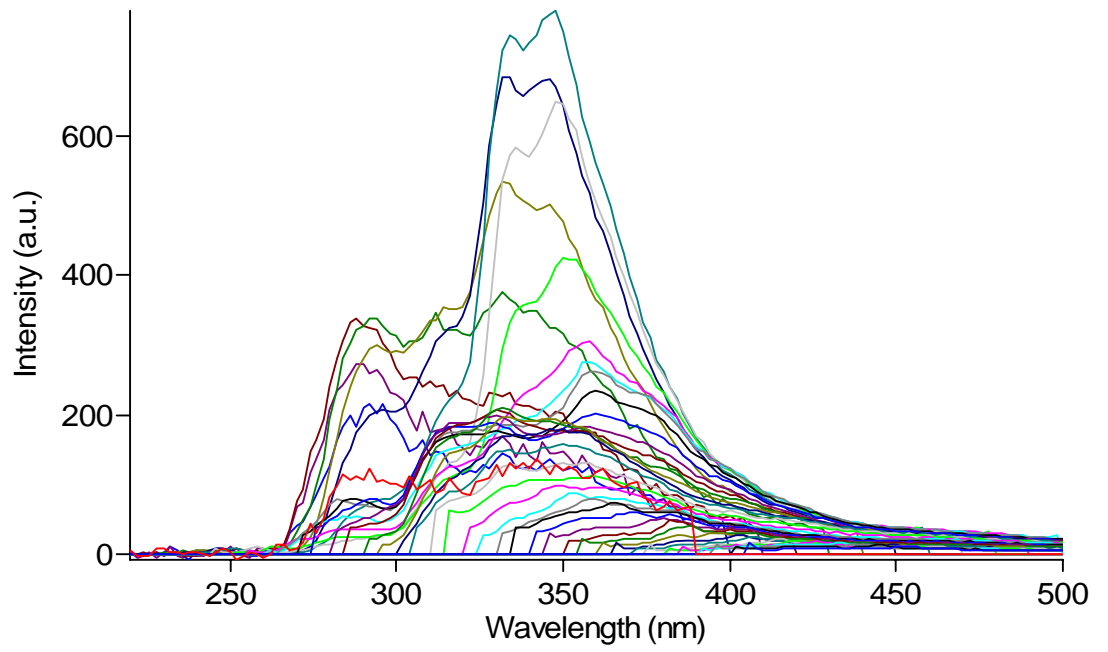
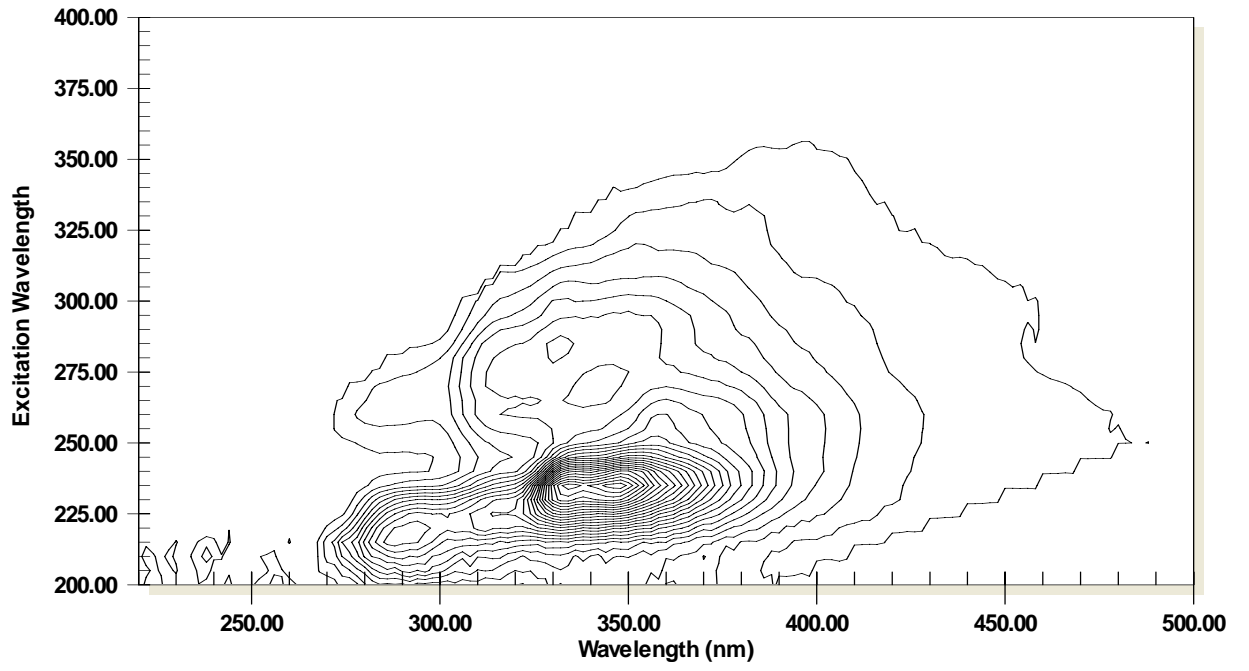
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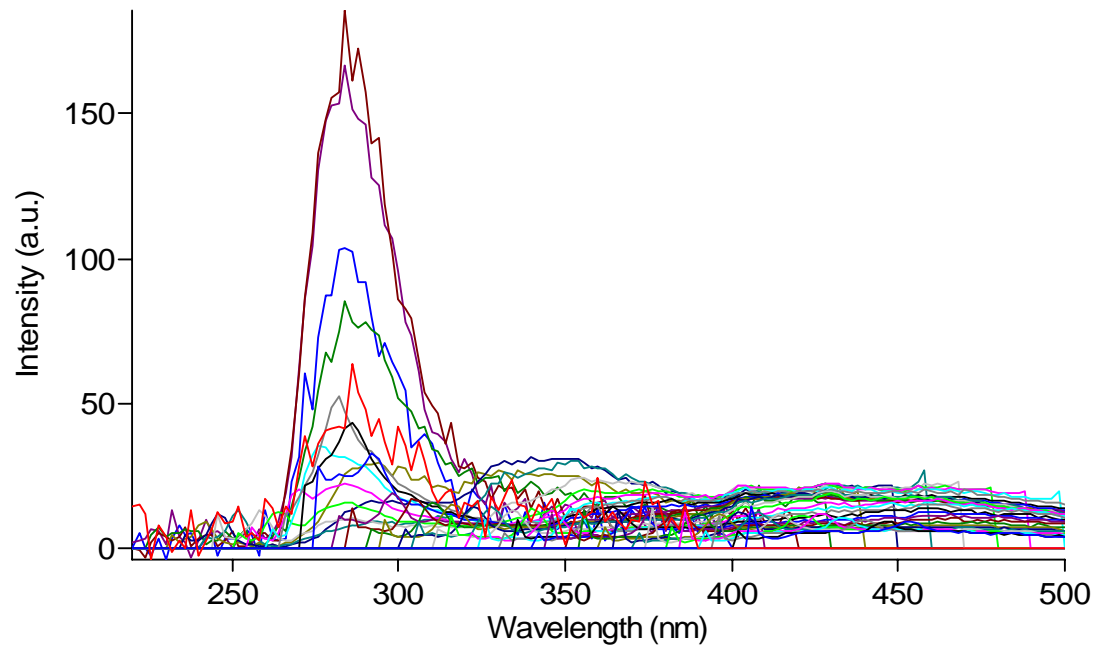
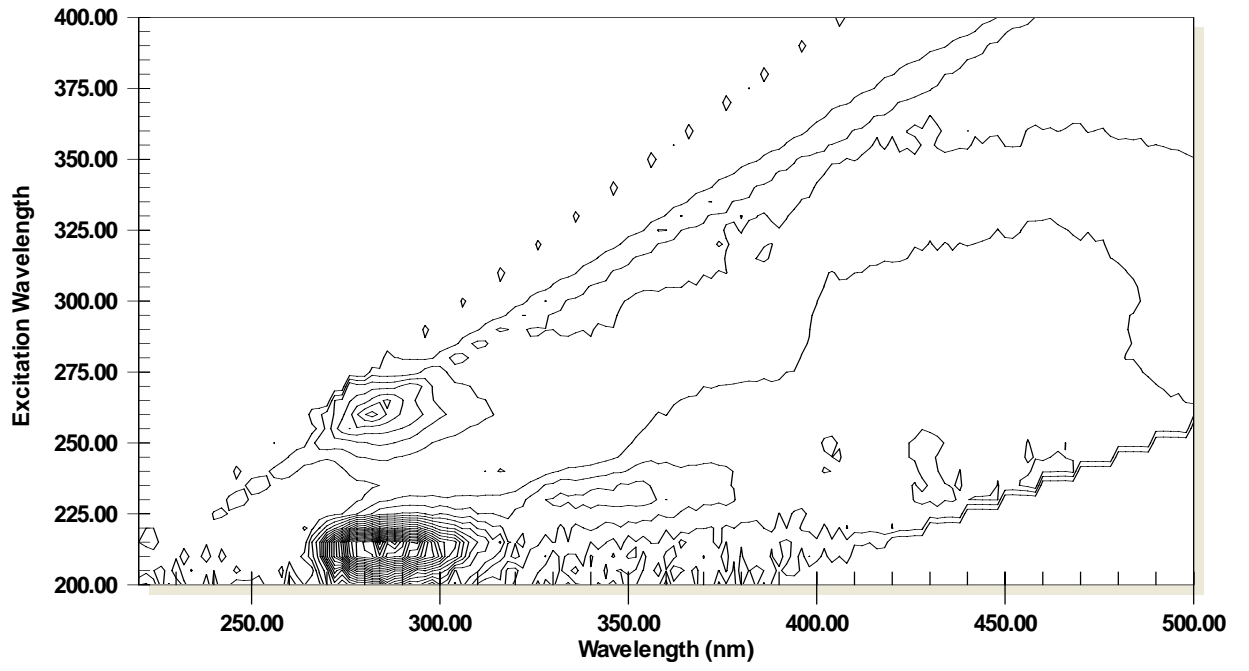
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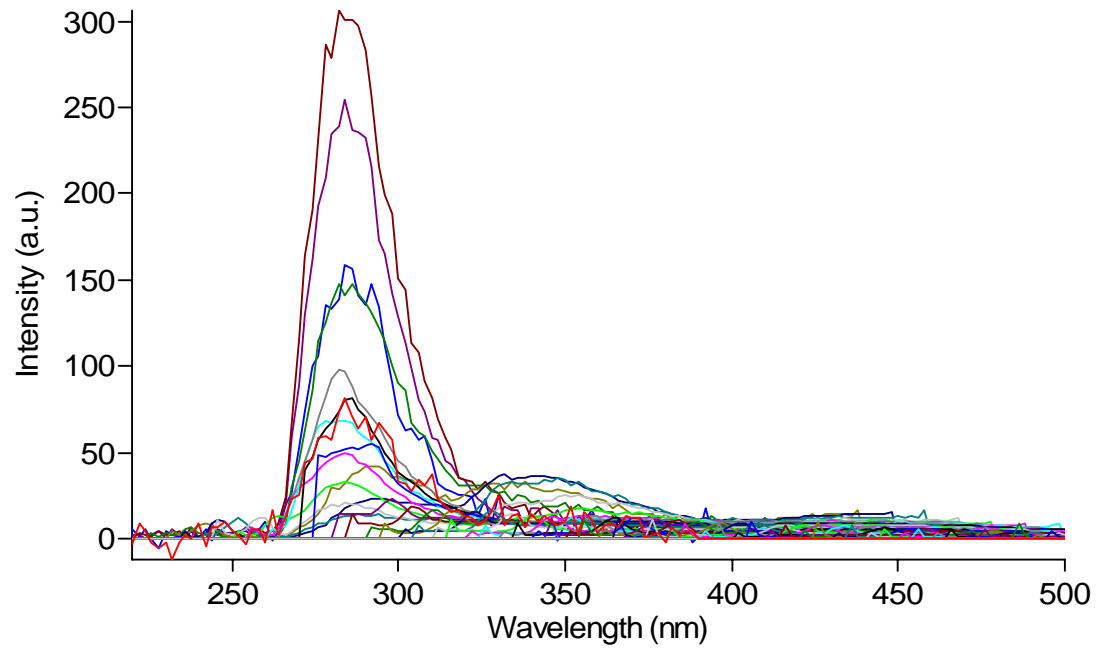
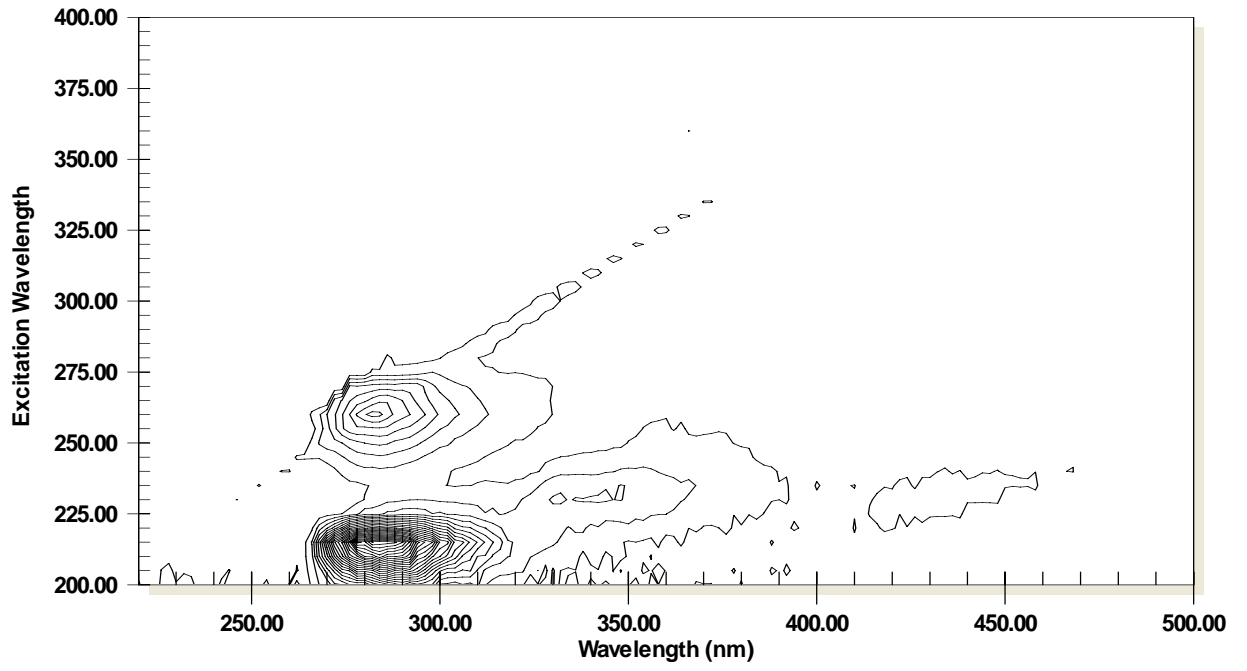
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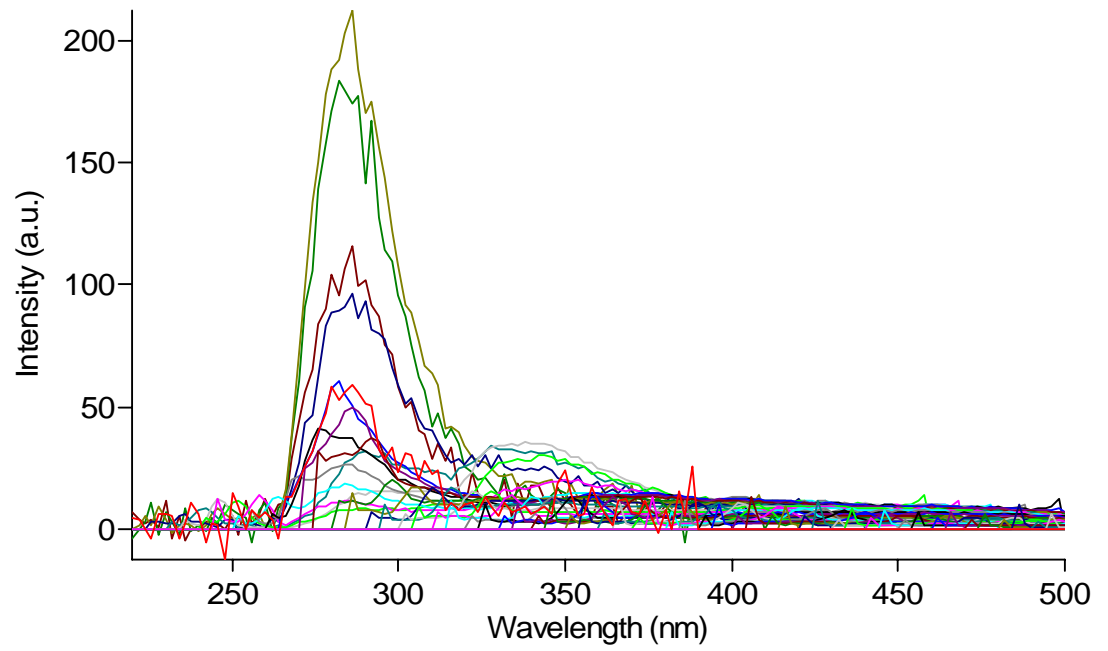
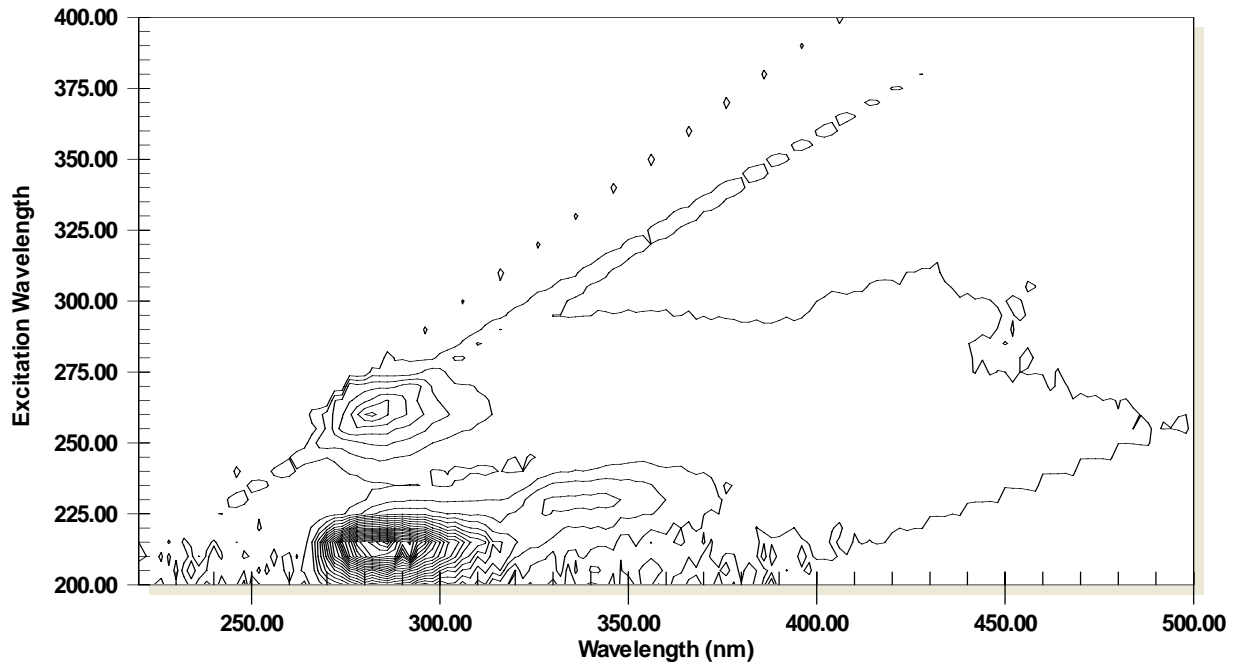
103BT



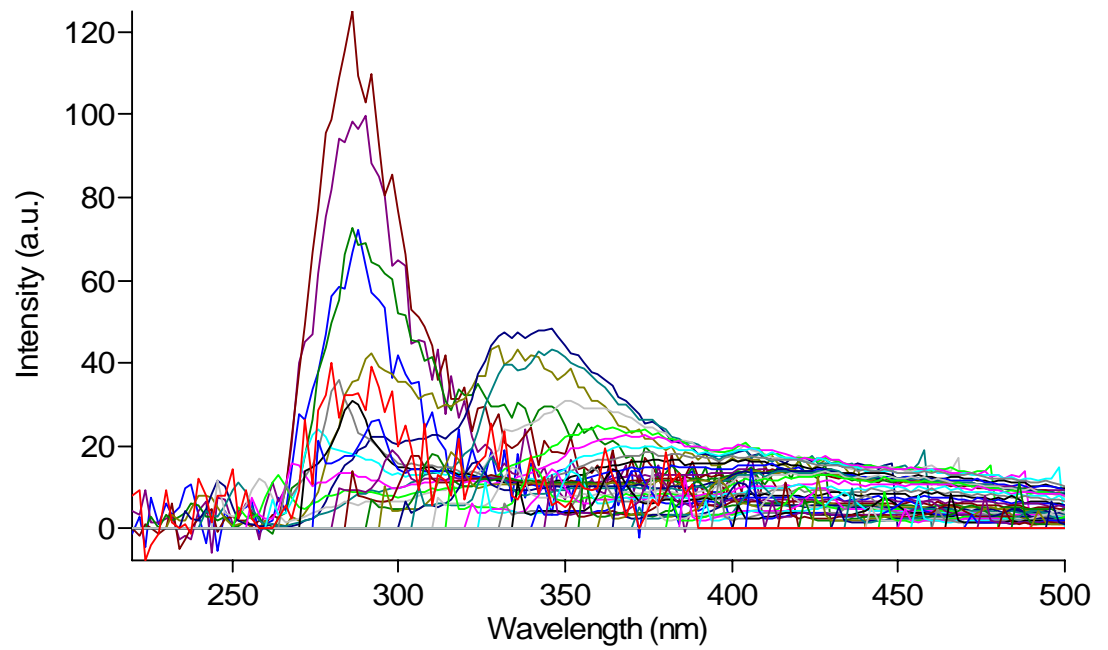
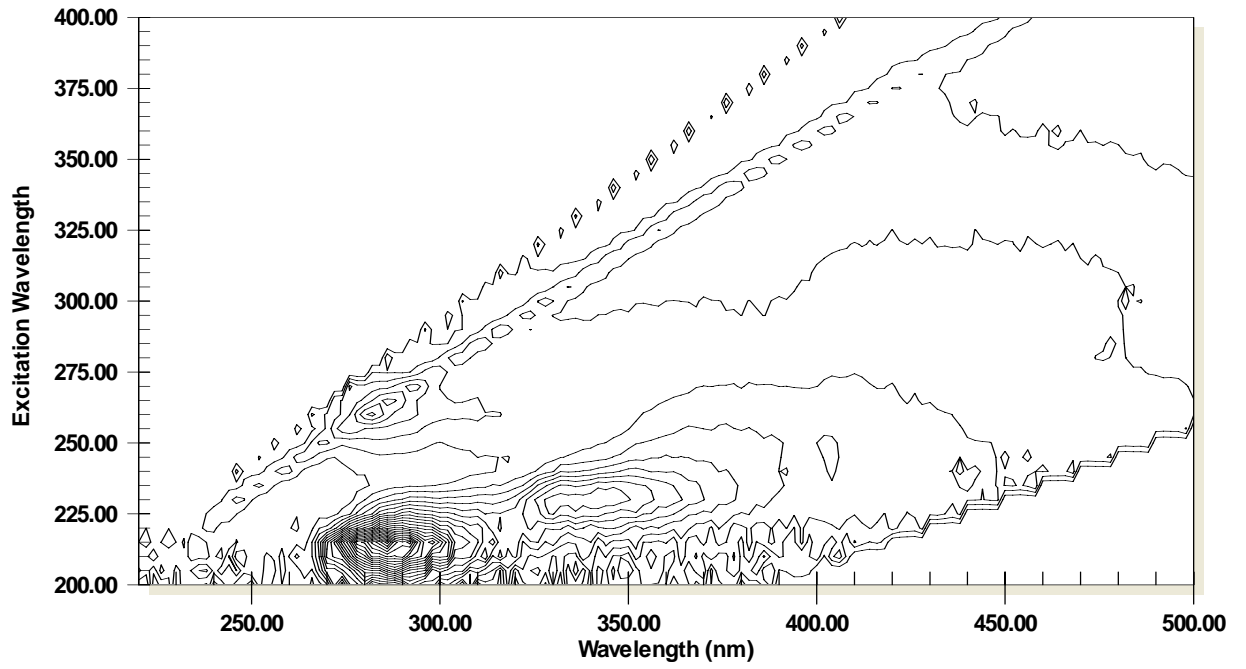
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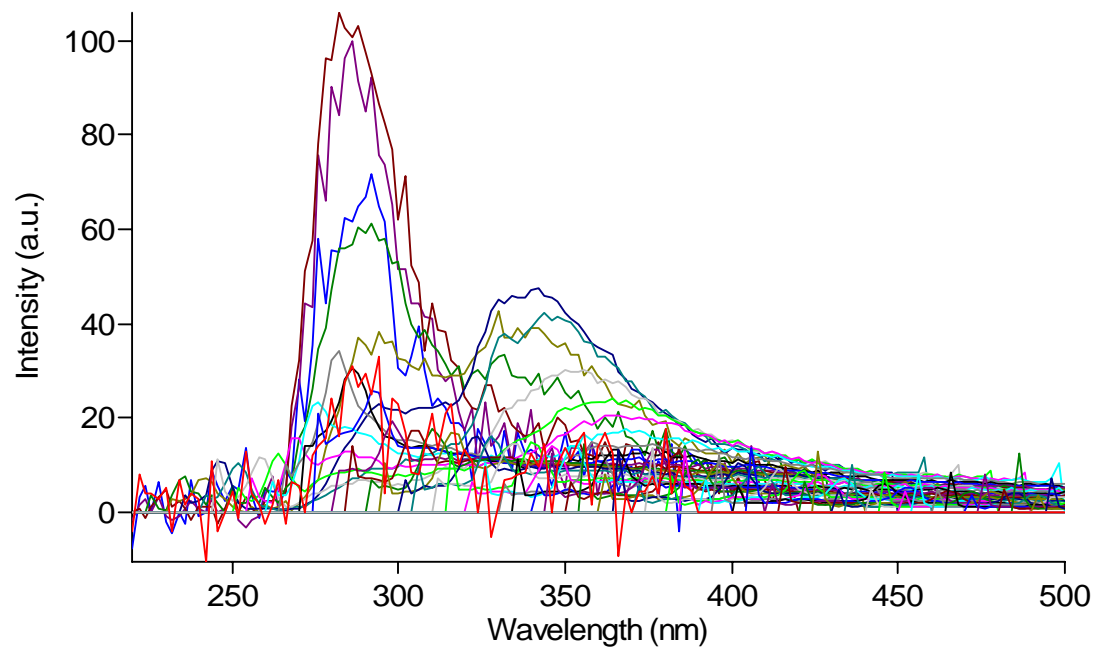
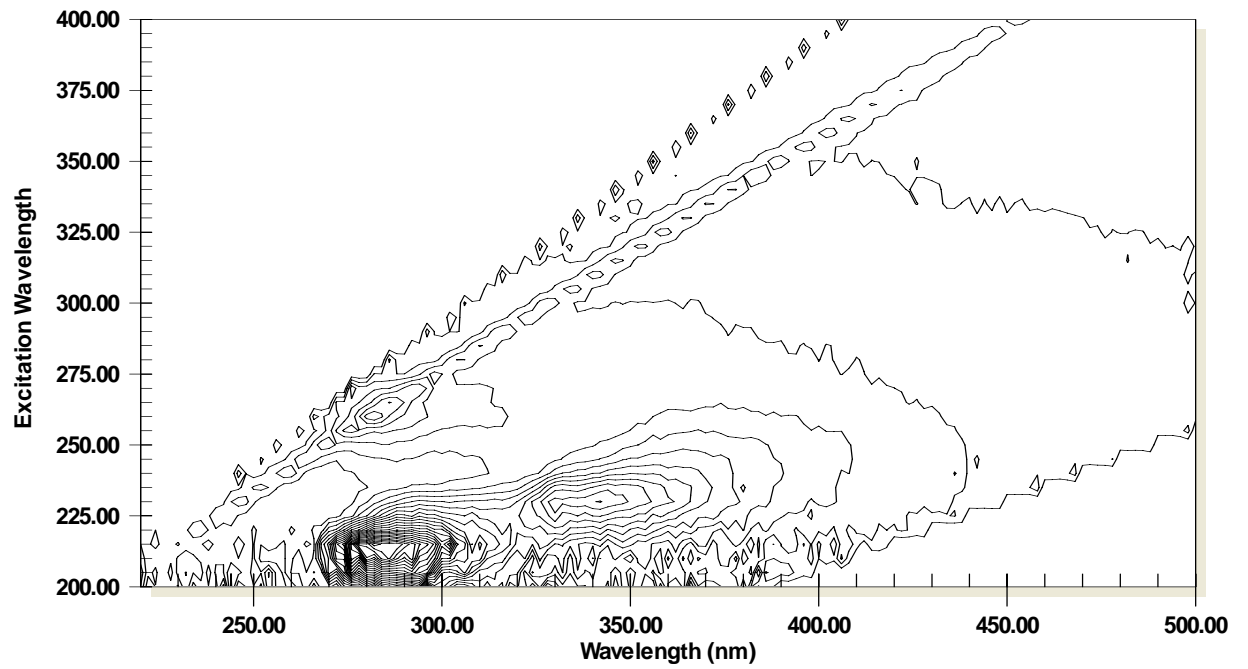
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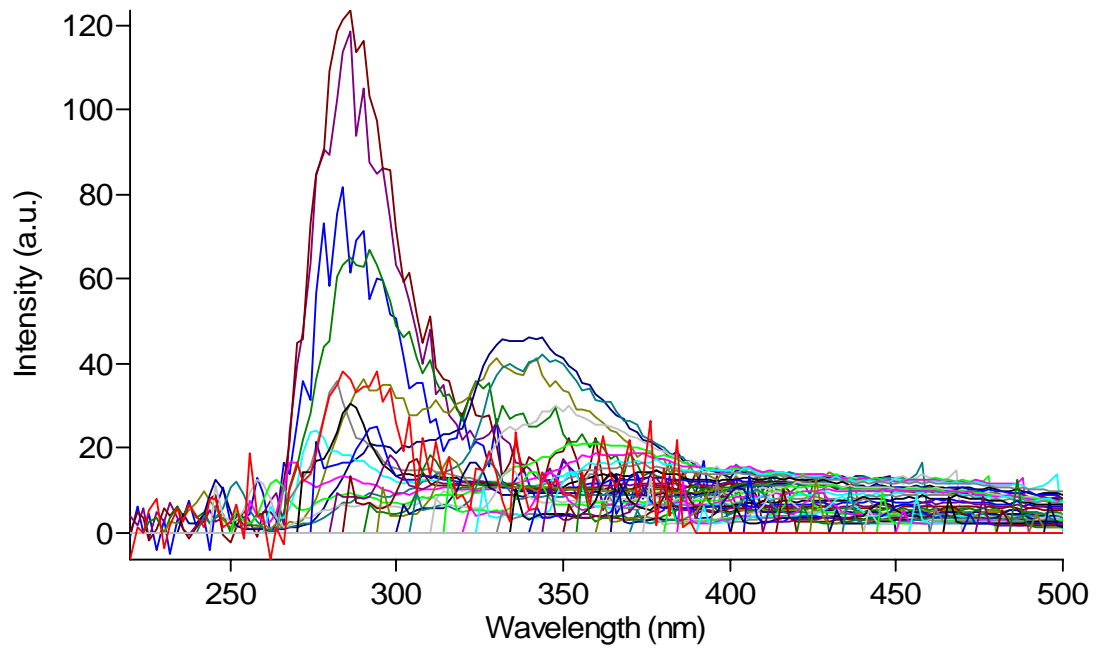
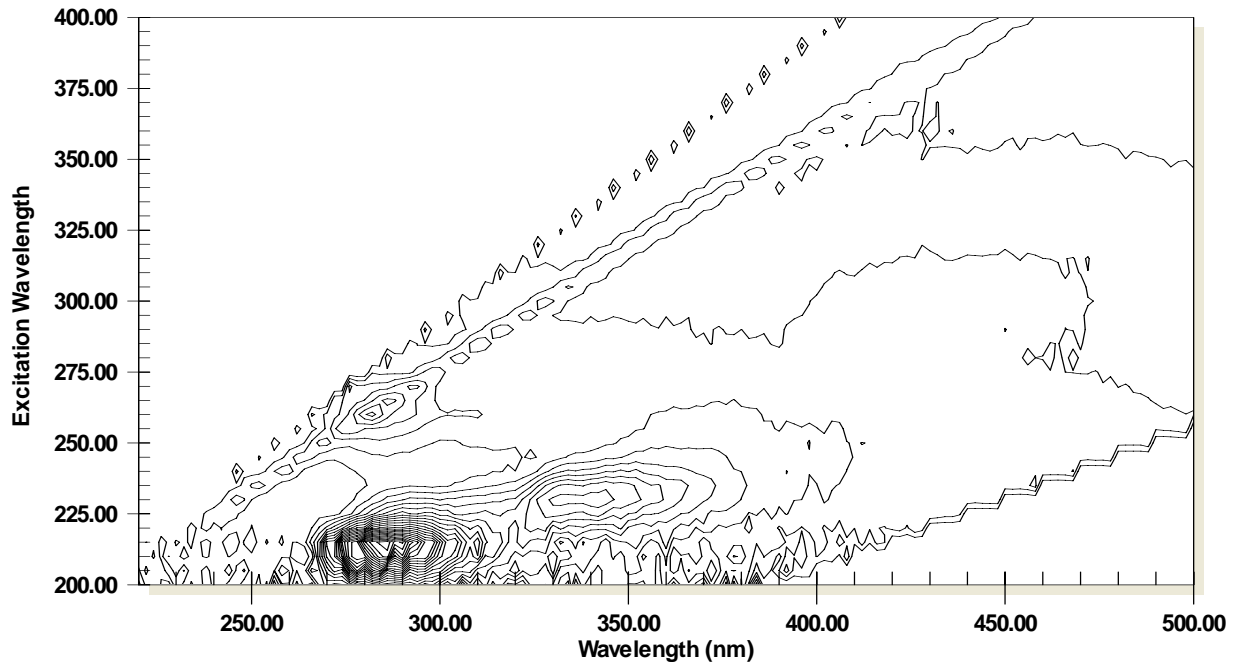
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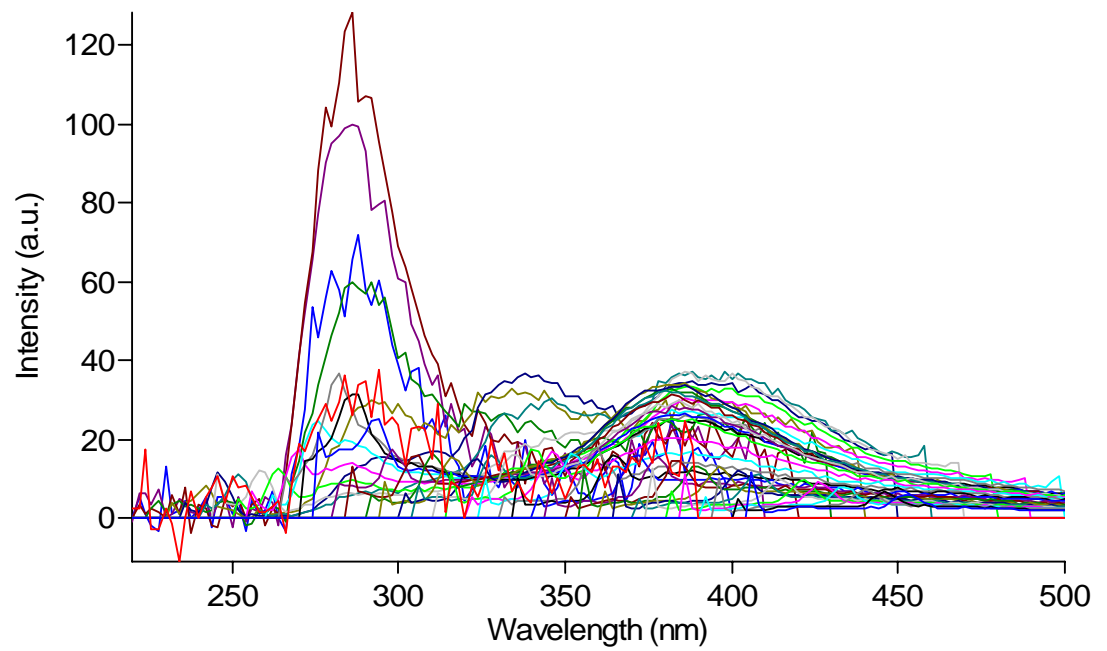
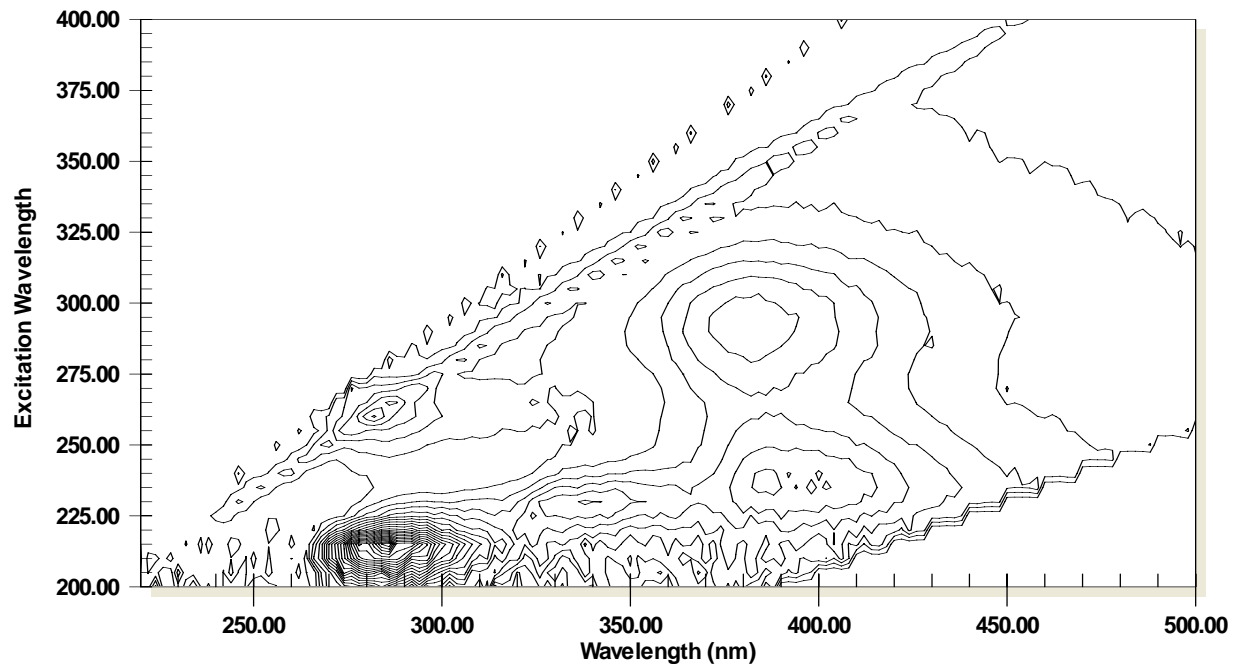
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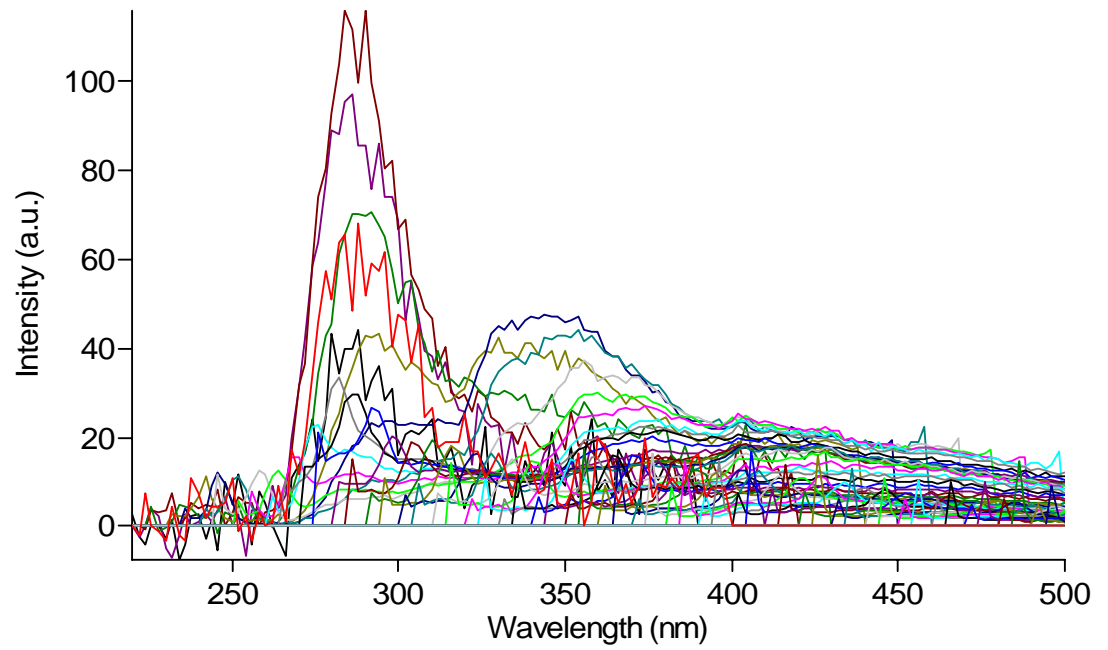
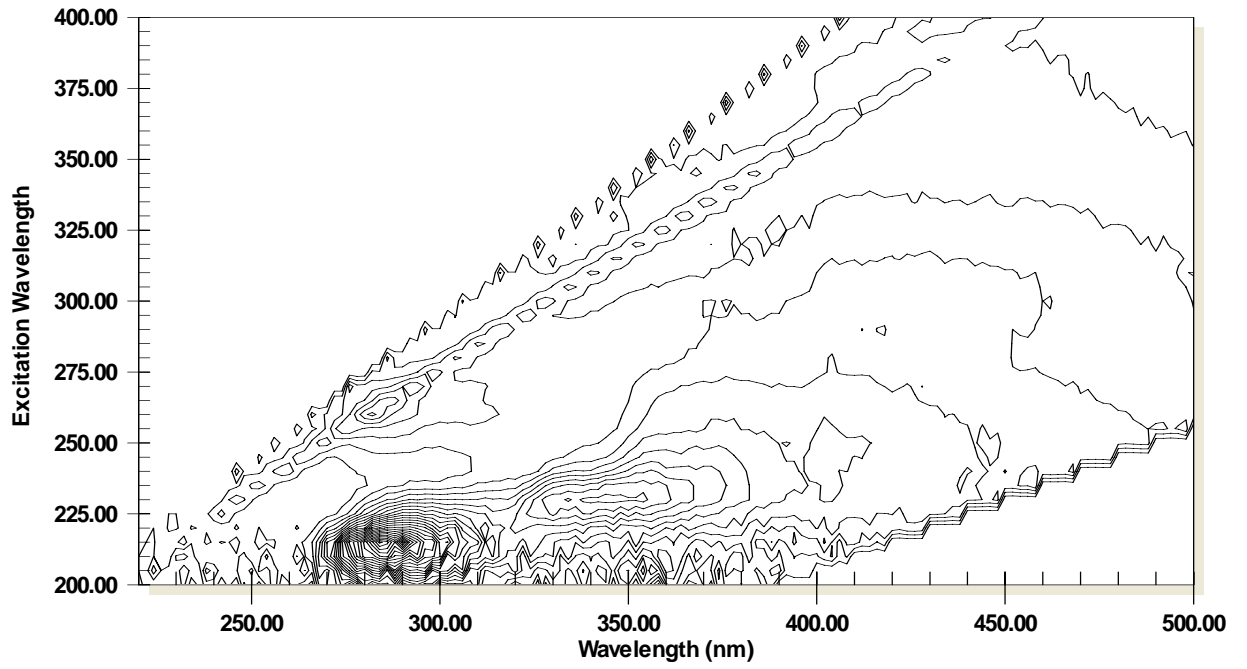
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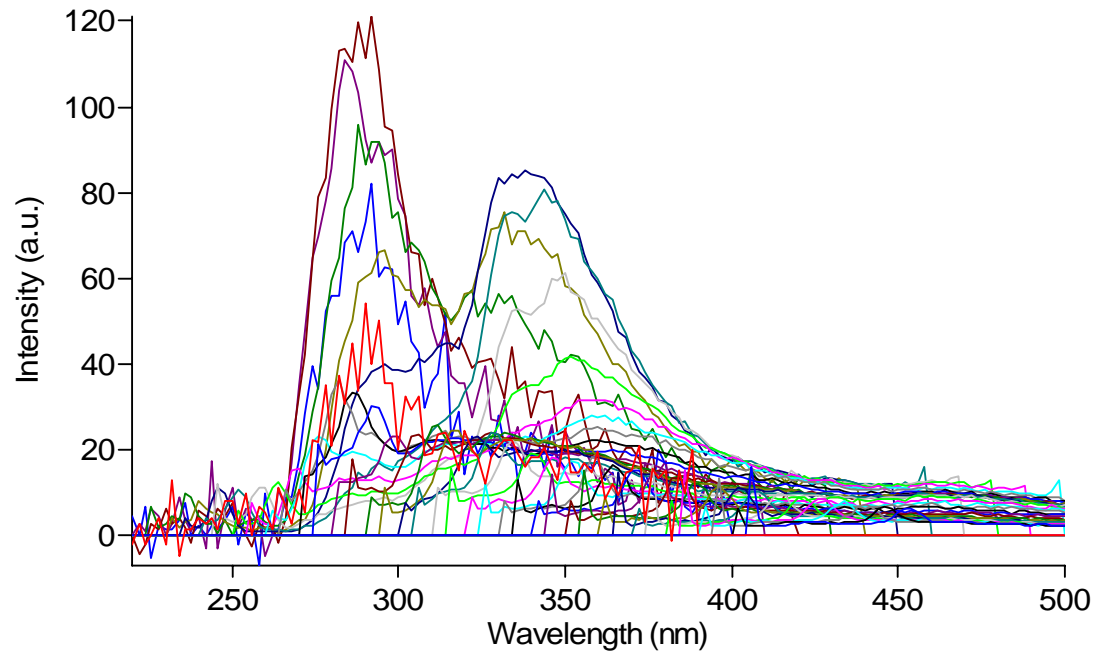
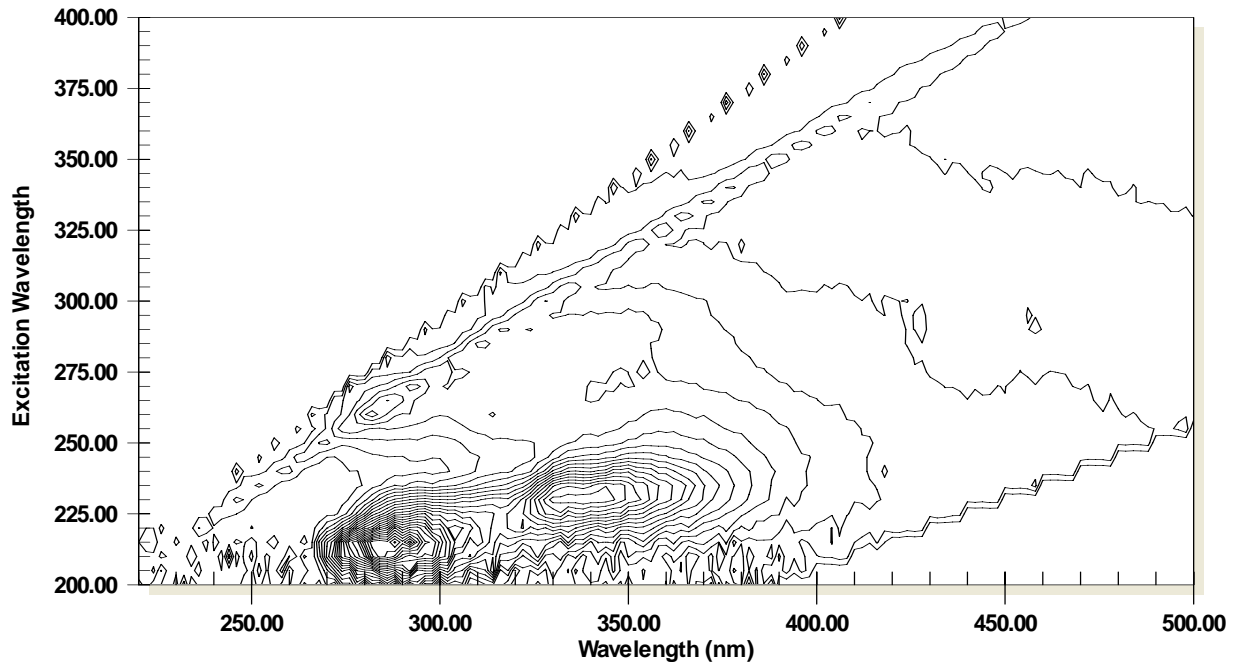
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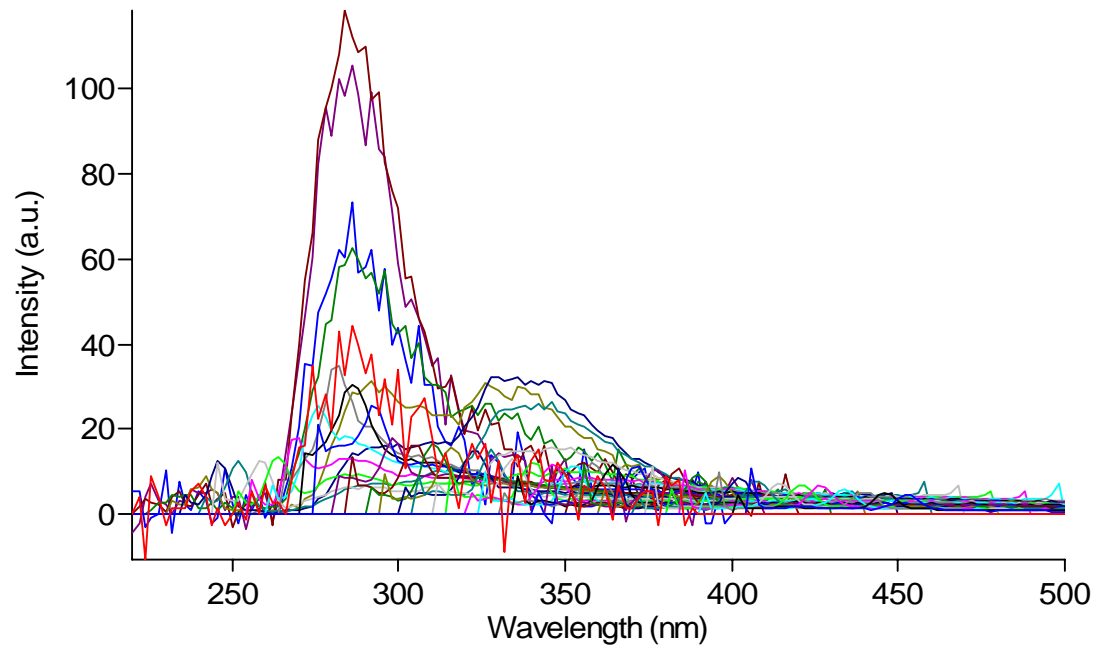
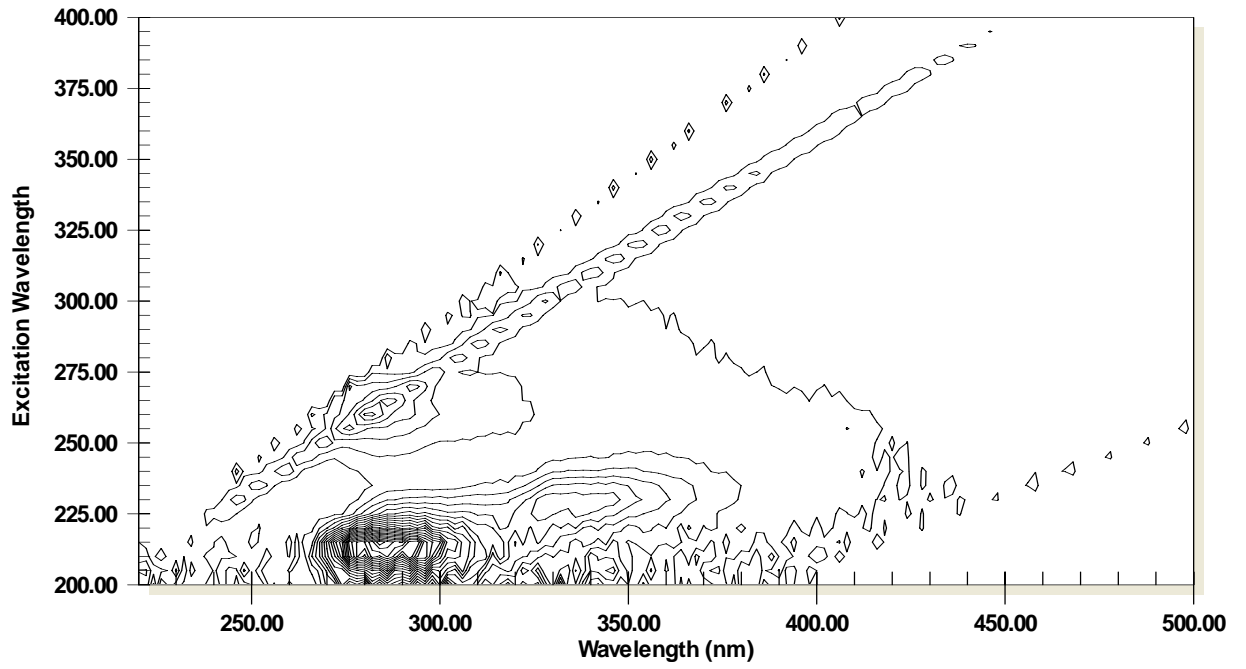
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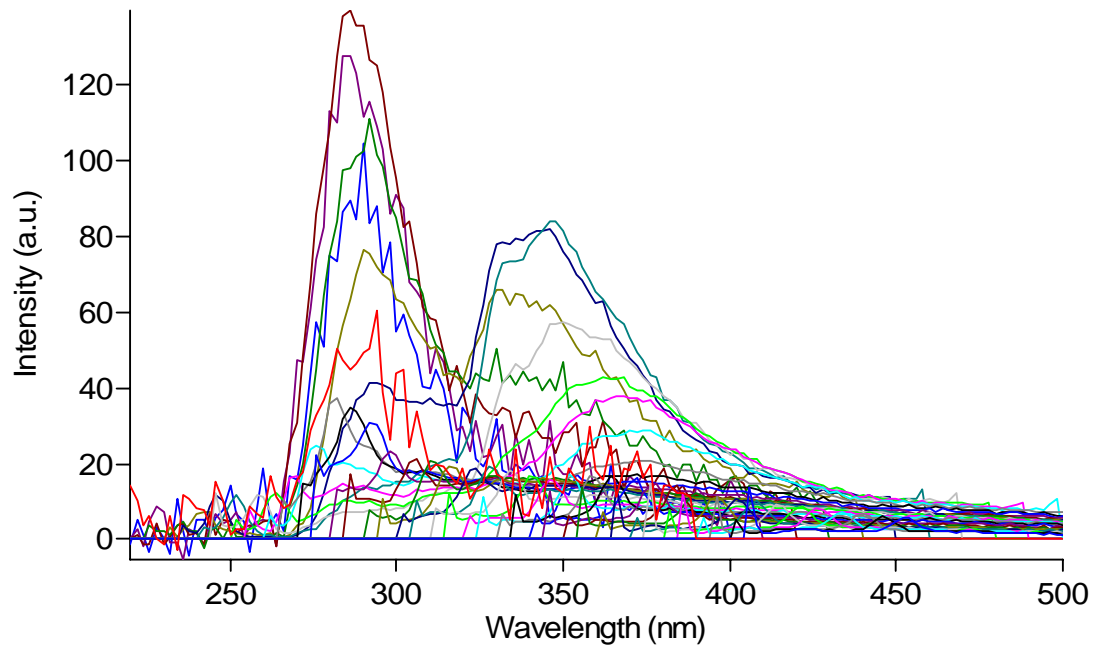
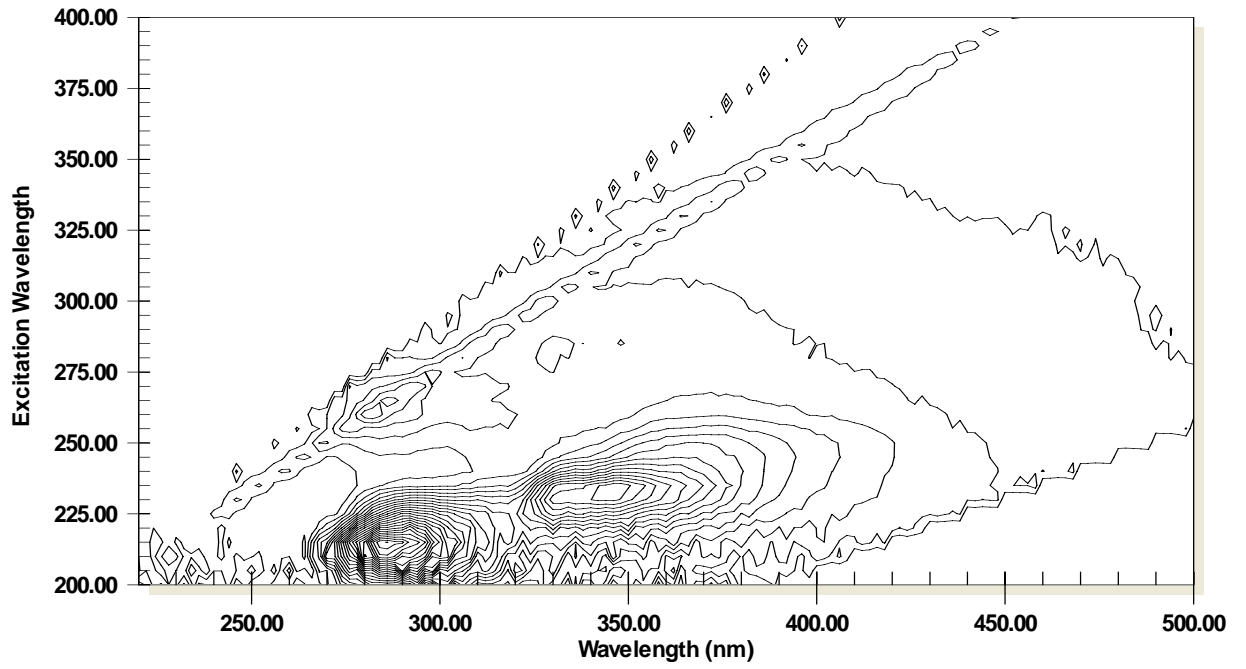
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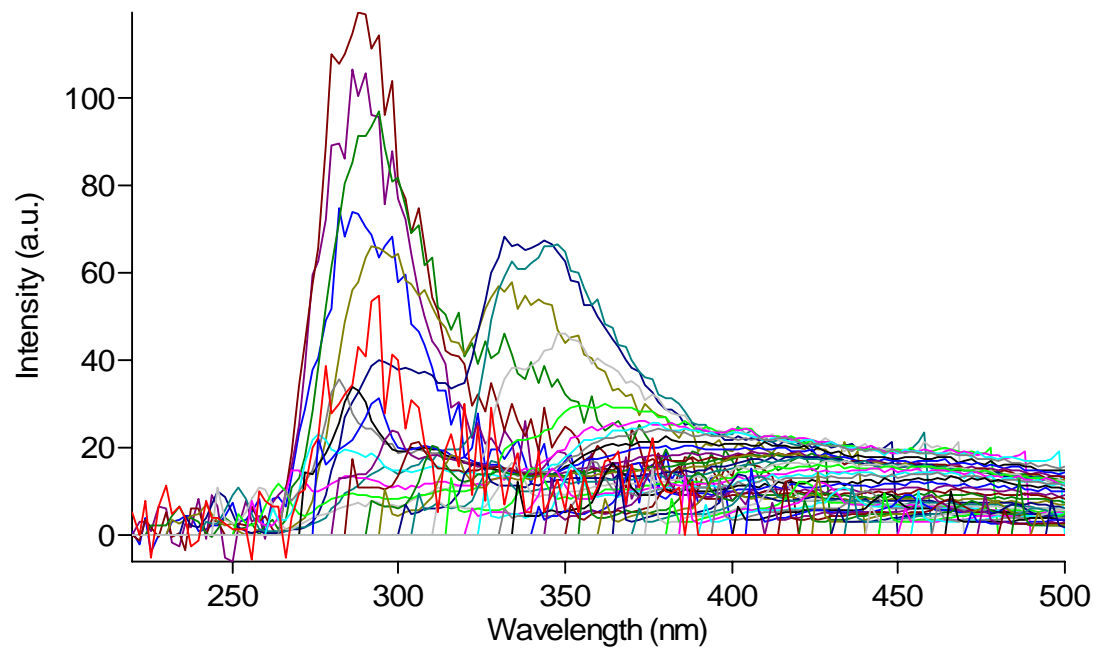
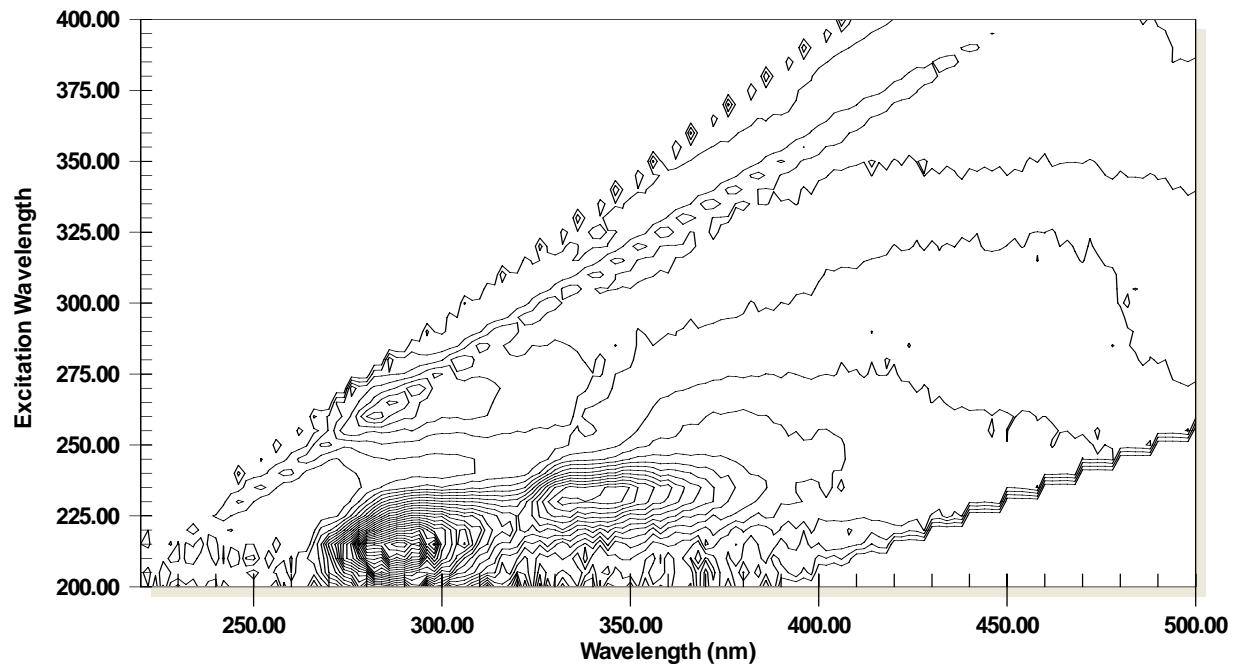
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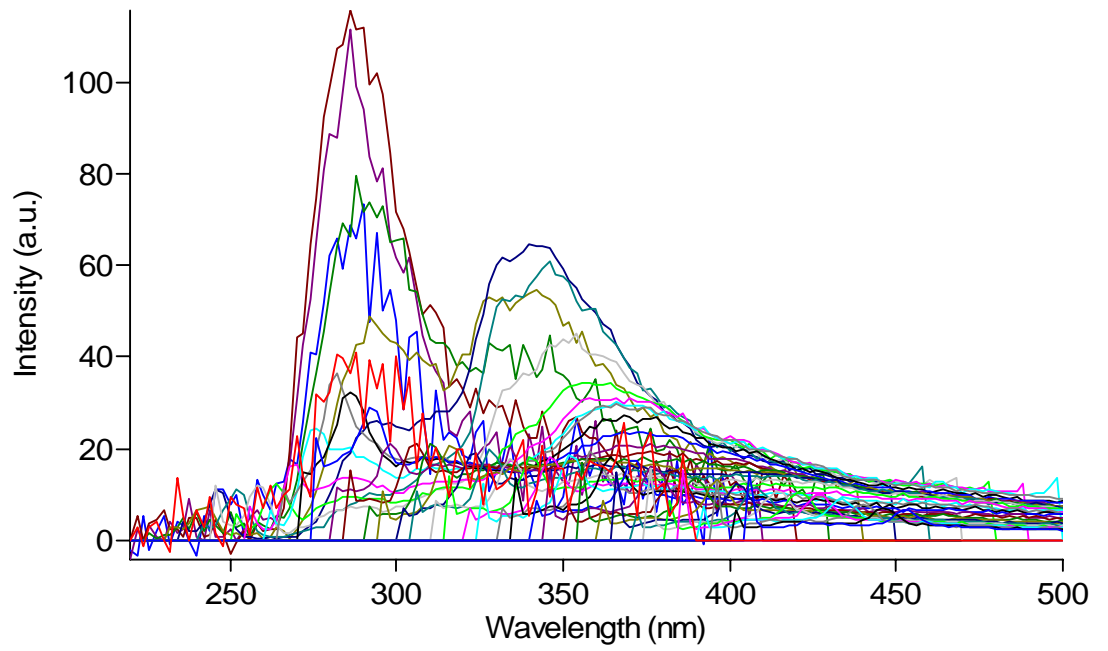
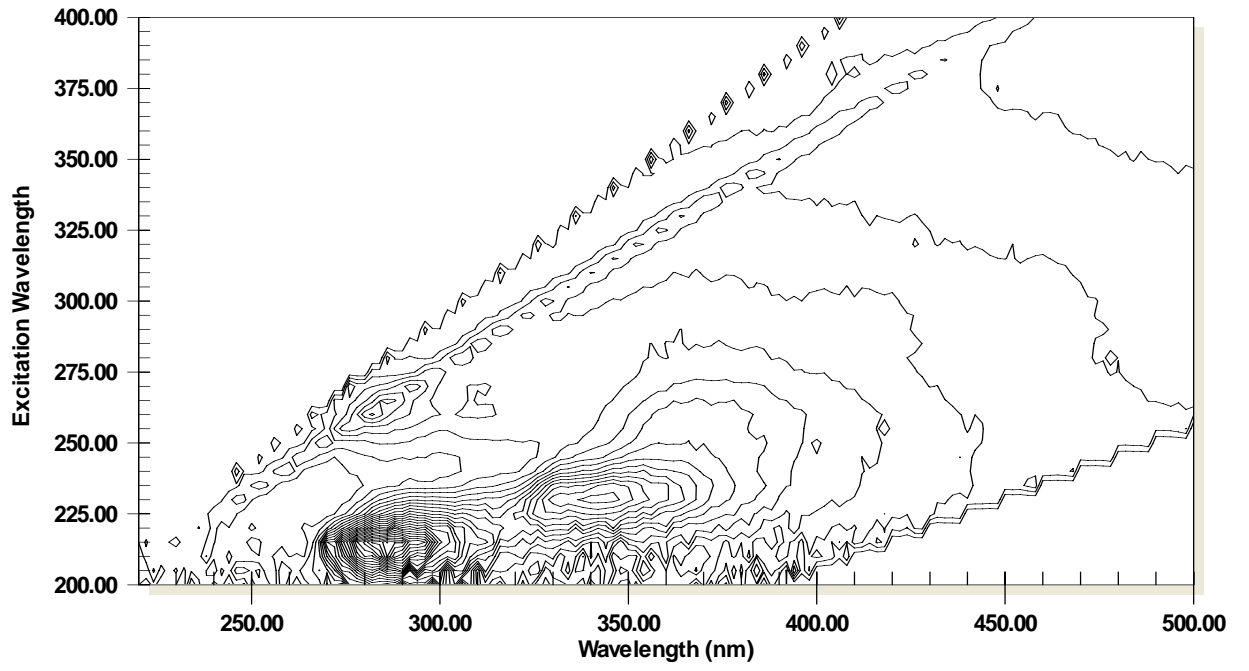
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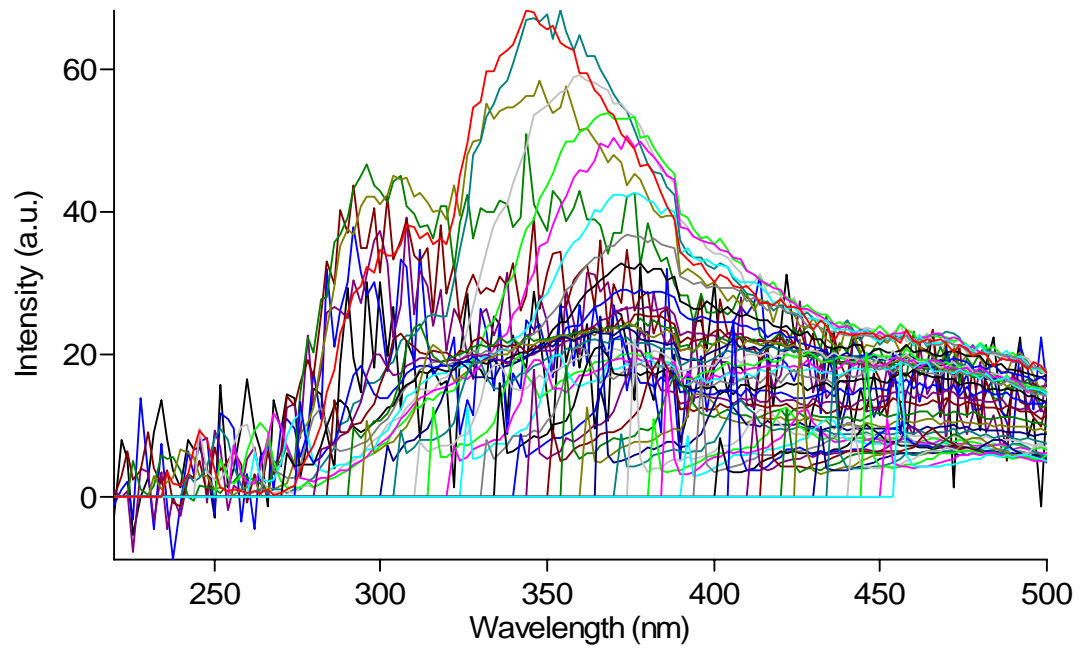
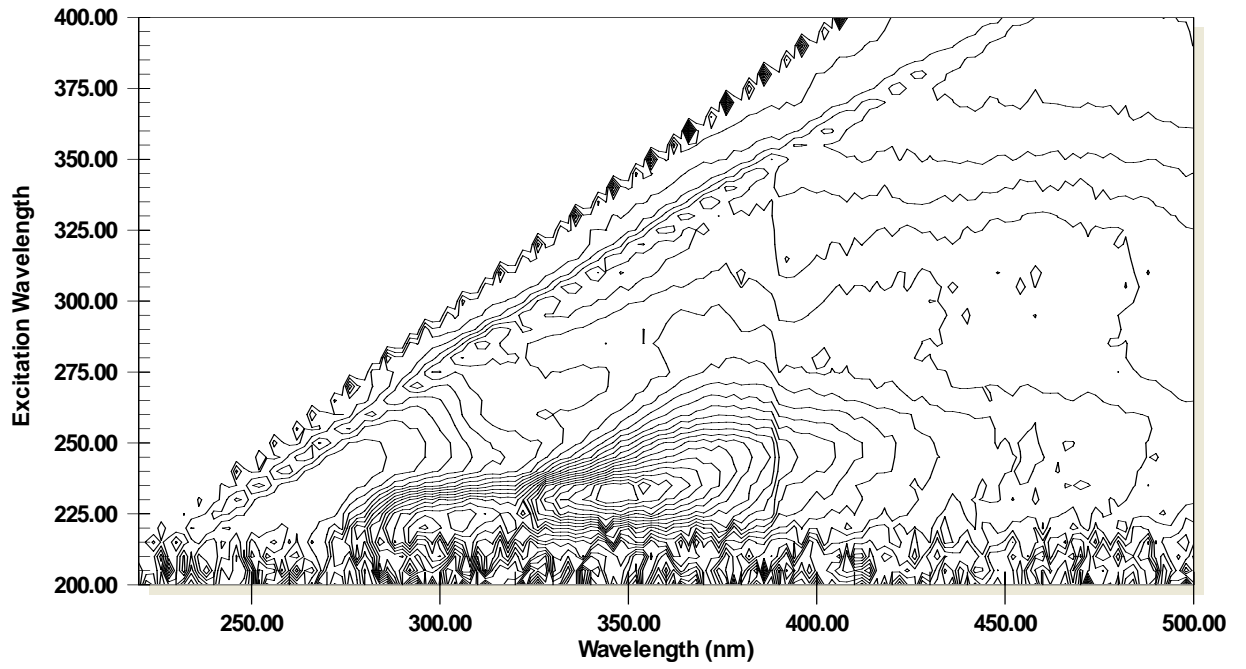
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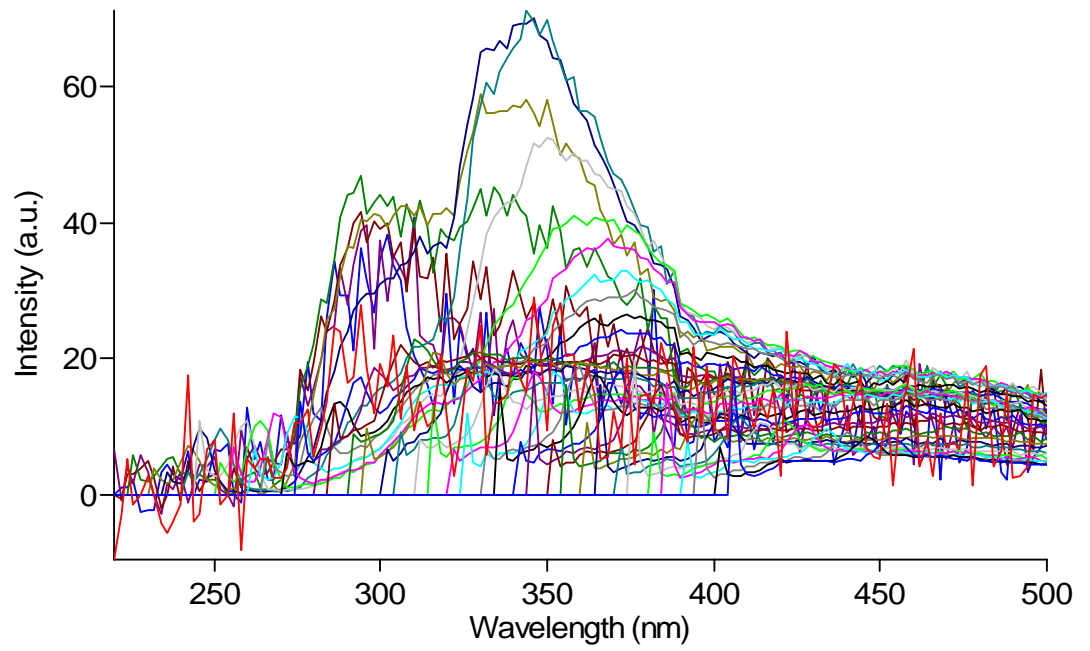
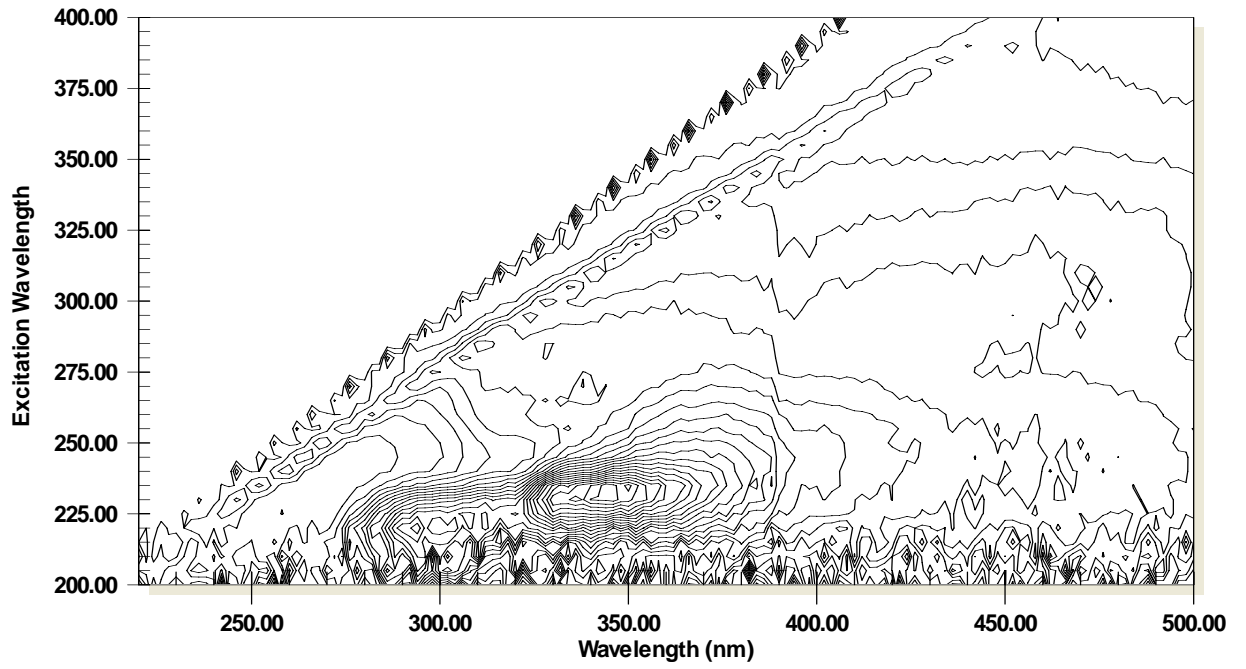
M46T



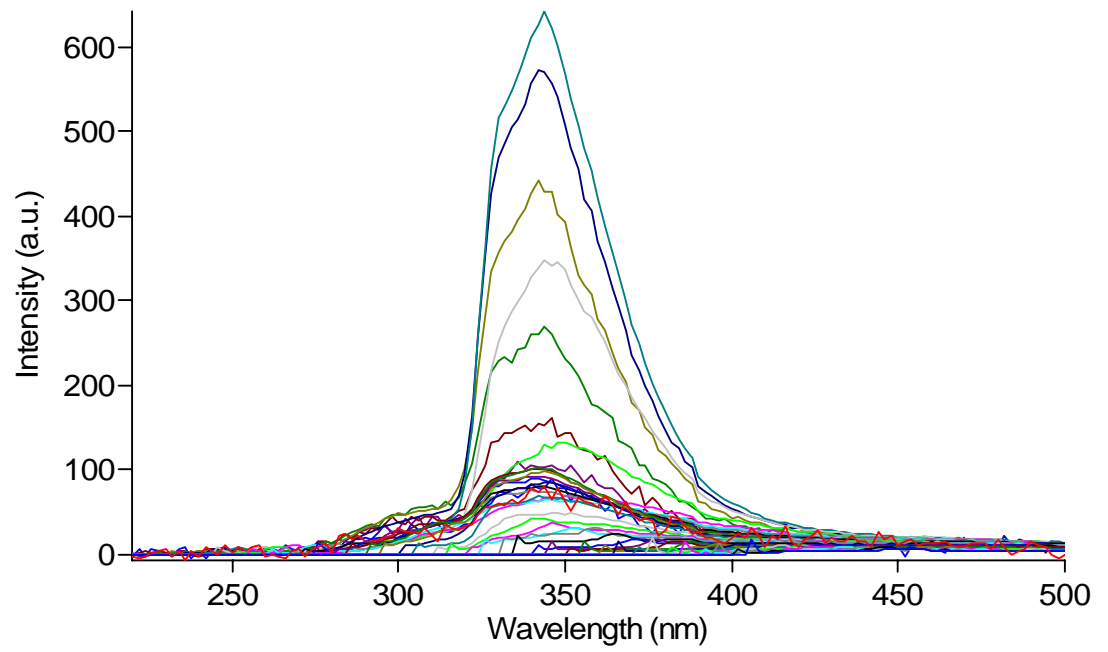
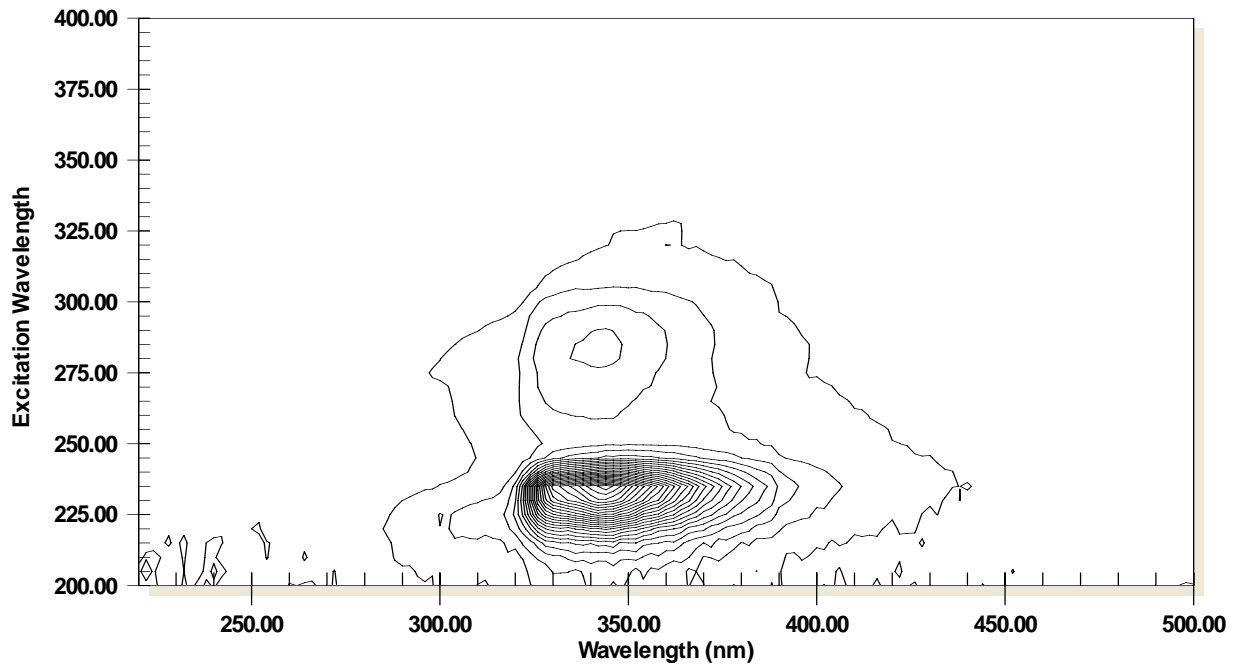
Pyridine Contour Plots and Excitation Scan Overlays
11Cy



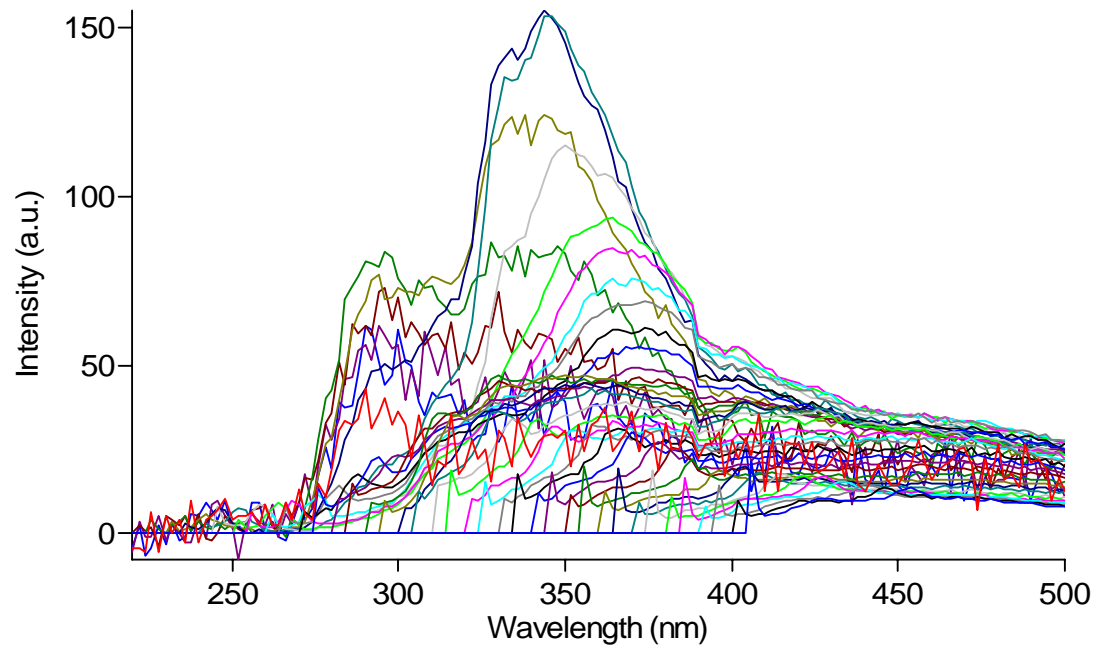
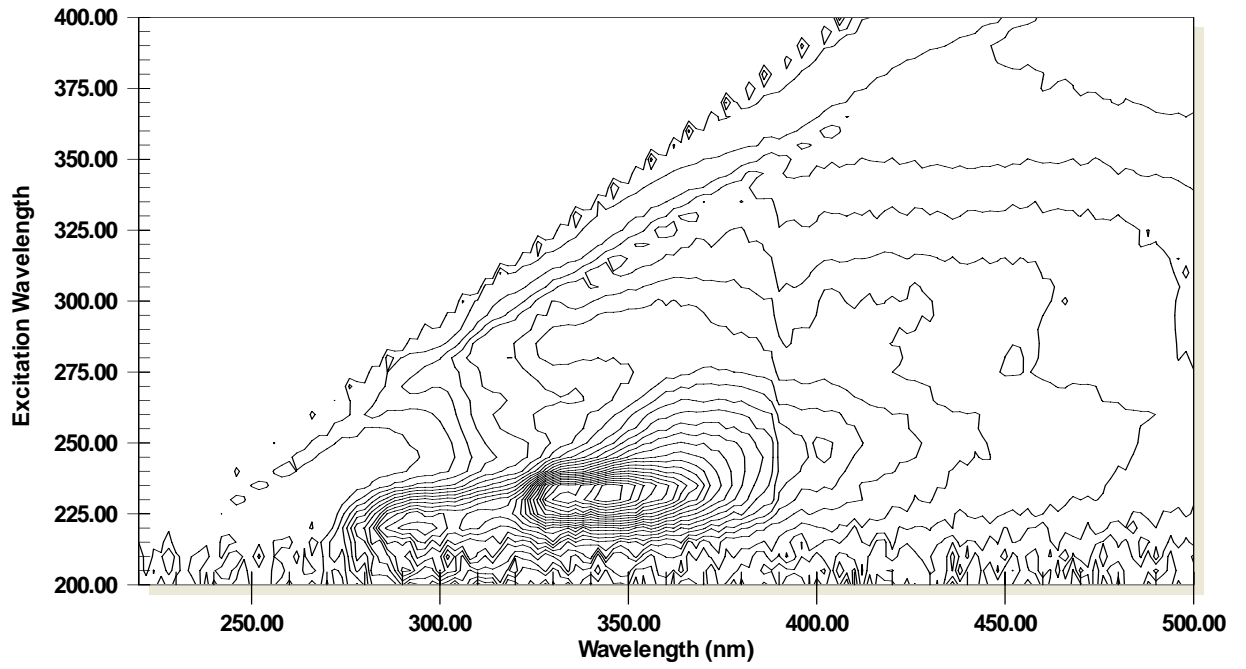
12Cy



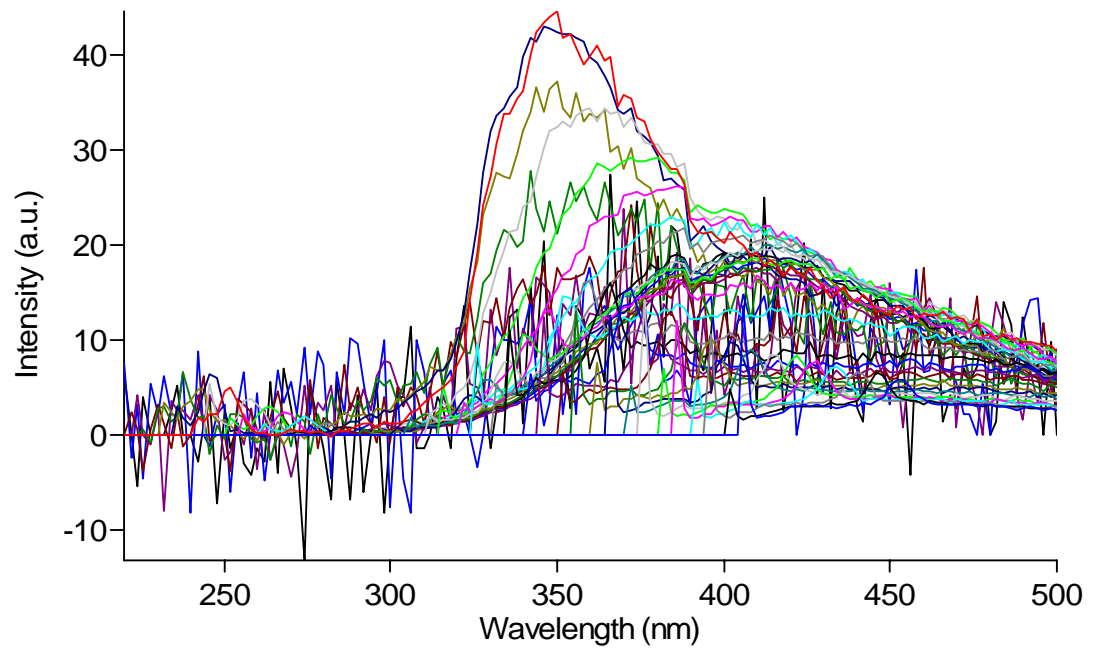
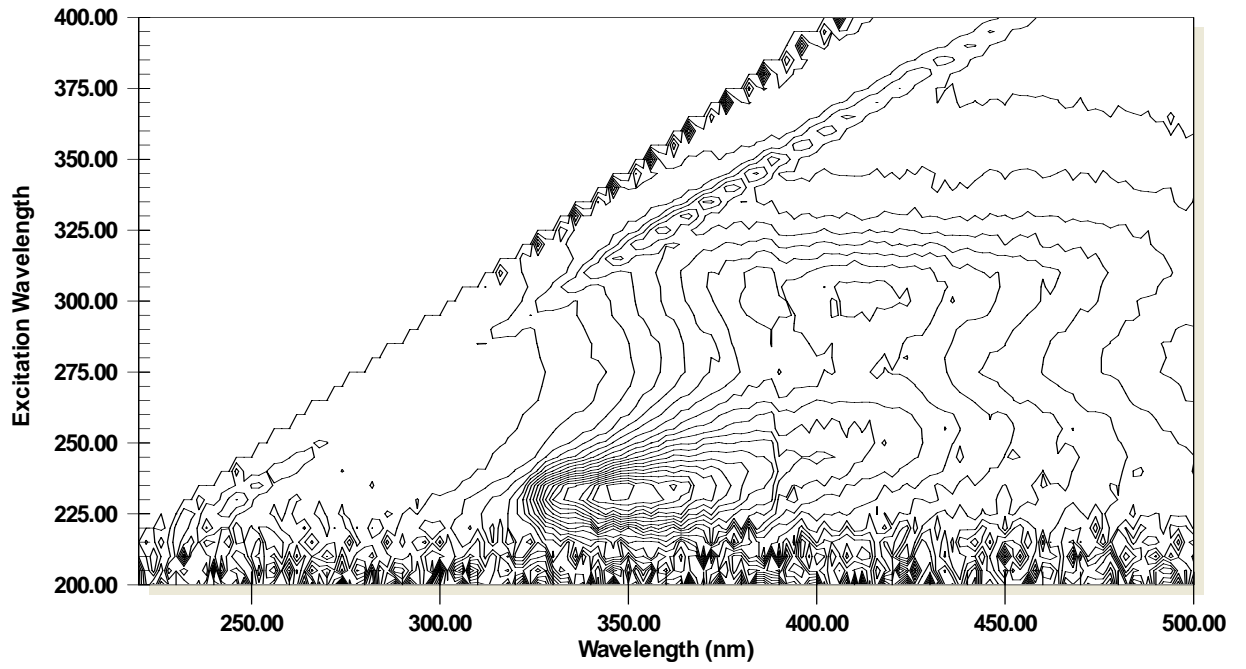
13By



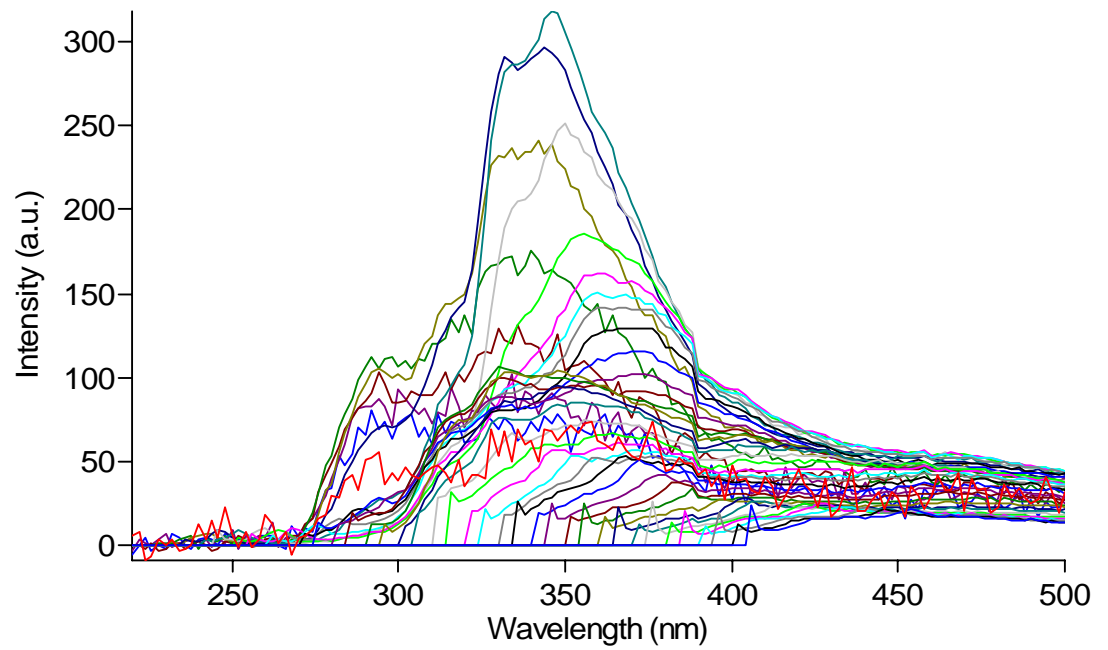
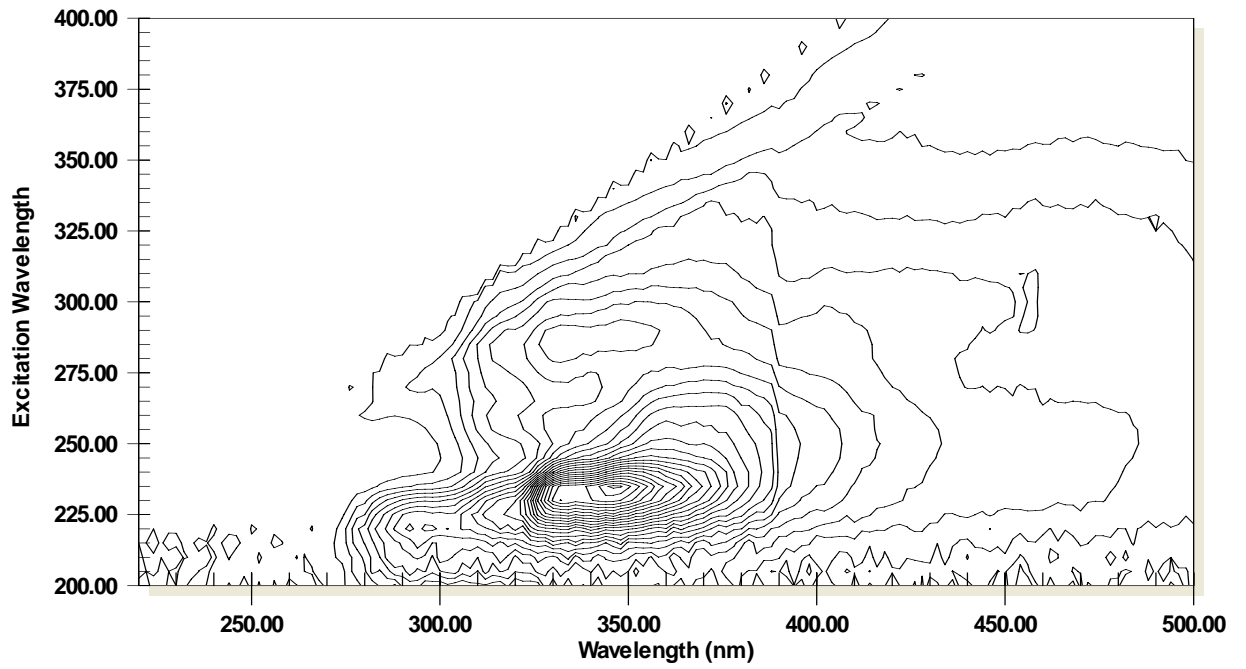
14By



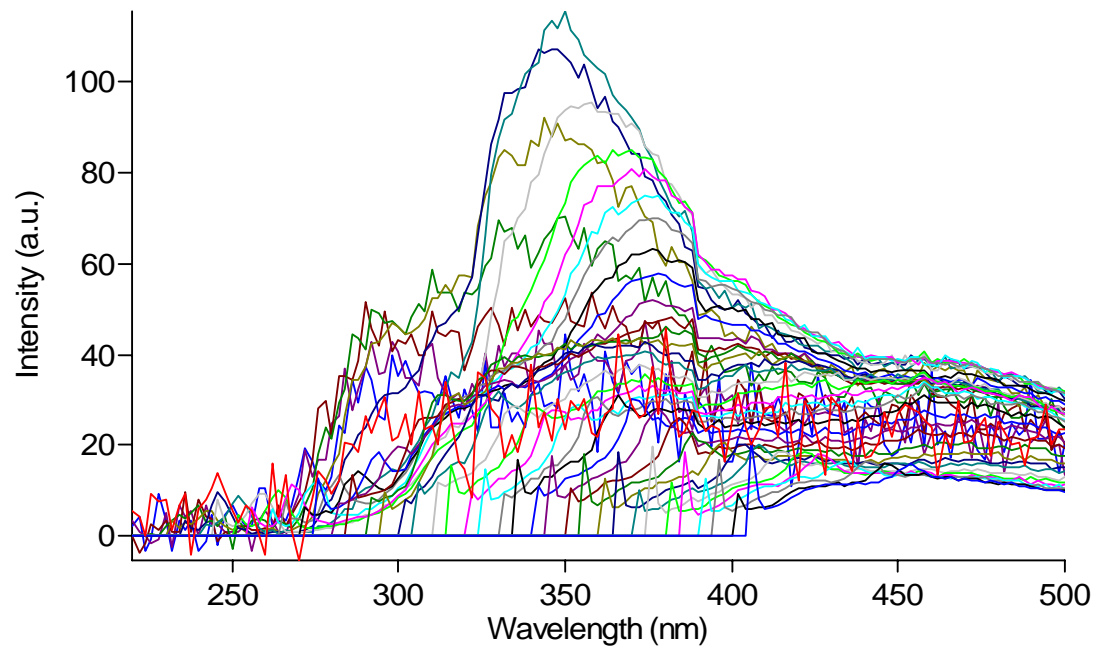
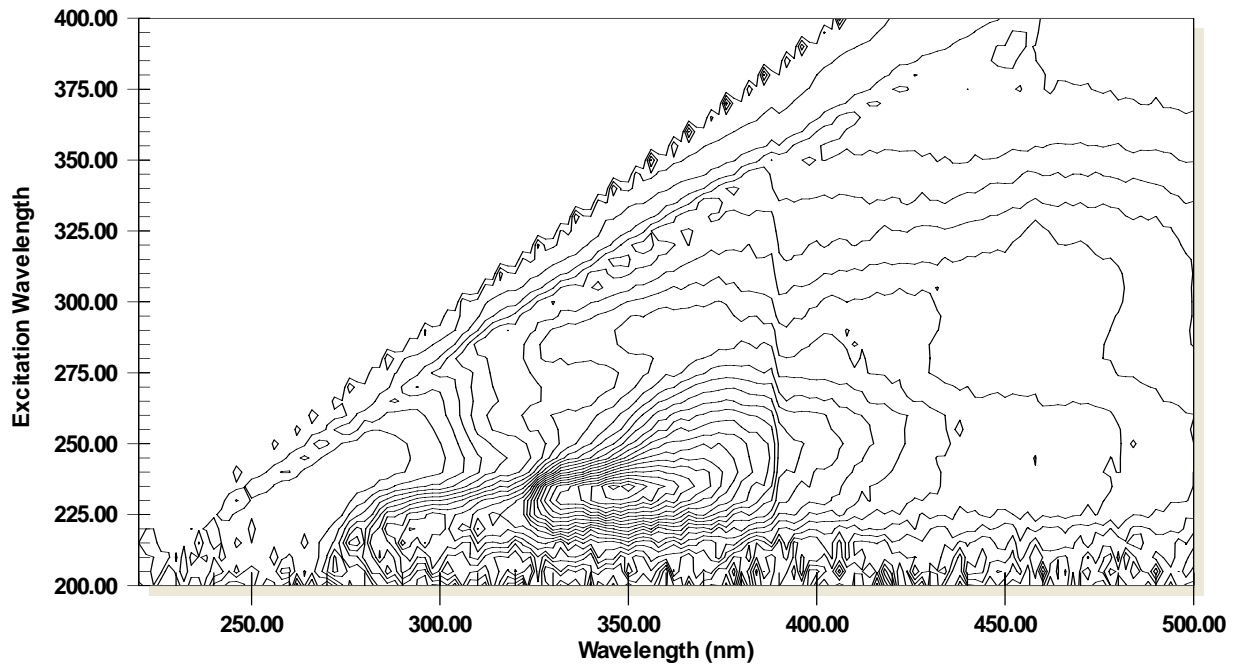
15By



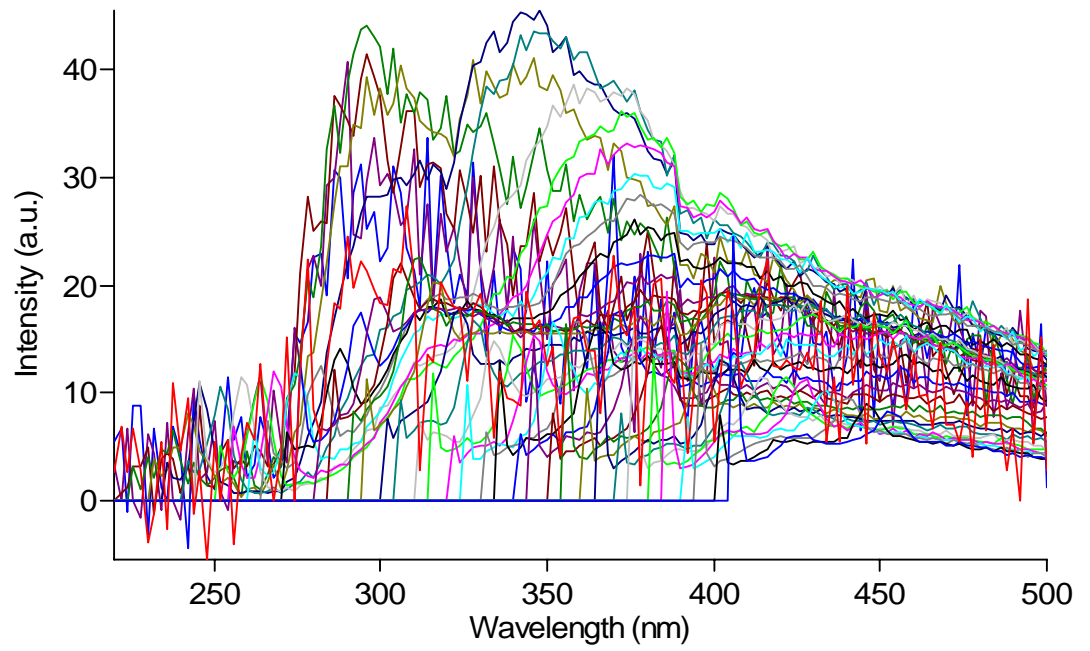
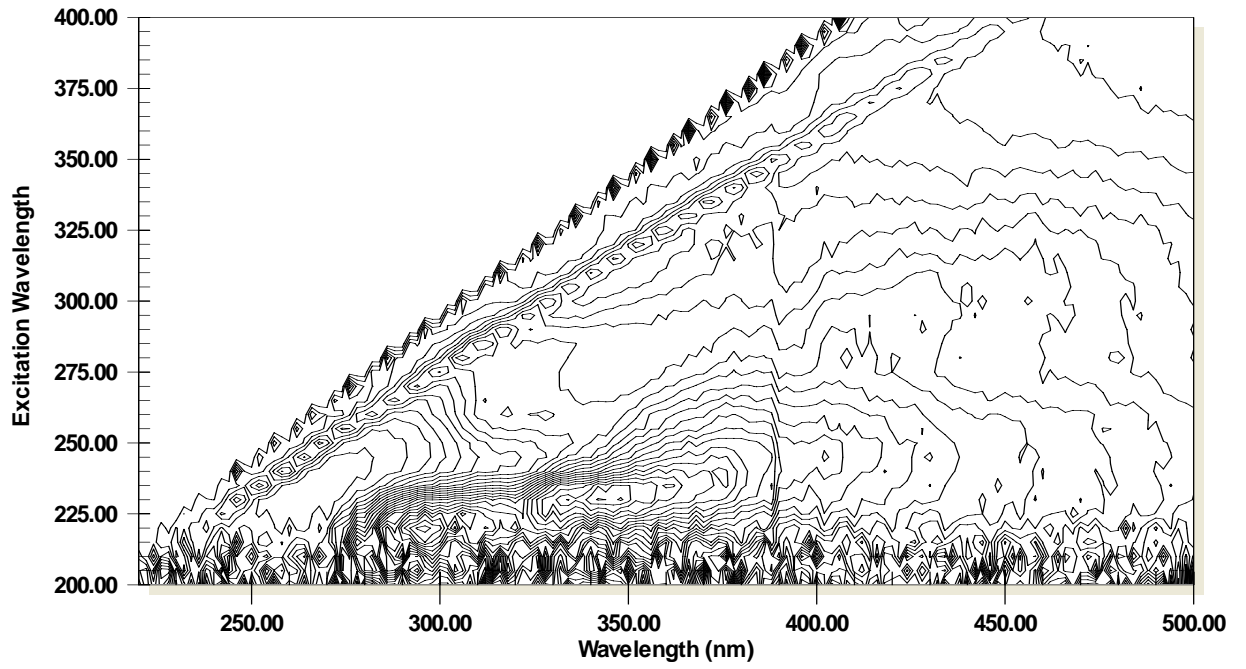
21By



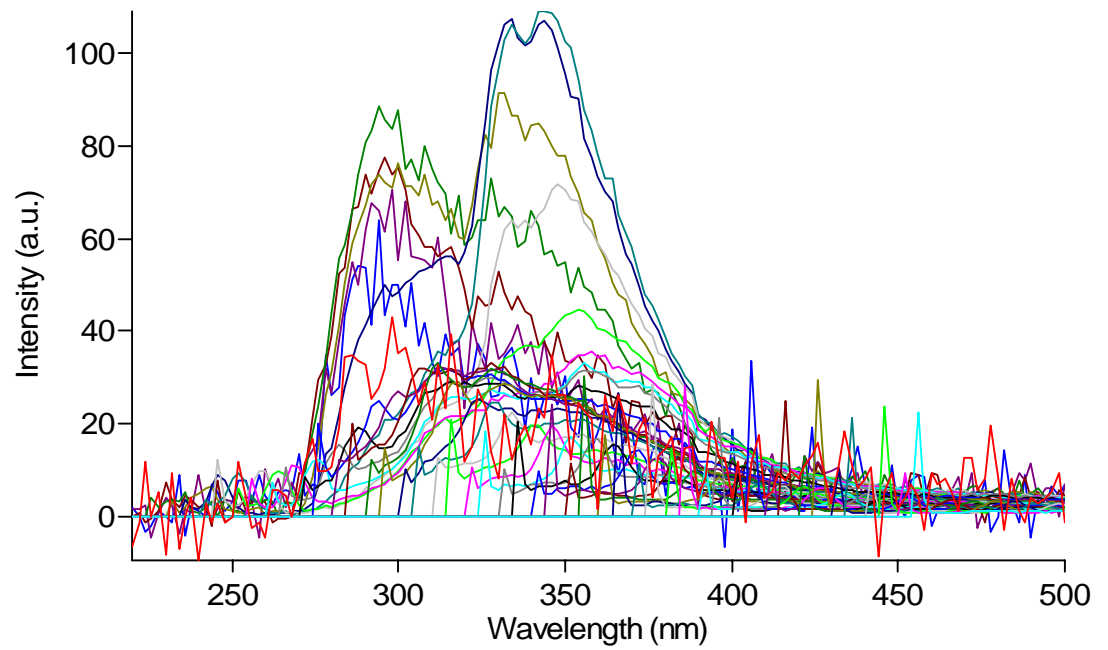
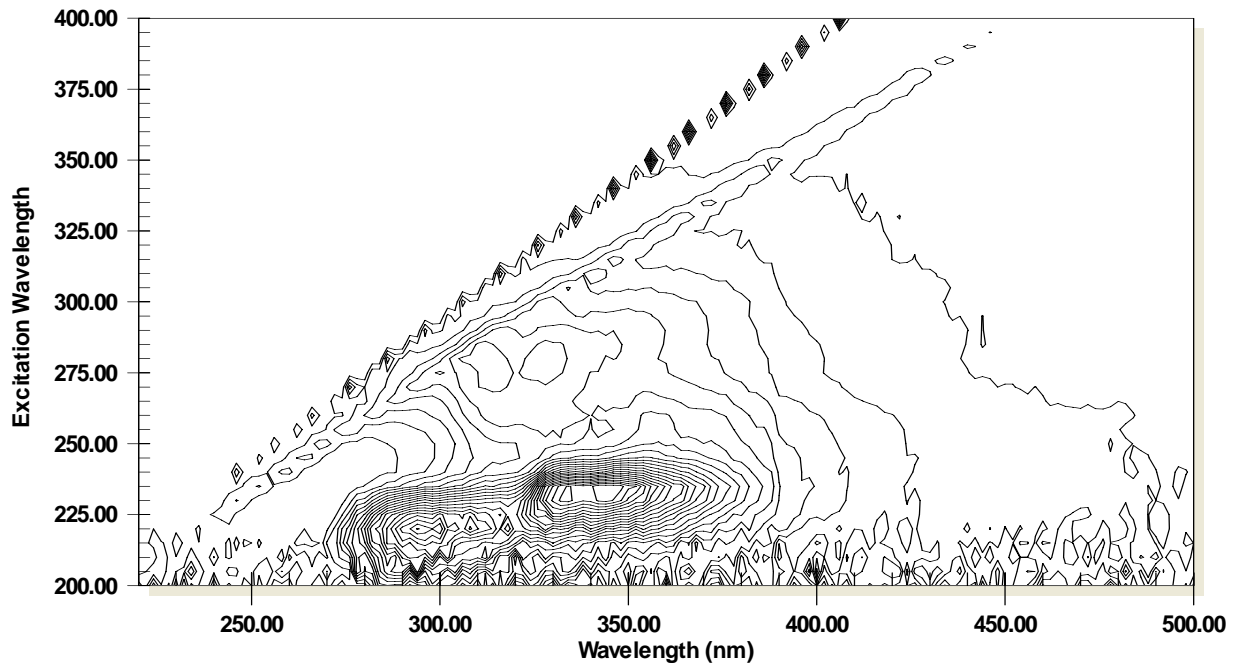
22Ay



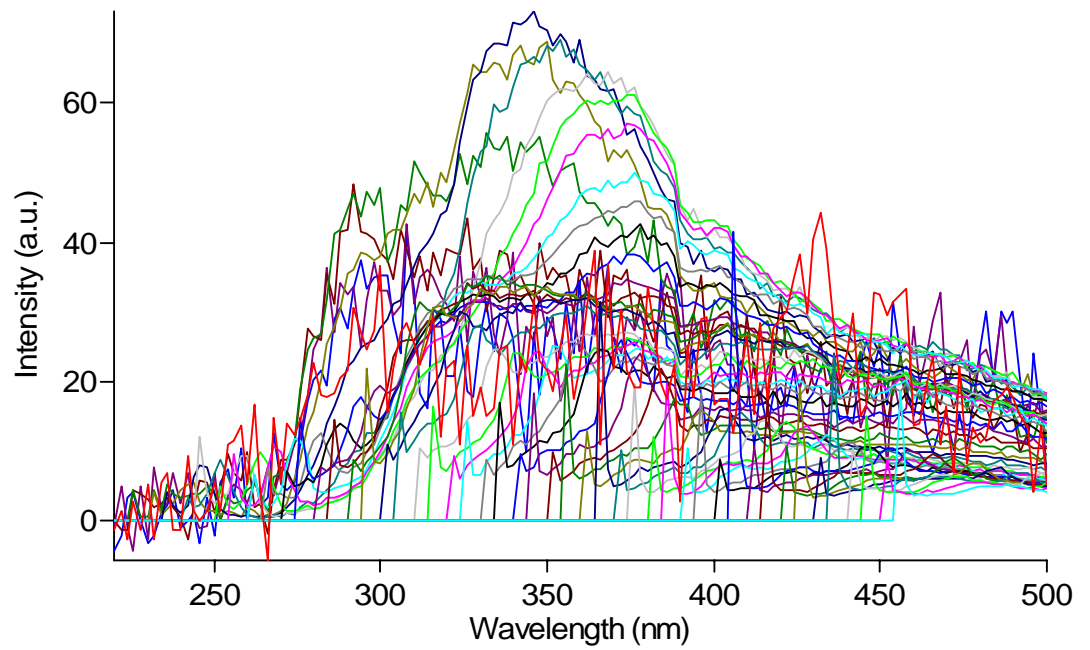
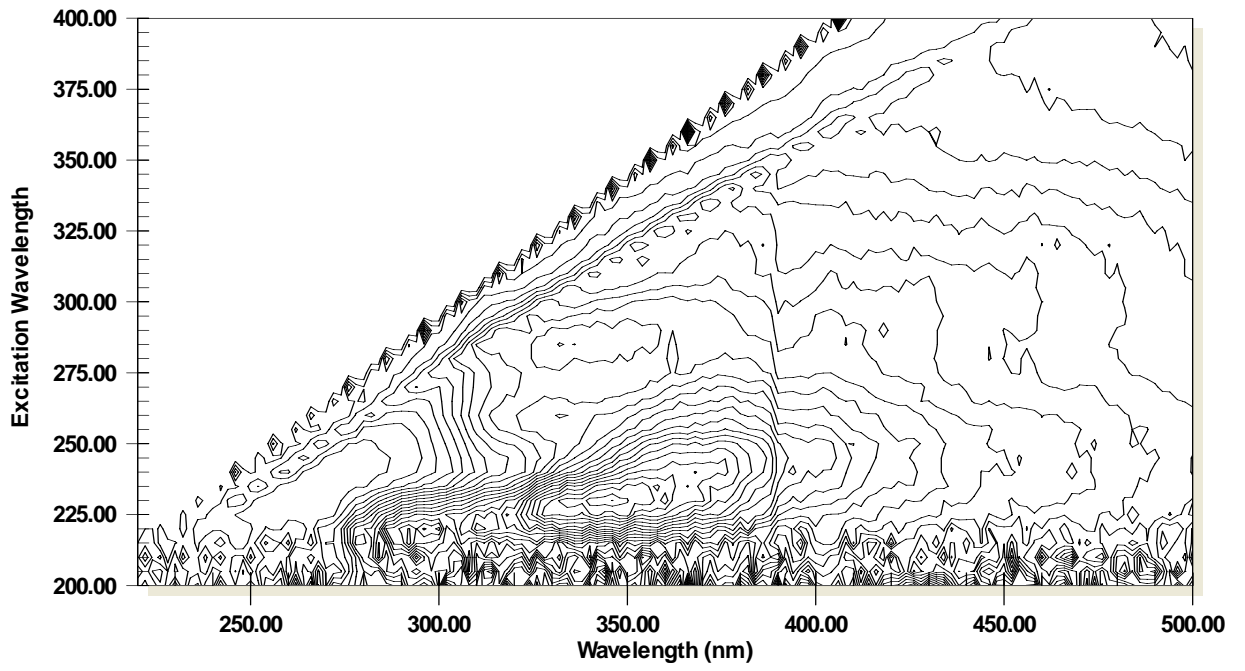
23Ay



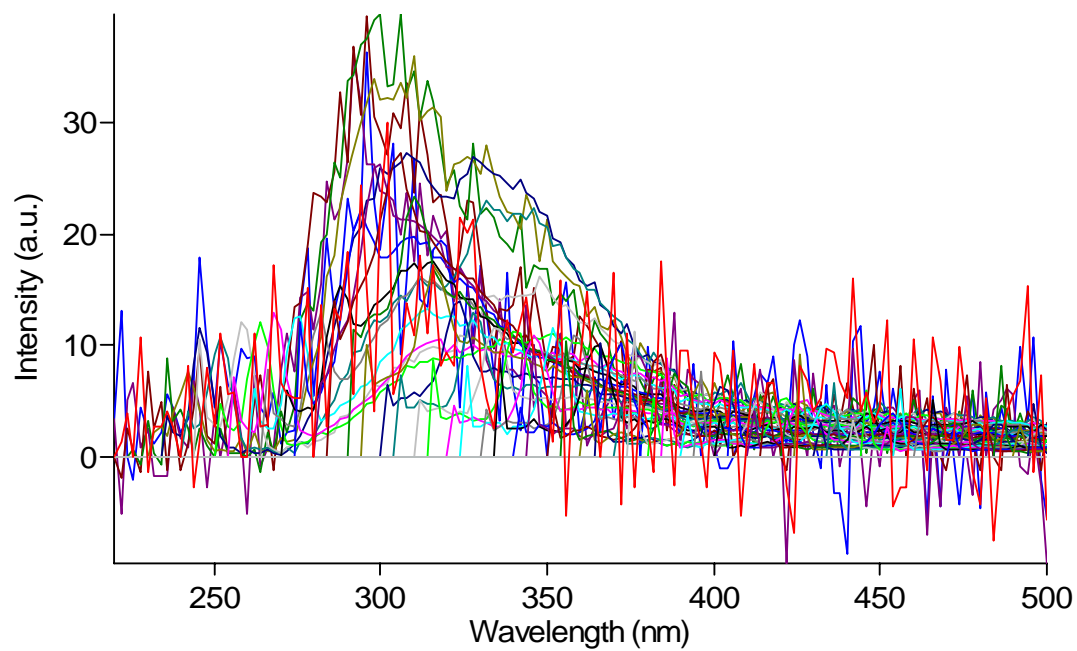
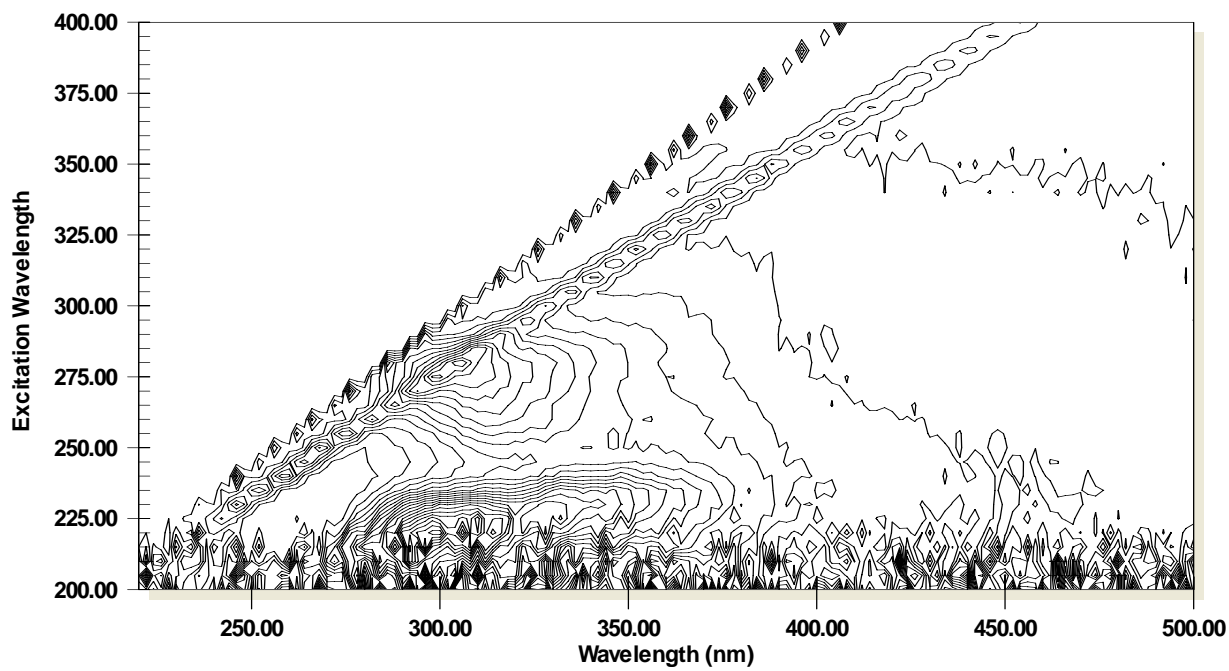
24Cy



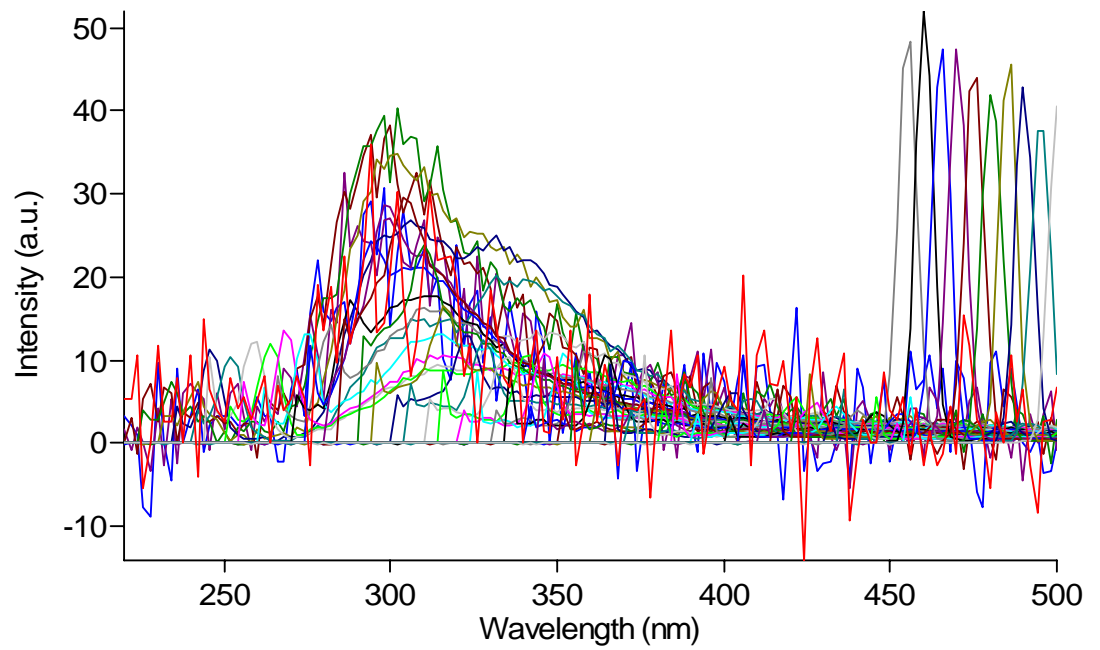
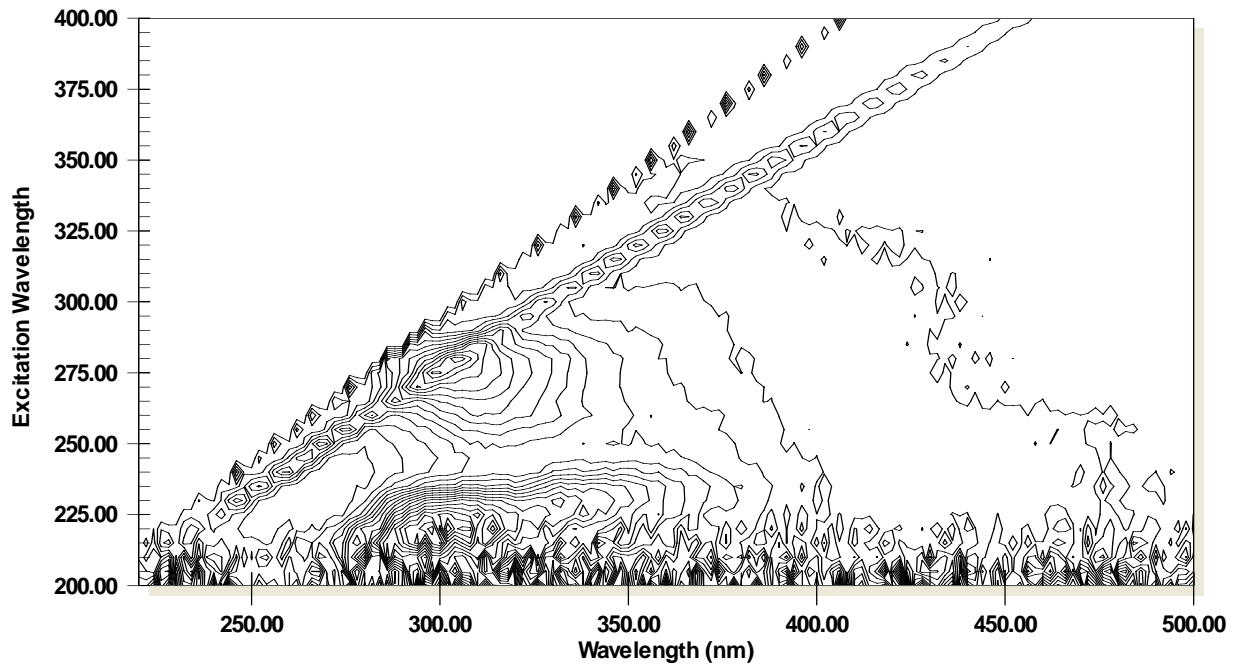
25Cy



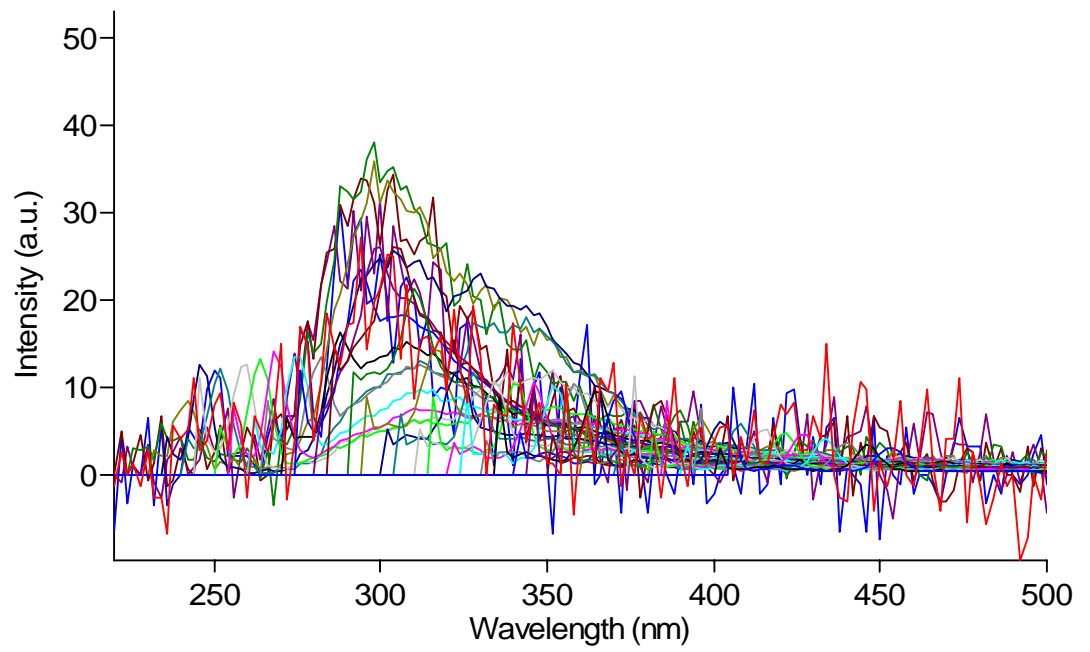
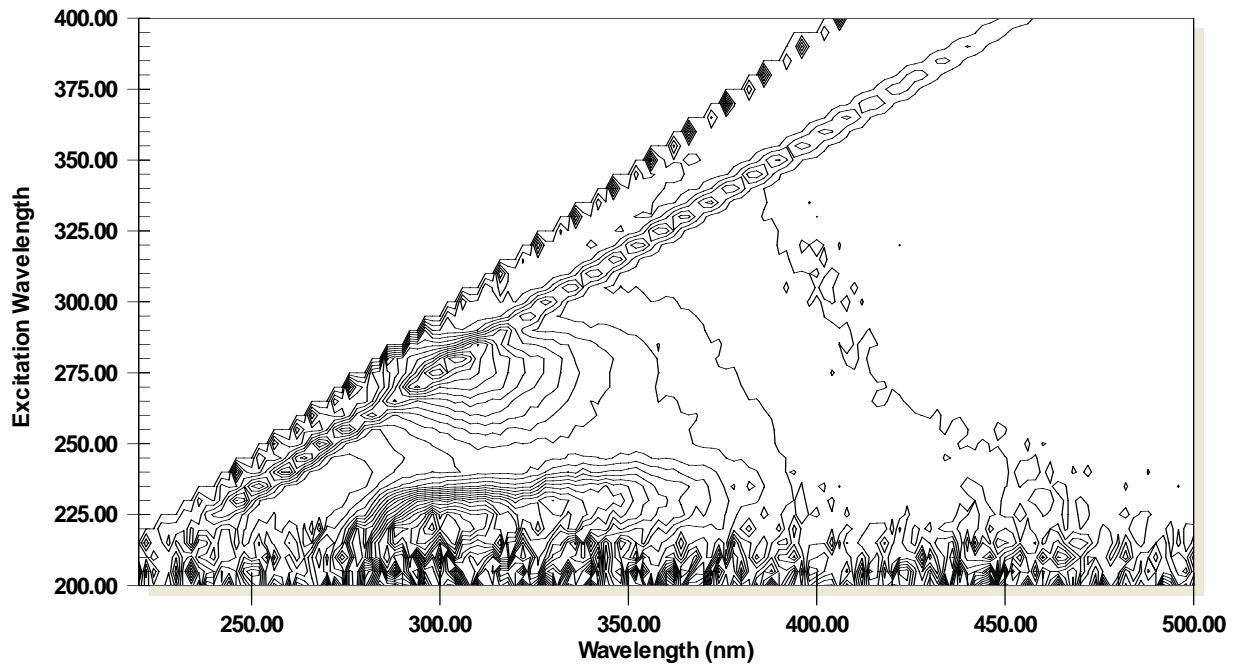
31Cy



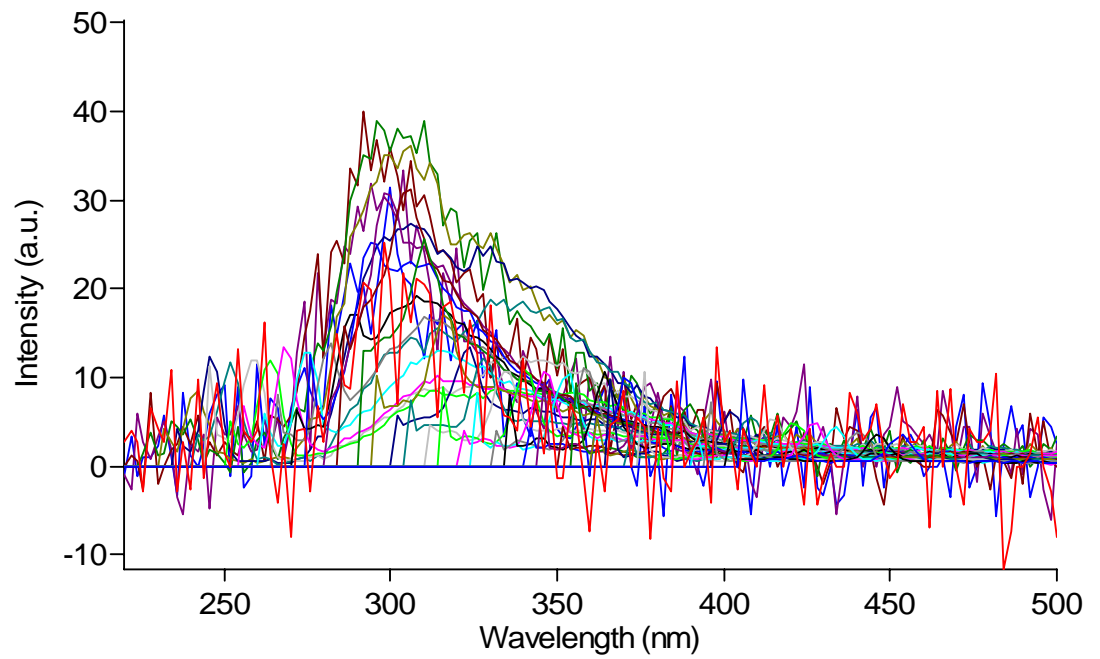
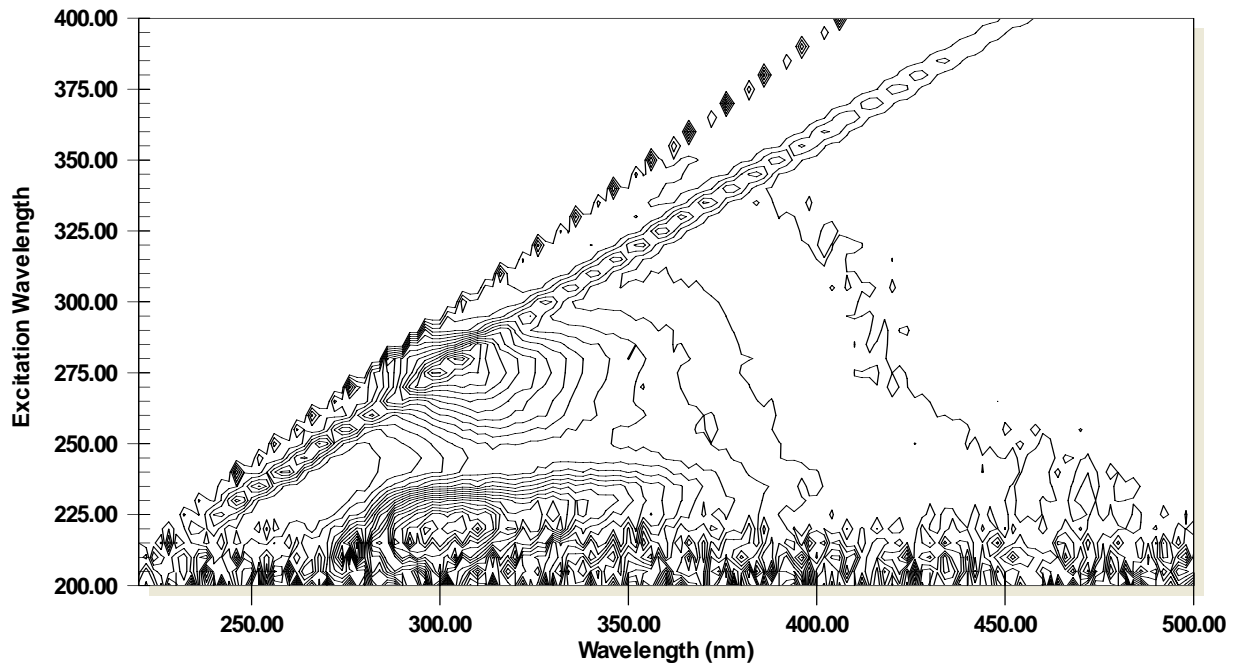
32Cy



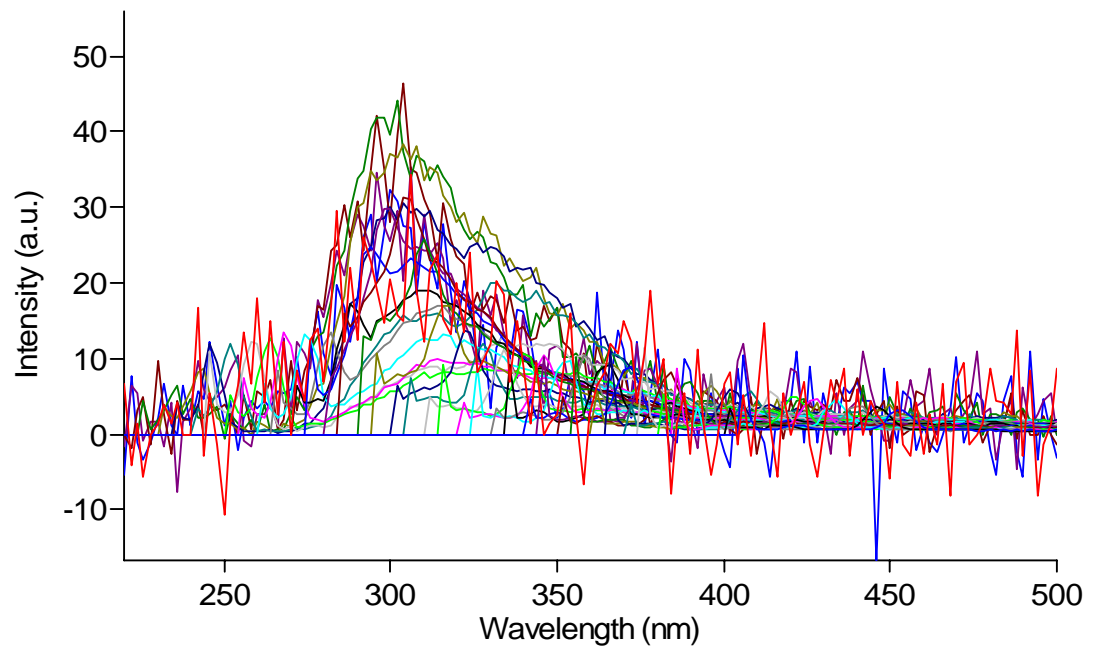
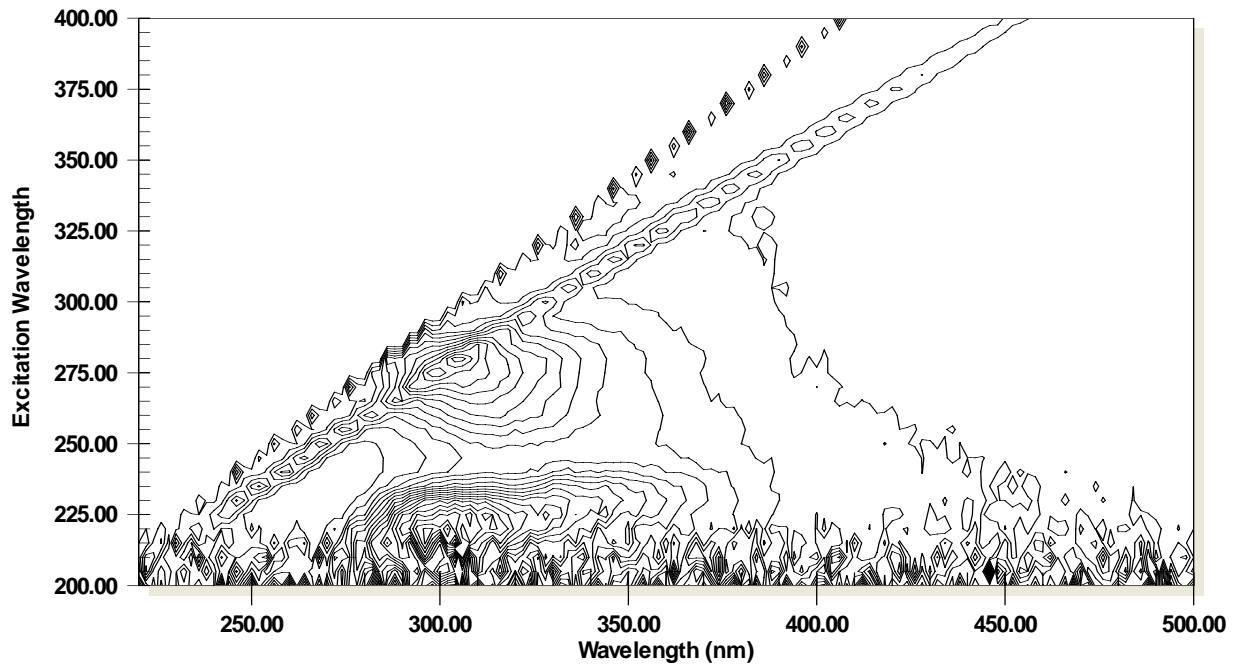
33Ay



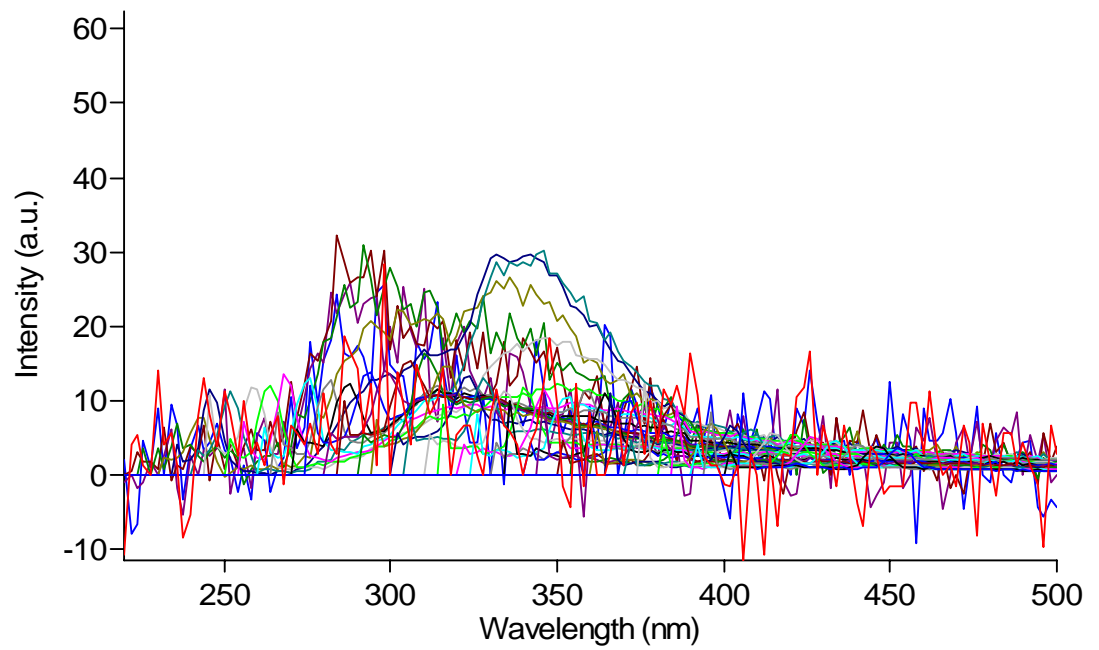
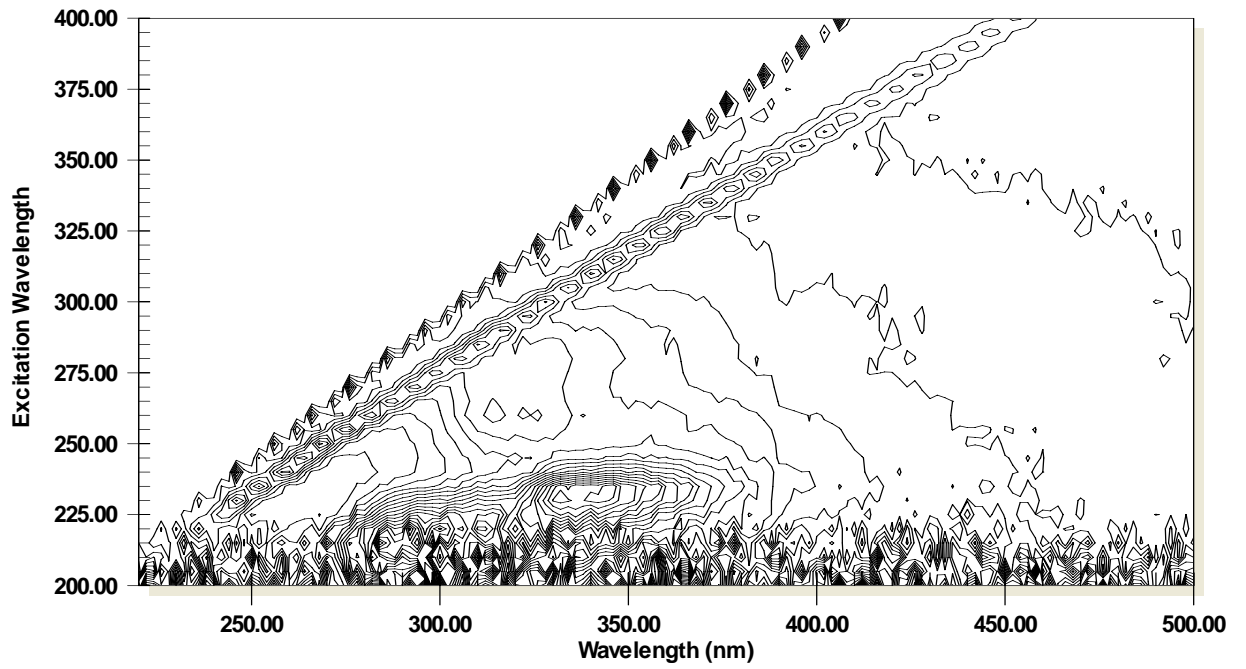
34By



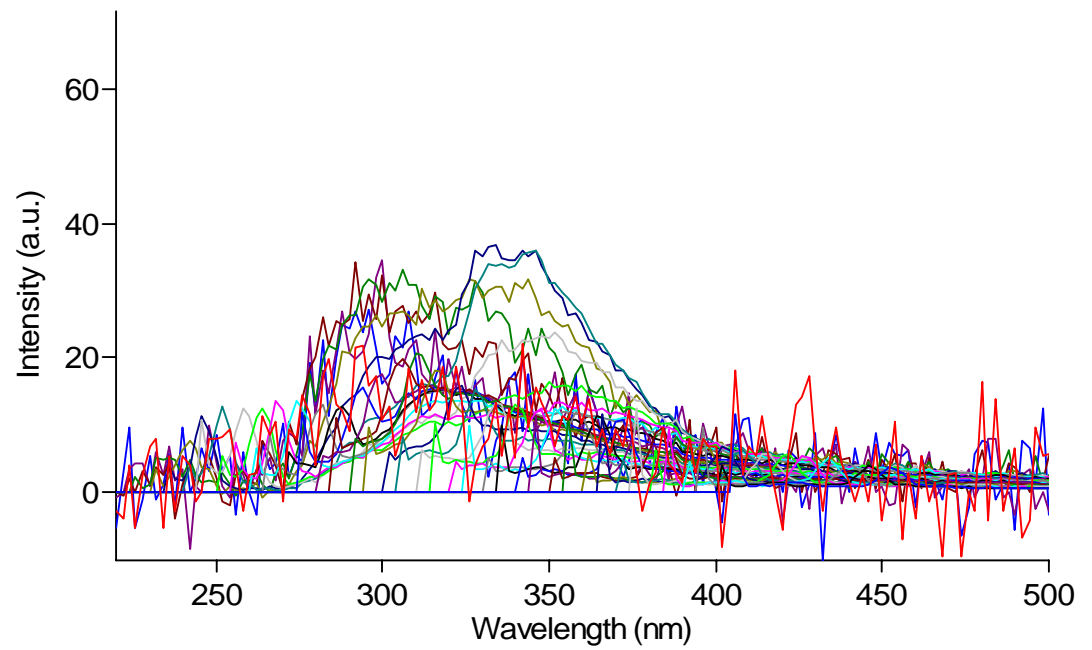
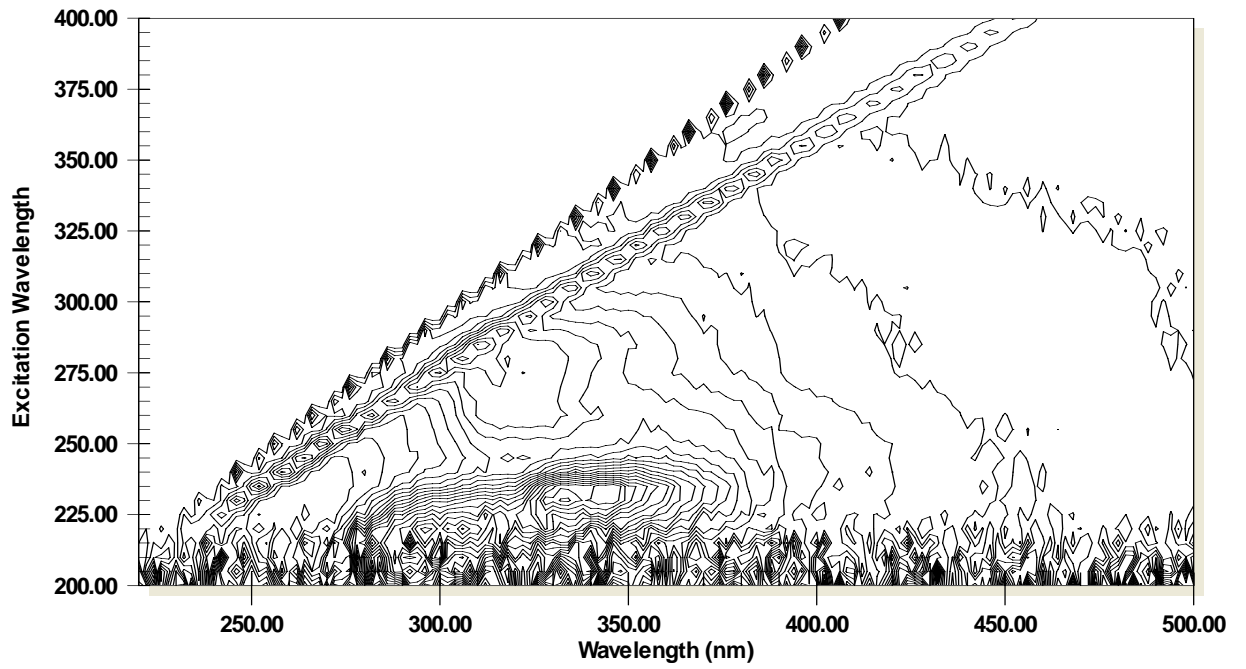
35Cy



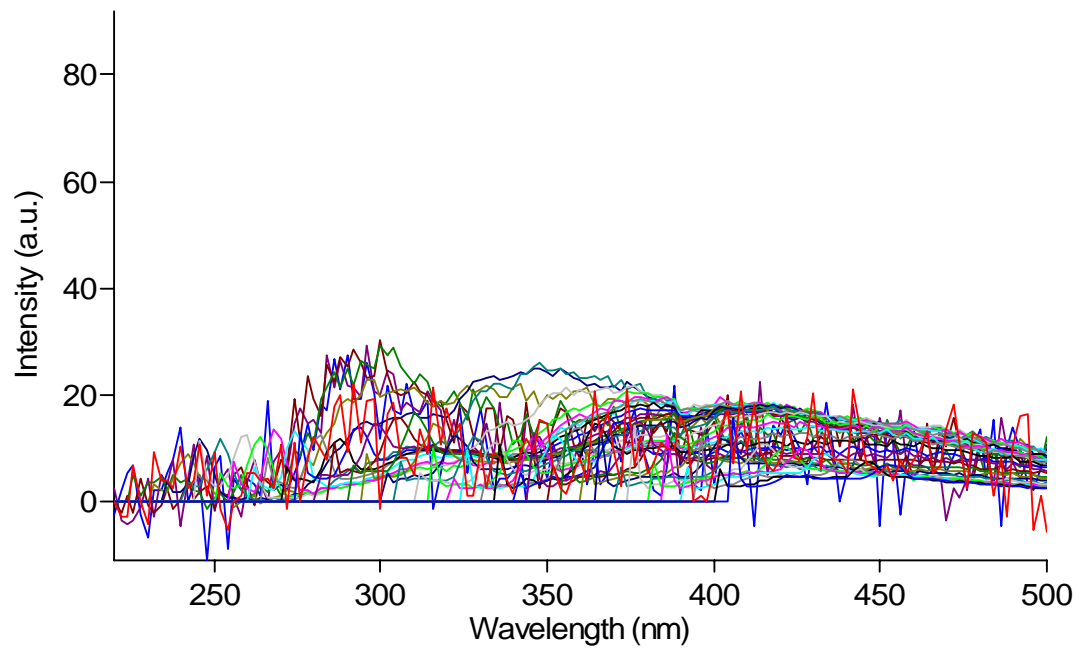
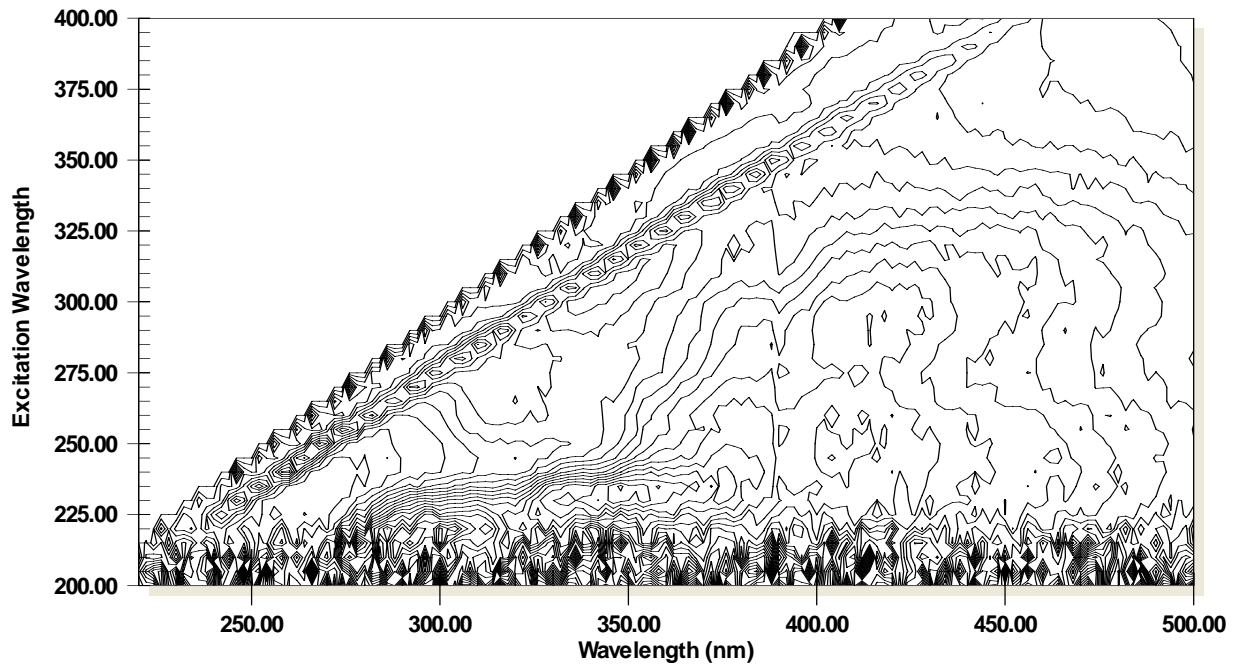
41Cy



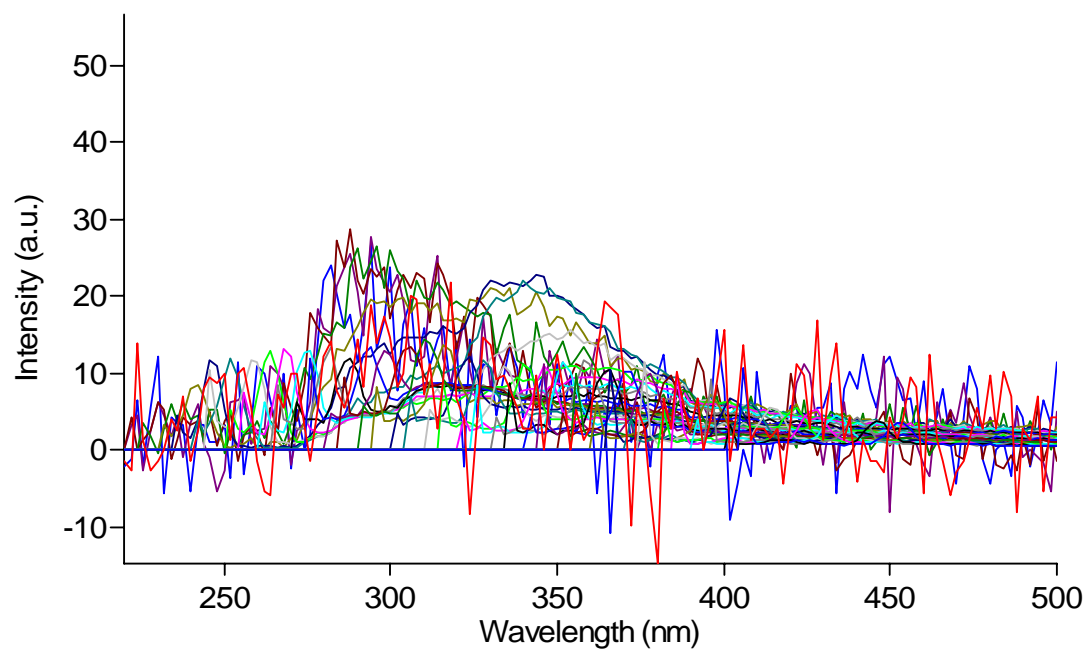
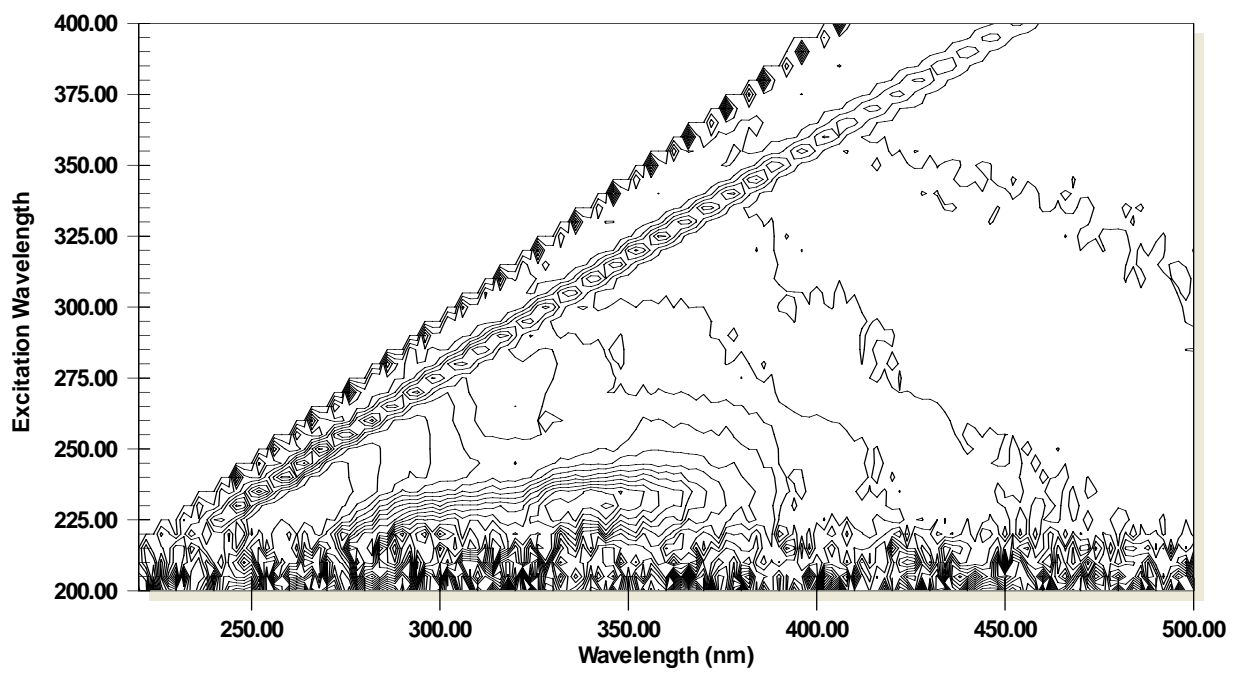
42By



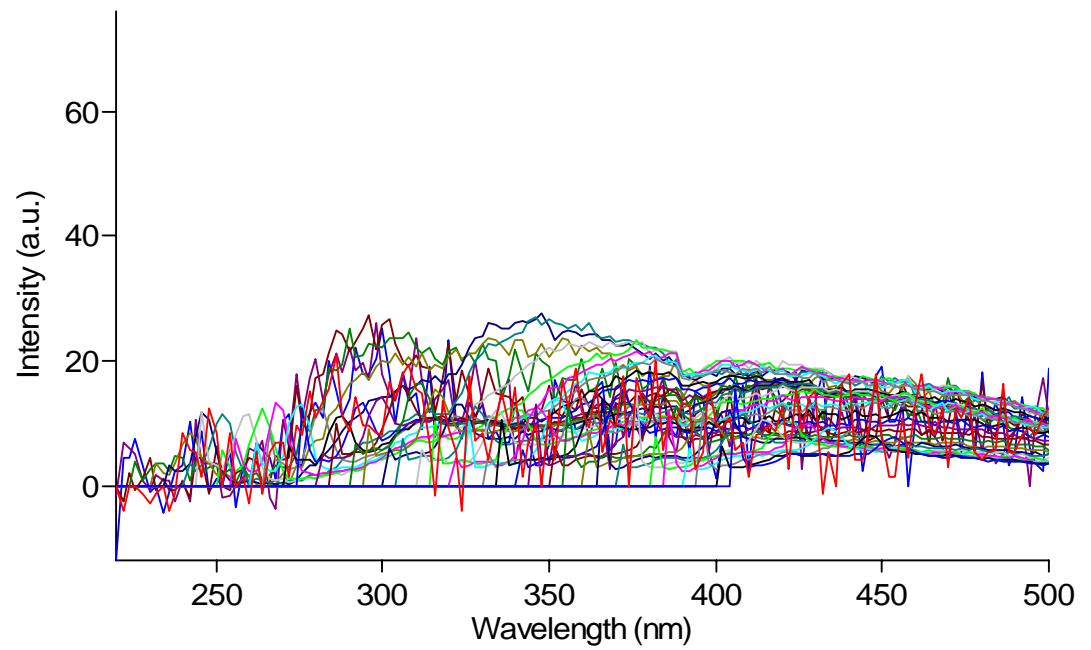
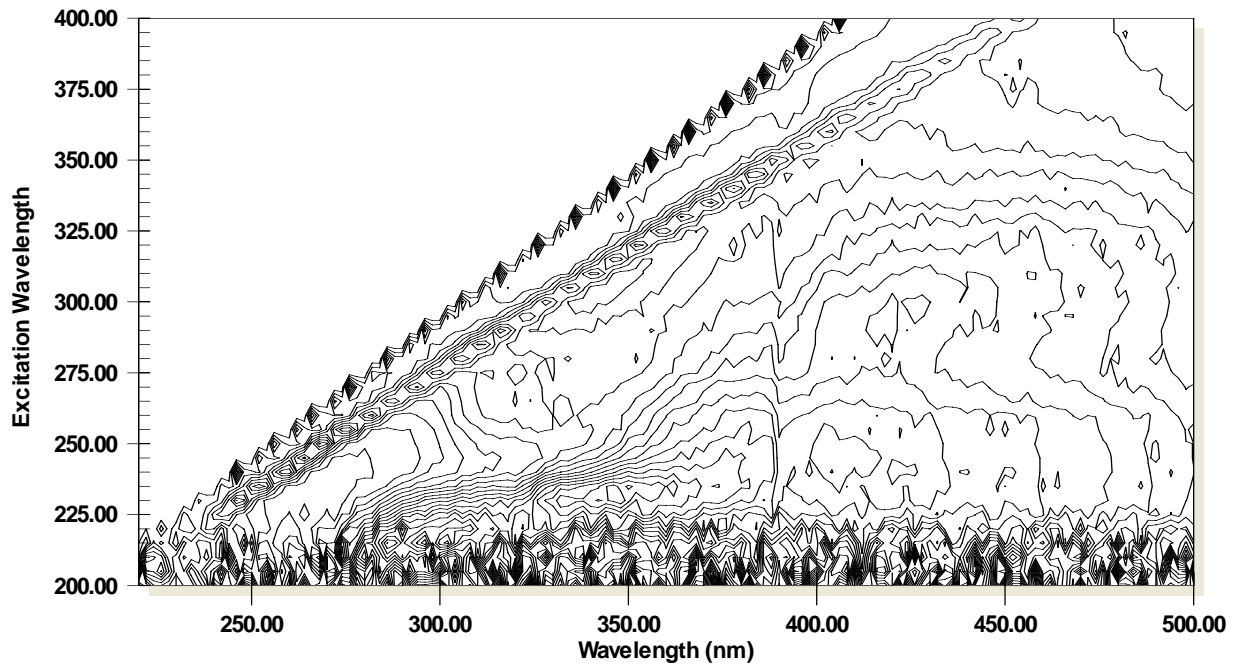
43Ay



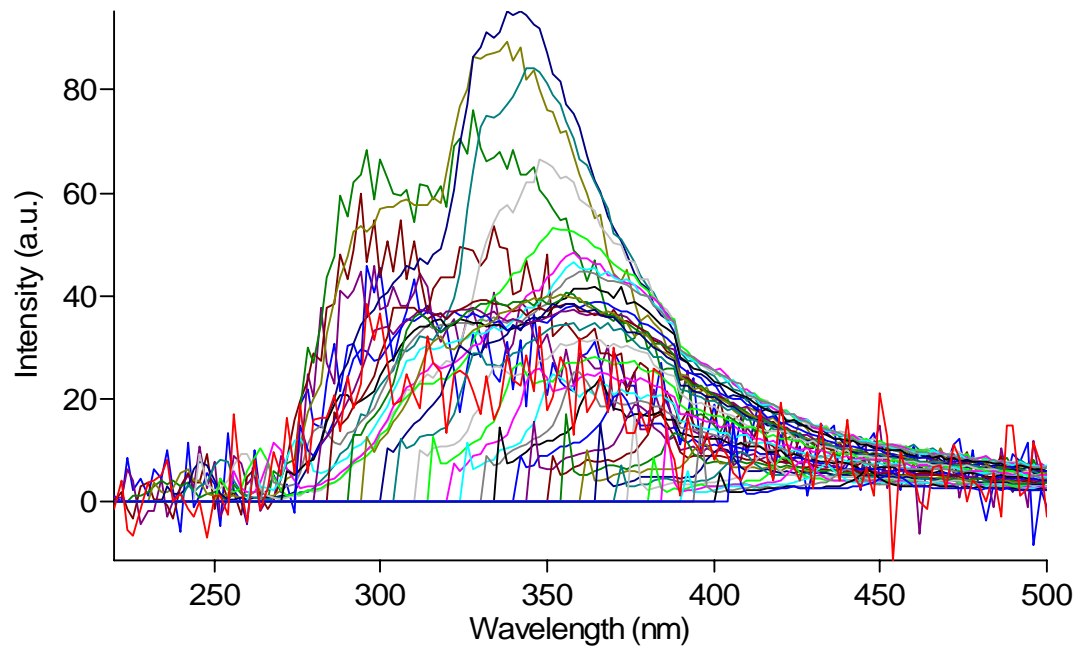
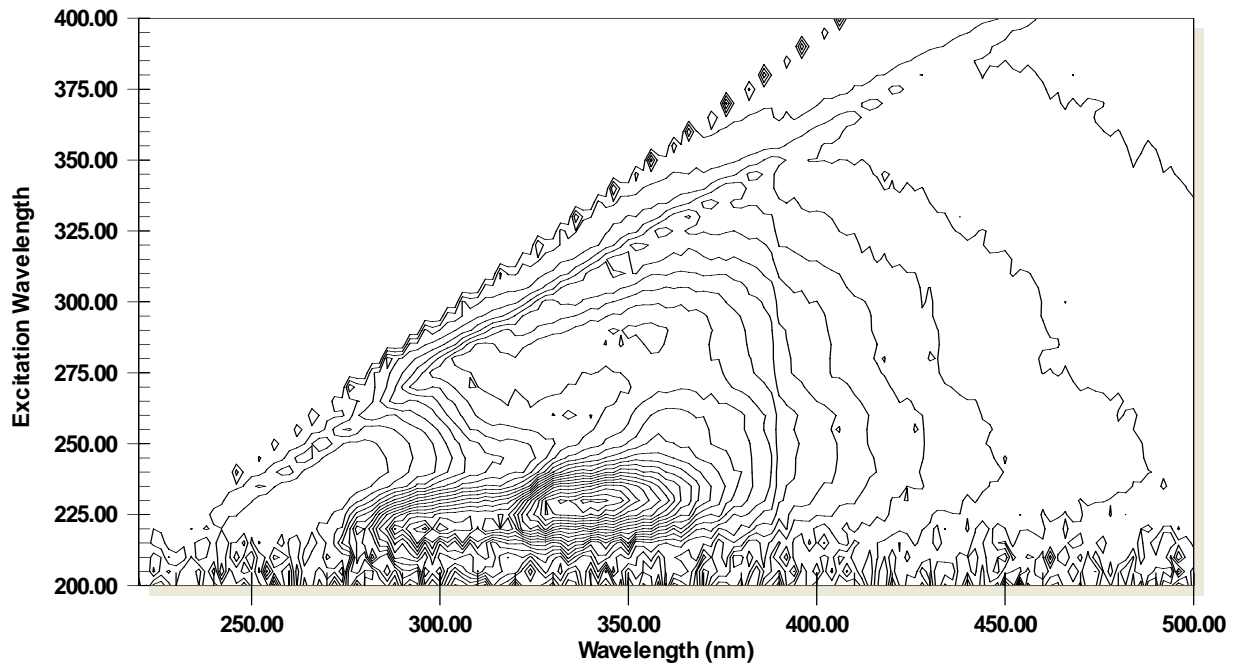
44Ay



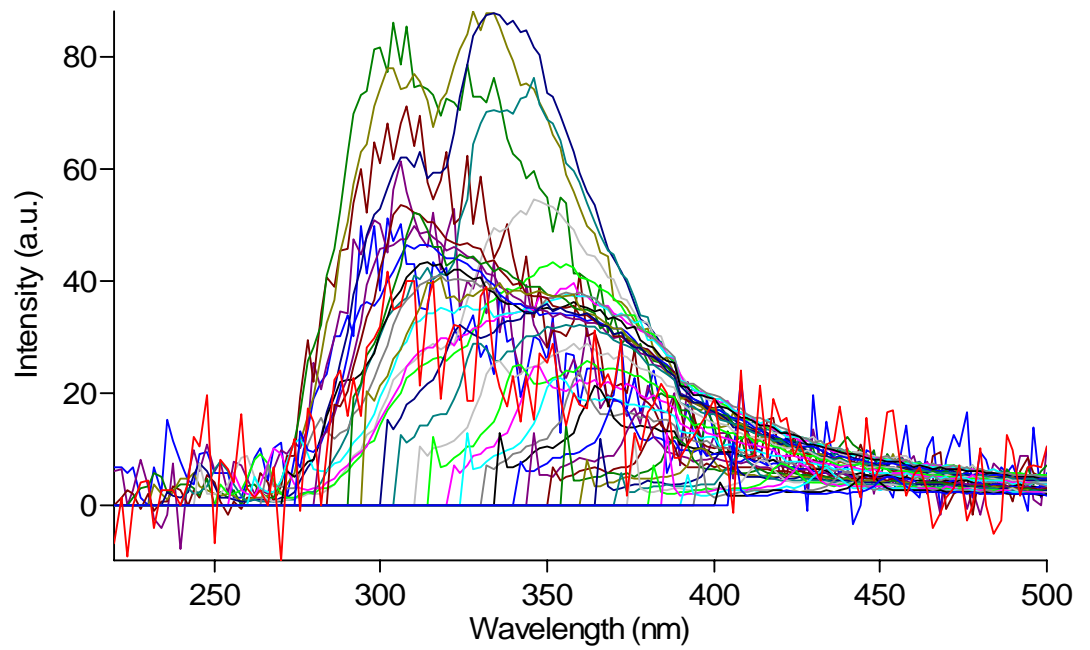
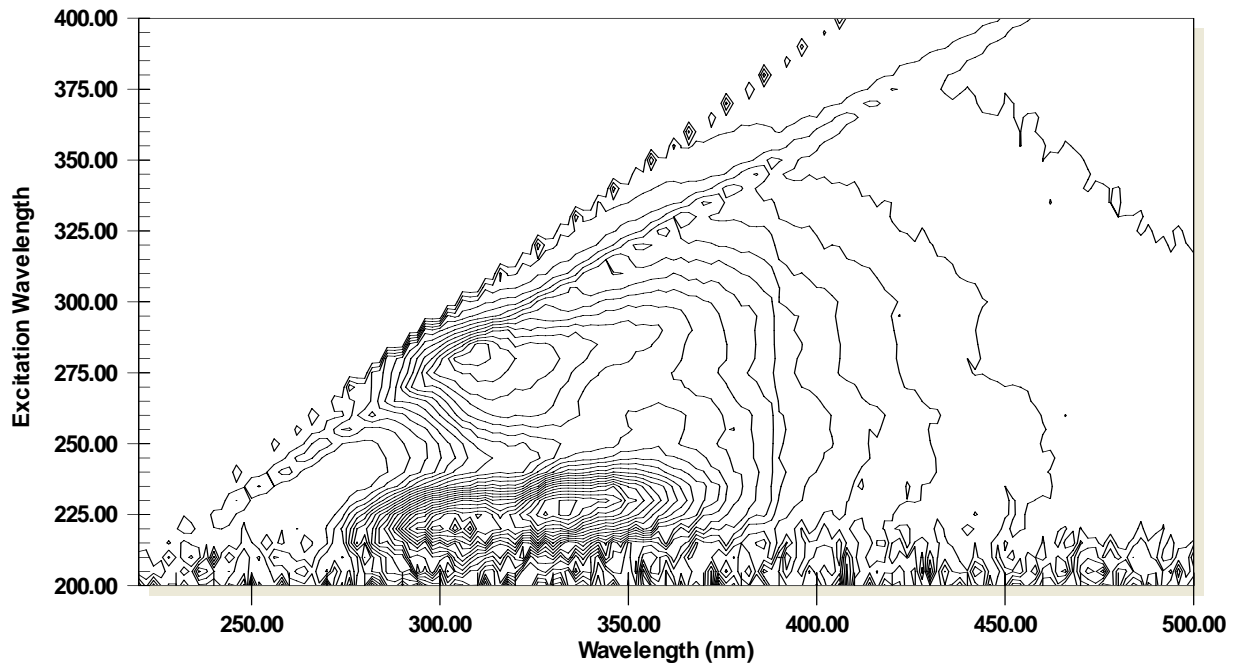
45Cy



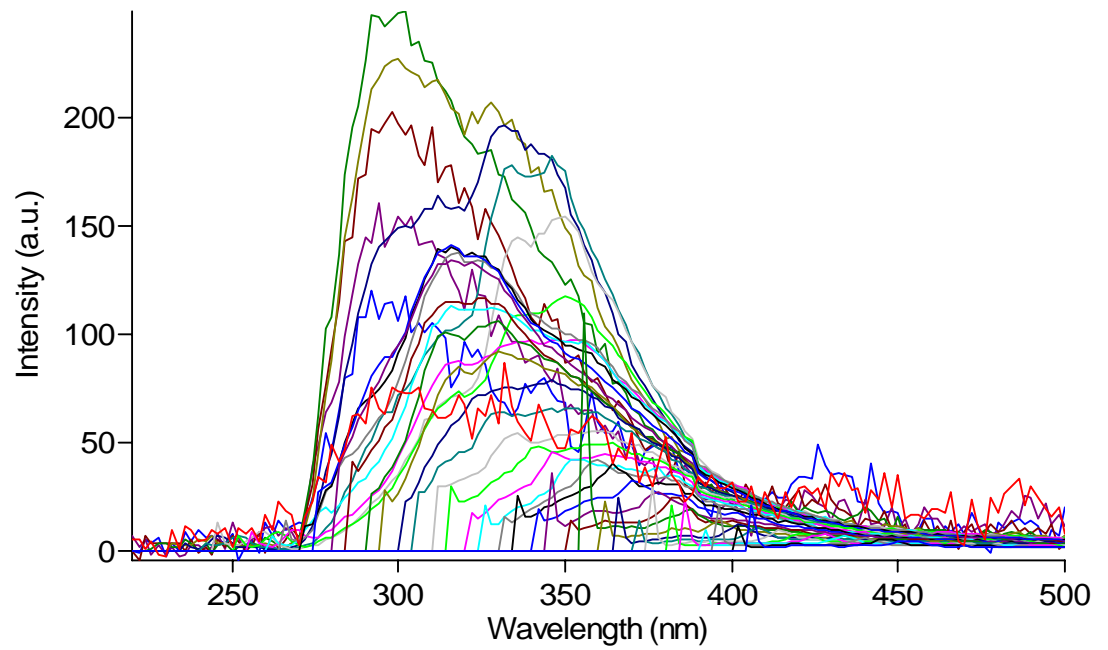
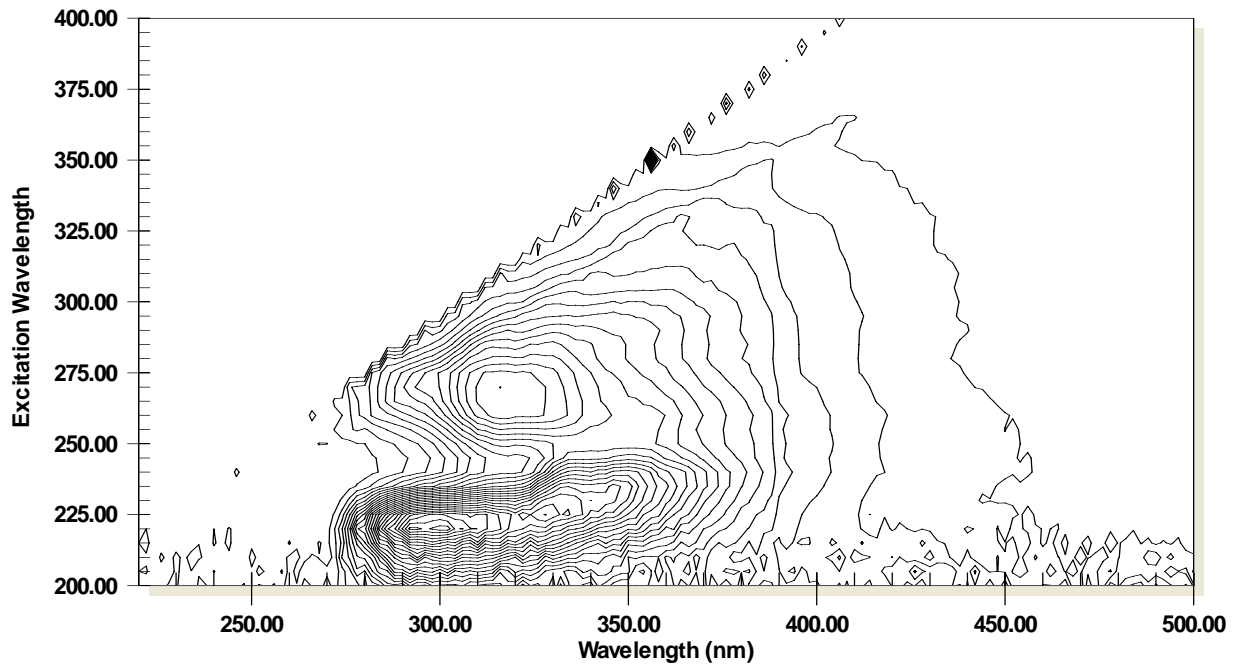
51Cy



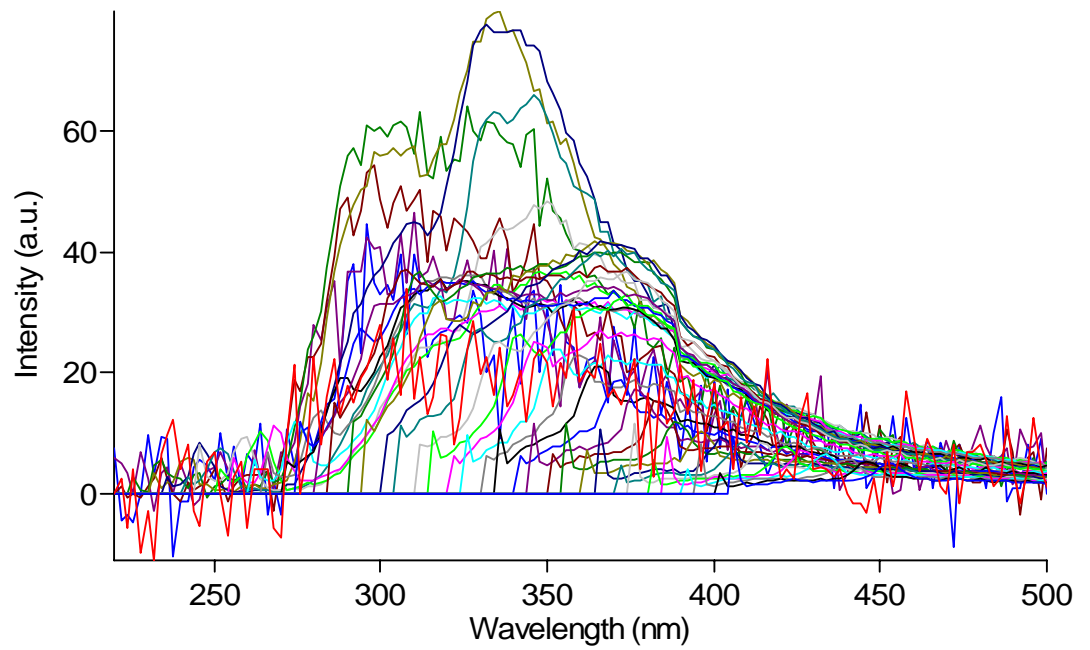
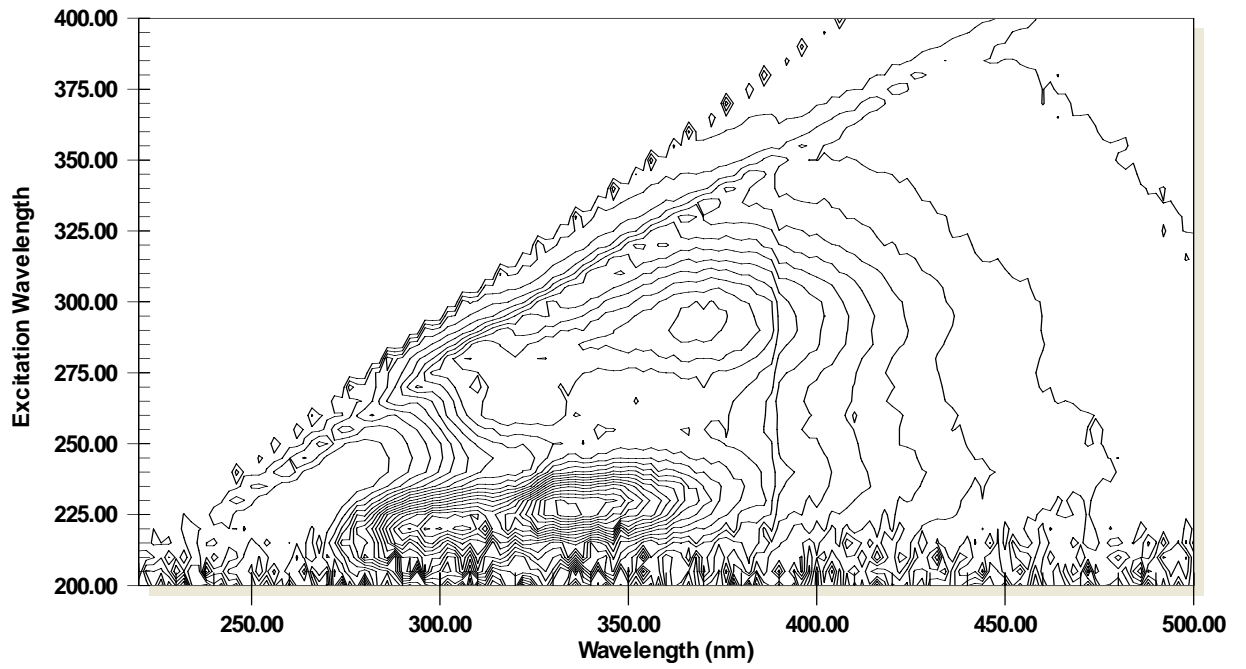
52By



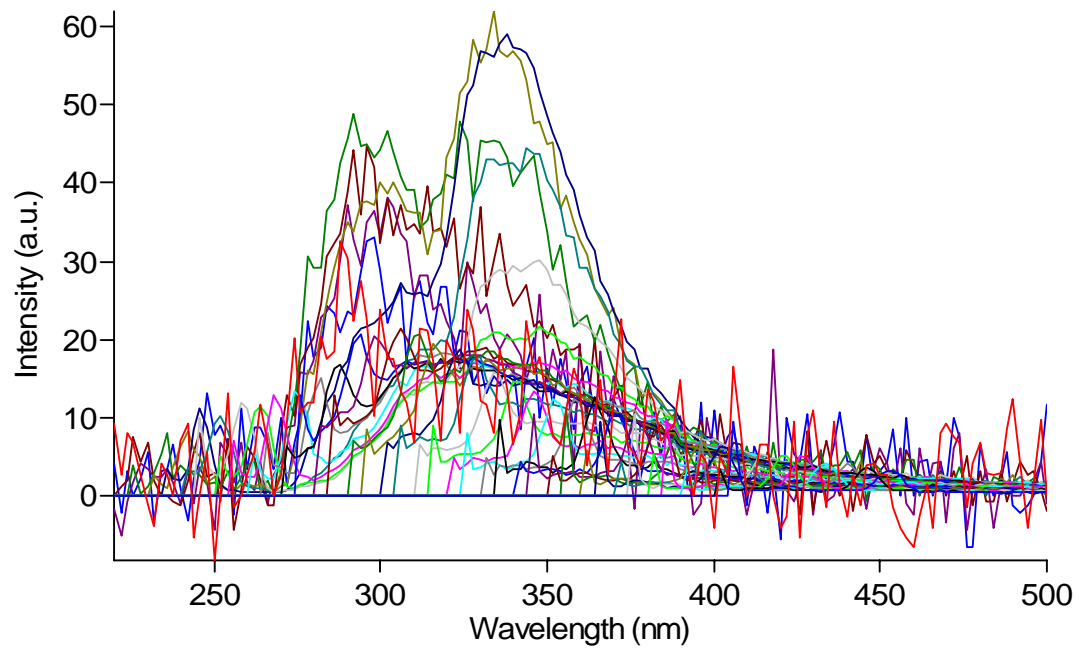
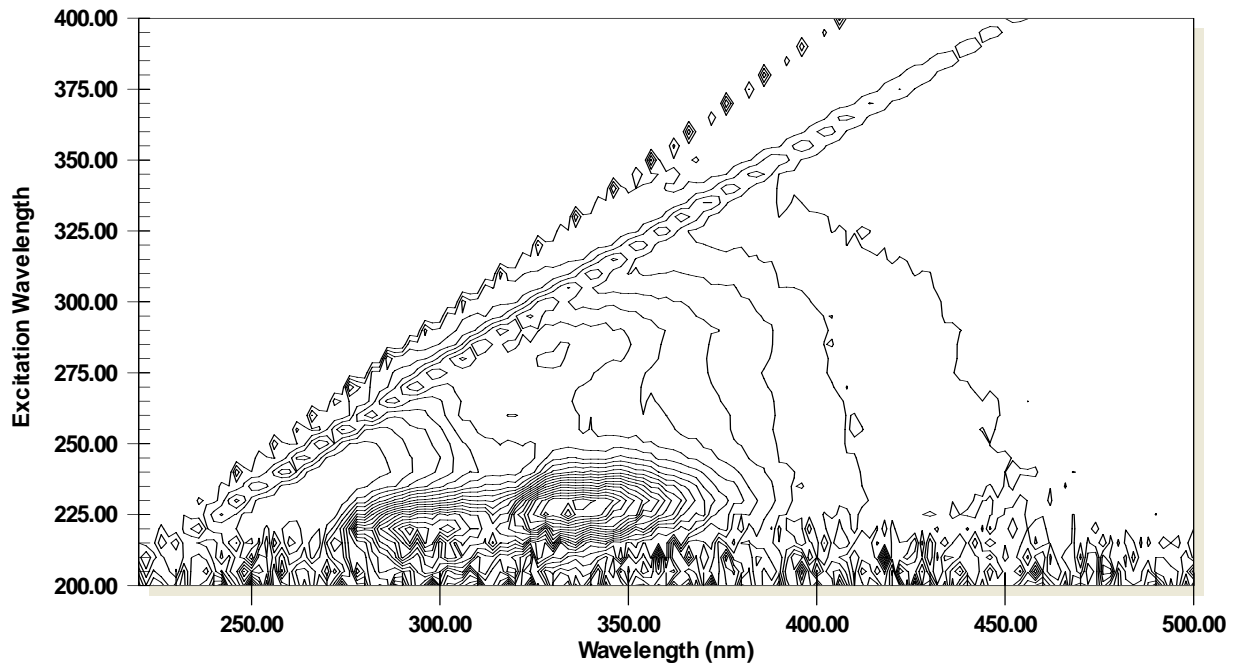
53Ay



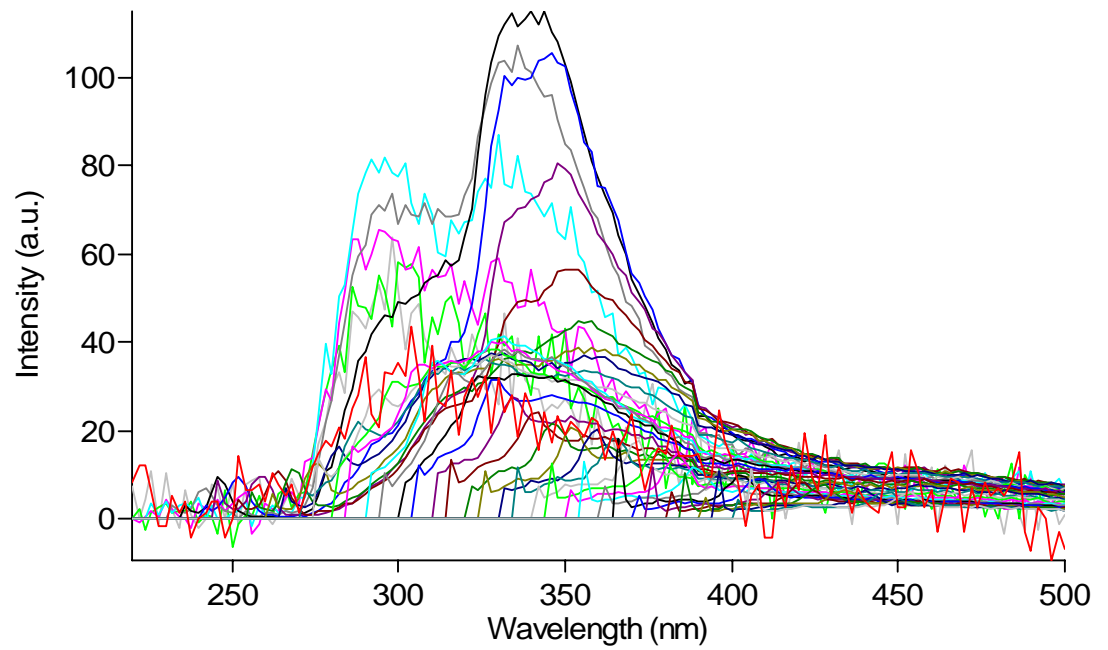
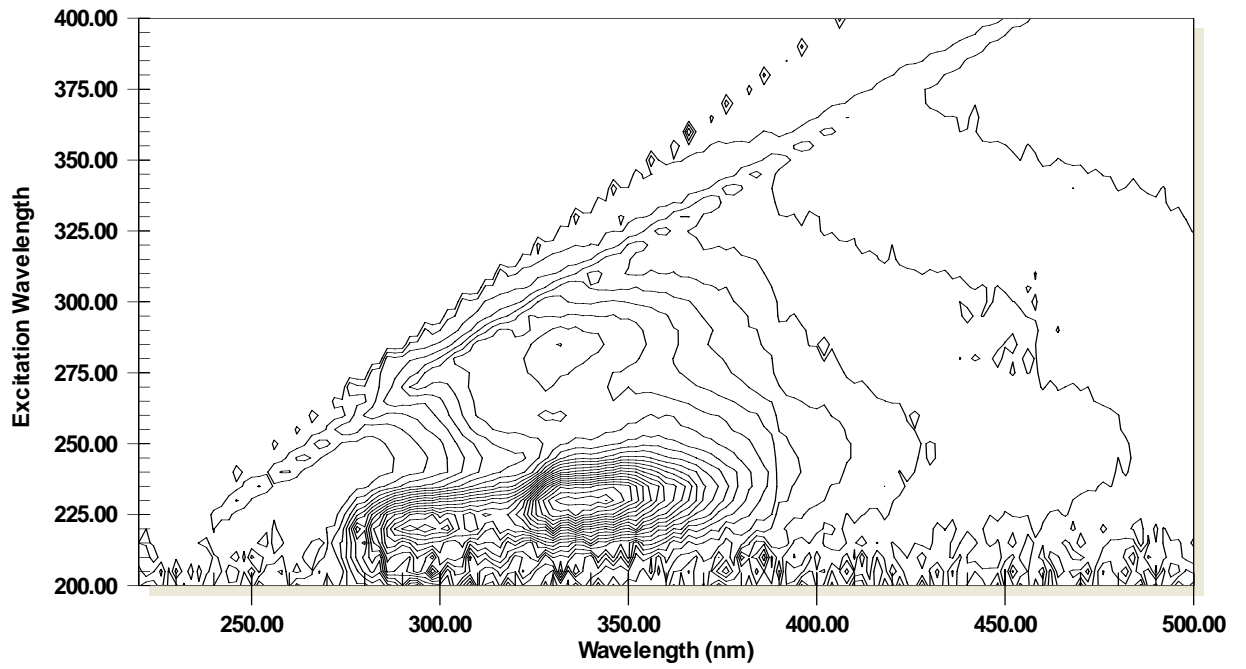
54By



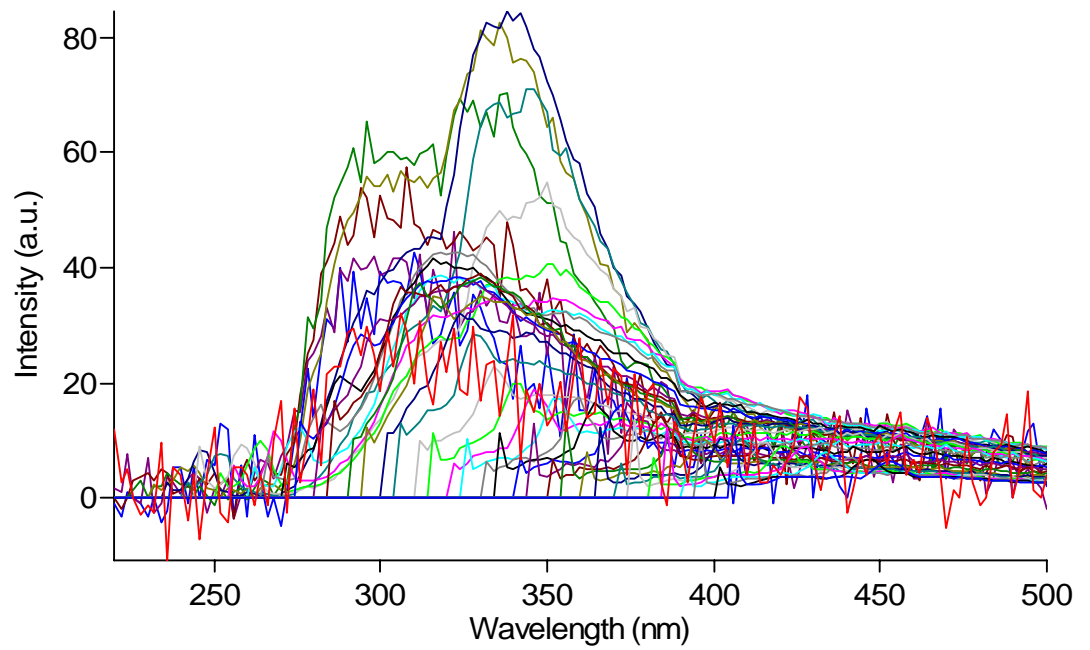
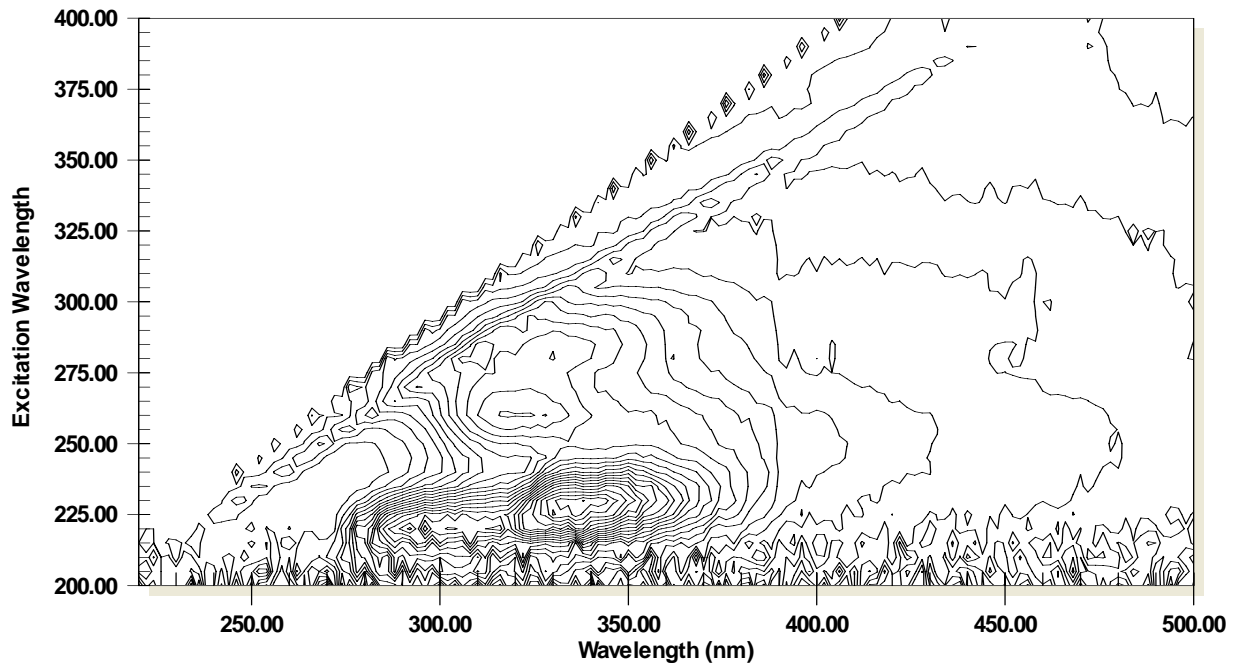
55Cy



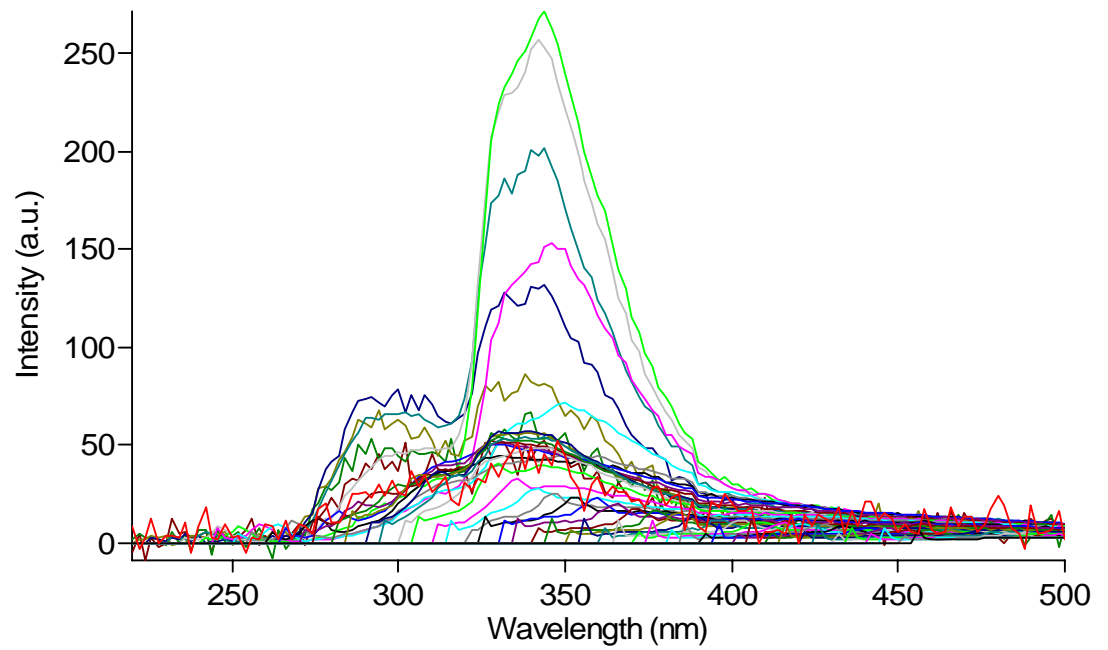
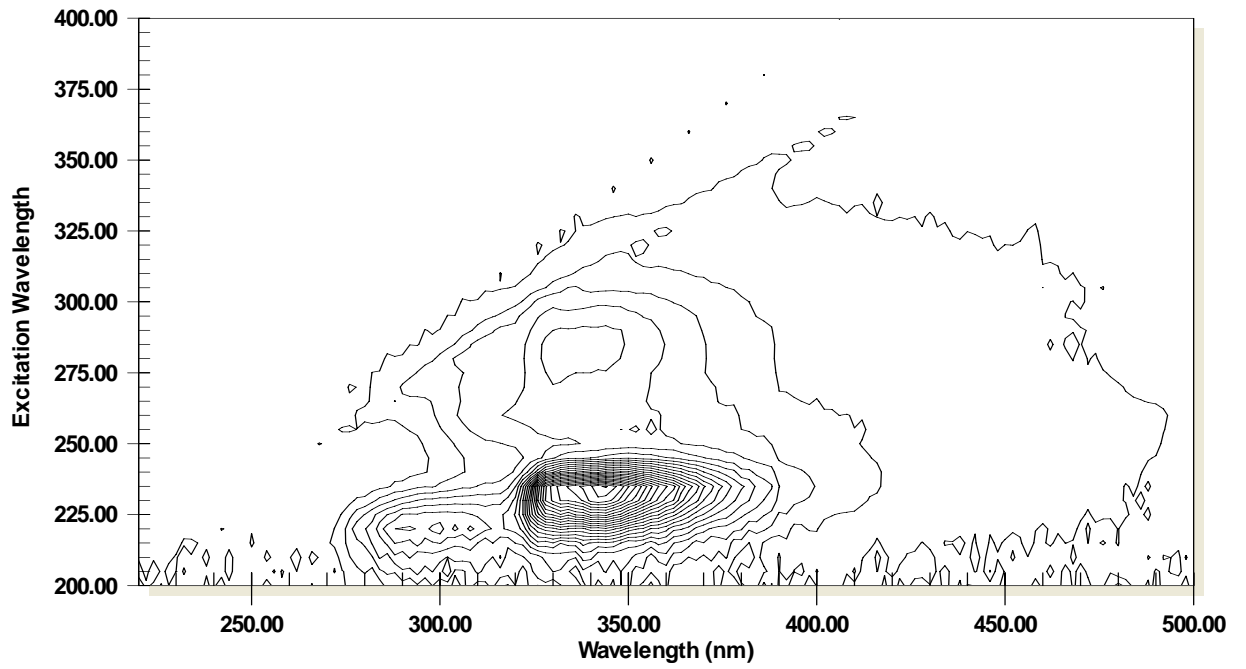
61Ay



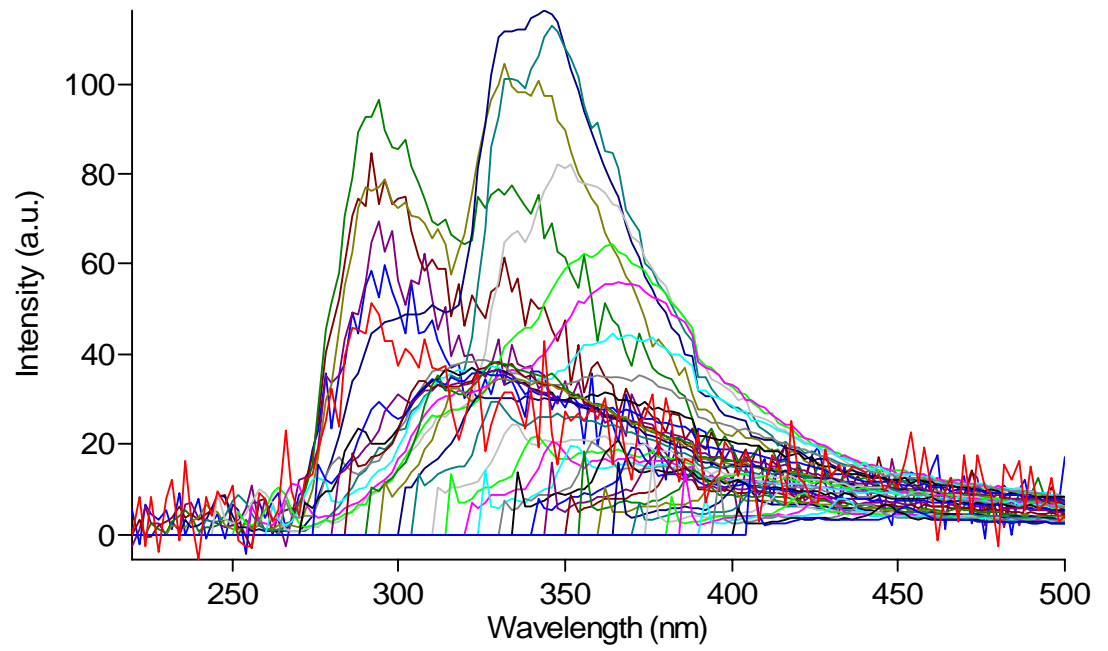
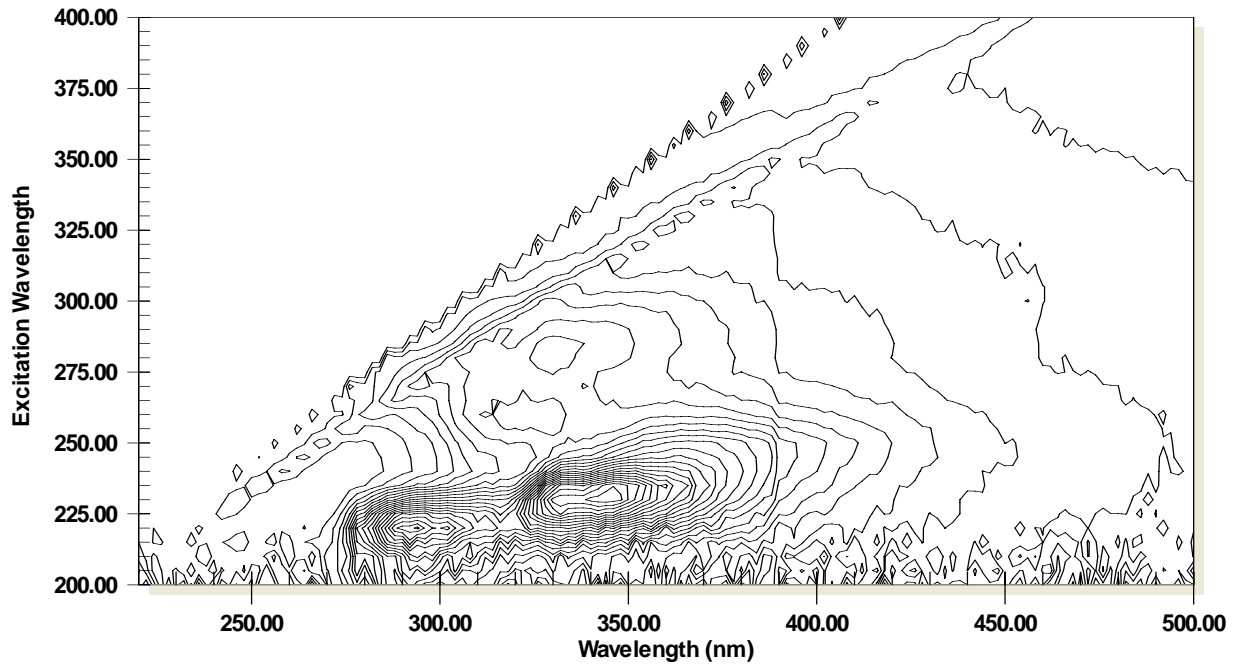
62Cy



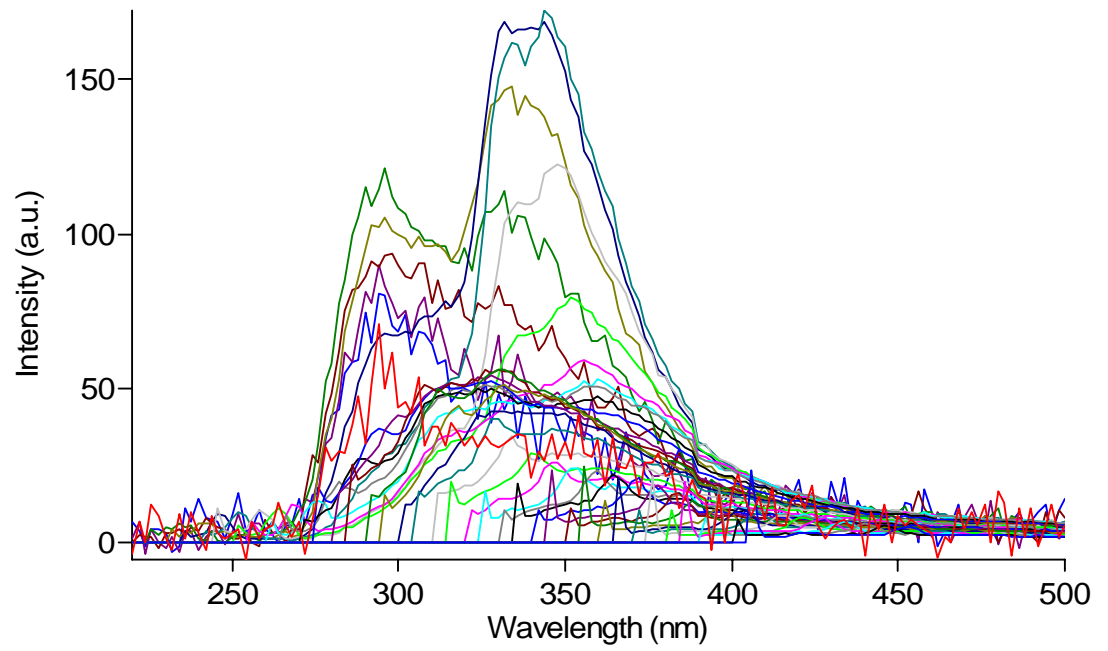
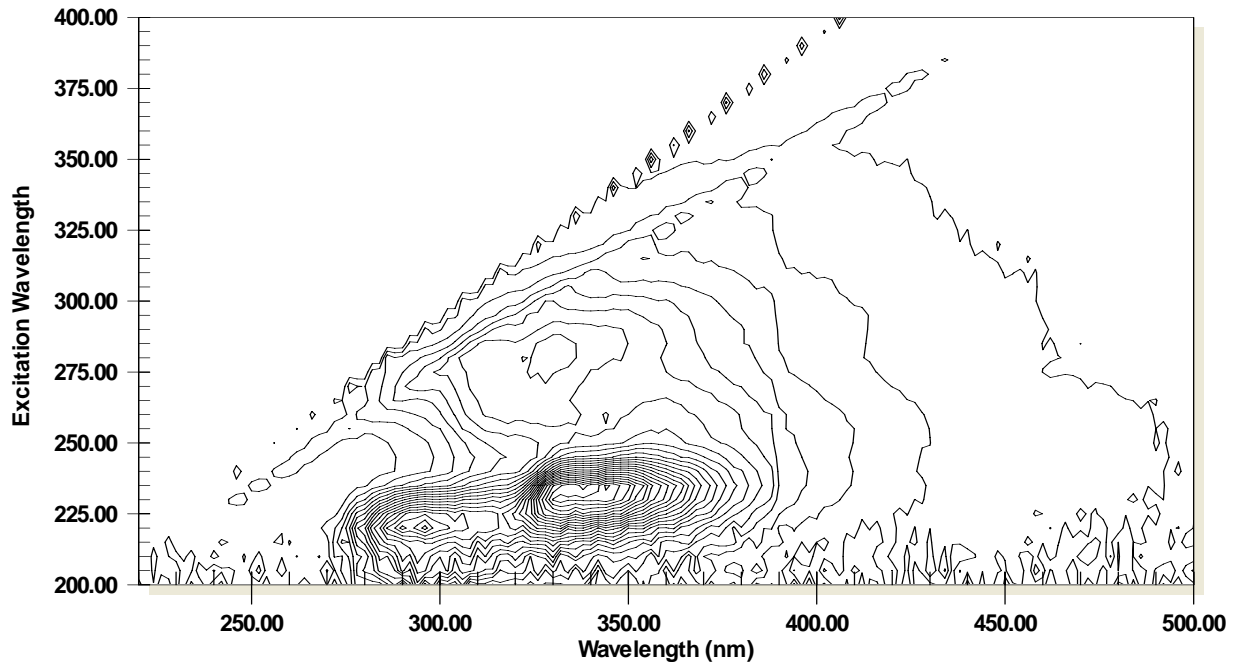
63By



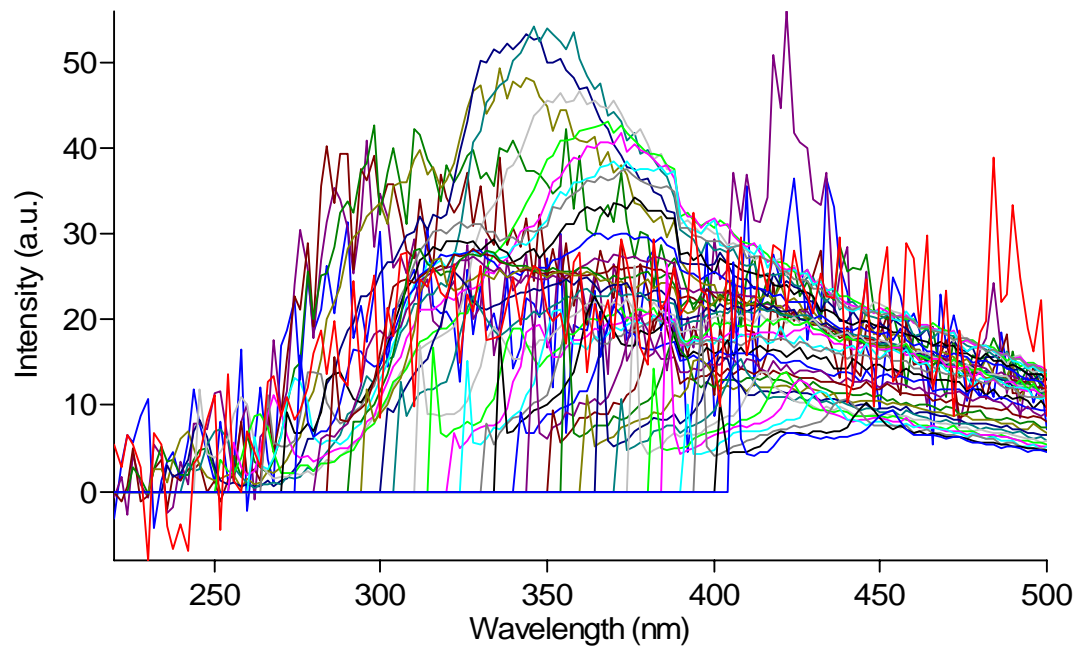
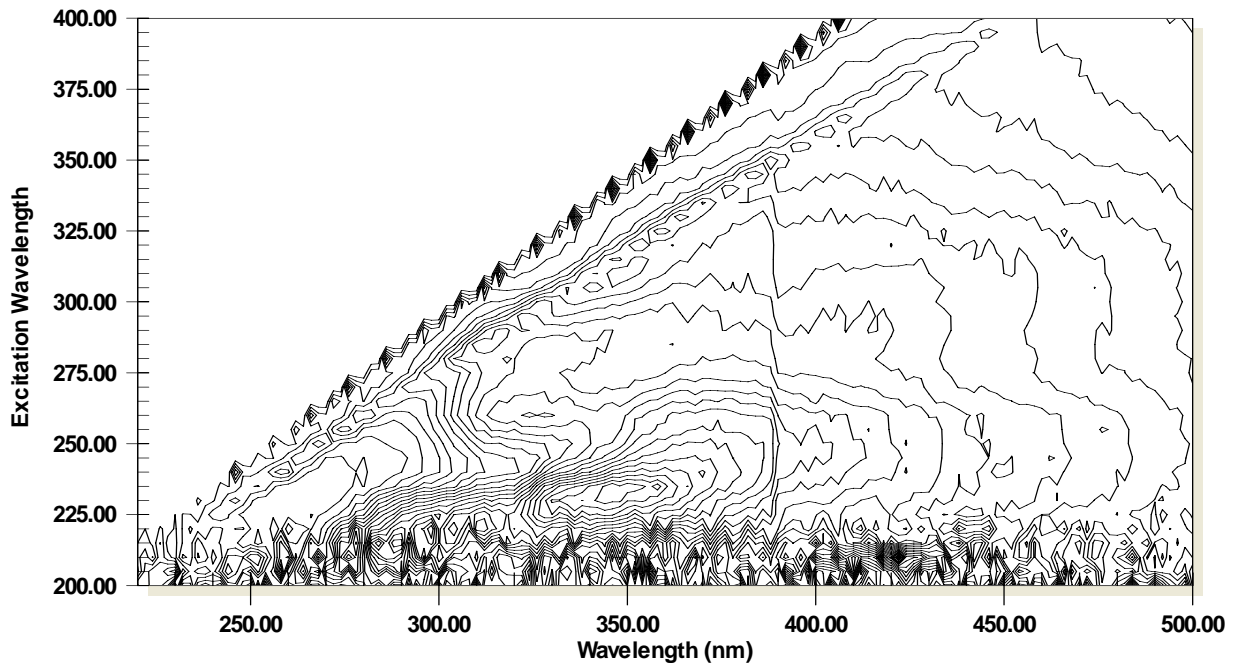
64Ay



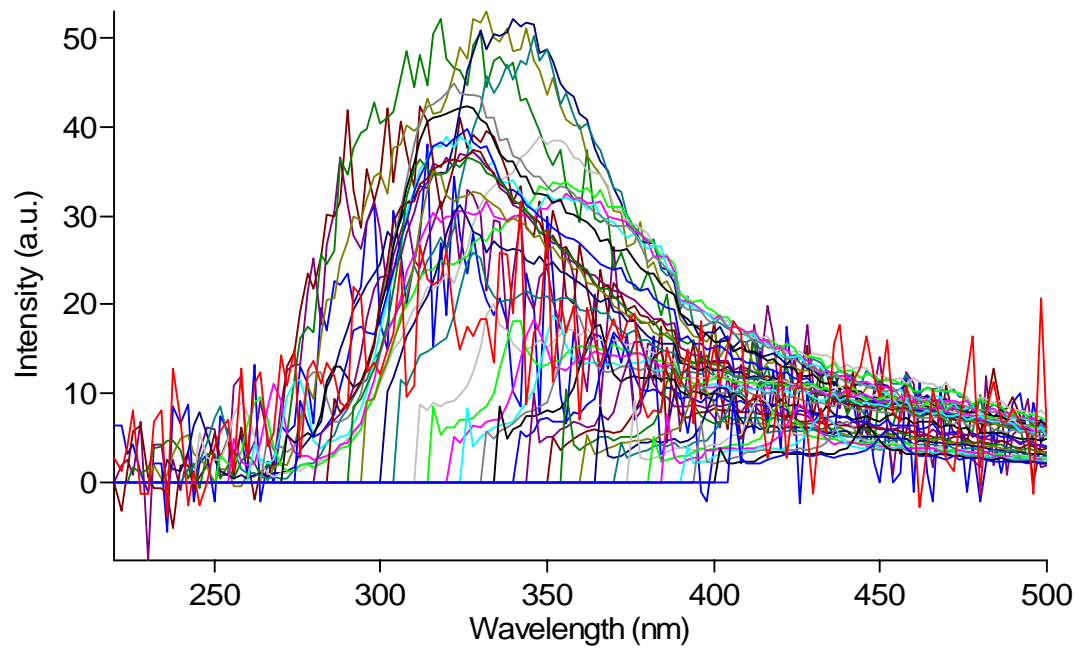
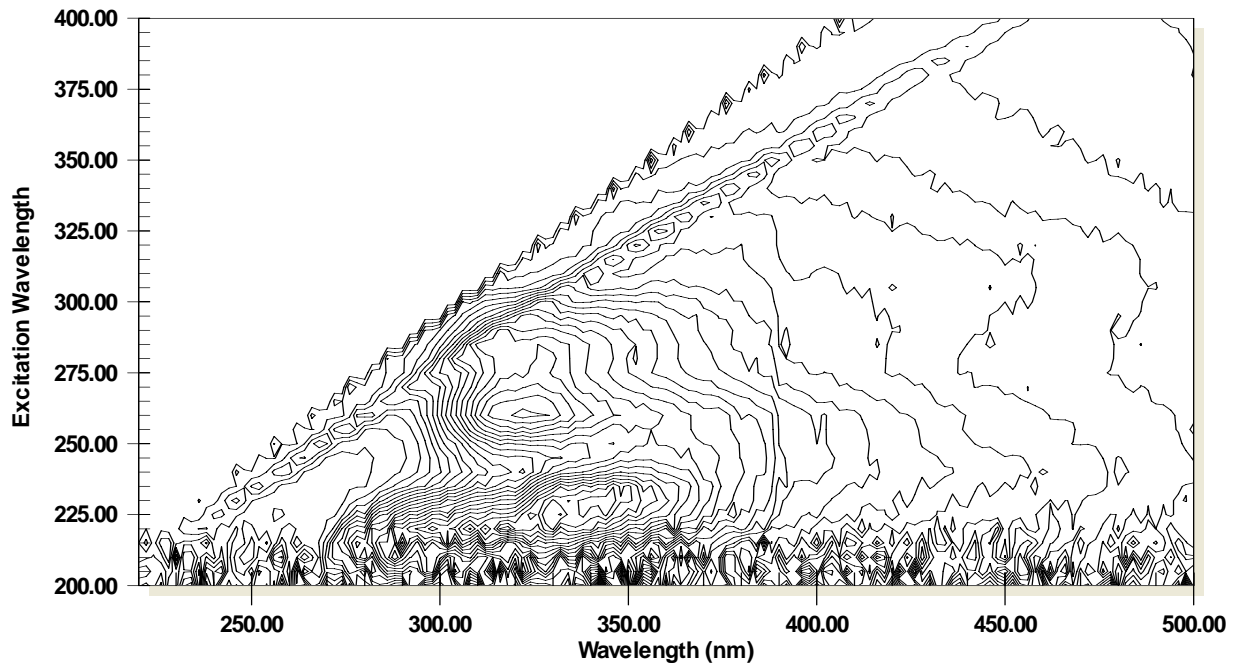
65By



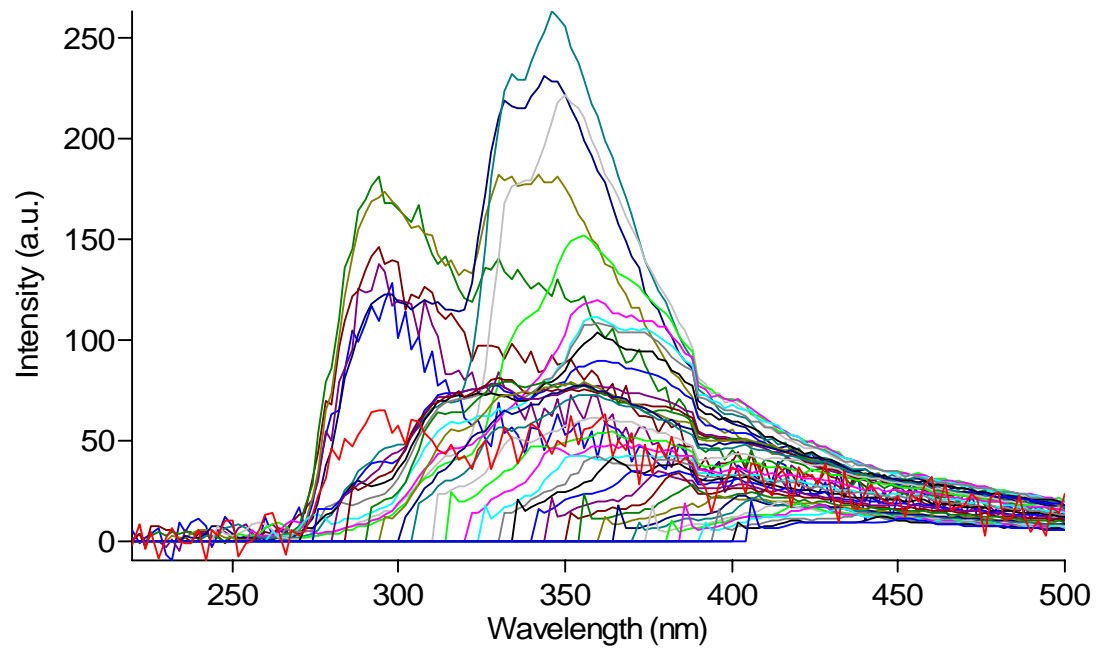
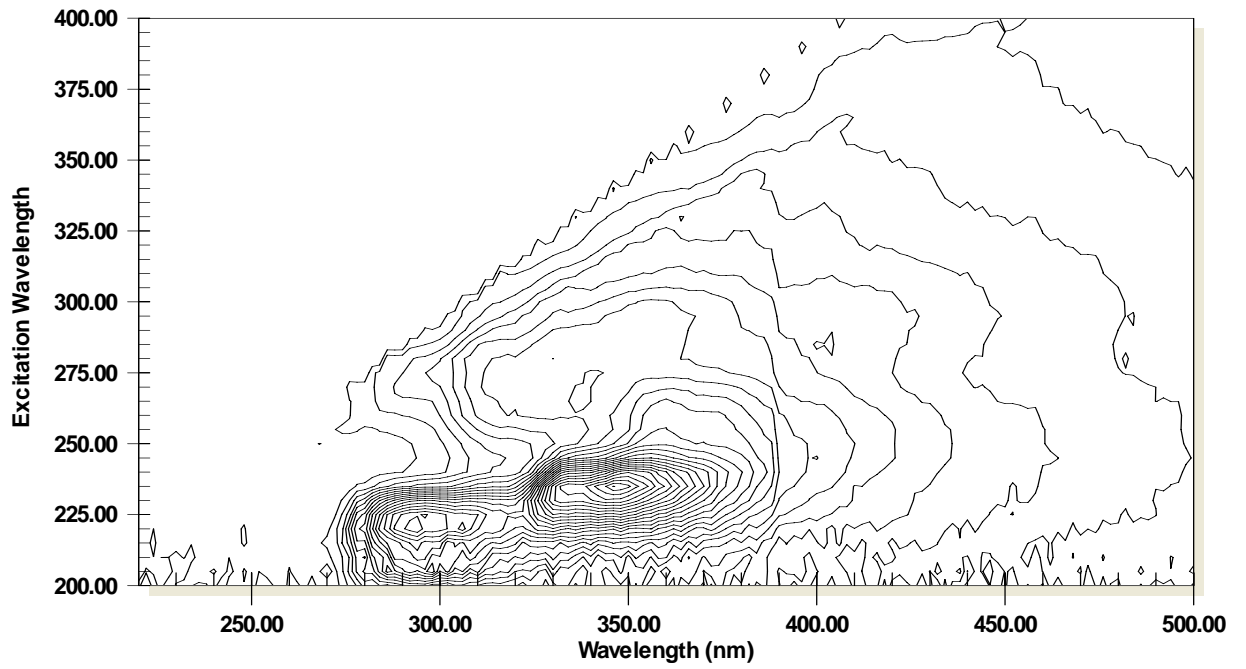
71By



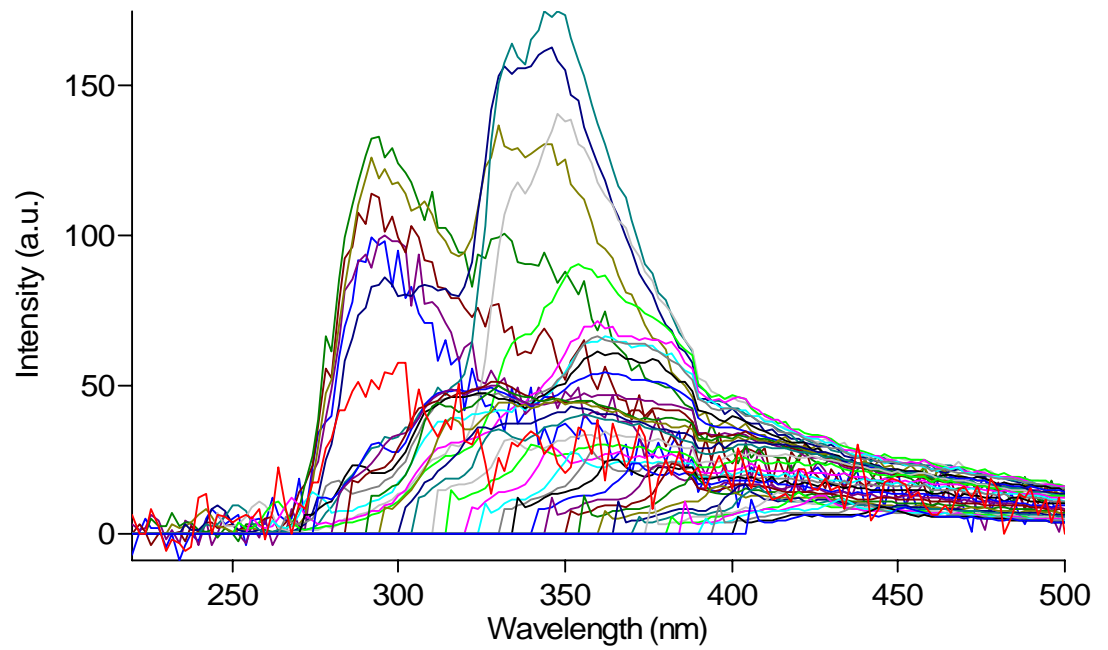
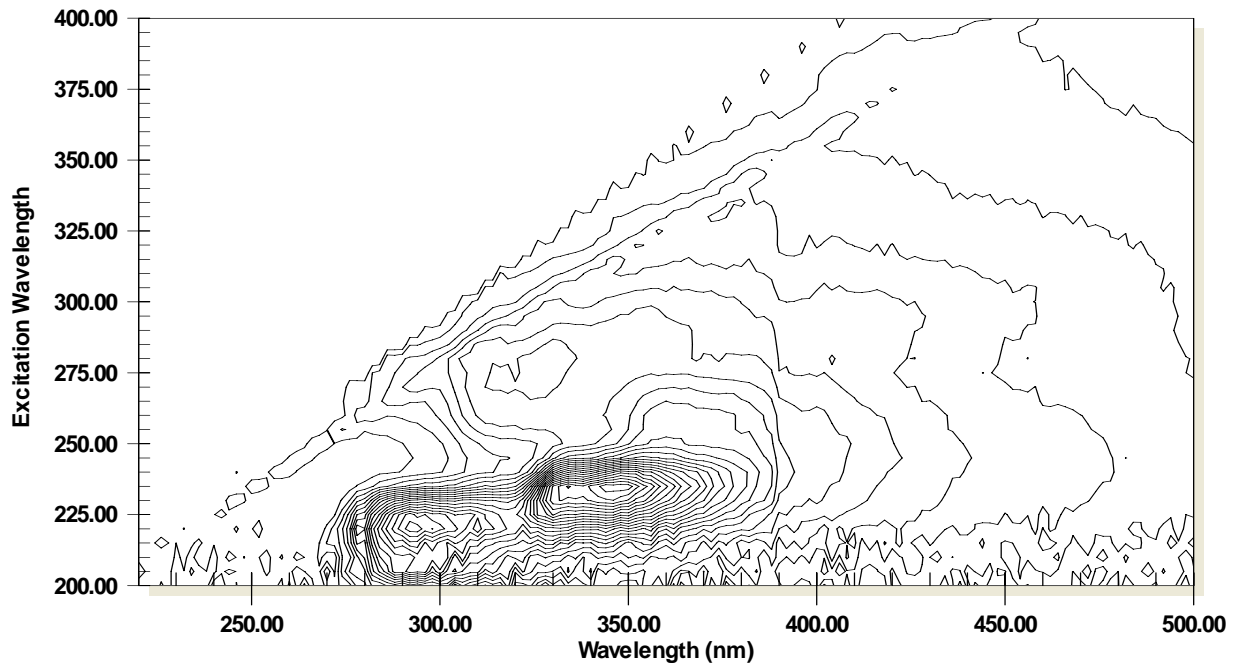
72Ay



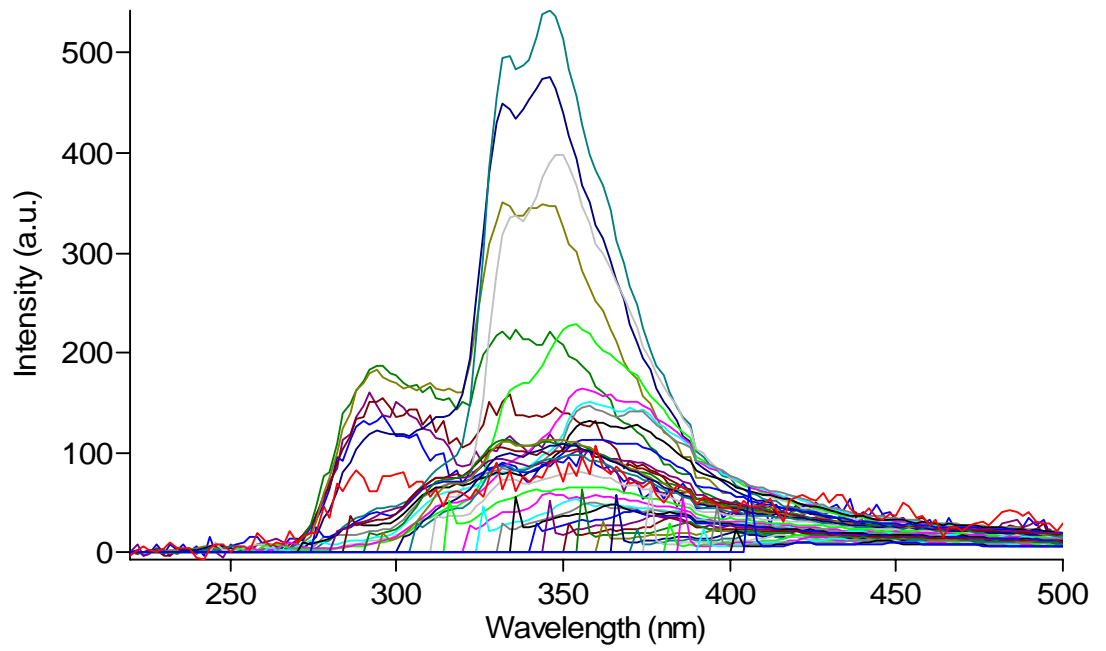
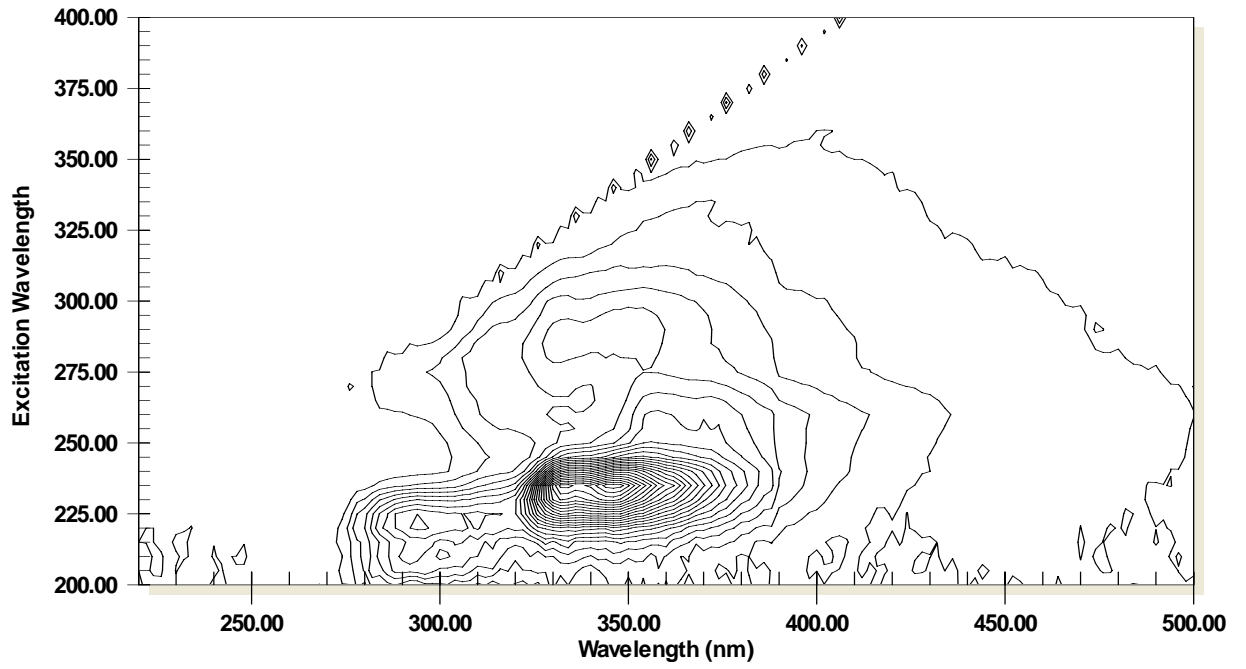
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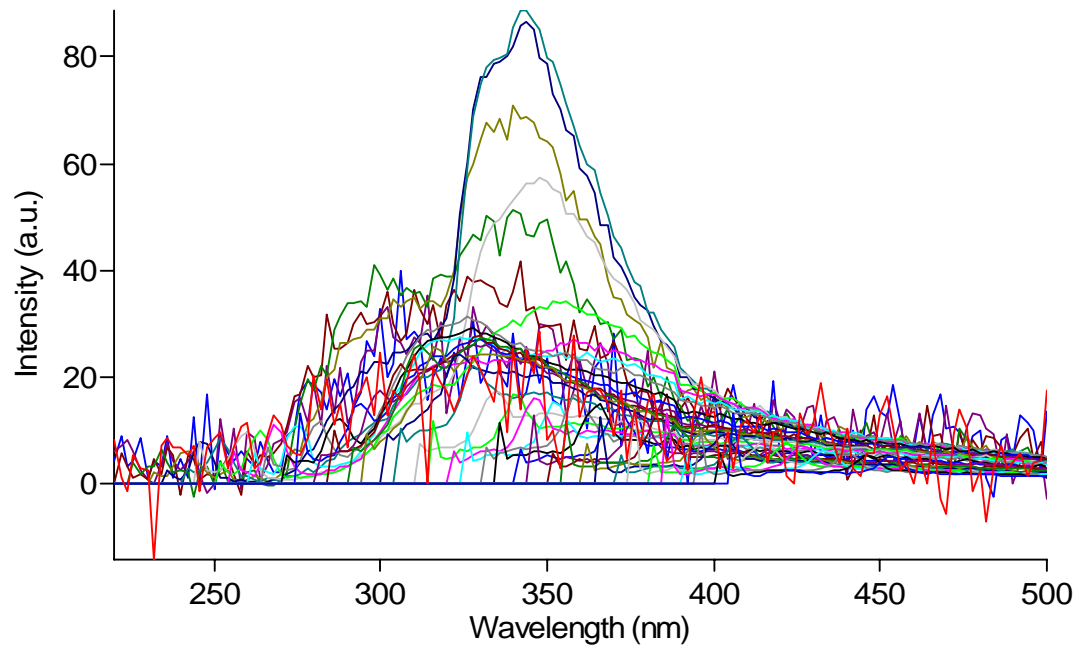
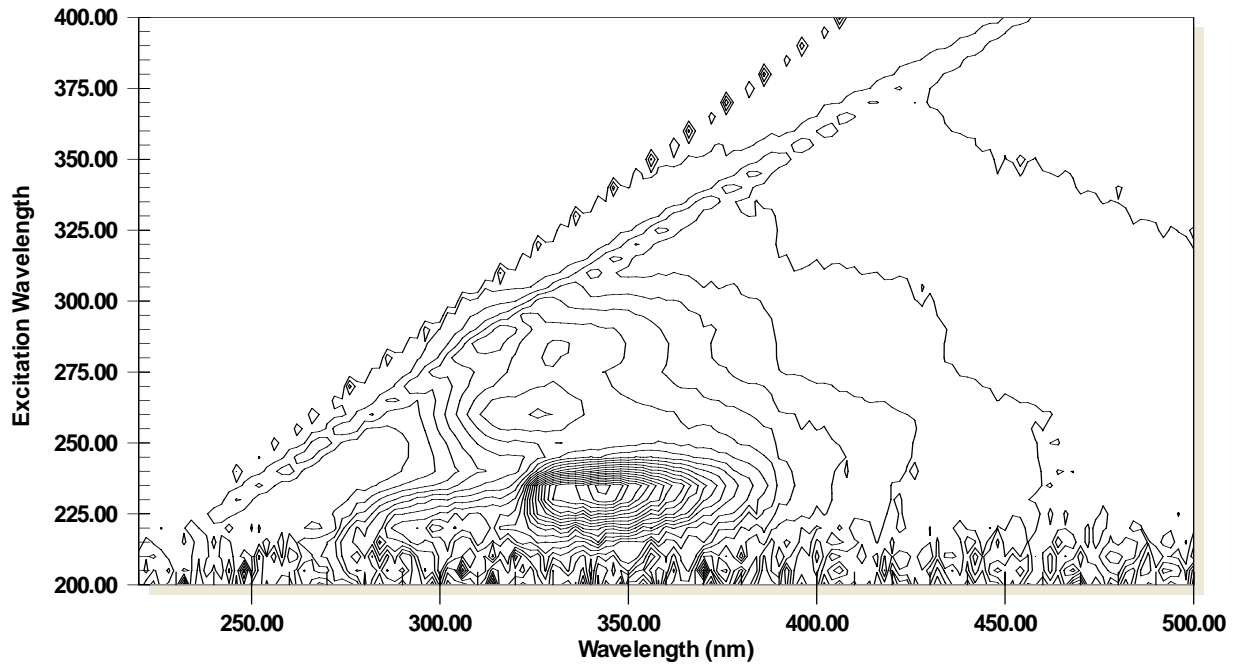
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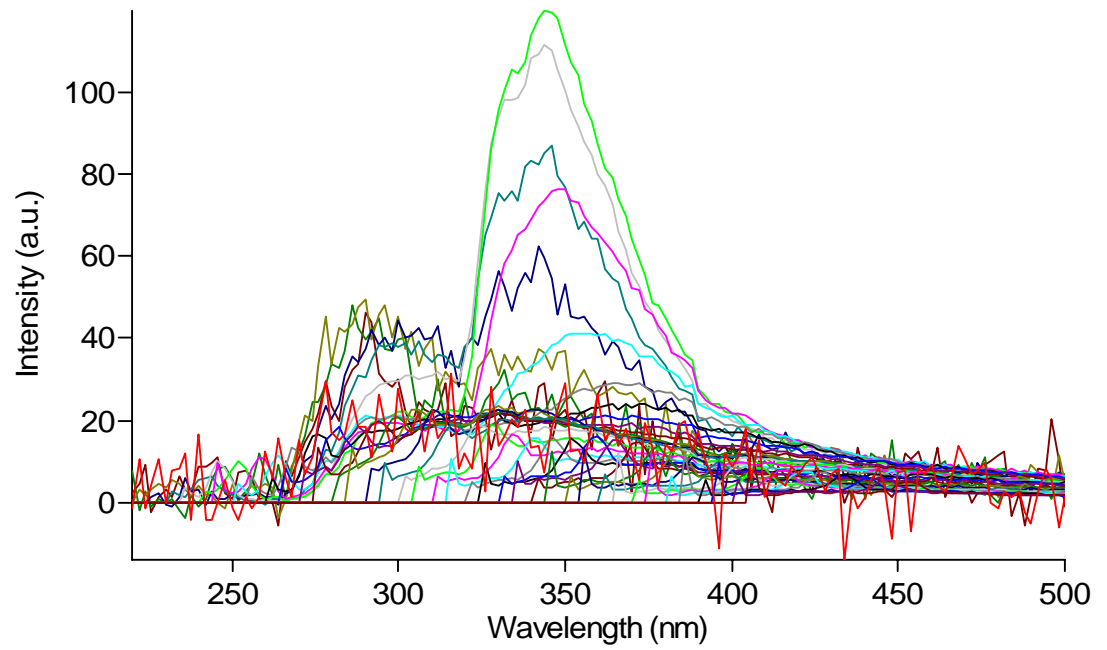
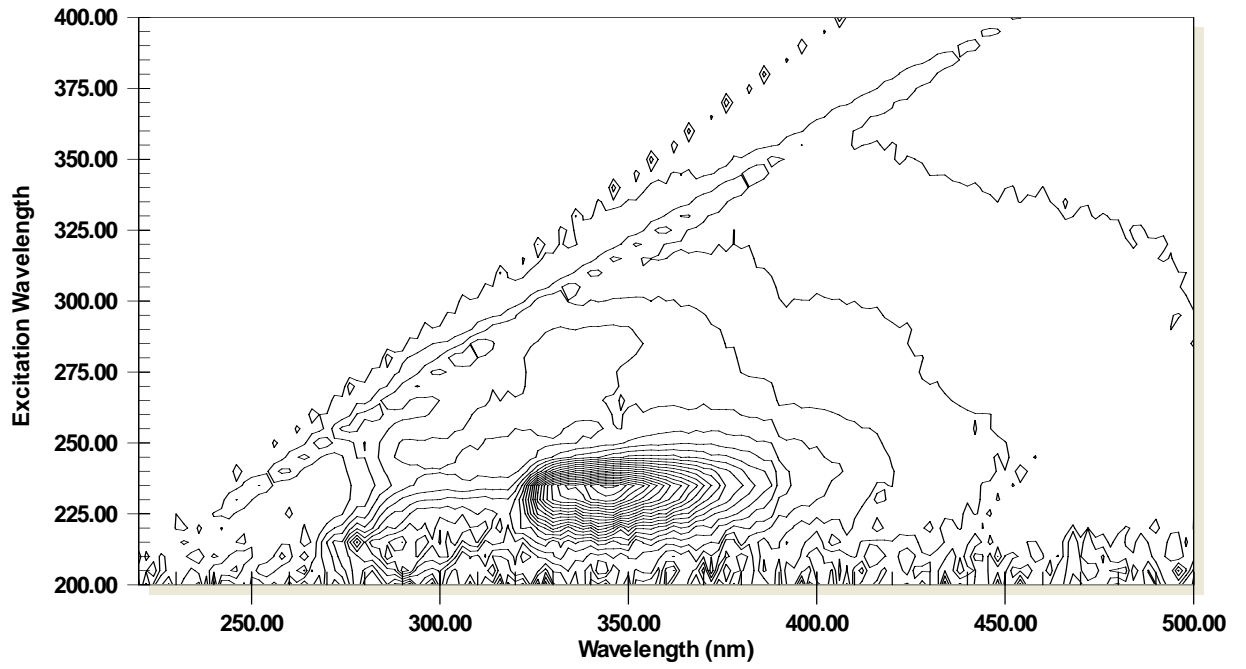
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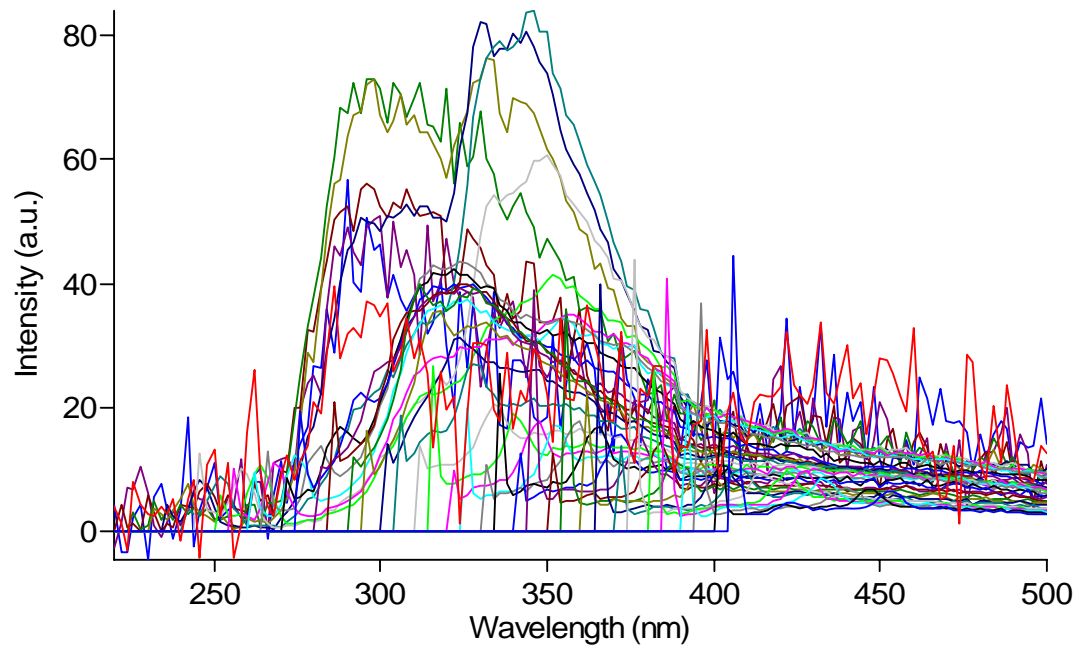
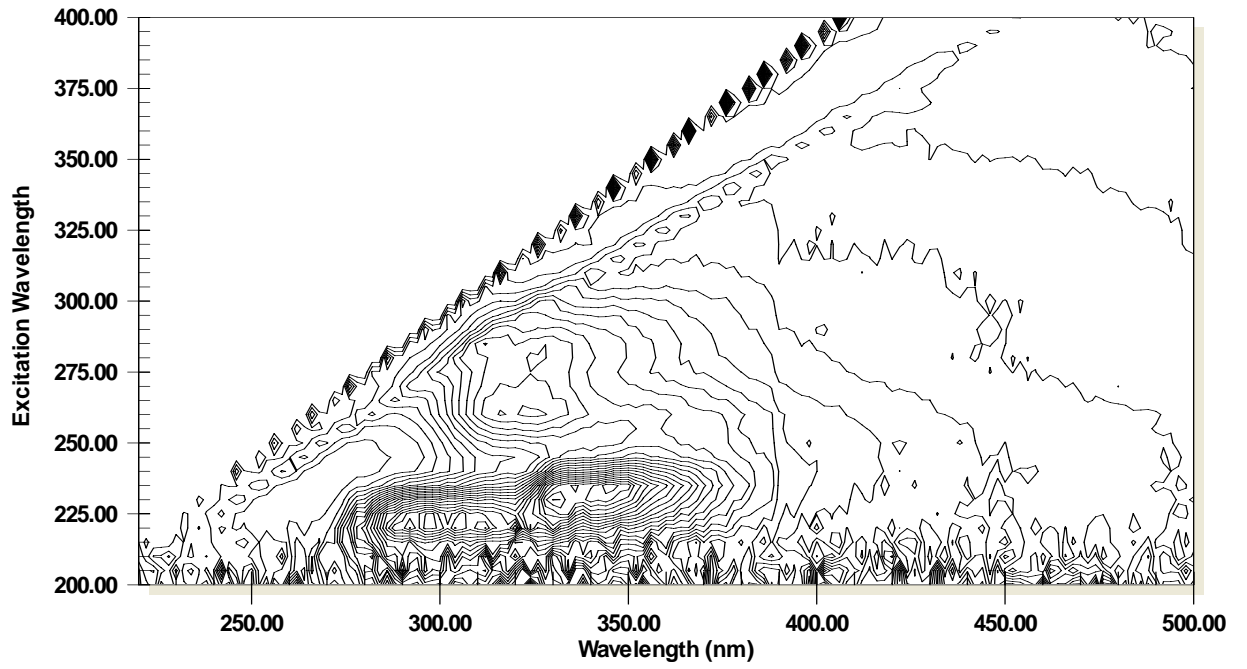
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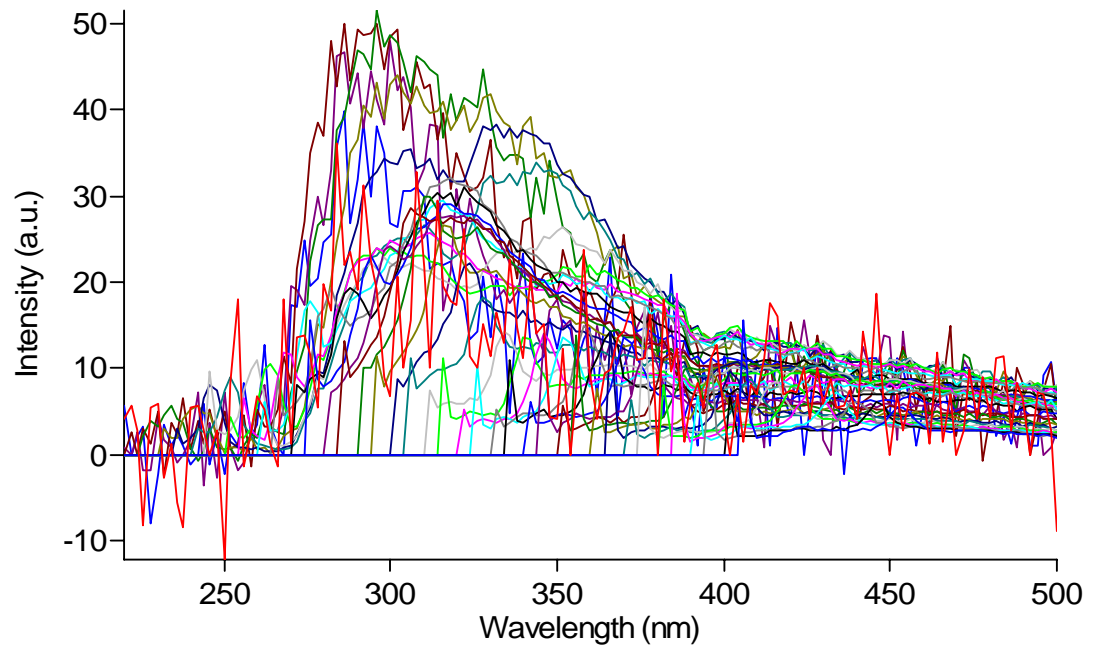
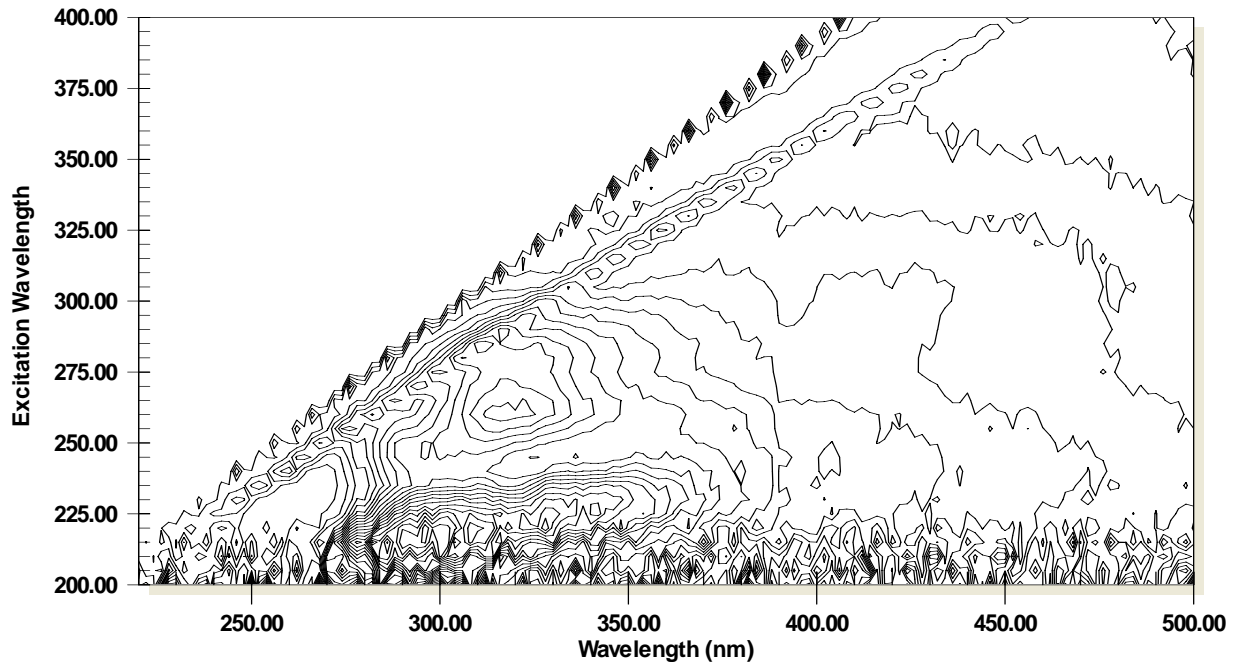
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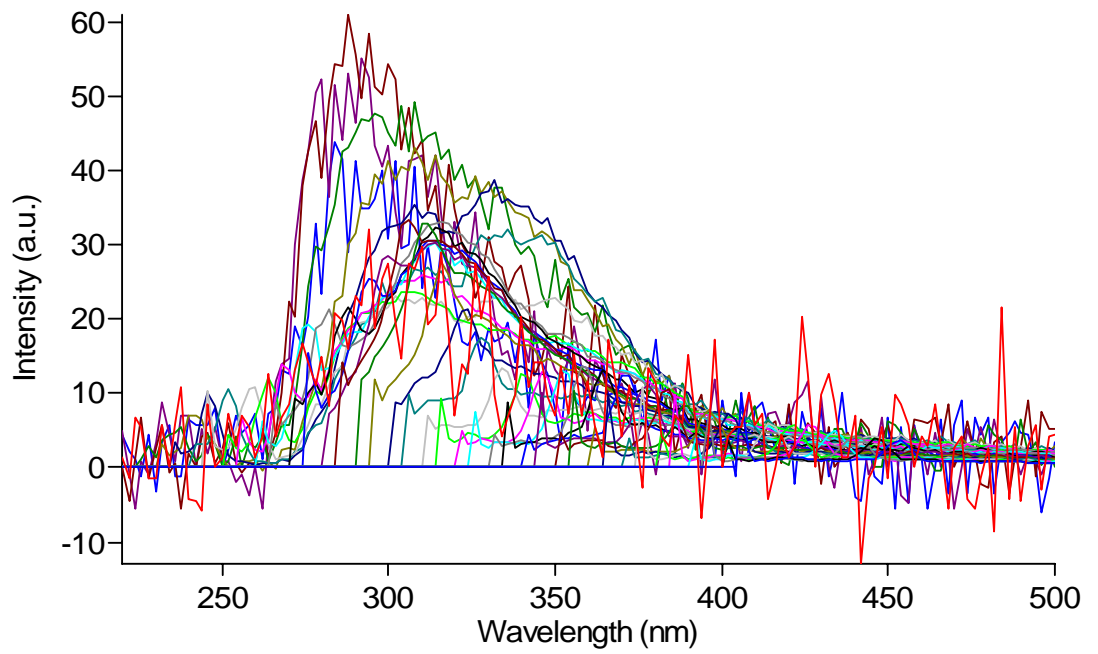
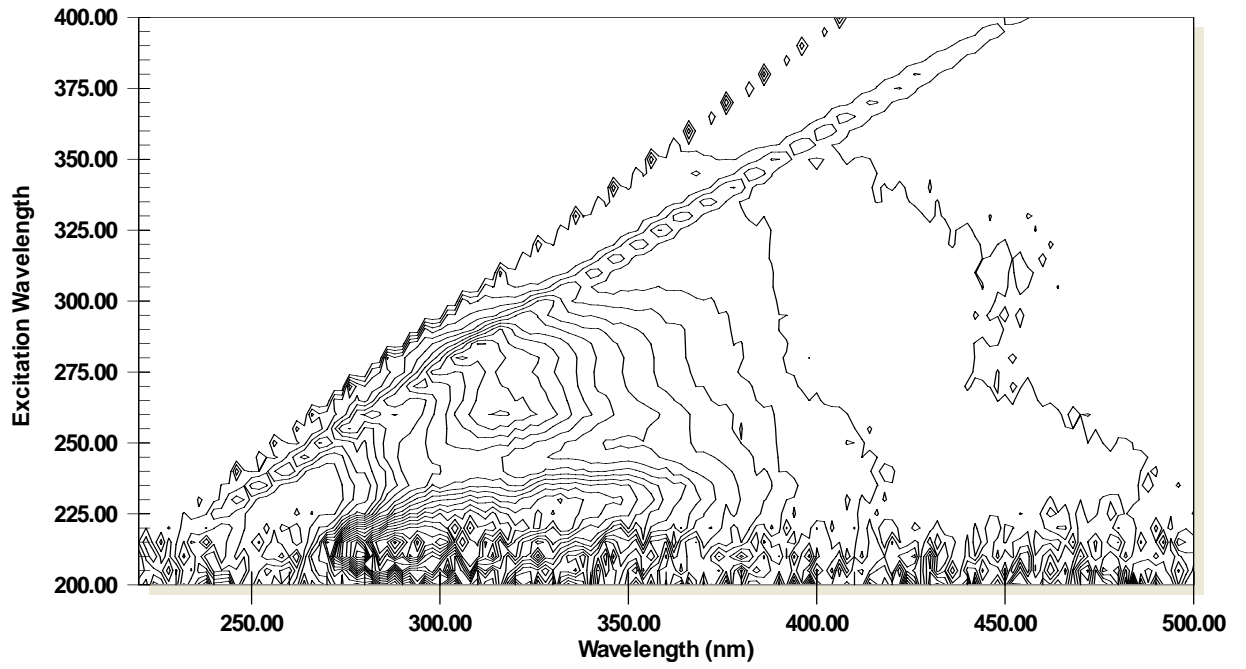
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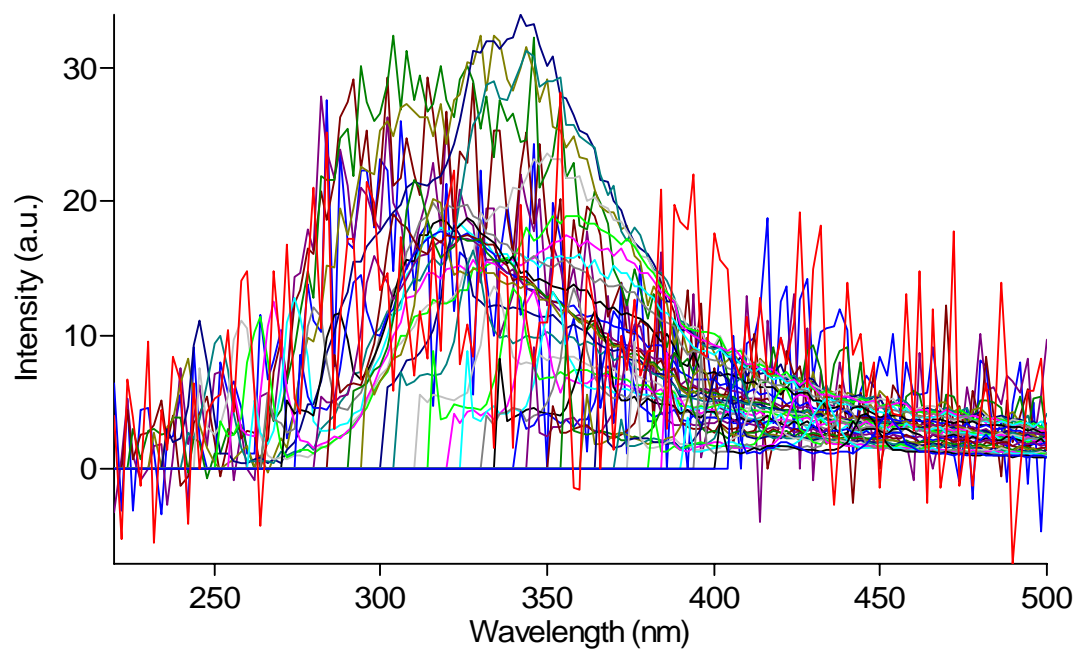
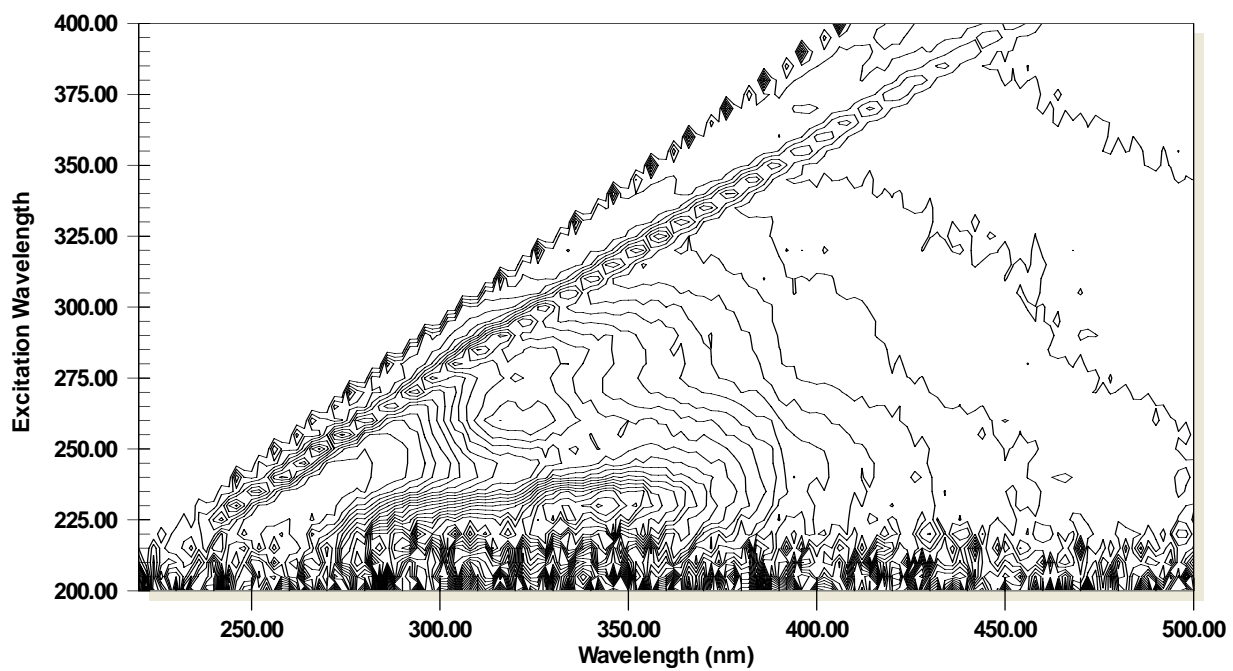
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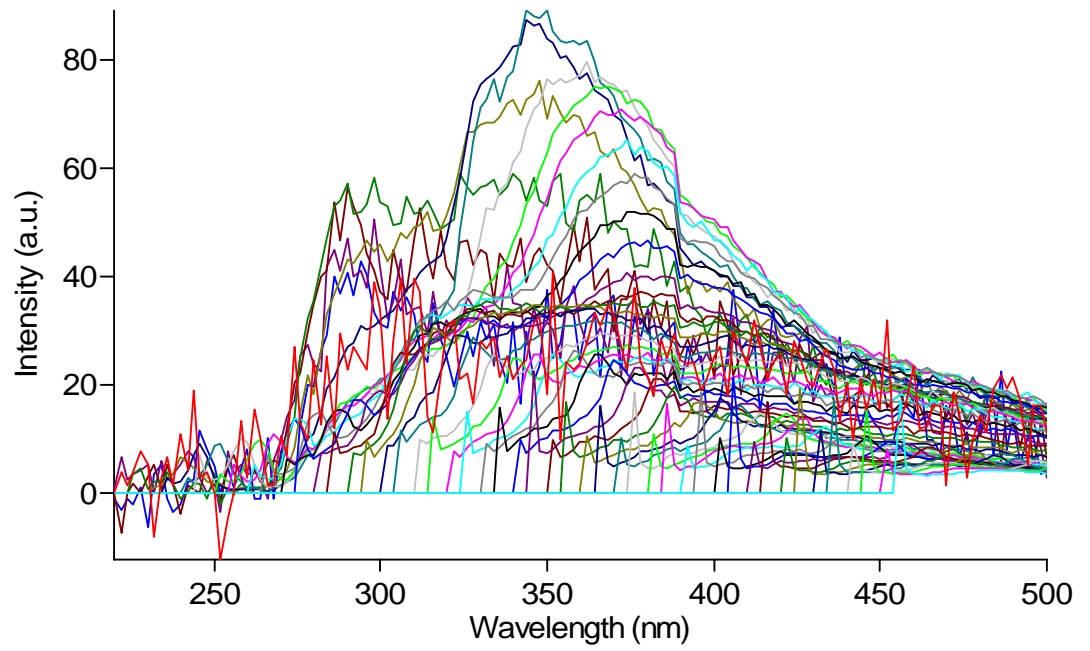
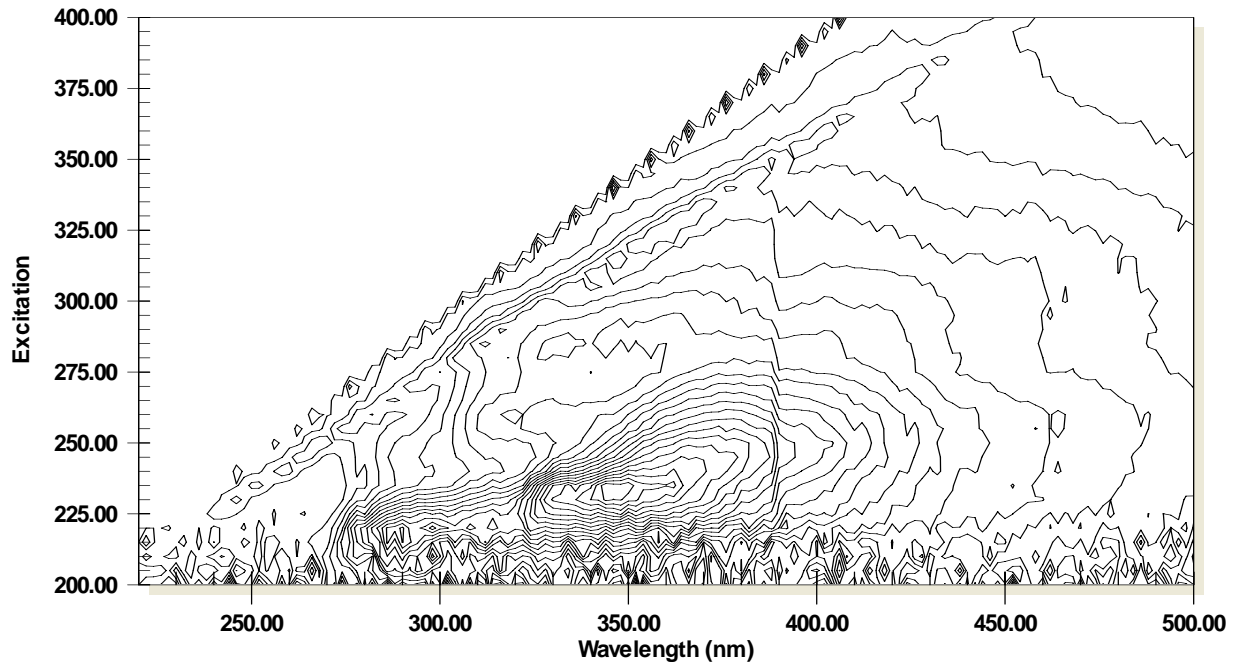
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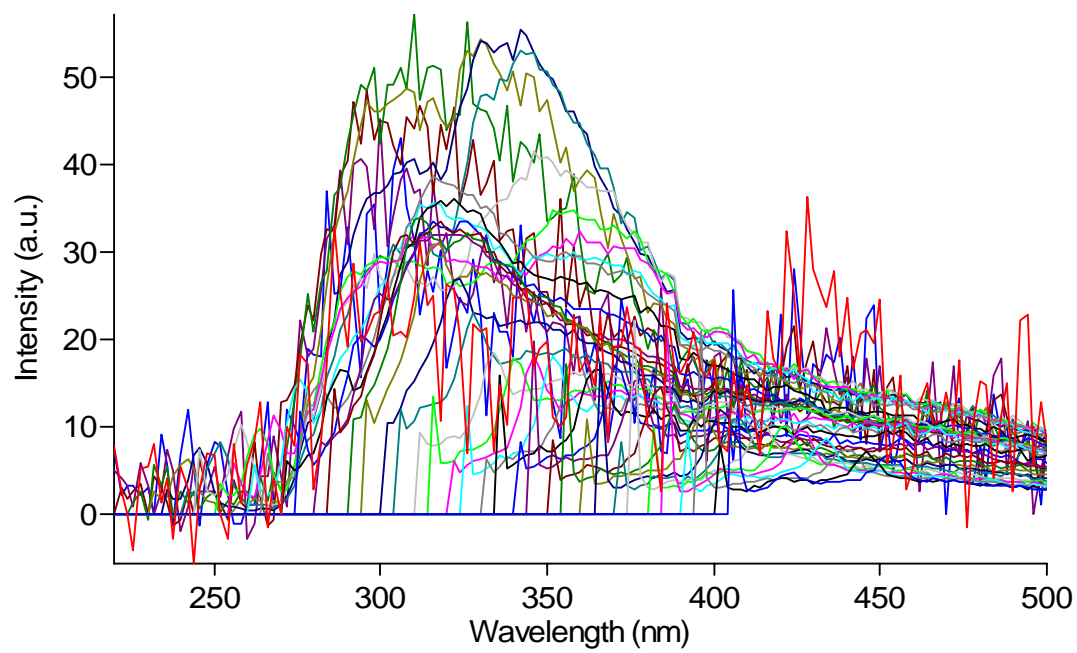
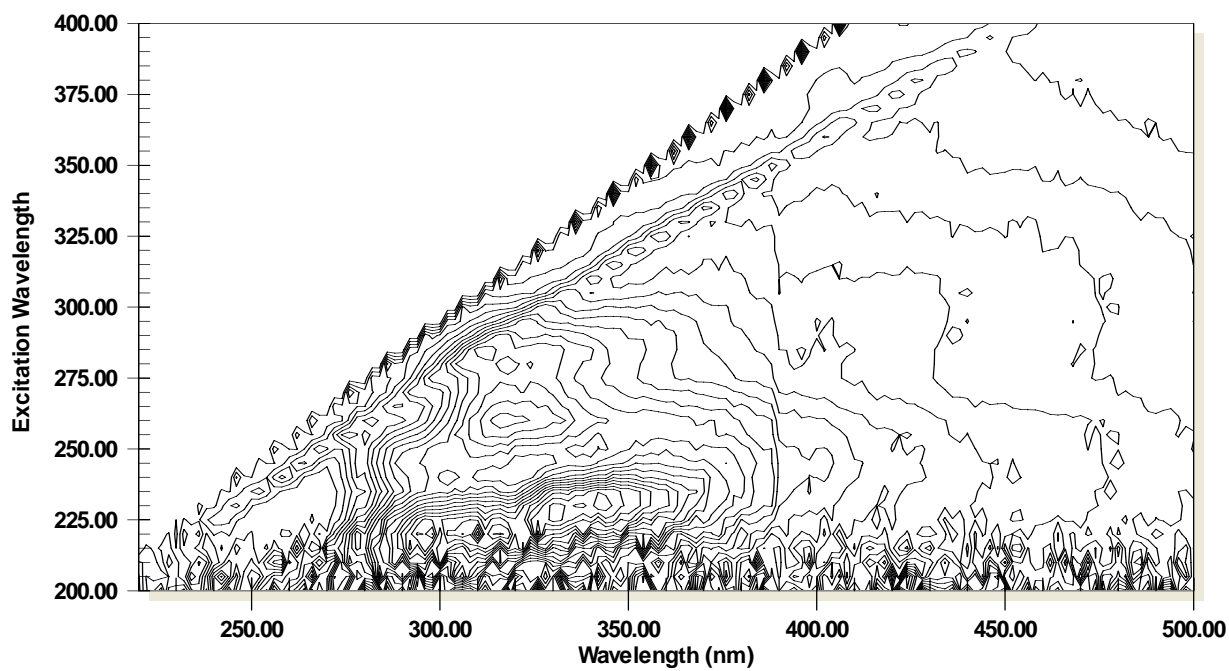
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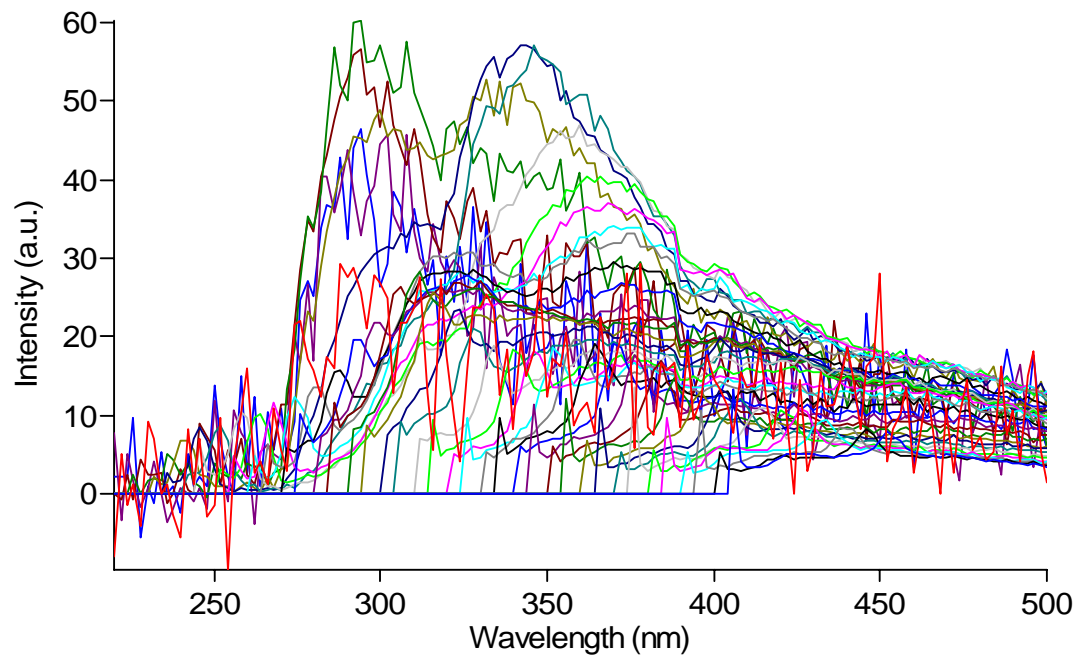
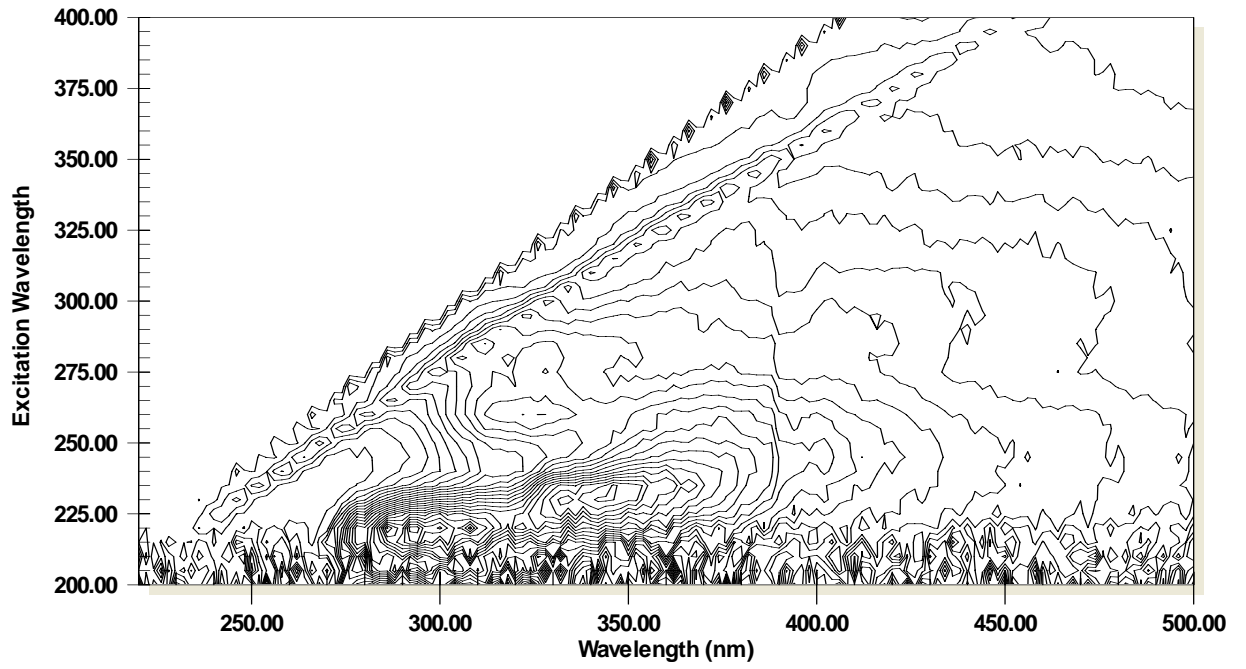
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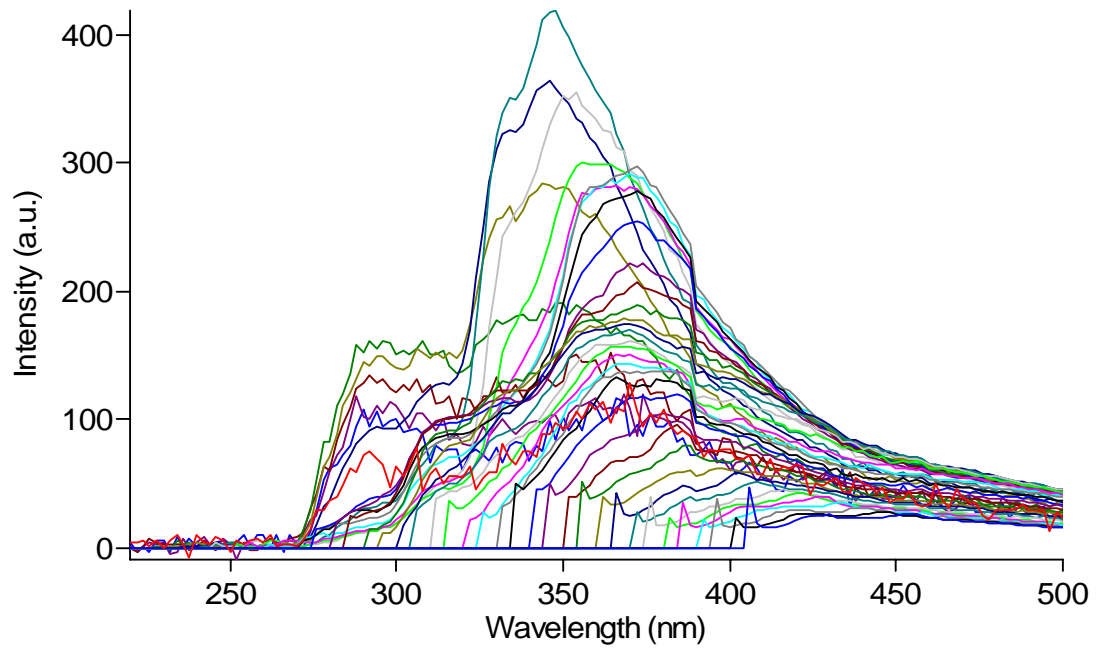
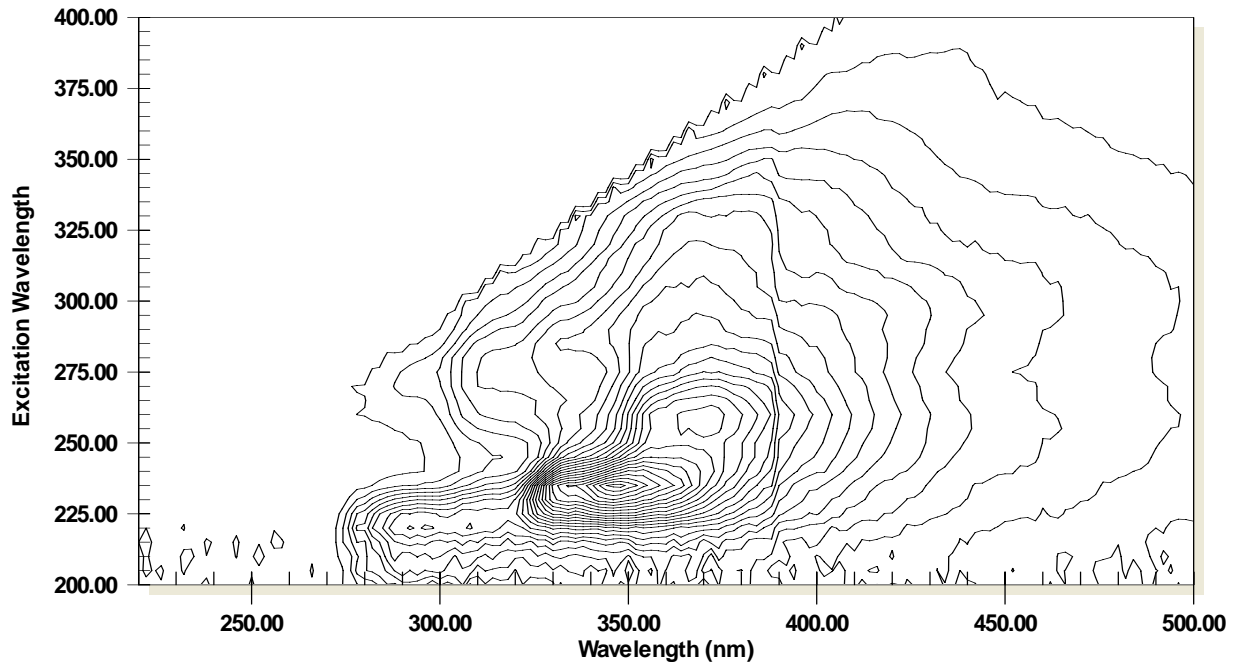
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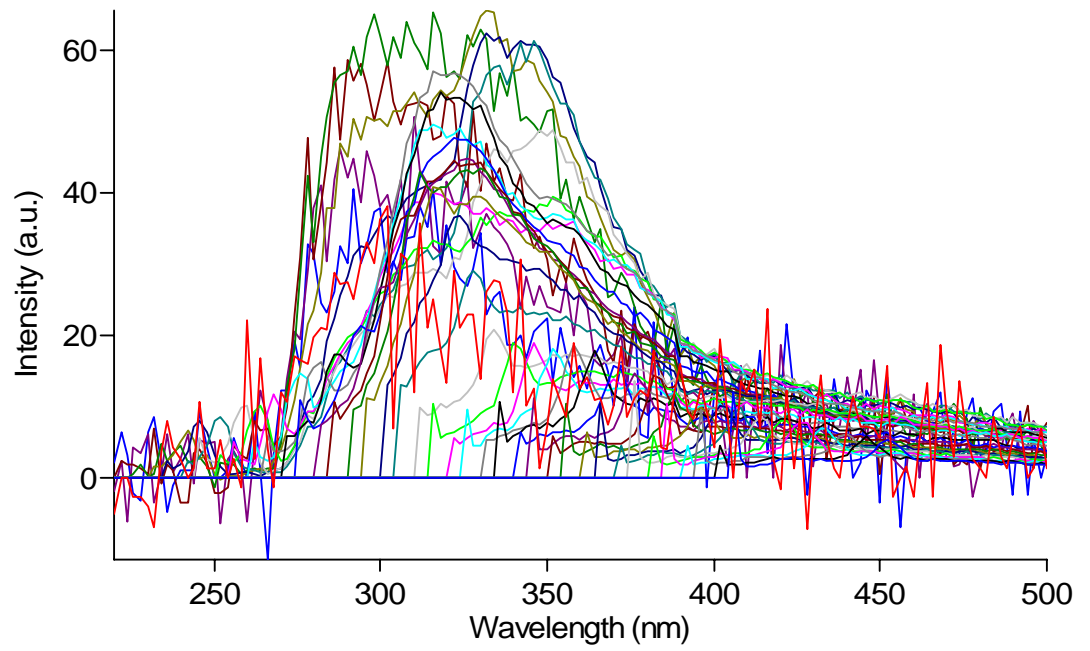
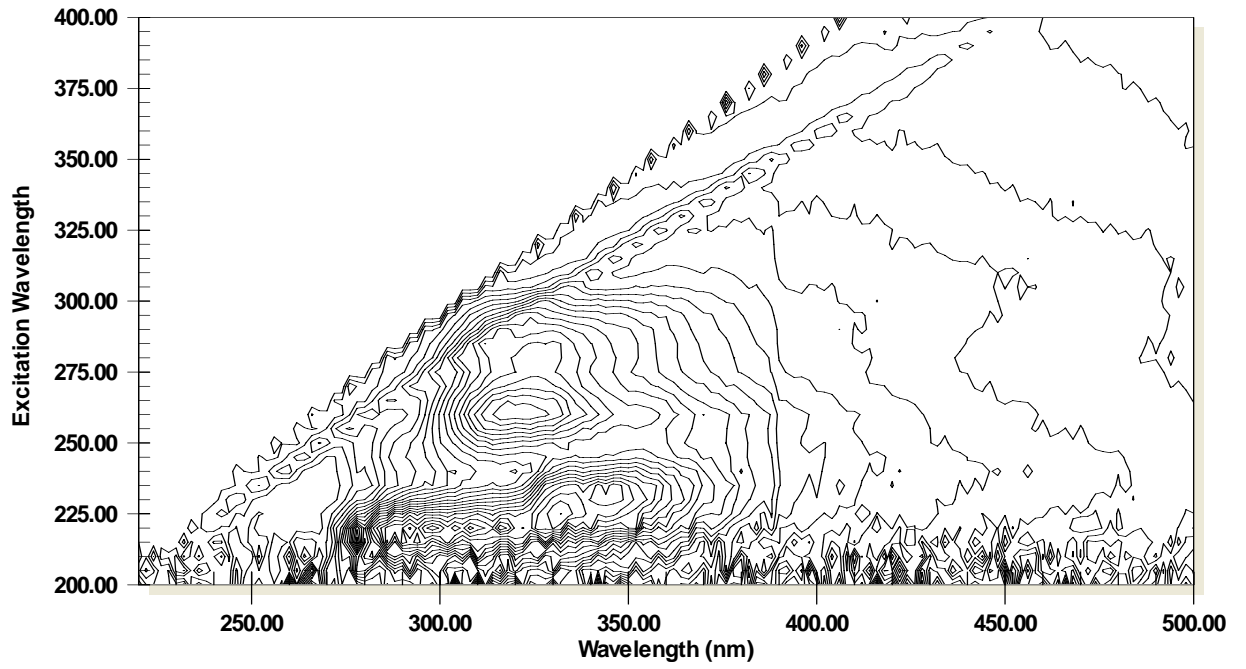
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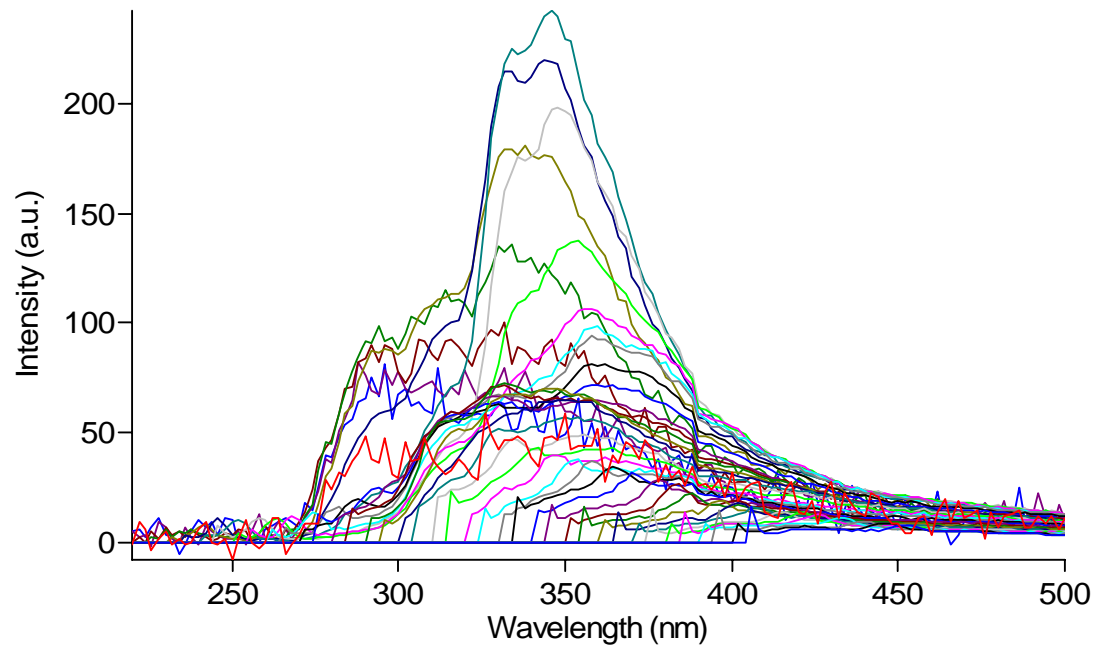
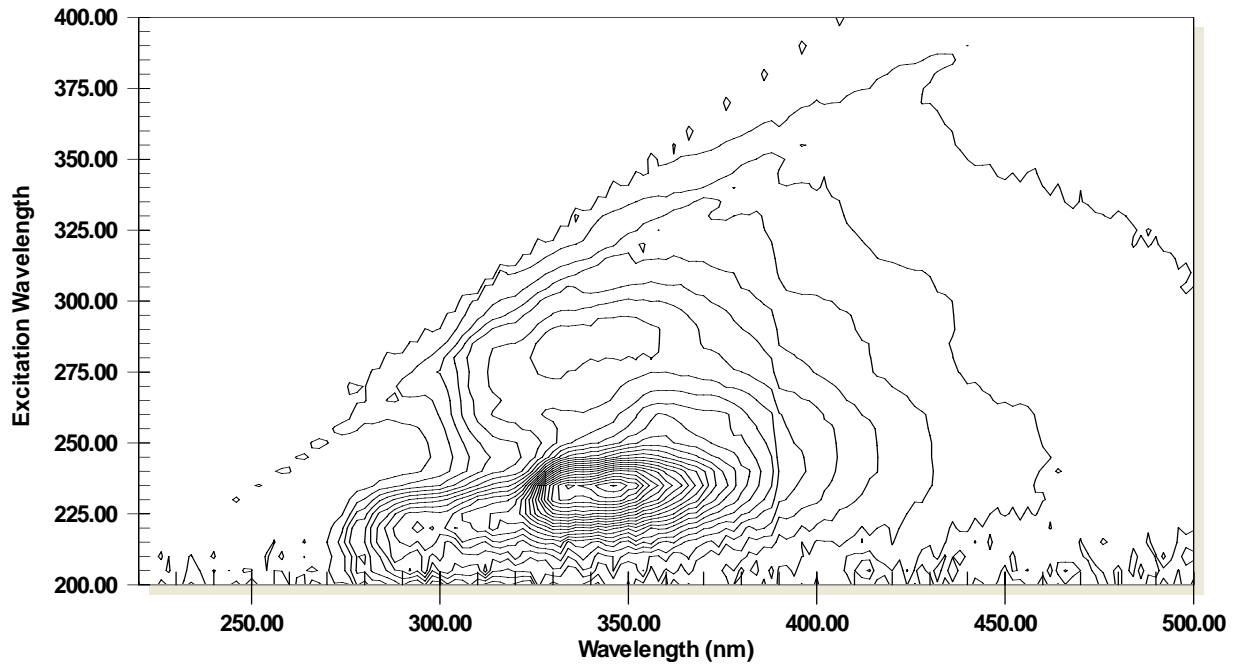
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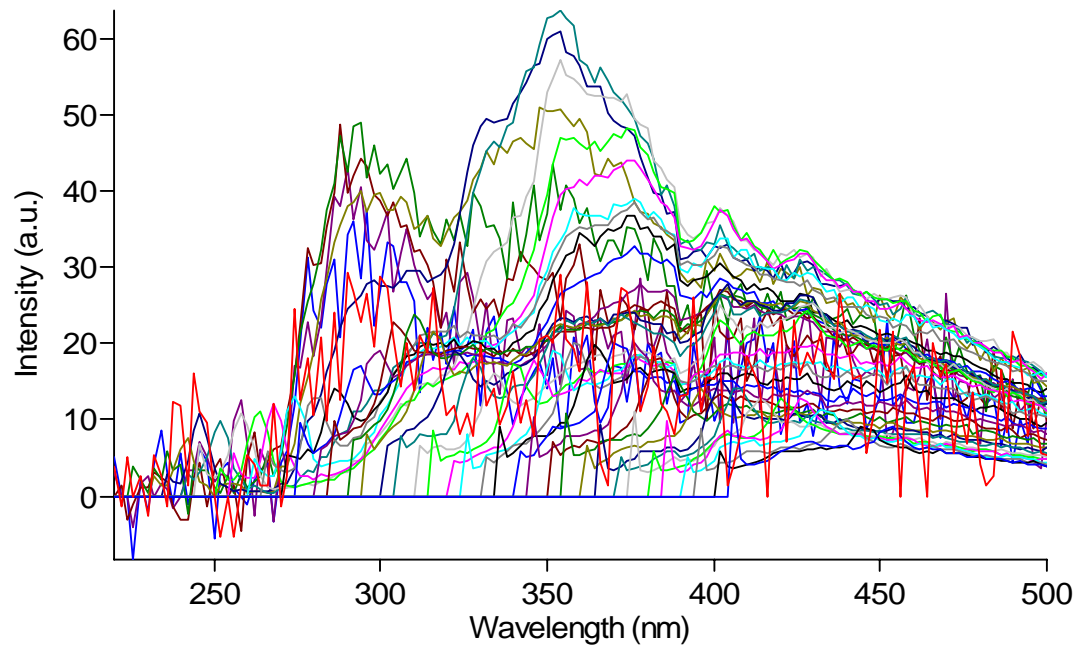
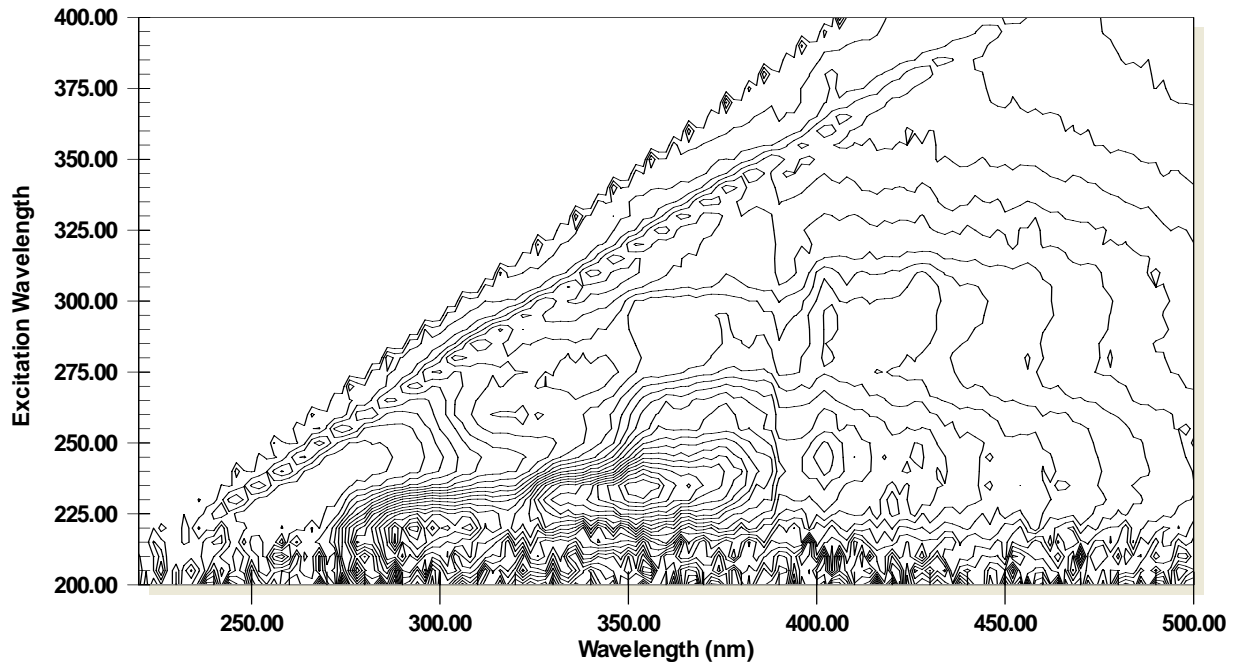
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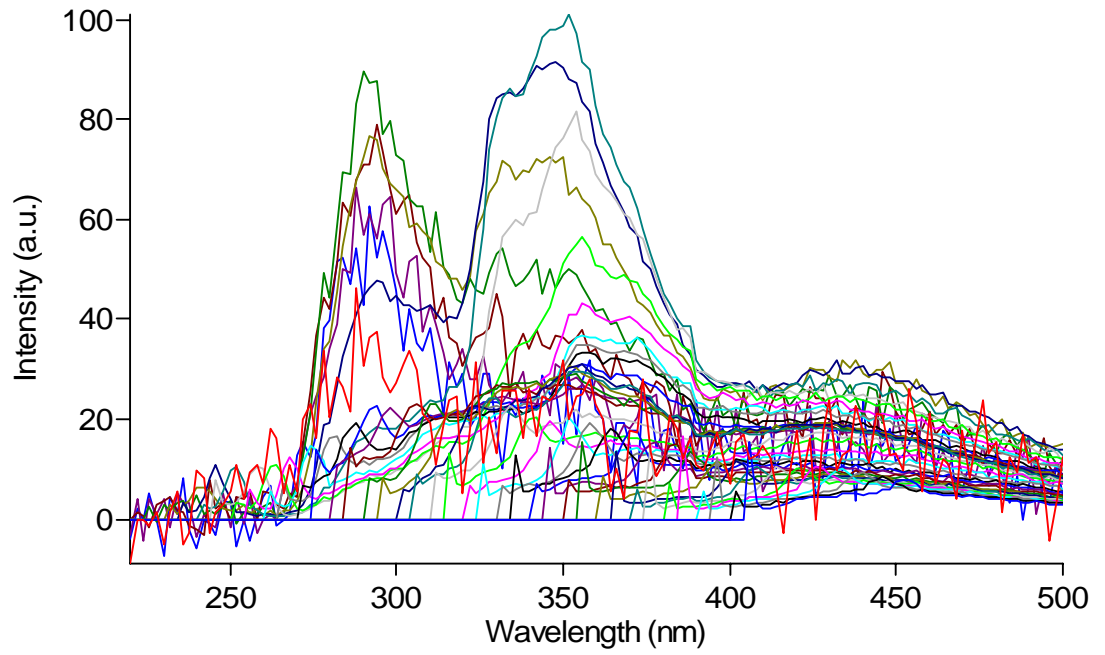
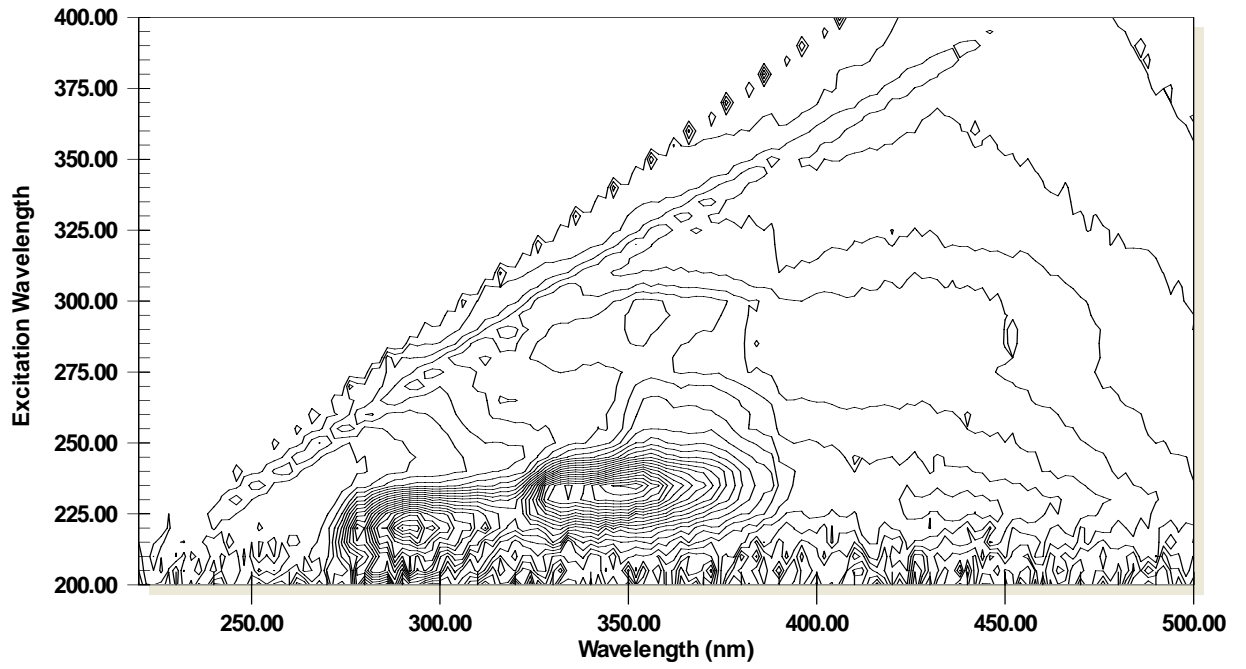
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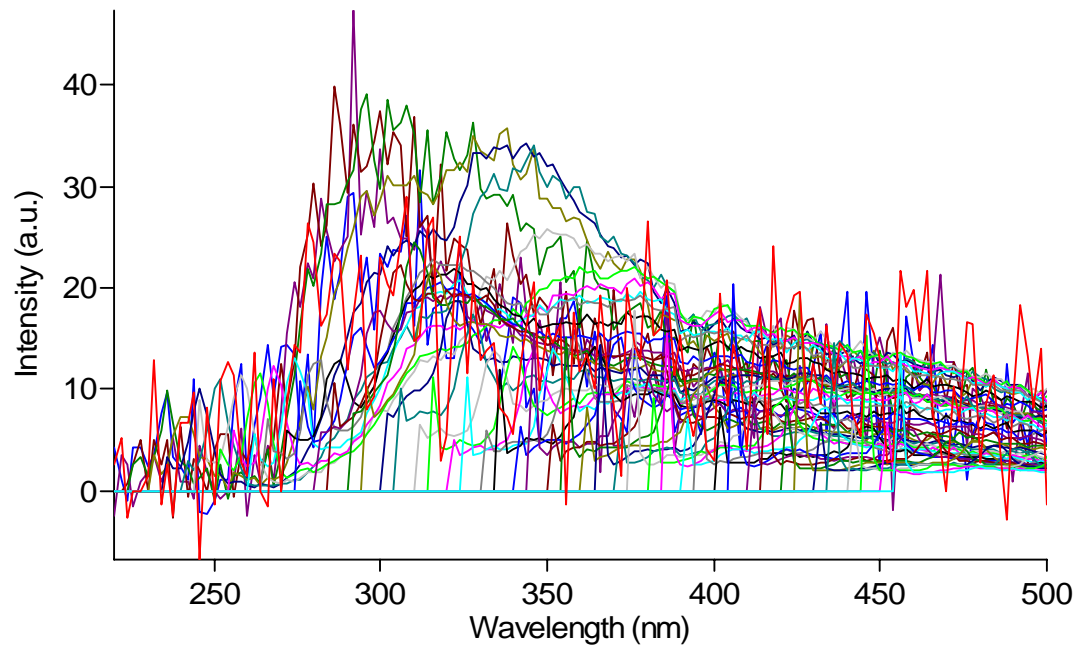
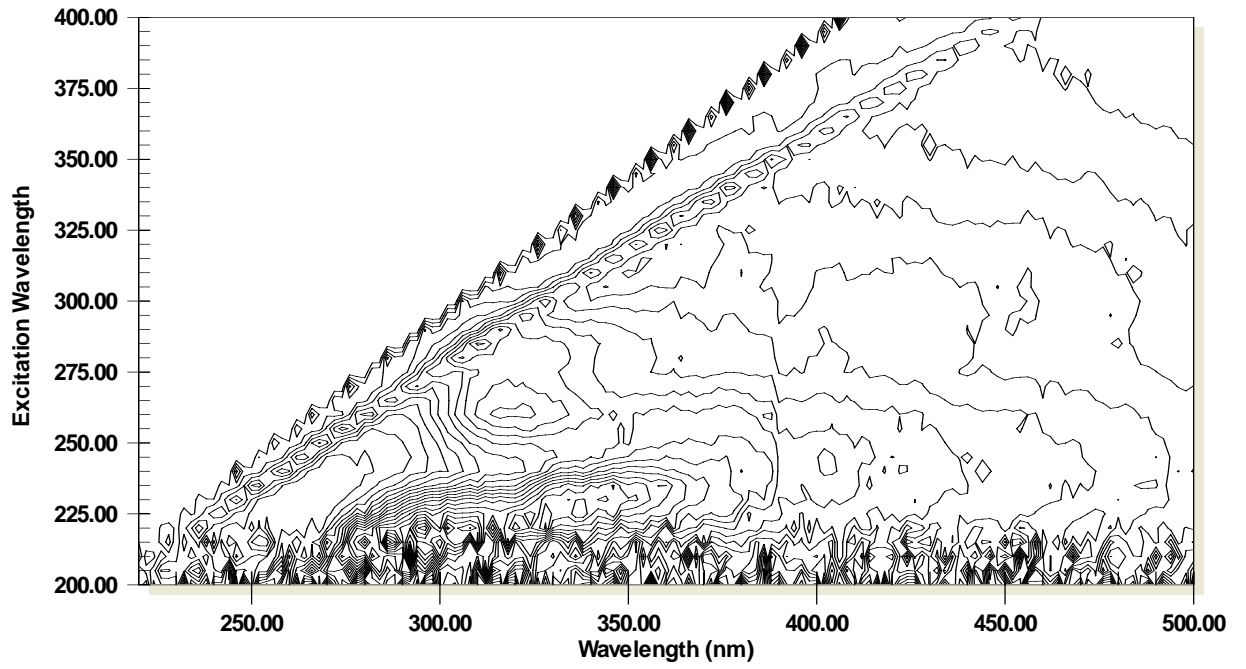
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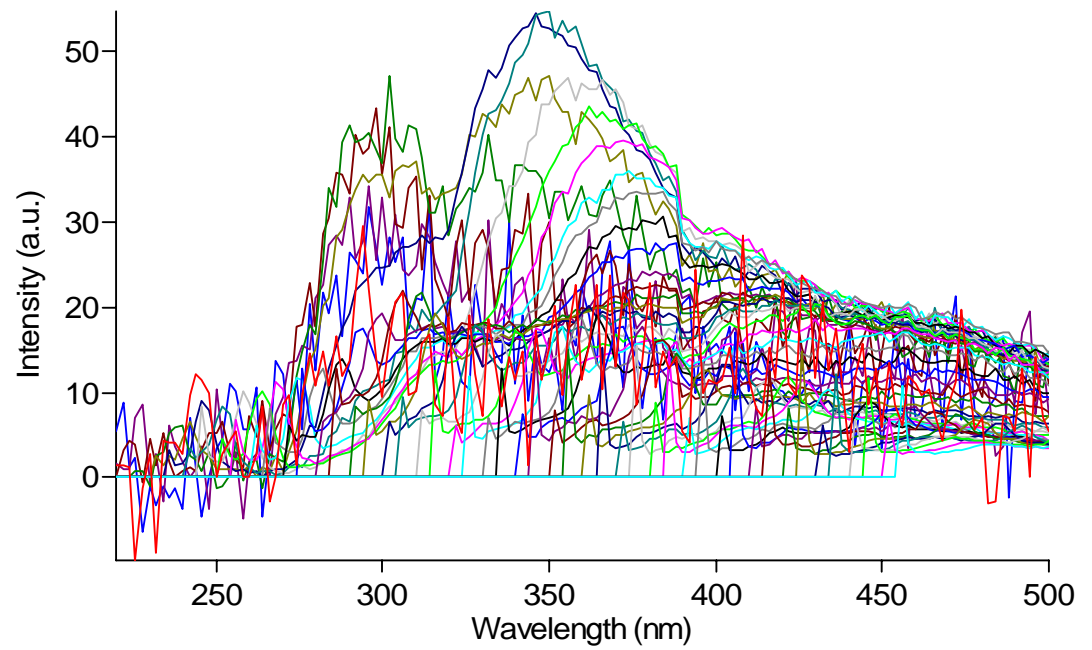
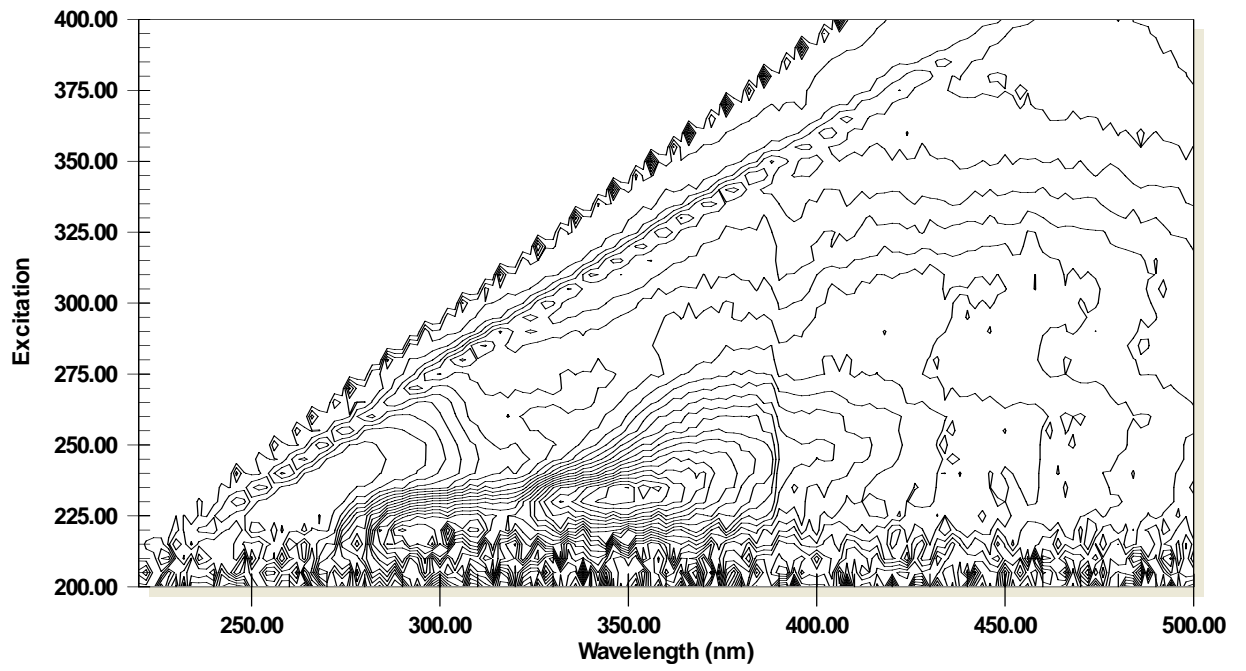
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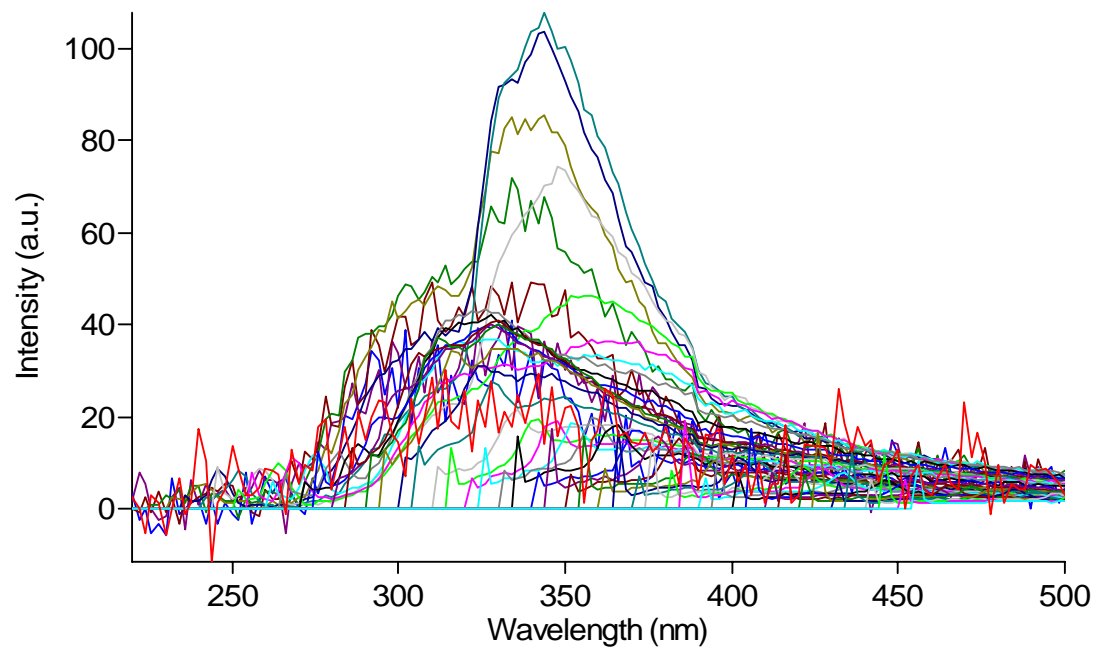
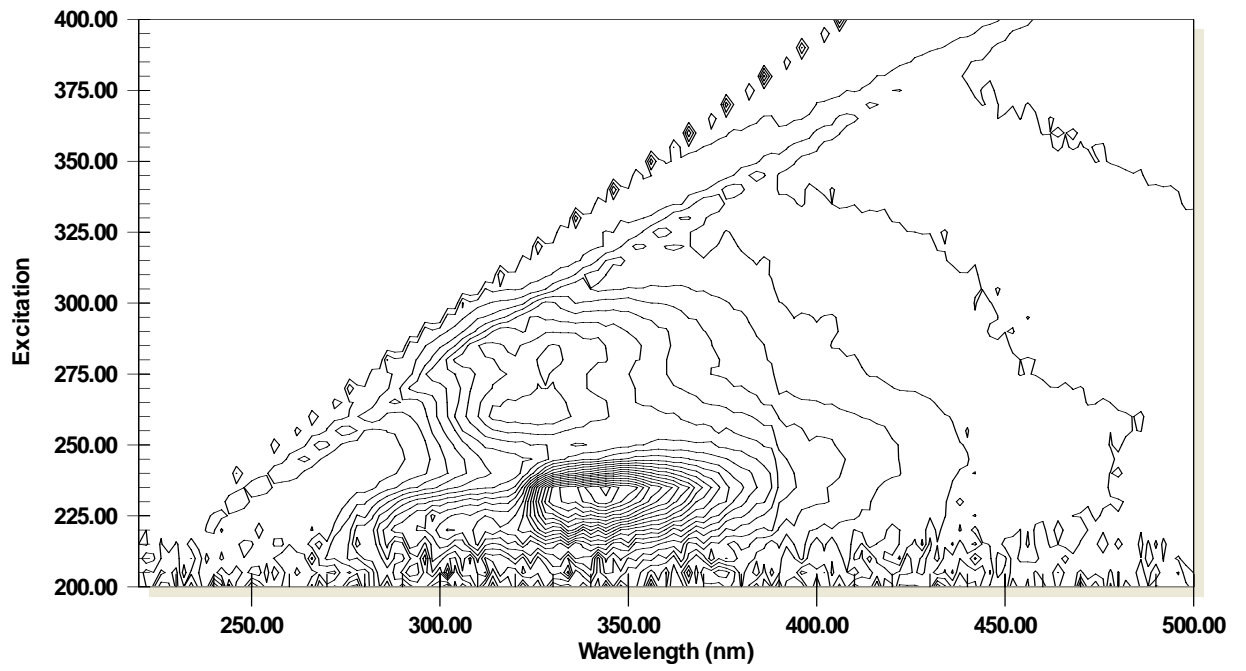
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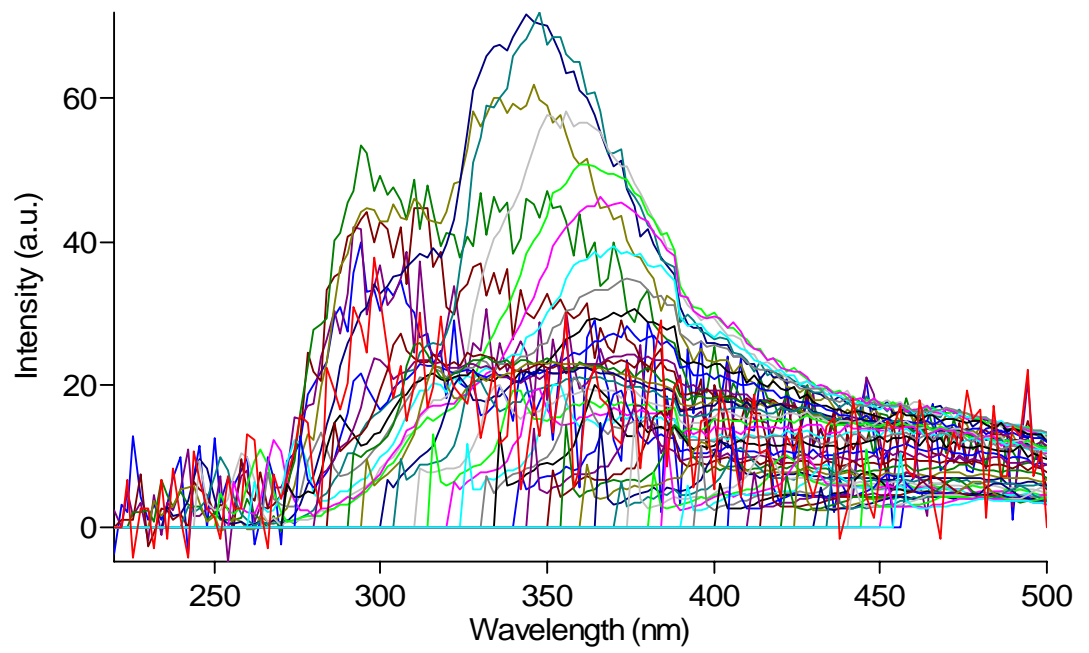
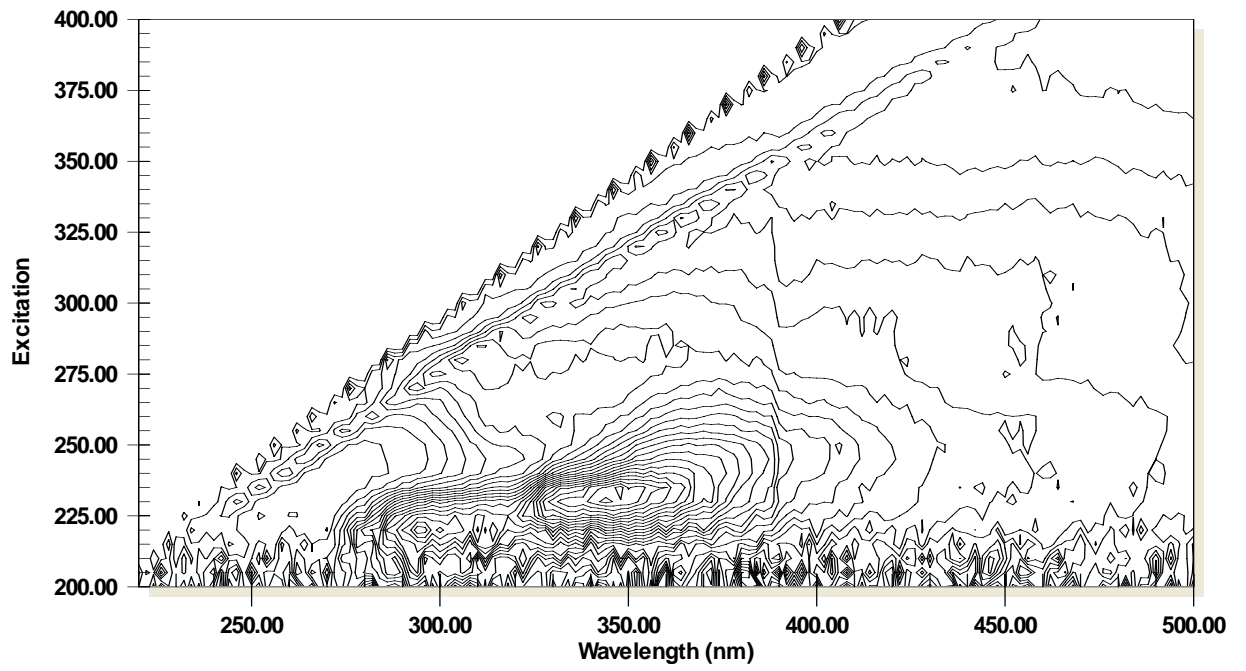
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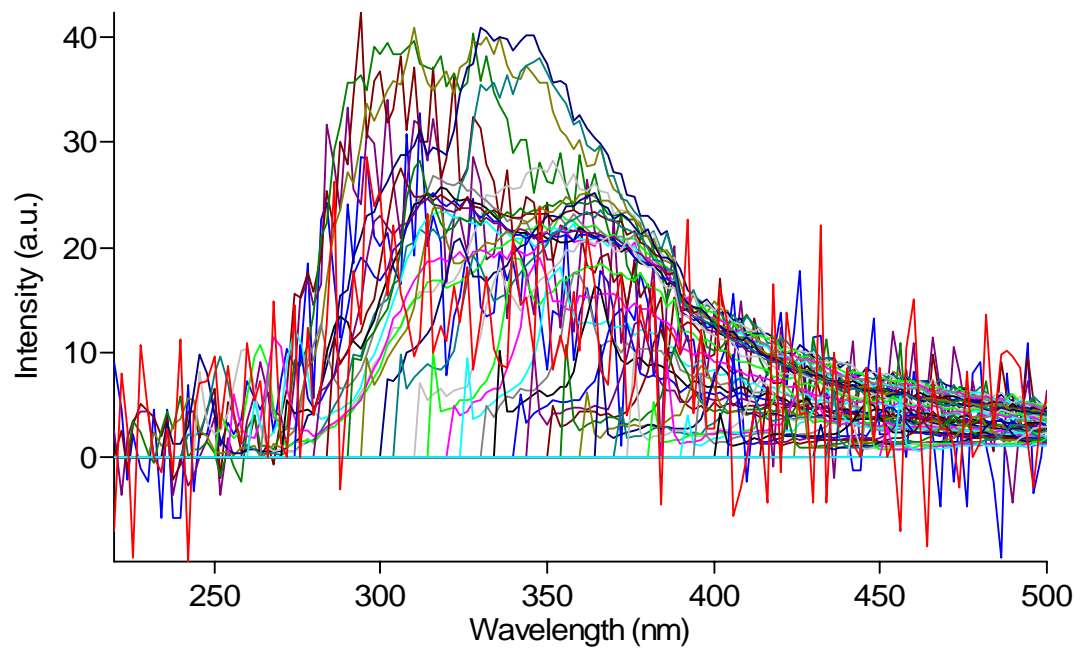
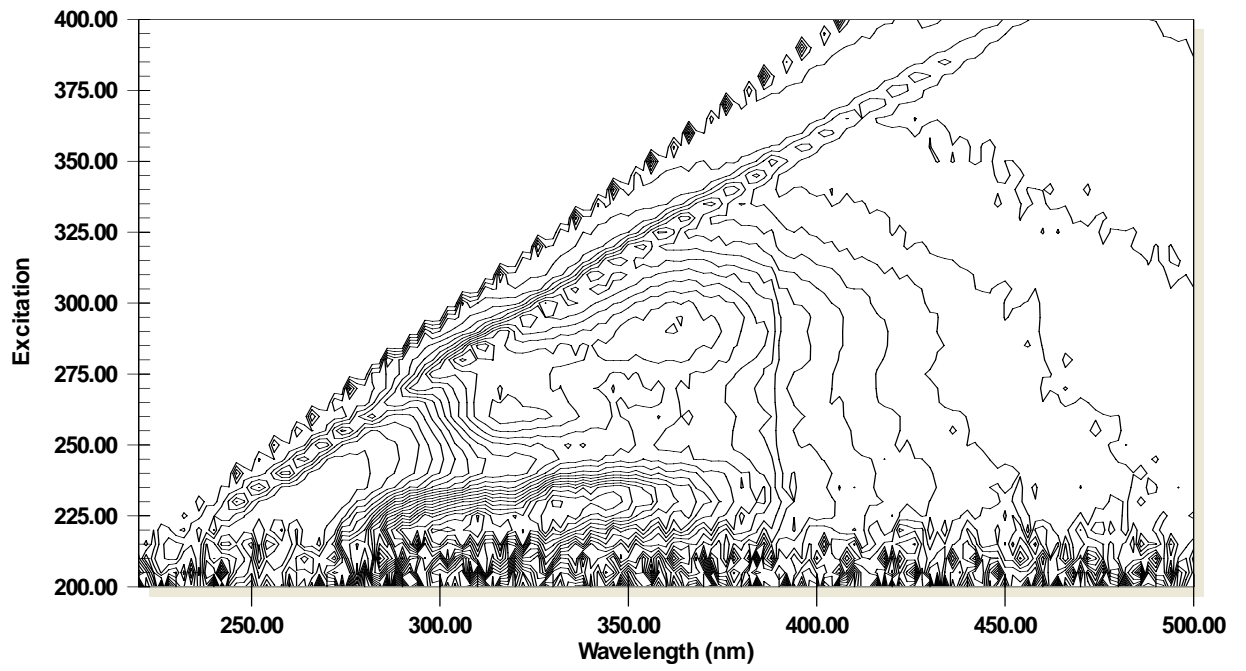
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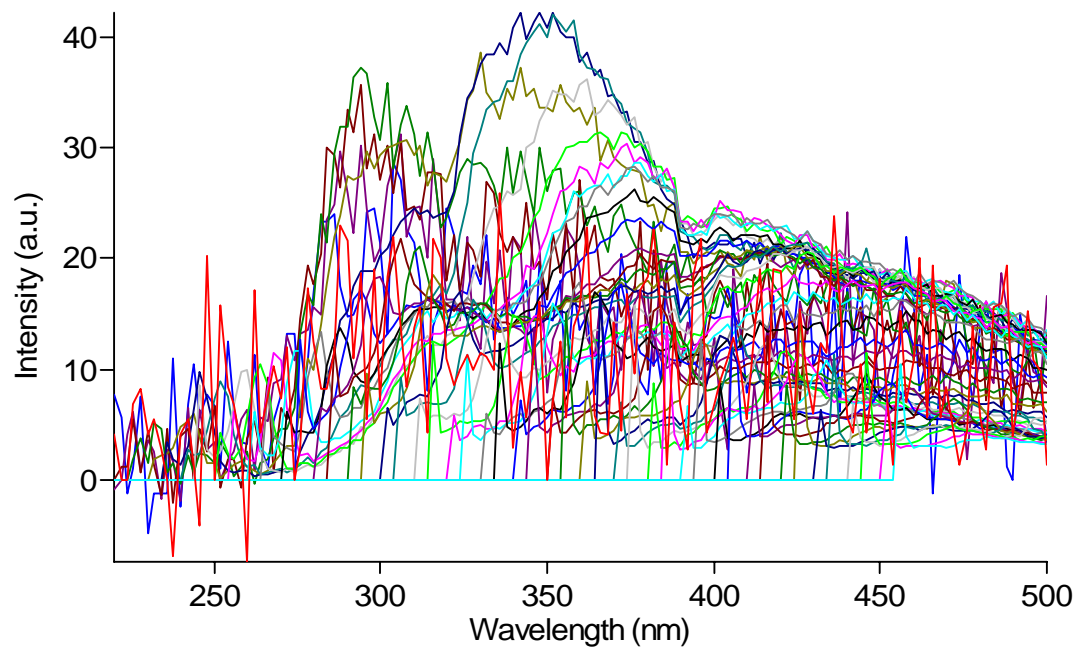
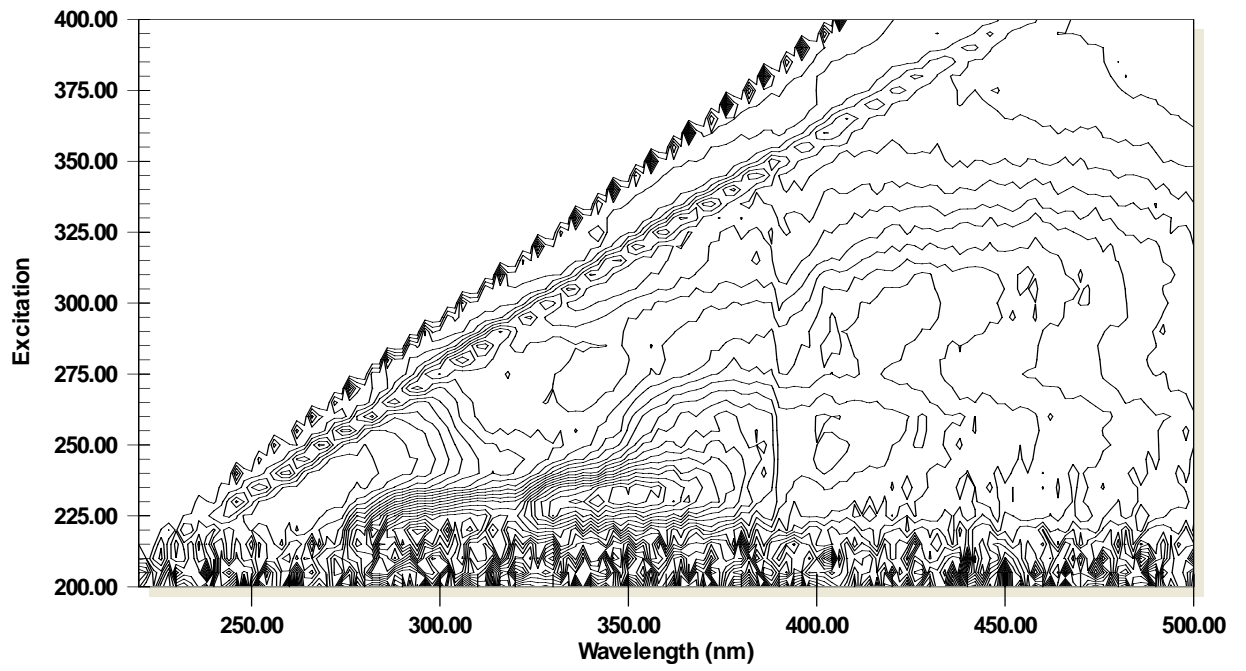
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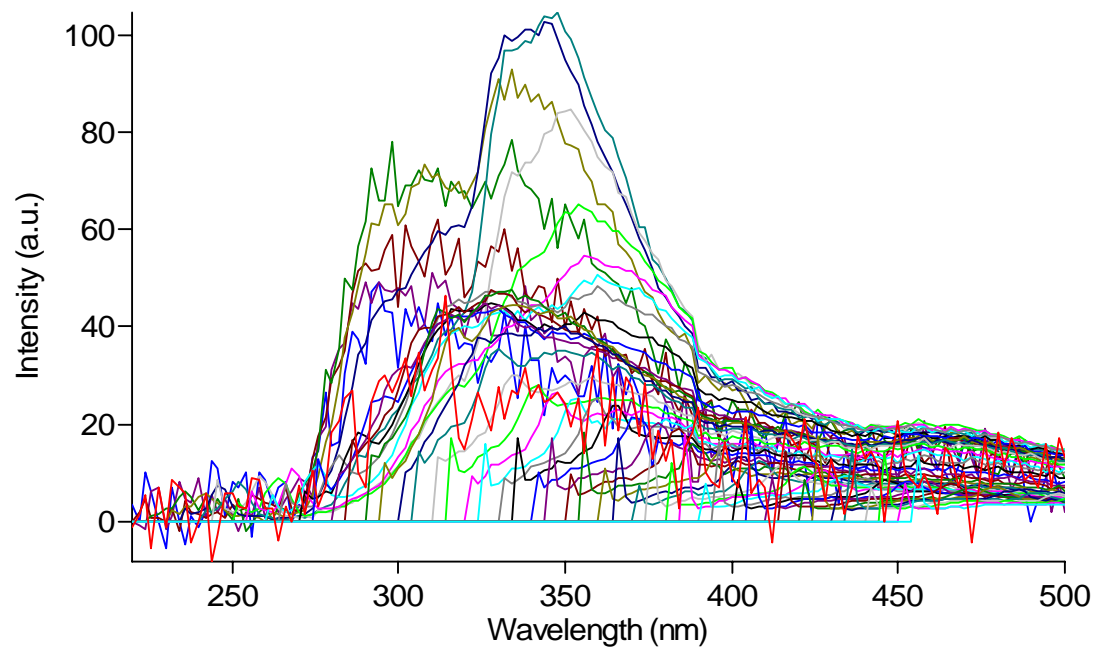
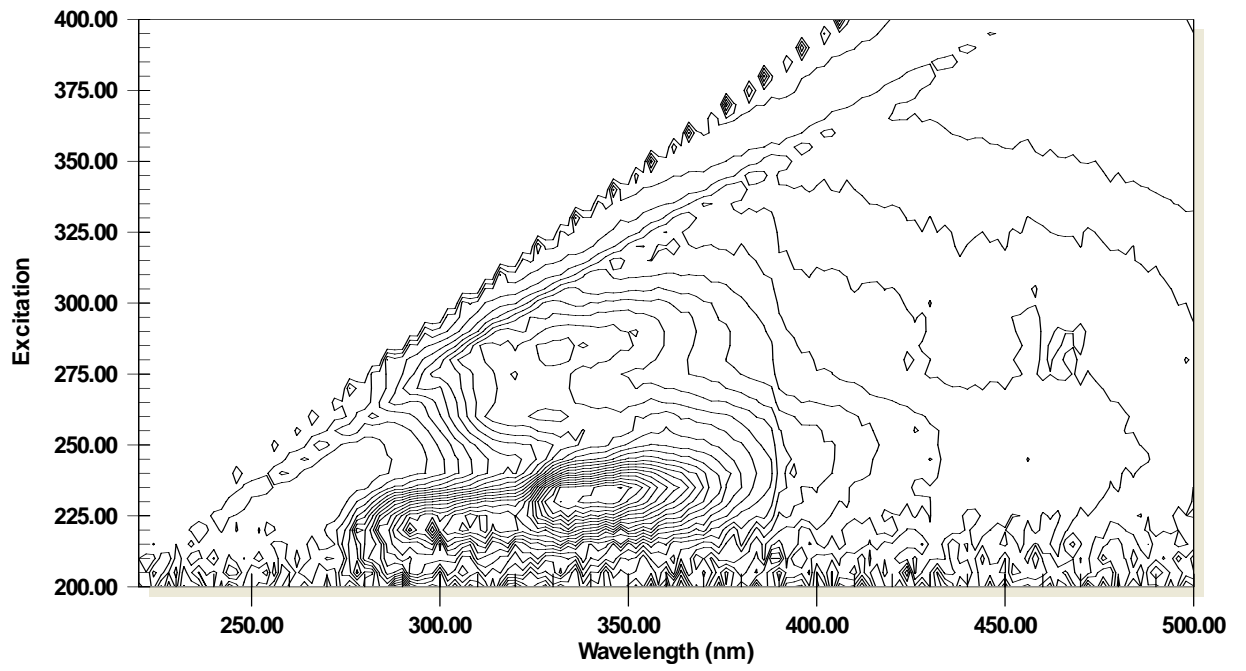
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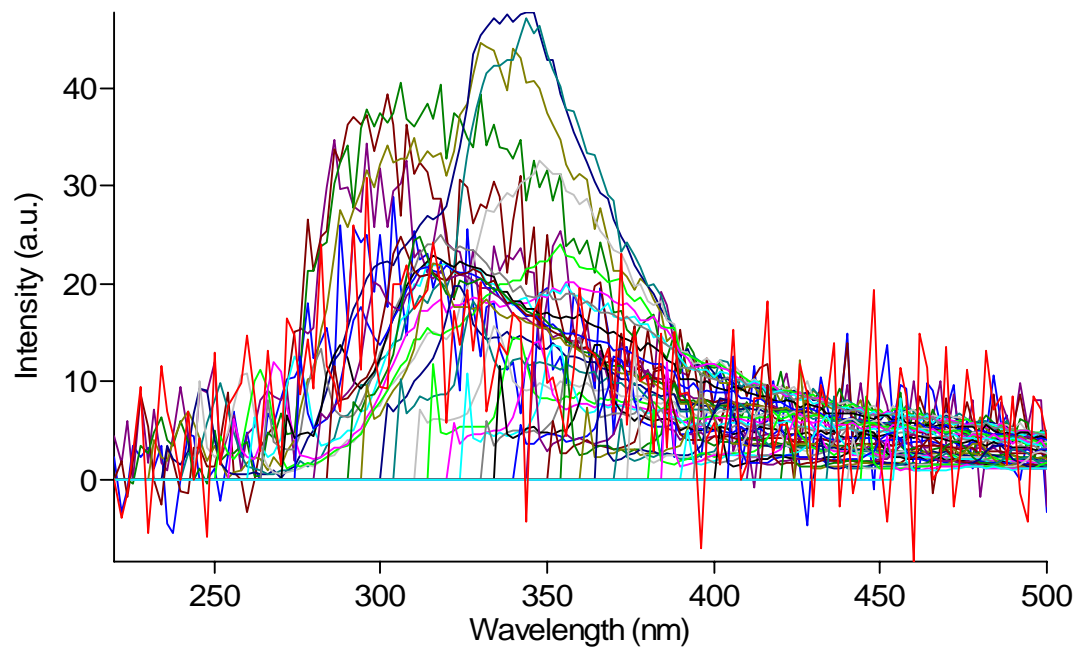
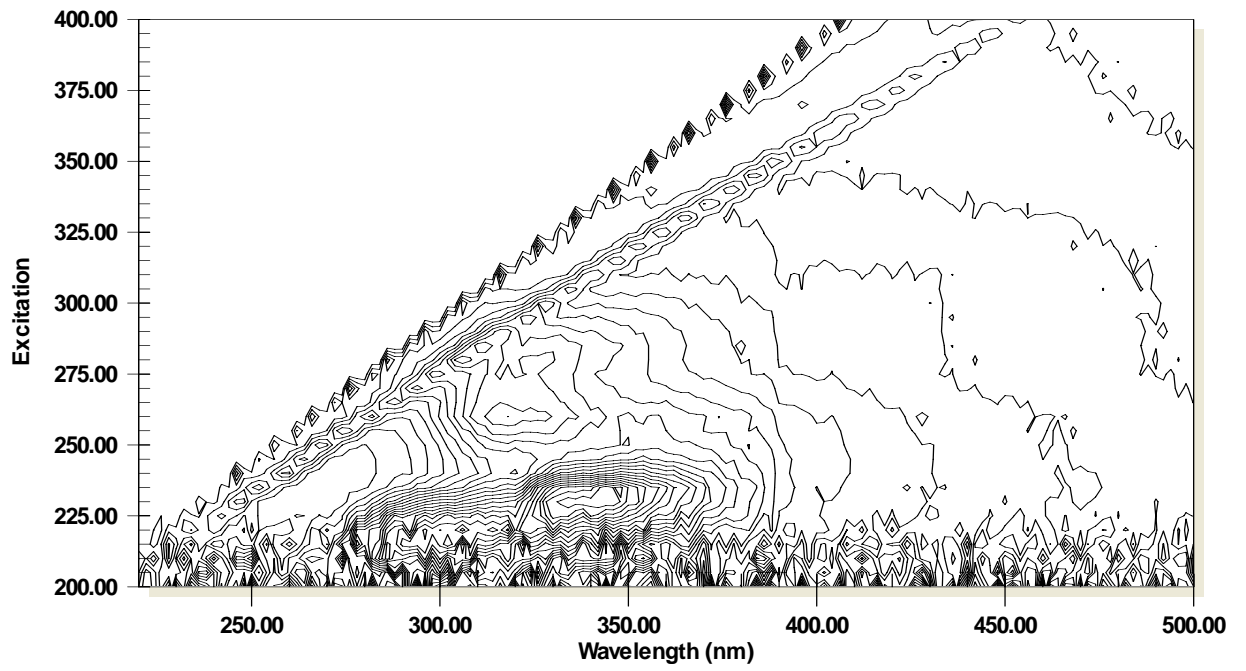
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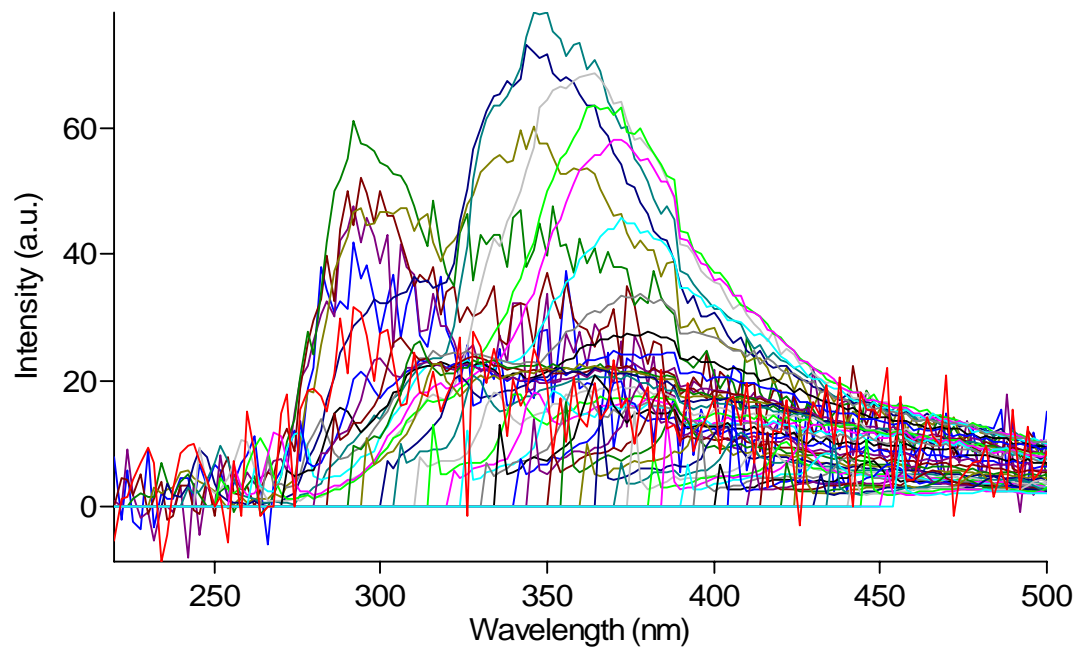
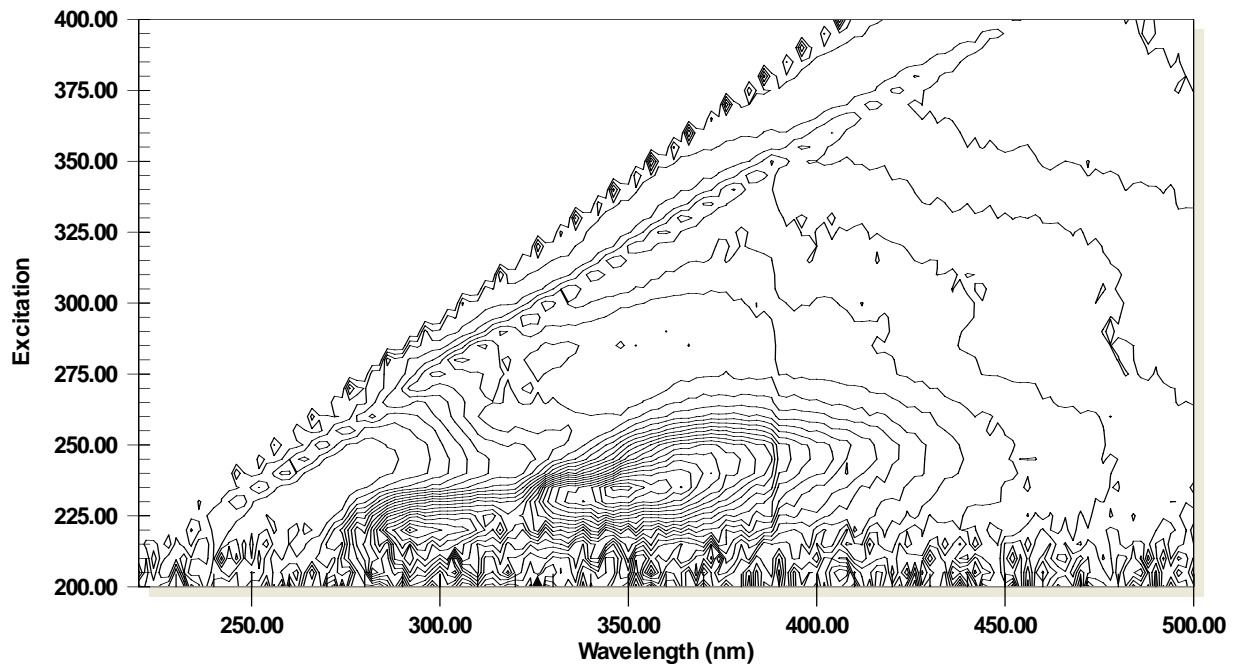
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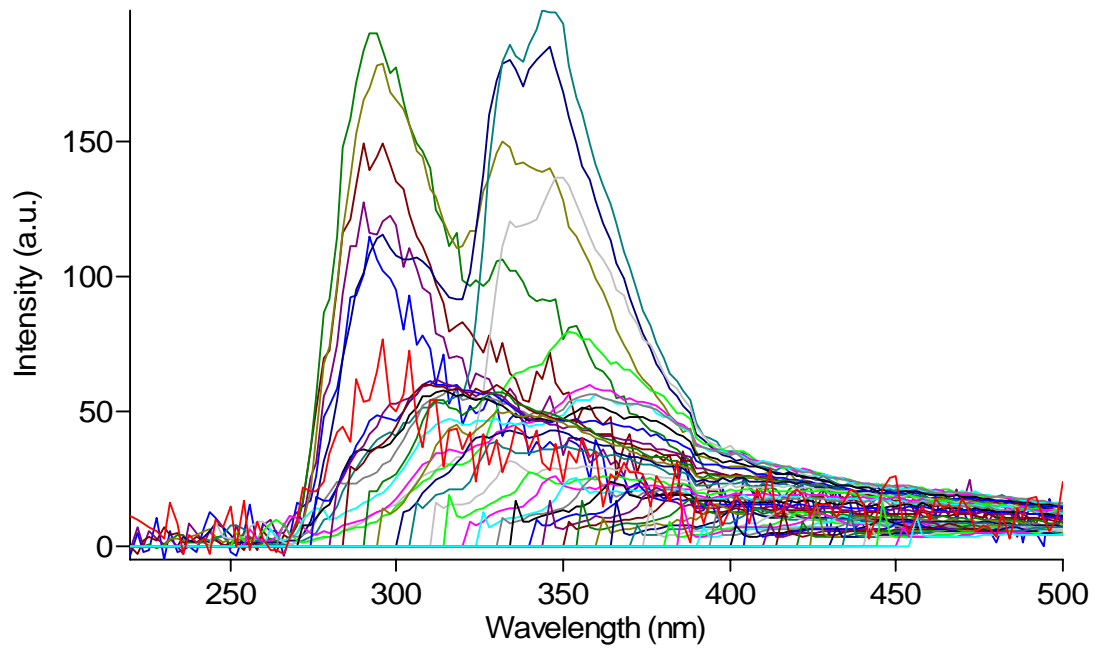
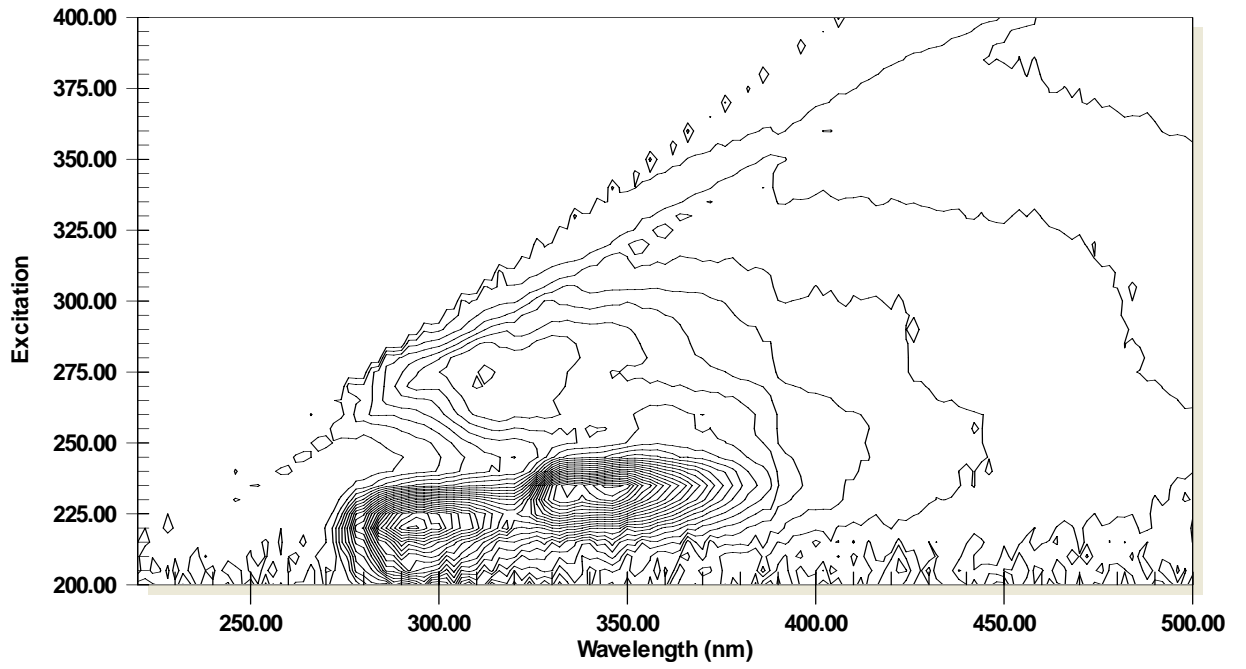
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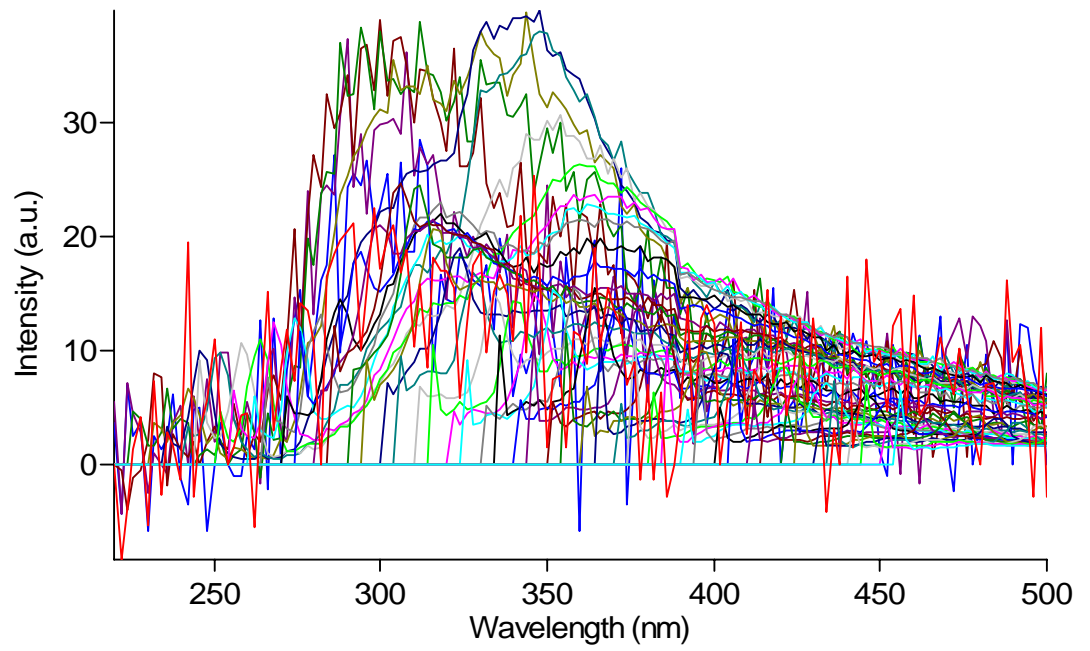
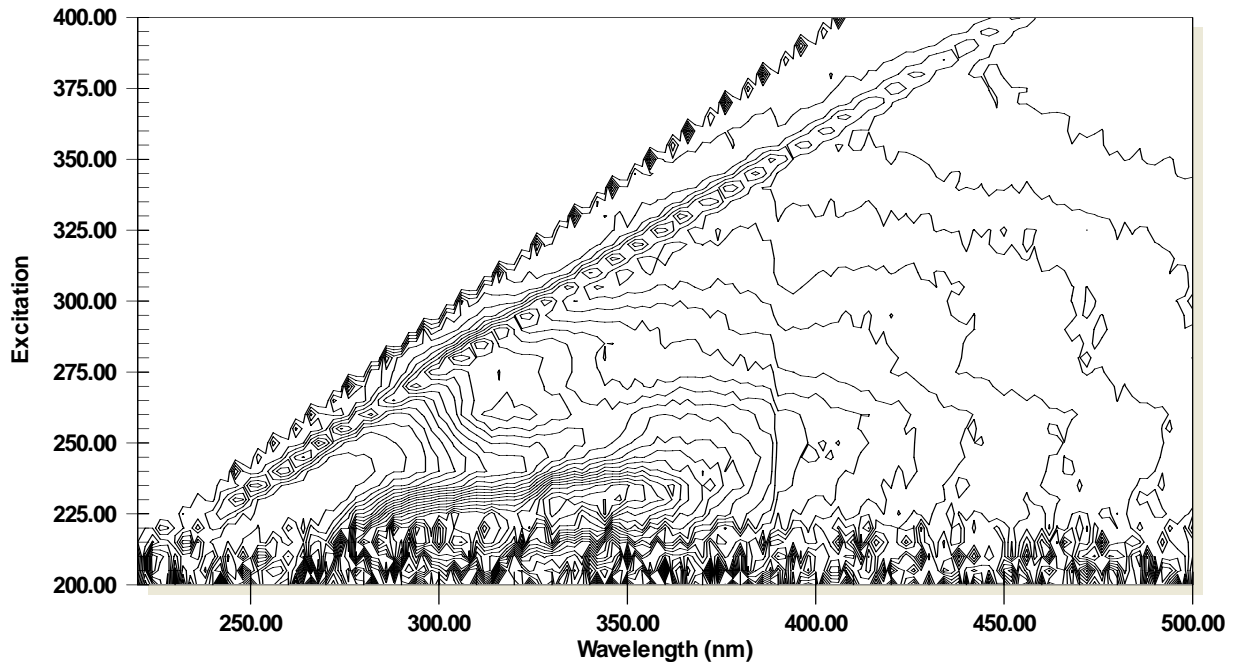
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M40y



M46y



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