

## **INFORMATION TO USERS**

The most advanced technology has been used to photograph and reproduce this manuscript from the microfilm master. UMI films the text directly from the original or copy submitted. Thus, some thesis and dissertation copies are in typewriter face, while others may be from any type of computer printer.

**The quality of this reproduction is dependent upon the quality of the copy submitted.** Broken or indistinct print, colored or poor quality illustrations and photographs, print bleedthrough, substandard margins, and improper alignment can adversely affect reproduction.

In the unlikely event that the author did not send UMI a complete manuscript and there are missing pages, these will be noted. Also, if unauthorized copyright material had to be removed, a note will indicate the deletion.

Oversize materials (e.g., maps, drawings, charts) are reproduced by sectioning the original, beginning at the upper left-hand corner and continuing from left to right in equal sections with small overlaps. Each original is also photographed in one exposure and is included in reduced form at the back of the book.

Photographs included in the original manuscript have been reproduced xerographically in this copy. Higher quality 6" x 9" black and white photographic prints are available for any photographs or illustrations appearing in this copy for an additional charge. Contact UMI directly to order.

# **U·M·I**

University Microfilms International  
A Bell & Howell Information Company  
300 North Zeeb Road, Ann Arbor, MI 48106-1346 USA  
313/761-4700 800/521-0600



**Order Number 9108107**

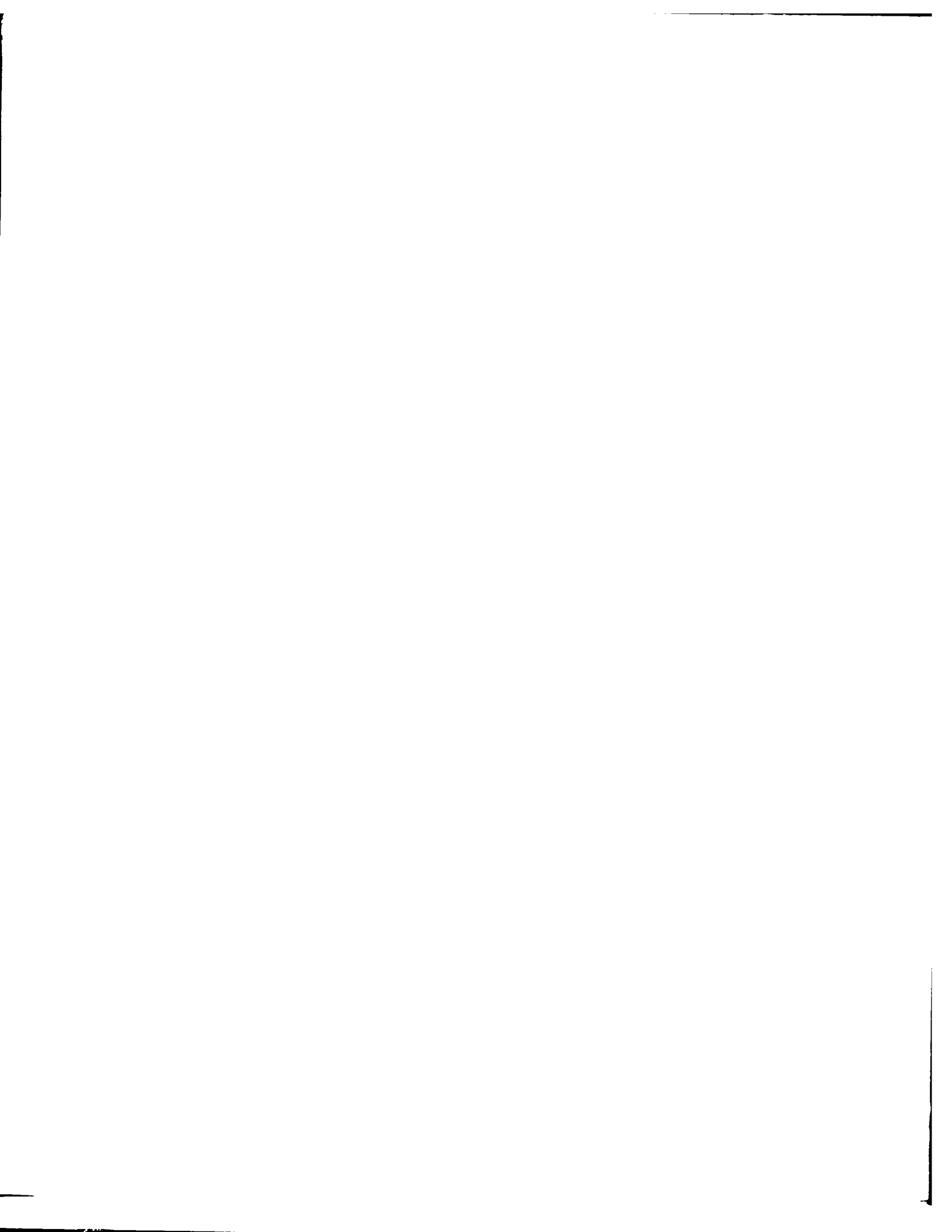
**Studies of the correlation energy**

**Gao, Xue-Qin, Ph.D.**

**City University of New York, 1990**

**Copyright ©1990 by Gao, Xue-Qin. All rights reserved.**

**U·M·I**  
300 N. Zeeb Rd.  
Ann Arbor, MI 48106



STUDIES OF THE CORRELATION ENERGY

BY

XUE-QIN GAO

A dissertation submitted to the graduate faculty in chemistry in partial fulfillment of the requirements for the degree of doctor of philosophy, The City of University of New York.

1990

• 1990

XUE-QIN GAO

All rights reserved

This manuscript has been read and accepted for the Graduate Faculty in Chemistry in satisfaction of the dissertation requirement for the degree of Doctor of Philosophy.

9/27/90  
Date

Lon Massa  
Chair of Examining Committee

9/27/90  
Date

[Signature]  
Executive Officer

Lon Massa  
[Signature]  
[Signature]  
Supervisory Committee

The City University of New York

**ABSTRACT****STUDIES OF THE CORRELATION ENERGY**

BY

XUE-QIN GAO

**ADVISOR: PROFESSOR LOU MASSA**

Colle and Salvetti introduced a method to calculate the correlation energy  $E_c$ , based upon a correlated wavefunction which is obtained by multiplying a single determinant by a correlation factor.

In this thesis the Colle and Salvetti theoretical and numerical formulas are tested with He, Be, Ne and their isoelectronic series up to  $z=36$ . We find the Colle-Salvetti formulas break down for ions with high value of  $Z$ .

We investigate a new formalism for calculating the correlation energy  $E_c$ . It is inspired by Colle and Salvetti's work, but differs in emphasizing the  $N$ -representability of the density matrix. The correlation factor is studied in a generalized form. This new formalism is tested with He and its isoelectronic ions. The calculations of  $E_c$  integrals employ our 3 dimensional Simpson integration program without any approximate series expansion. 70% Of the exact correlation energies have been obtained using all terms of the usual Hamiltonian.

Further work to extend our method is suggested.

### Acknowledgement

I would like to express special thanks to professor Lou Massa whose ideas have been most helpful to me. I am indebted to the members of my thesis committee, professor Avigdor Ronn and professor John Lombardi. I thank professor Mihaly Mezei for his help with computational work; Doctor Maria Flocco for discussions at an early stage of the work. I thank my fellow graduate student Cathy Abrams for her considerable help and for her kind support.

CONTENTS

	PAGE
CHAPTER I .....	(1)
INTRODUCTION	
CHAPTER II .....	(4)
BACKGROUND	
CHAPTER III .....	(20)
THE COLLE-SALVETTI FORMALISM FOR CORRELATION ENERGY	
(A) THE COLLE-SALVETTI FORMULA .....	(20)
(B) CALCULATIONS OF THE CORRELATION ENERGY FOR	
He, Be, Ne AND THEIR ISOELECTRONIC IONS .....	(28)
(C) REMARKS ON COLLE-SALVETTI FORMULA .....	(35)
CHAPTER IV .....	(42)
A FORMALISM FOR CORRELATION ENERGY	
(A) DERIVATION OF ONE-BODY AND TWO-BODY	
REDUCED DENSITY MATRICES .....	(43)
(B) DERIVATION OF THE FORMULA FOR	
THE CORRELATION ENERGY .....	(47)
(C) CALCULATION OF THE CORRELATION ENERGY	
FOR He AND ITS ISOELECTRONIC IONS .....	(53)
(D) CONCLUSION .....	(81)
BIBLIOGRAPHY .....	(137)

**APPENDIX**

	<b>PAGE</b>
<b>APPENDIX 1</b> .....	<b>(85)</b>
<b>NUMERICAL COORDINATES IN 3 DIMENSIONS</b>	
<b>AND SIMPSON NUMERICAL INTEGRATION</b>	
<b>APPENDIX 2</b> .....	<b>(94)</b>
<b>TAYLOR EXPANSION OF SECOND ORDER</b>	
<b>DENSITY MATRICES IN THE COLLE</b>	
<b>AND SALVETTI NUMERICAL FORMULA</b>	
<b>APPENDIX 3</b> .....	<b>(121)</b>
<b>THE FORMULAS FOR THE KINETIC</b>	
<b>CORRELATION ENERGY</b>	

LIST OF TABLES

	PAGE
(1) TABLE 2-1 .....	(19)
A comparison of exact HF and X-ray orbitals single-determinant on same basis by some expectation values of Li and Be atoms	
(2) TABLE 3-1 .....	(26)
The Correlation Energy of Some Atoms and Small Molecules Calculated by Colle-Salvetti	
(3) TABLE 3-2 .....	(27)
The Excitation Energy, Ionization Potential, Dissociation Energy Calculated by Colle-Salvetti with Their Numerical Formula	
(4) TABLE 3-3 .....	(29)
The Correlation Energy of He Isoelectronic Series Calculated by 3D SI with $\frac{1}{2}\langle P_2^b/r \rangle$ and Compared one from C-S Numerical Formula and One from Clementi	
(5) TABLE 3-4 .....	(31)
The Correlation Energy of Be Isoelectronic Series Calculated by 3D SI with $\frac{1}{2}\langle P_2^b/r \rangle$ and Compared one from C-S Numerical Formula and One from Clementi	
(6) TABLE 3-5 .....	(33)
The Correlation Energy of Ne Isoelectronic Series Calculated by 3D SI with $\frac{1}{2}\langle P_2^b/r \rangle$ and Compared one from C-S numerical Formula and One from Clementi	

- (7) TABLE 3.6 ..... (38)  
 $\langle P_2 b \rangle$  of He Atom Calculated by  
 3D SI with the C-S Functional
- (8) TABLE 3.7 ..... (40)  
 $-Z \langle P_2 b / r_1 \rangle$  of He Atom Calculated  
 by 3D SI with the C-S Functional
- (8) TABLE 4.1 ..... (55)  
 The Correlation Energy of He Atom and Its  
 Isoelectronic Ions Calculated by Varying  
 The  $u$  Parameter ( $q=1.4068$ )  
 (A):He (page 55)            (B):Li<sup>+</sup> (PAGE 58)  
 (C):Be<sup>+2</sup> (PAGE 61)        (D):B<sup>+3</sup> (PAGE 64)
- (9) TABLE 4.2 ..... (67)  
 The Correlation Energy of He and Its Isoelectronic  
 Ions at The Minimum ( $q=1.4068$ )
- (10) TABLE 4.3 ..... (68)  
 The Correlation Energy of He Atom Calculated  
 by Varying Both  $u$  and  $q$  Parameters  
 (A)  $u=1.0$  (PAGE 68)        (C)  $u=0.93$  (PAGE 70)  
 (B)  $u=0.97$  AND  $u=0.95$  (PAGE 69)
- (11) TABLE 4.4 ..... (72)  
 The Correlation Energy of Li<sup>+</sup> Ion Calculated  
 by Varying Both  $u$  and  $q$  Parameters  
 (A)  $u=1.0$  (PAGE 72)        (B)  $u=0.96$  (PAGE 73)  
 (C)  $u=1.10$  (PAGE 73)

- (12) TABLE 4.5 .....(75)  
The Correlation Energy of  $\text{Be}^{+2}$  Ion Calculated  
by Varying Both  $u$  and  $q$  Parameters  
(A)  $u=1.0$  AND  $u=0.97$  (PAGE 75)  
(B)  $u=0.96$  (PAGE 76)
- (13) TABLE 4.6 .....(78)  
The Best Values of the Correlation Energy  
of He and its Isoelectronic Ions by 3D SI  
with Varying Both  $q$  and  $u$  Parameters
- (14) TABLE 4.7 .....(79)  
The Correlation Energy  $E_c$  of He and its  
Isoelectronic Ions Calculated by 3D SI and  
Compared with the Nearly Exact Values
- (15) TABLE 4.8 .....(84)  
The Percentage Correlation for He Atom

LIST OF FIGURES

- (1) Figure 2.1 ..... (30)  
 A Comparison of Ec: He Isoelectronic Series
- (2) Figure 2.2 ..... (32)  
 A Comparison of Ec: Be Isoelectronic Series
- (3) Figure 2.3 ..... (34)  
 A Comparison of Ec: Ne Isoelectronic Series
- (4) Figure 2.4 ..... (39)  
 A Calculation  $\langle b \cdot P_2 \rangle$  of He
- (5) Figure 2.5 ..... (41)  
 A Calculation  $V_c(2)$  of He
- (6) Figure 3.1A ..... (56)  
 The Correlation Energy He ( $q=1.4068$ )
- (7) Figure 3.1B ..... (57)  
 A calculation of Ec for He ( $q=1.4068$ )
- (8) Figure 3.2A ..... (59)  
 The Correlation Energy  $Li^+$  ( $q=1.4068$ )
- (9) Figure 3.2B ..... (60)  
 A calculation of Ec for  $Li^+$  ( $q=1.4068$ )
- (10) Figure 3.3A ..... (62)  
 A calculation of Ec for  $Be^{+2}$  ( $q=1.4068$ )
- (11) Figure 3.3B ..... (63)  
 A calculation of Ec for  $Be^{+2}$  ( $q=1.4068$ )
- (12) Figure 3.4A ..... (65)  
 A calculation of Ec for  $B^{+3}$  ( $q=1.4068$ )

- (13) Figure 3.4B .....(66)  
A calculation of  $E_c$  for  $B^{+3}$  ( $q=1.4068$ )
- (14) Figure 4.1 .....(71)  
Ec of He Atom from double parameters  $q$  and  $u$
- (15) Figure 4.2 .....(74)  
Ec of  $Li^+$  Ion from double parameters  $q$  and  $u$
- (16) Figure 4.3 .....(77)  
Ec of  $Be^{+2}$  Ion from double parameters  $q$  and  $u$
- (17) Figure 5.1 .....(80)  
A Comparison of  $E_c$  Values Calculated by 3D SI

## CHAPTER I INTRODUCTION

A definition of correlation energy  $E_c$  given by Löwdin [1] is

$$E_c = E_{\text{exact}} - E_{\text{HF}} \quad (1-1)$$

where HF means the Hartree-Fock value. The electron correlation error of the HF approximation has been recognized as a central problem in atomic structure theory. There are two basic methods for calculating the correlation energy: one involves CI (configuration interaction), another involves correlation factors including interparticle distance vectors.

Colle and Salvetti introduced a method [2] to calculate the correlation energy  $E_c$  of a closed-shell system with a correlation factor starting from the known HF orbitals. A number of their results are reported to be close to exact values.

In this dissertation the Colle and Salvetti formalism (C-S formalism) is analyzed using our 3 dimensional Simpson integration program (3D SI). The C-S numerical and theoretical formulas (discussed in chapter III) are used to calculate the correlation energies of He, Be, Ne and their isoelectronic series up to  $Z=36$ . In most cases for high values of  $Z$  the results from the C-S formalism are not in agreement with nearly exact values from Clementi [3,4], and some results are much lower than the nearly exact value, so that the

Variational principle is violated.

In this thesis the following aspects of the C-S formalism were studied:

<1> The normalization of the density matrix.

<2> Contributions of the correlation energy from kinetic and potential terms in the usual Hamiltonian.

<3> Empirical constants fixed in the C-S fitting procedure.

<4> Results from their theoretical formula and their numerical formula.

In this thesis a quantum chemistry formalism is introduced to improve the C-S procedure.

<1> A parameter  $t$  is introduced into the correlation functional to satisfy the normalization of the wave function. The available  $t$  value is obtained automatically by our optimizing procedure program.

<2> Kinetic correlation energy  $T_c$  and potential correlation energy  $V_c(2)$  from the electron-nuclear attraction term were included to improve  $E_c$ .

<3> Two variational parameters  $u$  and  $q$  are used to obtain the optimum  $E_c$ , which satisfies the Variational principle.

<4> Highly accurate approximately  $N$ -representable density matrices are constructed from a single determinant wave function of the HF orbitals for the calculations [5].

<5> All integrations are carried out using a new 3 dimensional Simpson integration program without any empirical constants and without any fitting procedure. This method

avoids the difficulties arising from a Taylor expansion of the density matrix.

This new formalism is tested with the He atom and its isoelectronic ions. The results supported our formalism as indicated by:

<1> The Calculated  $E_c$  has reached 70% and more of the exact correlation energy for the He atom and its isoelectronic series.

<2> Variational parameters for calculating the optimum value of  $E_c$  for the He isoelectronic series have been determined.

<3> The relative contribution of all terms contributing to the correlation energy  $E_c$ , have been determined. Each term has a clear physical meaning and a reasonable magnitude.

Future work in this area could improve  $E_c$  by varying  $q$  in smaller intervals. More example calculations of  $E_c$  could include the Be and Ne atoms and their isoelectronic series.

## CHAPTER II BACKGROUND

## (A). The correlation concepts.

If two electrons are non-interactive, from the multiplicative law for the simultaneous occurrence of independent events we have

$$P_2(12) = p_1(1)p_1(2) \quad (2-1)$$

where  $p_1(1)$  is the diagonal element of the one-body density matrix, and has the meaning of the probability density of finding one electron at point  $r_1$ ;  $p_1(2)$  is the diagonal element of the one body density matrix for another electron, with a similar physical meaning;  $P_2(12)$  has the meaning of the probability density of finding one electron at point  $r_1$  and another electron at point  $r_2$  simultaneously.

In an interactive situation involving two electrons, a correlation factor is introduced so that

$$P_2(12) = p_1(1)p_1(2)(1+f(12)) \quad (2-2)$$

(2-2) could be written as

$$\frac{P_2(12)}{p_1(1)p_1(2)} = 1 + f(12) \quad (2-3)$$

According to probability theory the left side of (2-3) is the

conditional probability of finding an electron at point  $r_1$  when another electron is at point  $r_2$ .

For point  $r_1$  approaching point  $r_2$  the correlation factor "f" is negative. There is a correlation "hole" in the one-particle probability surrounding any point at which there is already a particle. McWeeny described the correlation hole by a function [6,7]

$$[ F(1) ]_{r_2} = p_1(1)f(12) \quad (2-4)$$

Usually the density matrices are defined in terms of wave functions  $\Psi(1, \dots, N)$ , which depend on both space and spin coordinates. Under the antisymmetric requirement the wave function  $\Psi$  may be a single determinant of the orbital  $\phi_m(i)$  with the form

$$\Psi(1, \dots, N) = \frac{1}{N!} \begin{vmatrix} \phi_1(1) & \dots & \phi_1(N) \\ \vdots & & \vdots \\ \phi_N(1) & \dots & \phi_N(N) \end{vmatrix} \quad (2-5)$$

$$\phi_m = \sum C_k \Omega_k \quad (2-6)$$

where  $\Omega_k$  is an atomic basis, and

$$\langle \phi_m | \phi_n \rangle = \delta_{mn} \quad (2-7)$$

The Fock-Dirac density matrix [8] is

$$\hat{p}_1(11') = \sum_{i=1}^N \psi_i(1) \cdot \psi_i^*(1') \quad (\text{A}) (2-8)$$

$$\hat{p}_1(12') = \sum_{i=1}^N \psi_i(1) \cdot \psi_i^*(2') \quad (\text{B}) (2-8)$$

where  $\hat{p}(11')$  and  $\hat{p}(12')$  are with spin.

The spinless density matrices which are sometimes desirable, may be obtained by integrating over the spins. For antisymmetric wave functions  $\Psi(1, \dots, N)$ , the spinless density matrices  $p_1(11')$  assume the form

$$p_1(11') = p_1^{\alpha\alpha}(11') + p_1^{\beta\beta}(11') \quad (2-9)$$

where  $p_1^{\alpha\alpha}(11')$  consists of all the terms in the Fock-Dirac density matrix with spin  $\alpha$ , and  $p_1^{\beta\beta}(11')$  is the corresponding quantity for spin  $\beta$ . For diagonal elements

$$p_1(1) = p_1^{\alpha}(1) + p_1^{\beta}(1) \quad (2-10)$$

where  $p_1^{\alpha}(1) = p_1^{\alpha\alpha}(11)$  and  $p_1^{\alpha}(1) \cdot d\vec{r}_1$  is the probability of finding an electron with  $\alpha$ -spin in the volume element  $dr_1$ ;  $p_1^{\beta}(1) = p_1^{\beta\beta}(11)$  and  $p_1^{\beta}(1) \cdot d\vec{r}_1$  has similar meaning.

The diagonal elements of the spinless two-particle

density matrix are obtained as

$$P_2(12) = P_2^{\alpha\alpha}(12) + P_2^{\alpha\beta}(12) + P_2^{\beta\alpha}(12) + P_2^{\beta\beta}(12) \quad (2-11)$$

where

$$P_2^{\alpha\alpha}(12) = p_1^\alpha(1)p_1^\alpha(2) - p_1^{\alpha\alpha}(12)p_1^{\alpha\alpha}(21) \quad (A) (2-12)$$

$$P_2^{\beta\beta}(12) = p_1^\beta(1)p_1^\beta(2) - p_1^{\beta\beta}(12)p_1^{\beta\beta}(21) \quad (B) (2-12)$$

$$P_2^{\alpha\beta}(12) = p_1^\alpha(1)p_1^\beta(2) \quad (C) (2-12)$$

$$P_2^{\beta\alpha}(12) = p_1^\beta(1)p_1^\alpha(2) \quad (D) (2-12)$$

$P_2^{\alpha\alpha}(12)$  is the probability density for finding an electron with  $\alpha$ -spin at  $r_1$ , and another electron with  $\alpha$ -spin at  $r_2$ ;  $P_2^{\alpha\beta}(12)$  is the probability density for an electron with  $\alpha$ -spin at  $r_1$ , and another with  $\beta$ -spin at  $r_2$ , and so forth. The sum  $P_2(12)$  is the probability density for an electron at  $r_1$ , and another at  $r_2$  irrespective of their spins.

When  $r_1 = r_2$ , both  $P_2^{\alpha\alpha}(12)$  and  $P_2^{\beta\beta}(12)$  vanish,  $P_2^{\alpha\beta}(12)$  and  $P_2^{\beta\alpha}(21)$  do not. This is meaningful according to the Pauli principle which prohibits two electrons with parallel spin from occupying the same point in space but does not impose any restrictions on two electrons with opposite spin.

From formula (2-2), when a correlation factor is

brought in, we have

$$P_2^{\alpha\alpha}(12) = p_1^\alpha(1)p_1^\alpha(2)(1+f^{\alpha\alpha}(1,2)) \quad (\text{A}) (2-13)$$

$$P_2^{\alpha\beta}(12) = p_1^\alpha(1)p_1^\beta(2)(1+f^{\alpha\beta}(1,2)) \quad (\text{B}) (2-13)$$

From the antisymmetry of the wave function,

$$P_2^{\alpha\alpha}(12) \Big|_{r_1 \rightarrow r_2} = 0 \quad (\text{A}) (2-14)$$

$$P_2^{\beta\beta}(12) \Big|_{r_1 \rightarrow r_2} = 0 \quad (\text{B}) (2-14)$$

This means  $f^{\alpha\alpha}(1,2)$  and  $f^{\beta\beta}(1,2)$  tend to  $-1$  when  $r_1, r_2$ . That means, there is zero probability of finding any two electrons with the same spin at the same spatial point, just as the Pauli Principle requires. Correlation arising from the antisymmetry of the electronic wavefunction is usually called the "Fermi correlation".

This restriction does not act on  $P_2^{\alpha\beta}(12)$  and  $P_2^{\beta\alpha}(12)$ , but the electrostatic repulsion still keep electrons with opposite spins apart on the average. This effect is called "Coulomb correlation".

Equation (2-12) show that this single determinant wavefunction only provides correlation for the same spin

$P_2^{\alpha\alpha}(1,2)$  and  $P_2^{\beta\beta}(1,2)$ , but for electrons of opposite spin the correlation is absent. In other words, there is only Fermi correlation here, but it fails to describe Coulomb correlation.

(B). The methods with correlation factors including interparticle distance vectors.

This kind of method has been greatly developed since the work of Hylleraas [9,10]. The interelectronic variables were introduced into a two particle wave function and have given solutions with high accuracy in many cases [11-15].

Boys [16] proposed a correlated wave function with a correlation factor as

$$\Psi = C \Psi_0 \quad (2-15)$$

where  $\Psi_0$  may be a single Slater determinant, the factor C is a symmetric function containing the interelectronic variables

$$C = \prod_{i < j} F(i,j) \quad (2-16)$$

Boys and Handy [17,18] suggested a modified variational procedure

$$\langle \delta \Psi^+ | (H-E) | \Psi \rangle = 0 \quad (2-17)$$

where

$$\Psi^+ = C^{-1}\Psi_0 \quad (2-18)$$

The equation (2-17) vanishes only when  $H\Psi = E\Psi$ , and it is not sensitive to the specific correlation factor used in an arbitrary variation. The functional

$$\langle \Psi_0 | C^{-1} H C | \Psi_0 \rangle \quad (2-19)$$

can be regarded as the expectation value of a "transcorrelated Hamiltonian", and is easier to handle than  $\langle \Psi | H | \Psi \rangle$ .

Under suitable forms of  $F(i,j)$ , very accurate wave functions and highly accurate values of energy have been obtained for the Ne atom [19] and the LiH molecule [20]. Many difficulties may arise when the method is applied to larger systems [7]. The method of Colle and Salvetti, discussed in Chapter 3, is inspired by wavefunctions of the type discussed above.

#### (C).The CI methods

CI methods have also been highly developed [21,22]. In quantum theory, to expand an N-electron antisymmetric wave function one needs a complete set of Slater determinants containing N spin orbitals, which are selected from a complete set. The N-electron antisymmetrizer "A" is defined as [23]

$$A = -\frac{1}{N!} \sum_P (-1)^P \hat{P} \quad (2-20)$$

From the antisymmetry of the N-body wave function  $\Psi$ , we have

$$\Psi = A \Psi \quad (2-21)$$

An infinite CI expansion is obtained as

$$\Psi(1, \dots, N) = \sum_K^M C_K \Phi_K \quad (2-22)$$

where  $M \rightarrow \infty$ , K represents  $k_1, k_2, \dots, k_M$ ,  $k_1 < k_2 < \dots < k_M$ , and

$$C_K = (N!)^{-\frac{1}{2}} c_{k_1 \dots k_M} \quad (2-23)$$

$$\Phi_K = (N!)^{-\frac{1}{2}} \det(\sigma_{k_1}(1) \dots \sigma_{k_M}(N)) \quad (2-24)$$

Each of the above Slater determinants correspond to a single electron configuration.  $\Phi_K$  may represent an orthonormal set of occupied orbitals or an orthonormal set of unoccupied orbitals. Most common methods of CI include the ground state and all single and double excitations to get a limited CI expansion [24].

The most important procedure in CI obtains the

coefficients  $C_k$  by minimizing the energy with the variational method. This leads to a system of linear equations

$$\sum_j (h_{ij} - E \delta_{ij}) C_K = 0 \quad (2-25)$$

where

$$h_{ij} = \langle \phi_i^* | \hat{H} | \phi_j \rangle \quad (2-26)$$

The secular determinant is

$$| h_{ij} - E \delta_{ij} | = 0 \quad (2-27)$$

The  $C_k$  are obtained by solving these equations. Although conceptually the CI method should give an arbitrarily accurate  $N$ -electron wave function, the convergence is usually slow and the terms which are neglected, may be important.

(D). Correlated determinants from X-ray diffraction.

A correlated single determinant wave function, whose reference state is the single determinant of X-ray orbitals which delivers the exact density, includes correlation effects. Density matrices for many-electron systems which are  $N$ -representable by such wave functions are highly accurate.

The Hohenberg and Kohn (HK) theorem [25] states that the many-particle ground state is a unique functional of the

electronic density, this means that the electron density contains all the information necessary to evaluate every electronic property of the system. Fortunately, the single crystal coherent X-ray diffraction experiment gives an image of the electron density of crystalline solids. A method has been developed, that constructs the quantum mechanically valid density matrices from the data of the single crystal coherent X-ray diffraction experiment, and then calculates properties of the electrons of the system with these density matrices, which are constructed by single determinant wave functions [26-33]. Remarkably good results for properties of some test systems have been obtained by the calculations with a single determinant of X-ray orbitals [34-37].

When a beam of X-rays shine on a crystal, the radiation is coherently scattered by the electrons in the atoms of the crystal [38], we have

$$F(\vec{k}) = \int e^{i\vec{k}\vec{r}} p(\vec{r}) d\vec{r} \quad (2-28)$$

where  $\vec{k}$  is the wave vector,  $F(\vec{k})$  is the wave amplitude scattered by the electron distribution  $p(\vec{r})$  relative to the scattering by a single electron. This relationship is expressed by a Fourier transformation [39]. The square of the amplitude is proportional to the intensity of the scattered wave from the measurement of the experiment. Certainly the experimentally measured intensity has to be corrected for

other effects [40], such as X-ray absorption by the crystal, temperature effect associated with the vibration and so on.

Quantum Mechanics requires that only the densities which are obtainable from the square of an antisymmetric wave functions may be appropriate. The antisymmetric N-body wave function  $\Psi$  of (2-5) [41-43] gives an one body reduced density matrix  $p_1(11')$  according to the rule

$$p_1(11') = N \int \Psi(1, \dots, N) \Psi(1', \dots, N) d2 \dots dN \quad (2-29)$$

where the integration is over all the spatial and spin coordinates except the coordinates of one particle. The diagonal element of the first-order density matrix represents the electron density

$$p_1(11') \Big|_{1' \rightarrow 1} = p(1) \quad (2-30)$$

For a single determinant (2-5) of orthonormal orbitals, the first order density matrix may be written in a convenient matrix notation as

$$p_1(\vec{r}, \vec{r}') = \text{tr} \underline{\phi}(\vec{r}) \underline{\phi}^+(\vec{r}') \quad (2-31)$$

$$\underline{\phi}(\vec{r}) = \underline{c} \underline{\Omega}(\vec{r}) \quad (2-32)$$

where  $\underline{C}$  is a rectangular matrix. From (2-31) we have

$$p_1(\vec{r}, \vec{r}') = \text{tr } \underline{C}^+ \underline{C} \underline{\Omega}(\vec{r}) \underline{\Omega}(\vec{r}') \quad (2-33)$$

where  $\underline{C}^+ \underline{C}$  is defined as the population  $\underline{P}$

$$\underline{P} = \underline{C}^+ \underline{C} \quad (2-34)$$

The above formulas could be rewritten with the population matrix  $\underline{P}$

$$p_1(\vec{r}, \vec{r}') = \text{tr } \underline{P} \underline{\Omega}(\vec{r}) \underline{\Omega}^+(\vec{r}') \quad (2-35)$$

$$p_1(\vec{r}) = \text{tr } \underline{P} \underline{\Omega}(\vec{r}) \underline{\Omega}(\vec{r}) \quad (2-36)$$

$$F(\vec{k}) = \text{tr } \underline{P} \underline{f}(\vec{k}) \quad (2-37)$$

$$f_{mn}(\vec{k}) = \int e^{i \vec{k} \cdot \vec{r}} \Omega_m(\vec{r}) \Omega_n^*(\vec{r}) d\vec{r} \quad (2-38)$$

$\underline{P}$  in the equation (2-35) can treat as parameter to be fit to the experimental structure factors. Certainly any fitting procedure should include appropriate Quantum Mechanical restrictions.

For  $N$ -representability of the one-body density matrix one has

$$\int p_1(r, r') \phi_i(r') dr' = \dot{E}_i \phi_i \quad (2-39)$$

where eigenvalue  $\dot{E}_i$  is a number between zero to one,  $0 \leq \dot{E}_i \leq 1$ . The physical interpretation of  $\dot{E}_i$  is the occupation number of a molecular orbital. In particular, the  $E_i$  values 0 and 1 are necessary and sufficient for  $p_1(r, r')$  to be  $N$ -representable by a single determinant of orbitals of the type of equation (2-32) [44].

Furthermore, Gilbert's theorem [45] states that any non-negative density which is normalized to the number of electrons  $N$ , is  $N$ -representable by a single-determinant wavefunction. Harriman [46] has given a procedure to construct such a determinant from the density, which has single determinant  $N$ -representability for  $p_1(\vec{r}, \vec{r}')$ .

In the single determinant case [47] one has that

$$\int p_1(\vec{r}', \vec{r}'') p_1(\vec{r}'', \vec{r}) d\vec{r}'' = p_1(\vec{r}', \vec{r}) \quad (2-40)$$

The hermiticity and the normalization are respectively

$$p_1(\vec{r}', \vec{r}) = p_1^*(\vec{r}, \vec{r}') \quad (2-41)$$

$$\int p_1(\vec{r}, \vec{r}) d\vec{r} = N \quad (2-42)$$

For these conditions to be satisfied by a single determinant of orthonormal orbitals, the population matrix  $\underline{p}$  is a symmetric normalized projector, and satisfies the experimental structure factors, that is

$$\underline{p} = \underline{p}^+ \quad (A) (2-43)$$

$$\text{tr } \underline{p} = N \quad (B) (2-43)$$

$$\underline{p}^2 = \underline{p} \quad (C) (2-43)$$

$$\text{tr } \underline{p} f(\vec{k}) = F(\vec{k}) \quad (D) (2-43)$$

The  $\underline{p}$  matrix if it is a solution of Clinton's equations (2-44) [48], will satisfy the conditions (2-43)

$$\underline{p}_{n+1} = 3 \underline{p}_n^2 - 2 \underline{p}_n^3 + \lambda_K f(k) + \lambda_N \underline{1} \quad (2-44)$$

where  $\lambda_K$  and  $\lambda_N$  are Lagrangian multipliers determined by constraints satisfying structure factors and normalization.

Frishberg and Massa have used the structure factors of Stewart, Davidson and Simpson [49] for the spherical hydrogen atom in a hydrogen molecule as constraints in the determination of the idempotent  $\underline{p}$  matrix. They showed

Clinton's equations are solvable and yield physically meaningful orbitals.

Frishberg and Massa also investigated the application of the X-ray orbital formalism to bonding effects of  $H_2$  [35]. Their results reproduced the known charge distribution and bond distance of  $H_2$ . The structure factors of Li and Be were calculated with X-ray orbitals and HF orbitals in the same basis. In a fit to the exact structure factors, X-ray orbitals gave good quantum properties. HF and X-ray wavefunctions in the same basis were used to calculate a number of properties of Li and Be atoms. The X-ray case gave excellent results (Table 1-1) [34].

One may conclude a single determinant of X-ray orbitals will provide an excellent reference state for correlated determinant wave functions to be used in calculation of correlation energy.

TABLE 2-1 A COMPARISON OF EXACT, HF AND X-RAY  
 ORBITALS SINGLE-DETERMINANT ON SAME BASIS BY  
 SOME EXPECTATION VALUES OF Li AND Be ATOMS [34]

ATOM	$\langle \frac{1}{r} \rangle$	$\langle r^2 \rangle$	$\langle \frac{1}{r_{12}} \rangle$	Kinetic energy	Total energy
<b>Li</b>					
Exact	5.71822	18.35034	2.199	7.478	-7.478025
HF	5.71549	18.62610	2.281	7.433	-7.432749
X-ray	5.71866	18.34609	2.285	7.438	-7.432694
<b>Be</b>					
Exact	8.4246		4.3803	14.66	-14.66090
HF	8.4088	17.32008	4.4891	14.57	-14.57302
X-ray	8.4255	16.32010	4.5376	14.59	-14.57150

CHAPTER III COLLE-SALVETTI FORMALISM  
FOR THE CORRELATION ENERGY

(A). THE COLLE-SALVETTI FORMULA (C-S Formula)

Colle and Salvetti gave a formalism for approximate calculation of the correlation energy  $E_c$  of a closed-shell system, by starting from knowledge of the Hartree-Fock determinant [50-55].

They introduced an approximate N-body wavefunction

$$\Psi(1, \dots, N) = \Psi^{\text{HF}}(1, \dots, N) \cdot C(r_1, r_2, \dots, r_N) \quad (3-1)$$

where  $\Psi^{\text{HF}}(1, 2, \dots, N)$  is the known HF wave function with spin,  $C$  is the correlation factor which is spinless, and is taken to be of the form [50]

$$C = \prod_{i < j} [ 1 - \phi(i, j) ] \quad (3-2)$$

where  $\phi(i, j)$  is a symmetric two-body function which is spinless, and brings in the correlation effects. Colle and Salvetti gave  $\phi(i, j)$  a specific form, viz

$$\phi(i, j) = e^{-\beta^2 r^2} \left( 1 - \epsilon(R) \left( 1 + \frac{r}{2} \right) \right) \quad (3-3)$$

where  $\epsilon(R)$  is an unknown functional to be determined and

$$r = |\vec{r}_i - \vec{r}_j| \quad R = -\frac{1}{2} |\vec{r}_i + \vec{r}_j| \quad (3-4)$$

Equation (3-3) satisfies the appropriate cusp conditions when  $\vec{r}_i - \vec{r}_j = 0$  as has been demonstrated by Bingle [56]. When the electrons are far apart, the function  $\Psi$  is reduced to that of HF; when a pair of electrons collide, the function becomes

$$\Psi = \Psi_{\text{HF}} \epsilon^{\left(-\frac{1}{2} \frac{\vec{r}_i + \vec{r}_j}{r}\right)} \quad (3-5)$$

The region in which the function  $\phi(i,j)$  is appreciably different from zero depends on the value of  $\beta$ . The volume of this region may be taken to be

$$V = 4\pi \int_0^\infty e^{-\beta^2 r^2} r^2 dr = \left(\frac{\sqrt{\pi}}{\beta}\right)^3 \quad (3-6)$$

Colle and Salvetti assume that  $V$  is proportional to the volume of exclusion in Wigner's formula [57] with a proportionality constant  $k$ ,

$$V = k \frac{1}{p(R)} \quad (3-7)$$

so that

$$\beta = \sqrt{\pi} \left( -\frac{p(R)}{k} \right)^{\frac{1}{3}} = q \cdot p^{\frac{1}{3}}(R) \quad (3-8)$$

where  $q$  is constant, and from (3-8) we can see that

$$q = \sqrt{\pi} \left( -\frac{1}{k} \right)^{\frac{1}{3}} \quad (3-9)$$

Colle and Salvetti empirically put  $q=2.29$  [50]. It is clear that  $\beta$  depends on the local density.

The second order density matrix for a two-electron system without spin may be written as [5]

$$P_2(121'2') = N(N-1) \cdot \int \Psi^{HF}(1, \dots, N) \prod_{i < j} [1 - \phi(ij)] [1 - \phi(i'j')] \Psi^{* HF}(1', 2', 3, \dots, N) dr_3 \dots dr_N ds_1 \dots ds_N \quad (3-10)$$

where  $r_m$  and  $s_m$  represent the spatial and the spin coordinates of particles respectively. Because  $\phi(12) \ll 1$  because of its exponential factor, we get approximately

$$P_2(121'2') = P_2^{HF}(121'2') (1 - \phi(12) - \phi(1'2') + \phi(12)\phi(1'2')) \quad (3-11)$$

An assumption is made that

$$p_1(11') = p_1^{HF}(11') \quad (3-12)$$

where  $p_1$  represents the first order density matrix. The

formula (3-12) means the first-order density matrix of HF approximates well enough the exact first-order density matrix. From the N-representability of the reduced density matrices, we have

$$p_1(11') = \int P_2(121'2) d2 \quad (\text{A})(3-13)$$

$$p_1^{\text{HF}}(11') = \int P_2^{\text{HF}}(121'2) d2 \quad (\text{B})(3-13)$$

it follows

$$\int P_2(121'2) d2 = \int P_2^{\text{HF}}(121'2) d2 \quad (3-14)$$

Combining this with (3-11) we have

$$\int P_2^{\text{HF}}(121'2') \cdot b(121'2') d2 = 0 \quad (3-15)$$

where

$$b(121'2') = -\phi(12) - \phi(1'2') + \phi(12)\phi(1'2') \quad (3-16)$$

Colle and Salvetti approximately solve the  $\epsilon(R)$  from the equation (3-15) as

$$\epsilon(R) \approx \frac{\sqrt{\pi} \cdot B}{1 + \sqrt{\pi} \cdot B} \quad (3-17)$$

Colle and Salvetti calculate  $E_c$  according to

$$E_c = E - E^{HF} = \frac{1}{2} \int P_2(12) (\phi'(12) - 2\phi(12)) \frac{1}{r} d1 d2 \quad (3-18)$$

This is the C-S theoretical formula.

Colle and Salvetti made a further simplification for the calculation of  $E_c$ . They proposed an empirical formula involving a Taylor expansion of  $P_2$ , so that the numerical integration only contains one variable  $R$ . The C-S numerical formula is

$$E_c = -0.04918 \int p(R) \frac{1 + 0.173 W e^{(-0.58/B)}}{1 + 0.8/B} d\vec{R} \quad (3-19)$$

with

$$B = 2.29 p^{\frac{1}{3}}(R) \quad (3-20)$$

$$W = 0.3814 p^{\frac{-8}{3}}(R) \left[ \nabla^2 P_2(\vec{R} - \frac{\vec{r}}{2}, \vec{R} + \frac{\vec{r}}{2}) \right]_{r=0} \quad (3-21)$$

where  $\nabla^2$  will operate on  $r$  and is defined as

$$\nabla^2 = \frac{\partial^2}{\partial x^2} + \frac{\partial^2}{\partial y^2} + \frac{\partial^2}{\partial z^2} \quad (A) (3-22)$$

where

$$r^2 = x^2 + y^2 + z^2 \quad (B) (3-23)$$

Colle and Salvetti listed the results of a number of ground state systems treated by the formula (3-19), which compared to the experimental values very well; the highest error reported is eight per cent of the correlation energy [50]. (Table 3.1)

The Colle and Salvetti ideas have been extended to calculate the excitation energy, ionization potentials and dissociation energies for some atoms and small molecules (Table 3.2) [58-60].

TABLE 3.1 THE CORRELATION ENERGY FOR SOME  
 ATOMS AND SMALL MOLECULES calculated by Colle  
 & Salvetti with comparing one by Clementi [50]

TRIAL	Ec (a.u.) COLLE-SALVETTI	Ec (a.u.) CLEMENTI	Error   %
He	-0.0416	-0.0420	1.0
Li <sup>+</sup>	-0.0438	-0.0435	0.7
Be <sup>2+</sup>	-0.0441	-0.0443	0.4
Be	-0.0926	-0.0940	1.5
B <sup>+</sup>	-0.106	-0.1123	5.4
Ne	-0.374	-0.372	0.5
CH <sub>4</sub>	-0.289	-0.283	2.1
H <sub>2</sub> O	-0.336	-0.364	7.7

TABLE 3.2 THE EXCITATION ENERGY, IONIZATION POTENTIAL,  
DISSOCIATION ENERGY CALCULATED BY COLLE & SALVETTI  
WITH THEIR NUMERICAL FORMULA [58] AND ADD TO THE HF  
OR THE MC-SCF VALUES COMPARED TO THE EXACT VALUES [3]

SYSTEM			$\Delta E$ (ev) COLLE-SALVETTI	$\Delta E$ (ev) EXACT
Be	$^1S$	(2s $\rightarrow$ 3s)	6.7	6.8
	$^1S$	(2s $\rightarrow$ 4s)	8.1	8.1
C <sub>2</sub> H <sub>4</sub>	$^1B_1$	(1b <sub>3</sub> $\rightarrow$ 1b <sub>2</sub> )	7.8	7.66
CH <sub>2</sub> O	$^1B_2$	(2b <sub>2</sub> $\rightarrow$ 6a <sub>1</sub> )	7.14	7.10
	$^1B_2$	(2b <sub>2</sub> $\rightarrow$ 7a <sub>1</sub> )	8.02	8.14
	$^1B_2$	(2b <sub>2</sub> $\rightarrow$ 8a <sub>1</sub> )	8.96	9.24
	$^1B_2$	(2b <sub>2</sub> $\rightarrow$ 9a <sub>1</sub> )	9.30	9.62
	$^2B_3$	Ionization Potential	10.65	10.88
H <sub>2</sub>	X $^1\Sigma^+$	Dissociation Energy	4.72	4.75

(B). Calculation of the correlation energy for three series of Isoelectronic Ions

The Colle-Salvetti numerical formula (3-19) and their functional (3-17) are tested by calculating the correlation energy  $E_c$  of He, Be, Ne atoms, as well as the corresponding isoelectronic series with a single determinant of the HF orbitals [61]. The details of the formulas used in the calculation are shown in Appendix 2.

Also the calculations of  $E_c$  are performed with the formula (3-18), by using our program of 3 Dimensional Simpson Numerical Integration (3D SI) and same HF orbitals. The correlation function  $\phi(i,j)$  and the  $\epsilon(R)$  were taken from the formula (3-3) and (3-17) which are used by Colle-Salvetti.

The results are compared with ones of Ab initio calculations from Clementi [3] as a reference, which are considered to be accurate correlation energies. These are shown in Table 3.3 to 3.5 with Figure 2.1 to 2.3 respectively. We see that the results of only some trials from the Colle-Salvetti numerical formula are close to those from Clementi. By increasing the nuclear charge  $Z$  the results go far from the Clementi results.

TABLE 3.3 THE CORRELATION ENERGY OF He ISOELECTRONIC  
 SERIES CALCULATED BY 3D SI WITH FORMULA  
 $\frac{1}{2}\langle P_2 b/r \rangle$  AND COMPARED ONE FROM C-S  
 NUMERICAL FORMULA AND ONE FROM CLEMENTI

Z	SYSTEM	C-S NUMERICAL (a.u.)	CLEMENTI (a.u.)	FORMULA $\frac{1}{2}\langle P_2 b/r \rangle$ BY 3D SI (a.u.)
Z=2	He	-0.0420	-0.0421	-0.0423
Z=3	Li <sup>+1</sup>	-0.0441	-0.0435	-0.0491
Z=4	Be <sup>+2</sup>	-0.0442	-0.0443	-0.0531
Z=5	B <sup>+3</sup>	-0.0438	-0.0448	-0.0557
Z=6	C <sup>+4</sup>	-0.0432	-0.0451	-0.0575
Z=7	N <sup>+5</sup>	-0.0426	-0.0453	-0.0590
Z=8	O <sup>+6</sup>	-0.0419	-0.0455	-0.0600
Z=9	F <sup>+7</sup>	-0.0413	-0.0456	-0.0608
Z=10	Ne <sup>+8</sup>	-0.0406	-0.0457	-0.0615
Z=11	Na <sup>+9</sup>	-0.0400	-0.0458	-0.0621
Z=12	Mg <sup>+10</sup>	-0.0394	-0.0459	-0.0626
Z=13	Al <sup>+11</sup>	-0.0389	-0.0459	-0.0630
Z=14	Si <sup>+12</sup>	-0.0384	-0.0460	-0.0634
Z=15	P <sup>+13</sup>	-0.0379	-0.0461	-0.0637
Z=16	S <sup>+14</sup>	-0.0375	-0.0461	-0.0639
Z=17	Cl <sup>+15</sup>	-0.0370	-0.0462	-0.0641
Z=18	Ar <sup>+16</sup>	-0.0367	-0.0463	-0.0643
Z=19	K <sup>+17</sup>	-0.0363	-0.0463	-0.0645
Z=20	Ca <sup>+18</sup>	-0.0359	-0.0463	-0.0647
Z=21	Sc <sup>+19</sup>	-0.0356	-0.046	-0.0648
Z=22	Ti <sup>+20</sup>	-0.0353	-0.046	-0.0650
Z=23	V <sup>+21</sup>	-0.0350	-0.047	-0.0651
Z=24	Cr <sup>+22</sup>	-0.0347	-0.047	-0.0652
Z=25	Mn <sup>+23</sup>	-0.0344	-0.047	-0.0653
Z=26	Fe <sup>+24</sup>	-0.0341	-0.047	-0.0654
Z=27	Co <sup>+25</sup>	-0.0339	-0.047	-0.0655
Z=28	Ni <sup>+26</sup>	-0.0336	-0.047	-0.0656
Z=29	Cu <sup>+27</sup>	-0.0334	-0.047	-0.0657
Z=30	Zn <sup>+28</sup>	-0.0332	-0.047	-0.0658
Z=31	Ga <sup>+29</sup>	-0.0330	-0.047	-0.0659
Z=32	Ge <sup>+30</sup>	-0.0327	-0.047	-0.0659
Z=33	As <sup>+31</sup>	-0.0326	-0.047	-0.0659
Z=34	Se <sup>+32</sup>	-0.0324	-0.047	-0.0660
Z=35	Br <sup>+33</sup>	-0.0322	-0.047	-0.0661
Z=36	Kr <sup>+34</sup>	-0.0320	-0.047	-0.0661

Figure 2.1 A COMPARISON OF  $E_c$

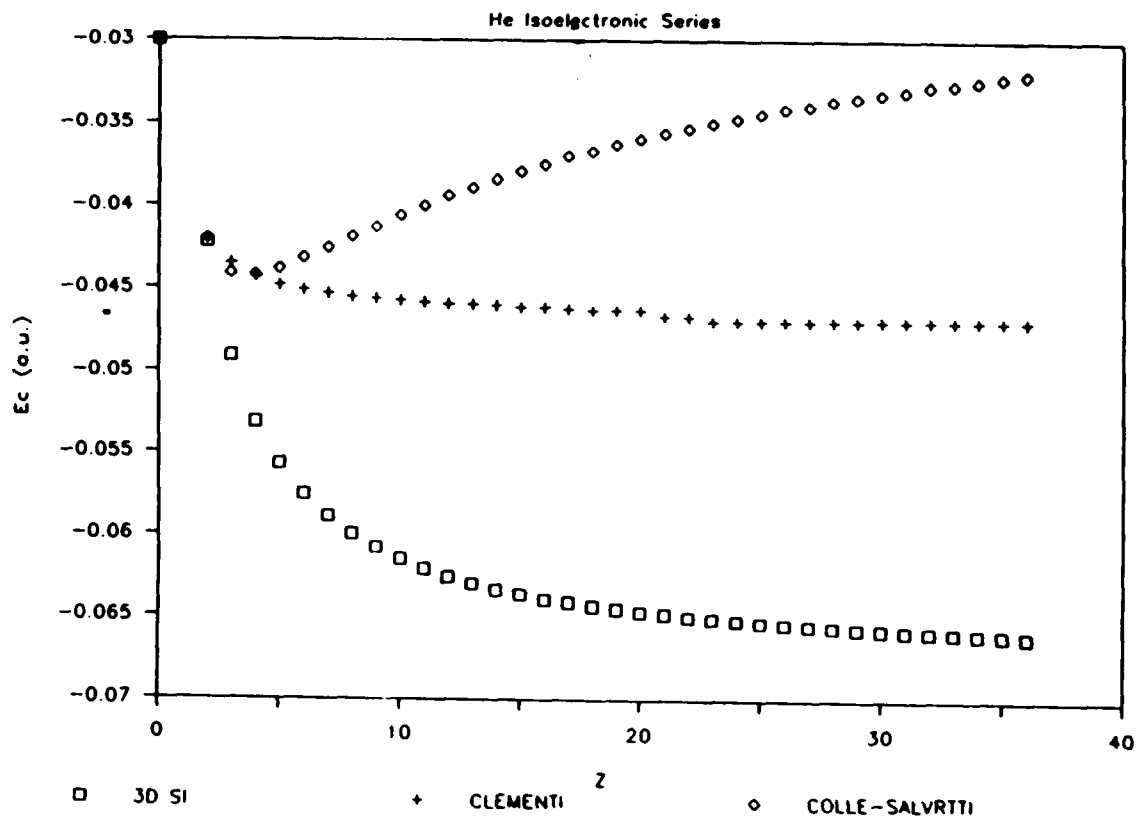


TABLE 3.4 THE CORRELATION ENERGY OF Be ISOELECTRONIC  
 SERIES CALCULATED BY 3D SI WITH FORMULA  
 $\frac{1}{2}\langle P_2b/r \rangle$  AND COMPARED ONE FROM C-S  
 NUMERICAL FORMULA AND ONE FROM CLEMENTI

Z	SYSTEM	C-S NUMERICAL (a.u.)	CLEMENTI (a.u.)	FORMULA $\frac{1}{2}\langle P_2b/r \rangle$ BY 3D SI (a.u.)
Z=4	Be	-0.0932	-0.0944	-0.0585
Z=5	B <sup>+1</sup>	-0.1060	-0.1123	-0.0700
Z=6	C <sup>+2</sup>	-0.1163	-0.1268	-0.0771
Z=7	N <sup>+3</sup>	-0.1270	-0.1412	-0.0808
Z=8	O <sup>+4</sup>	-0.1281	-0.1551	-0.0846
Z=9	F <sup>+5</sup>	-0.1312	-0.1684	-0.0891
Z=10	Ne <sup>+6</sup>	-0.1332	-0.1814	-0.0936
Z=11	Na <sup>+7</sup>	-0.1348	-0.1941	-0.0969
Z=12	Mg <sup>+8</sup>	-0.1360	-0.2066	-0.0995
Z=13	Al <sup>+9</sup>	-0.1372	-0.2190	-0.1014
Z=14	Si <sup>+10</sup>	-0.1380	-0.2313	-0.1029
Z=15	P <sup>+11</sup>	-0.1392	-0.2435	-0.1042
Z=16	S <sup>+12</sup>	-0.1402	-0.2556	-0.1053
Z=17	Cl <sup>+13</sup>	-0.1410	-0.2677	-0.1062
Z=18	Ar <sup>+14</sup>	-0.1417	-0.2797	-0.1070
Z=19	K <sup>+15</sup>	-0.1406	-0.2917	-0.1078
Z=20	Ca <sup>+16</sup>	-0.1425	-0.3037	-0.1086
Z=21	Sc <sup>+17</sup>	-0.1428	-0.316	-0.1092
Z=22	Ti <sup>+18</sup>	-0.1431	-0.327	-0.1098
Z=23	V <sup>+19</sup>	-0.1433	-0.339	-0.1104
Z=24	Cr <sup>+20</sup>	-0.1435	-0.351	-0.1109
Z=25	Mn <sup>+21</sup>	-0.1437	-0.363	-0.1114
Z=26	Fe <sup>+22</sup>	-0.1438	-0.375	-0.1118
Z=27	Co <sup>+23</sup>	-0.1439	-0.387	-0.1122
Z=28	Ni <sup>+24</sup>	-0.1440	-0.398	-0.1126
Z=29	Cu <sup>+25</sup>	-0.1440	-0.411	-0.1130
Z=30	Zn <sup>+26</sup>	-0.1441	-0.423	-0.1133
Z=31	Ga <sup>+27</sup>	-0.1441	-0.434	-0.1136
Z=32	Ge <sup>+28</sup>	-0.1441	-0.446	-0.1149
Z=33	As <sup>+29</sup>	-0.1441	-0.458	-0.1142
Z=34	Se <sup>+30</sup>	-0.1441	-0.470	-0.1144
Z=35	Br <sup>+31</sup>	-0.1440	-0.482	-0.1147
Z=36	Kr <sup>+32</sup>	-0.1440	-0.494	-0.1149

Figure 2.2 A COMPARISON OF  $E_c$

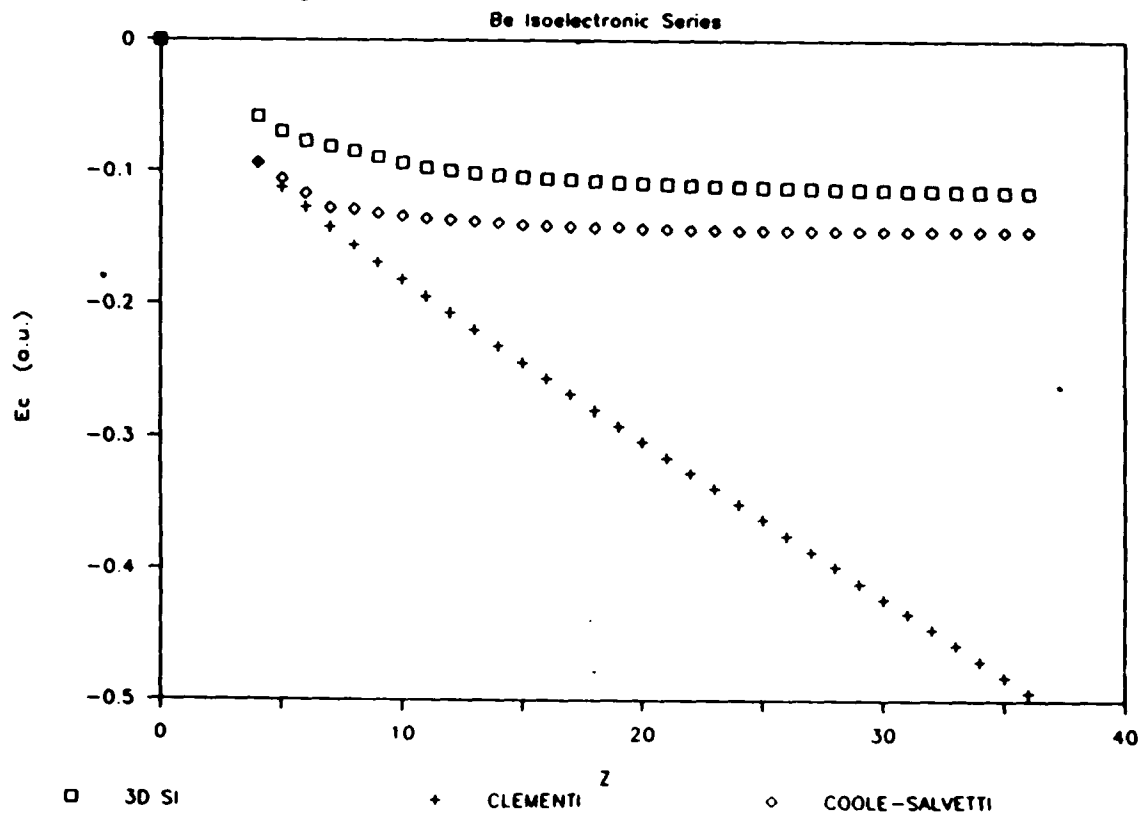
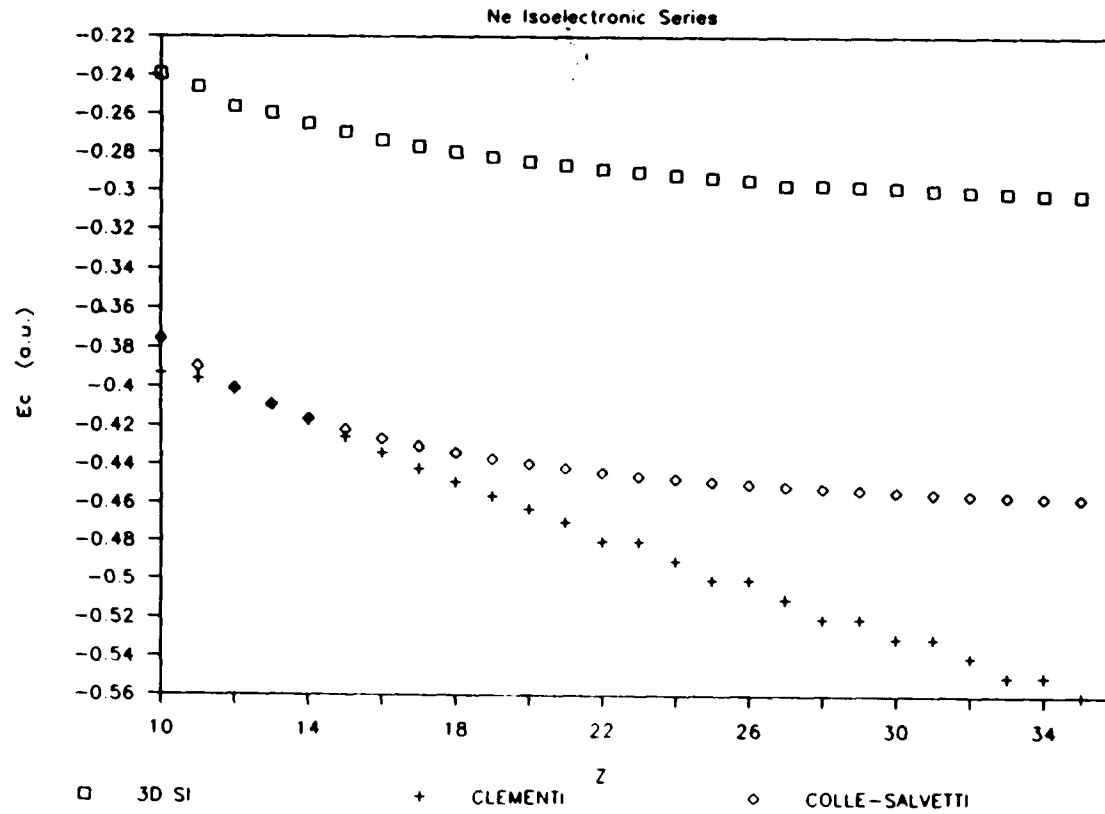


TABLE 3.5 THE CORRELATION ENERGY OF Ne ISOELECTRONIC  
 SERIES CALCULATED BY 3D SI WITH FORMULA  
 $\frac{1}{2}\langle P_2b/r \rangle$  AND COMPARED ONE FROM C-S  
 NUMERICAL FORMULA AND ONE FROM CLEMENTI

Z	SYSTEM	C-S NUMERICAL (a.u.)	CLEMENTI (a.u.)	FORMULA $\frac{1}{2}\langle P_2b/r \rangle$ BY 3D SI (a.u.)
Z=10	Ne	-0.375	-0.393	-0.0423
Z=11	Na <sup>+1</sup>	-0.390	-0.396	-0.0491
Z=12	Mg <sup>+2</sup>	-0.401	-0.402	-0.0531
Z=13	Al <sup>+3</sup>	-0.409	-0.409	-0.0557
Z=14	Si <sup>+4</sup>	-0.416	-0.417	-0.0575
Z=15	P <sup>+5</sup>	-0.422	-0.426	-0.0590
Z=16	S <sup>+6</sup>	-0.427	-0.434	-0.0600
Z=17	Cl <sup>+7</sup>	-0.431	-0.442	-0.0608
Z=18	Ar <sup>+8</sup>	-0.434	-0.449	-0.0615
Z=19	K <sup>+9</sup>	-0.437	-0.456	-0.0621
Z=20	Ca <sup>+10</sup>	-0.440	-0.463	-0.0626
Z=21	Sc <sup>+11</sup>	-0.442	-0.47	-0.0630
Z=22	Ti <sup>+12</sup>	-0.444	-0.48	-0.0634
Z=23	V <sup>+13</sup>	-0.446	-0.48	-0.0637
Z=24	Cr <sup>+14</sup>	-0.447	-0.49	-0.0639
Z=25	Mn <sup>+15</sup>	-0.449	-0.50	-0.0641
Z=26	Fe <sup>+16</sup>	-0.450	-0.50	-0.0643
Z=27	Co <sup>+17</sup>	-0.451	-0.51	-0.0645
Z=28	Ni <sup>+18</sup>	-0.452	-0.52	-0.0647
Z=29	Cu <sup>+19</sup>	-0.453	-0.52	-0.0648
Z=30	Zn <sup>+20</sup>	-0.454	-0.53	-0.0650
Z=31	Ga <sup>+21</sup>	-0.455	-0.53	-0.0651
Z=32	Ge <sup>+22</sup>	-0.456	-0.54	-0.0652
Z=33	As <sup>+23</sup>	-0.456	-0.55	-0.0653
Z=34	Se <sup>+24</sup>	-0.457	-0.55	-0.0654
Z=35	Br <sup>+25</sup>	-0.457	-0.56	-0.0655
Z=36	Kr <sup>+26</sup>	-0.458	-0.56	-0.0656

Figure 2.3 A COMPARISON OF  $E_c$



(C). REMARKS ON THE COLLE-SALVETTI FORMALISM

Three groups of the results of the correlation energy are compared in Table 3.3, 3.4, 3.5 and Figure 2.1, 2.2, 2.3. The first is from our calculations with the Colle-Salvetti numerical formula (3-19); the second group is calculated by Clementi which is a standard reference; and the last group is from the calculation with the formula (3-18) and our 3 dimensional Simpson integration (3D SI). Most results are far away from the Clementi results, which are considered to be nearly exact. It is clear that the Colle-Salvetti numerical formalism (3-19) introduced several constants which depend on an empirical fitting procedure, does give some results close to the Clementi results.

Considerations of with the Colle and Salvetti formalism are given below:

<1> As a consequence of their approximations, the density matrices derived from correlated wavefunction are not  $N$ -representable. Cohen, Santhanam and Frishberg [62] have argued that when a second-order density matrix is approximated by a  $N$ -representable part plus a non  $N$ -representable component, the resultant density matrix might still be reasonable if the non  $N$ -representable part is small. In Colle-Salvetti two-body density matrix, only the Hartree-Fock density matrix is  $N$ -representable, the non  $N$ -representability is introduced by the approximations of adding a correlation part as a small contribution.

There has been a study of the Colle-Salvetti formula substituting for the HF determinant a determinant of X-ray orbitals, which delivers the exact density in the article of Cohen, Frishberg, Lee and Massa [63]. The Be atom was a trial for a test calculation with Colle-Salvetti numerical formula (3-19) by Cohen et al. The result  $E_c = -0.093618$  a.u., is very close to the result from Veillard and Clementi by less than 2% [4].

<2> The two body Hamiltonian usually has a form

$$H = \sum h_i + \sum h_{ij} \quad (3-25)$$

where

$$h_i = -\frac{1}{2} \nabla_i^2 - \frac{Z}{r_i} \quad (A) (3-26)$$

$$h_{ij} = -\frac{1}{r_{ij}} \quad (B) (3-26)$$

Then, the expectation value of H for the two-body case is

$$E = \int h_1 p(11') \Big|_{1' \rightarrow 1} d1 + \frac{1}{2} \int h_{12} P_2(12) d1 d2 \quad (3-27)$$

The terms of the usual Hamiltonian in general should all contribute to the total correlation energy  $E_c$ . But in the Colle-Salvetti formalism only the term with  $h_{ij}$  contributes. They assume Hartree-Fock one-body density matrix to be equal

to the exact one-body density matrix as in equation (3-12). This assumption is consistent with

$$I = \int P_2(12) (\phi'(12) - 2\phi(12)) d1 d2 = 0 \quad (3-28)$$

In the C-S approximation one may neglect the contribution from some terms in the kinetic and potential energy to  $E_c$ . For example, the correlation from the electron-nuclear attraction has been neglected

$$\begin{aligned} V_c(2) &= -Z \int b(12) \frac{-1}{r_1} P_2(12) d1 d2 \\ &= \int \frac{-Z}{r_1} d1 \int b(12) \cdot P_2(12) d2 = 0 \end{aligned} \quad (3-29)$$

Although neglected in the Colle-Salvetti case is in general not zero.

<3> The functional  $\epsilon(R)$  (3-17) of C-S formalism does not satisfy the original equation (3-15), so that the "I" values in (3-28) and  $V_c(2)$  in (3-29) can be different from zero. Calculations of I values and  $V_c(2)$  with the functional  $\epsilon(R)$  of (3-17) are performed by 3D SI program. The results from different q values are listed on the Table 3.6 and the Table 3.7. When  $q=2.29$  which is the value taken by Colle-Salvetti, for the He atom  $I = -0.0242$  a.u. and  $V_c(2) = 0.0621$  a.u., the exact  $E_c$  is  $-0.042$  a.u. for He atom. Notice that the

magnitude of  $I$  is large compared to the value of  $E_c$ .

<4> The normalization of the density is exact if (3-15) holds exactly. The normalization of the density is then given by

$$\int p_1(11') \Big|_{1' \rightarrow 1} d1 = \int p_1^{HF}(11') \Big|_{1' \rightarrow 1} d1 = N \quad (3-30)$$

TABLE 3.6  $\langle b \cdot P_2 \rangle$  OF He ATOM CALCULATED BY  
3D SI WITH THE C-S FUNCTIONAL

$$\epsilon(R) = \frac{\sqrt{\pi} \cdot B}{1 + \sqrt{\pi} B}$$

---

q	$\langle b \cdot P_2 \rangle$	q	$\langle b \cdot P_2 \rangle$
0.69	-0.6880	1.69	-0.0758
0.79	-0.5515	1.79	-0.0619
0.89	-0.4405	1.89	-0.0508
0.99	-0.3512	1.99	-0.0418
1.09	-0.2800	2.09	-0.0347
1.19	-0.2235	2.19	-0.0289
1.29	-0.1788	2.29	-0.0242
1.39	-0.1434	2.39	-0.0203
1.49	-0.1155	2.49	-0.0172
1.59	-0.0933		

---

Figure 2.4 A CALCULATION  $\langle b^*P^2 \rangle$  OF He

BY 3D SI WITH COLLE-SALVETTI FUNCTIONAL

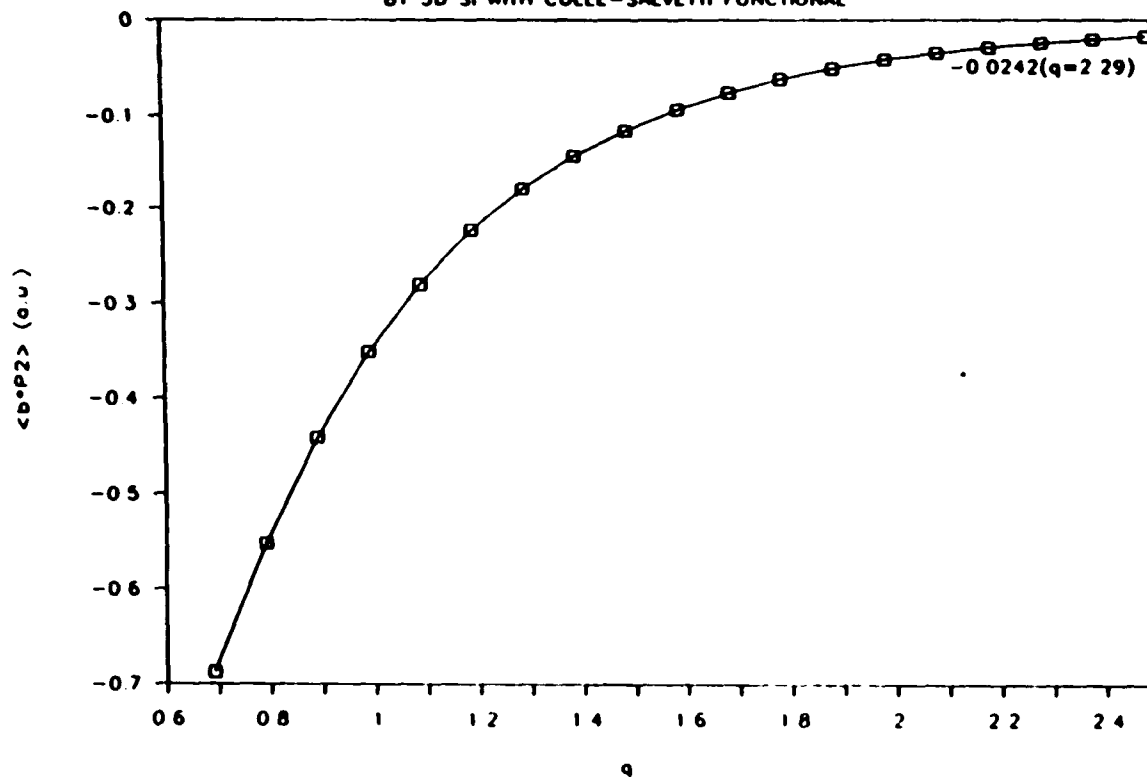


TABLE 3.7  $-Z \langle b \cdot P_2/r_1 \rangle$  OF He ATOM CALCULATED  
BY 3D SI WITH COLLE-SALVETTI FUNCTIONAL

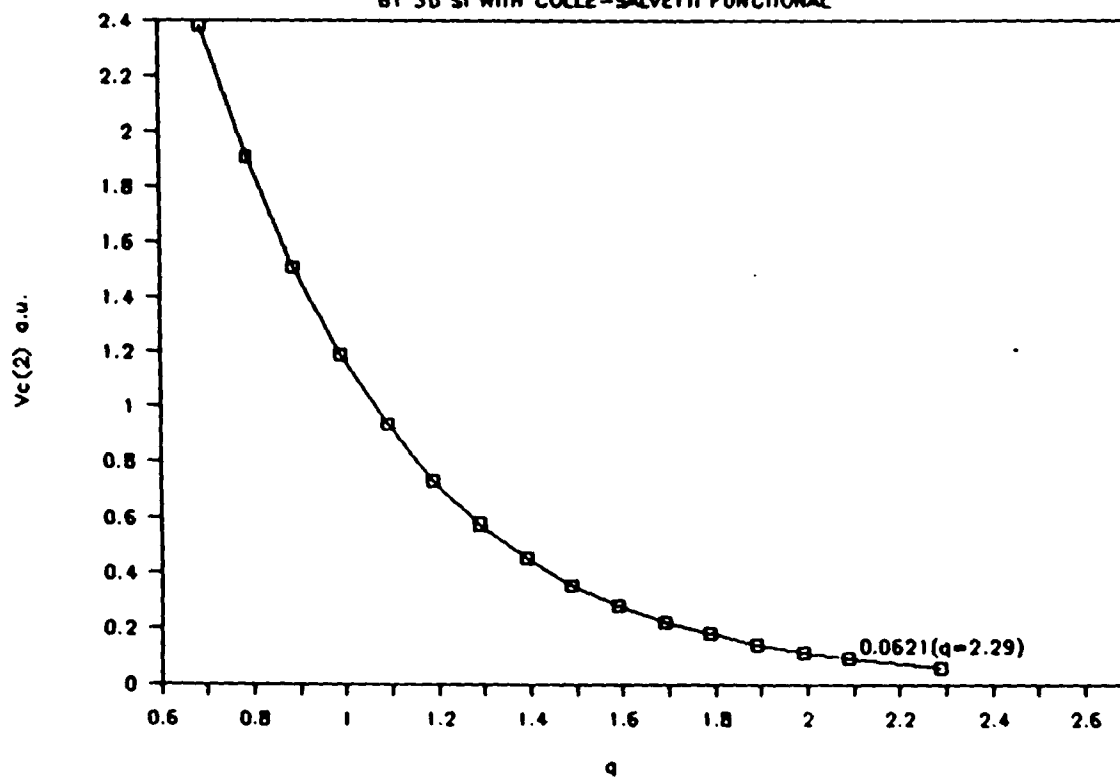
$$\epsilon(R) = \frac{\sqrt{\pi} \cdot B}{1 + \sqrt{\pi} B}$$

---

q	$-Z \langle b \cdot P_2/r_1 \rangle$	q	$-Z \langle b \cdot P_2/r_1 \rangle$
0.69	2.3816	0.79	1.9018
0.89	1.5076	0.99	1.1889
1.09	0.9342	1.19	0.7327
1.29	0.5744	1.39	0.4507
1.49	0.3545	1.59	0.2797
1.69	0.2217	1.79	0.1766
1.89	0.1414	1.99	0.1140
2.09	0.0925	2.29	0.0621

---

Figure 2.5 A CALCULATION  $V_c(2)$  OF He  
BY 30 SI WITH COLLE-SALVETTI FUNCTIONAL



## CHAPTER IV A FORMALISM FOR CORRELATION ENERGY

In this chapter, a formalism for correlation energy [5] is considered with:

<1> Highly accurate approximately N-representable density matrices from a single determinant wave function, which is constructed with HF orbitals.

<2> Correlation contributions from both kinetic and potential energy.

<3> A functional  $\epsilon(R)$  which ensures normalization of the wave function.

This new formalism leads to a procedure which applies our 3 Dimensional Simpson Integration (3D SI) computer program to calculate the correlation energy of a close-shell system without any approximate Taylor expansion. Testing of the correlation energy for He and its isoelectronic ions has been performed. The results have reached about 70% of the exact  $E_c$  value. The optimum values of variational parameters giving  $E_c$  for the He isoelectronic series have been found. The percentage correlation contribution of each term in  $E_c$  have been determined. All of our results give reasonable values with clear physical meaning and support our theoretical picture of correlation.

## (A) DERIVATION OF THE REDUCED DENSITY MATRICES

In this thesis the N-representability of density matrices is ensured, since this is required by quantum mechanics [62]. The problem of N-representability is that of finding conditions by which to recognize the reduced density matrices which are related to an N-body wave function according to the rule (4-1)

$$P_m(1, \dots, N, 1', \dots, N') = N(N-1) \dots (N-m+1) \int \Psi \Psi^* dm+1 \dots dN \quad (4-1)$$

where  $\Psi$  is an antisymmetric N-body wavefunction.

For convenience  $p_1(ij')$  or  $p(ij')$  represent a first order density matrix, and  $P_2(iji'j')$  represents a second order density matrix.

All higher order density matrices are known functionals of first order density matrices in the single determinant case. The diagonal elements of  $p(11')$  delivers by the density itself

$$p(1) = p(11') \Big|_{1' \rightarrow 1} \quad (4-2)$$

Furthermore, it occurs by Gilbert's theorem [45] that any normalized, well behaved density is N-representable by a single Slater determinant of orbitals. The density matrices

$P_{\text{det}}^{(11')}$ , are N-representable by a single Slater determinant

$\Psi_{\text{det}}$ , ie,

$$\Psi_{\text{det}}(1, \dots, N) = \frac{1}{N!} \begin{vmatrix} \phi_1(1) & \dots & \phi_1(N) \\ \vdots & & \vdots \\ \phi_N(1) & \dots & \phi_N(N) \end{vmatrix} \quad (4-3)$$

and

$$\phi_m = \sum C_k \Omega_k \quad (4-4)$$

where  $\Omega_k$  is a basis, and the population matrix  $\underline{P}$  is

$$\underline{P} = \underline{C}^+ \underline{C} \quad (4-5)$$

The population matrix  $\underline{P}$  is required to be a Hermitian, normalized projector and satisfies the experimental scattering factors [32], ie.,

$$\underline{P} = \underline{P}^+ \quad (\text{A}) (4-6)$$

$$\text{tr } \underline{P} = N \quad (\text{B}) (4-6)$$

$$\underline{P}^2 = \underline{P} \quad (\text{C}) (4-6)$$

$$F(\vec{k}) = \text{tr } \underline{P} \underline{f}(\vec{k}) \quad (\text{D}) (4-6)$$

If the  $\underline{P}$  matrix is a solution of Clinton's equation, the above conditions should be satisfied. Clinton's equation is

$$\underline{P}_{n+1} = 3 \underline{P}_n^2 - 2 \underline{P}_n^3 + \lambda_K \underline{f}(k) + \lambda_N \underline{1} \quad (1-48)$$

where  $\lambda_K$  and  $\lambda_N$  are Lagrangian multipliers. So

$$P_{\text{det}}(11') = \text{tr } \underline{P} \underline{\Omega}(1) \underline{\Omega}^+(1') \quad (4-8)$$

will be single determinant N-representable, and have diagonal elements which deliver the experimental density

[47] i.e.,

$$p(1) = p_{\text{det}}(11') \Big|_{1' \rightarrow 1} = \text{tr } \underline{P} \underline{\Omega}(1) \underline{\Omega}^+(1) \quad (4-9)$$

A highly accurate wavefunction can be generated by the product of the above determinant wavefunction and a spinless symmetric correlation function [16,62,63]

$$\Psi = \Psi_{\text{det}}(1, \dots, N) \prod_{i>j} [1-\phi(ij)] \quad (4-10)$$

The correlation function  $\phi(i,j)$  is defined symmetric to permutation in the coordinates of particle pairs so that the

total wave function of (4-10) is antisymmetric as required. The density matrices which flow from correlated determinant wave functions (4-10), may be obtained by integration directly from the fundamental definition (4-1) of density matrices, and contain a finite number of expansion terms in the form [64]

$$p(11') = p^{(0)}\{11'\} + p^{(1)}\{11'\} + p^{(2)}\{11'\} + \dots + p^{(N(N-1)/2)}\{11'\} \quad (\text{A}) (4-11)$$

$$p_2(12) = p_2^{(0)}\{12\} + p_2^{(1)}\{12\} + p_2^{(2)}\{12\} + \dots + p_2^{(N(N-1)/2)}\{12\} \quad (\text{B}) (4-11)$$

For convenience define  $b(iji'j')$  as

$$b(iji'j') = -\phi(ij) - \phi(i'j') + \phi(ij) \cdot \phi(i'j') \quad (4-12)$$

The properties of the system are expected to be calculated to be accurately enough up to second order in these terms because of the general smallness of the  $b(iji'j')$ . Since the Hamiltonian contains only one and two-body terms usually, the reduction from an  $N$ -body to a two-body description occurs [5]. Within this analysis for a two body system the density matrices are obtained as

$$P_2(121'2') = P_{2\text{det}}(121'2') + P_{2\text{det}}(121'2') \cdot b(121'2') \quad (4-13)$$

These highly accurate density matrices are approximately  $N$ -

representable.

From (4-1) we have

$$P_1(11') = \frac{1}{N-1} \int P_2(121'2) d2 \quad (4-14)$$

From (4-13) and (4-14),  $P_1(11')$  could be

$$P_1(11') = P_{1\text{det}}(11') + \int P_{2\text{det}}(121'2) \cdot b(121'2) d2 \quad (4-15)$$

The above density matrices are approximately N-representable, and may be used to calculate the correlation energy. The Hohenberg and Kohn (HK) theorem [25] states that the many-particle ground state is a unique functional of the electronic density, this means that the electron density which can be used to determine the above density matrices, contains all the information necessary to evaluate every electronic property of the system.

#### (B) DERIVATION OF A CORRELATION ENERGY FORMULA

The two body Hamiltonian has the usual form

$$H = \sum h_i + \sum h_{ij} \quad (4-16)$$

where

$$h_i = -\frac{1}{2} \nabla_i^2 - \frac{Z}{r_i} \quad (\text{A}) (4-17)$$

$$h_{ij} = -\frac{1}{r_{ij}} \quad (B) (4-17)$$

From the HK theorem, the expectation value of H for two bodies is

$$E = \int h_{1p} (11') \Big|_{1' \rightarrow 1} d1 + \frac{1}{2} \int h_{12} P_2 (12) d1 d2 \quad (4-18)$$

The correlation energy is calculated using the exact density determinant as

$$E_c = E_{\text{exa}} - E_{\text{det}} \quad (4-19)$$

where

$$E_{\text{det}} = \int h_{1p} P_{1\text{det}} (11') \Big|_{1' \rightarrow 1} d1 + \frac{1}{2} \int h_{12} P_{2\text{det}} (12) d1 d2 \quad (4-20)$$

where  $\nabla$  operates on  $r_1$  ie,

$$\nabla^2 = \frac{\partial^2}{\partial x_1^2} + \frac{\partial^2}{\partial y_1^2} + \frac{\partial^2}{\partial z_1^2} \quad (A) (4-21)$$

$$r_1^2 = x_1^2 + y_1^2 + z_1^2 \quad (B) (4-21)$$

From (4-15), (4-18) and (4-20) we have

$$\begin{aligned}
E_c &= -\frac{1}{4} \int \nabla^2 (b(121'2) \cdot P_{2\det}^{(121'2)}) \Big|_{1' \rightarrow 1} d1 d2 \\
&\quad - Z \int b(12) \cdot P_{2\det}^{(12)} \frac{-1}{r_1} d1 d2 \\
&\quad + \frac{1}{4} \int b(12) \cdot P_{2\det}^{(12)} \frac{-1}{r} d1 d2 \tag{4-22}
\end{aligned}$$

where

$$\begin{aligned}
r &= \sqrt{x^2 + y^2 + z^2} \\
&= \sqrt{(x_1 - x_2)^2 + (y_1 - y_2)^2 + (z_1 - z_2)^2} \tag{4-23}
\end{aligned}$$

For convenience, we list definitions

$$V_c(1) = \frac{1}{4} \int b(12) \cdot P_{2\det}^{(12)} \frac{-1}{r} d1 d2 \tag{A} (4-24)$$

$$V_c(2) = - \int b(12) \cdot P_{2\det}^{(12)} \frac{-Z}{r_1} d1 d2 \tag{B} (4-24)$$

$$T_c(1) = -\frac{1}{4} \int b(121'2) \cdot \nabla^2 P_{2\det}^{(121'2)} \Big|_{1' \rightarrow 1} d1 d2 \tag{C} (4-24)$$

$$T_c(2) = - \int \nabla b(121'2) \cdot \nabla P_{2\det}^{(121'2)} \Big|_{1' \rightarrow 1} d1 d2 \tag{D} (4-24)$$

$$Tc(3) = - \frac{1}{2} \int P_{2\text{det}}(121'2) \cdot v^2 b(121'2) \Big|_{1' \rightarrow 1} d1 d2 \quad (\text{E}) (4-24)$$

The correlation function form is taken as [2]

$$\phi(12) = e^{-\beta^2 R^2} \left[ 1 - \epsilon(R) \left( 1 + \frac{R}{2} \right) \right] \quad (4-25)$$

where

$$\vec{R} = \frac{1}{2} (\vec{r}_1 + \vec{r}_2) \quad (\text{A}) (4-26)$$

$$R = \frac{1}{4} \sqrt{(x_1+x_2)^2 + (y_1+y_2)^2 + (z_1+z_2)^2} \quad (\text{B}) (4-26)$$

$$\vec{r} = (\vec{r}_1 - \vec{r}_2) \quad (\text{C}) (4-26)$$

$$r = \sqrt{(x_1-x_2)^2 + (y_1-y_2)^2 + (z_1-z_2)^2} \quad (\text{D}) (4-26)$$

The definition of  $\phi(12)$  fulfills all the conditions of the cusp for  $r_1 \rightarrow r_2$  [56]. From (4-10) and (4-25), it is clear when the electrons are far apart the wave function (4-10) is reduced to the single determinant, and when a pair of electrons collide the function  $\phi(12)$  becomes

$$\Psi = \Psi_{\text{det}} \epsilon(R) \quad (4-27)$$

The region in which the function  $\phi(12)$  is appreciably different from zero depends on the value of  $\beta$ . the volume of this region may be taken to be

$$V = 4\pi \int_0^\infty e^{-\beta^2 r^2} r^2 dr = \left( \frac{\sqrt{\pi}}{\beta} \right)^3 \quad (4-28)$$

For reasonable  $\beta$  value a assumption is made as that the volume  $V$  in (4-28) should be proportional to the volume of exclusion in Wigner's formula [57]. So we have

$$V = k \frac{1}{p(R)} \quad (4-29)$$

where  $p(R)$  is the electronic density and  $k$  is proportional constant. This formula indicates the physical meaning of  $k$ , which is the average number of the electrons in volume  $V$ . From (4-28) and (4-29) we have

$$\beta = \sqrt{\pi} \cdot k^{-1/3} p(R)^{1/3} = q \cdot p(R)^{1/3} \quad (4-30)$$

where

$$q = \sqrt{\pi} \cdot k^{-1/3} \quad (4-31)$$

For the He atom and its isoelectronic series each trial has 2 electrons, so  $k \leq 2$ . we assume the average number in  $V$

is equal to 2 for He and its series, so  $q$  should be

$$q(\text{He}) \geq \sqrt{\pi} \cdot 2^{-1/3} \approx 1.4068 \quad (4-32)$$

Also the  $q$  can be considered a parameter, finding the optimum value of  $E_c$ .

The functional  $\epsilon(R)$  should be consistent with the density normalization, ie,

$$\int P_{1\text{exact}}^{(11')} |_{1' \rightarrow 1} d1 = \int P_{1\text{det}}^{(11)} |_{1' \rightarrow 1} d1 = N \quad (4-33)$$

For the He isoelectronic series  $N = 2$ . From equation (4-15)

$$I = \int b(12) P_{2\text{det}}(12) d2 d1 = 0 \quad (4-34)$$

The functional  $\epsilon(R)$  should satisfy the equation (4-34), for density normalization. In general all terms of the correlation energy from the usual Hamiltonian contribute to  $E_c$ . Here  $T_c(1)$ ,  $T_c(2)$ ,  $T_c(3)$ ,  $V_c(1)$  and  $V_c(2)$  all have contributions to  $E_c$ . The Colle-Salvetti formalism does not require the terms  $T_c(1)$  and  $V_c(2)$ , which here contribute to the total correlation energy.

In this thesis the form of  $\epsilon(R)$  is introduced as

$$\epsilon(R) = \frac{u}{t} \frac{\sqrt{\pi \cdot B}}{\sqrt{\pi \cdot B}} \quad (4-35)$$

This formula is inspired by the formula of Colle-Salvetti, and its use make easy satisfaction the equation of normalization in (4-34) by adjustment of the parameter  $t$ . The variational parameter  $u$  in (4-35) is introduced to obtain greater flexibility in optimizing  $E_c$ .

#### (C) THE CALCULATIONS OF CORRELATION ENERGY

All of the numerical integrations are performed by our 3D SI program without any approximate series expansions. There are two ways for determining  $q$ . One way is by taking  $q = 1.4068$  which is its value in (4-32) for the He isoelectronic series. That means we assume the average electronic number in the region, in which the function  $\phi(12)$  is appreciably different from zero, is equal to total number of the electrons in the system. Another is to take  $q$  as a variational parameter and let both  $q$  and  $u$  change in order to bring  $E_c$  to its best value at the minimum of  $E_c$ .

A summary of the procedure follows:

<1> Take a  $u$  value  $u_i$  and find a  $t$  value  $t_0(i)$ , which satisfies the equation (4-34) under this  $u_i$  by the premium selection method in mathematics. The  $u_i$  value should be limited by  $t \geq 0$ . This premium selection process may be completed by a computer program automatically.

<2> Use both  $u_i$  and  $t_0(i)$  values to calculate the correlation energy  $E_c(i)$  by the equations (4-22).

<3> Repeat above steps <1> and <2> to get more values of correlation energy  $E_c(i+1), E_c(i+2) \dots$  under different  $u_{i+1}, u_{i+2} \dots$  and  $t_0(i+1), t_0(i+2) \dots$  values.

<4> The values of  $E_c$  are plotted vs.  $u$ , then take the minimum of  $E_c$ .

<5> The parameter  $q$  may be varied similarly to  $u$ . It is not necessary just take  $q = 1.4068$ , although this value consistent with Colle and Salvetti formalism. The double parameters  $u$  and  $q$  can be varied to find the minimum of  $E_c$  as the best value, which of cause will give better values, when only a single parameter  $u$  is varied.

When  $q = 1.4068$  and the  $u$  parameter is varied, the results of the calculations for He,  $Li^+$ ,  $Be^{+2}$  and  $B^{+3}$  are listed in the Table (4.1), the Table (4.2) and Figure 4.1(A) to 4.5(B). The computer program's zero for Epsilon is  $Epsilon \leq 0.00001$  in the calculations. The results showed over 50% of correlation energy have been recovered.

The  $E_c$  of He,  $Li^+$  and  $Be^{+2}$  are calculated by varying both parameter  $u$  and  $q$ . The results have reached about 70% of the exact value and are listed in Table 4.3 and Figures 4.6A to 4.7B. These results strongly support our formalism.

TABLE 4.1A THE CORRELATION ENERGY OF He ATOM CALCULATED  
BY VARYING THE u PARAMETER (q = 1.4068)

He u	t	Tc	Vc(1)/Vc(2)	Vc	Ec
0.90	0.4829350	0.0100	-0.0595 0.0289	-0.0306	-0.0206
0.94	0.5698257	0.0319	-0.0554 0.0020	-0.0534	-0.0216
0.95	0.5920193	0.0368	-0.0545 -0.0038	-0.0585	-0.0217
0.96	0.6128494	0.0428	-0.0532 -0.1213	-0.0653	-0.0226
0.97	0.6350412	0.0476	-0.0523 -0.0178	-0.0702	-0.0226
0.98	0.6594372	0.0509	-0.0522 -0.0195	-0.0717	-0.0209
1.00	0.7046678	0.0598	-0.0500 -0.0238	-0.0801	-0.0203
1.10	0.9299361	0.1025	-0.0433 -0.0791	-0.1224	-0.0198
1.20	1.1573688	0.1400	-0.0366 -0.1228	-0.1594	-0.0194
1.30	1.3919221	0.1697	-0.0321 -0.1535	-0.1856	-0.0159

\*\*\*\*\*

Figure 3.1A THE CORRELATION ENERGY(He)

$q = 1.4068$ , MINIMUM  $E_c = -0.0226$  a.u.

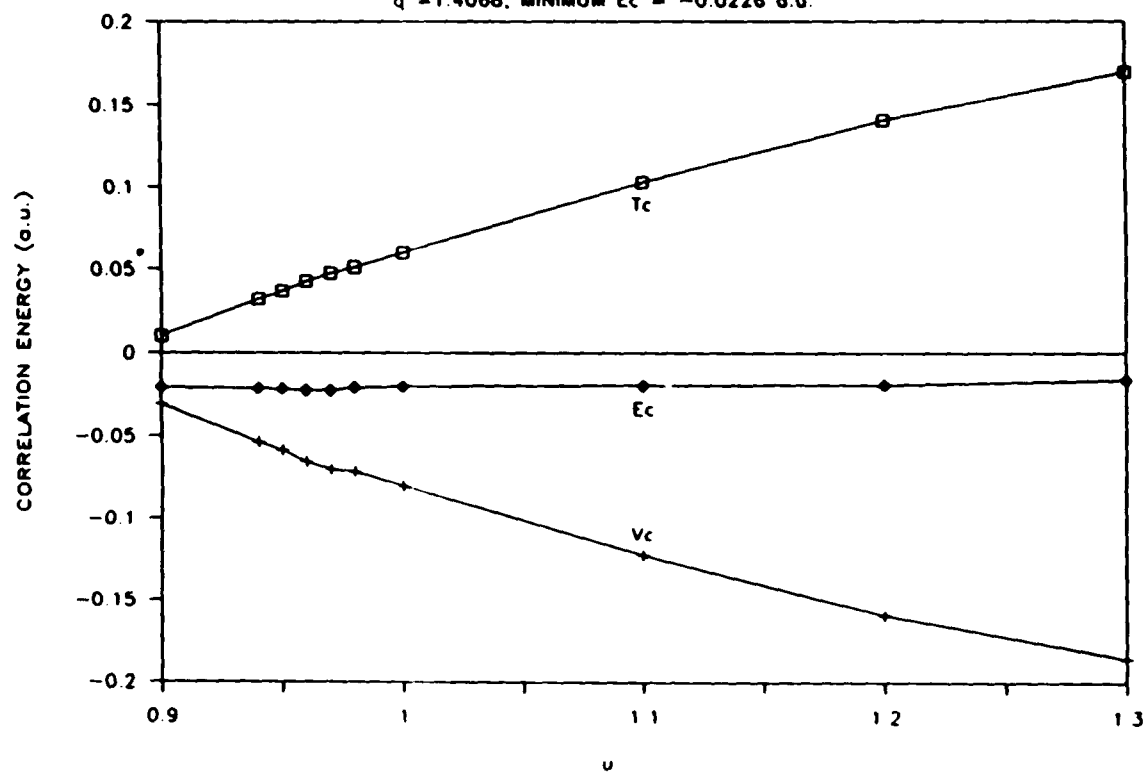


Figure 3.1B A CALCULATION FOR  $E_c$  OF He

$q = 1.4068$ , MINIMUM  $E_c = -0.0226$  a.u.

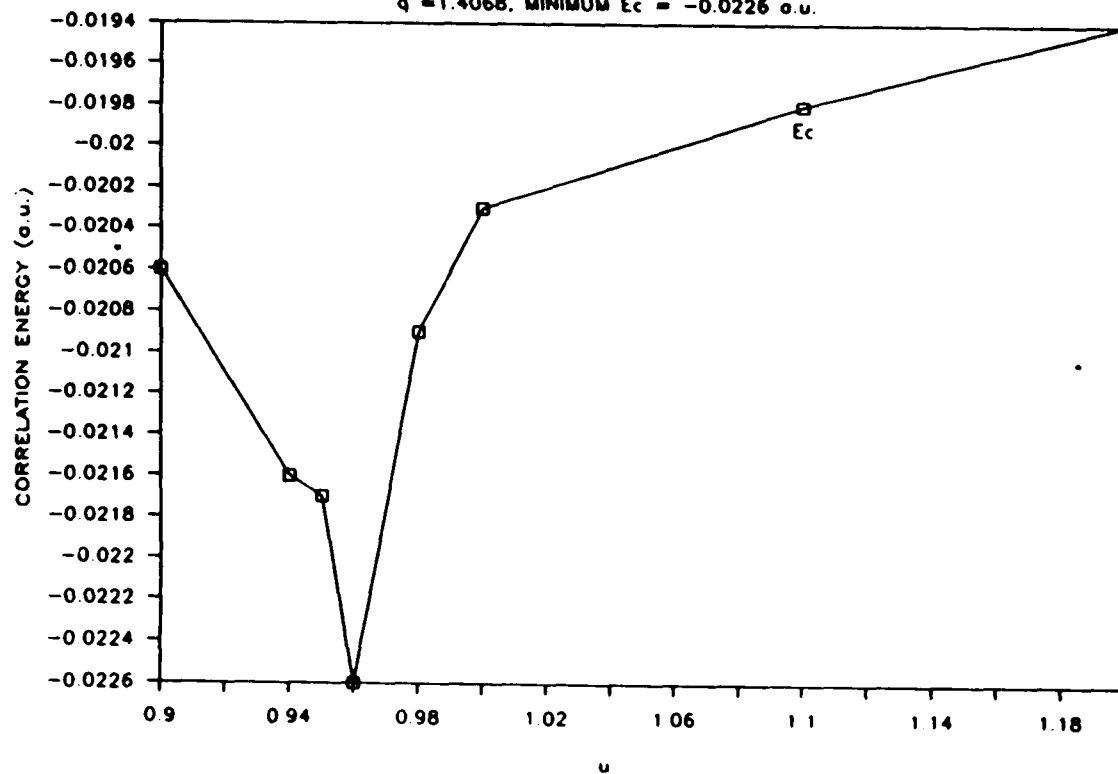


TABLE 4.1B THE CORRELATION ENERGY OF He ISOELECTRONIC  
 ION Li<sup>+</sup> CALCULATED BY VARYING THE u PARAMETER  
 (q = 1.4068)

Li <sup>+</sup> u	t	Tc	Vc(1)/Vc(2)	Vc	Ec
0.94	0.5126082	0.0128	-0.0674 0.0359	-0.0315	-0.0188
0.96	0.5712420	0.0450	-0.0630 -0.0051	-0.0681	-0.0231
0.97	0.6030703	0.0577	-0.0618 -0.0176	-0.0794	-0.0217
1.00	0.6962351	0.0997	-0.0572 -0.0623	-0.1209	-0.0213
1.10	1.0119278	0.2200	-0.0449 -0.1861	-0.2352	-0.0152
1.20	1.3303915	0.3270	-0.0331 -0.3012	-0.3395	-0.0125
1.30	1.6576583	0.4151	-0.0227 -0.3972	-0.4199	-0.0

\*\*\*\*\*

Figure 3.2A THE CORRELATION ENERGY(Li+)

$q = 1.4068$  minimum =  $-0.0231$  a.u.

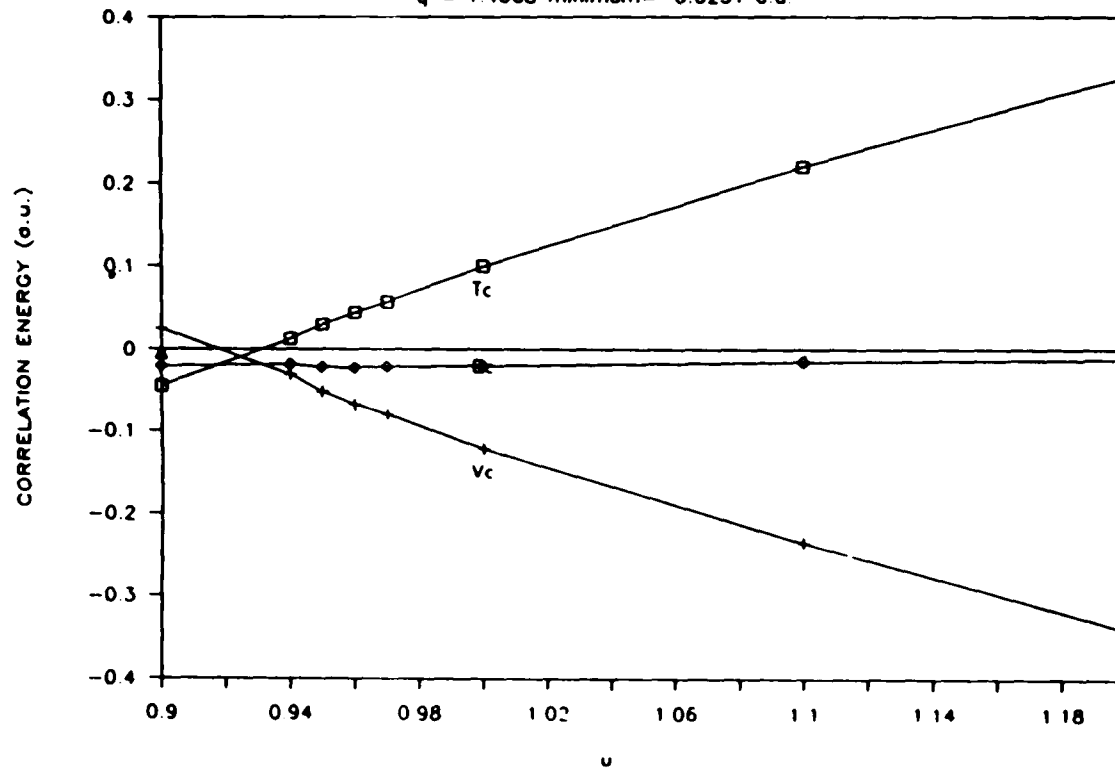


Figure 3.2B A CALCULATION FOR  $E_c$  OF  $Li^+$

$q = 1.4068$  minimum =  $-0.0231$  a.u.

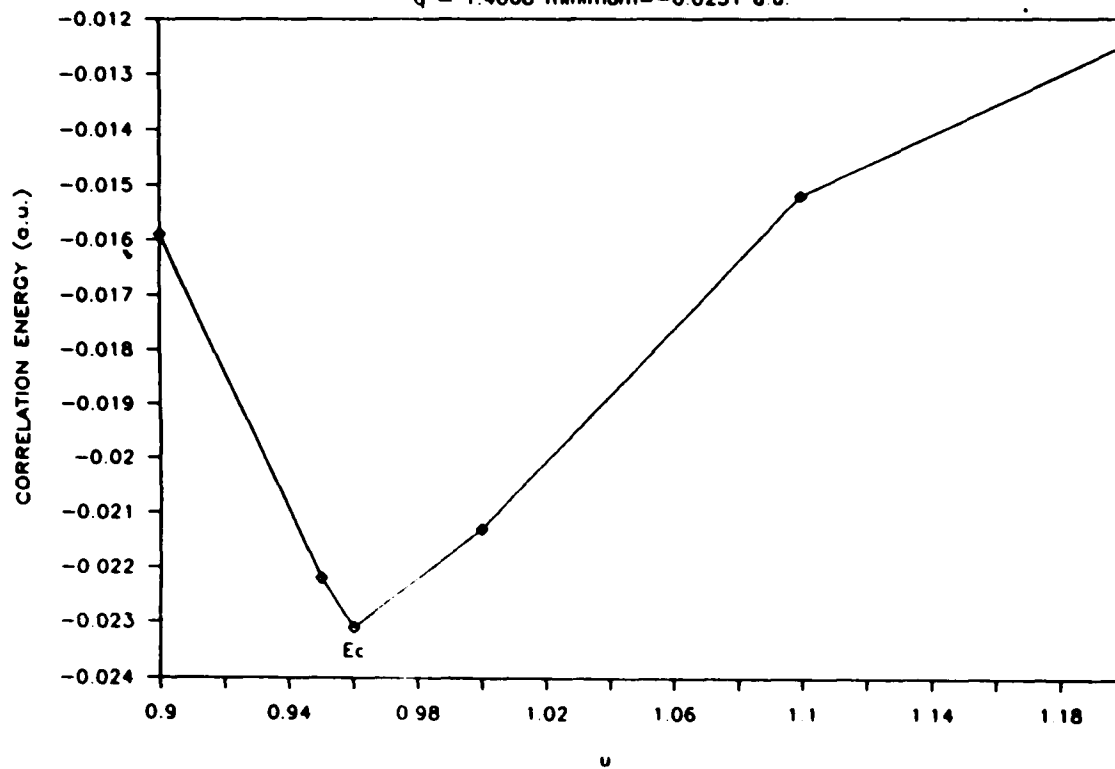


TABLE 4.1C THE CORRELATION ENERGY OF He ISOELECTRONIC  
 ION  $\text{Be}^{+2}$  CALCULATED BY VARYING THE  $u$  PARAMETER  
 ( $q = 1.4068$ )

$\text{Be}^{+2}$ $u$	$t$	$T_c$	$V_c(1)/V_c(2)$	$V_c$	$E_c$
0.96	0.5343898	0.0256	-0.0703 0.0230	-0.0472	-0.0216
0.97	0.5723571	0.0562	-0.0673 -0.0130	-0.0803	-0.0240
0.98	0.6125605	0.0826	-0.0653 -0.0392	-0.1045	-0.0219
1.00	0.6929689	0.1350	-0.0613 -0.0926	-0.1540	-0.0190
1.10	1.0985960	0.3757	-0.0420 -0.3427	-0.3847	-0.0090
1.20	1.5100455	0.5869	-0.0246 -0.5653	-0.5899	-0.0030

Figure 3.3A CORRELATION ENERGY Be(+2)

$q = 1.4068$  minimum =  $-0.0240$  a.u.

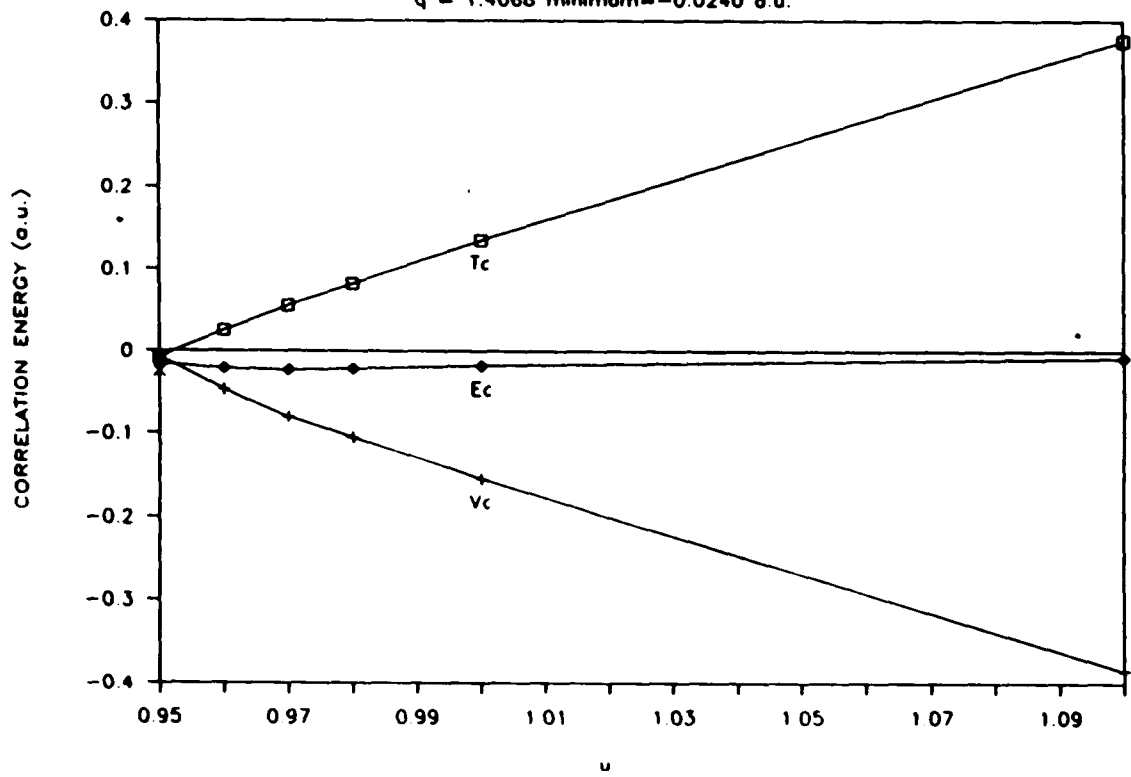


Figure 3.3B A CALCULATION FOR  $E_c$  Be(+2)

$q = 1.4068$  minimum =  $-0.0240$  a.u.

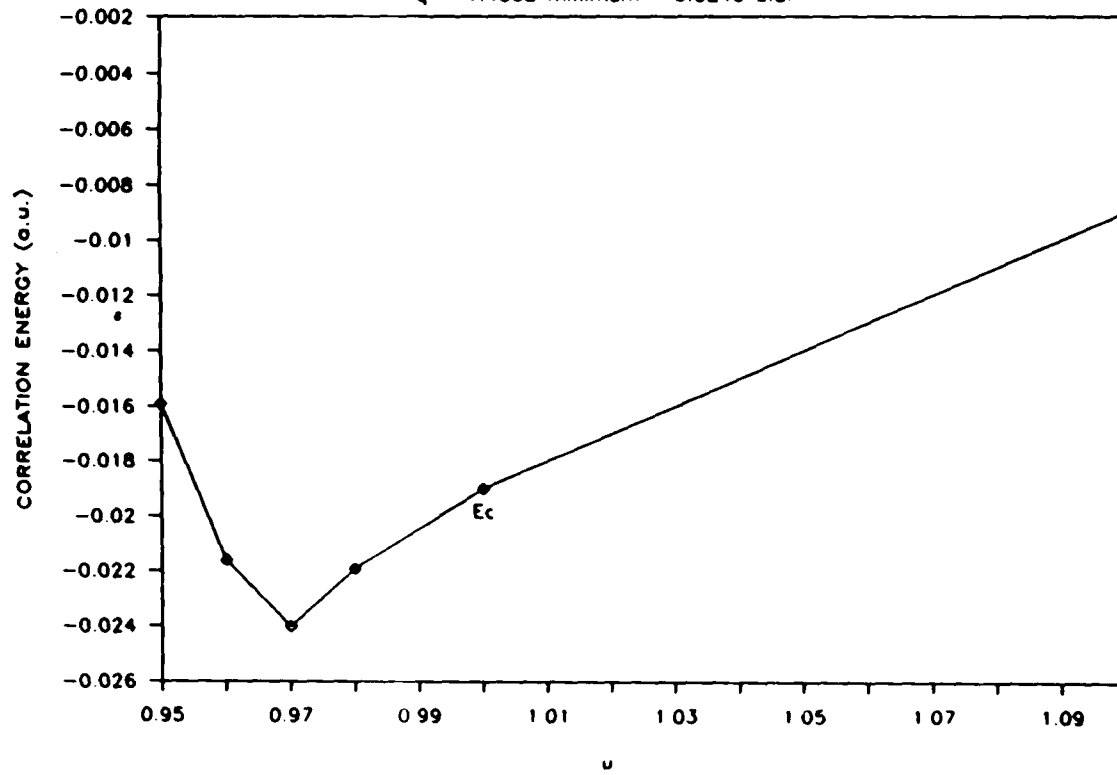


TABLE 4.1D THE CORRELATION ENERGY OF He ISOELECTRONIC  
 ION  $B^{+3}$  CALCULATED BY VARYING THE  $u$  PARAMETER  
 ( $q = 1.4068$ )

$B^{+3}$	$u$	$t$	$Tc$	$Vc(1)/Vc(2)$	$Vc$	$Ec$
	0.965	0.5224462	0.0082	-0.0750 0.0484	-0.0267	-0.0184
	0.970	0.5454866	0.0341	-0.0730 0.0176	-0.0554	-0.0213
	0.980	0.5942741	0.0796	-0.0702 -0.0278	-0.0980	-0.0183
	1.000	0.6918364	0.1689	-0.0646 -0.1190	-0.1835	-0.0146
	1.070	1.0332787	0.4707	-0.0443 -0.4401	-0.4844	-0.0136
	1.100	1.1840386	0.5844	-0.0373 -0.5535	-0.5908	-0.0064

Figure 3.4A CORRELATION ENERGY OF B(+3)

$q = 1.4068$  minimum =  $-0.0213$  e.u.

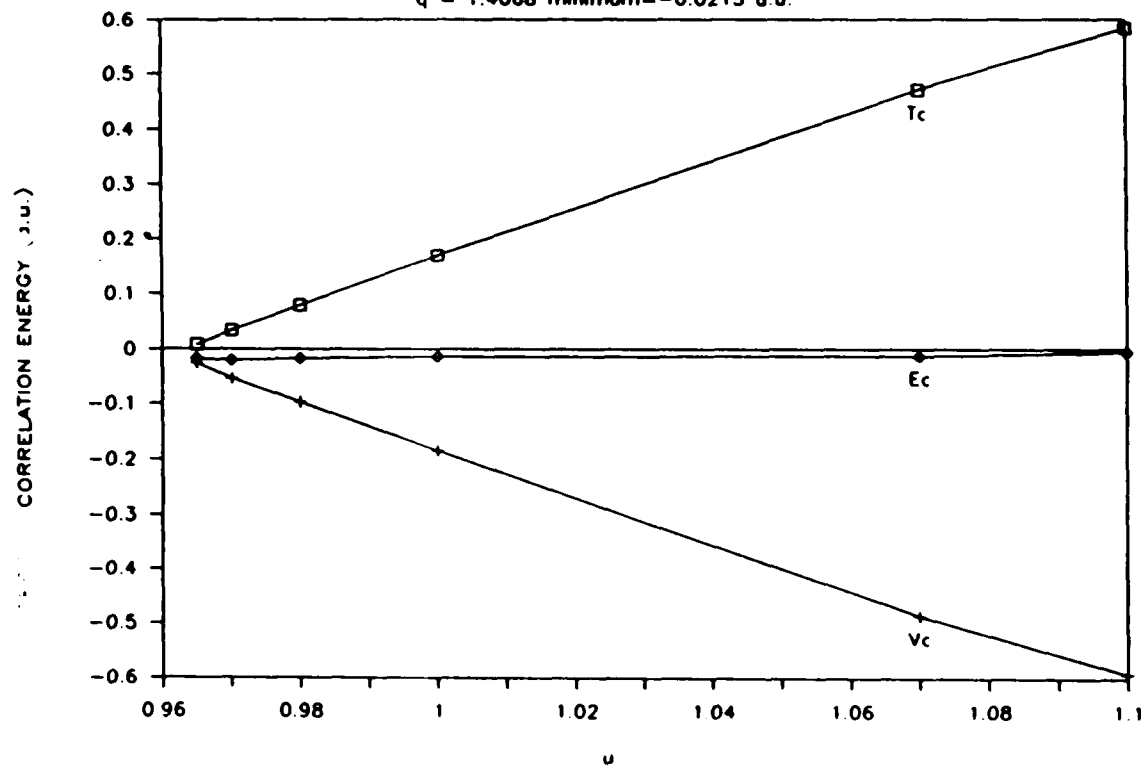


Figure 3.4B A CALCULATION FOR  $E_c/B(+3)$

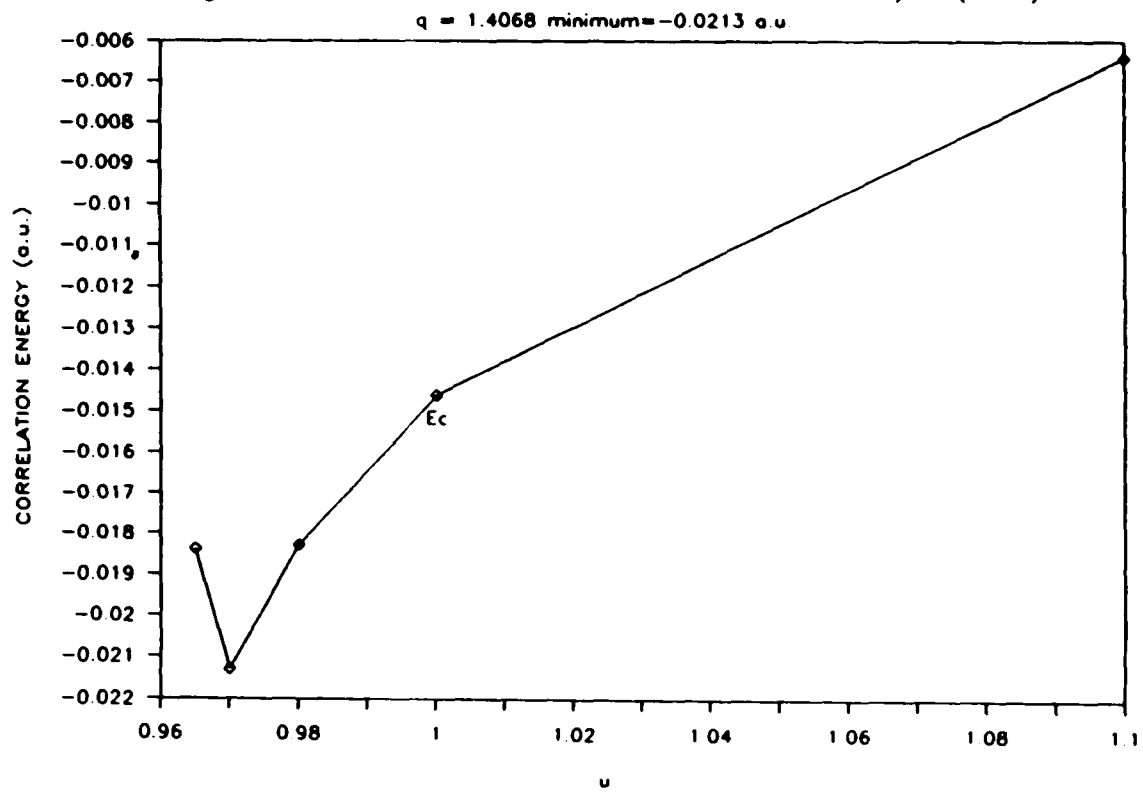


TABLE 4.2 THE LIST OF THE CORRELATION ENERGY OF He AND  
ITS ISOELECTRONIC IONS AT THE MINIMUM  
( $q = 1.4068$ )

	He	Li <sup>+</sup>	Be <sup>+2</sup>	B <sup>+3</sup>
u	0.96	0.96	0.97	0.97
t	0.612849	0.571242	0.572357	0.545487
Tc	0.0428	0.0450	0.0562	0.0341
Vc(1)	-0.0532	-0.0630	-0.0673	-0.0730
Vc(2)	-0.0121	-0.0051	-0.0130	0.0176
Vc	-0.0702	-0.0681	-0.0803	-0.0554
Ec	-0.0226	-0.0231	-0.0240	-0.0213
Ec (Clementi)	-0.0421	-0.0435	-0.0443	-0.0448
Ec(Cal.)				
-----	54 %	53%	54%	48%
Ec(Clementi)				

Table 4.3A THE CALCULATION OF THE CORRELATION ENERGY  
OF He ATOM BY VARYING BOTH u AND q PARAMETERS

-----					
He u=1.00					
q	t	Vc(1)/Vc(2)	Vc	Tc	Ec (Unit: a.u.)
-----					
0.70	0.46233469	-0.0996 0.0537	-0.0460	0.0210	-0.0250
0.77	0.49412486	-0.0928 0.0303	-0.0624	0.0354	-0.0271
0.80	0.50728059	-0.0901 0.0228	-0.0674	0.0401	-0.0272
0.81	0.51157230	-0.0893 0.0203	-0.0689	0.0416	-0.0273
0.90	0.54865986	-0.0821 0.0031	-0.0790	0.0521	-0.0268
1.00	0.58547395	-0.0743 -0.0133	-0.0871	0.0606	-0.0265
1.10	0.61992025	-0.0679 -0.0205	-0.0884	0.0635	-0.0249
1.20	0.64934248	-0.0612 -0.0291	-0.0903	0.0655	-0.0248
*****					

Table 4.3B THE CALCULATION OF THE CORRELATION ENERGY  
OF He ATOM BY VARYING BOTH u AND q PARAMETERS

-----					
He u=0.97					
q	t	Vc(1)/Vc(2)	Vc	Tc	Ec (Unit: a.u.)
-----					
0.80	0.46397403	-0.0927 0.0417	-0.0511	0.0221	-0.0289
0.90	0.50049251	-0.0843 0.0196	-0.0647	0.0356	-0.0291
1.00	0.53317547	-0.0766 0.0033	-0.0732	0.0444	-0.0288
1.10	0.56190836	-0.0693 -0.0094	-0.0787	0.0500	-0.0286
1.20	0.58752078	-0.0628 -0.0174	-0.0802	0.0523	-0.0279
1.30	0.61046767	-0.0569 -0.0215	-0.0785	0.0520	-0.0265
*****					
u=0.95					
-----					
0.80	0.43602830	-0.0953 0.0596	-0.0357	0.0078	-0.0279
0.90	0.46951395	-0.0866 0.0358	-0.0508	0.0222	-0.0286
1.00	0.49902606	-0.0785 0.0172	-0.0613	0.0323	-0.0291
1.10	0.52507025	-0.0712 0.0038	-0.0674	0.0394	-0.0298
1.20	0.54634184	-0.0637 -0.0097	-0.0734	0.0432	-0.0301
1.30	0.56867921	-0.0587 -0.0094	-0.0681	0.0415	-0.0266
1.40	0.58696079	-0.0535 -0.0111	-0.0647	0.0399	-0.0248
*****					

Table 4.3C THE CALCULATION OF THE CORRELATION ENERGY  
OF He ATOM BY VARYING BOTH u AND q PARAMETERS

-----					
He u=0.93					
q	t	Vc(1)/Vc(2)	Vc	Tc	Ec (Unit: a.u.)
-----					
0.90	0.43823978	-0.0887 0.0507	-0.0380	0.0091	-0.0289
1.00	0.46503213	-0.0805 0.0317	-0.0489	0.0197	-0.0292
1.10	0.48849687	-0.0732 0.0179	-0.0553	0.0263	-0.0289
1.20	0.50907588	-0.0665 0.0088	-0.0578	0.0298	-0.0280
1.30	0.52709681	-0.0606 0.0033	-0.0573	0.0307	-0.0266

\*\*\*\*\*

Figure 4.1  $E_c$  of He ATOM

From double parameters  $q$  and  $u$

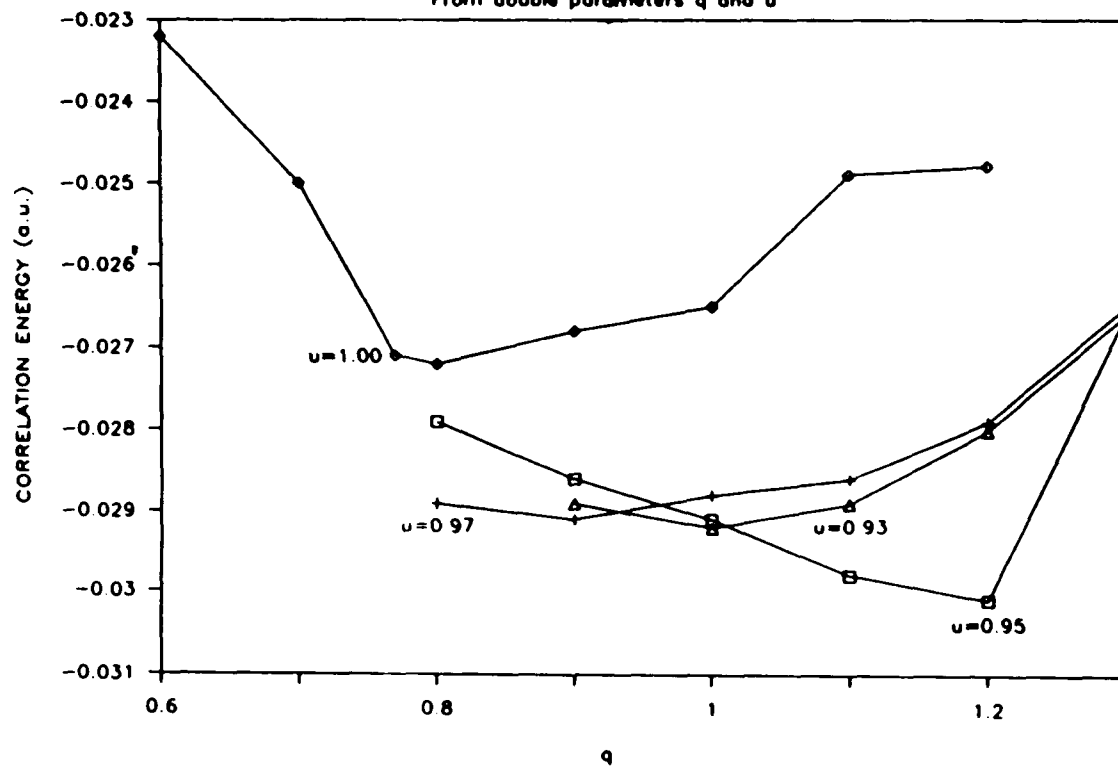


Table 4.4A THE CALCULATION OF THE CORRELATION ENERGY  
OF Li(+) ION BY VARYING BOTH u AND q PARAMETERS

Li <sup>+</sup> q	u=1.00 t	Vc(1)/Vc(2)	Vc	Tc	Ec (Unit: a.u.)
0.70	0.45761567	-0.1166 0.0683	-0.0483	0.0303	-0.0180
0.80	0.50229573	-0.1051 0.0187	-0.0864	0.0634	-0.0230
0.90	0.54215688	-0.0939 -0.0206	-0.1145	0.0903	-0.0242
1.00	0.57838076	-0.0845 -0.0469	-0.1315	0.1045	-0.0270
1.10	0.61089557	-0.0761 -0.0644	-0.1405	0.1119	-0.0286
1.20	0.64026922	-0.0685 -0.0737	-0.1422	0.1137	-0.0286
1.30	0.66679412	-0.0618 -0.0772	-0.1391	0.1109	-0.0281
1.41	0.6962351		-0.1209	0.0997	-0.0213

\*\*\*\*\*

Table 4.4B THE CALCULATION OF THE CORRELATION ENERGY  
OF Li(+) ION BY VARYING BOTH u AND q PARAMETERS

Li <sup>+</sup> q	u=0.96 t	Vc(1)/Vc(2)	Vc	Tc	Ec (Unit: a.u.)
0.90	0.45776379	-0.1026 0.0704	-0.0322	0.0121	-0.0200
1.00	0.48526198	-0.0922 0.0327	-0.0595	0.0330	-0.0264
1.10	0.51157230	-0.0843 0.0185	-0.0658	0.0411	-0.0247

\*\*\*\*\*

Table 4.4C THE CALCULATION OF THE CORRELATION ENERGY  
OF Li(+) ION BY 3D SI

Li <sup>+</sup> q	u=1.10 t	Vc(1)/Vc(2)	Vc	Tc	Ec (Unit: a.u.)
0.90	0.76062948	-0.0780 -0.1907	-0.2687	0.2511	-0.0176
1.00	0.81680536	-0.0690 -0.2107	-0.2797	0.2585	-0.0212
1.10	0.86864668	-0.0609 -0.2210	-0.2819	0.2584	-0.0235
1.20	0.91664743	-0.0538 -0.2238	-0.2776	0.2524	-0.0252
1.30	0.96132886	-0.0475 -0.2189	-0.2664	0.2419	-0.0245

\*\*\*\*\*

Figure 4.2  $E_c$  of  $\text{Li}(+)$  ION

From double parameters  $q$  and  $u$

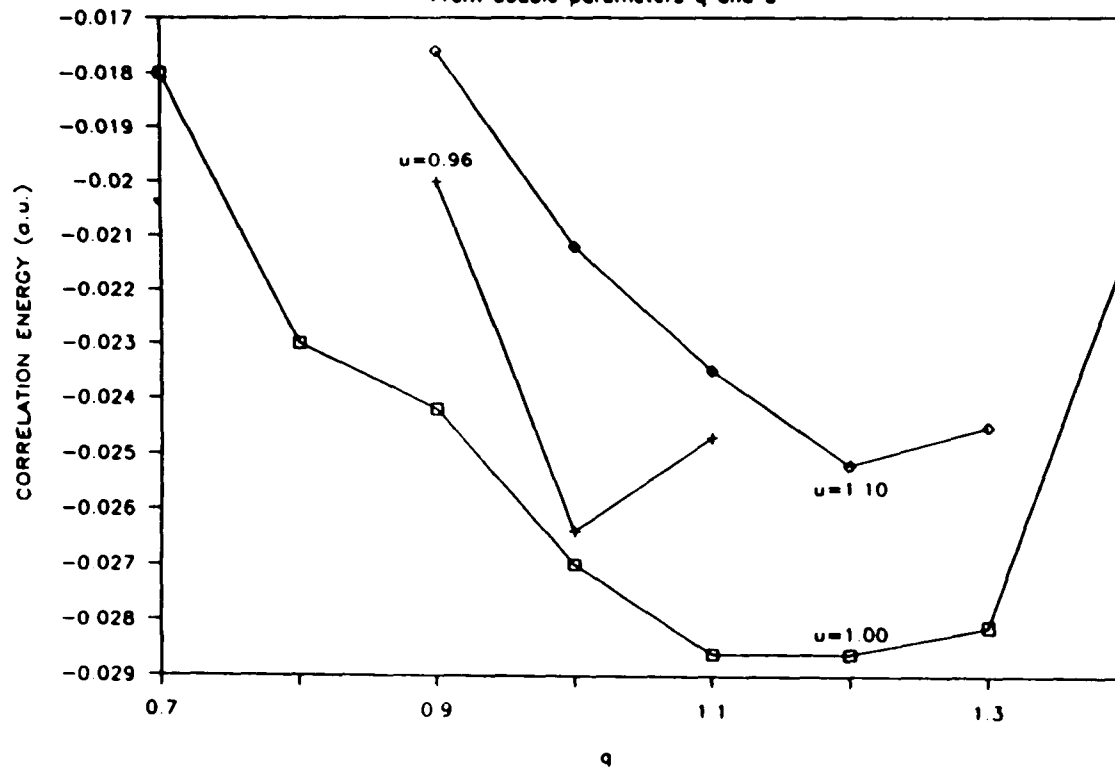


Table 4.5A THE CALCULATION OF THE CORRELATION ENERGY  
OF Be(+2) ION BY VARYING BOTH u AND q PARAMETERS

Be <sup>2+</sup> q	u=1.00 t	Vc(1)/Vc(2)	Vc	Tc	Ec (Unit: a.u.)
0.70	0.45399186	-0.1248 0.0701	-0.0547	0.0450	-0.0097
0.90	0.54025471	-0.1027 -0.0281	-0.1308	0.1159	-0.0149
1.00	0.57683742	-0.0926 -0.0611	-0.1537	0.1355	-0.0182
1.10	0.60705656	-0.0816 -0.1031	-0.1847	0.1545	-0.0303
1.20	0.63621616	-0.0734 -0.1143	-0.1878	0.1567	-0.0311
1.30	0.66295594	-0.0663 -0.1155	-0.1819	0.1518	-0.0301
*****					
Be <sup>2+</sup> q	u=0.97 t	Vc(1)/Vc(2)	Vc	Tc	Ec (Unit: a.u.)
1.00	0.48500440	-0.0988 0.0297	-0.0691	0.0390	-0.0301
1.10	0.50998199	-0.0893 0.0021	-0.0871	0.0550	-0.0322
1.20	0.53210580	-0.0809 -0.0141	-0.0949	0.0623	-0.0326
1.30	0.55151433	-0.0733 -0.0227	-0.0960	0.0641	-0.0319
*****					

Table 4.5B THE CALCULATION OF THE CORRELATION ENERGY  
OF Be(+2) ION BY VARYING BOTH u AND q PARAMETERS

Be <sup>+2</sup> q	u=0.96 t	Vc(1)/Vc(2)	Vc	Tc	Ec (Unit: a.u.)
1.00	0.45519423	-0.1015 0.0672	-0.0343	0.0033	-0.0310
1.10	0.47788617	-0.0920 0.0390	-0.0530	0.0206	-0.0323
1.20	0.49755937	-0.0834 0.0205	-0.0629	0.0304	-0.0325
1.30	0.51480752	-0.0758 0.0106	-0.0653	0.0334	-0.0319

\*\*\*\*\*

Figure 4.3 Ec. of Be(+2) ion

From double parameters q and u

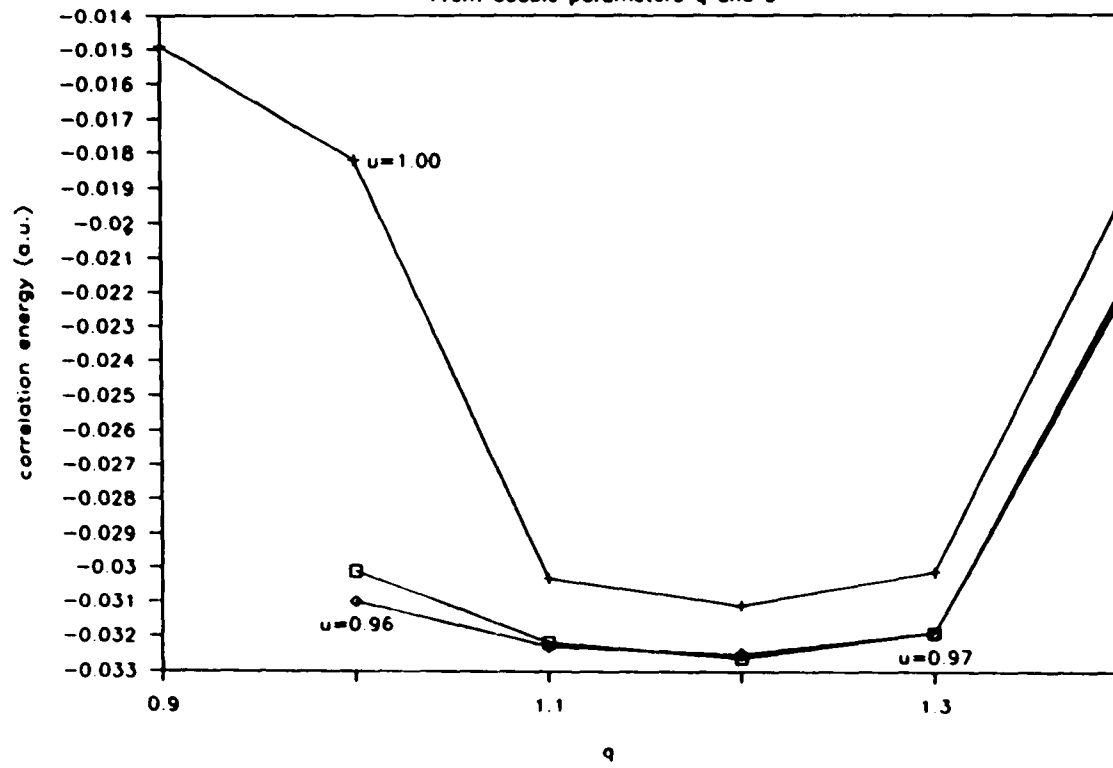


Table 4.6 THE BEST VALUES OF THE CORRELATION ENERGY  
 OF He AND ITS ISOELECTRONIC IONS BY 3D SI  
 WITH VARYING BOTH  $q$  AND  $u$  PARAMETERS

	He	Li <sup>+</sup>	Be <sup>+2</sup>
$q$	1.20	1.20	1.20
$u$	0.95	1.00	0.97
$t$	0.54634184	0.64026922	0.53210580
$Tc$	0.0432	0.1137	0.0623
$Vc(1)$	-0.0637	-0.0685	-0.0809
$Vc(2)$	-0.0097	-0.0737	-0.1143
$Vc$	-0.0734	-0.1422	-0.0949
$Ec$	-0.0301	-0.0286	-0.0326
$Ec(\text{Clementi})$	-0.0421	-0.0435	-0.0443
$Ec(\text{Cal.})$			
-----	73%	66%	74%
$Ec(\text{Clementi})$			

Table 4.7 THE CORRELATION ENERGY  $E_c$  OF He AND ITS  
 ISOELECTRONIC IONS CALCULATED BY 3D SI AND  
 COMPARED WITH THE NEARLY EXACT VALUES [3,4]

\*\*\*\*\*

	He	Li <sup>+</sup>	Be <sup>+2</sup>	B+3
$E_c(\text{Clementi})$	-0.0421	-0.0435	-0.0443	-0.0448

---

One Variational Parameter u

u	0.96	0.96	0.97	0.97
t	0.612849	0.571242	0.572357	0.545487
$E_c$	-0.0226	-0.0231	-0.0240	-0.0213

$E_c(\text{Cal.})$				
-----	54 %	53%	54%	48%
$E_c(\text{Clementi})$				

---

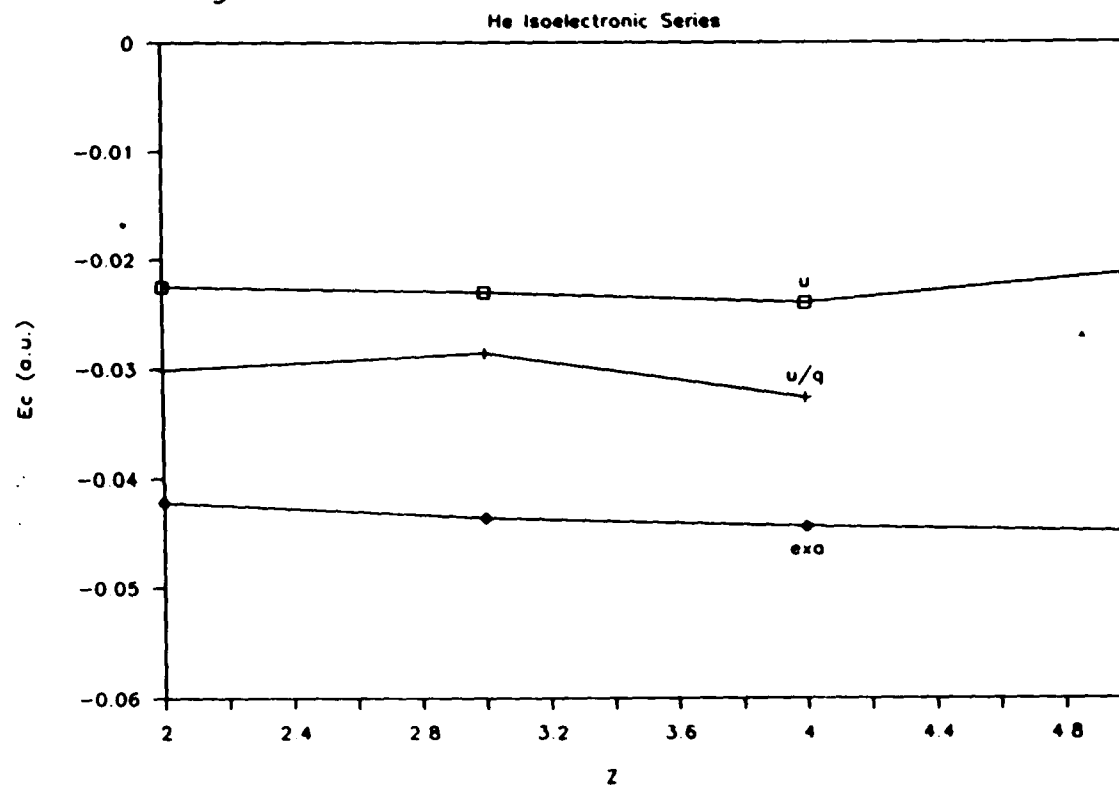
Two Variational Parameters u and q

q	1.20	1.20	1.20
u	0.95	1.00	0.96
t	0.5463418	0.6402692	0.5321058
$E_c$	-0.0301	-0.0286	-0.0326

$E_c(\text{Cal.})$			
-----	73%	66%	74%
$E_c(\text{Clementi})$			

\*\*\*\*\*

Figure 5.1 A COMPARISON OF  $E_c$  VALUES



## (D) CONCLUSION

The correlation energies for He and some of its isoelectronic ions are found to be about 70% of the exact  $E_c$ . The correlation energy in this thesis is nonrelativistic regarding the usual Hamiltonian. It includes the potential and kinetic terms of the correlation energy. Our calculations give values from -0.031 to -0.033 a.u.. The exact  $E_c$  of He and its isoelectronic series fall within the range -0.042 to -0.047 a.u. [3,4].

We define the percentage correlation to be the percentage of the calculated correlation energy over the HF energy in percentage.

$$J = -\frac{E_{COR}}{E_{HF}} \times 100\% \quad (4-36)$$

where  $E_{COR}$  represents any term in the correlation energy,  $E_{HF}$  represents the terms of the HF energy, which is correlated by the  $E_{COR}$ . The percentage correlation for He atom is listed on Table 4.8.

$V_c(1)$  is the most important contribution for  $E_c$  giving largest percentage correlation over -3%. The percentage correlation of other terms are 0.15% to 1.5%. This phenomenon reflects the fact [65] that Coulombic interaction between pairs of electrons, especially electrons with antiparallel spins as in the closed-shell case of the ground state of He isoelectronic series, is not properly accounted for by the HF method. The HF method just allows each electron to move in a

"smeared-out" field of the others. Electrons of parallel spins are kept apart by the antisymmetry principle; the electrons with antiparallel spins avoid each other also. The HF method overrides the Coulombic repulsion by considering the electronic effect alone. So  $V_c(1)$  gives a negative value to correct this problem introduced by the HF method.

The contribution of  $V_c(2)$  is important to reach the exact  $E_c$ . However, the percentage correlation from  $V_c(2)$  is less than 0.2%. This reflects the fact that the HF method and its central potential field model is very effective to calculate the potential energy  $V(2)$  of electron-nuclear attraction.

When the calculations are improved by varying the double parameters  $q$  and  $u$ , the Table 4.8 shows that only the percentage correlation of double electron term  $V_c(1)$  significantly change from -3.2% to -3.8%. The percentage correlation from  $T_c$  remains constant. This means the above improvement is only effective in the potential correlation energy.

The results also show  $T_c$  is positive and total  $V_c$  is negative. This means the HF method calculated the kinetic energy and the total potential energy insufficiently [65]. From Virial theorem

$$T^{HF} = -\frac{1}{2} V^{HF} = -E^{HF} \quad (4-37)$$

which is satisfied by HF method, if the one electron system was calculated by HF method, the  $T$  value should be lower than

the exact kinetic energy, so that the total energy  $E$  could be higher than the exact total energy which is required by the Variational principle.

The percentage correlation from  $T_c$  stays at 1.5%. It is clear this is a reasonable  $J$  value for  $T_c$ .

When  $q = 1.4068$ , the minimum values of  $E_c$  for our trials are around  $u=0.97$  (Table 4.2). This  $u$  value perhaps is helpful to find the minimum of  $E_c$  for the He isoelectronic series. Actually the minimum of  $E_c$  for  $B^{+3}$  was found by calculating  $E_c$  at  $u=0.97$  first, then calculating three or more points to make sure.

When both  $q$  and  $u$  are varied, the minimum values of  $E_c$  from the calculations have reached about 70% at  $u \approx 0.97$  and  $q=1.20$  for our trials (Table 4.6). The point ( $u \approx 0.97$  and  $q=1.20$ ) also is helpful to find out the  $E_c$  of more trials in the He isoelectronic series. So this point ( $u \approx 0.97$  and  $q=1.20$ ) may be called a characteristic point of He isoelectronic series.

Further work could improve the calculations of  $E_c$  by varying  $q$  in smaller intervals and calculating  $E_c$  from more example from the He isoelectronic series, as well as from Be and Ne isoelectronic series.

TABLE 4.8 THE PERCENTAGE CORRELATION  
FOR He ATOM [66]

CALCULATED CORRELATION ENERGY	HF ENERGY (unit:a.u.)	PERCENTAGE CORRELATION %
-----		
*****		
<u>By changing single parameter u</u>		
Vc(1)= -0.0532	V(1)= 1.68728	-3.2%
Vc(2)= -0.0121	V(2)= -7.41064	0.16%
Tc= 0.0428	T = 2.86168	1.5%
Ec=-0.0226	E = -2.86168	0.8%
*****		
<u>By changing double parameters q and u</u>		
Vc(1)= -0.0637	V(1)= 1.68728	-3.8%
Vc(2)= -0.0097	V(2)= -7.41064	0.13%
Tc=0.0432	T = 2.86168	1.5%
Ec=-0.0301	E = -0.28168	1.1%
*****		

APPENDIX 1    NUMERICAL COORDINATES IN 3 DIMENSIONS  
AND SIMPSON NUMERICAL INTEGRATION

(1) COORDINATES

The system of He atom and its isoelectronic series has 2 1s electrons in the ground state. They would be characterized by 6 dimensions.

Graph a1-1 gives coordinates (1) for this system. In these coordinates our hf orbital and correlation function only contain 3 independent variables  $r_1$ ,  $r_2$  and  $\theta'$ , which is the angle between  $r_1$  and  $r_2$ .

Graph A1-2 gives similar coordinates (2) for this system. In these coordinates our HF orbitals and correlation function only contains 3 independent variables  $R$ ,  $r$  and  $\theta$ , which is the angle between  $R$  and  $r$ .

The integrals for the system can be preformed only over the 3 dimensions above, other dimensions are not contained the density matrix or the correlation function, so that their contributions just give an over all multiplicative constant. All integration with 3 dimensions are performed by our 3D Simpson integration program.

Both sets coordinates (1) and (2) are used for convenient integration of various different functions.

In the above coordinates the following relationship holds:

$$\vec{r}_1 = \vec{i}x_1 + \vec{j}y_1 + \vec{k}z_1 \quad (\text{A1-1})$$

where  $\vec{i}, \vec{j}, \vec{k}$  are the unit vectors on the  $\vec{x}, \vec{y}, \vec{z}$  axes of the coordinate system.

$$\vec{r}_2 = \vec{i}x_2 + \vec{j}y_2 + \vec{k}z_2 \quad (\text{A1-2})$$

$$\vec{R} = \frac{1}{2} (\vec{r}_1 + \vec{r}_2) = \vec{i}x + \vec{j}y + \vec{k}z \quad (\text{A1-3})$$

$$= \frac{\vec{i}}{2} (x_1 + x_2) + \frac{\vec{j}}{2} (y_1 + y_2) + \frac{\vec{k}}{2} (z_1 + z_2)$$

$$\vec{r} = \vec{r}_1 - \vec{r}_2 = \vec{i}x + \vec{j}y + \vec{k}z \quad (\text{A1-4})$$

$$= \vec{i}(x_1 - x_2) + \vec{j}(y_1 - y_2) + \vec{k}(z_1 - z_2)$$

$$\vec{r}_1 = \vec{R} + \frac{\vec{r}}{2} \quad (\text{A1-5})$$

$$\vec{r}_1 = \vec{i}(x + \frac{x}{2}) + \vec{j}(y + \frac{y}{2}) + \vec{k}(z + \frac{z}{2}) \quad (\text{A1-6})$$

$$\vec{r}_2 = \vec{R} - \frac{\vec{r}}{2} \quad (\text{A1-7})$$

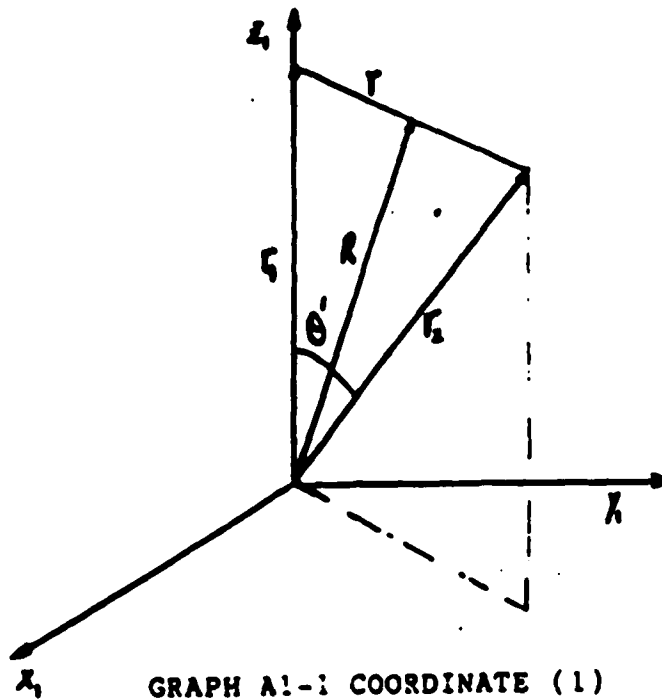
$$\vec{r}_2 = \vec{i}(x - \frac{x}{2}) + \vec{j}(y - \frac{y}{2}) + \vec{k}(z - \frac{z}{2}) \quad (\text{A1-8})$$

$$R^2 = \frac{1}{4} (r_1^2 + r_2^2 + 2 r_1 r_2 \cos \theta') \quad (\text{A1-9})$$

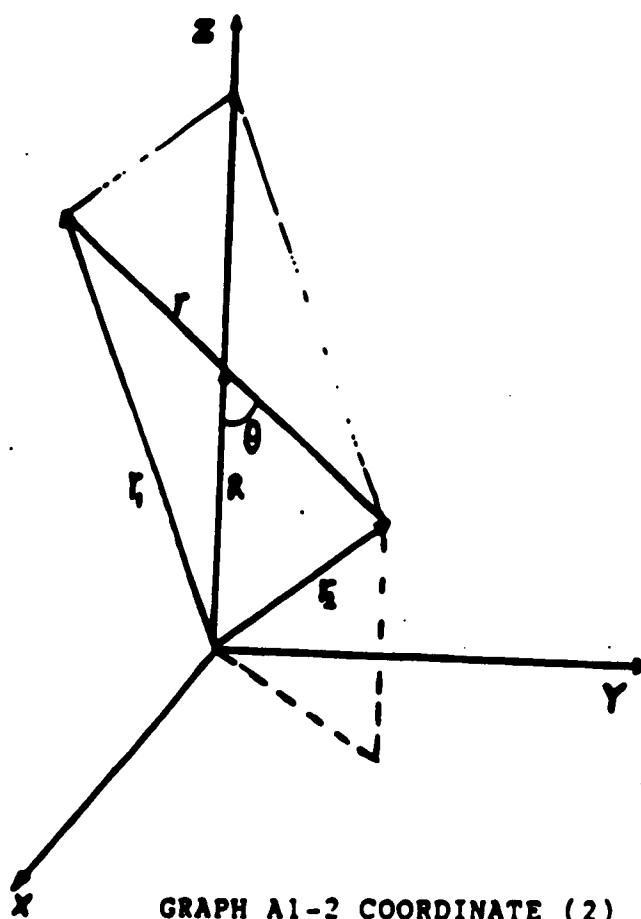
$$r^2 = r_1^2 + r_2^2 - 2 r_1 r_2 \cos \theta' \quad (\text{A1-10})$$

$$r_1^2 = R^2 + \frac{r^2}{4} + R r \cos \theta \quad (\text{A1-11})$$

$$r_2^2 = R^2 + \frac{r^2}{4} - R r \cos \theta \quad (\text{A1-12})$$



GRAPH A1-1: COORDINATE (1)



GRAPH A1-2 COORDINATE (2)

## (2) 3 DIMENSIONAL SIMPSON NUMERICAL INTEGRATION

Simpson numerical integration with variable step size for one dimension is discussed in what follows

$$I = \int_a^b f(x) dx \quad (A1-13)$$

One may start with

$$T_1 = \frac{1}{2} [f(a) + f(b)] \cdot (b - a) \quad (A1-14)$$

Then the area may be separated according to following intervals

$$h_k = \frac{b-a}{2^k} \quad k = 1, 2, 3, \dots \quad (A1-15)$$

For each interval, half of the area is improved by the formula:

$$T_{2^k} = \frac{1}{2} T_{2^{k-1}} + h_k \sum_{i=1}^{2^{k-1}} f(x_i) \quad (A1-16)$$

where

$$x_i = a + h_i (2i-1) \quad (A1-17)$$

Finally the results of the integration could come from the summation

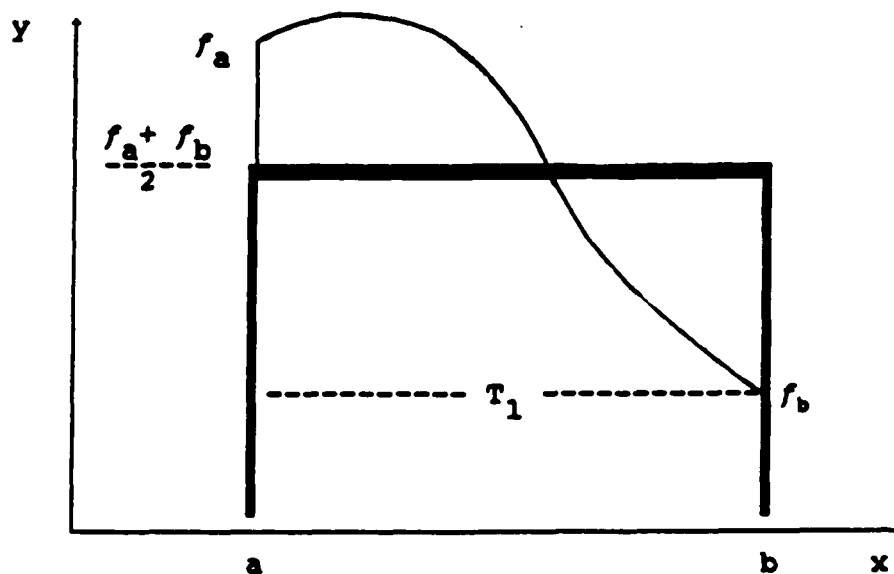
$$S_{2^k} = \frac{1}{3} (4T_{2^k} - T_{2^{k-1}}) \quad (A1-18)$$

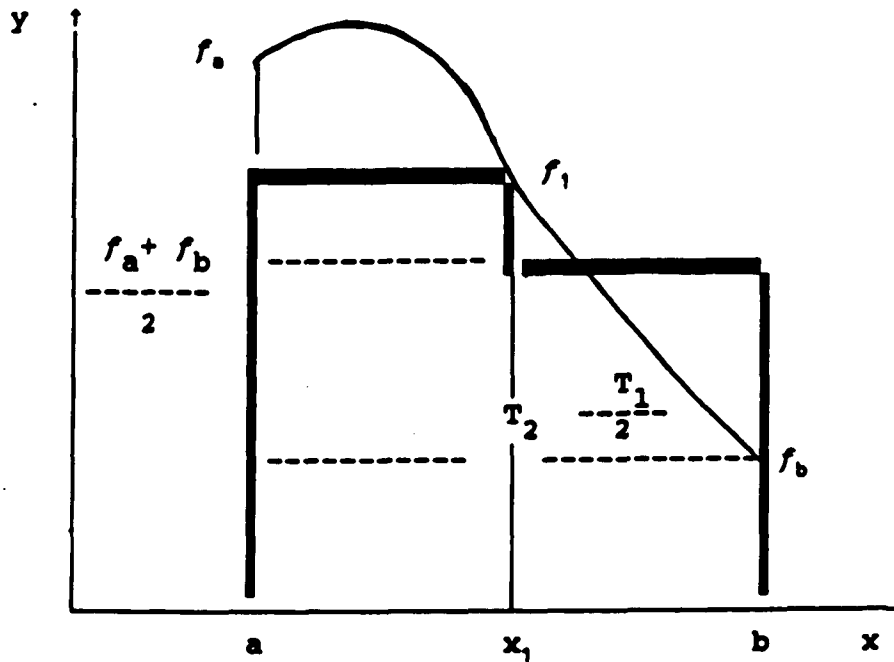
The calculation is stopped by a small enough value Epsilon  $\epsilon$  being reached. When

$$|S_{2k} - S_{2k-1}| \leq \epsilon \cdot (1 + |S_{2k}|) \quad (\text{A1-19})$$

the integration is completed.

The procedure is explained by the following graphs:





$T_2$  includes a half of  $T_1$  and  $f(x) \cdot \frac{1}{2} (b - a)$ .

More points may be used to separate  $T_1$  into more pieces, with 50% of the pieces improved by  $f(x_i)$  each time. This is the basic idea of the Simpson method with variable steps.

Three dimensional integration by the Simpson method connect 3 single Simpson integrations. We have the integration

$$E = \int_0^\pi \sin\theta \, d\theta \int_0^\infty r^2 \, dr \int_0^\infty f(R, r, \cos\theta) R^2 \, dR \quad (\text{A1-20})$$

The integration over  $\theta$  may be transformed to integration over

$\cos\theta$ , this is convenient for the integration of our function  $f$ , because our  $f$  only contains  $\cos\theta$ .

$$\int_0^\pi \sin\theta \, d\theta = \int_1^{-1} -d\cos\theta = \int_{-1}^1 d\cos\theta \quad (\text{A1-21})$$

The integration over  $R$  is transformed also

$$\int_0^\infty f R^2 \, dR = \int_0^1 f R^2 \, dR + \int_1^\infty f R^2 \, dR \quad (\text{A1-22})$$

Put

$$\tau = \frac{1}{R} \quad \text{i.e.} \quad R = \frac{1}{\tau} \quad (\text{A1-23})$$

$$dR = d\left(\frac{1}{\tau}\right) = -\frac{1}{\tau^2} d\tau \quad (\text{A1-24})$$

$$R^2 \, dR = -\frac{1}{\tau^3} d\tau \quad (\text{A1-25})$$

When  $R \rightarrow 1$ ,  $\tau \rightarrow 1$ ; when  $R \rightarrow \infty$ ,  $\tau \rightarrow 0$ , so we have

$$\begin{aligned} \int_1^\infty f(R, r, \cos\theta) R^2 \, dR &= \int_1^0 f(\tau, r, \cos\theta) \left(-\frac{1}{\tau^3}\right) d\tau \\ &= \int_0^1 f(\tau, r, \cos\theta) \frac{1}{\tau^3} d\tau \end{aligned} \quad (\text{A1-26})$$

The entire integration over  $R$  could be performed as

$$\int_0^\infty f(R, r, \cos\theta) R^2 \, dR = \quad (\text{A1-27})$$

$$\int_0^1 f(R, r, \cos\theta) R^2 \, dR + \int_0^1 f(\tau, r, \cos\theta) \frac{1}{\tau^3} d\tau$$

This formula is convenient for programming as it removes the

infinity limit of integration. The integration over  $r$  is transformed in the same way. Suppose

$$\mu = \frac{1}{r} \quad \text{i.e.} \quad r = \frac{1}{\mu} \quad (\text{A1-28})$$

$$d r = d \frac{1}{\mu} = -\frac{1}{\mu^2} d\mu \quad (\text{A1-29})$$

$$\int_0^{\infty} f(R, r, \cos\theta) r^2 dr = \quad (\text{A1-30})$$

$$\int_0^1 f(R, r, \cos\theta) r^2 dr + \int_0^1 f(\mu, r, \cos\theta) -\frac{1}{\mu^2} d\mu$$

In the 3D Simpson integration, put a value  $[\cos\theta]$ , and a value of  $[r]_1$  into the integration over  $R$ . When the integration over  $R$  under  $[\cos\theta]_1$  and  $[r]_1$  is completed, move  $[r]_1$  to  $[r]_2$  according to the Simpson procedure, and do the integration over  $R$  under the  $[\cos\theta]_1$  and  $[r]_2$  ..... Repeat this until the integration over  $r$  is completed, then move  $[\cos\theta]_1$  to  $[\cos\theta]_2$  according to the Simpson procedure, continue the integration until the integration over  $\cos\theta$  is completed. A computer program for the 3D Simpson integration has been written and applied in this thesis.

By the way, the other 3 dimensions  $(\theta', \phi, \phi')$  which are independent of the function  $f$ , may be dealt with easily as follows

$$\int_0^{\pi} \sin\theta' d\theta' \int_0^{2\pi} d\phi' \int_0^{2\pi} d\phi = -\cos\theta' \Big|_0^{\pi} \cdot 2\pi \cdot 2\pi = 8\pi^2 \quad (\text{A1-31})$$

## APPENDIX (2) COLLE-SALVITTE NUMERICAL FORMULA

## (1) THE NUMERICAL FORMULA

$$E_c = -0.04918 \int \frac{p(R) \left(1 + 0.173W * e^{\frac{-0.58}{B}}\right)}{1 + \frac{0.8}{B}} d\vec{R} \quad (A2-1)$$

where  $d\vec{R}$  means a volume element:  $R^2 dR \cdot \sin\theta d\theta \cdot d\phi$

$$W = 0.3814 P^{\frac{-8}{3}}(R) [\nabla^2 P_2(1122)]_{r=0} \quad (A2-2)$$

$$\nabla^2 = \frac{\partial^2}{\partial x^2} + \frac{\partial^2}{\partial y^2} + \frac{\partial^2}{\partial z^2} \quad (A2-3)$$

$$r^2 = x^2 + y^2 + z^2 = |\vec{r}_1 - \vec{r}_2|^2 \quad (A2-4)$$

$P_2$  is the second order density matrix with spinless

$$P_2(1212) = p(11)p(22) - \frac{1}{2} p(12)p(21) \quad (A2-5)$$

$p$  is the first order density matrix.

(2). Hartree-Fock orbitals are taken from Clementi HF orbital table [57].

For He and its isoelectronic series

$$p(11) = 2 \phi(1)\phi(1) \quad p(12) = 2 \phi(1)\phi(2)$$

$$p(21) = 2 \phi(2)\phi(1) \qquad p(22) = 2 \phi(2)\phi(2) \qquad (A2-6)$$

where  $\phi$  is a HF 1s orbital function

$$\phi(1) = \sum_{i=1}^4 C_i(1s) * A_i(1s) * e^{-\alpha_i r_1} \qquad (A) (A2-7)$$

$$\phi(2) = \sum_{i=1}^4 C_i(1s) * A_i(1s) * e^{-\alpha_i r_2} \qquad (B) (A2-7)$$

$C_i(1s)$  are the coefficients of the 1s orbital, and the normalization constant  $A_i(1s)$  is

$$A_i(1s) = \left( \frac{\alpha_i^3}{\pi} \right)^{1/2} \qquad (A2-8)$$

For Be and its isoelectronic series

$$p(11) = 2 \mu(1)\mu(1) + 2 \Omega(1)\Omega(1) \qquad (A) (A2-9)$$

$$p(12) = 2 \mu(1)\mu(2) + 2 \Omega(1)\Omega(2) = p(21) \qquad (B) (A2-9)$$

$$p(22) = 2 \mu(2)\mu(2) + 2 \Omega(2)\Omega(2) \qquad (C) (A2-9)$$

where  $\mu$  represents HF 1s orbitals of the Be series,  $\Omega$  represents HF 2s orbitals of the Be series

$$\mu(1) = \sum_{i=1}^2 C_i(1s) * A_i(1s) * e^{-\alpha_i r_1} + r_1 * \sum_{j=3}^5 C_j(1s) * A_j(1s) * e^{-\alpha_j r_1} \qquad (A) (A2-10)$$

$$\mu(2) = \sum_{i=1}^2 C_i(1s) * A_i(1s) * e^{-\alpha_i r_2} + r_2 * \sum_{j=3}^5 C_j(1s) * A_j(1s) * e^{-\alpha_j r_2}$$

(B) (A2-10)

$$\Omega(1) = \sum_{i=1}^2 C_i(2s) * A_i(2s) * e^{-\alpha_i r_1} + r_1 * \sum_{j=3}^5 C_j(2s) * A_j(2s) * e^{-\alpha_j r_1}$$

(A) (A2-11)

$$\Omega(2) = \sum_{i=1}^2 C_i(2s) * A_i(2s) * e^{-\alpha_i r_2} + r_2 * \sum_{j=3}^5 C_j(2s) * A_j(2s) * e^{-\alpha_j r_2}$$

(B) (A2-11)

where

$$A_i(1s) = \left( \frac{\alpha_i^3}{\pi} \right)^{\frac{1}{2}}$$

(A) (A2-12)

$$A_i(2s) = \left( \frac{\alpha_i^3}{\pi} \right)^{\frac{1}{2}}$$

(B) (A2-12)

$$A_j(1s) = \left( \frac{\alpha_j^5}{3\pi} \right)^{\frac{1}{2}}$$

(A) (A2-13)

$$A_j(2s) = \left( \frac{\alpha_j^5}{3\pi} \right)^{\frac{1}{2}}$$

(B) (A2-13)

For Ne and its isoelectronic series

$$p(11) = 2 \mu(1)\mu(1) + 2 \Omega(1)\Omega(1)$$

$$+ 2 \sigma_x(1)\sigma_x(1) + 2 \sigma_y(1)\sigma_y(1) + 2 \sigma_z(1)\sigma_z(1)$$

$$p(12) = 2 \mu(1)\mu(2) + 2 \Omega(1)\Omega(2) \quad (\text{A}) (\text{A2-14})$$

$$+ 2 \sigma_x(1)\sigma_x(2) + 2 \sigma_y(1)\sigma_y(2) + 2 \sigma_z(1)\sigma_z(2) \quad (\text{B}) (\text{A2-14})$$

$$p(22) = 2 \mu(2)\mu(2) + 2 \Omega(2)\Omega(2) \\ + 2 \sigma_x(2)\sigma_x(2) + 2 \sigma_y(2)\sigma_y(2) + 2 \sigma_z(2)\sigma_z(2) \quad (\text{C}) (\text{A2-14})$$

where  $\mu$  and  $\Omega$  have the same form but with different coefficients than in the Be series.  $\sigma_x, \sigma_y, \sigma_z$  represent the 2p orbitals of HF. They are

$$\sigma_x(1) = X_1 * \sum_{k=1}^4 C_k(2P) * A_k(2P) * e^{-\alpha_k r_1} \quad (\text{A}) (\text{A2-15})$$

$$\sigma_y(1) = Y_1 * \sum_{k=1}^4 C_k(2P) * A_k(2P) * e^{-\alpha_k r_1} \quad (\text{B}) (\text{A2-15})$$

$$\sigma_z(1) = Z_1 * \sum_{k=1}^4 C_k(2P) * A_k(2P) * e^{-\alpha_k r_1} \quad (\text{C}) (\text{A2-15})$$

$$\sigma_x(2) = X_2 * \sum_{k=1}^4 C_k(2P) * A_k(2P) * e^{-\alpha_k r_2} \quad (\text{A}) (\text{A2-16})$$

$$\sigma_y(2) = Y_2 * \sum_{k=1}^4 C_k(2P) * A_k(2P) * e^{-\alpha_k r_2} \quad (\text{B}) (\text{A2-16})$$

$$\sigma_z(2) = Z_2 * \sum_{k=1}^4 C_k(2P) * A_k(2P) * e^{-\alpha_k r_2} \quad (\text{C}) (\text{A2-16})$$

where  $C_k(2P)$  and  $\alpha_k$  are the HF orbital coefficients [57].

And the normalization constant  $A_k(2P)$  should be

$$A_k(2P) = \left( -\frac{\alpha_k^5}{\pi} \right)^{\frac{1}{2}} \quad (\text{A2-17})$$

### (3). DERIVATIVE

The second order derivative  $\nabla^2 P_2(1212)$  are

$$\begin{aligned} \nabla^2 P_2(1212) &= p(11) \cdot \nabla^2 p(22) + p(22) \cdot \nabla^2 p(11) + 2 \nabla p(11) \cdot \nabla p(22) \\ &- \frac{1}{2} p(12) \cdot \nabla^2 p(21) - \frac{1}{2} p(21) \cdot \nabla^2 p(12) - \nabla p(12) \cdot \nabla p(21) \end{aligned} \quad (\text{A2-18})$$

The basic vector relations are

$$\vec{r}_1 = \left( \vec{R} + \frac{\vec{r}}{2} \right) \quad \vec{r}_2 = \left( \vec{R} - \frac{\vec{r}}{2} \right) \quad (\text{A}) (\text{A2-19})$$

$$\vec{R} = \frac{1}{2} (\vec{r}_1 + \vec{r}_2) = \vec{i} x + \vec{j} y + \vec{k} z \quad (\text{B}) (\text{A2-19})$$

$$\vec{r} = \vec{r}_1 - \vec{r}_2 = \vec{i} \cdot x + \vec{j} \cdot y + \vec{k} \cdot z \quad (\text{C}) (\text{A2-19})$$

where  $\vec{i}, \vec{j}, \vec{k}$  are the unit vectors in the cartesian coordinate of Appendix 1. According to the above relationship we have

$$\vec{r}_1 = \vec{i} \left( x + \frac{x}{2} \right) + \vec{j} \left( y + \frac{y}{2} \right) + \vec{k} \left( z + \frac{z}{2} \right) \quad (\text{A}) (\text{A2-20})$$

$$\vec{r}_2 = \vec{i} \left( x - \frac{x}{2} \right) + \vec{j} \left( y - \frac{y}{2} \right) + \vec{k} \left( z - \frac{z}{2} \right) \quad (\text{B}) (\text{A2-20})$$

$$\begin{aligned} r_1^2 &= x_1^2 + y_1^2 + z_1^2 \quad (\text{A}) (\text{A2-21}) \\ &= \left( x + \frac{x}{2} \right)^2 + \left( y + \frac{y}{2} \right)^2 + \left( z + \frac{z}{2} \right)^2 \end{aligned}$$

$$\begin{aligned} r_2^2 &= x_2^2 + y_2^2 + z_2^2 \quad (\text{B}) (\text{A2-21}) \\ &= \left( x - \frac{x}{2} \right)^2 + \left( y - \frac{y}{2} \right)^2 + \left( z - \frac{z}{2} \right)^2 \end{aligned}$$

The derivative of basis functions are:

$$\begin{aligned} \nabla r &= \left[ \vec{i} \frac{\partial}{\partial x} + \vec{j} \frac{\partial}{\partial y} + \vec{k} \frac{\partial}{\partial z} \right] (x^2 + y^2 + z^2)^{\frac{1}{2}} \\ &= -\frac{\vec{r}}{r} \quad (\text{A}) (\text{A2-22}) \end{aligned}$$

$$\begin{aligned} \nabla r_1 &= \left[ \vec{i} \frac{\partial}{\partial x} + \vec{j} \frac{\partial}{\partial y} + \vec{k} \frac{\partial}{\partial z} \right] \left( \left( x + \frac{x}{2} \right)^2 + \left( y + \frac{y}{2} \right)^2 + \left( z + \frac{z}{2} \right)^2 \right)^{\frac{1}{2}} \\ &= -\frac{\vec{r}_1}{2r_1} \quad (\text{B}) (\text{A2-22}) \end{aligned}$$

$$\begin{aligned} \nabla r_2 &= \left[ \vec{i} \frac{\partial}{\partial x} + \vec{j} \frac{\partial}{\partial y} + \vec{k} \frac{\partial}{\partial z} \right] \left( \left( x - \frac{x}{2} \right)^2 + \left( y - \frac{y}{2} \right)^2 + \left( z - \frac{z}{2} \right)^2 \right)^{\frac{1}{2}} \\ &= -\frac{\vec{r}_2}{2r_2} \quad (\text{C}) (\text{A2-22}) \end{aligned}$$

The second order derivatives are

$$\nabla^2 r = \nabla \cdot \frac{\vec{r}}{r} = \frac{2}{r} \quad (\text{A}) \text{ (A2-23)}$$

$$\nabla^2 r_1 = \nabla \cdot \frac{\vec{r}_1}{2r_1} = \frac{1}{2r_1} \quad (\text{B}) \text{ (A2-23)}$$

$$\nabla^2 r_2 = \nabla \cdot \frac{-\vec{r}_2}{2r_2} = \frac{1}{2r_2} \quad (\text{C}) \text{ (A2-23)}$$

(A) For He isoelectronic series ( $1s^2$ ):  
The first order derivatives are

$$\nabla p(11) = 4 \phi(1) \cdot \nabla \phi(1) \quad (\text{A}) \text{ (A2-24)}$$

$$\nabla p(22) = 4 \phi(2) \cdot \nabla \phi(2) \quad (\text{B}) \text{ (A2-24)}$$

$$\nabla p(12) = 2 \phi(1) \cdot \nabla \phi(2) + 2 \phi(2) \cdot \nabla \phi(1) \quad (\text{C}) \text{ (A2-24)}$$

$$\begin{aligned} \nabla \phi(1) &= \frac{\vec{r}_1}{2r_1} \sum_{i=1}^4 C_i(1s) * A_i(1s) * (-\alpha_i) * e^{-\alpha_i r_1} \\ &= \frac{\vec{r}_1}{2r_1} \bar{B}_1(1s) \end{aligned} \quad (\text{A2-25})$$

where

$$\bar{B}_1(1s) = \sum_{i=1}^4 C_i(1s) * A_i(1s) * (-\alpha_i) * e^{-\alpha_i r_1} \quad (\text{A2-26})$$

$$\begin{aligned} \nabla\phi(2) &= -\frac{\vec{r}_2}{2r_2} \sum_{i=1}^4 C_i(1s) * A_i(1s) * (-\alpha_i) * e^{-\alpha_i r_2} \\ &= -\frac{\vec{r}_2}{2r_2} \bar{B}_2(1s) \end{aligned} \quad (\text{A2-27})$$

where

$$\bar{B}_2(1s) = \sum_{i=1}^4 C_i(1s) * A_i(1s) * (-\alpha_i) * e^{-\alpha_i r_2} \quad (\text{A2-28})$$

$$\nabla\phi(1) \cdot \nabla\phi(1) = -\frac{1}{4} [\bar{B}_1(1s)]^2 \quad (\text{A}) (\text{A2-29})$$

$$\nabla\phi(2) \cdot \nabla\phi(2) = -\frac{1}{4} [\bar{B}_2(1s)]^2 \quad (\text{B}) (\text{A2-29})$$

$$\nabla\phi(1) \cdot \nabla\phi(2) = -\frac{1}{4} \frac{\vec{r}_1}{r_1} \frac{\vec{r}_2}{r_2} \bar{B}_1(1s) \cdot \bar{B}_2(1s) \quad (\text{C}) (\text{A2-29})$$

Second order derivatives

$$\begin{aligned} \nabla^2 p(11) &= \nabla (4 \phi(1) \nabla\phi(1)) \\ &= 4 \phi(1) \nabla^2 \phi(1) + 4 \nabla\phi(1) \nabla\phi(1) \end{aligned} \quad (\text{A}) (\text{A2-30})$$

$$\begin{aligned} \nabla^2 p(22) &= \nabla (4 \phi(2) \nabla\phi(2)) \\ &= 4 \phi(2) \nabla^2 \phi(2) + 4 \nabla\phi(2) \nabla\phi(2) \end{aligned} \quad (\text{B}) (\text{A2-30})$$

$$\begin{aligned} \nabla^2 p(12) &= \nabla (2\phi(1) \nabla\phi(2) + 2\phi(2) \nabla\phi(1)) \\ &= 2\phi(1) \nabla^2 \phi(2) + 2\phi(2) \nabla^2 \phi(1) + 4 \nabla\phi(1) \nabla\phi(2) \\ &= \nabla^2 p(21) \end{aligned} \quad (\text{A2-31})$$

$$\begin{aligned}
\nabla^2 \psi(1) &= \nabla \left( \frac{\vec{r}_1}{2r_1} \bar{B}_1(1s) \right) \\
&= \bar{B}_1(1s) \nabla \frac{\vec{r}_1}{2r_1} + \frac{\vec{r}_1}{2r_1} \nabla \bar{B}_1(1s) \\
&= \frac{1}{2r_1} \bar{B}_1(1s) + -\frac{1}{4} \bar{B}_1(1s)
\end{aligned} \tag{A2-32}$$

where

$$\bar{B}_1(1s) = \sum_{i=1}^4 C_i(1s) * A_i(1s) * (-\alpha_i)^2 * e^{-\alpha_i r_1} \tag{A2-33}$$

$$\begin{aligned}
\nabla^2 \psi(2) &= \nabla \left( \frac{-\vec{r}_2}{2r_2} \bar{B}_2(1s) \right) \\
&= \bar{B}_2(1s) \nabla \frac{-\vec{r}_2}{2r_2} + \frac{-\vec{r}_2}{2r_2} \nabla \bar{B}_2(1s) \\
&= \frac{1}{2r_2} \bar{B}_2(1s) + -\frac{1}{4} \bar{B}_2(1s)
\end{aligned} \tag{A2-34}$$

where

$$\bar{B}_2(1s) = \sum_{i=1}^4 C_i(1s) * A_i(1s) * (-\alpha_i)^2 * e^{-\alpha_i r_2} \tag{A2-35}$$

(B) For Be isoelectronic series ( $1s^2 2s^2$ ):

First order derivatives are:

$$\nabla p(11) = 4 \mu(1) \nabla \mu(1) + 4 \Omega(1) \nabla \Omega(1) \tag{A) (A2-36)}$$

$$\nabla p(22) = 4 \mu(2) \nabla \mu(2) + 4 \Omega(2) \nabla \Omega(2) \quad (\text{B}) (\text{A2-36})$$

$$\begin{aligned} \nabla p(12) &= 2 \mu(1) \nabla \mu(2) + 2 \mu(2) \nabla \mu(1) \\ &+ 2 \Omega(1) \nabla \Omega(2) + 2 \Omega(2) \nabla \Omega(1) \end{aligned} \quad (\text{C}) (\text{A2-36})$$

$$\nabla \mu(1) = \frac{\vec{r}_1}{2r_1} ( \bar{B}_1(1s) + D_1(1s) + r_1 * \bar{D}_1(1s) ) \quad (\text{A2-37})$$

where

$$\bar{B}_1(1s) = \sum_{i=1}^2 C_i(1s) * A_i(1s) * (-\alpha_i) * e^{-\alpha_i r_1} \quad (\text{A2-38})$$

$$D_1(1s) = \sum_{j=3}^5 C_j(1s) * A_j(1s) * e^{-\alpha_j r_1} \quad (\text{A2-39})$$

$$\bar{D}_1(1s) = \sum_{j=3}^5 C_j(1s) * A_j(1s) * (-\alpha_j) * e^{-\alpha_j r_1} \quad (\text{A2-40})$$

$$\nabla \mu(2) = \frac{-\vec{r}_2}{2r_2} ( \bar{B}_2(1s) + D_2(1s) + r_2 * \bar{D}_2(1s) ) \quad (\text{A2-41})$$

where

$$\bar{B}_2(1s) = \sum_{i=1}^2 C_i(1s) * A_i(1s) * (-\alpha_i) * e^{-\alpha_i r_2} \quad (\text{A2-42})$$

$$D_2(1s) = \sum_{j=3}^5 C_j(1s) * A_j(1s) * e^{-\alpha_j r_2} \quad (\text{A2-43})$$

$$\bar{D}_2(1s) = \sum_{j=3}^5 C_j(1s) * A_j(1s) * (-\alpha_j) * e^{-\alpha_j r_2} \quad (\text{A2-44})$$

$$\nabla\Omega(1) = \frac{\vec{r}_1}{2r_1} ( \bar{B}_1(2s) + D_1(2s) + r_1 * \bar{D}_1(2s) ) \quad (\text{A2-45})$$

where

$$\bar{B}_1(2s) = \sum_{i=1}^2 C_i(2s) * A_i(2s) * (-\alpha_i) * e^{-\alpha_i r_1} \quad (\text{A2-46})$$

$$D_1(2s) = \sum_{j=3}^5 C_j(2s) * A_j(2s) * e^{-\alpha_j r_1} \quad (\text{A2-47})$$

$$\bar{D}_1(2s) = \sum_{j=3}^5 C_j(2s) * A_j(2s) * (-\alpha_j) * e^{-\alpha_j r_1} \quad (\text{A2-48})$$

$$\nabla\Omega(2) = -\frac{\vec{r}_2}{2r_2} ( \bar{B}_2(2s) + D_2(2s) + r_2 * \bar{D}_2(2s) ) \quad (\text{A2-49})$$

where

$$\bar{B}_2(2s) = \sum_{i=1}^2 C_i(2s) * A_i(2s) * (-\alpha_i) * e^{-\alpha_i r_2} \quad (\text{A2-50})$$

$$D_2(2s) = \sum_{j=3}^5 C_j(2s) * A_j(2s) * e^{-\alpha_j r_2} \quad (\text{A2-51})$$

$$\bar{D}_2(2s) = \sum_{j=3}^5 C_j(2s) * A_j(2s) * (-\alpha_j) * e^{-\alpha_j r_2} \quad (\text{A2-52})$$

The second order derivatives for the Be series are

$$\begin{aligned} \nabla^2 p(11) &= \nabla ( 4 \mu(1) \nabla\mu(1) + 4 \Omega(1) \nabla\Omega(1) ) \\ &= 4 \mu(1) \nabla^2 \mu(1) + 4 \nabla\mu(1) \nabla\mu(1) \\ &\quad + 4 \Omega(1) \nabla^2 \Omega(1) + 4 \nabla\Omega(1) \nabla\Omega(1) \end{aligned} \quad (\text{A2-53})$$

$$\begin{aligned}
\nabla^2 p(22) &= \nabla ( 4 \mu(2) \nabla \mu(2) + 4 \Omega(2) \nabla \Omega(2) ) \\
&= 4 \mu(2) \nabla^2 \mu(2) + 4 \nabla \mu(2) \nabla \mu(2) \\
&+ 4 \Omega(2) \nabla^2 \Omega(2) + 4 \nabla \Omega(2) \nabla \Omega(2) \quad (B) (A2-53)
\end{aligned}$$

$$\begin{aligned}
\nabla^2 p(12) &= \nabla ( 2\mu(1) \nabla \mu(2) + 2\mu(2) \nabla \mu(1) \\
&+ 2\Omega(1) \nabla \Omega(2) + 2\Omega(2) \nabla \Omega(1) ) \\
&= 2\mu(1) \nabla^2 \mu(2) + 2\mu(2) \nabla^2 \mu(1) + 4 \nabla \mu(1) \nabla \mu(2) \\
&+ 2\Omega(1) \nabla^2 \Omega(2) + 2\Omega(2) \nabla^2 \Omega(1) + 4 \nabla \Omega(1) \nabla \Omega(2) \quad (A2-54)
\end{aligned}$$

$$\nabla^2 p(12) = \nabla^2 p(21) \quad (A2-55)$$

$$\begin{aligned}
\nabla^2 \mu(1) &= \nabla \left( \frac{\vec{r}_1}{2r_1} [ \bar{B}_1(1s) + D_1(1s) + r_1 \bar{D}_1(1s) ] \right) \\
&= -\frac{2}{r_1} [ \bar{B}_1(1s) + D_1(1s) + r_1 \bar{D}_1(1s) ] \\
&+ \frac{1}{4} [ \bar{B}_1(1s) + 2\bar{D}_1(1s) + r_1 \bar{D}_1(1s) ] \quad (A2-56)
\end{aligned}$$

where

$$\bar{B}_1(1s) = \sum_{i=1}^2 C_i(1s) * A_i(1s) * (-\alpha_i)^2 * e^{-\alpha_i r_1} \quad (A2-57)$$

$$\bar{D}_1(1s) = \sum_{j=3}^5 C_j(1s) * A_j(1s) * (-\alpha_j)^2 * e^{-\alpha_j r_1} \quad (A2-58)$$

$$D_1(zs) = \sum_{j=1}^3 C_j(zs) * A_j(zs) * (-a_j)^{-1} * e^{-a_j r_1} \quad (A2-64)$$

$$B_1(zs) = \sum_{j=1}^3 C_j(zs) * A_j(zs) * (-a_j)^{-1} * e^{-a_j r_1} \quad (A2-63)$$

where

$$+ \frac{1}{r_1} [ B_1(zs) + 2D_1(zs) + r_1 * D_1(zs) ] \quad (A2-62)$$

$$= -\frac{r_1}{2} [ B_1(zs) + D_1(zs) + r_1 * D_1(zs) ]$$

$$u(1) = \frac{r_1}{2} [ B_1(zs) + D_1(zs) + r_1 * D_1(zs) ]$$

$$D_2(1s) = \sum_{j=1}^3 C_j(1s) * A_j(1s) * (-a_j)^{-1} * e^{-a_j r_2} \quad (A2-61)$$

$$B_2(1s) = \sum_{j=1}^3 C_j(1s) * A_j(1s) * (-a_j)^{-1} * e^{-a_j r_2} \quad (A2-60)$$

where

$$+ \frac{1}{r_2} [ B_2(1s) + 2D_2(1s) + r_2 * D_2(1s) ] \quad (A2-59)$$

$$= -\frac{r_2}{2} [ B_2(1s) + D_2(1s) + r_2 * D_2(1s) ]$$

$$u(2) = \frac{r_2}{2} [ B_2(1s) + D_2(1s) + r_2 * D_2(1s) ]$$

$$\begin{aligned}
\nabla^2 \Omega(2) &= \nabla \left( \frac{-\vec{r}_2}{2r_2^2} [ \bar{B}_2(2s) + D_2(2s) + r_2 * \bar{D}_2(2s) ] \right) \\
&= -\frac{2}{r_2^2} [ \bar{B}_2(2s) + D_2(2s) + r_2 * \bar{D}_2(2s) ] \\
&\quad + \frac{1}{4} [ \bar{B}_2(2s) + 2\bar{D}_2(2s) + r_2 * \bar{D}_2(2s) ]
\end{aligned} \tag{A2-65}$$

where

$$\bar{B}_2(2s) = \sum_{i=1}^2 C_i(2s) * A_i(2s) * (-\alpha_i)^2 * e^{-\alpha_i r_2} \tag{A2-66}$$

$$\bar{D}_2(2s) = \sum_{j=3}^5 C_j(2s) * A_j(2s) * (-\alpha_j)^2 * e^{-\alpha_j r_2} \tag{A2-67}$$

(C) For the Ne series ( $1s^2 2s^2 2p^6$ )

The first order derivatives of the Ne series are:

$$\begin{aligned}
\nabla p(11) &= 4 \mu(1) \nabla \mu(1) + 4 \Omega(1) \nabla \Omega(1) & (A) \text{ (A2-68)} \\
&\quad + 4 \sigma_x(1) \nabla \sigma_x(1) + 4 \sigma_y(1) \nabla \sigma_y(1) + 4 \sigma_z(1) \nabla \sigma_z(1)
\end{aligned}$$

$$\begin{aligned}
\nabla p(22) &= 4 \mu(2) \nabla \mu(2) + 4 \Omega(2) \nabla \Omega(2) & (B) \text{ (A2-68)} \\
&\quad + 4 \sigma_x(2) \nabla \sigma_x(2) + 4 \sigma_y(2) \nabla \sigma_y(2) + 4 \sigma_z(2) \nabla \sigma_z(2)
\end{aligned}$$

$$\begin{aligned}
\nabla p(12) &= 2 \mu(1) \nabla \mu(2) + 2 \mu(2) \nabla \mu(1) \\
&\quad + 2 \Omega(1) \nabla \Omega(2) + 2 \Omega(2) \nabla \Omega(1) \\
&\quad + 2 \sigma_x(1) \nabla \sigma_x(2) + 2 \sigma_y(1) \nabla \sigma_y(2) + 2 \sigma_z(1) \nabla \sigma_z(2) \\
&\quad + 2 \sigma_x(2) \nabla \sigma_x(1) + 2 \sigma_y(12) \nabla \sigma_y(1) + 2 \sigma_z(2) \nabla \sigma_z(1)
\end{aligned} \tag{A2-69}$$

$$\begin{aligned} \nabla \sigma_x(1) &= \nabla \left( x_1 * \sum_{k=1}^4 C_k(P) * A_k(P) * e^{-\alpha_k r_1} \right) \\ &= -\frac{\vec{1}}{2} F_1(P) + x_1 * \frac{\vec{r}_1}{2r_1} \bar{F}_1(P) \end{aligned} \quad (\text{A2-70})$$

where

$$F_1(P) = \sum_{k=1}^4 C_k(P) * A_k(P) * e^{-\alpha_k r_1} \quad (\text{A2-71})$$

$$\bar{F}_1(P) = \sum_{k=1}^4 C_k(P) * A_k(P) * (-\alpha_k) * e^{-\alpha_k r_1} \quad (\text{A2-72})$$

$$\begin{aligned} \nabla \sigma_y(1) &= \nabla \left( y_1 * \sum_{k=1}^4 C_k(P) * A_k(P) * e^{-\alpha_k r_1} \right) \\ &= -\frac{\vec{1}}{2} F_1(P) + y_1 * \frac{\vec{r}_1}{2r_1} \bar{F}_1(P) \end{aligned} \quad (\text{A2-73})$$

$$\begin{aligned} \nabla \sigma_z(1) &= \nabla \left( z_1 * \sum_{k=1}^4 C_k(P) * A_k(P) * e^{-\alpha_k r_1} \right) \\ &= -\frac{\vec{k}}{2} F_1(P) + z_1 * \frac{\vec{r}_1}{2r_1} \bar{F}_1(P) \end{aligned} \quad (\text{B2-73})$$

$$\begin{aligned} \nabla \sigma_x(2) &= \nabla \left( x_2 * \sum_{k=1}^4 C_k(P) * A_k(P) * e^{-\alpha_k r_2} \right) \\ &= -\frac{\vec{1}}{2} F_2(P) - x_2 * \frac{\vec{r}_2}{2r_2} \bar{F}_2(P) \end{aligned} \quad (\text{A2-74})$$

where

$$F_2(P) = \sum_{k=1}^4 C_k(P) * A_k(P) * e^{-\alpha_k r_2} \quad (A2-75)$$

$$\bar{F}_2(P) = \sum_{k=1}^4 C_k(P) * A_k(P) * (-\alpha_k) * e^{-\alpha_k r_2} \quad (A2-76)$$

$$\begin{aligned} \nabla \sigma_y(2) &= \nabla \left( y_2 * \sum_{k=1}^4 C_k(P) * A_k(P) * e^{-\alpha_k r_2} \right) \\ &= -\frac{\vec{y}}{2} F_2(P) - y_2 * \frac{\vec{r}_2}{2r_2} \bar{F}_2(P) \end{aligned} \quad (A) (A2-77)$$

$$\begin{aligned} \nabla \sigma_z(2) &= \nabla \left( z_2 * \sum_{k=1}^4 C_k(P) * A_k(P) * e^{-\alpha_k r_2} \right) \\ &= -\frac{\vec{z}}{2} F_2(P) - z_2 * \frac{\vec{r}_2}{2r_2} \bar{F}_2(P) \end{aligned} \quad (B) (A2-77)$$

The second order derivatives for the 2p orbitals of the Ne series are

$$\begin{aligned} \nabla^2 \sigma_x(1) &= \nabla \left( -\frac{\vec{x}}{2} F_1(P) + x_1 * \frac{\vec{r}_1}{2r_1} \bar{F}_1(P) \right) \\ &= -\frac{x_1}{r_1} \bar{F}_1(P) + -\frac{x_1}{4} \bar{F}_1(P) \end{aligned} \quad (A2-78)$$

where

$$\vec{F}_1(P) = \sum_{k=1}^4 C_k(P) * A_k(P) * (-\alpha_k)^2 * e^{-\alpha_k r_1} \quad (\text{A2-79})$$

$$\begin{aligned} \nabla^2 \sigma_y(1) &= \nabla \cdot \left( -\frac{\vec{j}}{2} F_1(P) + y_1 * \frac{\vec{r}_1}{2r_1} \vec{F}_1(P) \right) \\ &= -\frac{y_1}{r_1} \vec{F}_1(P) + \frac{y_1}{4} \vec{F}_1(P) \end{aligned} \quad (\text{A) (A2-80)}$$

$$\begin{aligned} \nabla^2 \sigma_z(1) &= \nabla \cdot \left( -\frac{\vec{k}}{2} F_1(P) + z_1 * \frac{\vec{r}_1}{2r_1} \vec{F}_1(P) \right) \\ &= -\frac{z_1}{r_1} \vec{F}_1(P) + \frac{z_1}{4} \vec{F}_1(P) \end{aligned} \quad (\text{B) (A2-80)}$$

$$\begin{aligned} \nabla^2 \sigma_x(2) &= -\nabla \cdot \left( -\frac{\vec{i}}{2} F_2(P) + x_2 * \frac{\vec{r}_2}{2r_2} \vec{F}_2(P) \right) \\ &= -\frac{x_2}{r_2} \vec{F}_2(P) + \frac{x_2}{4} \vec{F}_2(P) \end{aligned} \quad (\text{A2-81})$$

where

$$\vec{F}_2(P) = \sum_{k=1}^4 C_k(P) * A_k(P) * (-\alpha_k)^2 * e^{-\alpha_k r_2} \quad (\text{A2-82})$$

$$\begin{aligned} \nabla^2 \sigma_y(2) &= -\nabla \cdot \left( -\frac{\vec{j}}{2} F_2(P) + y_2 * \frac{\vec{r}_2}{2r_2} \vec{F}_2(P) \right) \\ &= -\frac{y_2}{r_2} \vec{F}_2(P) + \frac{y_2}{4} \vec{F}_2(P) \end{aligned} \quad (\text{A) (A2-83)}$$

$$\begin{aligned}
 \nabla^2 \sigma_2(2) &= -\nabla \cdot \left( -\frac{\vec{k}}{2} F_2(P) + z_2^* \frac{\vec{r}_2}{2r_2^2} \bar{F}_2(P) \right) \\
 &= -\frac{z_2}{r_2} \bar{F}_2(P) + -\frac{z_2}{4} \bar{F}_2(P) \qquad (B) (A2-83)
 \end{aligned}$$

(4).  $\nabla^2 P$  (1212) at  $r=0$

$r=0$ , implies  $x=0$ ,  $y=0$  and  $z=0$ , so it follows that

$$\begin{aligned}
 r_1^2 \Big|_{r=0} &= \left( \left(x + \frac{-x}{2}\right)^2 + \left(y + \frac{-y}{2}\right)^2 + \left(z + \frac{-z}{2}\right)^2 \right)_{r=0} \\
 &= X^2 + Y^2 + Z^2 = R^2 \qquad (A) (A2-84)
 \end{aligned}$$

Hence

$$r_1 \Big|_{r=0} = R \qquad (B) (A2-84)$$

$$\vec{r}_1 \Big|_{r=0} = \vec{R} \qquad (C) (A2-84)$$

For similar reasons

$$\begin{aligned}
 r_2^2 \Big|_{r=0} &= \left( \left(x - \frac{-x}{2}\right)^2 + \left(y - \frac{-y}{2}\right)^2 + \left(z - \frac{-z}{2}\right)^2 \right)_{r=0} \\
 &= X^2 + Y^2 + Z^2 = R^2 \qquad (A) (A2-85)
 \end{aligned}$$

So

$$r_2 \Big|_{r=0} = R \qquad (B) (A2-85)$$

$$\vec{r}_2 \Big|_{r=0} = \vec{R} \qquad (C) (A2-85)$$

$[\nabla^2 P_2(1212)]_{r=0}$  is only the function of R

The above HF orbitals are functions of R only, so the integration over  $\Theta$  and  $\phi$  associated with R can be carried out easily.

$$\psi(1) \Big|_{r=0} = \sum_{i=1}^4 C_i(1s) * A_i(1s) * e^{-\alpha_i R} = B_R(1s) \quad (\text{A2-86})$$

where

$$B(1s) = \sum_{i=1}^4 C_i(1s) * A_i(1s) * e^{-\alpha_i R} \quad (\text{A2-87})$$

$$\psi(2) \Big|_{r=0} = \sum_{i=1}^4 C_i(1s) * A_i(1s) * e^{-\alpha_i R} = B_R(1s) \quad (\text{A2-88})$$

$$\psi(1) \Big|_{r=0} = \psi(2) \Big|_{r=0} = B_R(1s) \quad (\text{A2-89})$$

The first order derivatives for  $\psi(1)$  and  $\psi(2)$  at  $r=0$  are

$$\begin{aligned} \nabla \psi(1) \Big|_{r=0} &= \frac{\vec{R}}{2R} \sum_{i=1}^4 C_i(1s) * A_i(1s) * (-\alpha_i) * e^{-\alpha_i R} \\ &= \frac{\vec{R}}{2R} \bar{B}_R(1s) \end{aligned} \quad (\text{A2-90})$$

where

$$\bar{B}_R(1s) = \sum_{i=1}^4 C_i(1s) * A_i(1s) * (-\alpha_i) * e^{-\alpha_i R} \quad (\text{A2-91})$$

$$\begin{aligned} \nabla \phi(2) \Big|_{r=0} &= -\frac{\vec{R}}{2R} \sum_{i=1}^4 C_i(1s) * A_i(1s) * (-\alpha_i) * e^{-\alpha_i R} \\ &= -\frac{\vec{R}}{2R} \bar{B}_R(1s) \end{aligned} \quad (\text{A2-92})$$

It is clear that

$$\nabla \phi(1) \Big|_{r=0} = -\nabla \phi(2) \Big|_{r=0} = -\frac{\vec{R}}{2R} \bar{B}_R(1s) \quad (\text{A2-93})$$

A check on the term  $\nabla p(12)$ :

For He and its isoelectronic series

$$\nabla p(12) = 2 \phi(1) \cdot \nabla \phi(2) + 2 \phi(2) \cdot \nabla \phi(1) \quad (\text{A2-94})$$

$$\begin{aligned} \nabla p(12) \Big|_{r=0} &= 2 \vec{B}_R(1s) \cdot \left(-\frac{\vec{R}}{2R} \bar{B}_R(1s)\right) + 2 \bar{B}_R(1s) \cdot \left(-\frac{\vec{R}}{2R} \vec{B}_R(1s)\right) \\ &= 0 \end{aligned} \quad (\text{A2-95})$$

From

$$\nabla p(21) = \nabla p(12) \quad (\text{A2-96})$$

So

$$\nabla p(21) \Big|_{r=0} = \nabla p(12) \Big|_{r=0} = 0 \quad (\text{A2-97})$$

The term  $\nabla p(12) \nabla p(21)$  in  $\nabla^2 P_2(1212)$  would disappear when  $r=0$ . And

$$\nabla^2 \phi(1) \Big|_{r=0} = -\frac{1}{2R} \bar{B}_R(1s) + -\frac{1}{4} \bar{B}_R(1s) \quad (\text{A2-98})$$

where

$$\bar{B}_R(1s) = \sum_{i=1}^4 C_i(1s) * A_i(1s) * (-\alpha_i)^2 * e^{-\alpha_i R} \quad (\text{A2-99})$$

$$\nabla^2 \phi(2) \Big|_{r=0} = -\frac{1}{2R} \bar{B}_R(1s) + -\frac{1}{4} \bar{B}_R(1s) \quad (\text{A2-100})$$

$$\nabla^2 \phi(1) \Big|_{r=0} = \nabla^2 \phi(2) \Big|_{r=0} \quad (\text{A2-101})$$

In summary

$$\nabla^2 p(12) = \nabla^2 p(21) \quad (\text{A2-102})$$

$$\nabla^2 p(11) \Big|_{r=0} = \nabla^2 p(22) \Big|_{r=0} \quad (\text{A2-103})$$

$$\begin{aligned} \nabla^2 P_2(1212) \Big|_{r=0} &= p(11) \cdot \nabla^2 p(22) \Big|_{r=0} + p(22) \cdot \nabla^2 p(11) \Big|_{r=0} \\ &+ 2 \nabla p(11) \cdot \nabla p(22) \Big|_{r=0} - \nabla p(12) \cdot \nabla p(21) \Big|_{r=0} \\ &- \frac{1}{2} p(12) \cdot \nabla^2 p(21) \Big|_{r=0} - \frac{1}{2} p(21) \cdot \nabla^2 p(12) \Big|_{r=0} \\ &= 2 p(11) \cdot \nabla^2 p(22) \Big|_{r=0} + 2 \nabla p(11) \cdot \nabla p(22) \Big|_{r=0} \\ &- p(12) \cdot \nabla^2 p(21) \Big|_{r=0} \quad (\text{A2-104}) \end{aligned}$$

Formulas (A2-102) to (A2-104) also are available for both the Be and the Ne series. For Be and their isoelectronic series the similar forms are

$$\mu(1) \Big|_{r=0} = \sum_{i=1}^2 C_i(1s) * A_i(1s) * e^{-\alpha_i R} + R * \sum_{j=3}^5 C_j(1s) * A_j(1s) * e^{-\alpha_j R} \quad (\text{A}) \text{ (A2-105)}$$

$$\mu(2) \Big|_{r=0} = \sum_{i=1}^2 C_i(1s) * A_i(1s) * e^{-\alpha_i R} + R * \sum_{j=3}^5 C_j(1s) * A_j(1s) * e^{-\alpha_j R} \quad (\text{B}) \text{ (A2-105)}$$

$$\Omega(1) \Big|_{r=0} = \sum_{i=1}^2 C_i(2s) * A_i(2s) * e^{-\alpha_i R} + R * \sum_{j=3}^5 C_j(2s) * A_j(2s) * e^{-\alpha_j R} \quad (\text{A}) \text{ (A2-106)}$$

$$\Omega(2) \Big|_{r=0} = \sum_{i=1}^2 C_i(2s) * A_i(2s) * e^{-\alpha_i R} + R * \sum_{j=3}^5 C_j(2s) * A_j(2s) * e^{-\alpha_j R} \quad (\text{B}) \text{ (A2-106)}$$

It is clear that

$$\mu(1) \Big|_{r=0} = \mu(2) \Big|_{r=0} \quad (\text{A2-107})$$

$$\Omega(1) \Big|_{r=0} = \Omega(2) \Big|_{r=0} \quad (\text{A2-108})$$

$$\nabla \mu(1) \Big|_{r=0} = -\frac{\bar{R}}{2R} ( \bar{B}_R(1s) + D_R(1s) + R * \bar{D}_R(1s) )$$

where (A2-109)

$$\bar{B}_R(1s) = \sum_{i=1}^2 C_i(1s) * A_i(1s) * (-\alpha_i) * e^{-\alpha_i R} \quad (\text{A2-110})$$

$$D_R(1s) = \sum_{j=3}^5 C_j(1s) * A_j(1s) * e^{-\alpha_j R} \quad (A2-111)$$

$$\bar{D}_R(1s) = \sum_{j=3}^5 C_j(1s) * A_j(1s) * (-\alpha_j) * e^{-\alpha_j R} \quad (A2-112)$$

$$\nabla \mu(2) \Big|_{r=0} = -\frac{\vec{R}}{2R} ( \bar{B}_R(1s) + D_R(1s) + R * \bar{D}_R(1s) ) \quad (A2-113)$$

$$\nabla \mu(1) \Big|_{r=0} = - \nabla \mu(2) \Big|_{r=0} \quad (A2-114)$$

$$\nabla \Omega(1) \Big|_{r=0} = -\frac{\vec{R}}{2R} ( \bar{B}_R(2s) + D_R(2s) + R * \bar{D}_R(2s) ) \quad (A2-115)$$

where

$$\bar{B}_R(2s) = \sum_{i=1}^2 C_i(2s) * A_i(2s) * (-\alpha_i) * e^{-\alpha_i R} \quad (A2-116)$$

$$D_R(2s) = \sum_{j=3}^5 C_j(2s) * A_j(2s) * e^{-\alpha_j R} \quad (A2-117)$$

$$\bar{D}_R(2s) = \sum_{j=3}^5 C_j(2s) * A_j(2s) * (-\alpha_j) * e^{-\alpha_j R} \quad (A2-118)$$

$$\nabla \Omega(2) \Big|_{r=0} = -\frac{\vec{R}}{2R} ( \bar{B}_R(2s) + D_R(2s) + R * \bar{D}_R(2s) ) \quad (A2-119)$$

$$\nabla \Omega(1) \Big|_{r=0} = - \nabla \Omega(2) \Big|_{r=0} \quad (A2-120)$$

So we still have

$$\nabla p(21) \Big|_{r=0} = \nabla p(12) \Big|_{r=0} = 0 \quad (\text{A2-121})$$

And for second order derivatives

$$\begin{aligned} \nabla^2 \mu(1) \Big|_{r=0} &= -\frac{2}{R} [ \bar{B}_R(1s) + D_R(1s) + R * \bar{D}_R(1s) ] \\ &+ -\frac{1}{4} [ \bar{B}_R(1s) + 2\bar{D}_R(1s) + R * \bar{D}_R(1s) ] \end{aligned} \quad (\text{A2-122})$$

where

$$\bar{B}_R(1s) = \sum_{i=1}^2 C_i(1s) * A_i(1s) * (-\alpha_i)^2 * e^{-\alpha_i R} \quad (\text{A2-123})$$

$$\bar{D}_R(1s) = \sum_{j=3}^5 C_j(1s) * A_j(1s) * (-\alpha_j) * e^{-\alpha_j R} \quad (\text{A2-124})$$

$$\begin{aligned} \nabla^2 \mu(2) \Big|_{r=0} &= -\frac{2}{R} [ \bar{B}_R(1s) + D_R(1s) + R * \bar{D}_R(1s) ] \\ &+ -\frac{1}{4} [ \bar{B}_R(1s) + 2\bar{D}_R(1s) + R * \bar{D}_R(1s) ] \end{aligned} \quad (\text{A2-125})$$

It is clear that

$$\nabla^2 \mu(1) \Big|_{r=0} = \nabla^2 \mu(2) \Big|_{r=0} \quad (\text{A2-126})$$

$$\begin{aligned} \nabla^2 \Omega(1) \Big|_{r=0} &= -\frac{2}{R} [ \bar{B}_R(2s) + D_R(2s) + R * \bar{D}_R(2s) ] \\ &+ -\frac{1}{4} [ \bar{B}_R(2s) + 2\bar{D}_R(2s) + R * \bar{D}_R(2s) ] \end{aligned} \quad (\text{A2-127})$$

where

$$\bar{B}_R(2s) = \sum_{i=1}^2 C_i(2s) * A_i(2s) * (-\alpha_i)^2 * e^{-\alpha_i R} \quad (A2-128)$$

$$\bar{D}_R(2s) = \sum_{j=3}^5 C_j(2s) * A_j(2s) * (-\alpha_j)^2 * e^{-\alpha_j R} \quad (A2-129)$$

$$\begin{aligned} \nabla^2 \Omega(2) \Big|_{r=0} = & -\frac{2}{R} [ \bar{B}_R(2s) + D_R(2s) + R * \bar{D}_R(2s) ] \\ & + \frac{1}{4} [ \bar{B}_R(2s) + 2\bar{D}_R(2s) + R * \bar{D}_R(2s) ] \end{aligned} \quad (A2-130)$$

It is clear that

$$\nabla^2 \Omega(1) \Big|_{r=0} = \nabla^2 \Omega(2) \Big|_{r=0} \quad (A2-131)$$

All orbitals of the Be series satisfy the equations (A2-102) to (A2-104).

The 1s and 2s orbitals of the Ne isoelectronic series have the same form as the 1s and 2s orbitals of the Be series, only the orbital coefficients are different.

When  $r=0$ , the p orbitals of Ne series became

$$\sigma_x(1) \Big|_{r=0} = X * \sum_{k=1}^4 C_k(2P) * A_k(2P) * e^{-\alpha_k R} = X * F_R(p) \quad (A2-132)$$

where

$$F_R(p) = X * \sum_{k=1}^4 C_k(2P) * A_k(2P) * e^{-\alpha_k R} \quad (A2-133)$$

$$\sigma_y(1) \Big|_{r=0} = Y * F_R(p) \quad (A) (A2-134)$$

$$\sigma_z(1) \Big|_{r=0} = Z * F_R(p) \quad (B) (A2-134)$$

$$\sigma_x(2) \Big|_{r=0} = X * \sum_{k=1}^4 C_k(2P) * A_k(2P) * e^{-\alpha_k R} = X * F_R(p) \quad (A) (A2-135)$$

$$\sigma_y(2) \Big|_{r=0} = Y * F_R(p) \quad (B) (A2-135)$$

$$\sigma_z(2) \Big|_{r=0} = Z * F_R(p) \quad (C) (A2-135)$$

It is clear that

$$\sigma_x(1) \Big|_{r=0} = X * F_R(p) = \sigma_x(2) \Big|_{r=0} \quad (A) (A2-136)$$

$$\sigma_y(1) \Big|_{r=0} = Y * F_R(p) = \sigma_y(2) \Big|_{r=0} \quad (B) (A2-136)$$

$$\sigma_z(1) \Big|_{r=0} = Z * F_R(p) = \sigma_z(2) \Big|_{r=0} \quad (C) (A2-136)$$

$$\nabla \sigma_x(1) \Big|_{r=0} = -\frac{\vec{1}}{2} F_R(p) + X * \frac{\vec{R}}{2R} \bar{F}_R(p) \quad (A2-137)$$

where

$$\bar{F}_R(p) = \sum_{k=1}^4 C_k(p) * A_k(p) * (-\alpha_k) * e^{-\alpha_k R} \quad (A2-138)$$

$$\nabla \sigma_x(2) \Big|_{r=0} = -\frac{\vec{1}}{2} F_R(p) - X * \frac{\vec{R}}{2R} \bar{F}_R(p) \quad (A2-139)$$

So

$$\nabla \sigma_x(1) \Big|_{r=0} = -\frac{1}{2} F_R(P) + X * \frac{1}{2R} \bar{F}_R(P) = - \nabla \sigma_x(2) \Big|_{r=0}$$

(A) (A2-140)

$$\nabla \sigma_y(1) \Big|_{r=0} = -\frac{1}{2} F_R(P) + Y * \frac{1}{2R} \bar{F}_R(P) = - \nabla \sigma_y(2) \Big|_{r=0}$$

(B) (A2-140)

$$\nabla \sigma_z(1) \Big|_{r=0} = -\frac{1}{2} F_R(P) + Z * \frac{1}{2R} \bar{F}_R(P) = - \nabla \sigma_z(2) \Big|_{r=0}$$

(C) (A2-140)

We still have

$$\nabla p(12) \Big|_{r=0} = \nabla p(12) \Big|_{r=0} = 0$$

(A2-141)

Also

$$\nabla^2 \sigma_x(1) \Big|_{r=0} = -\frac{X}{R} \bar{F}_R(P) + -\frac{1}{4} \bar{F}_R(P) = \nabla^2 \sigma_x(2) \Big|_{r=0}$$

(A2-142)

where

$$\bar{F}_R(P) = \sum_{k=1}^4 C_k(P) * A_k(P) * (-\alpha_k) * e^{-\alpha_k R}$$

(A2-143)

$$\nabla^2 \sigma_y(1) \Big|_{r=0} = -\frac{Y}{R} \bar{F}_R(P) + -\frac{1}{4} \bar{F}_R(P) = \nabla^2 \sigma_y(2) \Big|_{r=0}$$

(A) (A2-144)

$$\nabla^2 \sigma_z(1) \Big|_{r=0} = -\frac{Z}{R} \bar{F}_R(P) + -\frac{1}{4} \bar{F}_R(P) = \nabla^2 \sigma_z(2) \Big|_{r=0}$$

(B) (A2-144)

It is clear that All 2P orbitals satisfy the equations (A2-102) to (A2-104).

## APPENDIX (3) KINETIC CORRELATION ENERGY FORMULAS

The general formula shown as (4-16,A-C) in chapter (IV),  
is

$$T_c = - \frac{1}{2} \int_{1' \rightarrow 1} \nabla^2 (b(121'2) \cdot P_2(121'2)) \bigg|_{1' \rightarrow 1} d1 d2 \quad (A3-1)$$

$$T_c(1) = - \frac{1}{2} \int_{1' \rightarrow 1} b(121'2) \cdot \nabla^2 P_2(121'2) \bigg|_{1' \rightarrow 1} d1 d2 \quad (A) (A3-2)$$

$$T_c(2) = - \int_{1' \rightarrow 1} \nabla b(121'2) \cdot \nabla^2 P_2(121'2) \bigg|_{1' \rightarrow 1} d1 d2 \quad (B) (A3-2)$$

$$T_c(3) = - \frac{1}{2} \int_{1' \rightarrow 1} P_2(121'2) \cdot \nabla^2 b(121'2) \bigg|_{1' \rightarrow 1} d1 d2 \quad (C) (A3-2)$$

$$T_c = T_c(1) + T_c(2) + T_c(3) \quad (A3-3)$$

## (1) Basic variables

It is convenient for the calculation of  $T_c(1)$  to put the system into the coordinates (2); and for the calculation of  $T_c(2)$  and  $T_c(3)$  to put the system into the coordinates (1) in Appendix (1). The 6 dimensional integration for two bodies may be simplified to 3 dimensional numerical integration and the formula (A3-2) becomes

$$T_c(1) = - \frac{1}{2} 8\pi^2 \int_{-1}^1 \int_0^\infty \int_0^\infty b(121'2) \cdot \nabla^2 P_2(121'2) \bigg|_{1' \rightarrow 1} dr_1 dr_2 d\cos\theta' \quad (A) (A3-4)$$

$$Tc(2) = - 8\pi^2 \int_{-1}^1 \int_0^\infty \int_0^\infty \nabla b(121'2) \cdot \nabla P_2(121'2) \Big|_{1' \rightarrow 1} dR dr d\cos\theta \quad (B) (A3-4)$$

$$Tc(3) = - \frac{1}{2} 8\pi^2 \int_{-1}^1 \int_0^\infty \int_0^\infty P_2(121'2) \cdot \nabla^2 b(121'2) \Big|_{1' \rightarrow 1} dR dr d\cos\theta \quad (C) (A3-4)$$

The details of the transformation are in Appendix (1). The selection of the coordinates (1) or (2) is based on the fact that more variables in the denominator may be canceled in the coordinates chosen before the numerical integration, so that the integration of the formula (A3-4) can be completed more easily with our 3D Simpson numerical integration program.

Some of the basic variables are

$$\nabla = \vec{i} \frac{\partial}{\partial x_1} + \vec{j} \frac{\partial}{\partial y_1} + \vec{k} \frac{\partial}{\partial z_1} \quad (A) (A3-5)$$

$$\nabla^2 = \frac{\partial^2}{\partial x_1^2} + \frac{\partial^2}{\partial y_1^2} + \frac{\partial^2}{\partial z_1^2} \quad (B) (A3-5)$$

$$\vec{r}_1 = \vec{i}x_1 + \vec{j}y_1 + \vec{k}z_1 \quad (A) (A3-6)$$

$$\vec{R} = \frac{1}{2} (\vec{r}_1 + \vec{r}_2) \quad (B) (A3-6)$$

$$\vec{r} = \vec{r}_1 - \vec{r}_2 \quad (\text{C}) (\text{A3-6})$$

(2) Derivatives of the for basic variables

$$\begin{aligned} \nabla r_1 &= \nabla [x_1^2 + y_1^2 + z_1^2]^{\frac{1}{2}} = [\vec{i}x_1 + \vec{j}y_1 + \vec{k}z_1] \cdot [x_1^2 + y_1^2 + z_1^2]^{-\frac{1}{2}} \\ &= \frac{\vec{r}_1}{r_1} \end{aligned} \quad (\text{A}) (\text{A3-7})$$

$$\begin{aligned} \nabla R &= \nabla \left( \frac{1}{2} [(x_1+x_2)^2 + (y_1+y_2)^2 + (z_1+z_2)^2] \right)^{\frac{1}{2}} \\ &= \frac{1}{2} \frac{2[\vec{i}(x_1+x_2) + \vec{j}(y_1+y_2) + \vec{k}(z_1+z_2)]}{2[(x_1+x_2)^2 + (y_1+y_2)^2 + (z_1+z_2)^2]^{\frac{1}{2}}} \\ &= \frac{\vec{R}}{2R} \end{aligned} \quad (\text{B}) (\text{A3-7})$$

$$\begin{aligned} \nabla r &= \nabla [(x_1-x_2)^2 + (y_1-y_2)^2 + (z_1-z_2)^2]^{\frac{1}{2}} \\ &= \frac{\vec{i}(x_1-x_2) + \vec{j}(y_1-y_2) + \vec{k}(z_1-z_2)}{[(x_1-x_2)^2 + (y_1-y_2)^2 + (z_1-z_2)^2]^{\frac{1}{2}}} \\ &= \frac{\vec{r}}{r} \end{aligned} \quad (\text{C}) (\text{A3-7})$$

$$\nabla \vec{r}_1 = \nabla [\vec{i}x_1 + \vec{j}y_1 + \vec{k}z_1] = 3 \quad (\text{A}) (\text{A3-8})$$

$$\begin{aligned} \nabla \cdot \vec{R} &= \nabla \cdot \frac{1}{2} [\vec{i}(x_1+x_2) + \vec{j}(y_1+y_2) + \vec{k}(z_1+z_2)] \\ &= -\frac{3}{2} \end{aligned} \quad \text{(B) (A3-8)}$$

$$\begin{aligned} \nabla \cdot \vec{r} &= \nabla \cdot [\vec{i}(x_1-x_2) + \vec{j}(y_1-y_2) + \vec{k}(z_1-z_2)] \\ &= 3 \end{aligned} \quad \text{(C) (A3-8)}$$

$$\vec{R} \cdot \vec{r} = Rr \cdot \cos\theta \quad \text{(A) (A3-9)}$$

$$\vec{r}_1 \cdot \vec{r}_2 = r_1 r_2 \cos\theta' \quad \text{(B) (A3-9)}$$

$$\vec{R} \cdot \vec{r}_1 = \vec{R} \cdot (\vec{R} + \frac{\vec{r}}{2}) = R^2 + \frac{1}{2} Rr \cdot \cos\theta \quad \text{(A) (A3-10)}$$

$$\vec{R} \cdot \vec{r}_1 = \frac{1}{2} (\vec{r}_1 + \vec{r}_2) \cdot \vec{r}_1 = \frac{1}{2} (r_1^2 + r_1 r_2 \cos\theta') \quad \text{(B) (A3-10)}$$

$$\vec{r} \cdot \vec{r}_1 = \vec{r} \cdot (\vec{R} + \frac{\vec{r}}{2}) = Rr \cdot \cos\theta + \frac{1}{2} r^2 \quad \text{(A) (A3-11)}$$

$$\vec{r} \cdot \vec{r}_1 = (\vec{r}_1 - \vec{r}_2) \cdot \vec{r}_1 = r_1^2 - r_1 r_2 \cos\theta' \quad \text{(B) (A3-11)}$$

For convenience put

$$V_R = -\frac{\vec{r}_1}{r_1} \cdot \frac{\vec{R}}{R} = -\frac{1}{r_1} (R + \frac{1}{2} r \cdot \cos\theta) \quad \text{(A) (A3-12)}$$

$$V_R = -\frac{\vec{r}_1}{r_1} \cdot -\frac{\vec{R}}{R} = -\frac{1}{2R} (r_1 + r_2 \cos \theta') \quad (\text{B}) (\text{A3-12})$$

$$V_r = -\frac{\vec{r}_1}{r_1} \cdot -\frac{\vec{r}}{r} = -\frac{1}{r_1} (R \cos \theta + \frac{1}{2} r) \quad (\text{A}) (\text{A3-13})$$

$$V_r = -\frac{\vec{r}_1}{r_1} \cdot -\frac{\vec{r}}{r} = -\frac{1}{r} (r_1 - r_2 \cos \theta') \quad (\text{B}) (\text{A3-13})$$

$$\begin{aligned} \nabla^2_{r_1} &= \nabla \cdot -\frac{\vec{r}_1}{r_1} = -\frac{1}{r_1} \nabla \cdot \vec{r}_1 + \vec{r}_1 \cdot \nabla \left( -\frac{1}{r_1} \right) = -\frac{3}{r_1} - \frac{\vec{r}_1}{r_1^2} \cdot -\frac{\vec{r}_1}{r_1} \\ &= -\frac{2}{r_1} \quad (\text{A}) (\text{A3-14}) \end{aligned}$$

$$\begin{aligned} \nabla^2_R &= \nabla \cdot -\frac{\vec{R}}{2R} = -\frac{1}{2R} \nabla \cdot \vec{R} + \vec{R} \cdot \nabla \left( -\frac{1}{2R} \right) = -\frac{3}{4R} - \frac{\vec{R}}{2R^2} \cdot -\frac{\vec{R}}{2R} \\ &= -\frac{1}{2R} \quad (\text{B}) (\text{A3-14}) \end{aligned}$$

$$\begin{aligned} \nabla^2_r &= \nabla \cdot -\frac{\vec{r}}{r} = -\frac{1}{r} \nabla \cdot \vec{r} + \vec{r} \cdot \nabla \left( -\frac{1}{r} \right) = -\frac{3}{r} - \frac{\vec{r}}{r^2} \cdot -\frac{\vec{r}}{r} \\ &= -\frac{2}{r} \quad (\text{C}) (\text{A3-14}) \end{aligned}$$

(2) Derivatives of basis functions of the He atom and its isoelectronic series

$$\phi(1) = \sum_{i=1}^4 C_i A_i e^{-\alpha_i r_1} \quad (\text{A}) \text{ (A3-15)}$$

$$\begin{aligned} \nabla \phi(1) &= \nabla \left( \sum_{i=1}^4 C_i A_i e^{-\alpha_i r_1} \right) \\ &= -\frac{\vec{r}_1}{r_1} \sum_{i=1}^4 C_i A_i (-\alpha_i) e^{-\alpha_i r_1} \end{aligned} \quad (\text{B}) \text{ (A3-15)}$$

$$\nabla^2 \phi(1) = \nabla \left( -\frac{\vec{r}_1}{r_1} \sum_{i=1}^4 C_i A_i (-\alpha_i) e^{-\alpha_i r_1} \right) \quad (\text{C}) \text{ (A3-15)}$$

$$\begin{aligned} &= -\frac{\vec{r}_1}{r_1} \nabla \left( \sum_{i=1}^4 C_i A_i (-\alpha_i) e^{-\alpha_i r_1} \right) + \left( \sum_{i=1}^4 C_i A_i (-\alpha_i) e^{-\alpha_i r_1} \right) \nabla \frac{\vec{r}_1}{r_1} \\ &= \left( \sum_{i=1}^4 C_i A_i (-\alpha_i)^2 e^{-\alpha_i r_1} \right) + \frac{2}{r_1} \left( \sum_{i=1}^4 C_i A_i (-\alpha_i) e^{-\alpha_i r_1} \right) \end{aligned}$$

For convenience, put

$$B(1) = \sum_{i=1}^4 C_i A_i e^{-\alpha_i r_1} \quad (\text{A}) \text{ (A3-16)}$$

$$\bar{B}(1) = \sum_{i=1}^4 C_i A_i (-\alpha_i) e^{-\alpha_i r_1} \quad (\text{B}) \text{ (A3-16)}$$

$$\bar{\bar{B}}(1) = \sum_{i=1}^4 C_i A_i (-\alpha_i)^2 e^{-\alpha_i r_1} \quad (\text{C}) \text{ (A3-16)}$$

$$\nabla \phi(1) = -\frac{\vec{r}_1}{r_1} \bar{B}(1) \quad (\text{A}) \text{ (A3-17)}$$

$$\nabla^2 \phi(1) = \bar{B}(1) + \frac{-2}{r_1} \bar{B}(1) \quad (\text{B}) (\text{A3-17})$$

$$\phi(R) = \sum_{i=1}^4 C_i A_i e^{-\alpha_i R} \quad (\text{A}) (\text{A3-18})$$

$$\begin{aligned} \nabla \phi(R) &= \nabla \left( \sum_{i=1}^4 C_i A_i e^{-\alpha_i R} \right) \\ &= -\frac{\vec{R}}{2R} \sum_{i=1}^4 C_i A_i (-\alpha_i) e^{-\alpha_i R} \end{aligned} \quad (\text{B}) (\text{A3-18})$$

$$\begin{aligned} \nabla^2 \phi(R) &= \nabla \left( \frac{\vec{R}}{2R} \sum_{i=1}^4 C_i A_i (-\alpha_i) e^{-\alpha_i R} \right) \quad (\text{C}) (\text{A3-18}) \\ &= \frac{\vec{R}}{2R} \nabla \left( \sum_{i=1}^4 C_i A_i (-\alpha_i) e^{-\alpha_i R} \right) + \left( \sum_{i=1}^4 C_i A_i (-\alpha_i) e^{-\alpha_i R} \right) \nabla \frac{\vec{R}}{2R} \\ &= -\frac{1}{4} \left( \sum_{i=1}^4 C_i A_i (-\alpha_i)^2 e^{-\alpha_i R} \right) + -\frac{1}{2R} \left( \sum_{i=1}^4 C_i A_i (-\alpha_i) e^{-\alpha_i R} \right) \end{aligned}$$

For convenience, put

$$B(R) = \sum_{i=1}^4 C_i A_i e^{-\alpha_i R} \quad (\text{A}) (\text{A3-19})$$

$$\bar{B}(R) = \sum_{i=1}^4 C_i A_i (-\alpha_i) e^{-\alpha_i R} \quad (\text{B}) (\text{A3-19})$$

$$\bar{\bar{B}}(R) = \sum_{i=1}^4 C_i A_i (-\alpha_i)^2 e^{-\alpha_i R} \quad (\text{C}) (\text{A3-19})$$

$$\nabla \phi(R) = -\frac{\vec{R}}{2R} \bar{B}(R) \quad (\text{A}) (\text{A3-20})$$

$$\nabla^2 \phi(R) = -\frac{1}{4} \ddot{B}(R) + -\frac{1}{2R} \dot{B}(R) \quad (\text{B}) \text{ (A3-20)}$$

$$p(R) = 2\phi(R)\dot{\phi}(R) \quad (\text{A}) \text{ (A3-21)}$$

$$\begin{aligned} \nabla p(R) &= \nabla \{2\phi(R)\dot{\phi}(R)\} = 4\dot{\phi}(R)\nabla\phi(R) \\ &= 2\dot{\phi}(R)\ddot{B}(R) - \frac{\dot{R}}{R} \\ &= \dot{p}(R) - \frac{\dot{R}}{R} \end{aligned} \quad (\text{B}) \text{ (A3-21)}$$

where

$$\dot{p}(R) = 2\dot{\phi}(R)\ddot{B}(R) \quad (\text{A3-22})$$

$$\begin{aligned} \nabla^2 p(R) &= \nabla \{4\dot{\phi}(R)\nabla\phi(R)\} = 4\dot{\phi}(R)\nabla^2\phi(R) + 4[\nabla\dot{\phi}(R)]^2 \\ &= \dot{\phi}(R) \left[ \ddot{B}(R) + -\frac{2}{R} \dot{B}(R) \right] + [\dot{B}(R)]^2 \end{aligned} \quad (\text{A}) \text{ (A3-23)}$$

put

$$\ddot{p}(R) = \nabla^2 p(R) \quad (\text{B}) \text{ (A3-23)}$$

$$\begin{aligned} \nabla p(R)^{1/3} &= -\frac{1}{3} p(R)^{-2/3} \nabla p(R) = \frac{p(R)^{1/3}}{3p(R)} \nabla p(R) \\ &= -\frac{\dot{R}}{R} - \frac{2\dot{\phi}(R)\ddot{B}(R)}{3p(R)} p(R)^{1/3} \\ &= -\frac{\dot{R}}{R} p'(R) \cdot p(R)^{1/3} \end{aligned} \quad (\text{A3-25})$$

where

$$p'(R) = \frac{2\dot{\phi}(R)\ddot{B}(R)}{3p(R)} \quad (\text{A}) \text{ (A3-26)}$$

$$\begin{aligned}
\nabla p'(R) &= \nabla \left( \frac{2\phi(R)\bar{B}(R)}{3p(R)} \right) \\
&= \frac{2\bar{B}(R)}{3p(R)} \nabla \phi(R) + \frac{2\phi(R)}{3p(R)} \nabla \bar{B}(R) + \frac{2\phi(R)\bar{B}(R)}{3} \nabla \frac{1}{p(R)} \\
&= \frac{\vec{R}}{R} \left( \frac{[\bar{B}(R)]^2}{3p(R)} + \frac{\phi(R)\bar{B}(R)}{3p(R)} + \frac{-2[\bar{B}(R)]^2}{3p(R)} \right) \\
&= \frac{\vec{R}}{R} \frac{1}{3p(R)} \left( \phi(R)\bar{B}(R) - [\bar{B}(R)]^2 \right) \\
&= \frac{\vec{R}}{R} p''(R) \tag{B} \text{ (A3-26)}
\end{aligned}$$

where

$$p''(R) = \frac{1}{3p(R)} \left( \phi(R)\bar{B}(R) - [\bar{B}(R)]^2 \right) \tag{A3-27}$$

$$\beta = q p(R)^{1/3} \tag{A} \text{ (A3-28)}$$

$$\begin{aligned}
\nabla \beta &= q \nabla p(R)^{1/3} = q \cdot p(R)^{1/3} p'(R) \frac{\vec{R}}{R} \\
&= \frac{\vec{R}}{R} \beta p'(R) \tag{B} \text{ (A3-28)}
\end{aligned}$$

$$[\nabla \beta]^2 = \beta^2 [p'(R)]^2 \tag{C} \text{ (A3-28)}$$

$$\nabla^2 \beta = \nabla \left( -\frac{\vec{R}}{R} \beta p'(R) \right) \quad (D) (A3-28)$$

$$= -\frac{\vec{R}}{R} p'(R) \nabla \beta + -\frac{\vec{R}}{R} \beta \nabla p'(R) + \beta p'(R) \nabla -\frac{\vec{R}}{R}$$

$$= \beta [p'(R)]' + \beta p''(R) + \beta p'(R) -\frac{1}{R}$$

$$= \beta \left( p''(R) + [p'(R)]' + p'(R) -\frac{1}{R} \right)$$

(3) Derivatives of various functional and density functions

$$p(11') = 2\phi(1)\phi(1') \quad (A) (A3-29)$$

$$p(12) = 2\phi(1)\phi(2) \quad (B) (A3-29)$$

$$\begin{aligned} P_2(121'2) &= p(11')p(22) - \frac{1}{2} p(12)p(21') \\ &= 2\phi(1)\phi(1') \cdot p(22) - \frac{1}{2} \cdot 2\phi(1)\phi(2) \cdot p(21') \\ &= (2\phi(1') \cdot p(22) - \phi(2) \cdot p(21')) \cdot \phi(1) \quad (A) (A3-30) \end{aligned}$$

$$\begin{aligned} \nabla P_2(121'2) &= (2\phi(1') \cdot p(22) - \phi(2) \cdot p(21')) \nabla \phi(1) \\ &= -\frac{\vec{R}}{R} \frac{1}{R} \bar{B}(1) (2\phi(1') \cdot p(22) - \phi(2) \cdot p(21')) \quad (B) (A3-30) \end{aligned}$$

$$\begin{aligned} \nabla^2 P_2(121'2) &= (2\phi(1') \cdot p(22) - \phi(2) \cdot p(21')) \nabla^2 \phi(1) \\ &= (2\phi(1') \cdot p(22) - \phi(2) \cdot p(21')) \nabla \left[ -\frac{\vec{R}}{R} \frac{1}{R} \bar{B}(1) \right] \\ &= [\bar{B}(1) + -\frac{2}{R} \bar{B}(1)] (2\phi(1') \cdot p(22) - \phi(2) \cdot p(21')) \quad (C) (A3-30) \end{aligned}$$

$$\phi(12) = [1 - \epsilon(R) (1 + \frac{r}{2})] \cdot e^{-\beta^2 r^2} \quad (\text{A}) (\text{A3-31})$$

$$\begin{aligned} \nabla \phi(12) &= \nabla \left( [1 - \epsilon(R) (1 + \frac{r}{2})] e^{-\beta^2 r^2} \right) \\ &= [1 - \epsilon(R) (1 + \frac{r}{2})] \nabla e^{-\beta^2 r^2} - e^{-\beta^2 r^2} \nabla [ \epsilon(R) (1 + \frac{r}{2}) ] \end{aligned} \quad (\text{B}) (\text{A3-31})$$

$$\begin{aligned} \nabla^2 \phi(12) &= [1 - \epsilon(R) (1 + \frac{r}{2})] \nabla^2 e^{-\beta^2 r^2} \\ &+ 2 \nabla [1 - \epsilon(R) (1 + \frac{r}{2})] \cdot \nabla e^{-\beta^2 r^2} + e^{-\beta^2 r^2} \nabla^2 [1 - \epsilon(R) (1 + \frac{r}{2})] \end{aligned} \quad (\text{C}) (\text{A3-31})$$

$$\begin{aligned} \nabla [\beta^2 r^2] &= 2\beta r [r \nabla \beta + \beta \nabla r] \\ &= 2\beta^2 r \left[ r p'(R) \frac{\vec{R}}{R} + \frac{\vec{r}}{r} \right] \end{aligned} \quad (\text{A3-32})$$

$$\begin{aligned} \nabla e^{-\beta^2 r^2} &= e^{-\beta^2 r^2} \nabla [-\beta^2 r^2] = -2\beta r \cdot e^{-\beta^2 r^2} [r \nabla \beta + \beta \nabla r] \\ &= -2\beta r \cdot e^{-\beta^2 r^2} \left[ r \beta p'(R) \frac{\vec{R}}{R} + \beta \frac{\vec{r}}{r} \right] \end{aligned} \quad (\text{A3-33})$$

Put

$$L = (\nabla [-\beta^2 r^2])^2 \quad (\text{A}) (\text{A3-34})$$

$$\begin{aligned} (\nabla [-\beta^2 r^2])^2 &= (2\beta^2 r [r p'(R) \frac{\vec{R}}{R} + \frac{\vec{r}}{r}])^2 \\ &= 4\beta^4 r^2 \{ [r p'(R)]^2 + 2r p'(R) \cdot \cos\theta + 1 \} \end{aligned} \quad (\text{B}) (\text{A3-34})$$

Put

$$M = \nabla^2 [\beta^2 r^2] \quad (\text{A}) \text{ (A3-35)}$$

$$\begin{aligned} \nabla^2 [\beta^2 r^2] &= r^2 \nabla^2 \beta^2 + 2 \nabla \beta^2 \nabla r^2 + \beta^2 \nabla^2 r^2 \\ &= 2r^2 (\beta \nabla^2 \beta + [\nabla \beta]^2) + 8\beta r \nabla \beta \nabla r + 2\beta^2 (r \nabla^2 r + [\nabla r]^2) \\ &= 2\beta^2 r^2 \left( p''(R) + [p'(R)]^2 + \frac{1}{R} p'(R) \right) \\ &\quad + 2\beta^2 r^2 [p'(R)]^2 + 8\beta^2 r \cdot p'(R) \cdot \cos \theta + 6\beta^2 \\ &= 2\beta^2 r^2 \left( p''(R) + 2[p'(R)]^2 + \frac{1}{R} p'(R) \right) \\ &\quad + 8\beta^2 r \cdot p'(R) \cdot \cos \theta + 6\beta^2 \end{aligned} \quad (\text{B}) \text{ (A3-35)}$$

$$\begin{aligned} \nabla^2 e^{-\beta^2 r^2} &= \nabla \left( e^{-\beta^2 r^2} \nabla [-\beta^2 r^2] \right) \\ &= e^{-\beta^2 r^2} \nabla^2 [-\beta^2 r^2] + \nabla [-\beta^2 r^2] \cdot \nabla e^{-\beta^2 r^2} \\ &= -M \cdot e^{-\beta^2 r^2} + (\nabla [-\beta^2 r^2])^2 \cdot e^{-\beta^2 r^2} \\ &= [L-M] \cdot e^{-\beta^2 r^2} \end{aligned} \quad (\text{A3-36})$$

$$\begin{aligned} [1 - \epsilon(R) \left(1 + \frac{r}{2}\right)] \cdot \nabla^2 e^{-\beta^2 r^2} \\ = [1 - \epsilon(R) \left(1 + \frac{r}{2}\right)] \cdot [L-M] \cdot e^{-\beta^2 r^2} \end{aligned} \quad (\text{A3-37})$$

$$\epsilon(R) = \frac{u \cdot \sqrt{\pi B}}{t + \sqrt{\pi B}} \quad (\text{A}) \text{ (A3-38)}$$

$$\begin{aligned}
\nabla \epsilon(R) &= \nabla \frac{u \cdot \sqrt{\pi B}}{t + \sqrt{\pi B}} = u \frac{(t + \sqrt{\pi B}) \nabla [\sqrt{\pi B}] - \sqrt{\pi B} \nabla (t + \sqrt{\pi B})}{(t + \sqrt{\pi B})^2} \\
&= u \frac{t \nabla [\sqrt{\pi B}]}{(t + \sqrt{\pi B})^2} = u \frac{t \cdot \frac{\sqrt{\pi B} p'(R)}{(t + \sqrt{\pi B})^2} - \frac{\vec{R}}{R}}{(t + \sqrt{\pi B})^2} \\
&= \frac{\vec{R}}{R} \frac{ut \sqrt{\pi B}}{(t + \sqrt{\pi B})^2} p'(R) \tag{B} (A3-38)
\end{aligned}$$

Put

$$W = \nabla^2 \epsilon(R) \tag{A} (A3-39)$$

$$\begin{aligned}
\nabla^2 \epsilon(R) &= \nabla^2 \left( \frac{u \sqrt{\pi B}}{t + \sqrt{\pi B}} \right) = \nabla \left( \frac{ut \nabla [\sqrt{\pi B}]}{(t + \sqrt{\pi B})^2} \right) \\
&= \frac{(t + \sqrt{\pi B})^2 ut \nabla^2 [\sqrt{\pi B}] - ut \nabla [\sqrt{\pi B}] \nabla (t + \sqrt{\pi B})^2}{(t + \sqrt{\pi B})^4} \\
&= ut \frac{(t + \sqrt{\pi B}) \nabla^2 [\sqrt{\pi B}] - 2 \nabla [\sqrt{\pi B}]^2}{(t + \sqrt{\pi B})^3} \\
&= ut \frac{(t + \sqrt{\pi B}) \sqrt{\pi B} (p''(R) + [p'(R)]^2 + p'(R) \frac{1}{R}) - 2\pi B^2 [p'(R)]^2}{(t + \sqrt{\pi B})^3} \tag{B} (A3-39)
\end{aligned}$$

$$\nabla [1 - \epsilon(R) (1 + \frac{r}{2})] = -\nabla [\epsilon(R) (1 + \frac{r}{2})] \tag{A} (A3-40)$$

$$\begin{aligned}
&= - (1 + \frac{r}{2}) \nabla \epsilon(R) - \epsilon(R) \nabla (1 + \frac{r}{2}) \\
&= - (1 + \frac{r}{2}) \frac{\vec{R}}{R} \frac{ut \sqrt{\pi B}}{(t + \sqrt{\pi B})^2} p'(R) - \frac{1}{2} \epsilon(R) \frac{\vec{r}}{r}
\end{aligned}$$

$$\nabla^2 [1 - \epsilon(R) (1 + \frac{r}{2})] \quad (B) (A3-40)$$

$$= -(1 + \frac{r}{2}) \nabla^2 \epsilon(R) - 2 \nabla \epsilon(R) \nabla (1 + \frac{r}{2}) - \epsilon(R) \nabla^2 (1 + \frac{r}{2})$$

$$= -(1 + \frac{r}{2}) \cdot W - 2 \frac{\vec{R}}{R} \frac{ut}{(t + \sqrt{\pi\beta})^2} \frac{\sqrt{\pi\beta}}{R} p'(R) \cdot \frac{\vec{r}}{2R} - \frac{1}{R} \epsilon(R)$$

$$= -(1 + \frac{r}{2}) \cdot W - \frac{ut}{(t + \sqrt{\pi\beta})^2} \frac{\sqrt{\pi\beta}}{R} p'(R) \cdot \cos\theta - \frac{1}{R} \epsilon(R)$$

$$e^{-\beta^2 r^2} \nabla^2 [1 - \epsilon(R) (1 + \frac{r}{2})] \quad (A3-41)$$

$$= - e^{-\beta^2 r^2} ( (1 + \frac{r}{2}) \cdot W + \frac{1}{2} \frac{ut}{(t + \sqrt{\pi\beta})^2} \frac{\sqrt{\pi\beta}}{R} p'(R) \cdot \cos\theta + \epsilon(R) )$$

$$2 \nabla [1 - \epsilon(R) (1 + \frac{r}{2})] \cdot \nabla e^{-\beta^2 r^2} \quad (A3-42)$$

$$= 2 ( - (1 + \frac{r}{2}) \frac{\vec{R}}{R} \frac{ut}{(t + \sqrt{\pi\beta})^2} \frac{\sqrt{\pi\beta}}{R} p'(R) - \frac{1}{2} \epsilon(R) \frac{\vec{r}}{R} )$$

$$\cdot ( -2\beta r \cdot e^{-\beta^2 r^2} [r\beta p'(R) \frac{\vec{R}}{R} + \beta \frac{\vec{r}}{R} ] )$$

$$= 4\beta r e^{-\beta^2 r^2} ( \beta r \cdot [p'(R)]^2 (1 + \frac{r}{2}) \frac{ut}{(t + \sqrt{\pi\beta})^2} \frac{\sqrt{\pi\beta}}{R} + \frac{1}{2} \beta \epsilon(R) )$$

$$+ \frac{1}{2} \beta r \cdot p'(R) \cdot \epsilon(R) \cdot \cos\theta$$

$$+ \beta \cdot p'(R) \cdot (1 + \frac{r}{2}) \frac{ut}{(t + \sqrt{\pi\beta})^2} \frac{\sqrt{\pi\beta}}{R} \cos\theta )$$

## (4). Summary

$$\nabla^2 ( b(121'2) \cdot P_2(121'2) ) \quad (A3-43)$$

$$= b(121'2) \nabla^2 P_2(121'2) + P_2(121'2) \nabla^2 b(121'2)$$

$$+ 2 \nabla b(121'2) \nabla P_2(121'2)$$

$$b(121'2) = \phi(12)\phi(1'2) - \phi(12) - \phi(1'2)$$

$$P_2(121'2) = p(11')p(22) - \frac{1}{2} p(12)p(21')$$

$$= \phi(1) ( 2\phi(1')p(22) - \phi(2)p(21') )$$

$$b(121'2) \nabla^2 P_2(121'2) \quad (A3-44)$$

$$= b(121'2) \cdot ( 2\phi(1')p(22) - \phi(2)p(21') ) \nabla^2 \phi(1)$$

$$= b(121'2) \cdot [ \bar{B}(1) + \frac{-2}{r} \bar{B}(1) ] ( 2\phi(1')p(22) - \phi(2)p(21') )$$

$$P_2(121'2) \cdot \nabla^2 b(121'2) = P_2(121'2) \cdot ( \phi(1'2) - 1 ) \nabla^2 \phi(12)$$

(A3-45)

$$\nabla^2 \phi(12) = [ 1 - \epsilon(R) ( 1 + \frac{-r}{2} ) ] \nabla^2 e^{-\beta^2 r^2} \quad (A3-46)$$

$$+ 2 \nabla [ 1 - \epsilon(R) ( 1 + \frac{-r}{2} ) ] \cdot \nabla e^{-\beta^2 r^2} + e^{-\beta^2 r^2} \nabla^2 [ 1 - \epsilon(R) ( 1 + \frac{-r}{2} ) ]$$

$$\begin{aligned}
&= [1 - \epsilon(R) (1 + \frac{r}{2})] \cdot [L-M] \cdot e^{-\beta^2 r^2} \\
&+ 4\beta r e^{-\beta^2 r^2} \{ \beta r \cdot [p'(R)]^2 (1 + \frac{r}{2}) \frac{ut}{(t+\sqrt{\pi\beta})^2} + \frac{1}{2}\beta\epsilon(R) \\
&+ \frac{1}{2} \beta r \cdot p'(R) \cdot \epsilon(R) \cdot \cos\theta + \beta \cdot p'(R) \cdot (1 + \frac{r}{2}) \frac{ut}{(t+\sqrt{\pi\beta})} \cos\theta \} \\
&- e^{-\beta^2 r^2} \{ (1 + \frac{r}{2}) \cdot W + \frac{ut}{(t+\sqrt{\pi\beta})} p'(R) \cdot \cos\theta + \frac{1}{r} \epsilon(R) \}
\end{aligned}$$

$$2 \nabla b(121'2) \nabla P_2(121'2) \quad (A3-47)$$

$$= 2 \{ \phi(1'2) - 1 \} \nabla \phi(12) \cdot \{ 2\phi(1') p(22) - \phi(2) p(21') \} \nabla \phi(1)$$

$$= 2 \{ \phi(1'2) - 1 \} \cdot \{ 2\phi(1') p(22) - \phi(2) p(21') \} \cdot \nabla \phi(12) \cdot \nabla \phi(1)$$

$$\nabla \phi(12) \cdot \nabla \phi(1) \quad (A3-48)$$

$$= [1 - \epsilon(R) (1 + \frac{r}{2})] \nabla e^{-\beta^2 r^2} - e^{-\beta^2 r^2} \nabla [\epsilon(R) (1 + \frac{r}{2})] \nabla \phi(1)$$

$$= \{ -2\beta r \cdot e^{-\beta^2 r^2} [r\beta p'(R) \frac{\vec{R}}{R} + \beta \frac{\vec{r}}{r}] \cdot [1 - \epsilon(R) (1 + \frac{r}{2})]$$

$$- e^{-\beta^2 r^2} [ (1 + \frac{r}{2}) \frac{\vec{R}}{R} - \frac{ut}{(t+\sqrt{\pi\beta})} p'(R) + \frac{1}{2}\epsilon(R) \frac{\vec{r}}{r} ] \} \frac{\vec{r}}{r_1} B(1)$$

$$= - e^{-\beta^2 r^2} \cdot \bar{B}(1) \{ 2\beta^2 r [r p'(R) \cdot V_R + V_r] \cdot [1 - \epsilon(R) (1 + \frac{r}{2})]$$

$$+ [ (1 + \frac{r}{2}) \frac{ut}{(t+\sqrt{\pi\beta})} p'(R) \cdot V_R + \frac{1}{2}\epsilon(R) \cdot V_r ] \}$$

BIBLIOGRAPHY

- [1] Löwdin P.O. Adv. Chem. Phys. 1959, 2 207
- [2] Colle R.; Salvetti O.  
Theoret. Chim. Acta. (Berl.) 1975 37, 329-334
- [3] Clementi E. J. Chem. Phys. 1963 38, 2248
- [4] Veillard A.; Clementi E. J. Chem. Phys. 1968 49, 2415
- [5] Massa L.; Flocco M.; Soirat A. Journal of Molecular Structure 1989 199, 337-342
- [6] Mcweeny R. Int. J. Quantum Chem. Symp. 1967, 1, 351
- [7] Mcweeny R. "Present Status of the correlation problem".  
In "The New World of Quantum Chemistry" ed. Pullman B.;  
Parr R.; D.reidel Company, Dordrecht 1976, PAGE 3
- [8] Weissbluth M. "Density Matrices". In "Atoms and Molecules" ed. Academic Press Inc. 1978, New York.  
PAGE 268-289
- [9] Hylleraas E.A. Z. Phys. 1929 54, 825
- [10] Hylleraas E.A. Adv. Quantum Chem. 1964 1, 1
- [11] James H.; Coolidge A.S. J. Chem. Phys. 1933 1, 825
- [12] Pekeris C.L. Phys. Rev. 1958 112, 1649
- [13] Pekeris C.L. Phys. Rev. 1959 115, 1216
- [14] Kolos W.; Wolniewicz L. J. Chem. Phys. 1968 49, 404
- [15] Fischer C.F. "Correlation in many-electron atomic system" In "The Hartree-Fock Method for Atoms" ed.  
John Wiley & Sons, Inc., 1977, New York. PAGE 117-148
- [16] Boys S.F. Proc. Roy. Soc. 1969 A309, 195

- [17] Boys S.F.; Handy N.C. Proc. Roy. Soc. 1969 A309, 209
- [18] Boys S.F.; Handy N.C. Proc. Roy. Soc. 1969 A310, 43
- [19] Boys S.F.; Handy N.C. Proc. Roy. Soc. 1969 A310, 63
- [20] Boys S.F.; Handy N.C. Proc. Roy. Soc. 1969 A310, 309
- [21] Sinanoglu O. Riv. Mod. Phys. 1963 35, 517
- [22] W.Kutzelnigg "Pair Correlation Theories" In "Method of the Electronic Structure Theory" ed. Plenum Press, New York. 1977. PAGE 134
- [23] Xu G.X.; Li L.M.; Wang D.M. "Configuration Interaction" In "Quantum Chemistry" Vol.B. ed. Academic Publishing Inc. 1985, Beijing. Page 847-885
- [24] Weissbluth M. "Density Matrices". In "Atoms and Molecules" ed. Academic Press Inc. 1978, New York. PAGE 412
- [25] Hohenberg P.; Kohn W. Phys. Rev. 1964, 136B, 864
- [26] Clinton W.L.; Massa L.J.  
Phys. Rev. Lett. 1972, 29, 1363
- [27] Massa L.J.; Clinton W.L.  
Trans. America Crystallography Assn. 1972, 8 149
- [28] Clinton W.L.; Frishberg C.A.; Massa L.J.; Oldfield P.A.  
Int. J. Quantum Chem Symp. 1973, 7, 505
- [29] Frishberg C.A.; Massa L.J.  
Int. J. Quantum Chem. 1978, 13, 801

- [30] Frishberg C.A.; Goldberg M.J.; Massa L.J. "Quantum Model of the Coherent Diffraction Experiment: Recent Generalizations and Applications" In "Electron Distributions and the Chemical Bond" ed. Coppens P.; Hall M.B., Plenum, New York 1982, PAGE 101
- [31] Clinton W.L.; Frishberg C.A.; Goldberg M.J.; Massa L.J. Oldfield P.A. Int. J. Quantum Chem. Symp. 1983, 17, 517
- [32] Goldberg M.J.; Massa L.J. Int. J. Quantum Chem. 1983, 24, 113
- [33] Massa L.J.; Boehme R.F.; La Placa S.J. "X-ray Imaging of Quantum Electron Structure" In "Patterson and Pattersons: Fifth Years of the Patterson Function" Glusker J.P.; Patterson B.K.; Rossi M., Oxford University Press, New York, 1987 PAGE 427
- [34] Frishberg C.A.; Massa L.J. Phys. Rev. B. 1981, 24, 7018
- [35] Frishberg C.A.; Massa L.J. Acta Cryst. 1982, A38, 93
- [36] Massa L.J.; Goldberg M.J.; Frishberg C.A. Boehme R.F.; La Placa S.J. Phys. Rev. Lett. 1985, 55, 622
- [37] Boehme R.F.; La Placa S.J. Phys. Rev. Lett. 1987, 59, 985
- [38] Glusker J.R.; Trueblood K.N. Crystal Structure Analysis 2nd ed. Oxford University Press, New York 1985, PAGE 21
- [39] Kittel C. Introduction to Solid State Physics 2nd ed. PAGE 44
- [40] Stout G.H.; Jensen L.H.; X-ray Structure Determination: A practical Guide Mc Millan Company, New York 1968
- [41] Mcweeny R. Rev. Mod. Phys. 1960, 32, 335

- [42] Löwdin P.O. Phys. Rev. 1955, 97, 1474
- [43] Löwdin P.O. Phys. Rev. 1955, 97, 1490
- [44] Coleman J. Rev. Mod. Phys. 1963, 35 668
- [45] Gilbert T.L. Rev. Mod. Phys. 1975, B12, 2111
- [46] Harriman J. Phys. Rev. 1981, A24, 680
- [47] Flocco M. X-ray correlated wavefunction  
The Library of The City University of New York
- [48] Clinton W.L.; Galli A.J.; Massa L.J.  
Phys. Rev. 1969, 177, 7
- [49] Stewart R.F.; Davidson E.R.; Simpson W.T.  
J. Chem. Phys 1965, 42, 3175
- [50] Colle R.; Salvetti O.  
Theoret. Chim. Acta.(Berl.) 1975 37, 329
- [51] Montagnani R.; Riani P.; Salvetti O.  
Theoret. Chim. Acta. (Berl.) 1982 60, 399
- [52] Colle R.; Salvetti O. Mol. Physics 1982 47, 959
- [53] Colle R.; Salvetti O. J. Chem. Phys. 1983 79, 1404
- [54] Colle R.; Salvetti O. "A functional of the Two  
-Particle Density Matrix for the Approximate  
Calculation of the Electronic Correlation Energy".  
In "Density Matrices and Density Functionals", ed.  
Erdahl R.; Smith V.A.; D. Reidel Company, Dordrecht  
1986, PAGE 545
- [55] Flocco M.;Gao X.;Massa L. Int.J.Quantum Chem. 1990
- [56] Bingel W.A. Theoret. Chim. Acta. (Berl.) 1967 8, 54
- [57] Wigner E.; Seitz F. Phys. Rev. 1933 43, 804

- [58] Colle R.; Moscardo F.; Riani P.; Salvetti O.  
Theoret. Chim. Acta. 1977 44. 1
- [59] Colle R.; Montagnani R.; Riani P.; Salvetti O.  
Theoret. Chim. Acta. 1978 49. 37
- [60] Colle R.; Salvetti O. Theoret. Chim. Acta. 1979 53. 55
- [61] Clementi E. Tables of Atomic Functions, a supplement to the paper by Clementi E. "Ab Initio Computations in Atoms and Molecules". IBM J.Res. Dev. 1965 9. 2
- [62] Cohen L.; Santhanam P.; Frishberg C.  
Int. J. Quantum Chem. Symp. 1980 14. 143
- [63] Cohen L.; Frishberg C.; Lee C.; Massa L.J.  
Int. J. Quantum Chem. Symp. 1986 19. 525
- [64] Davidson E.R. "Reduced Density Matrices" In "Reduced Density Matrices in Quantum Chemistry" ed. Academic Press Inc. 1976, New York, PAGE 9
- [65] Pilar F.L. "The Electron Correlation Energy" In "Elementary Quantum Chemistry" ed. McGraw-Hill Book Company, New York. 1968 PAGE 351-354
- [66] Fischer C.F. "The Single Configuration Hartree-Fock Approximation" In "The Hartree-Fock Method for Atoms" ed. John Wiley Sons Inc. in Canada 1977. PAGE 28