

**Synthesis of C-Glycoside Probes for Sialyl Lewis X-
Selectin Recognition**

By

Richard W. Denton

A dissertation submitted to the Graduate Faculty in Chemistry in partial fulfillment of the requirements for the degree of Doctor of Philosophy, The City University of New York

2007

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This manuscript has been read and accepted for the
Graduate Faculty in Chemistry in satisfaction of the
dissertation requirement for the degree of Doctor of Philosophy.

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Abstract

Synthesis of C-Glycoside Probes for Sialyl Lewis X-Selectin Recognition

By

Richard W. Denton

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Interest in C-glycosides as glycomimetics has led to the development of several methodologies for their synthesis. The Mootoo's group has shown that 1-thio-1,2-O-isopropylidene acetals (TIAs) are precursors to β -C-galactodisaccharides. In this thesis, application of the TIA methodology to five C-glycoside analogs of the Gal1 β ->1 α Man O-disaccharide mimetic of sialyl Lewis X is described: two conformationally restrained analogs which have different conformational properties with respect to the intersaccharide linkages and three unrestrained structures with one or two fluorines on the intersaccharide carbon. These C-glycoside analogs are important because they are more hydrolytically stable than O-glycosides and could be more practical for drug development. The restrained C-glycosides are designed to restrict the conformational mobility around the Gal pseudoglycosidic linkage and only allow free rotation around the Man linkage. The fluorine substituent in the mono- and difluoro C-glycosides are also expected to effect conformational bias about the intersaccharide bond. Therefore these compounds could be useful as probes for interrogation of the conformational requirements for binding.

Conformational analysis of these C-glycosides was performed by a combination of NMR spectroscopy and molecular mechanics (MM) and molecular dynamics (MD)

calculations. The P-selectin binding of these mimetics in a Biacore assay was also determined. At 12 mM, the O-glycoside showed 48% inhibition of binding, while the C-glycoside analogs exhibited between 25 - 39% inhibition. The conformational and selectin binding data is discussed within the context of the crystal structure of sLe^x-P-selectin.

The application of the TIA methodology to the preparation of the C-glycoside of methyl α -D-altropyranosyl-(1 \rightarrow 4)- α -D-glucoopyranoside is also presented. This synthesis illustrates the suitability of the C-glycoside methodology to C-disaccharides with unusual glycone segments, and is also pertinent to the altromycins, a naturally occurring class of aryl- α -C-altrosides.

Acknowledgements

At times one may wonder when this journey will end; this experience opened the doors to valuable relationships and carved out wonderful friendships that cannot go unappreciated. I would like to express sincere thanks to everyone who has contributed to making my years as a graduate student memorable.

First of all, I am deeply thankful to my Savior Jesus Christ who has given me the impetus and the strength to complete this thesis. I know that He has led me to the best mentor any graduate student could ever desire.

To Dr. David R. Mootoo, I am truly grateful to you for your deepest and most insightful mentoring techniques. Your advice to me is very precious. Because of you I have learned a lot about carbohydrate and synthetic chemistry, not to mention life outside of the lab. You have instilled in me the discipline and skills that are needed to become a successful researcher. This I know will be an impetus in my career. Certainly you will never be forgotten.

I would like to also express my gratitude to my committee members, Dr. Richard Franck and Dr. Alexander Greer, for the inspiration and the invaluable comments they contributed throughout my academic studies.

Special thanks are also in line for Dr. Michael Blumenstein, for his in-depth knowledge and well needed expertise in NMR experiments. Also to my friend Dr. Clifford E. Soll for his instruction and contributions in MS analysis. I would also like thank Professor Jesús Jiménez-Barbero at Centro de Investigaciones Biológicas, CSIC in Madrid, Spain for the conformational studies done.

I am grateful for the financial support provided by the Minority Biomedical Research Support (MBRS) and the Minority Access/Graduate Networking (MAGNET) foundations, and the warmth provided by the staff of the respective programs: Dr. Victoria Luine, Mrs. Barbara Thorsen, Janerie Rodriguez and Gertude Rivera (MBRS) and Dr. Gail Smith (MAGNET).

I am also appreciative of the staff of the chemistry department at both Hunter College and the Graduate Center CUNY for their remarkable support in helping me to complete this doctoral degree.

To my colleagues and fellow students at Hunter and the Graduate Center, thank you for providing such a wonderful environment where we can communicate and learn from each other. This truly helped me to accomplishing my goal. Thanks guys for just being there: Darren Dabidene, Lei Zu, Fatou Camara, Line Augustin, Sunej Hans, Xiaohua Li, Stewart Bachan, Wang Feng, Clayton Mattis, Xuhong Cheng, Kurissery A. Tony, Anna Dilhas.

Finally, I am indebted to my family: My parents Lascelles and Wilhel Denton for their love and guidance. My mother in-law and pastor Rev. Joyce Hutchinson for her prayers and assistance in every way possible. My brothers, sisters and well wishers for their encouragements. To Jennifer Denton my wife for her patience, love, kindness, support, which assisted me when I needed it most. To my sons, Malachi (who's aspiration is to be a chemist), Christian and baby Jordan Denton thanks for being a motivational force during my graduate studies.

Brethren, I count not myself to have apprehended: but this one thing I do, forgetting those things that which are behind, and reach forth unto those things which are before, I press towards the mark for the prize of the high calling of God in Christ.

Philippians 4 vs.13-14.

To the people dearest to me

My beautiful wife

My adorable sons

My encouraging parents

My prayerful pastor

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Symbols and Abbreviations

Ac	acetyl
Ac ₂ O	acetic anhydride
Avg.	average
BF ₃ .EtO	boron trifluoride etherate
Bn	benzyl
br	broad
Brine	saturated aqueous sodium chloride solution
Bu ₄ NI	tetrabutyl ammonium iodide
Bz	benzoyl
¹³ C NMR	carbon-13 nuclear magnetic resonance
ca.	about
calcd	calculated
COSY	correlation spectroscopy
CR	complement regulatory-like unit
CSA	camphorsulfonic acid
δ	chemical shift in ppm
d	doublet
DAST	(Diethylamino)sulfur trifluoride
DCC	dicyclohexylcarbodiimide
DDQ	2,3-dichloro-5,6-dicyano-1,4-benzoquinone
DEAD	diethyl azodicarboxylate
DIAD	diisopropyl azodicarboxylate

DIB	iodobenzene diacetate
DMAP	4-dimethylaminopyridine
DTBMP	2,6-di- <i>tert</i> -butyl-4-methylpyridine
EGF	epidermal growth factor domain
ELAM-1	endothelial leukocyte adhesion molecule
eq	equivalent
ESL-1	E-selectin ligand-1
ESMS	electrospray mass spectroscopy
Et ₂ O	Diethyl ether
EtOAc	Ethyl acetate
FABMS	Fast Atom Bombardment Mass Spectroscopy
FCC	Flash column chromatography
Fuc	L-fucose
g	gram
Gal	D-galactose
Glc	glucose
GluNAc	N-acetyl-D-glucosamine
GLYCAM-1	glycoprotein cell adhesion molecule-1
h	hour
HOAc	acetic acid
HRMS	High resolution mass spectroscopy
¹ H NMR	Proton nuclear magnetic resonance spectroscopy
Hz	hertz
IC ₅₀	Concentration of inhibitor resulting in the reduction of bonding to 50% of

	maximum.
J	coupling constant in hertz
L	liter
Le ^x	Lewis X
m	multiplet
MadCAM-1	mucosal addressin cell adhesion molecule-1
man	mannopyranoside
<i>m</i> -CPBA	meta-chloroperoxybenzoic acid
MD	Molecular dynamics
Me	Methyl
mg	milligram
min	minute
MM3	Force fields for molecular mechanics calculations
mmol	millimole
MOM	methoxy methyl
ms	molecular sieves
NaH	Sodium Hydride
neu	Sialic acid
NMO	N-Methylmorpholine-N-oxide
NMR	Nuclear magnetic resonance
NOE	Nuclear overhauser effect
°C	degree Celsius
Ph	phenyl
PhSH	thiophenol

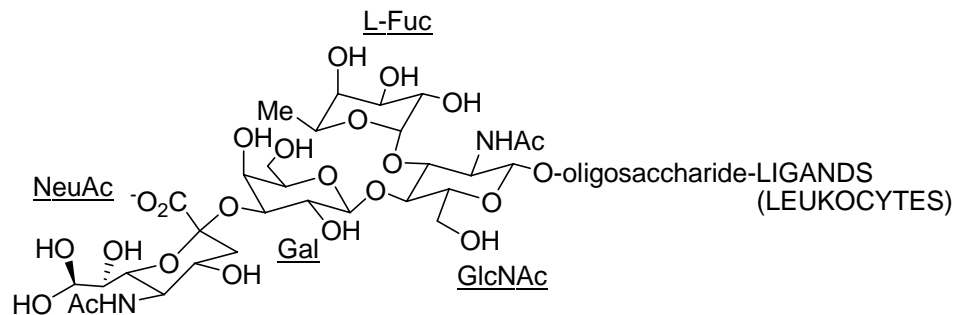
PMB	<i>para</i> -methoxy benzyl
ppm	parts per million
PSGL-1	P-selectin glycoprotein ligand-1
pyr	pyridine
q	quartet
rt	room temperature
s	singlet
sd	standard deviation
sLe ^x	sialyl Lewis X
t	triplet
TBDPS	<i>tert</i> -butyl diphenyl silyl
TfOH	trifluoromethanesulfonic acid
THF	tetrahydrofuran
TIA	1-thio-1,2- <i>O</i> -isopropylidene acetal
TLC	thin layer chromatography
TMEDA	<i>N,N,N',N'</i> -tetramethylethylenediamine
TMS	trimethyl silyl
TMSOTf	trimethyl trifluoromethanesulfonate
Trityl	triphenylmethyl
TsCl	<i>p</i> -toluenesulfonyl chloride (<i>p</i> -CH ₃ C ₆ H ₄ SO ₂ Cl)
vs	versus

Chapter 1

Introduction

1.1 Background

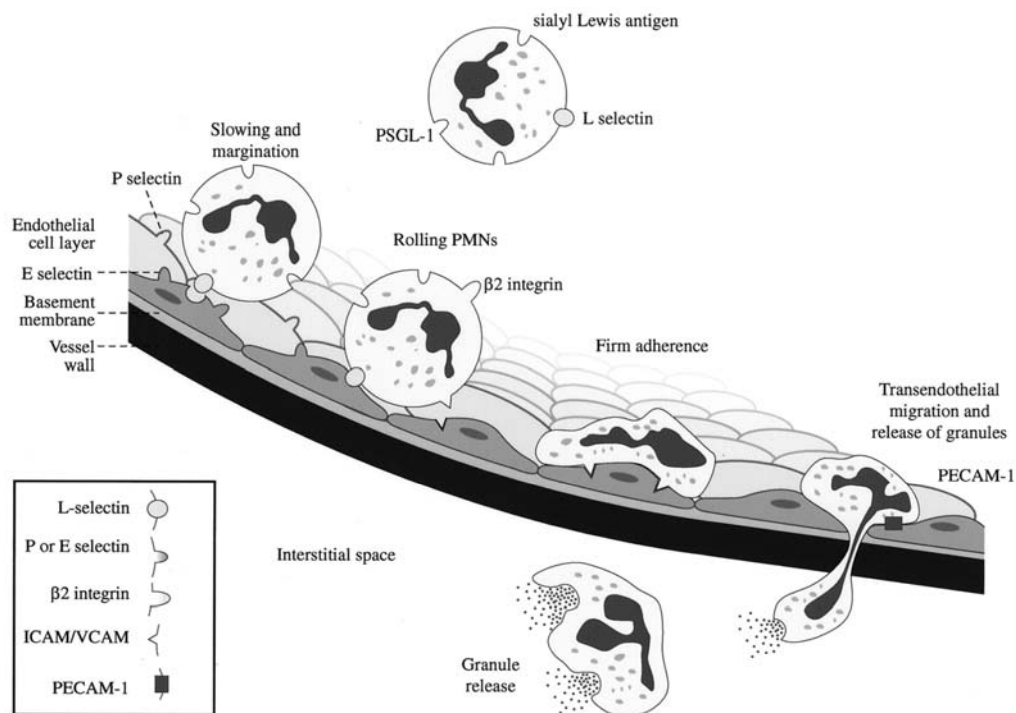
Figure 1.1: Sialyl Lewis X (sLe^x)



In the response to injury, the inflammation cascade begins with the release of cytokines at the site of injury within the tissue.¹ This stimulates the upregulation of two proteins E- and P-selectin on the endothelial surface. E- and P-selectin recognize the tetrasaccharide sialyl Lewis x (sLe^x) (Figure 1.1), and its glycoprotein derivatives on the surface of leukocytes and promote leukocyte adhesion to endothelial cells. A third selectin, L-selectin is constitutively expressed on leukocytes, and recognizes similar carbohydrate ligands found on the endothelium. This relatively weak, sLe^x mediated adhesion processes result in “rolling” of leukocytes across the endothelium leading to higher affinity adhesion between integrins on the leukocytes and endothelial proteins, ICAM-1 (intercellular adhesion molecule-1) and VCAM-1. Proteolytic cleavage of L-

selectin from the surface of the leukocytes occurs and PECAM receptors on endothelial cell surfaces mediate migration of the leukocytes through the endothelial layer to the damaged or infected tissue, where accumulation of leukocytes occurs (Figure 1.2). While inflammation plays a key role in controlling infections and repairing injuries, excessive accumulation of leukocytes may lead to damage of healthy cells and result in disease states such as reperfusion injuries, stroke, psoriasis, rheumatoid arthritis and respiratory diseases.^{2, 3} Since E- and P-selectin are activated in response to injury, regulation of sLe^x-selectin binding has been examined as a new strategy for treating inflammation disorders.²

Figure 1.2: Inflammation cascade

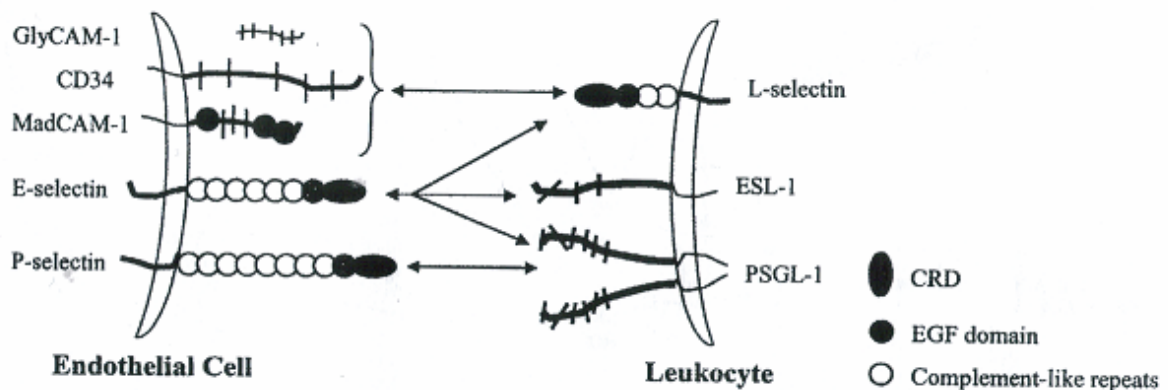


1.2 Selectin structure and their ligands:

The selectins are composed of five domains. A N-terminal calcium dependent lectin domain, an epidermal growth factor domain (EGF), a variable number of complement regulatory-like unit (CR), a transmembrane domain, and an intracellular tail.^{4,5}

L-selectin interacts with three glycoprotein ligands on the endothelial cell surface. These are the glycoprotein cell adhesion molecule-1 (GLYCAM-1), CD34 and the mucosal addressin cell adhesion molecule-1 (MadCAM-1) ligands. E-Selectin recognize by P-selectin glycoprotein ligand-1 (PSGL-1) and E-selectin ligand-1 (ESL-1). P-selectin only interacts with PSGL-1, which is the best characterized ligand to date (figure 1.3).^{6,7} PSGL-1 is a glycoprotein with N- and O-linked glycans containing sLe^x, and a cluster of three N-terminal tyrosine sulfated groups. A truncated form of PSGL-1 was found to be effective against P-selectin and this led to numerous syntheses of analogs of PSGL-1 as potential anti-inflammatory agents.^{8,9}

Figure 1.3: The interaction between the selectins and their ligands.⁷



were almost electrostatic in nature. Both structures indicated that the 3- and 4- hydroxyl groups of the fucose bound to calcium. This mode of binding was in contrast to the earlier prediction of the fucose 2- and 3- hydroxyl groups being ligated to calcium, a concept originating from the structure of the rat mannose binding protein complexed with oligomannose.¹¹ It was also shown that the fucose hydroxyl groups form hydrogen bonds with specific selectin residues; the 4-OH group shows hydrogen bond to Asn82 and Glu80, while the 3-OH group hydrogen bonds to Asn85. In E-selectin, additional hydrogen bonding interactions are observed between Asn83 and Fuc 2-, 3- hydroxyl groups and Glu107. These interactions are not seen in the P-selectin/sLe^x complex. It was suggested that the overall fucose interactions provide a large amount of the sLe^x binding energy. In both P- and E- complexes, the sLe^x galactose residue hydrogen bonds to Tyr94 and Glu92, and the carboxylate group of the sialic acid moiety hydrogen bonds to Tyr48. These interactions combined with differences in the fucose binding seem to be the reason for the higher affinity of E-selectin/sLe^x interaction.

This binding model has been widely used to design small molecule mimetics of sLe^x. The majority are carbohydrate derived mimetics that appear to adapt similar contact points as sLe^x. As O-glycosides are prone to enzymatic and chemical degradation, and possess poor pharmacokinetic and bioavailability profiles, there has been interest in non-carbohydrate type mimetics.⁷ The latter may also be easier to synthesize and more compatible with combinatorial chemistry.

A brief summary of sLe^x mimetics grouped according to tri-, di- and mono-saccharide and non-carbohydrate structures will be presented.

1.3 Trisaccharide mimetics

Trisaccharide mimetics were essentially used to identify the key hydroxyl groups of galactose and fucose residues. The expensive sialic acid was substituted by simpler species like sulfate **1.1**,⁴ carboxymethyl and cyclohexyl lactic acid groups. These provided some active mimetics of which compound **1.2** has one of the highest potency against E-selectin.^{7, 12} Interestingly the Gal 2-OH group compound **1.3** was less active than **1.2**. This increase in activity was believed to be due to preorganization of the bioactive conformation **1.2** compared to sLe^x and not additional interactions between the benzoate group and E-selectin.¹²

1.4 Disaccharide mimetics

An important disaccharide mimetic reported by Hiruma and coworkers was the O-glycoside **1.4**.¹³ This was a β -D-galactopyranosyl-(1 \rightarrow 1)- α -D-mannopyranoside with a flexible carboxymethyl group. It was reported to be five and forty times more active than sLe^x against E- and P-selectin respectively, in a cell based assay. Existing models for sLe^x-selectin binding suggested that the galactose and fucose recognition sites for sLe^x are homogeneous with the binding regions for the galactose and mannose residues in **1.4**.

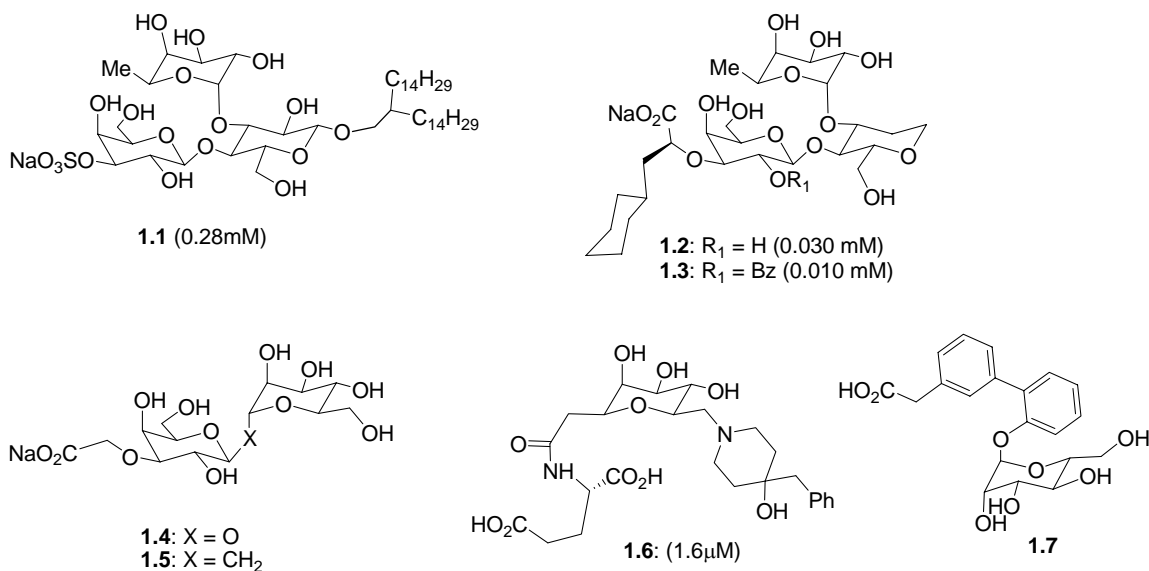
These encouraging results led to the synthesis of the exact C-glycoside **1.5** by the Mootoo group,¹⁴ as the C-glycoside was more hydrolytically stable than O-glycosides and therefore more attractive for drug development. Conformational studies, performed in collaboration with Jiménez-Barbero on **1.4** and **1.5** revealed that the unbound O-glycoside existed essentially in a single conformation, while the C-glycoside displayed five significant conformations about its intersaccharide linkages.¹⁵ The P-selectin

binding of **1.4** and **1.5** using the BIAcore surface plasmon resonance technique showed good dosage response with similar activity to sLe^x. The similarity in binding of **1.4** and **1.5** was interesting because these compounds were found to be very different in their conformational behavior with respect to their intersaccharide bonds.

1.5 Monosaccharide mimetics

In these types of sLe^x mimetics the lactosamine moiety was replaced with a flexible or rigid scaffold. The majority of these analogs retained the fucose residue but *D*-galactose or *D*-mannose sugar substitutes have also been popular. Two of the more potent groups of monosaccharide mimetics contained glycopeptide (**1.6**) or biphenyl residues (**1.7**)⁷ as lactose-amine replacements (figure 1.5)

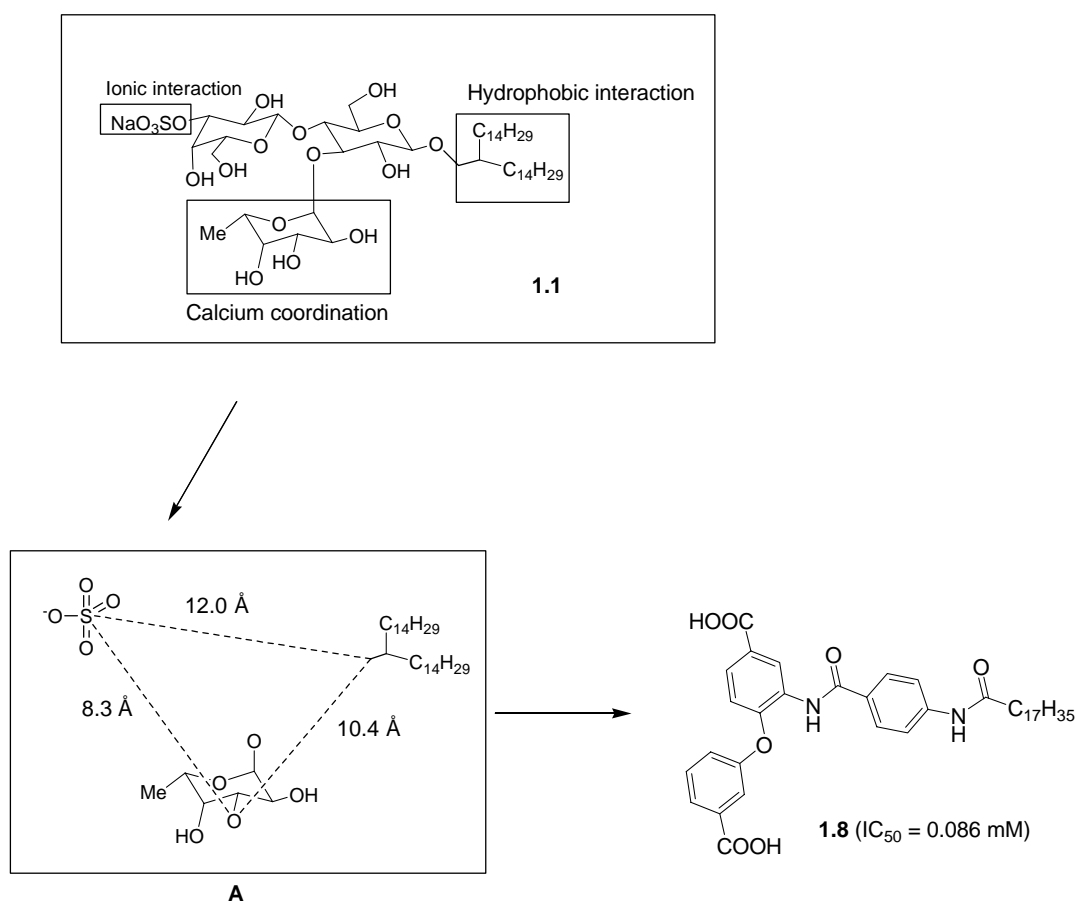
Figure 1.5: Selected carbohydrate mimetics of sLe^x and their binding activities



1.6 Non-carbohydrate mimetics

The analogs in this group are attractive because of their synthetic accessibility, stability, and their pharmacokinetic and bioavailability profiles. A potent E-selectin antagonist, compound **1.8** ($IC_{50} = 0.086$ mM) was designed using a 3D-pharmacophore model from the sulfated trisaccharide **1.1**/E-selectin complex model.^{7, 16} In this approach, the fucose unit and the negatively charged sulfonic acid were replaced with the diaryletherdioic acid residue, and the hydrophobic branched chain was replaced with a single alkyl chain (Figure 1.6).

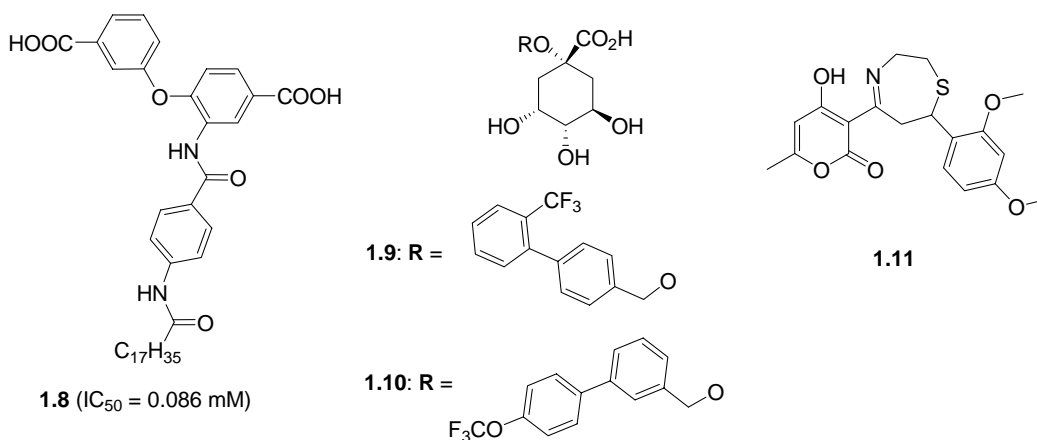
Figure 1.6: Design of **1.8** via a pharmacophore model approach.¹⁶



A cocrystal structure of a small quinic acid derivative with E-selectin led to the hypothesis that two of the hydroxyl groups of quinic acid mimicked the calcium-bound fucose of sLe^x.¹⁷ The X-ray structure data was used to design related **1.9** and **1.10**, which showed IC₅₀'s of approximately 10 mM in a Biacore P-selectin assay. The IC₅₀ of sLe^x was 15 mM.

The non-carbohydrate mimetic 7-phenyl-1,4-thiazepine **1.11** (KF38789) with IC₅₀ of 1.97 μM for P-selectin, and weaker binding for E and L-selectin, was fortuitously discovered.

Figure 1.7: Non-carbohydrate mimetics of sLe^x and their binding activities.



The aforementioned examples represent a fraction of the large number of sLe^x mimetics synthesized to date. The promising activity reported for the relatively simple *O*-disaccharide **1.4** and its *C*-glycoside **1.5** prompt a more extensive investigation on the synthesis, conformational behavior and selecting binding of related disaccharide analogs. This is the thrust of this thesis.

Chapter 2

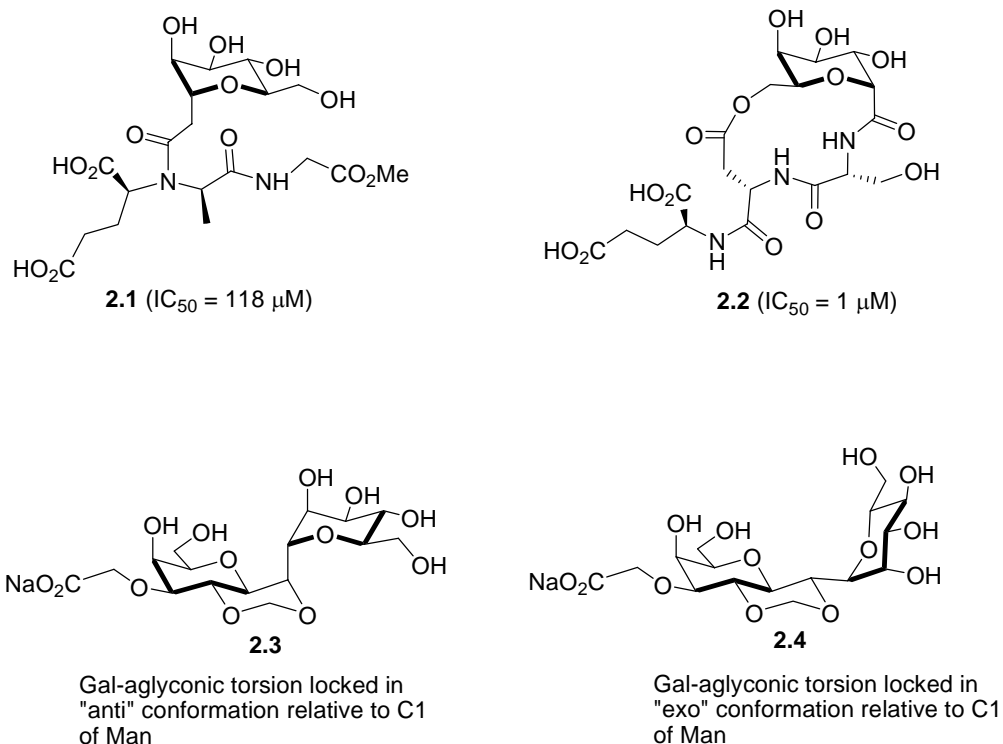
Synthesis of conformationally restrained C-disaccharide sLe^x mimetics.

2.10 Introduction

As previously mentioned, the exact C-glycoside **1.5** of O-glycoside **1.4** has been synthesized by the Mootoo group.¹⁴ There were two reasons for interest in this C-glycoside; first, C-glycosides are more hydrolytically stable than O-glycosides and therefore more attractive for drug development.¹⁸ Second, the presence of the carbon linker in the C-glycoside allows for design of analogs with different conformational properties with respect to the intersaccharide linkages¹⁹, thereby leading to disaccharide analogs with different spatial orientation of the galactose and mannose segments. Evaluation of the selectin affinity of such structures could provide insight on the most active conformation of the disaccharide framework and in general, a clearer picture on the optimal requirements for binding to the selectins. Rigid compounds with all the recognition elements pre-organized in a conformation that is favorable for binding are expected to show higher affinity because of entropy considerations.²⁰ An example of this concept is illustrated in the relative binding of sLe^x mimetic **2.1** and the rigid macrocyclic glycoprotein **2.2**.²⁰ The more rigid analog was approximately 120 fold more active in a P-selectin assay.

Against this backdrop, the synthesis of conformationally restrained C-glycoside **2.3** and **2.4** was undertaken. These analogs present the Man residue in more defined regions of conformational space compared to **1.5**, and could provide more precise structure activity information.

Figure 2.1: Conformationally restrained sLe^x mimetics.



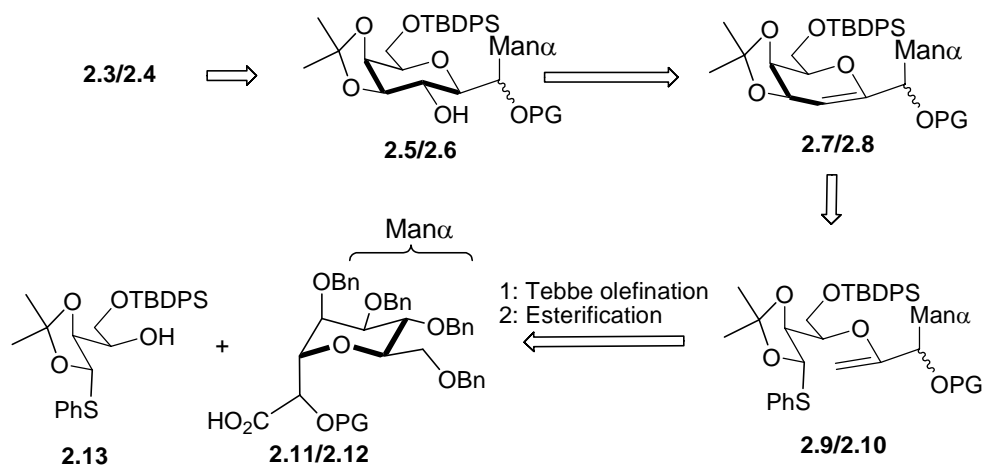
Results and Discussion

2.2 Retrosynthesis

The synthesis of **2.3** and **2.4** entailed the *de novo* construction of the galactose residue and centered on the oxocarbenium ion-enol ether cyclization¹⁴ that was previously applied to **1.5** (Scheme 2.1).^{14, 21} Thus, **2.3** and **2.4** could be obtained from the elaboration of C-disaccharides like **2.5** and **2.6** with an appropriate protecting group on the alcohol in the intersaccharide position. Compounds **2.5** and **2.6** should be available from the stereoselective hydroboration of C-1 substituted glycals **2.7** and **2.8**, which are the expected products of the oxocarbenium cyclization on thioacetal - enol ether substrates **2.9** and **2.10**, respectively. Precursors **2.9** and **2.10** would be obtained in a

convergent fashion from thioacetal fragment **2.13** and mannose derived α -alkoxy acids **2.11** and **2.12**.

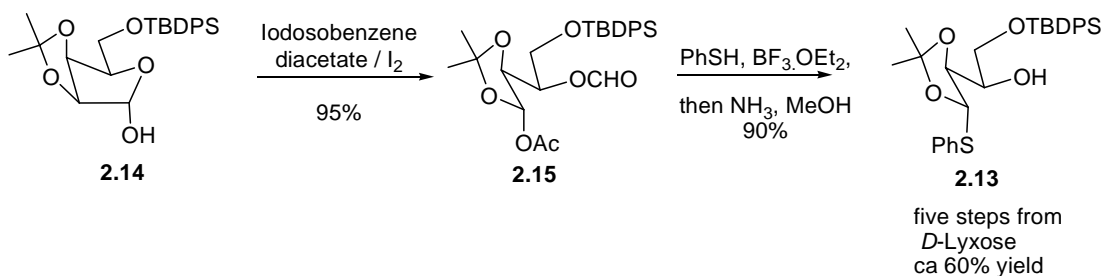
Scheme 2.1: Oxocarbenium ion cyclization strategy for hydroxymethyl linked C-disaccharides.



2.3 Synthesis of the thioacetal glycone and the acid segments.

The synthesis of the glycone **2.13** involved the Suarez radical fragmentation of the 1,2-O-isopropylidene furanose **2.14** to give the 1-O-acetyl-1,2-isopropylidene **2.15**.²² Acetal exchange of **2.15** with thiophenol followed by basic hydrolysis of the resulting formate ester gave the 1-thiophenyl-1,2-isopropylidene alcohol **2.13** (Scheme 2.2).¹⁴

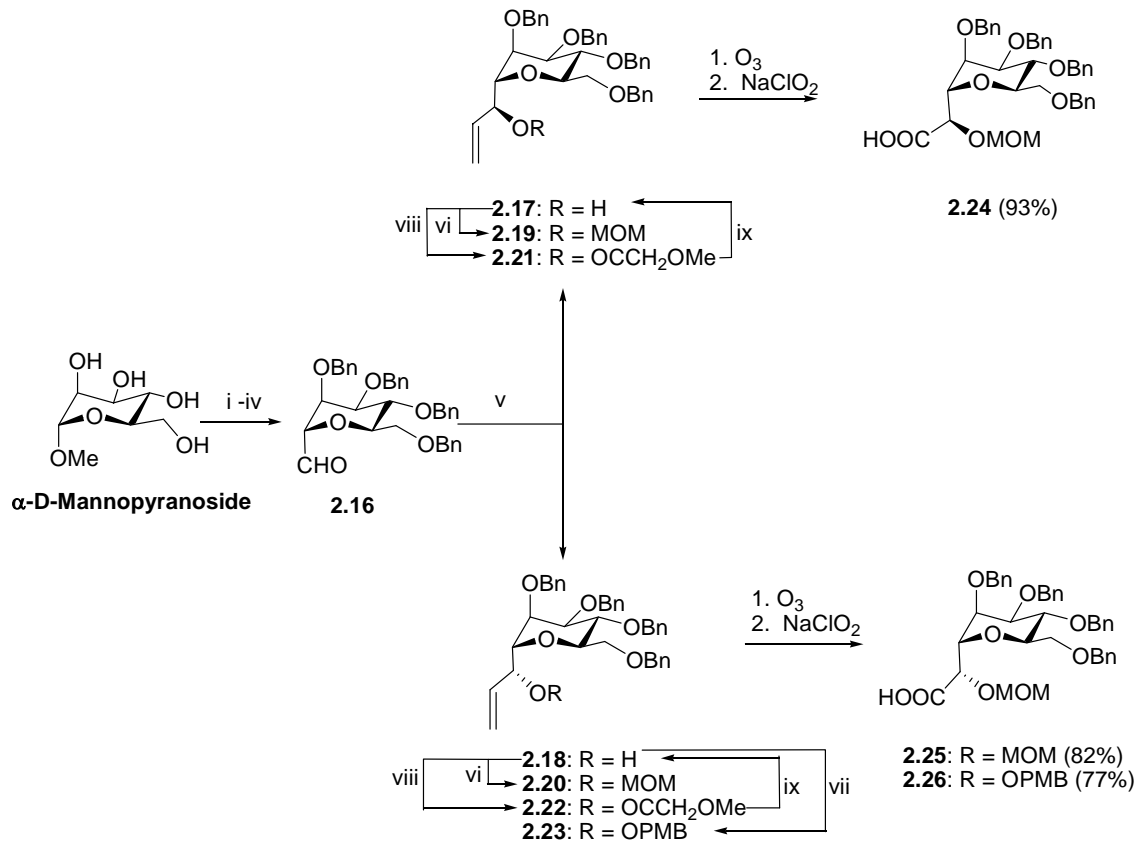
Scheme 2.2: Synthesis of the glycone synthon



The acid components were prepared *via* a straightforward sequence of reactions from the known C-formylmannoside **2.16**, available in four steps from methyl α -*D*-mannopyranoside.¹⁰ Treatment of **2.16** with vinylmagnesium bromide provided a mixture of alcohols **2.17** and **2.18** that was separated as the methoxymethyl ether derivatives **2.19** and **2.20** (66% from **2.16**, 3:2 respectively, scheme 2.3). Ozonolysis of MOM-protected allylic alcohols **2.19** and **2.20** followed by sodium chlorate oxidation of the resulting aldehydes led to the acids **2.24** and **2.25** (Scheme 2.3). The *p*-methoxybenzyl acid derivative **2.26** was also synthesized from the vinyl alcohol **2.18**.

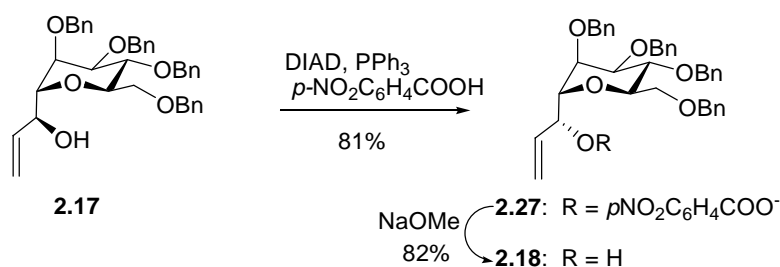
Two points in the synthesis of **2.19** and **2.20** are noteworthy. First, for easier chromatographic separation of these compounds, the alcoholic mixture **2.17/2.18** was converted to the methoxyacetate esters **2.21** and **2.22**. These esters were easily separated and individually hydrolyzed to **2.17** and **2.18** respectively, which was then transformed to **2.19** and **2.20** as before (Scheme 2.3).

Scheme 2.3: Synthesis of the C-mannoside acid components



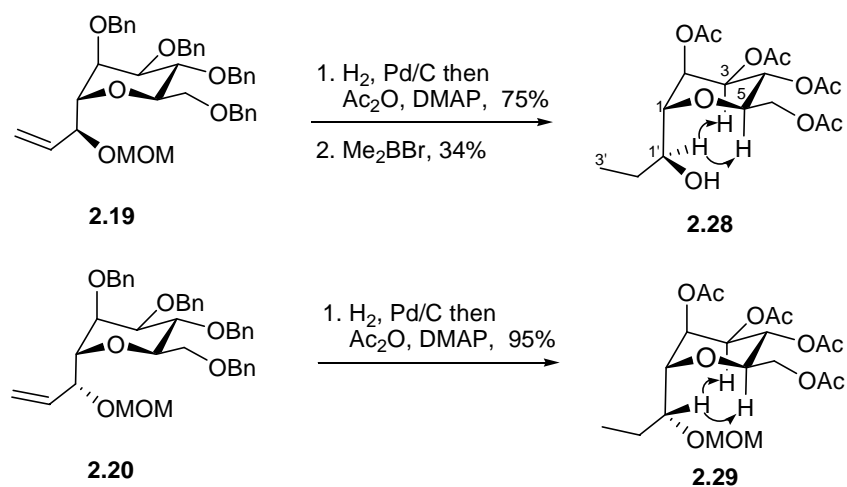
Second, a procedure for conversion of the major alcohol **2.17** to the minor isomer **2.18** was developed. This transformation was achieved *via* the Mitsunobu protocol²³ followed by base hydrolysis of the resulting ester (Scheme 2.4).

Scheme 2.4: Conversion of compound **2.17** to **2.18**



In order to confirm the axial orientation of the C1 substituent in the C-mannoside segments, **2.19** and **2.20** were converted to their tetraacetate-dihydro derivatives **2.28** and **2.29** respectively, and their stereochemical integrity verified by NOESY experiments (Scheme 2.5). The configuration at the allylic position of **2.19** and **2.20** was later assigned from ^1H NMR analysis of the conformationally restrained derivatives **2.45** and **2.46** (*vide infra*).

Scheme 2.5: Conformation of stereochemistry in C-mannoside segments



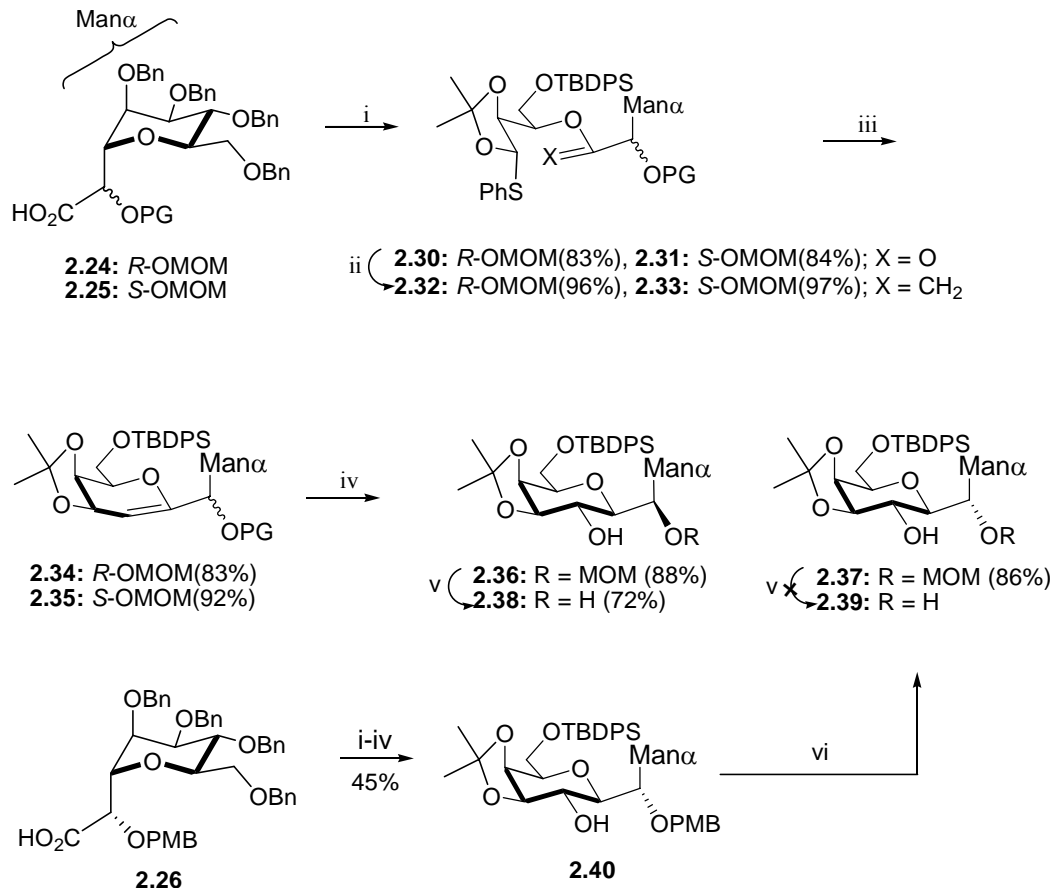
2.4 C-glycoside synthesis

The aglycone segments **2.24** and **2.25** were individually partnered with the alcohol **2.13** for the C-glycoside sequence (Scheme 2.6). DCC mediated esterification of **2.13** with **2.24** and **2.25** followed by treatment of the resulting esters with Tebbe reagent gave enol ethers **2.32** and **2.33** respectively, in 80% and 81% yield from **2.13**. The key cyclization reactions on **2.32** and **2.33** were promoted by methyl triflate in the presence of 2,6-di-tert-butyl-4-methylpyridine (DTBMP). This gave the C-1 substituted glycols **2.34**

(83%) and **2.35** (92%) respectively as the only cyclization products. Hydroboration of the glycals provided the β -C-galactosides **2.36** (88%) and **2.37** (86%), each as a single diastereomer.

The next step was the introduction of the methylene acetal residue in **2.38** and **2.39** (Scheme 2.7). The initial plan was to convert, these C-glycosides **2.36** and **2.37** directly to their cyclic methylene acetal derivatives using the conditions developed by Roush and co-workers.²⁴ However, the acidic conditions required for this transformation led to competing cleavage of the isopropylidene and methoxymethyl ether protecting groups. Therefore, in an alternative stepwise strategy selective removal of MOM acetals in **2.36** and **2.37** to diols **2.38** and **2.39**, was next attempted. Dimethylboronbromide was found to be the optimal acid promoter for this transformation, but this procedure was only useful for the conversion of **2.36** to **2.38**. Application of these conditions to **2.37** led to a mixture of several products, which afforded a low yield of **2.39** after purification. A more practical route to **2.39** was to use the PMB protected hydroxy-acid precursor **2.26** (instead of MOM ether **2.23**) in the C-glycoside synthesis. Thus, **2.26** was taken uneventfully, through the standard four-step C-glycosidation protocol to the PMB protected C-disaccharide **2.40** (45% overall yield from **2.13**). Conversion of **2.40** to diol **2.39** was best accomplished by first converting **2.40** to the acetate derivative, followed by DDQ mediated removal of the *p*-methoxybenzyl ether, and deacetylation of the product (Scheme 2.6).

Scheme 2.6: Synthesis of diols **2.38** and **2.39**



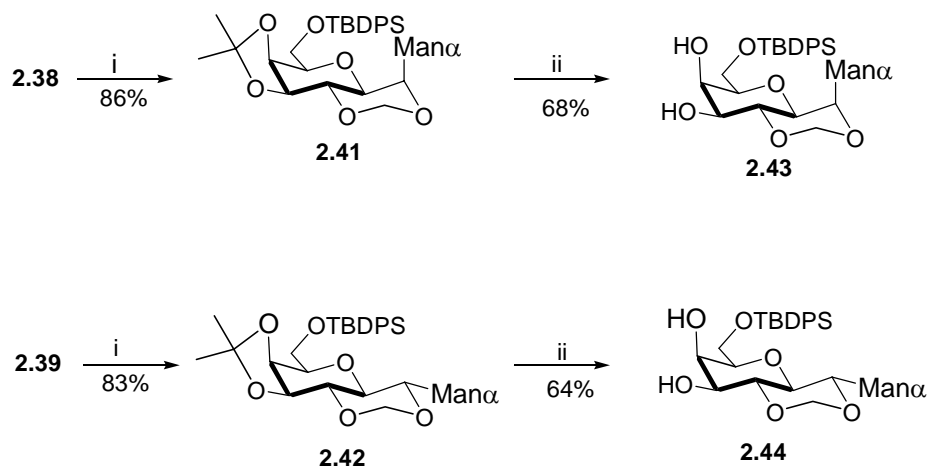
(i) DCC, DMAP, PhH; **2.13** (ii) Tebbe; (iii) MeOTf, DTBMP, CH₂Cl₂; (iv) BH₃.DMS then Na₂O₂; (v) Me₂BBr, DTBMP, CH₂Cl₂, -78 - 0 °C; (vi) (a) Ac₂O, DMAP, EtOAc; (b) DDQ, CH₂Cl₂-H₂O; (c) NaOMe, MeOH, 74%

2.5 Introduction of conformational restraints

The methylene acetal derivatives **2.41** and **2.42** were obtained by individual treatment of **2.38** and **2.39** with a mixture of dibromomethane and aqueous sodium hydroxide under phase transfer conditions.²⁵ These reactions were sometimes accompanied by small amounts of desilylated products, which could be easily reprotected. It was also found that small amounts of methyl ether side products were obtained and these were suppressed by performing the reaction in the presence of 2,3-

dimethylbutene. In preparation for the final alcohol protecting group modifications, the isopropylidene residue in **2.41** and **2.42** was selectively removed to give the respective diols **2.43** and **2.44** (Scheme 2.7).

Scheme 2.7: Synthesis of diols **2.43** and **2.44**



(i) $n\text{-Bu}_4\text{NBr}$, CH_2Br_2 , 50% aq. NaOH ; (ii) MeOH , HCl .

2.6 Stereochemical analysis of C-glycoside products

For characterization purposes **2.43** and **2.44** were transformed *via* straightforward alcohol protecting group changes to their peracetates **2.45** and **2.46** (Scheme 2.8, Table 2.1). The stereochemistry of the aglycone segment and the configuration at the intersaccharide carbon were assigned on the basis of vicinal J values. Thus, $J_{1,2} = 10.0$, $J_{2,3} = 10.0$, $J_{3,4} = 3.0$, $J_{4,5} = 0$ Hz for **2.45**, and $J_{1,2} = 9.8$, $J_{2,3} = 9.8$, $J_{3,4} = 3.0$, $J_{4,5} = 0$ Hz for **2.46** are consistent with the 3,4-O-isopropylidene- β -C-galacto motif.²⁶ A $J_{1,1'}$ value of 9.8 Hz and a NOE between H-2 of the galactose residue and the intersaccharide proton for **2.45** pointed strongly to an equatorial like attachment of the mannose residue onto a

chair-like dioxane ring. The corresponding $J_{1,1}$ value for **2.45** (6.5 Hz), is somewhat larger than expected for equatorial-axial arrangement of vicinal protons on a chair-like dioxane. It appears that in this case, the bulky pseudo-axial substituent results in a distorted, half-chair-like geometry for ring **B** leading to the unexpectedly large J value. Similar NMR data was obtained for the later derivatives **2.45** and **2.46** (*vide infra*).

Scheme 2.8: Synthesis of restrained peracetylated C-glycosides

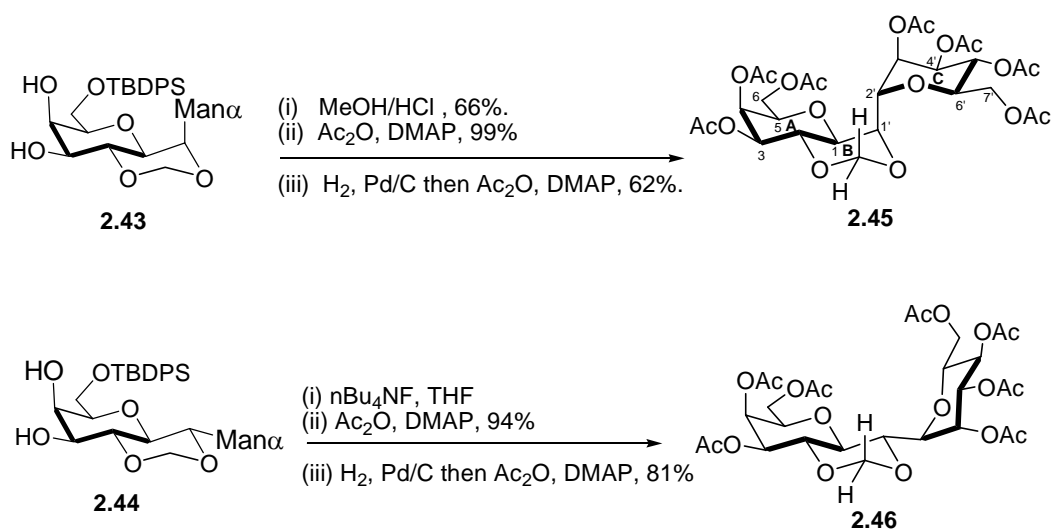


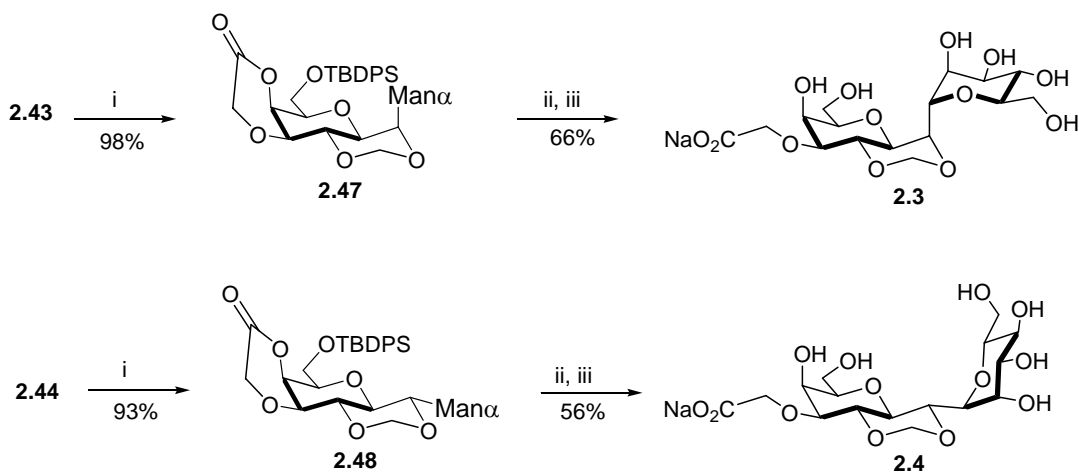
Table 2.1: $^1\text{H-NMR}$ of C-glycosides **2.45** and **2.46**.

Position	2.45/ ^1H NMR (J, Hz)	2.46/ ^1H NMR (J, Hz)
1'	4.19 (m)	3.77 (dd, J = 2.2, 9.8)
2'	4.56 (m)	4.46 (s)
3'	5.78 (dd, J = 3.0, 6.0)	5.84 (bs)
4'	5.70 (dd, J = 3.5, 5.8)	5.96 (dd, J = 4.0, 9.0)
5'	5.47 (t, J = 5.8)	5.72 (t, J = 9.0)
6'	4.07, (q, J = 5.7)	4.28 (m)
7'a	4.56 (m)	4.56 (dd, J = 5.0, 11.0)
7'b	4.56 (m)	4.28 (m)
1	3.47 (dd, J = 6.5, 10.0)	3.23 (t, J = 9.8)
2	4.51 (t, J = 10.0)	3.69 (t, J = 9.8)
3	5.19 (dd, J = 3.0, 10.0)	5.15 (dd, J = 3.0, 9.8)
4	5.55 (t, J = 3.0)	5.47 (d, J = 3.0)
5	3.33 (t, J = 6.0)	3.30 (t, J = 6.5)
6a	4.00 (dd, J = 6.0, 11.6)	3.99 (dd, J = 6.5, 10.8)
6b	4.19 (m)	4.10 (m)
O-CHa-O	5.06 (d, J = 6.0)	4.62 (d, J = 6.5)
O-CHbO	4.71 (d, J = 6.0)	buried at 4.10
CH ₃ CO-	1.56, 1.65, 1.67, 1.68, 1.75, 1.96	1.63, 1.68, 1.73, 1.74, 1.75, 1.78

2.7 “End games” for conformationally restrained mimetics **2.3** and **2.4**

Finally, diols **2.43** and **2.44** were transformed to **2.3** and **2.4** *via* a reaction sequence that involved the selective dibutyltin oxide mediated alkylation of diol **2.43** and **2.44** with methyl bromoacetate.¹⁴ This led to selective 3-O-alkylation followed by *in situ* lactonization to give **2.47** and **2.48** respectively. Exposure of these products to aqueous sodium hydroxide led to concomitant saponification and desilylation to give the corresponding dihydroxy acids, which were subjected to hydrogenolysis (Scheme 2.9). The target compounds **2.3** and **2.4** were obtained after purification using both reverse and normal phase chromatography, and lyophilization from aqueous solutions.

Scheme 2.9: Synthesis of the **2.3** and **2.4**



(i) Bu_2SnO , PhCH_3 , $\text{BrCH}_2\text{CO}_2\text{Me}$, $n\text{-Bu}_4\text{NI}$; (ii) aq. NaOH , EtOH ; (iii) H_2 , Pd/C , HCOOH , MeOH , then aq. NaOH

2.8 Summary

The conformationally restrained C-glycosides **2.3/2.4** were prepared using the oxocarbenium ion-enol cyclization methodology that was developed in our group. These structures were designed to probe the optimal conformation requirements for binding of

their disaccharide framework to P-selectin. The conformational behavior and P-selectin binding of **2.3** and **2.4** will be discussed in chapter 4. The stereochemistry was assigned through analysis of vicinal J values from ^1H -NMR data.

2.9 Experimental section

General

Unless otherwise stated, all reactions were carried out under nitrogen atmosphere in oven-dried glassware using standard syringe and septa technique. Chemical shifts are relative to the deuterated solvent peak or the tetramethylsilane (TMS) peak at (δ 0.00) and are in parts per million. The ^1H and ^{13}C NMR spectra were recorded on 300, 400 or 500 MHz instruments. Signals for selected nuclei were assigned through ^1H COSY experiments. High resolution mass spectroscopic (HRMS) data was obtained at the Mass Spectrometry Facility at University of Illinois, Urbana. Optical rotations are given in units of 10^{-1} deg/cm² g at 589 nm (sodium D-line).

Thin layer chromatography (TLC) was done on 0.25 mm thick precoated silica gel HF₂₅₄ alumina sheets (Whatman). The chromatograms were observed under UV light and, or were visualized by heating plates that were dipped in ammonium molybdate solution. Unless otherwise stated, flash column chromatography (FCC) was performed using silica gel 60 (230-400 mesh) and employed a stepwise solvent polarity gradient, correlated with TLC mobility.

1-Thio-1,2-*O*-isopropylidene acetal 2.13.

Compound **2.13** was prepared by the previously reported procedures.¹⁴ ¹H NMR (300 MHz, CDCl₃) δ 1.07 (s, 9H), 1.47, 1.49 (both s, 6H), 2.32 (br s, 1H, D₂O), 3.80 (m, 3H), 4.18 (dd, J = 2.0, 7.0 Hz, 1H), 5.44 (d, J = 7.0 Hz, 1H), 7.20-7.80 (m, 15H); ¹³C NMR (75 MHz, CDCl₃) δ 19.5, 26.3, 27.1, 27.5, 65.3, 70.1, 80.4, 85.4, 111.5, 127.6, 127.9, 129.1, 129.9, 132.0, 133.3, 134.0, 135.7; ESMS 531 (M + Na). FABHRMS calcd for C₂₃H₃₁O₄Si (M – SC₆H₆) 399.1992, found 399.1992.

Aldehyde 2.16

To a solution of methyl- α -D-mannopyranoside (5.0 g, 25.8 mmol) in dry DMF (50 mL) at 0 °C was added NaH (5.15 g, 60% in mineral oil, 0.13 mol) and Bu₄Ni (0.95 mg, 0.26 mol) followed by BnBr (15.3 mL, 0.13 mmol). The reaction was stirred for 0.5 hour at rt under an argon atmosphere, then quenched with water and extracted with ether (2×50 mL). The organic layer was washed with water (5 × 10 mL), dried (Na₂SO₄) and concentrated *in vacuo*. FCC of the crude material gave the tetrabenzylated product (13.6 g, 95%), R_f = 0.80 (15% EtOAc/petroleum ether); ¹H NMR (300 MHz, CDCl₃) δ 3.30 (s, 3H), 3.38 (m, 6H), 4.70 (m, 9H), 7.40 (m, 20H).

Trimethylsilyl triflate (1.8 mL, 10.1 mmol) was added dropwise to a solution of the tetrabenzylated methyl- α -D-mannoside (11.2 g, 20.2 mmol) and allyltrimethylsilane (6.5 mL, 40.5 mmol) in anhydrous acetonitrile (50 mL) at 0 °C under an atmosphere of argon. The solution was warmed to rt and stirred for an additional 16 hours. The resulting deep-orange solution was diluted with CH₂Cl₂. The combined organic phase was washed with brine, dried (Na₂SO₄) and concentrated *in vacuo*. FCC of the crude

material gave the α -C-allyl derivative (9.4 g, 90%); colorless oil; $R_f = 0.20$ (5% EtOAc/petroleum ether); $^1\text{H NMR}$ (300 MHz, CDCl_3) δ 2.29-2.38 (m, 2H), 3.62 (dd, $J = 3.5, 4.6$ Hz, 1H), 3.70 (dd, $J = 3.5, 10.3$ Hz, 1H), 3.82 (m, 2H), 3.82 (m, 1H), 3.87 (dd, $J = 6.8, 13.0$ Hz, 1H), 4.05 (ddd, $J = 4.8, 6.2, 7.8$ Hz, 1H), 4.51-4.61 (m, 7H), 4.70 (d, $J = 11.3$ Hz, 1H), 5.00-5.75 (m, 1H), 7.18-7.34 (m, 20H); $^{13}\text{C NMR}$ (75 MHz, CDCl_3) δ 34.6, 69.1, 71.5, 72.2, 72.3, 73.2, 73.6, 73.9, 74.8, 76.9, 117.2, 127.6, 128.0, 128.1, 128.3, 134.3, 138.2, 138.4.

The product from the previous step (1.27 g, 2.25 mmol) and $\text{PdCl}_2(\text{CH}_3\text{CN})_2$ (18 mg, 0.07 mmol) were dissolved in anhydrous benzene (75 mL) and refluxed for 16 hours.²⁶ Evaporation of the solvent *in vacuo* followed by FCC of the crude mixture gave the α -C-1-propenyl derivative (11 g, 88%) as a mixture of *E/Z* isomers (4:1 from $^1\text{H-NMR}$); $R_f = 0.45$ (10% EtOAc/petroleum ether); $^1\text{H NMR}$ (300 MHz, CDCl_3) δ 1.64, 1.71 (d, $J = 6.4$ Hz, 3H ea), 4.53-4.55 (m), 4.53-4.91(m), 5.46 (dd, $J = 1.5, 4.03$ Hz, 1H), 5.52 (dd, $J = 1.4, 5.1$ Hz, 1H), 5.59-5.71 (m, 2H), 7.21-7.42 (m, 40H); $^{13}\text{C NMR}$ (75 MHz, CDCl_3) δ 13.7, 18.2, 69.8, 70.7, 71.9, 72.2, 73.5, 74.0, 74.5, 75.5, 76.6, 78.8, 79.4, 127.4-128.2 (several lines), 138.4, 138.5; FABMS 587.2 (M + Na).

A solution of the product from the previous step (1.44 g, 2.55 mmol) in CH_2Cl_2 (50 mL) was cooled to -78 °C. Then a mixture of O_3/O_2 was bubbled through the solution until TLC indicated complete disappearance of the starting material. The solution was then purged with nitrogen, warmed to rt and treated with triphenylphosphine (0.73 g, 2.81 mmol) and stirred vigorously for 30 min. Evaporation of the solvent *in vacuo* gave a yellow oil. FCC of the crude product afforded the aldehyde **2.16** (1.31 g, 93%), $R_f = 0.44$ (30% EtOAc/petroleum ether); $^1\text{H NMR}$ (300 MHz, CDCl_3) δ 3.63 (m,

1H), 3.93 (m, 2H), 4.11 (t, J = 8.4 Hz, 1H), 4.36 (d, J = 2.6 Hz, 1H), 4.54-4.56 (m, 8H), 4.97 (d, J = 11.0 Hz, 1H), 7.44 (m, 20 H), 9.91 (d, J = 1.8 Hz, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 69.3, 72.0, 72.1, 72.9, 73.3, 74.1, 74.5, 76.6, 79.1, 79.8, 127.4, 127.5, 127.6, 127.8, 128.1, 128.2, 137.7, 137.8, 138.1, 202.9; FABMS 575.2 (M + Na).

Allylic alcohols **2.17** and **2.18**

Vinylmagnesium bromide (31.0 mL, 1M in THF, 31.0 mmol) was slowly added to a solution of aldehyde **2.16** (5.78 g, 10.0 mmol) in dry THF (25 mL) at -15 °C. The reaction mixture was stirred at this temperature for 1.75 hours then at rt for 0.5 hour. The reaction mixture was poured into saturated aqueous NH₄Cl at 0 °C and extracted with ether. The organic phase was dried (Na₂SO₄), filtered and evaporated *in vacuo*. FCC of residue gave the mixture of alcohols **2.17/2.18** (rel. ratio ca 3/2, 4.45 g, 73%) as a colorless oil; R_f = 0.62 (15% EtOAc/petroleum ether). For characterization purposes a sample of this mixture was separated by repeated FCC.

For allylic alcohol 2.17: ¹H NMR (500 MHz, CDCl₃) δ 2.50 (s, D₂O ex, 1H), 3.70 (dd, J = 4.6, 10.3 Hz, 1H), 3.75-3.89 (m, 5H), 4.01 (m, 1H), 4.33 (apparent t, J = 5.3 Hz, 1H), 4.49-4.62 (m, 8H), 5.18 (dt, J = 1.6, 10.5 Hz, 1H), 5.33 (dt, J = 1.6, 17.2 Hz, 1H), 5.90 (ddd, J = 5.8, 10.7, 16.1 Hz, 1H), 7.24-7.41 (m, 20H); ¹³C NMR (75 MHz, CDCl₃) δ 69.0, 70.6, 72.1, 72.7, 72.9, 73.4, 74.1, 74.9, 76.0, 116.6, 127.6-128.4 (several lines), 137.9, 138.2, 138.4; FABMS 603.2 (M + Na).

For allylic alcohol 2.18: ¹H NMR (500 MHz, CDCl₃) δ 2.36 (brd s, D₂O ex, 1H), 3.72 (dd, J = 5.1, 10.3 Hz, 1H), 3.85 (m, 2H), 3.94 (m, 2H), 4.02 (t, J = 6.2 Hz, 1H), 4.10 (m, 1H), 4.49-4.64 (m, 8H), 5.25 (d, J = 10.5 Hz, 1H), 5.35 (d, J = 17.6 Hz, 1H), 6.04 (ddd, J

= 6.4, 10.6, 17.2 Hz, 1H), 7.24-7.43 (m, 20H); ^{13}C NMR (125 MHz, CDCl_3) δ 68.7, 71.4, 72.5, 72.85, 72.93, 73.4, 73.6, 74.3, 74.6, 75.2, 75.3, 116.8, 127.7-128.6 (several lines), 137.1, 138.1, 138.3, 138.4; ESMS 598.4 ($\text{M}+\text{NH}_4^+$).

MOM-protected alcohols 2.19 and 2.20

MOMCl (0.91 mL, 11.9 mmol) was added to a mixture of **2.17/2.20** (2.30 g, 3.97 mmol), *i*-Pr₂NEt (2.76 mL, 15.9 mmol) and anhydrous CH_2Cl_2 (40 mL) at 0 °C. The mixture was refluxed for 4 hours at 50 °C, then washed with brine and dried (Na_2SO_4). Evaporation of the solvent under reduced pressure and FCC of the residue provided **2.19** (0.65 g), **2.20** (0.88 g) and a mixture of MOM products **2.19/2.10** (0.62 g) (ca. 90% overall yield).

For **2.19**: R_f = 0.63 (20% EtOAc/petroleum ether); ^1H NMR (300 MHz, CDCl_3) δ 3.29 (s, 3H), 3.42 (m, 1H), 3.72-4.19 (m, 6H), 4.40 (t, J = 7.7 Hz, 1H), 4.53-4.89 (m, 10H), 5.11 (d, J = 10.6 Hz, 1H), 5.35 (d, J = 14.6 Hz, 1H), 5.89 (ddd, J = 6.7, 8.0, 11.7 Hz, 1H), 7.28-7.51 (m, 20 H); ^{13}C NMR (75 MHz, CDCl_3) δ 55.8, 69.6, 71.7, 72.1, 73.4, 74.1, 75.1, 75.6, 76.1, 78.4, 94.1, 119.6, 127.4-128.3 (several lines), 134.9, 138.5; FABHRMS calcd for $\text{C}_{39}\text{H}_{44}\text{O}_7\text{Na}$ ($\text{M} + \text{Na}$) 647.2985, found 647.2985.

For **2.20**: R_f = 0.66 (20% EtOAc/petroleum ether); ^1H NMR (300 MHz, CDCl_3) δ 3.34 (s, 3H), 3.72-4.03 (m, 7H), 4.32 (t, J = 7.1 Hz, 1H), 4.53-4.77 (m, 10H), 5.15 (d, J = 11.4 Hz, 1H), 5.20 (d, J = 18.0 Hz, 1H), 5.61 (ddd, J = 5.3, 9.1, 15.9 Hz, 1H), 7.24-7.38 (m, 20 H); ^{13}C NMR (75 MHz, CDCl_3) δ 56.0, 69.9, 71.8, 72.3, 73.6, 73.9, 74.0, 75.1, 75.5, 76.1, 94.2, 119.3, 127.5-128.5 (several lines), 138.5, 138.7, 138.9; FABHRMS calcd for $\text{C}_{39}\text{H}_{44}\text{O}_7\text{Na}$ ($\text{M} + \text{Na}$) 647.2985, found 647.2985.

MOM-protected acid 2.24

Alkene **2.19** (3.90 g, 6.25 mmol) was subjected to the ozonolysis procedure described for the preparation of **2.16**. The aldehyde derivative was obtained (1.47 g, 94%). $R_f = 0.63$ (20% EtOAc/petroleum ether); $^1\text{H NMR}$ (500 MHz, CDCl_3) δ 3.34 (s, 3H), 3.62 (dd, $J = 6.0, 10.0$ Hz, 1H), 3.66 (t, $J = 3.8$ Hz, 1H), 3.75 (m, 3H), 3.97 (dd, $J = 3.0, 8.5$ Hz, 1H), 4.05 (dt, $J = 2.5, 6.5$ Hz, 1H), 4.16 (m, 1H), 4.26 (dd, $J = 3.0, 8.0$ Hz, 1H), 4.35-4.52 (m, 8H), 4.58 (dd, $J = 3.3, 8.6$ Hz, 1H), 4.74 (q, $J = 6.5$ Hz, 2H), 7.13-7.32 (m, 20H), 9.48 (s, 1H); $^{13}\text{C NMR}$ (75 MHz, CDCl_3) δ 56.2, 68.5, 71.7, 72.2, 72.8, 73.4, 74.4, 74.9, 75.4, 82.7, 97.5, 127.6, 127.7, 128.1, 128.2, 128.4, 138.0, 138.2, 138.4, 202.4; ESIMS 649.2 (M+Na).

To a mixture of the aldehyde from the previous step (1.48 g, 2.36 mmol) and $\text{NaH}_2\text{PO}_4 \cdot \text{H}_2\text{O}$ (3.26 mg, 23.6 mmol), in 5:1 $\text{CH}_3\text{CN}-\text{H}_2\text{O}$ (30 mL) were successively added 30% H_2O_2 solution (1.0 mL) and a solution of NaClO_2 (1.1 g, 12.2 mmol) in water (10 mL) at 0 °C. The reaction mixture was stirred for 1 hour at rt, then quenched by addition of Na_2SO_3 (1.50 g) and extracted with EtOAc. The organic phase washed with water, brine, dried (Na_2SO_4) and concentrated under reduced pressure. FCC of the residue afforded acid **2.24** (1.50 g, 99%): colorless oil; $R_f = 0.51$ (10% MeOH/ CHCl_3); $^1\text{H NMR}$ (300 MHz, C_6D_6) δ 3.23 (s, 3H), 3.85 (d, $J = 5.5$ Hz, 1H), 3.88 (m, 1H), 3.95 (t, $J = 4.8$ Hz, 1H), 4.30-4.44 (m, 8H), 4.51 (t, $J = 11.0$ Hz, 1H), 4.61-4.71 (m, 5H), 4.81 (dd, $J = 4.0, 7.0$ Hz, 1 H), 7.01-7.45 (m, 20H); $^{13}\text{C NMR}$ (75 MHz, CDCl_3) δ 56.0, 68.7, 71.7, 72.3, 72.6, 72.9, 73.1, 74.7, 75.1, 75.8, 76.8, 77.2, 77.6, 96.4, 127.4-128.2 (several lines), 138.0, 173.4; EIHRMS calcd for $\text{C}_{38}\text{H}_{46}\text{O}_9\text{SiN}$ (M + NH_4^+) 660.2; FABHRMS calcd for $\text{C}_{38}\text{H}_{42}\text{O}_9\text{Na}$ (M + Na) 665.2724, found 665.2727.

MOM-protected acid 2.25

Alkene **2.20** (3.90 g, 6.25 mmol) was subjected to the ozonolysis procedure described for the preparation of **2.24**. The aldehyde derivative (3.31 g, 85%) was obtained: $R_f = 0.44$ (30% EtOAc/petroleum ether); $^1\text{H NMR}$ (300 MHz, CDCl_3) δ 3.40 (s, 3H), 3.74 (m, 3H), 3.93 (t, $J = 3.3$ Hz, 1H), 4.01 (dd, $J = 3.0, 6.0$ Hz, 1H), 4.17 (m, 2H), 4.32-4.53 (m, 8H), 4.58 (dd, $J = 3.3, 8.6$ Hz, 1H), 4.73 (d, $J = 6.9$ Hz, 1H), 7.17-7.37 (m, 20H), 9.72 (d, $J = 1.1$ Hz, 1H); $^{13}\text{C NMR}$ (75 MHz, CDCl_3) δ 56.3, 68.3, 70.8, 71.3, 71.9, 72.3, 72.4, 73.2, 73.5, 73.9, 75.0, 82.1, 97.7, 127.4-128.3 (several lines), 137.7, 137.8, 137.9, 138.2, 202.6 FABHRMS calcd for $\text{C}_{38}\text{H}_{42}\text{O}_8\text{Na}$ ($M + \text{Na}$) 649.2769, found 649.2777.

The aldehyde (416 mg, 0.66 mmol) from the previous step was converted, following the procedure that was used for **2.24**, to acid **2.25** (410 mg, 97%): colorless oil; $R_f = 0.47$ (10% MeOH/ CHCl_3); $^1\text{H NMR}$ (300 MHz, CDCl_3) δ 3.49 (s, 3H), 3.74 (m, 3H), 3.91 (m, 2H), 4.01 (t, $J = 3.7$ Hz, 1H), 4.17 (m, 2H), 4.39 (m, 1H), 4.55-4.71 (m, 10H), 4.77 (d, $J = 7.3$ Hz, 1H), 4.85 (d, $J = 6.9$ Hz, 1H), 7.31-7.49 (m, 20H), 11.22 (s, 1H); $^{13}\text{C NMR}$ (75 MHz, CDCl_3) δ 56.5, 68.6, 70.8, 72.1, 72.2, 72.8, 73.2, 74.1, 74.4, 74.9, 75.7, 97.2, 127.4-128.3 (several lines), 137.8, 137.9, 138.0, 138.2, 174.5; FABHRMS calcd for $\text{C}_{38}\text{H}_{42}\text{O}_9\text{Na}$ ($M + \text{Na}$) 665.2727, found 665.2722.

***p*-Methoxybenzyl-protected acid 2.26**

To a solution of alcohol **2.18** (1.56 g, 2.69 mmol) in dry DMF (10 mL) at 0 °C under an argon atmosphere was added NaH (0.54 g, 60% in mineral oil, 13.4 mmol) and Bu_4NI (0.40 g, 1.08 mol) followed by *p*-methoxybenzyl chloride (2.20 mL, 16.1 mmol).

The reaction was stirred for 1 hour, then quenched with methanol (1 mL) and extracted with ether. The organic layer was washed with water, dried (Na_2SO_4) and concentrated *in vacuo*. FCC of the residue gave *p*-methoxybenzylated derivative **2.23** (1.70 g, 90%); $R_f = 0.58$ (20% EtOAc/petroleum ether); ^{13}C NMR (75 MHz, CDCl_3) 55.4, 70.0, 71.7, 72.0, 73.0, 74.7, 75.3, 76.4, 78.2, 79.5, 114.1, 119.4, 127.5-129.7 (several lines), 130.4, 135.7, 138.7, 138.8, 159.4; FABHRMS calcd for $\text{C}_{45}\text{H}_{48}\text{O}_7\text{Na}$ ($M + \text{Na}$) 723.3298, found 723.3300.

Treatment of **2.23** (2.64 g, 3.77 mmol) according to the standard ozonolysis procedure provided the derived aldehyde (2.20 g, 83%); $R_f = 0.31$ (20% EtOAc/petroleum ether); ^1H NMR (500 MHz, CDCl_3) δ 3.64 (dd, $J = 5.0, 10.0$ Hz, 1H), 3.70-3.84 (m, 4H), 3.78 (s, 3H buried), 4.04 (m, 2H), 4.32 (m, 2H), 4.47-4.57 (m, 8H), 4.65 (d, $J = 9.0$ Hz, 1H), 6.83 (d, $J = 8.5$ Hz, 2H), 7.20-7.37 (m, 22H), 9.52 (d, $J = 1.0$ Hz, 1H); ^{13}C NMR (75 MHz, CDCl_3) 55.1, 68.5, 71.5, 72.3, 72.6, 72.8, 73.0, 73.1, 75.1, 83.4, 113.9, 127.5-129.7 (several lines), 138.0, 138.2, 138.4, 159.4, 200.2; ESIMS 741.0 ($M + \text{K}$)

The aldehyde from the previous step (2.20 g, 3.13 mmol), was oxidized to the acid derivative, following the procedure that was used for acid **2.24**. Compound **2.26** (2.10 g, 93%) was obtained as a colorless oil; $R_f = 0.36$ (10% MeOH/ CHCl_3); ^1H NMR (500 MHz, CDCl_3) δ 3.64 (dd, $J = 5.0, 10.0$ Hz, 1H), 3.72 (m, 2H), 3.78 (s, 3H), 4.06 (dd, $J = 2.5, 7.5$ Hz, 1H), 4.12 (m, 1H), 4.30 (d, $J = 3.0$ Hz, 1H), 4.34 (m, 1H), 4.37 (bs, 1H), 4.42-4.64 (m, 10H), 6.81 (d, $J = 9.0$ Hz, 1H), 7.31-7.49 (m, 22H); ^{13}C NMR (75 MHz, CDCl_3) δ 55.2, 68.9, 71.7, 72.4, 72.8, 73.0, 75.0, 75.5, 76.4, 114.0, 127.6-129.9, 138.1,

138.3, 159.6, 172.9; ESIHRMS calcd for $C_{44}H_{46}O_9Na$ (M + Na) 741.3037, found 741.3040.

4-Nitrobenzoate ester **2.27**

A mixture of alcohol **2.17** (2.78 g, 4.79 mmol), triphenylphosphine (5.04 g, 19.2 mmol), and 4-nitrobenzoic acid (3.21 g, 19.2 mmol) was dissolved in dry toluene (150 mL). DIAD (3.78 mL, 19.2 mmol) was then added dropwise at 0 °C under an atmosphere of nitrogen. The mixture was allowed to warm to rt, and stirred for an additional 1 hour. After concentration under reduced pressure the mixture was neutralized with aqueous $NaHCO_3$, and extracted with ether. The ethereal extracts were combined, dried (Na_2SO_4) and concentrated *in vacuo*. The crude product was purified by FCC to give the ester **2.27** (2.83 g, 81%), colorless oil, $R_f = 0.44$ (20% EtOAc/petroleum ether); 1H NMR (300 MHz, $CDCl_3$) δ 3.73 (dd, J = 3.0, 10.3 Hz, 1H), 3.86-4.11(m, 5H), 4.33 (t, J = 5.7 Hz, 1H), 4.50-4.71 (m, 8H), 5.41 (d, J = 7.3 Hz, 1H), 5.46 (d, J = 16.1 Hz, 1H), 6.14 (ddd, J = 7.3, 7.9, 16.1 Hz, 1H), 7.29-7.39 (m, 20H), 8.12 (m, 4H); ^{13}C NMR (75 MHz, $CDCl_3$) δ 68.7, 71.3, 72.7, 73.1, 73.3, 74.6, 75.0, 75.2, 75.5, 120.5, 123.4, 127.5-128.3 (several lines), 130.7, 132.0, 135.3, 137.8, 137.9, 138.1, 138.2, 150.4, 163.6; ESIMS 752.2 (M + Na).

Tetra-O-acetate **2.28**

MOM protected alkene **2.19** (86 mg, 0.14 mmol), 10% Pd on carbon (170 mg), formic acid (0.25 mL) and methanol (5 mL) was stirred under an atmosphere of hydrogen (balloon), for 12 hours. The reaction mixture was purged with argon and filtered through

a bed of Celite. The filtrate was concentrated *in vacuo*, dissolved in EtOAc (5 mL) and treated with acetic anhydride (0.1 mL) and DMAP (5 mg). FCC of the crude mixture afforded the dihydro-tetraacetate derivative (57 mg, 92%) as a colorless oil; $R_f = 0.50$ (40% EtOAc/petroleum ether); $^1\text{H NMR}$ (500 MHz, C_6D_6) δ 0.90 (t, $J = 7.3$ Hz, 3H, H-3'), 1.49, 1.60 (m, 1H ea, H-2'), 1.65, 1.66, 1.73, 1.74, (all s, 3H ea, $\text{CH}_3\text{CO} \times 4$), 3.33 (s, 3H, CH_3O -), 3.60 (m, 1H, H-5), 3.82 (ddd, $J = 6.5, 8.5, 12.5$ Hz, 1H, H-1'), 4.01 (dd, $J = 2.0, 6.5$ Hz, 1H, H-1), 4.16 (dd, $J = 2.0, 12.0$ Hz, 1H, H-6a), 4.36 (dd, $J = 6.0, 12.0$ Hz, 1H, H-6b), 4.53 (q, $J = 6.0$ Hz, 2H, O- CH_2 -O), 5.57 (dd, $J = 3.5, 10.0$ Hz, 1H, H-3), 5.63 (apparent t, $J = 9.5$ Hz, 1H, H-4), 6.02 (br s, 1H, H-2); FABHRMS calcd for $\text{C}_{19}\text{H}_{31}\text{O}_{11}$ (M + Na) 458.1765, found 458.1764.

To a solution of the product from the previous step (40 mg, 0.09 mmol) in anhydrous CH_2Cl_2 (2 mL), at -78 °C, was added dimethylboron bromide (1.14 M, 0.29 mL, 0.33 mmol). The reaction mixture was stirred at rt for 30 min, then poured into a 1:1 mixture of THF in saturated NaHCO_3 and extracted with ether. The organic layer was washed with aqueous NaHSO_4 and brine, dried (Na_2SO_4), filtered and concentrated under reduced pressure. FCC of the residue afforded **2.28** (12 mg, 34%): colorless oil; $^1\text{H NMR}$ (500 MHz, C_6D_6) δ 0.84 (t, $J = 7.5$ Hz, 3H, H-3'), 1.47 (m, 2H, H-2'), 1.61, 1.67, 1.71 (all s, 3H, 3H 6H resp. CH_3CO), 1.86 (br s, 1H, D_2O exchange) 3.38 (m, 1H, H-1'), 3.78 (t, $J = 4.0$ Hz, 1H, H-1), 4.10 (dd, $J = 3.5, 12.0$ Hz, 1H, H-6a), 4.24 (dt, $J = 3.0, 7.0$ Hz, 1H, H-5), 4.55 (dd, $J = 6.5, 12.0$ Hz, 1H, H-6b), 5.44 (t, $J = 7.5$ Hz, 1H, H-4), 5.69 (t, $J = 4.0$ Hz, 1H, H-2), 5.91 (dd, $J = 3.5, 7.5$ Hz, 1H, H-3); FABHRMS calcd for $\text{C}_{17}\text{H}_{27}\text{O}_{10}$ (M + H) 391.1604, found 391.1604.

MOM-tetra-O-acetate 2.29

Treatment of MOM protected alkene **2.20** (70 mg, 0.11 mmol) following the hydrogenation-acetylation sequence described in the synthesis of **2.28** provided MOM-tetraacetate **2.29** (36 mg, 75%) as a colorless oil; ^1H NMR (500 MHz, C_6D_6) δ 0.79 (t, J = 7.5 Hz, 3H, H-3'), 1.65, 1.78 (m, 2H, H-2'a, b), 1.63, 1.70, 1.73, 1.74 (all s, 3H ea, $\text{CH}_3\text{CO} \times 4$), 3.26 (s, 3H, CH_2O -), 3.52 (m, 1H, H-1'), 4.00 (t, J = 3.3 Hz, 1H, H-1), 4.23 (dd, J = 3.0, 12.5 Hz, 1H, H-6a), 4.36 (m, 1H, H-5), 4.46 (A of ABq, $\Delta\delta$ = 0.13 ppm, J = 6.5 Hz, 1H, O- CH_a -O), 4.49 (dd, J = 6.0, 12.0 Hz, 1H, H-6b), 4.59 (B of ABq, $\Delta\delta$ = 0.13 ppm, J = 6.5 Hz, 1H, O- CH_b -O), 5.62 (t, J = 9.0 Hz, 1H, H-4), 5.69 (t, J = 3.0 Hz, 1H, H-2), 5.91 (dd, J = 3.5, 9.0 Hz, 1H, H-3).

Thioacetal ester 2.30

DCC (324 mg, 1.58 mmol) was added at 0 °C to a solution of alcohol **2.13** (478 mg, 0.94 mmol), acid **2.24** (400 mg, 0.63 mmol), and DMAP (16 mg, 0.13 mmol) in anhydrous benzene (50 mL). The reaction was warmed to rt and stirred for 2 hours at this temperature. The mixture was diluted with ether (10 mL) and filtered. The filtrate was washed with 0.1 N aqueous HCl and brine, dried (Na_2SO_4), filtered, and evaporated under reduced pressure. The residue was purified by FCC to give ester **2.30** (525 mg, 74% based on recovered **2.13**): colorless oil; R_f = 0.32 (10% EtOAc/petroleum ether); IR (neat) 1759 cm^{-1} ; ^1H NMR (300 MHz, CDCl_3) δ 1.16 (s, 9H), 1.47, 1.54, (both s, 3H ea), 3.38 (s, 3H), 3.62 (dd, J = 5.2, 9.4 Hz, 1H), 3.88-3.97 (m, 3H), 4.08-4.24 (m, 3H), 4.30-4.80 (m, 14H), 5.51 (m, 1H), 5.57 (d, J = 6.6 Hz, 1H), 7.34-7.49 (m, 29H), 7.66 (m, 2H), 7.81 (m, 4H); ^{13}C NMR (75 MHz, CDCl_3) δ 26.5, 27.0, 27.4, 56.4, 62.5, 69.5, 71.2, 72.2,

72.6, 72.9, 73.2, 74.9, 75.8, 79.2, 84.8, 96.1, 111.6, 127.3-129.8 (several lines), 132.8, 133.0, 133.4, 135.6, 135.7, 138.2, 138.4 (three signals), 138.8, 169.7; FABHRMS calcd for $C_{67}H_{76}O_{12}NaSiS$ ($M + Na$) 1155.4724, found 1155.4709.

Thioacetal ester **2.31**

Thioacetal ester **2.31** (1.63 g, 84% based on **2.13**): colorless oil; $R_f = 0.48$ (15% EtOAc/petroleum ether); 1H NMR (500 MHz, C_6D_6) δ 1.15 (s, 9H), 1.48, 1.49 (both s, 3H ea.), 3.08 (s, 3H), 3.66 (dd, $J = 4.0, 10.3$ Hz, 1H), 3.76 (m, 2H), 4.09-4.28 (m, 6H), 4.39-4.62 (m, 10H), 4.71 (t, $J = 6.0$ Hz, 1H), 4.78, (dd, $J = 2.5, 7.0$ Hz, 1H), 5.71 (m, 1H), 5.89 (d, $J = 7.0$ Hz, 1H), 6.90-7.34 (m, 31H), 7.69-7.81 (m, 4H); ^{13}C NMR (75 MHz, $CDCl_3$) δ 26.6, 27.0, 27.4, 56.2, 61.6, 69.1, 71.6, 72.2, 73.2, 73.3, 73.7, 75.3, 75.5, 76.8, 77.0, 77.2, 77.7, 78.7, 84.1, 96.8, 111.4, 127.4, 127.6-129.6 (several lines), 133.0, 133.3, 135.5, 135.6, 138.3, 138.4, 138.6, 169.5; FABHRMS calcd for $C_{67}H_{76}O_{12}SiSK$ ($M + K$) 1171.4464, found 1171.4464.

Thioacetal enol ether **2.32**

Tebbe reagent in THF (20.8 mL, 0.5 M, 10.4 mmol), was added dropwise under an atmosphere of argon at -78 °C to a solution of ester **2.30** (3.36 g, 2.97 mmol) and pyridine (0.3 mL) in anhydrous 3:1 toluene/THF (40 mL). The reaction mixture was warmed to rt, stirred at this temperature for 3 hours, and then slowly poured into a solution of 1 N aqueous NaOH at 0 °C. The resulting suspension was extracted with ether, and the organic phase washed with brine, dried (Na_2SO_4), filtered, and concentrated under reduced pressure. The residue was purified by FCC on basic alumina

(Brockmann I, 150 mesh) to give the thioacetal enol ether **2.32** (2.56 g, 96% based on recovered **2.30**): colorless oil, $R_f = 0.56$ (15% EtOAc/petroleum ether); ^1H NMR (300 MHz, C_6D_6) δ 1.13 (s, 9H), 1.45, 1.46 (both s, 3H ea.), 3.15 (s, 3H), 3.82 (m, 5H), 3.91 (bs, 1H), 3.95 (dd, $J = 4.5, 10.0$ Hz, 1H), 4.05 (m, 1H), 4.12 (t, $J = 9.5$ Hz, 1H), 4.24 (d, $J = 7.0$ Hz, 1H), 4.34 (m, 3H), 4.42 (d, $J = 12.0$ Hz, 1H), 4.47 (d, $J = 12.0$ Hz, 1H), 4.51 (d, $J = 12.0$ Hz, 1H), 4.52-4.62 (m, 4H), 4.74 (m, 3H), 4.97 (d, $J = 11.0$ Hz, 1H), 5.92 (d, $J = 7.5$ Hz, 1H), 7.00-7.38 (m, 28H), 7.50 (d, $J = 7.0$ Hz, 2H), 7.71 (d, $J = 7.0$ Hz, 2H), 7.77 (m, 3H); ^{13}C NMR (75 MHz, C_6D_6) δ 27.1, 27.5, 28.0, 56.1, 62.1, 71.2, 72.1, 72.3, 74.2, 75.4, 75.6, 75.7, 75.8, 76.0, 76.3, 80.4, 80.7, 84.8, 88.0, 94.6, 112.1, 127.8-130.6 (several lines), 132.3, 133.6, 133.8, 135.2, 136.3, 139.5, 139.7, 139.9, 158.1; FABHRMS calcd for $\text{C}_{68}\text{H}_{78}\text{O}_{11}\text{NaSiS}$ ($M + \text{Na}$) 1153.4932 found 1153.4966.

Thioacetal enol ether **2.33**

Thioacetal enol ether **2.33** (0.88 g, 96% based on recovered **2.32**): colorless oil, $R_f = 0.67$ (15% EtOAc/petroleum ether); ^1H NMR (500 MHz, C_6D_6) δ 1.16 (s, 9H), 1.56, 1.59 (both s, 3H ea.), 2.94 (s, 3H), 3.55 (dd, $J = 5.5, 10.5$ Hz, 1H), 3.81 (t, $J = 6.5$ Hz, 1H), 3.96 (s, 1H), 4.01 (d, $J = 9.5$ Hz, 1H), 4.15 (m, 2H), 4.20-4.29 (m, 5H), 4.36-4.46 (m, 4H), 4.57-4.71 (m, 5H), 4.81-4.88 (m, 3H), 4.99 (d, $J = 11.0$ Hz, 1H), 6.31 (d, $J = 6.0$ Hz, 1H), 6.94-7.39 (m, 29H), 7.53 (d, $J = 7.0$ Hz, 2H), 7.75 (m, 4H); ^{13}C NMR (75 MHz, C_6D_6) δ 27.2, 27.7, 55.7, 62.0, 70.8, 71.7, 71.9, 73.6, 73.9, 75.2, 75.7, 75.9, 76.0, 76.1, 80.5, 80.7, 84.3, 94.4, 112.4, 127.0-130.1 (several lines), 131.7, 133.4, 133.7, 135.0, 136.0, 139.3, 139.6, 139.7, 156.8; FABHRMS calcd for $\text{C}_{68}\text{H}_{78}\text{O}_{11}\text{NaSiS}$ ($M + \text{Na}$) 1153.4938, found 1153.4932.

Glycal 2.34

A mixture of enol ether **2.32** (2.56 g, 2.27 mmol), 2,6-di-*tert*-butyl-4-methylpyridine (5.57 g, 27.3 mmol), and freshly activated, powdered 4A molecular sieves (3.0 g) in anhydrous CH₂Cl₂ (100 mL) were stirred for 15 min at rt under an argon atmosphere and then cooled to 0 °C. Methyl triflate (2.56 mL, 22.7 mmol) was then introduced, and the mixture was warmed to rt and stirred for an additional 2 days, at which time triethylamine (4.75 mL) was added. The mixture was diluted with ether, washed with saturated aqueous NaHCO₃ and brine, dried (Na₂SO₄), filtered and evaporated under reduced pressure. The residue was purified by FCC over basic alumina (Brockmann I, 150 mesh) to give glycal **2.34** (1.91 g, 83%), clear oil, R_f = 0.60 (20% EtOAc/petroleum ether); ¹H NMR (300 MHz, C₆D₆) δ 1.20 (s, 9H), 1.38, 1.53, (both s, 3H ea), 3.33 (s, 3H), 3.82 (t, J = 7.0 Hz, 1H), 3.89-3.93 (m, 7H), 4.18 (m, 2H), 4.32-4.71 (m, 11H), 4.87 (m, 2H), 4.95 (d, J = 11.4 Hz, 1H), 7.12-7.39 (m, 24H), 7.45 (d, J = 7.0 Hz, 2H), 7.82 (m, 4H); ¹³C NMR (75 MHz, C₆D₆) δ 27.4, 27.6, 29.2, 56.3, 63.9, 69.6, 71.0, 72.1, 72.4, 72.7, 73.5, 74.2, 75.1, 75.6, 76.3, 76.6, 80.8, 95.3, 102.8, 110.8, 127.8-130.5 (several lines), 134.0, 136.3, 136.4, 139.7, 139.8, 139.9, 151.9; FABHRMS calcd for C₆₂H₇₂O₁₁NaSi (M + Na) 1043.4742, found 1043.4734.

Glycal 2.35

Glycal **2.35** (435 mg, 92% based on recovered **2.33**): clear oil, R_f = 0.43 (20% EtOAc/petroleum ether); ¹H NMR (500 MHz, C₆D₆) δ 1.18 (s, 9H), 1.35, 1.50 (both s, 3H ea), 3.10 (s, 3H), 3.81 (d, J = 10.0 Hz, 1H), 3.89 (dd, J = 4.0, 10.8 Hz, 1H), 4.05-4.11 (m, 3H), 4.17 (m, 2H), 4.32-4.47 (m, 6H), 4.50 (br s, 1H), 4.57 (d, J = 7.0 Hz, 1H), 4.66

(m, 2H), 4.95 (m, 2H), 7.00-7.43 (m, 26 H), 7.80 (m, 4H); ^{13}C NMR (75 MHz, C_6D_6) δ 27.1, 27.2, 28.8, 56.0, 63.7, 69.6, 70.3, 72.2, 72.4, 73.8, 74.5, 74.7, 75.5, 75.8, 75.9, 76.1, 76.6, 76.7, 80.2, 95.1, 102.0, 110.4, 127.4-130.0 (several lines), 133.7, 133.8, 136.0, 139.4, 139.6, 152.8; FABHRMS calcd for $\text{C}_{62}\text{H}_{72}\text{O}_{11}\text{NaSi}$ (M + Na) 1043.4740, found 1043.4742.

1,3,4,5-Tetra-*O*-benzyl-13-*O*-*tert*-butyldiphenylsilyl-2,6:8,12-dianhydro-10,11-*O*-isopropylidene-7-*O*-methoxymethyl-*D*-lyxo-*D*-galacto-*D*-manno-tridecitol 2.36.

$\text{BH}_3\cdot\text{Me}_2\text{S}$ (0.02 mL, 0.23 mmol) was added at 0 °C to a solution of the glycal **2.34** (24 mg, 0.02 mmol) in anhydrous THF (2 mL) under an atmosphere of argon. The mixture was warmed to rt and stirred for an additional 1.5 hours at this temperature. At that time the solution was cooled to 0 °C and treated with a mixture of 3N NaOH (0.12 mL) and 30% aqueous H_2O_2 (0.05 mL) for 30 min. The mixture was diluted with ether, washed with saturated aqueous NaHCO_3 and brine, dried (Na_2SO_4), filtered and evaporated under reduced pressure. The residue was purified by FCC to give MOM-protected C-galactoside as the only product: **2.36** (21 mg, 88% from **2.34**), colorless oil, $R_f = 0.61$ (30% EtOAc/petroleum ether); ^1H NMR (500 MHz, CDCl_3) δ 1.00 (s, 9H), 1.33, 1.44, (both s, 3H ea), 2.29 (d, $J = 8.5$ Hz, 1H, D_2O ex), 3.23 (s, 3H), 3.33 (br s, 1H), 3.43 (d, $J = 8.8$ Hz, 1H), 3.58-3.73 (m, 5H), 3.76 (d, $J = 4.5$ Hz, 1H), 3.80 (t, $J = 8.0$ Hz, 1H), 3.83 (dd, $J = 2.0, 7.5$ Hz, 1H), 3.90 (t, $J = 9.5$ Hz, 1H), 3.95 (dd, $J = 2.0, 7.5$ Hz, 1H), 4.14 (dd, $J = 2.0, 7.5$ Hz, 1H), 4.25 (m, 2H), 4.34 (d, $J = 12.0$ Hz, 1H), 4.46 (m, 4H), 4.53 (d, $J = 6.0$ Hz, 1H), 4.65 (d, $J = 6.0$ Hz, 1H), 4.72 (d, $J = 6.0$ Hz, 1H), 4.75 (d, $J = 12.0$ Hz, 1H), 5.52 (d, $J = 6.2$ Hz, 1H), 7.13-7.31 (m, 26H), 7.60 (m, 4H); ^{13}C NMR (75

MHz, CDCl₃) δ 26.6, 28.7, 56.1, 62.5, 69.4, 70.0, 71.9, 72.6, 73.2, 73.5, 74.6, 74.8, 75.7, 76.0, 77.0, 78.7, 78.9, 98.6, 109.4, 127.7-129.9 (several lines), 133.6, 135.6, 135.8, 138.4, 138.5, 138.9; FABHRMS calcd for C₆₂H₇₄O₁₂NaSi (M + Na) 1061.4847, found 1061.4840.

1,3,4,5-Tetra-*O*-benzyl-13-*O*-*tert*-butyldiphenylsilyl-2,6:8,12-dianhydro-10,11-*O*-isopropylidene-7-*O*-methoxymethyl-*D*-lyxo-*D*-gulo-*D*-manno-tridecitol 2.37.

Application of the procedure that was used for **2.36** to glycal **2.35** (558 mg, 0.55 mmol) provided MOM protected C-glycoside **2.37** (415 mg, 86% based on recovered **2.35**): oil, R_f = 0.61 (30% EtOAc/petroleum ether); ¹H NMR (500 MHz, C₆D₆) δ 1.18 (s, 9H), 1.35, 1.56 (both s, 3H ea), 3.13 (s, 3H), 3.52 (t, J = 9.0 Hz, 1H), 3.77 (d, J = 9.5 Hz, 1H), 3.81 (dd, J = 2.0, 10.5 Hz, 1H), 3.86 (m, 2H), 3.94 (t, J = 9.5 Hz, 1H), 4.12 (t, J = 6.0 Hz, 1H), 4.16-4.27 (m, 5H), 4.32-4.62 (m, 12H), 4.88 (d, J = 11.0 Hz, 1H), 4.96 (d, J = 8.0 Hz, 1H), 7.00-7.38 (m, 26H), 7.82 (m, 4H); ¹³C NMR (75 MHz, CDCl₃) δ 26.7, 27.0, 28.6, 55.9, 63.0, 69.9, 70.0, 71.9, 72.3, 73.0, 73.6, 73.7, 73.9, 74.3, 74.5, 74.7, 74.8, 75.2, 77.2, 78.2, 79.4, 80.4, 97.1, 109.2, 127.5, 127.7, 128.0, 128.2, 128.4, 133.6, 135.6, 135.6, 135.7, 138.2, 138.4, 138.6; ESIHRMS calcd for C₆₂H₇₄O₁₂NaSi (M + Na) 1061.4847, found 1061.4847.

1,3,4,5-Tetra-*O*-benzyl-13-*O*-*tert*-butyldiphenylsilyl-2,6:8,12-dianhydro-10,11-*O*-isopropylidene-*D*-lyxo-*D*-galacto-*D*-manno-tridecitol 2.38.

Dimethylboron bromide (0.18 mL, 1.85 mmol) was added at -78 °C, to a mixture of MOM-protected C-glycoside **2.36** (320 mg, 0.31 mmol), 2,6-di-*tert*-butyl-4-

methylpyridine (252 mg, 1.23 mmol), freshly activated, powdered 4A molecular sieves (624 mg) and anhydrous CH₂Cl₂ (20 mL). The reaction mixture was stirred at this temperature for 30 min, at rt for an additional 30 min, quenched by addition of a 1:1 mixture of THF and saturated aqueous NaHCO₃ (10 mL), and extracted with ether. After washing of the organic layer with 10% aqueous NaHSO₄ and brine, the organic layer was dried (Na₂SO₄), filtered and concentrated under reduced pressure. FCC of the crude product afforded **2.38** (221 mg, 72% yield based on **2.36**) as a colorless oil; R_f = 0.17 (30% EtOAc/petroleum ether); ¹H NMR (500 MHz, C₆D₆) δ 0.95 (s, 9H), 1.25 (s, 3H), 1.40 (s, 3H), 3.13 (bs, 1H, D₂O ex), 3.25 (dd, J = 5.5, 9.0 Hz, 1H), 3.45 (dd, J = 3.0, 10.0 Hz, 1H), 3.49 (t, J = 5.0 Hz, 1H), 3.56 (d, J = 6.0 Hz, 1H), 3.75-3.93 (m, 5H), 4.00 (m, 2H), 4.11 (m, 1H), 4.27 (dd, J = 2.0, 4.8 Hz, 1H), 4.30 (dd, J = 4.37 Hz, 1H), 4.45 - 4.63 (m, 9H), 7.25-7.42 (m, 26H), 7.70 (m, 4H); ¹³C NMR (75 MHz, CDCl₃) δ 26.6, 27.1, 28.6, 63.1, 69.4, 71.0, 71.2, 71.5, 72.4, 72.8, 73.1, 73.3, 73.8, 74.2, 75.0, 75.8, 76.3, 77.9, 80.3, 109.4, 127.7 - 129.8 (several lines), 133.7, 135.7, 137.9, 138.2, 138.5, 138.6; FABHRMS calcd for C₆₀H₇₀O₁₁NaSi (M + Na) 1017.4585, found 1017.4557.

1,3,4,5-Tetra-O-benzyl-13-O-tert-butyl-diphenylsilyl-2,6:8,12-dianhydro-10,11-O-isopropylidene-D-lyxo-D-gulo-D-manno-tridecitol 2.39.

Acetylation of PMB-protected C-glycoside **2.40** (155 mg, 0.13 mmol) following the procedure described in the synthesis of **2.28** provided the acetate derivative (156 mg, 98%) as a colorless oil; R_f = 0.34 (20% EtOAc/petroleum ether); ¹H NMR (500 MHz, CDCl₃) δ 1.10 (s, 9H), 1.36 (s, 3H), 1.63 (s, 3H), 2.06 (s, 3H), 3.38 (dd, J = 3.0, 9.0 Hz, 1H), 3.60 (t, J = 8.8 Hz, 1H), 3.68 (dd, J = 6.8, 11.0 Hz, 1H), 3.75 (dt, J = 2.5, 7.5 Hz,

1H), 3.83-4.00 (m, 7H), 4.02 - 4.09 (m, 4H), 4.18 (dd, $J = 2.0, 3.3$ Hz, 1H), 4.32-4.85 (m, 8H), 5.34 (dd, $J = 7.5, 9.7$ Hz, 1H), 6.75 (d, $J = 8.6$ Hz, 2H), 7.11 (d, $J = 8.6$ Hz, 2H), 7.21 - 7.43 (m, 26H), 7.75 (m, 4H); ^{13}C NMR (75 MHz, CDCl_3) δ 21.5, 26.6, 27.1, 27.9, 55.2, 70.3, 70.4, 71.8, 72.8, 73.6, 73.8, 74.7, 75.0, 75.2, 75.4, 75.9, 76.5, 77.4, 78.3, 80.1, 109.9, 113.9, 127.5-130.5 (several lines), 133.7, 133.8, 135.7, 138.5, 138.6, 138.7, 159.4, 169.2.

DDQ (62 mg, 0.27 mmol) was added to a solution of the product from the previous step (155 mg, 0.13 mmol) in CH_2Cl_2 (10 mL). The reaction mixture was stirred at rt for 2 hours, then diluted with saturated aqueous NaHCO_3 , extracted with CH_2Cl_2 , dried (Na_2SO_4), and evaporated under reduced pressure. FCC of the residue afforded the derived alcohol (107 mg, 77%) as a colorless oil; $R_f = 0.33$ (20% EtOAc/petroleum ether); ^1H NMR (500 MHz, CDCl_3) δ 1.05 (s, 9H), 1.34 (s, 3H), 1.46 (s, 3H), 1.99 (s, 3H), 3.06 (d, $J = 3.5$ Hz, D_2O exchange, 1H) 3.66-3.89 (m, 9H), 3.94 (dd, $J = 3.5, 9.0$ Hz, 1H), 4.00 (m, 1H), 4.24 (bs, 1H), 4.30 (d, $J = 6.0$ Hz, 1H), 4.48-4.63 (m, 7H), 4.77 (d, $J = 11.5$ Hz, 1H), 5.30 (t, $J = 5.5$ Hz, 1H), 7.21 - 7.41 (m, 26H), 7.68 (m, 4H); ^{13}C NMR (75 MHz, CDCl_3) δ 21.3, 26.1, 27.1, 62.8, 68.1, 69.8, 71.2, 71.9, 72.1, 73.3, 73.5, 74.1, 74.2, 75.1, 75.3, 76.2, 78.1, 78.3, 81.4, 82.2, 110.1, 127.6 - 129.8 (several lines), 133.6, 133.7, 135.7, 138.5, 138.7, 170.0; ESIHRMS calcd for $\text{C}_{62}\text{H}_{72}\text{O}_{12}\text{SiNa}$ ($M + \text{Na}$) 1059.4695, found 1059.4691.

NaOMe (50 mg, 1.14 mmol) was added to a solution of the product from the previous step (106 mg, 0.11 mmol) in anhydrous CH_3OH (5 mL). The reaction was stirred at rt for 2 hours and the pH was then adjusted to 7 by addition of 2N methanolic HCl . The mixture was concentrated under reduced pressure and the residue purified by

FCC to give the **2.39** (100 mg, 98%) as a colorless oil; $R_f = 0.51$ (25% EtOAc/petroleum ether); ^1H NMR (500 MHz, CDCl_3) δ 1.07 (s, 9H), 1.34 (s, 3H), 1.51 (s, 3H), 2.42 (d, $J = 9.5$ Hz, D_2O exchange, 1H), 3.34 (dd, $J = 2.0, 9.0$ Hz, 1H), 3.44 (t, $J = 9.0$ Hz, 1H), 3.63 (m, 2H), 3.77 (dd, $J = 2.5, 8.5$ Hz, 1H), 3.78-4.00 (m, 6H), 4.09 (m, 1H), 4.15 (t, $J = 2.5$ Hz, 1H), 4.21 (dd, $J = 2.5, 9.5$ Hz, 1H), 4.26 (d, $J = 3.0$ Hz, 1H), 4.48 - 4.71 (m, 8H), 7.13-7.45 (m, 26H), 7.71 (m, 4H); ^{13}C NMR (125 MHz, CDCl_3) δ 26.6, 27.0, 28.5, 63.1, 69.2, 69.9, 72.2, 72.6, 73.6, 73.7, 74.0, 74.1, 75.1, 80.0 (two signals), 109.5, 127.8-129.9 (several lines), 133.5, 133.7, 135.7, 135.9, 137.3, 138.2, 138.4; FABHRMS calcd for $\text{C}_{60}\text{H}_{70}\text{O}_{11}\text{SiNa}$ ($M + \text{Na}$) 1017.4588, found 1017.4585.

1,3,4,5-Tetra-*O*-benzyl-13-*O*-*tert*-butyldiphenylsilyl-2,6:8,12-dianhydro-10,11-*O*-isopropylidene-7-*O*-*p*-methoxybenzyl-*D*-lyxo-*D*-gulo-*D*-manno-tridecitol 2.40.

OPMB protected thioacetal ester (2.61 g, 87%) as a colorless oil; $R_f = 0.55$ (15% EtOAc/petroleum ether); ^1H NMR (500 MHz, C_6D_6) δ 1.05 (s, 9H), 1.41 (s, 3H), 1.47, (s, 3H), 3.42 (dd, $J = 3.0, 10.0$ Hz, 1H), 3.51 (dd, $J = 3.0, 7.7$ Hz, 1H), 3.61 (dd, $J = 4.7, 10.0$ Hz, 1H), 3.74 (m, 4H), 3.87 (m, 3H), 3.99 (t, $J = 3.4$ Hz, 1H), 4.14-4.45 (m, 10H), 4.57 (m, 3H), 5.34 (m, 2H), 6.75 (d, $J = 8.6$ Hz, 2H), 7.12-7.55 (m, 28H), 7.65 (m, 4H); ^{13}C NMR (75 MHz, CDCl_3) δ 26.7, 27.0, 27.5, 55.3, 62.2, 69.3, 71.8, 72.0, 72.2, 72.9, 73.4, 74.3, 74.7, 75.7, 78.0, 79.1, 84.7, 111.6, 114.0, 127.3-129.8 (several lines), 132.7, 132.8, 133.0, 133.4, 135.5, 135.6, 138.3, 138.5, 138.7, 159.5, 169.6; FABHRMS calcd for $\text{C}_{73}\text{H}_{80}\text{O}_{12}\text{SiSNa}$ ($M + \text{Na}$) 1231.5038, found 1231.5037.

OPMB protected thioacetal enol ether (1.11 g, 79%) as a colorless oil; $R_f = 0.60$ (10% EtOAc/petroleum ether); ^1H NMR (500 MHz, C_6D_6) δ 1.17 (s, 9H), 1.56 (s, 3H),

1.60 (s, 3H), 3.25 (s, 3H), 3.57 (dd, $J = 5.0, 11.0$ Hz, 1H), 3.68 (d, $J = 11.0$ Hz, 1H), 3.75 (dd, $J = 5.0, 9.5$ Hz, 1H), 3.80 (m, 1H), 3.96 (d, $J = 9.5$ Hz, 1H), 4.03 (m, 2H), 4.19-4.29 (m, 5H), 4.36-4.52 (m, 6H), 4.56 (A, of ABq, $J = 11.5$ Hz, $\Delta\delta = 0.16$ ppm, 1H), 4.61 (A of ABq, $J = 12.5$ Hz, $\Delta\delta = 0.41$ ppm, 1H), 4.78 (B of ABq, $J = 11.5$ Hz, $\Delta\delta = 0.16$ ppm, 1H), 4.81 (d, $J = 9.5$ Hz, 1H), 4.85 (d, $J = 6.0$ Hz, 1H), 4.97 (B of ABq, $J = 11.0$ Hz, $\Delta\delta = 0.41$ ppm, 1H), 6.27 (d, $J = 6.0$ Hz, 1H), 6.71 (d, $J = 8.0$ Hz, 2H), 6.95-7.28 (m, 29H), 7.32 (s, 1H), 7.39 (t, $J = 7.0$ Hz, 2H), 7.46 (d, $J = 7.0$ Hz, 1H), 7.76 (m, 4H); ^{13}C NMR (75 MHz, C_6D_6) δ 27.6, 28.1, 55.3, 62.4, 70.6, 71.3, 71.9, 74.0, 75.0, 75.4, 75.8, 76.1, 76.4, 78.4, 81.0, 85.0, 88.2, 112.8, 114.5, 126.8-129.5 (several lines), 130.5, 132.2, 133.3, 133.8, 134.0, 135.4, 136.4, 139.7, 140.0, 157.8, 160.2; ESIHRMS calcd for $\text{C}_{72}\text{H}_{82}\text{O}_{13}\text{SiNa}$ ($\text{M} + \text{Na}$) 1229.5414, found 1229.5422.

OPMB protected glycal (0.92 g, 87% based on recovered starting material) as a clear oil, $R_f = 0.27$ (10% EtOAc/petroleum ether); ^1H NMR (500 MHz, C_6D_6) δ 1.30 (s, 9H), 1.50 (s, 3H), 1.69 (s, 3H), 3.38 (s, 3H), 3.91 (dd, $J = 2.3, 13.0$ Hz, 1H), 3.98 (dd, $J = 4.7, 13.0$ Hz, 1H), 4.08 (dd, $J = 3.0, 9.0$ Hz, 1H), 4.19 (m, 3H), 4.32 (m, 3H), 4.40 (t, $J = 5.3$ Hz, 1H), 4.46 (d, $J = 9.0$ Hz, 1H), 4.50 (d, $J = 7.2$ Hz, 1H), 4.58-4.66 (m, 5H), 4.69 - 4.77 (m, 5H), 5.06 (d, $J = 11.4$ Hz, 1H), 5.18 (d, $J = 2.0$ Hz, 1H), 6.86 (d, $J = 8.6$ Hz, 2H), 7.11-7.50 (m, 28H), 7.90 (m, 4H); ^{13}C NMR (125 MHz, C_6D_6) δ 26.8, 26.9, 28.6, 54.5, 63.4, 69.5, 70.3, 71.0, 71.8, 71.9, 72.2, 73.5, 73.8, 74.5, 75.5, 76.1, 76.5, 77.4, 80.2, 101.2, 110.2, 114.0, 127.2-129.9 (several lines), 135.8 (two signals), 139.3, 153.2, 159.7; ESIHRMS calcd for $\text{C}_{68}\text{H}_{76}\text{O}_{11}\text{SiNa}$ ($\text{M} + \text{Na}$) 1119.5050, found 1119.5047.

OPMB protected C-glycoside **2.40** (0.52 g, 75% based on recovered starting material) as a clear oil; $R_f = 0.40$ (30% EtOAc/petroleum ether); ^1H NMR (500 MHz,

C_6D_6) δ 1.08 (s, 9H), 1.40 (s, 3H), 1.55 (s, 3H), 3.56 (m, 3H), 3.71 - 3.83 (m, 9H), 3.90 - 4.00 (m, 4H), 4.07 (apparent t, $J = 5.8$ Hz, 1H), 4.22 - 4.36 (m, 4H), 4.40 (dd, $J = 2.5, 8.0$ Hz, 1H), 4.47 - 4.68 (m, 7H), 4.81 (d, $J = 11$ Hz, 1H), 6.77 (d, $J = 8.5$ Hz, 2H), 7.11 (d, $J = 8.5$ Hz, 2H), 7.17 - 7.44 (m, 26H), 7.75 (m, 13.5 Hz, 4H); ^{13}C NMR (75 MHz, $CDCl_3$) δ 26.7, 27.1, 28.6, 55.3, 63.1, 70.1, 70.5, 71.8, 72.1, 72.5, 73.5, 74.6, 74.8, 75.2, 79.6, 80.5, 82.2, 109.3, 114.0, 127.7 - 129.8 (several lines), 130.1, 133.6, 133.8, 135.6, 135.7, 138.0, 138.4, 138.6, 159.4; ESIHRMS calcd for $C_{68}H_{78}O_{12}SiNa$ ($M + Na$) 1137.5161, found 1137.5160.

1,3,4,5-Tetra-*O*-benzyl-13-*O*-*tert*-butyldiphenylsilyl-2,6,8,12-dianhydro-10,11-*O*-isopropylidene-7,9-*O*-methylene-*D*-lyxo-*D*-galacto-*D*-manno-tridecitol 2.41.

A mixture of diol **2.38** (260 mg, 0.26 mmol), nBu_4Br (42 mg, 0.13 mmol), CH_2Br_2 (0.91 mL, 13.3 mmol) and 2,3-dimethylbutene (0.70 mL) was stirred at 65 °C for 15 min. 20% aqueous NaOH (8 mL) was then added dropwise to the reaction mixture and stirring continued at 65 °C for 3 hours. The mixture was then cooled to rt and extracted with ether. The organic layer was washed with water, dried (Na_2SO_4) and concentrated under reduced pressure. FCC afforded **2.41** (204 mg, 86% based on recovered **2.38**) as a colorless oil; $R_f = 0.44$ (15% EtOAc/petroleum ether); 1H NMR (500 MHz, $CDCl_3$) δ 1.05 (s, 9H), 1.39 (s, 3H), 1.51 (s, 3H), 3.49 (dd, $J = 6.9, 10.6$ Hz, 1H), 3.63 (dd, $J = 5.7, 10.1$ Hz, 1H), 3.74 (dd, $J = 5.7, 10.1$ Hz, 1H), 3.81 (m, 3H), 3.90 (m, 2H), 3.96 (m, 2H), 4.06 (dd, $J = 5.4, 7.6$ Hz, 1H), 4.24 (dd, $J = 7.8, 10.5$ Hz, 1H), 4.30 - 4.58 (m, 11H), 4.82 (A of ABq, $J = 5.9$ Hz, $\Delta\delta = 0.59$ ppm, 1H), 5.42 (B of ABq, $J = 5.9$ Hz, $\Delta\delta = 0.59$ ppm, 1H), 7.15-7.40 (m, 26H), 7.68 (m, 4H); ^{13}C NMR (125 MHz, $CDCl_3$) 26.7, 27.0, 28.6,

62.0, 62.8, 68.9, 70.5, 71.5, 72.1, 72.6, 72.7, 73.5, 73.8, 73.9, 74.7, 74.8, 75.2, 76.0, 76.8, 77.6, 91.2, 109.9, 127.6 - 130.0 (several lines), 133.6, 133.7, 135.7, 135.8 (two signals), 138.5, 138.6, 138.7, 138.9; ESIHRMS calcd for C₆₁H₇₁O₁₁Si (M + H) 1007.4766, found 1007.4810.

1,3,4,5-Tetra-*O*-benzyl-13-*O*-*tert*-butyldiphenylsilyl-2,6:8,12-dianhydro-10,11-*O*-isopropylidene-7,9-*O*-methylene-*D*-lyxo-*D*-gulo-*D*-manno-tridecitol 2.42.

Application of the procedure that was used for **2.41** to diol **2.39** (260 mg, 0.26 mmol) provided **2.42** (175 mg, 83% based on recovered **2.39**); colorless oil; R_f = 0.50 (20% EtOAc/petroleum ether); ¹H NMR (500 MHz, CDCl₃) δ 1.05 (s, 9H), 1.34 (s, 3H), 1.55 (s, 3H), 3.03 (t, J = 9.0 Hz, 1H), 3.36 (t, J = 9.0 Hz, 1H), 3.66 (m, 2H), 3.74 (dd, J = 2.0, 9.5 Hz, 1H), 3.80 (m, 3H), 3.86 - 4.00 (m, 4H), 4.04 (q, J = 5.0 Hz, 1H), 4.18 (d, J = 3.5 Hz, 1H), 4.35 (d, J = 4.0 Hz, 1H), 4.44 - 4.58 (m, 7H), 4.63 (d, J = 10.5 Hz, 1H), 4.65 (A of ABq, J = 6.5 Hz, Δδ = 0.39 ppm, 1H), 5.04 (B of ABq, J = 6.5 Hz, Δδ = 0.39 ppm, 1H), 7.17-7.43 (m, 26H), 7.69 (m, 4H); ¹³C NMR (75 MHz, CDCl₃) δ 26.6, 27.2, 28.8, 62.7, 69.7, 69.3, 70.6, 71.9, 72.2, 72.8, 73.4, 73.8, 75.4, 76.0, 77.0, 80.1, 80.4, 93.7, 109.8, 127.5 - 129.9 (several lines), 133.7, 133.8, 135.7, 135.8, 138.7, 138.8; FABHRMS calcd for C₆₁H₇₀O₁₁SiNa (M + Na) 1029.4584, found 1029.4585.

1,3,4,5-Tetra-*O*-benzyl-13-*O*-*tert*-butyldiphenylsilyl-2,6:8,12-dianhydro-7,9-*O*-methylene-*D*-lyxo-*D*-galacto-*D*-manno-tridecitol 2.43.

A saturated solution of HCl in ether (0.1 mL) was added to a solution of **2.41** (200 mg, 1.29 mmol) in dry CH₃OH (30 mL). The reaction mixture was stirred at rt for 2.5

hours then neutralized with a solution of NaOMe in methanol. Removal of the volatiles under reduced pressure and FCC of the residue provided **2.43** (130 mg, 68%) as a colorless oil; $R_f = 0.53$ (40% EtOAc/petroleum ether); $^1\text{H NMR}$ (500 MHz, CDCl_3) δ 0.95 (s, 9H), 2.49 (d, $J = 7.5$ Hz, 1H, $\text{D}_2\text{O ex}$), 2.92 (d, $J = 11$ Hz, 1H), 3.02 (d, $J = 10.0$ Hz, 1H, $\text{D}_2\text{O ex}$), 3.21 (s, 1H), 3.52 (m, 2H), 3.59 (m, 1H), 3.78 (m, 3H), 3.83 (t, $J = 10.0$ Hz, 1H), 3.88 - 4.00 (m, 3H), 4.05 (dd, 3.0, 9.8 Hz, 1H), 4.20 - 4.52 (m, 10H), 4.83 (A of ABq, $J = 5.5$ Hz, $\Delta\delta = 0.45$ ppm, 1H), 5.38 (B of ABq, $J = 5.5$ Hz, $\Delta\delta = 0.45$ ppm, 1H), 7.00-7.38 (m, 26H), 7.58 (m, 4H); $^{13}\text{C NMR}$ (75 MHz, CDCl_3) 27.1, 63.0, 66.4, 69.4, 69.7, 70.4, 71.9, 72.4, 72.8, 73.2, 73.4, 73.5, 74.7, 74.8, 75.0, 75.4, 80.0, 92.1, 127.6 - 129.9 (several lines), 133.4, 133.5, 135.7, 137.9, 138.1, 138.3, 138.7; FABHRMS calcd for $\text{C}_{58}\text{H}_{66}\text{O}_{11}\text{NaSi}$ ($M + \text{Na}$) 989.4272, found 989.4251.

1,3,4,5-Tetra-*O*-benzyl-13-*O*-*tert*-butyldiphenylsilyl-2,6:8,12-dianhydro-7,9-*O*-methylene-*D*-lyxo-*D*-gulo-*D*-manno-tridecitol 2.44.

Application of the procedure that was used for **2.43** to **2.42** (175 mg, 0.17 mmol) provided **2.44** (92 mg, 64% based on recovered **2.42**) as a colorless oil; $R_f = 0.44$ (40% EtOAc/petroleum ether); $^1\text{H NMR}$ (500 MHz, CDCl_3) δ 0.95 (s, 9H), 2.44 (d, $J = 6.0$ Hz, 1H, $\text{D}_2\text{O ex}$), 2.83 (b s, 1H, $\text{D}_2\text{O ex}$), 2.94 (t, $J = 9.0$ Hz, 1H), 3.25 (t, $J = 5.0$ Hz, 1H), 3.50 (m, 2H), 3.59 (m, 2H), 3.73 (m, 4H), 3.86 (m, 2H), 4.07 (m, 1H), 4.11 (d, $J = 3.0$ Hz, 1H), 4.35-4.61 (m, 10H), 4.97 (d, $J = 6.0$ Hz, 1H), 7.08-7.35 (m, 26H), 7.56 (m, 4H); $^{13}\text{C NMR}$ (75 MHz, CDCl_3) δ 27.2, 63.6, 69.8, 70.0, 72.0, 72.2, 72.3, 72.6, 72.9, 73.6, 75.5, 76.1, 76.6, 77.4, 77.5, 78.0, 78.8, 80.4, 93.8, 127.5 - 130.1 (several lines), 132.9,

133.1, 135.7, 135.8, 138.6, 138.8; FABHRMS calcd for $C_{58}H_{66}O_{11}NaSi$ ($M + Na$) 989.4272, found 989.4254.

1,3,4,5,10,11,13-Hepta-O-acetyl-2,6:8,12-dianhydro-7,9-O-methylene-D-lyxo-D-galacto-D-manno-tridecitol 2.45.

A solution of conc HCl in ether (0.1 mL) was added to diol **2.44** (110 mg, 0.14 mmol) in dry methanol (10 mL). The reaction was stirred at rt for approximately 2 hours and neutralized with sodium methoxide in methanol. Removal of the volatiles under reduced pressure and FCC of the residue provided the corresponding triol (69 mg, 66%), colorless oil, $R_f = 0.20$, (70% EtOAc/petroleum ether); 1H NMR (500 MHz, $CDCl_3$) δ 2.02 (dd, $J = 4.5, 8.0$ Hz, 1H, D_2O ex), 2.50 (d, $J = 8.0$ Hz, 1H, D_2O ex) 3.16 (d, $J = 9.0$ Hz, 1H), 3.33 (d, $J = 9.0$ Hz, 1H, D_2O ex), 3.36 (d, $J = 3.5$ Hz, 1H), 3.59 (dt, $J = 3.0, 8.8$ Hz, 1H), 3.67 (m, 2H), 3.74 (m, 1H), 3.82 (m, 1H), 3.87 (m, 1H), 3.93 (dd, $J = 3.5, 9.5$ Hz, 1H), 4.05 (dd, $J = 3.0, 9.8$ Hz, 1H), 4.08-4.15 (m, 3H), 4.33-4.64 (m, 10H), 4.93 (d, $J = 5.5$ Hz, 1H), 5.50 (d, $J = 5.5$ Hz, 1H), 7.18 (t, $J = 7.0$ Hz, 6H), 7.25-7.39 (m, 14H); ^{13}C NMR (75 MHz, $CDCl_3$) 63.0, 66.8, 69.1, 71.0, 72.0, 72.5, 73.0, 73.1, 73.1, 73.2, 73.3, 74.6, 75.1, 79.5, 92.0, 128.3, 128.4, 128.5, 128.6, 137.9, 138.1, 138.4; FABHRMS calcd for $C_{42}H_{48}O_{11}Na$ ($M + Na$) 751.3094, found 751.3094.

A solution of the above triol (14.0 mg, 0.02 mmol), DMAP (1.0 mg, 0.01 mmol), and acetic anhydride (0.07 mL, 0.64 mmol) in EtOAc (3 mL), was stirred at rt for 20 min. Then MeOH was added and the reaction mixture evaporated to dryness under reduced pressure. The residue was purified by FCC to afford the triacetate (16.3 mg, 99%), $R_f = 0.34$ (30 % EtOAc/petroleum ether); 1H NMR (500 MHz, C_6D_6) δ 1.73, 1.76, 1.60 (all, s, 9 H), 3.32 (t, $J = 6.3$ Hz), 3.52 (dd, $J = 7.00, 9.8$ Hz, 1H), 3.91 (dd, $J = 7.0, 9.8$ Hz, 1H),

4.03 (m, 3H), 4.15-4.60 (m, 13H), 4.67 (dd, $J = 2.5, 7.8$ Hz, 1H), 4.89 (d, $J = 5.5$ Hz, 1H), 4.95, (t, $J = 10.0$ Hz, 1H), 5.19 (dd, $J = 3.0, 11.0$ Hz, 1H), 5.56 (d, $J = 2.0$ Hz, 1H), 5.65 (d, $J = 5.5$ Hz, 1H), 7.07-7.29 (m, 18H), 7.36 (d, $J = 7.5$ Hz, 2H); FABHRMS calcd for $C_{42}H_{54}O_{14}Na$ ($M + Na$) 877.3411, found 877.3412.

The above triacetate was subjected to the hydrogenation-acetylation sequence outlined for the synthesis of **2.28** to give the peracetylated compound **2.45** (7.9 mg, 62%) as a colorless oil; $R_f = 0.30$ (40% EtOAc/petroleum ether); 1H NMR (500 MHz, C_6D_6) δ 1.56, 1.65, 1.67, 1.68, 1.75, 1.96 (all s, 21H), 3.33 (t, $J = 6.0$ Hz, 1H), 3.47 (dd, $J = 6.5, 10.0$ Hz, 1H), 4.00 (dd, $J = 6.0, 11.6$ Hz, 1H), 4.07 (q, $J = 5.7$ Hz, 1H), 4.19 (m, 2H), 4.51 (t, $J = 10.0$ Hz, 1H), 4.56 (m, 3H), 4.71 (d, $J = 6.0$ Hz, 1H), 5.06 (d, $J = 6.0$ Hz, 1H), 5.19 (dd, $J = 3.0, 10.0$ Hz, 1H), 5.47 (t, $J = 5.8$ Hz, 1H), 5.55 (d, $J = 3.0$ Hz, 1H), 5.70 (dd, $J = 3.5, 5.8$ Hz, 1H), 5.78 (dd, $J = 3.0, 6.0$ Hz, 1H); ^{13}C NMR (75 MHz, $CDCl_3$) 20.7, 20.9, 21.0 (two signals), 21.1 (two signals), 61.8, 62.3, 67.4, 68.0, 68.1, 68.2, 68.7, 70.7, 71.1, 71.9, 72.6, 73.1, 75.5, 90.5, 169.3, 169.7, 170.0 (two signals), 170.2, 170.5 (two signals); FABHRMS calcd for $C_{28}H_{38}O_{18}Na$ ($M + Na$) 685.1956, found 685.1952.

1,3,4,5,10,11,13-Hepta-O-acetyl-2,6:8,12-dianhydro-7,9-O-methylene-D-lyxo-D-gulo-D-manno-tridecitol 2.46.

A solution of diol **2.44** (12.0 mg, 0.01 mmol) in THF (3 mL) was treated with tetrabutyl ammonium fluoride in THF (1M, 0.05 mL, 0.05 mmol) for 1.5 hours. The solvent was then evaporated *in vacuo* to give the crude triol. The crude product, DMAP (1.0 mg, 0.01 mmol), and acetic anhydride (0.10 mL, 1.06 mmol) in EtOAc (3 mL), was stirred at rt for 30 min. Then MeOH was added and the reaction mixture evaporated to

dryness under reduced pressure. The residue was purified by FCC to give the triacetate (10.0 mg, 94%): $R_f = 0.62$ (50 % EtOAc/petroleum ether); $^1\text{H NMR}$ (500 MHz, C_6D_6) δ 1.62 (s, 3H), 1.73 (s, 6 H), 3.36 (t, $J = 6.5$ Hz, 1H), 3.59 (t, $J = 9.5$ Hz, 1H), 3.79 (t, $J = 9.8$ Hz, 1H), 3.92 (m, 2H), 3.98-4.10 (m, 6H), 4.28 (m, 2H), 4.44 (m, 1H), 4.53-4.64 (m, 8H), 4.83 (d, $J = 6.5$ Hz, 1H), 5.27 (dd, $J = 3.0, 10.3$ Hz, 1H), 5.48 (d, $J = 3.0$ Hz, 1H), 7.10-7.33 (m, 18H), 7.45 (d, $J = 7.5$ Hz, 2H). FABHRMS calcd. for $\text{C}_{42}\text{H}_{54}\text{O}_{14}\text{Na}$ ($\text{M} + \text{Na}$) 877.3411, found 877.3405.

Peracetylated compound **2.46** (6.3 mg, 81%) colorless oil; $R_f = 0.29$ (50 % EtOAc/petroleum ether); $^1\text{H NMR}$ (500 MHz, C_6D_6) δ 1.63, 1.68, 1.73, 1.74, 1.75, 1.78 (all s, 21H), 3.23 (t, $J = 9.8$ Hz, 1H), 3.30 (t, $J = 6.5$ Hz, 1H), 3.69 (t, $J = 9.8$ Hz, 1H), 3.77 (dd, $J = 2.2, 9.8$ Hz, 1H), 3.99 (dd, $J = 6.5, 10.8$ Hz, 1H), 4.10 (m, 2H), 4.28 (m, 2H), 4.46 (s, 1H), 4.56 (dd, $J = 5.0, 11.0$ Hz, 1H), 4.62 (d, $J = 6.5$ Hz, 1H), 5.15 (dd, $J = 3.0, 9.8$ Hz, 1H), 5.47 (d, $J = 3.0$ Hz, 1H), 5.72 (t, $J = 9.0$ Hz, 1H), 5.84 (br s, 1H), 5.96 (dd, $J = 4.0, 9.0$ Hz, 1H); $^{13}\text{C NMR}$ (75 MHz, CDCl_3) δ 20.8 (two signals), 20.9 (two signals), 21.0, (two signals), 21.2, 61.7, 62.9, 67.3, 68.4 (two signals), 69.9, 71.1, 72.8, 73.7, 74.6, 75.5, 75.7, 80.4, 93.5, 169.7, 168.8 (two signals), 169.9, 170.0, 170.3, 170.7.

Lactone **2.47**.

A mixture of diol **2.44** (54 mg, 0.06 mmol), Bu_2SnO (27 mg, 0.11 mmol), and anhydrous toluene (5 mL) was heated at reflux using a Dean-Stark set-up for 1 hour. The solution was then evaporated *in vacuo* and the residue was dissolved in dry toluene (3 mL). $n\text{-Bu}_4\text{NI}$ (22 mg, 0.06 mmol) and methyl 2-bromoacetate (0.10 mL, 1.08 mmol) were added and the solution heated at reflux for 1 hour, at which time the volatiles were

removed under reduced pressure. FCC of the residue afforded lactone **2.47** (55 mg, 98%) as a colorless oil; $R_f = 0.35$ (20 % EtOAc/petroleum ether); ^1H NMR (500 MHz, CDCl_3) δ 1.41 (s, 9H), 3.42 (t, $J = 7.0$ Hz, 1H), 3.38 (dd, $J = 6.5, 9.0$ Hz, 1H), 3.58 (dd, $J = 3.5, 10.0$ Hz, 1H), 3.84 - 4.48 (m, 16H), 4.51 (d, $J = 12.0$ Hz, 1H), 4.56 (d, $J = 3.0$ Hz, 1H), 4.61 (dd, $J = 1.5, 6.5$ Hz, 1H), 4.66 (dd, $J = 1.5, 9.5$ Hz, 1H), 4.76 (d, $J = 12$ Hz, 1H), 4.93 (d, $J = 6.0$ Hz, 1H), 4.98 (t, $J = 10.0$ Hz, 1H), 5.74 (d, $J = 6.0$ Hz, 1H), 7.04 - 7.36 (m, 26H), 7.82 (m, 4H); ^{13}C NMR (75 MHz, CDCl_3) δ 27.1, 60.9, 61.2, 68.4, 69.0, 70.5, 71.9, 72.1, 72.7, 73.2, 73.6, 74.2, 74.3, 74.7, 78.2, 91.7, 127.2 - 130.0 (several lines), 133.0, 133.2, 135.6, 138.3, 138.4, 138.5, 138.7, 166.5; FABHRMS calcd for $\text{C}_{60}\text{H}_{66}\text{O}_{12}\text{NaSi}$ (M + Na) 1029.4221, found 1029.4197.

Lactone **2.48**.

Application of the procedure that was used for **2.47** to diol **2.45** (90 mg, 0.09 mmol) provided **2.48** (75 mg, 93% based on recovered **2.45**) as a colorless oil; $R_f = 0.33$ (20 % EtOAc/petroleum ether); ^1H NMR (500 MHz, CDCl_3) δ 1.06 (s, 9H), 3.55 (t, $J = 9.5$ Hz, 1H), 3.61-3.82 (m, 6H), 3.90 - 3.97 (m, 4H), 4.14 (d, $J = 7.0$ Hz, 1H), 4.22 (s, 1H), 4.35 - 4.51 (m, 11H), 4.68 (d, $J = 6.5$ Hz, 1H), 4.88 (d, $J = 3.0$ Hz, 1H), 5.10 (d, $J = 6.5$ Hz, 1H), 7.15 - 7.44 (m, 26H), 7.67 (m, 4H); ^{13}C NMR (75 MHz, CDCl_3) δ 27.2, 60.6, 61.1, 68.4, 69.5, 71.0, 71.9, 72.3, 72.8, 73.5, 73.6, 74.1, 74.2, 75.2, 75.8, 76.1, 77.5, 79.4, 93.8, 127.6 - 130.1 (several lines), 132.9, 133.2, 135.6, 135.7, 138.3, 138.5, 138.6, 138.7, 166.8; FABHRMS calcd for $\text{C}_{60}\text{H}_{66}\text{O}_{12}\text{SiNa}$ (M + Na) 1029.4222, found 1029.4221.

(2,6:8,12-dianhydro-7,9-O-methylene-D-lyxo-D-galacto-D-manno-tridecitol-10-yloxy)-ethanoic acid sodium salt 2.3.

Lactone **2.47** (99 mg, 0.10 mmol) was dissolved in ethanol (5 mL) and treated with 3N NaOH (2 mL). After 2 hours the solvent was removed under reduced pressure, and the residue purified by FCC to give the dihydroxy sodium salt resulting from saponification of the lactone and cleavage of the silyl ether (56 mg, 73%), as a colorless oil; $R_f = 0.46$ (30 % CH₃OH/acetone); ¹H NMR (500 MHz, CD₃OD) δ 3.54 (dd, $J = 3.0, 9.5$ Hz, 1H), 3.59 (t, $J = 5.3$ Hz, 1H), 3.66 (dd, $J = 7.0, 9.5$ Hz, 1H), 3.73 (m, 3H), 3.80 (m, 2H), 3.86 (m, 1H), 3.91 (q, $J = 5.3$ Hz, 1H), 4.11 (m, 2H), 4.27 (A of ABq, $\Delta\delta = 0.11$ ppm, $J = 16.5$ Hz, 1H), 4.36 (apparent t, $J = 5.5$ Hz, 1H), 4.38 (B of ABq, $\Delta\delta = 0.11$ ppm, $J = 16.5$ Hz, 1H), 4.51-4.60 (m, 13H), 4.77 (d, $J = 6.0$ Hz, 1H), 5.29 (d, $J = 6.0$ Hz, 1H), 7.20-7.37 (m, 20H); ¹³C NMR (100 MHz, CD₃OD) δ 61.7, 68.1, 68.4, 68.7, 70.1, 71.7, 72.7, 72.9, 73.1, 73.6, 74.0, 74.2, 74.5, 76.4, 79.3, 80.3, 90.4, 127.4, 128.1, 138.6, 174.1; ESIHRMS calcd for C₄₄H₅₁O₁₃ 787.3330, found 787.3358.

A mixture of the product from the previous step (52 mg, 0.06 mmol), 10% Pd on carbon (100 mg), formic acid (0.1 mL) and CH₃OH (3 mL) was stirred under an atmosphere of hydrogen (balloon), for 12 hours. The reaction mixture was purged with argon, filtered through a bed of Celite and the filtrate concentrated under reduced pressure. The residue was dissolved in ethanol (5 mL) and treated with 3M NaOH (1 mL). After stirring at rt for 2 hours the mixture was evaporated *in vacuo* and residue purified by sequential FCC on C18 silica gel (CH₃OH/H₂O) and Sephadex LH-20 (H₂O). Lyophilization of the eluate provided **2.3** as an amorphous solid (27 mg, 90%); $[\alpha]_D + 13.5$ (*c* 0.40 H₂O); ¹H NMR (500 MHz, D₂O) δ 3.61 (dt, $J = 2.6, 7.4$ Hz, 1H), 3.65-3.74

(m, 5H), 3.75 (d, $J = 2.6$ Hz, 1H), 3.81 (m, 2H), 3.87 (dd, $J = 3.5, 8.3$ Hz, 1H), 4.13 (ABq, $J = 16.3$ Hz, $\Delta\delta = 0.09$ ppm, 2H), 4.16 (d, $J = 3.1$ Hz, 1H), 4.20 (t, $J = 3.1$ Hz, 1H), 4.26 (t, $J = 9.9$ Hz, 1H), 4.31 (dd, $J = 3.4, 9.3$ Hz, 1H), 4.50 (dd, $J = 5.9, 9.3$ Hz, 1H), 4.91 (A of ABq, $J = 6.8$ Hz, $\Delta\delta = 0.18$ ppm, 1H), 5.09 (B of ABq, $J = 6.8$ Hz, $\Delta\delta = 0.18$ ppm, 1H); ^{13}C NMR (125 MHz, D_2O) δ 60.7, 61.7, 66.5, 67.6, 68.0, 68.5, 69.0, 70.6, 72.3, 72.4, 72.5, 76.4, 79.5, 80.4, 89.6, 178.3; FABHRMS calcd for $\text{C}_{16}\text{H}_{26}\text{O}_{13}\text{Na}$ 449.1288, found 449.1271.

(2,6:8,12-dianhydro-7,9-O-methylene-D-lyxo-D-gulo-D-manno-tridecitol-10-yloxy)-ethanoic acid sodium salt 2.4.

Lactone **2.48** (75 mg, 0.07 mmol) was subjected to the similar saponification procedure that was described for lactone **2.47**. The corresponding dihydroxy sodium salt (38 mg, 67%) was obtained as a colorless oil; $R_f = 0.28$ (30 % $\text{CH}_3\text{OH}/\text{acetone}$); ^1H NMR (500 MHz, CD_3OD) δ 3.16 (t, $J = 9.0$ Hz, 1H), 3.45 (dd, $J = 3.0, 9.5$ Hz, 1H), 3.53 (t, $J = 6.3$ Hz, 1H), 3.66 (d, $J = 5.0$ Hz, 1H), 3.71 (dd, $J = 5.0, 11.5$ Hz, 1H), 3.77 (m, 2H), 3.91 (d, $J = 9.5$ Hz, 1H), 3.96 (q, $J = 5.0$ Hz, 1H), 4.00 (m, 1H), 4.06 (m, 3H), 4.28 (m, 2H), 4.48-4.69 (m, 10H), 4.73 (d, $J = 6.5$ Hz, 1H), 5.04 (d, $J = 6.5$ Hz, 1H), 7.25-7.43 (m, 20H); ^{13}C NMR (75 MHz, CD_3OD) δ 63.0, 68.7, 70.8, 71.0, 73.1, 73.6, 73.8, 74.0, 74.5, 74.9, 76.6, 77.2, 78.7, 79.1, 80.7, 81.2, 81.9, 94.5, 128.5, 128.7, 128.8, 129.0, 129.1, 129.4, 139.7, 139.8, 178.4; FABHRMS calcd for $\text{C}_{44}\text{H}_{51}\text{O}_{13}$ (M + H) 787.3330, found 787.3327.

The product from the previous step (37 mg, 0.05 mmol) was subjected to the hydrogenation procedure that was described for **2.3**. Compound **2.4** (17 mg, 84%) was

obtained as an amorphous solid; $[\alpha]_D + 41.6$ (*c* 0.30 H₂O); ¹H NMR (500 MHz, D₂O) δ 3.32 (t, *J* = 9.4 Hz, 1H), 3.60 (t, *J* = 8.6 Hz, 1H), 3.65 (dd, *J* = 2.5, 12.4 Hz, 1H), 3.68 - 3.71 (m, 8H), 3.97 (dd, *J* = 3.6, 8.6 Hz, 1H), 4.02 (dd, *J* = 2.7, 9.6 Hz, 1H), 4.10 - 4.15 (m, 5H), 4.17 (m, 3H), 4.79 (A of ABq *J* = 6.5 Hz, $\Delta\delta$ = 0.29 ppm, 1H), 5.08 (B of ABq, *J* = 6.5 Hz, $\Delta\delta$ = 0.29 ppm, 1H); ¹³C NMR (125 MHz, D₂O) δ 61.2, 61.5, 66.4, 67.5 (two signals), 68.6, 71.3, 72.7, 76.1, 77.1, 77.4, 79.2, 79.3, 79.7, 93.2, 178.3; FABHRMS calcd for C₁₆H₂₆O₁₃Na (*M* + Na) 449.1286, found 449.1295.

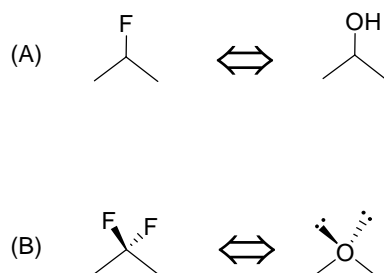
Chapter 3

Synthesis of fluorinated C-disaccharide sLe^x mimetics.

3.1 Introduction

Fluorinated carbohydrates are increasingly being utilized as tools for the investigation of various biological processes.²⁸⁻³² Interest in these compounds is related to the intrinsic properties of fluorine. The van der Waals radii of fluorine (1.47 Å) lies between oxygen (1.57 Å) and hydrogen (1.2 Å) and thus fluorine may behave as a less sterically demanding isostere of oxygen, or replaces hydrogen without any considerable sterical or geometrical demands.³³ Hence, fluoroalkanes and difluoromethylene groups are isosteric substituents of alkanols and ethers, respectively (Figure 3.1).³⁴

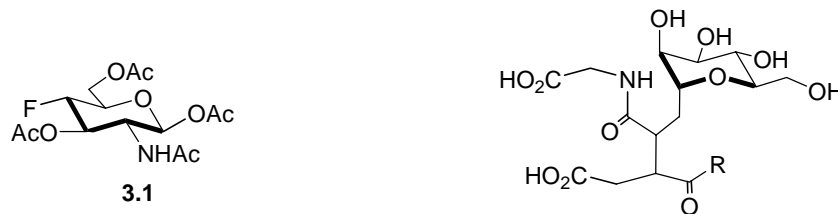
Figure 3.1: Isosteric fluorine substituents



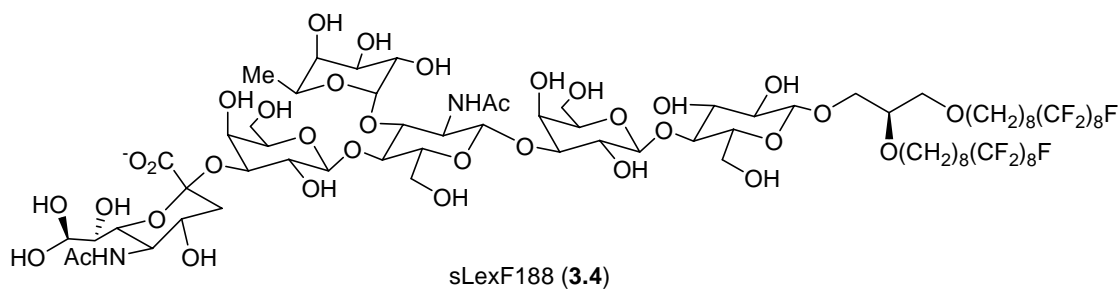
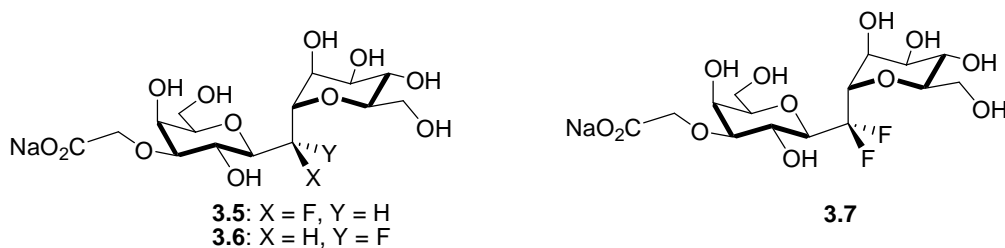
Fluorinated carbohydrates have previously been examined for their effects on selectin mediated cellular adhesion. A recent study by Descheny and coworkers revealed that 2-acetamido-1,3,6-tri-O-acetyl-4-deoxy-4-fluoro-D-glucopyranose **3.1** was able to inhibit selectin binding activity.³⁵ This effect was related to the ability of **3.1** to function as a potent metabolic inhibitor of N-acetylglucosamine and sLe^x biosynthesis.

Other studies imply that the introduction of fluorine substituents on selectin ligands may increase their selectin binding affinity.^{36,37} This property of such fluorinated compounds may be due to their increased lipophilicity in comparison to their non-fluorinated analogs. Such characteristic seems to rationalize the two fold increase in binding of the fluorinated sLe^x mimetic **3.2**, to its non-fluoro analog **3.3** in a cell-free sLe^a-polymer/ E-selectin assay.³⁶ The fluorinated glycolipid sLeXF188 (**3.4**) was used to gain deeper insight into the effect of clustering of sLe^x glycolipids and their ability to induce cell rolling.³⁷ This study seems to imply that fluorination in the alkyl chain of **3.4** leads to an increase in its cluster size. This causes an impediment of cell rolling when compared to its unfluorinated analog at higher ligand concentrations in a dynamic cell assay.

Fluorine substitution on the pseudo-anomeric carbon in C-glycosides is expected to alter the conformational behavior in the intersaccharide torsions. This situation is analogous to distortion of the natural conformation of the sugar residue in nucleosides by introduction of a 2-fluoro substituents.³⁸ These expected characteristics of fluoro carbohydrates led to the synthesis of the epimeric fluoro C-glycosides **3.5** and **3.6**, and the difluoro-C-glycoside **3.7** (Figure 3.2).

Figure 3.2: Fluorinated carbohydrate mimetics of sLe^x**3.1**

3.2: R = CF₃ (42% inhibition of E-selectin at 3 mM)
3.3: R = CH₃ (28% inhibition of E-selectin at 3 mM)

sLexF188 (**3.4**)

3.5: X = F, Y = H
3.6: X = H, Y = F

3.7

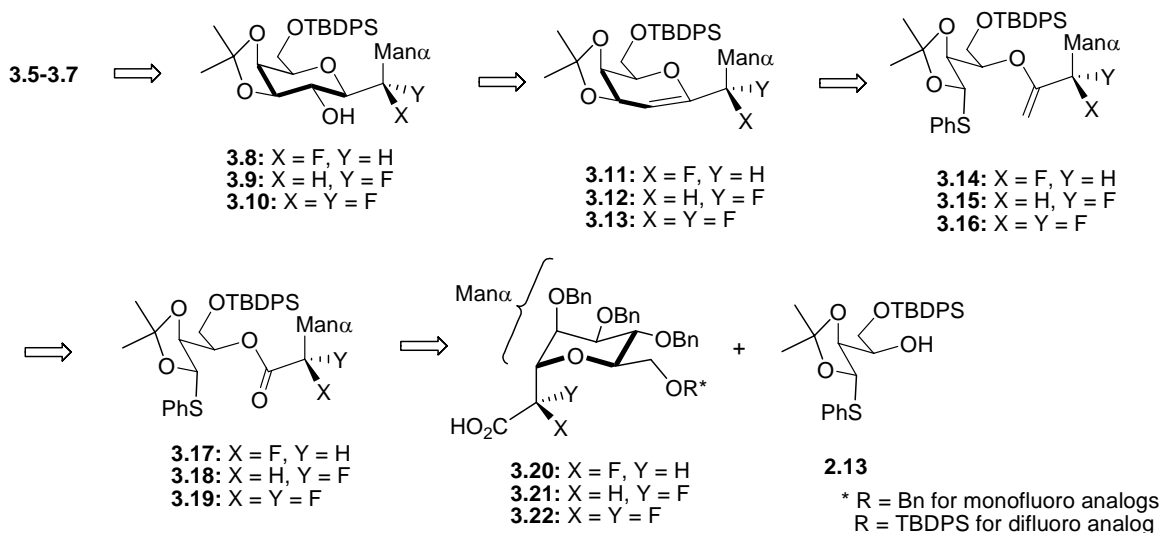
Results and Discussion

3.2 Retrosynthesis

We envisioned that the TIA methodology previously described could be used for the synthesis of the fluoro-C-glycosides **3.5-3.7**. Stereoselective hydroboration of the fluoro glycals **3.11-3.13** should provide **3.8-3.10**. These glycals may be obtained from the methyl triflate mediated cyclization of the enol ethers **3.14 - 3.16**. An important question about the later reaction is whether the electronegativity of the fluorine substituent would have an unfavorable effect on the key oxocarbenium cyclization step.

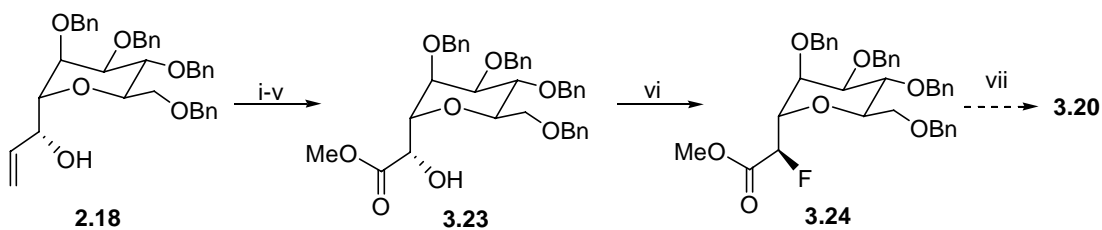
The enol ethers **3.14-3.16** could be synthesized from olefination of the corresponding α -fluoro esters **3.17-3.19**. These esters may be obtained from the DCC coupling of the fluoro acids **3.20-3.22** and the known TIA alcohol **2.13** (Scheme 3.1).

Scheme 3.1: Retrosynthesis of the fluoro-C-glycosides **3.5-3.7**.



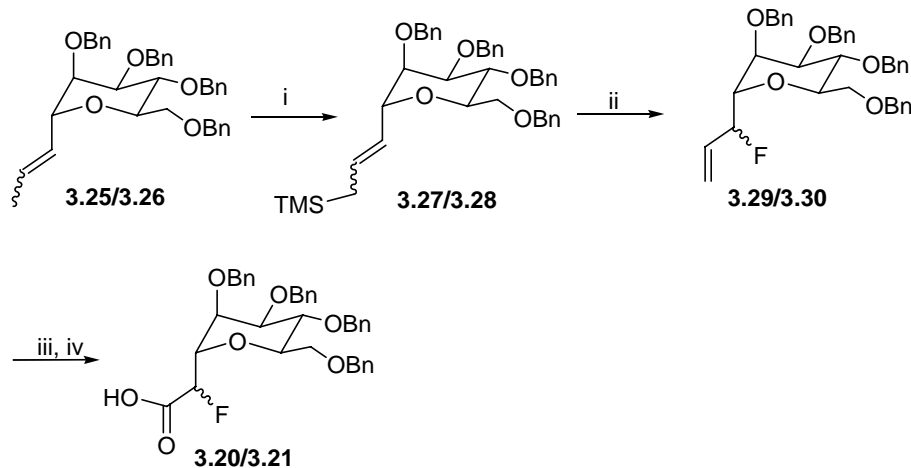
3.3 Synthesis of the fluoro acid precursors **3.20-3.22**.

We first attempted the synthesis of the acid **3.20** by treatment of the α -hydroxy ester derivative **3.23** with diethylaminosulfur trifluoride (DAST).³¹ Previously prepared vinyl alcohol **2.18** was transformed to **3.23** in five steps (Scheme 3.2). The reaction of **3.23** with DAST gave the desired fluoro ester **3.24** in 16% yield. Both the ^1H and ^{13}C NMR spectra of **3.24** revealed a $^2J_{\text{H,F}}$ coupling of 48.8 Hz at δ 5.18 and $^1J_{\text{C,F}}$ coupling of 191.8 Hz at δ 89.6 respectively, that was indicative of the α -fluoro ester. However, the low yield of the reaction prompted the development of an alternative synthesis of the fluoro acids **3.20/3.21**.

Scheme 3.2: Attempted synthesis of fluoro acid **3.20**.

i. AcO₂, DMAP, quant. yield; ii. O₃, PPh₃, 97%; iii. NaClO₂, 99%; iv. TMSCHN₂, ether, quant. yield; v. NaOMe/MeOH, 80%; vi. DAST, CH₂Cl₂, 16%; vii. NaOH/EtOH, HCl

A more practical method for the synthesis of the fluoro acids **3.20/3.21** involved the coupling of the known α -C-1-propenyl derivatives **3.25/3.26** with allyltrimethylsilane using a cross metathesis protocol. An *E/Z* (3/1) mixture of allylsilane **3.27/3.28** was obtained in 88% yield when the reaction was done under an ethylene atmosphere with 15 mol % of second-generation Grubbs' catalyst.³⁹ The mixture **3.27/3.28** was identified mainly as the *E* isomer in 88% yield. Treatment of **3.27/3.28** with SelectfluorTM led to a 1:1 mixture of the epimeric allylic fluorides **3.29/3.30** in 60% yield. In this reaction, the trimethylsilyl group enhances the reactivity of the alkene toward the addition of SelectfluorTM and controls the regioselectivity because of stabilization of the incipient cation β to the silicon.⁴⁰ Ozonolysis of **3.29/3.30** followed by sodium chlorate oxidation of the resulting mixture of aldehyde afforded the required fluoro acids **3.20/3.21** as an inseparable mixture (Scheme 3.3).

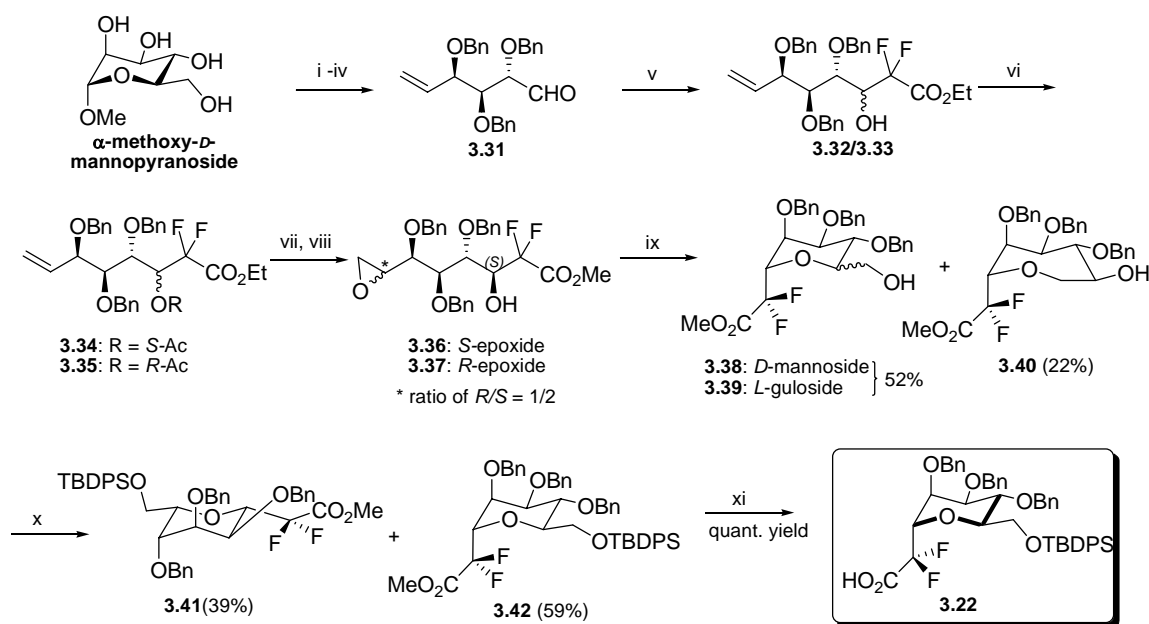
Scheme 3.3: Synthesis of fluoro acids **3.20/3.21**.

i. Grubbs 2, CH₂=CH₂ atm CH₂CHCH₂Si(CH₃)₃, 2 days, 89%; ii. selectfluorTM, CH₃CN, 60%; iii. O₃, PPh₃, 97%; iv. NaClO₂, 99%

The synthesis of the difluoro acid **3.22** was accomplished using the protocol developed by Quirion and coworkers.⁴¹ Starting with the known enal **3.31**⁴² (available in five steps from methyl α -D-mannopyranoside), Reformatsky reaction was done using the enolate ion generated from ethyl bromodifluoroacetate. This provided a diastereomeric mixture of alcohols **3.32/3.33** in 54% yield.⁴¹ Attempts to separate this mixture by column chromatography was unsuccessful. However, acetylation enabled their separation as the acetates **3.34** and **3.35** in a ratio of ca. 3:1. At this juncture, the major product **3.34** was assumed to be the *anti* isomer based on evidence provided by literature.⁴³ The outcome of such transformation was later established in the ¹H NMR analysis of the later derivatives **3.45** and **3.46** (*vide infra*) and the major product was in fact the *anti* isomer. Therefore, the Reformatsky reaction probably preceded by the Felkin-Anh transition state.⁴³ Deacetylation of **3.34** followed by treatment of the resulting alcohol with *m*-CPBA yielded a 2:1 mixture of C-5 epimeric epoxides

3.36/3.37. Cyclization of the mixture of epoxides was initially performed under acidic conditions (CSA, CH₂Cl₂). This gave a mixture of several compounds. However, under basic conditions (NaOMe, MeOH), the crude products obtained were treated with TMSN₂CH to mainly give the difluoro-C-glycoside mixture **3.38/3.39**, and the difluoro-C-heptoside derivative **40** as the minor product (Scheme 3.4). Protection of **3.38/3.39** as the *tert*-butyldiphenylsilyl ethers enabled their separation as the *D*- and *L*-sugars **3.41** and **3.42** respectively.

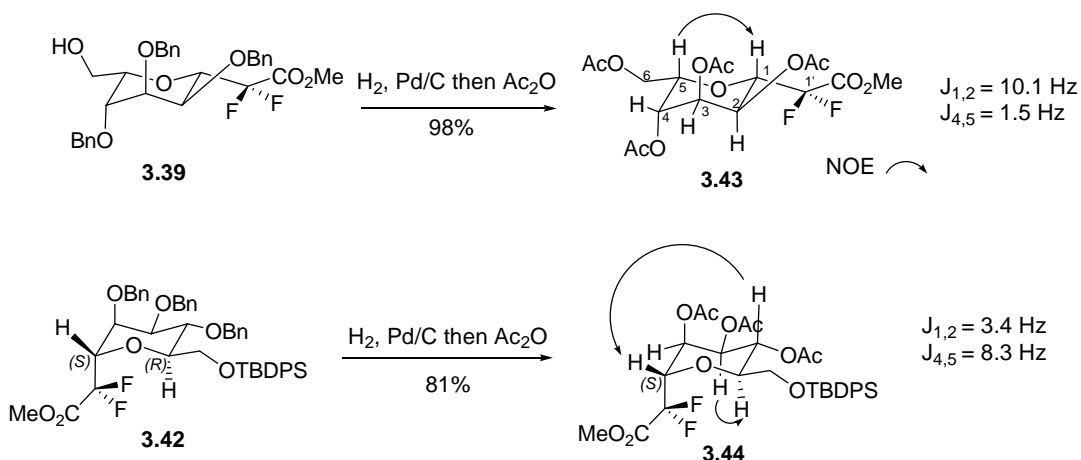
Scheme 3.4: Synthesis of the difluoro acid **3.22**.



i. Ph₃CCl, py, 99%; ii. (a) BnBr, NaH (b) TsOH, MeOH, 86% (2 steps); iii. Ph₃P, I₂, imdZl, 93%; iv. n-BuLi/THF, 94%; v. BrCF₂COOEt, Zn, 54%; vi. Ac₂O, DMAP, 85% (isomer separation, 3/1 ratio); vii. NaOMe/MeOH, 83%; viii. m-CPBA, Na₂SO₄/NaHSO₄ buffer, 74%, (2:1 ratio); ix. (a), NaOMe/MeOH then TMSN₂CH in PhMe/MeOH; x. TBDPSCI, imid. isomer separation; xi. NaOH, EtOH then HCl.

The axial orientation at C-1 and the stereochemistry at C-5 in the difluoro-C-glycosides **3.41** and **3.42** were verified from the ¹H-¹H coupling constants and NOESY experiments of their acetylated derivatives **3.43** and **3.44** respectively (Scheme 3.5).

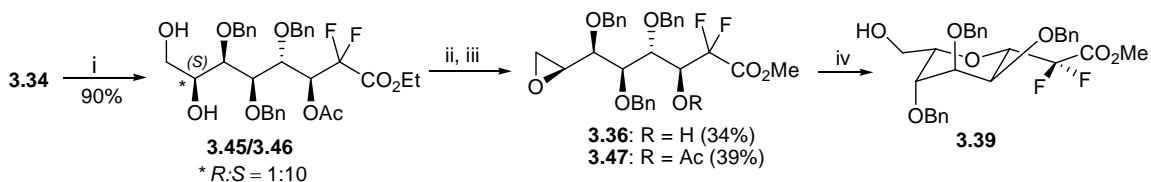
Scheme 3.5: Stereochemical analysis of difluoro-*D*-mannoside **3.39** and difluoro-*L*-gulose **3.41**.



Base hydrolysis of the required difluoro-C-mannoside **3.42** followed by acid workup led to the isolation of acid **3.22**. (Scheme 3.4)

An attempt was made to increase the selectivity of the required epoxide **3.37** (Scheme 3.6). Thus, the acetate **3.34** was subjected to dihydroxylation with osmium tetroxide and the product obtained selectively tosylated and treated under basic conditions to give the hydroxyl epoxide **3.36** and its acetylated derivative **3.47** instead. These compounds were individually cyclized to give the difluoro-C-glycoside **3.39** (Scheme 3.6).

Scheme 3.6: Synthesis of *L*-guloside **3.39**

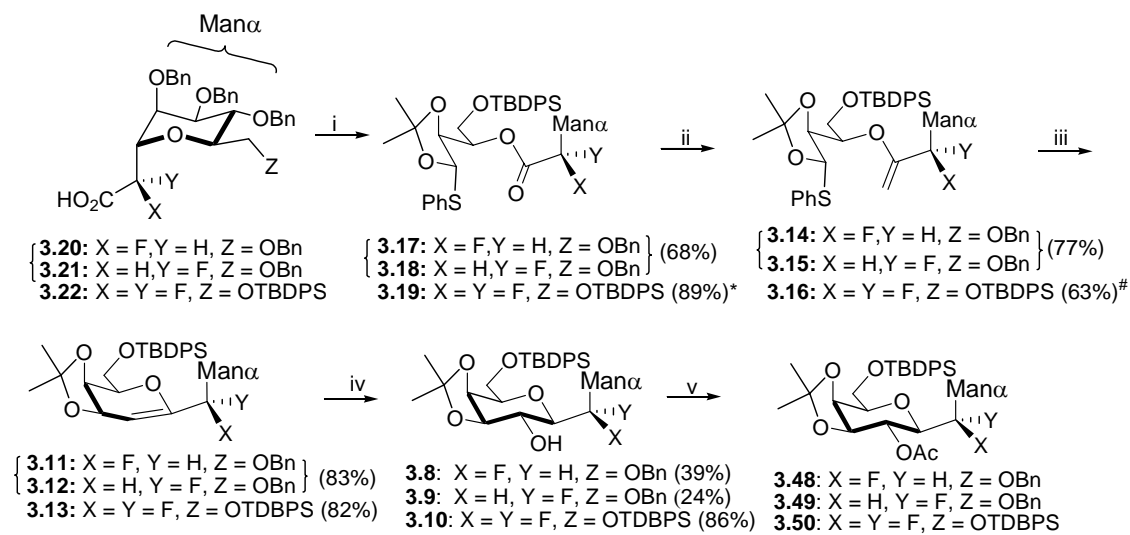


i. OsO_4/NMO , 90%; ii. TsCl/pyr , 87%; iii. 2eq. NaOMe , MeOH , 18h; iv. NaOMe/MeOH then HCl , 62% from **3.36** and 69% from **3.47**.

3.4 Synthesis of fluoro-C-glycosides 3.8-3.10.

The acid segments **3.20/3.21** and **3.22** were coupled with the alcohol **2.13** for the fluoro-C-glycoside sequence (Scheme 3.7). DCC mediated esterification of **3.20/3.21** and **3.22** with **2.13** gave a mixture of α -fluoro esters **3.17/3.18** (68%) and the difluoro ester **3.19** respectively. As the conversion of **3.19** from acid **3.22** was very low, the Yamaguchi esterification protocol⁴⁴ was attempted. Thus, **3.19** was prepared in 91% yield (based on recovered **2.13**) with 76% conversion from **3.22**. Tebbe methylenation on esters **3.17/3.18** provided a mixture of enol ethers **3.14/3.15**. However the reaction was not successful for difluoro ester **3.19**. The Takai methylenation⁴⁵ on **3.19** afforded the difluoro enol ether **3.16** in 63% yield. It should be noted that the difluoro ester **3.19** was labile to basic condition upon purification of the crude product. To overcome this, the purification process was performed with silica gel chromatography. The key cyclization reaction on **3.14/3.15** and **3.16**, was promoted by methyl triflate in the presence of 2,6-di-tert-butyl-4-methylpyridine (DTBMP). The feasibility of this reaction was of concern because the presence of the electronegative fluorine atom could reduce the nucleophilicity of the alkene in the oxocarbenium ion cyclization. In the event, the reaction proceeded to give the fluoro glycals **3.11/3.12** and **3.13** in high yields. Selective hydroboration of **3.11/3.12** and **3.13** provided the fluoro-C-galactosides **3.8** (39%), **3.9** (24%) and **3.10** (86%) respectively, each as a single diastereomer (Scheme 3.7). Alcohols **3.8** and **3.9** were readily separated on silica gel column chromatography (R_f 's: 0.63 and 0.75 respectively, 30% ethyl acetate/petroleum ether).

Scheme 3.7: Synthesis of the fluorinated C-glycosides **3.8-3.10**.

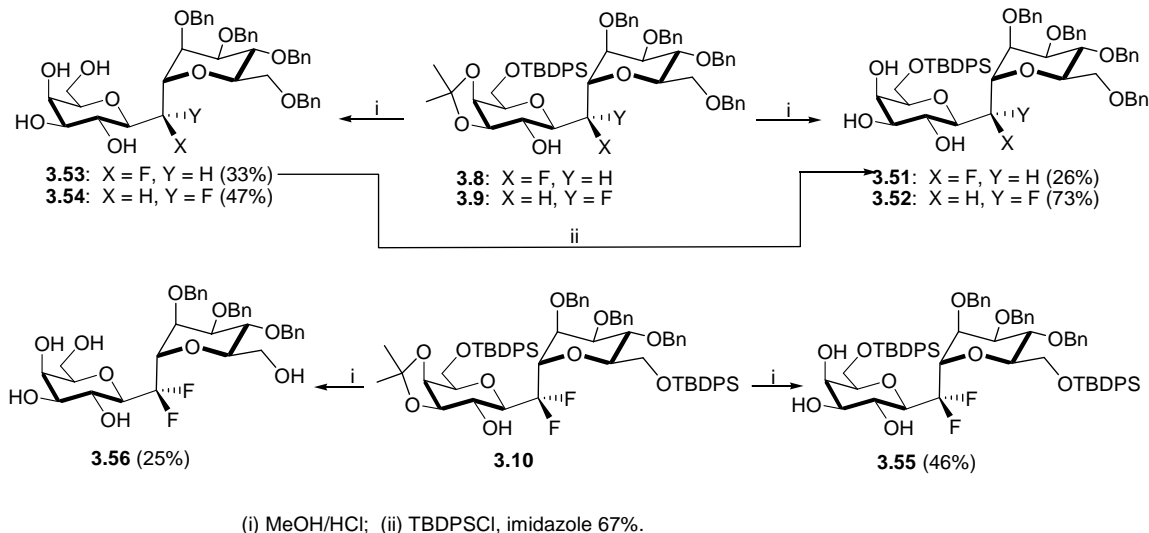


i. (a) DCC, DMAP, **2.13**, PhH* and/or (b) Et₃N, DMAP, 2,4,6-trichlorobenzoyl chloride, then **2.13**, DMAP, **3.22** (91%);
 ii. (a) Tebbe or (b) TiCl₄, CH₂Br₂, Zn, TMEDA[#]; iii. MeOTf, DTBMP, CH₂Cl₂; iv. BH₃.DMS then Na₂O₂; v. Ac₂O, DMAP.

The stereochemistry at the C-1 position of galactose in these fluoro-C-glycosides was confirmed from the $J_{H-1,H-2}$ couplings of 8.4, 8.2 and 9.5 Hz for the acetates **3.48**, **3.49** and **3.50** respectively,⁴ and subsequent derivatives **3.59** and **3.60** (*vide infra*).

Compounds **3.8**, **3.9** and **3.10** were converted to their respective triols **3.51** (26%), **3.52** (73%) and **3.55** (46%) upon treatment with methanolic HCl. These reactions on **3.8** and **3.9** were also accompanied by desilylation to give the tetrol **3.52** (33%) and the pentol **3.55** (25%) respectively. Resilylation of **3.52** gave the required **3.50** in a moderate yield of 67% (Scheme 3.8).

Scheme 3.8: Acid hydrolysis of fluoro-C-glycosides **3.8-3.10**.

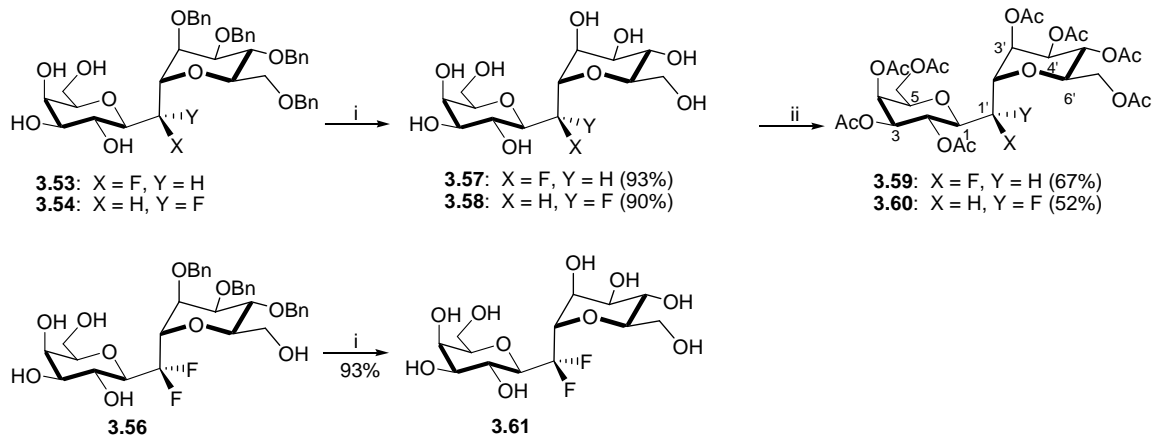


3.5 Stereochemical analysis of fluoro-C-glycosides 3.59 and 3.60.

The structures **3.53** and **3.54** were confirmed by NMR analysis of their peracetate derivatives **3.59** and **3.60** (Scheme 3.9, Table 3.1). The stereochemistry of the aglycone segment was assigned on the basis of vicinal J values. Thus, $J_{1,2} = 10.0$ Hz, $J_{2,3} = 10.0$ Hz, $J_{3,4} = 3.3$ Hz, $J_{4,5} = 0$ Hz for both **3.59** and **3.60** were consistent with the 3,4,6-acetylated- β -C-galacto motif of recently synthesized restrain disaccharides.²⁶

The stereochemistry of the fluorine substituents at the C-1' could not be determined at this point. This was subsequently assigned from NMR experiments performed by our collaborator Jiménez-Barbero in Spain. (See chapter 4)

Scheme 3.9: Synthesis of the peracetylated fluoro-C-glycosides **3.36** and **3.37**



(i) Pd/C, H₂, MeOH; (ii) DMAP, Ac₂O, EtOAc

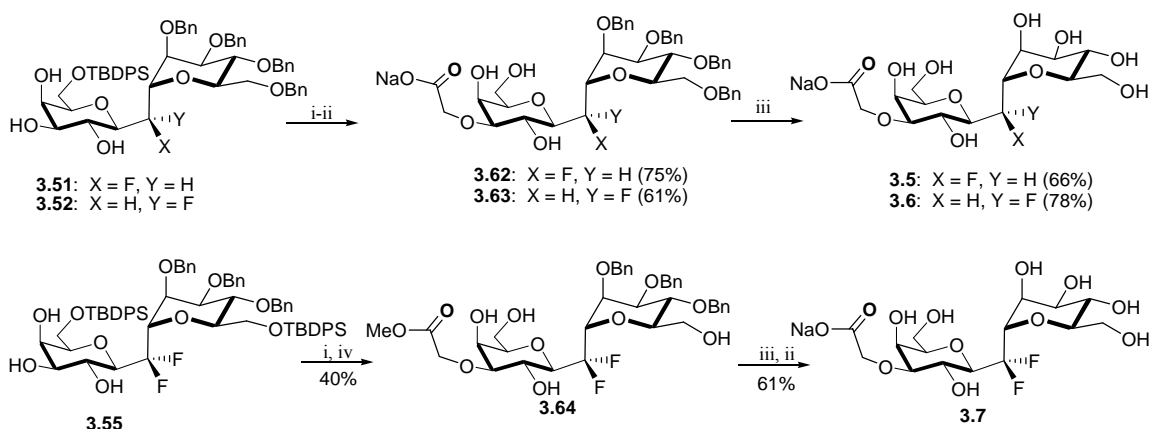
Table 3.1: ¹H-NMR of the peracetylated fluoro-C-glycosides **3.59** and **3.60**.

Positions	¹H NMR (J, Hz) fluoro derivative 3.59	¹H NMR (J, Hz) fluoro derivative 3.60
1'	4.85 (ddd, J = 2.2, 7.7, 46.6)	5.01 (ddd, J = 2.7, 5.5, 47.4)
2'	4.66 (ddd, J = 2.6, 7.7, 17.7)	4.61 (dt, J = 3.5, 31.5)
3'	5.80 (t, 2.6)	6.02 (t, J = 3.0)
4'	5.41 (dd, J = 2.6, 9.4)	5.78 (dd, J = 3.5, 8.8)
5'	5.66 (t, J = 9.4)	5.69 (t, J = 8.8)
6'	4.03 (m)	4.22 (m)
7'a	4.45 (dd, J = 5.3, 12.2)	4.49 (q, J = 5.9)
7'b	4.15 (dd, J = 6.9, 11.4)	4.10 (m)
1	3.72 (ddd, J = 2.4, 9.8, 20.1)	3.66 (m)
2	5.91 (t, J = 10.0)	5.62 (t, J = 10.0)
3	5.20 (dd, J = 3.3, 10.0)	5.08 (dd, J = 3.3, 10.0)
4	5.49 (d, J = 3.3)	5.45 (d, J = 3.3)
5	3.37 (t, J = 6.3)	3.27 (t, J = 6.5)
6a	4.22 (dd, J = 6.3, 11.4)	4.10 (m)
6b	4.24 (dd, J = 2.6, 12.2)	4.22 (m)
CH ₃ CO-	1.60, 1.65, 1.67, 1.69 (2), 1.730, 1.733, 1.76	1.61, 1.64, 1.667, 1.671, 1.71, 1.72, 1.76

3.6 Synthesis of the fluorinated sLe^x mimetics 3.5-3.7.

The triols **3.51**, **3.52** and **3.55** were next transformed to the targeted fluorinated sLe^x **3.5**, **3.6**, and **3.7** respectively (Scheme 3.10). Thus, selective dibutyltin oxide mediated alkylation of **3.51**, **3.52** and **3.55** with methyl bromoacetate led to 3-O-alkylation followed by *in situ* lactonization to give in each case a mixture of 2-O and 4-O-lactone products. Exposure of the mono fluoro crude products to aqueous sodium hydroxide led to concomitant saponification and desilylation to give the corresponding trihydroxy sodium salts **3.62** and **3.63**, which were subjected to hydrogenolysis. Exposure of the crude difluoro lactonized products to excess tetrabutyl ammonium fluoride followed by acid work up afforded the methyl ester **3.64**. The latter was then subjected to hydrogenolysis and base hydrolysis (Scheme 3.10). The target compounds **3.5**, **3.6** and **3.7** were obtained after purification using reverse and normal phase chromatography, and lyophilization from aqueous solutions.

Scheme 3.10: Final synthesis of the fluoro sLe^x mimetics **3.5-3.7**.



(i) Bu₂SnO then BrCH₂CO₂Me, nBu₄Nl; (ii) EtOH/NaOH; (iii) H₂, Pd/C, MeOH; (iv) TBAF, THF, then HCl in MeOH.

3.7 Conclusion:

The fluorinated C-glycosides **3.5**, **3.6** and **3.7** were prepared, *via* a convergent *de novo* synthesis of the β -galacto- residue starting from the readily available 1-thio-1,2-*O*-isopropylidene precursor **2.13** and the mono- and difluoro acids **3.20**, **3.21** and **3.22**. The key step in the synthesis of the monofluoro acids **3.20/3.21** was the reaction of an allylsilane precursor with SelectfluorTM to give a mixture of allylic fluorides. The synthesis of the difluoro acid **3.22** proceeded *via* the Reformatsky type addition of ethyl bromodifluoroacetate on the known aldehyde **3.31**. The mono-fluoro-C-glycosides **3.5** and **3.6** were tested for their P-selectin binding affinities as well as their conformational properties (See chapter 4).

3.8 Experimental section

Hydroxy ester **3.23**

The α -acetoxy acid (0.74 g, 1.16 mmol), derived from the vinyl alcohol **2.18** was dissolved in methanol (10 mL) and toluene (25 mL) and to it was added TMS diazomethane (3.0 mL, 6.00 mmol, 2.0 M solution in ether) dropwise. After 5 min, the reaction was quenched with acetic acid (2 drops) and the solvent removed under reduced pressure. FCC of the crude residue provided the methoxy ester (0.75 g, quantitative yield) as a colorless oil. This ester was then dissolved in dry methanol (10 mL) and NaOMe in methanol (1M, 2.0 mL, 2.0 mmol) was added to the solution. The mixture was stirred for 1 hour and the volatiles removed *in vacuo*. FCC provided the hydroxy ester **3.23** (567 mg, 80%) as colorless oil: $R_f = 0.43$ (30% Ethyl acetate/petroleum ether);

^1H NMR (500 MHz, CDCl_3) δ 2.98 (d, $J = 7.0$ Hz, 1H), 3.57 (m, 1H), 3.60 (dd, $J = 6.3, 10.3$ Hz, 1H), 3.63 (s, 3H), 3.66 (dd, $J = 7.1, 10.2$ Hz, 1H), 3.76 (t, $J = 3.1$ Hz, 1H), 3.97 (dd, $J = 2.7, 9.5$ Hz, 1H), 4.06 (t, $J = 6.5$ Hz, 1H), 4.18 (dd, $J = 1.4, 9.5$ Hz, 1H), 4.27-4.52 (m, 9H), 7.11-7.26 (m, 20H); ^{13}C -NMR (125 MHz, CDCl_3) δ 52.7, 68.3, 70.5, 71.0, 71.8, 72.2, 72.5, 72.9, 73.3, 73.9, 74.3, 75.2, 127.7-128.6 (several lines), 138.1, 138.2, 138.3, 138.4, 173.8.

α -Fluoro ester 3.24.

To a solution of alcohol **3.23** (490 mg, 0.81 mmol) in dichloromethane (10 mL) was added DAST (0.23 mL, 2.42 mmol) at rt. The mixture was then stirred at rt for 1 hour. The reaction was then quenched with a saturated solution of NaHCO_3 in water (5 mL) and extracted with CH_2Cl_2 (3×10 mL). The organic layer was washed with brine, dried (Na_2SO_4) and evaporated *in vacuo*. FCC of the residue gave two unidentified compounds and **3.24** (60 mg, 16%) as a colorless oil: $R_f = 0.78$ (30% Ethyl acetate/petroleum ether); ^1H NMR (500 MHz, CDCl_3) δ 3.58 (s, 3H), 3.69 (dd, $J = 5.5, 10.3$ Hz, 1H), 3.72 (t, $J = 3.7$ Hz, 1H), 3.79 (dd, $J = 6.7, 10.3$ Hz, 1H), 3.82 (t, $J = 6.8$ Hz, 1H), 4.39 (dd, $J = 3.2, 8.7$ Hz, 1H), 4.42-4.59 (m, 8H), 5.18 (dd, $J = 1.9, 48.8$ Hz, 1H), 7.21-7.23 (m, 20H); ^{13}C -NMR (125 MHz, CDCl_3) δ 52.4, 68.4, 71.5 (d, $J = 19.9$ Hz), 71.8, 72.2, 72.3, 72.9, 73.4, 74.2, 74.6, 75.5, 89.6 (d, $J = 191.8$ Hz), 127.8-128.6 (several lines), 138.0, 138.2, 138.3, 138.4, 168.0 (d, $J = 24.4$ Hz).

Allyl trimethylsilanyl derivatives 3.27/3.28.

Grubbs II catalyst (0.45 g, 0.53 mmol) was added as a solid (in three portions over 24 hours) to a solution of the α -C-1-propenyl derivative **3.25/3.26** (5.60 g, 9.93 mmol), and allyltrimethylsilane (6.24 mL, 39.7 mmol) in dichloromethane (60 mL) under an ethylene atmosphere, and the mixture maintained at reflux for 3 days. Anhydrous DMSO (3 mL) was then introduced and stirring continued at reflux for an additional 30 min. The mixture was then concentrated *in vacuo* and the residue purified by FCC to afford allylTMS mannoside **3.27/3.28** (4.41 g, 89% based on recovered **3.25/3.26**) (E/Z ratio: 3/1) as a colorless oil; $R_f = 0.50$ (10% EtOAc/petroleum ether); $^1\text{H NMR}$ (CDCl_3) δ 0.03 (s, Me_3Si , 9H), 1.45 (m, 1H, H-3'), 3.69 (t, $J = 2.9$ Hz, 1H, H-6_b), 3.71-3.82 (m, 4H, H-2, 3, 5, 6_a), 3.94 (t, $J = 8.6$ Hz, 1H, H-4), 4.52 (A of ABq, $J = 10.9$ Hz, $\Delta\delta = 0.33$ ppm, 1H, Ar-CH), 4.52-4.66 (m, 5H, H-1, Ar- $\text{CH}_2 \times 2$), 4.73 (ABq, $J = 12.4$ Hz, $\Delta\delta = 0.07$ ppm, 2H, Ar- CH_2), 4.84 (B of ABq, $J = 10.9$ Hz, $\Delta\delta = 0.33$ ppm, 1H, Ar-CH), 5.32 (dd, $J = 6.2, 15.5$ Hz, 1H, H-1'), 5.61 (ddt, $J = 1.0, 7.7, 25.4$ Hz, 1H, H-2'), 7.19 (m, 2H, Ar-H), 7.25-7.37 (m, 16H, Ar-H) 7.40 (m, 2H, Ar-H); $^{13}\text{C NMR}$ (125 MHz) CDCl_3) δ 1.90, 23.5, 69.9, 71.8, 72.1, 73.5, 73.7, 74.8, 74.9, 75.6, 76.5, 79.3, 123.7-129.1 (several lines), 138.3, 138.6, 138.7; ESIHRMS calcd for $\text{C}_{40}\text{H}_{49}\text{O}_5\text{Si}$ (M + H) 637.3349, found 637.3345.

Fluoro-C-glycoside derivatives 3.29/3.30.

A solution of **3.27/3.28** (4.40 g, 6.92 mmol) and SelectfluorTM (3.18 g, 8.98 mmol) in CH_3CN (70 mL) was stirred at rt for 16 hours. The mixture then concentrated *in vacuo*. Purification by FCC afforded **3.29/3.30** (R/S ratio : 1/1) (2.35 g, 60% based on recovered starting material) as a colorless oil: $R_f = 0.32$ (30% EtOAc/petroleum ether);

^1H NMR (500 MHz, CDCl_3) δ 3.69 (dd, 5.1, 10.7 Hz, 0.5H), 3.72 (dd, $J = 4.0, 10.5$ Hz, 0.5H), 3.74-3.82 (m, 2H), 3.87 (m, 1H), 3.95 (apparent dt, $J = 4.4, 20.1$ Hz, 0.5H), 4.11 (m, 0.5H), 4.49-4.70 (m, 8H), 5.16 (dm, $J = 45.4$ Hz, 1H), 5.29 (d, $J = 10.5$ Hz, 0.5H), 5.31 (d, $J = 10.5$ Hz, 0.5H), 5.33 (dt, $J = 3.2, 17.3$ Hz, 0.5H), 5.39 (dm, 17.3 Hz, 0.5H), 5.92 (dddd, $J = 6.2, 10.1, 16.7, 18.9$ Hz, 0.5H), 6.03 (dddd, $J = 6.2, 10.9, 17.3, 20.2$ Hz, 0.5H), 7.15-7.37 (m, 20H); ^{13}C NMR (125 MHz, CDCl_3) δ 68.5, 69.1, 71.8, 72.3, 72.5, 72.6 (d, $J = 4.2$ Hz), 72.9 (d, $J = 9.2$ Hz), 73.0 (d, $J = 4.0$ Hz), 73.4, 73.5, 73.8, 74.4, 74.6, 75.3 ($\times 2$), 75.5, 91.6 (d, $J = 175.2$ Hz), 93.5 (d, $J = 175.2$ Hz), 119.0 (d, $J = 11.7$ Hz), 119.4 (d, $J = 12.1$ Hz), 127.7-128.6 (several lines), 132.9 (d, $J = 19.4$ Hz), 133.4 (d, $J = 20.2$ Hz), 138.9, 138.2, 138.3 ($\times 2$), 138.4, 138.5; ESIHRMS calcd for $\text{C}_{37}\text{H}_{40}\text{O}_5\text{F}$ ($M + \text{H}$) 583.2860, found 583.2904.

Fluoro acids 3.20/3.21.

Fluoro alkenes **3.29/3.30** (2.35 g, 4.04 mmol) were subjected to the ozonolysis procedure described for the preparation of **2.16**. The aldehyde derivatives (2.30 g, 97%) were produced: $R_f = 0.36$ (30% EtOAc/petroleum ether); ^1H NMR (CDCl_3) δ 3.63-3.79 (m, 2.4H), 3.86-3.92 (m, 1.6H), 4.07 (dd, $J = 2.6, 9.6$ Hz, 0.4H), 4.12 (dd, $J = 2.8, 9.7$ Hz, 0.6H), 4.18 (t, $J = 5.9$ Hz, 0.6H), 4.24 (t, $J = 6.2$ Hz, 0.4H), 4.36-4.69 (m, 9H), 5.01 (dd, $J = 0.8, 49.6$ Hz, 0.4H), 5.07 (dd, $J = 1.4, 48.0$ Hz, 0.6H), 7.19-7.38 (m, 20H), 9.59 (d, $J = 6.4$ Hz, 0.4H), 9.82 (d, $J = 5.9$ Hz, 0.6H); ^{13}C NMR (CDCl_3) δ 67.9, 68.0, 70.5 (d, $J = 17.4$ Hz), 71.1 (d, $J = 19.2$ Hz), 71.9 (3 peaks), 72.0, 73.0 (2 peaks), 73.3, 73.4, 73.6, 74.1, 74.4, 75.3, 75.7, 93.6 (d, $J = 187.9$ Hz), 95.5 (d, $J = 187.0$ Hz), 127.8-128.7 (several

lines), 137.6, 137.8, 138.0 (2 peaks), 138.1, 138.3, 138.4, 196.8 (d, $J = 34.8$ Hz), 199.8 (d, $J = 35.7$ Hz).

The aldehyde mixture from the previous step (2.70 g, 3.94 mmol) was subjected to the procedure that was used to synthesize **2.24**, to give the fluoro acid mixture **3.20/3.21** (2.75 g, 99%): colorless oil; $R_f = 0.29$ (10% MeOH/ CHCl_3); ^1H NMR (CDCl_3) δ 3.66-3.77 (m, 3H), 3.82 (dd, $J = 6.9, 10.3$ Hz, 0.4H), 3.89 (t, $J = 3.1$ Hz, 0.6H), 4.03 (dd, $J = 2.7, 9.4$ Hz, 0.6H), 4.10 (dd, $J = 2.8, 9.1$ Hz, 0.4H), 4.17 (m, 1H), 4.36-4.57 (m, 9H), 5.17 (dd, $J = 1.7, 48.4$ Hz, 0.4H), 5.01 (dd, $J = 1.7, 47.5$ Hz, 0.6H), 7.16-7.38 (m, 20H); ^{13}C NMR (CDCl_3) δ 68.1, 68.2, 70.4 (d, $J = 18.3$ Hz), 71.1 (d, $J = 20.2$ Hz), 71.9, 72.0, 72.2 (d, $J = 5.5$ Hz), 72.3 (d, $J = 2.7$ Hz), 72.9 (2 peaks), 73.3, 74.0, 74.2, 74.3, 75.2, 75.6, 87.8 (d, $J = 192.5$ Hz), 89.0 ($J = 190.6$ Hz), 127.8-128.6 (several lines), 137.8 (2 peaks), 138.0 (2 peaks), 138.2 (2 peaks), 171.3 (d, $J = 24.7$ Hz), 172.0 (d, $J = 25.7$ Hz); ESIHRMS calcd for $\text{C}_{36}\text{H}_{38}\text{O}_7\text{F}$ ($\text{M} + \text{H}$) 601.2602, found 601.2623.

2,3,4-tris(benzyloxy)hex-5-enal 3.31.

A solution of methyl- α -D-mannopyranoside (25.0 g, 0.13 mol), trityl chloride (43.1 g, 1.55 mol), DMAP (0.16 g, 0.13 mmol) in dry DMF (50 mL) and pyridine (50 mL) was stirred for 12 hours. Removal of the solvents *in vacuo* gave the crude product. FCC of the residue yielded the 6-O-trityl derivative (58 g, 99%): colorless oil; $R_f = 0.43$ (100% Ethyl acetate); ^1H NMR (500 MHz, CDCl_3) δ 2.35 (broad s, 1H, D_2O exchange), 2.75 (broad s, 1H, D_2O exchange), 3.29 (s, 3H), 3.35 (m, 2H), 3.55-3.64 (m, 2H), 3.68 (dd, $J = 3.0, 8.7$ Hz, 1H), 3.82 (m, 1H), 4.63 (s, 1H), 7.15-7.34 (m, 15H); ^{13}C NMR (125

MHz, CDCl₃) δ 55.0, 65.0, 70.2, 70.3, 71.9, 85.4, 100.9, 127.3-128.9 (several lines), 143.9, 144.0.

To a solution of methyl 6-O-trityl- α -D-mannopyranoside (53.2 g, 0.12 mol) in dry DMF (300 mL) at 0 °C was added NaH (19.5 g, 60% in mineral oil, 0.48 mol) and Bu₄Ni (4.50 mg, 0.01 mol) followed by BnBr (49.4 mL, 0.41 mmol). The reaction was stirred for 14 hours at rt under an argon atmosphere, then diluted with water (280 mL) and extracted with ether (2 \times 50 mL). The organic layer was washed with water (3 \times 250 mL), dried (Na₂SO₄) and concentrated *in vacuo*. FCC of the crude methyl-2,3,4-trio-O-benzyl-6-O-trityl- α -D-mannoside. This crude product was dissolved in 9:1 MeOH:EtOAc (430 mL) and *p*-toluenesulfonic acid (4.9 g, 0.02 moles) was added to the solution. The reaction was stirred for 3 hours at rt. Triethylamine (3 mL) was added and the solvents was evaporated under reduced pressure, and the residue was purified by FCC to provide methyl-2,3,4-trio-O-benzyl- α -D-mannopyranoside (48.7 g, 86% over two steps); R_f = 0.47 (30% EtOAc/petroleum ether); ¹H NMR (500 MHz, CDCl₃) δ 2.04 (broad s, 1H, D₂O exchange), 3.32 (s, 3H), 3.64 (ddd, J = 3.0, 4.6, 9.5 Hz, 1H), 3.77-3.87 (m, 3H), 3.92 (dd, J = 3.0, 9.4 Hz, 1H), 3.99 (t, J = 9.4 Hz, 1H), 4.60-4.82 (m, 6H), 4.95 (d, J = 11.0 Hz, 1H), 7.20-7.44 (m, 15H); ¹³C NMR (125 MHz, CDCl₃) δ 55.0, 62.8, 72.3, 72.5, 73.2, 75.0, 75.2, 75.4, 80.5, 99.6, 127.8-128.6 (several lines), 138.5, 138.67, 138.71.

A mixture of the product from the previous step (36.0 g, 0.08 mol), triphenylphosphine (50.8 g, 0.19 mol), imidazole (10.6 g, 0.16 mol) and iodine (49.0 g, 0.19 mol) in anhydrous toluene (300 mL) was heated at reflux for 2 hours. The mixture was then cooled to rt and diluted with ether (400 ml), filtered through a bed of Celite and

evaporated under reduced pressure. FCC of the residue gave the iodide (41.5 g, 93%) as colorless oil: $R_f = 0.37$ (10% EtOAc/petroleum ether); ^1H NMR (500 MHz, CDCl_3) δ 3.34 (dd, $J = 7.7, 10.3$ Hz, 1H), 3.39 (s, 3H), 3.54 (ddd, $J = 2.3, 7.9, 9.1$ Hz, 1H), 3.58 (dd, $J = 2.3, 10.3$ Hz, 1H), 3.79 (m, 2H), 3.91 (dd, $J = 3.1, 9.3$ Hz, 1H), 4.63-4.79 (m, 6H), 5.00 (d, $J = 11.0$ Hz, 1H), 7.28-7.40 (m, 15H); ^{13}C NMR (125 MHz, CDCl_3) δ 7.2, 55.3, 71.7, 72.3, 73.0, 74.9, 75.6, 78.9, 80.2, 99.3, 127.8-128.7 (several lines), 138.4, 138.50, 138.53; ESIHRMS calcd for $\text{C}_{28}\text{H}_{35}\text{O}_5\text{IN}$ $[\text{M} + \text{NH}_4]^+$ 592.1554, found 592.1557.

To a solution of the above iodide (44.5 g, 0.08 mol) in THF (200 mL) was added *n*-BuLi (37.2 mL of 2.5 M solution in THF, 0.09 mol) dropwise at -78 °C, under an atmosphere of argon. The reaction was stirred at this temperature for 1.5 hours, then poured into saturated aqueous NaHCO_3 and extracted with ether. The combined organic phase was dried (Na_2SO_4), filtered and evaporated *in vacuo*. FCC of the residue afforded 2,3,4-tris(benzyloxy)hex-5-enal **3.31** (30.5 g, 94%), $R_f = 0.29$ (10% EtOAc/petroleum ether); ^1H NMR (500 MHz, CDCl_3) δ 3.92 (dd, $J = 3.8, 5.6$ Hz, 1H), 4.10 (dd, $J = 1.3, 3.4$ Hz, 1H), 4.13 (dd, $J = 6.7, 7.4$ Hz, 1H), 4.38-4.74 (m, 6H), 5.37 (d, $J = 10.1$ Hz, 1H), 5.41 (d, $J = 17.1$ Hz, 1H), 5.90 (ddd, $J = 7.8, 10.3, 17.7$ Hz, 1H), 7.27-7.39 (m, 15H), 9.69 (d, $J = 1.3$ Hz, 1H); ^{13}C NMR (125 MHz, CDCl_3) δ 70.9, 72.9, 74.2, 80.7, 83.0, 83.8, 119.9, 127.7-128.6 (several lines), 135.3, 137.5, 138.1, 138.3, 201.7; ESIHRMS calcd for $\text{C}_{27}\text{H}_{32}\text{O}_4\text{N}$ $[\text{M} + \text{NH}_4]^+$ 434.2326, found 434.2250.

(3R,3S)-Ethyl 4,5,6-tris(benzyloxy)-2,2-difluoro-3-hydroxyoct-7-enoate 3.48/3.49.

To a suspension of activated zinc dust (13.2 g, 0.21 mol) in dry THF (30 mL) at reflux was added ethyl bromodifluoroacetate (30.0 g, 19.1 mL, 0.15 mol). After 10 min, a solution of **3.31** (20.0 g, 0.05 mmol) in THF (60 mL) was added dropwise over 30 min and the reaction mixture heated to reflux for a further 3 hours. The mixture was cooled to rt and carefully poured into 1M HCl (40 mL) and ice (40 g). The mixture was extracted with EtOAc (3 × 150 mL) and the organic layer washed with saturated aqueous NaHCO₃ and brine. This was then dried (Na₂SO₄) and concentrated *in vacuo* to give a dark brown residue. FCC of the crude gave (3R,3S)-ethyl 4,5,6-tris(benzyloxy)-2,2-difluoro-3-hydroxyoct-7-enoate (**3.32/3.33**) as a mixture of epimeric alcohols (13.9 g, 54%) in a 3:1 ratio (¹H-NMR): colorless oil; R_f = 0.29 (10% EtOAc/ petroleum ether); ¹H NMR (500 MHz, CDCl₃) δ 1.74 (t, J = 7.1 Hz, 2.3H), 1.32 (t, J = 7.1 Hz, 0.7H), 3.38 (dd, J = 3.2, 10.0 Hz, 0.3H), 3.58 (m, 0.7H) 3.77 (m, 0.3H), 3.86 (m, 1.7H), 3.94-4.08 (m, 1.7H), 4.11-4.22 (m, 0.7H), 4.31 (q, J = 7.1 Hz 0.6H), 4.40-4.87 (m, 7H), 5.37-5.50 (m, 2H), 5.92 (m, 0.7H), 6.03 (m, 0.3H), 7.27-7.39 (m, 15H); major isomer, ¹³C NMR (125 MHz, CDCl₃) δ 13.8, 62.5, 70.4, 75.2, 78.9 (d, J = 3.9 Hz), 81.6, 82.9, 114.8 (dd, J = 257.3, 257.4 Hz), 120.0, 127.9, 128.1-128.6 (several lines), 135.3, 137.4, 138.01, 138.4, 163.4 (t, J = 30.8 Hz). Minor isomer; 14.0, 63.2, 70.4, 70.6, 73.9 (d, J = 3.4 Hz), 74.3, 75.3, 80.0, 81.7, 119.5, 127.8, 135.8, 137.6, 137.99, 138.2, 163.6 (dd, J = 30.0, 33.1 Hz); ESIHRMS calcd for C₃₁H₃₈O₆F₂N [M + NH₄]⁺ 558.2662, found 558.2651.

(3S)- and (3R)-ethyl 3-acetoxy-4,5,6-tris(benzyloxy)-2,2-difluorooct-7-enoate 3.34 and 3.35.

Treatment of the mixture **3.32/3.33** (0.54 g, 1.09 mmol) following the acetylation procedure described in the synthesis of **2.28** provided compounds **3.34** (0.40 g, 64%) and **3.35** (0.13 g, 21%) as colorless oils.

For (3S)-ethyl 3-acetoxy-4,5,6-tris(benzyloxy)-2,2-difluorooct-7-enoate (3.34): $R_f = 0.38$ (10% EtOAc/ petroleum ether); $^1\text{H NMR}$ (CDCl_3) δ 1.08 (t, $J = 7.1$ Hz, 3H), 2.20 (s, 3H), 3.74 (d, $J = 8.1$ Hz, 1H), 3.92 (q, $J = 7.1$ Hz, 1H), 4.06 (m, 3H), 4.41 (A of ABq, $J = 10.8$ Hz, $\Delta\delta = 0.15$ ppm, 1H), 4.44 (A of ABq, $J = 12.2$ Hz, $\Delta\delta = 0.25$ ppm, 1H), 4.56 (B of ABq, $J = 10.8$ Hz, $\Delta\delta = 0.15$ ppm, 1H), 4.68 (B of ABq, $J = 12.2$ Hz, $\Delta\delta = 0.25$ ppm, 1H), 4.87 (ABq, $J = 12.0$ Hz, $\Delta\delta = 0.04$ ppm, 2H), 5.39 (d, $J = 17.1$ Hz, 1H), 5.46 (dd, $J = 1.2, 10.5$ Hz, 1H), 5.81 (ddd, $J = 7.8, 10.5, 17.4$ Hz, 1H), 6.00 (ddd, $J = 3.2, 10.5, 20.8$ Hz, 1H), 7.19-7.44 (m, 15H); $^{13}\text{C NMR}$ (125 MHz, CDCl_3) δ 13.7, 20.6, 62.8, 69.4 (t, $J = 22.0$ Hz), 70.5, 73.0, 75.0, 77.3 (d, $J = 2.7$ Hz), 81.2, 81.8, 113.3 (dd, $J = 251.1, 257.5$ Hz), 120.4, 127.9-128.7 (several lines), 135.0, 137.1, 138.4, 138.8, 162.3 (dd, $J = 28.9, 33.1$ Hz), 168.4; ESIHRMS calcd. for $\text{C}_{33}\text{H}_{40}\text{O}_7\text{F}_2\text{N}$ $[\text{M} + \text{NH}_4]^+$ 600.2767, found 600.2761.

For (3R)-ethyl 3-acetoxy-4,5,6-tris(benzyloxy)-2,2-difluorooct-7-enoate (3.35): $R_f = 0.28$ (10% EtoAc/ petroleum ether); $^1\text{H NMR}$ (500 MHz, CDCl_3) δ 1.23 (t, $J = 7.1$ Hz, 3H), 2.07 (s, 3H), 3.78 (dd, $J = 4.2, 6.5$ Hz, 1H), 4.20 (m, 4H), 4.38-4.25 (m, 2H), 4.64-4.80 (m, 4H), 5.42 (d, $J = 10.4$ Hz, 1H), 5.48 (d, $J = 17.3$ Hz, 1H), 5.80 (ddd, $J = 2.2, 12.7, 19.9$ Hz, 1H), 6.01 (ddd, $J = 7.8, 10.4, 17.6$ Hz, 1H), 7.28-7.43 (m, 15H); $^{13}\text{C NMR}$ (125 MHz, CDCl_3) δ 13.9, 20.7, 63.3, 70.0 (dd, $J = 24.6, 28.6$ Hz) 73.2, 74.2, 75.1, 80.6, 81.8,

113.2 (dd, $J = 255.2, 257.4$ Hz), 119.4, 127.7-128.5 (several lines), 135.9, 138.1, 138.4, 138.5, 162.8 (dd, $J = 30.3, 32.9$ Hz), 168.5; ESIHRMS calcd for $C_{33}H_{40}O_7F_2N$ [$M + NH_4$]⁺ 600.2767, found 600.2753.

Epoxides 3.36/2.37

A solution of acetate **3.34** (1.11 g, 1.90 mmol) in dry MeOH (20 mL) was treated with MeONa/MeOH solution (1M, 5.7 mL, 5.70 mmol). After stirring for 1 hour at rt, the reaction was neutralized with 2M HCl and the solvent evaporated at reduced pressure. FCC of the crude product gave (3S)-methyl 4,5,6-tris(benzyloxy)-2,2-difluoro-3-hydroxyoct-7-enoate (0.79 g, 83%) as a colorless oil: $R_f = 0.54$, (10% EtOAc/petroleum ether); 1H NMR (500 MHz, $CDCl_3$) δ 3.50 (s, 3H), 3.52 (m, 1H), 3.84 (m, 2H), 4.11 (t, $J = 7.1$ Hz, 1H), 4.36-4.73 (m, 5H), 4.83 (s, 2H), 5.37 (d, $J = 18.6$ Hz, 1H), 5.41 (d, $J = 18.6$ Hz, 1H), 5.90 (ddd, $J = 8.3, 10.0, 17.6$ Hz, 1H), 7.14-7.46 (m, 15H); ^{13}C NMR (125 MHz, $CDCl_3$) δ 52.9, 70.7 (t, $J = 22.9$ Hz), 70.8, 72.9, 75.3, 77.3 (d, $J = 2.7$ Hz), 81.2, 81.8, 113.3 (dd, $J = 251.1, 257.5$ Hz), 120.4, 127.9-128.7 (several lines), 135.0, 137.1, 138.4, 138.8, 163.8 (dd, $J = 30.2, 32.1$ Hz); ESIHRMS calcd for $C_{30}H_{36}O_6F_2N$ [$M + NH_4$]⁺ 544.2505, found 544.2505.

To a solution of the above alcohol (0.79 g, 1.50 mmol) in CH_2Cl_2 (20 mL) was added a mixture of *m*-CPBA (2.61 g, 15.1 mmol), in CH_2Cl_2 (20 mL), NaH_2PO_4 (4.30 g, 30.3 mmol) and Na_2HPO_4 (4.14 g, 30.0 mmol) in water (40 mL). The suspension was stirred for 26 hours at rt and poured into 10% Na_2SO_3 in saturated $NaHCO_3$ solution. After stirring for 1 hour, the organic layer was separated, washed with brine, dried (Na_2SO_4) and evaporated under reduced pressure. FCC of this residue afforded a mixture

of epimeric epoxides **3.36/3.37** (0.50 g, 74% based on recovered alkene, 2:1 ratio) as an oil: $R_f = 0.30$ (15% EtOAc/petroleum ether); ^1H NMR (500 MHz, CDCl_3) δ 2.47 (dd, $J = 2.7, 4.9$ Hz, 0.6H), 2.60 (t, $J = 4.5$ Hz, 0.6H), 2.68 (dd, $J = 2.6, 4.9$ Hz, 0.4H), 2.84 (t, $J = 4.6$ Hz, 0.4H), 3.09 (m, 0.4H), 3.36 (m, 0.6H), 3.30 (m, 0.6H), 3.53 -3.60 (m, 3.8H), 3.68 (m, 0.6H), 3.91 (m, 1.6H), 4.01 (d, $J = 5.9$ Hz, 0.4H), 4.45-4.90 (m, 7H), 7.18-7.40 (m, 15H); ^{13}C NMR (CDCl_3) δ (major isomer) 43.9, 52.7, 53.1, 70.6 (t, $J = 22.5$ Hz), 73.9, 74.4, 78.9 (d, $J = 3.3$ Hz), 80.8, 80.9, 114.9 (dd, $J = 250.0, 258.1$ Hz), 122.9-128.7 (several lines), 137.4, 137.9, 138.1, 138.4, 138.8, 163.8 (dd, $J = 30.2, 32.3$ Hz); ESIHRMS calcd for $\text{C}_{30}\text{H}_{36}\text{O}_7\text{F}_2\text{N}$ $[\text{M} + \text{NH}_4]^+$ 560.2454, found 560.2452.

Difluoro-C-glycosides 3.38/3.39.

A solution of the epoxide mixture **3.36/3.37** in dry MeOH (400 mL) was treated with MeONa/MeOH solution (1M, 23.2 mL, 23.2 mmol). After stirring for 23 hours at rt, the reaction was acidified to pH 2 with a solution of HCl in ether and the solvent evaporated at reduced pressure. The crude product was then dissolved in MeOH (20 mL) and toluene (60 mL) and treated with TMSN_2CH (5.8 mL, 11.6 mmol, 2M solution in ether) at 0 °C. After 30 min, acetic acid (1 mL) was added to the reaction and the solvents evaporated at the pump. FCC of the crude product yielded the difluorinated monosaccharides **3.38/3.39** (2.16 g, 52%) as an inseparable mixture of *D/L*- sugars and a second compound that was presumed to be the heptoside derivative **3.40** (0.95 g, 22%).

For 3.38/3.39. Colorless oil: $R_f = 0.49$ (30% EtOAc/petroleum ether); ^1H NMR (500 MHz, CDCl_3) δ 1.66 (d, $J = 8.9$ Hz, D_2O exchange, 0.5H), 1.87 (t, $J = 6.2$ Hz, D_2O exchange, 0.5H), 3.42 (dd, $J = 1.2, 3.3$ Hz, 0.5H), 3.50 (m, 0.5H), 3.65 (m, 2H), 3.75-

3.98 (m, 5.5H), 4.08 (dd, $J = 3.2, 5.2$ Hz, 0.5H), 4.28-4.72 (m, 7H), 7.15-7.38 (m, 15H); ^{13}C NMR (125 MHz, CDCl_3) δ 53.0, 53.4, 61.7, 62.0, 71.4, 71.9, 72.0, 72.4, 72.5, 72.7-73.2 (several lines), 73.4, 74.3, 74.8, 75.7, 76.9, 114.0 (t, $J = 254.7$ Hz), 114.9 (dd, $J = 256.3, 260.3$ Hz), 127.7-128.6 (several lines), 137.4, 137.6, 137.9, 138.0, 163.37 (t, $J = 30.9$ Hz), 163.40 (dd, $J = 27.0, 31.8$ Hz); ESIHRMS calcd for $\text{C}_{30}\text{H}_{36}\text{O}_7\text{F}_2\text{N}$ $[\text{M} + \text{NH}_4]^+$ 560.2454, found 560.2489.

For compound 3.40: colorless oil: $R_f = 0.60$ (30% EtOAc/petroleum ether); ^1H NMR (500 MHz, CDCl_3) δ 3.13 (d, $J = 8.3$ Hz, D_2O exchange, 1H), 3.62 (s, 3H), 3.71 (m, 1H), 3.77 (dd, $J = 4.9, 12.6$ Hz, 1H), 3.83 (t, $J = 5.2$, 1H), 4.02 (d, $J = 6.0$ Hz, 1H), 4.06 (d, $J = 12.7$ Hz, 1H), 4.33-4.44 (m, 4H), 4.60 (s, 2H), 4.70 (ABq, $J = 11.8$ Hz, $\Delta\delta = 0.03$ ppm, 2H), 7.25-7.37 (m, 15H); ^{13}C NMR (125 MHz, CDCl_3) δ 53.3, 66.5, 67.2, 72.0, 72.5, 73.3, 74.1, 75.2, 80.1, 115.7 (t, $J = 255.7$ Hz), 127.1-128.8 (several lines), 137.5, 137.7, 138.0, 163.8 (t, $J = 31.5$ Hz); ESIHRMS calcd for $\text{C}_{30}\text{H}_{36}\text{O}_7\text{F}_2\text{N}$ $[\text{M} + \text{NH}_4]^+$ 560.2454, found 560.2451.

β -C-L-gulo pyranosides 3.41 and α -C-D-manno pyranosides 3.42.

The mixture **3.38/3.39** (213.0 mg, 0.39 mmol), TBDPSCl (0.03 mL, 1.18 mmol), and imidazole (106.0 mg, 1.56 mmol) in anhydrous DMF (5 mL) was stirred at 50 °C for 2.5 hours. The reaction was then quenched with methanol (1 mL) and extracted with ether. The combined organic phase was washed with brine, dried (Na_2SO_4), filtered, and evaporated under reduced pressure. The residue was purified by gravity column chromatography to give the *L*-guloside **3.41** (119.0 mg, 39%) and *D*-mannoside **3.42** (180.0 mg 59%) as colorless oils.

For β -C-L-*gulo* pyranosides 3.41: $R_f = 0.62$ (10% EtOAc/petroleum ether); ^1H NMR (500 MHz, CDCl_3) δ 1.09 (s, 9H), 3.61 (s, 3H), 3.67 (dd, $J = 1.1, 3.7$ Hz, 1H), 3.75 (t, $J = 2.7$ Hz, 1H), 3.81 (m, 2H), 3.96 (dd, $J = 2.6, 10.0$ Hz, 1H), 4.03 (t, $J = 6.8$ Hz, 1H), 4.32-4.70 (m, 5H), 4.48 (A of ABq, $J = 12.1$ Hz, $\Delta\delta = 0.21$ ppm, 1H), 4.69 (B of ABq, $J = 12.1$ Hz, $\Delta\delta = 0.21$ ppm, 1H), 7.16 (m, 2H), 7.28-7.47 (m, 19H), 7.67 (m, 4H); ^{13}C NMR (125 MHz, CDCl_3) δ 27.0, 53.2, 62.3, 72.0, 72.1, 73.1, 73.2, 73.4 (two signals, a singlet and an apparent t, $J = 23.0$ Hz), 74.8, 75.7, 114.3 (t, $J = 254.5$ Hz), 127.9-128.8 (several lines), 133.4, 133.7, 135.7, 135.8, 137.8, 138.4 (two signals), 163.7 (t, $J = 31.3$ Hz); ESIHRMS calcd for $\text{C}_{46}\text{H}_{54}\text{O}_7\text{SiF}_2\text{N}$ $[\text{M} + \text{NH}_4]^+$ 798.3632 found 798.3630.

For α -C-D-*manno* pyranosides 3.42: $R_f = 0.68$ (10% EtOAc/petroleum ether); ^1H NMR (500 MHz, CDCl_3) δ 1.11 (s, 9H), 3.69 (s, 3H), 3.91 (dd, $J = 6.4, 12.5$ Hz, 1H), 3.94-4.00 (m, 3H), 4.07 (t, $J = 5.9$ Hz, 1H), 4.13 (dd, $J = 3.2, 5.6$ Hz, 1H), 4.51 (dd, $J = 5.6, 10.5, 19.0$ Hz, 1H), 4.60-4.71 (m, 6H), 7.26 (m, 2H), 7.34-7.50 (m, 19H), 7.72 (m, 4H); ^{13}C NMR (125 MHz, CDCl_3) δ 26.9, 53.4, 62.8, 72.3, 72.87, 72.94 (dd, $J = 22.3, 27.0$ Hz), 73.5, 74.1, 77.4, 115.2 (dd, $J = 256.5, 258.8$ Hz), 127.8-129.8 (several lines), 133.5, 133.8, 135.8, 136.0, 137.9, 138.3, 138.4, 163.6 (t, $J = 31.4$ Hz); ESIHRMS calcd for $\text{C}_{46}\text{H}_{54}\text{O}_7\text{SiF}_2\text{N}$ $[\text{M} + \text{NH}_4]^+$ 798.3632 found 798.3637.

β -C-L-tetra-O-guloside 3.43

Treatment of *L*-guloside **3.39** (50.0 mg, 0.09 mmol) following the hydrogenation-acetylation sequence described in the synthesis of **2.28** provided the tetra-O-acetylated derivative **3.43** (40.0 mg, 98%): colorless oil; $R_f = 0.42$ (40% EtOAc/petroleum ether); ^1H NMR (500 MHz, C_6D_6) δ 1.98, 2.06, 2.20 (all s, 12H, $\text{CH}_3\text{CO} \times 4$), 3.91 (s, 3H,

CH₃O-), 4.14-4.66 (m, 2H, H-5, 6a), 4.17 (t, J = 5.9 Hz, 1H, H-6b), 4.34 (ddd, J = 7.8, 8.3, 14.4 Hz, 1H, H-1), 5.00 (dd, J = 1.5, 3.9 Hz, 1H, H-4), 5.40 (dd, J = 3.2, 10.1 Hz, 1H, H-2), 5.43 (t, J = 3.2 Hz, 1H, H-3); ¹³C NMR (125 MHz, CDCl₃) δ 20.6, 20.8, 20.9 (two signals), 53.6, 61.9, 63.9 (d, J = 4.2 Hz), 66.6, 67.8, 72.9 (dd, J = 23.8, 28.4 Hz), 73.0, 113.6 (dd, J = 252.0, 261.5 Hz), 163.2 (dd, J = 30.2, 32.2 Hz), 169.0 (two signals), 169.7, 170.5; ESIHRMS calcd for C₁₇H₂₆O₁₁F₂N [M + NH₄]⁺ 458.1468, found 458,1467.

6-O-tertbutyldiphenylsilyl-α-C-D-tri-O-acetyl mannoside 3.44.

Treatment of *D*-mannoside **3.42** (45.0 mg, 0.06 mmol) following the hydrogenation-acetylation sequence described in the synthesis of **2.28** provided the tri-O-acetate derivative **3.44** (30.0 mg, 81%): colorless oil; R_f = 0.44 (30% EtOAc/petroleum ether); ¹H NMR (500 MHz, CDCl₃) δ 1.08 (s, 9H, (CH₃)₃-), 1.99, 2.00, 2.12 (all s, 9H, CH₃CO × 3), 3.71 (dd, J = 3.2, 11.7 Hz, 1H, H-6a), 3.75 (dd, J = 4.7, 11.7 Hz, 1H, H-6b), 3.82 (s, 3H, CH₃O-), 3.98 (m, 1H, H-5), 4.42 (ddd, J = 3.6, 6.6, 23.4 Hz, 1H, H-1), 5.38 (dd, J = 3.6, 8.3 Hz, 1H, H-3), 5.43 (t, J = 8.3 Hz, 1H, H-4), 5.65 (t, J = 3.4 Hz, 1H, H-2), 7.36-7.42 (m, 6H), 7.64-7.71 (m, 4H); ¹³C NMR (125 MHz, CDCl₃) δ 20.8 (two signals), 20.9, 26.9, 53.9, 62.5, 65.7, 66.1, 69.4 (d, J = 4.3 Hz), 74.3 (dd, J = 22.0, 30.0 Hz), 76.4, 114.5 (dd, J = 258.1, 262.1 Hz), 127.9 (two signals), 128.5, 130.0, 133.2, 133.4, 135.8, 135.9, 162.9 (dd, J = 29.2, 32.7 Hz), 169.5, 169.7, 169.9; ESIHRMS calcd for C₃₁H₄₂O₁₀SiF₂N [M + NH₄]⁺ 654.2541, found 654. 2537.

**(3*S*,7*R*)-ethyl-3-acetoxy-4,5,6-tris(benzyloxy)-2,2-difluoro-7,8-dihydroxyoctanoate
3.45/3.46.**

To a solution of **3.34** (0.70 g, 1.72 mmol) in acetone (30 mL) was added OsO₄ (0.1 mL, 0.10 mmol, 1M in butanol) and NMO (1.20 mL, 5.16 mmol, 50% wv in water). The mixture was stirred for 5 hours at rt then quenched with solid Na₂SO₃ (0.40 g). Ethyl acetate (20 mL) was added to the reaction mixture, and after separation of the layers, the aqueous phase was further extracted with the ethyl acetate (2 × 10 mL). The combined organic phase was washed with brine and dried over Na₂SO₄. After removal of the solvents *in vacuo*. FCC afforded **3.45/3.46** (0.67 g, 90%) as a 10:1 mixture of epimeric alcohol: colorless oil; R_f = 0.48 (50% EtOAc/petroleum ether); ¹H NMR (500 MHz, CDCl₃) δ 1.04 (t, J = 7.1 Hz, 0.3H), 1.08 (t, J = 7.1 Hz, 2.7H), 1.98 (t, J = 6.0 Hz, 1H), 2.09 (s, 0.3H), 2.12 (s, 2.7 H), 2.59 (d, J = 5.1 Hz, 0.1H), 2.61 (d, J = 6.1 Hz, 0.9H), 3.65 (m, 3H), 3.80 (m, 1H), 3.80-4.01 (m, 3H), 4.10 (dd, J = 1.9, 9.1 Hz, 1H), 4.50-4.83 (m, 6H), 6.08 (ddd, 3.2, 9.2, 20.6 Hz, 1H), 7.21-7.38 (m, 15H); ¹³C NMR (125 MHz, CDCl₃) δ (major isomer) 13.7, 20.8, 63.0, 63.4, 69.3 (t, J = 24.1 Hz), 71.6, 74.0, 74.4, 74.7, 77.2, 79.1, 78.0, 114.6 (t, J = 257.0 Hz), 127.8-128.9 (several lines), 136.9, 137.8, 138.0, 163.4 (t, J = 31.7 Hz), 168.9; ESIHRMS calcd for C₃₃H₄₂O₉F₂N [M + NH₄]⁺ 634.2822 found 634.2825.

Epoxides 3.36 and 3.47.

To a solution of the diol mixture **3.45/3.46** (0.65 g, 1.04 mmol) in pyridine (10 mL) at 0 °C was added *p*-toluenesulfonyl chloride (0.24 g, 1.24 mmol) at 0 °C. The reaction was then stirred at rt 4 hours. The solvent was removed under reduced pressure,

and the residue purified by FCC to give a mixture of mono tosylates (0.53 g, 87% based on recovered diol): colorless oil, $R_f = 0.44$ (30% EtOAc/petroleum ether); ^1H NMR (500 MHz, CDCl_3) δ 1.12 (t, $J = 7.1$ Hz, 3H), 2.12 (s, 3H), 2.45 (s, 3H), 2.62 (s, 1H, D_2O exchange, 1H), 3.65 (t, $J = 6.3$ Hz, 1H), 3.93-4.02 (m, 4H), 4.14 (m, 2H), 4.24 (dd, $J = 2.6, 10.1$ Hz, 1H), 4.46-4.84 (m, 6H), 6.12 (ddd, 2.4, 8.5, 20.7 Hz, 1H), 7.17-7.41 (m, 17H), 7.81 (m, 2H); ^{13}C NMR (125 MHz, CDCl_3) δ (major isomer) 13.7, 20.7, 21.7, 62.9, 68.9 (t, $J = 22.1$ Hz), 70.1, 71.3, 73.9, 74.6, 77.1, 78.3, 79.5, 113.1 (dd, $J = 251.3, 258.8$ Hz), 128.0-128.7 (several lines), 130.1, 132.8, 136.9, 137.8 (two signals), 145.2, 162.2 (dd, $J = 29.6, 32.7$ Hz), 168.8; ESIHRMS calcd for $\text{C}_{40}\text{H}_{48}\text{O}_{11}\text{SF}_2\text{N}$ $[\text{M} + \text{NH}_4]^+$ 788.2911 found 788.2910.

The mixture from the previous step (0.53 g, 0.69 mmol) was treated with base following the procedure described for the synthesis of **3.38/3.39**. The hydroxy epoxide **3.36** (0.16 g, 34%) and the acetylated derivative **3.47** (0.13 g, 39%) were produced.

For hydroxyl epoxide 3.36: colorless oil; $R_f = 0.38$ (30% EtOAc/petroleum ether); ^1H NMR (500 MHz, CDCl_3) δ 2.69 (dd, $J = 2.7, 4.9$ Hz, 1H), 2.84 (t, $J = 4.7$ Hz, 1H), 3.09 (m, 1H), 3.53 (s, 3H), 3.56 (t, $J = 6.2$ Hz, 1H), 3.58 (dd, $J = 1.6, 8.5$ Hz, 1H), 3.93 (d, $J = 9.1$ Hz, 1H), 4.02 (d, $J = 5.9$ Hz, 1H), 4.47-4.88 (m, 7H), 7.27-7.41 (m, 15H); ^{13}C NMR (125 MHz, CDCl_3) δ 47.3, 51.1, 52.9, 70.3 (t, $J = 22.0$ Hz), 73.3, 73.9, 75.2, 78.3 (d, $J = 3.4$ Hz), 79.5, 81.5, 114.9, (dd, $J = 249.9, 256.8$ Hz), 127.7-128.7 (several lines), 137.2, 137.7, 138.3, 163.8 (t, $J = 31.6$ Hz); ESIHRMS calcd for $\text{C}_{30}\text{H}_{36}\text{O}_7\text{F}_2\text{N}$ $[\text{M} + \text{NH}_4]^+$ 560.2454, found 560.2457.

For epoxide 3.47: $R_f = 0.41$ (30% EtOAc/petroleum ether); ^1H NMR (500 MHz, CDCl_3) δ 2.06 (s, 3H), 2.71 (dd, $J = 2.6, 5.2$ Hz, 1H), 2.78 (m, 1H), 3.16 (m, 1H), 3.49 (s, 3H),

3.59 (t, $J = 5.2$ Hz, 1H), 3.79 (dd, $J = 2.9, 5.4$ Hz, 1H), 4.16 (dd, $J = 2.9, 8.4$ Hz, 1H), 4.55-4.83 (m, 6H), 6.02 (ddd, $J = 4.2, 8.4, 19.3$ Hz, 1H), 7.28-7.42 (m, 15H); ^{13}C NMR (75 MHz, CDCl_3) δ 20.6, 45.9, 51.1, 53.1, 69.6 (t, $J = 22.0$ Hz), 73.9, 74.1, 74.4, 77.1, 77.7, 80.6, 113.2 (dd, $J = 251.4, 258.2$ Hz), 127.8-128.6 (several lines), 137.1, 138.0, 138.1, 138.2, 162.7 (dd, $J = 29.6, 32.7$ Hz), 168.7.

Difluoro *L*-guloside **3.39**

Hydroxy epoxide **3.36** (128.0 mg, 0.24 mmol) was treated with base as described for the synthesis of **3.38/3.39**. This afforded the *L*-guloside **3.39** (58.0 mg, 62%) and the heptose sugar **40** (12.7 mg, 14%) based on recovered epoxides. The acetoxy derivative **3.47** (158.0 mg, 0.27 mmol) also gave **3.39** (108.6 mg, 69%): colorless oil; ^1H NMR (500 MHz, CDCl_3) δ 1.75 (d, $J = 7.1$ Hz, D_2O exchange, 1H), 3.45 (s, 3H), 3.71 (dd, $J = 1.7, 4.2$ Hz, 1H), 3.53 (m, 1H), 3.80 (apparent t, $J = 3.4$ Hz, 1H), 3.84 (dd, $J = 7.8, 11.4$ Hz, 1H), 3.97 (m, 2H), 4.32-4.72 (m, 7H), 7.19-7.38 (m, 15H); ^{13}C NMR (125 MHz, CDCl_3) δ 53.2, 62.3, 71.7, 72.2, 72.7, 73.3 (apparent d, $J = 2.0$ Hz), 73.5 (two signals, s and an apparent t, $J = 23.1$ Hz), 75.1, 76.0, 114.2 (t, $J = 254.8$ Hz), 127.9-128.8 (several lines), 137.6, 137.8, 138.3, 163.6 (t, $J = 31.3$ Hz); ESIHRMS calcd for $\text{C}_{30}\text{H}_{36}\text{O}_7\text{F}_2\text{N}$ [$\text{M} + \text{NH}_4$] $^+$ 560.2454, found 560.2456.

Difluoro acid **3.22**.

The difluoro-C-mannoside **3.42** (1.70 g, 2.18 mmol) was treated with a mixture of 3N NaOH (2.2 mL, 6.54 mmol) and ethanol (40 mL). After 1 hour the reaction mixture concentrated *in vacuo* and acidified with 2N HCl (20 mL). The mixture was then

extracted with ethyl acetate (3 × 20 mL) and the organic phase washed with water and dried (Na₂SO₄). The solvent was removed under reduced pressure, to provide the acid **3.22** (1.66 g, quantitative yield): colorless oil, R_f = 0.34 (20% MeOH/CHCl₃); ¹H NMR (500 MHz, CDCl₃) δ 1.10 (s, 9H), 3.87-4.05 (m, 5H), 4.14 (dd, J = 2.9, 6.2 Hz, 1H), 4.48-4.66 (m, 7H), 7.24-7.48 (m, 21H), 7.71 (m, 4H), 9.49 (br s, D₂O exchange, 1H); ¹³C NMR (125 MHz, CDCl₃) δ 27.0, 27.2, 62.6, 72.1 (t, J = 24.1 Hz), 72.37, 72.43, 72.9, 73.1, 73.8, 76.2, 77.5, 114.6 (t, J = 257.0 Hz), 127.7-129.9 (several lines), 133.4, 133.6, 135.8, 135.9, 137.6, 138.0, 138.2, 166.4 (t, J = 31.7 Hz); ESIHRMS calcd for C₄₅H₅₂O₇SiF₂N [M + NH₄]⁺ 784.3476 found 784.3464.

Mono-fluoro-thioacetal esters 3.17/3.18.

TIA alcohol **2.13** (2.27 g, 4.46 mmol) and acid mixture **3.20/3.21** (2.69 g, 4.51 mmol) was subjected to the same procedure used to synthesize compound **2.30**. An inseparable mixture of fluoro esters **3.17/3.18** (2.63 g, 68% based on recovered alcohol) was obtained: colorless oil; R_f = 0.20 (10% EtOAc/petroleum ether); ¹H NMR (500 MHz, CDCl₃) δ 1.03 (s, 9H), 1.34, 1.39, 1.43, 1.46 (all s, 6H), 3.40 (dd, J = 5.0, 9.0 Hz, 0.4H), 3.60 (m, 1H), 3.70 (m, 1H), 3.72-3.88 (m, 4H), 3.99 (m, 1.6H), 4.27 (m, 2H), 4.35-4.55 (m, 7H), 5.18 (dm, J = 47.2 Hz, 1H), 5.20-5.45 (m, 2H), 7.11(m, 1H), 7.20-7.40 (m, 29H), 7.53 (m, 1H), 7.66 (m, 4H); ¹³C NMR (125 MHz, CDCl₃) δ 26.4, 26.6, 26.9, 27.0, 27.3, 27.4, 61.5, 62.3, 68.3, 68.8, 70.9, 71.1, 71.5, 71.5, 71.9, 72.0, 72.4 (d, J = 7.1 Hz), 72.5, 72.6, 72.7, 73.1, 73.3, 74.1 (d, J = 23.6 Hz), 74.8, 75.5, 78.6, 78.8, 84.4, 85.1, 88.1 (d, J = 194.3 Hz), 89.2 (d, J = 193.4 Hz), 111.7, 111.9, 127.6-129.2 (several lines), 130.0, 132.0, 132.4, 132.5, 135.7, 135.8, 135.9, 138.0 (×2), 138.2, 138.3, 138.4, 138.7, 166.9 (d,

$J = 24.8$ Hz), 167.5 (d, $J = 25.7$ Hz); ESIHRMS calcd for $C_{65}H_{75}O_{10}NFSiS$ ($M + NH_4$) 1108.4865, found 1108.4918.

Difluoro thioacetal ester 3.19.

Procedure 1: TIA alcohol **2.13** (85.5 mg, 0.17 mmol) and difluoro acid **3.22** (140.0 mg, 0.18 mmol) was subjected to the same procedure used to synthesize compound **2.30**. This provided ester **3.19** (83.2 mg, 89% based on recovered **2.13** and 34% conversion based on **3.22**).

Procedure 2: A mixture of difluoro acid **3.22** (43.0 mg, 0.06 mmol), 2,4,6-trichlorobenzoyl chloride (0.01 mL, 0.06 mmol) and triethylamine (0.02 mL, 0.12 mmol) in THF (3 mL) was stirred for 3.5 hours at $0^\circ C$. TIA alcohol **2.13** (30.7 mg, 0.06 mmol) and DMAP (10.0 mg, 0.08 mmol) in toluene were then added, and the mixture stirred for another 1 hour. The mixture was then diluted with diethyl ether (10 mL) and washed with saturated aqueous $NaHCO_3$ solution, brine, dried (Na_2SO_4), filtered, and evaporated under reduced pressure. The residue was purified by FCC to give ester **3.19** (53.9 mg, 91% based on recovered **2.13** and 76% conversion based on **3.22**): colorless oil; $R_f = 0.33$ (10% EtOAc/petroleum ether); 1H NMR (500 MHz, $CDCl_3$) δ 1.04 (two singlets, 18H), 1.31, 1.43, (both s, 6H), 3.77-3.89 (m, 5H), 3.91 (q, $J = 5.1$ Hz, 1H), 3.99 (t, $J = 5.9$ Hz, 1H), 4.07 (dd, $J = 3.0, 5.6$ Hz, 1H), 4.39-4.60 (m, 8H), 5.22 (m, 1H), 5.42 (d, $J = 6.6$ Hz, 1H), 7.16-7.42 (m, 30H), 7.52 (m, 2H), 7.65 (m, 8H); ^{13}C NMR (125 MHz, $CDCl_3$) δ 26.7, 26.9, 27.0, 27.2, 61.6, 62.8, 72.5, 72.6, 72.7, 72.9, 73.2, 74.0, 74.5, 77.1, 77.6, 78.7, 84.7, 112.1, 114.9 (t, $J = 257.4$ Hz), 127.8-129.8 (several lines), 130.0, 132.6, 132.9, 133.0, 133.4, 133.5, 133.8, 135.8 (three signals), 135.9, 136.0, 137.8, 138.4, 138.5, 162.4

(t, $J = 31.8$ Hz); ESIHRMS calcd for $C_{74}H_{86}O_{10}SSi_2F_2N$ $[M + NH_4]^+$ 1274.5474, found 1274.5476.

Mono-fluoro thioacetal enol ethers 3.14/3.15.

Treatment of ester **3.17/3.18** (1.37 g, 1.26 mmol) under similar conditions used to prepare **2.32** afforded the fluoro enol ether mixture **3.14/3.15** (1.06 g, 77%): colorless oil, $R_f = 0.58$ (15% EtOAc/petroleum ether); 1H NMR (500 MHz, C_6D_6) δ 1.14, 1.15 (both s, 9H), 1.39, 1.47, 1.49, 1.51 (all s, 6H), 3.57 (dd, $J = 5.5, 10.8$ Hz, 0.6H), 3.65 (m, 1H), 3.81 (ddd, $J = 2.8, 6.2, 8.9$ Hz, 0.6H), 3.88-3.91 (m, 1.6H), 4.02 (dd, $J = 5.1, 10.3$ Hz, 0.4H), 4.05-4.10 (m, 5H), 4.33-4.65 (m, 9.2H), 4.71 (dd, $J = 2.4, 6.7$ Hz, 0.6H), 4.77 (dd, $J = 2.2, 6.2$ Hz, 0.6H), 4.80-4.86 (m, 1H), 4.89 (d, $J = 12.0$ Hz, 0.6H), 4.94 (d, $J = 7.8$ Hz, 0.4H), 5.11 (dd, $J = 2.6, 46.4$ Hz, 0.4H), 5.95 (d, $J = 6.8$ Hz, 0.4H), 6.05 (d, $J = 6.2$ Hz, 0.6H), 6.92-7.42 (m, 29H), 7.68-7.76 (m, 6H); ^{13}C NMR (C_6D_6) δ 26.9, 27.2, 27.4 (2 peaks), 27.6, 27.9, 62.0, 70.4, 70.8, 72.0, 72.3, 72.5, 72.8, 73.5 (d, $J = 9.3$ Hz), 73.6 (d, $J = 16.7$ Hz), 73.9, 74.6, 75.4, 75.5, 75.6, 76.5, 76.7, 80.3, 80.5, 80.8, 84.8, 85.3, 88.5 (d, $J = 6.1$ Hz), 90.5 (d, $J = 183.5$ Hz), 91.6 (d, $J = 178.4$ Hz), 127.6-129.5 (several lines), 130.5, 130.6, 132.2, 132.9, 133.6, 133.8, 134.8, 135.2, 136.4, 136.5, 139.3, 139.4, 139.5, 139.8, 139.9, 156.4 (d, $J = 17.9$ Hz), 156.9 (d, $J = 24.7$ Hz); ESIHRMS calcd for $C_{66}H_{77}O_9NFSiS$ (M + NH_4) 1106.5072, found 1106.5103.

Difluoro thioacetal enol ethers 3.16.

A solution of titanium tetrachloride (4.98 mL, 4.98 mmol, 1M) was added to THF (6 mL) at 0 °C. The mixture was stirred for 30 min at which point TMEDA (1.2 mL,

7.96 mmol) was added in one portion. The resulting yellow-brown suspension was allowed to warm to rt and stirred for an additional 30 min. At this time freshly activated zinc dust (546.0 mg, 8.36 mmol) and lead (II) chloride (27.7 mg, 0.10 mmol) were added in portion and stirred at rt for a further 10 min. To the resulting bluish-green mixture was added a solution of ester **3.19** (257.0 mg, 0.20 mmol) and dibromomethane (0.28 mL, 3.98 mmol) in THF (6 mL) in one portion. The mixture was stirred at 60 °C for 3.5 hours and then diluted with brine. The mixture was stirred for 30 min at rt, diluted with ether (20 mL) and stirred vigorously for an additional 20 min. The resulting suspension was filtered through neutral alumina and the residue washed with ether. The ethereal extract was concentrated *in vacuo* and gravity chromatography of the residue over silica gel afforded enol ether **3.16** (100.9 mg, 63%, based on recovered **3.19**): colorless oil, $R_f = 0.53$ (5% EtOAc/petroleum ether); ^1H NMR (500 MHz, C_6D_6) δ 1.15, 1.17 (both s, 18H), 1.47, 1.55, (both s, 6H), 3.81 (dd, $J = 5.9, 11.0$ Hz, 1H), 4.01-4.15 (m, 5H), 4.18 (dd, $J = 7.3, 11.0$ Hz, 1H), 4.31 (apparent t, $J = 2.5$ Hz, 1H), 4.35 (t, $J = 3.7$ Hz, 1H), 4.43-4.49 (m, 4H), 4.62 (A of ABq, $J = 11.9$ Hz, $\Delta\delta = 0.14$ ppm, 1H), 4.69 (B of ABq, $J = 11.9$ Hz, $\Delta\delta = 0.14$ ppm, 1H), 4.75 (d, $J = 3.7$ Hz, 1H), 4.79 (dd, $J = 2.7, 5.6$ Hz, 1H), 4.87 (d, $J = 11.3$ Hz, 1H), 5.08 (ddd, $J = 1.7, 7.6, 26.7$ Hz, 1H), 6.89-7.43 (m, 30H), 7.65-7.88 (m, 10H); ^{13}C NMR (125 MHz, C_6D_6) δ 27.3, 27.4, 27.6, 27.9, 62.1, 65.2, 72.6, 73.0, 73.5, 74.7 (dd, $J = 22.0, 32.1$ Hz), 75.0, 75.2, 78.9 (d, $J = 4.6$ Hz), 81.0, 84.9, 88.3 (t, $J = 4.6$ Hz), 113.1, 119.8 (dd, $J = 248.4, 253.9$ Hz), 127.3-129.5 (several lines), 130.2 (two signals), 130.6 (two signals), 131.6, 133.6, 133.8, 134.5, 134.6, 135.5, 136.4, 136.5, 139.1, 139.5, 139.7, 154.3 (dd, $J = 23.8, 31.2$ Hz); ESIHRMS calcd for $\text{C}_{75}\text{H}_{88}\text{O}_9\text{SSi}_2\text{F}_2\text{N}$ $[\text{M} + \text{NH}_4]^+$ 1272.5681, found 1272.5671.

Mono-fluoro glycals 3.11/3.12.

Treatment of the fluoro enol ether mixture **3.14/3/15** (1.06 g, 0.98 mmol) under similar conditions used to prepare **2.34** afforded the mono-fluoro glycal mixture **3.10/3.11** (0.68 g, 83% based on recovered **3.14/3/15**): clear oil; $R_f = 0.39$ (15% EtOAc/petroleum ether); $^1\text{H NMR}$ (500 MHz, C_6D_6) δ 1.17 (s, 9H), 1.32, 1.36, 1.47, 1.60 (all s, 6H), 3.76 (dd, $J = 2.9, 10.7$ Hz, 0.5H), 3.85 (m, 1.5H), 3.93 (m, 1H), 4.01 (m, 1H), 4.04-4.14 (m, 3H), 4.18 (dd, $J = 2.8, 7.6$ Hz, 0.5H), 4.24 (m, 1.5H), 4.30 (m, 1H), 4.38-4.62 (m, 9.5H), 4.84 (d, $J = 11.5$ Hz, 0.5H), 4.97 (m, 0.5H), 5.03 (dd, $J = 5.3, 46.6$ Hz, 0.5H), 5.18 (dd, $J = 2.0, 46.1$ Hz, 0.5H), 5.38 (m, 0.5H), 7.03-7.34 (m, 26H), 7.78 (m, 4H); $^{13}\text{C NMR}$ (125 MHz, C_6D_6) δ 27.4 ($\times 2$), 27.8, 28.9, 63.8, 64.1, 69.6, 69.8, 70.3, 71.3 (d, $J = 19.0$ Hz), 72.2, 72.4, 72.7, 72.9, 73.0, 73.9 ($\times 2$), 74.0 (d, $J = 4.0$ Hz), 74.6, 74.7, 74.8, 88.6 (d, $J = 182.0$ Hz), 91.6 (d, $J = 180.0$ Hz), 100.8 (d, $J = 8.8$ Hz), 102.2 (d, $J = 8.8$ Hz), 111.0, 111.1, 127.9-128.9 (several lines), 133.9, 134.0, 134.1 ($\times 2$), 136.3, 136.4, 139.2, 139.3, 139.4, 139.5 ($\times 2$), 139.6, 139.7, 139.8, 150.7 (d, $J = 25.6$ Hz), 150.9 (d, $J = 22.0$ Hz); ESIHRMS calcd for $\text{C}_{60}\text{H}_{68}\text{O}_9\text{FSi}$ (M + H) 979.4617, found 979.4645.

Difluoro glycal 3.13.

The enol ether **3.16** (0.20 g, 0.16 mmol) was subjected to similar conditions used to prepare **2.34**. This provided the difluoro glycal **3.13** (0.12 g, 82% based on recovered **3.16**): clear oil, $R_f = 0.54$ (10% EtOAc/petroleum ether); $^1\text{H NMR}$ (500 MHz, C_6D_6) δ 1.16, 1.18 (both s, 18H), 1.27, 1.28, (both s, 6H), 3.86 (t, $J = 6.9$ Hz, 1H), 4.03-4.17 (m, 7H), 4.30 (t, $J = 2.9$ Hz, 1H), 4.38 (m, 1H), 4.50-4.60 (m, 5H), 4.72 (A of ABq, $J = 11.5$ Hz, $\Delta\delta = 0.23$ ppm, 1H), 4.79 (apparent ddd, $J = 2.9, 10.0, 22.7$ Hz, 1H), 4.98 (B of ABq,

$J = 11.5$ Hz, $\Delta\delta = 0.23$ ppm, 1H), 5.41 (d, $J = 2.7$ Hz, 1H), 7.08-7.38 (m, 27H), 7.70-7.87 (m, 8H); ^{13}C NMR (125 MHz, C_6D_6) δ 27.4, 27.5, 28.8, 63.7, 63.8, 69.3, 72.2, 73.0, 73.3, 74.0, 74.9, 75.2 (dd, $J = 24.7, 30.2$ Hz), 77.5, 78.5, 80.3 (d, $J = 2.7$ Hz), 102.7 (t, $J = 5.5$ Hz), 111.3, 118.8 (dd, $J = 247.4, 252.0$ Hz), 127.9-128.9 (several lines), 133.9, 134.1 (two signals), 136.2, 136.3, 136.4, 136.7, 139.2, 139.5, 139.9, 148.5 (dd, $J = 26.6, 31.2$ Hz); ESIHRMS calcd for $\text{C}_{69}\text{H}_{82}\text{O}_9\text{Si}_2\text{F}_2\text{N}$ [$\text{M} + \text{NH}_4$] $^+$ 1162.5491, found 1162.5497.

1,3,4,5-Tetra-*O*-benzyl-13-*O*-*tert*-butyldiphenylsilyl-2,6:8,12-dianhydro-10,11-*O*-isopropylidene-7-fluoro-*D*-lyxo-*D*-galacto-*D*-manno-tridecitol 3.8 and 1,3,4,5-Tetra-*O*-benzyl-13-*O*-*tert*-butyldiphenylsilyl-2,6:8,12-dianhydro-10,11-*O*-isopropylidene-7-fluoro-*D*-lyxo-*D*-gulo-*D*-manno-tridecitol 3.9.

Application of the procedure that was used for **2.36** to the glycal mixture **3.11/3.12** (0.91 g, 0.94 mmol) provided the mono-fluoro- β -C-glycosides **3.8** (0.36 g, 39%) and **3.9** (0.23 g, 24%).

For C-glycoside 3.8: colorless oil; $R_f = 0.63$ (30% EtOAc/petroleum ether); ^1H NMR (CDCl_3) δ 1.07 (s, 9H, $\text{CH}_3)_3\text{CSi}$), 1.36, 1.49 (both s, 6H, $\text{C}(\text{CH}_3)_2$), 3.34 (d, $J = 6.9$ Hz, 1H, D_2O exchange, OH), 3.40 (dd, $J = 3.5, 10.6$ Hz, 1H, H-6a), 3.47 (ddd, $J = 6.1, 9.2, 14.0$ Hz, 1H, H-1'), 3.53 (t, $J = 4.4$ Hz, 1H, H-4), 3.69-3.79 (m, 3H, H-2', 3, 4), 3.84-3.95 (m, 5H, H-2, 3', 5' 6'a, 6'b), 4.09 (m, 1H, H-5), 4.30 (dd, $J = 2.2, 5.6$ Hz, 1H, H-4'), 4.40 (ddd, $J = 2.7, 8.1, 27.1$ Hz, 1H, H-1), 4.31-4.61 (m, 8H, 4 x PhCH_2), 4.84 (ddd, $J = 2.5, 6.1, 46.7$ Hz, 1H, H-1''), 7.05-7.47 (m, 26H, Ph), 7.70 (m, 4H, Ph); ^{13}C NMR (CDCl_3) δ 26.4, 26.9, 28.3, 62.8, 68.1, 69.6 (d, $J = 17.8$ Hz), 69.9 (d, $J = 3.7$ Hz), 72.3, 73.1, 73.2, 73.4, 75.0, 75.1, 76.9, 77.7, 79.9, 91.1 (d, $J = 181.2$ Hz), 109.6, 127.8-129.8 (several

lines), 133.6 (2 peaks), 135.9 (2 peaks), 137.8, 138.0, 138.1, 138.3; FABHRMS calcd for $C_{60}H_{70}O_{10}FSi$ (M + H) 997.4722, found 997.5031.

For C-glycoside 3.9: colorless oil; $R_f = 0.75$ (30% EtOAc/petroleum ether); 1H NMR ($CDCl_3$) δ 1.07 (s, 9H, $(CH_3)_3CSi$), 1.38, 1.52, (both s, 6H, $C(CH_3)_2$), 3.43 (dd, $J = 8.6, 9.8$ Hz, 1H, H-6a), 3.51 (ddd, $J = 2.5, 10.0, 16.0$ Hz, 1H, H-1'), 3.66 (m, 2H, H-4, 6b), 3.72 (dd, $J = 2.6, 8.6$ Hz, 1H, H-3), 3.82 (dt, $J = 2.0, 8.8$ Hz, 1H, H-5), 3.84 - 4.00 (m, 5H, H-2', 3', 5, 6'a, 6'b), 4.02 (q, $J = 2.8$ Hz, 1H, H-2), 4.05 (d, $J = 5.9$ Hz, D_2O exchange, 1H, OH), 4.34 (dd, $J = 1.7, 4.9$ Hz, 1H, H-4'), 4.41-4.64 (m, 8H, H-1, 7 x $PhCH$), 4.81 (B of ABq, $J = 11.0$ Hz, $\Delta\delta = 0.38$ ppm, 1H, $PhCH$), 5.02 (ddd, $J = 2.5, 9.1, 44.8$ Hz, 1H, H-1'), 7.11-7.47 (m, 26H, Ph), 7.72 (m, 4H, Ph); ^{13}C NMR ($CDCl_3$) δ 26.6, 26.9, 28.6, 62.7, 69.3 (d, $J = 29.1$ Hz), 69.9, 71.2 (d, $J = 23.5$ Hz), 72.1, 72.8, 73.2 (d, $J = 3.9$ Hz), 73.7, 74.6, 74.7, 74.9, 76.9, 78.2 (d, $J = 18.7$ Hz), 79.8, 80.0, 86.6 (d, $J = 178.6$ Hz), 109.4, 127.8-129.9 (several lines), 133.6, 133.7, 135.8, 135.9, 137.3, 138.1 (2 peaks), 138.2; FABHRMS calcd for $C_{60}H_{70}O_{10}FSi$ (M + H) 997.4722, found 997.4759.

3,4,5-Tri-O-benzyl-1,13-O-di-tert-butyl-diphenylsilyl-2,6:8,12-dianhydro-10,11-O-isopropylidene-7-deoxy-7,7-difluoro-D-threo-L-gulo-D-manno-tridecitol 3.10.

Application of the procedure that was used for **2.36** to the glycal **3.13** (0.12 g, 0.11 mmol) provided the difluoro- β -C-galactoside **3.10** (0.11 g, 86%) as colorless oil, colorless oil; $R_f = 0.48$ (20% EtOAc/petroleum ether); 1H NMR ($CDCl_3$) δ 1.07 (s, 18H, 2 x $(CH_3)_3CSi$), 1.35, 1.47, (both s, 6H, $C(CH_3)_2$), 2.93 (d, $J = 5.3$ Hz, 1H, D_2O exchange, -OH), 3.54 (m, 1H, H-1'), 3.68 (t, $J = 7.6$ Hz, 1H, H-4), 3.79 (ddd, $J = 2.2, 6.1, 7.8$ Hz, 1H, H-5'), 3.80-3.97 (m, 8H, H-2', 3', 6'a, 6'b, 3, 5, 6a, 6b), 4.09 (t, $J = 3.4$ Hz, 1H, H-

2), 4.27 (dd, $J = 2.5, 5.1$ Hz, 1H, H-4'), 4.36 (d, $J = 11.3$ Hz, 1H, PhCH), 4.50-4.62 (m, 6H, H-1, 5 x PhCH), 7.06 (m, 2H, Ph), 7.22-7.42 (m, 25H, Ph), 7.65-7.73 (m, 8H, Ph); ^{13}C NMR (CDCl_3) δ 26.5, 27.0, 27.3, 28.5, 62.7, 64.3, 69.7, 72.8 (two signals), 72.9, 73.4, 74.0, 74.5, 76.6 (apparent t, $J = 31.9$ Hz), 78.2, 78.7, 80.2, 109.9, 121.2 (t, $J = 254.2$ Hz), 127.8-130.0 (several lines), 133.2, 133.5, 133.7, 135.8, 135.9, 136.0, 138.2, 138.3, 138.5; ESIHRMS calcd for $\text{C}_{69}\text{H}_{84}\text{O}_{10}\text{Si}_2\text{F}_2\text{N}$ [$\text{M} + \text{NH}_4$] $^+$ 1180.5596, found 1180.5601

1,3,4,5-Tetra-*O*-benzyl-9-*O*-acetyl-13-*O*-*tert*-butyldiphenylsilyl-2,6:8,12-dianhydro-10,11-*O*-isopropylidene-7-fluoro-*D*-lyxo-*D*-galacto-*D*-manno-tridecitol 3.48 and 1,3,4,5-Tetra-*O*-benzyl-9-*O*-acetyl-13-*O*-*tert*-butyldiphenylsilyl-2,6:8,12-dianhydro-10,11-*O*-isopropylidene-7-fluoro-*D*-lyxo-*D*-gulo-*D*-manno-tridecitol 3.49.

Fluoro- β -C-galactoside acetate **3.48**: colorless oil; $R_f = 0.29$ (20% EtOAc/petroleum ether); ^1H NMR (500 MHz, CDCl_3) δ 1.07 (s, 9H), 1.37, 1.54, (both s, 3H ea), 2.02 (s, 3H), 3.22 (ddd, $J = 3.8, 7.8, 20.9$ Hz, 1H), 3.63 (m, 3H), 3.72 (m, 2H), 3.78 (dd, $J = 5.6, 14.9$ Hz, 1H), 3.88 (q, $J = 5.0$ Hz, 1H), 3.95 (m, 2H), 4.07 (t, $J = 6.2$ Hz, 1H), 4.18 (dt, $J = 5.3, 23.0$ Hz, 1H), 4.35 (dd, $J = 1.9, 5.5$ Hz, 1H), 4.40-4.64 (m, 9H), 5.14 (dd, $J = 7.1, 8.4$ Hz, 1H), 7.12-7.48 (m, 26H), 7.75 (m, 4H); ^{13}C NMR (125 MHz, CDCl_3) δ 21.3, 26.4, 27.0, 27.5, 62.3, 68.9, 69.1 (d, $J = 5.5$ Hz), 71.9, 73.0, 73.1, 73.4, 73.8, 74.2 (d, $J = 6.0$ Hz), 74.7, 75.1 (d, $J = 19.6$ Hz), 76.0, 76.6, 87.4 (d, $J = 186.1$ Hz), 110.4, 127.6-129.9 (several lines), 133.2, 133.5, 135.7, 135.8, 169.4; ESMS 1056.3 ($\text{M} + \text{NH}_4$)

Fluoro- β -C-galactoside acetate **3.49**: colorless oil; $R_f = 0.29$ (20% EtOAc/petroleum ether); ^1H NMR (500 MHz, CDCl_3) δ 1.03 (s, 9H), 1.34, 1.51, (both s,

3H ea), 2.02 (s, 3H), 3.60 (m, 1H), 3.67 (dd, $J = 3.2, 7.1$ Hz, 1H), 3.69-3.87 (m, 7H), 3.94 (m, 1H), 4.13 (dd, $J = 5.1, 10.9$ Hz, 1H), 4.25 (dt, $J = 4.2, 23.4$ Hz, 1H), 4.31 (dd, 1.5, 5.6 Hz, 1H), 4.41-4.68 (m, 8H), 4.77 (dt, $J = 47.5, 4.9$ Hz, 1H), 5.16 (dd, $J = 9.9, 8.2$ Hz, 1H), 7.14-7.40 (m, 26H), 7.68 (m, 4H); ^{13}C NMR (125 MHz, CDCl_3) δ 21.3, 26.2, 26.9, 27.6, 69.4, 70.2 (d, $J = 4.4$ Hz), 71.9, 72.4, 73.0, 73.5, 74.1, 74.7, 75.9 (d, $J = 10.6$ Hz), 76.5, 93.3 (d, $J = 180.1$ Hz), 110.2, 127.7-129.9 (several lines), 130.0, 133.3, 133.5, 135.7, 135.9, 138.3, 138.4, 138.5 ($\times 2$), 138.6, 169.9; ESMS 1056.3 ($\text{M} + \text{NH}_4$)

3,4,5-Tri-*O*-benzyl-9-*O*-acetyl-1,13-*O*-di-*tert*-butyldiphenylsilyl-2,6:8,12-dianhydro-10,11-*O*-isopropylidene-7-deoxy-7,7-difluoro-*D*-threo-*L*-gulo-*D*-manno-tridecitol

3.50:

Difluoro- β -C-galactoside Acetate **3.50**: colorless oil; $R_f = 0.29$ (10% EtOAc/petroleum ether), ^1H NMR (500 MHz, CDCl_3) δ 1.02, 1.04 (both s, 18H), 1.36, 1.52, (both s, 6H), 1.99 (s, 3H), 3.65 (m, 2H), 3.78-3.92 (m, 7H), 3.98 (t, $J = 3.4$ Hz, 1H), 4.02 (t, $J = 7.1$ Hz, 1H), 4.10 (dd, $J = 5.6, 7.1$ Hz, 1H), 4.24 (m, 1H), 4.31 (dd, $J = 2.0, 5.4$ Hz, 1H), 4.47-4.73 (m, 6H), 5.26 (dd, $J = 7.3, 9.5$ Hz, 1H), 7.18-7.41 (m, 27H), 7.66 (m, 8H); ESIHRMS calcd for $\text{C}_{71}\text{H}_{86}\text{O}_{11}\text{Si}_2\text{F}_2\text{N}$ [$\text{M} + \text{NH}_4$] $^+$ 1222.5702, found 1222.5696

1,3,4,5-Tetra-*O*-benzyl-13-*O*-*tert*-butyldiphenylsilyl-2,6:8,12-dianhydro-7-fluoro-*D*-lyxo-*D*-galacto-*D*-manno-tridecitol 3.51 and 1,3,4,5-Tetra-*O*-benzyl-2,6:8,12-dianhydro-7-fluoro-*D*-lyxo-*D*-galacto-*D*-manno-tridecitol 3.53.

Application of the procedure that was used for **2.43** to **3.8** (407 mg, 0.41 mmol) provided the corresponding triol **3.51** (101 mg, 26%) and the tetrol **3.53** (78 mg, 33%).

The above tetrol **3.53** (78.0 mg, 0.11 mmol), TBDPSCl (0.04 mL, 0.14 mmol), and imidazole (15 mg, 0.22 mmol) in anhydrous DMF (3 mL) was stirred at 50 °C for 1.5 hours. The reaction mixture was then diluted with water (2 mL) and extracted with ether. The combined organic phase was washed with brine, dried (Na₂SO₄), filtered, and evaporated under reduced pressure. The residue was purified by FCC to give the triol **3.51** (70.0 mg, 67% yield).

For triol 3.51: colorless oil; R_f = 0.37 (40% EtOAc/petroleum ether); ¹H NMR (500 MHz, CDCl₃) δ 1.05 (s, 9H), 2.59 (d, J = 4.1 Hz, 1H, D₂O exchange), 2.67 (d, J = 3.0 Hz, 1H, D₂O exchange), 3.33 (dt, J = 3.7, 9.2 Hz, 1H), 3.41-3.47 (m, 3H), 3.50 (t, J = 4.8 Hz, 1H), 3.58 (d, J = 6.4 Hz, 1H), 3.78 (apparent t, J = 4.2 Hz, 1H), 3.79-3.96 (m, 5H), 4.13 (m, 2H), 4.35 (apparent ddd, J = 1.8, 7.8, 26.1 Hz, 1H), 4.38-4.64 (m, 8H), 4.84 (ddd, 2.0, 4.8, 45.1 Hz, 1H), 7.19 (m, 2H), 7.23-7.41 (m, 24H), 7.67-7.72 (m, 4H); ¹³C NMR (125 MHz, CDCl₃) δ 27.0, 63.5, 68.6, 69.0, 69.9 (d, J = 17.8 Hz), 72.3, 72.6, 73.2, 73.3, 73.5 (d, J = 3.0 Hz), 75.0, 75.1, 75.5, 75.6, 77.8, 78.0, 78.5, 91.2 (d, J = 181.5 Hz), 127.9-130.0 (several lines), 133.1, 133.4, 135.8, 135.9, 137.8, 138.0 (×2), 138.3; ESMS 979.3 (M + Na).

For tetrol 3.53: colorless oil: R_f = 0.13 (40% EtOAc/petroleum ether); ¹H NMR (500 MHz, CDCl₃) δ 2.41 (dd, J = 3.8, 8.0 Hz, 1H, D₂O exchange), 2.75, 2.80 (both br s, 1H ea, D₂O exchange), 3.28 (dt, J = 3.4, 9.3 Hz, 1H), 3.43 (m, 1H), 3.47 (t, J = 4.6 Hz, 1H), 3.50 (ddd, J = 4.8, 9.7, 14.6 Hz, 1H), 3.65 (d, J = 6.8 Hz, 1H), 3.72 (ddd, J = 4.1, 8.5, 12.0 Hz, 1H), 3.82-3.94 (m, 5H), 3.97 (dd, J = 2.6, 7.9 Hz, 1H), 4.16 (m, 1H), 4.39 (ddd, J = 1.9, 8.0, 29.2 Hz, 1H), 4.34-4.64 (m, 9H), 4.89 (ddd, J = 1.9, 4.8, 45.2 Hz, 1H), 7.25 (m, 2H), 7.26-7.38 (m, 18H); ¹³C NMR (125 MHz, CDCl₃) δ 63.4, 63.5, 68.4, 68.8, 69.7,

69.9 (d, $J = 17.8$ Hz), 72.4, 72.5, 73.2, 73.4 (d, $J = 3.2$ Hz), 74.9, 75.1, 75.3, 75.4, 77.9, 78.0, 78.6, 91.2 ($J = 181.5$ Hz), 128.0-128.7 (several lines), 137.7, 137.9, 138.0, 138.2.

1,3,4,5-Tetra-*O*-benzyl-13-*O*-*tert*-butyldiphenylsilyl-2,6:8,12-dianhydro-7-fluoro-D-lyxo-D-gulo-D-manno-tridecitol 3.52.

The fluoro disaccharide **3.9** (280 mg, 0.28 mmol) was subjected to similar conditions as used in the synthesis of **3.51**. This gave the triol **3.52** (152 mg, 73 % based on recovered **3.9**): colorless oil; $R_f = 0.22$ (40% EtOAc/petroleum ether); ^1H NMR (500 MHz, CDCl_3) δ 1.09 (s, 9H), 2.27 (d, $J = 4.1$ Hz, 1H, D_2O exchange), 2.67 (d, $J = 3.0$ Hz, 1H, D_2O exchange), 3.39 (dt, $J = 3.7, 9.2$ Hz, 1H), 3.52-3.59 (m, 2H), 3.61 (apparent t, $J = 8.8$ Hz, 1H), 3.75-3.84 (m, 5H), 3.97 (apparent dd, $J = 6.9, 10.0$ Hz, 1H), 4.01 (m, 1H), 4.09 (m, 2H), 4.15 (t, $J = 2.8$ Hz, 1H), 4.49-4.91 (m, 9H), 5.04 (ddd, 2.0, 9.1, 44.4 Hz, 1H), 7.24-7.46 (m, 26H), 7.76 (m, 4H); ^{13}C NMR (125 MHz, CDCl_3) δ 26.9, 63.0, 67.3 (d, $J = 7.6$ Hz), 68.9, 70.6, 71.1 (d, $J = 23.5$ Hz), 72.0, 72.5, 72.8, 73.4 (d, $J = 3.9$ Hz), 73.9, 74.4, 74.8, 75.0, 75.2, 78.1, 79.2, 79.4, 79.8, 86.6 ($J = 179.6$ Hz), 127.7-130.0 (several lines), 133.2, 133.4, 135.7, 135.8, 137.2, 138.2 ($\times 4$); ESMS 979.3 ($\text{M} + \text{Na}$)

1,3,4,5-Tetra-*O*-benzyl-13-2,6:8,12-dianhydro-7-fluoro-D-lyxo-D-gulo-D-manno-tridecitol 3.54.

Compound **3.9** (62 mg, 0.06 mmol) was again subjected to similar acid hydrolysis conditions as used in synthesis of **3.51**. Thus after a much longer time period provided the fluoro tetrol derivative **3.54** (21 mg, 47%), colorless oil: $R_f = 0.39$ (30% MeOH/ CHCl_3); ^1H NMR (500 MHz, CDCl_3) δ 2.45 (br s, 2H, D_2O exchange), 2.89 (br s,

1H, D₂O exchange), 3.36 (m, 1H), 3.45 (m, 1H), 3.56 (m, 1H), 3.62 (t, J = 9.0 Hz, 1H), 3.74-3.96 (m, 7H), 4.10 (m, 2H), 4.47-4.74 (m, 9H), 5.01 (ddd, J = 2.0, 8.8, 44.5 Hz, 1H), 7.25-7.42 (m, 20H); ¹³C NMR (125 MHz, CDCl₃) δ 62.8, 67.3 (d, J = 7.3 Hz), 69.6, 70.5, 71.3, 72.1, 72.7, 73.2 (d, J = 3.7 Hz), 73.9, 74.4, 74.8, 74.8 (d, J = 8.4 Hz), 78.2, 79.0 (d, J = 18.8 Hz), 79.5, 86.9 (d, J = 178.2 Hz), 127.9-128.8 (several lines), 137.2, 138.1 (×2), 138.2.

3,4,5-Tri-*O*-benzyl-1,13-di-*O*-*tert*-butyldiphenylsilyl-2,6:8,12-dianhydro-7-deoxy-7,7-difluoro-*D*-*threo*-*L*-*gulo*-*D*-*manno*-tridecitol 3.55 and **3,4,5-Tri-*O*-benzyl-2,6:8,12-dianhydro-7-deoxy-7,7-difluoro-*D*-*threo*-*L*-*gulo*-*D*-*manno*-tridecitol 3.56**.

Application of the procedure that was used for **2.43** to C-glycoside **3.10** (80.0 mg, 0.07 mmol) provided the triol **3.55** (30 mg, 46%) and the desilylated product **3.56** (9.3 mg, 25% yield) based on recovered **3.10**, both as colorless oils;

For triol 3.54: R_f = 0.51 (40% EtOAc/petroleum ether); ¹H NMR (500 MHz, CDCl₃) δ 1.07, 1.08 (both s, 18H), 2.78 (broad s, D₂O exchange, 3H), 3.42 (dd, J = 3.2, 9.5 Hz, 1H), 3.49 (t, J = 5.9 Hz, 1H), 3.60 (q, J = 9.5 Hz, 1H), 3.68 (t, J = 7.6 Hz, 1H), 3.83-3.92 (m, 4H), 3.95 (dd, J = 5.9, 10.8 Hz, 1H), 4.03 (m, 1H), 4.09 (t, J = 9.3 Hz, 1H), 4.11 (d, J = 2.9 Hz, 1H), 4.14 (t, J = 3.1 Hz, 1H), 4.40 (d, J = 11.3 Hz, 1H), 4.54-4.70 (m, 6H), 7.10 (m, 2H), 7.20-7.47 (m, 25H), 7.63-7.77 (m, 8H); ¹³C NMR (125 MHz, CDCl₃) δ 27.0, 27.2, 63.3, 64.7, 67.8, 68.9, 72.8, 72.9, 72.8, 72.9, 74.1, 74.5, 75.4, 77.4, 78.3 (d, J = 7.4 Hz), 78.5, 78.7, 121.1 (t, J = 254.8 Hz), 127.8-128.6 (several lines), 130.0 (two signals), 130.1, 133.1, 133.2, 133.4, 133.5, 135.7, 135.8 (two signals), 135.9 (two signals), 138.2,

(two signals), 138.4; ESIHRMS calcd for $C_{66}H_{76}O_{10}Si_2F_2Na$ $[M + Na]^+$ 1145.4837, found 1145.4838.

For pentol 3.56: $R_f = 0.28$ (95% EtOAc/petroleum ether); 1H NMR (500 MHz, $CDCl_3$) δ 2.24 (br s, D_2O exchange, 2H), 3.41 (s, 2H), 3.53 (m, 2H), 3.70 (m, 3H), 3.79 (m, 1H), 3.85 (t, $J = 9.5$ Hz, 1H), 3.96 (apparent t, $J = 5.0$ Hz, 1H), 4.02 (s, 1H), 4.11 (m, 2H), 4.38-4.61 (m, 7H), 4.84 (br s, 1H), 7.17-7.33 (m, 15H); ESIHRMS calcd for $C_{34}H_{40}O_{10}F_2Na$ $[M + Na]^+$ 669.2482, found 669.2480.

1,3,4,5,9,10,11,13-Octa-O-acetyl-2,6:8,12-dianhydro-7-fluoro-D-lyxo-D-galacto-D-manno-tridecitol 3.59.

The tetrol **3.53** (28 mg, 0.04 mmol), 10% Pd on carbon (60 mg), formic acid (0.05 mL) and methanol (2 mL) were stirred under an atmosphere of hydrogen (balloon), for 12 hours. The reaction mixture was purged with argon and filtered through a bed of Celite. The filtrate was then concentrated and the crude fluoro disaccharide **3.56** (13 mg, 93%) was collected as an amorphous solid; $R_f = 0.43$ (60% MeOH/ $CHCl_3$); 1H NMR (500 MHz, D_2O) δ 3.39 (dd, $J = 2.2, 13.2$ Hz, 1H), 3.59-3.77 (m 7H), 3.82 (apparent t, $J = 9.7$ Hz, 1H), 3.87 (m, 2H), 3.94 (d, $J = 3.3$ Hz, 1H), 4.25 (m, 1H), 4.36 (ddd, $J = 1.6, 7.4, 15.3$ Hz, 1H), 5.16 (ddd, $J = 1.6, 7.4, 46.6$ Hz, 1H); ^{13}C NMR (125 MHz, D_2O) 61.8 ($\times 2$), 67.3, 67.4 (d, $J = 6.4$ Hz), 69.1, 69.4, 71.3, 74.7, 76.2 (d, $J = 23.1$ Hz), 76.8, 78.7 (d, $J = 20.5$ Hz), 79.4, 91.7 (d, $J = 176.8$ Hz); ESIHRMS calcd for $C_{13}H_{23}O_{10}Na$ (M + Na) 381.1173, found 381.1182.

Compound **3.56** (13 mg, 0.04 mmol) was then taken up in ethyl acetate (5 mL), DMAP (3.0 mg, 0.02 mmol), acetic anhydride (0.2 mL, 2.12 mmol). This was stirred for

2 hours and the solvent removed *in vacuo*. FCC of the crude residue gave fluoro peracetate **3.59** (17 mg, 67%), colorless oil: $R_f = 0.50$ (60% EtOAc /petroleum ether); ^1H NMR (500 MHz, C_6D_6) δ 1.60, 1.65, 1.67, 1.69 (2), 1.730, 1.733, 1.76 (all s, 3H ea), 3.37 (t, $J = 6.3$ Hz, 1H, H-5), 3.72 (ddd, $J = 2.4, 9.8, 20.1$ Hz, 1H, H-1), 4.03 (m, 1H, H-6'), 4.15 (dd, $J = 6.9, 11.4$ Hz, 1H, H-7'b), 4.22 (dd, $J = 6.3, 11.4$ Hz, 1H, H-6a), 4.24 (dd, $J = 2.6, 12.2$ Hz, 1H, H-6b), 4.45 (dd, $J = 5.3, 12.2$ Hz, 1H, H-7'a), 4.66 (ddd, $J = 2.6, 7.7, 17.7$ Hz, 1H, H-2'), 4.85 (ddd, $J = 2.2, 7.7, 46.6$ Hz, 1H, H-1'), 5.20 (dd, $J = 3.3, 10.0$ Hz, 1H, H-3), 5.41 (dd, $J = 2.6, 9.4$ Hz, 1H, H-4'), 5.49 (d, $J = 3.3, 1\text{H}$, H-4), 5.66 (t, $J = 9.4$ Hz, 1H, H-5'), 5.80 (t, $J = 2.6$ Hz, 1H, H-3), 5.91 (t, $J = 10.0$ Hz, 1H, H-2); ^{13}C -NMR (125 MHz, C_6D_6) δ 20.4, 20.5 ($\times 3$), 20.6 ($\times 3$), 20.7, 61.9, 62.7, 66.1 (d, $J = 2.8$ Hz), 67.0, 68.0, 68.5 (d, $J = 6.9$ Hz), 70.5, 72.8, 73.1, 75.3 (d, $J = 22.0$ Hz), 75.5, 76.6 (d, $J = 20.9$ Hz), 86.5 (d, $J = 185.6$ Hz), 169.5, 169.7, 170.1, 170.2 ($\times 2$), 170.3, 170.5 170.6; ESIHRMS calcd for $\text{C}_{29}\text{H}_{40}\text{O}_{18}\text{F}$ (M + H) 695.2199, found 695.2216.

1,3,4,5,9,10,11,13-Octa-O-acetyl-2,6:8,12-dianhydro-7-fluoro-D-lyxo-D-gulo-D-manno-tridecitol 3.60.

The tetrol **3.54** (20 mg, 0.03 mmol) was subjected to hydrogenation condition as described in the synthesis of **3.56** to give crude fluoro disaccharide **3.58** (9 mg, 90%); amorphous solid; $R_f = 0.29$ (60% MeOH/ CHCl_3); ^1H NMR (500 MHz, D_2O) δ 3.70 (ddd, $J = 1.1, 9.6, 28.8$ Hz, 1H), 3.63-3.77 (m 8H), 3.90 (m, 2H), 3.95 (d, $J = 3.1$ Hz, 1H), 4.02 (t, $J = 1.3$ Hz, 1H), 4.37 (ddd, $J = 1.4, 8.9, 16.8$ Hz, 1H), 5.20 (dd, $J = 8.9, 47.6$ Hz, 1H); ^{13}C NMR (125 MHz, D_2O) 61.5, 61.9, 66.5 (d, $J = 5.2$ Hz), 67.4, 68.5 (d, $J = 7.6$ Hz), 69.4, 71.6, 74.4, 74.7, 75.5, 77.8 (d, $J = 20.1$ Hz), 78.0 (d, $J = 18.1$ Hz), 85.8 (d, $J =$

179.1 Hz); ESIHRMS calcd for $C_{13}H_{24}O_{10}F$ (M + H) 359.1354, found 359.1367.

Treatment of the fluoro C-glycoside **3.58** (9 mg, 0.03 mmol) following the acetylation procedure as described in the synthesis of **3.59**, yielded the fluoro peracetate **3.60** (13 mg, 52%): colorless oil; $R_f = 0.21$ (60% Ethyl acetate/petroleum ether); 1H NMR (500 MHz, C_6D_6) δ 1.61, 1.64, 1.67 ($\times 2$), 1.71, 1.72, 1.76 (all s, 3H ea), 3.27 (t, $J = 6.5$ Hz, 1H, H-5), 3.66 (m, 1H, H-1), 4.10 (m, 2H, H-6a, 7'b), 4.22 (m, 2H, H-6b, 6'), 4.49 (q, $J = 5.9$ Hz, 1H, H-7'a), 4.61 (dt, $J = 3.5, 31.5$ Hz, 1H, H-2'), 5.01 (ddd, $J = 2.7, 5.5, 47.4$ Hz, 1H, H-1'), 5.08 (dd, $J = 3.3, 10.0$ Hz, 1H, H-3), 5.45 (d, $J = 3.3$ Hz, 1H, H-4), 5.62 (t, $J = 10.0$ Hz, 1H, H-2), 5.69 (t, $J = 8.8$ Hz, 1H, H-5'), 5.78 (dd, $J = 3.5, 8.8$ Hz, 1H, H-4'), 6.02 (t, $J = 3.0$ Hz, 1H, H-3'); ^{13}C -NMR (125 MHz, C_6D_6) δ 20.8 ($\times 3$), 20.9 ($\times 2$), 21.0 ($\times 2$), 21.1, 61.4, 62.8, 66.4, 67.1 (d, $J = 3.7$ Hz), 67.4, 68.1, 69.9 (d, $J = 6.0$ Hz), 71.9, 73.5 (d, $J = 1.8$ Hz), 74.4 (d, $J = 19.2$ Hz), 74.6, 76.1 (d, $J = 24.1$ Hz), 94.8 (d, $J = 181.4$ Hz), 169.9, 170.0, 170.1 ($\times 2$), 170.2, 170.3, 170.6, 170.9; ESIHRMS calcd for $C_{29}H_{39}O_{18}FNa$ (M + Na) 717.2018, found 717.2050.

2,6:8,12-dianhydro-7-deoxy-7,7-difluoro-D-threo-L-gulo-D-manno-tridecitol 3.61.

The pentol **3.56** (12.0 mg, 0.02 mmol) was then subjected to hydronolysis condition as above. This gave difluoro C-glycoside **3.61** (7.5 mg, 93%) as a colorless oil; $R_f = 0.32$ (50% MeOH/ CH_3Cl); 1H NMR (500 MHz, D_2O) δ 3.76-3.93 (m, 8H), 3.96 (dd, 2.1, 12.2 Hz, 1H), 4.00 (dd, $J = 3.6, 9.0$ Hz, 1H), 4.05 (t, $J = 9.5$ Hz, 1H), 4.07 (d, $J = 3.2$ Hz, 1H), 4.41 (t, $J = 2.2$ Hz, 1H), 4.59 (ddd, $J = 1.9, 14.6, 20.3$ Hz, 1H); ^{13}C NMR (125 MHz, D_2O) δ 61.2, 61.4, 66.3, 66.7, 66.9, 68.8, 71.1 (d, $J = 2.6$ Hz), 73.9, 76.3 (dd, $J =$

22.9, 26.6 Hz), 76.9 (t, J = 22.9 Hz), 77.9, 79.3, 121.6 (dd, J = 251.1, 253.9 Hz); ESIHRMS calcd for C₁₃H₂₂O₁₀F₂Na [M + Na]⁺ 399.1073, found 399.1070.

(1,3,4,5-tetra-*O*-benzyl-2,6:8,12-dianhydro-7-fluoro-D-lyxo-D-galacto-D-manno-tridecit-10-yloxy)-ethanoic acid sodium salt 3.62.

A mixture of triol **3.51** (100 mg, 0.10 mmol), dibutyltin oxide (39 mg, 0.16 mmol), and anhydrous toluene (20 mL) was heated at reflux in a Dean-Stark apparatus for 1 hour. The solution was then evaporated *in vacuo*, and the residue was dissolved in dry toluene (3 mL). n-Bu₄NI (58 mg, 0.16 mmol) and methyl 2-bromoacetate (0.20 mL, 1.08 mmol) were added and the solution heated at reflux for 1 hour, at which time the volatiles were removed. Partial purification of the crude material by FCC provided a mixture (75 mg) of three components; R_f's 0.18, 0.32 and 0.51 (30 % EtOAc/petroleum ether).

The above mixture (209 mg) was treated with a 1/1 mixture of aqueous 3N NaOH/ethanol (2 mL). After 4 hours, the solvent was removed under reduced pressure, and the residue was purified by chromatography to give **3.62** (106 mg, 75% from triol **3.51**): clear oil; R_f = 0.26 (30% MeOH/ CHCl₃); ¹H NMR (500 MHz, CD₃OD) δ 3.05 (ddd, J = 1.8, 9.2, 26.3 Hz, 1H), 3.37 (dd, J = 2.7, 9.4 Hz, 1H), 3.42 (t, J = 5.9 Hz, 1H), 3.68-3.78 (m 5H), 3.94 (J = 7.8 Hz, 1H), 3.99 (t, J = 9.4 Hz, 1H), 4.10 (d, J = 2.8 Hz, 1H), 4.32 (ABq, J = 16.8 Hz, Δδ = 0.05 ppm, 1H), 4.45-4.76 (m, 9H), 5.10 (ddd, J = 2.0, 7.2, 47.2 Hz, 1H), 7.23-7.35 (m, 18H), 7.42 (m, 2H); ¹³C NMR (125 MHz, CD₃OD) 62.8, 67.0, 68.6, 70.1, 73.0, 73.6, 74.3, 74.7 (d, J = 2.0 Hz), 75.3, 75.5, 76.0, 79.4, 79.6, (d, J =

18.2 Hz), 80.2, 85.4, 87.2 (d, $J = 176.8$ Hz), 128.8-130.2 (several lines), 139.6, 139.8 ($\times 2$), 139.9, 176.3; ESIHRMS calcd for $C_{43}H_{50}O_{12}F$ (M + H) 777.3287, found 777.3292.

(1,3,4,5-tetra-*O*-benzyl-2,6:8,12-dianhydro-7-fluoro-*D*-lyxo-*D*-gulo-*D*-manno-tridecit-10-yloxy)-ethanoic acid sodium salt 3.63.

The triol **3.52** (152 mg, 0.16 mmol) was subjected to similar alkylation and saponification conditions as described for **3.62** to give **3.63**. (78 mg, 61% from triol **3.52**); clear oil; $R_f = 0.22$ (30% MeOH/ $CHCl_3$); 1H NMR (500 MHz, CD_3OD) δ 3.45 (t, $J = 5.9$ Hz, 1H), 3.53-3.73 (m, 7H), 3.85 (m, 2H), 3.94 (m 1H), 4.13 (m, 2H), 4.24 (br s, 1H), 4.41-4.58 (m, 6H), 4.59 (ddd, $J = 1.7, 5.9, 17.7$ Hz, 1H), 4.66 (B of ABq, $J = 11.9$ Hz, $\Delta\delta = 0.20$ ppm, 1H), 4.73 (B of ABq, $J = 11.1$ Hz, $\Delta\delta = 0.30$ ppm, 1H), 5.04 (ddd, $J = 1.1, 5.8, 46.6$ Hz, 1H), 7.11 (m, 2H), 7.17-7.27 (m, 16H), 7.37 (m, 2H); ^{13}C NMR (125 MHz, CD_3OD) 62.9, 67.6, 67.7, 69.2, 70.7, 72.8, 73.2, 74.1 (d, $J = 22.6$ Hz), 74.5, 75.4, 75.8 (d, $J = 4.5$ Hz), 76.5, 80.3, 81.3, 81.4 (d, $J = 21.3$ Hz), 86.1, 93.2 (d, $J = 178.9$ Hz), 128.7-130.1 (several lines), 139.6, 139.8, 139.9, 140.0, 177.3; ESIHRMS calcd for $C_{43}H_{50}O_{12}F$ (M + H) 777.3287, found 777.3293.

(3,4,5-Tri-*O*-benzyl-2,6:8,12-dianhydro-7-deoxy-7,7-difluoro-*D*-threo-*L*-gulo-*D*-manno-tridecit-10-yloxy)-ethanoate methyl ester 3.64.

The triol **3.55** (50.0 mg, 0.05 mmol) was subjected to alkylation procedure as that of triol **3.62**. This afforded a mixture of two lactones (26.9 mg, 66% based on recovered **3.55**). To the above mixture of lactones in THF (3 mL) was added tetrabutyl ammonium fluoride (0.09 mL, 0.09 mmol, 1M solution in THF) and stirred for 2 hours. The solvent

was then evaporated and the residue taken up in concentrated methanolic HCl (0.5 mL) and stirred for 30 min at pH 2.0. Removal of the volatiles under reduced pressure and FCC of the residue provided the methanol ester **27** (10.0 mg, 40% based on recovered **3.55**): colorless oil; $R_f = 0.59$ (10% MeOH/CH₃Cl); ¹H NMR (500 MHz, CD₃OD) δ 3.45 (dd, $J = 3.2, 9.3$ Hz, 1H), 3.46 (t, $J = 6.1$ Hz, 1H), 3.68 (dd, $J = 2.6, 12.1$ Hz, 1H), 3.73-3.82 (m, 5H), 3.85 (s, 3H), 3.92 (m, 1H), 3.97 (dd, $J = 3.2, 7.3$ Hz, 1H), 4.16 (d, $J = 2.4$ Hz, 1H), 4.20 (t, $J = 9.3$ Hz, 1H), 4.25 (t, $J = 3.9$ Hz, 1H), 4.45 (ABq, $J = 16.8$ Hz, $\Delta\delta = 0.04$ ppm, 2H), 4.58-4.79 (m, 7H), 7.34-7.47 (m, 15H); ¹³C NMR (125 MHz, CDCl₃) δ 53.2, 62.4, 62.6, 67.6, 67.8, 68.5, 73.6 (two signals), 74.1, 74.9, 75.8, 79.2 (apparent t, $J = 35.2$ Hz), 79.4, 80.6, 85.9, 128.9-128.6 (several lines), 139.6, 139.8, 139.9, 174.2; ESIHRMS calcd for C₃₇H₄₈O₁₂F₂N [M + NH₄]⁺ 736.3139, found 736.31840.

(2,6:8,12-dianhydro-7-fluoro-D-lyxo-D-galacto-D-manno-tridecit-10-yloxy)-ethanoic acid sodium salt 3.5.

The sodium salt **3.62** (106 mg, 0.13 mmol), 20% Pd (OH)₂ (300 mg) and methanol (20 mL) were stirred under an atmosphere of hydrogen (balloon), for 24 hours. The reaction mixture was purged with argon and filtered through a bed of Celite. The filtrate was then concentrated and the residue taken up into methanol (10 mL) and 3N NaOH (1 mL). This was stirred for 2 hours and the solvent removed. This gave the crude compound **3.5** (38 mg, 66%) as a white solid. In order to obtain pure samples for bioassay, this crude was repeatedly columned using Sephadex LH-20 (water) to provide an amorphous solid (10.2 mg) after lyophilization. $[\alpha]_D + 40.8$ (c 0.25 H₂O); $R_f = 0.45$ (70% MeOH/CHCl₃); ¹H NMR (500 MHz, D₂O) δ 3.34 (ddd, $J = 1.4, 10.0, 29.1$ Hz, 1H,

H-1), 3.41 (dd, $J = 3.3, 9.4$ Hz, 1H, H-3), 3.50-3.72 (m, 7H, H-4', 5', 6', 7'a, 5, 6a, 6b), 3.77 (apparent, dd, $J = 1.6, 12.3$ Hz, 1H, H-7'b), 3.88 (t, $J = 9.6$ Hz, 1H, H-2), 3.92 (m, 1H, H-3'), 3.98 (m, 2H, O=C-CH₂-O), 4.01 (d, $J = 2.9$ Hz, 1H, H-4), 4.28 (ddd, $J = 1.4, 8.8, 16.9$ Hz, 1H, H-2'), 5.08 (ddd, $J = 1.4, 8.5, 47.5$ Hz, 1H, H-1'); ^{13}C NMR (125 MHz, D₂O) δ 61.5, 62.0, 65.5 (d, $J = 6.4$ Hz), 66.3, 67.4, 68.6 ($J = 7.8$ Hz), 68.7, 71.5, 75.5, 77.8 (d, $J = 19.7$ Hz), 78.1 (d, $J = 17.4$ Hz), 79.6, 83.6, 85.8 (d, $J = 178.7$ Hz), 179.1; ESIHRMS calcd for C₁₅H₂₅O₁₂FNa (M + Na) 439.1228, found 439.1238.

(2,6:8,12-dianhydro-7-fluoro-D-lyxo-D-gulo-D-manno-tridecit-10-yloxy)-ethanoic acid sodium salt 3.6:

The sodium salt **3.63** was subjected to similar hydrogenation and saponification conditions as described for the synthesis of **3.5** to give crude product **3.6** (30 mg, 78%). Repeated sephadex LH-20 (water) and reverse phase chromatography provided **3.6** as an amorphous solid after lyophilization. $[\alpha]_{\text{D}} + 14.4$ (c 0.25 H₂O); $R_{\text{f}} = 0.43$ (70% MeOH/CHCl₃); ^1H NMR (500 MHz, D₂O) δ 3.37 (dd, $J = 3.0, 9.4$ Hz, 1H, H-3), 3.50-3.77 (m, 7H, H-1, 5, 6a, 6b, 5', 6', 7'a), 3.76 (d, $J = 12.5$ Hz, 1H, H-7'b), 3.79 (dd, $J = 3.3, 9.2$ Hz, 1H, H-4'), 3.84 (t, $J = 9.7$ Hz, 1H, H-2), 3.97 (m, 1H, O=C-CH₂-O), 4.01 (d, $J = 2.7$ Hz, 1H, H-4), 4.16 (br s, 1H, H-3'), 4.29 (dd, $J = 7.2, 15.4$ Hz, 1H, H-2'), 5.04 (dd, $J = 7.1, 46.6$ Hz, 1H, H-1'); ^{13}C NMR (125 MHz, D₂O) δ 61.8, 61.9, 66.2, 66.4 (d, $J = 8.5$ Hz), 67.2, 68.8, 69.1, 71.4, 76.2 (d, $J = 23.6$ Hz), 76.8, 78.9 (d, $J = 20.4$ Hz), 79.1, 84.0, 91.7 (d, $J = 177.1$ Hz), 179.1; ESIHRMS calcd for C₁₅H₂₅O₁₂FNa (M + Na) 439.1228, found 439.1235.

(2,6:8,12-dianhydro-7-deoxy-7,7-difluoro-D-threo-L-gulo-D-manno-tridecit-10-yloxy)-ethanoic sodium salt 3.7.

The difluoro C-glycoside **3.64** (4.1 mg, 5.71 μmol) was subjected to the hydrogenation and saponification conditions as described for the synthesis of **3.5** to give crude product **3.7** (1.6 mg, 61%): an amorphous solid; ^1H NMR (500 MHz, D_2O) δ 3.62 (dd, $J = 3.1, 9.4$ Hz, 1H), 3.77-3.98 (m, 8H), 4.00 (dd, $J = 3.4, 9.0$ Hz, 1H), 4.14-4.19 (m, 4H), 4.42 (brd s, 1H), 4.61 (t, $J = 17.9$ Hz, 1H); ^{13}C NMR (125 MHz, D_2O) δ 61.2, 61.3, 65.2, 65.5, 66.5, 66.8, 68.5, 70.9, 76.2 (dd, $J = 14.6, 21.9$ Hz), 76.8 (t, $J = 23.2$ Hz), 78.9, 83.1, 121.4 (dd, $J = 251.1, 254.8$ Hz), 178.5; ESIHRMS calcd for $\text{C}_{15}\text{H}_{24}\text{O}_{12}\text{F}_2\text{Na}$ $[\text{M} + \text{H}]^+$ 457.1128, found 457.1125.

Chapter 4

Conformational and biological evaluation of sLe^x mimetics.

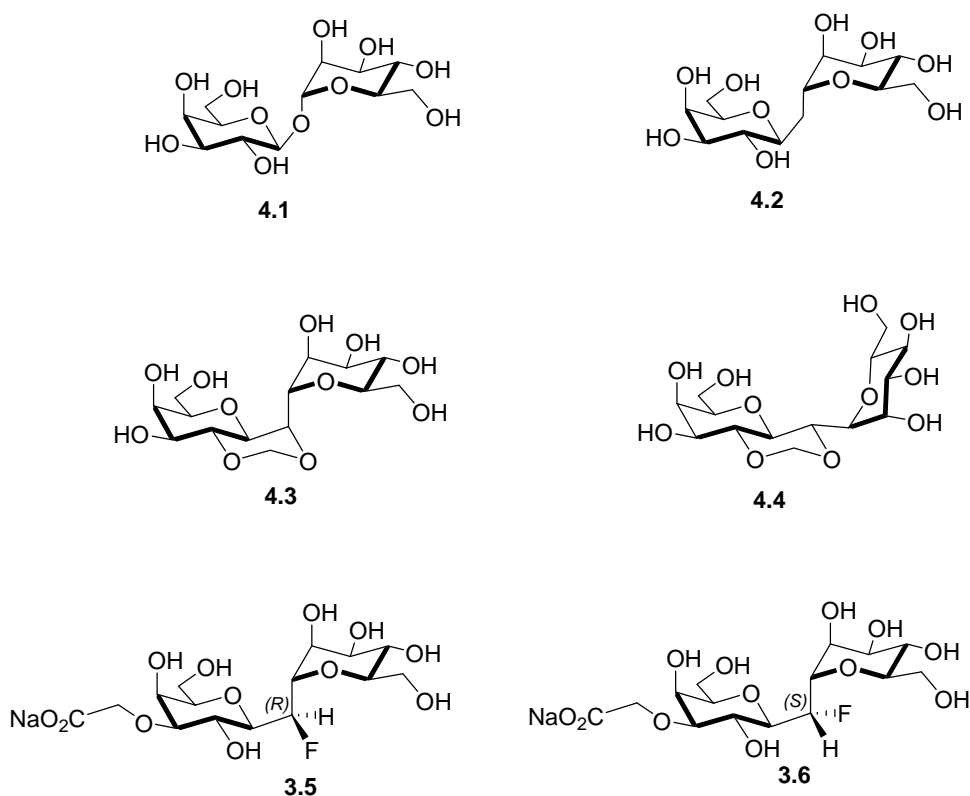
4.1 Introduction.

Carbohydrate-protein interactions are involved in a variety of important biological processes,^{46,47} including attachment of pathogens to host cells, inflammation, metastasis, and immunological defense mechanisms.⁴⁸ C-glycosides (where the *exo* glycosidic oxygen in the natural O-glycosides is replaced with a methylene) are attractive mimetic of O-glycosides because of their enzymatic and chemical stability.⁴⁸ The substitution of the acetal oxygen atoms by methylene groups also increases the flexibility and the number of low energy conformations with respect to the inter-residue linkages. This conformational behaviour has been explained on the basis of the *exo*-anomeric effect.⁴⁸ The smaller energy difference between low energy conformation of C-glycoside (compare to O-glycoside) may lead to increase binding affinity of C-glycoside ligands if there is an enthalpic gain that exceeds the entropic penalty.

Previously the Mootoo laboratory in collaboration with Professor Jiménez-Barbero used a combination of NMR spectroscopy, molecular mechanics (MM) and molecular dynamics (MD) calculations to investigate the conformational properties of **1.4** and **1.5**, and their analogs without the glycolate residue at position 3 of the galactose segment, **4.1** and **4.2** (Figure 4.1).^{49, 50} These studies indicated that the presence or absence of glycolate substituent did not have a significant effect on the conformational behavior with respect to the intersaccharide torsions. For simplicity in ¹H NMR signal

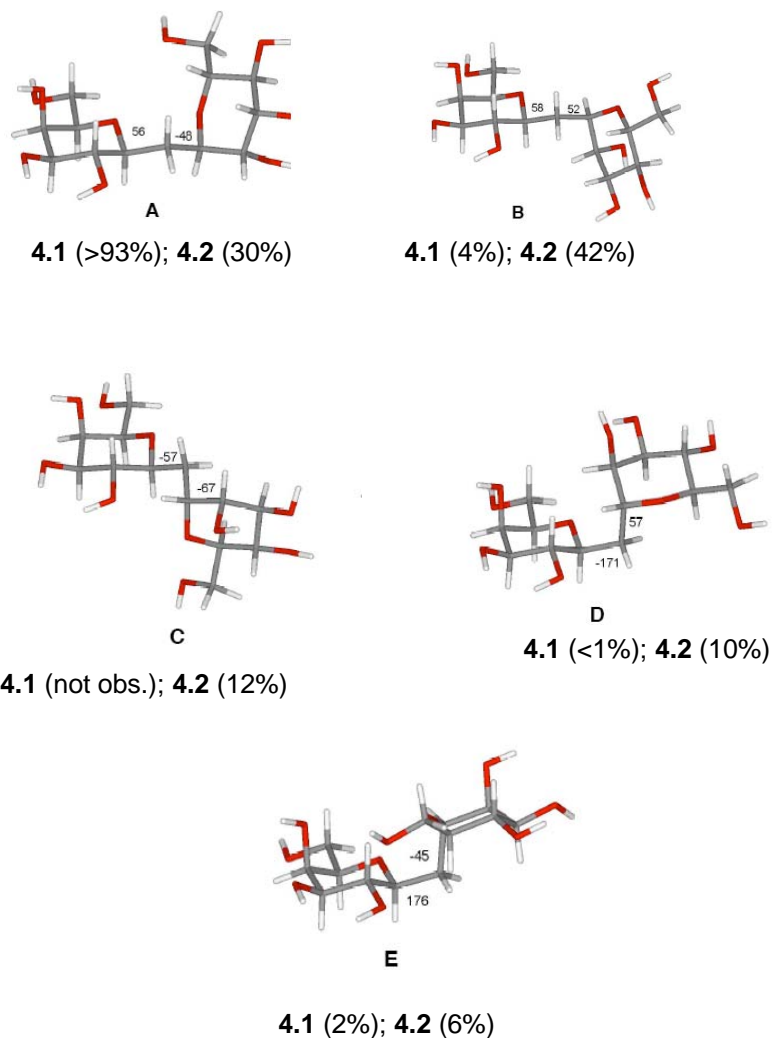
resolution a similar study on the restrained C-glycosides **4.3/4.4**⁵¹, and the fluoro C-glycosides **3.5** and **3.6** was undertaken.

Figure 4.1: Compounds for conformational analysis



4.9. Preliminary results for 4.1 and 4.2

C-glycoside **4.2** in water populates five different conformational families defined by the glyconic torsions (Φ_{gal}, Φ_{man}): **A** (56, -48; 30%), **B** (58, 52; 42%), **C** (-57, -67; 12%), **D** (-171, 57; 10%) and **E** (176, -45; 6%) (Figure 4.2). In comparison, O-glycoside **4.1** exists in greater than 93% in conformational family **A** with very minor populations of conformational families **B**, **D** and **E**. Structures and exact dihedrals are shown for C-glycoside **4.2**.

Figure 4.2: Conformations for **4.1** and **4.2**

Structures and exact dihedrals are shown for C-glycoside **4.2**. The % populations of the conformational families for **4.1** and **4.2** are given in parentheses.

The torsion angles Φ_{Gal} and Φ_{Man} is defined as H1Gal-C1Gal-X-C1Man and H1Man-C1Man-X-C1Gal for **4.2**. Where the Φ shows a $+60^\circ$ value adopts an exo-orientation in β -glycosides, Φ value of -60° value adopts an exo-orientation in α -glycoside and Φ value of $\pm 180^\circ$ value adopts an anti-type arrangement. These

conformers can also be depicted as **(A)** $\text{exo-}\Phi_{\text{Gal}}/\text{exo-}\Phi_{\text{Man}}$, **(B)** $\text{exo-}\Phi_{\text{Gal}}/\text{non-exo-}\Phi_{\text{Man}}$, **(C)** $\text{non-exo-}\Phi_{\text{Gal}}/\text{exo-}\Phi_{\text{Man}}$, **(D)** $\text{anti-}\Phi_{\text{Gal}}/\text{non-exo-}\Phi_{\text{Man}}$, **(E)** $\text{anti-}\Phi_{\text{Gal}}/\text{exo-}\Phi_{\text{Man}}$.

Results and discussion

4.3 Conformational analysis of restrained C-glycosides 4.3 and 4.4

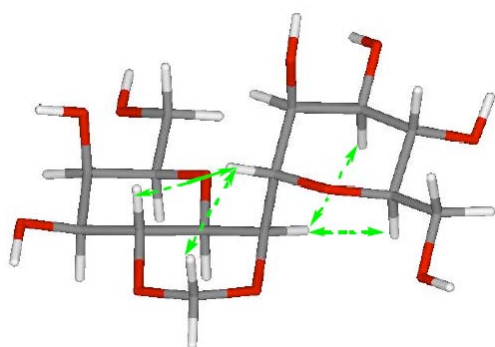
As for **4.1** and **4.2**, the conformationally restrained analogs **4.3** and **4.4**, molecular mechanics and dynamics calculations were performed using the MM3* force field,⁵² as implemented in MACROMODEL 7.1.⁵³ However, unlike the case of the more flexible C-glycoside **4.2**, for which a time-averaged restrained MD protocol was used, no restraints were used for modeling the conformational behavior of **4.3** and **4.4**. Clearly, the presence of the cyclic acetal in these latter structures severely restricted the conformational mobility around the Gal pseudoglycosidic linkage and only the torsional degree of freedom around the Man linkage remained. This made the problem highly simplified, and the conformers around Φ_{Man} were generated and optimized with MM3* with Φ_{Gal} left free during the minimization process. The GB/SA solvation model for water was used.⁵⁴ The probability distribution was calculated from the energy values according to a Boltzmann function at 300 K. Three low energy minima were obtained for **4.3** (Figure 4.3, Table 4.1). The lowest energy and one of the high-energy conformations (Φ_{gal} , Φ_{man} : 175, 51 and 151, -59), corresponded to conformational families **D** and **E** that were previously observed for **4.1** and **4.2**. Also, the highest energy conformation, **F** (Φ_{gal} , Φ_{man} : 168, 140) was detected. The latter was not observed for **4.1** and **4.2**. Intersaccharide coupling constants for these three conformations were obtained from the Karplus-Altona relationship⁵⁵ and compared with the values measured from the ^1H NMR

(Table 4.1). Accordingly **4.3** existed predominantly in conformation **D** (i.e. ca 85%), with minor populations of **E** and **F** (i.e. 10 and 5% respectively). Thus the relative population distribution was in qualitative agreement with the result predicted by MM3* calculations. The observation of strong key NOE contacts (H_{intersac} - $H3_{\text{man}}$, H_{intersac} - $H5_{\text{man}}$, $\text{OCH}_2\text{O}_{\text{ax}}$ - $H1_{\text{man}}$ and $H2_{\text{gal}}$ - $H1_{\text{man}}$) is also in agreement with a high population of conformation **D**.

Table 4.1: MM3* and ^1H NMR data for **4.3**

Conformation	D	C	F	Exp
ΔE (kJ/mol)	0	28	32	
Φ Gal ($^\circ$)	175	151	168	
Φ Man ($^\circ$)	51	-59	140	
J $H1_{\text{gal}}-H_{\text{intersac}}$ (Hz)	5.0	7.0	5.8	5.9
J $H1_{\text{man}}-H_{\text{intersac}}$ (Hz)	10.2	1.1	0.5	8.8
% population	85	10	5	

Figure 4.3: Conformation **D** for **4.3**



key NOE's indicated by double headed arrows

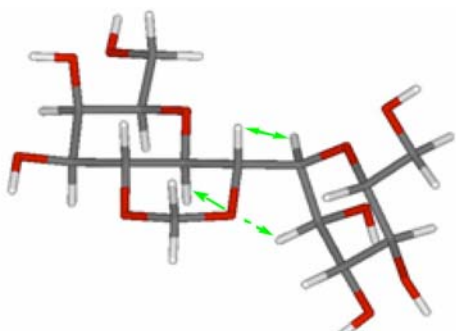
The MM3* calculations for **4.4** indicated low and high energy minima corresponding to conformational families **A** and **B** respectively (Figure 4.4). Analysis of the experimental J values as before indicated that **4.4** existed as a 20:80 ratio of **A**:**B**

(Table 4.2). Thus, in the case of **4.4**, the presence of conformation **A** was overestimated by the MM3* calculations. Observed strong key NOE contacts (H_{intersac} - $H_{1_{\text{man}}}$ and $H_{1_{\text{gal}}}$ - $H_{2_{\text{man}}}$) were also in agreement with a major presence of conformation **B**.

Table 4.2: MM3* and ^1H NMR data for **4.4**

Conformation	A	B	Exp
ΔE (kJ/mol)	0.0	6.0	
Φ Gal ($^\circ$)	54.7	58.2	
Φ Man ($^\circ$)	-50.8	42.0	
J $H_{1_{\text{gal}}}$ - H_{intersac} (Hz)	9.9	9.9	10.0
J $H_{1_{\text{man}}}$ - H_{intesac} (Hz)	9.8	0.9	3.0
% population (soln.)	20	80	

Figure 4.4: Conformation **B** for **4.4**



key NOE's indicated by double headed arrows

4.4 Conformational analysis and epimer identification of fluoro-C-glycosides **3.5** and **3.6**

At first, the configuration at the fluorinated carbon in **3.5** and **3.6** was not known. Again using the MM3* force field, the potential energy surfaces of both isomers were calculated by our collaborator Jesus Jiménez-Barbero. These calculations enabled the prediction of the major low energy conformers of **3.5** and **3.6** about the glycosidic torsion

angles Φ_{Gal} and Φ_{Man} (Table 4.3). These results showed that the compounds **3.5** and **3.6** had distinct conformational behavior depending on their stereochemical nature and five low-energy conformers with appreciable populations (Figure 4.4) were calculated for both compounds.

Table 4.3: Comparison between the inter-residue proton–proton distances deduced by MM3* calculations for the **A–E** conformers of 3.5-3.6 (approximated Φ_{Gal} and Φ_{Man} angles between brackets) and their observed NOEs (% , distance Å)

Conformer Φ/Ψ	(3.5) NOE exp.	(3.6) NOE exp.	A 50/–50 (R/S)	B 60/60 (R/S)	C –70/–70 (R/S)	D –170/60 (R/S)	E 180/– 70 (R/S)
ΔE (R)	—	—	0.97	0.0	4.17	5.74	10.54
ΔE (S)	—	—	0.0	3.57	0.54	5.14	2.72
1G-1M	—	—	2.47	3.15	3.39	3.85	3.83
1G-2M	S (13) /2.30	—	4.35	2.28	4.78	3.96	4.77
1G-3M	Overlap	Overlap	5.16	3.45	4.40	4.28	4.89
1G-5M	Overlap	Overlap	4.28	4.54	2.49	4.59	4.13
CHF-2G	M (3)/3.16	M (3) /3.07	3.18/ 2.57	3.14/3.24	2.53 /4.06	3.81/3.24	3.85/3.18
CHF-2M	M (2)/3.19	M (3) /3.07	2.49/3.27	3.17/3.13	2.69/3.02	3.37/3.90	2.45/2.90
CHF-3M	S (13) /2.32	MS (7)/2.66	3.36/2.49	2.44 /2.36	3.73/2.42	2.51 /4.05	3.43/2.39
CHF-5M	Overlap	S (11)/2.47	4.02/2.32	2.34 /2.20	4.06/2.39	2.11 /2.94	3.86/2.51
1M-2G	VW (0.5)/4.00	M (4) /3.03	4.71	4.35	3.00	2.52	3.28
1M-3G	—	—	5.13	5.42	4.95	5.10	5.53
1M-5G	—	—	4.14	3.98	5.48	5.44	4.75
2M-2G	—	—	5.34	5.02	4.82	2.73	5.11

According to the MM3* calculations, in the *S* isomer the $\text{exo-}\Phi_{\text{Gal}}/\text{exo-}\Phi_{\text{Man}}$ conformer **A** is the major one (40%), followed by the non- $\text{exo-}\Phi_{\text{Gal}}/\text{exo-}\Phi_{\text{Man}}$ conformer **C** (28%), $\text{anti-}\Phi_{\text{Gal}}/\text{exo-}\Phi_{\text{Man}}$ **E** (15%) and **B** (12%). For the *R* isomer the same energy minima was shown, although with different relative steric energies and populations. In this case the major conformer was **B** (65%) followed by the double exo conformer **A** (26%). In this isomer, *anti*-conformers were more destabilized.

A qualitative analysis of **3.5** and **3.6** using NMR experiments, MM3* and dynamics calculations, were done. Also analysis of the measured interglycosidic J values between the methylene proton or the fluorine atom and the H-1Gal and H-1Man protons at the glycosidic positions were carried out. For the theoretical values of the $J_{\text{H,F}}$ constants the Karplus-type equation developed by Chattopadhyaya was used.⁵⁵

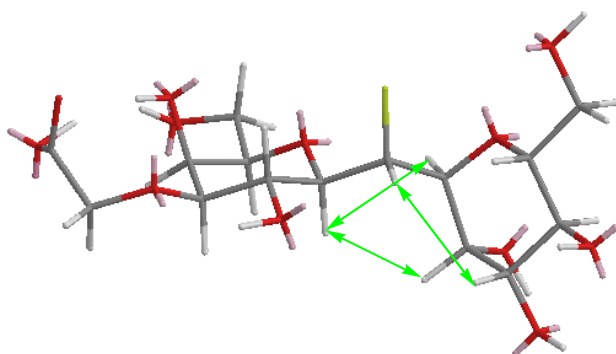
Table 4.4: Expected J values (Hz) for conformations **A-E** for the basic conformations around Φ and Ψ angles for **3.5** and **3.6**, deduced by applying the generalized Karplus⁵⁷ equation proposed by Altona to the geometries provided by MM3* molecular mechanics calculations and Karplus type equation for the $J_{\text{H,F}}$ constants.⁵⁶

Pair	Conformer (J , Hz)					Exp. (J , Hz)	
	<i>A</i>	<i>B</i>	<i>C</i>	<i>D</i>	<i>E</i>	3.5	3.6
<i>R epimer</i>							
G-1'H/CHF	-0.3	0.4	7.9	4.7	4.2	1.4	
G-1'H/F	36.2	34.2	15.1	12.2	16.2	29.1	overlap
M-1H/CHF	-0.4	9.7	0.8	8.0	0.4	8.8	7.2
M-1H/F	35.9	11.0	30.1	14.5	36.1	16.9	15.4
<i>S epimer</i>							
G-1'H/CHF	9.5	9.5	0.8	2.9	3.3	1.4	
G-1'H/F	11.0	8.3	33.5	14.5	14.5	29.1	overlap
M-1H/CHF	9.5	0.8	8.7	0.8	9.5	8.8	7.2
M-1H/F	11.0	32.1	14.5	33.5	15.0	16.9	15.4

From the calculated coupling constant data for the different conformations, compound **3.5** might be predominantly the conformer **B** of epimer R or **C/E** of epimer S (Table 4.4). The presence of a strong 1G-2M NOE which is exclusive for conformer **B** suggested that **3.5** was the epimer R. Therefore for **3.5**, the experimental J values suggest 90% of the *exo*-Gal/*non-exo*-Man conformer **B**, and less than 10% of the double *exo*-anomeric conformer **A**. Thus, the population of conformer **A** of epimer R seemed to be overestimated by the MM3* simulations (from 26% to less than 10%). A strong NOE observed between CHF and 3M indicated that conformer **B** was a significant conformation in solution and also supports the population distribution (Figure 4.5).

Table 4.5: MM3* and ¹H NMR data for *R* epimer **3.5**.

Conformation	<i>A</i>	<i>B</i>	Exp.
ΔE (kJ/mol)	0.97	0	
J G-1'H/CHF (Hz)	-0.3	0.4	1.4
J G-1'H/F (Hz)	36.2	34.2	29.1
J M-1H/CHF(Hz)	-0.4	9.7	8.8
J M-1H/F (Hz)	35.9	11.0	16.9
1G-1M (NOE, dist. Å)	2.47	3.15	-
1G-2M (NOE, dist. Å)	4.35	2.28	(s, 2.30)
CHF-3M (NOE, dist. Å)	3.3	2.44	(s, 2.32)
% population (soln.)	10	90	-

Figure 4.5: Conformation B for **3.5**

key NOE's indicated by doubly headed arrows

The values of the coupling constants for compound **3.6** were in the same range of that of **3.5**, therefore the same possibility; **B-R** epimer or **C/E-S** epimer. The strong 1G-2M NOE, which is exclusive for **B-R** epimer was not observed in **3.6** (where as it is for **3.5**). Thus **C/E-S** epimer were considered more likely possibilities. The presence of strong CHF-3M and CHF-5M NOE's are consistent with both conformers **C** and **E**, but an estimation of the ratio of each conformer was impossible because of the experimental J values were similar to the to that predicted for **C** and **E** (Table 4.6).

Table 4.6: MM3* and ^1H NMR data for *S* epimer **3.6**.

Conformation	C	E	Exp.
ΔE (kJ/mol)	0.54	2.72	
J G-1'H/CHF (Hz)	0.8	3.3	
J G-1'H/F (Hz)	33.5	14.5	overlap
J M-1H/CHF(Hz)	8.7	9.5	7.2
J M-1H/F (Hz)	14.5	15.0	15.4
1G-2M (NOE, dist. Å)	3.39	3.85	-
CHF-3M (NOE, dist. Å)	2.42	2.39	(MS, 2.66)
CHF-5M (NOE, dist. Å)	2.39	2.51	(S, 2.47)
1M-2G (NOE, dist. Å)	3.00	3.28	(M, 3.03)

4.5 P-Selectin binding data for sLe^x mimetics 1.4, 1.5, 2.3, 2.4, 3.5 and 3.6.

The competitive binding of O-disaccharide **1.4** and C-disaccharides **1.5**, **2.3**, **2.4**, **3.5** and **3.6** to a soluble truncated form of human P-selectin was next evaluated in a Biacore assay with an immobilized monomeric truncated form of human PSGL-1 as the reference ligand.¹⁷ The data revealed that at 12 mM, **1.4**, **1.5**, **2.3**, **2.4**, **3.5** and **3.6** showed 48, 26, 25, 31, 39, and 26% inhibition respectively (Table 4.7). The IC₅₀ of sLe^x under these conditions was previously found to be 15 mM.¹³ It should be noted that our result for O-disaccharide **1.4** is in disagreement with an earlier study in which the binding of **1.4** to P-selectin was found to be 40 times greater than sLe^x.¹³ This inconsistency could be due to the fact that the latter investigation used a cell-based assay,⁵⁷ but this explanation is not completely satisfying because the IC₅₀'s for sLe^x in both the Biacore and cell-based measurement were similar (ca 8 vs. 15 mM respectively).

Table 4.7: P-selectin Biacore inhibition assay results for sLe^x mimetics.

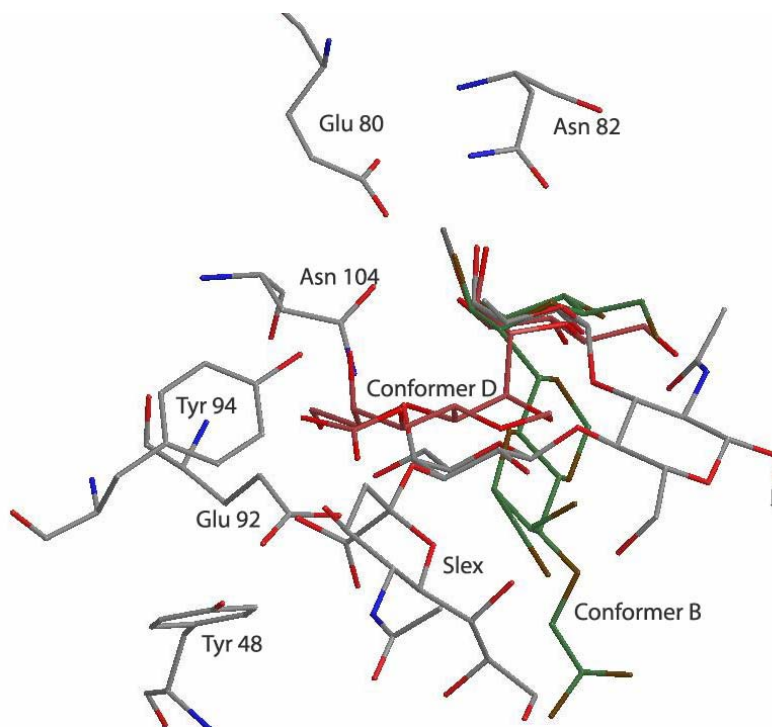
Samples	Concentration /mM	Avg. % inhib	sd
Glycyrrhizin (control)	1	52	0.99
O-Glycoside 1.4	12	48	0.53
	6	33	0.52
C-Glycoside 1.5	12	26	0.65
	6	12	0.63
Rigid C-glycoside 2.3	12	25	0.57
	6	10	0.60
Rigid C-glycoside 2.4	12	31	0.46
	6	13	0.50
R-Fluoro-C-glycoside 3.5	12	39	0.41
	6	19	0.56
S-Fluoro-C-glycoside 3.6	12	26	0.36
	6	13	0.45
Glycyrrhizin (control)	1	51	0.88

4.6 Discussion

It was anticipated that because of the different relative spatial positioning of the two sugar residues in **1.4**, **1.5**, **2.3**, **2.4**, **3.5** and **3.6**, these analogs would have very different activities. In so far as % inhibition could be used as a measure of binding, this information could provide insight on the optimal conformation requirements for binding of the disaccharide framework. However, the similarity of activities of these mimetics of sLe^x made any such conclusions somewhat speculative. Nevertheless, analysis of possible bound conformations in terms of the sLe^x-P-selectin recognition model may be useful for the design of more active P-selectin ligands (Figure 1.4). In this regard, the more conformationally rigid analogs **2.3** and **2.4** had more limited modes of binding and were an appropriate starting point. Accordingly, the galactose and mannose residues of **2.3** and **2.4** can individually, but not simultaneously mimic the interaction of the galactose and fucose residues in sLe^x, respectively (Figure 4.5). The selection of which of the two subunits of the disaccharide framework to use as a mimic of the analogous segment of sLe^x, was guided by the crystal structure of the sLe^x-P-selectin complex and existing structure activity data which suggested that the fucose residue accounted for the major part of the overall binding energy of sLe^x.^{4,58,59} Therefore, it was speculated that the binding of the mannose segment of these disaccharide frameworks was conserved, and closely mimics that of the fucose in sLe^x (i.e. the 2-, 3- and 4- OH of the mannose residue map to the 4-, 3- and 2- OH of fucose). In the case of the restrained mimetics **2.3** and **2.4** this meant that individual galactose substituents must occupy different spatial positions relative to the mannose segment, and would therefore interact differently with the receptor sites in the galactose binding domain of sLe^x. Inspection of the sLe^x-P

selectin complex indicated these to be Tyr48, Tyr94 and Glu92, which interacted with Neu-COO-, Gal-4-OH and Gal-6-OH of sLex.⁵⁹ The assumption that the mannose mimics the fucose of sLex, the conformational restraints on **2.3** and **2.4**, and the notion that the relative position of the carboxylate of neuraminic acid in sLex and the fucose is critical for binding,^{4,60} point to conformations like **D** ($\Phi_{gal}, \Phi_{man}; 180^\circ, 60^\circ$) and **B** ($\Phi_{gal}, \Phi_{man}; 60^\circ, 60^\circ$) as possible bound orientations for **2.3** and **2.4** respectively (Figure 4.6). Considering the P-selectin sites in the vicinity of the galactose binding domain of sLex, conformations like **B** and **D** appeared to give the best fit to the receptor, relative to their rotamers with respect to the unrestrained Φ_{man} torsion.

Figure 4.6: Comparison of binding of sLex, and conformations **B** and **D** using Chem3D program.



Thus with a restrained Φ_{gal} of approximately 170° , a Φ_{man} of 60° in **2.3** gave a **D**-like conformation (Φ_{gal} , Φ_{man} ; 170, 60), which positions the carboxylate oxygen in **2.3** close to the Tyr48 binding site for Neu-COO- in sLe^x. The respective COO-/OTyr48 distances for this conformation and that of bound sLe^x were 4.8 and 3.6 Å. In this **D**-like conformation Gal-6-O in **2.3**, is farther away from the carboxylate oxygen of Glu92, compared with Gal-6-O in sLe^x (3.5 vs. 2.5 Å), while the Gal-4-O/Tyr94-O distance is closer for **2.3** compared with sLe^x (2.5 vs 3.5 Å). For **2.4**, in which Φ_{gal} is restrained to 60° , a Φ_{man} of 60° leads to conformation **B** which places the carboxylate of the glycolate in very similar proximity to Tyr 48-O compared with the carboxylate in sLe^x (3.4 vs 3.6 Å respectively). However, in conformation **B** the oxygens at position 4 and 6 of the galactose segment were at a much further distance from Tyr94-O and Glu92-COO- respectively, compared with the corresponding galactose positions in sLe^x, and apparently out of binding range (i.e. ca 7 Å for gal-4-O/Tyr94-O, ca 9 Å for gal-6-O/Glu92-COO-). Instead, Gal-6-O in conformation **B** was relatively close to Tyr94-O (5.2 Å), and appeared more likely to interact with this residue. Therefore, to summarize, the most important points on the binding of **B** and **D** type frameworks were an essentially identical mode of binding of their mannose segments but differences in the receptor interactions in their galactose segments, i.e. **2.3** appeared capable of a relatively strong Gal-4-OH/Tyr94 and weak COO-/Tyr48 and Gal-6-OH/Glu92 interactions, whereas for **2.4** a strong COO-/Tyr48, and weak Gal-6-OH/Tyr94 appeared to be the case. Since unbound **2.3** and **2.4** favored conformations like **D** and **B** respectively, and therefore expected to incur relatively small reorganizational energetic costs on binding, their similar binding might be an indication that these galactose interactions are relatively

weak compared to the binding of the mannose segment. This situation is analogous to the apparent dominant binding of the fucose residue in sLe^x noted earlier. However, an alternative explanation is that the combined magnitude of the carboxylate and galactose interactions in **D** and **B** are similar. Analogs of **2.3** and **2.4** with appropriate deletions of alcohol groups may be useful in probing this issue.

If the active conformation for restrained glycosides **1.4** and **1.5** were considered in terms of conformers **B** and **D**, the O-glycoside; which was preorganized in conformation **A** (Φ_{gal} , Φ_{man} ; ca 60° , -60°), was closer in enthalpy to conformation **B** than **D** and may prefer a **B**-type bound conformation. Interestingly, minor reduction in the dihedral angles for the idealized **B** conformation (Φ_{gal} , Φ_{man} ; 60° , 60°), which would be energetically less demanding for **1.4** (compared with the more rigid framework **2.4**), allows for **B**-like conformations that have closer COO-/Tyr48 and Gal-6-O/Tyr94 contacts. Thus the larger enthalpy cost (compared to **2.4**), required for the O-glycoside **1.4** to adopt a **B** like conformation may be offset by a more intimate fit of **1.4** to the receptor. The larger entropy cost (relative to **2.4**) that the flexible C-glycoside **1.5** would incur on binding in a **B**-like conformation, may be compensated in a similar way. Thus, a **B**-like active conformation for **1.4** and **1.5** may also account for the similarity of their binding compared with **2.4** (and **2.3**).

In the case of the monofluoro-C-glycosides **3.5** and **3.6**, the binding affinity for P-selectin was also similar that of the natural O-glycoside **1.4** and its C-glycoside analogs **1.5**, **2.3** and **2.4**. This similarity may be an indication that all these analogs are not properly preorganized in an optimal conformation for binding and hence pay similar energetic penalties in order to do so. However, it should be noted that the R-fluoro-C-

glycoside **3.5** which existed predominantly (< 90%) in the **B** conformation (exo- Φ_{Gal} /non-exo- Φ_{Man}) was twice as active as the *S*-isomer **3.6** which does not possess any **B** conformers.

4.7 Summary

C-disaccharides with different conformational properties about the intersaccharide linker were synthesized and evaluated together with their parent O-disaccharide for binding to P-selectin. These analogs were found to have comparable activity to sLe^x. However, the small differences in their activity did not permit any clear conclusions on a favored bound conformation. Possible bound conformations were considered based on the torsional constraints of conformationally restrained analogs and the structure of the sLe^x-P-selectin complex. That such conformationally different analogs show similar activity as sLe^x, raises questions about the extent to which the substituents on the galactose residue of sLe^x contribute to binding. Studies with more finely tuned analogs of **2.3**, **2.4**, **3.5** and **3.6** could give a clearer understanding of this disaccharide framework. On a general note the C-glycosides used in this study represent conformational mimetics of O-glycosides that could find wider use as recognition probes.

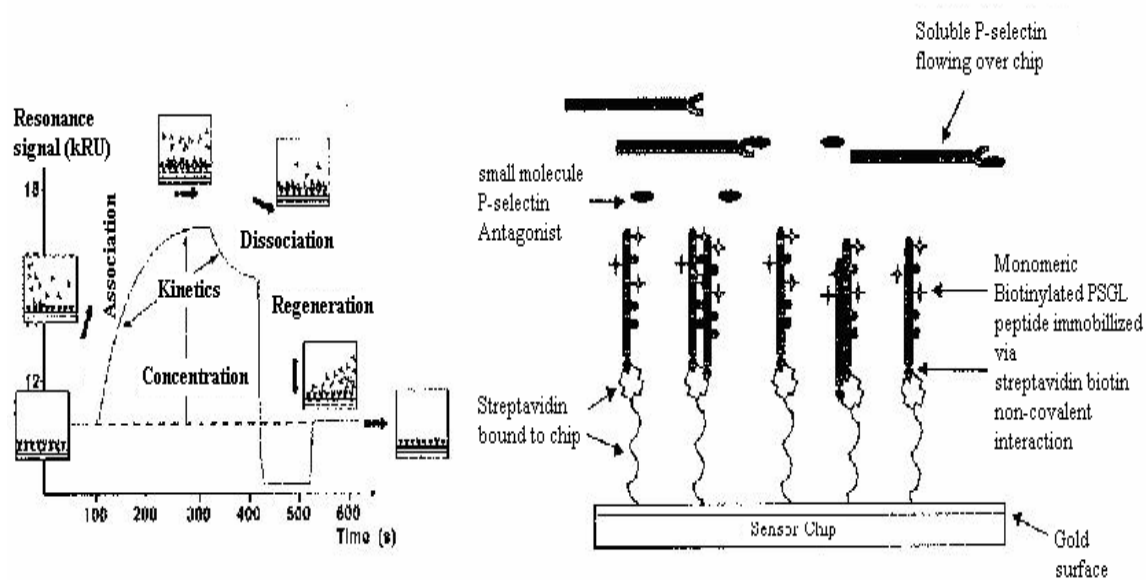
4.8 Experimental section

P-selectin-Biacore Inhibition Assay

P-selectin inhibition assays for **1.4**, **1.5**, **2.3**, **2.4**, **3.5** and **3.6** were performed using a surface plasmon (Biacore) assay on a Biacore instrument, following the published

protocol.¹⁷ Biotinylated 19ek (a purified monomeric truncated form of human PSGL-1) was immobilized on SA sensor chip 11 and a soluble recombinant truncated form of human P-selectin was delivered to the coated 19ek sensor chip at 30 $\mu\text{L}/\text{min}$ and 25 $^{\circ}\text{C}$ in the presence and absence of the test ligands (Figure 4.7).

Figure 4.7: Characteristics of a sensogram, the readout of the Biacore instrument, inset picture shows events at the sensor chip surface.⁴



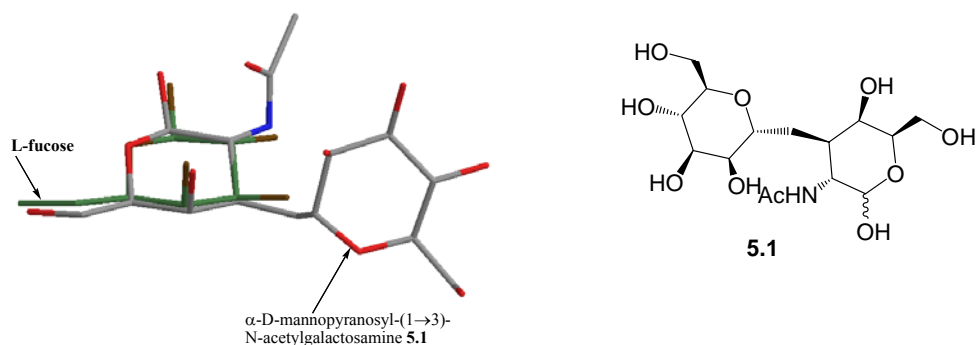
Chapter 5

Synthesis of the C-glycoside of methyl α -D-altropyranosyl-(1 \rightarrow 4)- α -D-glucopyranoside.

5.1 Introduction

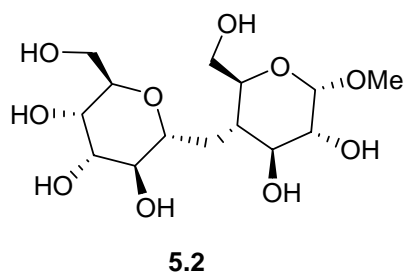
The interest in disaccharide analogues as specific inhibitors of glycosidases and glycotransferases is based on the idea that they may act as bisubstrate-type transition state mimetics.^{61, 62} C-disaccharides have attracted attention as hydrolytically stable glycomimetics because of their subtle conformational differences compared to O-glycosides.⁶³⁻⁶⁶ These concepts inspired a recent investigation on the C-disaccharide of α -D-mannopyranosyl-(1 \rightarrow 3)-N-acetylgalactosamine **5.1** (Figure 5.1).⁶¹ Compound **5.1** was found to be a good bisubstrate inhibitor of human α -(1,3)-fucosyltransferase VI (which catalyzes the transfer of GDP-fucose to N-acetyllactosamine). Presumably, the α -L-fucopyranosyl and the D-GlcNAc binding sites of the enzyme are analogous to the respective binding regions for the C- α -D-mannopyranosyl and the D-GalNAc residues of **5.1**. Thus as illustrated in this example, disaccharide analogues⁶⁷⁻⁷⁰ with “unnatural” glycone substituents are of interest. In this vein, we explored the use of our oxocarbenium ion cyclization methodology^{14, 51, 71} for the synthesis of methyl C- α -D-altropyranosides.

Figure 5.1: A Chem 3D overlay for the N-acetyl galactosamine moiety of the α -(1,3)-fucosyltransferase VI inhibitor **5.1** with L-fucose.



Accordingly, the C-glycoside of methyl- α -D-altropyranosyl-(1 \rightarrow 4)- α -D-glucopyranoside **5.2** was targeted (Figure 5.2). This synthesis may also be significant to the synthesis of the naturally occurring class of aryl- α -C-altroside, the altromycins.^{72, 73}

Figure 5.2: C-glycoside **5.2**

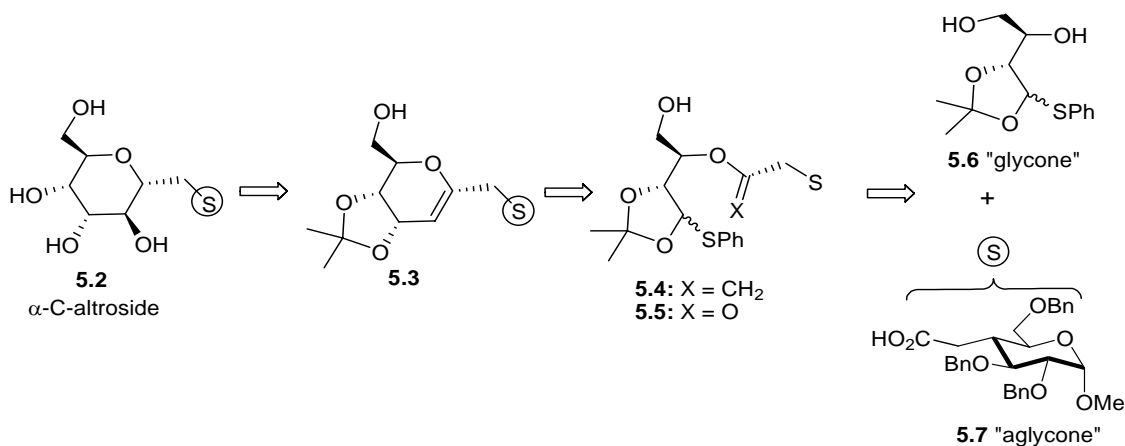


Results and discussion

5.2 Retrosynthesis

The synthesis of compound **5.2** requires a glycone segment **5.6**, and an aglycone **5.7**. Thus **5.2** can be derived from the selective hydroboration of the glycal **5.3**. Methyl triflate cyclization of **5.4** would give the glycal **5.3** as the only regioisomer. The enol ether **5.4** should be readily obtained from Tebbe olefination of the ester **5.5**. DCC mediated esterification of **5.6** and **5.7** would provide **5.5** (Scheme 5.1).

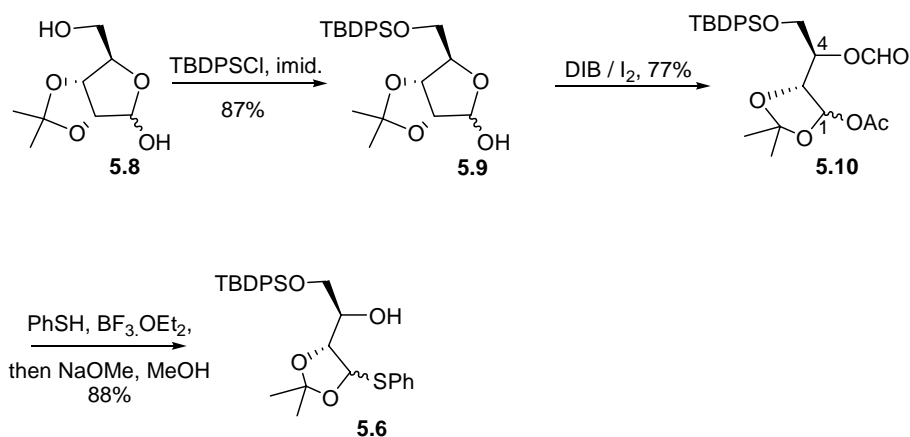
Scheme 5.1: Retrosynthesis of compound 5.2



5.3 Synthesis of 1-thio-1,2-isopropylidene acetals 5.6.

The TIA segment **5.6** was obtained from 2,3-*O*-isopropylidene-*D*-ribofurose **5.8**.⁷⁴ Thus conversion of **5.8** to the silyl ether **5.9**, and treatment of this material with diacetoxyiodosobenzene (DIB) and iodine gave the 1-*O*-acetyl-1,2-*O*-isopropylidene **5.10**, according to the Suarez protocol.²² Compound **5.10** was exposed to thiophenol and BF₃·OEt₂ at -78 °C and the crude subjected to base hydrolysis of the formate ester. Thus **5.6** was made in 59% overall yield from 2,3-*O*-isopropylidene-*D*-ribofurose **5.8**.

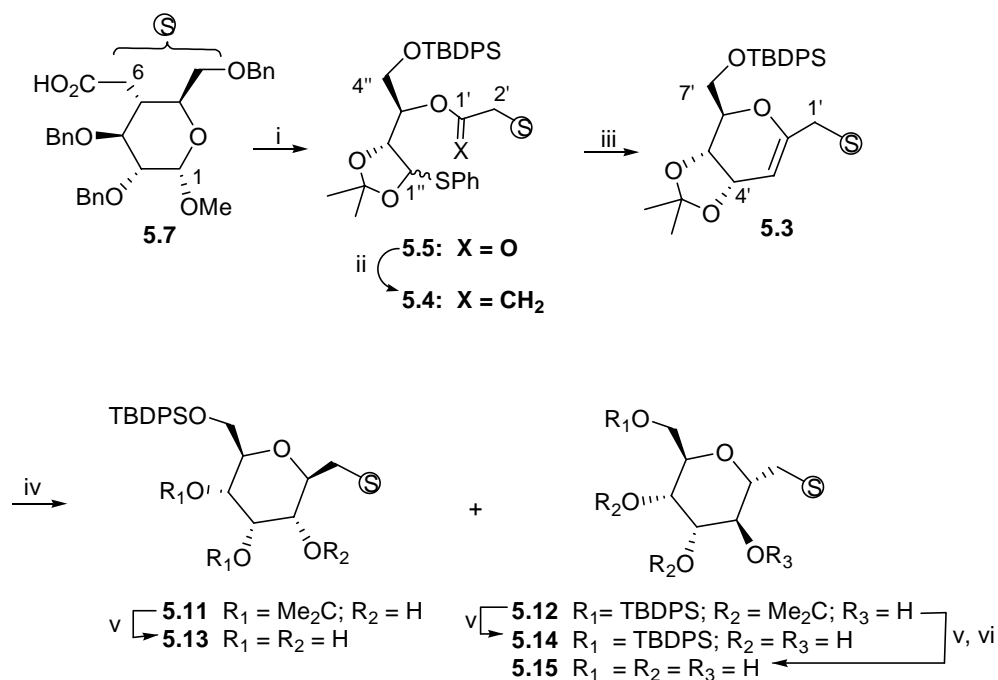
Scheme 5.2: Synthesis of the 1-thio-1,2-isopropylidene acetals 5.6



5.4 C-glycoside synthesis

The aglycone segment **5.7** was obtained from the hydrolysis of the known ethyl ester derivative.^{75, 76} The DCC coupling of alcohol **5.6** and acid **5.7** afforded the ester **5.5** which was transformed to the enol ether **5.4** by treatment with Tebbe reagent (Scheme 5.3). The key cyclization step on **5.4** was promoted by methyl triflate in the presence of 2,6-di-tert-butyl-4-methylpyridine, and led to C1 substituted glycal **5.3** in 70% yield. Hydroboration of **5.3** provided an approximately 1:5 ratio of two stereoisomers **5.11** and **5.12** respectively, in a combined yield of 84%. Chromatography of this mixture provided partial separation, giving a sample of pure **5.12** and an unseparated mixture of **5.11** and **5.12**.

Scheme 5.3: Synthesis of C-glycosides

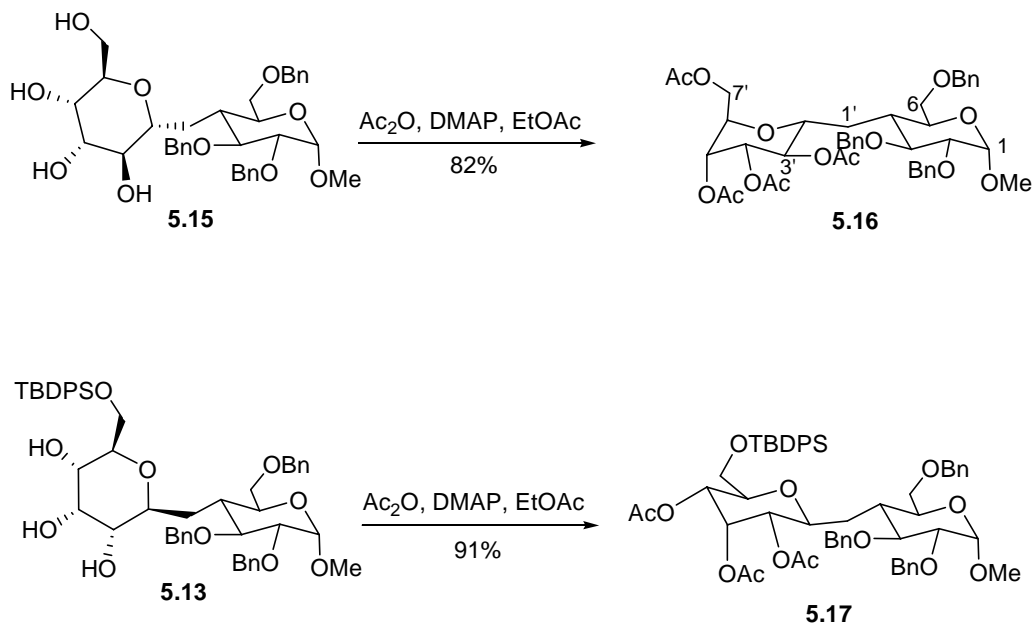


(i) **5.6**, DCC, DMAP, PhH, 85%; (ii) Tebbe, 60%; (iii) MeOTf, DTBMP, CH₂Cl₂, 70%;
 (iv) BH₃.DMS then Na₂O₂, 84%; (v) HCl, MeOH; (vi) Bu₄NF, THF.

5.5 Conformational analysis of hydroboration products.

The stereochemistry of **5.12** was determined by conversion to the tetraacetate **5.16**. The J values ($J_{1',2'} = J_{2',3'} = 8.5$, $J_{3',4'} = 2.5$, $J_{4',5'} < 2.0$ Hz) for **5.16** were in agreement with an α -C-altroside in primarily the 1C_4 conformation.⁷⁷ The minor product of the hydroboration reaction, **5.11** was characterized as the triacetate **5.17**. Acid hydrolysis of the mixture of hydroboration products **5.11** and **5.12** provided an easily separable mixture of the corresponding triols **5.13** and **5.14**. Acetylation of **5.13** provided **5.17**. The data for **5.17** ($J_{1',2'} = 10.0$, $J_{2',3'} = 2.7$, $J_{3',4'} = 2.7$, $J_{4',5'} = 9.5$ Hz) supported a β -alloside in the 4C_1 conformation,⁷⁷ a result that is self consistent with the stereochemistry of the hydroboration reaction (Scheme 5.4).

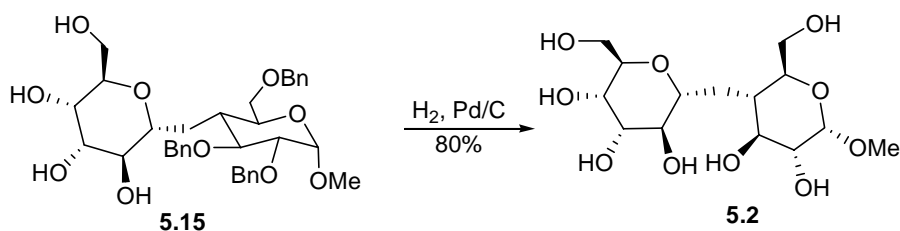
Scheme 5.4: Synthesis of acetylated products 5.16 and 5.17.



5.6 Synthesis of disaccharide 5.2

The targeted α -C-altroside **5.2** was obtained by hydrogenolysis of the tetraol **5.15**. The stereochemical integrity of **5.2** resides in the assignment of the aforementioned tetraacetate derivative **5.16**. Poor signal resolution in the $^1\text{H-NMR}$ resolution did not allow for conformational analysis of the altropyranoside ring.

Scheme 5.5: Final synthesis of α -C-altroside 5.2.



5.7 Summary

The synthesis of the **5.2** shows that the TIA C-glycoside methodology may also be used to prepare stereoselectively of α -C-altro-disaccharides. An attractive feature is the convergent nature of this strategy, which allows for incorporation of different aglycone subunits. Elaboration of the intermediate disaccharide glycal intermediates provides additional possibilities for variation of the glycone segment. These attributes are especially relevant to the synthesis of libraries of C-disaccharides with unusual substitution patterns.

5.8 Experimental section

2,3-*O*-Isopropylidene-5-*O*-*tert*-butyldiphenylsilyl- α/β -*D*-ribofuranose **5.9**.

A solution of 2,3-*O*-isopropylidene-*D*-ribofuranose **5.8 α/β** ⁷⁴ (7.86 g, 41.0 mmol), TBDPSCI (11.4 mL, 41.0 mmol), and imidazole (5.64 g, 85.0 mmol) in anhydrous DMF (100 mL) was stirred at 50 °C for 1.5 hours. The reaction mixture was then diluted with water and extracted with ether. The combined organic phase was washed with brine, dried (Na₂SO₄), filtered, and concentrated under reduced pressure. The residue was purified by FCC to give an inseparable mixture **5.9 α/β** (15.2 g, 87%) as a colorless oil; R_f = 0.42 (20% EtOAc/petroleum ether); IR (neat) 3423 cm⁻¹. Compound **5.9 α** had ¹H NMR (300 MHz, CDCl₃) δ 1.06 [s, 9H, (CH₃)₃CSi], 1.33, 1.48 [both s, 3H ea, C(CH₃)₂], 3.66 (dd, J = 2.6, 11.4 Hz, 1H, H-5a), 3.83 (dd, J = 2.6, 11.4 Hz, 1H, H-5b), 4.29 (bs, 1H, H-4), 4.54 (d, J = 10.6 Hz, D₂O ex, 1H, OH), 4.61 (d, J = 6.2 Hz, 1H, H-3), 4.72 (d, J = 5.9 Hz, 1H, H-2), 5.38 (d, J = 10.3 Hz, 1H, H-1), 7.43–7.80 (m, 10H, Ph); ¹³C NMR (75 MHz, CDCl₃) δ 25.3, 26.8 [C(CH₃)₂], 27.2 (CH₃)₃CSi], 65.8, 82.0, 87.4, 87.5 (4C, C-2, 3, 4, 5), 103.6 (C-1), 112.3 [C(CH₃)₂], 128.0–135.9 (Ph). Compound **5.9 β** had ¹H NMR (300 MHz, CDCl₃) δ (selected signals) 1.06 [s, 9H, (CH₃)₃CSi], 1.41, 1.56 [both s, 3H ea, C(CH₃)₂], 3.94 (d, J = 11.4 Hz, 1H, D₂O ex, OH), 4.16 (bs, 1H, H-4), 5.67 (dd, J = 4.0, 11.4 Hz, 1H, H-1); ¹³C NMR (75 MHz, CDCl₃) δ 25.1, 26.5 [C(CH₃)₂], 27.2 (CH₃)₃CSi], 66.2, 79.8, 81.6, 82.2 (4C, C-2, 3, 4, 5), 98.2 (C-1), 113.3 [C(CH₃)₂]; Mixture **5.9 α/β** had HRMS(EI) m/z calcd for C₂₃H₂₉O₅Si (M-CH₃) 413.1784. Found 413.1777.

1-*O*-Acetyl-4-*O*-*tert*-butyldiphenylsilyl-3-*O*-formyl-1,2-*O*-isopropylidene-*D*-erythro-hemiacetal **5.10.**

A solution of compound **5.9** (5.82 g, 13.6 mmol) in anhydrous cyclohexane (90 mL) containing diacetoxyiodosobenzene (4.39 g, 15.2 mmol) and iodine (3.45 g, 13.5 mmol) was stirred under an atmosphere of argon at rt for 4 hours. The reaction mixture was then diluted with water and extracted with ether. The organic phase was washed with aqueous Na₂S₂O₃ and brine, then dried (Na₂SO₄), filtered, and concentrated under reduced pressure. The residue was separated by FCC to give major (4.00 g, 61%) and minor (1.08 g, 16%) isomers of **5.10**. The major isomer was isolated as a white solid; R_f = 0.23 (5% EtOAc/petroleum ether); IR (film) 1732, 1752 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 1.09 (s, 9H, (CH₃)₃CSi), 1.41, 1.50 [both s, 3H ea, C(CH₃)₂], 2.04 (s, 3H, CH₃CO), 3.91 (m, 2H, H-4a, 4b), 4.52 (dd, J = 1.8, 5.9 Hz, 1H, H-2), 5.18 (q, J = 4.8 Hz, 1H, H-3), 6.27 (d, J = 1.8 Hz, 1H, H-1), 7.38-7.70 (m 10H, Ph), 8.02 (s, 1H, HCO); ¹³C NMR (75 MHz, CDCl₃) δ 21.9 (CH₃CO), 27.5, (CH₃)₃CSi, C(CH₃)₂, 1C], 28.1 [C(CH₃)₂, 1C), 62.9, 73.4, 81.2 (3C, C-2, 3, 4), 96.8 (C-1), 113.6 [C(CH₃)₂], 128.4, 130.4, 133.6, 136.2 (Ph), 160.3 (HCO), 170.2 (CH₃CO). HRMS (EI) m/z calcd for C₂₄H₃₁O₅Si (M-C₂H₃O₂) 427.1941, found 427.1932.

The minor isomer was isolated as a colorless oil; R_f = 0.28 (5% EtOAc/petroleum ether); ¹H NMR (300 MHz, CDCl₃) δ 1.09 (s, 9H), 1.42, 1.51 [both s, 3H ea, C(CH₃)₂], 2.04 (s, 3H, CH₃CO), 3.98 (m, 2H, H-4a,b), 4.54 (dd, J = 3.3, 9.2 Hz, 1H, H-2), 5.33 (m, 1H, H-3), 6.33 (d, J = 3.3 Hz, 1H, H-1), 7.40-7.70 (m 10H, Ph), 7.91 (s, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 21.3 (CH₃CO), 26.3 [C(CH₃)₂, 1C], 27.1 (CH₃)₃CSi], 28.4 [C(CH₃)₂,

1C], 63.4, 71.3, 76.0 (3C, C-2, 3, 4), 93.5 (C-1), 112.7 [C(CH₃)₂], 127.8, 129.9, 133.3, 135.7 (Ph), 159.5 (HC = O), 170.0 (CH₃C=O); ESMS m/z 504.3 [M + NH₄]⁺.

**4-*O*-*tert*-Butyldiphenylsilyl-1,2-*O*-isopropylidene-*D*-*erythro*-*S*-phenyl
monothiohemiacetal **5.6**.**

BF₃.OEt₂ (1.60 mL, 12.7 mmol) was slowly added to a solution of **5.10** (4.10 g, 8.4 mmol) and thiophenol (2.16 mL, 21.1 mmol) in anhydrous CH₂Cl₂ (40 mL) at -78 °C under an atmosphere of argon. The reaction was warmed to -40 °C and stirred at this temperature for 1 hour, or until TLC indicated complete disappearance of the starting material. Triethylamine (6.0 mL) was then added, and the reaction mixture was diluted with satd. aq NaHCO₃ and extracted with ether. The organic phase was washed with brine, dried (Na₂SO₄), filtered, and concentrated under reduced pressure. The crude material was dissolved in methanol (50 mL) and treated with a solution of NaOMe in methanol at rt for 30 min. Most of the solvent was then removed under reduced pressure. FCC of the residue provided **5.6** (3.49 g, 81%) as a colorless oil; R_f = 0.42 (10% EtOAc/petroleum ether); IR (neat) 3485 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 1.07 (s, 9H, (CH₃)₃CSi), 1.40, 1.53 [both s, 3H ea, C(CH₃)₂], 2.55 (d, J = 4.4 Hz, 1H, D₂O ex, OH), 3.81 (m, 3H, H-4a, 4b, H-3), 4.18 (t, J = 5.7 Hz, 1H, H-2), 5.58 (d, J = 5.5 Hz, 1H, H-1), 7.26–7.71 (m, 15H, Ph); ¹³C NMR (75 MHz, CDCl₃) δ 26.3 [C(CH₃)₂, 1C], 26.8 (CH₃)₃CSi), 27.1 (CH₃)₃CSi), 27.8 [C(CH₃)₂, 1C], 64.6, 72.4, 81.2 (3C, C-2, 3, 4), 86.0 (C-1), 111.7 [C(CH₃)₂], 127.2–135.6 (Ph); HRMS(EI) m/z calcd for C₂₃H₃₁O₄Si (M-SPh) 399.1992, found 399.1996.

Methyl 2,3,6-tri-*O*-benzyl-4-*C*-(carboxymethyl)-4-deoxy- α -*D*-glucopyranoside 5.7.

The ethyl ester of **5.7**⁷⁶ (7.20 g, 10.0 mmol) was treated with a 1:1 mixture of aq 3N NaOH:ethanol (120 mL). After 4 hours most of the ethanol was removed under reduced pressure. The resulting mixture was acidified with concentrated HCl to pH 4-5 and extracted with ethyl acetate. The combined organic layer was concentrated *in vacuo*. The residue was purified by FCC to give unreacted starting material and acid **5.7** (3.63g, 86% based on recovered starting material) as a colorless oil; $R_f = 0.49$ (40% EtOAc/petroleum ether); IR (film) 2500–3200 (br), 1707 cm^{-1} ; ^1H NMR (300 MHz, CDCl_3) δ 2.03 (m, 2H, H-4, H-2'a), 2.57 (dd, $J = 5.5, 16.5$ Hz, 1H, H-2'b), 3.38 (s, 3H, OCH_3), 3.61, 3.83 (both m, 3H, 2H resp, H-2, 3, 5, 6a, 6b), 4.47 (d, $J = 11.7$ Hz, 1H, PhCH), 4.57–4.81 (m, 5H, H-1, 4 \times PhCH), 5.02 (d, $J = 11.0$ Hz, 1H, PhCH), 7.20–7.35 (m, 15H, Ph); ^{13}C NMR (125 MHz, CDCl_3) δ 32.3, 40.6 (C-4, 2'), 55.5 (OCH_3), 69.8, 70.4, 73.1, 73.6, 75.5, 78.1, 81.8 (7C, C-2, 3, 5, 6, PhCH_2), 98.7 (C-1), 127.8–138.7 (Ph), 176.5 (C = O); HRMS (FAB) m/z calcd for $\text{C}_{30}\text{H}_{34}\text{O}_7\text{Na}$ $[\text{M} + \text{Na}]^+$ 529.2201, found 529.2202.

Methyl 2,3,6-tri-*O*-benzyl-[(4-*C*-(4-*O*-*tert*-butyl)diphenylsilyl-1,2-*O*-isopropylidene-*D*-*erythro*-*S*-phenyl monothiohemiacetal)-2-ethanoate]-4-deoxy- α -*D*-glucopyranoside 5.5.

DCC (2.71 g, 13.15 mmol) was added at 0 $^\circ\text{C}$ to a solution of alcohol **5.6** (2.67 g, 5.26 mmol), acid **5.7** (3.63 g, 7.17 mmol), and DMAP (128 mg, 1.05 mmol) in anhydrous benzene (50 mL). The reaction was warmed to rt and stirred for 2 hours. The mixture was then diluted with ether (20 mL) and filtered. The filtrate was washed with 0.1 N aq

HCl and brine, dried (Na₂SO₄), filtered, and concentrated under reduced pressure. The residue was purified by FCC to give **5.5** (4.47 g, 85%) as a colorless oil; R_f = 0.42 (20% EtOAc/petroleum ether); IR (neat) 1734 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) δ 1.00 [s, 9H, (CH₃)₃CSi], 1.31, 1.42 [both s, 3H ea, C(CH₃)₂], 2.26 (m, 2H, H-4, 2'a), 2.55 (m, 1H, H-2'b), 3.10 (s, 3H, OCH₃), 3.33 (m, 1H, H-1), 3.52 (m, 2H, H-6a, 6b), 3.79 (m, 4H, H-3, 5, 4a, 4b), 4.31 (t, J = 6.0 Hz, 1H, H-2''), 4.39 (d, J = 12.0 Hz, 1H, PhCH), 4.51 (m, 3H, 3 × PhCH), 4.61 (d, J = 11.0 Hz, 1H, PhCH), 4.62 (d, J = 4.0 Hz, 1H, H-1), 4.89 (m, 1H, H-3'''), 4.91 (d, J = 11.0 Hz, 1H, PhCH), 5.37 (d, J = 6.5 Hz, 1H, H-1''), 7.10–7.70 (m, 30 H, Ph); ¹³C NMR (75 MHz, CDCl₃) δ 26.7 [C(CH₃)₂, 1C], 27.6 (CH₃)₃CSi, 28.1 [C(CH₃)₂, 1C], 33.5 (C-2'), 41.2 (C-4), 55.9 (OCH₃), 62.8 (C-4''), 70.6, 71.1, 73.5, 74.1, 74.4, 75.9, 78.9, 79.5, 82.5 (9C, C-2, 3, 5, 6, 2'', 3'', PhCH₂), 85.6 (C-1''), 99.1 (C-1), 112.2 [C(CH₃)₂], 127.8–139.3 (Ph), 171.6 (C=O); HRMS(FAB)m/z calcd for C₅₉H₆₈O₁₀SSiNa [M + Na]⁺ 1019.4203, found 1019.4200.

Methyl 2,3,6-tri-*O*-benzyl-4-*C*-[(4-*O*-*tert*-butyl diphenylsilyl-1,2-*O*-isopropylidene-*D*-*erythro*-*S*-phenyl monothiohemiacetal)-2-propene]-4-deoxy- α -*D*-glucopyranoside

5.4.

A solution of Tebbe reagent in THF (9.3 mL, 0.5 M, 4.7 mmol) was added dropwise, under an atmosphere of argon at -78 °C to ester **5.5** (1.82 g, 1.86 mmol) and pyridine (0.15 mL) in anhydrous 3:1 toluene/THF (20 mL). The reaction mixture was warmed to rt, stirred for 2 hours, and then slowly poured into a solution of 1 N aq NaOH at 0 °C. The resulting suspension was extracted with ether, and the organic phase was washed with brine, dried (Na₂SO₄), filtered, and concentrated under reduced pressure.

The residue was purified by FCC on basic alumina (Brockmann I, 150 mesh) to give enol ether **5.4** (1.09 g, 60%) as a colorless oil, $R_f = 0.61$ (15% EtOAc/petroleum ether); ^1H NMR (300 MHz, C_6D_6) δ 1.10 [s, 9H, $(\text{CH}_3)_3\text{CSi}$], 1.47, 1.62 [both s, 3H ea, $\text{C}(\text{CH}_3)_2$], 2.30 (dd, $J = 4.8, 14.5$ Hz, 1H, H-2'a), 2.40 (m, 1H, H-4), 2.69 (dd, $J = 3.6, 14.5$ Hz, 1H, H-2'b), 3.16 (s, 3H, OCH_3), 3.54 (dd, $J = 3.5, 9.7$ Hz, 1H, H-2), 3.75–4.15, 4.28 (both m, 8H, 1H, resp, H-3, 5, 6a, 6b. 3'', 4''a, 4''b, = CH_2), 4.30–4.65 (m, 4H, $2 \times \text{PhCH}_2$), 4.74 (m, 2H, H-1, PhCH), 4.88 (bt, $J = 5.0$ Hz, 1H, H-2''), 5.19 (d, $J = 11.2$ Hz, 1H, PhCH), 6.10 (d, $J = 6.2$ Hz, 1H, H-1''), 6.94–7.84 (m, 30H, Ph); ^{13}C NMR (75 MHz, C_6D_6) δ 26.8 [$\text{C}(\text{CH}_3)_2$, 1C], 27.6 $(\text{CH}_3)_3\text{CSi}$], 28.2 [$\text{C}(\text{CH}_3)_2$, 1C], 33.8 (C-2'), 42.1 (C-4), 55.2 (OCH_3), 62.3 (C-4''), 71.5, 71.9, 72.7, 74.1, 75.2, 77.4, 78.6, 81.4, 83.8, 85.7 (10C, C-2, 3, 5, 6, 2'', 3'', = CH_2 , Ph CH_2) 86.1 (C-1''), 98.7 (C-1), 112.4 [$\text{C}(\text{CH}_3)_2$], 127.5–140.7 (Ph), 160.1 ($\text{OC} = \text{CH}_2$). HRMS (FAB) m/z calcd for $\text{C}_{60}\text{H}_{71}\text{O}_9\text{SSiH}$ [$\text{M} + \text{H}$] $^+$ 995.4587, found 995.4588.

Methyl 4-C-(2,6-anhydro-7-O-tert-butylidiphenylsilyl-4,5-O-isopropylidene-1,3-dideoxy-D-ribo-hept-2-enitol-1-C-yl)-2,3,6-tri-O-benzyl-4-deoxy- α -D-glucopyranoside 5.3.

A mixture of enol ether **5.4** (740 mg, 0.08 mmol), 2,6-di-tert-butyl-4-methylpyridine (1.54 g, 7.51 mmol), and freshly activated, powdered 4 Å molecular sieves (300 mg) in anhydrous CH_2Cl_2 (25 mL) was stirred for 15 min at rt under an argon atmosphere, then cooled to 0 °C. Methyl triflate (1.0 mL, 7.55 mmol) was introduced, and the mixture was warmed to rt and stirred for an additional 18 hours, or until all the starting material had disappeared, at which time triethylamine (2.5 mL) was added. The

mixture was diluted with ether, washed with satd aq NaHCO₃ and brine, dried (Na₂SO₄), filtered and concentrated under reduced pressure. The residue was purified by FCC over basic alumina (Brockmann I, 150 mesh) to give **5.3** (460 mg, 70%) as a clear oil, R_f = 0.60 (20% EtOAc/petroleum ether); IR (film) 1647 cm⁻¹; ¹H NMR (300 MHz, C₆D₆) δ 1.24 [s, 9H, (CH₃)₃CSi], 1.31, 1.37 [both s, 3H ea, C(CH₃)₂], 2.40 (dd, J = 5.7, 14.4 Hz, 1H, H-1'a), 2.56 (m, 2H, H1'b, 4), 3.24 (s, 3H, OCH₃), 3.70 (m, 2H, H-2, 7'a), 3.85–4.00–4.20 (both m, 2H, 5H resp., H-3, 5, 6a, 6b, 3', 6', 7'b), 4.27 (t, J = 5.0 Hz, 1H, H-5'), 4.60 (m, 4H, 2 × PhCH₂), 4.83 (m, 2H, H-1, PhCH), 4.93 (d, J = 5.6 Hz, 1H, H-4'), 5.27 (d, J = 11.7 Hz, 1H, PhCH), 7.11–7.46 and 7.90 (both m, 25H, Ph); ¹³C NMR (75 MHz, C₆D₆) δ 26.8 [C(CH₃)₂, 1C], 27.7 (CH₃)₃CSi], 28.2 [C(CH₃)₂, 1C], 33.9 (C-1'), 42.2 (C-4), 55.3 (OCH₃), 62.4 (C-7'), 71.6, 72.0, 72.8, 74.2, 75.3, 77.5, 78.7, 81.6, 83.9, 85.7, 86.2 (11C, C2, 3, 5, 6, 3', 4', 5', 6', 3 × PhCH₂), 100.0 (C-1), 112.5 [C(CH₃)₂], 128.0–140.8 (Ph), 160.2 (C-2'); HRMS (FAB) m/z calcd for C₅₄H₆₅O₉Si [M + H]⁺ 885.4399, found 885.4398.

Methyl 4-C-(2,6-anhydro-4,5-O-isopropylidene-7-O-tert-butylidiphenylsilyl-1-deoxy-D-glycero-D-allo-heptitol-1-C-yl)-2,3,6-tri-O-benzyl-4-deoxy-α-D-glucopyranoside (5.11) and Methyl 4-C-(2,6-anhydro-4,5-O-isopropylidene-7-O-tert-butylidiphenylsilyl-1-deoxy-D-glycero-D-manno-heptitol-1-C-yl)-2,3,6-tri-O-benzyl-4-deoxy-α-D-glucopyranoside (5.12).

BH₃.Me₂S (2.7 mL of 1 M solution in CH₂Cl₂, 2.7 mmol) was added at 0 °C to a solution of the glycal **5.3** (340 mg, 0.39 mmol) in anhydrous THF (15 mL) under an atmosphere of argon. The mixture was warmed to rt and stirred for an additional 1 hour.

At that time the solution was cooled to 0 °C and treated with a mixture of 3N NaOH (2.5 mL) and 30% aqueous H₂O₂ (2.5 mL) for 30 min. The mixture was diluted with ether, washed with satd aq NaHCO₃ and brine, dried (Na₂SO₄), filtered and concentrated under reduced pressure. FCC of the residue afforded a mixture of **5.11** and **5.12** (264 mg, ca. ratio 1:5, 84%) which had identical TLC mobilities (R_f = 0.25, 10% EtOAc/petroleum ether). For characterization purposes, a pure sample of **5.12** was obtained by resubjecting this material to a second chromatography and collection of a later eluting fraction. Compound **5.12** was isolated as a colorless oil; ¹H NMR (300 MHz, CDCl₃) δ 1.07 [s, 9H, (CH₃)₃CSi], 1.29, 1.46 [both s, 3H ea, C(CH₃)₂], 1.51 (m, 1H, H-1'a), 1.75 (m, 1H, H-1'b), 2.05 (m, 1H, H-4), 3.00 (bs, 1H, D₂O ex), 3.32 (m, 1H), 3.37 (s, 3H), 3.41 (m, 1H), 3.55 (m, 2H), 3.60–3.95 (m, 7H), 4.12 (m, 1H), 4.44 (s, 2H), 4.65–4.85 (m, 4H), 5.03 (d, J = 12.0 Hz, 1H), 7.26–7.78 (m, 25H); ¹³C NMR (75 MHz, CDCl₃) δ 27.7 (CH₃)₃CSi, C(CH₃)₂, 1C), 28.8 [C(CH₃)₂, 1C], 33.1 (C-1'), 40.5 (C-4), 55.9 (OCH₃), 65.4 (C-7'), 71.4, 72.4, 73.5, 74.0, 74.6, 75.5, 76.6, 78.8, 81.2, 82.6 (12C, C-2, 3, 5, 6, 2', 3', 4', 5' 6', 3 × PhCH₂), 99.0 (C-1), 109.5 [C(CH₃)₂], 128.1–139.1 (Ph); HRMS (FAB) m/z calcd for C₅₄H₆₆O₁₀SiNa [M + Na]⁺ 925.4321, found 925.4323. Compound **5.11** was characterized as triol **5.13** and triacetate **5.17** (*vide infra*).

Methyl 4-C-(2,6-anhydro-7-O-tert-butylidiphenylsilyl-1-deoxy-D-glycero-D-*allo*-heptitol-1-C-yl)-2,3,6-tri-O-benzyl-4-deoxy-α-D-glucopyranoside 5.13 and Methyl 4-C-(2,6-anhydro-7-O-tert-butylidiphenylsilyl-1-deoxy-D-glycero-D-*manno*-heptitol-1-C-yl)-2,3,6-tri-O-benzyl-4-deoxy-α-D-glucopyranoside 5.14.

To a portion of the mixture of **5.11** and **5.12** from the previous step (103 mg, 0.11 mmol) in dry methanol was added a 1 M solution of HCl in ether (0.7 mL). The reaction was stirred for 1.5 hours and then neutralized with a solution of NaOMe in methanol. Removal of the volatiles followed by FCC of the residue provided triols **5.13** (12 mg, 12%) and **5.14** (65 mg, 66%).

Compound **5.13** was isolated as a clear gum; $R_f = 0.64$ (50% EtOAc/petroleum ether); ^1H NMR (500 MHz, C_6D_6) δ 1.16 [s, 9H, $(\text{CH}_3)_3\text{CSi}$], 1.68 (m, 1H, H-1'a), 2.20 (m, 1H, H-1'b), 2.42 (m, 1H, H-4), 2.44 (bs, 1H, D_2O ex, OH), 2.58 (d, $J = 7.5$ Hz, 1H, D_2O ex, OH), 2.93 (d, $J = 5.5$ Hz, 1H, D_2O ex, OH), 3.07 (t, $J = 7.3$ Hz, 1H), 3.18 (s, 3H, OCH_3), 3.60 (m, 3H), 3.80 (m, 3H), 3.88–4.00 (m, 4H), 4.04 (dd, $J = 4.5, 11.0$ Hz, 1H), 4.42 (ABq, $J = 12.0$ Hz, $\Delta\delta = 0.05$ ppm, 2H, PhCH_2), 4.48 (ABq, $\Delta\delta = 0.04$ ppm, $J = 12.0$ Hz, 2H, PhCH_2), 4.76 (d, $J = 3.0$ Hz, 1H, H-1), 4.89 (ABq, $\Delta\delta = 0.44$ ppm, $J = 11.0$ Hz, 2H, PhCH_2), 7.05–7.82 (m, 25H); HRMS (FAB) m/z calcd for $\text{C}_{51}\text{H}_{62}\text{O}_{10}\text{NaSi}$ [$\text{M} + \text{Na}$] $^+$ 885.4010. Found 885.4030.

Compound **5.14** was isolated as a clear gum; $R_f = 0.50$ (50% EtOAc/petroleum ether); ^1H NMR (300 MHz, C_6D_6) δ 1.18 [s, 9H, $(\text{CH}_3)_3\text{CSi}$], 1.82 (m, 1H, H-1'a), 1.98 (m, 1H, H-1'b), 2.32 (m, 1H, H-4), 2.75 (s, 1H, D_2O ex, OH), 2.92 (s, 1H, D_2O ex, OH), 3.22 (s, 3H, OCH_3), 3.51 (s, 1H, D_2O ex, OH), 3.52 (dd, $J = 3.3, 9.9$ Hz, partially hidden by s at δ 3.51, 1H), 3.60 (dd, $J = 2.6, 8.4$ Hz, 1H), 3.72 (m, 2H), 3.82 (m, 4H), 3.93 (dd, $J = 6.3, 9.3$ Hz, 1H), 4.01 (t, $J = 9.6$ Hz, 1H), 4.09 (bs, 1H), 4.26 (t, $J = 6.2$ Hz, 1H), 4.44 (ABq, $\Delta\delta = 0.06$ ppm, $J = 11.7$ Hz, 1H), 4.49 (ABq, $\Delta\delta = 0.05$ ppm, $J = 12.4$ Hz, 1H), 4.74 (d, $J = 3.3$ Hz, 1H, H-1), 4.87 (ABq, $\Delta\delta = 0.41$ ppm, $J = 10.6$ Hz, 1H), 7.10–7.76 (m, 25H); ^{13}C NMR (75 MHz, CDCl_3) δ 27.2 ($(\text{CH}_3)_3\text{CSi}$), 33.4 (C-1'), 40.0 (C-4), 55.4 (OCH_3),

62.9 (C-7'), 69.0, 69.1, 70.8, 72.0, 72.9, 73.6, 73.7, 73.8, 74.4, 76.5, 81.3, 82.2, (12C, C-2, 3, 5, 6, 2', 3', 4' 5' 6', 3 × PhCH₂), 98.4 (C-1), 127.6–138.4 (Ph). HRMS (FAB) m/z calcd for C₅₁H₆₂O₁₀NaSi [M + Na]⁺ 885.4010, found 885.4013.

Methyl 4-C-(2,6-anhydro-1-deoxy-D-glycero-D-manno-heptitol-1-C-yl)-2,3,6-tri-O-benzyl-4-deoxy- α -D-glucopyranoside 5.15.

A solution of HCl in ether (0.3 mL of ca.1M) was added to **5.12** (419 mg, 0.49 mmol) in dry methanol (30 mL). The reaction was stirred at rt for approximately 4 hours, then neutralized with a solution of NaOMe in methanol. Addition of Bu₄NF (2.8 mL of a 1 M solution in THF, 2.8 mmol) to the reaction mixture, followed by stirring for an additional 1 hour, then removal of the volatiles under reduced pressure, and FCC of the residue, provided **5.15** (164 mg, 58%) as a colorless oil; R_f = 0.37, (20% acetone/EtOAc); ¹H NMR (300 MHz, CDCl₃) δ 1.62 (m, 1H, H-1'a), 1.77 (m, 1H, H-1'b), 2.07 (m, 1H, H-4), 2.86 (bs, 1H, D₂O ex, OH), 3.29 (m, 2H), 3.38 (s, 3H, OCH₃), 3.43–3.67 (m, 8H), 3.85 (m, 2H), 4.50–4.76 (m, 6H, H-1, 5 × PhCH), 5.03 (d, J = 10.6 Hz, 1H, PhCH), 7.24–7.37 (m, 15H, Ph); ¹³C NMR (75 MHz, CDCl₃) δ 32.3 (C-1'), 39.4 (C-4), 55.5 (OCH₃), 59.4 (C-7'), 68.6, 70.6, 71.5, 72.3, 72.5, 73.0, 73.4, 73.6, 76.3, 77.5, 81.6, 82.0 (12C, C-2, 3, 5, 6, 2', 3', 4', 5', 6', 3 × PhCH₂), 98.3 (C-1), 127.7–138.2 (Ph); HRMS (FAB) m/z calcd for C₃₅H₄₄O₁₀Na [M + Na]⁺ 647.2832, found 647.2834.

Methyl 4-C-(2,6-anhydro-3,4,5,7-tetra-O-acetyl-1-deoxy-D-glycero-D-mannoheptitol-1-C-yl)-2,3,6-tri-O-benzyl-4-deoxy- α -D-glucopyranoside 5.16.

A solution of tetraol **5.15** (25.5 mg, 0.04 mmol), DMAP (1.00 mg, 4 mmol), and acetic anhydride (0.04 mL, 0.40 mmol) in ethyl acetate (3.0 mL), was stirred at rt for 20 min. Methanol was then added and the reaction mixture concentrated under reduced pressure. The residue was purified by FCC to give **5.16** (26.7 mg, 82%) as a colorless oil; $R_f = 0.66$ (40% EtOAc/petroleum ether); $[\alpha]_D + 23^\circ$ (c 1.3 CHCl₃); IR (film) 1743 cm⁻¹; ¹H NMR (500 MHz, C₆D₆) δ 1.59 (s, 3H, CH₃CO), 1.64, (m, buried under s at d 1.64, 1H, H-1'a) 1.64, 1.69, 1.78 (all s, 3H ea, 3 \times CH₃CO), 2.12 (t, J = 10.5 Hz, 1H, H-1'b), 2.23 (apparent q, J = 7.5 Hz, 1H, H-4), 3.25 (s, 3H, OCH₃), 3.56 (bd, J = 9.0 Hz, 1H, H-2), 3.62 (m, 1H, H-6a), 3.70 (bd, J = 10.5 Hz, 1H, H-6b), 3.81 (bd, J = 10.5 Hz, 1H, H-5), 3.89 (bd, J = 11.0 Hz, 1H, H-7'a), 4.06 (m, 2H, H-3, 6'), 4.28 (m, 2H, H-2', 7'b), 4.43 (m, 4H, 2 \times PhCH₂), 4.74 (bs, 1H, H-1), 4.94 (ABq, $\Delta\delta = 0.49$ ppm, J = 10.5 Hz, 2H, PhCH₂), 5.33 (t, J = 8.5 Hz, 1H, H-3'), 5.43 (dd, J = 2.5, 8.5 Hz, 1H, H-4'), 5.49 (d, J = 2.5 Hz, 1H, H-5'), 7.10-7.51 (m, 15H, Ph); ¹³C NMR (75 MHz, C₆D₆) δ 20.7, 20.8, 20.9 (CH₃CO, 4C), 31.8 (C-1'), 41.1 (C-4), 55.5 (OCH₃), 62.3, 69.5, 70.6, 71.7, 72.0, 72.3, 72.9, 73.3 (2C), 74.2, 75.8, 81.4, 83.5 (13C, C-2, 3, 5, 6, 2', 3', 4', 5', 6', 7', 3 \times PhCH₂), 99.0 (C-1) 128.0-130.0 (Ph), 169.6, 169.7, 169.8, 170.1 (CH₃CO). HRMS (FAB) m/z calcd for C₄₃H₅₂O₁₄Na [M + Na]⁺ 815.8355, found 815.3256.

Methyl 4-C-(2,6-anhydro-7-O-tert-butyldiphenylsilyl-3,4,5-tri-O-acetyl-1-deoxy-D-glycero-D-allo-heptitol-1-C-yl)-2,3,6-tri-O-benzyl-4-deoxy- α -D-glucopyranoside 5.17.

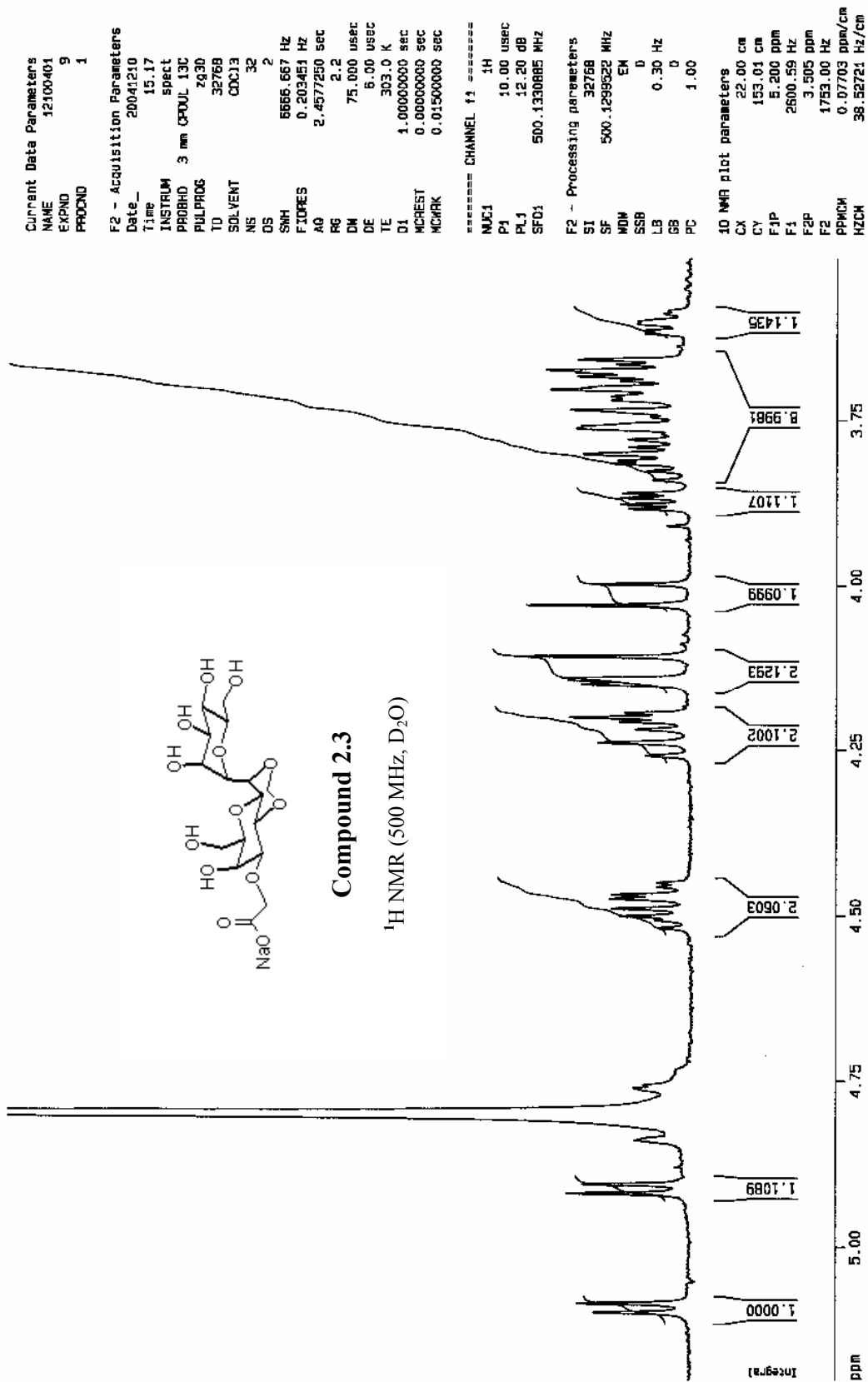
Triol **5.13** (4.0 mg, 0.005 mmol) was subjected to the identical acetylation procedure that was described for the preparation of **5.16**. FCC of the crude reaction product provided **5.13** (4.2 mg, 91%) as a clear gum; $R_f = 0.50$ (30% EtOAc/petroleum ether); $^1\text{H NMR}$ (500 MHz, C_6D_6) δ 1.72 [s, 9H, $(\text{CH}_3)_3\text{CSi}$], 1.50, 1.63, 1.64 (all s, 3H ea, CH_3CO), 1.75 (m, 1H, H-1'a), 2.15 (bdd, $J = 4.5, 14.0$ Hz, 1H, H-1'b), 2.62 (m, 1H, H-4), 3.16 (s, 3H, OCH_3), 3.70 (dd, $J = 3.0, 8.8$ Hz, 1H, H-2), 3.76 (dd, $J = 3.5, 11.5$ Hz, 1H, H-7'a), 3.84 (m, 2H, H-6a, 6'), 3.88 (bd, $J = 11.5$ Hz, 1H, H-7'b), 3.98 (dd, $J = 3.5, 11.0$ Hz, 1H, H-6b), 4.12, (t, $J = 10.5$ Hz, 1H, H-3), 4.23 (m, 2H, H-5, H-2'), 4.44 (ABq, $\Delta\delta = 0.03$ ppm, $J = 12.0$ Hz, 2H, PhCH_2), 4.54 (ABq, $\Delta\delta = 0.07$ ppm, $J = 12.0$ Hz, 2H, PhCH_2), 4.81, (d, $J = 3.0$ Hz, 1H, H-1), 4.90 (dd, $J = 2.0, 9.5$ Hz, 1H, H-3'), 4.99 (ABq, $\Delta\delta = 0.55$ ppm, $J = 11.5$ Hz, 2H, PhCH_2), 5.43 (dd, $J = 2.0, 9.9$ Hz, 1H, H-5'), 6.20 (bt, $J = 2.0$ Hz, 1H, H-4'), 7.20, 7.80 (both m, 25H, Ph); $^{13}\text{C NMR}$ (75 MHz, CDCl_3) δ 20.7, 20.8, 20.9 (CH_3CO), 27.2, 27.6 ($(\text{CH}_3)_3\text{CSi}$), 30.0 (C-1'), 40.4 (C-4), 55.3 (OCH_3), 63.1, 64.5, 66.8, 68.8, 70.3, 70.7, 71.0, 71.8, 73.1, 73.8, 74.7, 75.0, 82.5 (13C, C-2, 3, 5, 6, 2', 3', 4', 5', 6', 7', $3 \times \text{PhCH}_2$), 98.6 (C-1), 127.4-139.3 (Ph), 168.9, 169.3, 169.9 (CH_3CO). ESMS m/z 1006.5 $[\text{M} + \text{NH}_4]^+$.

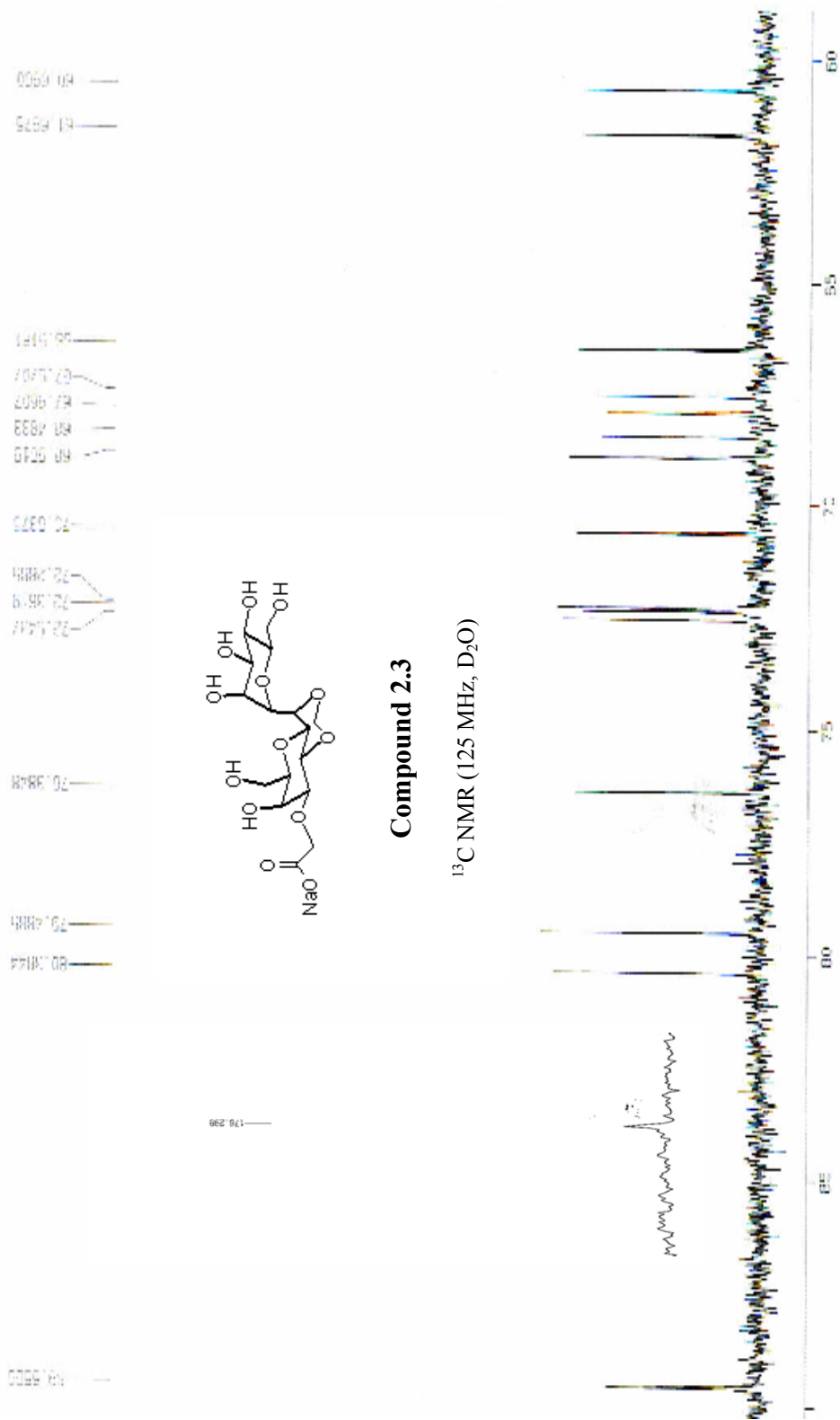
Methyl 4-C-(2,6-anhydro-1-deoxy-D-glycero-D-manno-heptitol-1-C-yl)-4-deoxy- α -D-glucopyranoside 5.2.

A mixture of **5.15** (124 mg, 0.20 mmol), 20% Pd on carbon (400 mg), formic acid (0.35 mL) and methanol (7.0 mL) was stirred under an atmosphere of hydrogen (balloon),

for 12 hours. The reaction mixture was purged with argon and filtered through a bed of Celite. The filtrate was concentrated *in vacuo*, and the residue was purified by FCC to provide **5.2** (55.0 mg, 83%) as a white solid; $R_f = 0.68$ (50% MeOH/CHCl₃); $[\alpha]_D + 93$ (c 0.54 MeOH); ¹H NMR (500 MHz, D₂O, 25 °C) δ 1.71 (ddd, $J = 2.8, 10.1, 15.1$ Hz, 1 H, H-1'a), 1.83 (m, 1H, H-4), 2.08 (ddd, $J = 2.0, 6.0, 15.1$ Hz, 1H, H-1'b), 3.42 (s, 3H, OCH₃), 3.50 (apparent t, $J = 9.0$ Hz, 1H, H-7'a), 3.64 (dd, $J = 3.7, 9.6$ Hz, 1H, H-2), 3.66–3.76 (m, 4H, H-5, 2', 3', 6'), 3.82 (m, 3H, H-4', 7'a, 7'b), 3.93 (bdd, $J = 9.1, 12.0$ Hz, H-6a), 3.99 (m, 2H, H-6b, 5'), 4.85 (d, $J = 3.7$ Hz, H-1); ¹³C NMR (125 MHz, D₂O) δ 29.9 (C-1'), 39.0 (C-4), 54.9 (OCH₃), 58.0, 61.5 (C-6, 7'), 68.5, 70.9, 71.4, 71.7, 72.4 (two carbons), 72.9, 78.4 (8C, C-2, 3, 5, 2', 3', 4', 5', 6'), 99.5 (C-1); HRMS (FAB) m/z calcd for C₁₄H₂₇O₁₀Na [M + Na]⁺ 355.1604, found 355.1602.

Appendix





Current Data Parameters
 NAME 01040501
 EXPNO 9
 PROCNO 1

F2 - Acquisition Parameters
 Date_ 20050104
 Time 16.13
 INSTRUM spect
 PROBHD 3 mm CPDUL 13C
 PULPROG zg30

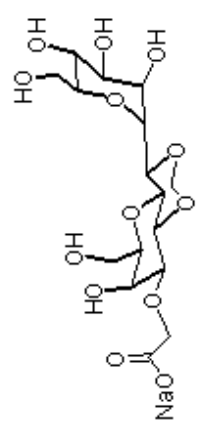
SOLVENT D2O
 NS 128
 DS 2
 SMH 6666.667 Hz
 FIDRES 0.203451 Hz
 AQ 2.4577250 sec

RG 2
 DM 75.000 usec
 DE 6.00 usec
 TE 300.5 K
 D1 1.00000000 sec
 MCREST 0.00000000 sec
 MCMRK 0.01500000 sec

***** CHANNEL f1 *****
 NUC1 1H
 P1 10.00 usec
 PL1 12.20 dB
 SFO1 500.1330685 MHz

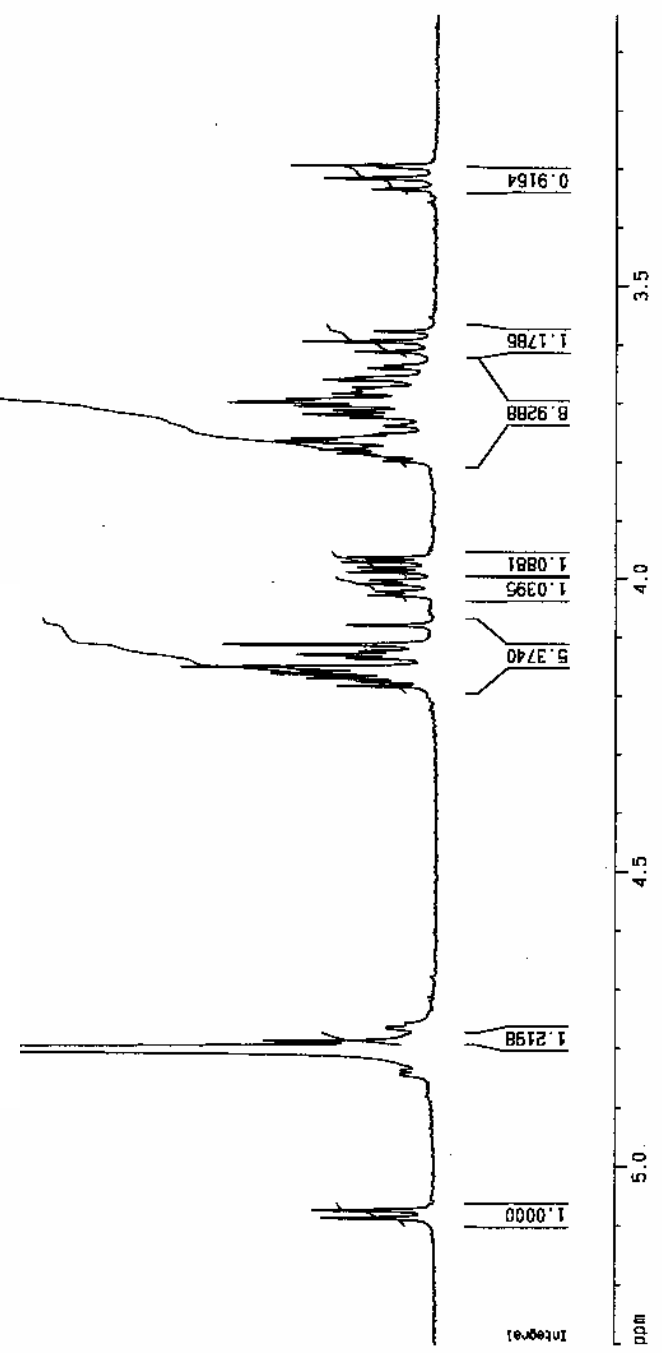
F2 - Processing parameters
 SI 32768
 SF 500.1289512 MHz
 MDW EM
 SSB 0
 LB 0.30 Hz
 GB 0
 PC 1.00

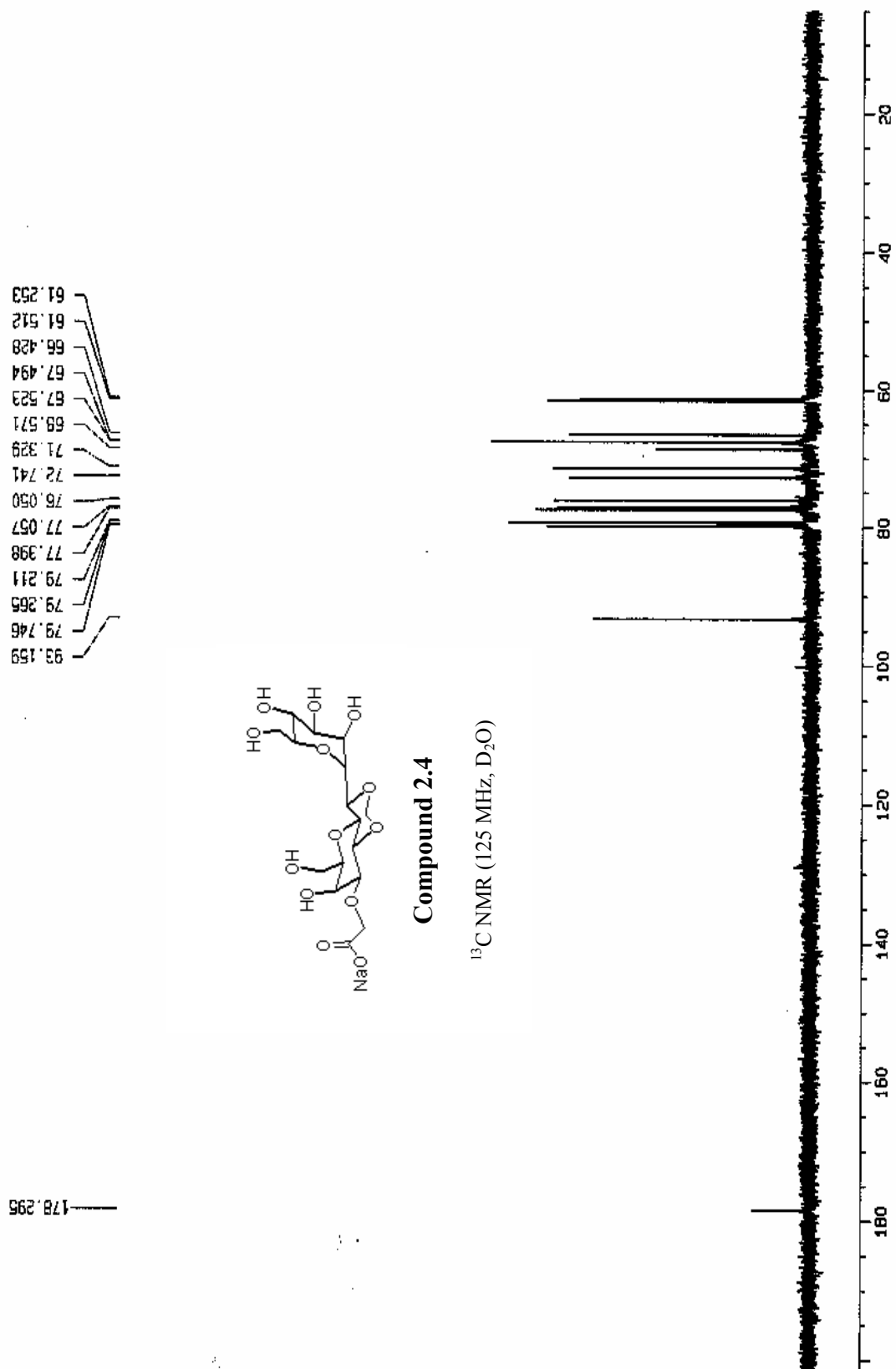
1D NMR plot parameters
 CX 22.00 cm
 CY 91.67 cm
 F1P 5.302 ppm
 F1 2661.88 Hz
 F2P 3.038 ppm
 F2 1519.19 Hz
 PPMCM 0.10284 ppm/cm
 MZCM 51.48573 Hz/cm



Compound 2.4

¹H NMR (500 MHz, D₂O)





Current Data Parameters
 NAME 01110601
 EXPNO 32
 PROCNO 1

F2 - Acquisition Parameters

Date_ 20060111
 Time 19.36
 INSTRUM spect
 PROBHD 5 mm DUL 13C-1
 PULPROG zg30
 TO 32768
 SOLVENT CDCl3
 NS 27
 DS 2
 SMH 8012.820 Hz
 FIDRES 0.244532 Hz
 AQ 2.0446355 sec
 RG 287.4
 DM 62.400 usec
 DE 6.00 usec
 TE 303.0 K
 O1 1.00000000 sec
 ACREST 0.00000000 sec
 MCNRK 0.01500000 sec

***** CHANNEL f1 *****

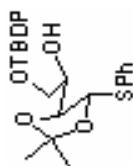
NUC1 1H
 P1 9.30 usec
 PL1 -3.00 dB
 SF01 500.1330685 MHz

F2 - Processing parameters

SI 32768
 SF 500.1300000 MHz
 NCH EN
 SSB 0
 LB 0.30 Hz
 GB 0
 PC 1.00

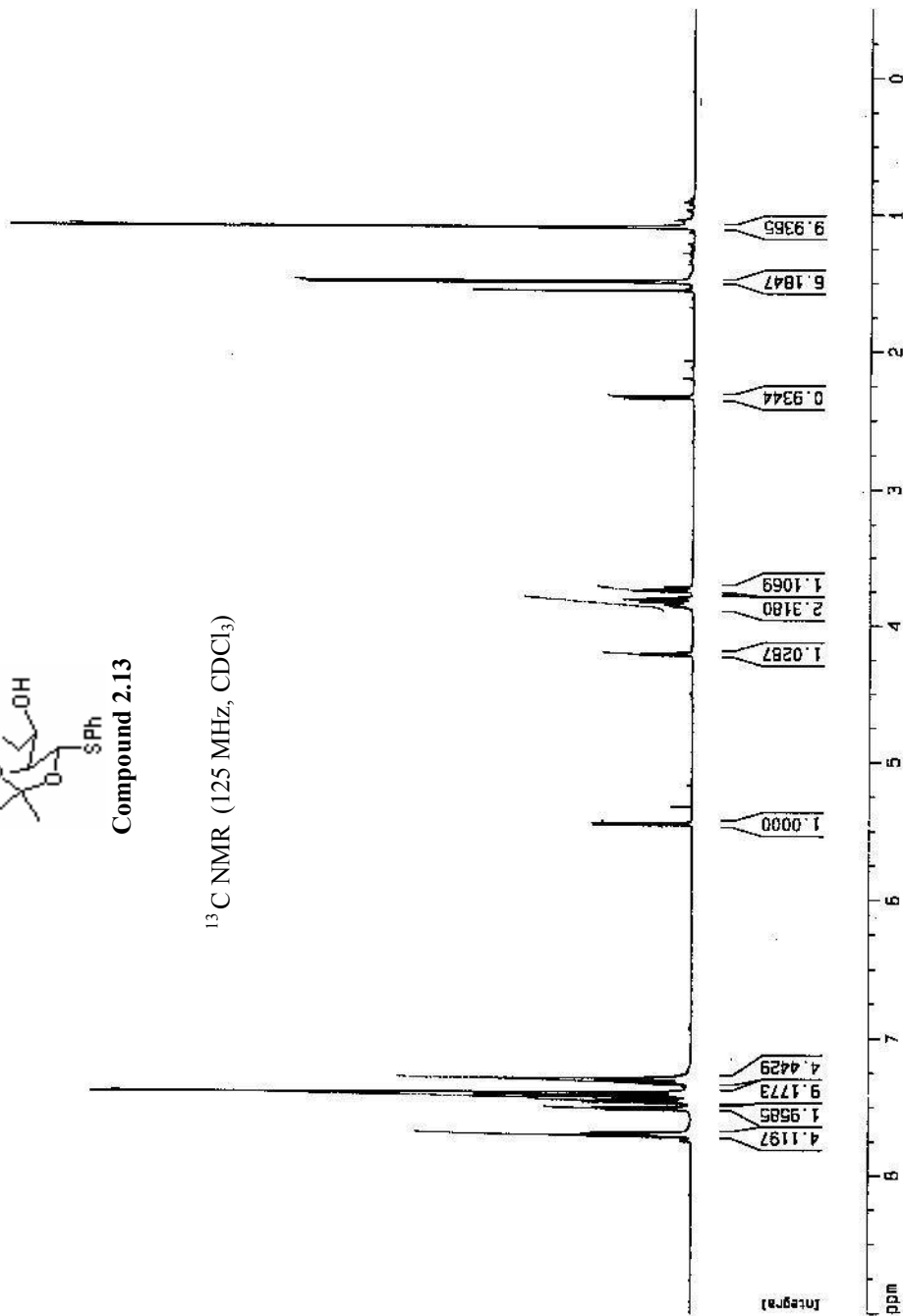
1D NMR plot parameters

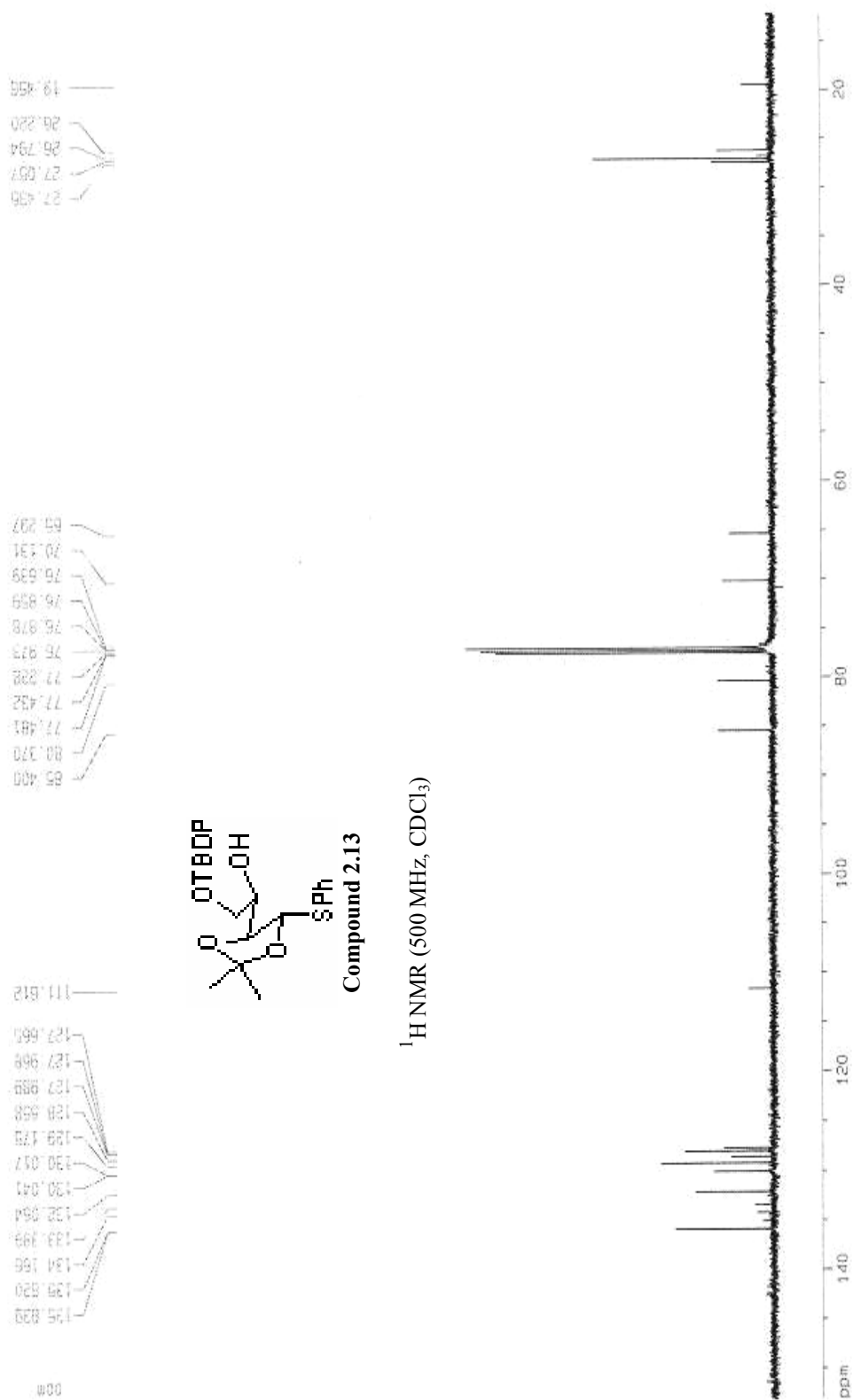
CK 22.00 cm
 CY 25.65 cm
 F1P 9.000 ppm
 F1 4501.17 Hz
 F2P -0.500 ppm
 F2 -250.07 Hz
 PPHCM 0.43182 ppm/c
 HZCM 215.96523 Hz/cm

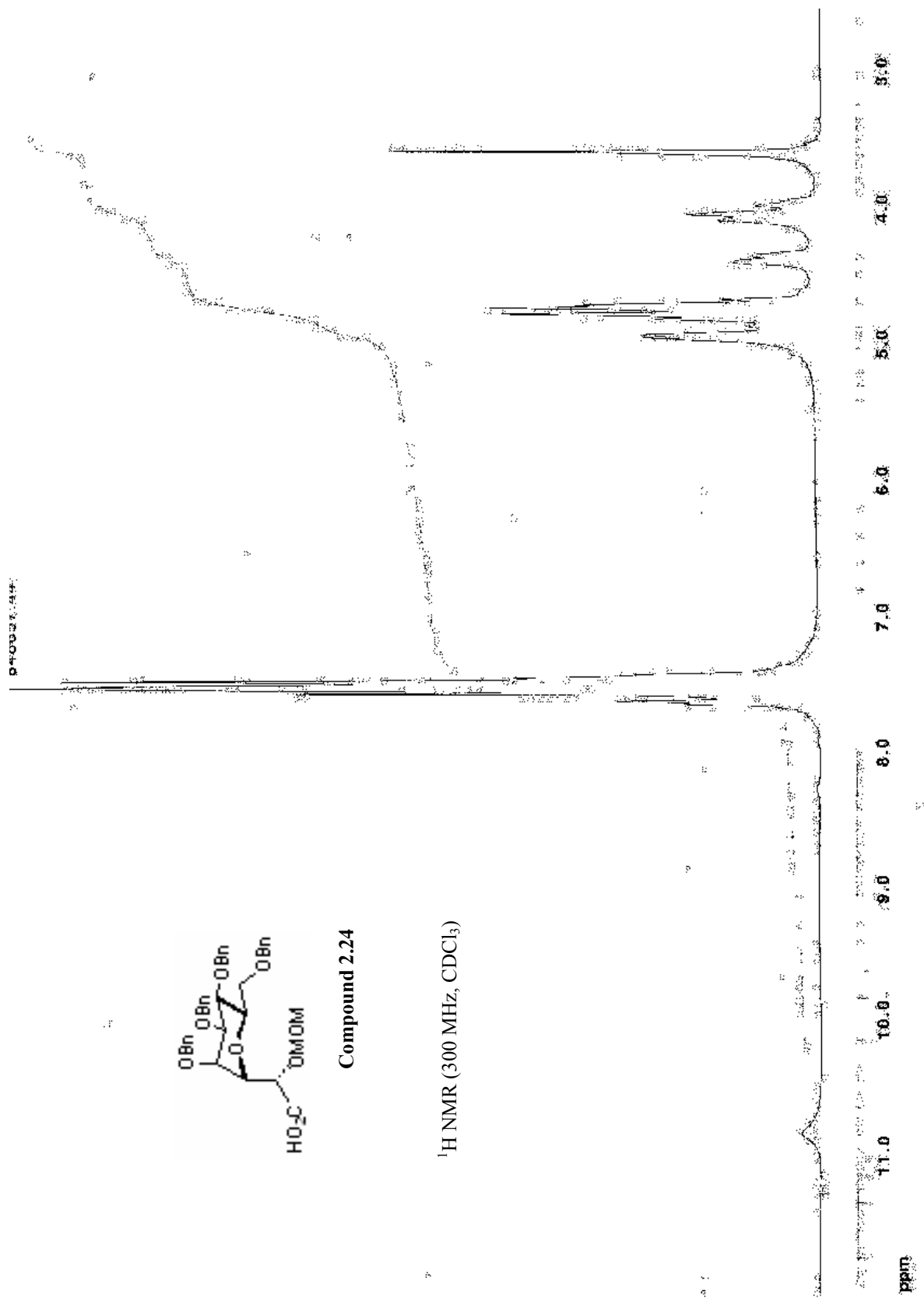


Compound 2.13

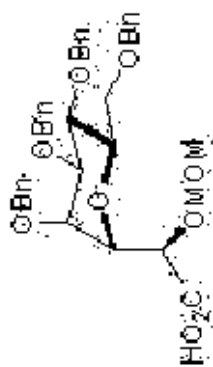
¹³C NMR (125 MHz, CDCl₃)





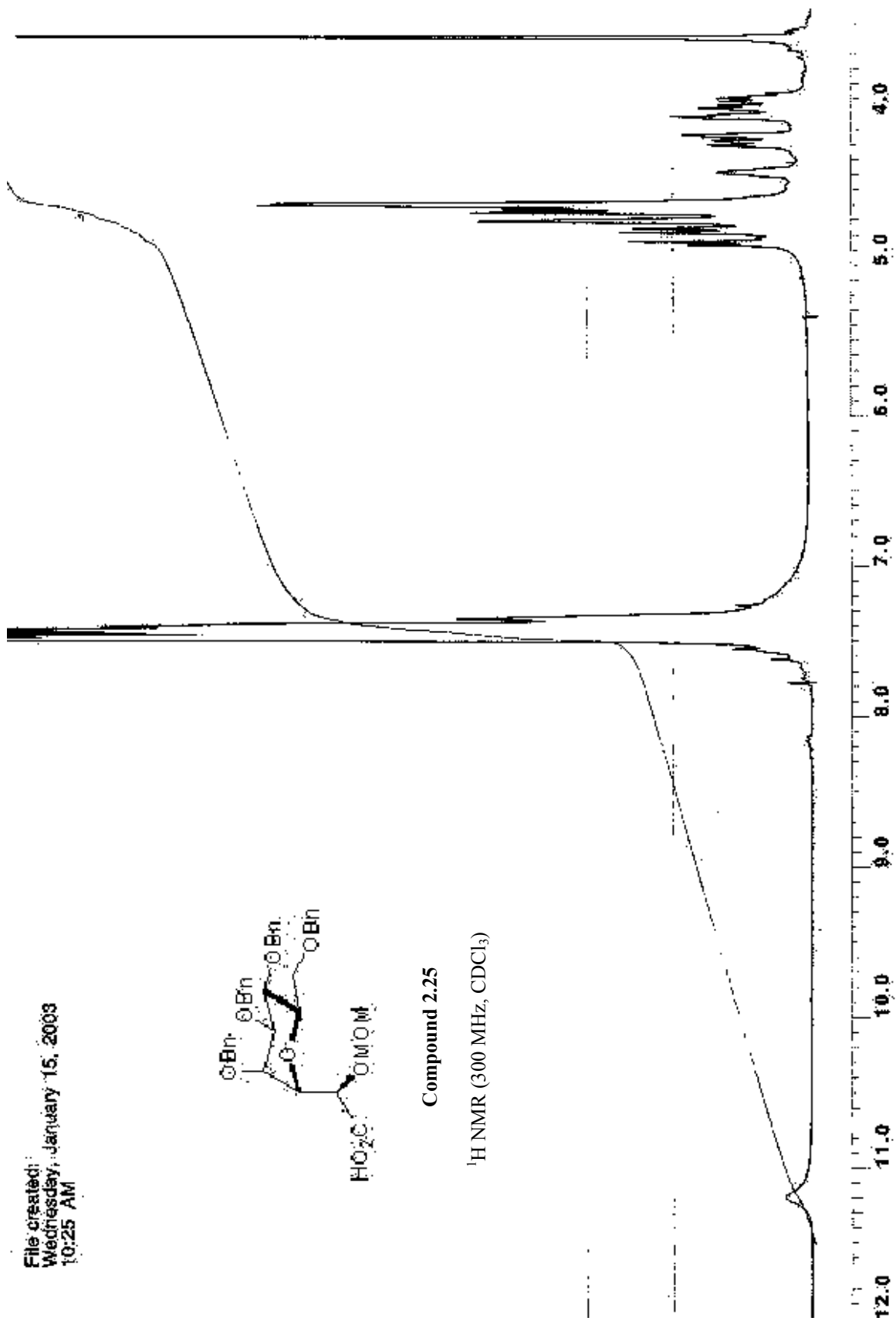


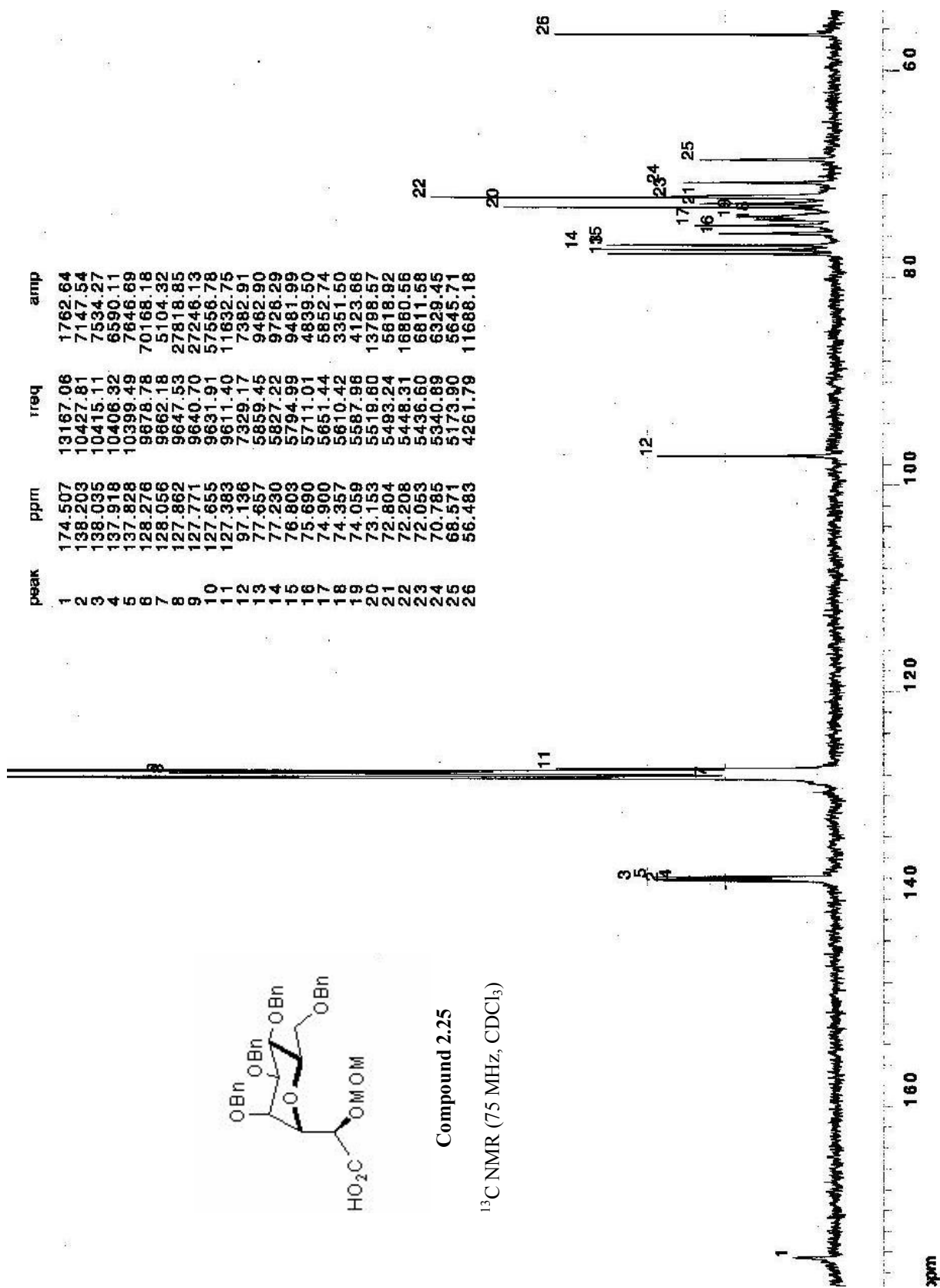
File created:
Wednesday, January 15, 2003
10:25 AM

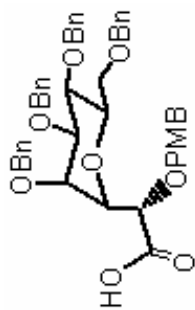


Compound 2.25

^1H NMR (300 MHz, CDCl_3)

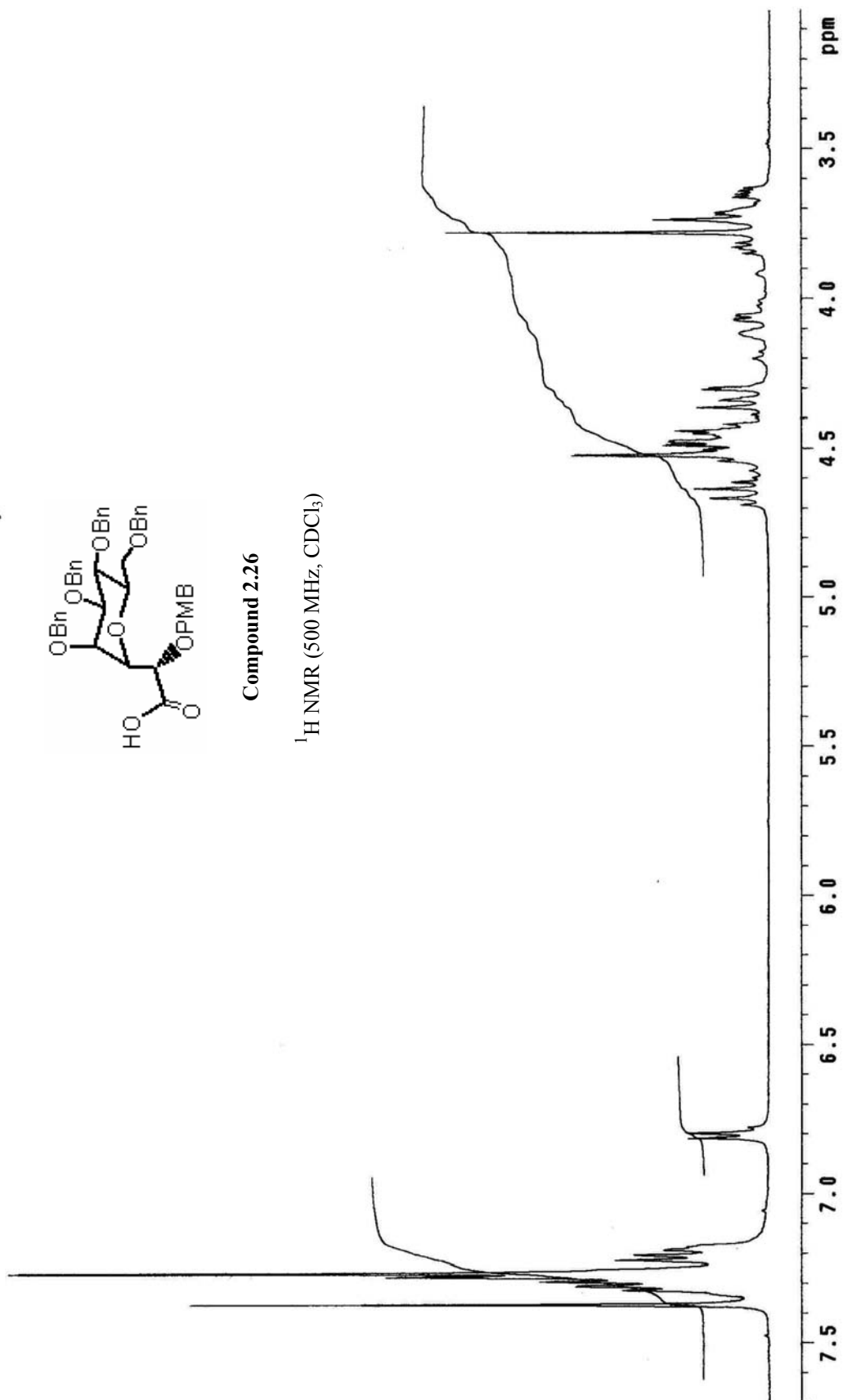


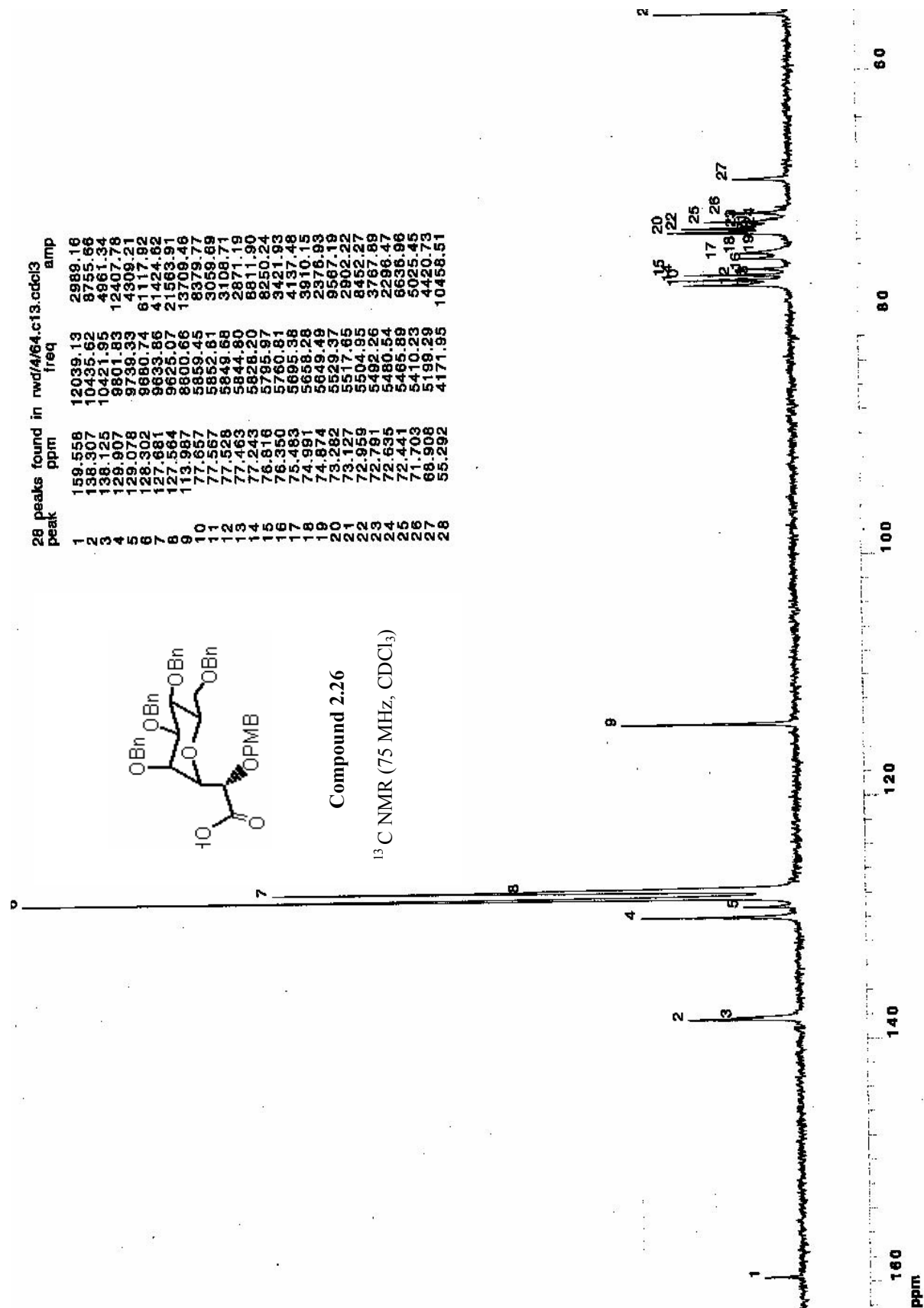


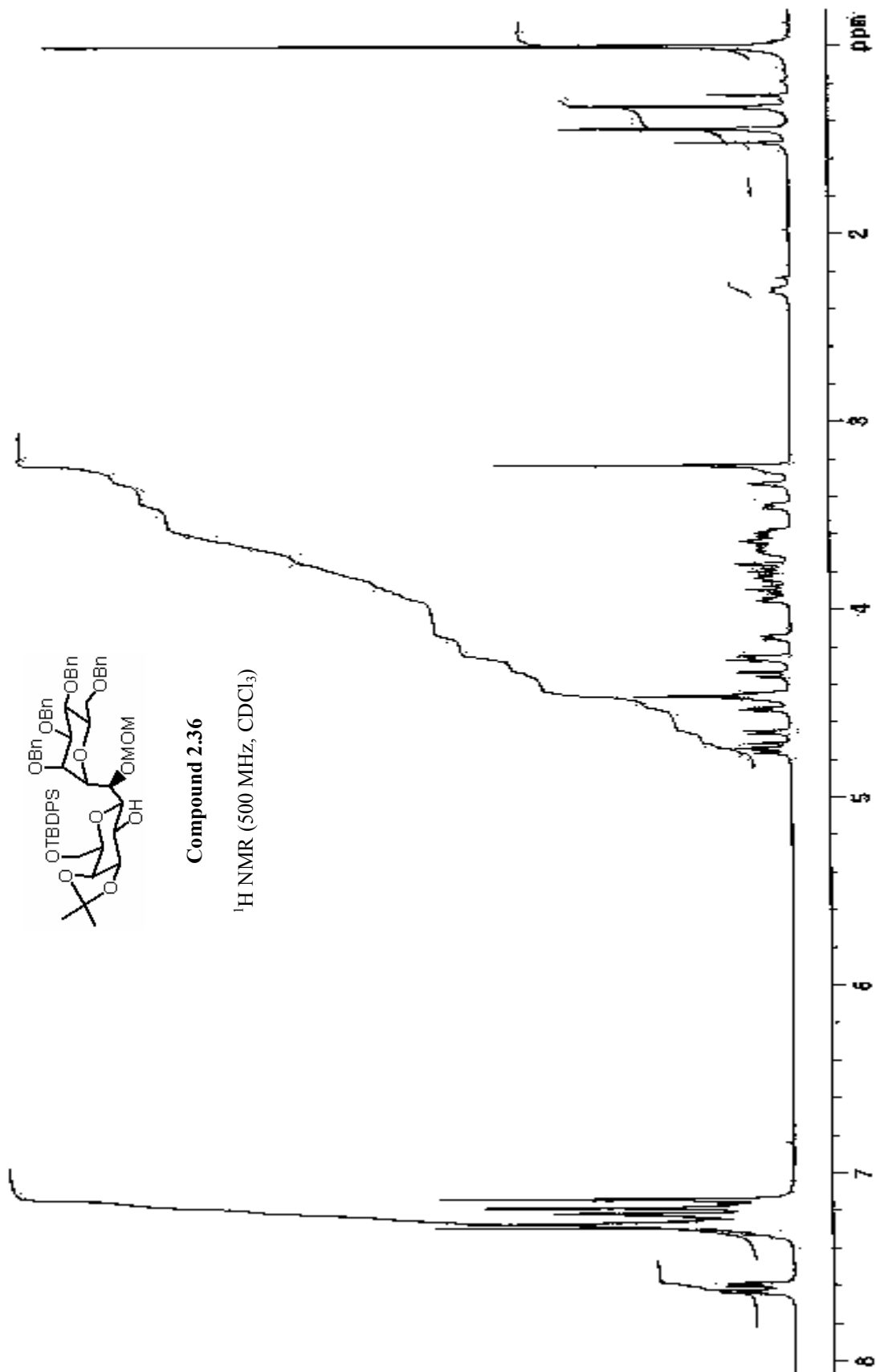


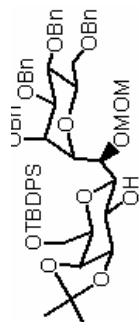
Compound 2.26

$^1\text{H NMR}$ (500 MHz, CDCl_3)



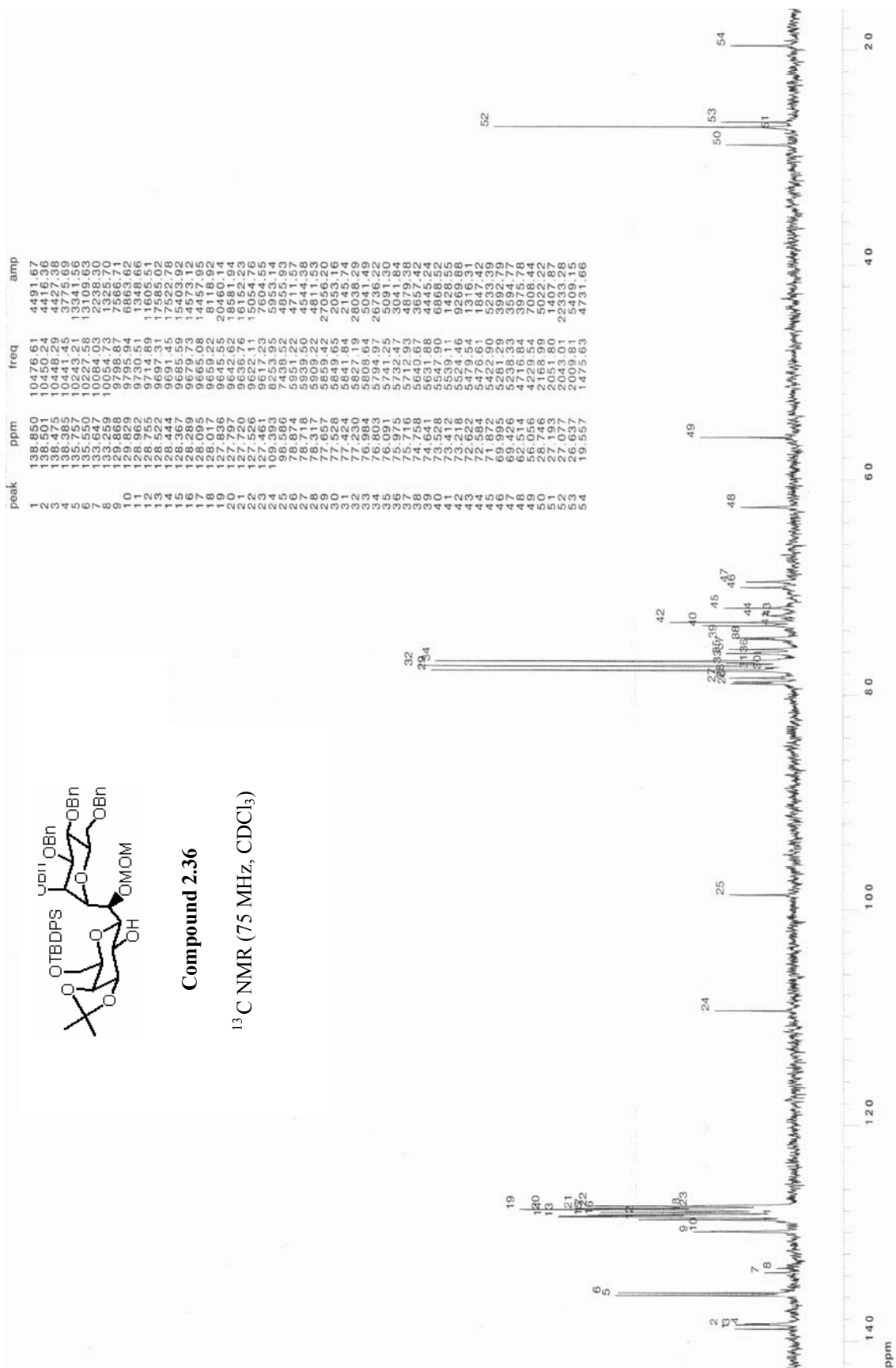


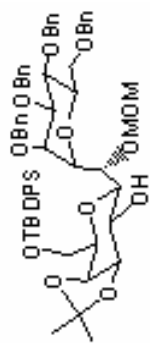




Compound 2.36

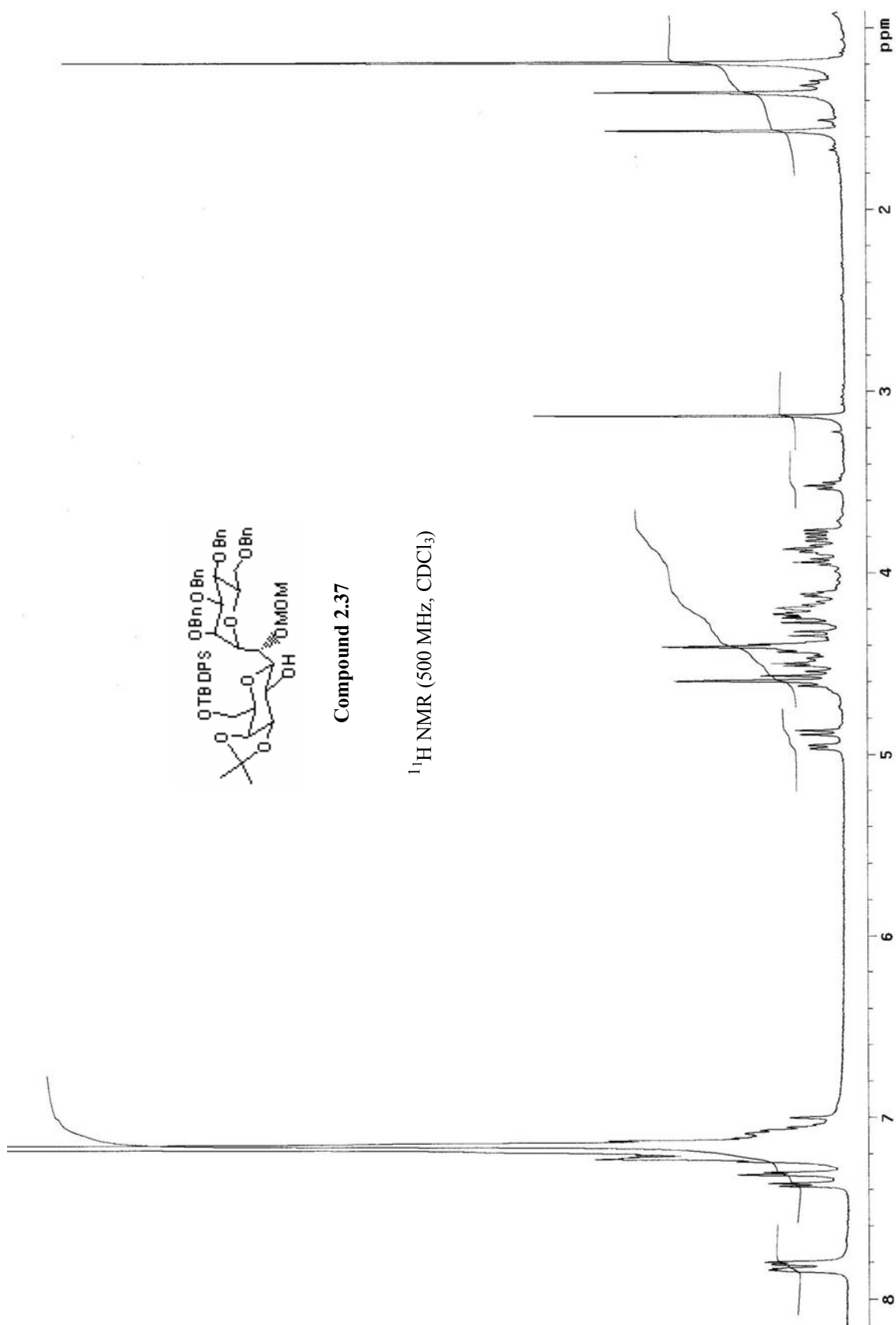
^{13}C NMR (75 MHz, CDCl_3)

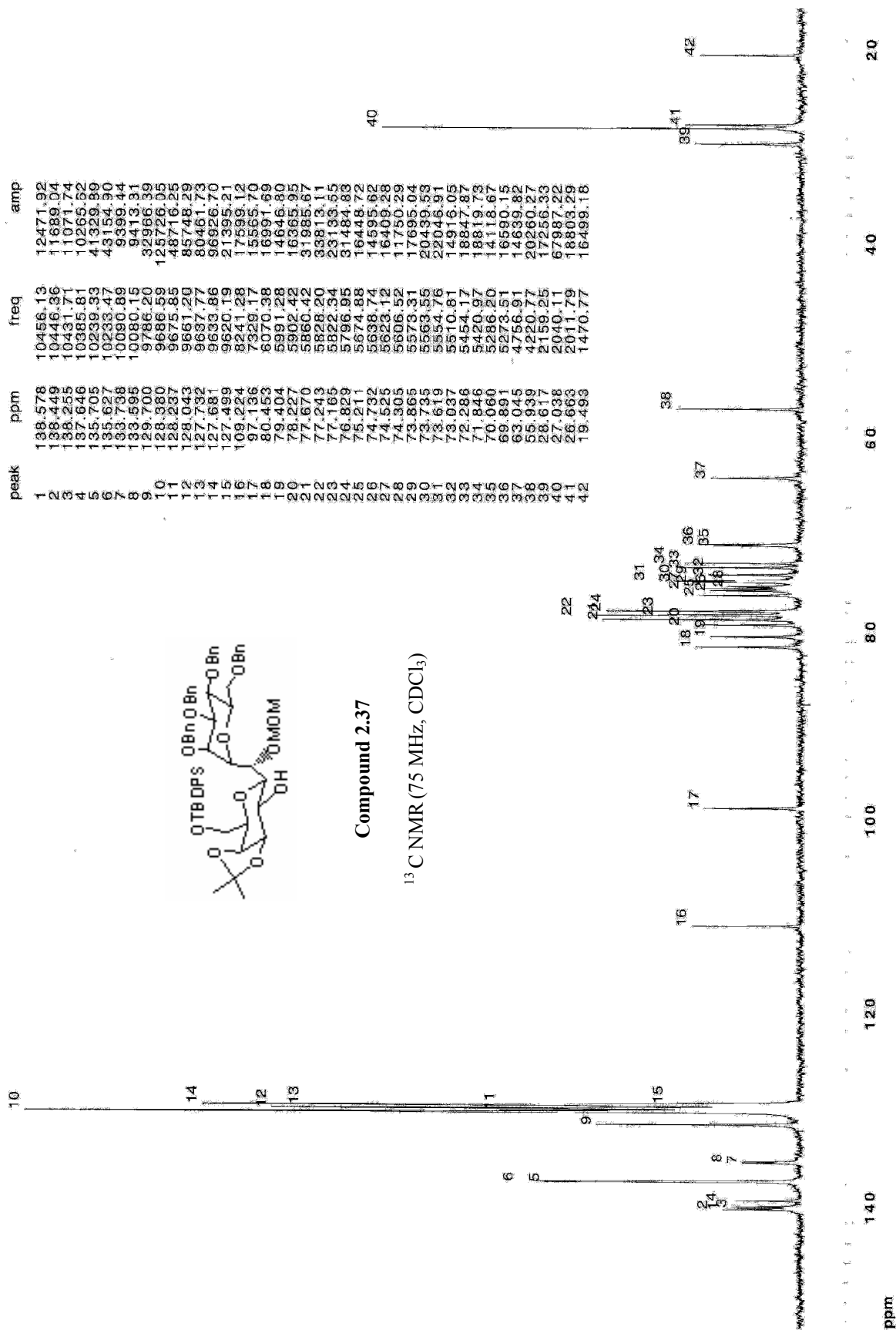


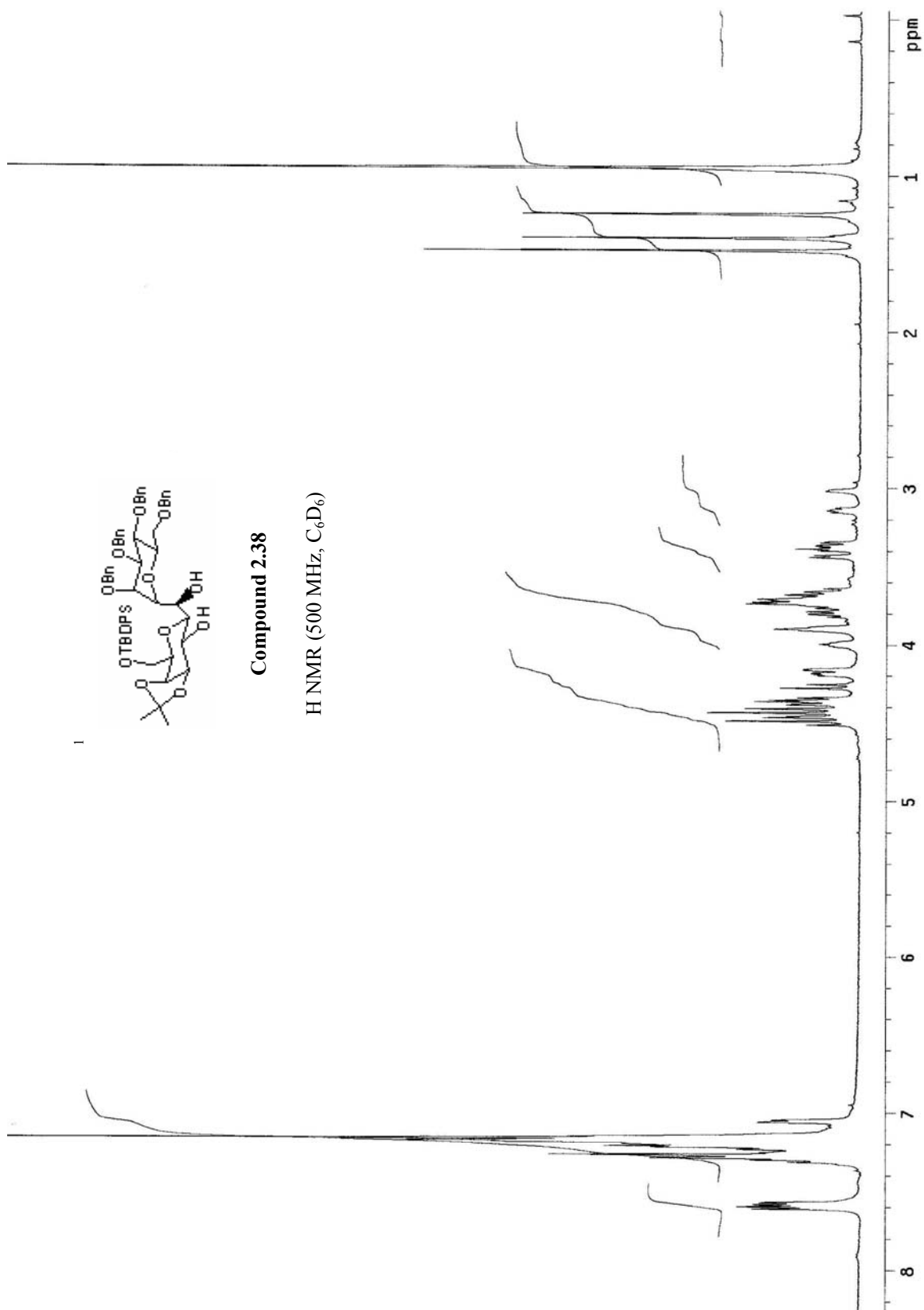


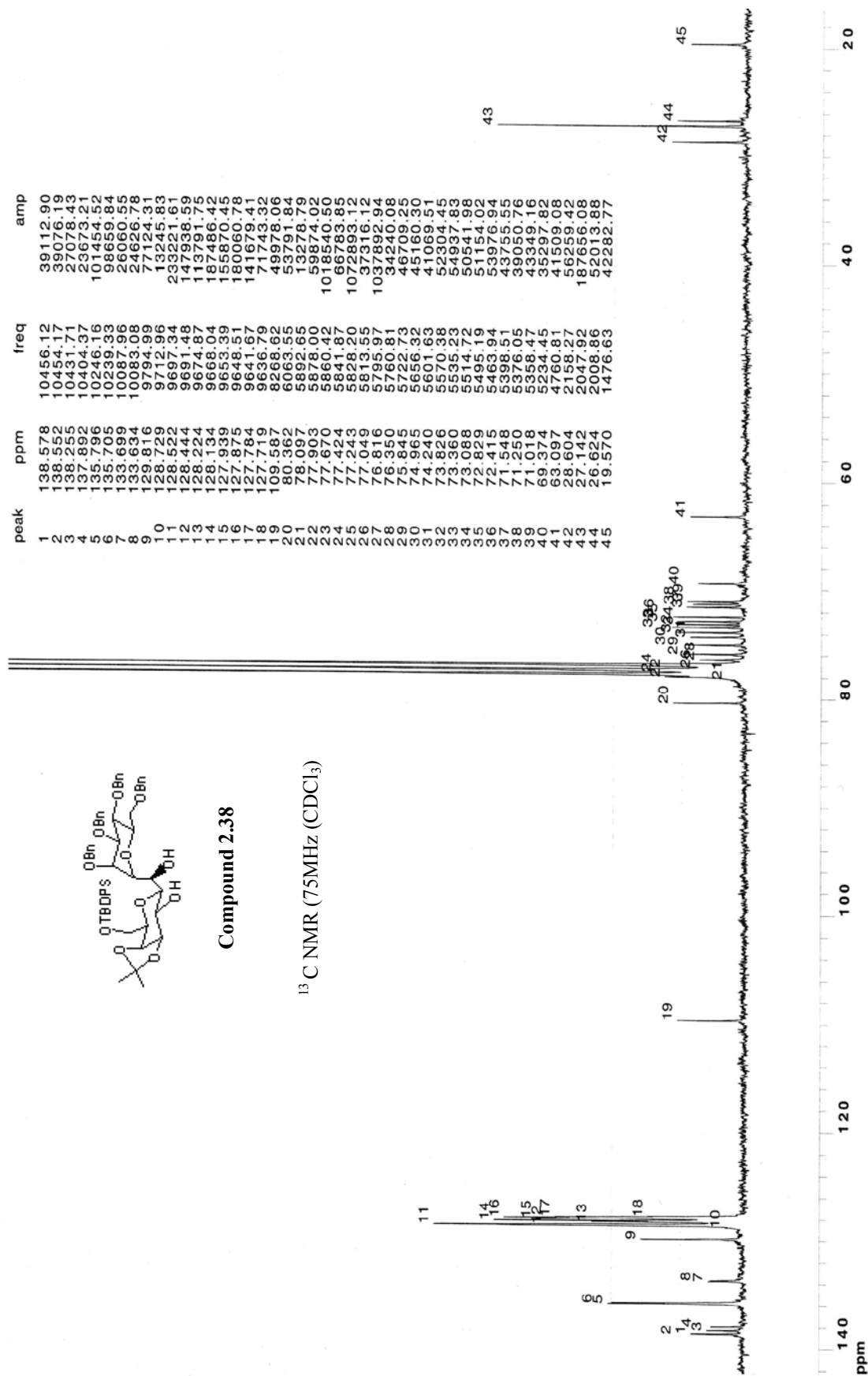
Compound 2.37

^1H NMR (500 MHz, CDCl_3)





**Compound 2.38**H NMR (500 MHz, C_6D_6)



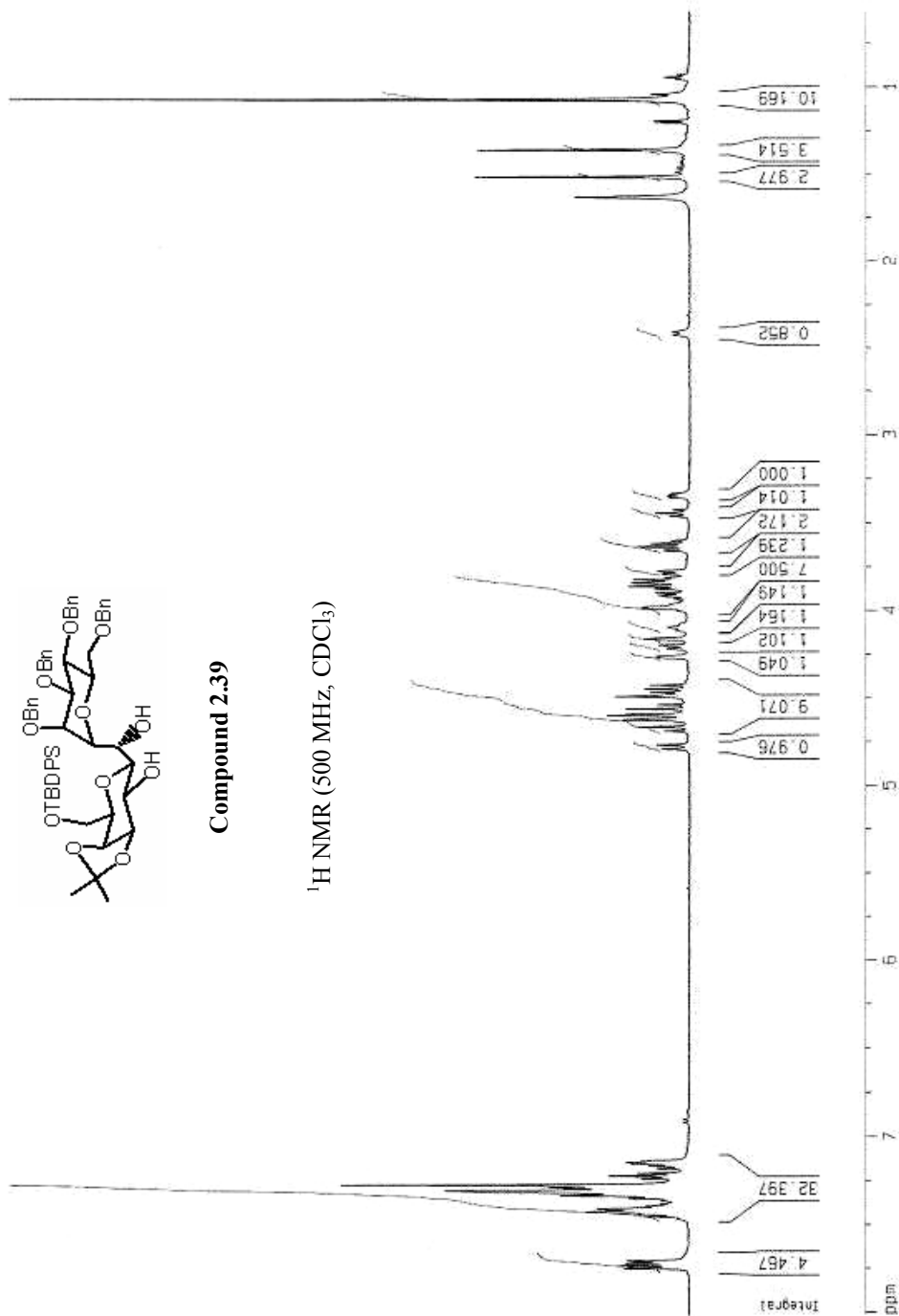
Current Data Parameters
 NAME 06280501
 EXPNO 23
 PROCNO 1

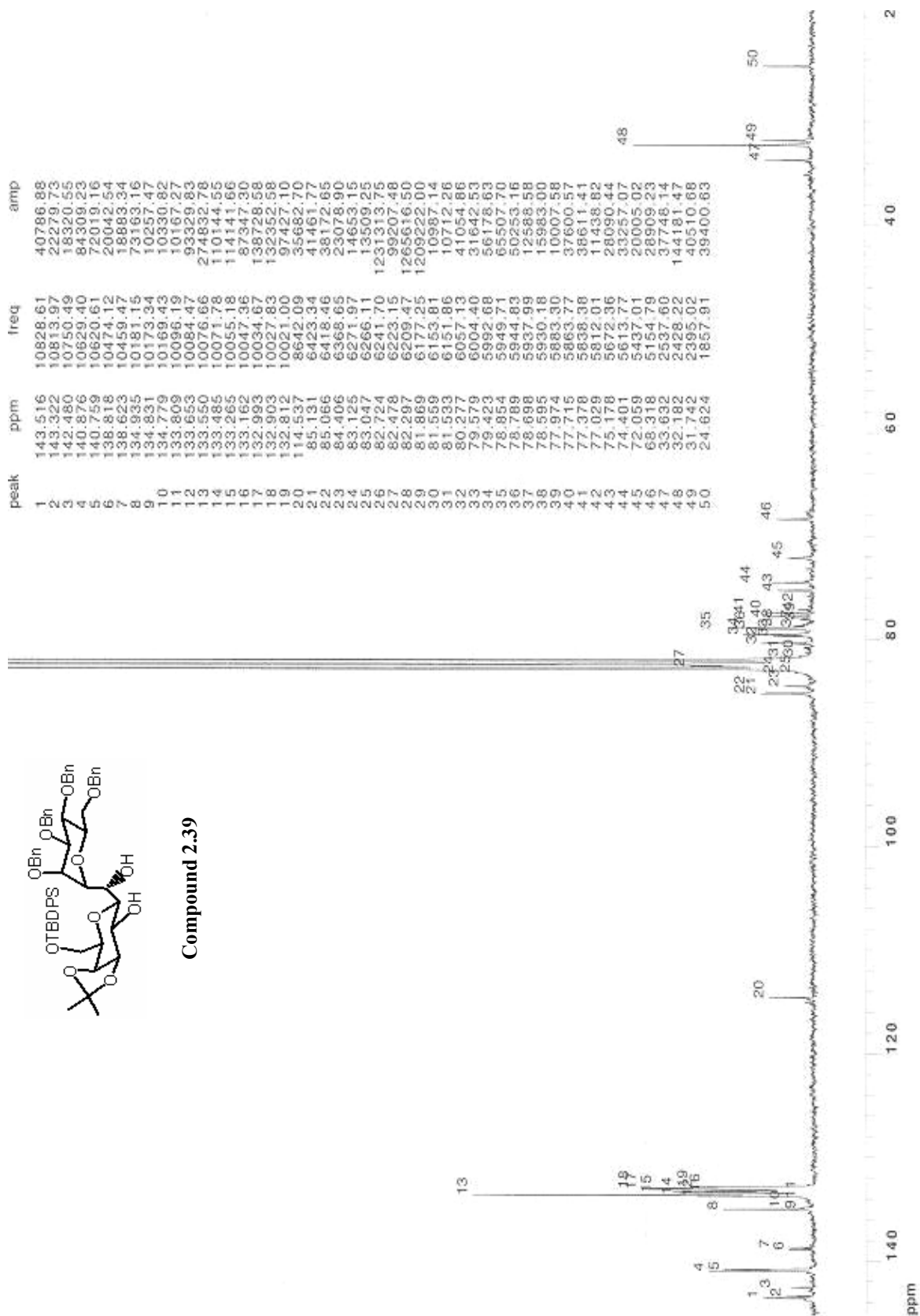
F2 - Acquisition Parameters
 Date_ 20050628
 Time 22.43
 INSTRUM spect
 PROBHD 3 mm CPDIL 13C
 PULPROG zg30
 TD 32768
 SOLVENT CDCl3
 NS 8
 DS 2
 SWH 8012.820 Hz
 FIDRES 0.244532 Hz
 AQ 2.0440356 sec
 RG 1
 DM 52.400 USEC
 DE 6.00 USEC
 TE 303.0 K
 D1 1.00000000 sec
 MCHRES 0.00000000 sec
 MCMRK 0.01500000 sec

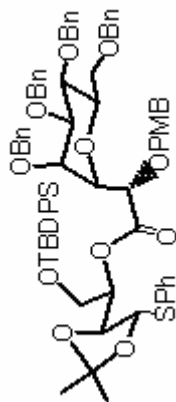
===== CHANNEL f1 =====
 NUC1 1H
 P1 10.00 USEC
 PL1 12.20 dB
 SFO1 500.1330885 MHz

F2 - Processing parameters
 S1 32768
 SF 500.1300089 MHz
 MD 64
 SSB 0
 LB 0.30 Hz
 GB 0
 PC 1.00

1D NMR plot parameters
 CX 22.00 cm
 CY 11.98 cm
 F1 8.015 ppm
 F2 4008.66 Hz
 F2 0.566 ppm
 PPMCM 282.89 Hz
 HZCM 169.35309 Hz/cm

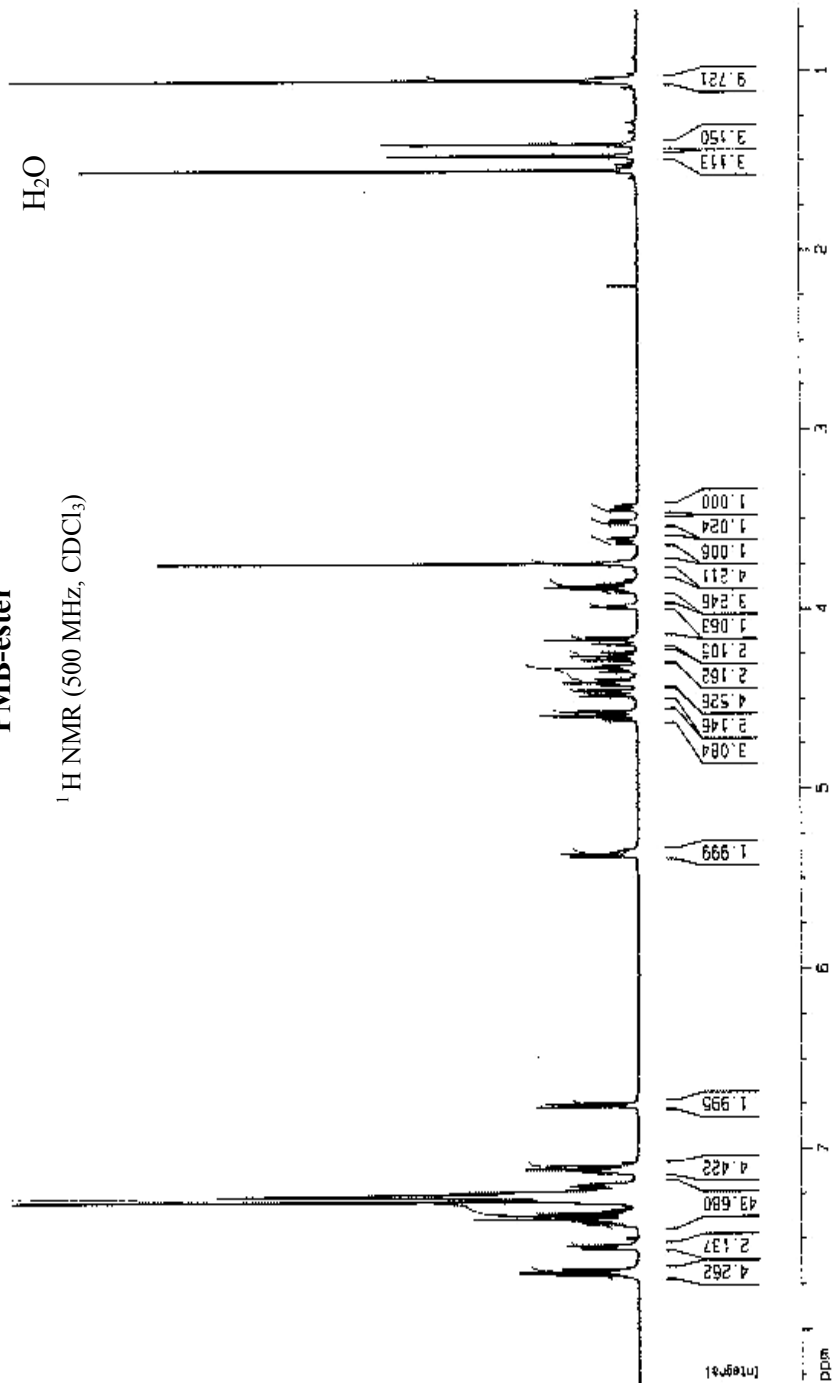






PMB-ester

¹H NMR (500 MHz, CDCl₃)



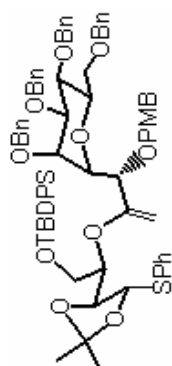
Current Data Parameters
 NAME 05070401
 EXPNO 5
 PROCNO 1

F2 - Acquisition Parameters
 Date_ 20040507
 Time 13.13
 INSTRUM spect
 PROBHD 5 mm DUL 13C-1
 PULPROG zg30
 TD 32768
 SOLVENT CDCl3
 NS 8
 DS 2
 SWH 6666.667 Hz
 FIDRES 0.203451 Hz
 AQ 2.4577250 sec
 RG 406.4
 CW 75.000 usec
 DE 6.00 usec
 TE 303.0 K
 DL 1.0000000 sec
 MCREST 0.0000000 sec
 MCARX 0.0150000 sec

***** CHANNEL f1 *****
 NUCL1 1H
 P1 8.45 usec
 PL1 -3.00 dB
 SFO1 500.1330685 MHz

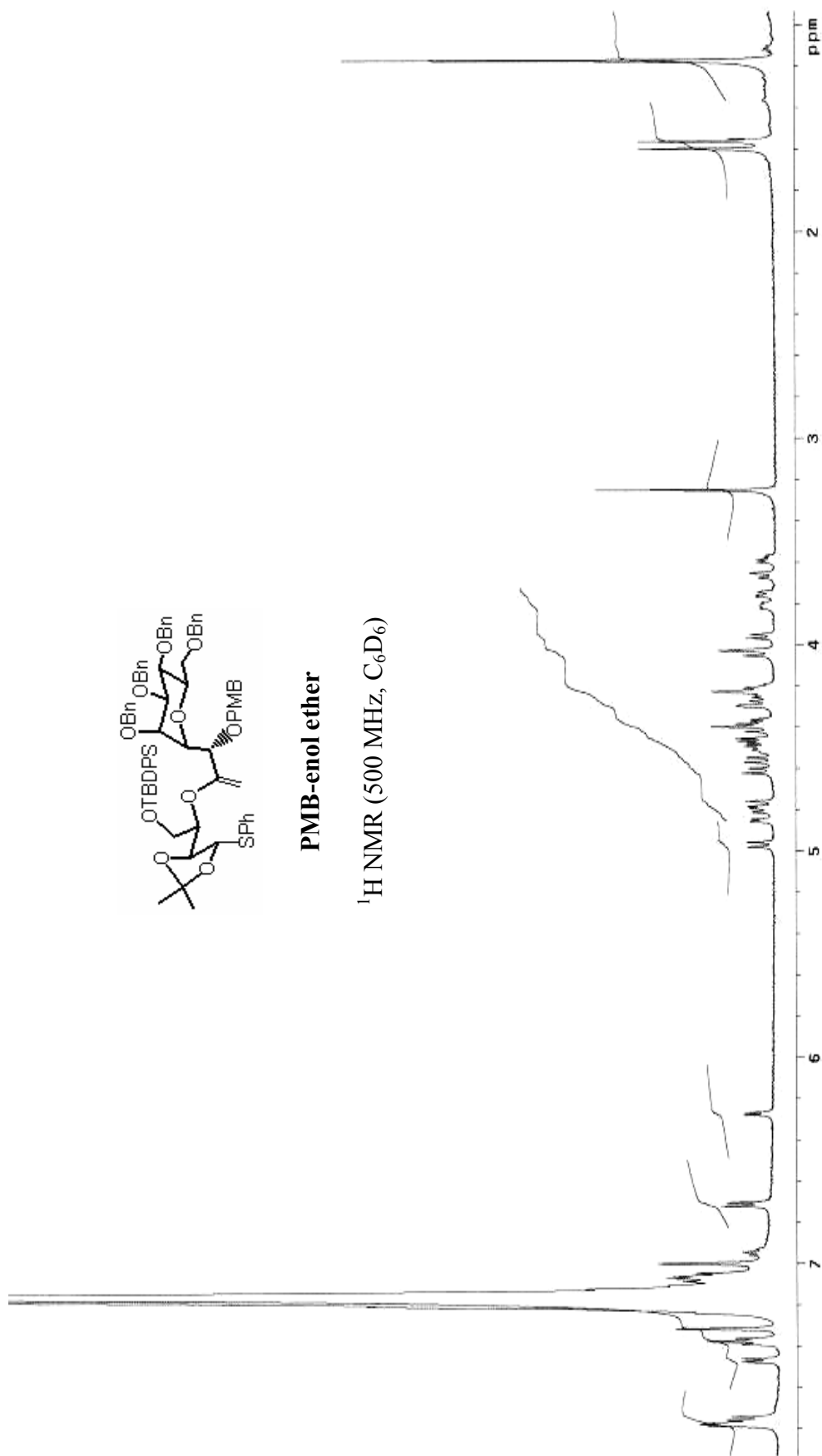
F2 - Processing parameters
 SI 32768
 SF 500.1300600 MHz
 WGM FM
 SSB 0
 LB 0.30 Hz
 GB 0
 PC 1.00

1D NMR plot parameters
 CX 22.00 cm
 CY 46.55 cm
 FJP B.286 dB
 FJ 4143.57 Hz
 FZP 0.654 ppm
 FZ 325.84 Hz
 PPMCH 0.34685 ppm/cm
 HZCH 173.48772 Hz/cm



PMB-enol ether

^1H NMR (500 MHz, C_6D_6)



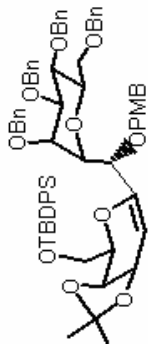
Current Data Parameters
 NAME 05100461
 EXPNO 6
 PROCNO 1

F2 - Acquisition Parameters
 Date_ 20040510
 Time 14:31
 INSTRUM spect
 PROBHD 5 mm QNP 13C-1
 PULPROG zgpg30
 TD 32768
 SOLVENT C505
 NS 8
 DS 2
 SWH 6656.667 Hz
 FIDRES 0.203451 Hz
 AQ 2.4577260 sec
 RG 256
 ON 75.000 usec
 DE 6.00 usec
 TE 303.0 K
 D1 1.00000000 sec
 ACQRES 0.06000000 sec
 ICNMRK 0.01500000 sec

===== CHANNEL f1 =====
 NUC1 1H
 P1 8.45 usec
 PL1 -3.00 dB
 SFO1 500.1330985 MHz

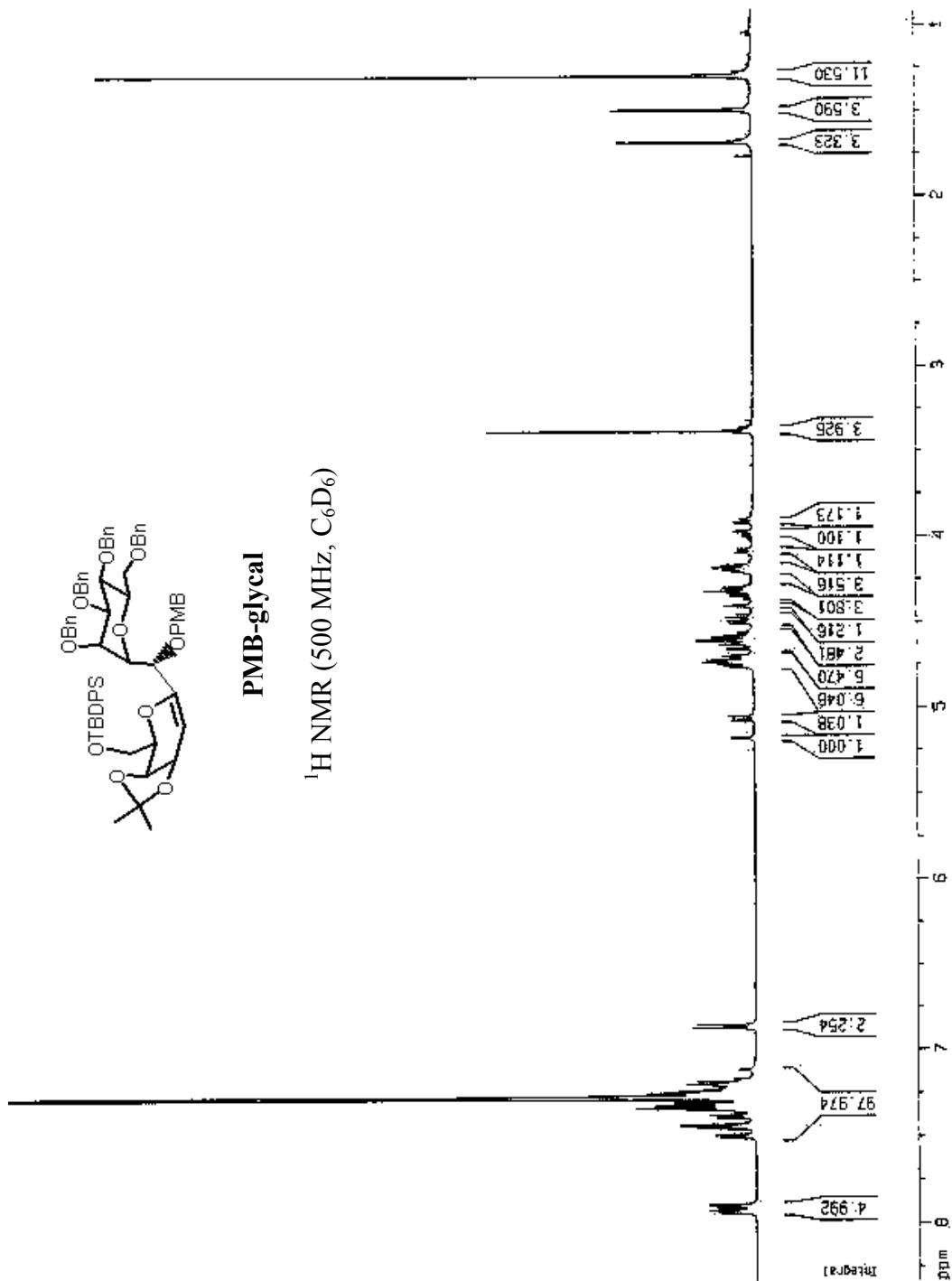
F2 - Processing Parameters
 SI 32768
 SF 500.1300000 MHz
 MDW EM
 SSB 0
 LB 0.30 Hz
 GB 0
 PC 1.60

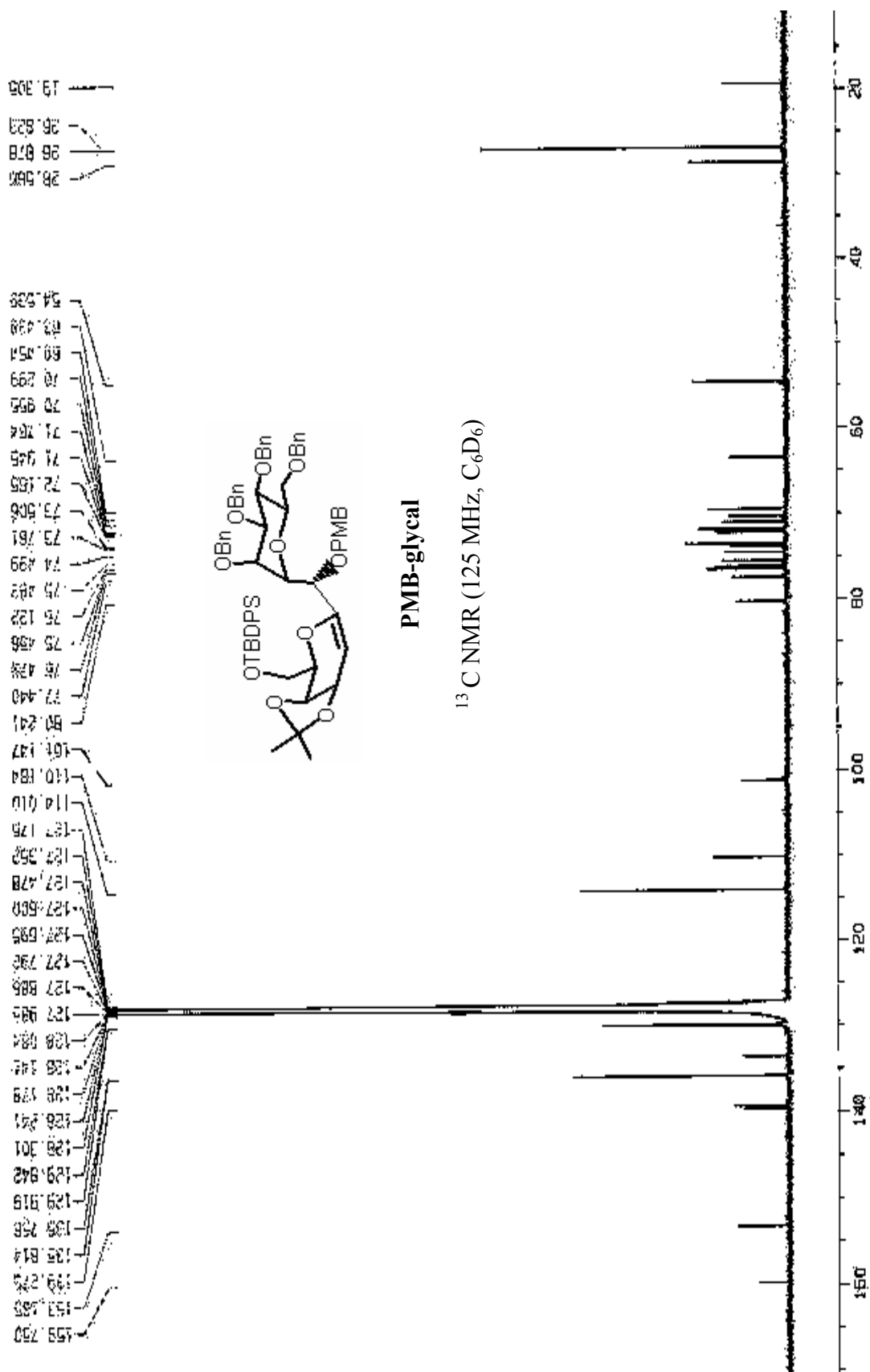
1D NMR plot parameters
 X1 22.00 cm
 X2 40.13 cm
 F1P 8.385 ppm
 F1 4168.79 Hz
 F2P 0.906 ppm
 F2 452.84 Hz
 PPGCM 0.33772 ppm/cm
 MZCM 168.90213 Hz/cm

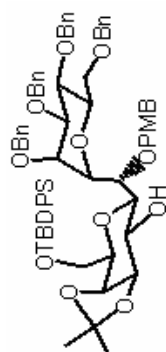


PMB-glycal

¹H NMR (500 MHz, C₆D₆)

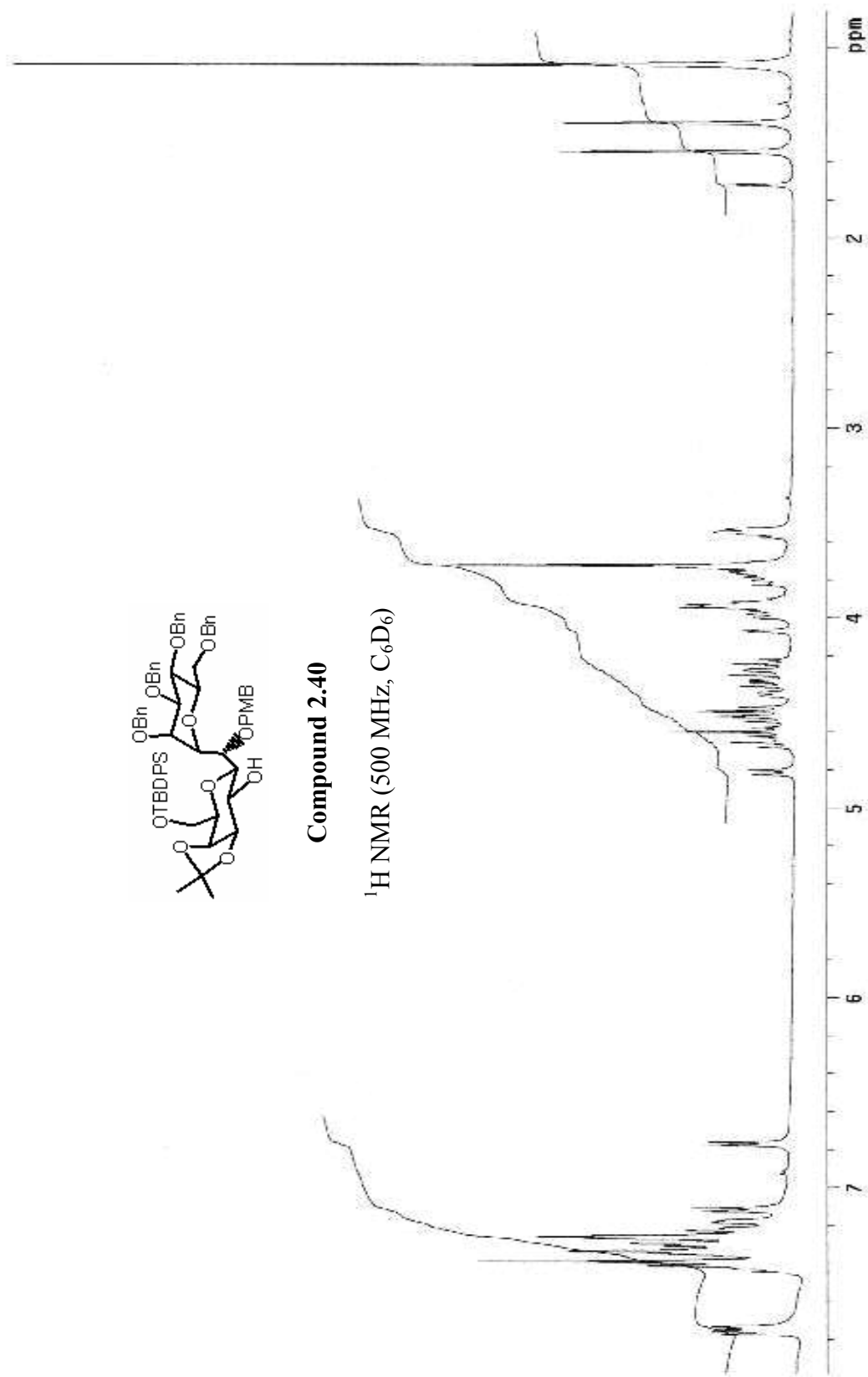


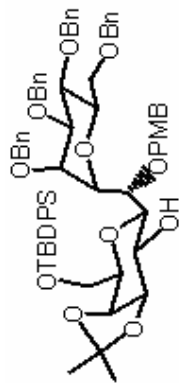




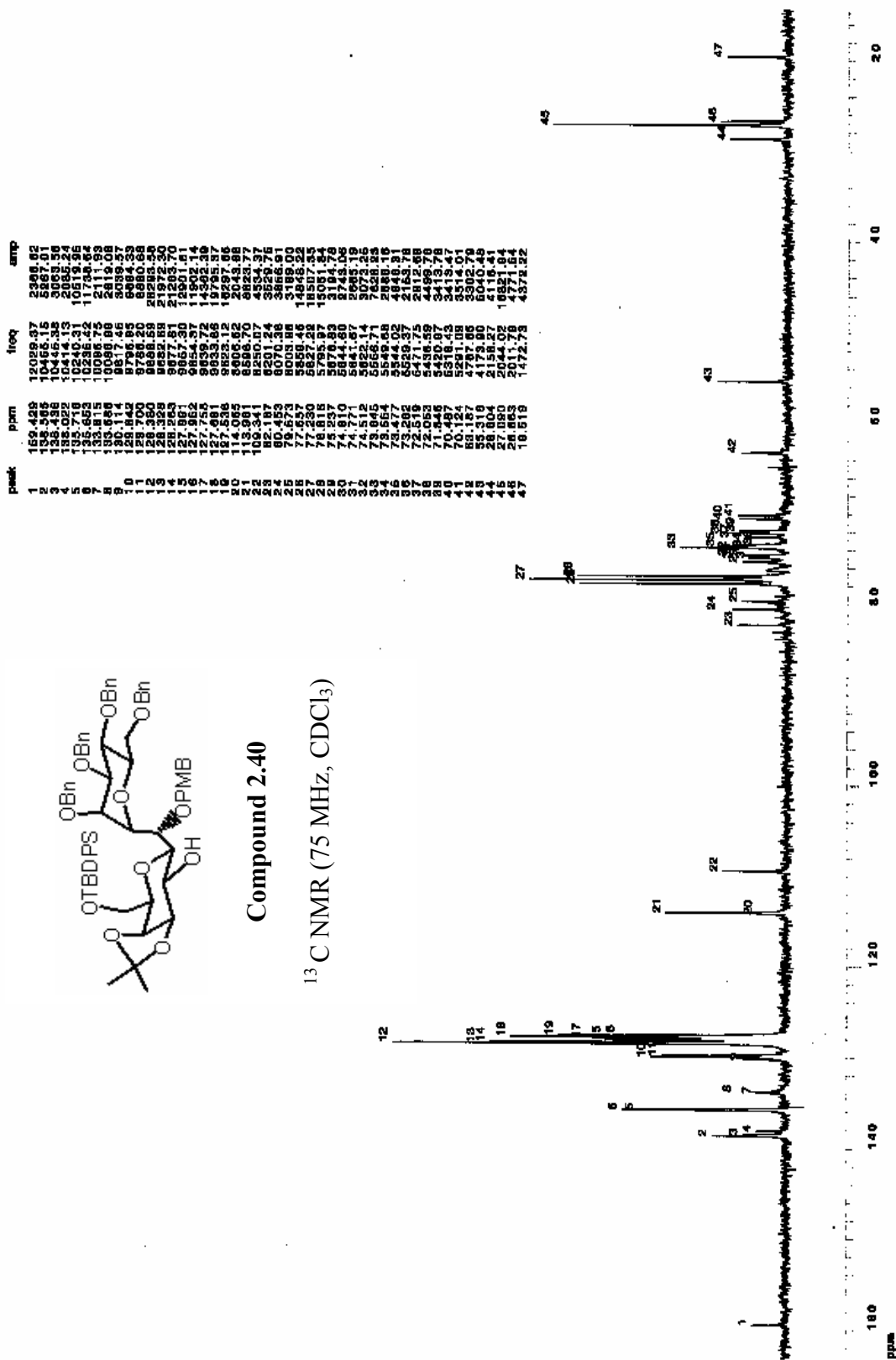
Compound 2.40

¹H NMR (500 MHz, C₆D₆)





Compound 2.40

 ^{13}C NMR (75 MHz, CDCl_3)

Current Data Parameters
 NAME 07120501
 EXPNO 20
 PROCNO 1

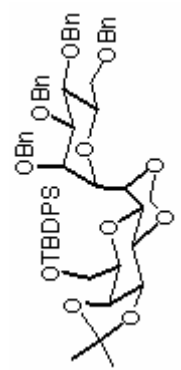
F2 - Acquisition Parameters
 Date_ 20050712
 Time 18.45
 INSTRUM spect
 PROBHD 5 mm DUL 13C-1
 PULPROG zg30
 TD 32768
 SOLVENT CDCl3
 NS 8
 DS 2

SWH 8012.820 Hz
 FIDRES 0.244532 Hz
 AQ 2.0448356 sec
 RG 161.3
 DW 62.400 usec
 DE 6.00 usec
 TE 303.0 K
 D1 1.00000000 sec
 MCREST 0.00000000 sec
 MCWPRK 0.01500000 sec

===== CHANNEL f1 =====
 NUC1 1H
 P1 9.30 usec
 PL1 -3.00 dB
 SF01 500.1330885 MHz

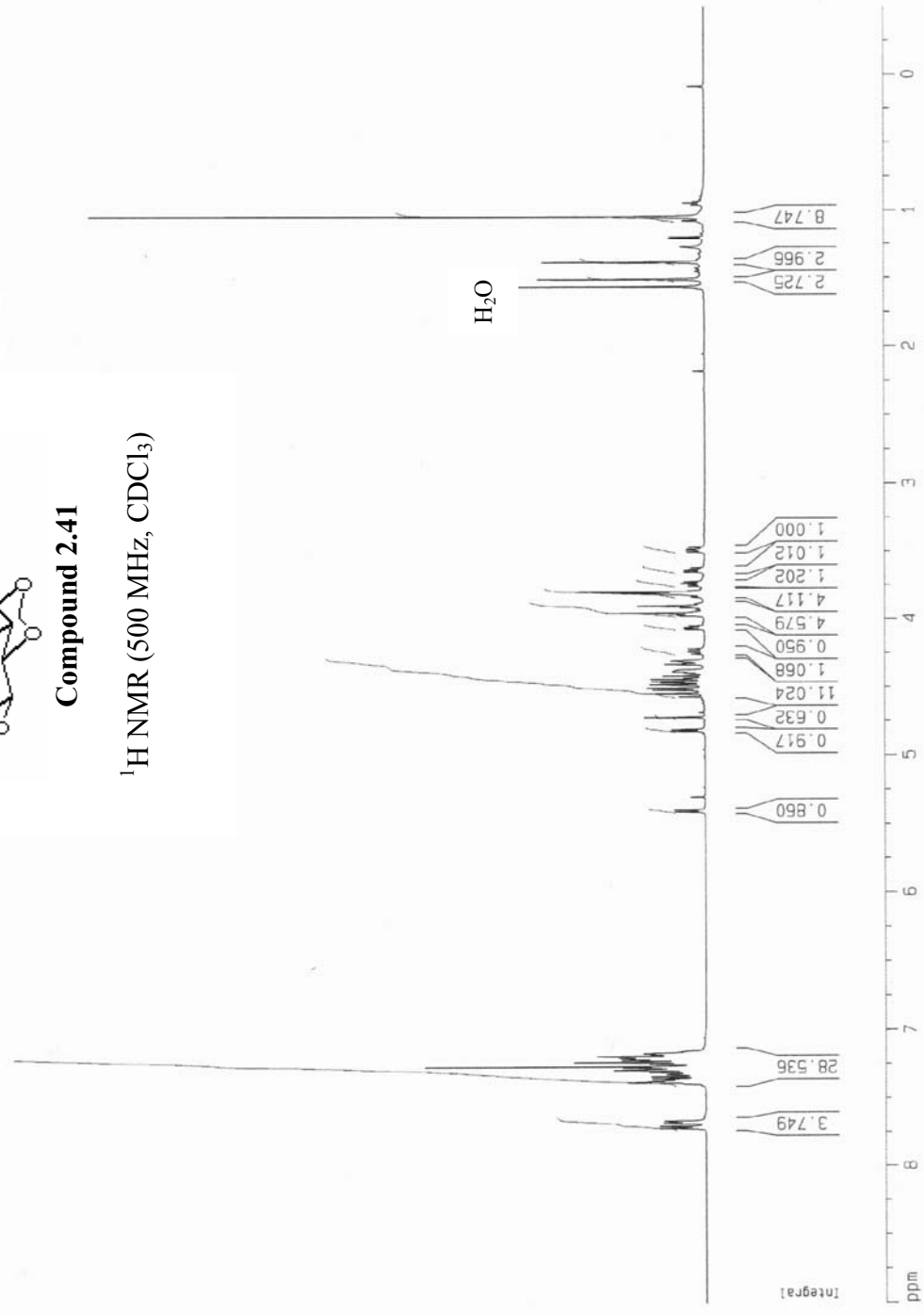
F2 - Processing parameters
 SI 32768
 SF 500.1300084 MHz
 NDW EM
 SSB 0
 LB 0.30 Hz
 GB 0
 PC 1.00

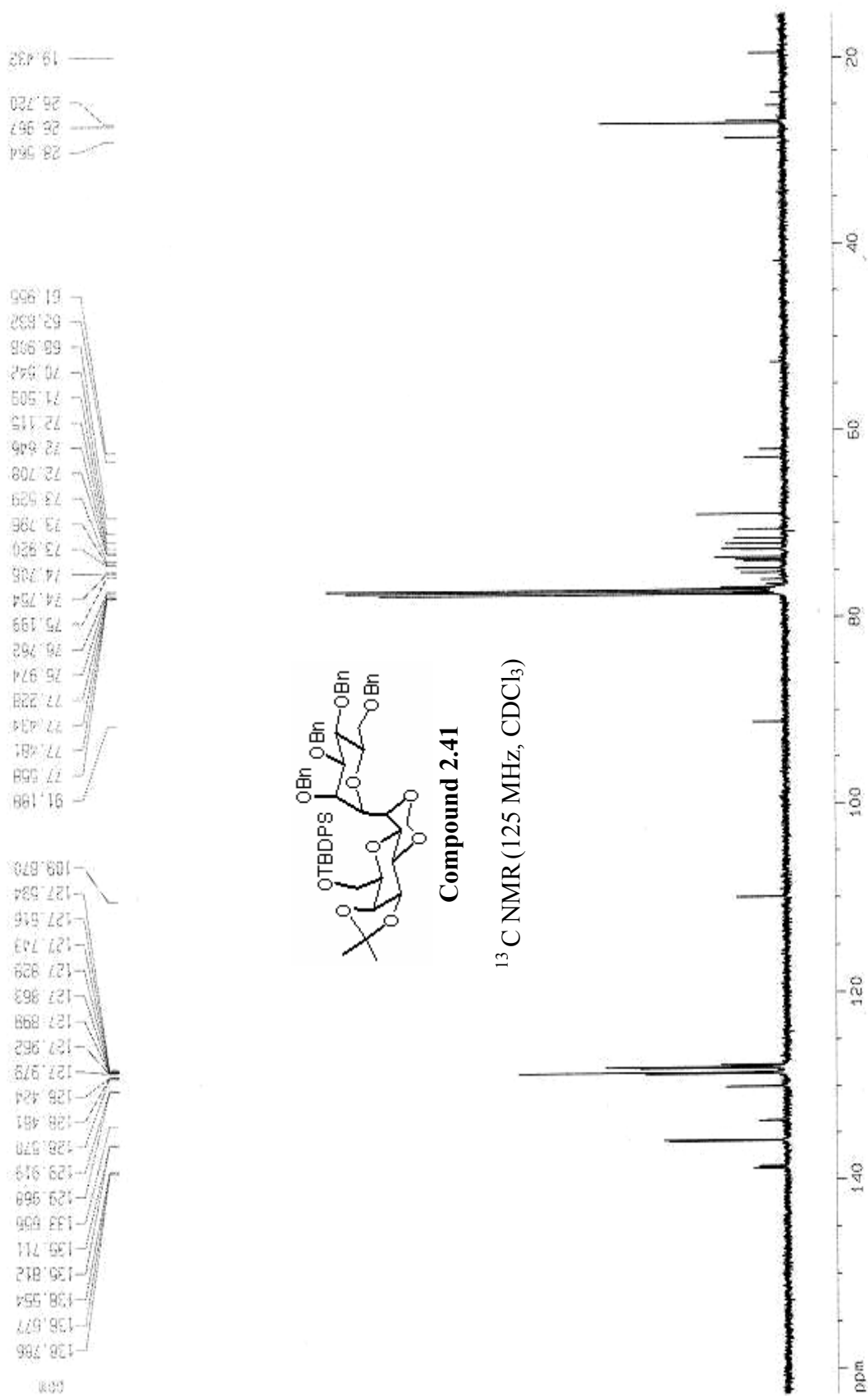
ID NMR plot parameters
 CX 22.00 cm
 CY 10.34 cm
 FIP 9.000 ppm
 F1 4501.17 Hz
 F2P -0.500 ppm
 F2 -250.07 Hz
 PPMCM 0.43182 ppm/cm
 HZCM 215.96523 Hz/cm

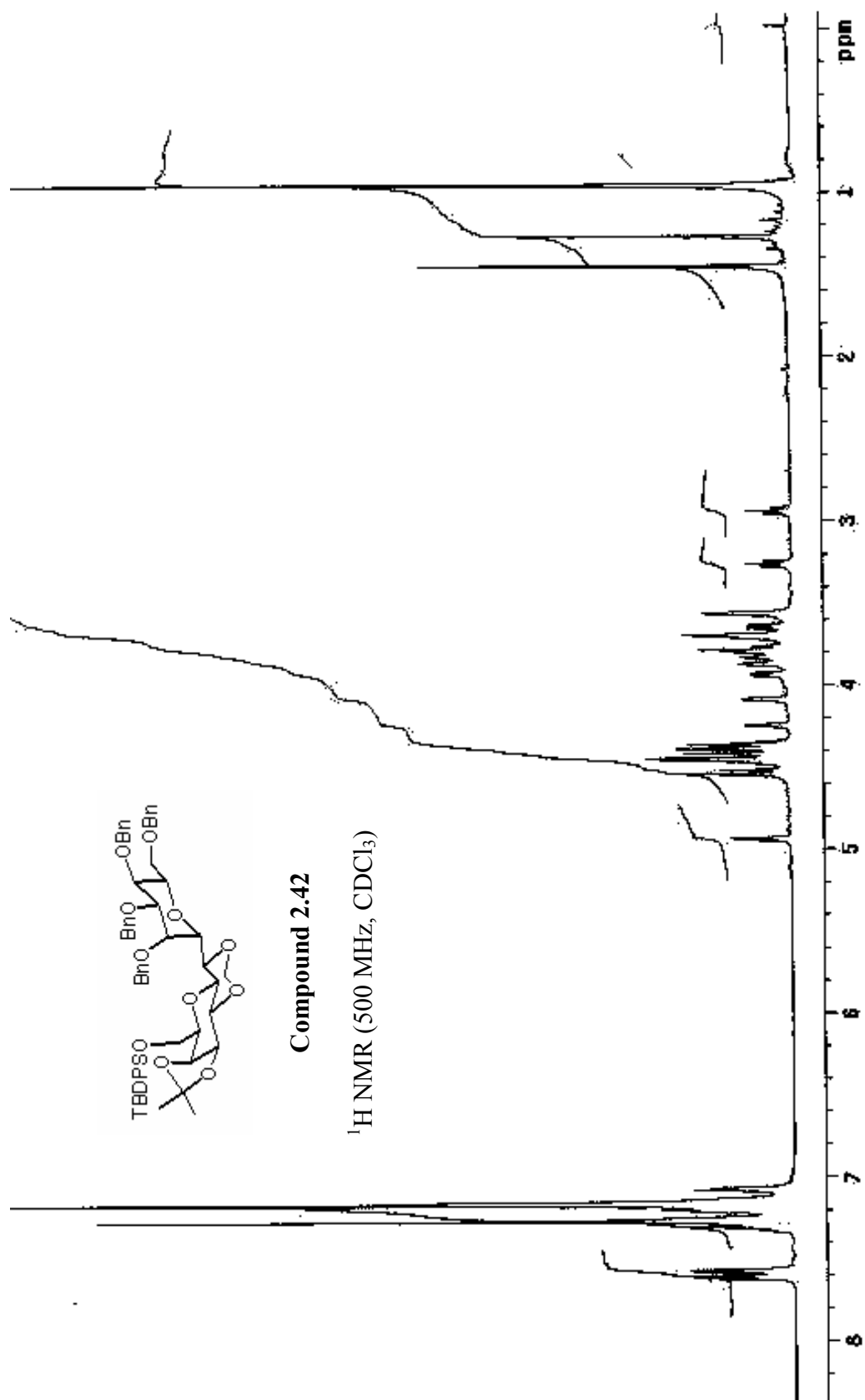


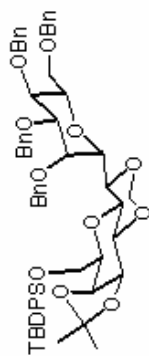
Compound 2.41

¹H NMR (500 MHz, CDCl₃)



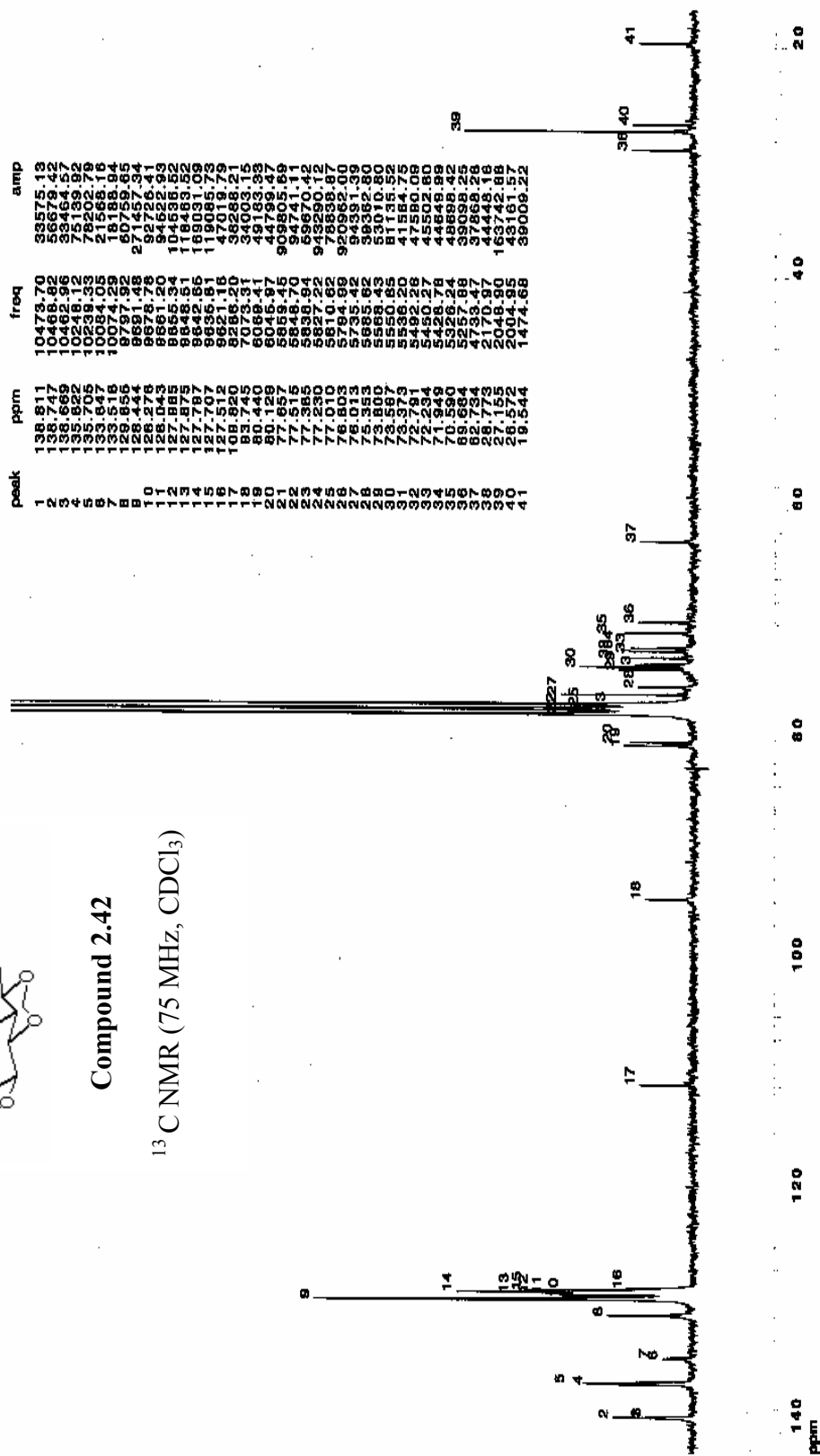


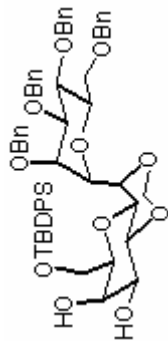




Compound 2.42

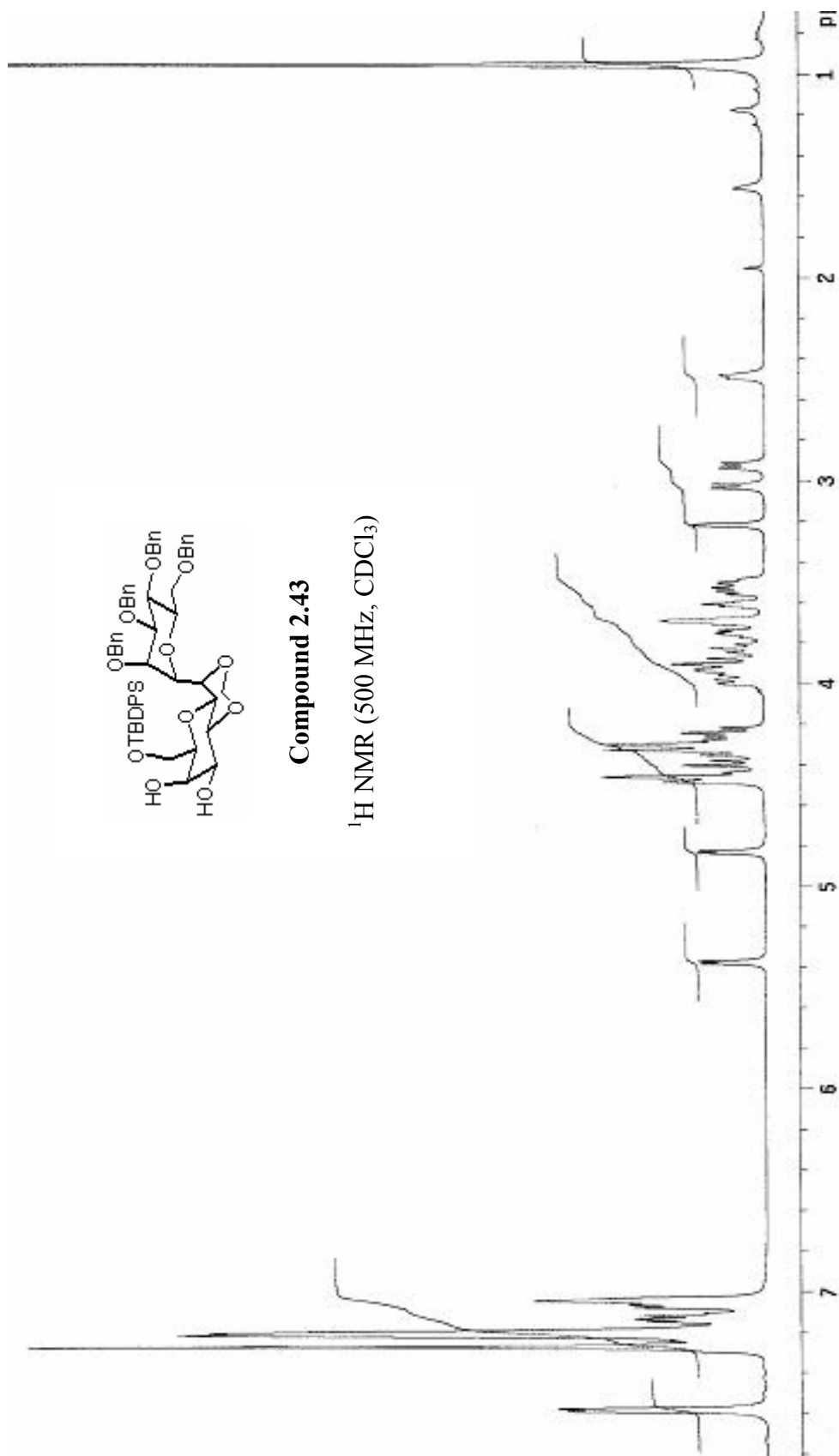
^{13}C NMR (75 MHz, CDCl_3)

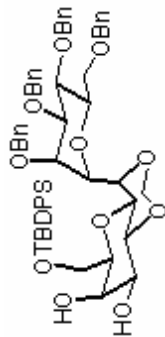




Compound 2.43

¹H NMR (500 MHz, CDCl₃)



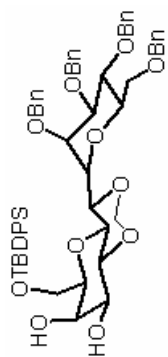


Compound 2.43

 ^{13}C NMR (75 MHz, CDCl_3)

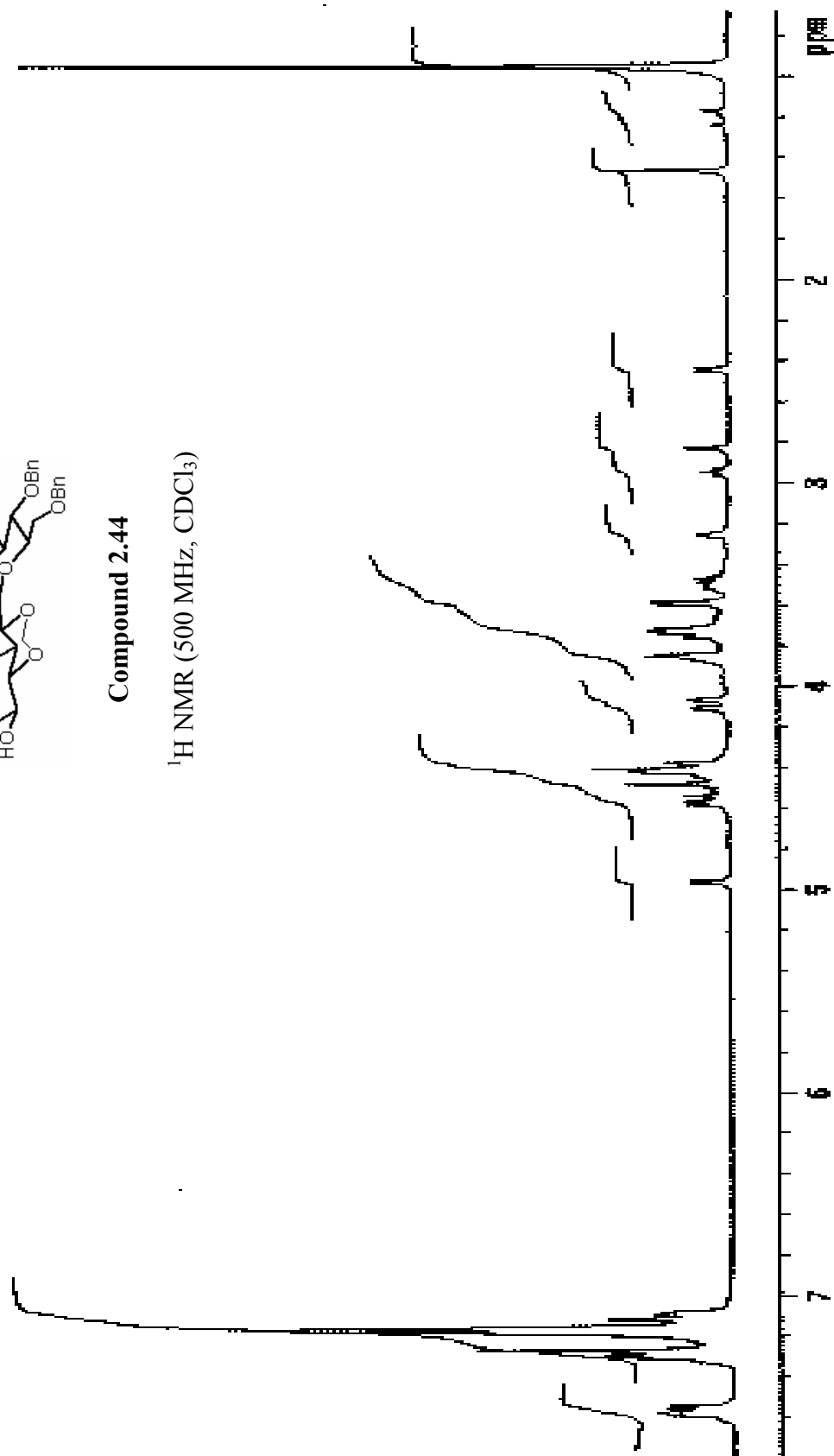
peak	ppm	freq	amp
1	138.688	10482.88	4027.44
2	138.345	10436.56	4367.12
3	138.138	10422.82	4087.12
4	137.881	10407.30	3640.91
5	135.866	10238.40	20127.92
6	133.478	10071.38	2897.21
7	133.375	10063.55	2831.04
8	128.865	9787.82	10878.82
9	128.545	9769.28	15078.03
10	128.417	9762.03	4821.97
11	127.924	9695.94	1634.70
12	127.862	9687.63	16682.41
13	127.787	9672.65	1707.38
14	127.681	9658.88	19286.88
15	127.616	9652.70	16032.70
16	127.511	9638.98	16028.98
17	122.075	8947.34	4865.68
18	80.026	8038.16	6008.14
19	77.857	5858.45	44703.08
20	77.457	5842.84	2805.88
21	77.388	5839.81	2780.82
22	77.230	5827.22	43737.88
23	76.816	5795.97	44012.81
24	75.453	5691.48	5888.18
25	75.017	5660.23	6397.15
26	74.784	5642.65	6208.23
27	74.745	5639.72	8172.07
28	73.477	5544.02	7943.45
29	73.425	5540.11	9888.14
30	73.360	5535.23	7692.28
31	73.231	5525.46	6203.58
32	72.752	5468.33	6203.58
33	72.415	5463.84	7091.16
34	71.885	5423.80	6724.26
35	70.357	5308.86	7261.78
36	68.659	5255.83	5854.59
37	68.361	5233.47	8112.58
38	66.367	5008.84	3869.49
39	63.008	4753.88	5231.35
40	27.142	2047.92	21445.58
41	18.519	1472.73	5761.18

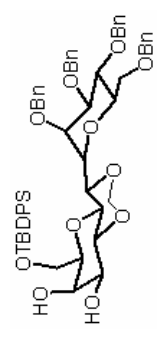




Compound 2.44

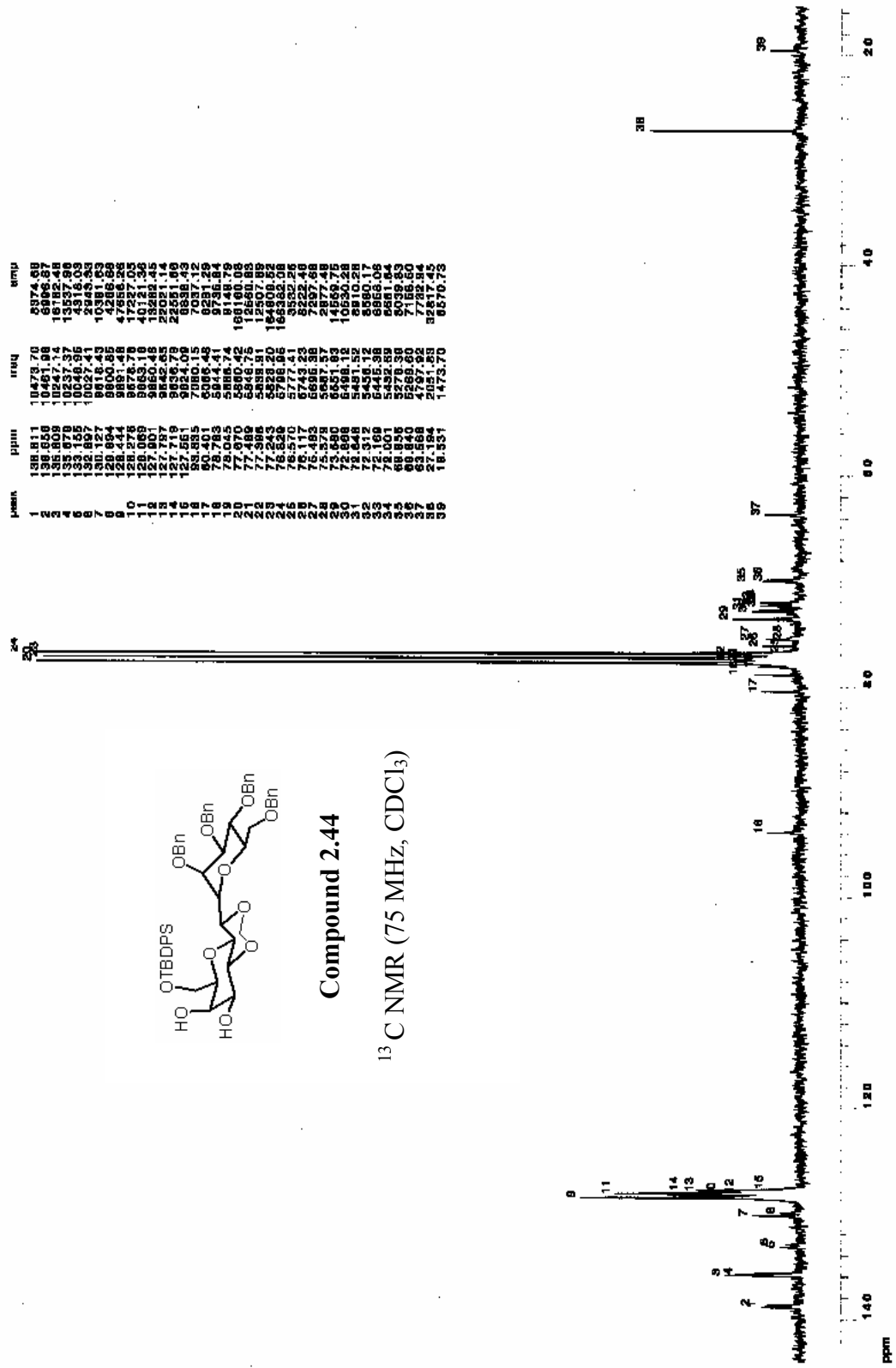
^1H NMR (500 MHz, CDCl_3)

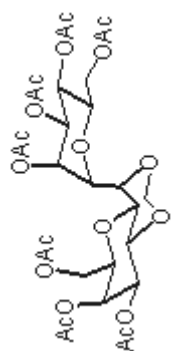




Compound 2.44

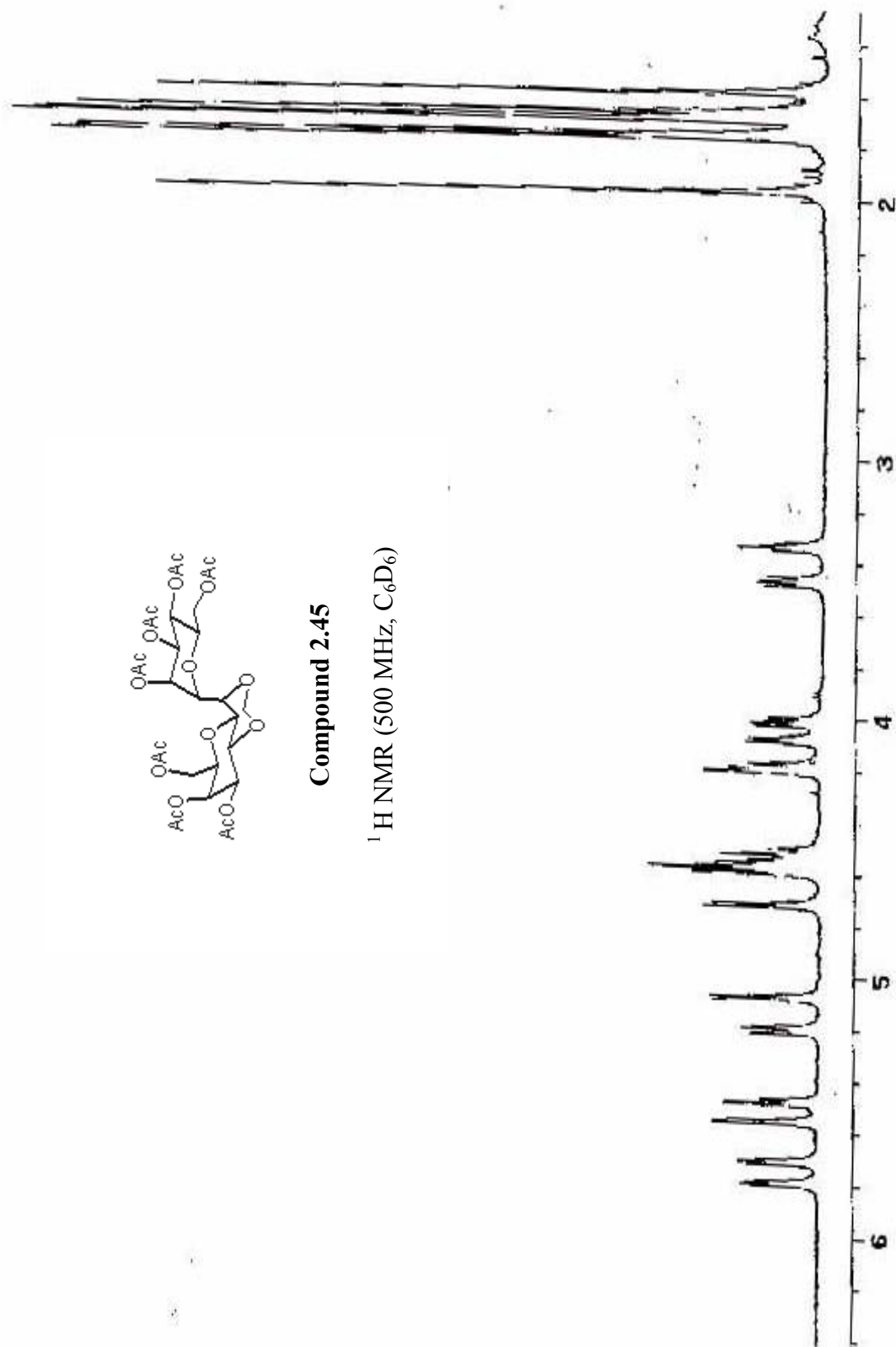
¹³C NMR (75 MHz, CDCl₃)





Compound 2.45

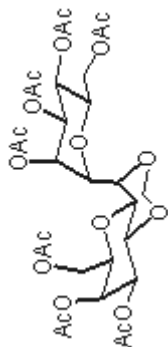
$^1\text{H NMR}$ (500 MHz, C_6D_6)



32 peaks found in 'RWD/2/225.C13.CDCL' amp

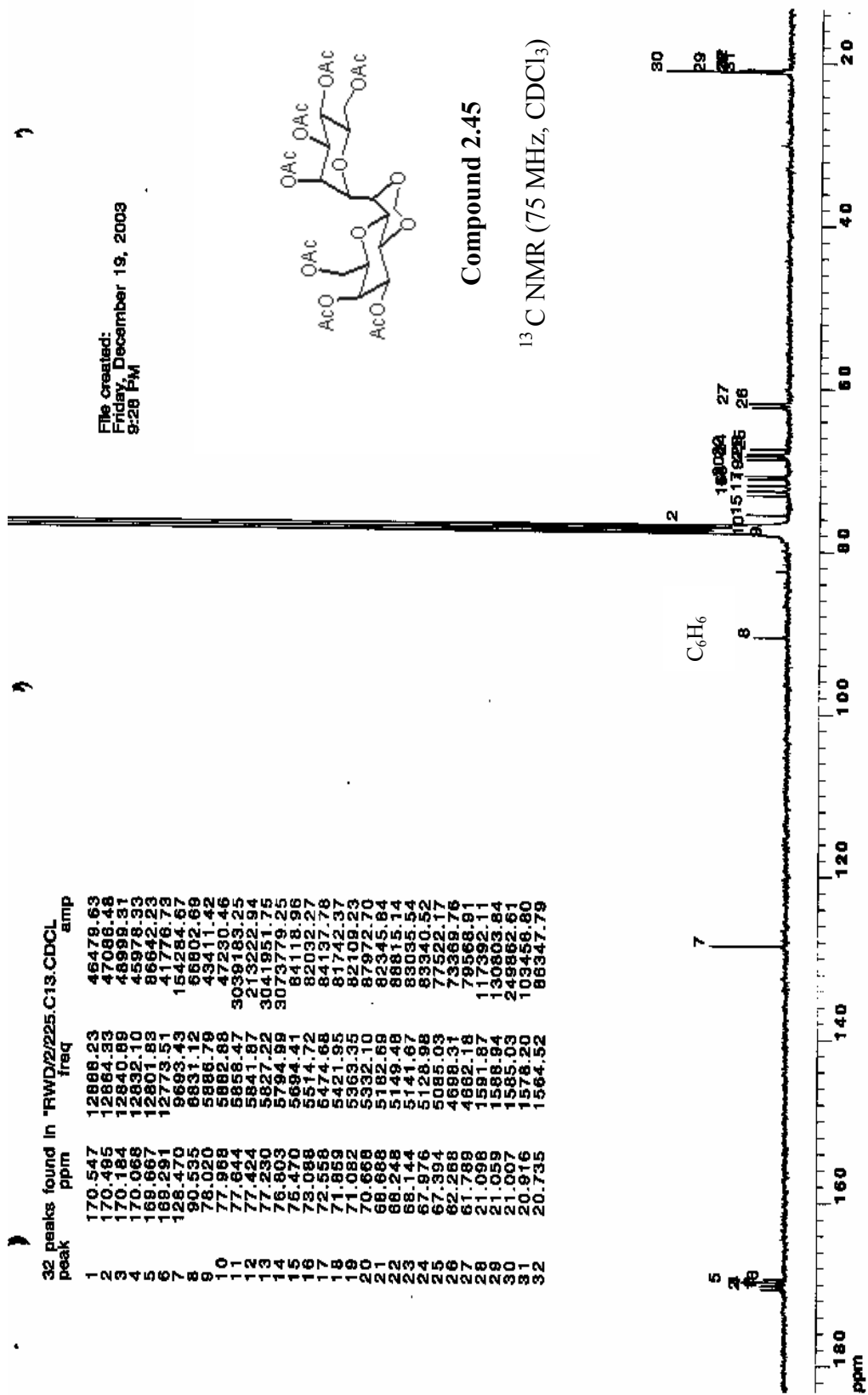
peak	ppm	freq	amp
1	170.547	12888.23	46478.63
2	170.465	12884.53	47086.48
3	170.184	12840.89	48998.31
4	170.068	12832.10	45978.33
5	168.867	12801.83	86642.23
6	168.291	12773.51	41776.73
7	128.470	9693.43	154284.67
8	90.535	6831.12	66802.69
9	78.020	5886.79	43411.42
10	77.968	5882.88	47230.46
11	77.644	5858.47	3039183.25
12	77.424	5827.22	213222.94
13	77.230	5794.99	3041951.75
14	76.803	5694.41	84118.98
15	76.470	5514.72	82037.27
16	73.038	5474.65	84137.76
17	72.559	5421.95	81742.37
18	71.852	5363.35	82108.23
20	70.668	5332.10	87972.70
21	68.868	5182.69	82345.84
22	68.248	5149.48	88815.14
23	68.144	5141.67	83035.54
24	67.876	5128.98	83340.52
25	67.394	5085.03	77522.17
26	62.288	4698.31	73368.76
27	61.789	4662.18	79568.91
28	21.098	1591.87	117392.11
29	21.059	1588.94	130803.84
30	21.007	1585.03	248862.61
31	20.916	1578.20	103456.80
32	20.735	1564.52	86347.73

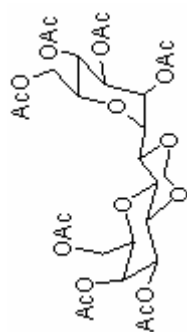
File created:
Friday, December 19, 2003
9:28 PM



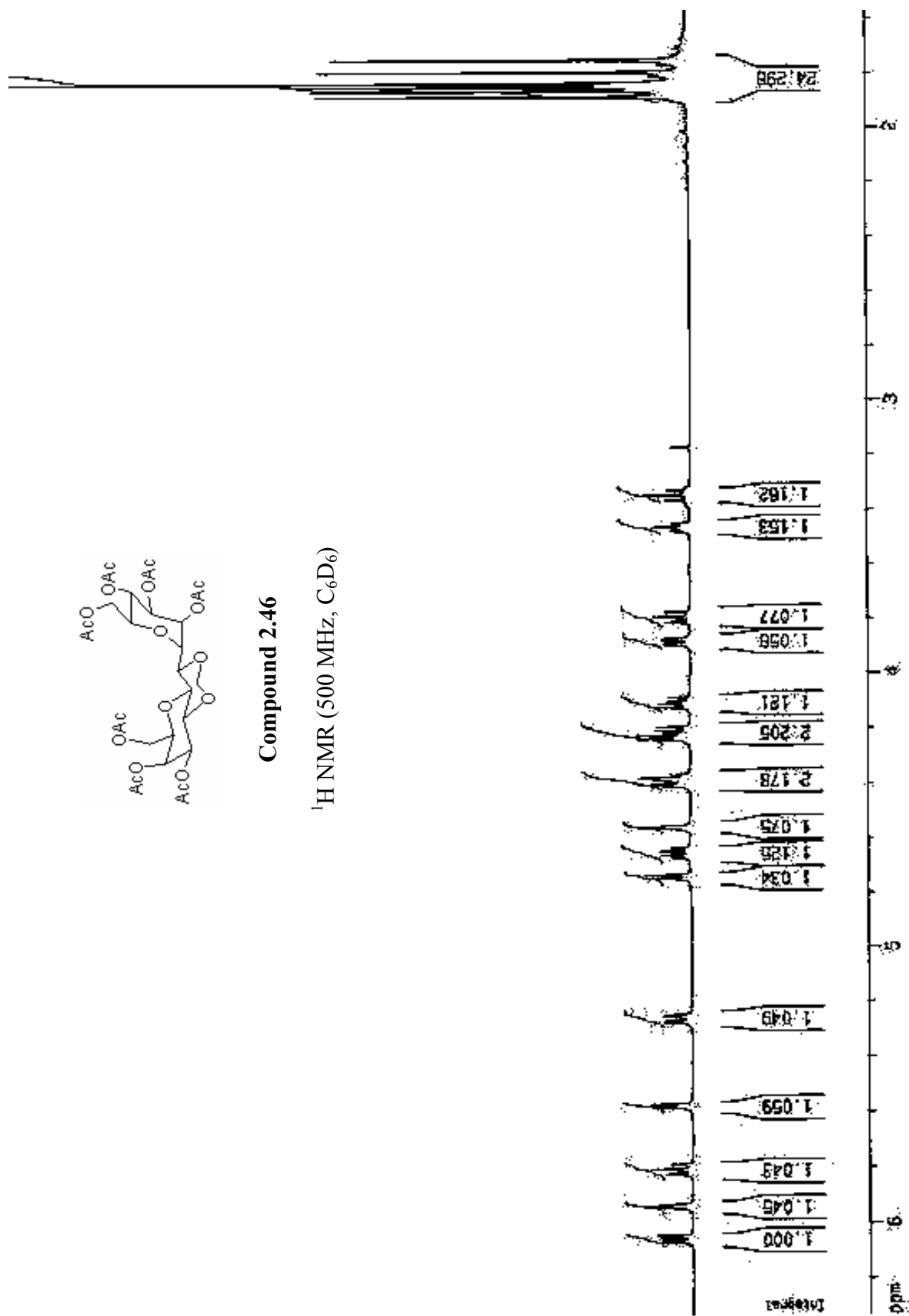
Compound 2.45

¹³C NMR (75 MHz, CDCl₃)



**Compound 2.46**

^1H NMR (500 MHz, C_6D_6)



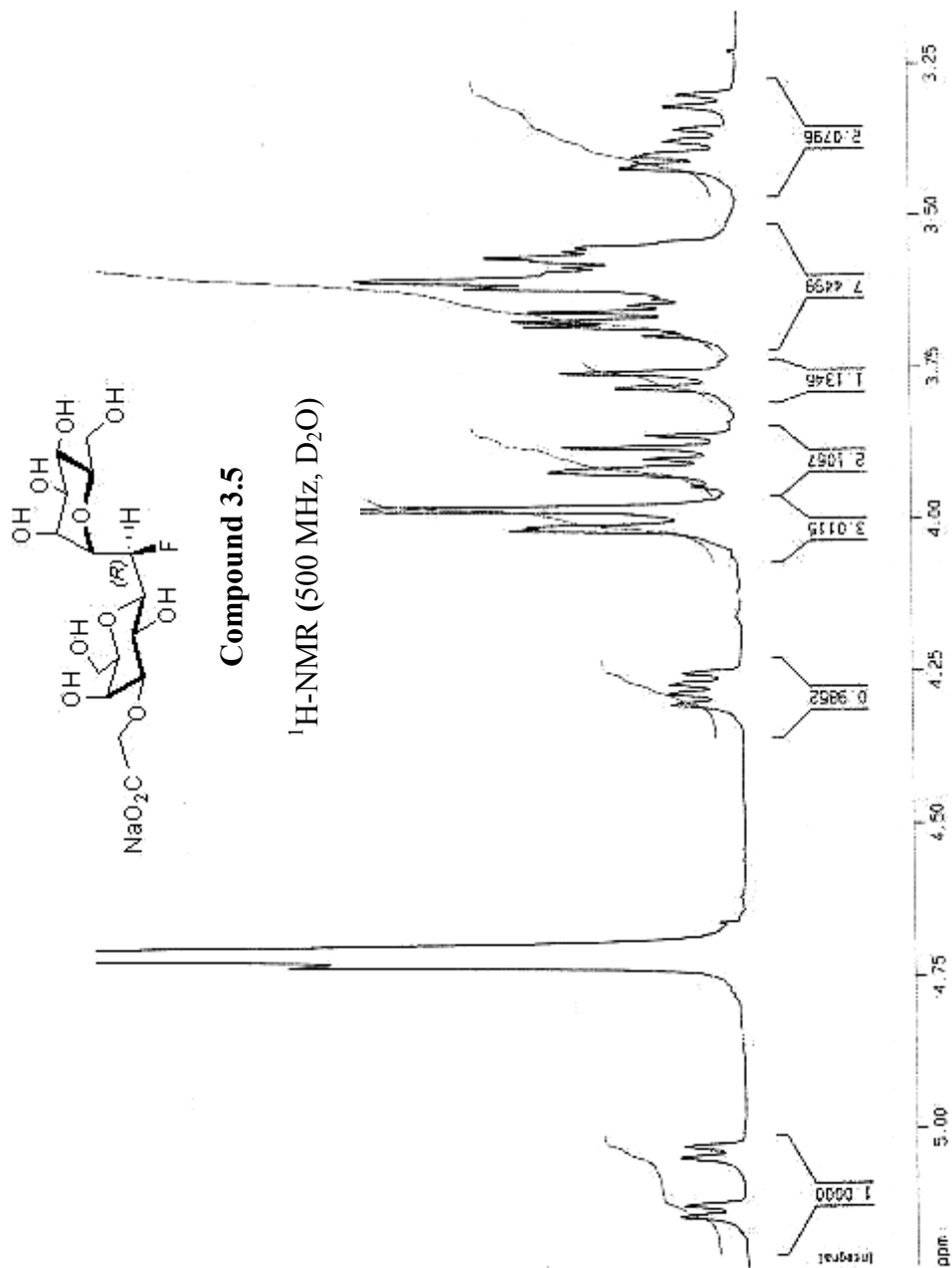
Current Data Parameters
 NAME 05090501
 EXPNO 22
 PROCNO 1

F2 - Acquisition Parameters
 Date_ 20090509
 Time 22.31
 INSTRUM spect
 PROBR40 3 mm CPDUL 13C
 PULPROG zgpg30
 TD 32768
 SOLVENT D2O
 NS 45
 DS 2
 SWH 8666.167 Hz
 FIDRES 0.203451 Hz
 AQ 2.457250 sec
 RG 161.3
 DM 75.000 usec
 DE 6.00 usec
 TE 303.15 K
 D1 1.0000000 sec
 NUC1 13C
 NUC2
 NUC3
 P1 18.00 usec
 PL1 12.20 dB
 SFO1 500.130000 MHz

F2 - Processing parameters
 SI 32768
 SF 500.130000 MHz
 MDW EM
 SSB 0
 LB 0.30 Hz
 GB 0
 PC 1.00

10 MHz plot parameters
 CX 22.00 cm
 CY 103.90 cm
 FIP 5.325 ppm
 F1 2613.21 Hz
 F2P 3.158 uPa
 F2 1579.37 Hz
 PRNCRW 0.09396 ppm/cm
 RZCM 46.99311 Hz/cm

***** CHANNEL f1 *****
 NUC1 13C
 P1 18.00 usec
 PL1 12.20 dB
 SFO1 500.130000 MHz



Current Data Parameters
 NAME 05090501
 EXPNO 98
 PROCNO 1

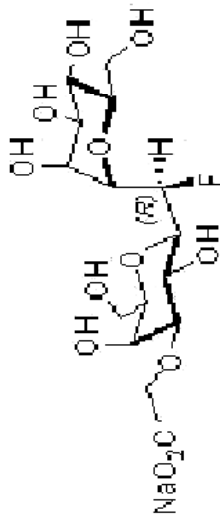
F2 - Acquisition Parameters

Date_ 20050509
 Time_ 21.44
 INSTRUM spect
 PROBHD 3 mm CPDUL 13C
 PULPROG zgpg
 TD 65418
 SOLVENT GBEHS
 NS 513
 DS 4
 SWH 30030.029 Hz
 FIDRES 0.459048 Hz
 AQ 1.0892763 sec
 RG 1000
 DW 16.650 usec
 DE 6.00 usec
 TE 303.0 K
 D1 2.0000000 sec
 d11 0.0300000 sec
 DELTA 1.8999998 sec
 MCREST 0.0000000 sec
 MCWRK 0.0150000 sec

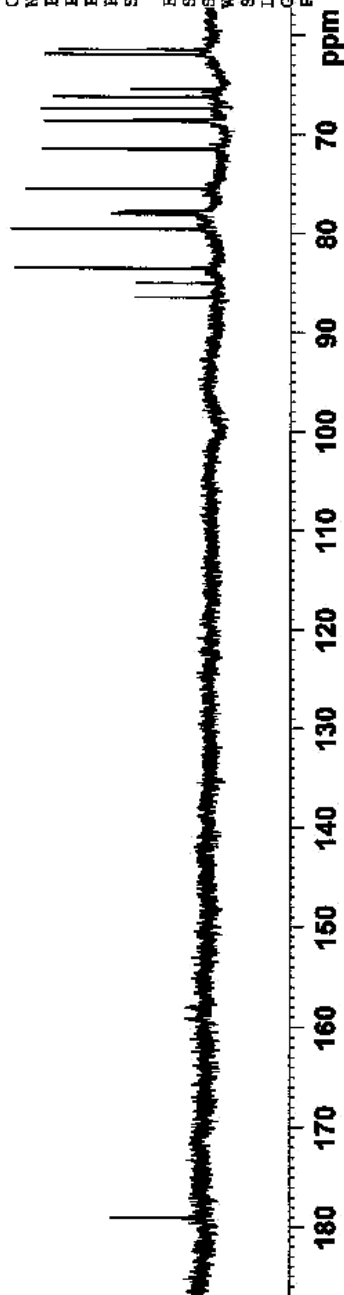
===== CHANNEL f1 =====
 NUC1 13C
 P1 10.00 usec
 PL1 18.00 dB
 SFO1 125.7703643 MHz

===== CHANNEL f2 =====
 CPDPRG2 waitz16
 NUC2 1H
 PCPD2 80.00 usec
 PL2 12.20 dB
 PL12 29.00 dB
 PL13 29.00 dB
 SFO2 500.1320005 MHz

F2 - Processing parameters
 SI 65536
 ST 125.7577201 MHz
 NDW EM
 SSB 0
 LB 2.00 Hz
 GB 0
 PC 0.20



Compound 3.5

¹³C-NMR (125 MHz, D₂O)

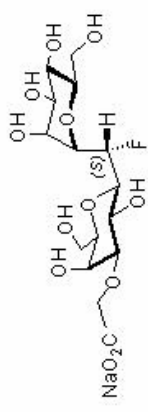
Current Data Parameters
 NAME 05130501
 EXPNO 27
 PROCNO 1

F2 - Acquisition Parameters
 Date_ 20050513
 Time 19.47
 INSTRUM spect
 PROBHD 3 mm CPDUL 13C
 PULPROG zg30
 TD 32768
 SOLVENT C6D6
 NS 21
 DS 2
 SWH 6666.667 Hz
 FIDRES 0.203451 Hz
 AQ 2.4577250 sec
 RG 64
 DW 75.000 usec
 DE 6.00 usec
 TE 303.0 K
 D1 1.0000000 sec
 MCREST 0.0000000 sec
 MCWPK 0.01500000 sec

***** CHANNEL f1 *****
 NUC1 1H
 P1 10.00 usec
 PL1 12.20 dB
 SFO1 500.1330885 MHz

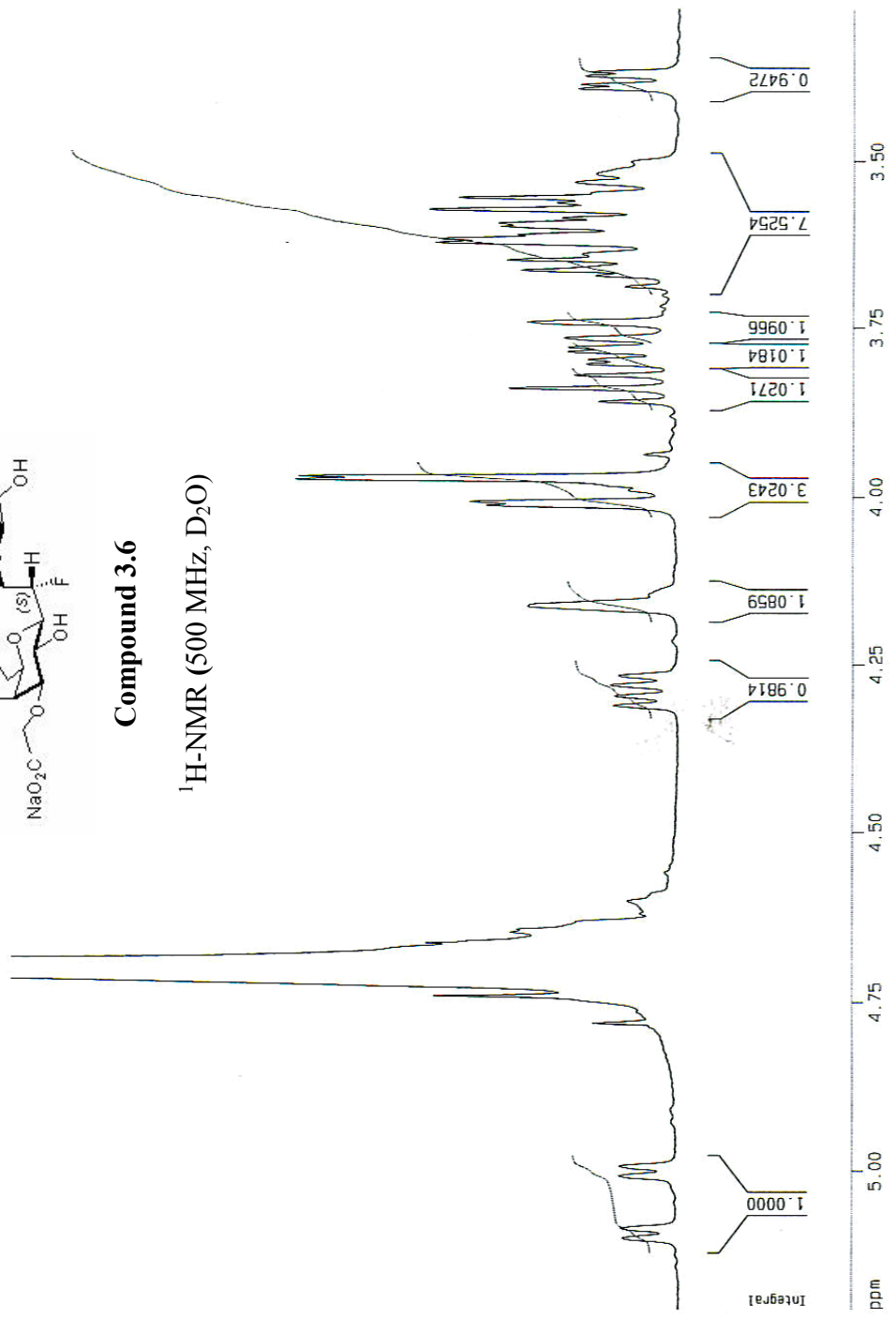
F2 - Processing parameters
 SI 32768
 SF 500.1300000 MHz
 KDW EM
 SSB 0
 LB 0.30 Hz
 GB 0
 PC 1.00

1D NMR plot parameters
 CX 22.00 cm
 CY 744.96 cm
 F1P 5.205 ppm
 F1 2603.40 Hz
 F2P 3.272 ppm
 F2 1636.21 Hz
 PPMCK 0.08790 ppm/cm
 HZCM 43.96318 Hz/cm



Compound 3.6

¹H-NMR (500 MHz, D₂O)



Daughter Data Parameters
 NAME 10659201
 EXPNO 46
 PROCNO 1

F2 - Acquisition Parameters

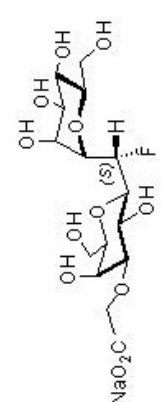
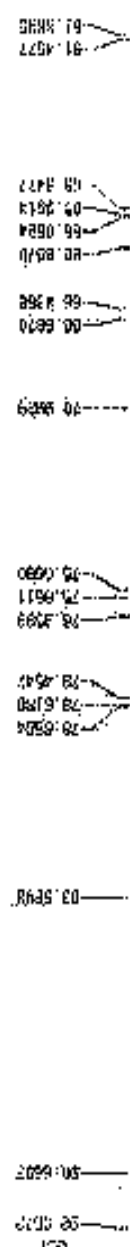
Date_ 20051008
 Time 9:18
 INSTRUM spect
 PULPROG zgpg30
 F2 == DMU-31C-1
 TD 65536
 SFO 500.1330005 MHz
 SOLVENT DMSO
 NS 4
 DS 4
 SWH 36039.083 Hz
 FIDRES 0.47832 Hz
 AQ 1.0818410 sec
 SFO 500.1330005 MHz
 C1 19.560 MHz
 DZ 0.00 MHz
 HZ 300.0 K
 D1 4.00000000 sec
 D11 0.02000000 sec
 D12 0.01000000 sec
 D13 0.01000000 sec
 D14 0.01000000 sec
 D15 0.01500000 sec

PROBHD 5MM QNP1H1
 P1 130
 PL1 0.00 dB
 PL2 0.00 dB
 PL3 0.00 dB
 PL4 120.7708603 MHz

===== CHANNEL f2 =====
 NUC1 13C
 P2 130
 PL2 0.00 dB
 PL3 0.00 dB
 PL4 120.7708603 MHz

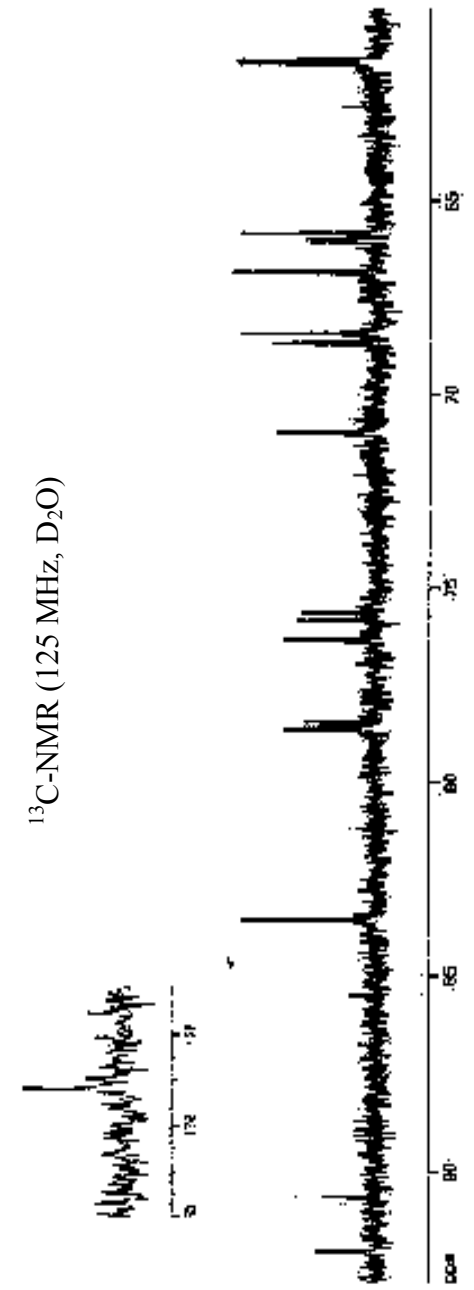
===== CHANNEL f1 =====
 NUC2 1H
 P1 90.00 MHz
 PL1 0.00 dB
 PL2 0.00 dB
 PL3 0.00 dB
 PL4 500.1330005 MHz

F2 - Processing parameters
 SI 32768
 SF 500.1330005 MHz
 WDF 0
 SFB 0
 LB 1.00 Hz
 GB 0
 PC 0.20
 EQ 80 MHz pulv. parametric
 CK 25.00 dB
 CX 3.00 dB
 F1 50.0000000 MHz
 F2 500.1330005 MHz
 PCYC 1
 JACO 101.84000 MHz



Compound 3.6

¹³C-NMR (125 MHz, D₂O)



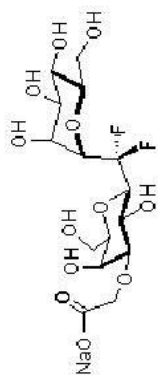
Current Data Parameters
 NAME 09190601
 EXPNO 42
 PROCNO 1

F2 - Acquisition Parameters
 Date_ 20060919
 Time 19.04
 INSTRUM spect
 PROBD 5 mm DUL 130-1
 PULPROG zg
 TD 32768
 SOLVENT D2O
 NS 725
 DS 2
 SMH 8012.820 Hz
 FIDRES 0.244532 Hz
 AQ 2.0446356 sec
 RG 362
 ON 62.400 usec
 DE 5.00 usec
 TE 303.0 K
 D1 1.00000000 sec
 MCREST 0.00000000 sec
 MCKRK 0.01500000 sec

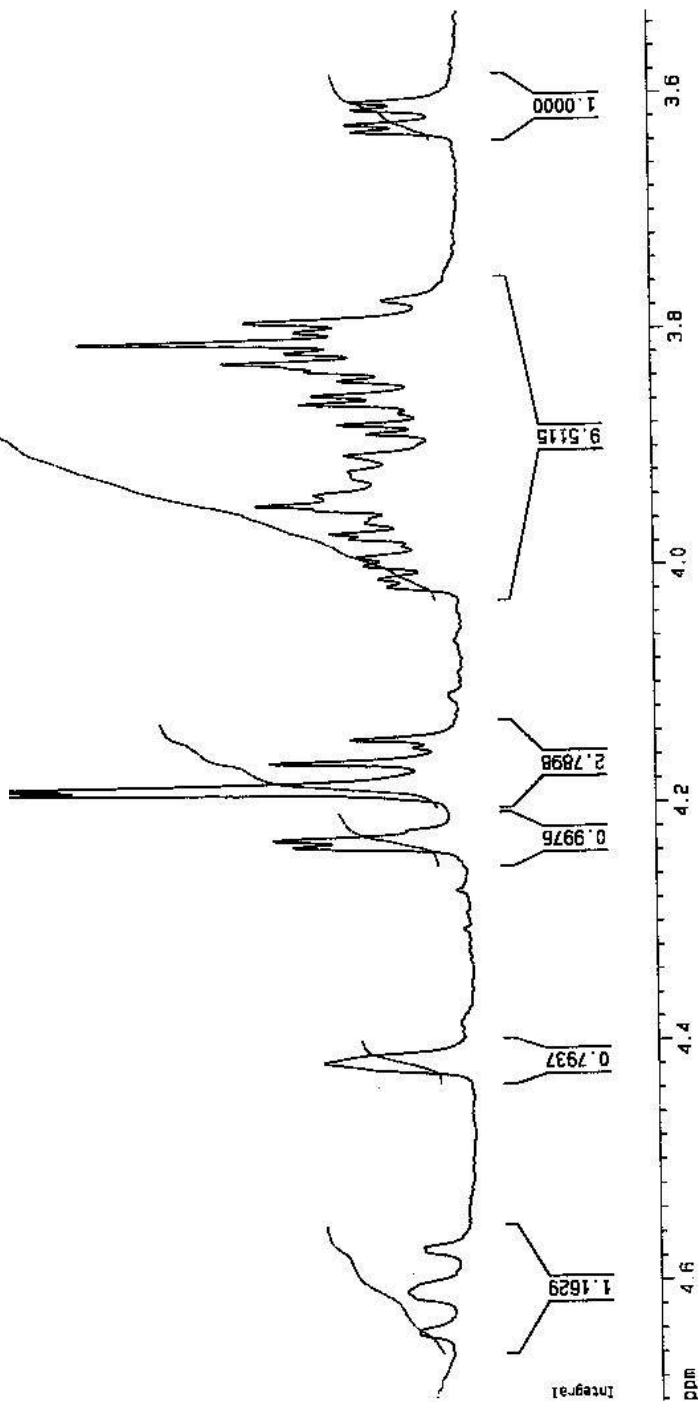
***** CHANNEL f1 *****
 NUC1 1H
 P1 6.20 usec
 PL1 -3.00 dB
 SFO1 500.1330885 MHz

F2 - Processing parameters
 SI 32768
 SF 500.1299226 MHz
 MDM EM
 SSB 0
 LB 0.80 Hz
 GB 0
 PC 1.00

1D NMR plot parameters
 CX 22.00 cm
 CY 691.81 cm
 F1P 4.702 ppm
 F1 2351.54 Hz
 F2P 3.529 ppm
 F2 1764.90 Hz
 PPMCK 0.05332 ppm/cm
 HZCM 26.66553 Hz/cm



Compound 3.7

¹H NMR (500 MHz, D₂O)

Current Data Parameters
 NAME 10130601
 EXPNO 41
 PROCNO 1

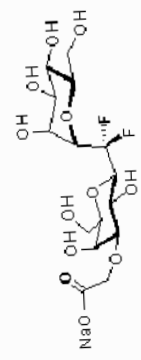
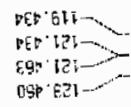
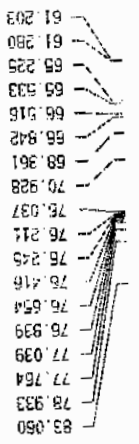
F2 - Acquisition Parameters
 Date_ 20061013
 Time 21.55
 INSTRUM spect
 PROBHD 5 mm CPOCH 13C
 PULPROG zgpg
 TD 65536
 SOLVENT D2O
 NS 23950
 DS 4
 SFO1 30030.029 Hz
 FIDRES 0.458222 Hz
 AQ 1.0912410 sec
 RG 724.1
 DIK 18.650 usec
 CE 6.00 usec
 TE 303.0 K
 D1 2.00000000 sec
 d11 0.03000000 sec
 DELTA 1.89999999 sec
 MCREST 0.00000000 sec
 MCNMRK 0.01500000 sec

CHANNEL f1 13C
 NUC1 13C
 P1 9.25 usec
 PL1 3.50 dB
 SFO1 125.7703643 MHz

CHANNEL f2 maltz16
 CPOPRG2 1H
 NUC2 1H
 P1 80.00 usec
 PL2 120.00 dB
 PL12 17.00 dB
 PL13 20.00 dB
 SFO2 500.1320005 MHz

F2 - Processing parameters
 SI 32768
 SF 125.7577690 MHz
 EQ 0
 NSB 0
 LB 1.00 Hz
 GB 0
 PC 0.20

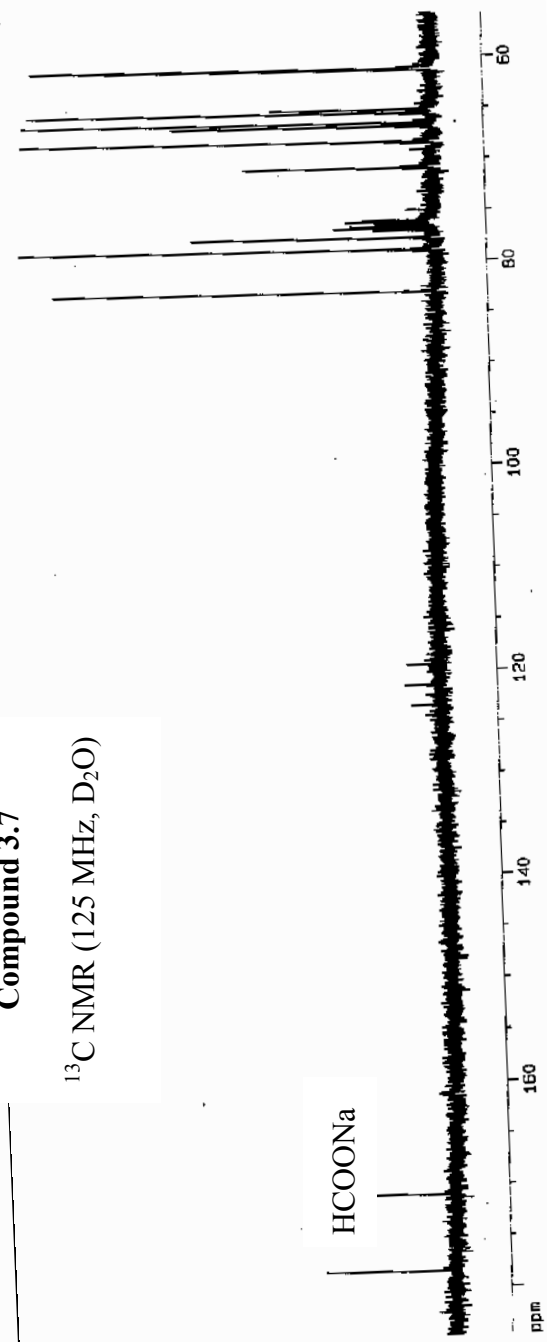
1D NMR plot parameters
 CX 22.00 cm
 CY 7.98 cm
 FIP 184.688 ppm
 F1 2328.95 Hz
 F2 55.740 ppm
 PPMCH 7009.74 Hz
 HZCH 5.86127 Hz/cm
 HZCN 737.10071 Hz/cm



Compound 3.7

¹³C NMR (125 MHz, D₂O)

HCOONa



Current Data Parameters
 NAME 07130601
 EXPNO 26
 PROCNO 1

F2 - Acquisition Parameters

Date_ 20050713
 Time 19.24
 INSTRUM spect
 PROBHD 5 mm DUL 13C-1
 PULPROG zg30
 TD 32768
 SOLVENT C6D6
 NS 70
 DS 2
 SWH 8012.820 Hz
 FIDRES 0.244532 Hz
 AQ 2.0448355 sec
 RG 362
 DM 62.400 usec
 DE 6.00 usec
 TE 303.0 K
 D1 1.00000000 sec
 MCREST 0.00000000 sec
 MCHWK 0.01500000 sec

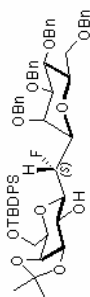
***** CHANNEL f1 *****

NUC1 ¹H
 P1 9.30 usec
 PL1 -3.00 dB
 SFO1 500.1330685 MHz

F2 - Processing parameters

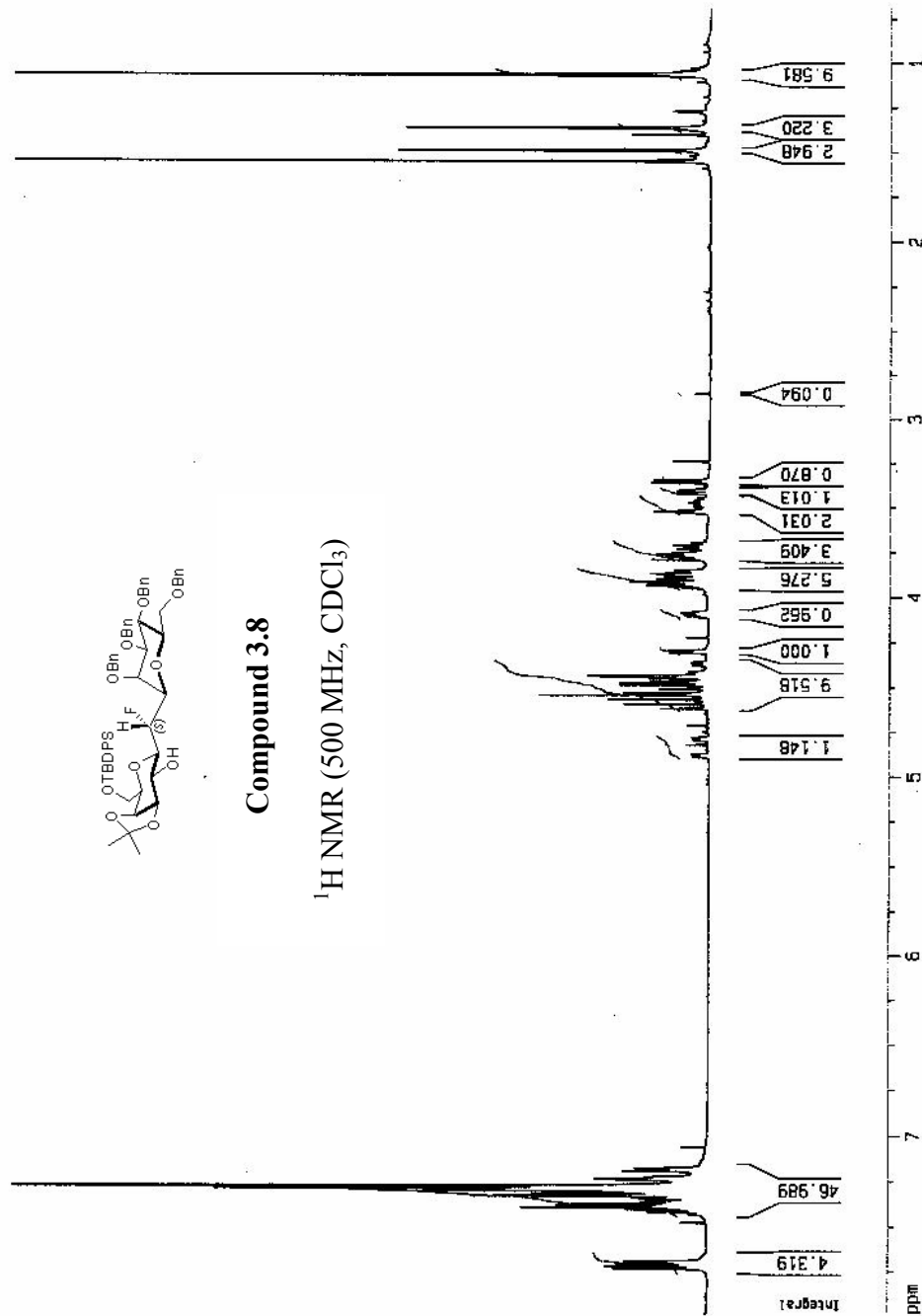
SI 32768
 SF 500.1300081 MHz
 WDW EM
 SSB 0
 LB 0.30 Hz
 GB 0
 PC 1.00

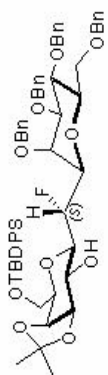
1D NMR plot parameters
 CX 22.00 cm
 CY 80.84 cm
 F1P 7.987 ppm
 F1 3994.30 Hz
 F2P 0.682 ppm
 F2 345.93 Hz
 PPMCM 0.33158 ppm/cm
 HZCM 165.83498 Hz/cm



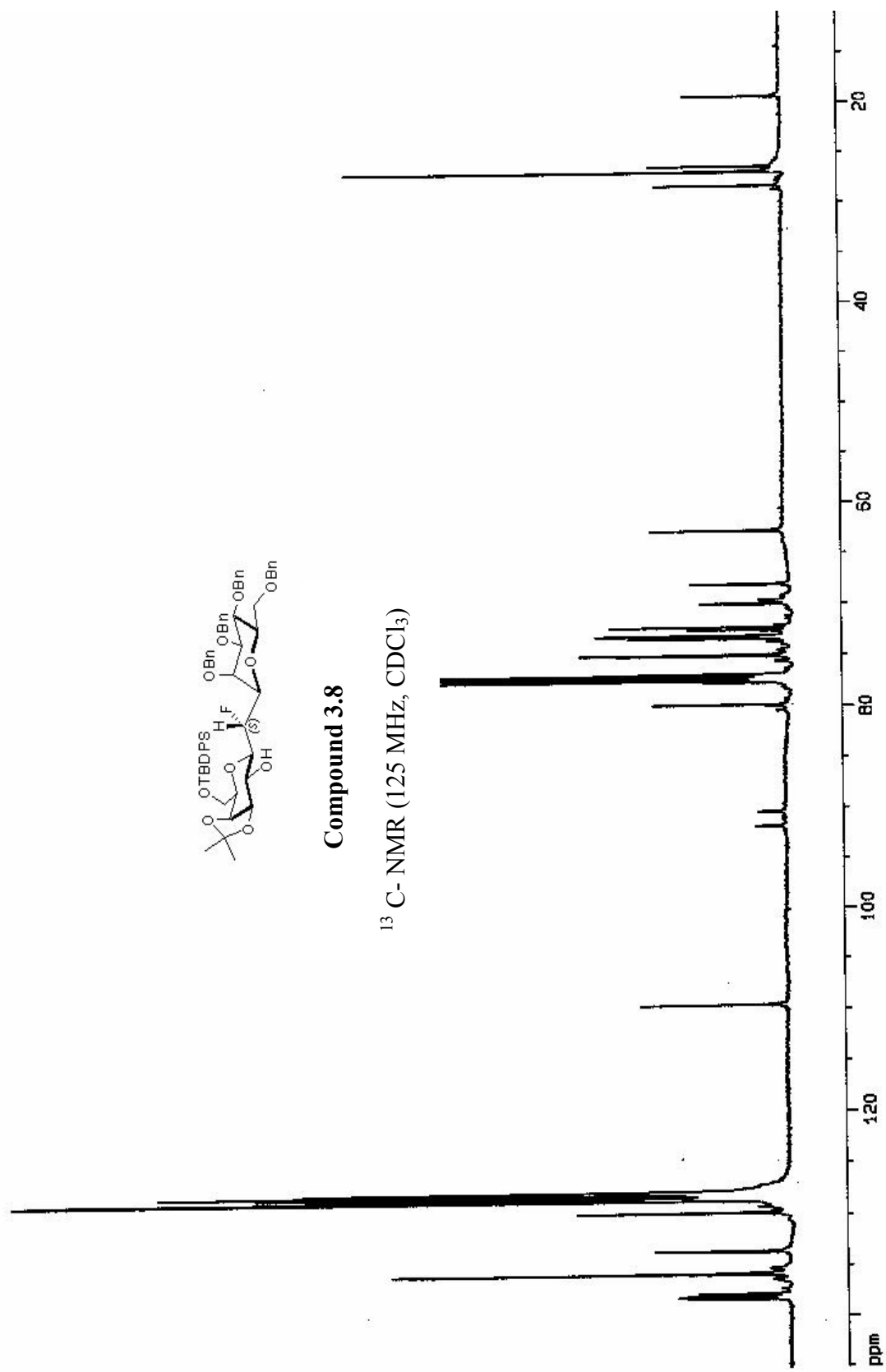
Compound 3.8

¹H NMR (500 MHz, CDCl₃)



**Compound 3.8**

^{13}C -NMR (125 MHz, CDCl_3)



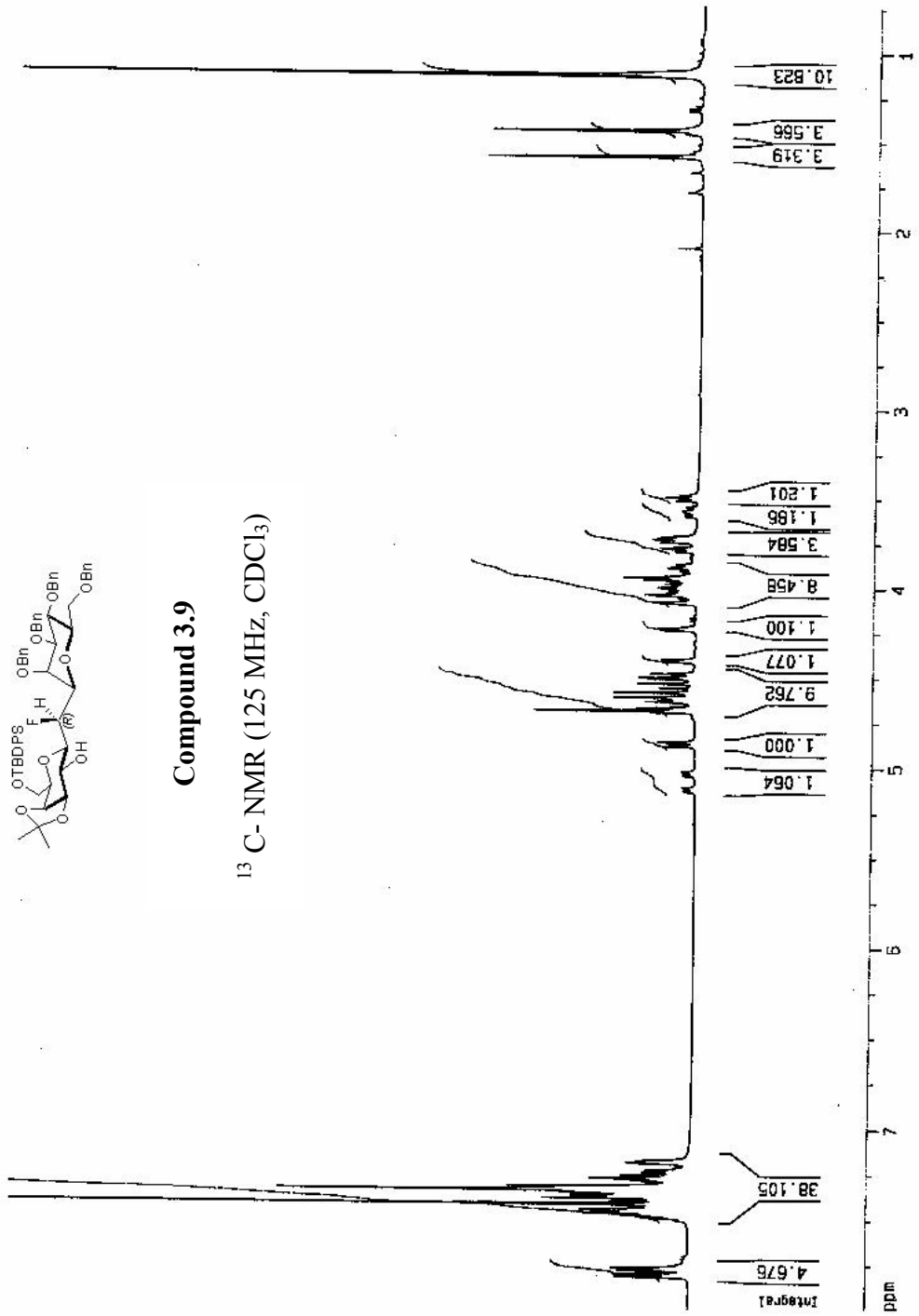
Current Data Parameters
 NAME 04140501
 EXPNO 16
 PROCNO 1

F2 - Acquisition Parameters
 Date_ 20050414
 Time 16.16
 INSTRUM spect
 PROBHD 3 mm CPUL 13C
 PULPROG zg30
 TD 32768
 SOLVENT CDCl3
 NS B
 DS 2
 SMH 6666.667 Hz
 FIDRES 0.203451 Hz
 AQ 2.4577250 sec
 RB 10.1
 DM 75.000 usec
 DE 6.00 usec
 TE 303.0 K
 D1 1.00000000 sec
 MCREST 0.00000000 sec
 MCMRK 0.01500000 sec

***** CHANNEL f1 *****
 NUC1 1H
 P1 10.00 usec
 PL1 12.20 dB
 SF01 500.1330885 MHz

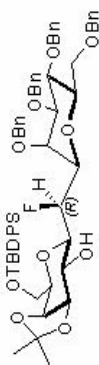
F2 - Processing parameters
 SI 32768
 SF 500.1300061 MHz
 WDM EM
 SSB 0
 LB 0.30 Hz
 GB 0
 PC 1.00

1D NMR plot parameters
 CX 22.00 cm
 CY 12.14 cm
 FIP 7.998 ppm
 F1 3999.67 Hz
 F2P 0.736 ppm
 F2 368.21 Hz
 PPMCH 0.33007 ppm/cm
 HZCM 165.07542 Hz/cm



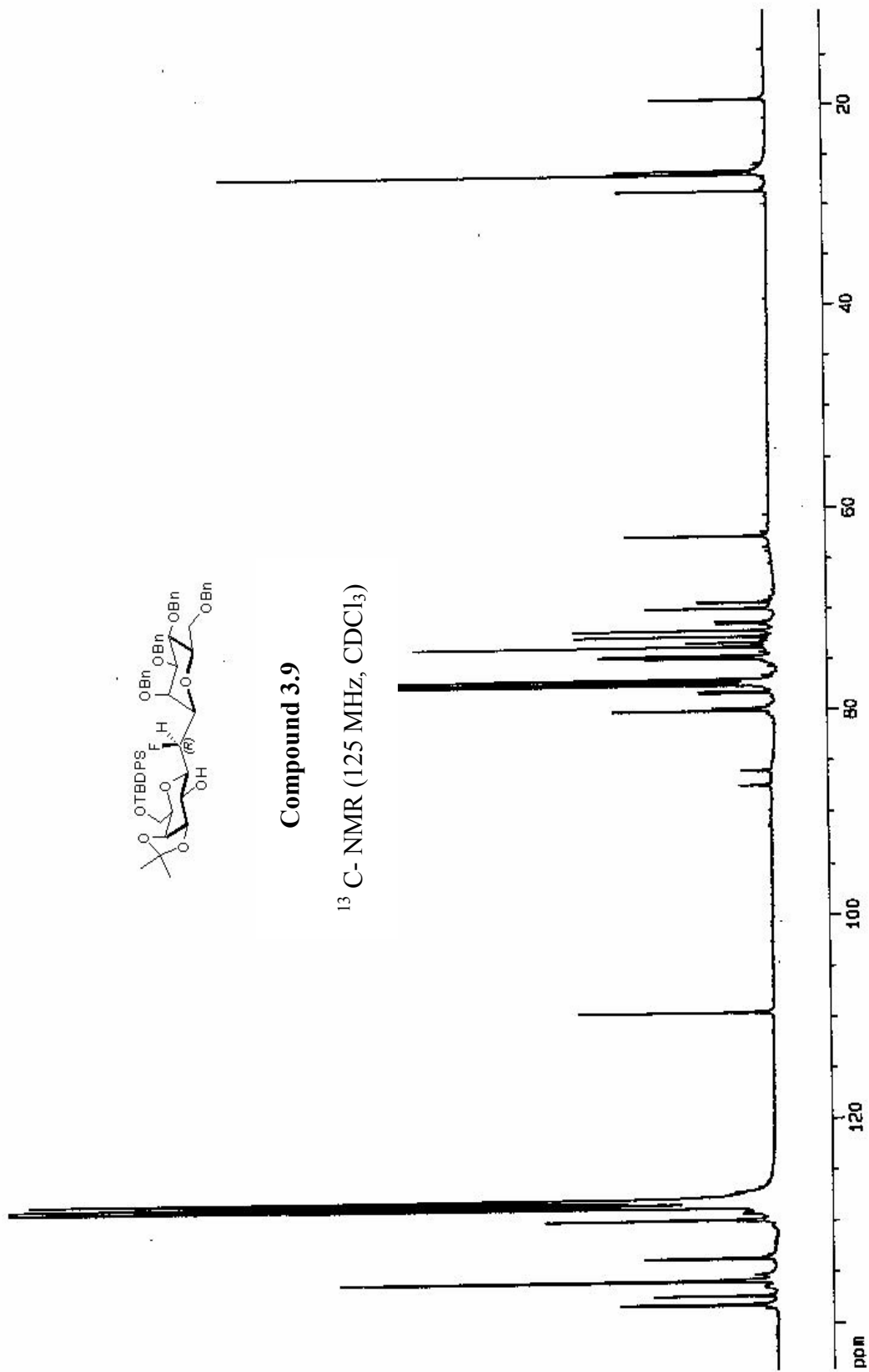
Compound 3.9

^{13}C -NMR (125 MHz, CDCl_3)



Compound 3.9

^{13}C -NMR (125 MHz, CDCl_3)

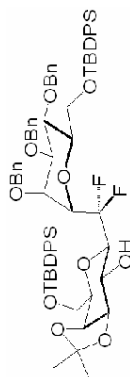


Current Data Parameters
 NAME 06160501
 EXPNO 14
 PROCNO 1

F2 - Acquisition Parameters
 Date_ 20060616
 Time 17.37
 INSTRUM spect
 PROBD 5 mm DUL 13C-1
 PULPROG zg30
 TD 32768
 SOLVENT DMS-D6
 NS 64
 DS 2
 SWH 6012.820 Hz
 FIDRES 0.244532 Hz
 AQ 2.0446355 sec
 RG 226.1
 DM 62.400 usec
 DE 6.00 usec
 TE 303.0 K
 .D1 1.00000000 sec
 MCHRES 0.00000000 sec
 MCHWK 0.01500000 sec

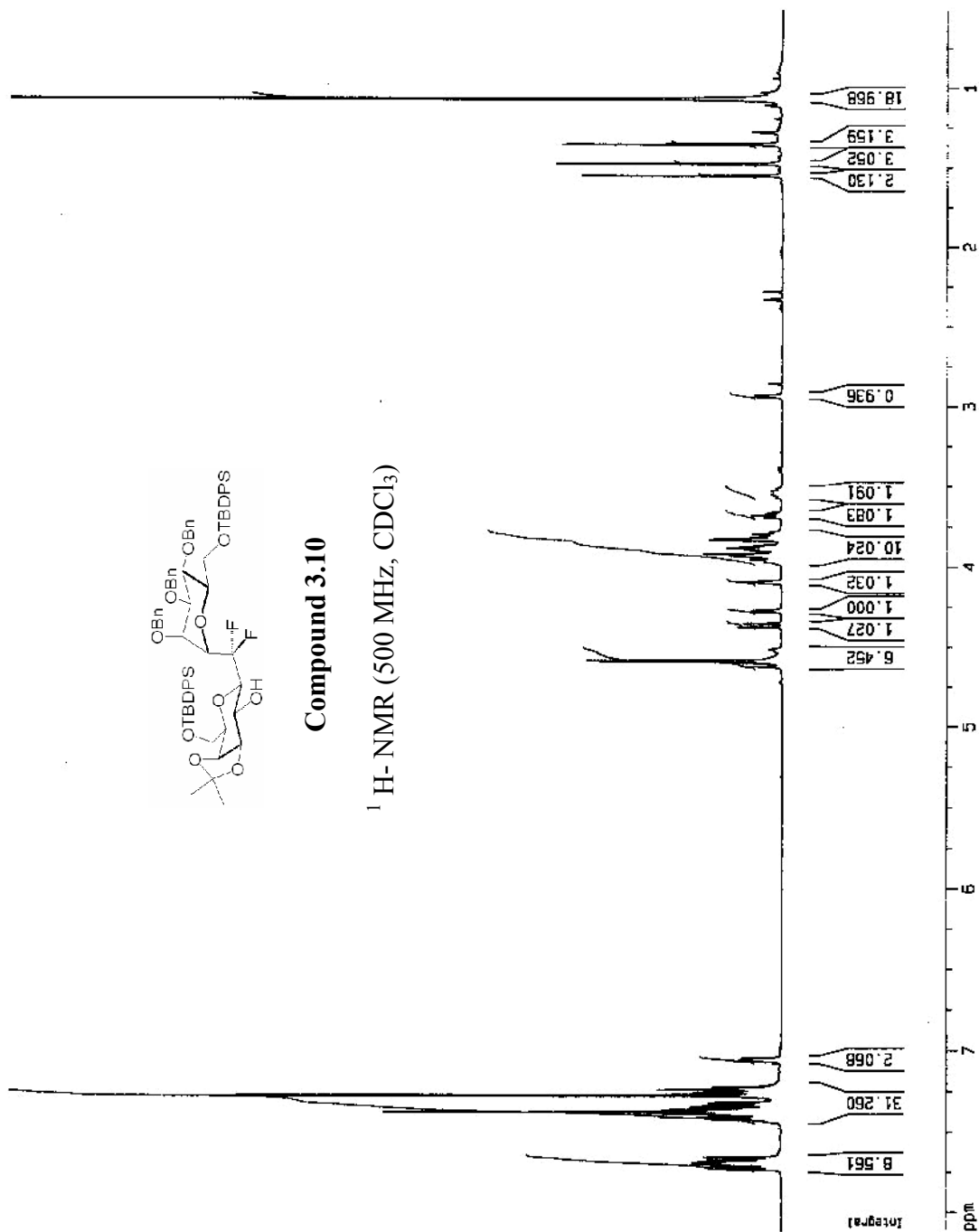
----- CHANNEL f1 -----
 NUC1 1H
 P1 9.30 usec
 PL1 -3.00 dB
 SF01 500.1350885 MHz
 F2 - Processing parameters
 SI 32768
 SF 500.1300079 MHz
 WDW EM
 SSB 0
 LB 0.30 Hz
 GB 0
 PC 1.00

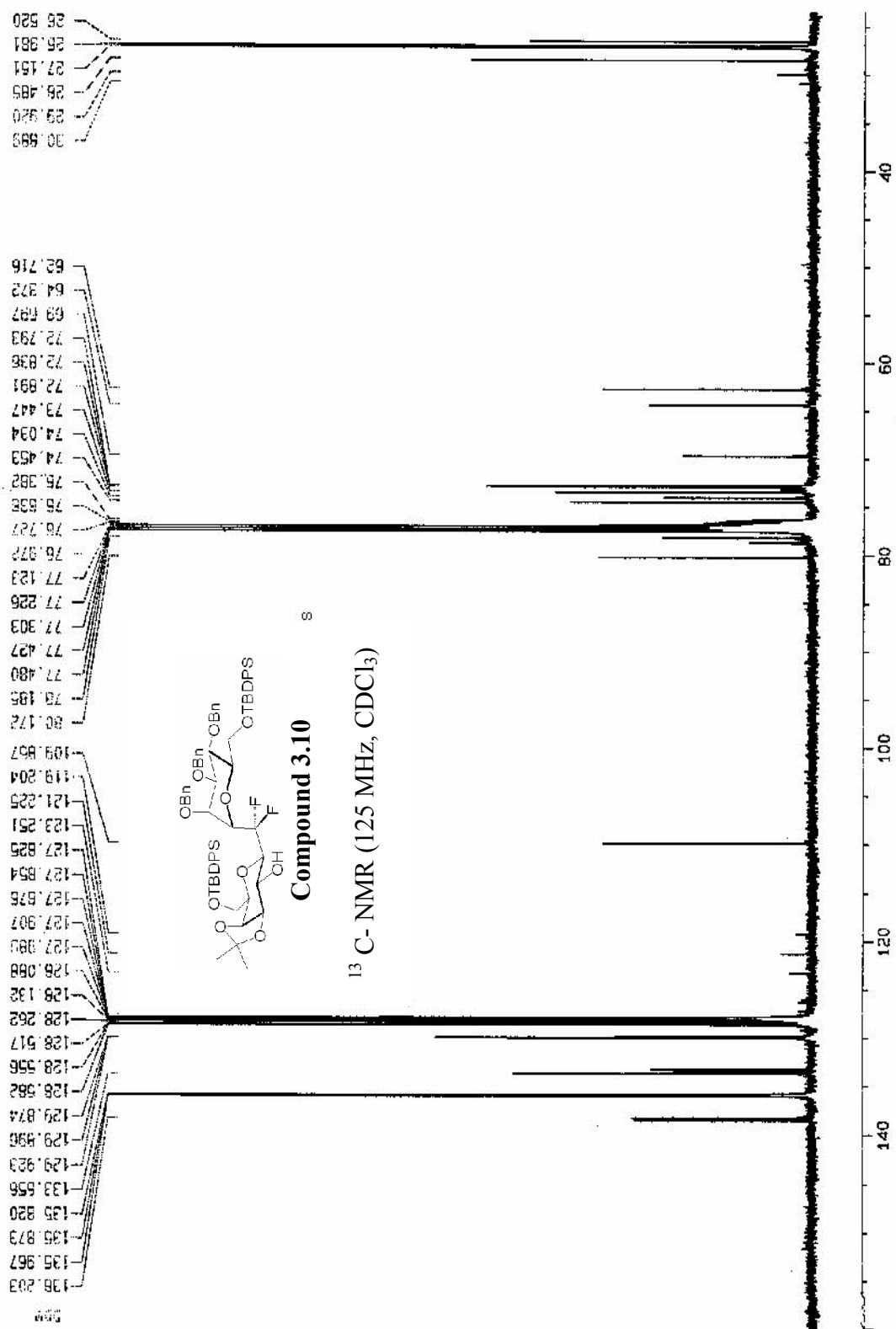
1D NMR plot parameters
 CX 22.00 cm
 CY 27.86 cm
 F1P 8.116 ppm
 F1 4059.26 Hz
 F2P 0.513 ppm
 F2 256.76 Hz
 PPMCM 0.34559 ppm/cm
 HZCM 172.84190 Hz/cm

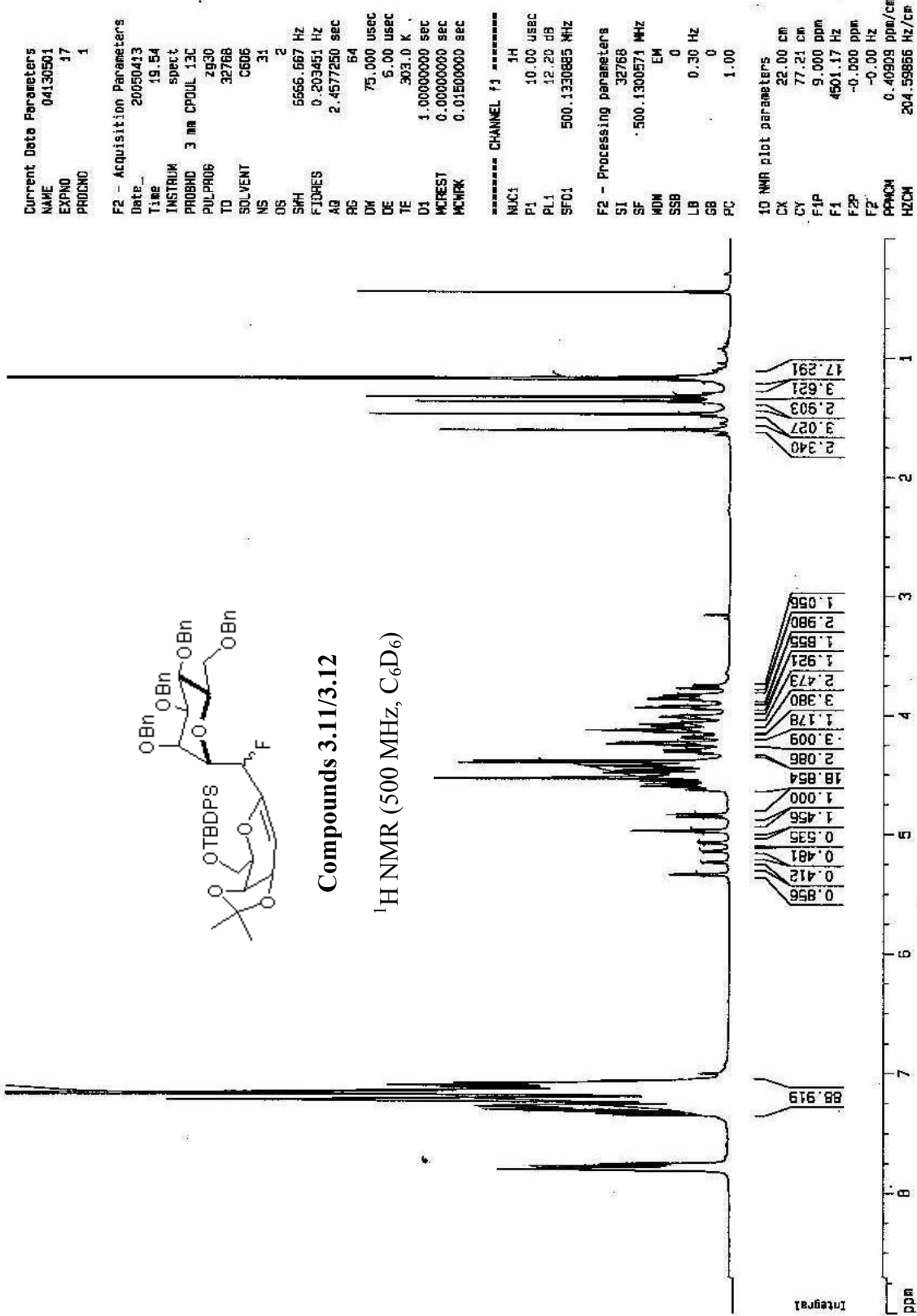


Compound 3.10

¹H-NMR (500 MHz, CDCl₃)







Current D7
 Parameters
 NAME 04130501
 EXPNO 99
 PROCNO 1

F2 - Acquisition Parameters

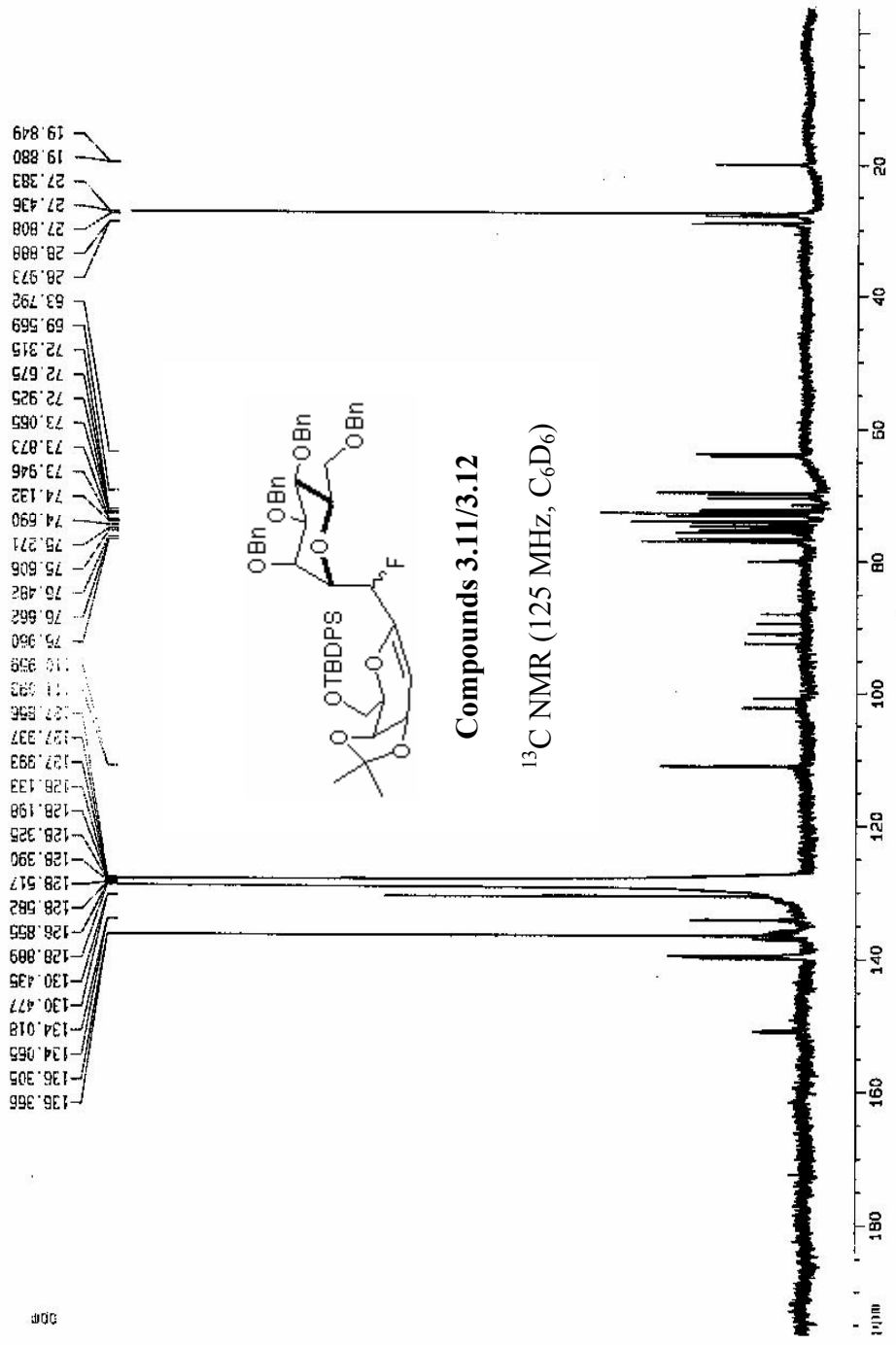
Date_ 20050413
 Time 20.19
 INSTRUM spect
 PROBHD 3 mm CPOLL 13C
 PULPROG zgpg
 TO 65418
 SOLVENT CDCl3
 NS 429
 DS 4
 SWH 30030.029 Hz
 FIDRES 0.459048 Hz
 AQ 1.0892763 sec
 RB 1000
 DN 16.650 usec
 DE 6.00 usec
 TE 303.0 K
 D1 2.00000000 sec
 d11 0.03000050 sec
 DELTA 1.89999988 sec
 PCREST 0.00000000 sec
 MCPRK 0.01500000 sec

===== CHANNEL f1 =====
 NU1 13C
 P1 10.00 usec
 PL1 18.00 dB
 SF01 125.7703643 MHz

===== CHANNEL f2 =====
 CPDPRG2 waltz16
 NU2 1H
 P2 80.00 usec
 PL2 12.20 dB
 PL12 29.00 dB
 PL13 29.00 dB
 SF02 500.1320005 MHz

F2 - Processing parameters
 SI 65536
 SF 125.7577182 MHz
 EQ EM
 SSB 0
 LB 2.00 Hz
 GB 0
 PC 0.20

3D NMR plot parameters
 CX 22.00 cm
 CY 780.85 cm
 F1P 196.349 ppm
 F1 24692.40 Hz
 F2P -3.704 ppm
 F2 -495.84 Hz
 PPMCF 9.09333 ppm/cm
 HZCM 1143.56627 Hz/cm



30P

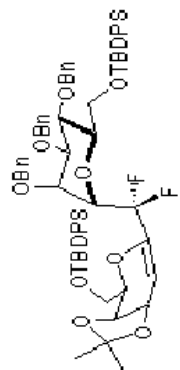
Current Data Parameters
 NAME 06090601
 EXPNO 22
 PROCNO 1

F2 - Acquisition Parameters
 Date_ 20060609
 Time 19:29
 INSTRUM spect
 PROBHD 5 mm DUL 13C-1
 PULPROG zg30
 TO 32768
 SOLVENT C606
 NS 8
 DS 2
 SWH 8012.820 Hz
 FIDRES 0.244532 Hz
 AQ 2.0446356 sec
 RG 71.8
 DM 62.400 usec
 DE 6.00 usec
 TE 303.0 K
 D1 1.00000000 sec
 MCREST 0.00000000 sec
 MCWRR 0.01500000 sec

===== CHANNEL f1 =====
 NUC1 1H
 P1 9.30 usec
 PL1 -3.00 dB
 SF01 500.1330885 MHz

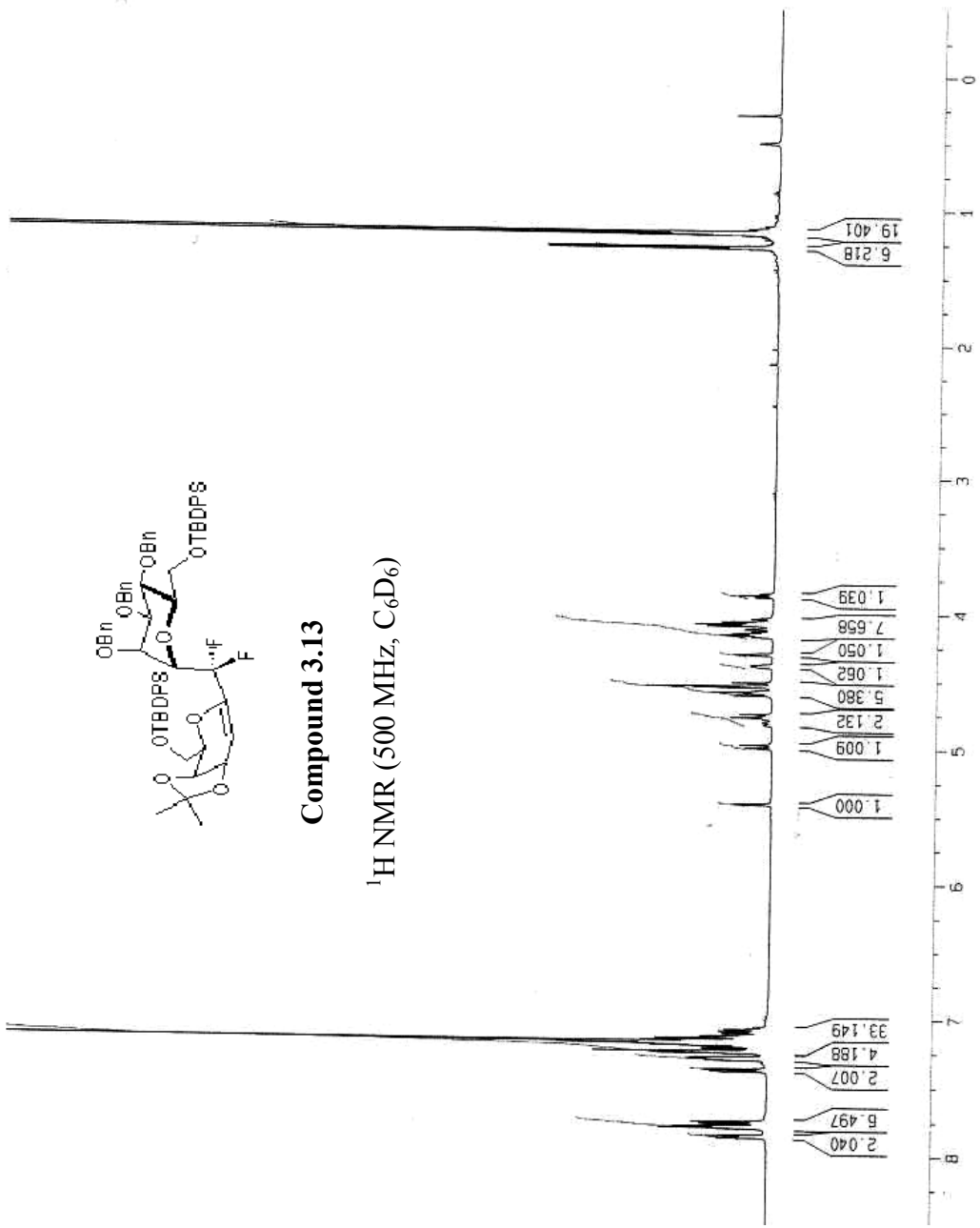
F2 - Processing parameters
 SI 32768
 SF 500.1300545 MHz
 WDW EM
 SSB 0
 LB 0.30 Hz
 GB 0
 PC 1.00

3D NMR plot parameters
 CX 22.00 cm
 CY 16.90 cm
 F1P 9.000 ppm
 F1 4501.17 Hz
 F2P -0.500 ppm
 F2 -250.07 Hz
 PPRCK 0.43182 ppm/cm
 HZCM 215.96526 Hz/cm



Compound 3.13

¹H NMR (500 MHz, C₆D₆)



Current Parameters
 NAME 05090601
 EXPNO 20
 PROCNO 1

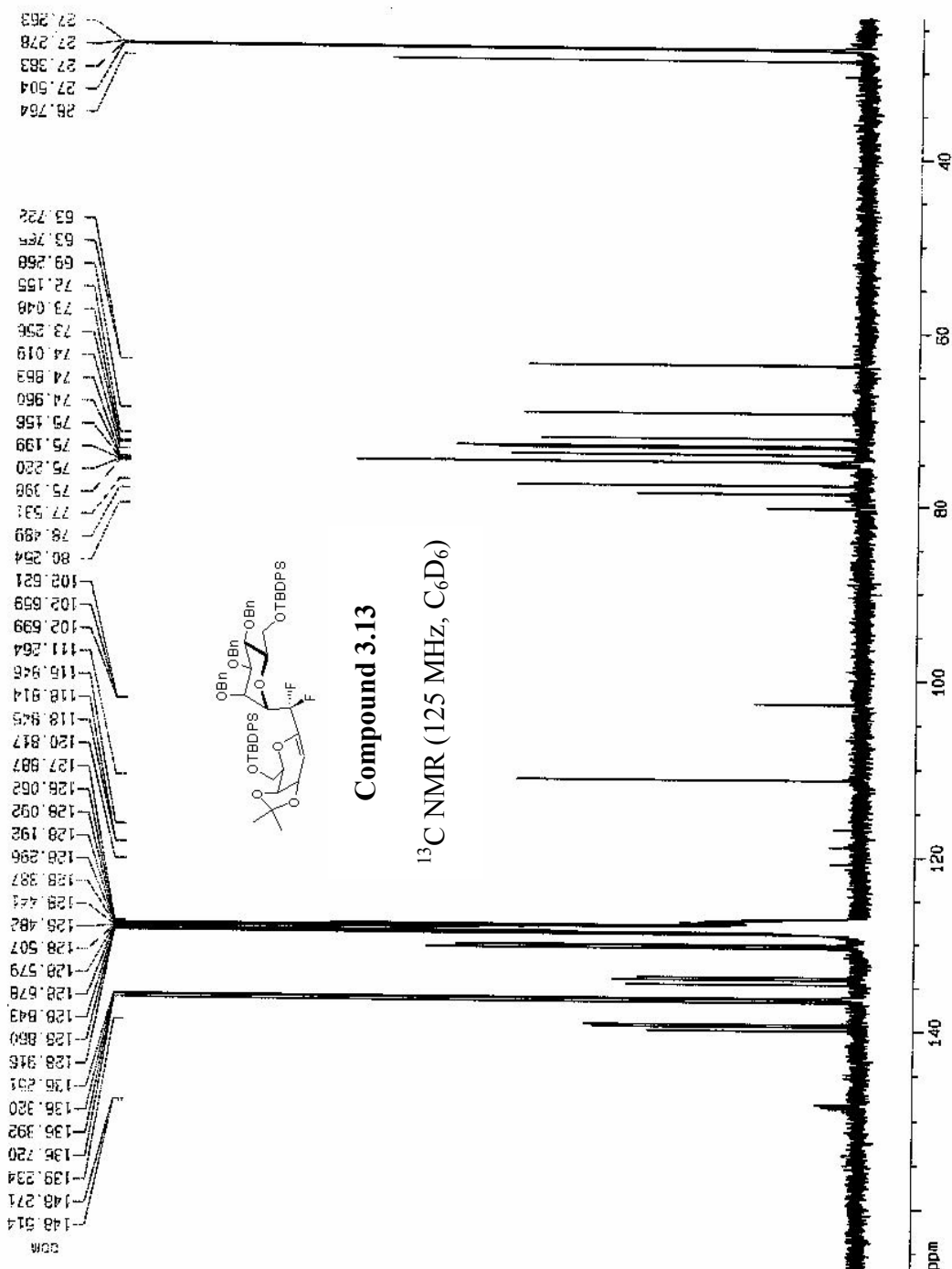
F2 - Acquisition Parameters
 Date_ 20060609
 Time 18.48
 INSTRUM spect
 PROBHD 5 mm QNP 13C-1
 PULPROG zgpg
 TD 65536
 SOLVENT CDCl3
 NS 801
 DS 4
 SWH 30030.029 Hz
 FIDRES 0.458222 Hz
 AQ 1.0912410 sec
 RG 724.1
 DM 16.650 usec
 DE 6.00 usec
 TE 303.0 K
 D1 2.00000000 sec
 d11 0.03000000 sec
 DELTA 1.89999995 sec
 MCREST 0.00000000 sec
 MCHPRK 0.01500000 sec

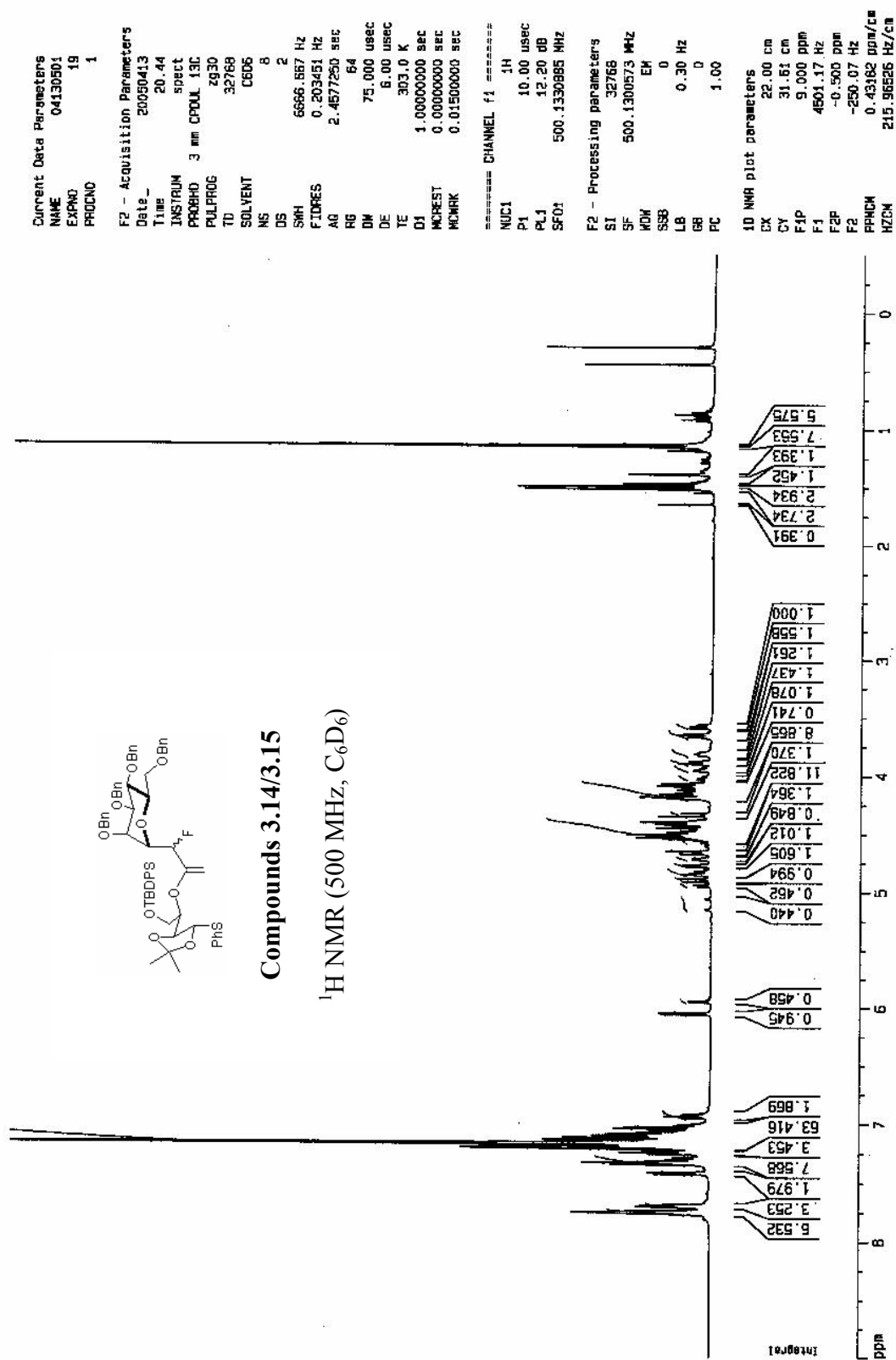
***** CHANNEL f1 *****
 NUC1 13C
 P1 5.00 usec
 PL1 0.00 dB
 SF01 125.7703643 MHz

***** CHANNEL f2 *****
 CPDPRG2 waltz16
 NUC2 1H
 PCDP2 80.00 usec
 PL2 -3.00 dB
 PL12 15.89 dB
 PL13 15.50 dB
 SF02 500.1320005 MHz

F2 - Processing parameters
 SI 32786
 SF 125.7577145 MHz
 MDW EN
 SSB 0
 LB 1.00 Hz
 GB 0
 PC 0.20

1D NMR plot parameters
 CX 22.00 cm
 CY 177.34 cm
 FJP 166.819 ppm
 F1 20978.75 Hz
 F2 23.725 ppm
 F3 2983.65 Hz
 PPMCM 6.50424 ppm/cm
 HZCM 817.95892 Hz/cm





Current Dir Parameters
 NAME 04130501
 EXPNO 99
 PROCNO 1

F2 - Acquisition Parameters

Date_ 20050413
 Time 21.02
 INSTRUM spect
 PROBHD 3 mm CPUL 13C
 PULPROG zgpg
 TD 65418
 SOLVENT CDCl3
 NS 512
 DS 4
 SMH 30030.029 Hz
 FIDRES 0.469048 Hz
 AQ 1.0892753 sec
 RG 1000
 DIH 15.950 usec
 DE 5.00 usec
 TE 303.0 K
 D1 2.00000000 sec
 d11 0.03000000 sec
 DELTA 1.89999998 sec
 MZTEST 0.00000000 sec
 MCNMR 0.01500000 sec

===== CHANNEL f1 =====
 NUC1 13C
 P1 10.00 usec
 PL1 18.00 dB
 SFO1 125.7703643 MHz

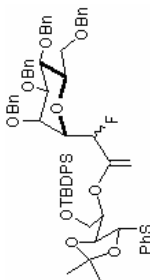
===== CHANNEL f2 =====
 CPDPRG2 waltz16
 NUC2 1H
 PCPD2 80.00 usec
 PL2 12.20 dB
 PL12 29.00 dB
 PL13 29.00 dB
 SFO2 500.1320005 MHz

F2 - Processing parameters

SI 65536
 SF 125.7577182 MHz
 MDM EM
 SSB 0
 LB 2.00 Hz
 BB 0
 PC 0.20

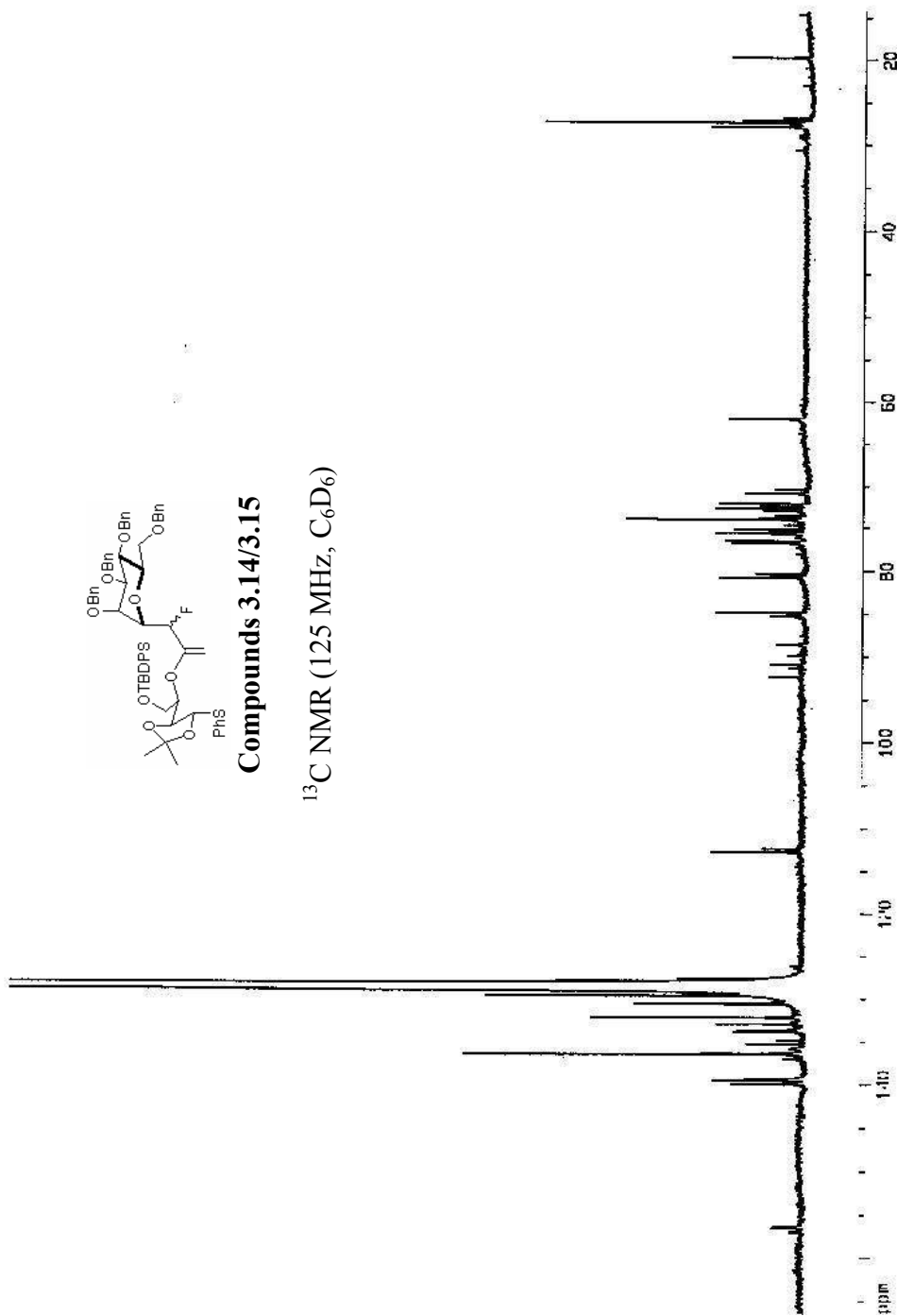
3D NMR plot parameters

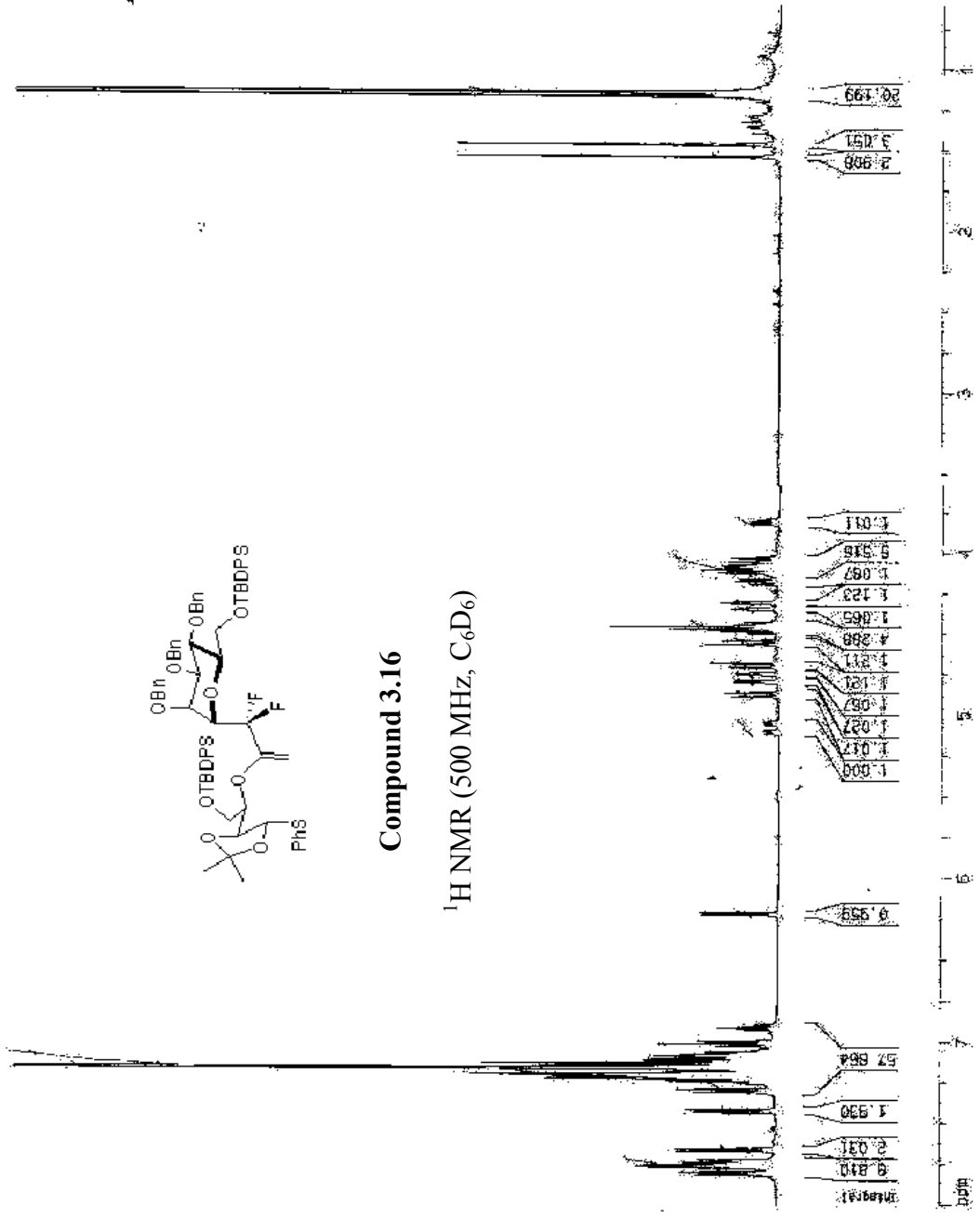
CX 22.00 cm
 CY 316.12 cm
 F1P 165.355 ppm
 F1 20920.49 Hz
 F2P 14.267 ppm
 F2 1795.70 Hz
 FWHM 6.91221 ppm/cm
 HZCM 869.26318 Hz/cm



Compounds 3.14/3.15

¹³C NMR (125 MHz, C₆D₆)





Compound 3.16

¹H NMR (500 MHz, C₆D₆)

Current Data Parameters
 Name: 05120804
 ExpNo: 23
 Procnm: 1

F2 - Acquisition Parameters
 Date_: 20060902
 Time: 17.35
 INSTRUM: spect
 PROGNO: 5 mp DUL 150-1
 PULPROG: zg30
 TD: 32768
 SFO1: 500.1301586
 AQ: 5
 F2: 8032.824 Hz
 SFO2: 0.240583 Hz
 FIDRES: 0.2348386 Hz
 AQRES: 111.3
 DM: 62.400.0584
 DE: 6.00.0845
 TE: 303.2 K
 D1: 1.00000000 sec
 MCHEBT: 0.00000000 sec
 ACQPRG: 0.00000000 sec
 NAMEPR: 0.00000000 sec

===== CHANNEL f1 =====
 NUC1: 1H
 P1: 9.00.0256
 PL1: 3.00.00
 SFO1: 500.1301586 MHz

F2 - Processing parameters
 SI: 32768
 SF: 500.1301586 MHz
 HN: 0
 HZ: 0.30 Hz
 L6: 0
 S6: 0
 PC: 0.00

10 NMR plot parameters
 G: 0.00 cm
 O: 35.27 cm
 FIP: 4.000.000
 FL: 4000.13 Hz
 P3P: 0.005.000
 FE: 302.35 Hz
 PRMR: 0.00000.00000
 HZP: 160.17182.16181

Current Dir Parameters
 NAME 06020501
 EXPNO 25
 PROCNO 1

F2 - Acquisition Parameters

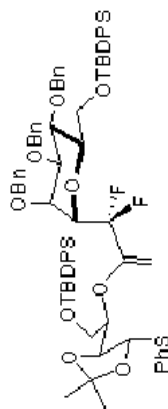
Date_ 20060602
 Time 28.33
 INSTRUM spect
 PROBHD 5 mm QNP 13C-1
 PULPROG zgpg30
 TD 65536
 SOLVENT CDCl3
 NS 21585
 DS 4
 SWH 30030.028 Hz
 FIDRES 0.458222 Hz
 AQ 1.0912410 sec
 RG 724.1
 DM 16.650 usec
 DE 6.00 usec
 TE 303.0 K
 D1 2.00000000 sec
 d11 0.03000000 sec
 DELTA 1.69999998 sec
 MCREST 0.00000000 sec
 MCNMR 0.01500000 sec

===== CHANNEL F1 =====
 NUC1 13C
 P1 5.00 usec
 PL1 0.00 dB
 SF01 125.7703643 MHz

===== CHANNEL F2 =====
 CPDPRG2 waltz16
 NUC2 1H
 P2 90.00 usec
 PL2 -3.00 dB
 PL12 15.69 dB
 PL13 18.50 dB
 SF02 500.1329095 MHz

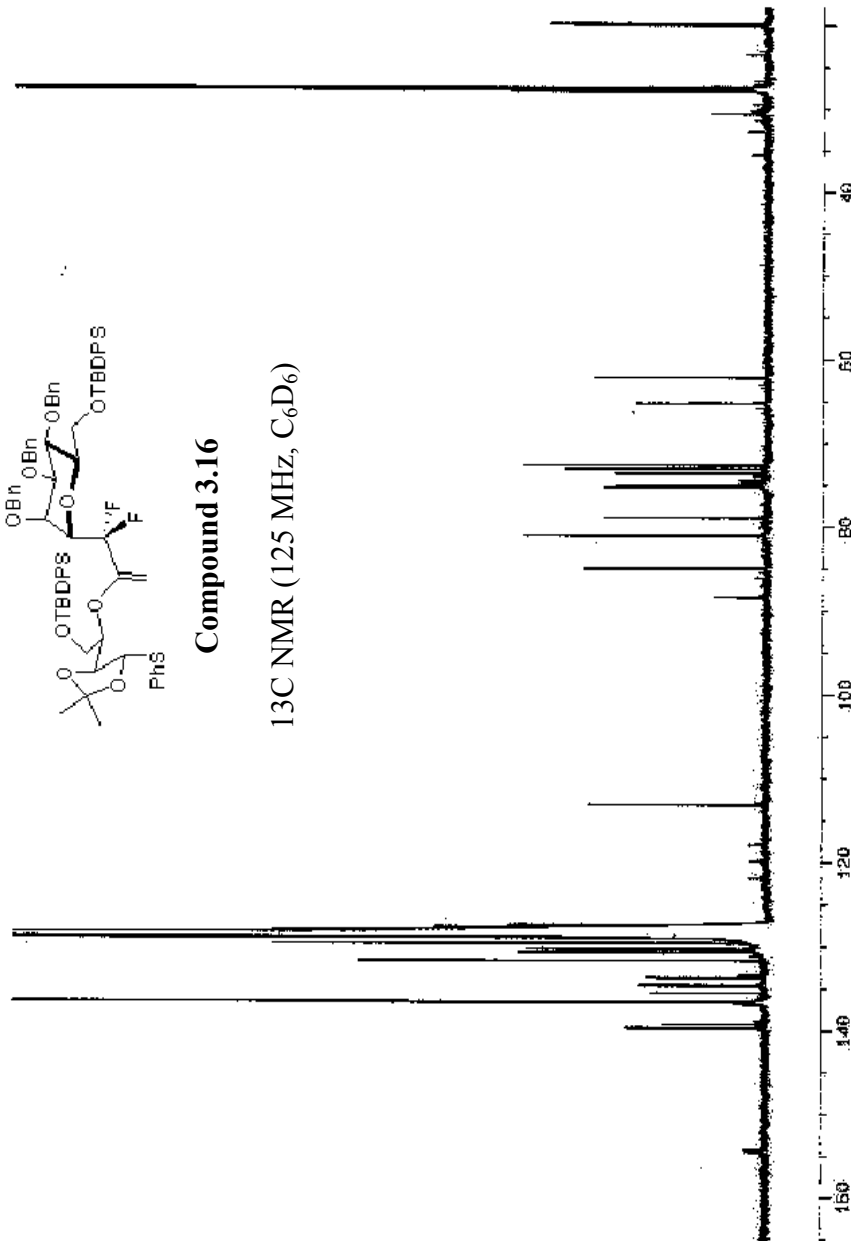
F2 - Processing parameters
 SI 32756
 SF 125.757135 MHz
 EQ
 NDN 0
 SSB 1.00 Hz
 LB 0
 GB 0
 PC 0.20

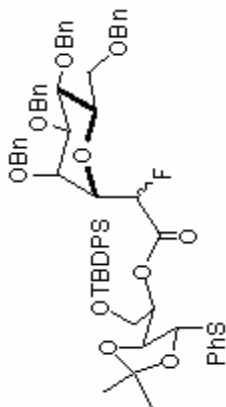
1D NMR plot parameters
 CX 22.00 cm
 CY 371.91 cm
 F1 187.492 00 Hz
 F2 23578.61 Hz
 F20 17.831 00 Hz
 F22 2242.40 Hz
 PPMX0 7.7188 ppm/cm
 HZCM 969.82606 Hz/cm



Compound 3.16

¹³C NMR (125 MHz, C₆D₆)





Compounds 3.17/3.18

¹H NMR (500 MHz, CDCl₃)

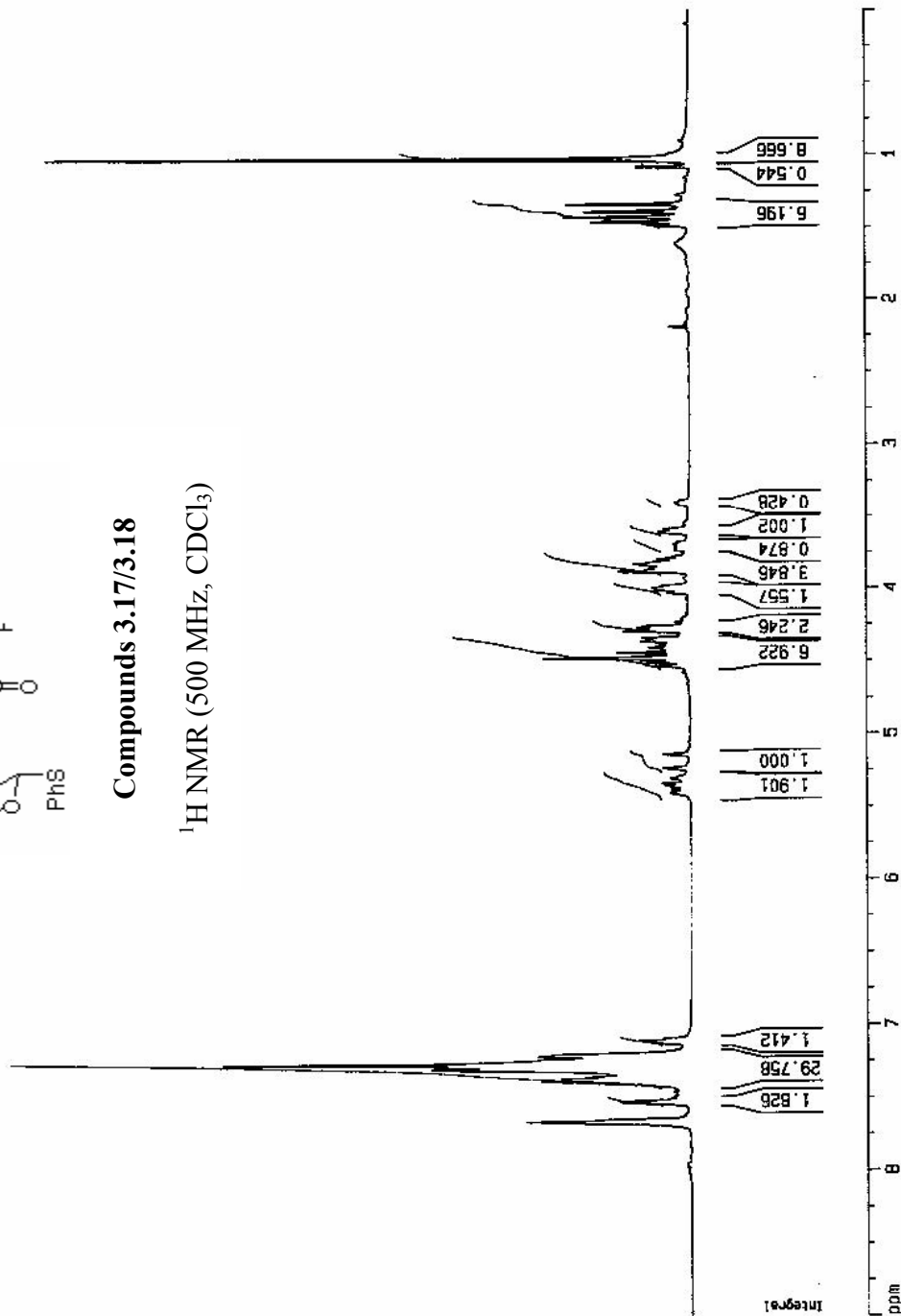
Current Data Parameters
 NAME 04040501
 EXPNO 27
 PROCNO 1

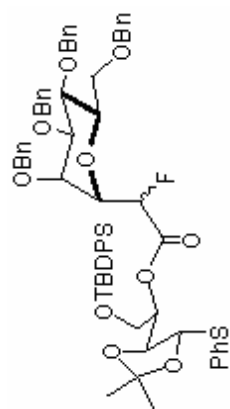
F2 - Acquisition Parameters
 Date_ 20050404
 Time 19.51
 INSTRUM spect
 PROBHD 3 mm CPDUL 13C
 PULPROG zg30
 TD 32768
 SOLVENT CDCl₃
 NS 8
 DS 2
 SWH 6566.667 Hz
 FIDRES 0.203451 Hz
 AQ 2.457250 sec
 RG 64
 DN 75.000 usec
 DE 6.00 usec
 TE 303.0 K
 D1 1.00000000 sec
 MCREST 0.00000000 sec
 MCWPK 0.01500000 sec

===== CHANNEL f1 =====
 NUC1 1H
 P1 10.00 usec
 PL1 12.20 dB
 SF01 500.1330885 MHz

F2 - Processing parameters
 SI 32768
 SF 500.1330886 MHz
 KW EM
 SSB 0
 LB 0.30 Hz
 BB 0
 PC 1.00

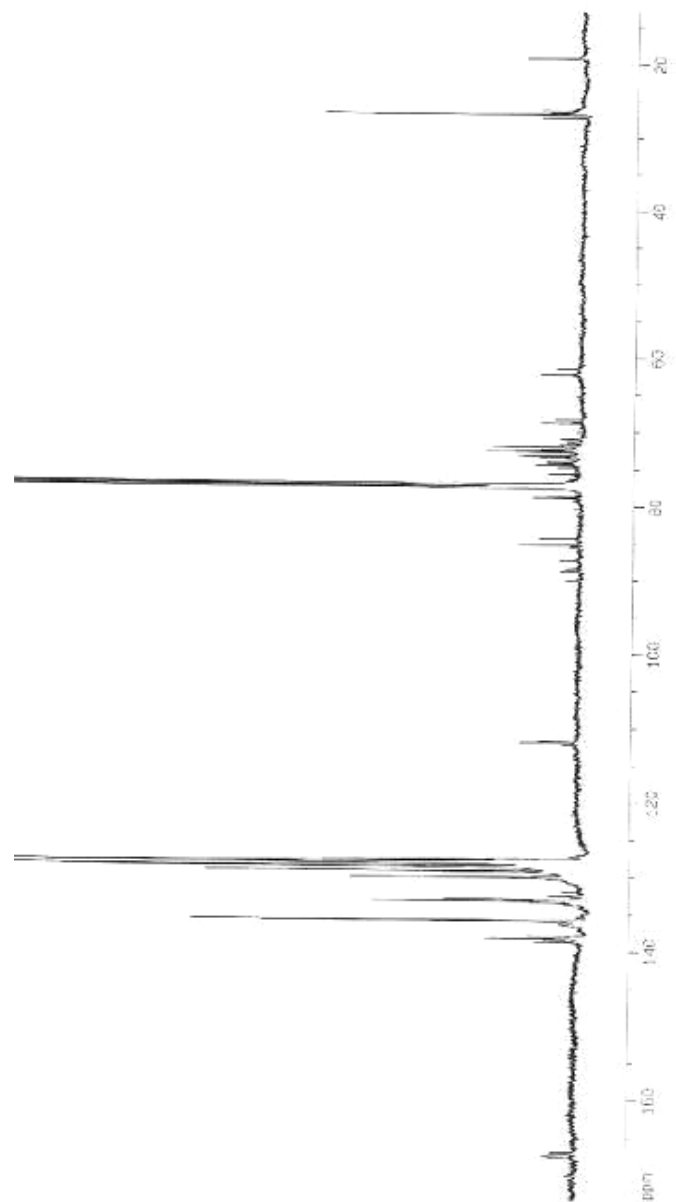
1D NMR plot parameters
 CX 22.00 cm
 CY 10.85 cm
 F1P 9.000 ppm
 F1 4501.17 Hz
 F2P -0.000 ppm
 F2 -0.00 Hz
 PPMCM 0.40509 ppm/cm
 HZCM 204.55865 Hz/cm





Compounds 3.17/3.18

¹H NMR (125 MHz, CDCl₃)



Current: 0 Parameters
 NAME / 0404503
 EXPNO 39
 PROCNO 1

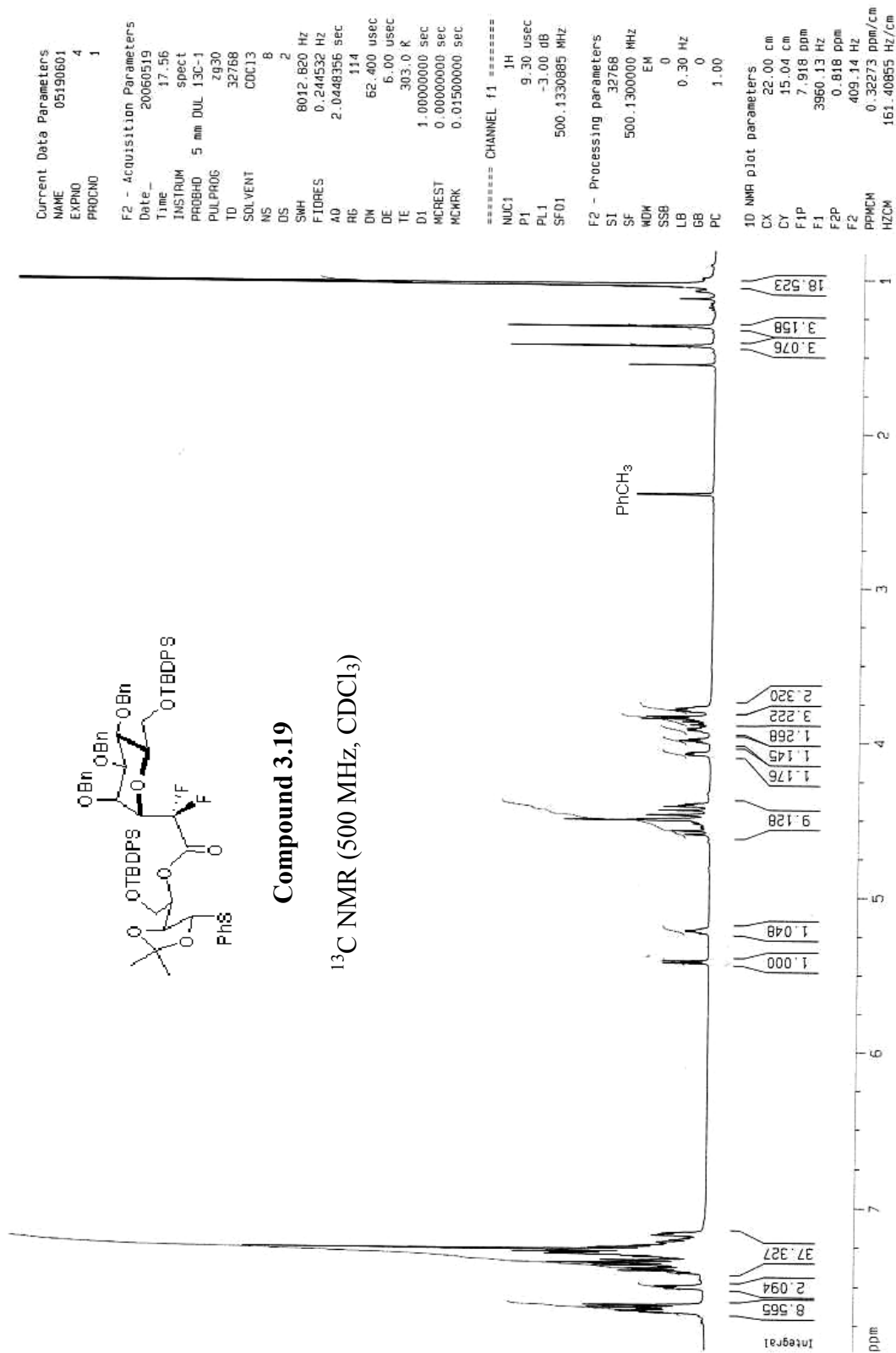
F2 - Acquisition Parameters
 Date_ 20050604
 Time 20.02
 INSTRUM spect
 PROBHD 3 mm CPDPR 13C
 PULPROG zgpg30
 TD 65544
 SFO 125.130
 DD 0.13
 NS 256
 DS 4
 SWH 30030.629 Hz
 FIDRES 0.456045 Hz
 AQ 1.0962783 sec
 SFO 125.130
 IN 15.000000 sec
 DE 1.0000000 sec
 TE 303.2 K
 D1 2.00000000 sec
 D2 2.00000000 sec
 DELTA 1.80000000 sec
 DELTA 1.80000000 sec
 DELTA 1.80000000 sec
 DELTA 1.80000000 sec
 DELTA 1.80000000 sec
 DELTA 1.80000000 sec

----- CHANNEL f1 -----
 NUC1 13C
 P1 18.00 usec
 PL1 05.00 dB
 SFO1 125.7705613 MHz

----- CHANNEL f2 -----
 CHANNEL Name: f1
 NUC2 1H
 P2 12.00 usec
 PL2 12.00 dB
 PL12 19.00 dB
 PL13 19.00 dB
 SFO2 500.136051900 MHz

F2 - Processing parameters
 SI 65536
 SF 125.757755 MHz
 XN 16
 EN 0
 GB 7.00 Hz
 BR 0
 BU 0.20

10 MS p131 parameters
 CX 22.00 Hz
 CY 26.50 Hz
 FZ 173.207 Hz
 FL 21793.47 Hz
 LP 13.000 Hz
 F2 1635.81 Hz
 PHA0 7.28569 deg/ci
 HZ00 915.25720 Hz/cm



Current D: Parameters
 NAME r_05230601
 EXPNO 14
 PROCNO 1

F2 - Acquisition Parameters

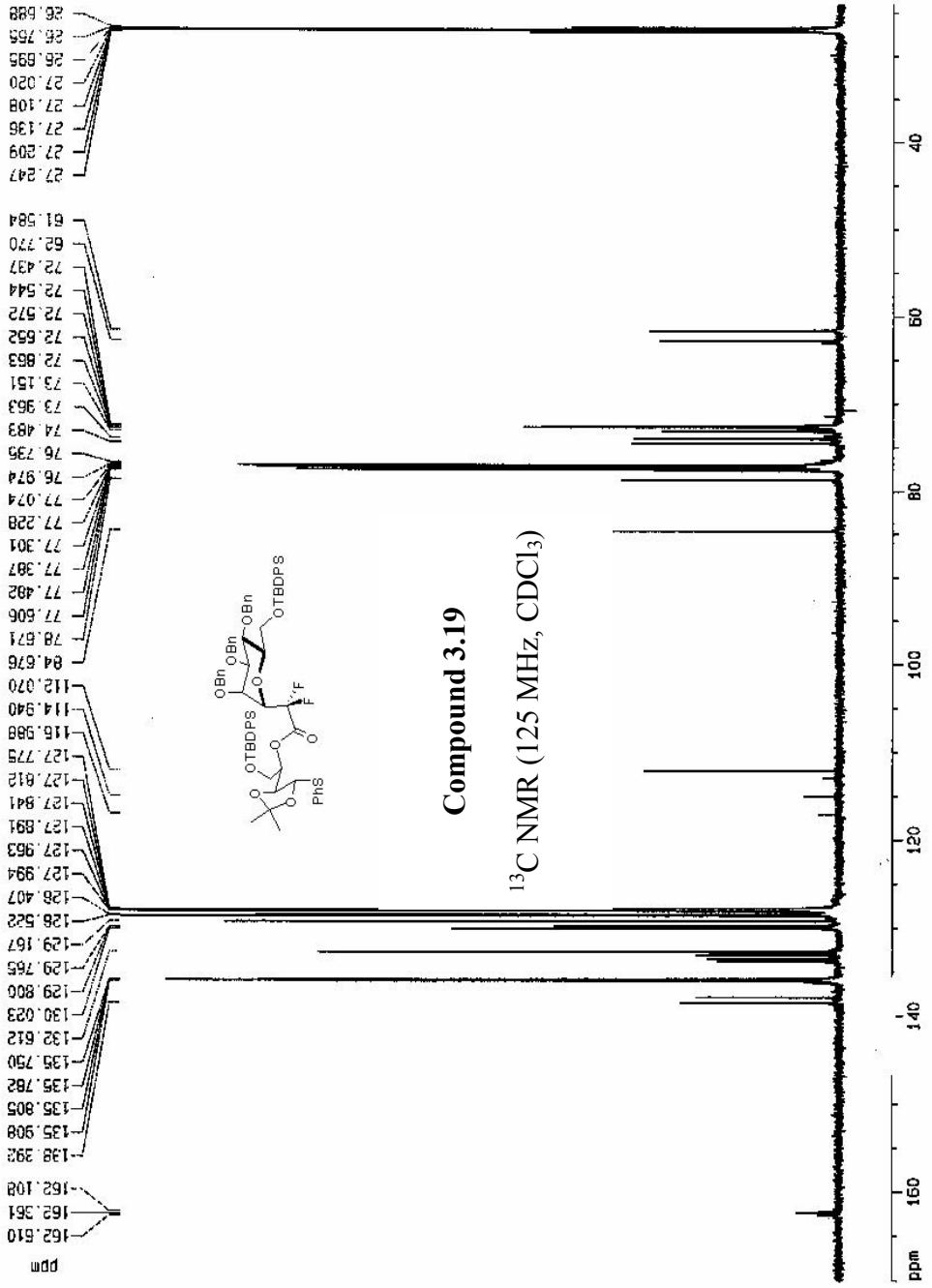
Date_ 20060523
 Time 17.02
 INSTRUM spect
 PROBHD 5 mm DUL 13C-1
 PULPROG zgpg
 TD 65536
 CQC13 CQC13
 NS 1000
 DS 4
 SWH 30030.029 Hz
 FIDRES 0.456222 Hz
 AQ 1.0912410 sec
 RG 1290.2
 DM 16.650 usec
 DE 16.650 usec
 TE 303.0 K
 D1 2.00000000 sec
 d11 0.03000000 sec
 DELTA 1.86999998 sec
 MCREST 0.00000000 sec
 MCWPK 0.01500000 sec

===== CHANNEL f1 =====
 NUC1 13C
 P1 5.00 usec
 PL1 0.00 dB
 SFO1 125.7703643 MHz

===== CHANNEL f2 =====
 CPDPRG2 waltz16
 NUC2 1H
 PCP02 80.00 usec
 PL2 -3.00 dB
 PL12 15.59 dB
 PL13 16.50 dB
 SFO2 500.1320005 MHz

F2 - Processing parameters
 SI 32768
 SF 125.7577661 MHz
 MVM EH
 SSB 0
 LB 1.00 Hz
 GB 0
 PC 0.20

1D NMR plot parameters
 CX 22.00 cm
 CY 24.70 cm
 F1P 170.000 ppm
 F1 21376.82 Hz
 F2P 24.000 ppm
 F2 3038.19 Hz
 PRNCM 5.63636 ppm/cm
 HZCM 834.57422 Hz/cm



```

Current Data Parameters
NAME      03022001
EXPNO    18
PROCNO   1

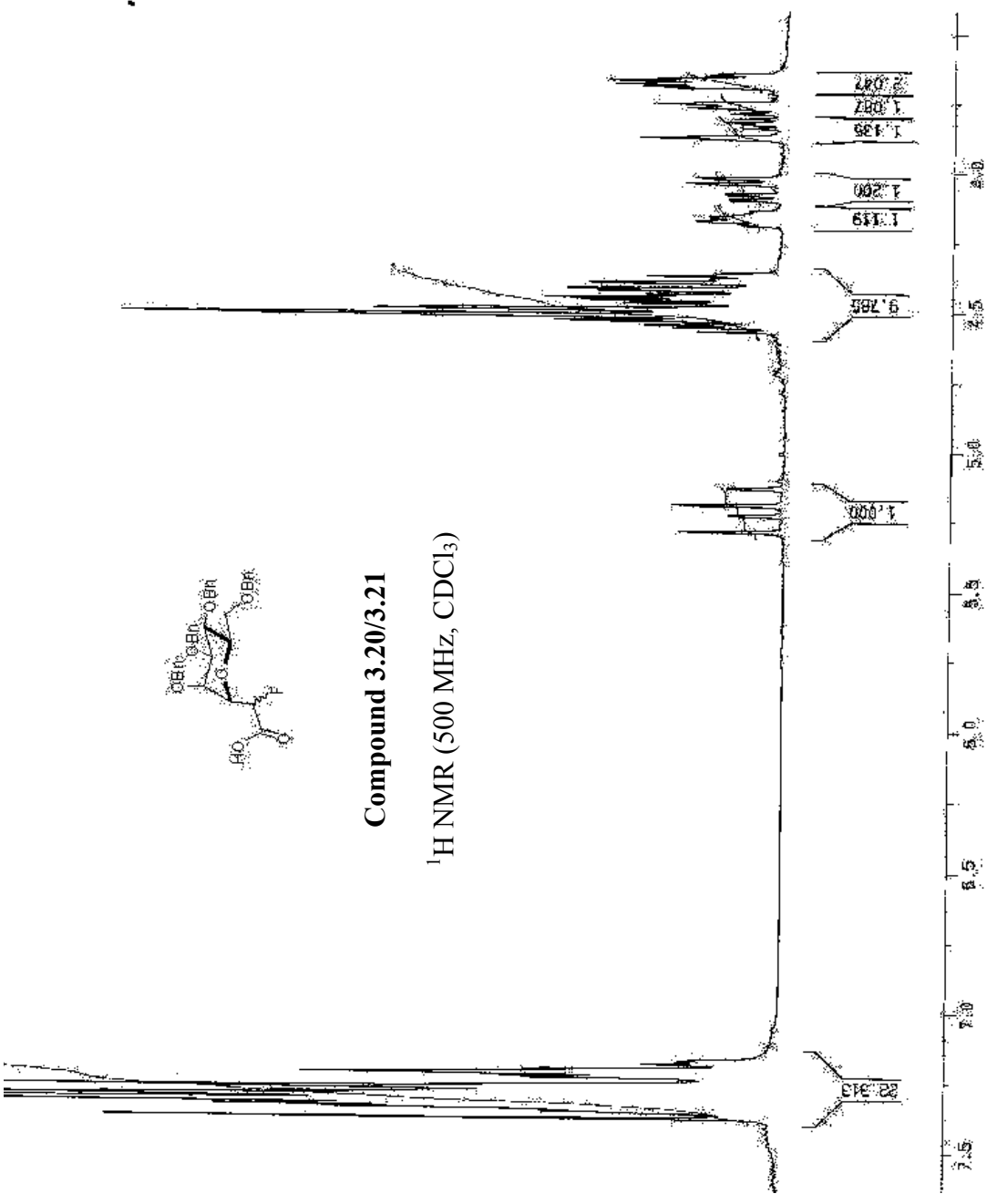
F2 - Acquisition Parameters
Date_    20050902
Time     23:57
INSTRUM  spect
PROBHD   5 mm BUI 125-1
PULPROG  zgpg30
TD       32768
SOLVENT  CDCl3
NS       328
DS       4
SWH       30246.00 Hz
FIDRES    0.244553 Hz
AQ        2.8480615 sec
RG         504
DM         65.400 usec
DE         6.00 usec
TE        303.2 K
D1         1.0000000 sec
ACQRES    6.0000000 sec
SFORES    0.01500000 sec
===== CHANNEL f1 =====
NUC1      1H
P1        9.00 usec
PL1       -2.00 dB
SFO1      500.1360995 MHz

F2 - Processing parameters:
SI        32768
SF        500.1360995 MHz
WDW       EM
SSB       0
LB        0.50 Hz
GB        0
PC        1.00

F0 - Name/plot parameters
BA        23.00 cm
CY        18.07 cm
FI        7.1419 ppm
PI        3818.47 Hz
PO        3.419 ppm
PS        1410.05 Hz
PWHW     0.15698 ppm/cm
AQRES    100.0187 Hz/cm
    
```



Compound 3.20/3.21
¹H NMR (500 MHz, CDCl₃)



Current Parameters
 NAME 09020501
 EXPNO 8
 PROCNO 1

F2 - Acquisition Parameters
 Date_ 20050902
 Time 15.13

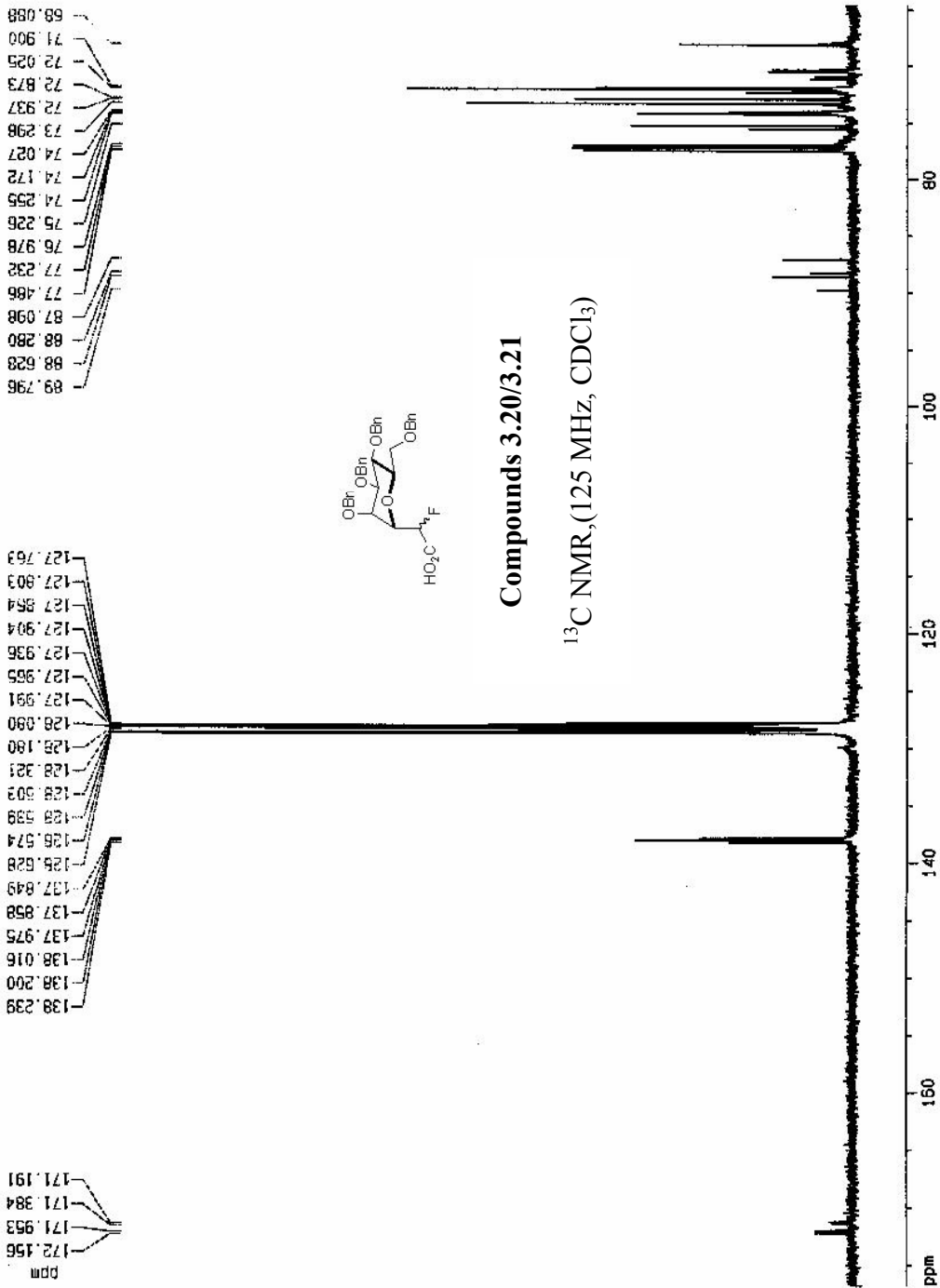
INSTRUM spect
 PROBRD 5 mm DML 13C-1
 PULPROG zgpg30
 TD 65536
 SOLVENT CDCl3
 NS 259
 DS 4
 SNUH 30030.029 Hz
 FIDRES 0.458222 Hz
 AQ 1.0912410 sec
 RB 724.1
 DM 16.650 usec
 DE 6.00 usec
 TE 303.0 K
 D1 2.0000000 sec
 d11 0.0300000 sec
 DELTA 1.8689998 sec
 MCREST 0.0000000 sec
 MCMRK 0.0350000 sec

CHANNEL f1
 NUC1 13C
 P1 5.00 usec
 PL1 0.00 dB
 SF01 125.7703643 MHz

CHANNEL f2
 CPDPRG2 waltz16
 NUC2 1H
 PCPD2 80.00 usec
 PL2 -3.00 dB
 PL12 15.69 dB
 PL13 16.50 dB
 SF02 500.1320005 MHz

F2 - Processing parameters
 SI 32768
 SF 125.7577753 MHz
 MDM EM
 SSB 0
 LB 1.00 Hz
 GB 0
 PC 0.20

1D NMR plot parameters
 CX 22.00 cm
 CY 26.66 cm
 F1P 175.757 ppm
 F2P 64.592 ppm
 F2 8122.86 Hz
 PPMH 5.09841 ppm/cm
 HZCM 641.16492 Hz/cm



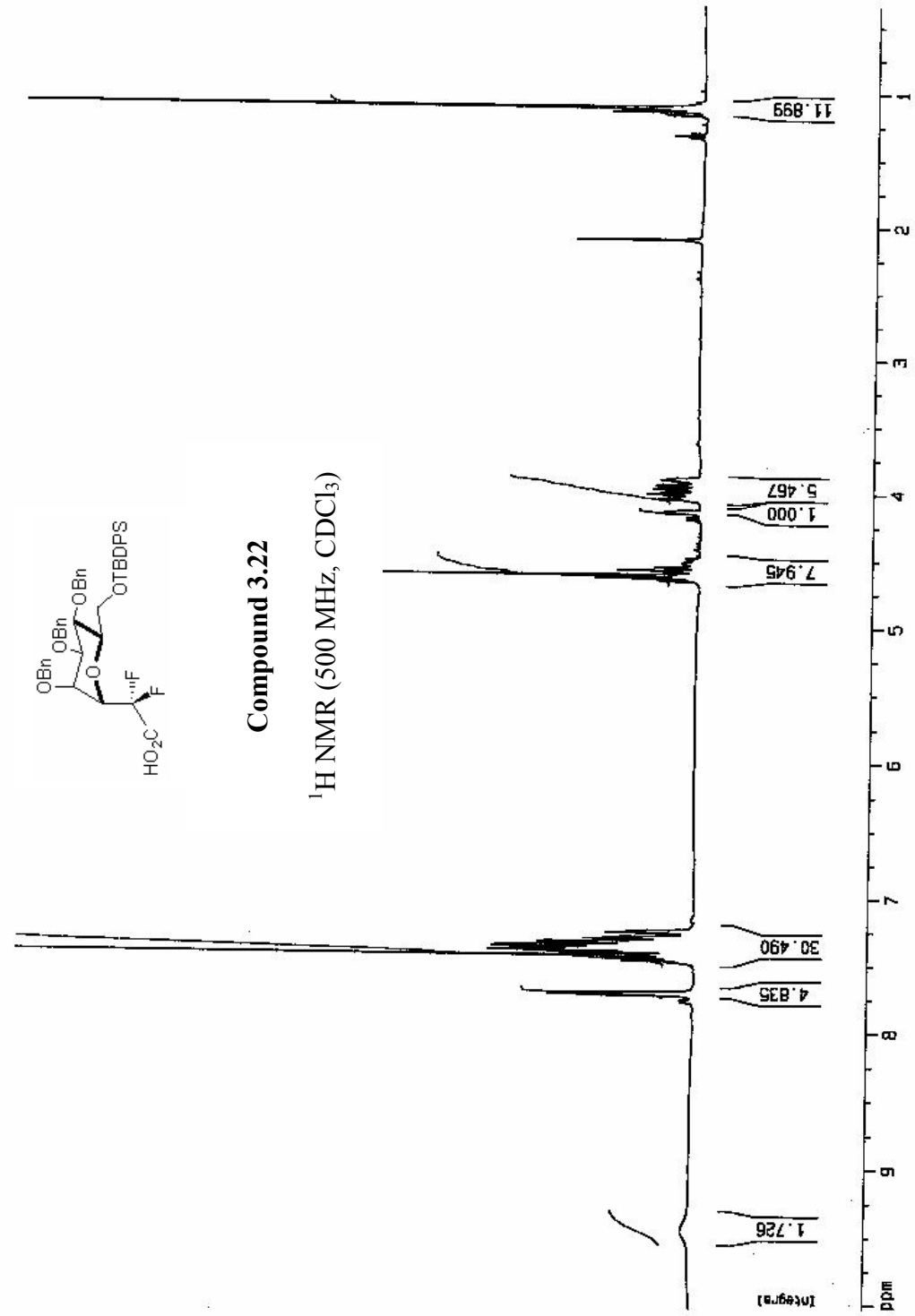
Current Data Parameters
 NAME 04110601
 EXPNO 30
 PROCNO 1

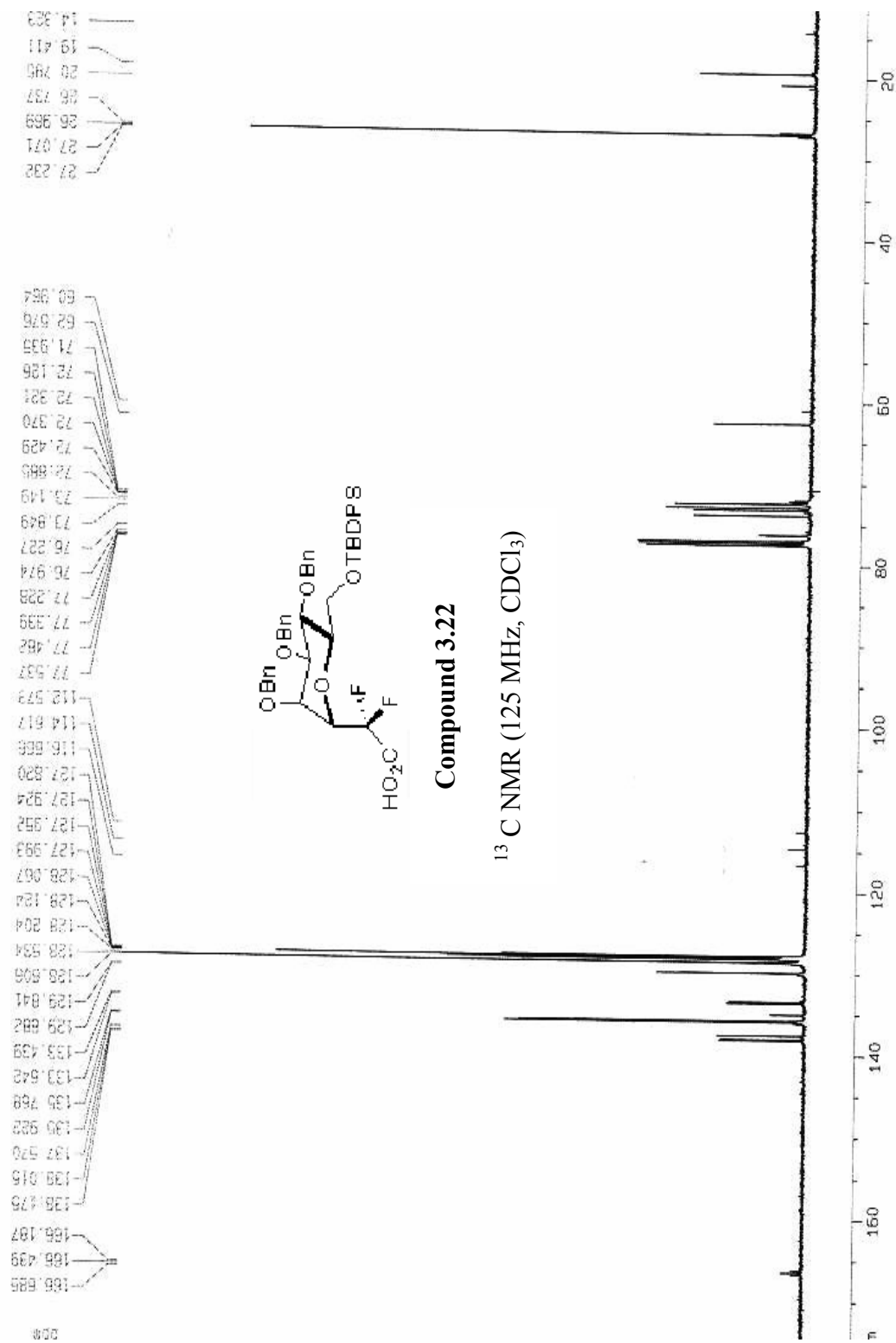
F2 - Acquisition Parameters
 Date_ 20060411
 Time 17.51
 INSTRUM spect
 PROBHD 5 mm QNP 13C-1
 PULPROG zg30
 TD 32768
 SOLVENT CDCl3
 NS 8
 DS 2
 SWH 8012.820 Hz
 FIDRES 0.244532 Hz
 AQ 2.0446355 sec
 RG 256
 OM 62.400 usec
 DE 6.00 usec
 TE 303.0 K
 O1 1.00000000 sec
 MCREST 0.00000000 sec
 MCMRK 0.01500000 sec

***** CHANNEL f1 *****
 NUC1 1H
 P1 9.30 usec
 PL1 -3.00 dB
 SF01 500.1330885 MHz

F2 - Processing parameters
 SI 32768
 SF 500.1300000 MHz
 MDW EM
 SSB 0
 LB 0.30 Hz
 GB 0
 PC 1.00

1D NMR plot parameters
 CX 22.00 cm
 CY 18.05 cm
 F1P 10.032 ppm
 F1 5017.14 Hz
 F2P 0.345 ppm
 F2 172.42 Hz
 PPMCM 0.44031 ppm/cm
 HZCM 220.21465 Hz/cm





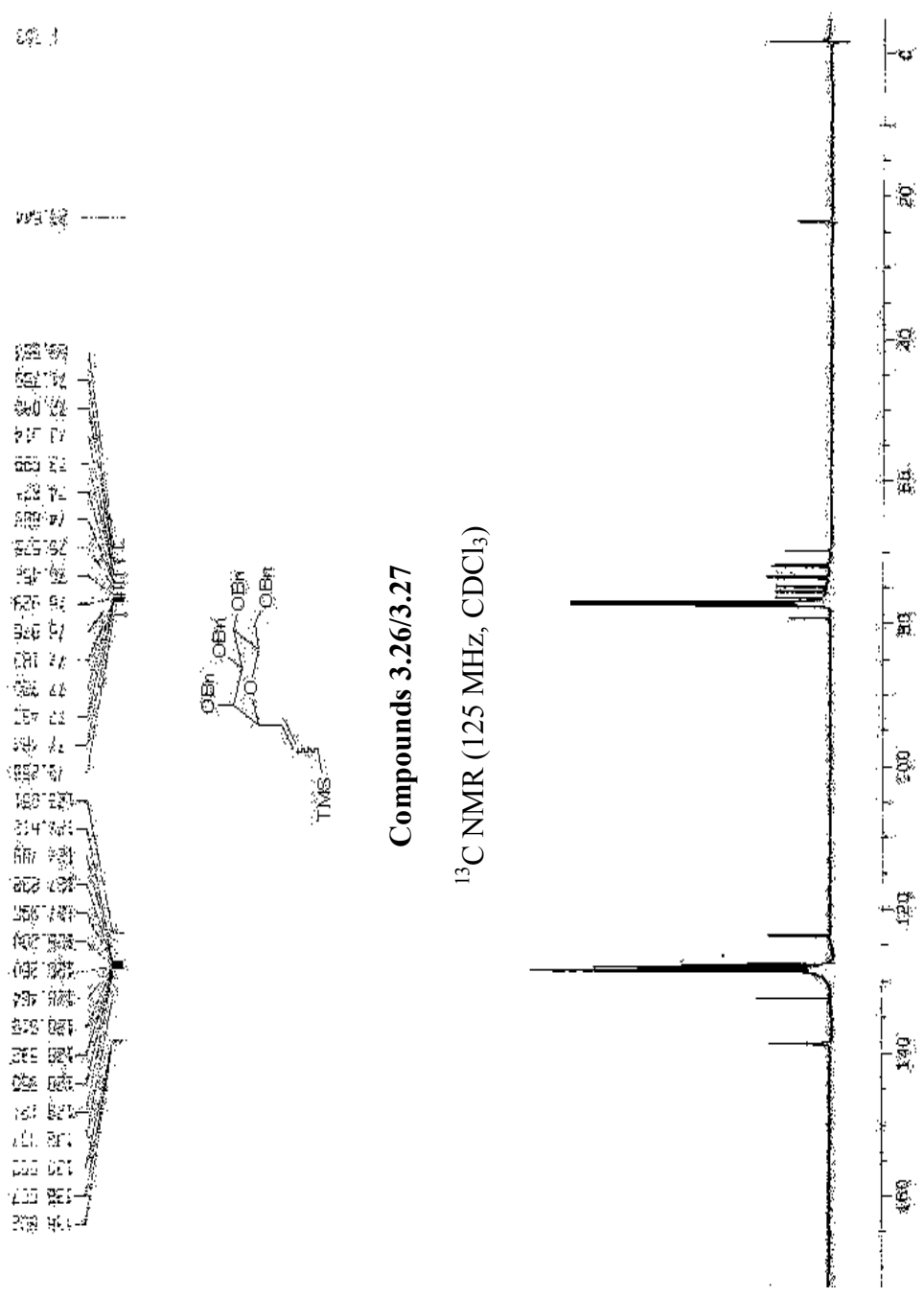

```

Experiment Parameters
NAME          EX010
PROCNO       09
1

===== Acquisition Parameters =====
Date_         20030323
Time          17:02
INSTRUM      spect
PROBHD       5mm TMA-130
PULPROG      zgpg30
TD           65536
SOLVENT      CDCl3
NS           209
DS           4
SWH           39030.029 MHz
FIDRES       0.269048 Hz
AQ           1.0893763 sec
RG           1024
DA           16.656 USec
DE           0.0965
TE           300.2 K
d1           2.00000000 sec
d11          0.02000000 sec
DELTA        1.00000000 sec
WDELSY       0.00000000 sec
WDELSX       0.00000000 sec
WDELRY       0.00000000 sec
WDELRY       0.00000000 sec

===== CHANNEL f1 =====
NUC1          13C
P1           12.00 USec
PL1          0.00 dB
RF00         16.00 MHz
===== CHANNEL f2 =====
NUC2          1H
P2           12.00 USec
PL2          0.00 dB
RF01         500.136099 MHz
===== Processing parameters =====
SI           32768
SF           125.7628090 MHz
RG           1024
WDW          EM
SSB          0
LB           0.00 Hz
GB           0
PC           0.00

===== 1D NMR list parameters =====
EX          1
AQ          1.0893763 sec
RG          1024
WDW          EM
SSB          0
LB           0.00 Hz
GB           0
PC           0.00
  
```



Compounds 3.26/3.27

¹³C NMR (125 MHz, CDCl₃)

Current Data Parameters
 NAME 10050501
 EXPNO 16
 PROCNO 1

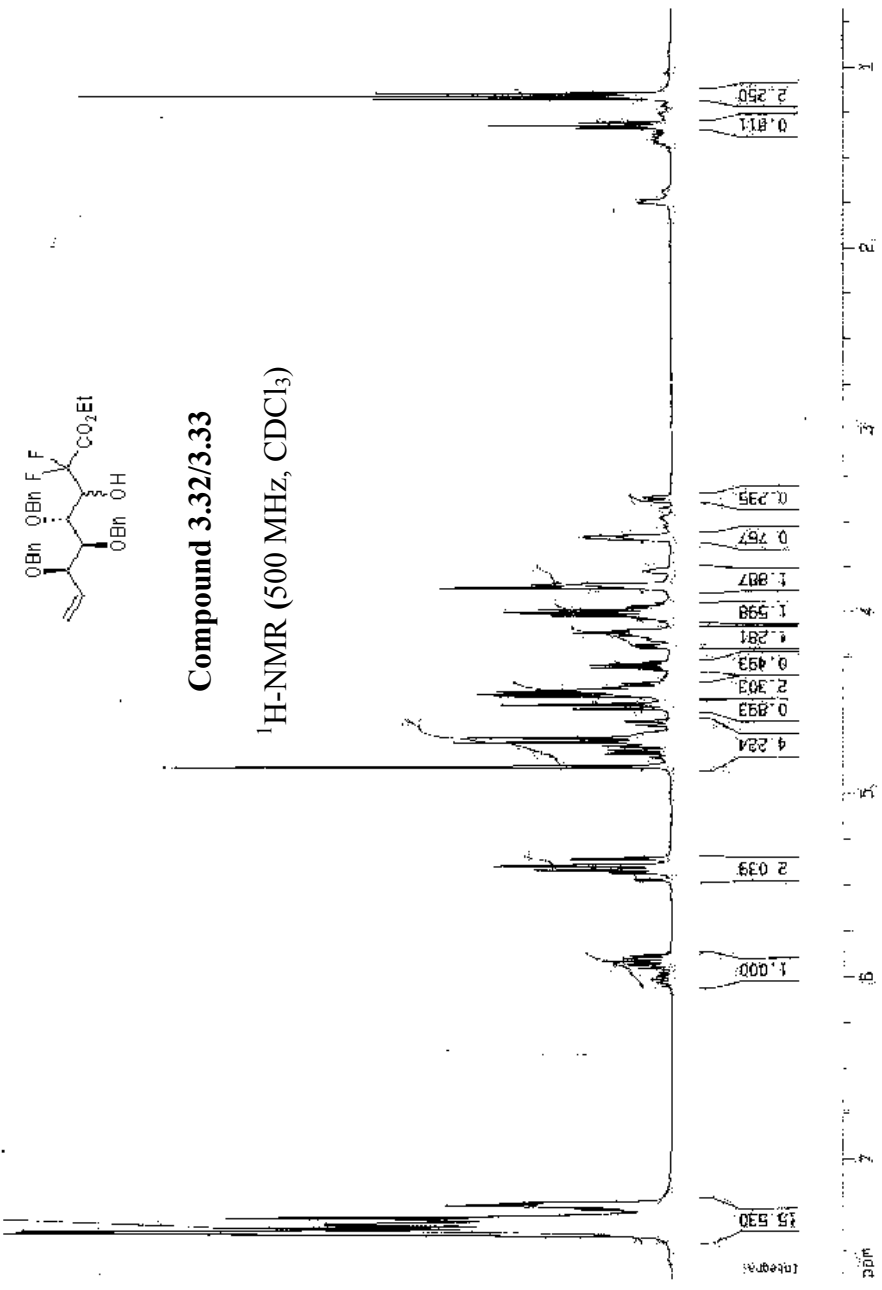
F2 - Acquisition Parameters
 Date_ 20051005
 Time 13:36
 INSTRUM spect
 PROBRD 5 mm QNP 1
 PULPROG zgpg30
 TD 32768
 SOLVENT CDCl3
 NS 8
 DS 2

SWH 8012.500 Hz
 FIDRES 3.244538 Hz
 AQ 2.0446358 sec
 RG 357.5
 DM 59.400 usec
 DE 6.70 usec
 TE 303.0 K
 D1 1.0000000 sec
 ACQRES 0.2600000 sec
 MCWRR 0.0100000 sec

===== CHANNEL f2 =====
 NUC1 1H
 P1 5.00 uses
 PL1 -3.00 dB
 SFO1 500.1335699 MHz

F2 - Processing parameters
 SI 32768
 SF 500.1335697 MHz
 WDR 64
 SSB 0
 LB 0.70 Hz
 GB 0
 PC 1.00

500 MHz plot parameters
 CX 22.00 cm
 CY 12.00 cm
 F1F 7.001 ppm
 F1 9628.59 Hz
 F2P 0.679 ppm
 F2 649.59 Hz
 REACH 0.31691 ppm/cm
 HZCM 158.48843 Hz/cm



Name: Parameters
 10050501
 EXPNO: 19
 PROCNO: 2

F2 - Acquisition Parameters
 Date_: 20051005
 Time: 14.10

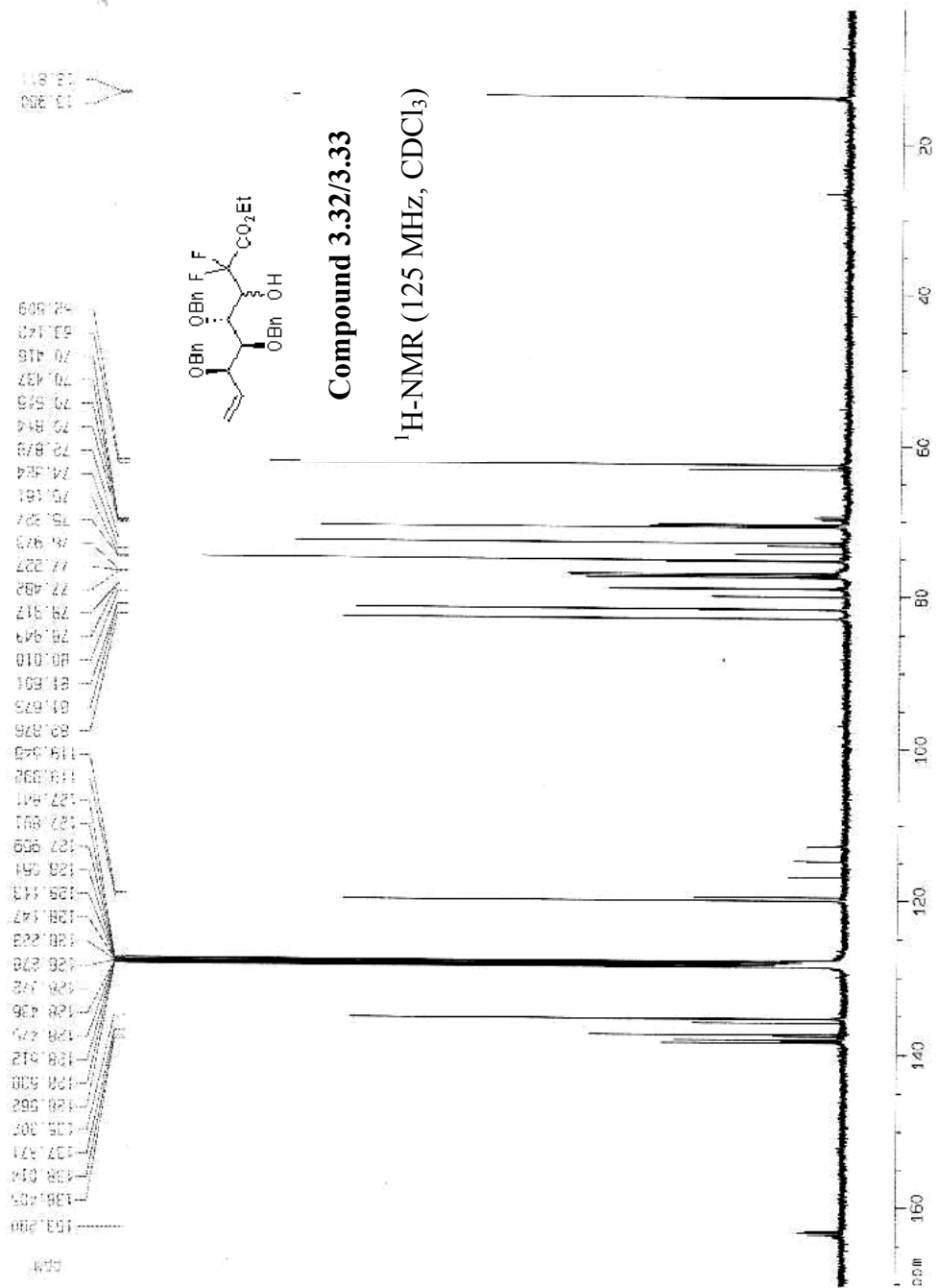
INSTRUM: spect
 PROBHD: 5 mm DUL 13C-1
 PULPROG: zgpg30
 TD: 65536
 SOLVENT: CDCl3
 NS: 267
 DS: 4
 SWH: 30030.029 Hz
 FIDRES: 0.456222 Hz
 AQ: 1.0312410 sec
 RG: 724.1
 DM: 16.656 usec
 DE: 6.00 usec
 TE: 303.0 K
 D1: 2.0000000 sec
 D11: 0.0300000 sec
 DELTA: 1.8999999 sec
 MREST: 0.0000000 sec
 MCMRK: 0.0190000 sec

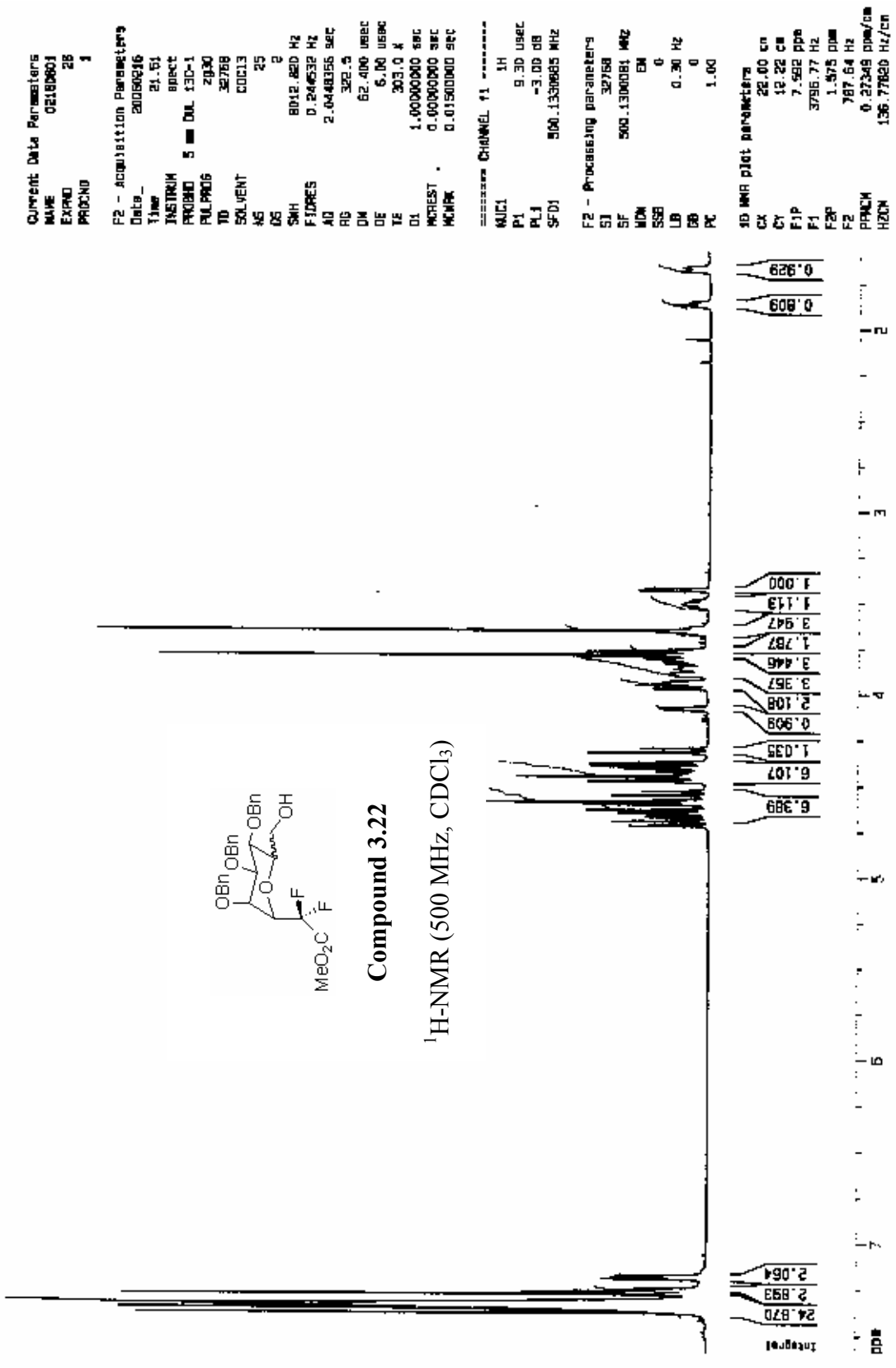
===== CHANNEL F1 =====
 NUC1: 13C
 P1: 5.00 usec
 PL1: 0.00 dB
 SFO1: 125.7703643 MHz

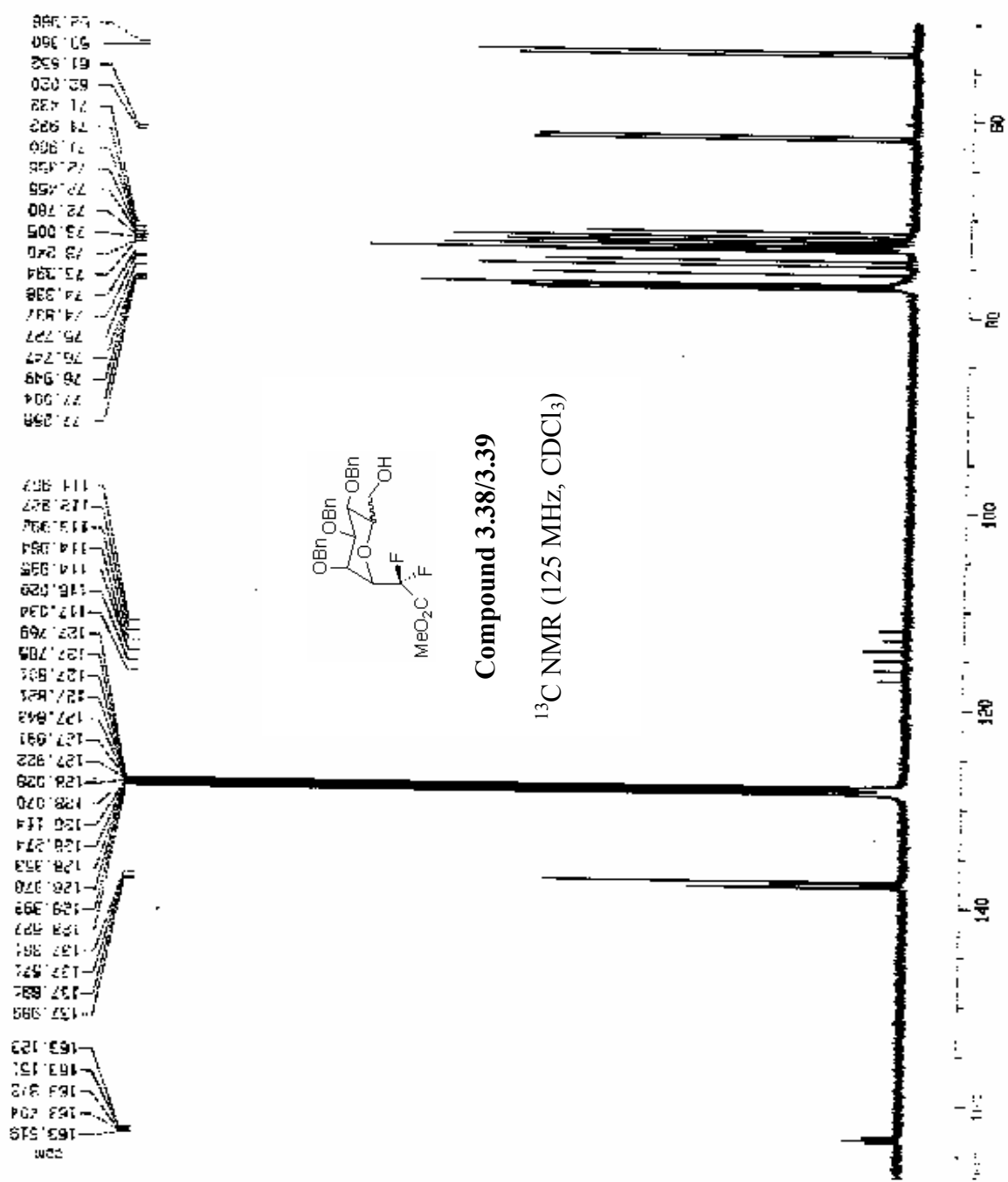
===== CHANNEL F2 =====
 CPOPRG2: zgpg30
 NUC2: 1H
 PCPD2: 80.00 usec
 PL2: -3.00 dB
 PL12: 15.69 dB
 PL13: 16.50 dB
 SFO2: 500.1320005 MHz

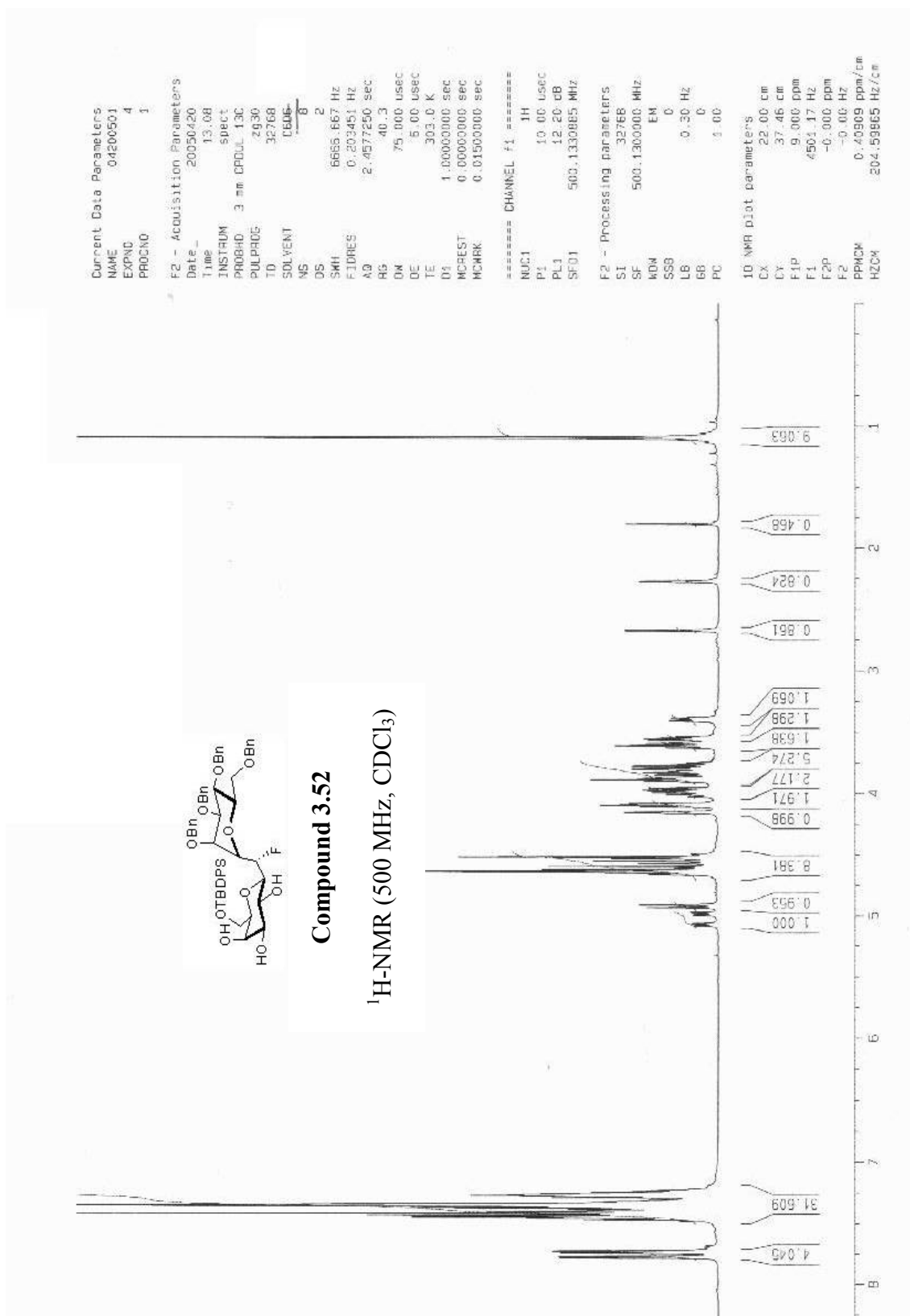
F2 - Processing parameters
 SI: 32768
 SF: 125.7577771 MHz
 NCM: EM
 SSB: 0
 LB: 1.00 Hz
 GB: 0
 PC: 0.20

1D NMR plot parameters
 CX: 22.00 cm
 CY: 37.46 cm
 F1P: 170.305 ppm
 F1: 21427.23 Hz
 F2P: 2.037 ppm
 F2: 256.17 Hz
 PPM1M: 7.54655 ppm/cm
 HZCY: 961.55521 Hz/cm









Current Def Parameters
 NAME 04200501
 EXPNO 1
 PROCNO 1

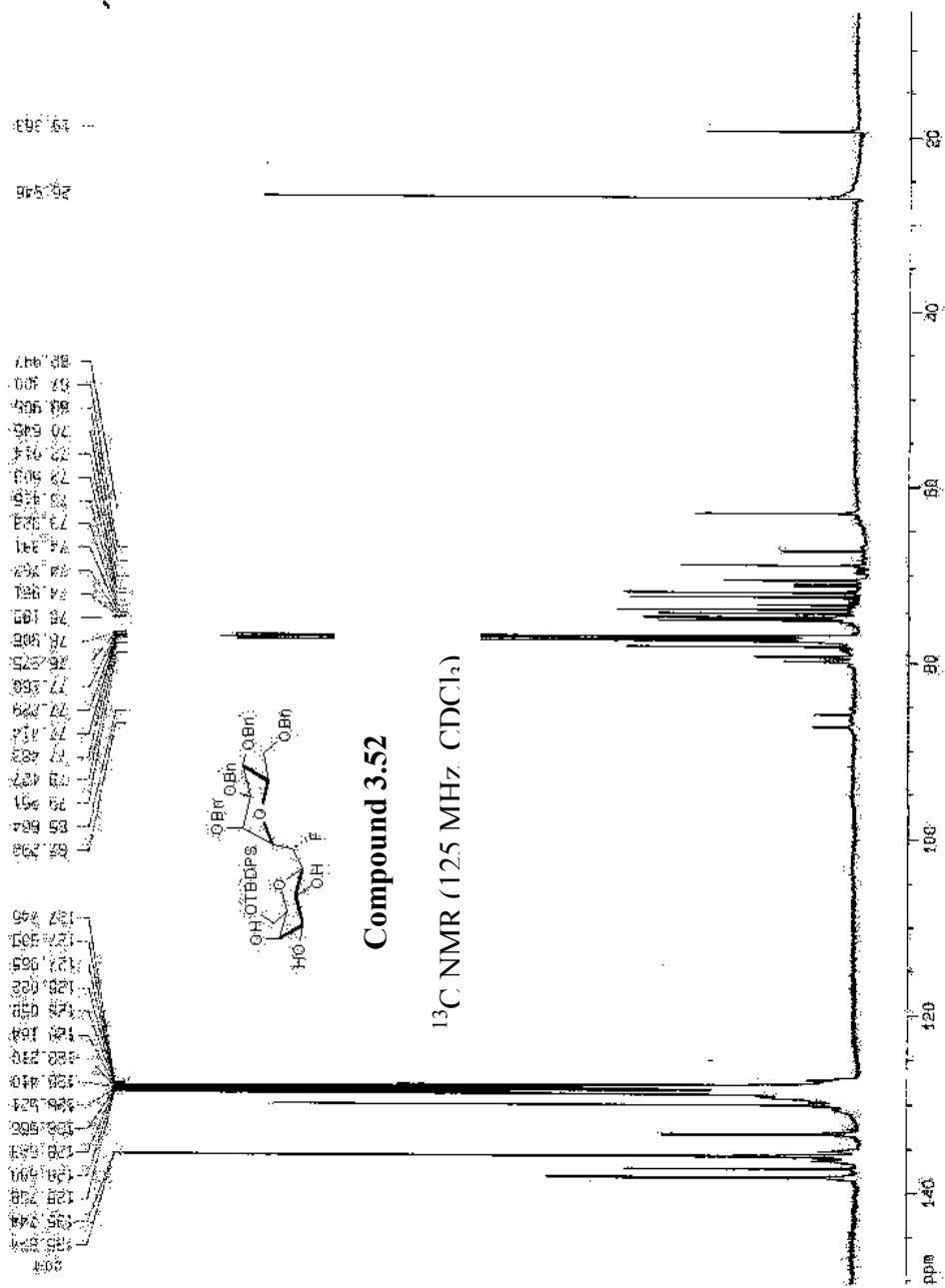
F2 - Acquisition Parameters
 Date_ 20050400
 Time 13:19
 INSTRUM spect
 PROBU0 3 mm QNP1 13C
 PULPROG zgpg
 TD 65408
 SFO 125.760
 FIDRES 0.0001000 Hz
 AQ 0.0002753 sec
 RG 1000
 OR 16.000 usec
 DE 5.00 usec
 TE 303.0 K
 D1 2.00000000 sec
 d11 0.05000000 sec
 DELTA 1.69999998 sec
 ACQRES 0.00010000 sec
 MCORR 0.01500160 sec

===== CHANNEL f2 =====
 NUC1 13C
 P1 10.00 usec
 PL1 18.00 dB
 SFO1 125.7703642 MHz

===== CHANNEL f1 =====
 CPDPRG2 waltz16
 NUC2 1H
 PPRG2 80.00 usec
 PL2 18.00 dB
 PL12 29.00 dB
 PL13 29.00 dB
 SFO2 500.1360005 MHz

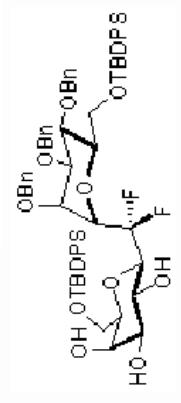
F2 - Processing parameters
 SI 65536
 SF 125.7577788 MHz
 EQ
 SSB 0
 LB 2.00 Hz
 GB 0
 PC 0.20

1D NMR list parameters
 CX 22.00 cm
 CY 25.03 cm
 F1P 150.299 900
 F2 10901.33 Hz
 F3P 5.744 900
 F2 722.37 Hz
 PPMH 5.570 880/56
 HZCM 825.34655 Hz/cm



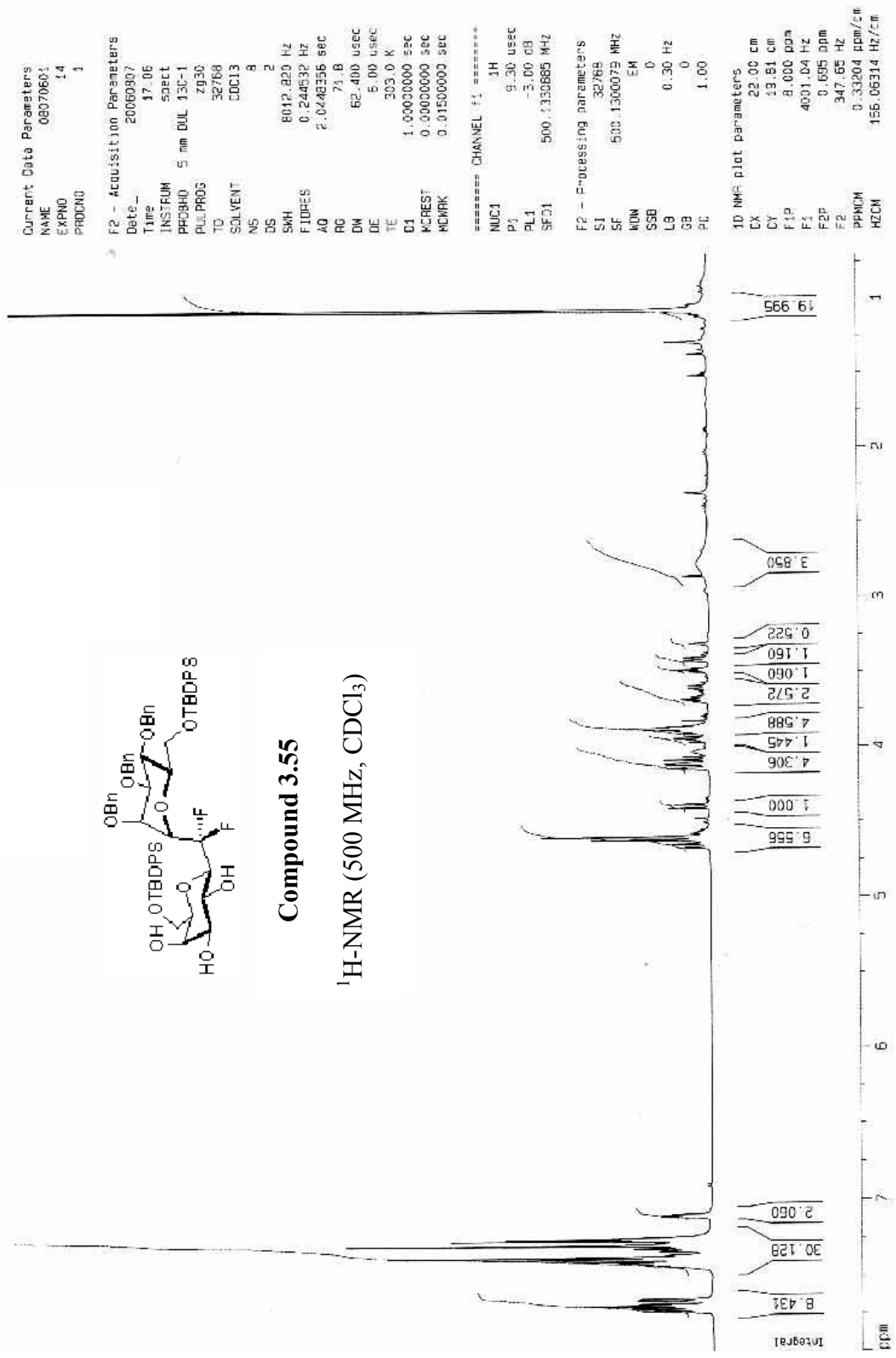
Compound 3.52

¹³C NMR (125 MHz, CDCl₃)



Compound 3.55

¹H-NMR (500 MHz, CDCl₃)

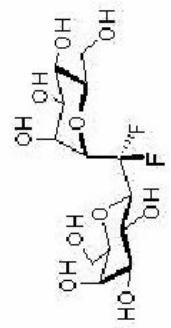


Current Data Parameters
 NAME 08070601
 EXPNO 14
 PROCNO 1

F2 - Acquisition Parameters
 Date_ 20060807
 Time 17.06
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 SOLVENT CDCl3
 NS 8
 DS 2
 SWH 8012.820 Hz
 FIDRES 0.244532 Hz
 AQ 2.0748356 sec
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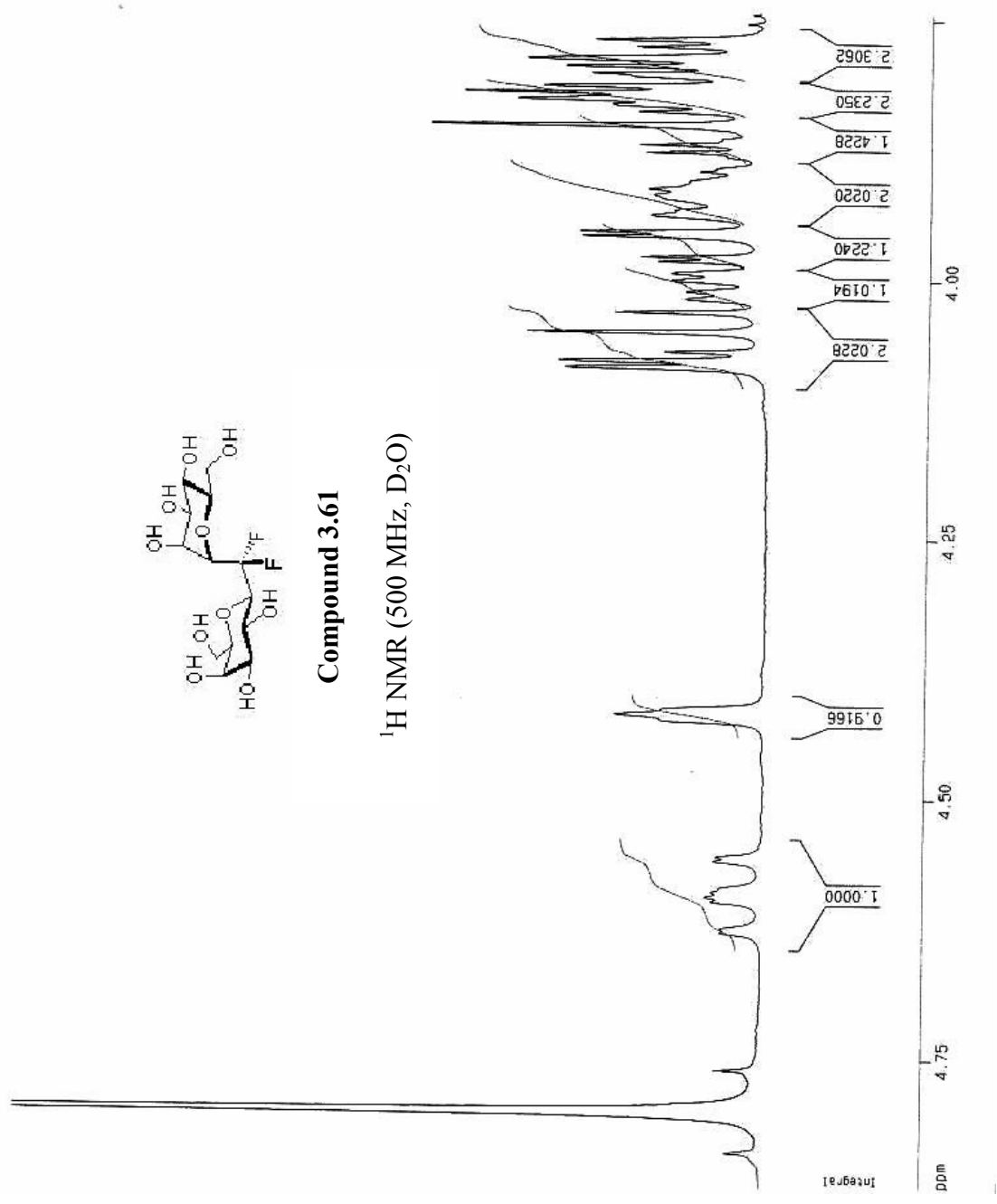
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 SSB 0
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 GB 0
 PC 1.00

1D NMR plot parameters
 CX 22.00 cm
 CY 19.81 cm
 FIP 8.000 ppm
 F1 4001.04 Hz
 F2 0.695 ppm
 F2 347.65 Hz
 PPNOM 0.33204 ppm/cm
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Compound 3.61

¹H NMR (500 MHz, D₂O)



Current Data Parameters
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 EXPNO 1
 PROCNO 1

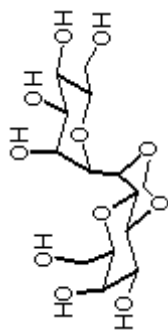
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 DS 2
 SWH 8012.820 Hz
 FIDRES 0.244532 Hz
 AQ 2.0448356 sec
 RG 406.4
 DW 62.400 usec
 DE 6.00 usec
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 MCPRK 0.01500000 sec

===== CHANNEL f1 =====
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 P1 9.30 usec
 PL1 -3.00 dB
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F2 - Processing parameters
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 PC 1.00

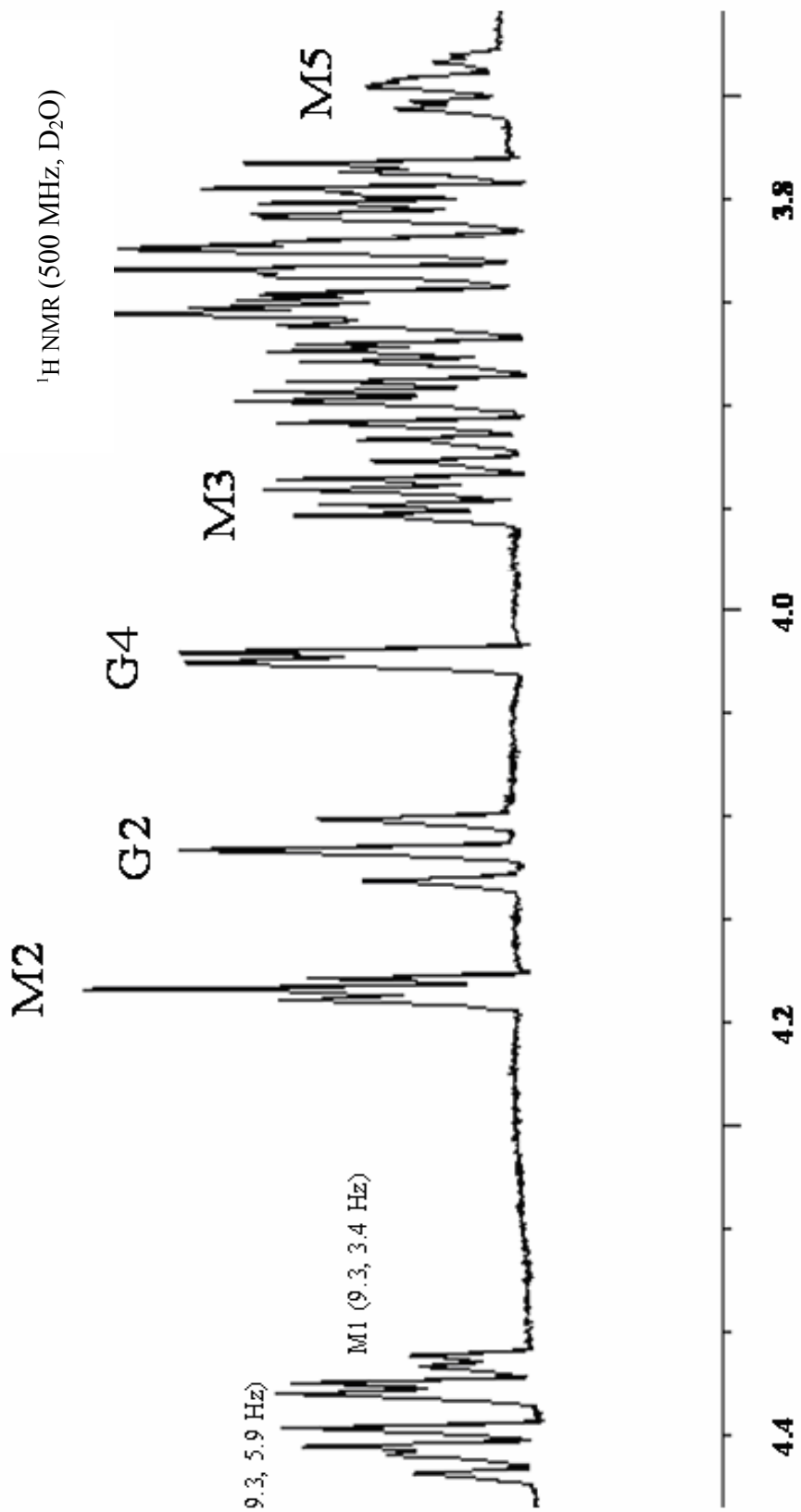
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 HZCM 25.68392 Hz/cm

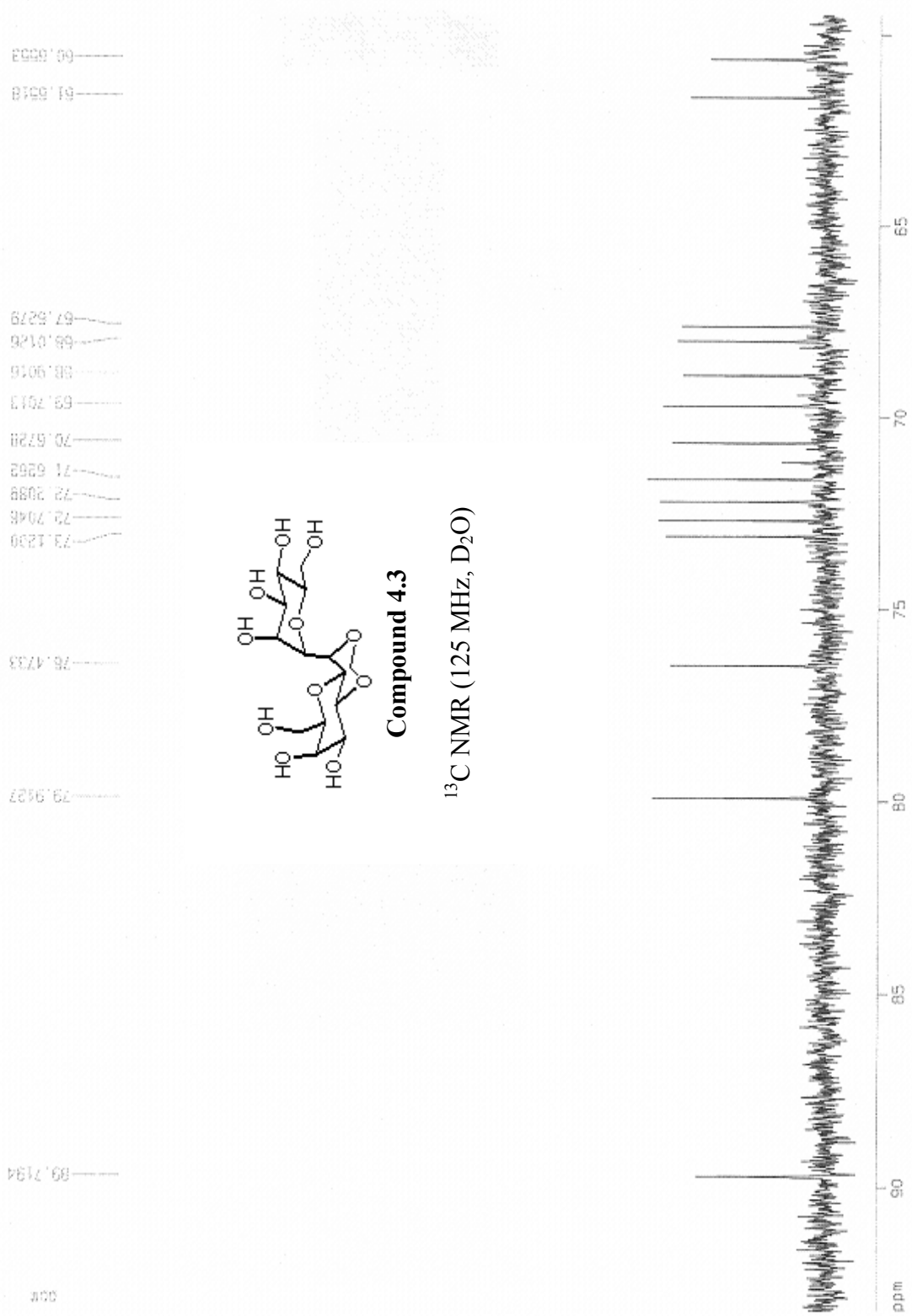
¹H NMR spectrum and key coupling constants

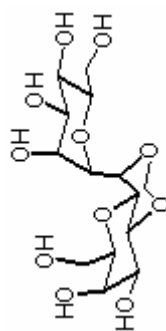


Compound 4.3

¹H NMR (500 MHz, D₂O)



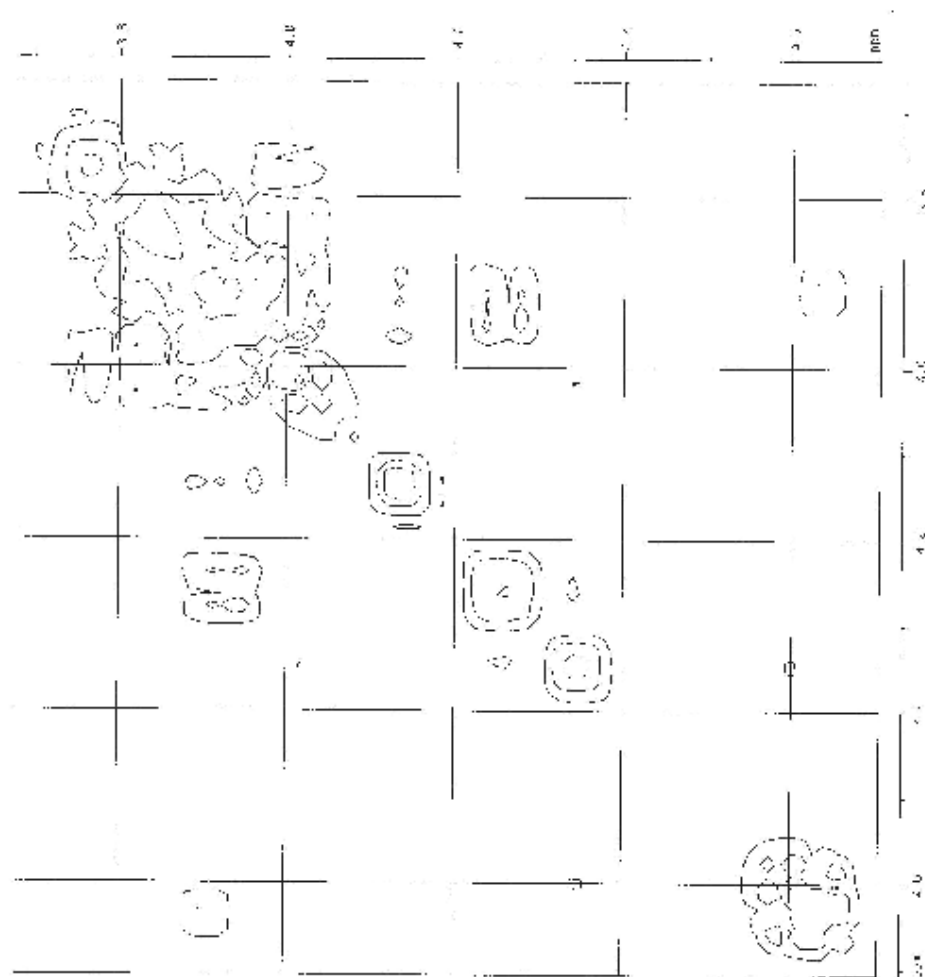


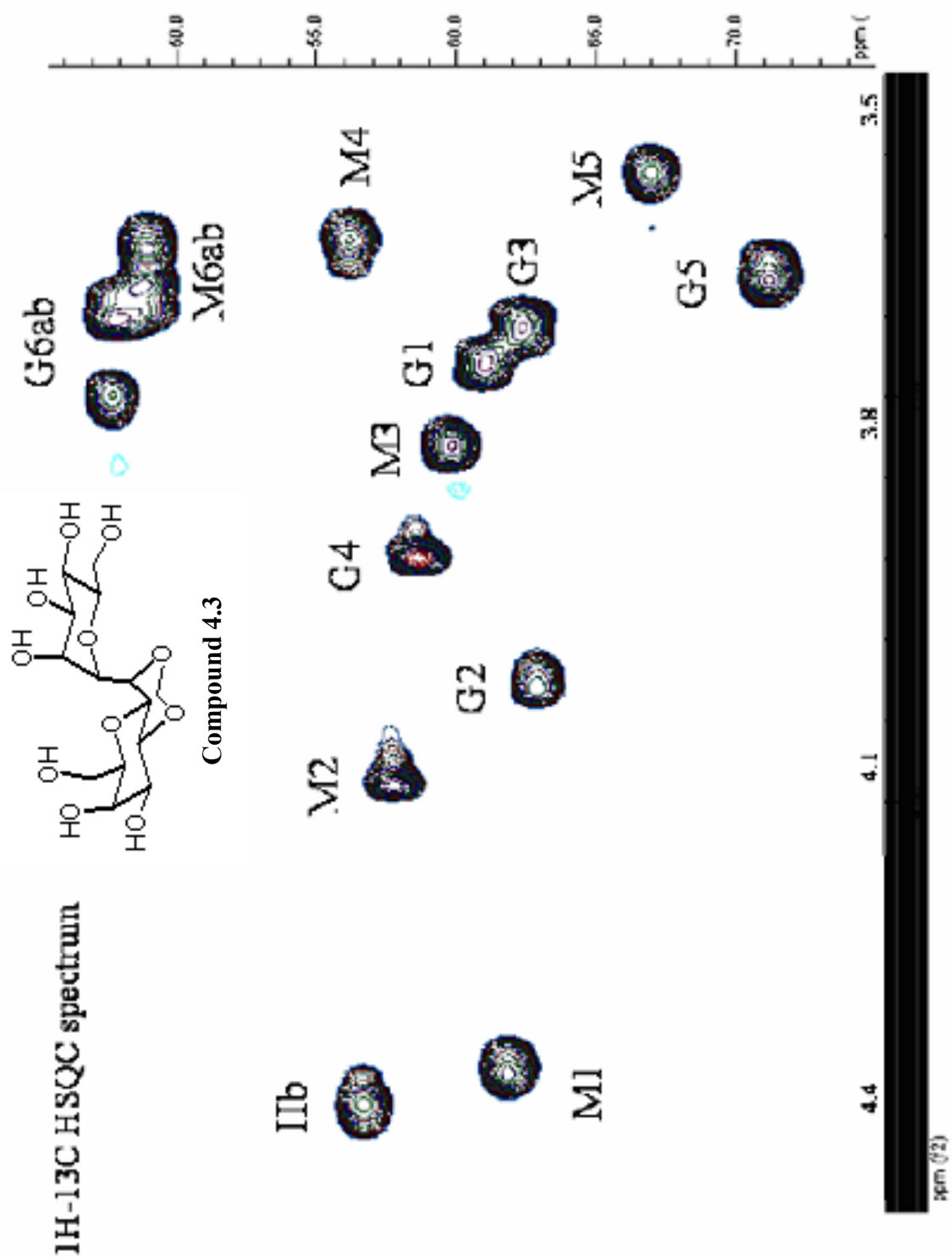


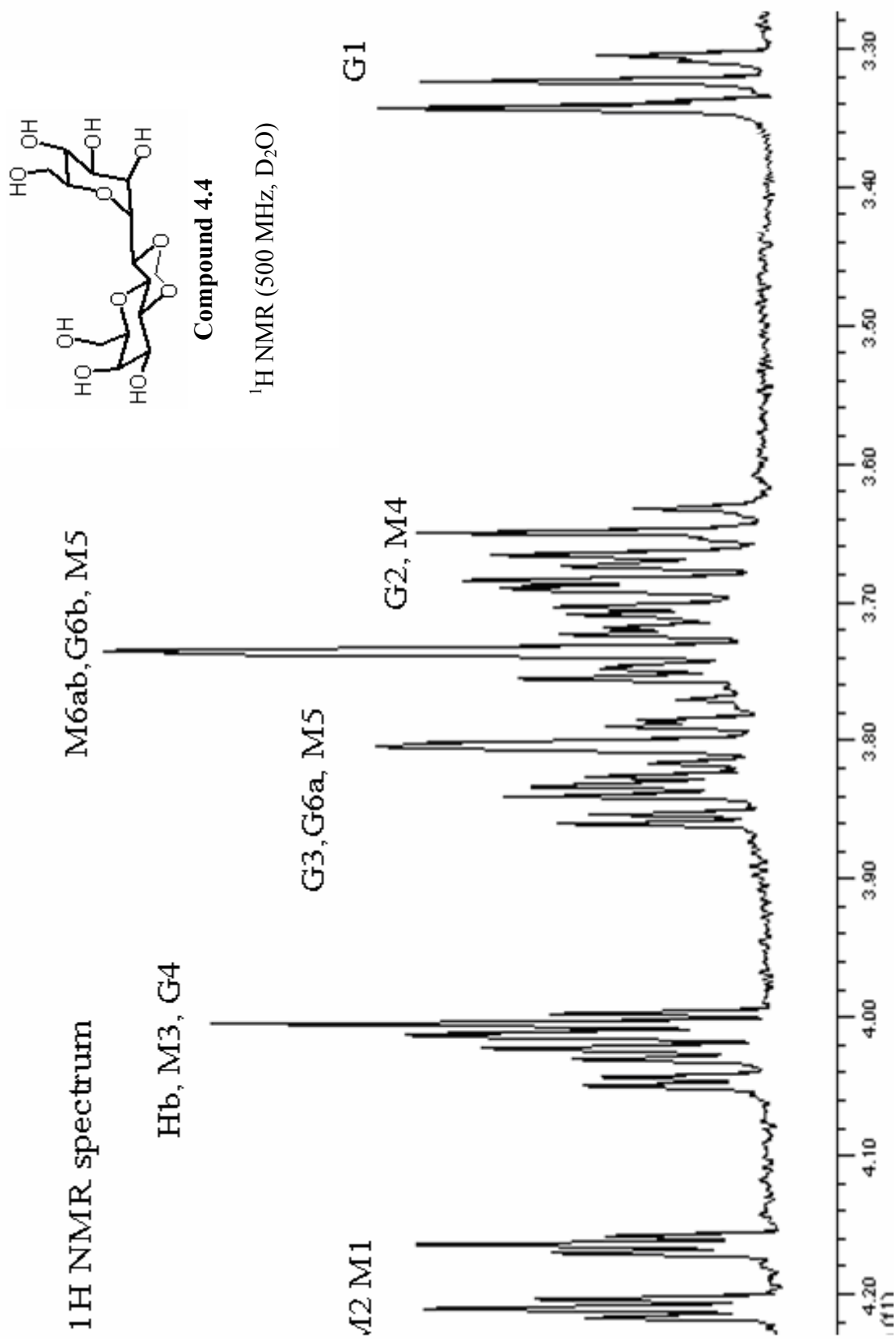
Compound 4.3

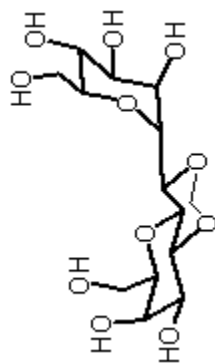
(1H-1H COSY)

	δ (ppm)	J (Hz)
M1	4.39	8.8
M2	4.14	3.3
M3	3.82	3.7
M4	3.63	
M5	3.56	2.4
M6a	3.69	
M6b	3.64	
Hintersac	4.44	8.8
G1	3.75	5.9
G2	4.05	9.2
G3	3.74	
G4	3.92	3.3
G5	3.67	
G6a	3.79	
G6b	3.70	
OCH ₂ Oax	5.04	6.5
OCH ₂ Oac	4.86	6.5



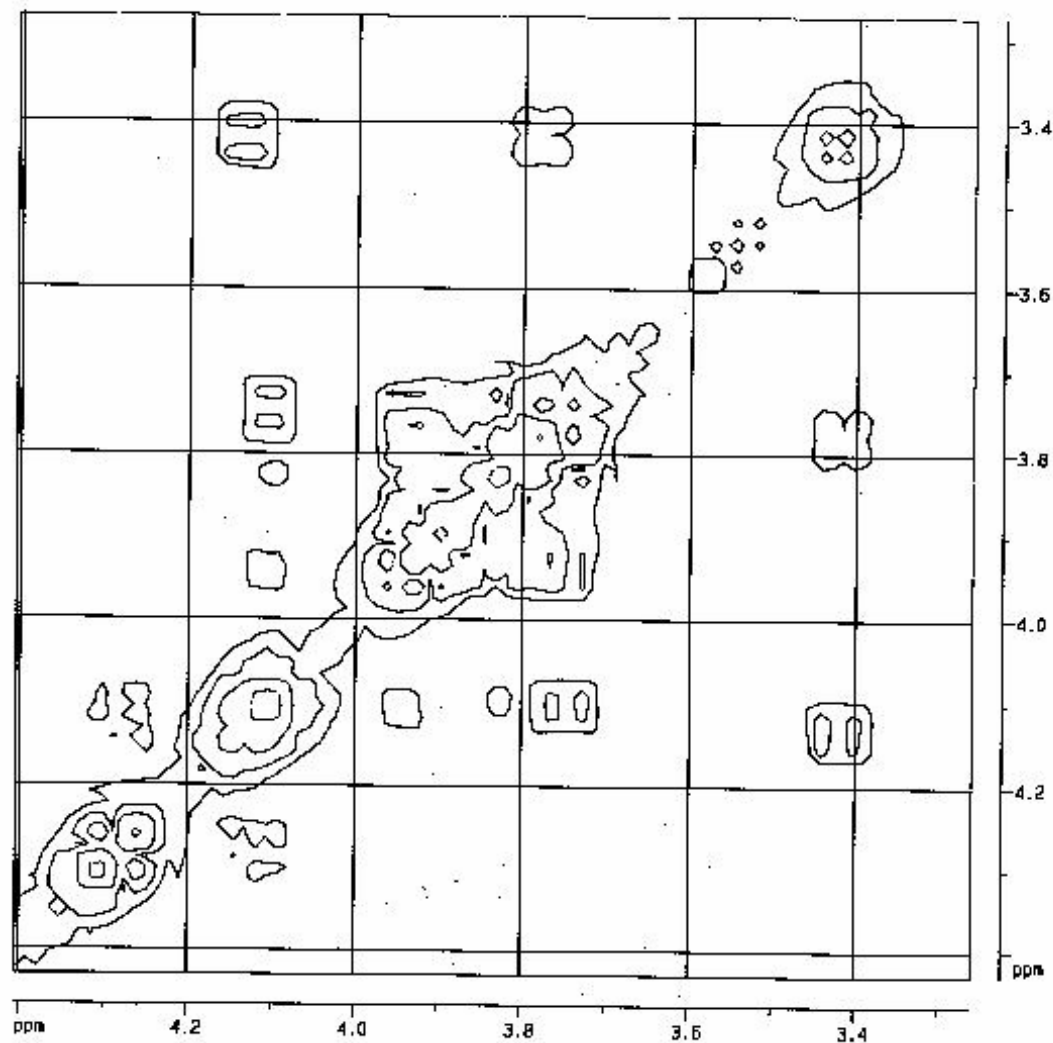




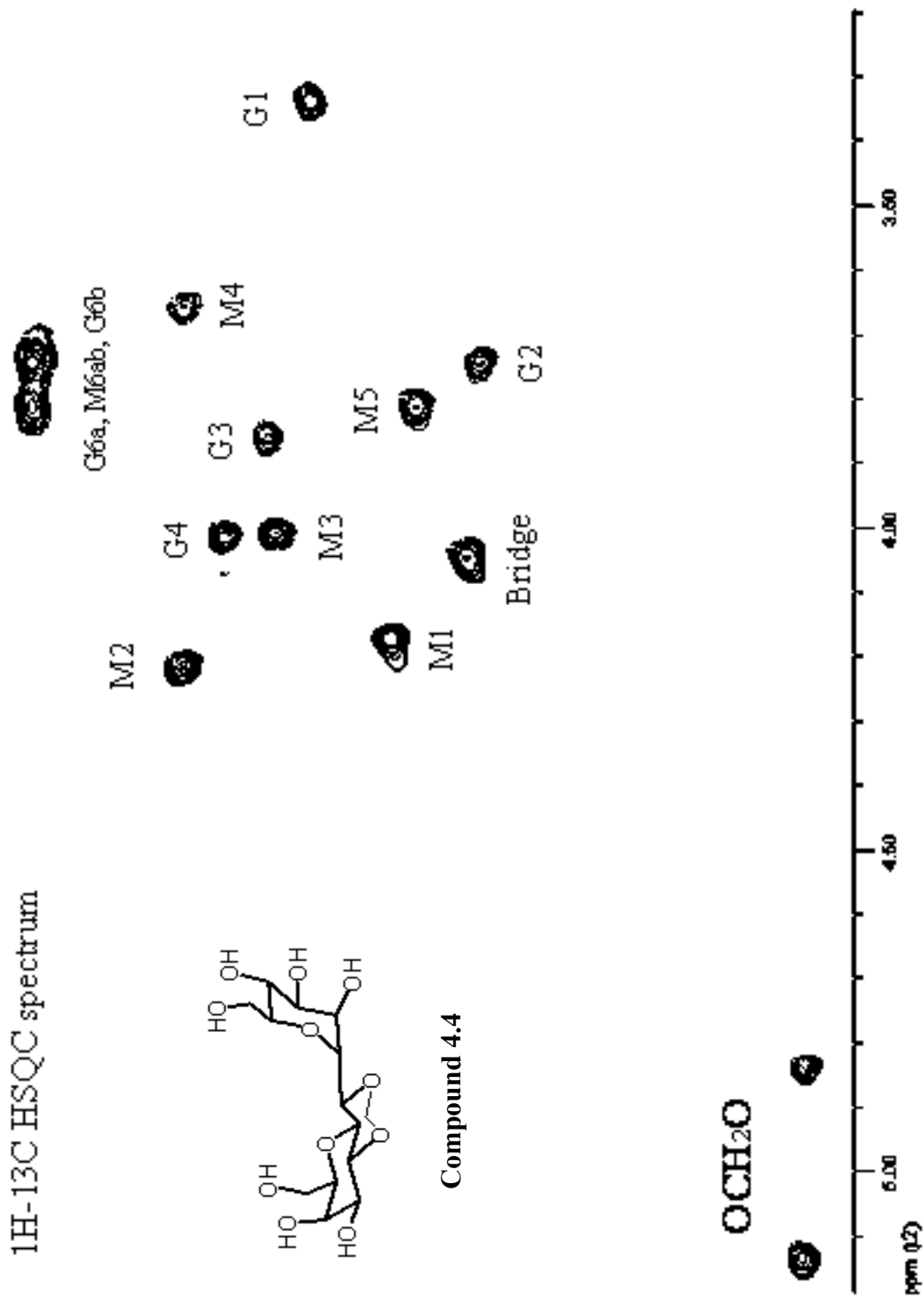


Compound 4.4
(1H-1H COSY)

	δ (ppm)	J (Hz)
M1	4.16	3.0
M2	4.21	3.3
M3	4.01	
M4	3.65	8.7
M5	3.80	
M6a	3.74*	
M6b	3.74*	
H bridge	4.04	3.3
G1	3.32	9.4
G2	3.69	
G3	3.85	
G4	4.01	
G5	3.73	
G6a	3.74*	
G6b	3.80*	
OCH ₂ Oax	4.84	6.5
OCH ₂ Oeq	5.15	6.5



¹H-¹³C HSQC spectrum



61.4099
61.2242

67.4750

69.4542

70.8816

71.3087

72.8353

75.0000

77.3949

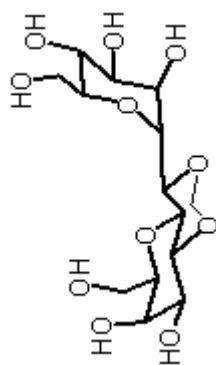
77.7994

78.1908

79.6195

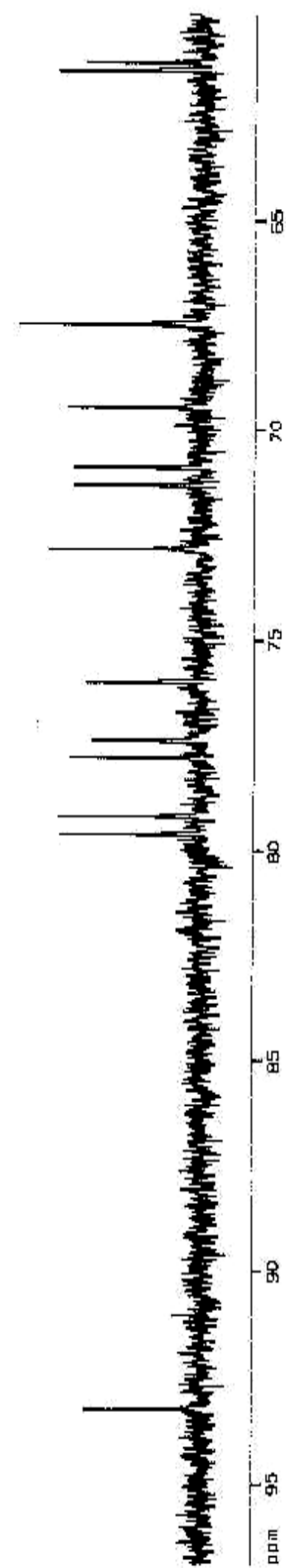
93.2425

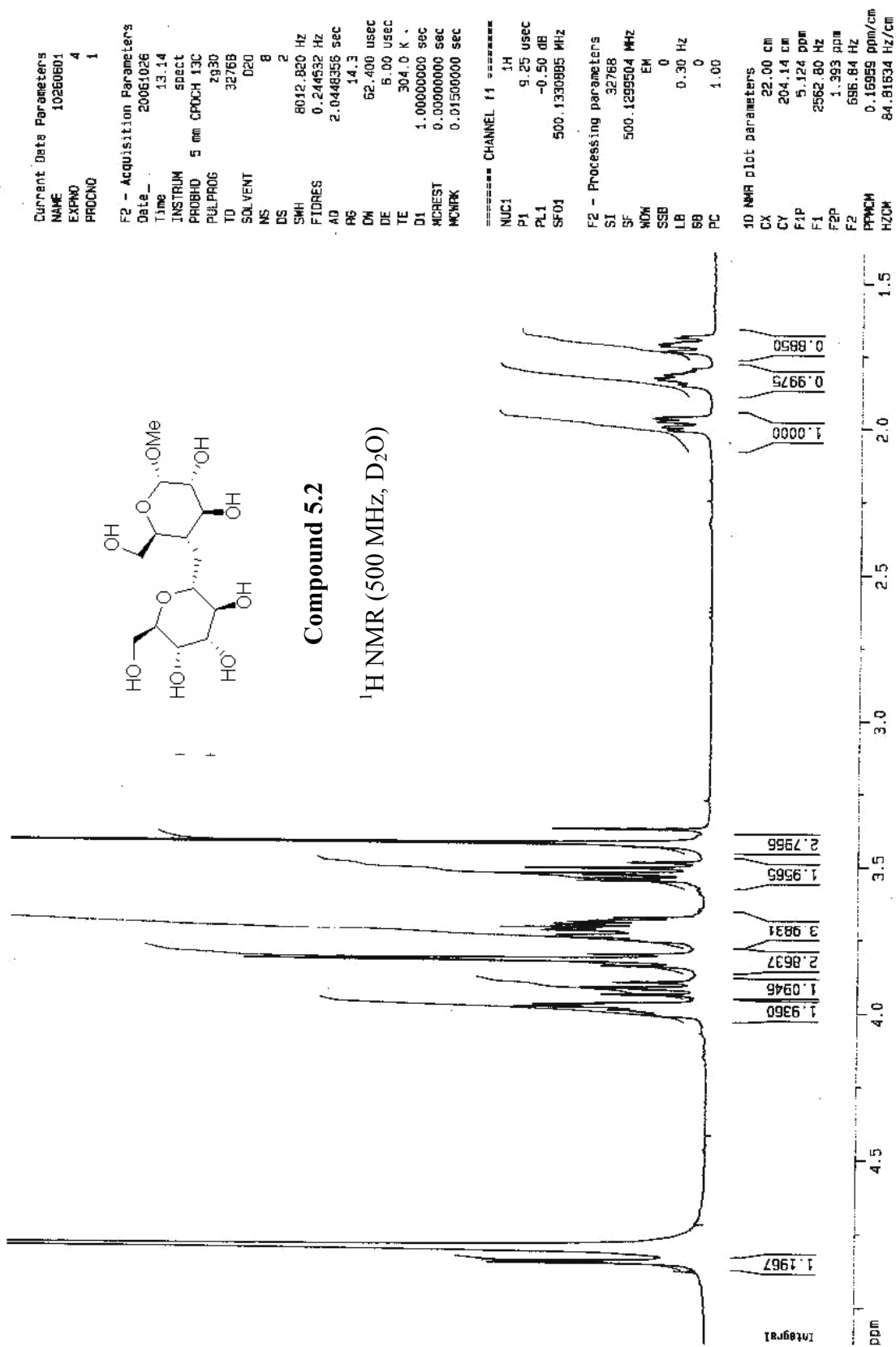
ppm



Compound 4.4

^{13}C NMR (125 MHz, D_2O)





119.6

102.6

91.9

97.6

90.9

82.6

87.6

71.4

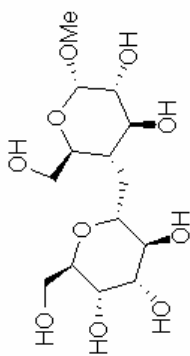
71.7

72.4

72.6

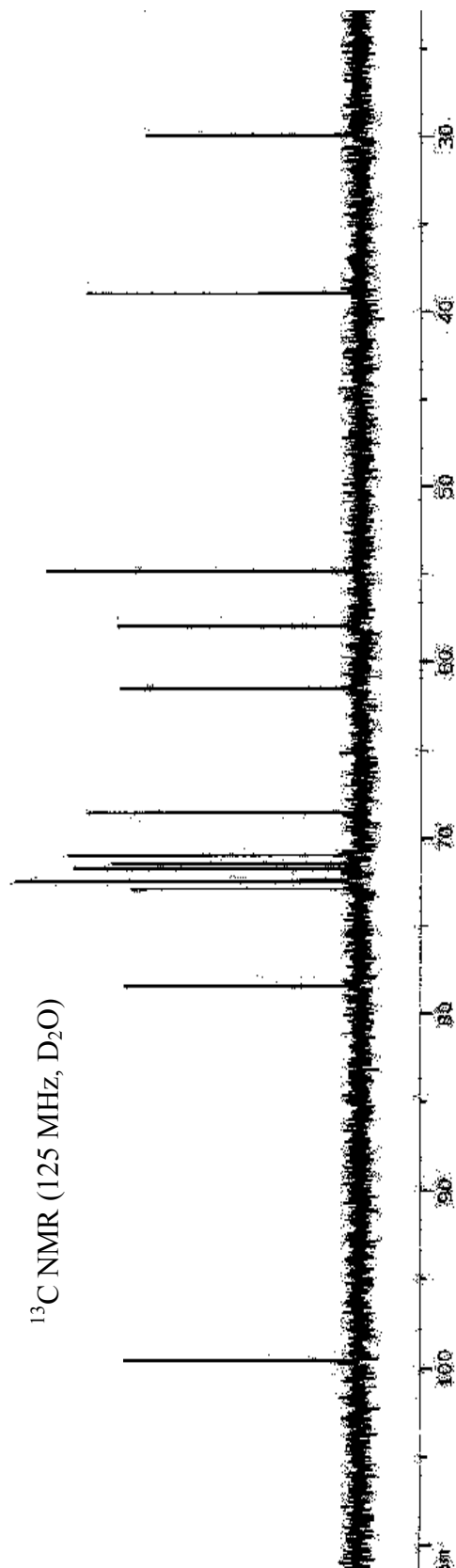
56.3

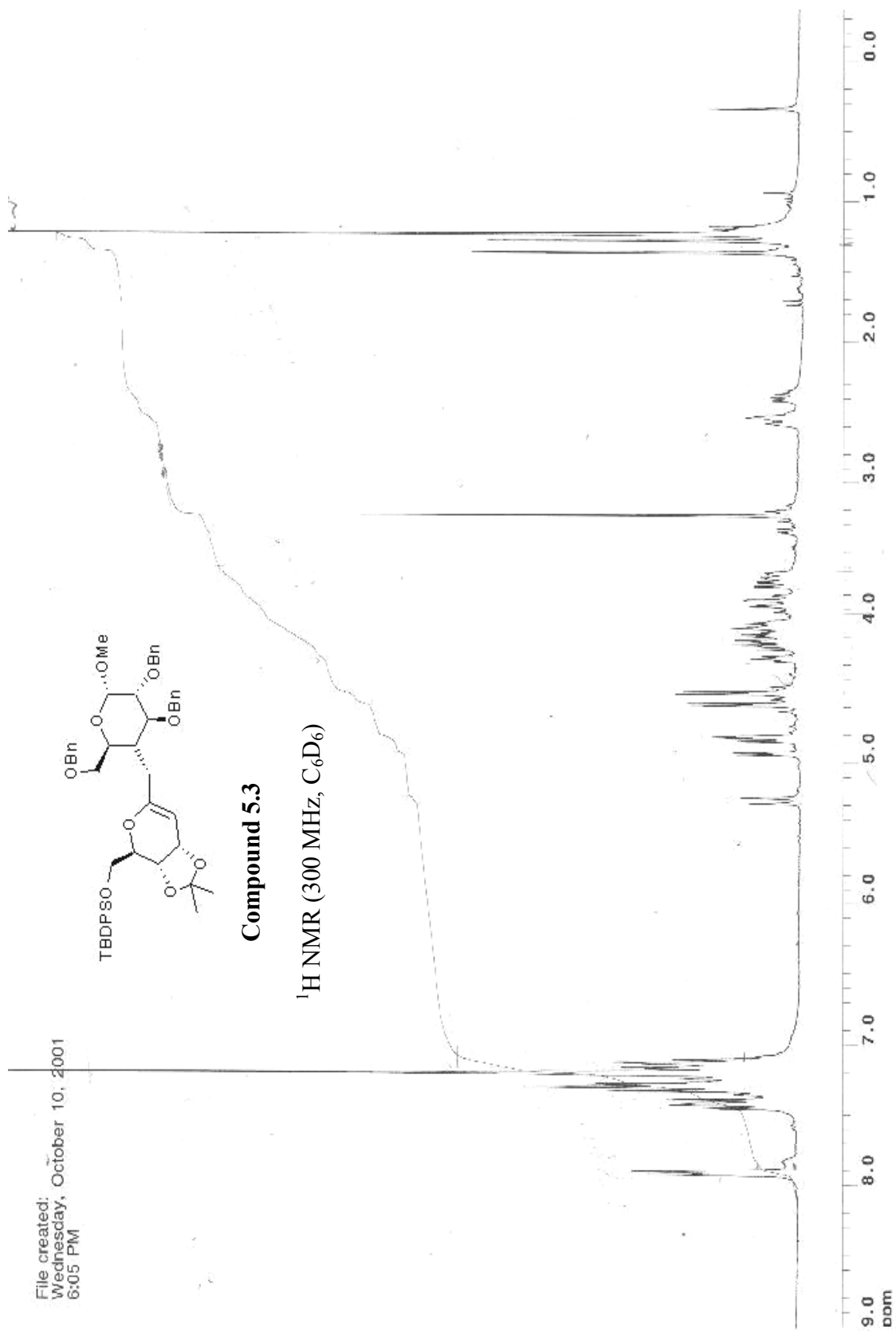
100.0



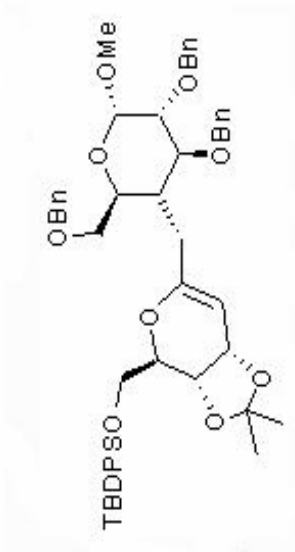
Compound 5.2

^{13}C NMR (125 MHz, D_2O)



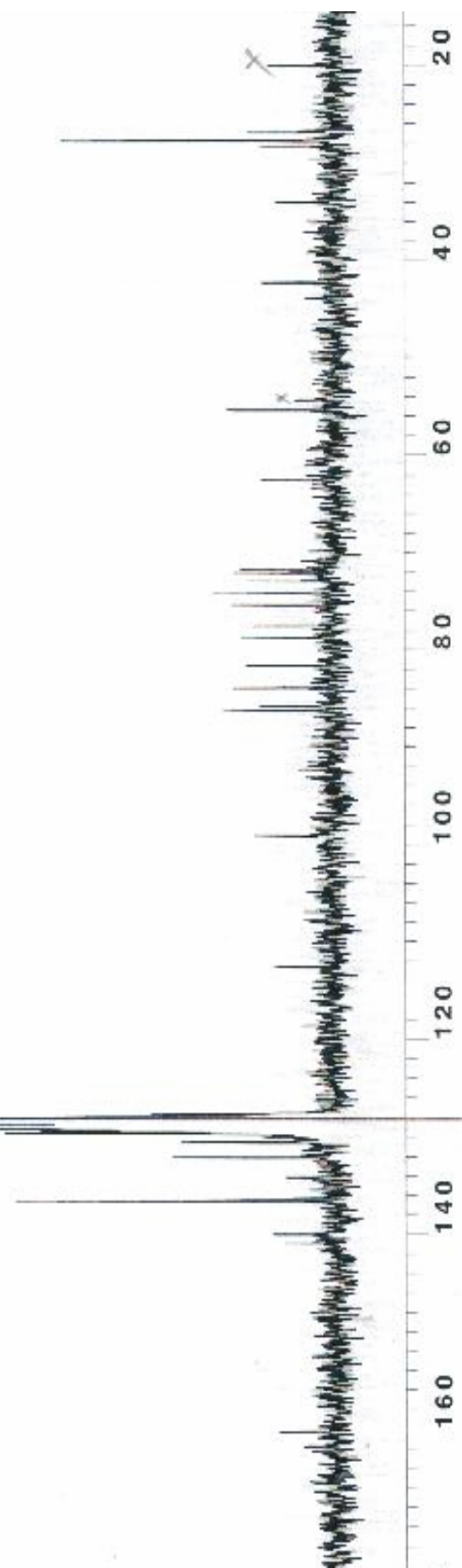


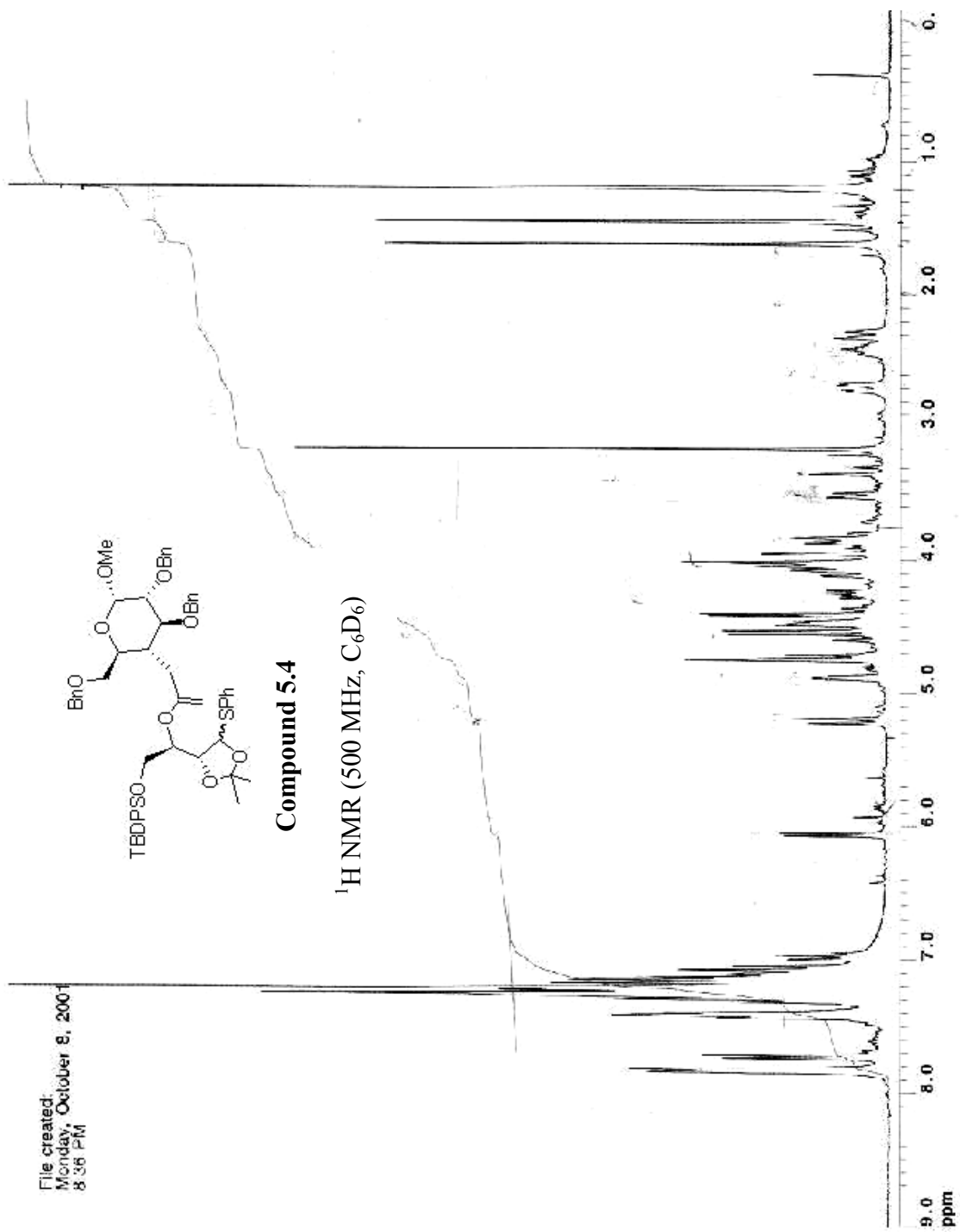
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Wednesday, October 9, 2002
7:47 PM

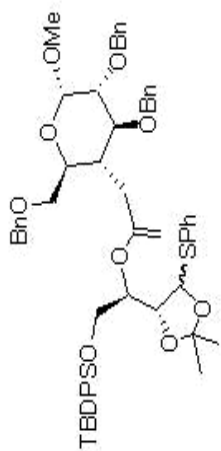


Compound 5.3

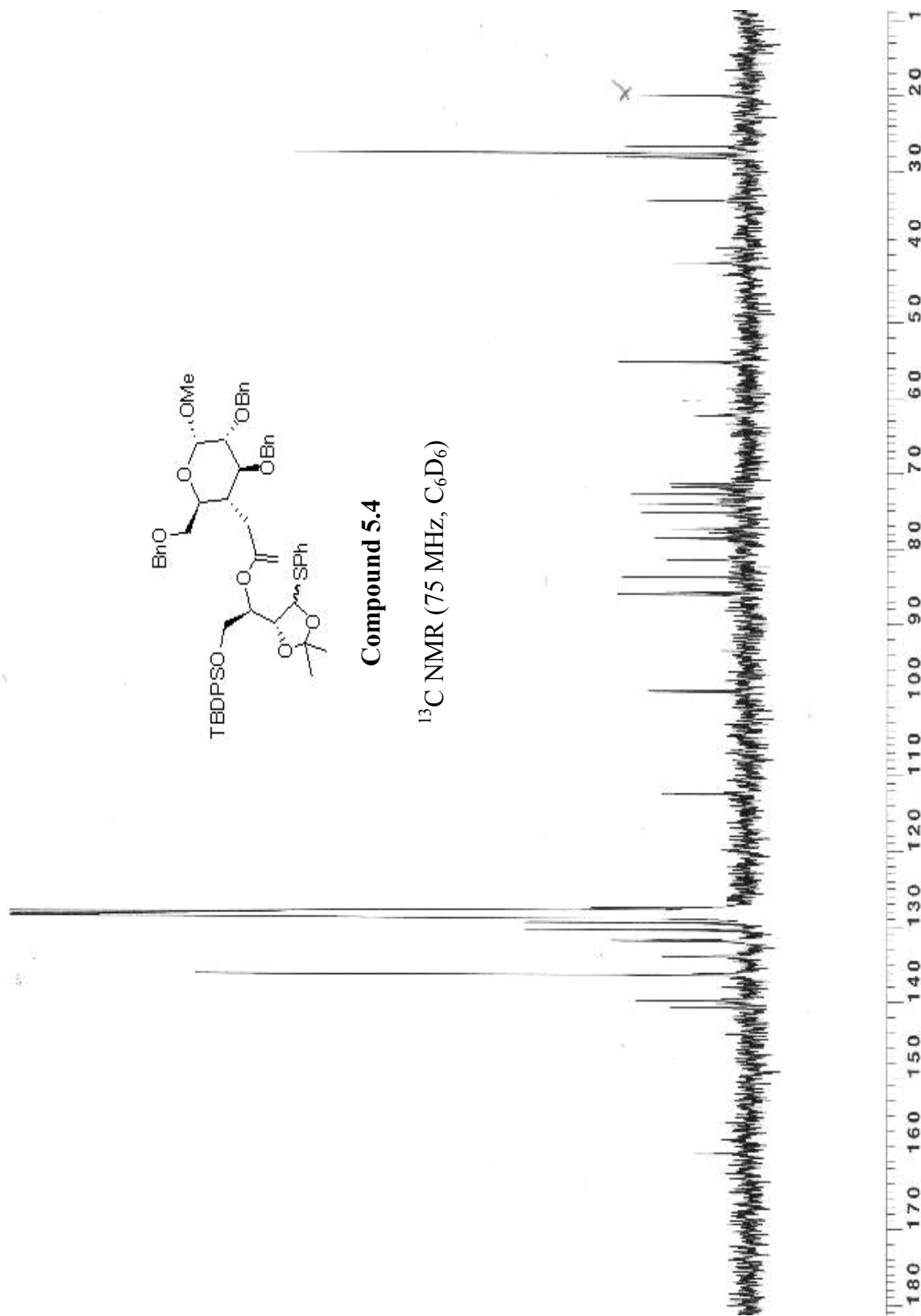
^{13}C NMR (75 MHz, C_6D_6)

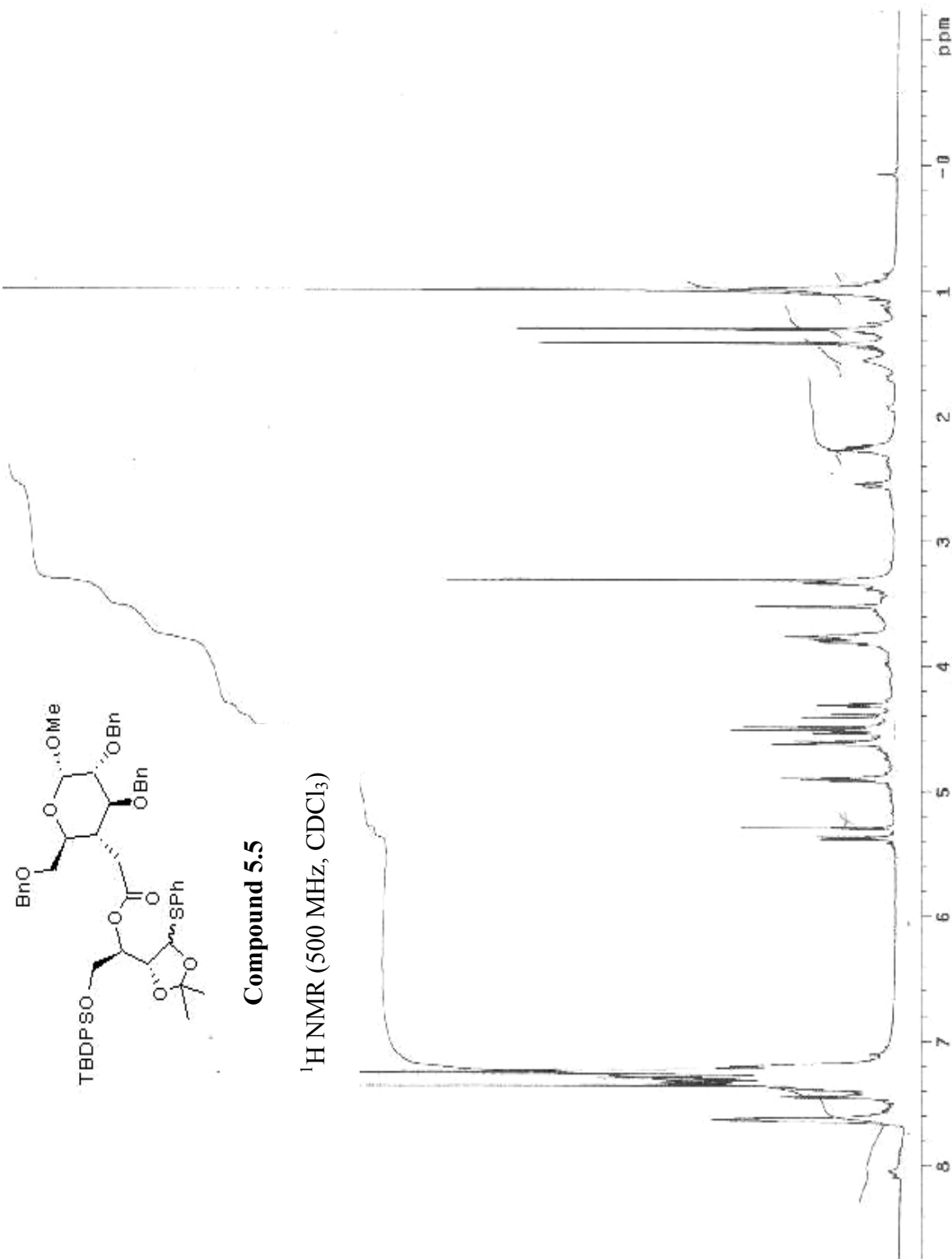




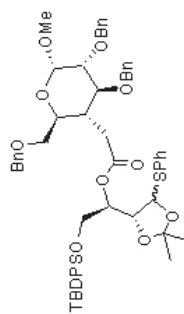
**Compound 5.4**

^{13}C NMR (75 MHz, C_6D_6)



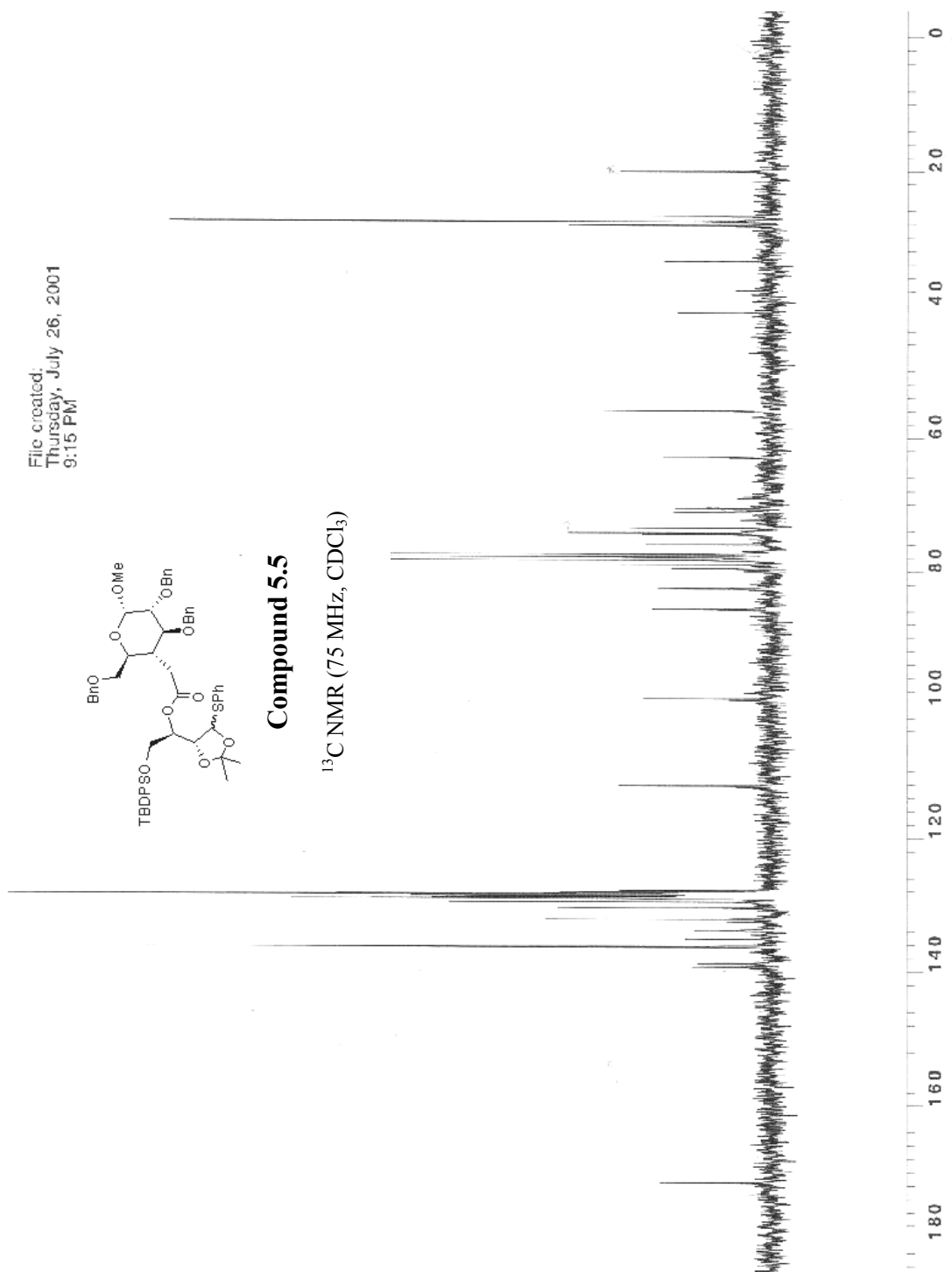


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Thursday, July 26, 2001
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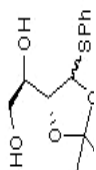


Compound 5.5

^{13}C NMR (75 MHz, CDCl_3)

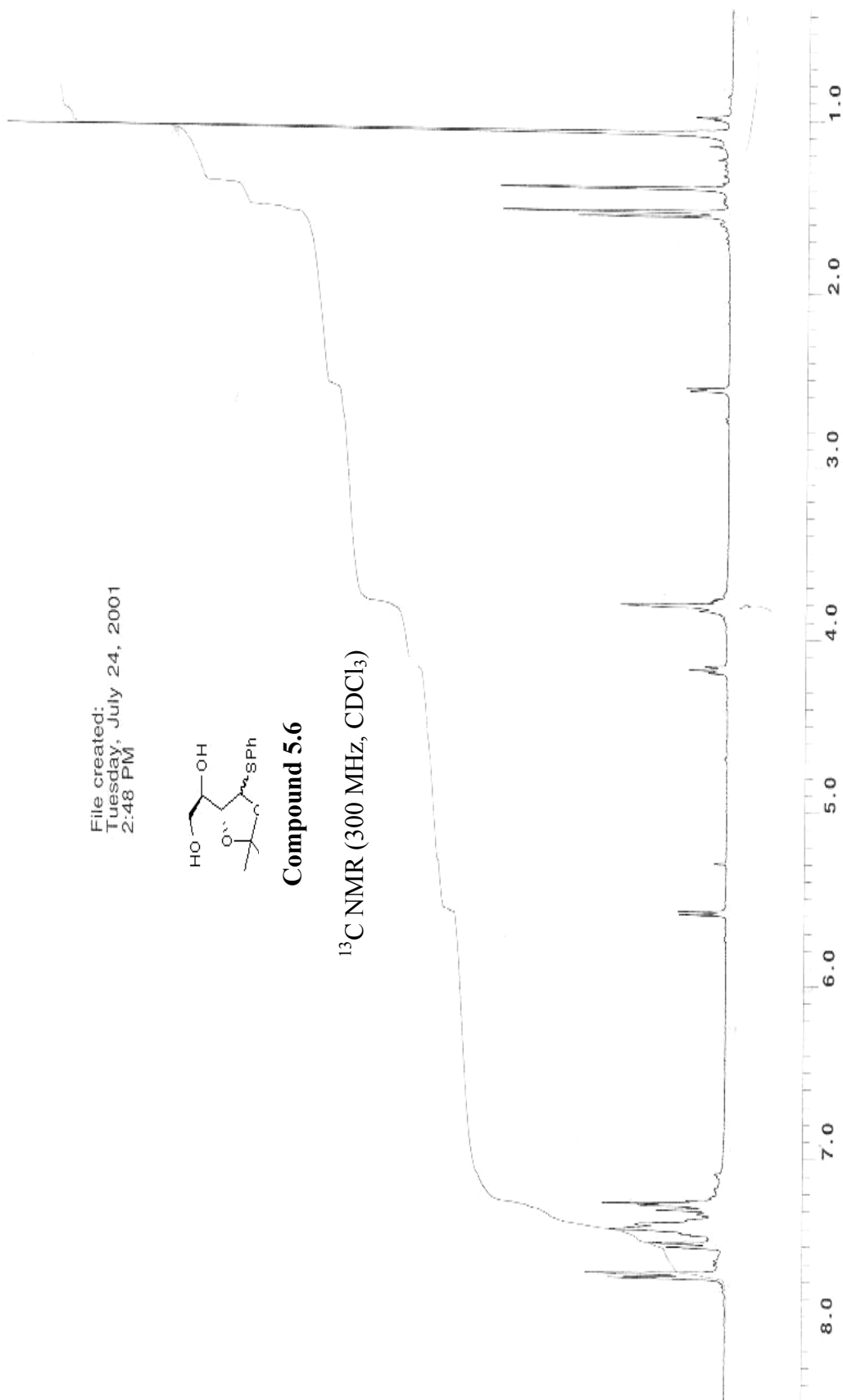


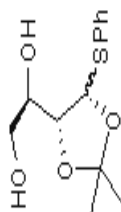
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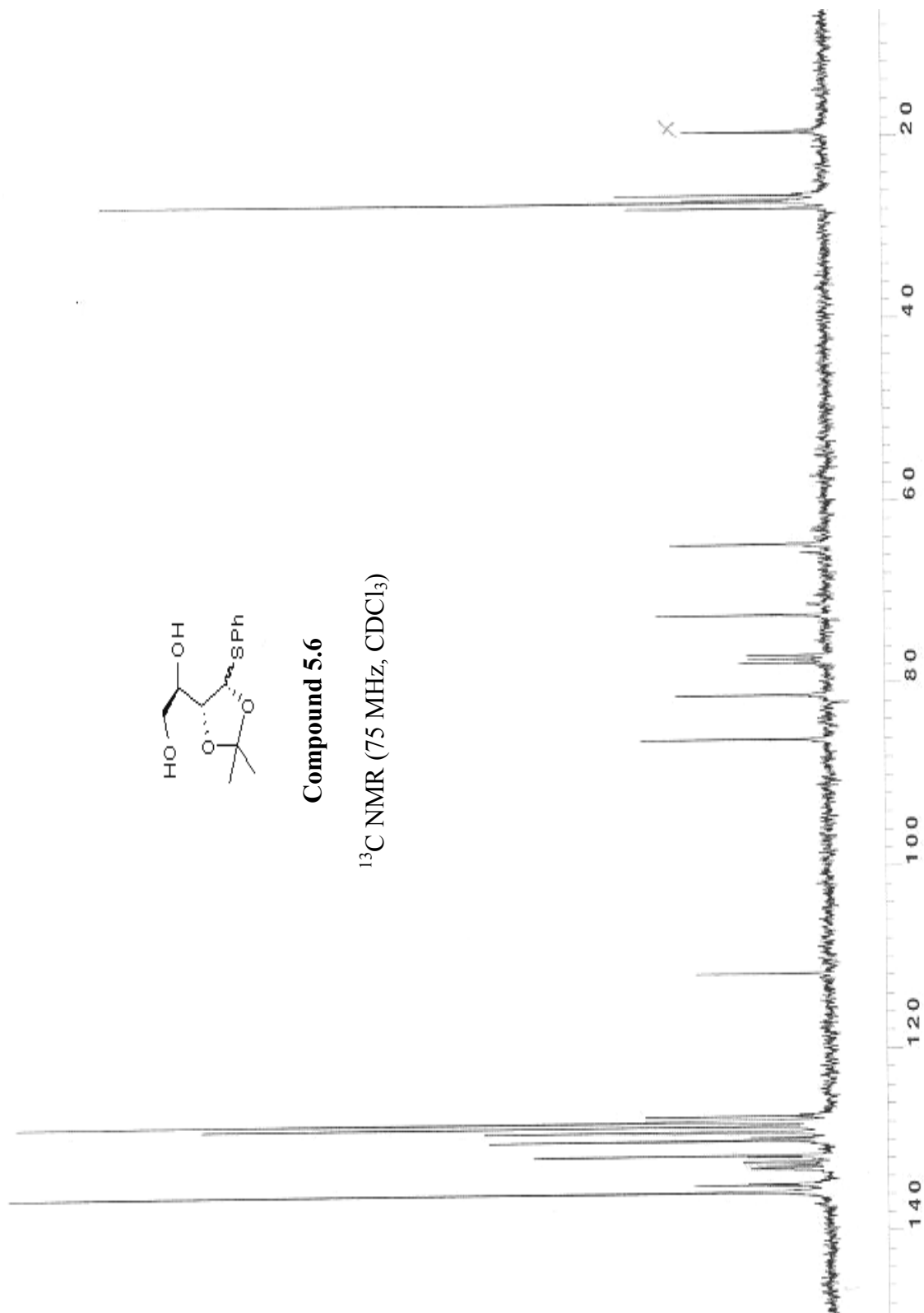
Compound 5.6

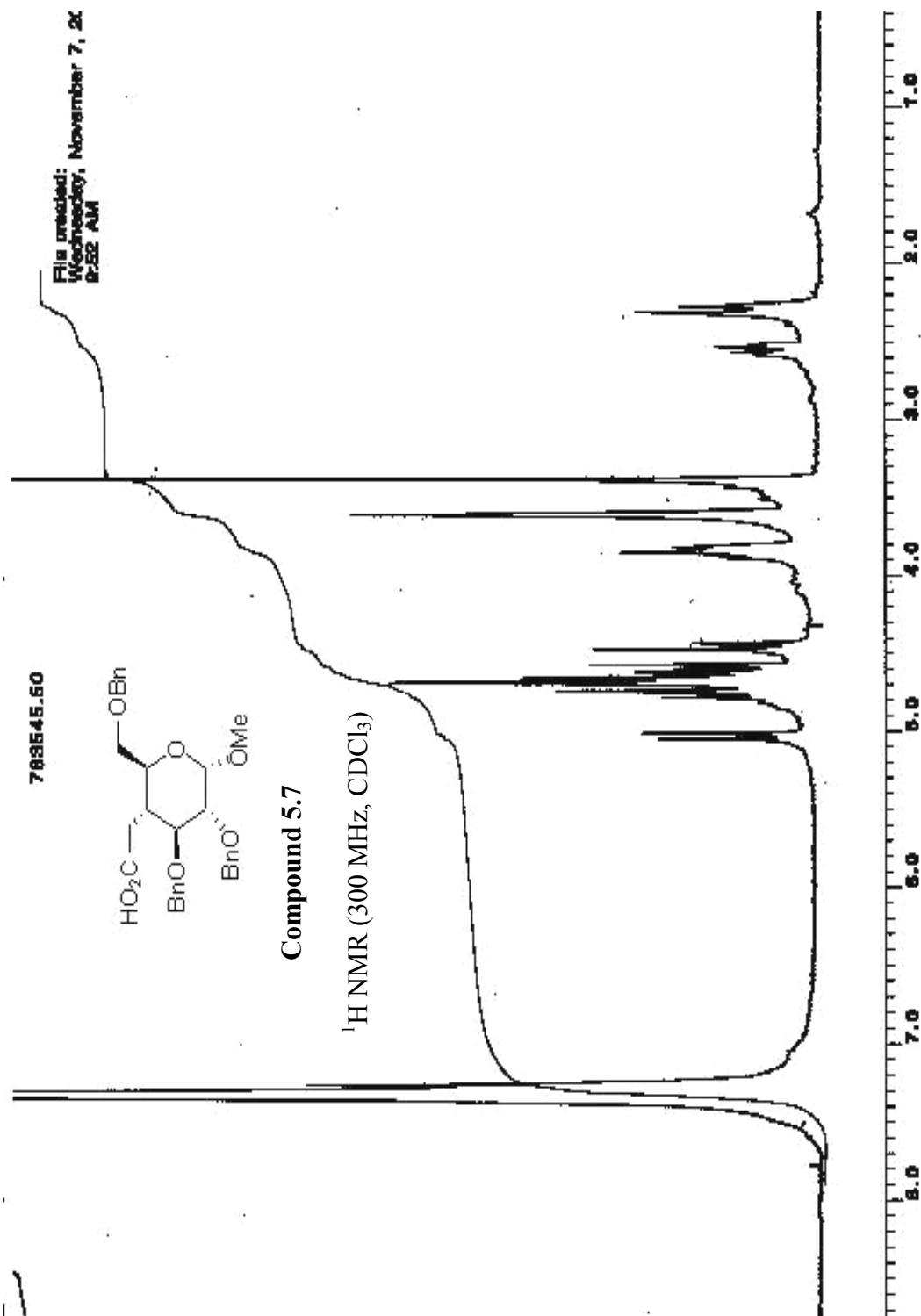
^{13}C NMR (300 MHz, CDCl_3)

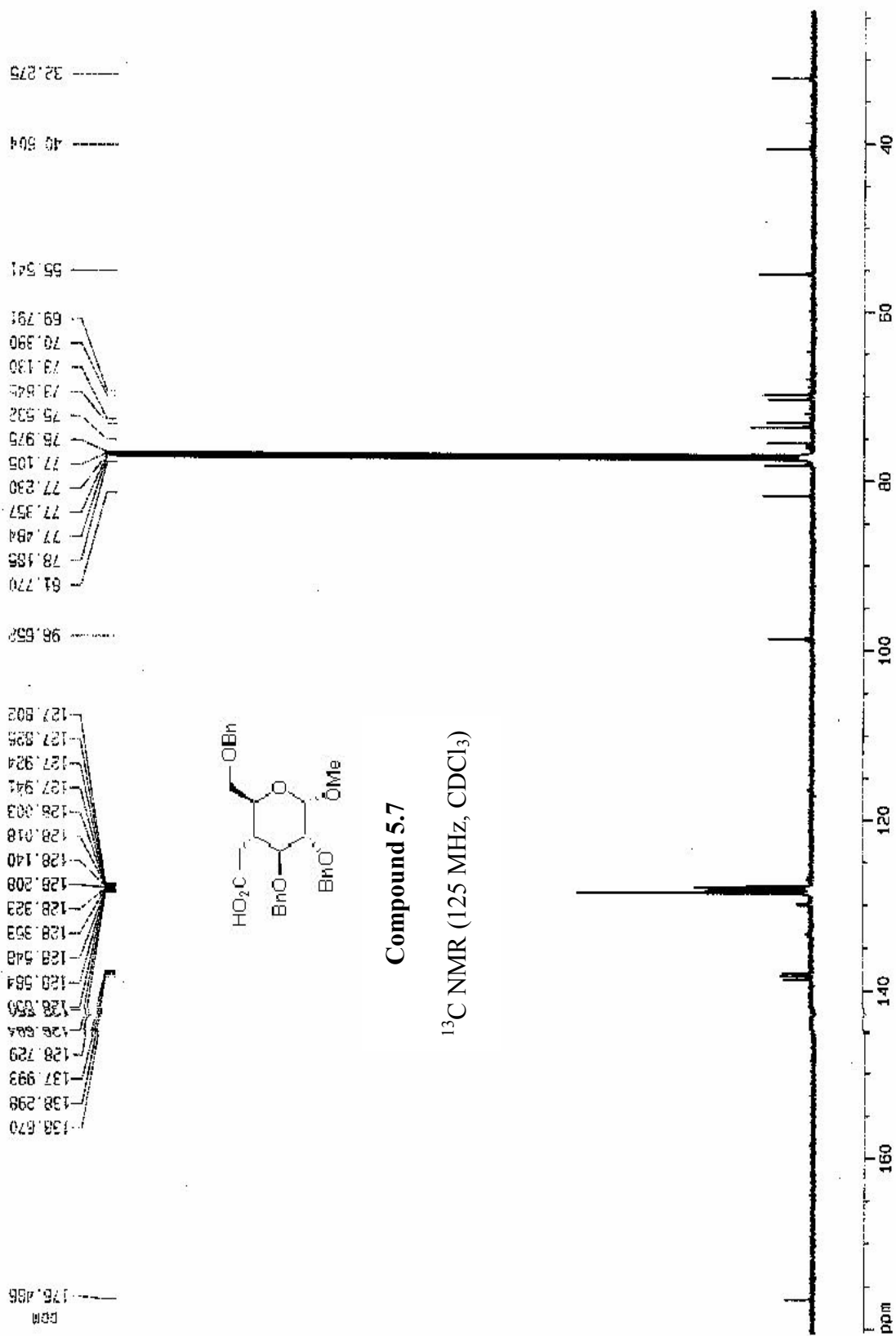


**Compound 5.6**

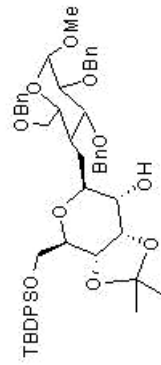
¹³C NMR (75 MHz, CDCl₃)





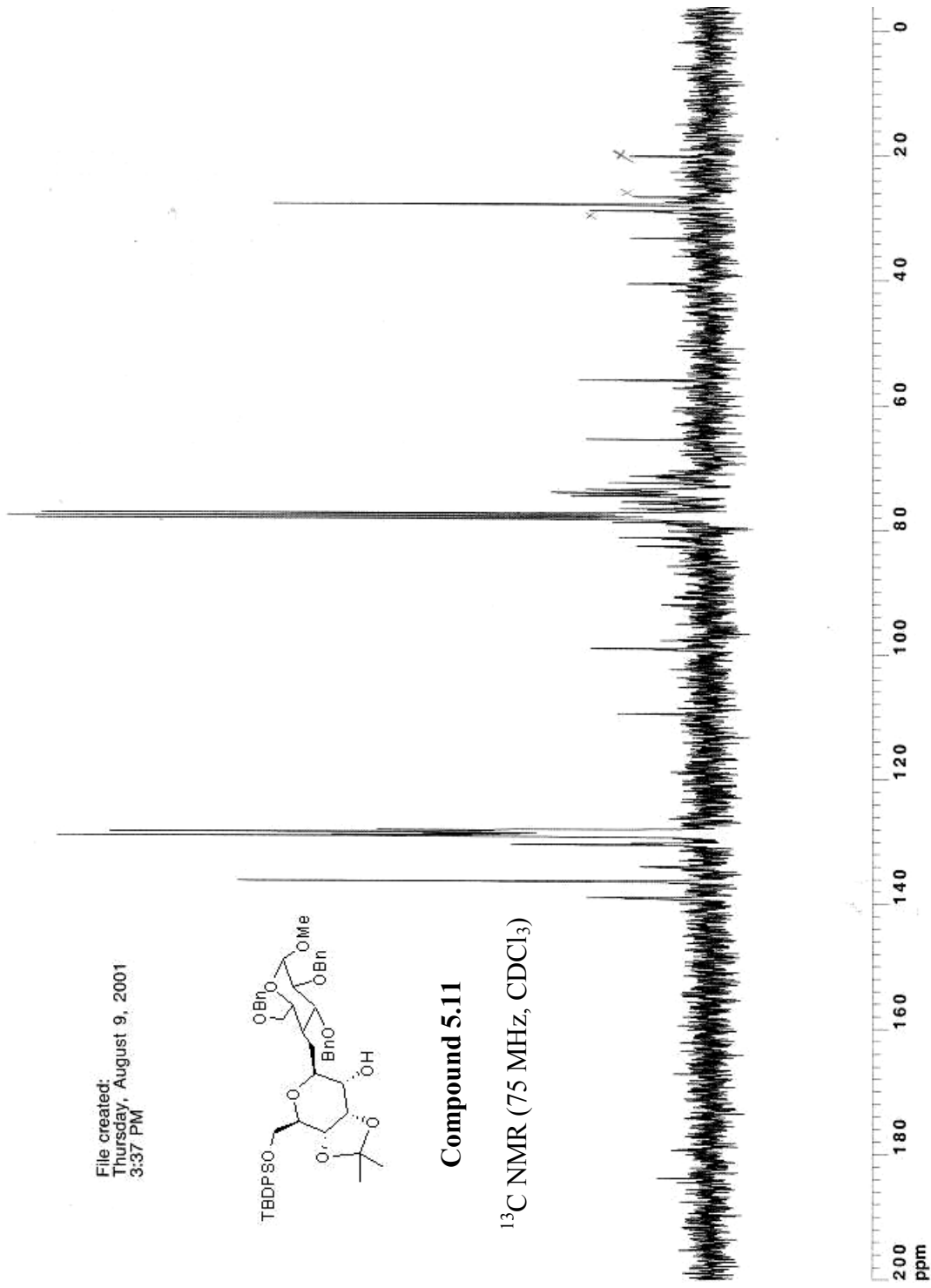


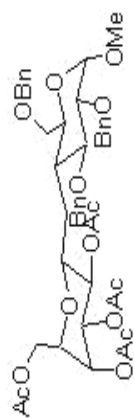
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Compound 5.11

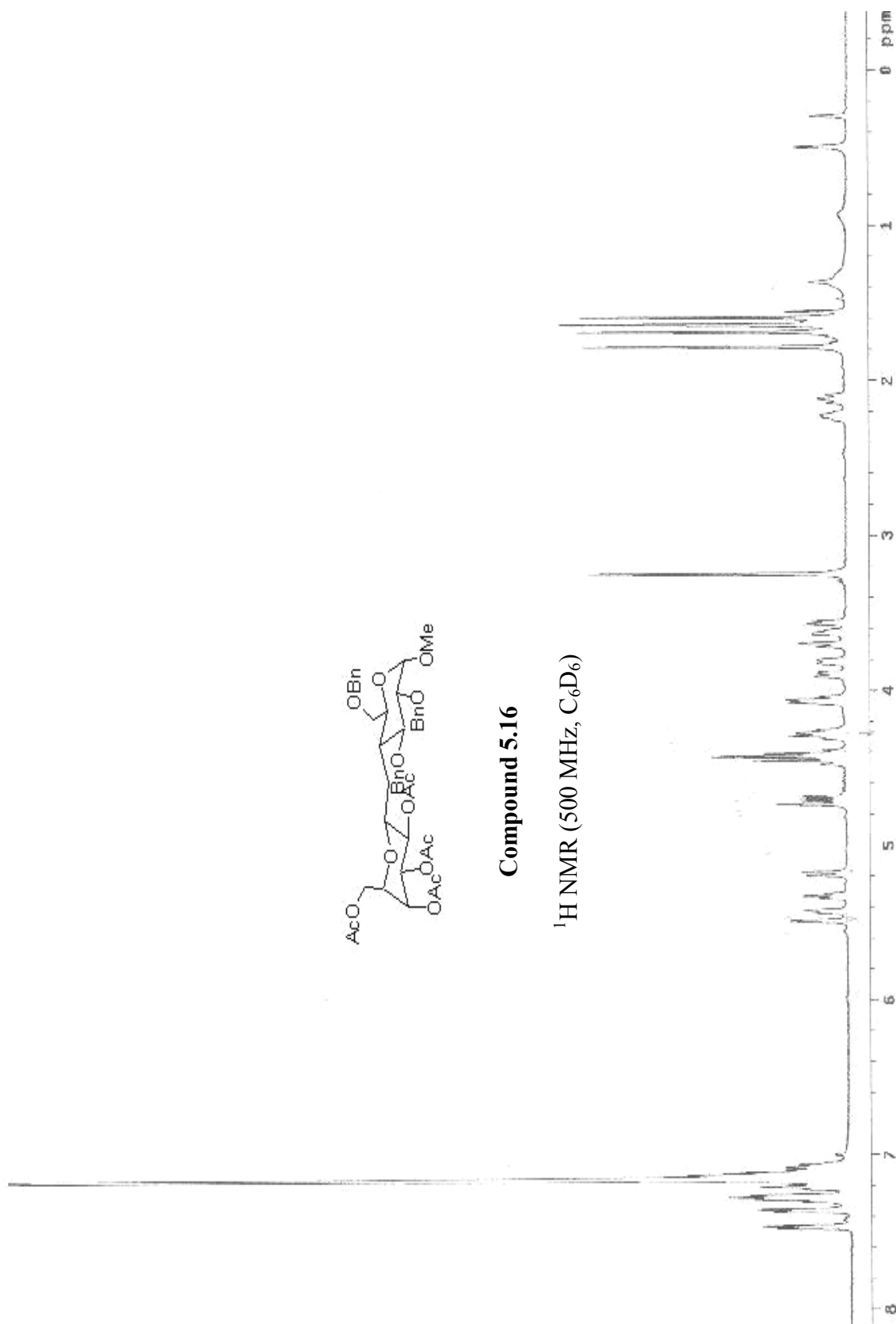
¹³C NMR (75 MHz, CDCl₃)





Compound 5.16

^1H NMR (500 MHz, C_6D_6)

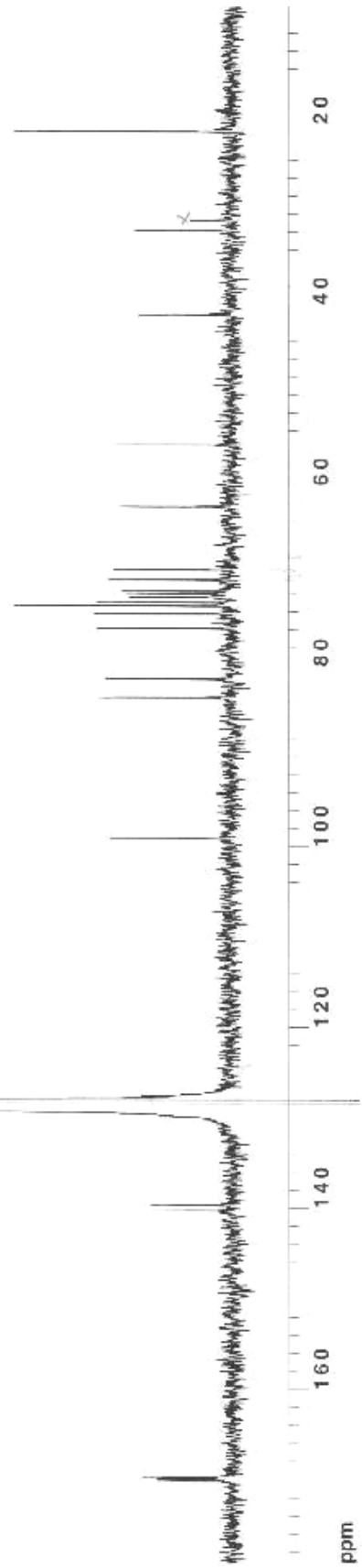


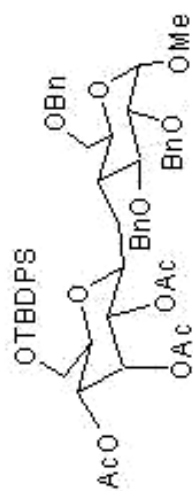
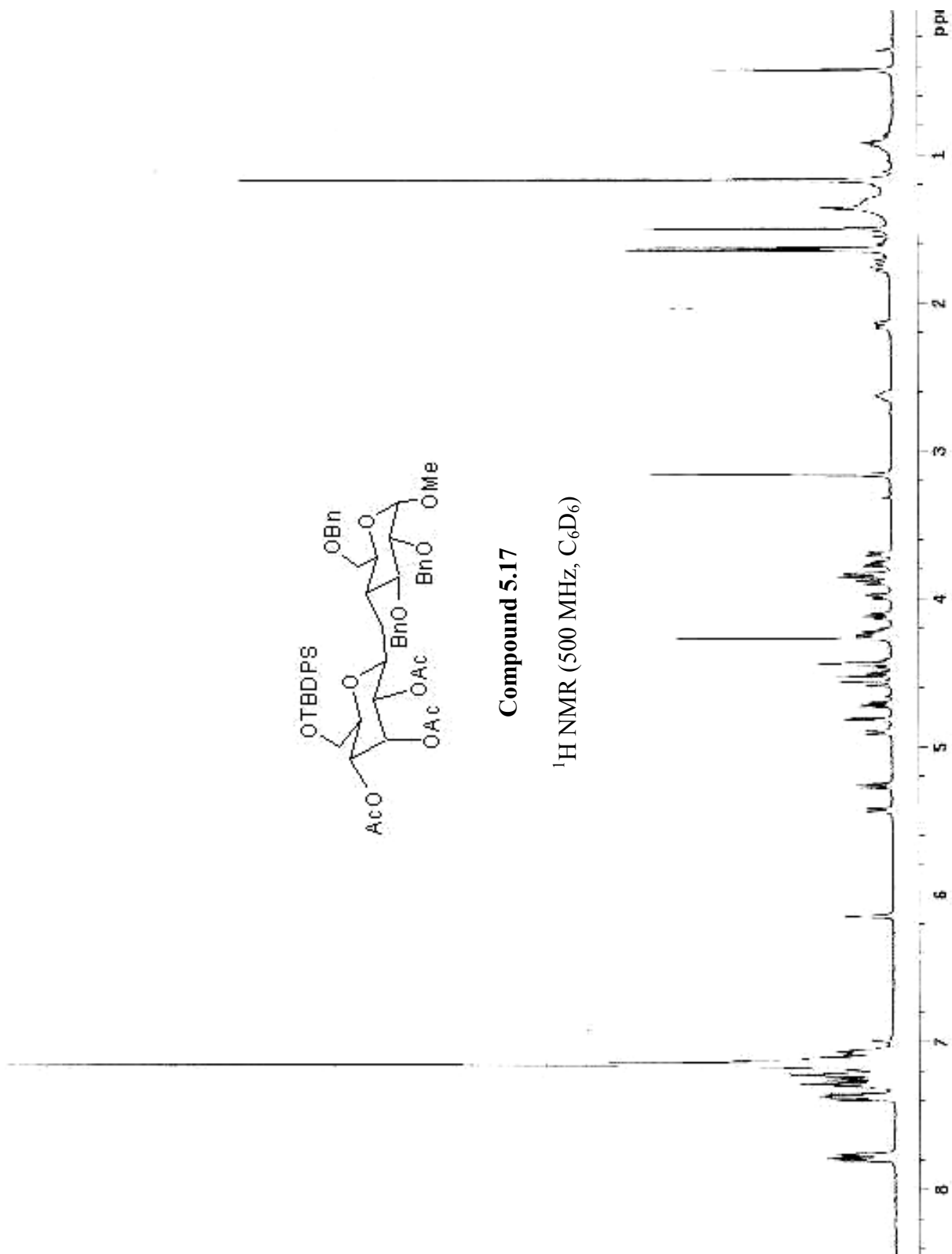
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Tuesday, October 22, 2002
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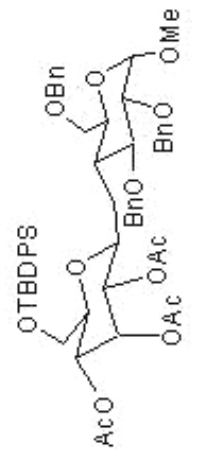
Compound 5.16

^{13}C NMR (75 MHz, C_6D_6)



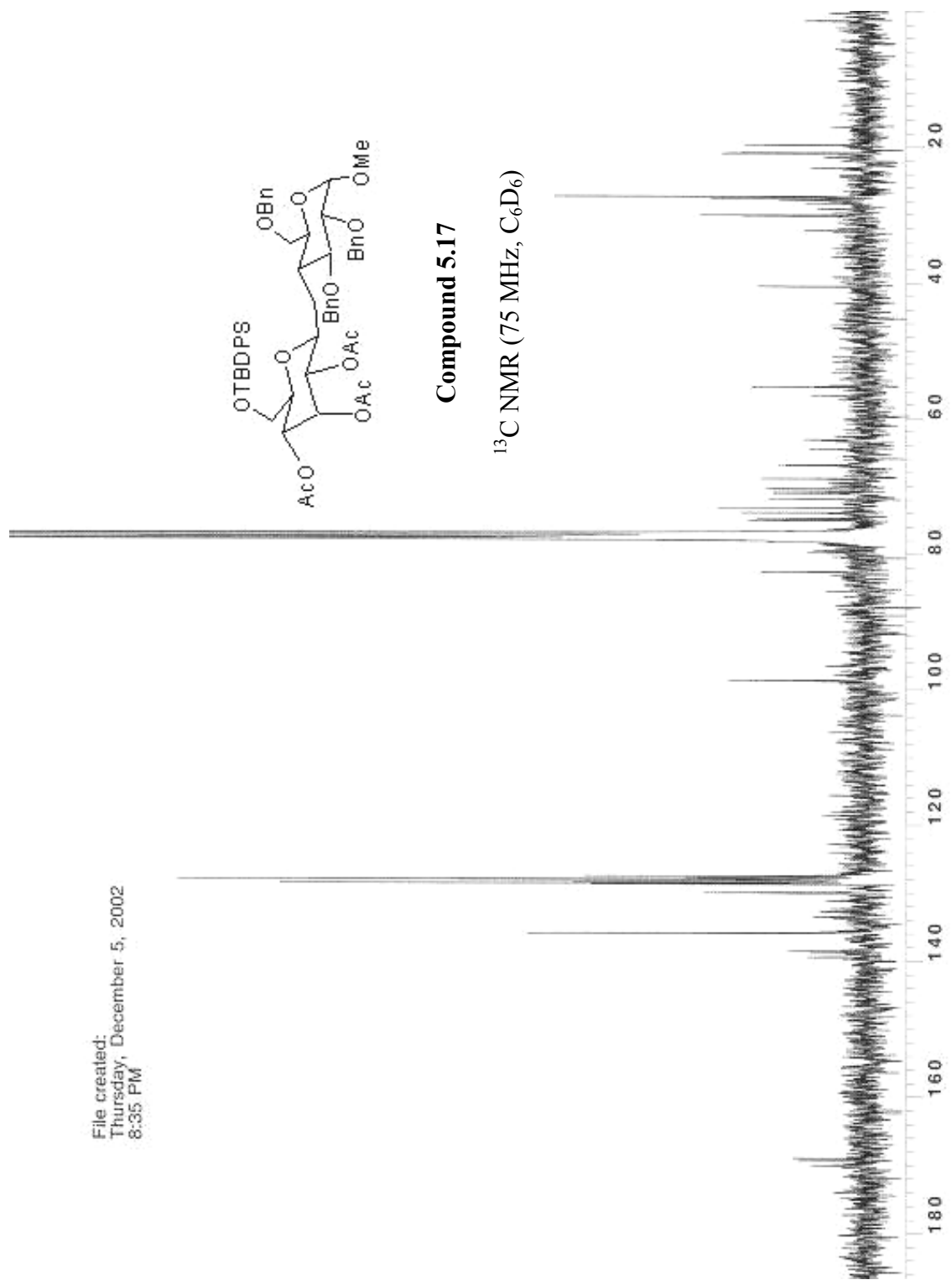
**Compound 5.17**¹H NMR (500 MHz, C₆D₆)

File created:
Thursday, December 5, 2002
8:35 PM



Compound 5.17

¹³C NMR (75 MHz, C₆D₆)



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