

MODULAR STRATEGIES FOR THE SYNTHESIS OF TETRAHYDROFURAN AND
TETRAHYDROPYRAN CONTAINING ACETOGENINS

by

FENG WANG

A dissertation submitted to the Graduate Faculty in Chemistry in partial fulfillment of the requirements for the degree of Doctor of Philosophy, The City University of New York

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Abstract

MODULAR STRATEGIES FOR THE SYNTHESIS OF TETRAHYDROFURAN AND TETRAHYDROPYRAN CONTAINING ACETOGENINS

by

Feng Wang

Advisor: Professor David R. Mootoo

Annonaceous acetogenins, a family of naturally occurring polyketides, are well known for their cytotoxic, antitumoral, insecticidal, and immunosuppressive activities. The majority of structures contain one or more stereochemically complex tetrahydrofuran (THF) rings. A smaller subset referred to as non-classical congeners contain a highly substituted tetrahydropyran (THP). This thesis describes modular strategies for both THF and THP containing acetogenins, using as test cases the mono-THF acetogenin, 4-deoxyannoreticuin and the THF-THP congener muconin. 4-Deoxyannoreticuin, a typical member of the mono-THF acetogenin with two flanking carbinol groups at C-15 and C-20, shows cytotoxicity at the micromolar level against several human cancer cell lines, including MCF-7, HT-29, A-498, PC-3 and PACA. Muconin shows micromolar and selective cytotoxicity against PACA-2 and MCF-7.

Chapter 1 presents a review of the different structural types and bioactivity of the acetogenins.

Chapter 2 describes an olefin cross metathesis strategy for coupling of cyclic ether and butenolide segments of the acetogenins. This approach was applied to alcohol epimers of 4-deoxyannoreticuin in an attempt to assign the configuration at the C9 carbinol in the natural product. The activity of these compounds against two human tumor cell lines,

PC3 and Jurkat, was also evaluated. Unfortunately, identification of one or the other epimeric structures with the natural product was not possible because of the closeness of the physical and cytotoxicity data for all three compounds.

Chapter 3 deals with a tandem, iodonium ion promoted, cyclization on acetals of dihydroxydienes diene-acetals. These precursors are assembled in a modular fashion from aldehydes and easily accessible dihydroxydienes and the reaction products are stereochemically complex THF-THP subunits as found in the non-classical acetogenins. Vicinal or 1,3- diol precursors produced adjacently or methylene linked cyclic ethers in good to moderate yield. In all cases the cascade appears to be initiated by a 5-exo trig type cyclization to give a 2,5-disubstituted THF connected to an oxocarbenium ion which undergoes attack by the second alkene to form the THP ring. The substitution pattern and connectivity of the second alkene determines the mechanism of formation and constitution of the THP, i.e. a 2,5,6-tri-alkylated or 2,6-dialkyl-4-halogenated. Altogether, three new stereogenic centers are generated in the THF-THP products. The feasibility of this tandem reaction for non-classical acetogenin synthesis was illustrated in the preparation of an advanced precursor to muconin.

Acknowledgment

It would not have been possible to write this doctoral thesis without the help and support of the kind people around me, to only some of whom it is possible to give particular mention here.

Above all, I would like to thank my mentor, Dr. David R. Mootoo, for his support and patience, not to mention his advice and unsurpassed knowledge of chemistry. He has showed me what kind of qualities a chemist should have and what the true professionalism is. I will continue the journey of chemistry without him nearby, but his advice would never be forgotten.

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献给我的父亲母亲: 王胜国和罗明杰

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CHAPTER ONE

Acetogenins: classification, biosynthesis and bioactivity

1.1 Structure

Annonaceous acetogenins, a family of naturally occurring polyketides, are well known for their cytotoxic, antitumoral, insecticidal, and immunosuppressive activities.¹ Since the discovery of uvaricin in 1982 (**Figure 1.1**),² 417 annonaceous acetogenins have been isolated. Their highly potent bioactivities and structural complexity have attracted the attention of synthetic and medicinal chemists. A notable feature of several acetogenins is the ability to inhibit multiple drug resistant (MDR) tumor cell lines.³ The cytotoxic effects exhibited by acetogenins are at least in part due to the inhibition of NADH-ubiquinone oxidoreductase (mitochondrial complex I) of the respiratory chain. This inhibition results in a depletion of ATP levels which may cause arrest in the cell cycle at the G1 phase, and consequently induce apoptosis.⁴ However, the exact details of this mechanism remains elusive.

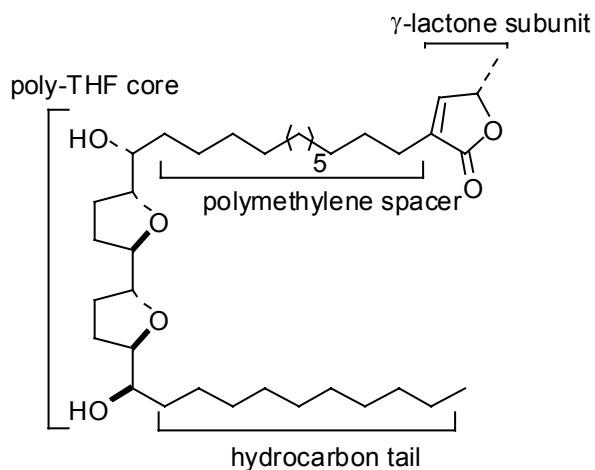


FIGURE 1.1 Structure of uvaricin.

Structurally, the simplest acetogenins contain linear C₃₂ or C₃₄ hydrocarbon chain ending in a γ -lactone (**Figure 1.2**). In the more complex structures, a variety of oxygenated functions, such as hydroxyl, ketone, epoxide, tetrahydrofuran (THF) and tetrahydropyran (THP) are present, as well as double and triple bonds. The classical cyclic ether acetogenins contain one or more epoxide, a mono- THF, an adjacently linked bis- or tris-THF, or non-adjacently linked bis- THF. The less common or non-classical acetogenins are those containing a THP ring. Among the 417 acetogenins isolated to date, 40 are linear, 16 are epoxy (E-1, E-2 and E-3), 206 are mono-THF (T-1 and T-2), 146 are bis-THF (T-3 ~ T6), one is a tris-THF (T-7), and the remaining eight are non-classical structures (T8 ~ T10) (**Figure 1.3**).⁴

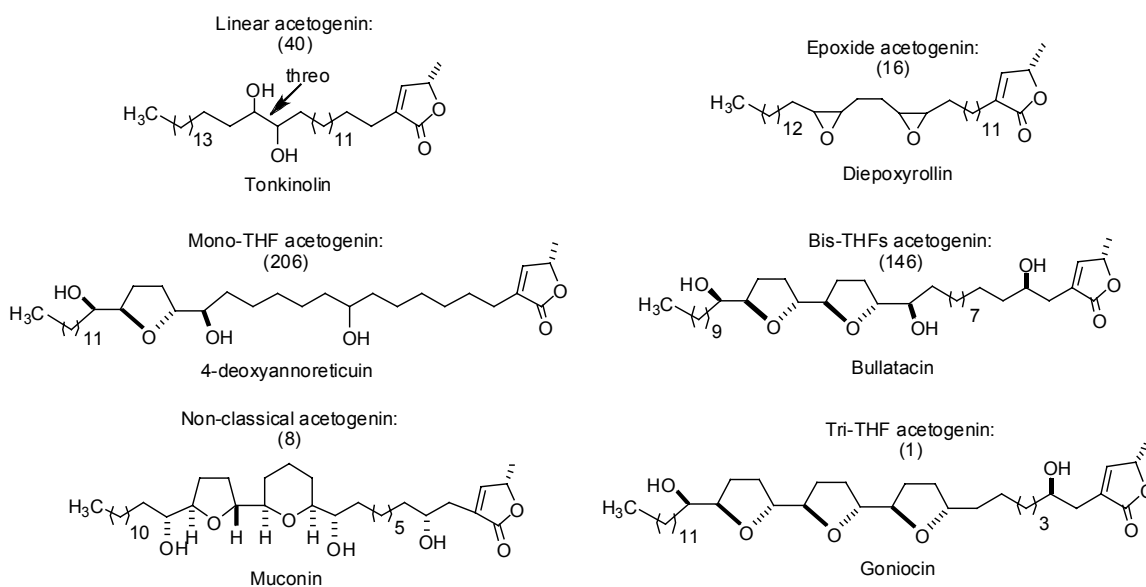


Figure 1.2 Representative acetogenins

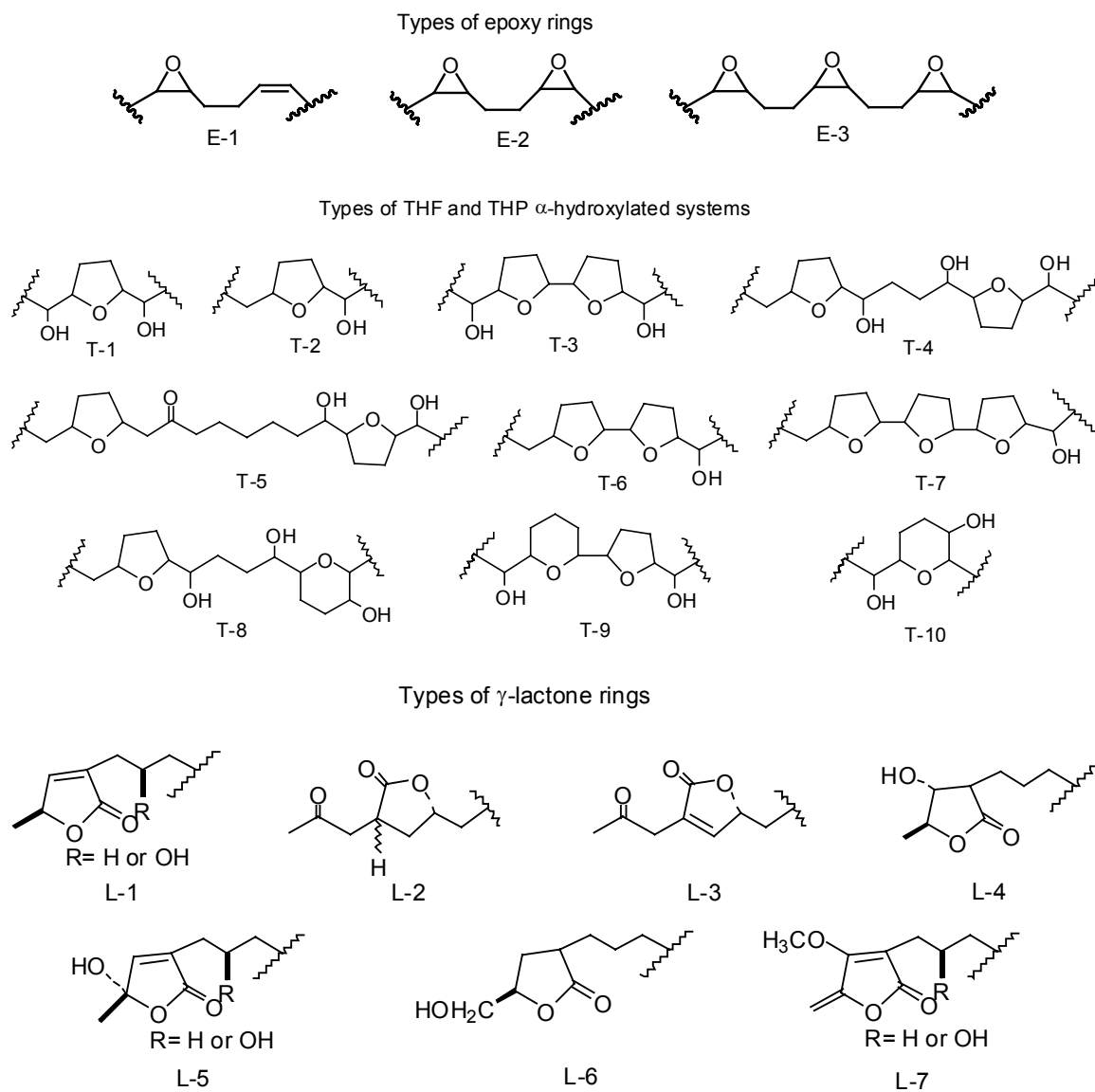


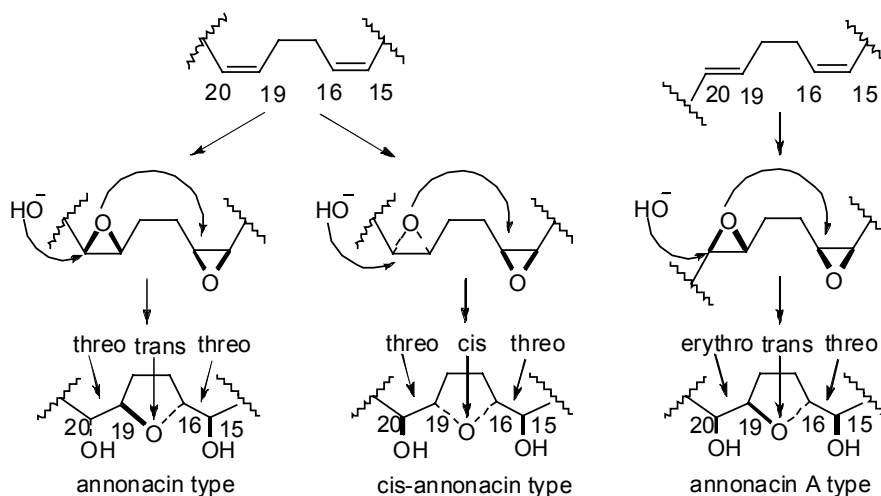
FIGURE 1.3 Subtypes of cyclic ether containing acetogenins with different lactone residues

1.2 Biosynthesis

Although acetogenins are very varied in terms of the position, number and the size of the cyclic ether ring, the number and the position of hydroxyl groups and the structures of the γ -lactones, they are believed to be all derived from related polyketide pathways.^{1c,5}

Recent discoveries of muridienins (presumed precursors of mono-THF acetogenins) and chatenaytrienins (presumed precursors of bis-THF acetogenins) provided strong evidence that the lactone moiety arises through enzymatic combination of lacceroic (C32) and ghedoic (C34) fatty acids with a three-carbon unit.^{6,7} However, these free acids or their esterified products have not been isolated, which might suggest that they are actually short-term intermediates.^{1c} The sequence of these processes remains unknown. The THF, THP and epoxide rings are suggested to arise from the epoxidation of isolated alkenes and subsequent epoxide cyclizations (**Scheme 1.1**).^{1c} This hypothesis is strongly supported by the discovery of possible linear acetogenin precursors with double bonds in appropriate positions.

SCHEME 1.1 Proposed biosynthetic pathways of three types of mono-THF acetogenins bearing flanking carbinols.

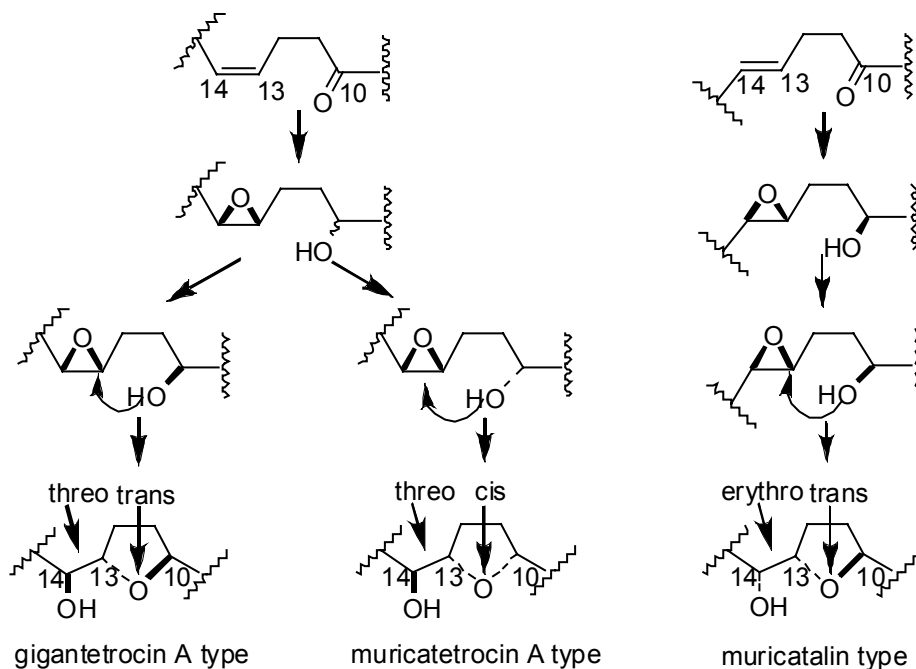


Acetogenins bearing a mono-THF ring with two flanking carbinols (the annonacin,^{8,9} cis-annonacin⁹ and annonacin A¹⁰ types) are considered to be formed from cis-cis or trans-cis

dienes, through epoxidation followed by cyclization starting from the left side of each precursor.

The three types of acetogenins bearing a mono-THF ring with one flanking carbinol (gigantetrocin A,¹¹ muricatetrocin A¹² and muricatelin¹³ types) can be formed from oxidation of one of the alkenes to give a keto *cis*-alkene or keto *trans*-alkene, followed by reduction of the keto group and cyclization of the resulting hydroxy epoxide (**Scheme 1.2**).

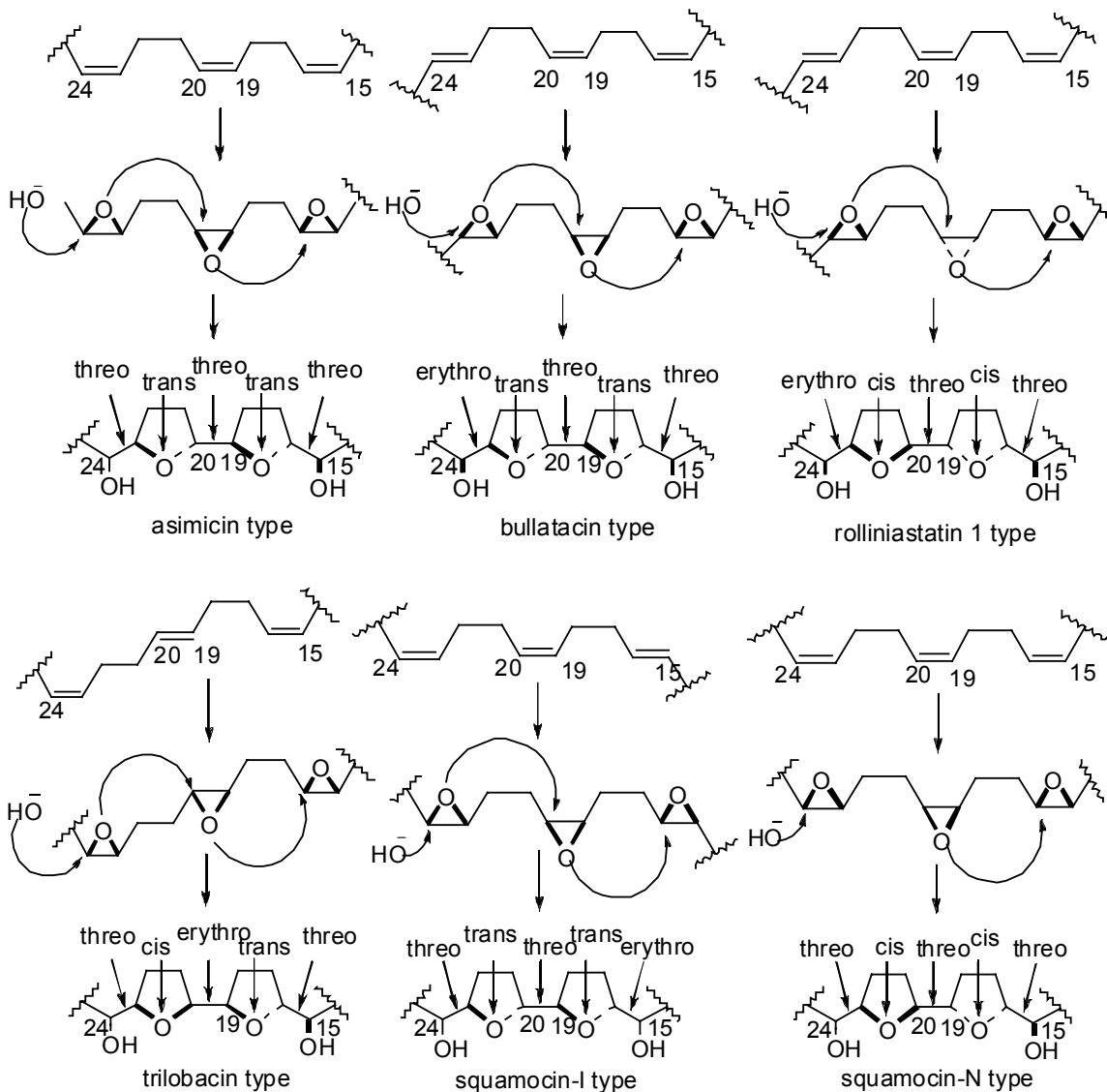
SCHEME 1.2 Proposed biosynthetic pathways of three types of mono-THF acetogenins.



Six types of adjacent bis-THF ring acetogenins having two flanking hydroxyls (asimicin,¹⁴ bullatacin,¹⁵ rolliniastatin,¹⁶ trilobacin,¹⁷ squamocin-I¹⁸ and squamocin-N¹⁸

types) can be generated via epoxide opening cascade reactions similar to those described for the annonacin series (**Scheme 1.3**).

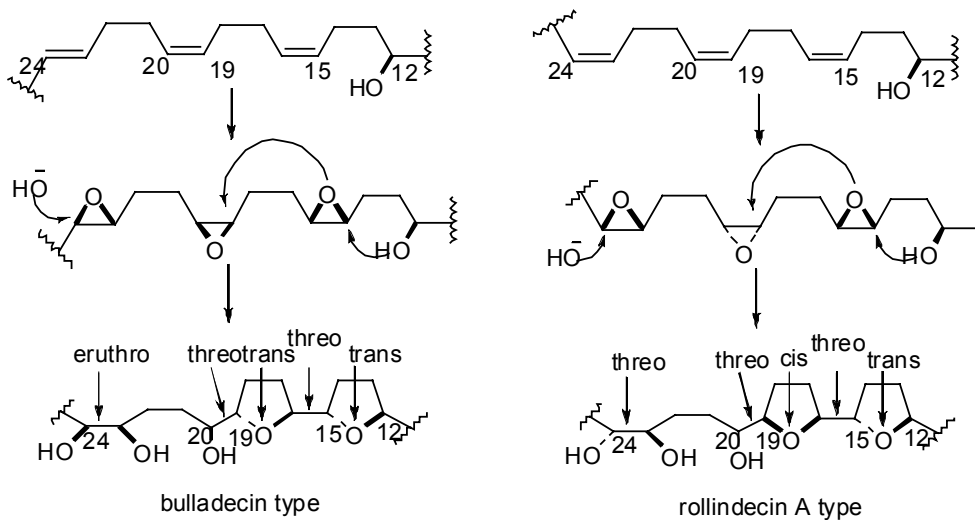
SCHEME 1.3 Proposed biosynthetic pathways of six types of bis-THF acetogenins.



Two other types of acetogenins, bearing adjacent bis-THF rings with one flanking carbinol (bulladecin and rollindecin A types), can be formed from hydroxylated *cis-cis-*

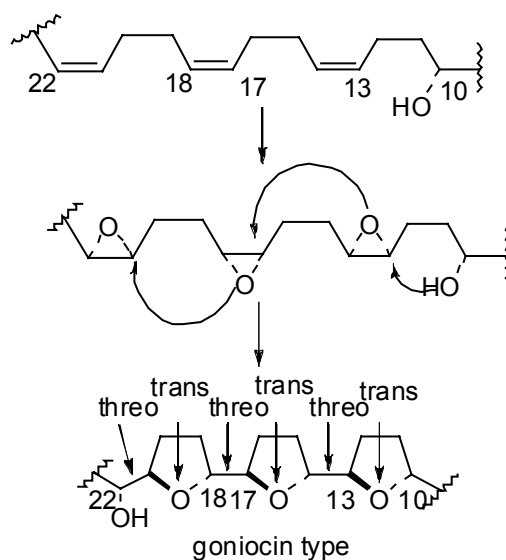
trans or hydroxylated *cis-cis-cis* trienes, through epoxidation and cyclization starting from the right side (**Scheme 1.4**).

SCHEME 1.4 Proposed biosynthetic pathways of bis-THF acetogenins with one flanking carbinol.



The formation of acetogenin, bearing adjacent tri-THF rings (goniocin type) which was found recently in *Goniothalamus giganteus*,¹⁹ can also be explained via an epoxide opening cascade, in this case on a *tris*-epoxy alcohol precursor (**Scheme 1.5**).

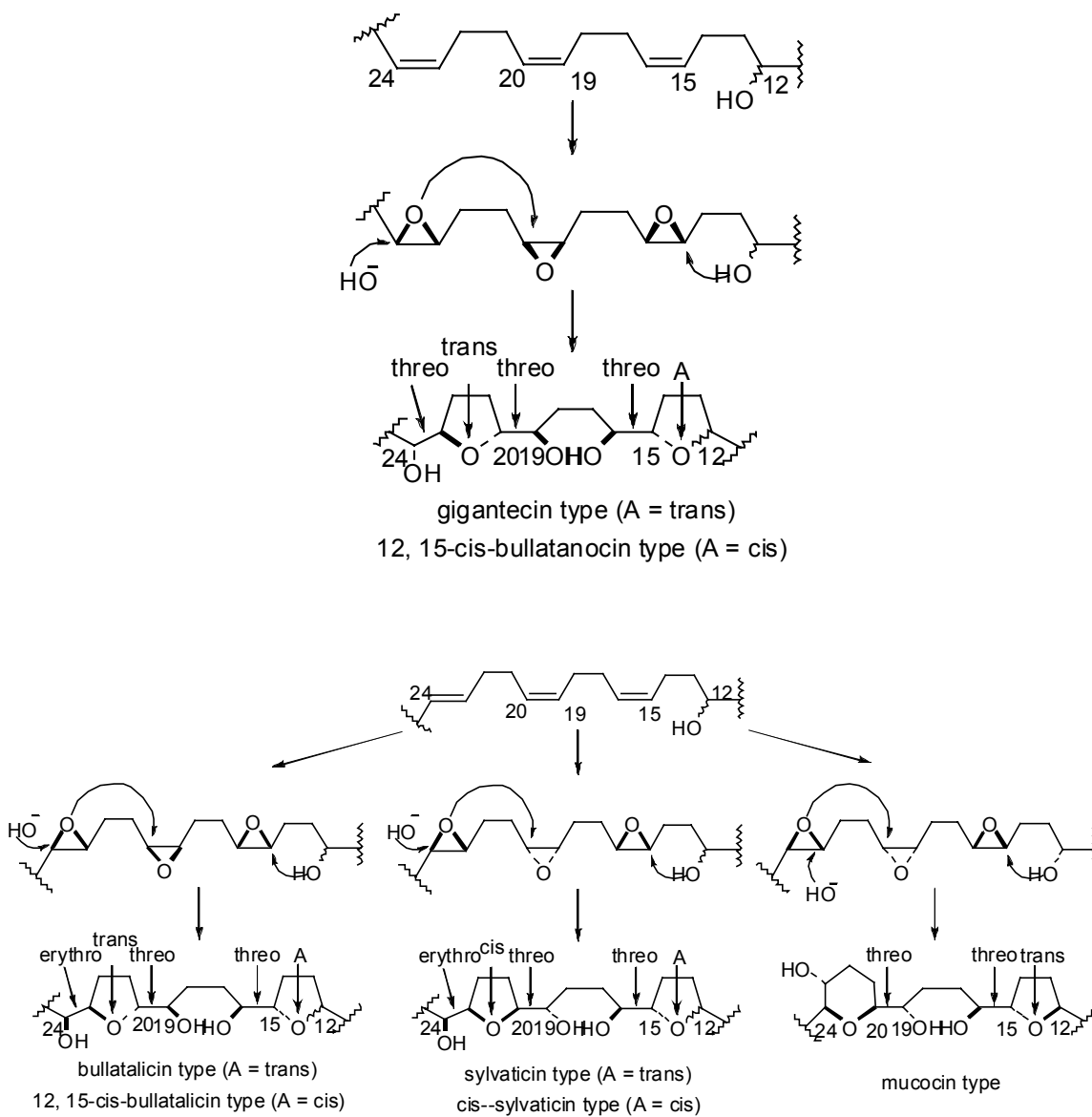
SCHEME 1.5 Proposed biosynthetic pathways of goniocin.



Six types of acetogenins having non-adjacent bis-THF rings (gigantecin,²⁰ 12,15-*cis*-bullatanocin,²¹ bullutalycin,²² 12,15-*cis*-bullatalycin,²¹ sylvaticin²³ and *cis*-sylvaticin.²¹) have been found. These compounds consist of one THF ring bearing one flanking carbinol and another THF ring bearing two flanking carbinols, separated by two methylenes. The formations of these acetogenins can be considered to result from two separate epoxide opening cascades that start from ends of a tris-epoxide precursor. Non-classical acetogenins, like mucocin, which have non-adjacent THF and tetrahydropyran (THP) rings, may arise from a similar biogenetic pathway (**Scheme 1.6**).

The biosynthesis of acetogenins remains elusive. The epoxide opening cascade sequence is one of the more plausible pathways. The laboratory syntheses of several THF containing acetogenins have mimicked this mechanism.²⁴

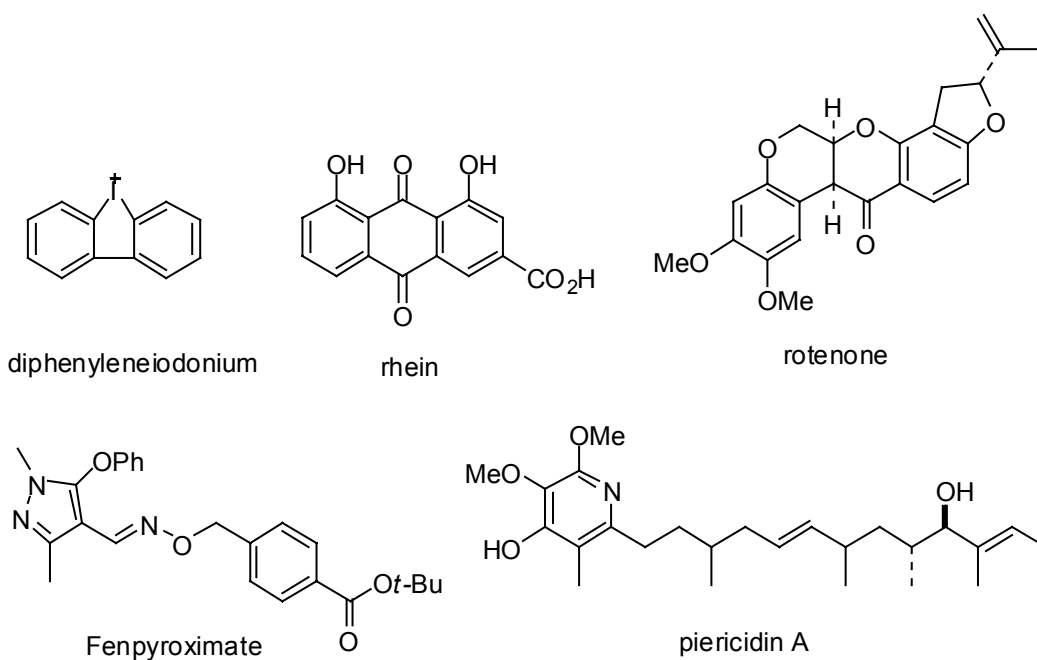
SCHEME 1.6 Proposed biosynthetic pathways of six types of bis-THF acetogenins and mucocin.



1.3 Bioactivity

Annonaceous acetogenins are the most potent inhibitors of mitochondrial NADH-ubiquinone oxidoreductase (complex I).²⁵ Except rhein²⁶ and diphenyleneiodonium,²⁷ which inhibit electron input into the enzyme, all inhibitors are believed to act on the terminal electron transfer step to ubiquinone. Ordinary complex I inhibitors such as piericidin A, rotenone, and several synthetic agrochemicals like Fenpyroximate have structural features in common, notably a polar and/or heterocyclic ring processing hydrogen bond ability and a hydrophobic “tail” structure (**Scheme 1.7**).²⁸ The former may play an important role in binding to the enzyme through specific interactions. The tail moiety may be important for attaching to and penetrating through the membrane-embedded segment of the enzyme. Therefore a critical degree of hydrophobicity may be required.

SCHEME 1.7 Some common inhibitors of mitochondrial complex I.



For acetogenins, the inhibitory mechanism might be different from that of the inhibitors mentioned above since two toxophores of acetogenins (the hydroxylated THF and γ -lactone moieties) are separated by a long alkyl spacer. Based on NMR spectroscopy and differential scanning calorimetry studies on acetogenins in liposomal membranes, MacLaughlin and colleagues proposed a model for an active conformation of acetogenins in the mitochondrial membrane environment, wherein the THF ring(s) with flanking carbinol groups resides near the glycerol backbone of phosphatidylcholine irrespective of the number of THF rings, and act as a hydrophilic anchor at the membrane surface. The γ -lactone ring directly interacts with the target site of complex I by lateral diffusion in the mitochondrial membrane interior.²⁹ This model has been challenged by recent structural-activity studies on acetogenin mimetics.³⁰ According to this model, the γ -lactone is the only species that interacts directly with the enzyme. Therefore structural modification of this moiety is expected to result in a drastic decrease in inhibitory potency. However, the structural modifications of γ -lactone moiety, such as deletion of the γ -methyl group, transformation of the γ -lactone ring to a six-membered δ -lactone ring and complete removal of γ -lactone ring did not significantly affect the inhibitory potency.²⁹ Moreover, the observation that OH groups adjacent to the THF ring is not crucial for the potent activity is in disagreement with the supposed important function of these OH groups as hydrophilic anchors at the membrane surface. On the contrary, OH groups and lactone may be crucial for cellular uptake and localization, because the degree and location of hydroxylation and the presence of lactone have a marked effect on cytotoxicity.

CHAPTER TWO

Synthesis and antitumor activity of
4-deoxyannoreticuin

2.1 Introduction

4-deoxyannoreticuin,³⁰ a white amorphous powder, is similar to annoreticuin³¹ except that it does not have a hydroxyl group at C-4 (**Figure 2.1**). It shows cytotoxicity at the micromolar level against several human cancer cell lines, including MCF-7, HT-29, A-498, PC-3 and PACA.³⁰ 4-deoxyannoreticuin is a typical member of the mono-THF acetogenin with two flanking carbinol groups at C-15 and C-20. The MH⁺ peak in the FABMS at m/z 581 established the molecular weight as 580. The elemental composition of C₃₅H₆₄O₆ was confirmed by high resolution EIMS of the molecular ion which gave an exact mass of m/z 581.4763 (calcd 581.4781). NMR data indicated the presence of an α,β -unsaturated γ -lactone which was corroborated by an absorption band in the IR at 1760 cm⁻¹ and the UV λ_{\max} at 215 nm.

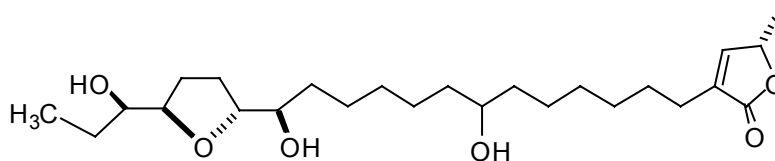


FIGURE 2.1 Structure of 4-deoxyannoreticuin.

The presence of a single THF ring with two flanking carbinols in a *threo/trans/threo* relative configuration was evidenced by ¹H NMR signals at δ 3.41 (H-15/20) and 3.80 (H-16/19), each integrating for two protons, as well as ¹³C NMR peaks at δ 74.06 (C-15/20), 82.58 (C-16/19), and 82.64 (C-16/19).³² The ring was located between C-15 and C-20 based on peaks in the EIMS as m/z 311 and 381. The hydroxyl located between THF ring and lactone was confirmed by EIMS at m/z 311. A small EIMS peak at m/z

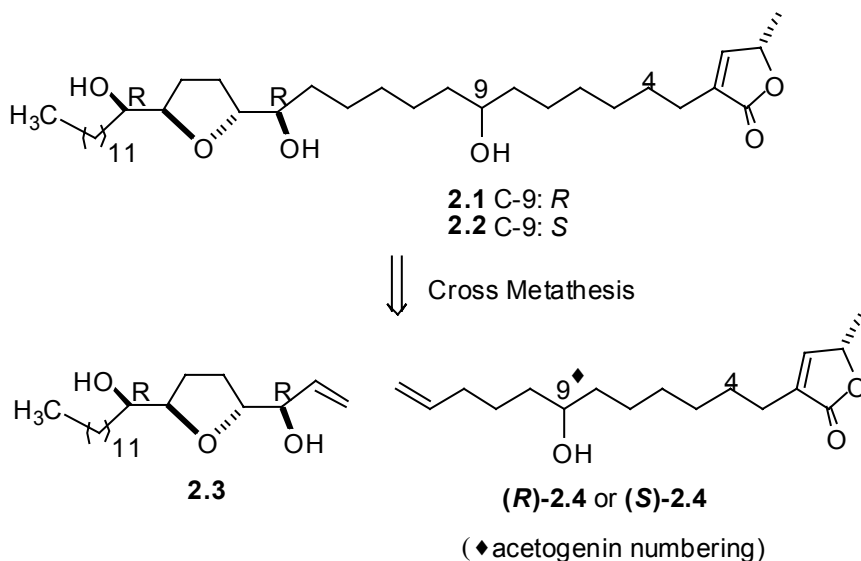
211 suggested that the location might be at C-9. A high resolution EIMS of the peak at m/z 211 gave the exact mass of m/z 211.1342 (calcd 211.1334), established the molecular composition of the fragment as $C_{12}H_{19}O_3$, and confirmed that hydroxyl group was at C-9. Unfortunately, insufficient material was available to complete Mosher ester derivatives, which would be helpful in resolving the stereochemistry of C-9.

2.2 Synthetic strategy

We were particularly in a synthetic plan that would be practical for assigning an unambiguous structure to 4-deoxyannoreticuin and in the longer term would be applicable to other mono-THF acetogenin congeners. The gross structure of 4-deoxyannoreticuin is represented as in **2.1** or **2.2** but the configuration at C-9 was not determined. The relative stereochemistry of the C-15-C20 portion was deduced as the threo-trans-threo motif on the basis of the known 1H and ^{13}C NMR trends. The absolute stereochemistry of the THF core was assigned as 15*R*, 16*R*, 19*R*, 20*R* because this configuration was determined for the analogous segment in 2,4-*trans*-squamoxinone, a co-isolate from the same plant material, and several other mono-THF containing acetogenins.³⁰ It was also assumed that the configuration at C-34 is *S* by analogy with all the other acetogenins for which this center has been determined.⁹ In order to establish the configuration at C-9, we set as synthetic goals the C-9 epimeric structures **2.1** and **2.2**, with the plan of matching one or the other to the natural product on the basis of its physical properties. Accordingly, a modular synthesis in which a fixed THF segment *x* could be paired with epimeric-C-9-hydroxy-butenolide components (*R*)-4 and (*S*)-4 (or

appropriately protected alcohol derivatives), via an olefin cross metathesis, was conceived (**Scheme 2.1**).^{33,34}

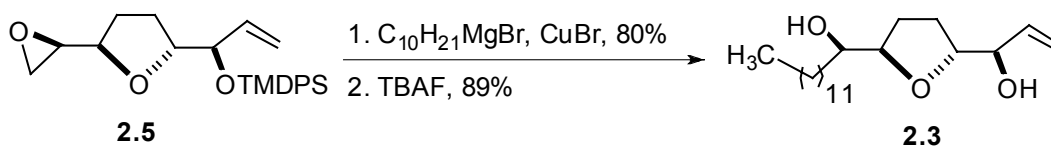
SCHEME 2.1 Retrosynthesis for 4-deoxyannoreticuin.



2.3 Synthesis

The synthesis of the THF segment followed our earlier work on the analog of **2.3** that contained a hydrocarbon chain with two carbons less.^{33b} Thus, the copper (I)-mediated reaction of undecylmagnesium bromide with the known epoxide **2.5**, followed by desilylation of the product provided diol **2.3** (**Scheme 2.2**).

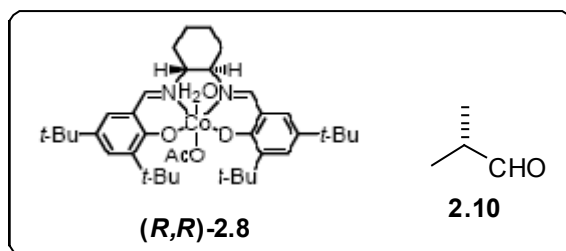
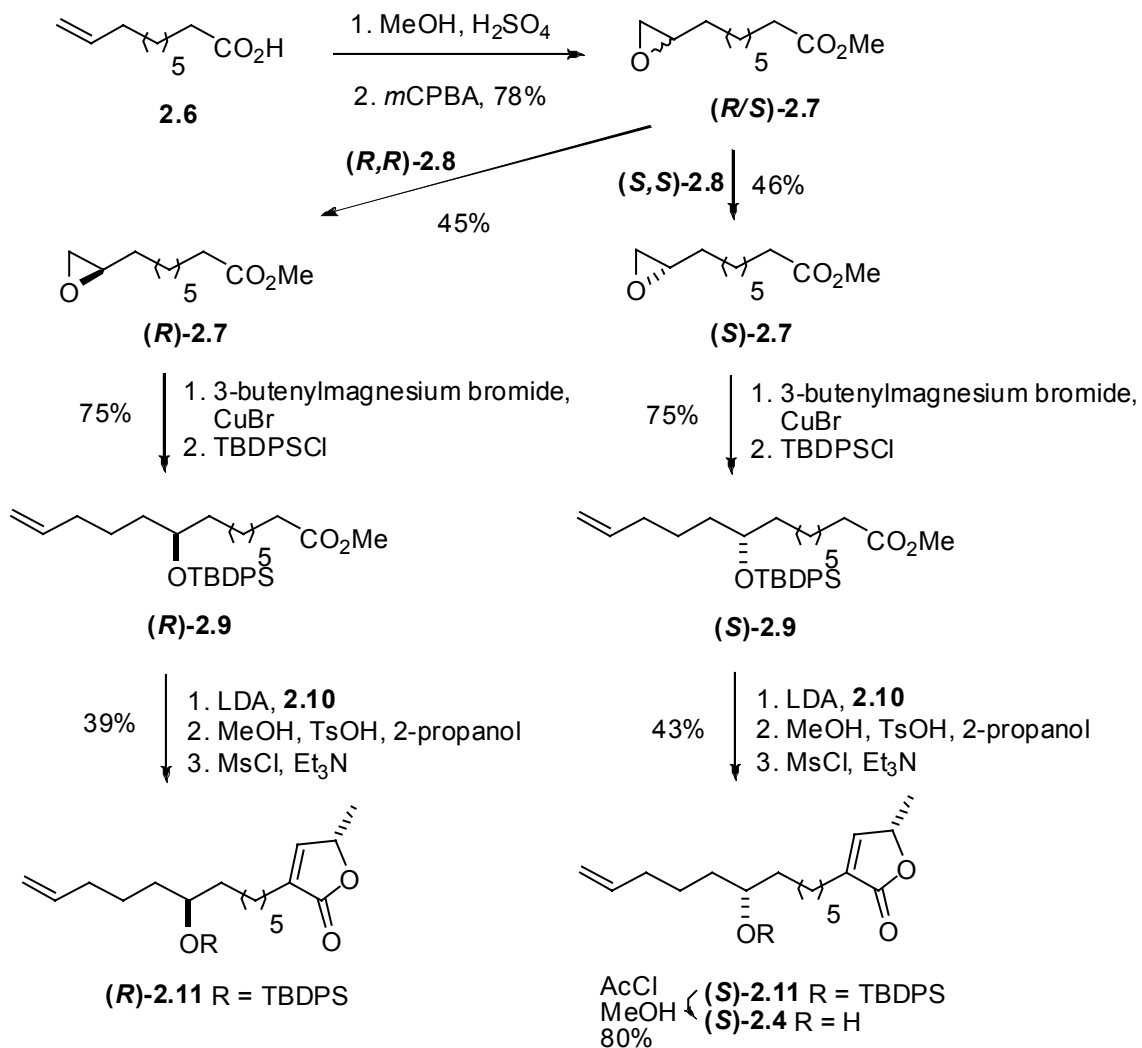
SCHEME 2.2 Synthesis of THF segment.



The epimeric butenolide segment (*R*)-**2.4** and (*S*)-**2.4**, originated from the racemic mixture of epoxides *rac*-**2.7** are derived from the reaction of methyl dec-9-enoate with *m*CPBA (**Scheme 2.3**). Treatment of *rac*-**2.7** under Jacobson's Hydrolytic Kinetic Resolution (HKR) conditions using (*R, R*)-**2.8**, or its enantiomer gave either (*R*)-**2.7** or (*S*)-**2.7** in 45% and 46% yields, respectively.³⁵ Epoxides (*R*)-**2.7** and (*S*)-**2.7** were next converted to the silyl ethers (*R*)-**2.9** and (*S*)-**2.9**, respectively, via a two-step epoxide opening-alcohol silylation sequence. The *ee* of these products were deduced to be over 95% by H¹ NMR analysis of their Mosher ester derivatives.³⁶ Silyl ether (*R*)-**2.9** and (*S*)-**2.9** were individually partnered with aldehyde **2.10** and forwarded into a known aldol reaction-elimination protocol to give butenolide alkenes (*R*)-**2.11** and (*S*)-**2.11**, respectively, in 39% and 43% overall yield from (*R*)-**2.9** and (*S*)-**2.9**.³⁷ To compare the use of protected versus deprotected alcohol butenolide partners in the CM and subsequent reactions (*vide infra*), one of the silyl ether (*S*)-**2.11** was converted to the derived alcohol (*S*)-**2.4**.

The CM and subsequent steps were first evaluated for TBDPS-protected butenolide (*R*)-**2.11** (**Scheme 2.4**). The optimal CM conditions were found to be methylene chloride as the solvent, a 1:2 molar ratio of **2.11**: (*R*)-**2.11**, 10 mol % Grubbs II catalyst, at room temperature for approximately 20h. The desired product (*R*)-**2.12** was obtained in 73% isolated yield based on THF segment **2.3**. The homodimer of (*R*)-**2.11** was also isolated in 20% yield together with approximately 15% of unreacted (*R*)-**2.11**. Selective hydrogenation of isolated alkene in (*R*)-**2.12** using RhCl(PPh₃)₃ as a catalyst provided an inseparable 7:1 mixture of the desired hydrogenated product and the completely hydrogenated material in 70% combined yield.

SCHEME 2.3 Synthesis of butenolide segments.

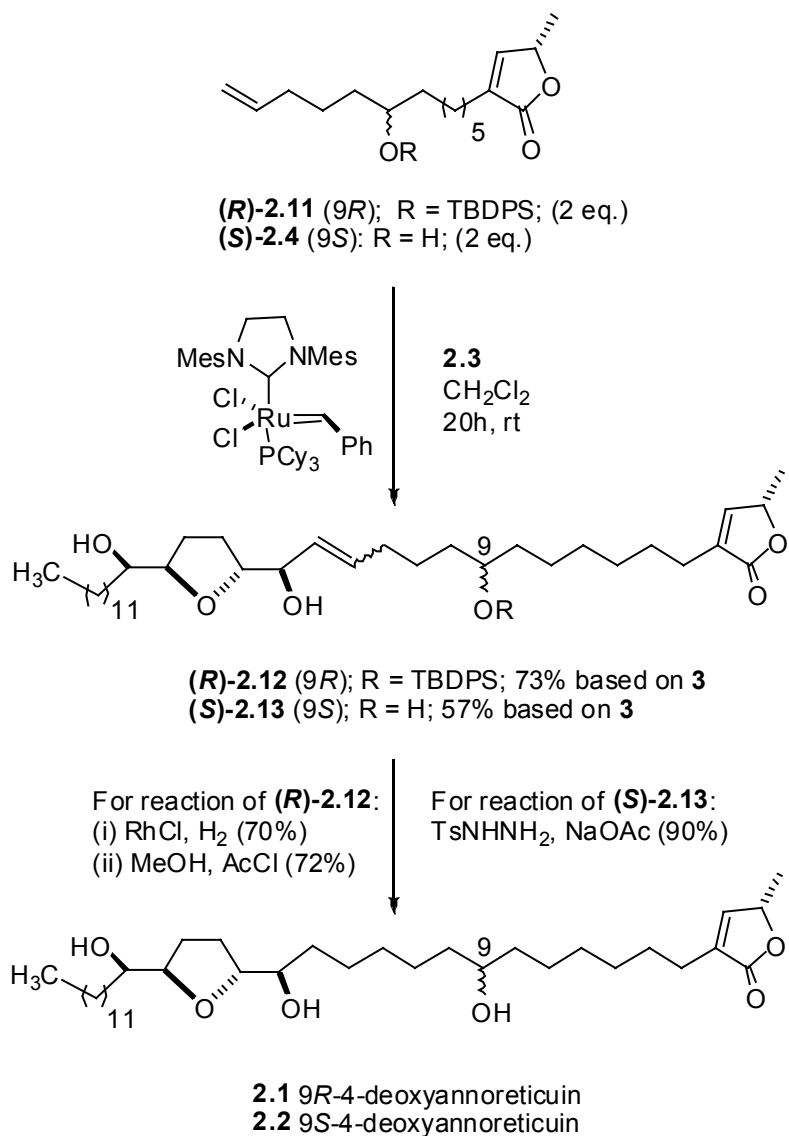


Exposure of this mixture to methanolic HCl led to 9*R*-4-deoxyannoreticuin (**2.1**) that contained approximately 10% of the corresponding over-reduced product. For the

synthesis of 9*S*-4-deoxyannorecticuin (**2.2**), the CM reaction was performed with the desilylated butenolide (*S*)-**2.4** thereby obviating the desilylation step in a more advanced THF-butenolide synthetic intermediate. Accordingly, application of the CM conditions to **2.3** and (*S*)-**2.4** gave (*S*)-**2.13** in 57% isolated yield based on **2.3**. Given the difficulty in controlling the RhCl(PPh₃)₃-catalyzed hydrogenation of (*R*)-**2.12**, the diimide reduction of (*S*)-**2.13** was explored and found to be superior, giving 9*S*-4-deoxyannorecticuin (**2.2**) in 90% yield with no evidence of the over-reduced product.

In the absence of an actual sample of 4-deoxyannorecticuin we attempted to assign the configuration at C-9 in the natural product by comparison of NMR data for synthetic **2.1** and **2.2** with the reported listings for the natural 4-deoxyannorecticuin. All three sets of data were essentially identical, which corroborated the gross structure assigned to the natural product, but at the same time made assignment of the C-9 configuration impossible. This scenario was not completely surprising giving the close similarity (albeit noticeably different) of the NMR data for the analogous epimers of corossolin, the C-10 regioisomer of 4-deoxyannorecticuin.^{38,39} Similar observations have been made in a recent rigorous investigation on acetogenin diastereomers that vary with respect to the configuration at the remote stereogenic centers. Examination of the optical rotations were also inconclusive because the values obtained for **2.1** and **2.2** were appreciably different from that reported for the natural product, ($[\alpha]_D = + 19^\circ$, $+ 25^\circ$ and $+ 6.8^\circ$ for **2.1**, **2.2** and natural 4-deoxyannorecticuin, respectively). Parenthetically, the optical rotation for **2.1** and **2.2** were essentially identical to the values observed for the respective C-10 epimers of corossolin.³⁸

SCHEME 2.4 Cross metathesis and final products.



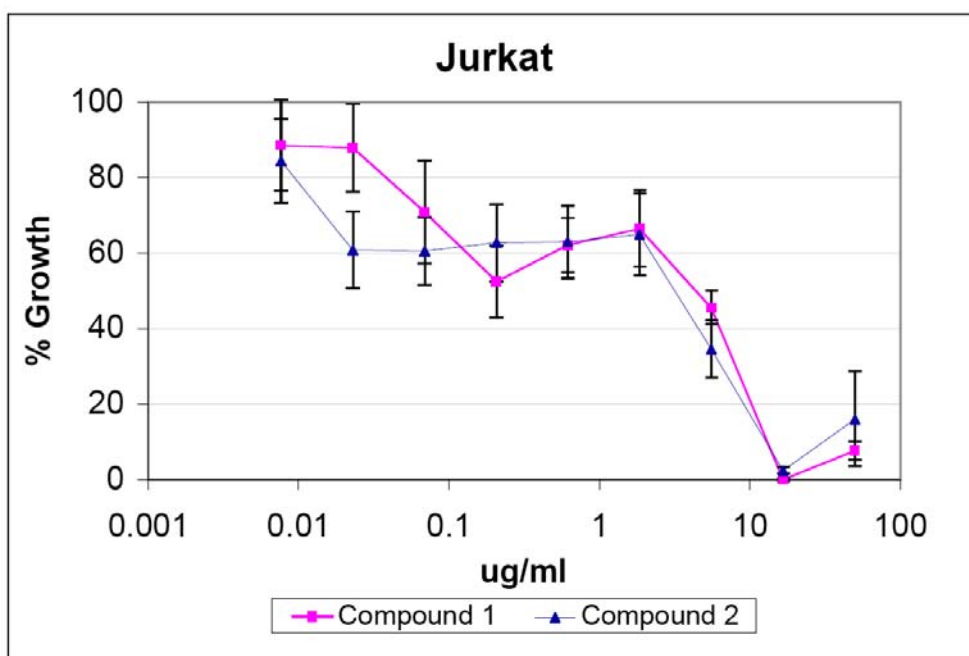
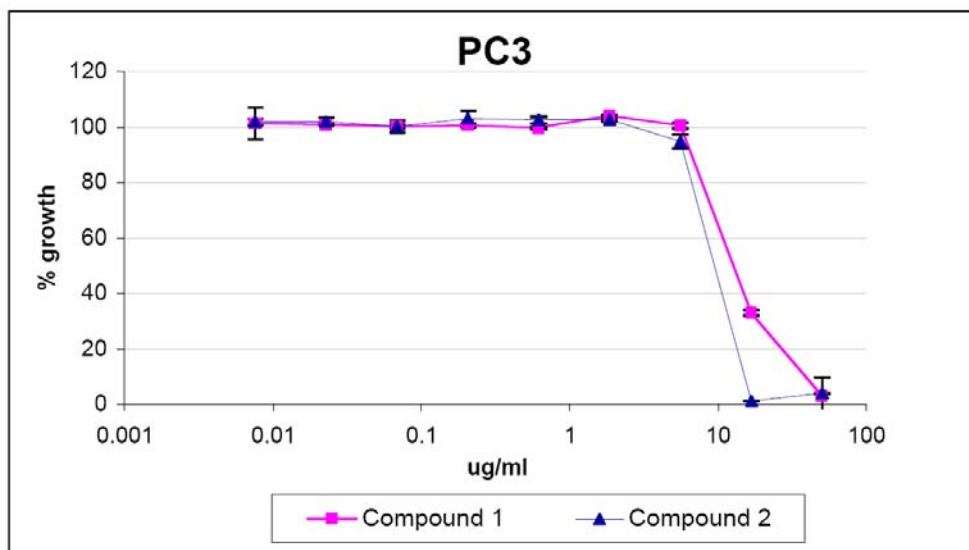
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respectively). Parenthetically, the optical rotation for **2.1** and **2.2** were essentially identical to the values observed for the respective C-10 epimers of corossolin.³⁸

2.4 Biological assay

In the case of corossolin,³⁸ the cytotoxic activities of the C-10 epimers were significantly different. Therefore, we next attempted to identify **2.1** or **2.2** with the natural product by comparing their cytotoxicity against PC-3 (human prostate) with the expectation that the activity of one or the other would more closely match the reported data for the natural product. The GI₅₀ values for **2.1** and **2.2** were found to be approximately 15 and 10 µg/mL in a 7-day MTT assay. Thus, preliminary biological evaluation was also not structurally insightful. Epimers **2.1** and **2.2** also showed similar activity against Jurkat cells (human T-cell leukemia), with GI₅₀ values in the range of 5-10 µg/mL. These data suggest that the notion a change in the configuration at such remoter carbinol centers leads to significant difference in cytotoxic activity may hold only for certain cell lines and is not general.

MTS assays for 1 and 2 against PC-3 and Jurkat cell lines



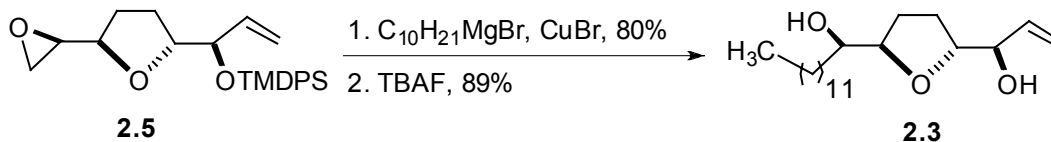
2.5 Conclusion

In conclusion we have developed a highly convergent synthesis for the mono-THF containing acetogenins that is based on the CM coupling of THF and butenolide components. This methodology was applied to the epimeric analogues of the natural product 4-deoxyannoreticuin, in an attempt to resolve the unassigned configuration at C-9. Unfortunately, identification of one or the other epimeric structures with the natural product was not possible because of the closeness of the physical data for all three compounds, illustrating the limitation of this strategy for the assignment of stereochemistry at remote stereogenic centers. In contrast to observations made for closely related THF acetogenins, the cytotoxicity of the epimeric analogues of 4-deoxyannoreticuin was found to be very similar.

2.6 Experimental section

General. Solvents were purified by standard procedures or used from commercial sources as appropriate. Petroleum ether refers to the fraction of petroleum ether boiling between 40 and 60 °C. Ether refers to diethyl ether. Unless otherwise stated thin layer chromatography (TLC) was done on 0.25 mm thick precoated silica gel 60 (HF-254, Whatman) aluminium sheets and flash column chromatography (FCC) was performed using Kieselgel 60 (32-63 mesh, Scientific Adsorbents). Elution for FCC usually employed a stepwise solvent polarity gradient, correlated with TLC mobility. Chromatograms were observed under UV (short and long wavelength) light, and/or were visualized by heating plates that were dipped in a solution of ammonium (VI) molybdate tetrahydrate (12.5 g) and cerium (IV) sulfate tetrahydrate (5.0 g) in 10% aqueous sulphuric acid (500 mL). Optical rotations ($[\alpha]_D$) were recorded using a Rudolph Autopol III polarimeter which has a thermally jacketed 10 cm cell (path length of 1 dm) and are given in units of 10^{-1} deg cm^2g^{-1} at 589 nm (sodium D-line). NMR spectra were recorded using either Varian Unity Plus 500 or Bruker Ultra Shield instruments (^1H and ^{13}C ; 500 and 125 MHz respectively). Spectra were recorded in CDCl_3 solutions with residual CHCl_3 as internal standard (H 7.27 and C 77.0 ppm). Chemical shifts are quoted in ppm relative to tetramethylsilane (H 0.00) and coupling constants (J) are given in Hertz. First order approximations are employed throughout. High-resolution mass spectrometry was performed on an Ultima Micromass Q-ToF instrument at the Mass Spectrometry Laboratory of the University of Illinois, Urbana-Champaign.

Synthesis of (2R, 5R, 1'R, 1''R)-2-[1'-Hydroxy-2'-propenyl]-5-[1''-hydroxytridecyl]-tetrahydrofuran (2.3)

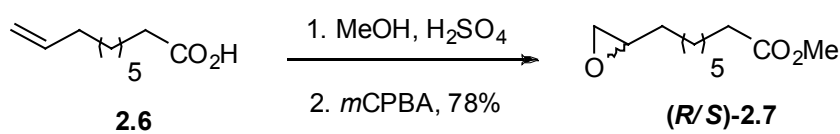


In a 50 mL round-bottom flask equipped with magnetic stirring bar, were placed CuBr (36.3 mg, 0.253 mmol) and anhydrous THF (10 mL). Pre-prepared, $C_{11}H_{23}MgBr$ (11.5 mL, ca 0.4 M in THF) was added dropwise at 0 °C, and then epoxide **2.5** (120 mg, 0.422 mmol) was introduced. The reaction was stirred at 0 °C for 3 h, then poured into ice cold saturated aqueous NH_4Cl , and extracted with ether. The organic phase was washed with brine, dried (Na_2SO_4), filtered, and evaporated under reduced pressure. The residue was purified by FCC (2-5% EtOAc:petroleum ether) to give the derived alcohol (160 mg, 86%): $R_f = 0.45$ (5% EtOAc: petroleum ether); 1H NMR ($CDCl_3$) δ 5.81 (m, 1H), 5.24-5.28 (dd, 2H, $J = 17.1$ Hz, 10.6 Hz), 4.10 (t, 1H), 3.88-3.89 (q, 1H), 3.75-3.76 (q, 1H), 3.35 (m, 1H), 2.31 (d, 1H, $J = 4.1$ Hz), 1.86-1.89 (m, 2H), 1.52-1.70 (m, 2H), 1.36 (m, 1H), 1.36-1.38 (m, 2H), 1.23 (s, br, 20H), 0.88 (s, br, 9H), 0.85 (t, $J = 6.7$ Hz, 3H), 0.05 (s, 3H), 0.03 (s, 3H); ^{13}C NMR ($CDCl_3$) δ 137.6, 115.7, 82.7, 82.4, 76.0, 74.0, 33.6, 31.9, 29.7, 29.7, 29.6, 29.6, 29.6, 29.4, 28.4, 27.8, 25.8, 25.6, 22.7, 18.3, 14.1, 1.0, -4.6, -4.8; HRMS (FAB) calcd for $C_{26}H_{53}O_3Si$ ($M + H^+$) 441.3764, found 441.3744.

To a solution of the product from the previous step (64 mg, 0.145 mmol) in THF (10 mL) was added Bu_4NF (1.45 mL, 1M in THF) at 0 °C. The reaction was stirred at rt until TLC indicated complete disappearance of the starting material, at which time the mixture was diluted with water and extracted with EtOAc. The organic phase was dried (Na_2SO_4),

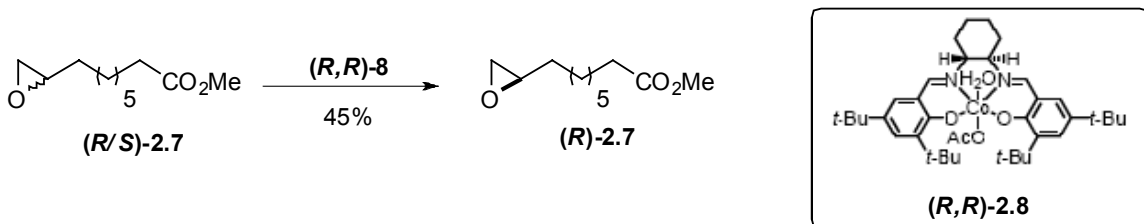
filtered and evaporated under reduced pressure. The residue was purified by FCC to give **2.3** (40 mg, 85%): white powder; $R_f = 0.3$ (25% EtOAc: petroleum ether); $^1\text{H NMR } \delta$ (CDCl_3) 5.78 (m, 1H), 5.18-5.36 (dd, 2H, $J = 17.5, 10.5$ Hz), 3.93 (m, 1H), 3.80-3.88 (m, 2H), 3.39 (m, 1H), 2.45 (b, 1H), 2.25 (b, 1H), 1.94-1.98 (m, 2H), 1.68-1.73 (m, 2H), 1.40-1.41 (m, 1H), 1.29-1.38 (m, 3H), 1.24 (s, br, 20H), 0.86 (t, $J = 6.8$ Hz, 3H); $^{13}\text{C NMR } \delta$ (CDCl_3) 136.7, 117.1, 82.9, 82.2, 75.5, 74.0, 33.6, 31.9, 29.7, 29.7, 29.7, 29.6, 29.6, 29.6, 29.3, 28.5, 28.4, 25.6, 22.7, 14.1; HRMS (FAB) calcd for $\text{C}_{20}\text{H}_{38}\text{O}_3\text{Na}$ ($\text{M} + \text{Na}^+$) 349.2719, found 349.2708.

Synthesis of methyl 8-(oxiran-2-yl)octanoate (**2.7**)



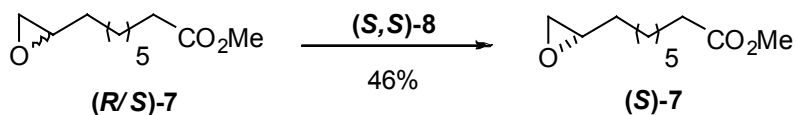
To a solution of methyl dec-9-enoate (1.40 g, 7.60 mmol) in CH_2Cl_2 (25 mL) was added *m*CPBA (4.26 g, 19 mmol, 77% purity) at room temperature. The mixture was stirred for 1h and quenched by adding saturated aqueous NaHCO_3 and saturated aqueous $\text{Na}_2\text{S}_2\text{O}_3$, and extracted with EtOAc. The organic phase was dried (Na_2SO_4), filtered and evaporated under reduced pressure. The residue was purified by FCC to give **2.7** (1.31 g, 86.4%): as a yellow oil; $R_f = 0.70$ (10% EtOAc: petroleum ether); $^1\text{H NMR } \delta$ (CDCl_3) 3.69 (s, 3H), 2.92 (m, 1H), 2.75-2.77 (m, 1H), 2.47-2.48 (m, 1H), 2.32 (t, $J = 7.5$ Hz, 2H), 1.60-1.62 (m, 2H), 1.34-1.55 (m, 10H); $^{13}\text{C NMR } \delta$ (CDCl_3) 174.2, 52.3, 51.4, 47.1, 34.1, 32.4, 29.2, 29.1, 29.0, 25.9, 24.9.

Synthesis of (R)- Methyl 8-(oxiran-2-yl)octanoate [(R)-2.7]



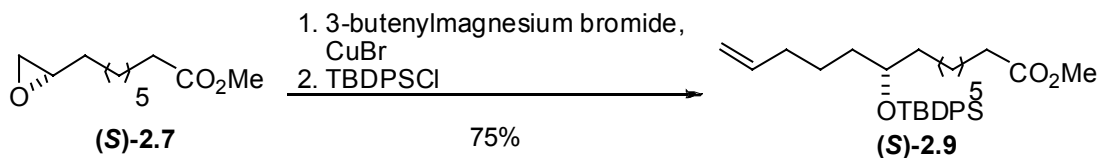
A solution of **(R/S)-2.7** (1.83 g, 9.17 mmol) in DME (10 mL) was treated with **(R,R)**-salen-Co(OAc) (182 mg, 0.275 mmol) and distilled water (0.091 mL, 5.06 mmol). The mixture was stirred at room temperature for 72 h, then purified by FCC to give **(R)-2.7** (677 mg, 37%) and the derived diol (765 mg, 38%). The TLC and NMR data for **(R)-2.7** was as listed for **(R/S)-2.7**.

Synthesis of (S)-Methyl 8-(oxiran-2-yl)octanoate [(S)-2.7]



(S)-2.7 (4.2 g, 46%) was obtained from **(R/S)-2.7** (9.2 g, 46 mmol) using **(S, S)-2.8**, following the procedure described for **(R)-2.7**. For **(S)-2.7**: The TLC and NMR data for **(S)-2.7** was as listed for **(R/S)-2.7**.

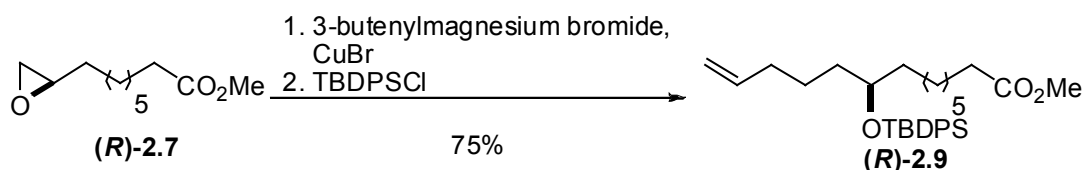
Synthesis of (S)-methyl 9-tert-butylidiphenylsilyloxytetradec-13-enoate [(S)-2.9]



To a suspension of CuBr (258 mg, 1.8 mmol) in THF (15 mL) at 0 °C was added 3-butenylmagnesium bromide (18 mL, 0.5 M in THF, 9.0 mmol) over 5 min. A solution of (**S**)-**2.7** (600 mg, 3.00 mmol) in anhydrous THF (3 mL) was then introduced, dropwise over 5 min. The reaction was stirred at 0 °C for 10 min at which time saturated aqueous NH₄Cl was added and the mixture extracted with ether. The organic phase was washed with brine, dried (Na₂SO₄), filtered, and evaporated under reduced pressure. The residue was purified by FCC to give the derived alcohol: R_f = 0.50 (20% EtOAc: petroleum ether).

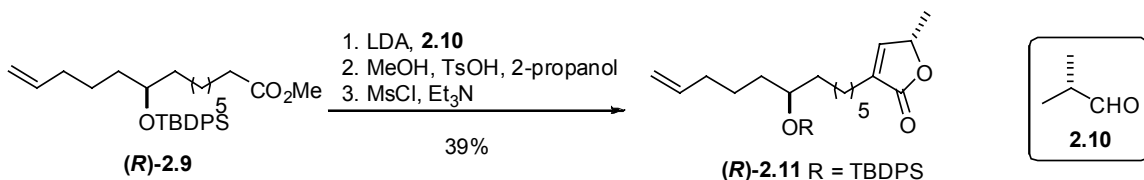
A solution of the product from the previous step (358 mg, 1.4 mmol), TBDPSCl (769 mg, 2.8 mmol) and imidazole (477 mg, 7.00 mmol) in anhydrous CH₂Cl₂ (20 mL) was stirred under nitrogen at room temperature for 2 h. The mixture was quenched with brine and extracted with ether (3 x 50 mL). The organic phase was dried (Na₂SO₄) filtered, and evaporated under reduced pressure. The residue was purified by FCC to afford **2.9** (600 mg, 87%): R_f = 0.50 (5% EtOAc: petroleum ether); ¹H NMR (CDCl₃) δ 7.69-7.70 (m, 4H), 7.37-7.44 (m, 6H), 4.91-4.96 (m, 2H), 3.75 (m, 1H), 3.70 (s, 3H), 2.30 (t, *J* = 7.5 Hz, 2H), 1.92-1.93 (m, 2H), 1.59-1.60 (m, 2H), 1.40-1.45 (m, 6H), 1.14-1.23 (m, 8H); ¹³C δ NMR (CDCl₃) 174.5, 139.2, 136.2, 135.1, 135.0, 129.6, 127.6, 114.5, 77.0, 73.3, 51.6, 36.5, 36.0, 34.3, 34.0, 29.7, 29.3, 29.3, 27.3, 25.2, 25.0, 24.4, 19.7.

Synthesis of (**R**)-methyl 9-tert-butylidiphenylsilyloxytetradec-13-enoate [(**R**)-**2.9**]



(*R*)-**2.9** (600 mg, 75%) was obtained from (*R*)-**2.7** (324 mg, 1.62 mmol) following the two step reaction sequence described for (*S*)-**2.9**. The TLC and NMR data for (*R*)-**2.9** was as listed for (*S*)-**2.9**.

Synthesis of butenolide (*R*)-**2.11**

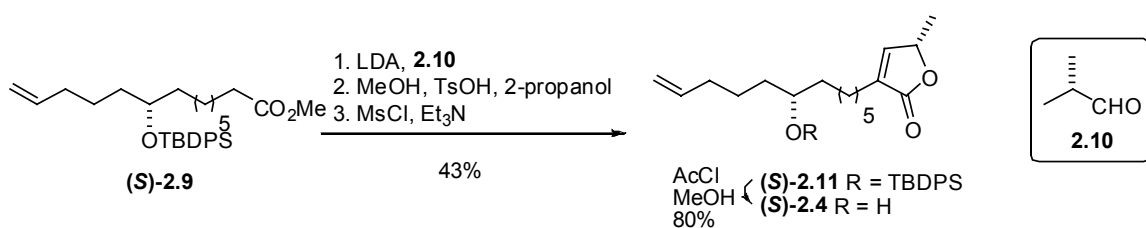


To a solution of diisopropylamine (1.2 mL, 8.5 mmol) in anhydrous THF (15 mL) at -78 °C, was added BuLi (2.4 mL, 2.5 M in hexane, 6 mmol). The mixture was warmed to 0 °C and stirred at this temperature for 10 min, then cooled to -78 °C, at which time a solution of (*R*)-**2.9** (600 mg, 1.21 mmol) in anhydrous THF (6 mL) was added. The reaction mixture was stirred at -78 °C for 1 h and a solution of aldehyde **2.10** (574 mg, 3.63 mmol) in anhydrous THF (6 mL) was slowly introduced. After 30 min, the reaction was quenched with saturated aqueous NH₄Cl and extracted with ether. The organic extract was dried (Na₂SO₄) and concentrated under reduced pressure. The residue was purified by FCC to afford the aldol product (400 mg, 70% based on consumed (*R*)-**2.9**).

To the solution of product from the previous step (400 mg, 0.613 mmol) in a mixture of MeOH (10 mL) and 2-propanol (1 mL) was added TsOH (12 mg, 0.06 mmol) at 0 °C. The reaction mixture was stirred at room temperature for 20 h, then concentrated under reduced pressure. The residue was purified by FCC to afford the derived lactone as a yellow oil (227 mg, 69%).

To a solution of the material from the previous step (227 mg, 0.42 mmol) in CH₂Cl₂ (10 mL) was added Et₃N (0.3 mL, 2.1 mmol) and MsCl (0.07 mL, 0.84 mmol) at 0 °C. After stirring at room temperature for 14 h, the reaction was diluted with saturated aqueous NH₄Cl and extracted with ether. The organic phase was dried (Na₂SO₄) and concentrated under reduced pressure. The residue was purified by FCC to afford (**R**)-**2.11** (176 mg, 81%): R_f = 0.75 (15% EtOAc: petroleum ether); ¹H NMR (CDCl₃) δ 7.66-7.68 (m, 4H), 7.35- 7.41 (m, 6H), 6.94 (d, *J* = 1.5 Hz, 1H), 5.69- 5.72 (m, 1H), 4.97-4.99 (m, 1H), 4.88- 4.94 (m, 2H), 3.70-3.74 (m, 1H), 2.20- 2.24 (m, 2H), 1.90-1.91 (m, 2H), 1.38-1.47 (m, 11H), 1.14- 1.123 (m, 6H), 1.05 (s, 9H); ¹³C NMR (CDCl₃) δ 173.8, 148.8, 138.9, 135.9, 134.8, 134.7, 134.3, 129.4, 129.4, 127.4, 114.3, 73.1, 36.3, 35.8, 33.8, 29.3, 29.1, 27.3, 27.1, 25.1, 24.7, 24.2, 19.4, 19.2; HRMS (FAB) calcd for C₃₃H₄₆O₃Si (M + H⁺) 519.3294, found 519.3282.

Synthesis of butenolide (**S**)-**2.11**



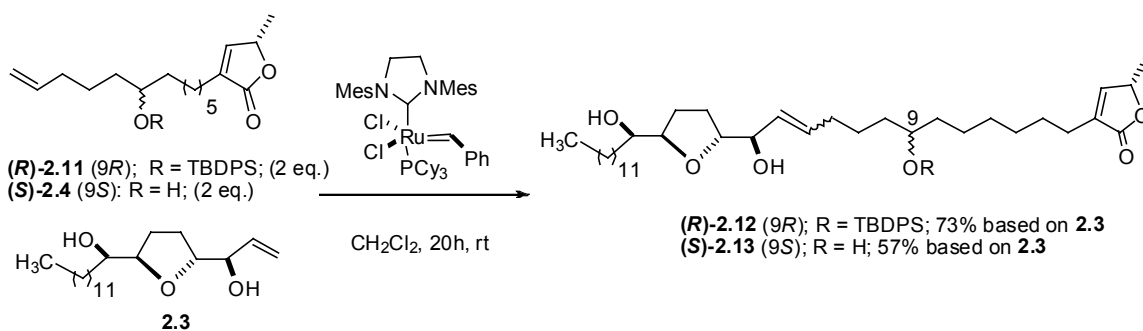
R_f = 0.70 (15% EtOAc: petroleum ether); ¹H NMR (CDCl₃) δ 7.69- 7.70 (m, 4H), 7.37- 7.45 (m, 6H), 6.96 (d, *J* = 1.4 Hz, 1H), 5.71- 5.76 (m, 1H), 5.00- 5.01 (m, 1H), 4.91- 4.96 (m, 2H), 3.74- 3.76 (m, 1H), 2.23- 2.26 (t, *J* = 8.1 Hz, 2H), 1.92- 1.93 (m, 2H), 1.25- 1.49 (m, 11H), 1.16- 1.23 (m, 6H), 1.08 (s, 9H); ¹³C NMR (CDCl₃) δ 173.8, 148.7, 138.9, 135.9, 134.8, 134.7, 134.4, 129.4, 127.4, 114.3, 73.1, 36.3, 35.8, 33.8, 29.3, 29.1, 27.3,

27.1, 25.1, 24.7, 24.2, 19.4, 19.2; HRMS (FAB) calcd for C₃₃H₄₆O₃Si (M + H⁺) 519.3294, found 519.3293.

Synthesis of butenolide (**S**)-2.4

5% AcCl in MeOH (2.1 mL) was added at room temperature to a solution of (**S**)-**2.11** in CH₂Cl₂ (5 mL). The mixture was stirred for 16 h, then diluted with CH₂Cl₂ and washed with saturated aqueous NaHCO₃. The organic phase was dried (Na₂SO₄) and concentrated under reduced pressure. FCC of the residue afforded (**S**)-**2.4** (50 mg, 82%): R_f = 0.45 (30% EtOAc: petroleum ether); ¹H NMR (CDCl₃) δ 7.00 (s, 1H), 5.81- 5.86 (m, 1H), 4.97- 5.06 (m, 3H), 3.61- 3.62 (m, 1H), 2.27- 2.31 (m, 2H), 2.02- 2.11 (m, 2H), 1.37- 1.58 (m, 17H); ¹³C NMR (CDCl₃) δ 173.8, 148.8, 138.7, 134.3, 114.6, 71.8, 37.4, 36.9, 33.7, 29.3, 29.1, 27.4, 25.5, 25.2, 24.9, 19.2; HRMS (FAB) calcd for C₁₇H₂₈O₃ (M + H⁺) 281.2117, found 281.2111.

Synthesis of THF-butenolide (**R**)-2.12



Grubb's II catalyst (6 mg, 0.007 mmol) in CH₂Cl₂ (1 mL) was injected, at rt, into a degassed solution of alcohol **2.3** (23 mg, 0.07 mmol) and (**R**)-**2.11** (74 mg, 0.14 mmol), in CH₂Cl₂ (2 mL). The reaction was stirred for 20 h at rt, then quenched by addition of

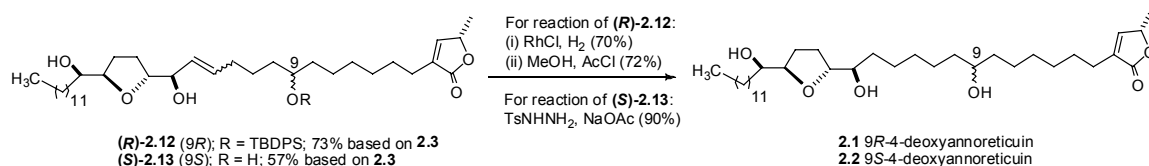
DMSO (50 μ L) and concentrated *in vacuo*. FCC of the residue afforded (**R**)-**2.12** (42 mg, 73% based on **2.3**), recovered (**R**)-**2.11** (ca 15%) and the homodimer (**R**)-**2.11** (ca 20 %). For (**R**)-**2.12**: R_f = 0.50 (30% EtOAc: petroleum ether); ^1H NMR (CDCl_3) δ 7.65- 7.67 (m, 4H), 7.34- 7.42 (m, 6H), 6.94 (s, 1H), 5.65- 5.68 (m, 1H), 5.32- 5.34 (m, 1H), 4.97- 4.99 (m, 1H), 3.81- 3.85 (m, 3H), 3.70- 3.73 (m, 1H), 3.41 (m, 1H), 2.43- 2.44 (m, 1H), 2.20- 2.21 (m, 3H), 1.88- 1.89 (m, 4H), 1.20- 1.47 (m, 36H), 1.04 (s, 9H), 0.89 (t, J = 3.2 Hz, 3H); ^{13}C NMR (CDCl_3) δ 173.8, 149.5, 149.3, 149.1, 148.7, 135.9, 134.7, 134.5, 134.3, 129.4, 128.3, 127.4, 123.1, 123.0, 122.7, 82.7, 82.6, 75.6, 74.0, 73.0, 33.6, 31.9, 29.7 (three signals), 29.6 (three signals), 29.3 (two signals), 29.1, 28.4 (two signals), 27.3, 27.1, 25.1, 22.7, 19.2, 14.1; HRMS (FAB) calcd for $\text{C}_{51}\text{H}_{80}\text{O}_6\text{Si}$ ($\text{M} + \text{Na}^+$) 839.5622, found 839.5613.

Synthesis of THF-butenolide (**S**)-**2.13**

Grubb's II catalyst (9.3 mg, 0.01 mmol) in CH_2Cl_2 (1 mL) was injected, at rt, into a degassed solution of alcohol **2.3** (36 mg, 0.11 mmol), and (**S**)-**2.4** (50 mg, 0.178 mmol), in CH_2Cl_2 (2 mL). The reaction was stirred for 20 h at rt, then quenched by addition of DMSO (50 μ L) and concentrated *in vacuo*. FCC of the residue provided (**S**)-**2.13** (36 mg, 57% based on **2.3**), and the homodimer of (**S**)-**2.4** (ca 27%). For (**S**)-**2.13**: R_f = 0.50 (80% EtOAc: petroleum ether); ^1H NMR (CDCl_3) δ 7.01 (s, 1H), 5.76- 5.82 (m, 1H), 5.41-5.46 (m, 1H), 5.02- 5.03 (m, 1H), 3.85- 3.92 (m, 3H), 3.61 (s, br, 1H), 3.42- 3.44 (m, 1H), 2.29 (t, J = 7.5 Hz, 2H), 2.10 (m, 2H), 1.97- 2.01 (m, 2H), 1.28-1.70 (m, 44H), 0.90 (t, J = 6.6 Hz, 3H); ^{13}C NMR (CDCl_3) δ 173.9, 148.9, 134.3, 128.6, 82.8, 82.5, 76.8, 76.7, 75.5, 74.0, 71.7, 37.4, 36.9, 32.3, 31.9, 29.7 (five signals), 29.6, 29.4, 29.3, 29.2, 27.4,

25.6, 25.5, 25.2, 22.7, 19.2, 14.2; HRMS (FAB) calcd for C₃₅H₆₂O₆ (M + H⁺) 579.4614, found 579.4625.

Synthesis of (9*R*)-4-deoxyannoreticuin (2.1)



Chlorotris(triphenylphosphine)-rhodium (I) (7.1 mg, 0.0077 mmol) was added to a degassed solution of (**R**)-**2.12** (23 mg, 0.028 mmol) in a mixture of benzene-EtOH (2 mL, 50% v/v). The mixture was stirred under an atmosphere of hydrogen for 12 h, at which time the solvent was removed under reduced pressure. FCC of the residue gave an inseparable mixture of the 13,14-dihydro- and tetrahydro derivatives of (**R**)-**2.12** (16 mg, 70%): R_f = 0.50 (30% EtOAc: petroleum ether).

A mixture of 5% AcCl in MeOH (0.5 mL) was added at room temperature to a solution of the material obtained in the previous step (10 mg, 12.2 mmol) in CH₂Cl₂ (1 mL). The mixture was stirred at this temperature for 3 h, diluted with CH₂Cl₂, and washed with a saturated aqueous NaHCO₃. The organic layer was dried (Na₂SO₄), filtered, and concentrated under reduced pressure. FCC of the residue afforded a mixture of **2.1** and the completely hydrogenated product (7.2 mg, 72%) in an approximate 7:1 ratio: R_f = 0.35 (60% EtOAc: petroleum ether); Repeated FCC provided a sample of **2.1** that contained less than 10% of the completely reduced material: [α]_D²² +19 (c 2.00, CH₂Cl₂).
¹H NMR (CDCl₃) δ 6.96 (s, 1H), 4.96- 4.98 (m, 1H), 3.76- 3.80 (m, 2H), 3.56 (s, br, 1H), 3.36- 3.41 (m, 2H), 2.25- 2.27 (m, 4H, CH₂, 2xOH, D₂O ex.), 1.96- 1.97 (m, 2H), 1.24-

1.52 (m, 48H), 0.87 (t, $J = 6.8$ Hz, 3H); ^{13}C NMR (CDCl_3) δ 173.8, 148.9, 134.3, 82.7, 77.4, 74.0 (two signals), 71.9, 37.4, 33.5, 33.4, 31.9, 29.7 (three signals), 29.6 (three signals), 29.3, 29.1, 28.7, 27.3, 25.6 (three signals), 25.5, 25.2, 22.7, 19.2, 14.1; HRMS (FAB) calcd for $\text{C}_{35}\text{H}_{64}\text{O}_6$ ($\text{M}+\text{H}^+$) 581.4781, found 581.4794.

Synthesis of (9S)-4-deoxyannoreticuin (2.2)

A solution of sodium acetate (280 mg, 3.41 mmol) in water (5 mL) was added via a syringe pump, over 4 h, to a mixture of (**S**)-**2.13** (24 mg, 0.041 mmol), p-toluenesulfonyl hydrazide (511 mg, 2.74 mmol) and DME (4 mL) at reflux. After cooling to rt, the reaction mixture was poured into water and extracted with EtOAc. The combined organic extract was washed with 2M HCl, water and brine, dried (Na_2SO_4), filtered, concentrated under reduced pressure and purified by FCC to give **2.2** as a white solid (21.5 mg, 90%). $R_f = 0.50$ (80% EtOAc: petroleum ether); $[\alpha]_D^{22} +25$ (c 4.00, CH_2Cl_2); ^1H NMR (CDCl_3) δ 6.96 (s, 1H), 4.95- 4.99 (m, 1H), 3.76- 3.80 (m, 2H), 3.56 (s, br, 1H), 3.36- 3.40 (m, 2H), 2.23- 2.24 (m, 2H), 1.95- 1.96 (m, 2H), 1.66 (m, 2H), 1.23- 1.53 (m, 48H), 0.86 (t, $J = 6.8$ Hz, 3H); ^{13}C NMR (CDCl_3) δ 173.8, 148.9, 134.3, 82.7, 82.6, 77.4, 74.0, 71.9, 37.4, 33.5, 33.4, 31.9, 29.7, 29.6, 29.6, 29.6, 29.3, 29.3, 29.1, 28.7, 27.3, 25.6, 25.6, 25.5, 22.7, 19.2, 14.1; HRMS (FAB) calcd for $\text{C}_{35}\text{H}_{64}\text{O}_6$ ($\text{M} + \text{H}^+$) 581.4781, found 581.4789.

2.7 Cytotoxicity Assay

Cells were cultured in RPMI medium supplemented with 10% fetal bovine serum and penicillin-streptomycin-fungizone mixture (100 units/mL, 100 µg/mL and 0.25 µg/mL respectively) and maintained in a 37 °C humidified 5% CO₂ incubator. On the day before the drug treatment, cells were plated onto each well of 96-well plate at 2,000 cells/well (200 µl of the medium per well). After 24 hours, cells were treated with different concentrations of compounds and incubated for 72 hours. After the incubation, cell growth was evaluated using a CellTiter 96 AQueous One Solution Cell Proliferation Assay (Promega). UV absorption (490 nm) of each well was quantified by SpectraMax Plus 384 microplate reader (Molecular Devices).

CHAPTER THREE

Tandem cyclization of diene-acetals to THF-THP subunits: synthetic studies on muconin

3.1 Syntheses of muconin and jimenezin

Non-classical acetogenins, which are characterized by the presence of a THP ring, represents a minor subgroup in the acetogenin family. They can be classified into three categories: mono-THP (pyrinicin and pyragoncin type),⁴⁰ non-adjacent-bis-THF-THP (mucocin type)⁴¹ and adjacent-bis-THF-THP (jimenezin and muconin type).⁴² The syntheses of these non-classical acetogenins is interesting to synthetic chemists because of their structural uniqueness and potent biological activities. Among the non-classical acetogenins, muconin stands out for its potent and selective in vitro cytotoxicity against PACA-2 ($ED_{50} = 5.4 \times 10^{-4} \mu\text{g/mL}$) and MCF-7 ($ED_{50} = 2.4 \times 10^{-4} \mu\text{g/mL}$).⁴³ Muconin was first isolated by the MacLaughlin group in 1996.⁴³ With eight stereogenic centers, six of which are found in the THF/THP core unit, muconin poses a significant synthetic challenge (**Figure 3.1**). Muconin has attracted considerable synthetic interest, with four published total syntheses.⁴⁴

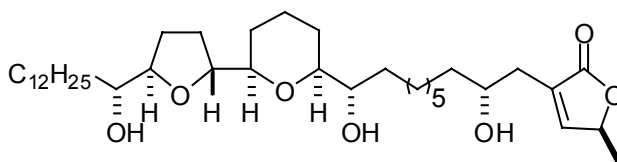
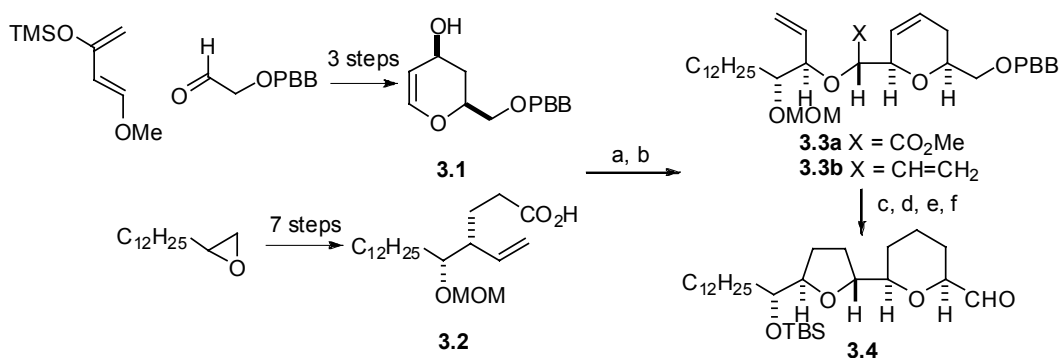


Figure 3.1 Structure of muconin

Jacobsen's group achieved the first total synthesis of muconin in 1998 (**Scheme 3.1**).^{44a} Building block **3.1** was readily accessed through a hetero-Diels-Alder reaction and intermediate **3.2** was obtained in seven steps, starting with the Kinetic Hydrolytic Resolution on racemic tetradecene oxide.³⁵ Esterification of **3.1** with acid **3.2** was effected and the silyl ketene acetal of the ester was generated. Ireland-Claisen

rearrangement⁴⁵ on the latter afforded a 2,6-*cis*-disubstituted dihydropyran, which was isolated as the methyl ester **3.3a** in 81% yield. The methyl ester was converted to the terminal olefin **3.3b** in 70% yield through a one-pot DIBALH reduction/Wittig olefination sequence. The THF ring was constructed in excellent yield through a ring-closing metathesis on **3.3b**. Deprotection of the silyl ether followed by alcohol oxidation with Dess-Martin periodinane furnished THF-THP subunit **3.4**.

SCHEME 3.1

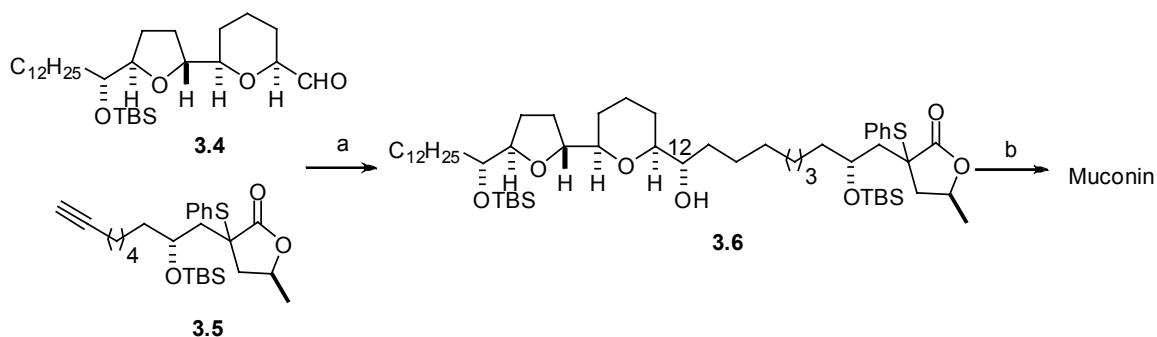


(a) (i) EDC, cat. DMAP, CH₂Cl₂; (ii) LDA, TMSCl, THF/HMPA 4:1; (iii) CH₂N₂, 82%; (b) (i) DIBALH, PhMe; (ii) CH₂=PPh₃, 70%; (c) (i) TMSBr, 4Å molecular sieves, 82%; (ii) TBSOTf, 2,6-lutidine, 99%; (d) 5 mol%, Mo(CHMe₂Ph)(NAr)(OCMe(CF₃)₂)₂, 99%; (e) 10% Pd/C, NaHCO₃, EtOH, H₂(5 atm), 95%; (f) DMP, pyridine, 95%.

The butenolide segment **3.5** was synthesized through a route which has been widely used in acetogenin syntheses (Scheme 3.2).⁴⁶ The key fragment coupling was accomplished by hydroboration of **3.5** and transmetalation, followed by addition of the resulting vinylzinc derivative to aldehyde **3.4**. However, this provided the undesired alcohol configuration at C-12. A swern oxidation/Zn(BH₄)₂ reduction sequence ensured the desired (*S*) stereochemistry at C-12. Oxidation of the thiophenyl group with *m*CPBA, followed by thermally induced elimination to the butenolide and alcohol deprotection

completed the synthesis of muconin. The longest linear sequence was 25 steps and had an overall yield of 7.3%.

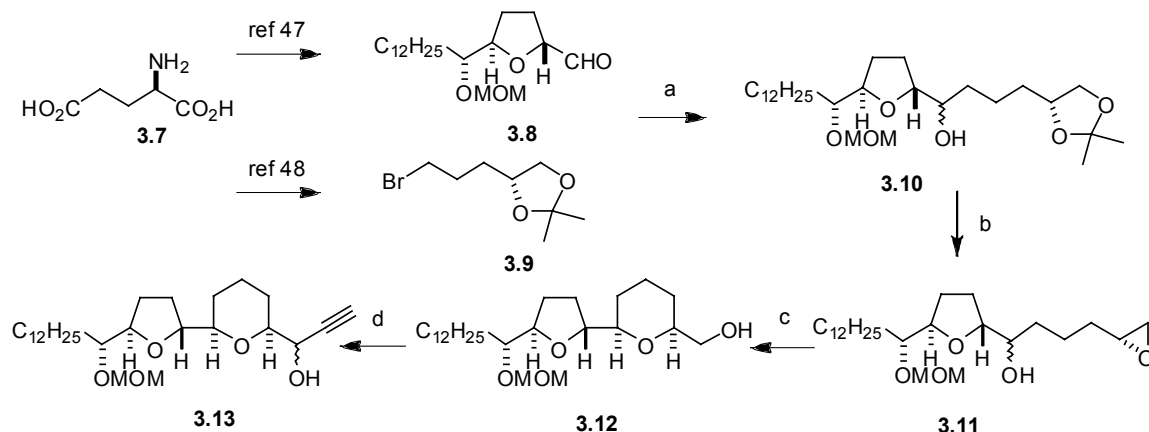
SCHEME 3.2



- (a) (i) (Cyclohexyl)₂BH, hexane; (ii) ZnEt₂, hexane, 73%; (iii) Swern oxidation; (iv) Zn(BH₄)₂, Et₂O, cyclohexane, CH₂Cl₂; (v) 10mol% PtO₂, THF, H₂, 63%; (b) (i) *m*-CPBA, CH₂Cl₂; (ii) PhMe, heat; (iii) 5% AcCl in MeOH, 70%.

Subsequently, the Kitahara's synthesis of muconin was reported (**Scheme 3.3**).^{44b, 44c} The two building blocks **3.8** and **3.9** were prepared from D-glutamic acid (**3.7**).^{47, 48} The chelation-controlled addition of a Grignard reagent derived from **3.9** gave the coupling product **3.10** as an inseparable mixture with a 3:1 diastereoselectivity favoring the desired α -epimer. The mixture was transformed to the corresponding epoxide mixture **3.11**. Treatment of **3.11** with CSA afforded the desired *cis*-THP **3.12** together with the *trans*-THP isomer. Dess-Martin periodinane oxidation of **3.12** gave the aldehyde, which was reacted with ethynylmagnesium bromide to afford the THF-THP segment **3.13** in 80% yield as a 1:1.4 mixture of alcohol epimers.

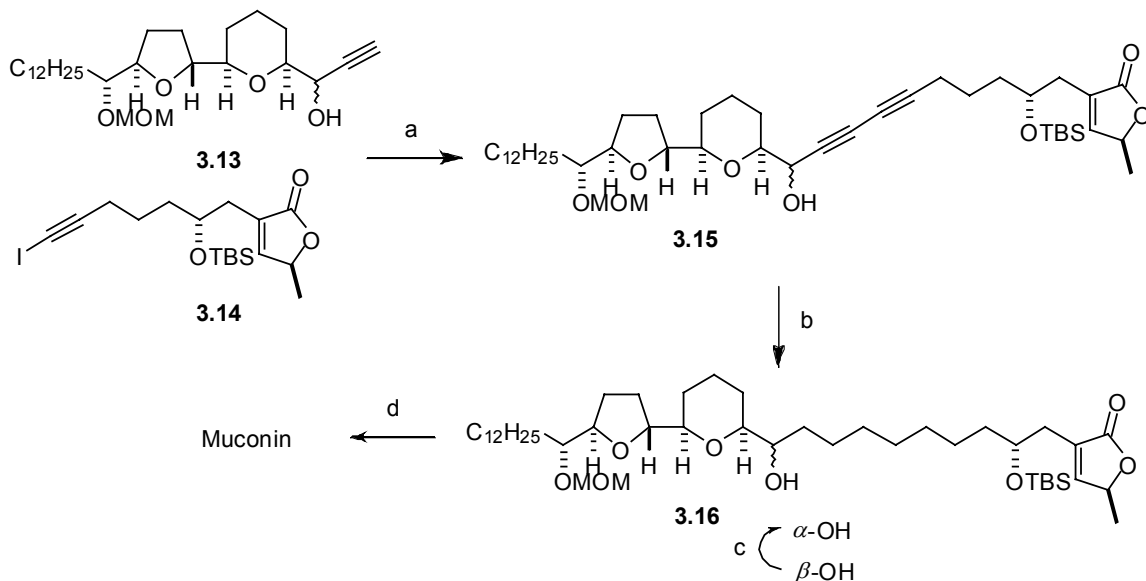
SCHEME 3.3



Palladium(0)-mediated coupling of the alkyne **3.13** with the iodoalkyne **3.14** using Schreiber's protocol afforded a separable mixture of diyne alcohols **3.15** with α/β selectivity of 1/1.4 (**Scheme 3.4**). Following hydrogenation of the diyne, the β alcohol epimer was converted to the desired α product by an oxidation-reduction sequence. Alcohol protecting group removal then furnished muconin with a longest linear step of 24 and 0.73% overall yield.

The Takahashi's synthesis of muconin was reported in 2002 (**Scheme 3.5**).^{44d, 44e} The intermediate **3.17** was prepared by the Sharpless asymmetric oxidation on a triene precursor, which was prepared in 10 steps from ethyl (*E*)-heptadec-4-enoate. Tetraol **3.18** was obtained via a double Sharpless dihydroxylation on the silyl ether derivative of **3.17**.

SCHEME 3.4

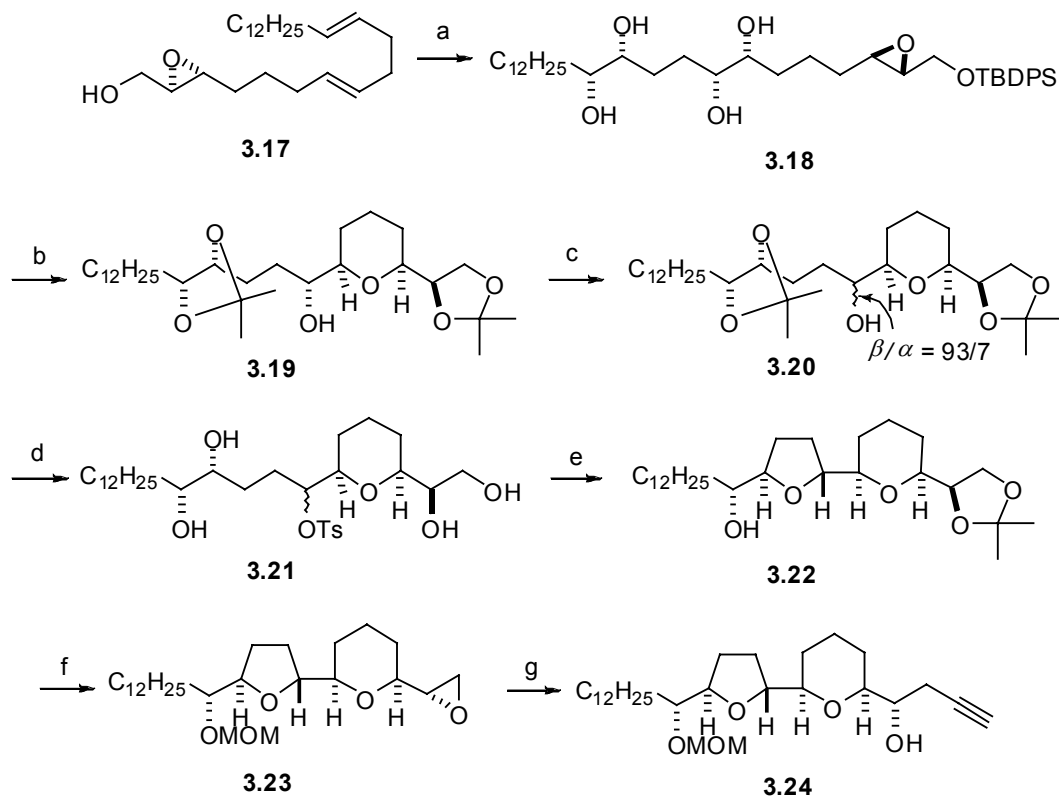


(a) Pd₂(dba)₃, CuI, *i*-Pr₂NH, (2-furyl)₃P, C₆H₆, 55%;
 (b) (Ph₃P)₃RhCl, H₂, 77%; (c) (i) DMP, DMC; (ii) LiAl(O*t*-Bu)₃H,
 THF, 80%; (d) BF₃• Et₂O, Me₂S, 82%.

Exposure of **3.18** to CSA gave the THP resulting from a 6-*exo* cyclization of the hydroxy epoxide. Treatment of this product with dimethoxypropane under acidic condition afforded **3.19** in 85% yield. DMP oxidation of **3.19** followed by Zn(BH₄)₂ reduction of the derived ketone afforded alcohol **3.20** with dominant β -selectivity. Formation of the THF ring in **3.22** was achieved by heating tosylate **3.21** with sodium methoxide in methanol. Compound **3.22** was transformed to **3.24** by a straightforward reaction sequence.

Coupling of fragment **3.24** and butenolide **3.25** provided **3.26**. Selective hydrogenation of **3.26** with Wilkinson's catalyst, followed by global deprotection gave the final product. The longest linear sequence was 28 steps and had an overall yield of 9.7% (**Scheme 3.6**).

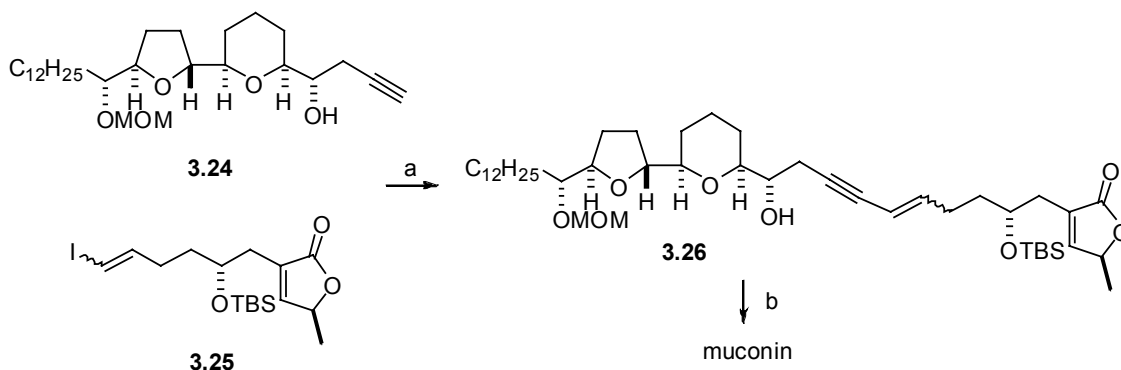
SCHEME 3.5



(a) (i) TBDPSCI, CH_2Cl_2 ; (ii) AD-mix- β , MeSO_2NH_2 , $t\text{-BuOH-H}_2\text{O}$, quant.; (b) CSA, then MeOH, conc., and $(\text{MeO})_2\text{CMe}_2$, 85%; (c) (i) DMP, NaHCO_3 ; (ii) $\text{Zn}(\text{BH}_4)_2$, Et_2O , 98%; (d) (i) $p\text{-TsCl}$, DMAP, Et_3N ; (ii) $\text{AcOH-H}_2\text{O}$, 97%; (e) (i) NaOMe, MeOH, 50°C ; (ii) $(\text{MeO})_2\text{CMe}_2$, CSA, 78%; (f) (i) MOMBr, $i\text{-Pr}_2\text{NEt}$; (ii) $\text{AcOH-H}_2\text{O}$; (iii) BzCl, pyridine, and then MsCl; (iv) aq. NaOH, MeOH-THF, 74%; (g) (i) Lithium (trimethylsilyl)acetylide, $\text{BF}_3\cdot\text{Et}_2\text{O}$, THF; (ii) K_2CO_3 , MeOH, 97%.

The most recent total synthesis of muconin was reported by Yoshimitsu in 2004 (**Scheme 3.7**).^{44f} The synthesis started with (-)-muricatacin (**3.27**), which was obtained through a novel radical $\alpha\text{-C-H}$ hydroxyalkylation on tetrahydrofuran.⁴⁹ Muricatacin was converted to the TBS ether which was subjected to a one-pot reduction/acetylation sequence to an acetylated lactol.

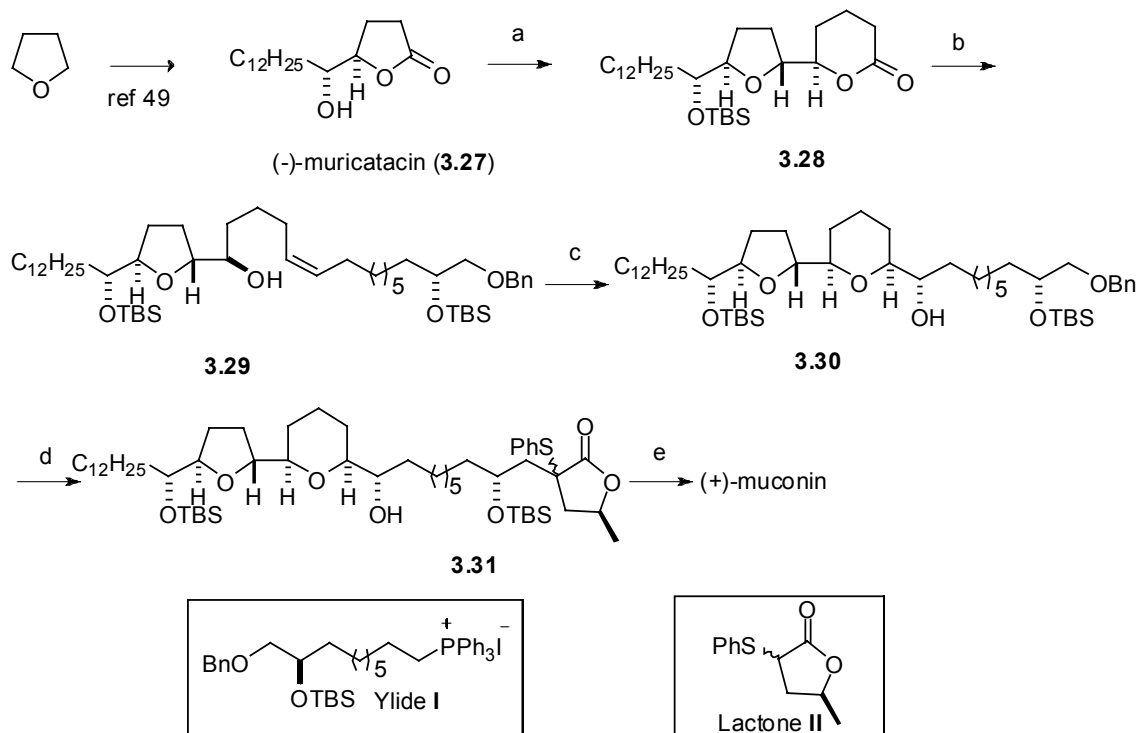
SCHEME 3.6



(a) $(Ph_3P)_2PdCl_2$, CuI, Et_3N , 79%; (b) (i) $(Ph_3P)_3RhCl$, H_2 , benzene-EtOH; (ii) 10% HCl-MeOH, CH_2Cl_2 , 66%.

The Lewis acid-catalyzed reaction of the product with 1-(trimethylsilyloxy)cyclopentene followed by Baeyer-Villiger oxidation of the resulting ketone afforded δ -lactone **3.28**. Reduction of **3.28** with DIBALH provided the lactol derivative, which subsequently underwent Wittig olefination with phosphonium ylide **I** to give the hydroxy *Z*-olefin **3.29** in 83% yield. Olefin **3.29** was oxidized with *m*CPBA to provide a mixture of epoxides, which upon exposure to CSA gave the desired tetrahydropyran **3.30** (39%) along with the diastereomeric THF-THP (59%). The latter was converted to **3.30** by a three-step sequence: alcohol oxidation/base catalyzed epimerization/ketone reduction. Treatment of a triflate derivative of **3.30** with the lithium enolate generated from the lactone **II** gave **3.31** in 86% yield. Compound **3.31** was transformed to the butenolide as described earlier and removal of the silyl ether provided muconin. The synthesis was completed with a longest linear sequence of 19 steps and 2.1% overall yield.

SCHEME 3.7



(a) (i) TBSCl, imidazole, DMF, 98%; (ii) DIBAL, CH₂Cl₂, then Ac₂O, Et₃N, DMAP, 96%; (iii) 1-(trimethylsilyloxy)cyclopentene, propionitrile, 93%; (iv) *mCPBA*, Na₂HPO₄, CH₂Cl₂, 89%; (b) (i) DIBAL, CH₂Cl₂, quant.; (ii) NaHMDS, THF, then ylide I, *n*-BuLi, THF; (c) (i) *mCPBA*, CH₂Cl₂, rt; (ii) CSA, CH₂Cl₂; (d) (i) TBSCl, imidazole, DMF, 98%; (ii) Pd(OH)₂-C, H₂, AcOEt, quant.; (iii) Tf₂O, 2,6-lutidine, CH₂Cl₂, 99%; (iv) LiHMDS, lactone II, HMPA, THF, 86%; (e) (i) *mCPBA*, CH₂Cl₂, (ii) toluene, reflux, 97% overall; (iii) cat. AcCl, MeOH, 96%.

Synthetic studies on the structurally related non-classical acetogenin jimenezin have also been reported (Figure 3.2). Jimenezin was isolated from the seeds of *Rollinia mucosa* and exhibited potent cytotoxic activity against six human solid tumor cell lines.⁴² In 1999, Takahashi's group reported the first total synthesis of jimenezin and corrected the stereochemistry of the THF ring from *trans* to *cis* (Scheme 3.8).⁵⁰

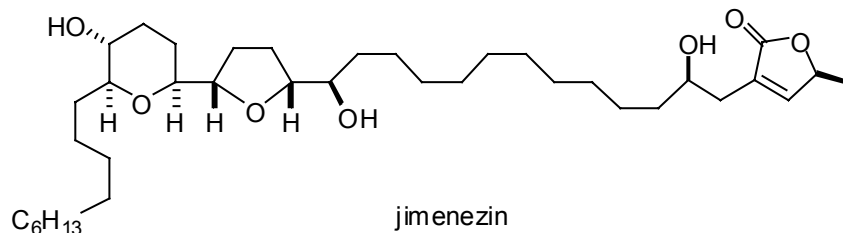
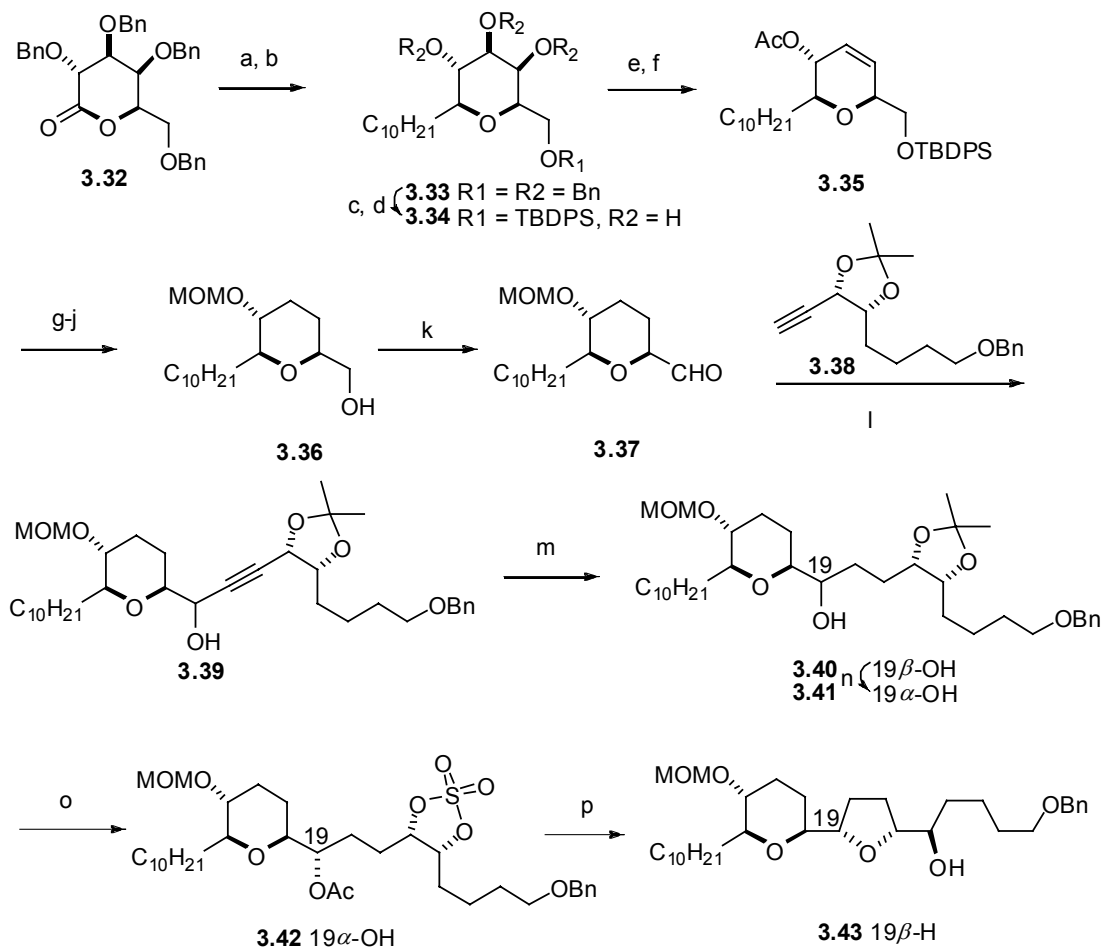


Figure 3.2 Structure of jimenezin

The Takahashi's synthesis was achieved by using carbohydrate **3.32** as the chiral building block.⁵¹ The key steps include an efficient construction of the THF-THP fragment **3.43** through a stereoselective condensation between the pyranyl aldehyde **3.37** and the acetylene derivative **3.38** and a sulfate cyclization to form THF ring in the presence of a ruthenium catalyst. The aldehyde **3.37** was prepared from 2,3,4,6-tetra-*O*-benzyl-D-galactono-1,5-lactone (**3.32**) as follows. Reaction of **3.32** with decylmagnesium bromide in ether afforded a hemiacetal, which was treated with triethylsilane to give a β -glycoside **3.33** in 80% yield. Subsequent debenzylation and mono-silylation of **3.33** afforded a triol **3.34** in 78% yield. Deoxygenation of **3.34** was accomplished through an orthoester to provide an olefin **3.35** in 78% yield, which was converted to a primary alcohol **3.36** by the following sequence: (i) hydrogenation, (ii) deacetylation, (iii) formation of a methoxymethyl ether, (iv) desilylation (93% overall yield). Swern oxidation of **3.36** gave the aldehyde **3.37** in quantitative yield. Coupling reaction of **3.37** and **3.38** was achieved by using THF-HMPA (6:1) as the solvent at -78°C , giving a 92:8 mixture of the undesired β -carbinol and its diastereoisomer in 72% yield. Dess-Martin oxidation of the mixture (**3.40** and **3.41**), and subsequent reduction with L-Selectride in THF produced the **3.41** (94% de) in 97% overall yield.

Scheme 3.8

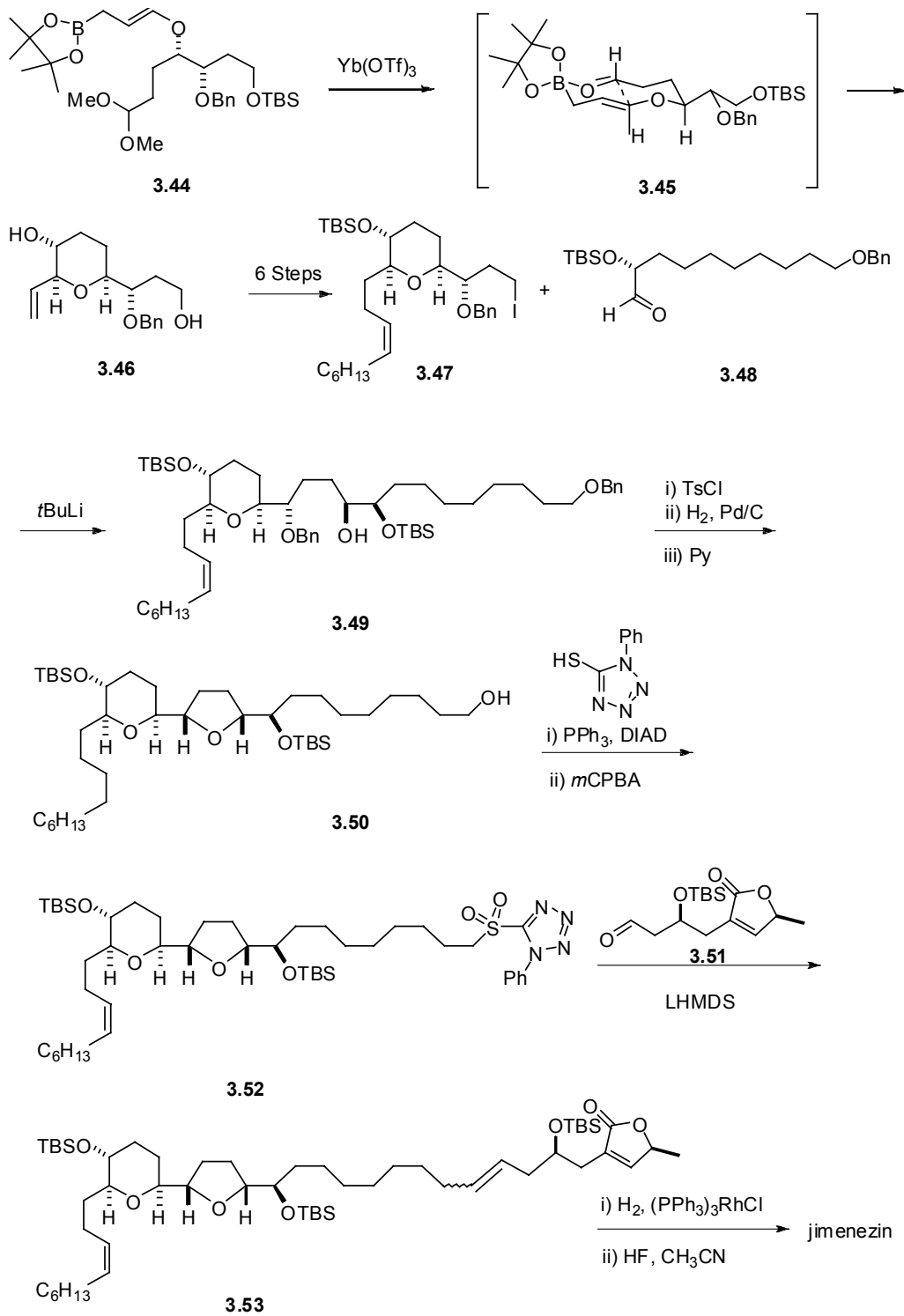


(a) Decylmagnesium bromide, Et₂O; (b) Et₃SiH, BF₃Et₂O, CH₂Cl₂, 80 %; (c) 10 % Pd/C, H₂, EtOAc-MeOH, rt; (d) TBDPSCI, imidazole, DMF, rt, 78 %; (e) HC(OMe)₃, CSA, CH₂Cl₂, rt; (f) Ac₂O, 78 %; (g) 0 % Pd/C, H₂, EtOAc, rt; (h) NaOMe, MeOH, rt; (i) MOMCl, *i*-Pr₂NEt, CH₂Cl₂, (j) TBAF, THF, rt, 93 %; (k) Swern oxidation, quant.; (l) *n*-BuLi, THF/HMPA; 72 %; (m) 10 % PtO₂, H₂, EtOAc, rt, 88 %; (n) (i) Dess-Martin periodinane, CH₂Cl₂, rt; (ii) L-Selectride, THF, 97%; (o) (i) Ac₂O, pyr, rt; (ii) aqu. AcOH, rt, 95%; (iii) SOCl₂, Et₃N, CH₂Cl₂, rt; (iv) cat. RuCl₃, NaIO₄, rt; (p) (i) NaOMe, MeOH, rt; (ii) aqu. H₂SO₄-Et₂O, rt, 83%.

The THF ring formation proceeded as follows. The 19 α -alcohol **3.41** was acetylated, and then hydrolysis under acidic conditions afforded a diol, which was then treated with thionyl chloride and triethylamine. The resulting sulfite was oxidized with NaIO₄-RuCl₃, giving a cyclic sulfate **3.42**. Upon treatment with a base, **3.42** afforded the THF-THP core structure of jiminezin **3.43** in 83% yield from **3.42**.

Hoffmann's group reported the second total synthesis of jimenezin in 2006.⁵² A highly stereoselective intramolecular allylboration was used to establish the THP ring and an intramolecular Williamson ether synthesis closed the THF ring (**Scheme 3.9**). Treatment of the *E*-allyl boronate **3.44** with Yb(OTf)₃ in acetonitrile led to the THP **3.46**. The organolithium derived from iodide **3.47** was treated with aldehyde **3.48** to afford the alcohol **3.49** as a single epimer. Tosylation of the secondary alcohol followed by one-pot alkene hydrogenation and benzylether cleavage afforded the derived dihydroxytosylate, which when treated with pyridine at reflux gave the THP-THF fragment **3.50**. The alcohol was converted to sulfone **3.51**, which was subjected to Julia-Kocienski olefination with butenolide aldehyde **3.52**, to give **3.53**. Chemoselective hydrogenation of the double bond followed by cleavage of the silyl ethers furnished jimenezin.

SCHEME 3.9



3.2 Summary of THF synthesis

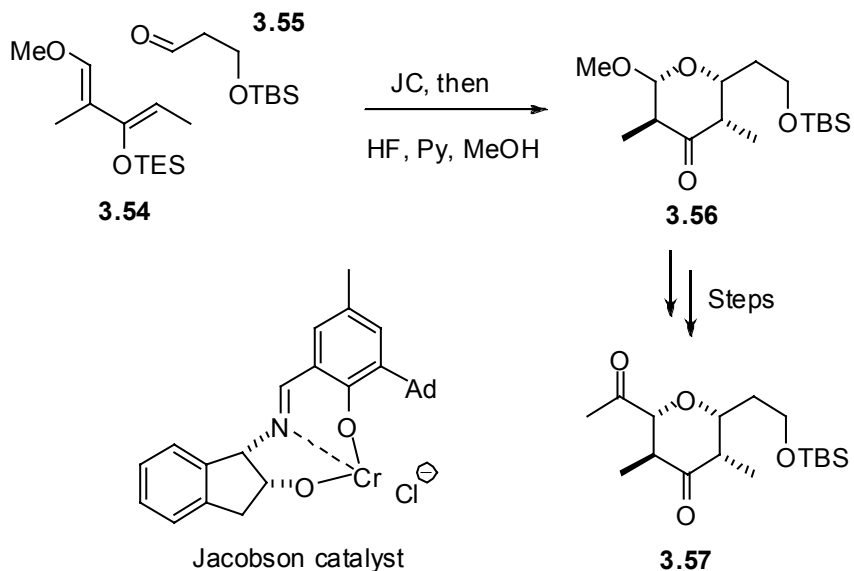
Our synthesis of 4-deoxyannoreticuin (Chapter 2) and the foregoing routes to the THF-THP acetogenins illustrate several of the more common strategies for introduction of the THF rings in the acetogenins: (i) halocyclization of hydroxyalkene derivatives,⁵³ (ii) cyclization of hydroxy epoxides,⁵⁴ (iii) Williamson ether synthesis on hydroxy tosylates⁵⁵ and (iv) Ring-Closing Metathesis on bis-allyl ethers.^{56a} The oxidative cyclization of 4-alken-1-ols has also been widely used.^{56b} This approach has also been applied to adjacently linked bis-THFs.⁵⁷

3.3 Summary of THP synthesis

As illustrated in the synthesis of muconin, the cyclization of hydroxy epoxides is one of the more popular approaches for THP synthesis (**Scheme 3.3**, **3.5** and **3.7**). The other common strategies for THP synthesis are based on the hetero-Diels-Alder⁵⁸ and Prins reactions.⁵⁹

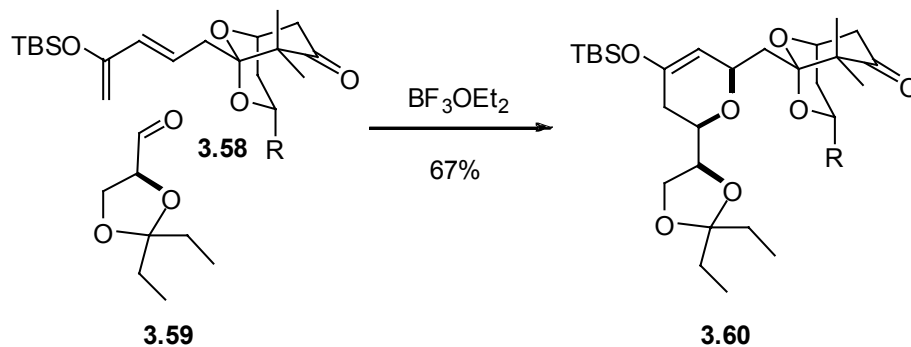
Jacobsen's synthesis of muconin is illustrative of the hetero-Diels-Alder approach (**Scheme 3.1**). A recent synthesis of phorboxazoles by Burke is another example of hetero-Diels-Alder reaction with Jacobsen catalyst.⁶⁰ The silyl enol ether, which was the product of the Diels-Alder reaction, was immediately treated with HF and pyridine to furnish the penta-substituted THP ring of the phorboxazoles in 77% yield and 91% ee (**Scheme 3.10**).

Scheme 3.10



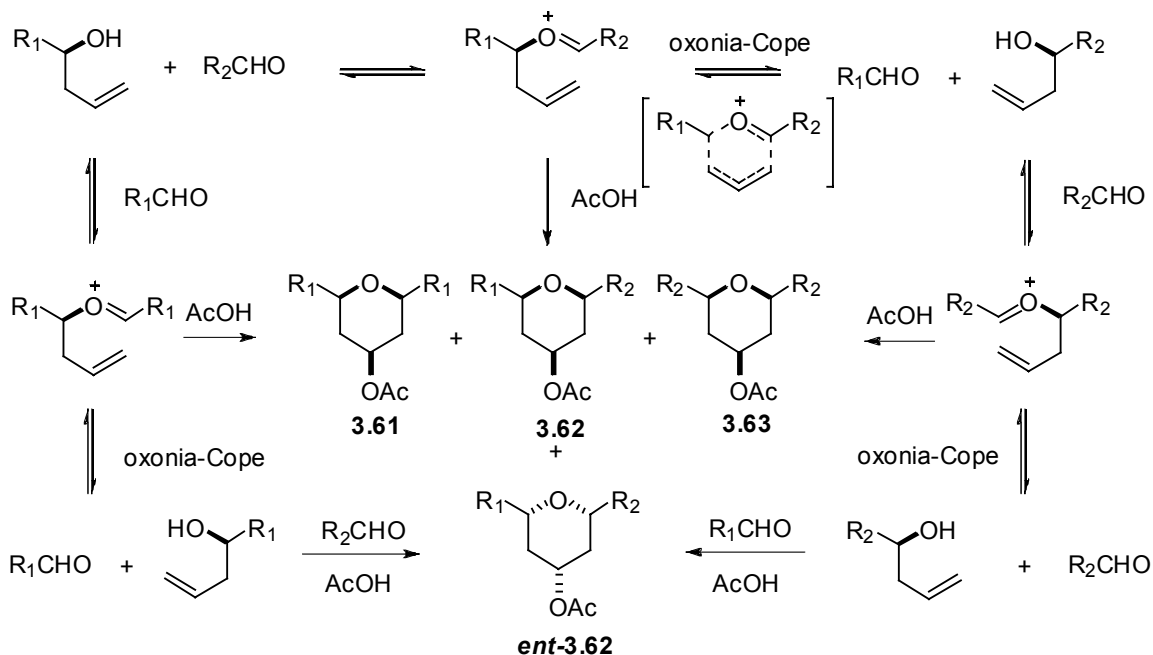
The hetero-Diels-Alder reaction has also been used in the synthesis of the B ring in the northern C1-C16 section of bryostatin I (Scheme 3.11).⁶¹ The 15:4:1 mixture of diastereoisomers obtained from this reaction was readily separated to give the major product as the needed isomer for continuation of the synthesis.

SCHEME 3.11



Prins cyclization has been widely used in the synthesis of THP containing natural products due to the convergent and efficient nature of the reaction.⁵⁹ However, a major drawback of the Prins reaction is that it requires acidic conditions, which are often not compatible with complex systems. Secondly, there is a potential for formation of diastomeric products because of a competing 2-oxonia Cope rearrangement.⁶² Alder proposed that 2-oxonia Cope rearrangement proceeds via a delocalised cationic transition state (Scheme 3.12). As illustrated by the scheme, a mixture of 4 cyclization products (3.61, 3.62, *ent*-3.62 and 3.63) was expected to form.

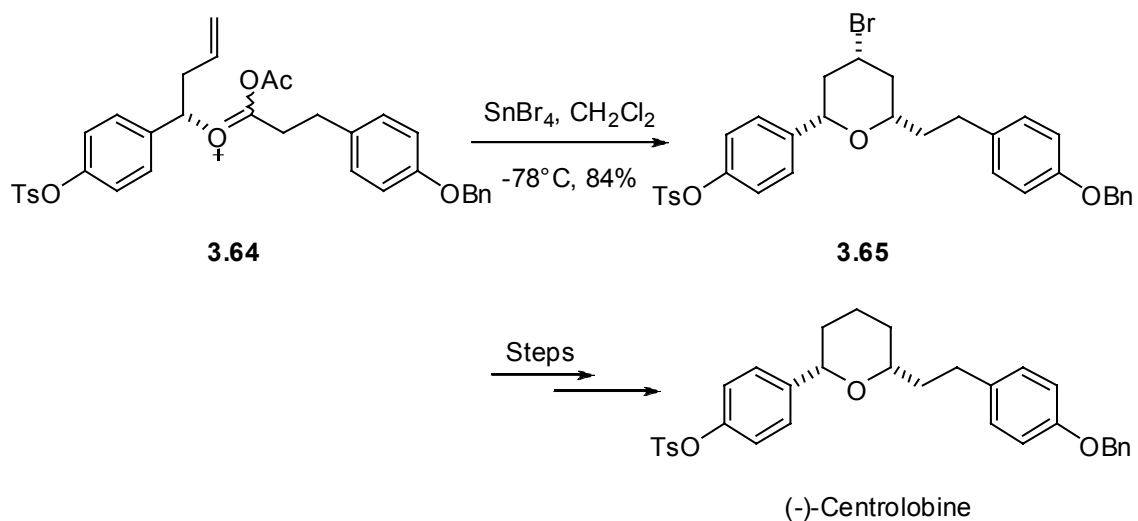
SCHEME 3.12



Rychnovsky overcame this problem by using SnBr₄ instead of traditional acid promoters such as BF₃OEt.⁶³ The possible explanation might be the cyclization promoted by SnBr₄ is much faster than that promoted by BF₃OEt and thus suppresses the competing 2-oxonia Cope rearrangement. The use of SnBr₄ to promote the α -acetoxy ether Prins cyclisation

in the synthesis of (-)-centrolobin showed significant improvement in both enantiomeric excess and overall yield (**Scheme 3.13**).

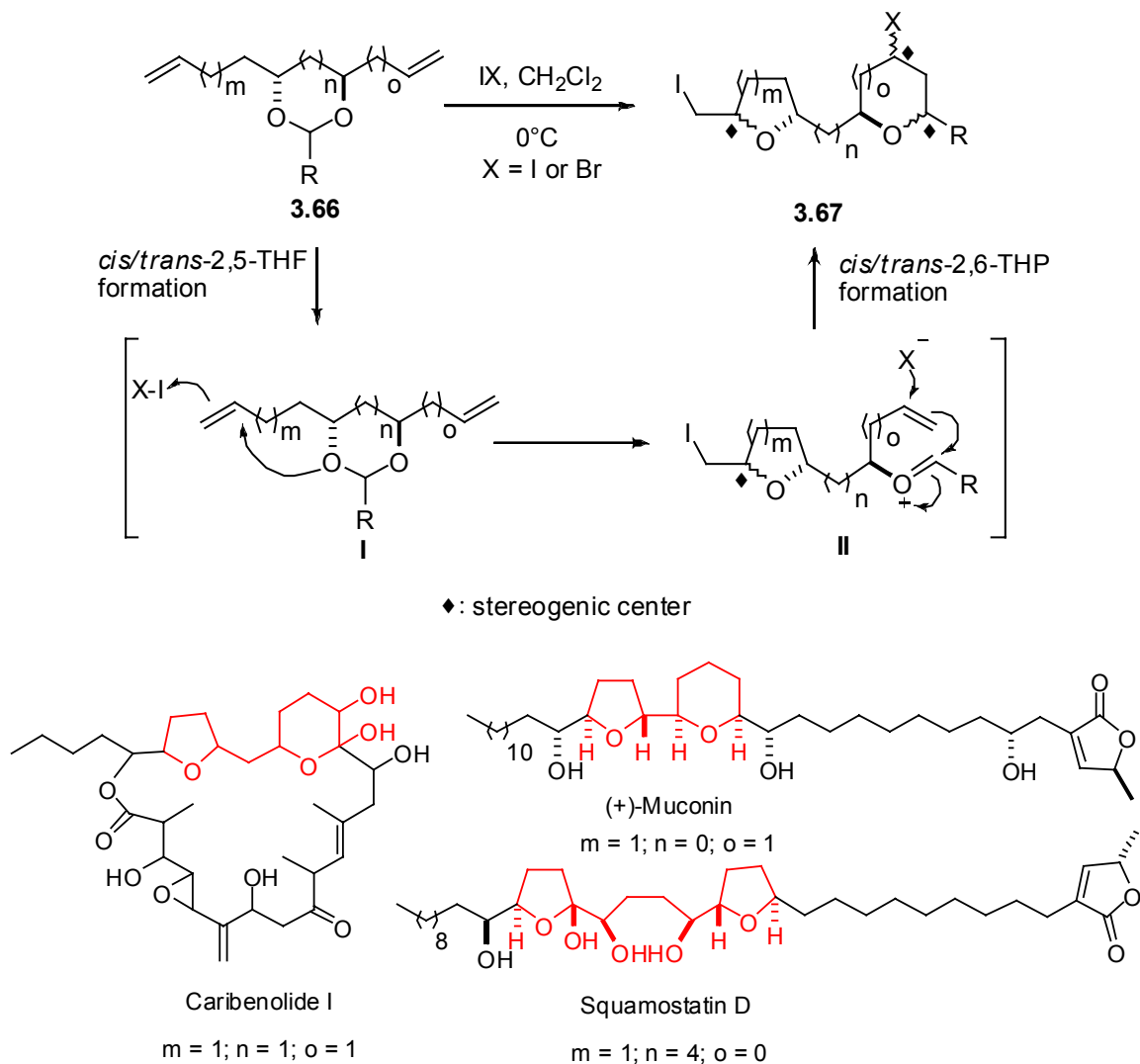
SCHEME 3.13



3.4 Tandem cyclization of diene-acetals to THF-THP segments

Our strategy for synthesis of the non-classical THF-THP subunits found in acetogenins such as muconin, centers on a tandem reaction for formation of the cyclic ethers (**Scheme 3.14**). This approach may be also applied to the THF-THP subunit found in other antitumor natural products such as caribenolide-I.⁶⁴ Thus reaction of an acetal-diene like **3.66** with an electrophile (I^+) proceeds via formation of the first cyclic ether and generation of an oxocarbenium ion (**I** \rightarrow **II**). An alkene-oxocarbenium ion cyclization on **II** gives the second ether ring leading to **3.67**. This reaction provides a convergent and rapid entry to stereochemically complex bis-cyclic ethers. Key questions on the feasibility of this plan are: (i) selective reaction of one of the two alkenes in the first step and (ii) stereoselectivity of the key C-O and C-C bond forming reactions.

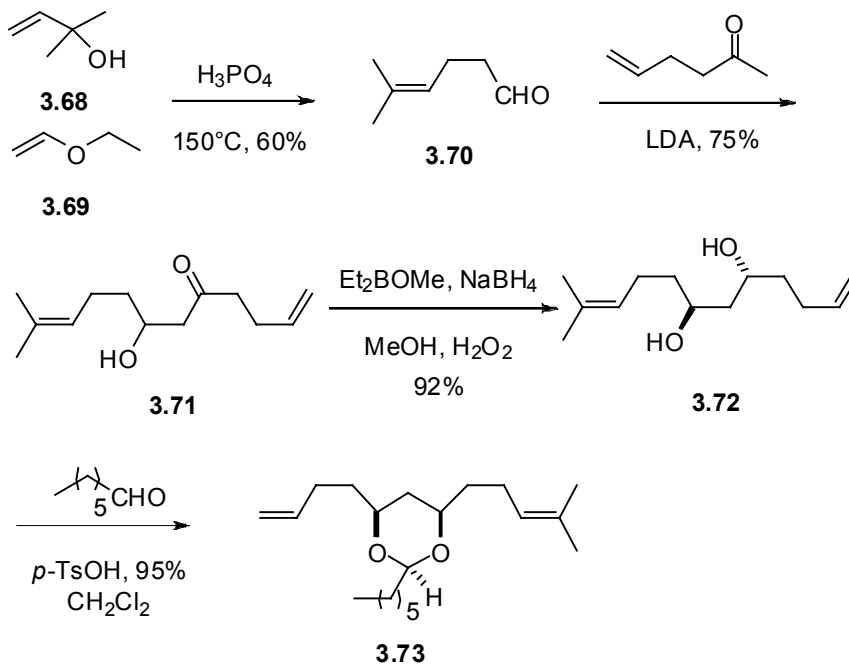
SCHEME 3.14



The tandem strategy was first tested on acetal alkene **3.73**. The precursor to **3.73** was diol **3.72**. The synthesis of **3.72** started from the alkylation of aldehyde **3.70**⁵² with the lithioenolate from hex-5-en-2-one. The aldol product **3.71** was subjected to stereoselective reduction using Et₂BOMe and NaBH₄ to give *syn*-diol **3.72**. Due to the closeness of chemical shifts of the carbinol hydrogens in **3.72**, the 1, 3-diol configuration solely relies on literature examples of this reaction.^{65, 66} Reaction of diol **3.72** with

heptaldehyde under acidic conditions gave the racemic acetal diene **3.73** as a single acetal diastereomer (**Scheme 3.15**).

SCHEME 3.15



The acetal configuration in **3.73** was presumed to correspond to the more stable diastereomer, i.e. the one in which the ring substituents were all equatorial. (**Figure 3.3**).

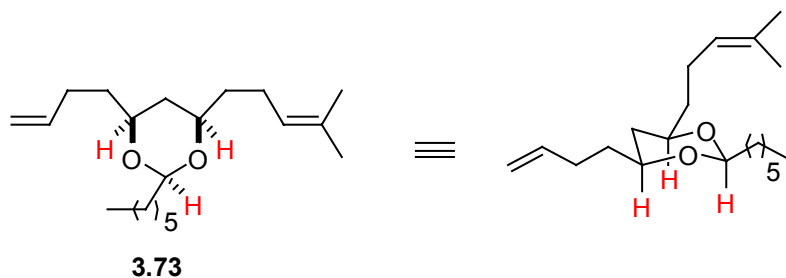
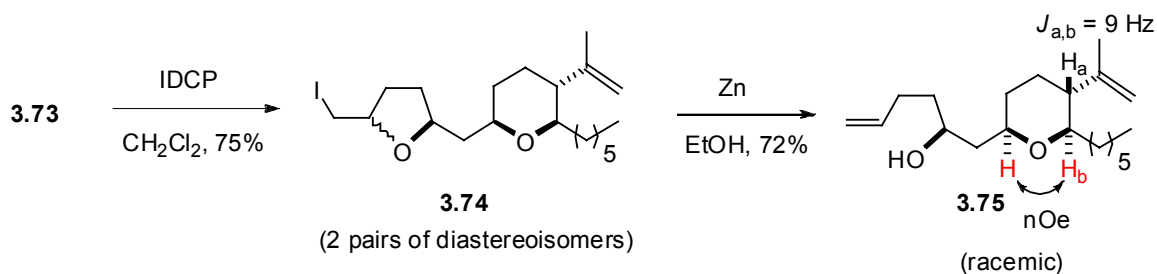


Figure 3.3 Presumed configuration of **3.73**

With acetal diene **3.73** in hand, the key cyclisation was next attempted. Treatment of **3.73** with IDCP gave the THF-THP ether **3.74** as an approximate 1:1 mixture of two diastereoisomers in 75 % yield. To simplify the stereochemical analysis, **3.74** was converted to **3.75** by exposure to zinc. A single diastereoisomer **3.75** was obtained, indicating that the mixture **3.74** was due to *cis* and *trans* THF rings. A 2, 6-*cis*-THP ring was deduced from a nOe. The isopropenyl substituent was assigned as *trans* to the other THP substituents from a vicinal $J_{a,b}$ value of 9 Hz (Scheme 3.16).

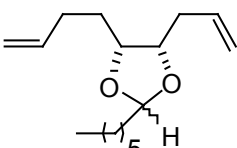
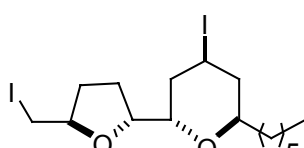
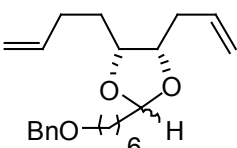
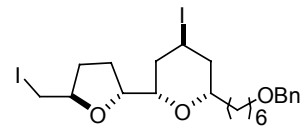
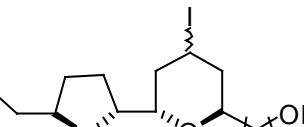
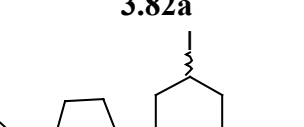
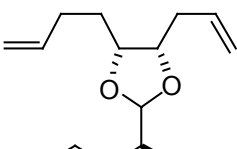
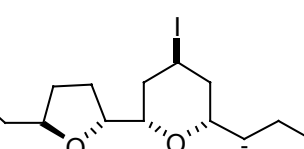
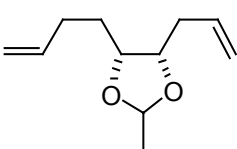
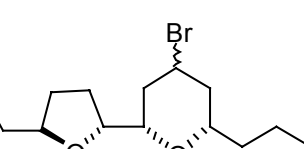
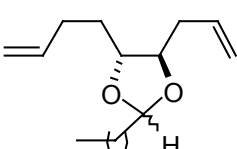
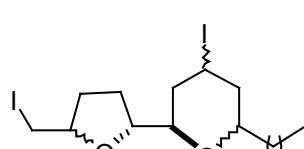
SCHEME 3.16

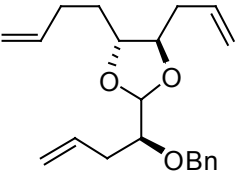
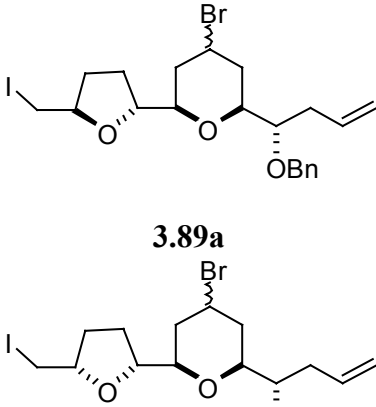


Inspired by this result, we next evaluated the tandem cyclisation of a variety of acetal diene substrates. The substrates, reaction conditions and cyclisation results are summarized in Table 3.1.

Table 3.1

Entry	Substrates	Conditions	Cyclised products	Yields
1	<p>3.76 (\pm) (3:1 ratio of 2 isomers)</p>	<p>IDCP CH_2Cl_2 rt</p>	<p>3.77 (Proposed Stereochemistry)</p>	50%

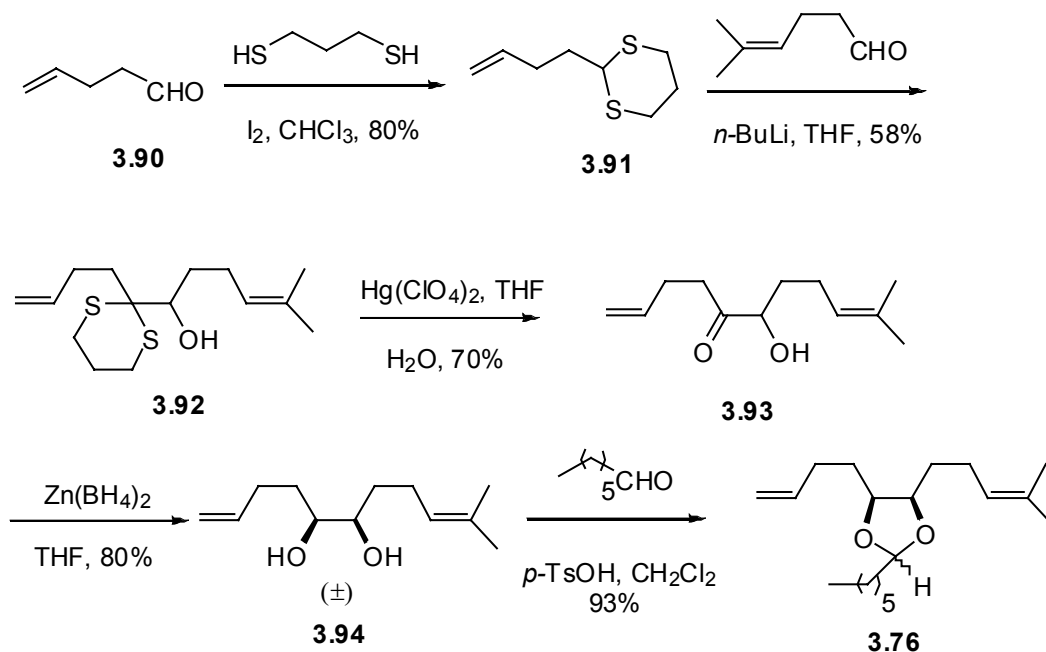
2	 3.78 (Single isomer)	I_2 CH_2Cl_2 $-40^\circ C$	 3.79	14%
3	 3.80 (3:1 ratio of 2 isomers)	I_2 CH_2Cl_2 $0^\circ C$	 3.81  3.82a  3.82b (3:7 ratio of 3.82a : 3.82b)	8.5% 43%
4	 3.83 (Single isomer)	I_2 CH_2Cl_2 $0^\circ C$	 3.84	15%
5	 3.83 (Single isomer)	IBr CH_2Cl_2 $0^\circ C$	 3.85	51%
6	 3.86 (1:1 ratio of 2 isomers)	I_2 CH_2Cl_2 $0^\circ C$	 3.87 (1:5:5 of 3 isomers)	60%

7	 <p>3.88 (1:1 ratio of 2 isomers)</p>	<p>I₂ CH₂Cl₂ 0°C</p>	 <p>3.89a 3.89b (2 isomers)</p>	45%
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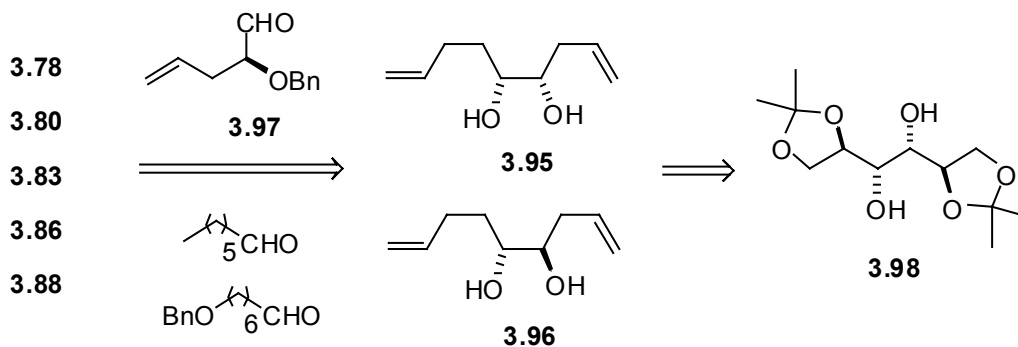
Synthesis of the 5, 6-*anti*-diol acetal **3.76** started from pent-4-enal (**3.90**) (Scheme 3.17). First, the aldehyde was protected as dithioacetal **3.91** by treating it with propane-1,3-dithiol in the presence of iodine. Deprotonation of **3.91** with *n*-butyllithium and alkylation with 5-methylhex-4-enal gave the product **3.92**. The α -hydroxy ketone **3.93** was obtained upon removal of the dithioacetal protecting group in **3.92**. Selective reduction of **3.93** with zinc borohydride gave the racemic **3.94** as the single diastereoisomer, which was presumed to be the anti diastereoisomer based on literature precedence.⁶⁶ Finally, A 3:1 ratio of acetal diastereoisomers **3.76** of undetermined configuration was obtained by treating diol **3.94** with heptaldehyde under acidic conditions.

The other acetal precursors listed in Table 3.1 were all synthesized from commercially available 1,2:5,6-di-*O*-isopropylidene-D-mannitol **3.98** (Scheme 3.18).

SCHEME 3.17



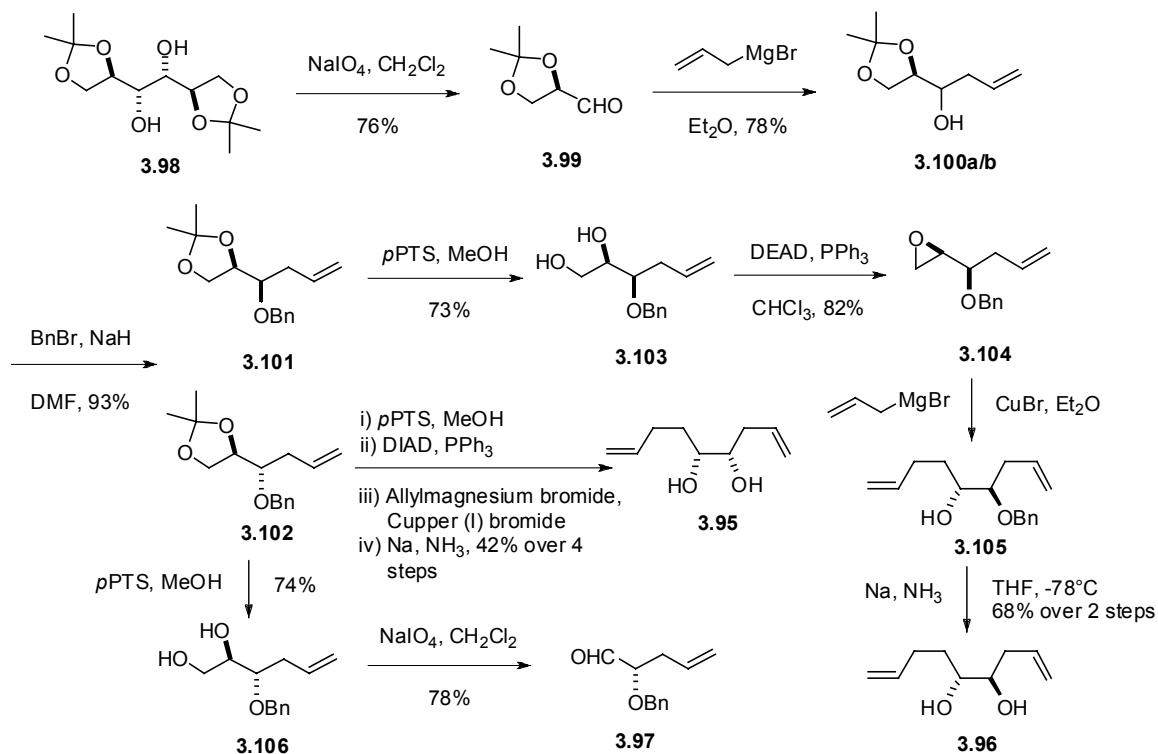
SCHEME 3.18



Treating **3.98** with NaIO_4 afforded aldehyde **3.99**, which was then treated with allylmagnesium bromide to give an inseparable 1:1 mixture of diastereoisomers **3.100a/b** (Scheme 3.19). Fortunately, the derived benzyl ethers **3.101** and **3.102** were chromatographically separable. Debenzylation of **3.101** and **3.102** gave the corresponding alcohols whose NMR information matches that of known alcohols

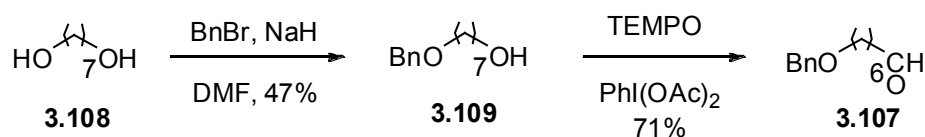
perfectly. The isopropylidene in **3.101** was easily removed under acidic conditions to give diol **3.103**. Diol **3.103** was then transformed to epoxide **3.104** under Mitsunobu conditions. Treatment of **3.104** with allylmagnesium bromide provided (4*R*, 5*R*)-4-benzyloxynona-1,8-dien-5-ol (**3.105**). Upon exposure of **3.105** to sodium in the liquid ammonia, the benzyl group was cleanly removed to give (4*R*, 5*R*)-nona-1,8-diene-4,5-diol (**3.96**). (4*S*, 5*R*)-nona-1,8-diene-4,5-diol (**3.95**) was obtained in 42 % yield through a similar reaction sequence from benzyl ether **3.102**. Aldehyde **3.97** was synthesized from **3.102** via the removal of the isopropylidene and oxidative cleavage of the resulting vicinal diol in **3.106**.

SCHEME 3.19



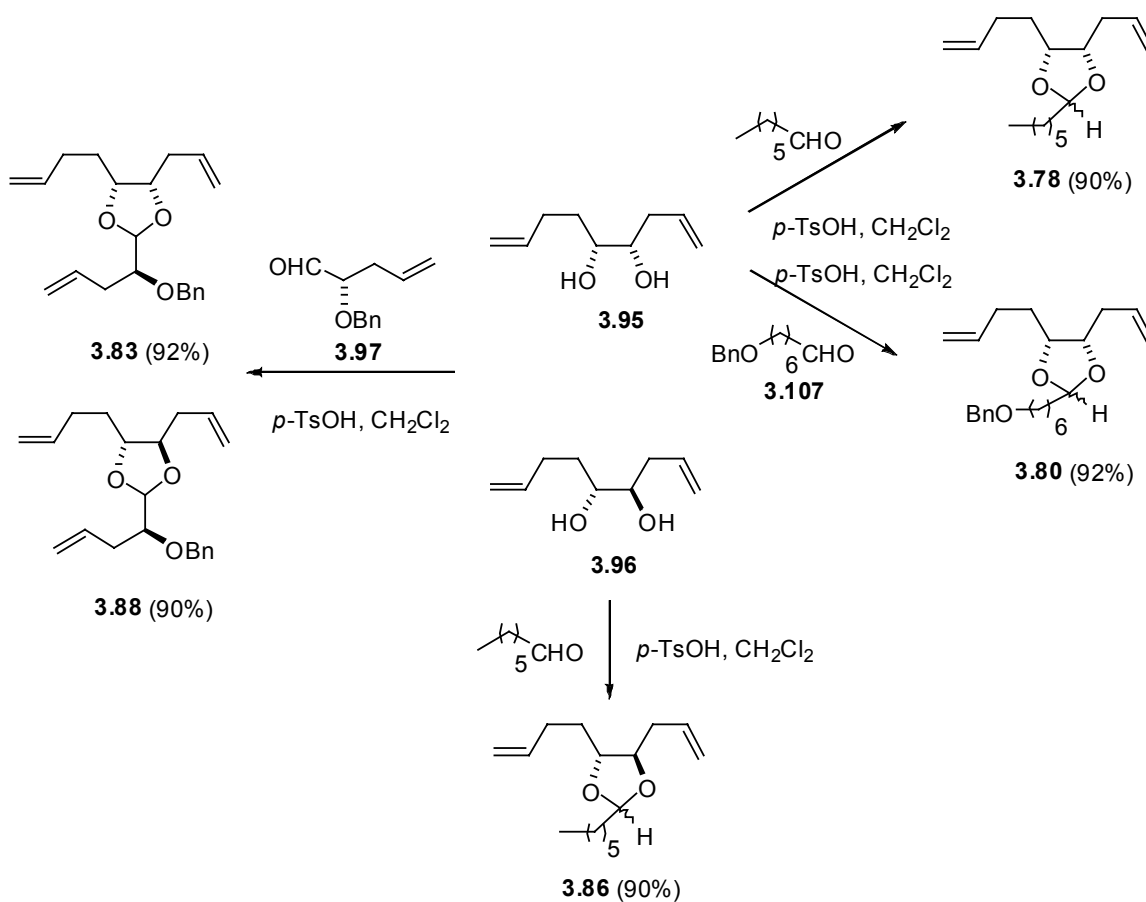
Aldehyde **3.107** was prepared from heptan-1, 7-diol (**3.108**) through oxidation of the mono-benzyl ether **3.109** (Scheme 3.20).

SCHEME 3.20



Acetal diene **3.78**, **3.80**, **3.83**, **3.86** and **3.88** were obtained by the acetalization of diols **3.95** and **3.96** with the corresponding aldehydes **3.97**, heptanal or 7-benzyloxyheptanal (Scheme 3.21).

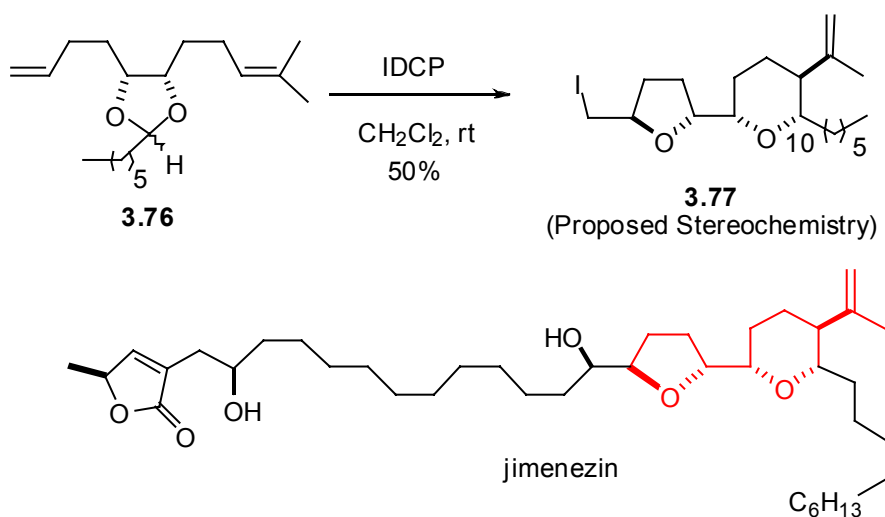
SCHEME 3.21



The diastereoselectivity of the acetalization reaction varied with the relative stereochemistry of the diol precursor. The *anti* diols gave predominantly a single acetal isomer whereas the *syn* diols gave an approximately 1:1 mixture of acetals. The configuration at the acetal carbons was not determined.

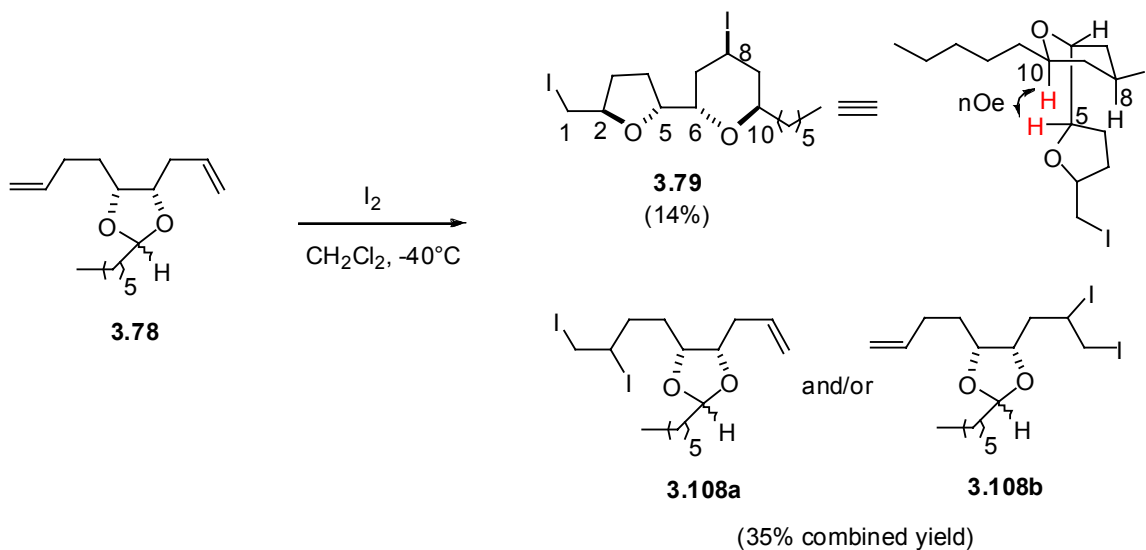
The tandem cyclization of the *anti* diol substrates were first examined. Treatment of acetal mixture **3.76** with IDCP in dichloromethane at room temperature gave a single THF-THP diastereoisomer **3.77** in 55 % yield (**Scheme 3.22**). The gross structure of **3.77** was assigned by NMR but signal overlap did not permit stereochemical assignment. The stereochemistry of the THP ring was tentatively assigned by comparison with the reaction of the related 1, 3-diol substrate **3.73**. The 2, 5-*trans* THF stereochemistry was assigned following our observation that the iodoetherification reactions of 1, 2-*O*-isopropylidene-5-alkenes give the 2,5-*trans* THF motif. The conversion of **3.76** to **3.77** is relevant to the synthesis of the THF-THP core structure of the jimenezin subgroup of the acetogenins.

SCHEME 3.22



The tandem cyclization on the enantiopure acetal diene **3.78** was next examined. In this case the second ether ring formation involves a Prins-like cyclisation. Treating **3.78** with iodine at -40°C gave THF-THP product **3.79** (14%) and uncyclized diiodides **3.108a** and/or **3.108b** as the major products (**Scheme 3.23**). Unexpectedly, the *trans*-THP ring was obtained instead of *cis*-THP ring, which is the normal product in Prins cyclization reactions. A nOe effect was observed between H5 and H10, which indicated that the THP ring was *trans*. The stereochemistry of THF ring was presumed as *trans* based on the cyclization result on **3.80** under the similar condition (*vide infra*). The equatorial iodide on C8 was deduced from a *J* value of 8 Hz, which was attributed to a *trans* diaxial relationship between H8 and one of the protons on the adjacent carbons.

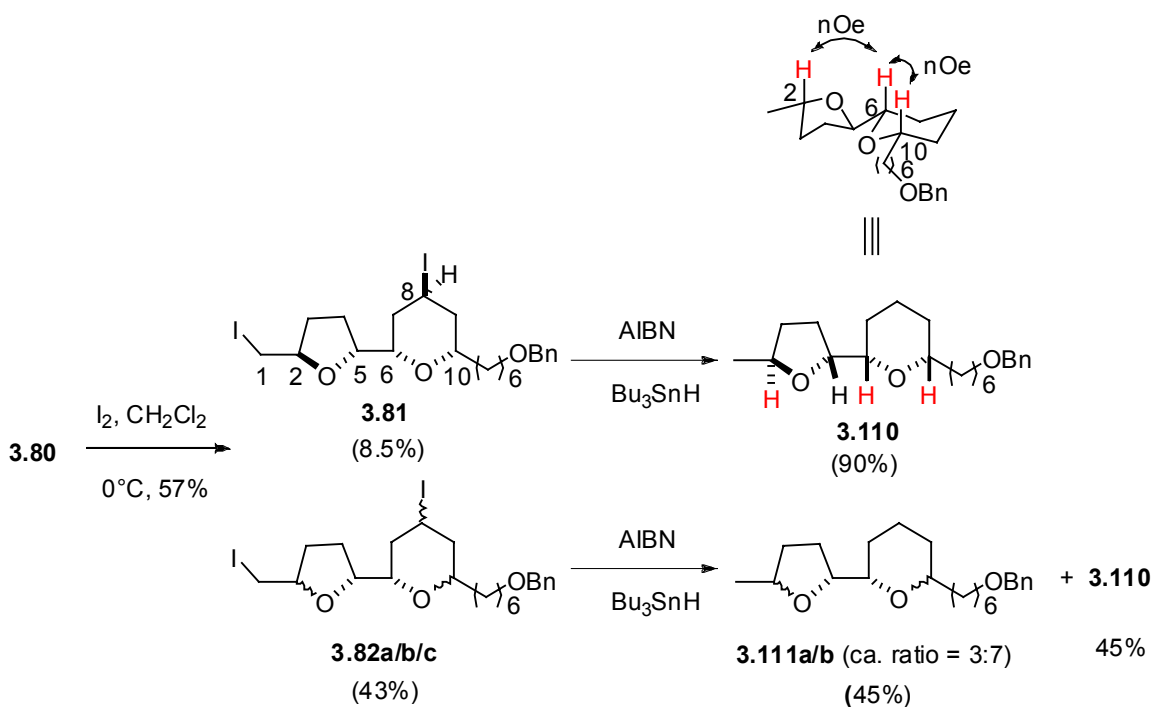
SCHEME 3.23



Application of the identical cyclization conditions that were used on **3.78** to **3.80** gave **3.81** (8.5%) and **3.82** as a mixture of diastereoisomers (43%) (**Scheme 3.24**). To simplify the analysis of stereochemistry of the products, **3.81** and **3.82** were treated with tributyltin

hydride to give **3.110** and **3.111** respectively. Configuration of **3.110** was confirmed by the nOe effects observed between H2 and H6, H6 and H10. The equatorial iodide at C8 in **3.81** was assigned on the basis of a *J*_{vic} value of 8 Hz, as in the case of **3.79**. Removal of iodines in **3.82** yielded three products, **3.110** and the inseparable mixture of two isomers **3.111a/b**.

SCHEME 3.24

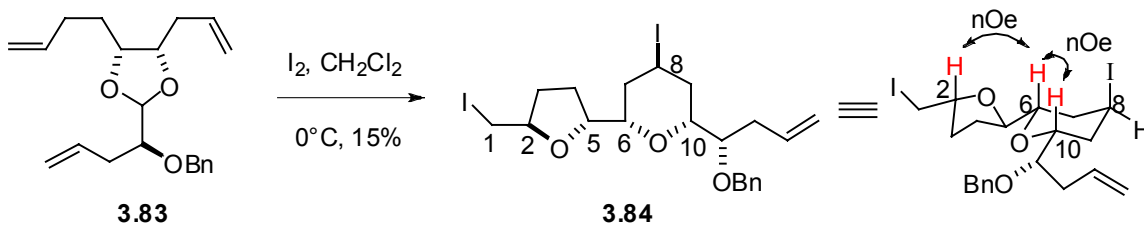


The results on **3.78** and **3.80** indicate that the tandem cyclization of *anti*-1, 2-diol acetal substrates with simple aldehyde components favor the formation of the 2, 6-*trans*-THP pattern with moderate stereoselectivity. The reaction of *anti*-1, 2-diol acetals with an α -alkoxy aldehyde component were next examined.

Acetal diene **3.83** was treated with iodine at $0^\circ C$ in CH_2Cl_2 (Scheme 3.25). To our pleasant surprise a single cyclization product **3.84** containing a *trans* THF and a *cis* THP

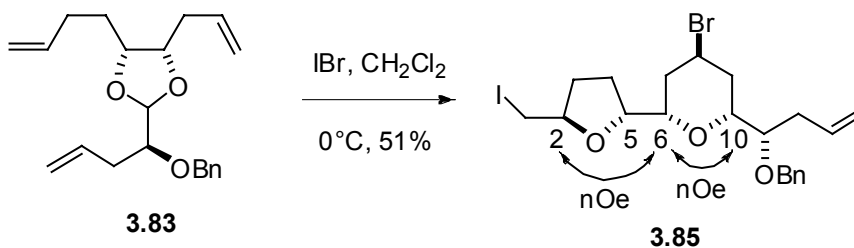
was obtained. The configurations at the newly formed stereogenic carbons in **3.84** were deduced from nOe and vicinal *J* values. However, this reaction was sluggish and low yielding.

SCHEME 3.25



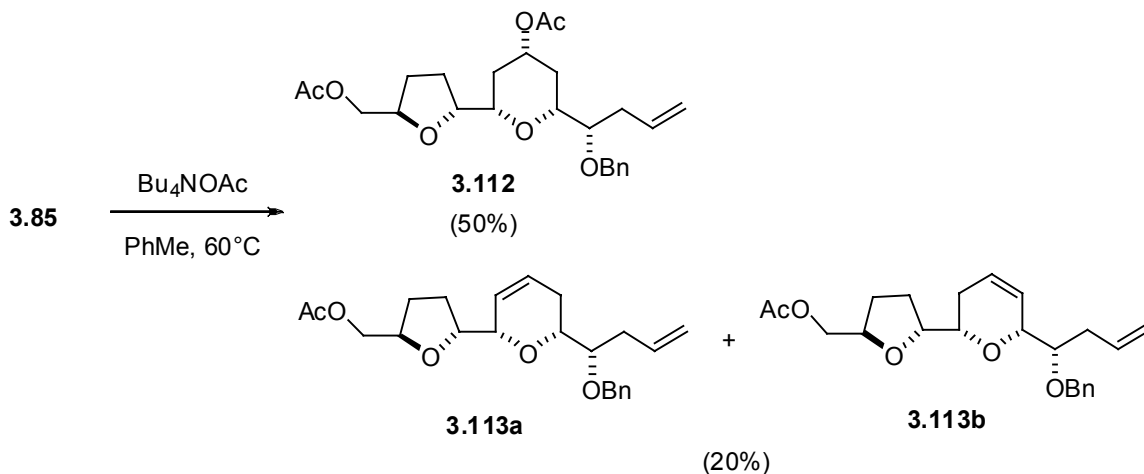
To improve the efficiency, we replaced iodine with iodine monobromide which is a more reactive specie of I^+ (Scheme 3.26). The reaction time was significantly shortened from 40 h to 10 min. The major product **3.85** was obtained in 51% yield as a single diastereoisomer. As for **3.84**, the stereochemistry of **3.85** was deduced from nOe effects observed between H2 and H6, H6 and H10.

SCHEME 3.26



In view of further elaboration of the product, **3.85** was treated with tetrabutylammonium acetate. A major single product **3.112** was obtained along with elimination products **3.113a** and **3.113b** in 20% yield (Scheme 3.27).

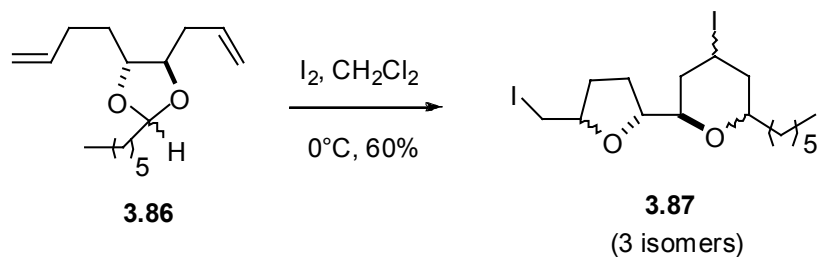
SCHEME 3.27



The afore-mentioned substrates are derived from 1, 2-*anti*-diols. To evaluate the effect of the stereochemistry of the 1, 2-diol, *syn*-diol substrates were next examined.

Treatment of 4, 5-*syn*-diol acetal **3.86** with iodine in dichloromethane at 0°C gave THF-THP **3.87** as a mixture of 3 diastereoisomers (Scheme 3.28). Although we were able to separate them, signal overlap in the NMR did not permit stereochemical assignment.

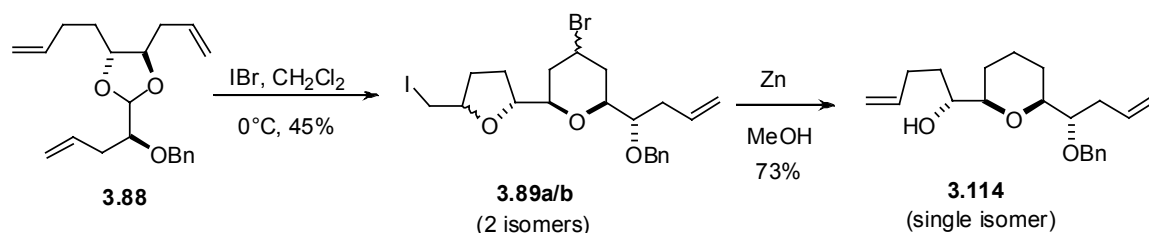
SCHEME 3.28



Treatment of 4, 5-*syn*-diol acetal **3.88** with iodine monobromide in dichloromethane at 0°C afforded a mixture of 2 diastereoisomers of very similar chromatographic mobility (Scheme 3.29). Exposure of **3.89a/b** to zinc provided a single product **3.114**. The THP

ring in **3.114** was assigned the *cis* stereochemistry from a nOe between the carbinol protons of the THP. As will be discussed later, selective removal of the bromide in a subsequent derivative of **3.89** confirmed that the mixture **3.89a/b** was due to *cis* and *trans* THFs.

SCHEME 3.29



3.5 Synthetic studies on muconin

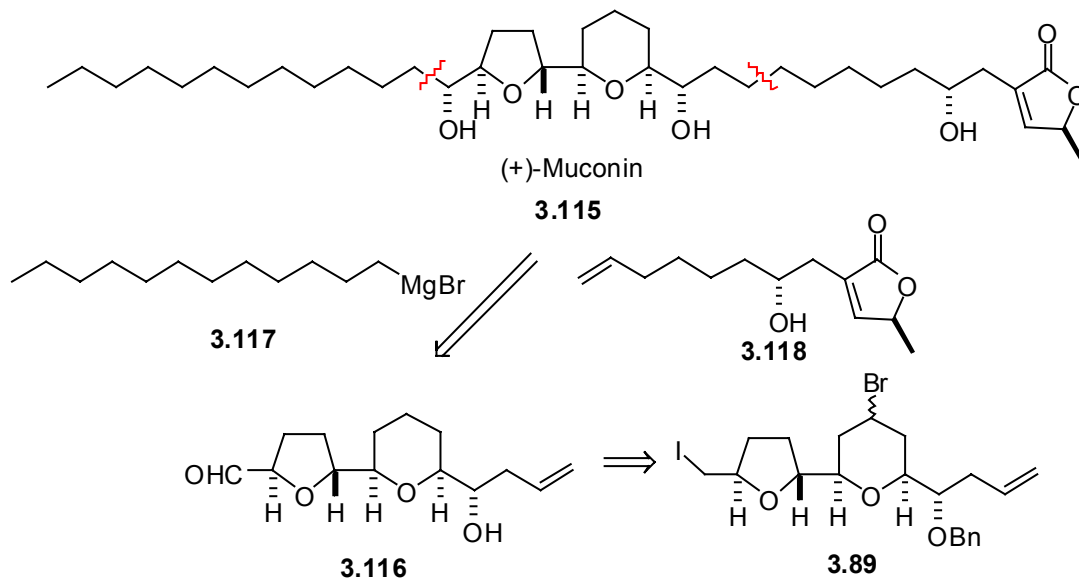
The results from the cyclization of acetal diene **3.88** was next applied to the synthesis of muconin (Scheme 3.30). Appropriate functionalization of **3.89** would transform **3.89** to aldehyde **3.116**, which can be converted to muconin (**3.115**) using established reactions (vide infra).

Treatment of the *cis/trans* THF mixture **3.89** with thiophenol led to selective substitution of the primary iodide to give **3.119** (Scheme 3.31). Reduction of **3.119** with zinc afforded mixture **3.120**. Compound **3.120** was then oxidized to sulfoxide **3.121** by *m*CPBA. Pummerer rearrangement went smoothly to give thioacetal **3.122**.

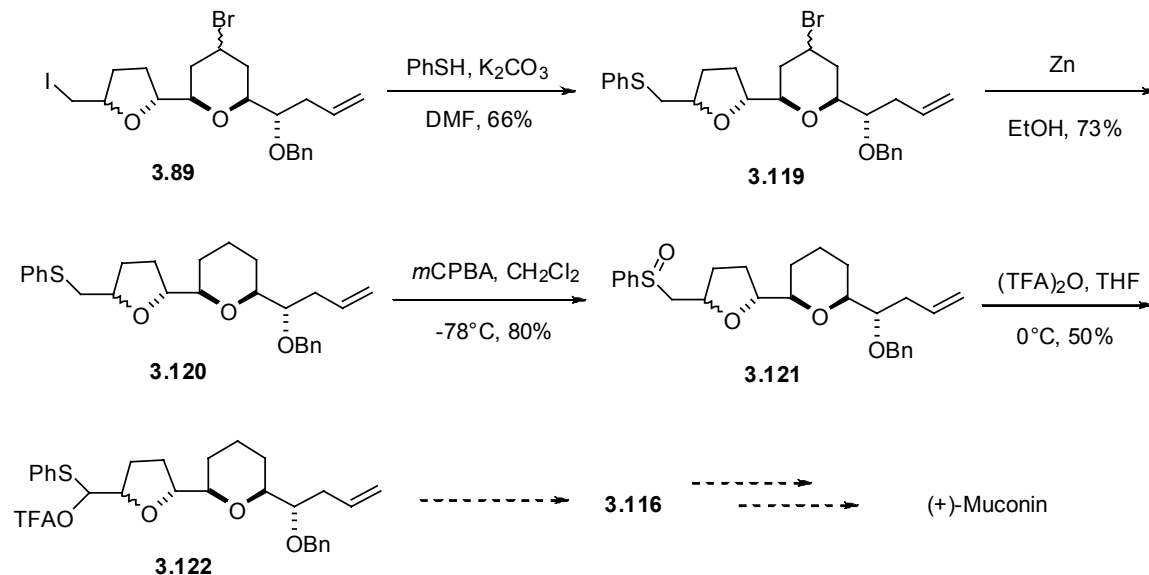
Completion of the synthesis of muconin requires the following key transformations: (i) conversion of thioacetal **3.122** to aldehyde **3.116**, (ii) alkylation of **3.116** with Grignard reagent **3.120**, (iii) olefin cross metathesis with butenolide **3.121** on the product from the

Grignard reaction and (iv) selective hydrogenation of the metathesis product (**Scheme 3.31**).⁶⁸ It may be possible to achieve the first two steps in a single reaction by treatment of **3.112** with excess Grignard reagent.⁶⁹

SCHEME 3.30



SCHEME 3.31

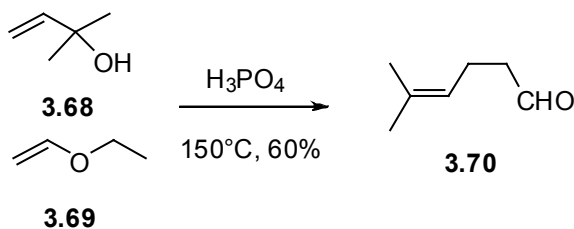


3.6 Conclusion

In conclusion a tandem, iodonium ion triggered, THF-THP forming reaction on acetals of dihydroxy-dienes was investigated. Vicinal or 1,3- diol precursors produced adjacently or methylene linked cyclic ethers in good to moderate yield. In all cases the cascade appears to be initiated by a 5-exo trig type cyclization to give a 2,5-disubstituted THF connected to an oxocarbenium ion which undergoes attack by the second alkene to form the THP ring. The substitution pattern and connectivity of the second alkene determines the mechanism of formation and constitution of the THP, i.e. a 2,5,6-tri-alkylated or 2,6-dialkyl-4-halogenated. IDCP was the reagent of choice for the first and IBr was optimal for the second. Altogether, three new stereogenic centers are generated in the THF-THP products. The anti-diol substrates showed high selectivity for the trans-2,5-disubstituted THF ring, whereas the selectivity for syn diol precursors was poor. Generally, the THP segments were formed in moderate to high selectivity depending on the operative mechanism in the oxocarbenium ion cyclization. Finally, the feasibility of using this tandem reaction for non-classical THF-THP containing acetogenins was illustrated by preparation of an advanced precursor to muconin. The modular nature of this synthesis and the rapid entry to such stereochemically rich frameworks from relatively simple precursors are noteworthy attributes in the context of future applications.

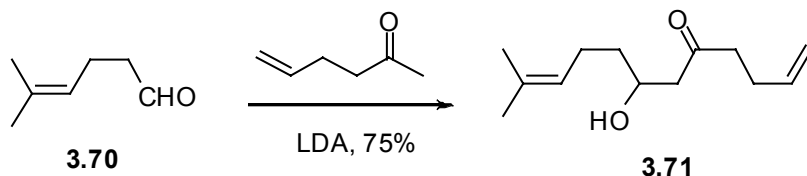
3.7 Experimental section

Synthesis of 5-methylhex-4-enal (**3.70**)



In a 500 mL high-pressure tolerant beer bottle with secured hitch cork were added in 2-methylbut-3-en-2-ol (**3.68**) (10 mL, 95.7 mmol) and ethylvinyl ether (**3.69**) (18.3 mL, 191 mmol) and 3 drops of H_3PO_4 . The reaction mixture was heated up to 150°C for 2 h. Then it was cooled down to 0°C and distilled at 110°C under the hood vacuum to give the desired product **3.70** as a colorless liquid (6.4 g, 60%). ^1H NMR (CDCl_3) δ 9.77 (s, 1H), 5.09 (s, 1H), 2.46 (s, 2H), 2.32 (s, 2H), 1.70 (s, 3H), 1.63 (s, 3H); ^{13}C NMR (CDCl_3) δ 202.7, 122.1, 70.6, 43.9, 25.6, 20.9, 17.7

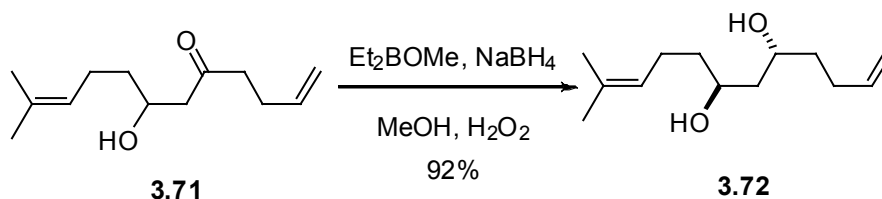
Synthesis of 7-hydroxy-11-methyldodeca-1, 10-dien-5-one (**3.71**)



To LDA prepared from *n*-BuLi (7.7 mL, 19.3 mmol, 2.5 M in hexane) and *i*-Pr₂NH (3.3 mL, 23.5 mmol) of in THF (53.2 mL) at -78°C was added a solution of hex-5-en-2-one (0.42 g, 4.28 mmol) in THF (32 mL). The reaction mixture was stirred at -78°C for 1 h, and to it was added a solution of aldehyde **3.70** (0.96 g, 8.56 mmol) of in THF (32 mL).

The reaction mixture was stirred at $-78\text{ }^{\circ}\text{C}$ for 1 h, quenched with saturated aqueous NH_4Cl , and diluted with ether. The aqueous layer was extracted with ether, and the combined extracts were dried (MgSO_4). The solvent was removed under reduced pressure. The residue was subjected to column chromatography to give **3.71** (0.67 g, 75%) as a colorless oil. ^1H NMR (CDCl_3) δ 5.78- 5.85 (m, 1H), 5.11- 5.15 (m, 1H), 5.00- 5.08 (m, 2H), 4.05- 4.09 (m, 1H), 2.93 (d, 1H, $J = 3.6$ Hz), 2.52- 2.63 (m, 4H), 2.34- 2.38 (m, 2H), 2.09- 2.15 (m, 2H), 1.71 (s, 3H), 1.64 (s, 3H), 1.45- 1.59 (m, 1H), 1.42- 1.44 (m, 1H); ^{13}C NMR (CDCl_3) δ 211.1, 136.8, 132.2, 123.8, 115.4, 67.3, 49.2, 36.5, 36.4, 27.5, 25.7, 24.1, 17.7.

Synthesis of 11-methyldodeca-1,10-diene-5,7-diol (**3.72**)



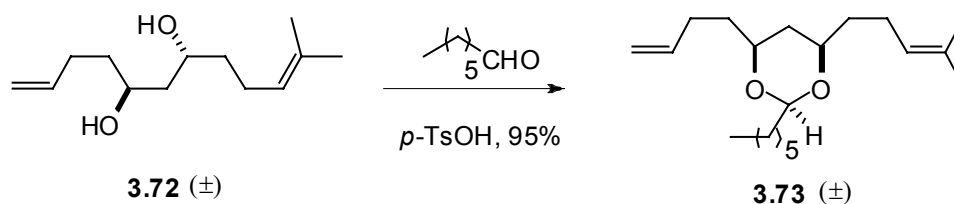
To a solution of ketone **3.71** (0.7 g, 3.33 mmol) in THF (37 mL) and MeOH (10 mL) was added in Et_2BOMe (3.8 mL, 3.83 mmol, 1.0 M in THF) at $-78\text{ }^{\circ}\text{C}$. The reaction mixture was stirring at $-78\text{ }^{\circ}\text{C}$ for 0.5 h. Then NaBH_4 (151 mg, 4 mmol) was introduced and mixture stirred at $-78\text{ }^{\circ}\text{C}$ for an additional hour. MeOH (14 mL) was added followed by 1N aqueous NaOH (8.6 mL), and the mixture was maintained at $25\text{ }^{\circ}\text{C}$ for 1 h. The mixture was cooled to $0\text{ }^{\circ}\text{C}$ and 30% of H_2O_2 (4.3 mL) was added dropwise and the reaction mixture was allowed to warm to rt overnight. Brine (14 mL) was added and the mixture was concentrated to approximately 30 mL prior to extraction with Et_2O . The solvent was removed under reduced pressure. The residue was subjected to column

chromatography. The product **3.72** (0.44 g, 92%) was obtained as colorless oil. $^1\text{H NMR}$ (CDCl_3) δ 5.81- 5.87 (m, 1H), 5.12- 5.15 (m, 1H), 4.97- 5.08 (m, 2H), 3.84- 3.90 (m, 2H), 3.39- 3.40 (m, 1H), 3.19- 3.20 (m, 1H), 2.06- 2.20 (m, 4H), 1.70 (s, 3H), 1.63 (s, 3H), 1.46- 1.60 (m, 6H); $^{13}\text{C NMR}$ (CDCl_3) δ 138.4, 132.1, 123.9, 114.8, 72.8, 72.4, 42.9, 38.1, 37.2, 29.7, 25.7, 24.0, 17.7. HRMS (ESI) calcd for $\text{C}_{13}\text{H}_{24}\text{O}_2$ ($\text{M} + \text{Na}^+$) 235.1669, found 235.1671.

General procedure for synthesis of cyclic acetals from diol and aldehyde precursors

To a solution of the diol (1 mmol) in CH_2Cl_2 was added the aldehyde (2 mmol) and *p*-TsOH $\cdot\text{H}_2\text{O}$ (0.1 mmol) at 0 °C. The reaction mixture was stirring at rt for an additional hour, then quenched by adding saturated aqueous NaHCO_3 , and extracted with CH_2Cl_2 . The organic extract was dried (Na_2SO_4) and the solvent was removed under reduced pressure. The residue was subjected to column chromatography to give the desired acetal.

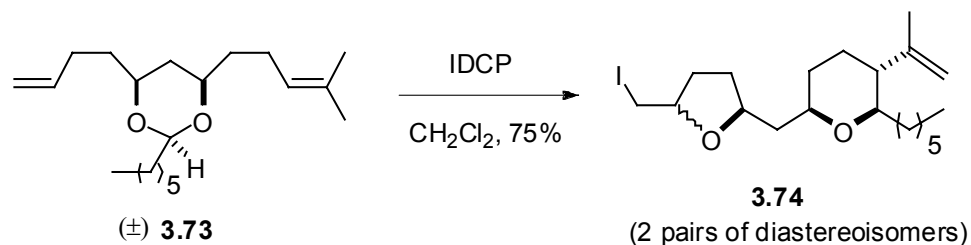
Synthesis of (2*S*, 4*S*, 6*R*)-4-(but-3-enyl)-2-ethyl-6-(4-methylpent-3-enyl)-1,3-dioxane (3.73)



$^1\text{H NMR}$ (CDCl_3) δ 5.80- 5.86 (m, 1H), 5.11- 5.14 (m, 1H), 5.05- 5.06 (d, $J = 5$ Hz, 1H), 5.05- 5.05 (d, $J = 1.6$ Hz, 1H), 4.47- 4.48 (m, 1H), 3.53- 3.58 (m, 2H), 2.09- 2.21 (m, 4H), 1.70 (s, 1H), 1.64- 1.69 (m, 2H), 1.60 (s, 1H), 1.22- 1.54 (m, 14H), 0.88- 0.92 (m, 3H); $^{13}\text{C NMR}$ (CDCl_3) δ 138.3, 131.8, 124.0, 114.6, 101.8, 75.5, 37.2, 36.0, 35.1, 35.1, 31.8,

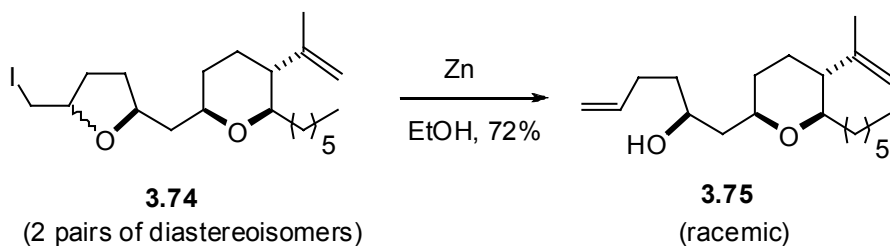
29.2, 29.2, 25.7, 24.2, 23.5, 22.6, 17.6, 14.0. HRMS (ESI) calcd for C₂₀H₃₆O₂ (M + H⁺) 309.2788, found 309.2790.

Synthesis of cyclised product 3.74



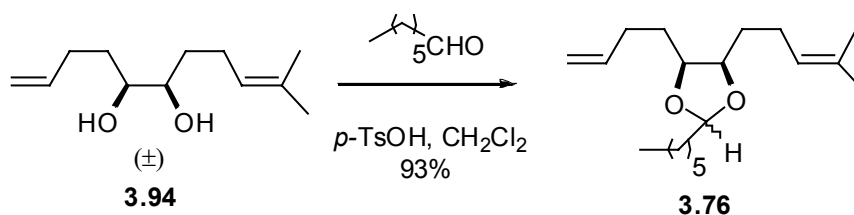
To a solution of diol acetal **3.73** (70 mg, 0.227 mmol) in CH₂Cl₂ (10 mL) were added grinded molecular sieves (pre-activated, 1 g) and IDCP (160 mg, 0.341 mmol) at 0 °C. The reaction mixture was stirring at rt for 2 h. Then it was quenched by adding saturated Na₂S₂O₃, extracted by Et₂O, and dried (Na₂SO₄). The solvent was removed under reduced pressure. The residue was subjected to column chromatography to give **3.74** (74 mg, 75%) as a colorless oil. ¹H NMR (CDCl₃) δ 4.74 (s, 2H), 4.76 (s, 2H), 4.25- 4.28 (m, 1H), 4.11- 4.15 (m, 2H), 3.98- 4.00 (m, 1H), 3.40- 3.45 (m, 2H), 3.29- 3.32 (m, 2H), 3.17- 3.22 (m, 4H), 2.21- 2.24 (m, 1H), 2.07- 2.12 (m, 2H), 1.90- 2.02 (m, 4H), 1.50- 1.80 (m, 21H), 1.28- 1.37 (m, 20H), 0.87- 0.92 (m, 6H); ¹³C NMR (CDCl₃) δ 147.2, 111.6, 79.8, 79.8, 75.0, 49.9, 42.1, 41.9, 33.5, 32.6, 32.1, 31.9, 31.9, 21.9, 31.5, 30.8, 30.2, 29.2, 25.6, 22.7, 20.1, 14.1, 11.2, 11.0.

Synthesis of (*S*)-1-((2*R*, 5*R*, 6*R*)-6-ethyl-5-(prop-1-en-2-yl)tetrahydro-2H-pyran-2-yl)hex-5-en-2-ol (**3.75**)



To a solution of **3.74** (15 mg, 0.035 mmol) in 95 % EtOH (5 mL) was added in Zinc (10.2 mg, 0.156 mmol). The mixture was kept refluxing for 1 h. The zinc was removed by filtration and solvent was removed under reduced pressure. The residue was subjected to column chromatography to give **3.75** (6.3 mg, 72%) as colorless oil. ^1H NMR (CDCl_3) δ 5.84- 5.90 (m, 1H), 5.04- 5.07 (d, $J = 17.0$ Hz, 1H), 4.96- 4.99 (d, $J = 11.1$ Hz, 1H), 4.78 (s, 1H), 4.75 (s, 1H), 3.98 (s, 1H), 3.88- 3.90 (m, 1H), 3.58- 3.62 (m, 1H), 3.33- 3.37 (dt, $J = 2.0, 9.0$ Hz, 1H), 2.15- 2.23 (m, 2H), 1.94- 2.00 (m, 1H), 1.73- 1.76 (m, 1H), 1.56- 1.65 (m, 6H), 1.28- 1.55 (m, 12H), 0.89- 0.90 (m, 3H); ^{13}C NMR (CDCl_3) δ 146.8, 138.8, 114.4, 112.0, 80.2, 79.3, 71.7, 70.6, 50.0, 49.3, 42.6, 36.7, 33.4, 32.4, 31.7, 29.8, 29.7, 29.3, 25.6, 20.1, 14.0.

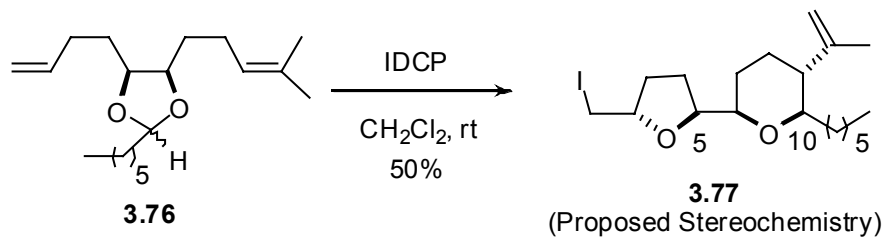
Synthesis of diol acetal **3.76**



^1H NMR (CDCl_3) δ 5.84- 5.89 (m, 1H), 5.13- 5.17 (m, 1H), 5.01- 5.10 (m, 2H), 4.90- 4.92 (m, 1H), 3.93- 3.96 (m, 2H), 2.07- 2.31 (m, 5H), 1.72 (s, 3H), 1.62- 1.70 (m, 6H), 1.31- 1.61 (m, 10H), 0.89- 0.92 (m, 3H); ^{13}C NMR (CDCl_3) δ 138.2, 132.2, 123.8, 123.7,

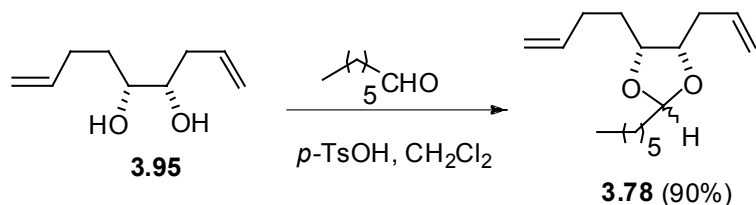
114.9, 114.8, 103.7, 102.5, 101.7, 77.9, 77.7, 77.6, 77.5, 35.3, 34.9, 34.4, 31.8, 31.8, 31.7, 31.7, 30.4, 30.1, 29.4, 29.2, 29.2, 29.0, 29.0, 28.9, 28.7, 27.9, 25.7, 24.8, 24.0, 22.6, 17.6, 17.6, 14.0.

Synthesis of compound **3.77**



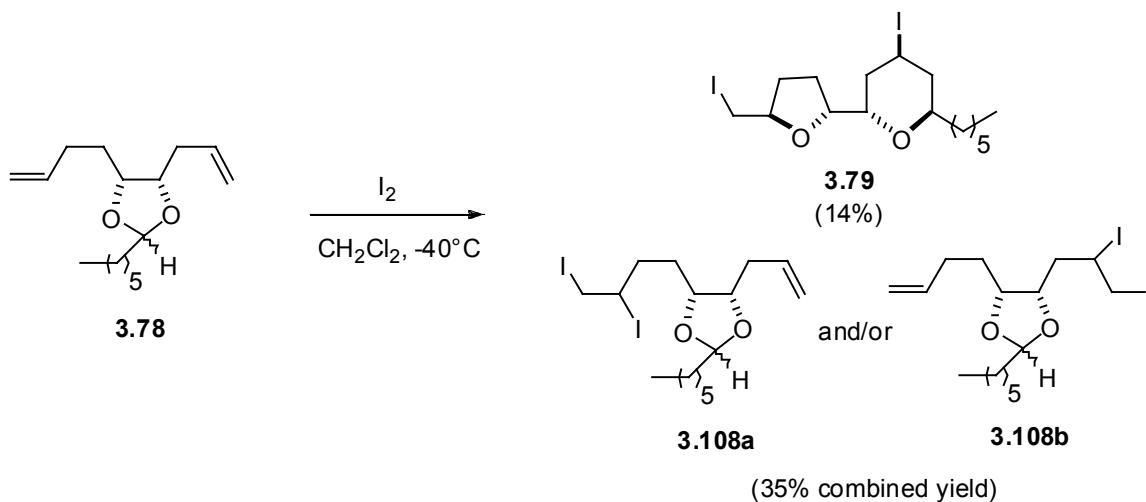
To a solution of acetal diene **3.76** (9 mg, 0.03 mmol) in CH₂Cl₂ (2 mL) were added grinded molecular sieves (pre-activated, 0.5 g) and IDCP (28.7 mg, 0.06 mmol) at 0 °C. The reaction mixture was stirring at rt for 2 h. Then it was quenched by adding saturated aqueous Na₂S₂O₃, extracted with Et₂O and dried (Na₂SO₄). The solvent was removed under reduced pressure and the residue subjected to column chromatography to give **3.77** (12 mg, 50 %) as colorless oil. ¹H NMR (CDCl₃) δ 4.76 (s, 1H), 4.73 (s, 1H), 4.09- 4.11 (m, 1H), 3.98- 4.02 (m, 1H), 3.28- 3.31 (m, 1H), 3.23- 3.26 (m, 2H), 3.17- 3.22 (m, 1H), 2.18- 2.25 (m, 2H), 2.07- 2.10 (m, 1H), 1.91- 1.98 (m, 2H), 1.74- 1.80 (m, 2H), 1.62- 1.68 (m, 2H), 1.25- 1.41 (m, 10H), 0.91- 0.99 (m, 6H); ¹³C NMR (CDCl₃) δ 147.2, 111.6, 101.7, 82.7, 79.9, 78.9, 70.6, 49.9, 34.4, 33.4, 32.1, 31.9, 29.8, 29.2, 29.0, 28.9, 28.4, 27.8, 25.6, 24.0, 23.5, 22.6, 22.5, 20.2, 14.1, 10.6. HRMS (ESI) calcd for C₁₉H₃₃IO₂ (M + NH₄⁺) 438.1863, found 438.1866.

Synthesis of acetal diene **3.78**



$^1\text{H NMR}$ (CDCl_3) δ 5.84- 5.89 (m, 2H), 5.00- 5.16 (m, 4H), 5.05- 5.06 (t, $J = 1.7$ Hz, 1H), 3.97- 4.04 (m, 2H), 2.24- 2.32 (m, 3H), 2.13- 2.15 (m, 1H), 1.62- 1.70 (m, 2H), 1.31- 1.58 (m, 10H), 0.91- 0.94 (m, 3H); $^{13}\text{C NMR}$ (CDCl_3) δ 138.1, 134.8, 117.0, 114.9, 114.8, 103.8, 103.6, 78.4, 77.8, 34.9, 34.7, 31.9, 31.8, 30.4, 30.4, 29.4, 29.3, 29.2, 24.0, 23.9, 22.6, 19.6, 14.1, 14.1.

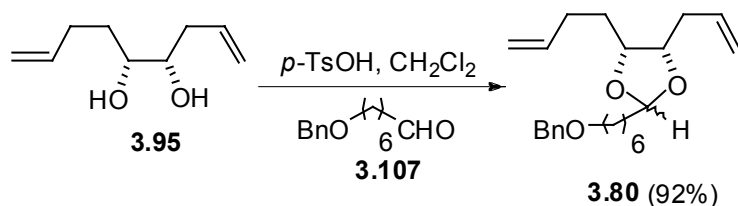
Synthesis of cyclized product **3.79**



To a solution of acetal diene **3.78** (40 mg, 0.159 mmol) in CH_2Cl_2 (10 mL) were added grinded molecular sieves (pre-activated, 0.5 g) and I_2 (60 mg, 0.238 mmol) in CH_2Cl_2 (5 mL) at -40°C . The reaction mixture was stirring at 0°C for 2 h. Then it was quenched

by adding saturated aqueous $\text{Na}_2\text{S}_2\text{O}_3$, extracted with Et_2O , and dried (Na_2SO_4). The solvent was removed under reduced pressure and the residue was subjected to column chromatography to give **3.79** (10 mg, 14 %) and **3.108a/b** (25 mg, 35 %). For **3.79**: ^1H NMR (CDCl_3) δ 4.60 - 4.64 (tt, $J = 4.0, 8.0$ Hz, 1H), 4.40- 4.45 (m, 1H), 4.04- 4.10 (m, 1H), 3.50- 3.56 (m, 1H), 3.47- 3.49 (m, 1H), 3.26- 3.29 (m, 1H), 3.20- 3.24 (m, 1H), 2.58- 2.61 (m, 1H), 2.29- 2.36 (m, 2H), 2.11- 2.22 (m, 2H), 1.93- 2.00 (m, 1H), 1.82- 1.89 (m, 1H), 1.29- 1.47 (m, 10H), 0.90- 0.92 (m, 3H); ^{13}C NMR (CDCl_3) δ 78.8, 78.2, 73.2, 45.2, 38.8, 36.2, 32.0, 31.8, 29.4, 29.2, 25.4, 22.6, 21.0, 14.1, 10.5, 1.0; For **3.108a/b**: ^1H NMR (CDCl_3) δ 5.86- 5.87 (m, 2H), 5.00- 5.17 (m, 4H), 4.91- 4.92 (m, 1H), 4.25- 4.32 (m, 1H), 3.97- 4.09 (m, 4H), 3.20- 3.30 (m, 3H), 3.17- 3.19 (m, 1H), 2.44- 2.45 (m, 1H), 2.09- 2.20 (m, 3H), 1.89- 1.92 (m, 2H), 1.29- 1.42 (m, 10H), 0.90- 0.97 (m, 3H); ^{13}C NMR (CDCl_3) δ 138.1, 134.8, 117.0, 116.9, 114.9, 103.8, 103.7, 81.7, 80.9, 79.0, 45.4, 41.9, 35.6, 34.9, 34.8, 34.7, 32.1, 31.8, 30.4, 29.4, 29.3, 29.2, 28.6, 27.8, 25.4, 23.9, 22.7, 22.6, 14.1, 14.0, 10.4, 1.0.

Synthesis of acetal diene **3.80**

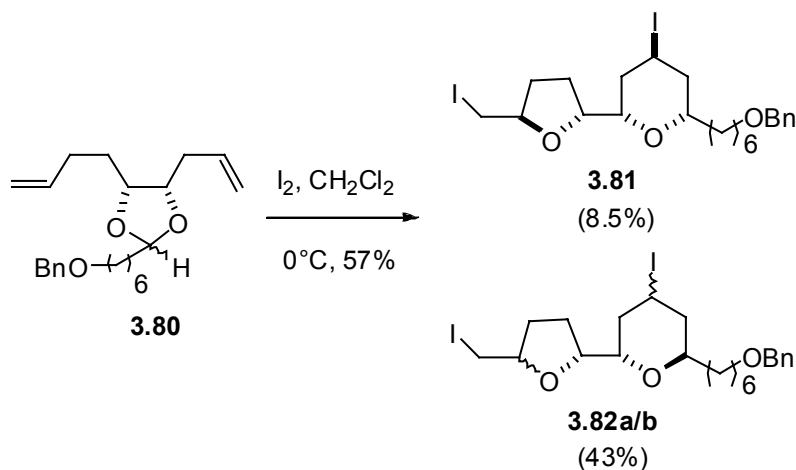


^1H NMR (CDCl_3) δ 7.30- 7.40 (m, 5H), 5.85- 5.88 (m, 2H), 5.00- 5.16 (m, 4H), 4.90- 4.92 (t, $J = 4.8$ Hz, 1H), 4.52 (s, 2H), 3.98- 4.04 (m, 2H), 3.47- 3.50 (t, $J = 6.7$ Hz, 2H), 2.20- 2.32 (m, 4H), 1.36- 1.69 (m, 10H); ^{13}C NMR (CDCl_3) δ 138.7, 138.1, 128.4, 127.7,

127.5, 117.1, 115.0, 103.7, 77.8, 72.9, 70.5, 34.7, 34.6, 30.5, 29.7, 29.4, 29.1, 26.1, 23.9.

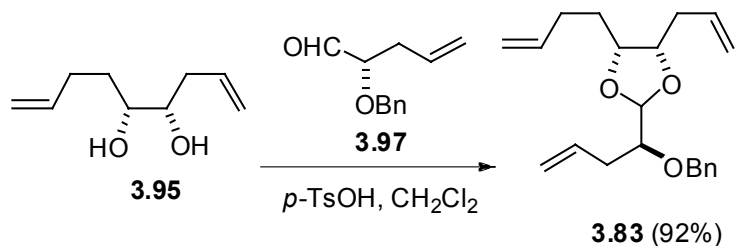
HRMS (ESI) calcd for C₂₃H₃₄O₃ (M + H⁺) 359.2581, found 359.2580.

Synthesis of **3.81** and **3.82a/b**



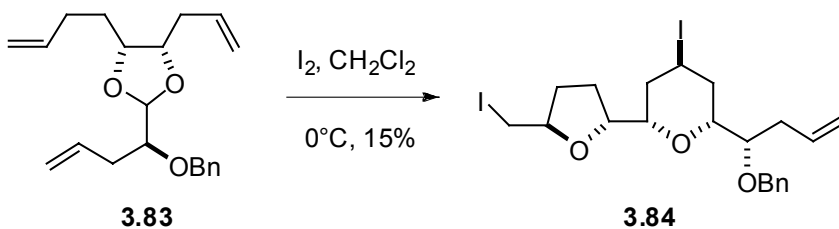
The procedure refers to the synthesis of **3.79**. For **3.81**: ¹H NMR (CDCl₃) δ 7.30- 7.38 (m, 5H), 4.53 (s, 2H), 4.25- 4.27 (tt, *J* = 4.0, 8.0 Hz, 1H), 4.06- 4.07 (m, 1H), 3.98- 4.00 (m, 1H), 3.47- 3.50 (m, 2H), 3.24- 3.30 (m, 3H), 3.17- 3.20 (m, 1H), 2.44- 2.47 (m, 1H), 2.29- 2.31 (m, 1H), 2.08- 2.22 (m, 2H), 1.88- 1.92 (m, 3H), 1.58- 1.72 (m, 6H), 1.28- 1.42 (m, 10H); ¹³C NMR (CDCl₃) δ 128.4, 127.7, 127.5, 81.7, 80.8, 79.0, 72.9, 70.4, 45.4, 41.8, 35.5, 32.1, 29.7, 29.3, 27.8, 26.2, 25.4, 22.6, 10.5. For **3.82a/b**: ¹H NMR (CDCl₃) δ 7.29- 7.37 (m, 5H), 4.89- 4.90 (m, 1H), 4.52- 4.53 (m, 2H), 4.06- 4.09 (m, 2H), 3.75- 3.82 (m, 2H), 3.47- 3.50 (m, 3H), 3.17- 3.30 (m, 3H), 1.74- 2.22 (m, 4H), 1.61- 1.74 (m, 4H), 1.32- 1.57 (m, 8H); ¹³C NMR (CDCl₃) δ 138.7, 128.4, 127.7, 127.5, 81.9, 78.8, 78.7, 78.2, 75.8, 73.6, 73.1, 72.9, 70.5, 45.2, 40.8, 38.8, 37.2, 36.1, 35.3, 34.1, 32.0, 30.9, 29.7, 29.4, 28.0, 26.2, 25.5, 22.4, 21.5, 21.0, 14.1, 10.7, 10.6, 10.3, 1.0.

Synthesis of acetal diene **3.83**



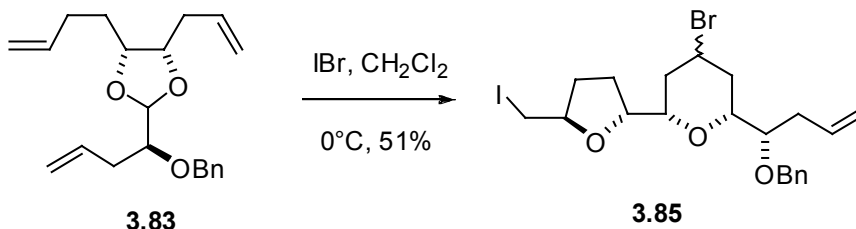
¹H NMR (CDCl₃) δ 7.27- 7.40 (m, 5H), 5.85- 5.92 (m, 3H), 5.00- 5.17 (m, 6H), 4.91- 4.92 (d, *J* = 4.75 Hz, 1H), 4.80- 4.83 (d, *J* = 11.7 Hz, 1H), 4.63- 4.65 (d, *J* = 11.7 Hz, 1H), 3.45- 3.46 (m, 1H), 2.20- 2.39 (m, 6H), 1.54- 1.55 (m, 2H); ¹³C NMR (CDCl₃) δ 137.9, 135.0, 134.7, 128.2, 128.0, 127.5, 117.2, 116.9, 115.1, 104.2, 79.7, 78.0, 73.1, 35.0, 34.6, 30.4, 29.0.

Synthesis of **3.84**



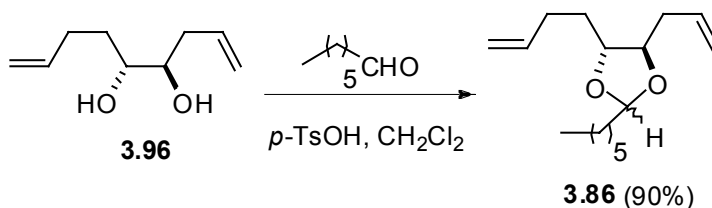
Compound **3.84** was prepared following the procedure used for **3.79**. ¹H NMR (CDCl₃) δ 7.30- 7.38 (m, 5H), 5.86- 5.90 (m, 1H), 5.11- 5.17 (m, 2H), 4.93- 4.94 (m, 1H), 4.62- 4.65 (m, 2H), 4.05- 4.09 (m, 2H), 3.87- 3.89 (m, 1H), 3.80- 3.83 (m, 1H), 3.58- 3.59 (m, 1H), 3.28- 3.31 (m, 1H), 3.19- 3.22 (m, 1H), 2.35- 2.37 (m, 2H), 2.09- 2.21 (m, 4H), 1.97- 1.99 (m, 1H), 1.67- 1.72 (m, 2H), 1.56- 1.59 (m, 1H); ¹³C NMR (CDCl₃) δ 138.6, 134.5, 128.3, 127.9, 127.6, 117.4, 81.8, 80.1, 78.8, 75.9, 75.5, 72.4, 37.3, 35.9, 35.5, 32.1, 30.2, 27.8, 22.4, 14.1, 10.6.

Synthesis of 3.85



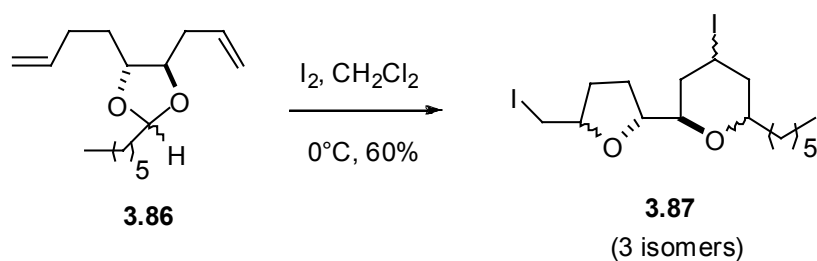
To a solution of acetal diene **3.83** (110 mg, 0.345 mmol) in CH₂Cl₂ (15 mL) were added grinded molecular sieves (pre-activated, 0.5 g) and IBr (107 mg, 0.517 mmol, 1 M in CH₂Cl₂) at 0 °C. The reaction mixture was stirring at rt for 10 min. Then it was quenched by adding saturated aqueous Na₂S₂O₃, extracted with Et₂O, and dried (Na₂SO₄). The solvent was removed under reduced pressure and the residue subjected to column chromatography to give product **3.85** (102 mg, 51 %). ¹H NMR (CDCl₃) δ 7.30- 7.38 (m, 5H), 5.86- 5.90 (m, 1H), 5.10- 5.16 (m, 2H), 4.81- 4.82 (m, 1H), 4.60- 4.66 (m, 2H), 4.02- 4.09 (m, 2H), 3.91- 3.93 (m, 1H), 3.86- 3.88 (m, 1H), 3.56- 3.57 (m, 1H), 3.28- 3.29 (m, 1H), 3.20- 3.22 (m, 1H), 2.36- 2.39 (m, 2H), 2.10- 2.20 (m, 4H), 1.89- 1.98 (m, 2H), 1.70- 1.79 (m, 2H); ¹³C NMR (CDCl₃) δ 135.2, 134.8, 134.6, 129.0, 128.8, 128.7, 128.6, 128.5, 128.4, 128.2, 128.1, 128.0, 127.9, 126.3, 117.5, 82.0, 81.9, 80.7, 80.2, 74.4, 74.1, 72.5, 50.4, 36.1, 36.0, 35.4, 35.3, 35.2, 34.7, 32.0, 31.9, 31.8, 27.7, 27.6, 27.5, 10.6. HRMS (ESI) calcd for C₂₁H₂₈BrIO₃ (M + H⁺) 554.0587, found 554.0594.

Synthesis of acetal diene 3.86



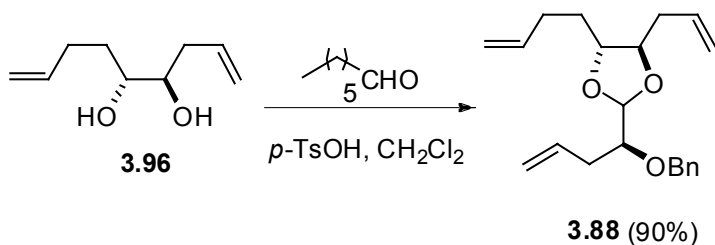
^1H NMR (CDCl_3) δ 5.82- 5.90 (m, 2H), 5.06- 5.18 (m, 3H), 5.00- 5.02 (m, 2H), 3.67- 3.72 (m, 2H), 2.19- 2.41 (m, 3H), 2.13- 2.18 (m, 1H), 1.63- 1.70 (m, 2H), 1.31- 1.60 (m, 10H), 0.89- 0.92 (m, 3H); ^{13}C NMR (CDCl_3) δ 138.0, 138.0, 117.6, 117.6, 114.9, 103.6, 103.3, 81.1, 80.9, 79.8, 79.7, 37.6, 37.4, 34.4, 34.4, 32.3, 32.2, 32.2, 31.8, 30.1, 30.1, 29.3, 23.9, 23.9, 22.6, 14.1. HRMS (ESI) calcd for $\text{C}_{16}\text{H}_{28}\text{O}_2$ ($\text{M} + \text{H}^+$) 253.2162, found 253.2158.

Synthesis of 3.87



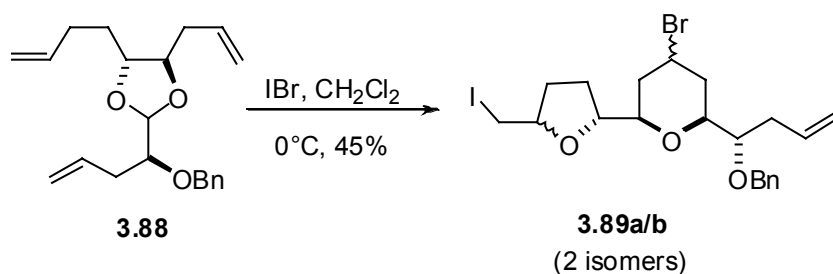
Compound **3.87** was prepared following the procedure used for **3.79**. ^1H NMR (CDCl_3) δ 4.27- 4.30 (m, 1H), 4.12- 4.14 (m, 1H), 4.03- 4.07 (m, 1H), 3.31- 3.34 (m, 1H), 3.26- 3.29 (m, 2H), 3.16- 3.20 (m, 1H), 2.31- 2.34 (m, 1H), 2.23- 2.27 (m, 2H), 2.09- 2.14 (m, 1H), 2.01- 2.03 (m, 1H), 1.83- 1.95 (m, 2H), 1.66- 1.70 (m, 1H), 1.29- 1.56 (m, 10H), 0.89- 0.92 (m, 3H); ^{13}C NMR (CDCl_3) δ 81.4, 81.0, 79.1, 79.0, 45.1, 41.1, 35.7, 32.4, 31.8, 29.2, 28.0, 25.3, 22.9, 22.6, 14.1, 10.6. HRMS (ESI) calcd for $\text{C}_{16}\text{H}_{28}\text{I}_2\text{O}_2$ ($\text{M} + \text{NH}_4^+$) 524.0517, found 524.0518.

Synthesis of acetal diene **3.88**



$^1\text{H NMR}$ (CDCl_3) δ 7.30- 7.40 (m, 5H), 5.82- 5.93 (m, 3H), 5.00- 5.18 (m, 7H), 4.78- 4.82 (t, $J = 11.5$ Hz, 1H), 4.63- 4.66 (d, $J = 11.7$ Hz, 1H), 3.70- 3.76 (m, 2H), 3.46- 3.49 (m, 1H), 2.35- 2.40 (m, 4H), 2.17- 2.20 (m, 2H), 1.66- 1.70 (m, 2H); $^{13}\text{C NMR}$ (CDCl_3) δ 138.7, 138.6, 137.9, 137.9, 135.0, 133.8, 133.7, 128.3, 128.0, 127.9, 127.5, 117.8, 117.7, 117.0, 115.0, 104.1, 104.0, 81.0, 80.8, 80.0, 79.8, 79.1, 73.1, 73.1, 37.2, 37.0, 34.9, 34.8, 31.9, 31.8, 30.2, 30.1. HRMS (ESI) calcd for $\text{C}_{21}\text{H}_{28}\text{O}_3$ ($\text{M} + \text{NH}_4^+$) 346.2377, found 346.2369.

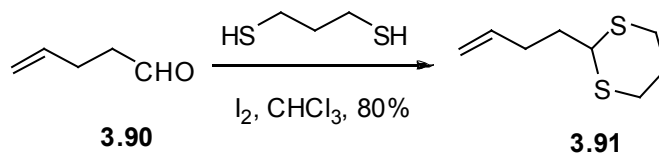
Synthesis of **3.89a/b**



Compound **3.89a/b** was prepared following the procedure used for **3.83**. $^1\text{H NMR}$ (C_6D_6) δ 7.36- 7.46 (m, 5H), 5.96- 5.97 (m, 1H), 5.25- 5.29 (m, 2H), 4.54- 4.56 (m, 2H), 4.40- 4.45 (m, 2H), 4.15- 4.17 (m, 1H), 4.03- 4.05 (m, 1H), 3.80- 3.82 (m, 1H), 3.69- 3.70 (m, 1H), 3.30- 3.31 (m, 1H), 3.22- 3.25 (m, 1H), 3.14- 3.16 (m, 1H), 2.50- 2.51 (m, 1H), 2.39- 2.40 (m, 1H), 1.96- 1.99 (m, 1H), 1.70- 1.81 (m, 5H), 1.48- 1.54 (m, 2H); $^{13}\text{C NMR}$ (C_6D_6) δ 139.1, 135.2, 117.1, 81.6, 80.3, 79.7, 74.2, 73.0, 72.3, 51.9, 36.0, 35.1, 24.7,

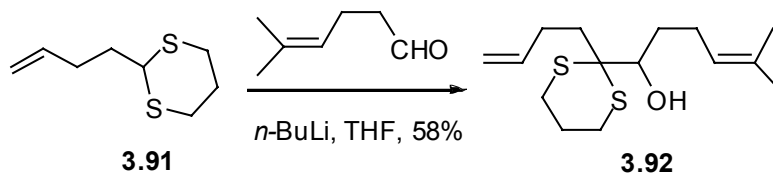
31.6, 27.4, 10.6. HRMS (ESI) calcd for $C_{21}H_{28}BrIO_3$ ($M + NH_4^+$) 552.0605, found 552.0608.

Synthesis of 2-(but-3-enyl)-1,3-dithiane (3.91)



To a solution of pent-4-enal (**3.90**) (450 mg, 5.36 mmol) in chloroform were added propane-1,3-dithiol (696 mg, 6.43 mmol) and iodine (136 mg, 0.536 mmol) at 0 °C. The reaction mixture was kept at rt for 2 h, then quenched with aqueous $Na_2S_2O_3$ and extracted with ether. The solvent was removed under reduced pressure and the residue was subjected to column chromatography to give **3.91** (750 mg, 80 %) as a yellow oil. 1H NMR ($CDCl_3$) δ 5.78- 5.86 (m, 1H), 5.07- 5.11 (d, $J = 20.0$ Hz, 1H), 5.02- 5.04 (d, $J = 9.0$ Hz, 1H), 4.05- 4.07 9 (t, $J = 7$ Hz, 1H), 2.80- 2.92 (m, 4H), 2.27- 2.32 (m, 2H), 2.11- 2.17 (m, 1H), 1.82- 1.99 (m, 3H); ^{13}C NMR ($CDCl_3$) δ 137.1, 115.6, 70.6, 46.7, 34.6, 30.5, 30.3, 26.0.

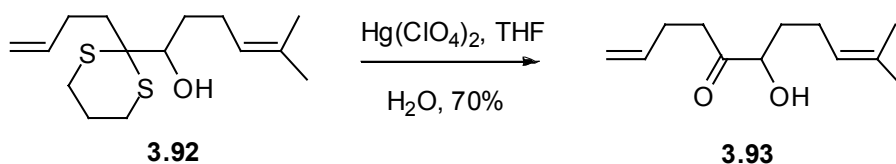
Synthesis of 1-(2-(but-3-enyl)-1,3-dithian-2-yl)-5-methylhex-4-en-1-ol (3.92)



To the solution of **3.91** (240 mg, 1.38 mmol) in THF was added $n-BuLi$ (0.44 mL, 1.1 mmol, 2.5 M in hexane) at -78 °C. The mixture was then warmed to 0 °C for 0.5 h and re-cooled to -78 °C, at which time 5-methylhex-4-enal (103 mg, 0.92 mmol) was added.

The reaction mixture was stirring at rt for 2 h, then quenched with saturated aqueous NH_4Cl and extracted with ether. The solvent was removed under reduced pressure and the residues subjected to column chromatography to give **3.92** (230 mg, 58 %) as colorless oil. ^1H NMR (CDCl_3) δ 5.80- 5.85 (m, 1H), 5.20- 5.22 (t, $J = 6.6$ Hz, 1H), 5.04- 5.08 (d, $J = 17.1$ Hz, 1H), 4.99- 4.97 (d, $J = 10.2$ Hz, 2H), 3.99- 4.01 (d, $J = 10.0$ Hz, 1H), 2.96- 3.01 (m, 2H), 2.73 (s, 1H), 2.63- 2.69 (m, 2H), 2.30- 2.38 (m, 3H), 2.18- 2.21 (m, 1H), 2.06- 2.09 (m, 1H), 1.94- 1.99 (m, 2H), 1.86- 1.88 (m, 1H), 1.74 (s, 3H), 1.71- 1.72 (m, 2H), 1.68 (s, 3H), 1.51- 1.53 (m, 1H); ^{13}C NMR (CDCl_3) δ 138.2, 132.4, 124.2, 114.7, 71.2, 59.0, 33.8, 30.4, 28.9, 26.0, 25.9, 25.7, 25.1, 24.5, 17.8.

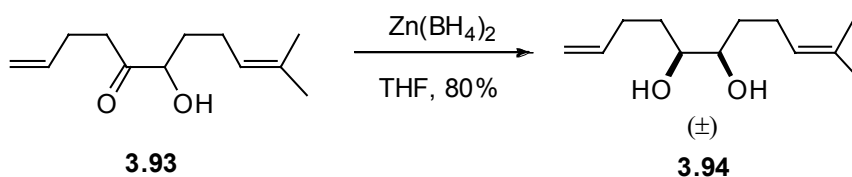
Synthesis of 6-hydroxy-10-methylundeca-1, 9-dien-5-one (**3.93**)



To a solution of **3.92** (85 mg, 0.3 mmol) in THF was added in $\text{Hg}(\text{ClO}_4)_2$ (143 mg, 0.36 mmol) at 0 °C. The mixture was stirring at 0 °C for 1 h and at rt for 2 h. Then it was quenched by saturated aqueous NaHCO_3 and extracted with ether. The solvent was removed under reduced pressure and the residue subjected to column chromatography to give **3.93** (41 mg, 70 %) as a colorless oil. ^1H NMR (CDCl_3) δ 5.79- 5.87 (m, 1H), 5.11- 5.14 (m, 1H), 5.06- 5.09 (d, $J = 18.6$ Hz, 1H), 5.02- 5.04 (d, $J = 9$ Hz, 1 H), 4.16- 4.20 (m, 1H), 3.44-3.45 (d, $J = 4.9$ Hz, 1H), 2.52- 2.67 (m, 2H), 2.38- 2.41 (m, 2H), 2.19- 2.22 (m, 1H), 2.08- 2.18 (m, 1H), 1.85- 1.91 (m, 1H), 1.72 (s, 3H), 1.68 (s, 3H), 1.55- 1.64 (m,

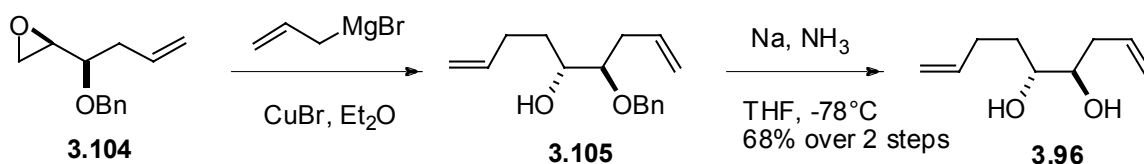
2H); ^{13}C NMR (CDCl_3) δ 211.6, 136.6, 133.3, 123.0, 115.8, 76.1, 37.0, 33.8, 27.5, 25.7, 23.4, 17.7.

Synthesis of (5*S*, 6*R*)-10-methylundeca-1, 9-diene-5, 6-diol (3.94)



To a solution of **3.93** (20 mg, 0.1 mmol) in THF (2 mL) was added $\text{Zn(BH}_4)_2$ (19 mg, 0.2 mmol) at 0 °C. The reaction mixture was stirring at rt for 0.5 h and quenched by adding 2 drops of H_2O at 0 °C. Then it was extracted with EtOAc and the solvent was removed under reduced pressure. The residue was subjected to column chromatography to give **3.94** (15.8 mg, 80 %) as a white powder. ^1H NMR (CDCl_3) δ 5.83- 5.92 (m, 1H), 5.16- 5.18 (m, 1H), 5.07- 5.11 (d, $J = 17.1$ Hz, 1H), 5.01- 5.03 (d, $J = 10.2$ Hz, 1H), 3.65- 3.72 (m, 1H), 2.28- 2.34 (m, 1H), 2.08- 2.24 (m, 3H), 1.85- 1.93 (m, 2H), 1.77 (s, 3H), 1.66 (s, 3H), 1.48- 1.60 (m, 4H); ^{13}C NMR (CDCl_3) δ 138.4, 132.6, 123.9, 115.1, 74.5, 74.6, 31.3, 30.4, 30.2, 25.7, 24.5, 17.7.

Synthesis of (4*R*, 5*R*)-nona-1, 8-diene-4, 5-diol (3.96)

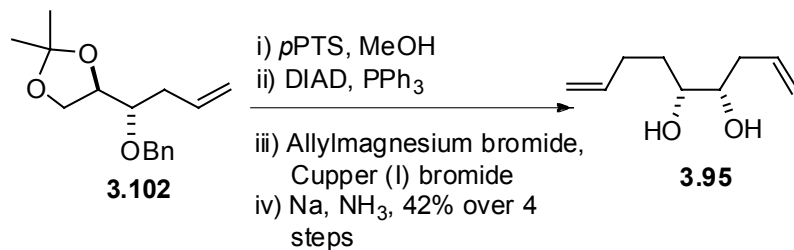


In a 100 mL round-bottom flask equipped with magnetic stirring bar, were placed CuBr (0.75 g, 5.22 mmol) and anhydrous THF (30 mL). Allylmagnesium bromide (87 mmol, 1

M in Et₂O) was added dropwise at 0 °C, and then epoxide **3.104** (1.77 g, 8.7 mmol) was introduced. The reaction was stirred at 0 °C for 3 h, then poured into ice cold saturated aqueous NH₄Cl, and extracted with ether. The organic phase was washed with brine, dried (Na₂SO₄), filtered, and evaporated under reduced pressure. The residue was purified by FCC to give the derived alcohol **3.105** (1.82 g, 85%); ¹H NMR (CDCl₃) δ 7.32- 7.40 (m, 5H), 5.82- 5.91 (m, 2H), 4.98- 5.19 (m, 4H), 4.73- 4.75 (m, 1H), 4.50- 4.75 (m, 1H), 3.59- 3.61 (m, 1H), 3.35- 3.39 (m, 1H), 2.50- 2.53 (m, 1H), 2.38- 2.40 (m, 1H), 2.29- 2.30 (m, 1H), 2.25- 2.28 (m, 1H), 2.15- 2.18 (m, 1H); ¹³C NMR (CDCl₃) δ 138.5, 138.1, 134.2, 128.5, 127.9, 127.8, 117.6, 114.8, 81.5, 72.3, 71.9, 34.8, 32.6, 29.9.

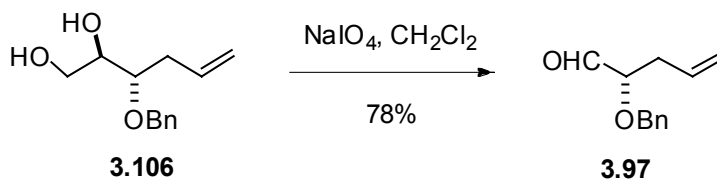
To a solution of **3.105** (294 mg, 1.2 mmol) in THF (20 mL) were added the liquidized NH₃ (c.a. 5 mL) and Na (55 mg, 2.4 mmol) at -78 °C. The reaction mixture was stirring for 10 min at -78 °C, and then solid NH₄Cl was added in to quench it until the dark blue color discharged. MeOH and H₂O were added in to ensure the absence of Na and it was extracted with EtOAc and dried (Na₂SO₄). The solvent was removed under reduced pressure and the residue subjected to column chromatography to give **3.96** (0.156 g, 84 %). ¹H NMR (CDCl₃) δ 5.84- 5.90 (m, 2H), 5.21- 5.22 (m, 1H), 5.19 (s, 1H), 5.07- 5.17 (m, 1H), 5.00- 5.03 (m, 1H), 3.51- 3.55 (m, 1H), 2.37-2.39 (m, 1H), 2.19- 2.31 (m, 3H), 2.12- 2.13 (d, *J* = 5.2 Hz, 1H), 2.08- 2.09 (d, *J* = 4.4 Hz, 1H), 1.61- 1.66 (m, 2H); ¹³C NMR (CDCl₃) δ 138.3, 134.4, 118.5, 115.1, 73.4, 73.2, 38.3, 32.7, 29.9. HRMS (ESI) calcd for C₉H₁₆O₂ (M + Na⁺) 179.1043, found 179.1036.

Synthesis of (4*S*, 5*R*)-nona-1, 8-diene-4, 5-diol (3.95)



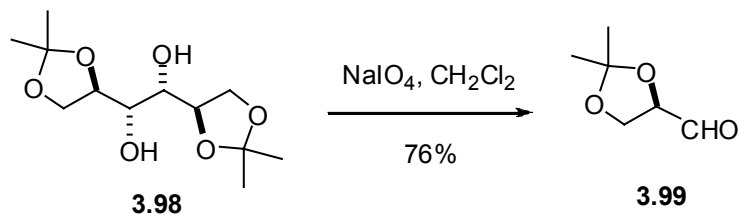
The procedure refers to the synthesis of **3.96**. ¹H NMR (CDCl₃) δ 5.83- 5.91 (m, 2H), 5.22- 5.23 (m, 1H), 5.21 (s, 1H), 5.08- 5.12 (m, 1H), 5.01- 5.04 (m, 1H), 3.71- 3.73 (m, 1H), 3.64- 3.67 (m, 1H), 2.17- 2.36 (m, 4H), 2.03- 2.04 (m, 2H), 1.58- 1.62 (m, 2H); ¹³C NMR (CDCl₃) δ 138.3, 134.8, 118.6, 115.2, 73.4, 73.1, 36.0, 30.8, 30.3. HRMS (ESI) calcd for C₉H₁₆O₂ (M + Na⁺) 179.1043, found 179.1041.

Synthesis of (*S*)-2-(benzyloxy)pent-4-enal (3.97)



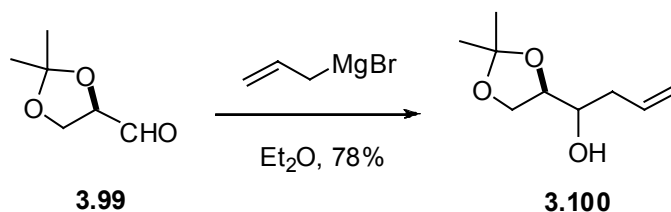
Compound **3.97** was prepared following the procedure used for **3.99**. ¹H NMR (CDCl₃) δ 9.69 (s, 1H), 7.33- 7.41 (m, 5H), 5.81- 5.90 (m, 1H), 5.15- 5.22 (m, 2H), 4.62- 4.72 (m, 2H), 3.86- 3.89 (m, 1H), 2.49- 2.57 (m, 2H); ¹³C NMR (CDCl₃) δ 203.3, 137.2, 132.4, 128.6, 128.2, 128.0, 118.5, 82.8, 72.5, 34.7.

Synthesis of (*R*)-2, 2-dimethyl-1, 3-dioxolane-4-carbaldehyde (**3.99**)



To a solution of mannitol derivative **3.98** (5 g, 19 mmol) in CH₂Cl₂ was added a solution of saturated NaHCO₃ (2.3 mL) 0 °C. NaIO₄ (6.12 g, 28.6 mmol) was added slowly and suspension was stirred vigorously for 2 h until TLC analysis indicated the reaction was complete. Anhydrous MgSO₄ was added, and the reaction mixture was filtered and concentrated to provide **3.99** (3.75 g, 76%) as clear oil. ¹H NMR (CDCl₃) δ 9.71 (s, 1H), 4.36- 4.40 (m, 1H), 4.08- 4.20 (m, 2H), 1.50 (s, 3H), 1.42 (s, 3H).

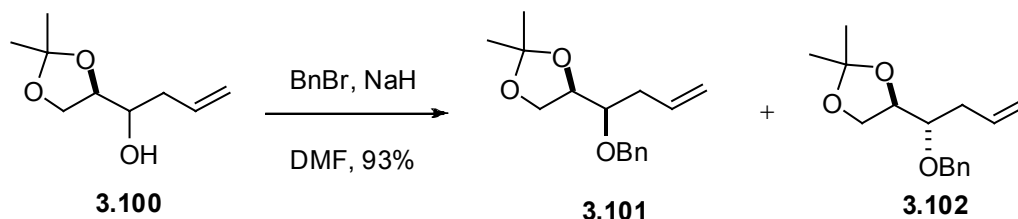
Synthesis of alcohol **3.100**



To a solution of aldehyde **3.99** (562 mg, 4.32 mmol) in anhydrous ether at 0 °C was added allylmagnesium bromide (8.65 mL, 1 M in Et₂O). The reaction mixture was stirred at rt for 0.5 h, and then was quenched by NH₄Cl and extracted with Et₂O and dried (Na₂SO₄). The solvent was removed under reduced pressure. The residue was subjected to column chromatography to give **3.100** (580 mg, 78 %) as a colorless oil. ¹H NMR (CDCl₃) δ 5.83- 5.91 (m, 1H), 5.14- 5.21 (m, 2H), 3.95- 4.07 (m, 2H), 3.77- 3.82 (m, 1H), 2.21- 2.37 (m, 2H), 2.00- 2.01 (d, J = 3.2 Hz, 1H), 1.46- 1.47 (m, 3H), 1.39- 1.40

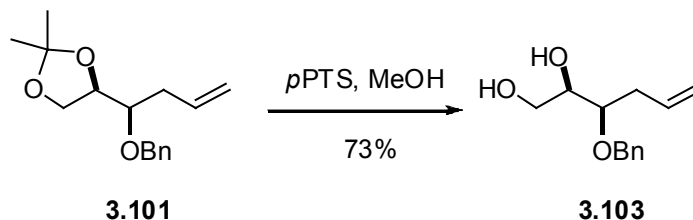
(m, 3H); ^{13}C NMR (CDCl_3) δ 134.0, 133.9, 117.9, 109.4, 109.1, 78.5, 77.2, 71.6, 70.4, 66.1, 38.3, 26.6, 25.3.

Synthesis of benzylether **3.101** and **3.102**



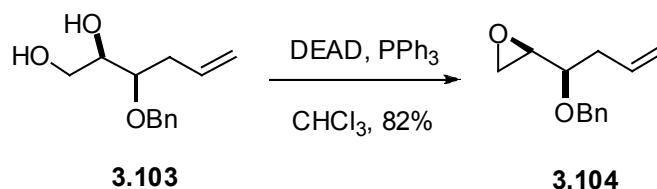
To a solution of alcohol **3.100** (4.1 g, 23.8 mmol) in DMF at 0 °C were added NaH (1.43 g, 35.7 mmol, 60 % in mineral oil) and BnBr (6.1 g, 35.7 mmol). The mixture was kept at rt for 2 h, and then quenched by saturated NH_4Cl solution, extracted with ether and dried (NaSO_4). The solvent was removed under reduced pressure. The residue was subjected to column chromatography to give **3.101** (2.68 g, 43 %) and **3.102** (3.1 g, 50 %). For **3.101**: ^1H NMR (CDCl_3) δ 7.29- 7.40 (m, 5 H), 5.86- 5.95 (m, 1H), 5.08- 5.15 (m, 2H), 4.68- 4.76 (m, 2H), 4.22- 4.26 (m, 1H), 4.00- 4.03 (m, 1H), 3.72- 3.75 (m, 1H), 3.52- 3.55 (m, 1H), 2.31- 2.37 (m, 1H), 2.20- 2.29 (m, 1H), 1.52 (s, 3H), 1.46 (s, 3H); ^{13}C NMR (CDCl_3) δ 138.6, 134.6, 128.3, 127.8, 127.6, 117.2, 109.3, 79.3, 77.9, 72.6, 65.8, 35.3, 26.5, 25.4; For **3.102**: ^1H NMR (CDCl_3) δ 7.28- 7.50 (m, 5H), 5.88- 5.95 (m, 1H), 5.12- 5.20 (m, 2H), 4.61- 4.69 (m, 1H), 4.08- 4.14 (m, 1H), 4.05- 4.07 (m, 1H), 3.92- 3.94 (m, 1H), 3.58- 3.62 (m, 1H), 2.44- 2.48 (m, 1H), 2.35- 2.40 (m, 1H), 1.45 (s, 3H), 1.39 (s, 3H); ^{13}C NMR (CDCl_3) δ 138.3, 134.2, 128.5, 127.9, 127.8, 117.8, 109.1, 78.7, 72.6, 66.4, 35.5, 26.7, 25.4.

Synthesis of (2*R*, 3*R*)-3-(benzyloxy)hex-5-ene-1,2-diol (3.103)



To a solution of **3.101** (2 g, 7.63 mmol) in MeOH (130 mL) was added *p*PTS (3.83 g, 15.3 mmol) at 0 °C. The mixture was kept stirring overnight and the solvent was removed under reduced pressure. The residue was subjected to column chromatography to give **3.103** (1.24 g, 73 %) as colorless oil. ¹H NMR (CDCl₃) δ 7.29- 7.39 (m, 5H), 5.85- 5.93 (m, 1H), 5.14- 5.24 (m, 1H), 4.70- 4.80 (m, 1H), 4.50- 4.56 (m, 1H), 3.57- 3.74 (m, 4H), 2.63- 2.64 (m, 1H), 2.50- 2.61 (m, 1H), 2.40- 2.45 (m, 1H), 2.13- 2.17 (m, 1H); ¹³C NMR (CDCl₃) δ 137.8, 133.8, 128.6, 128.1, 128.0, 118.0, 79.0, 72.5, 72.1, 63.8, 34.6.

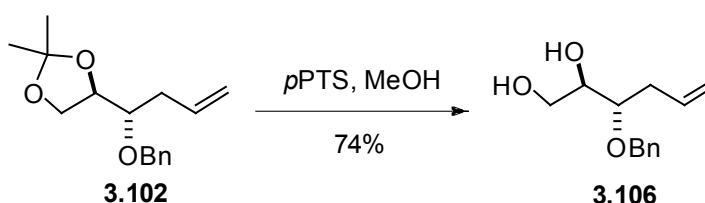
Synthesis of (*R*)-2-((*R*)-1-(benzyloxy)but-3-enyl)oxirane (3.104)



To a solution of **3.103** (1.15 g, 5.18 mmol) in CH₂Cl₂ (30 mL) at 0 °C were added DEAD (1.65 g, 9.5 mmol) and PPh₃ (2.5 g, 9.5 mmol). The reaction mixture was heated to 60 °C for 2 h. Then the solvent was removed under reduced pressure. The residue was subjected to column chromatography to give **3.104** (0.87 g, 82 %) as colorless oil. ¹H

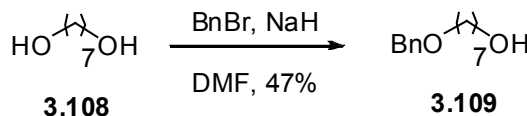
NMR (CDCl₃) δ 7.28- 7.41 (m, 5H), 5.84- 5.92 (m, 1H), 5.09- 5.17 (m, 2H), 4.81- 4.86 (m, 1H), 4.60- 4.62 (m, 1H), 3.12- 3.17 (m, 1H), 3.07- 3.09 (m, 1H), 2.80- 2.82 (m, 1H), 2.50- 2.54 (m, 1H), 2.37- 2.46 (m, 2H); ¹³C NMR (CDCl₃) δ 138.5, 133.9, 128.3, 127.7, 127.5, 117.5, 80.1, 71.7, 54.6, 43.4, 37.1.

Synthesis of (2*R*, 3*S*)-3-(benzyloxy)hex-5-ene-1,2-diol (**3.106**)



The procedure refers to the synthesis of **3.103**. ¹H NMR (CDCl₃) δ 7.31-7.40 (m, 5H), 5.86- 5.93 (m, 1H), 5.13- 5.21 (m, 2H), 4.70- 4.72 (d, *J* = 11.4 Hz, 1H), 4.54- 4.56 (d, *J* = 11.4 Hz, 1H), 3.79- 3.82 (m, 1H), 3.73- 3.77 (m, 1H), 3.65- 3.69 (m, 1H), 2.49- 2.52 (m, 1H), 2.48- 2.49 (m, 1H), 2.39- 2.46 (m, 1H); ¹³C NMR (CDCl₃) δ 138.0, 134.1, 128.6, 128.0, 127.9, 117.8, 80.6, 72.6, 72.3, 63.2, 35.1.

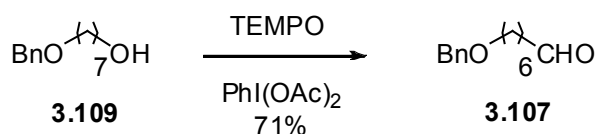
Synthesis of **3.109**



To a solution of **3.108** (1 g, 7.23 mmol) in DMF (30 mL) were added NaH (0.289 g, 7.23 mmol, 60 % in mineral oil) and BnBr (1.24 g, 7.23 mmol) at 0 °C. The reaction mixture was kept at rt for 2 h. Then it was quenched by adding saturated aqueous NH₄Cl, extracted with Et₂O, and dried (Na₂SO₄). The solvent was removed under reduced

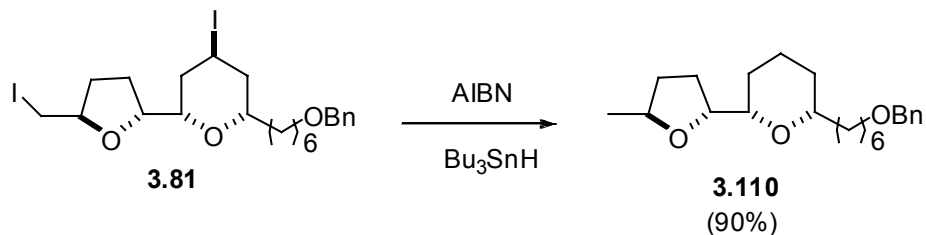
pressure and the residue was subjected to column chromatography to give **3.109** (0.75 g, 45 %). ^1H NMR (CDCl_3) δ 7.28- 7.31 (m, 5H), 4.53- 4.56 (m, 2H), 3.61- 3.66 (m, 2H), 3.44- 3.50 (m, 2H), 1.58- 1.77 (m, 4H), 1.29- 1.53 (m, 6H); ^{13}C NMR (CDCl_3) δ 138.7, 128.4, 127.7, 127.5, 72.9, 70.4, 63.1, 32.7, 29.7, 29.3, 26.2, 25.7.

Synthesis of **3.107**



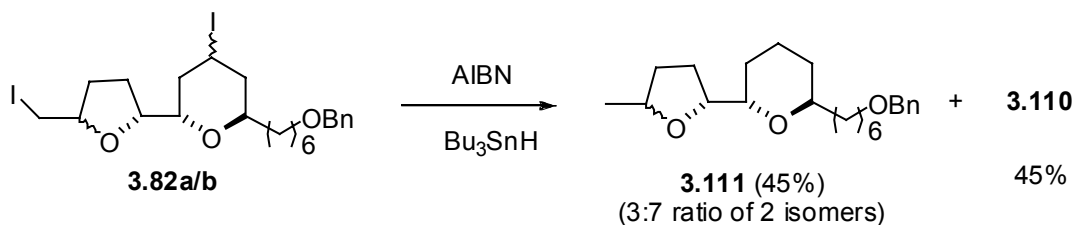
To a solution of **3.109** (0.75 g, 3.38 mmol) in pentane (20 mL) and CH_2Cl_2 (4 mL) were added TEMPO (53 mg, 0.338 mmol) and PhI(OAc)_2 (1.2 g, 3.72 mmol) at 0°C . The reaction mixture was stirring at rt for 1 h. Then it was quenched by adding saturated aqueous $\text{Na}_2\text{S}_2\text{O}_3$, extracted with Et_2O , and dried (Na_2SO_4). The solvent was removed under reduced pressure and the residue subjected to column chromatography to give **3.107** (0.53 g, 71 %). ^1H NMR (CDCl_3) δ 9.79 (s, 1H), 7.29- 7.44 (m, 5H), 4.56 (s, 2H), 3.48- 3.50 (m, 2H), 2.45- 2.47 (m, 2H), 1.63- 1.67 (m, 5H), 1.34- 1.40 (m, 5H); ^{13}C NMR (CDCl_3) δ 202.9, 238.6, 128.4, 127.7, 127.5, 72.9, 70.3, 43.9, 29.6, 29.0, 26.0, 22.0.

Synthesis of **3.110**



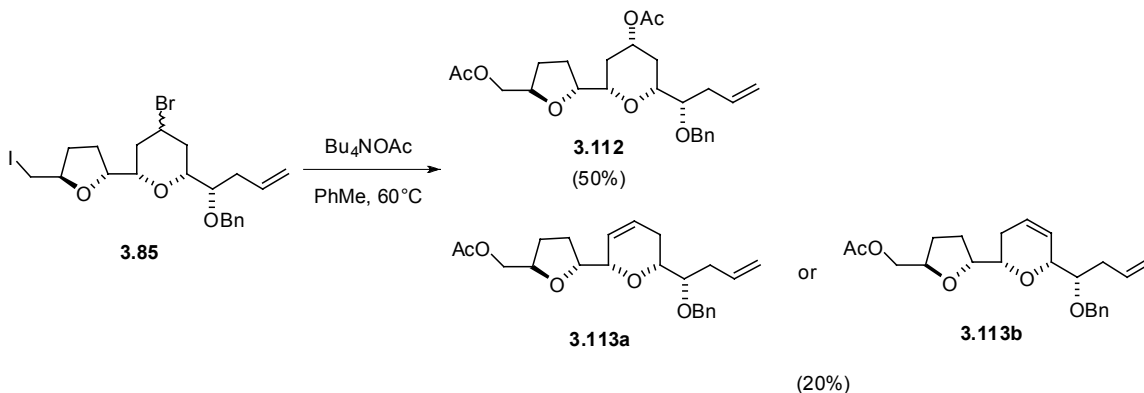
To a solution of **3.81** (21 mg, 0.034 mmol) in benzene (5 mL) were added AIBN (1.69 mg, 0.01 mmol) and Bu₃SnH (30 mg, 0.102 mmol). The reaction mixture was heated to reflux for 2 h and the solvent was removed under the reduced pressure. The residue was subjected to column chromatography to give **3.110** (11 mg, 90 %). ¹H NMR (CDCl₃) δ 7.30- 7.38 (m, 5H), 4.53 (s, 2H), 4.05- 4.07 (m, 1H), 3.89- 3.91 (m, 1H), 3.46- 3.49 (m, 2H), 3.17- 3.20 (m, 2H), 2.02- 2.05 (m, 2H), 1.84- 1.88 (m, 2H), 1.75- 1.76 (1H), 1.64- 1.67 (m, 3H), 1.32- 1.67 (m, 10H), 1.21-1.22 (m, 3H), 1.18- 1.19 (m, 2H); ¹³C NMR (CDCl₃) δ 138.6, 128.4, 127.7, 127.6, 81.4, 80.3, 78.0, 75.6, 72.9, 70.5, 36.5, 33.7, 31.8, 29.8, 29.6, 29.2, 28.5, 28.4, 26.8, 26.3, 25.7, 23.4, 21.2, 16.4, 14.2, 13.8. HRMS (ESI) calcd for C₂₃H₃₆O₃ (M + NH₄⁺) 378.3003, found 378.3009.

Synthesis of **3.111**



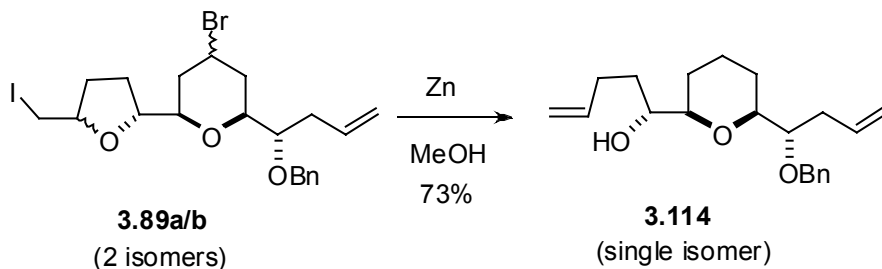
Compound **3.111** was prepared following the procedure used for **3.110**. For **3.111**: ¹H NMR (C₆D₆) δ 7.10- 7.33 (m, 5H), 4.35 (s, 2H), 4.15- 4.17 (m, 1H), 3.95- 4.03 (m, 1H), 3.52- 3.58 (m, 2H), 3.33- 3.35 (m, 2H), 1.70- 1.92 (m, 4H), 1.61- 1.68 (m, 4H), 1.22- 1.54 (m, 12H), 1.14- 1.15 (m, 3H); ¹³C NMR (C₆D₆) δ 139.3, 128.5, 128.4, 128.3, 128.2, 128.1, 127.4, 127.2, 79.5, 78.8, 75.5, 75.0, 73.8, 73.4, 72.6, 71.5, 71.4, 70.2, 33.8, 33.2, 33.1, 32.7, 30.1, 30.0, 29.6, 28.8, 27.6, 27.6, 26.4, 25.9, 21.2, 18.6, 18.5. HRMS (ESI) calcd for C₂₃H₃₆O₃ (M + NH₄⁺) 378.3003, found 378.3008.

Synthesis of 3.112 and 3.113a/b



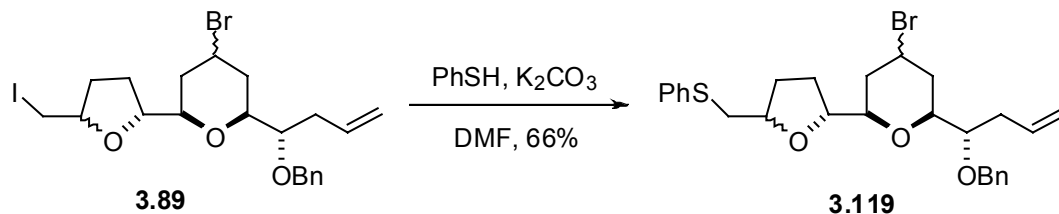
To a solution of **3.85** (65 mg, 0.106 mmol) in toluene (5 mL) was added Bu_4NOAc (224 mg, 0.743 mmol). The reaction was heated to 60°C for 1 h. Then solvent was removed under reduced pressure. The residue was subjected to column chromatography to give the products **3.112** (24 mg, 50 %) and **3.113a/b** (9 mg, 20 %). For **3.112**: ^1H NMR (CDCl_3) δ 7.30- 7.37 (m, 5H), 5.85- 5.88 (m, 1H), 5.08- 5.15 (m, 2H), 4.63- 4.94 (m, 1H), 4.60- 4.63 (m, 2H), 4.22- 4.23 (m, 1H), 4.14- 4.17 (m, 1H), 3.98- 4.03 (m, 2H), 3.53- 3.55 (m, 1H), 3.37- 3.47 (m, 2H), 2.37- 2.39 (m, 2H), 2.11- 2.20 (m, 6H), 2.04 (s, 3H), 2.02 (s, 3H), 1.93- 1.94 (m, 1H), 1.59- 1.67 (m, 1H), 1.42- 1.45 (m, 1H), 1.28- 1.30 (m, 1H); ^{13}C NMR (CDCl_3) δ 171.1, 170.5, 138.5, 134.5, 128.3, 127.8, 127.6, 117.4, 81.5, 80.4, 77.7, 72.5, 70.4, 66.6, 35.4, 34.1, 32.7, 28.2, 27.2, 21.3, 21.0. For **3.113a/b**: ^1H NMR (CDCl_3) δ 7.30- 7.37 (m, 5H), 5.86- 5.95 (m, 2H), 5.72- 5.74 (m, 1H), 5.08- 5.15 (m, 2H), 4.67 (s, 2H), 4.27- 4.30 (m, 1H), 4.14- 4.19 (m, 2H), 3.99- 4.04 (m, 2H), 3.59- 3.65 (m, 2H), 2.40- 2.41 (m, 2H), 2.23- 2.25 (m, 1H), 1.97- 2.04 (m, 3H), 1.65- 1.68 (m, 2H); ^{13}C NMR (CDCl_3) δ 171.2, 138.8, 134.8, 128.3, 127.8, 127.6, 126.9, 126.0, 117.1, 81.6, 80.8, 75.2, 72.6, 66.7, 35.6, 28.3, 26.7, 26.5, 21.0.

Synthesis of 3.114



Compound **3.114** was prepared following the procedure used for **3.75**. ^1H NMR (CDCl_3) δ 7.30- 7.37 (m, 5H), 5.82- 5.92 (m, 2H), 4.98- 5.15 (m, 4H), 4.60- 4.68 (m, 2H), 3.42- 3.52 (m, 3H), 3.17- 3.21 (m, 1H), 2.16- 2.48 (m, 5H), 1.93- 1.95 (m, 1H), 1.45- 1.66 (m, 6H), 1.33- 1.35 (m, 1H); ^{13}C NMR (CDCl_3) δ 138.6, 135.1, 128.4, 127.9, 127.7, 117.0, 114.7, 81.2, 80.7, 79.1, 73.6, 72.7, 34.8, 31.9, 29.7, 27.2, 26.6, 23.0.

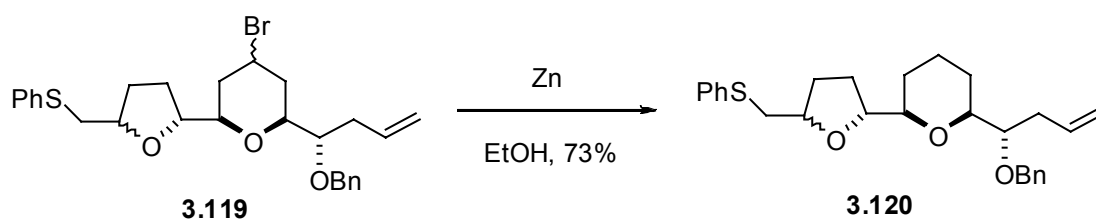
Synthesis of 3.119



To a solution of **3.89** (80 mg, 0.142 mmol) in DMF (5 mL) were added in PhSH (17.2 mg, 0.156 mmol) and K_2CO_3 (21.5 mg, 0.156 mmol). The reaction mixture was kept at rt overnight. The solvent was removed under reduced pressure. The residue was subjected to column chromatography to give the products **3.119** (51 mg, 66 %). ^1H NMR (CDCl_3) δ 7.30- 7.40 (m, 10H), 5.83- 5.87 (m, 1H), 5.08- 5.15 (m, 2H), 4.83- 4.84 (m, 1H), 4.62- 4.65 (m, 2H), 3.96- 4.28 (m, 3H), 3.82- 3.85 (m, 1H), 3.50- 3.52 (m, 1H), 3.20- 3.30 (m,

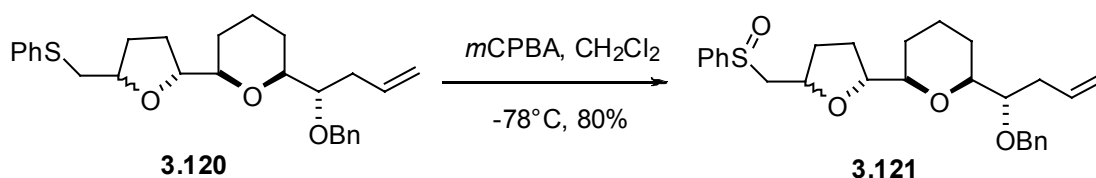
2H), 2.97- 2.99 (m, 1H), 2.10- 2.50 (m, 3H), 1.86- 2.05 (m, 7H); ^{13}C NMR (CDCl_3) δ 135.0, 129.3, 129.1, 128.9, 128.8, 128.3, 128.3, 127.8, 127.8, 127.6, 126.0, 125.8, 117.2, 117.1, 81.3, 80.8, 80.1, 78.8, 78.5, 74.8, 74.5, 73.6, 73.1, 72.4, 72.3, 51.5, 51.3, 38.9, 38.6, 36.0, 35.9, 34.7, 34.6, 31.4, 30.8, 27.6, 27.4.

Synthesis of 3.120



Compound **3.120** was prepared following the procedure used for **3.75**. ^1H NMR (C_6D_6) δ 7.02- 7.53 (m, 10H), 6.07- 6.10 (m, 1H), 5.17- 5.29 (m, 2H), 4.64- 4.73 (m, 2H), 4.18- 4.40 (m, 1H), 3.89- 4.06 (m, 1H), 3.50- 3.53 (m, 2H), 3.18- 3.40 (m, 2H), 2.86- 3.06 (m, 1H), 2.50- 2.68 (m, 1H), 2.38- 2.48 (m, 1H), 1.73- 1.85 (m, 3H), 1.32- 1.65 (m, 7H); ^{13}C NMR (C_6D_6) δ 139.7, 137.5, 136.0, 129.0, 127.3, 125.6, 125.5, 116.5, 116.4, 81.8, 81.3, 81.2, 80.3, 79.8, 79.5, 79.2, 78.3, 78.2, 72.6, 72.5, 38.8, 38.6, 35.2, 31.6, 30.8, 27.5, 27.2, 27.0, 26.3, 23.4. HRMS (ESI) calcd for $\text{C}_{27}\text{H}_{34}\text{O}_3\text{S}$ ($\text{M} + \text{H}^+$) 439.2301, found 349.2290.

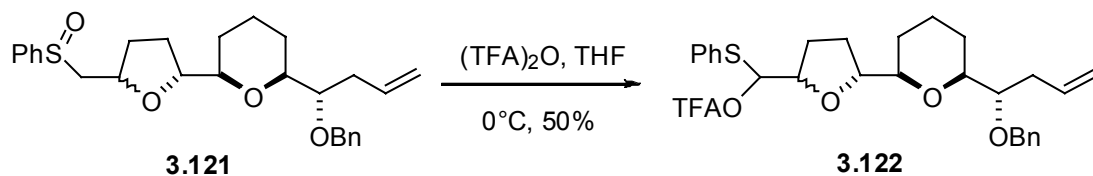
Synthesis of 3.121



To a solution of **3.120** (32 mg, 0.068 mmol) in CH_2Cl_2 (5 mL) was added in *m*CPBA (15.3 mg, 0.068 mmol, 77 %) at -78°C . The reaction mixture was stirring for 10 min at

-78 °C before it was quenched by saturated aqueous Na₂S₂O₃. Then it was extracted with CH₂Cl₂, and dried (Na₂SO₄). The solvent was removed under reduced pressure and the residue subjected to column chromatography to give product **3.121** (27 mg, 80 %). ¹H NMR (C₆D₆) δ 7.03- 7.77 (m, 10H), 5.92- 6.19 (m, 1H), 5.05- 5.18 (m, 2H), 4.54- 4.81 (m, 2H), 3.78- 4.21 (m, 2H), 3.40- 3.64 (m, 2H), 3.02- 3.38 (m, 1H), 2.68- 3.00 (m, 1H), 2.28- 2.62 (m, 2H), 1.30- 2.00 (m, 9H); ¹³C NMR (C₆D₆) δ 135.9, 135.8, 135.7, 135.7, 130.1, 130.0, 130.0, 129.9, 128.9, 128.9, 128.8, 128.8, 128.8, 128.3, 128.3, 127.4, 127.3, 127.3, 124.4, 124.2, 123.7, 123.6, 116.6, 116.5, 116.5, 81.7, 81.5, 81.2 (3), 81.1 (2), 81.0, 79.7 (2), 79.6 (2), 79.5, 79.4, 79.1, 75.3, 75.0, 74.3, 74.2, 74.0, 73.7, 73.6, 72.7, 72.6, 72.4 (2), 65.0, 64.8, 62.8, 62.7, 35.2 (2), 35.1 (2), 32.4, 1.8, 31.4, 31.0, 27.4 (2), 27.3, 27.2, 27.1, 27.0.

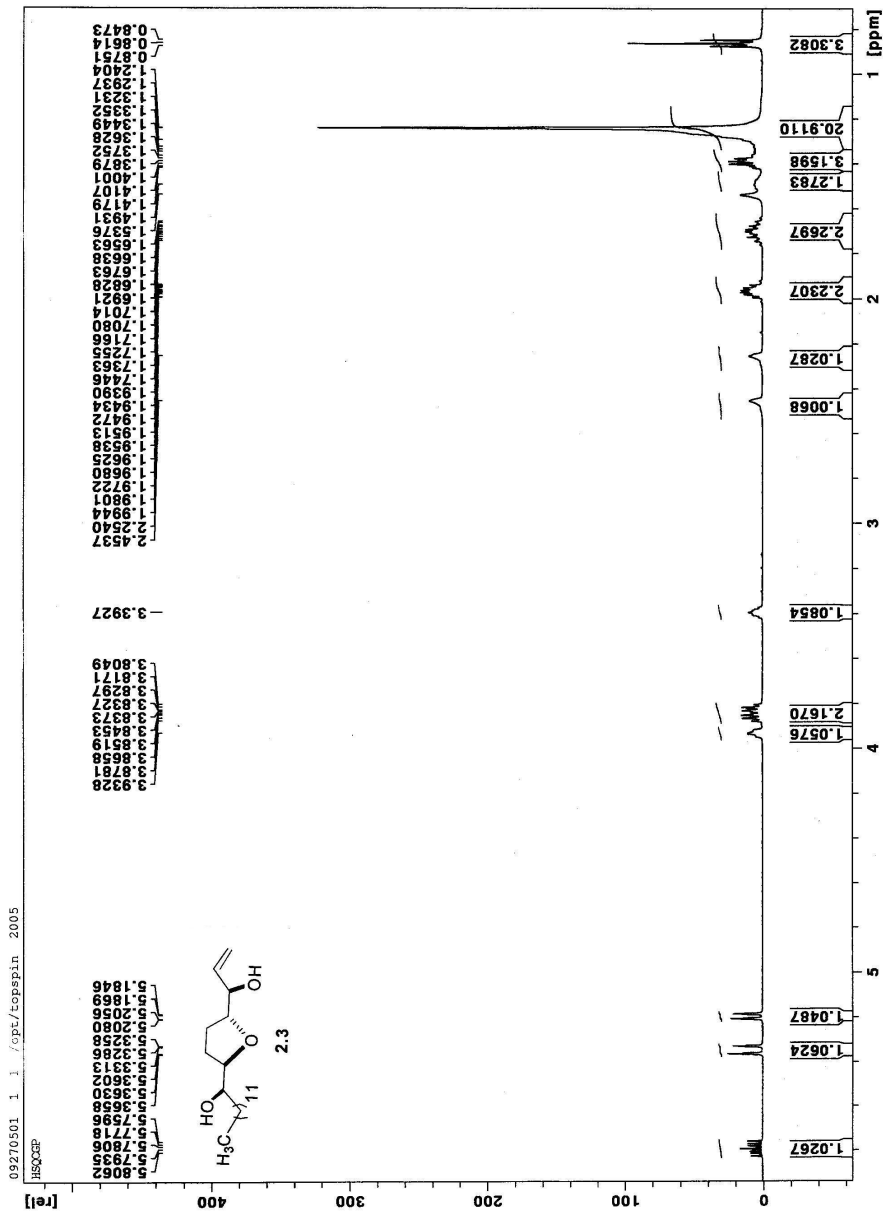
Synthesis of **3.122**

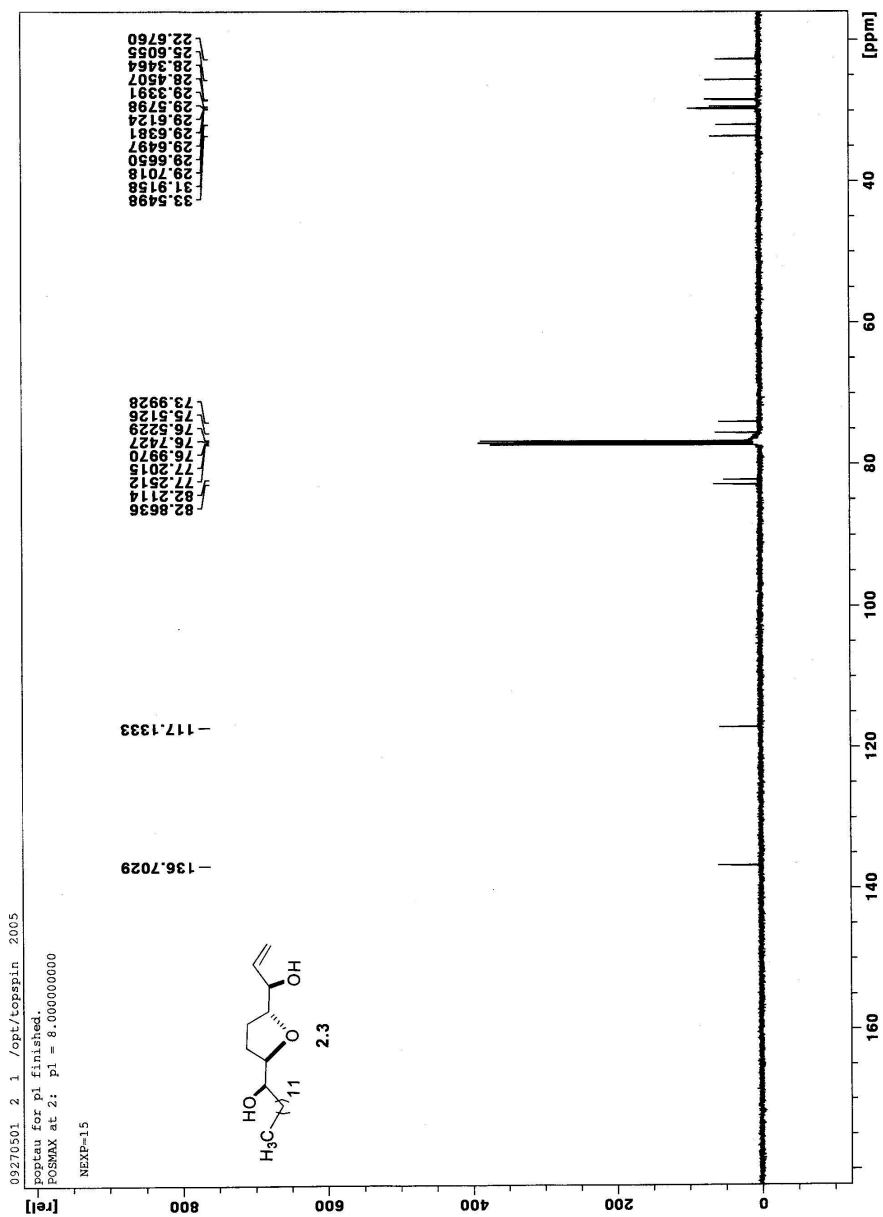


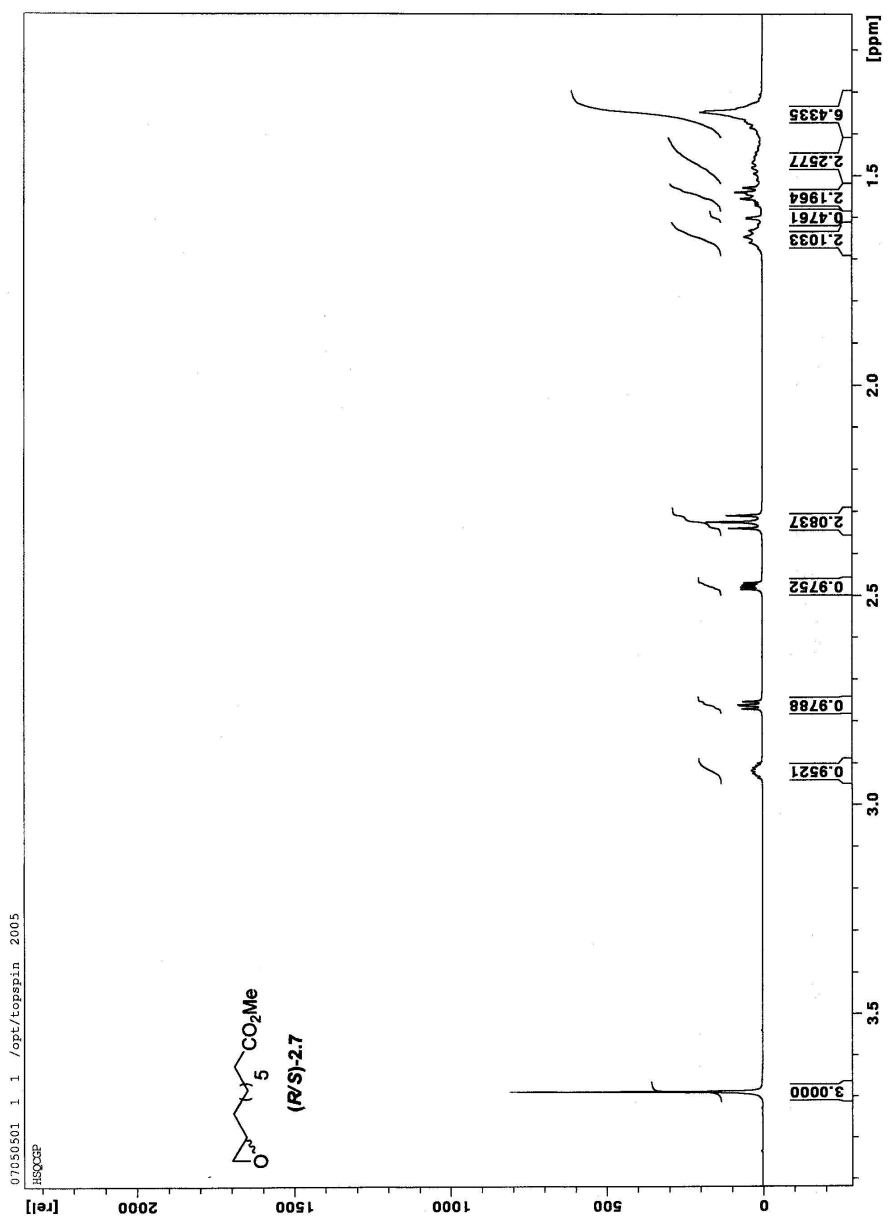
To a solution of **3.121** (3 mg, 0.006 mmol) in THF (3 mL) was added in (TFA)₂O (25 μL) at 0 °C. The reaction mixture was kept stirring at 0 °C for 10 min before it was quenched by saturated aqueous NaHCO₃. Then it was extracted with CH₂Cl₂, and dried (Na₂SO₄). The solvent was removed under reduced pressure and the residue subjected to column chromatography to give product **3.122** (1.6 mg, 50 %). ¹H NMR (C₆D₆) δ 7.00- 7.70 (m, 10H), 6.68- 6.70 (m, 1H), 6.02- 6.04 (m, 1H), 5.17- 5.27 (m, 2H), 4.69 (s, 2H), 4.19- 4.24 (m, 1H), 3.74- 3.81 (m, 1H), 3.50- 3.53 (m, 1H), 3.38- 3.42 (m, 1H), 2.97- 3.01 (m, 1H), 2.40- 2.68 (m, 2H), 2.17- 2.22 (m, 1H), 1.75- 1.82 (m, 2H), 1.50- 1.64 (m, 2H), 1.21-

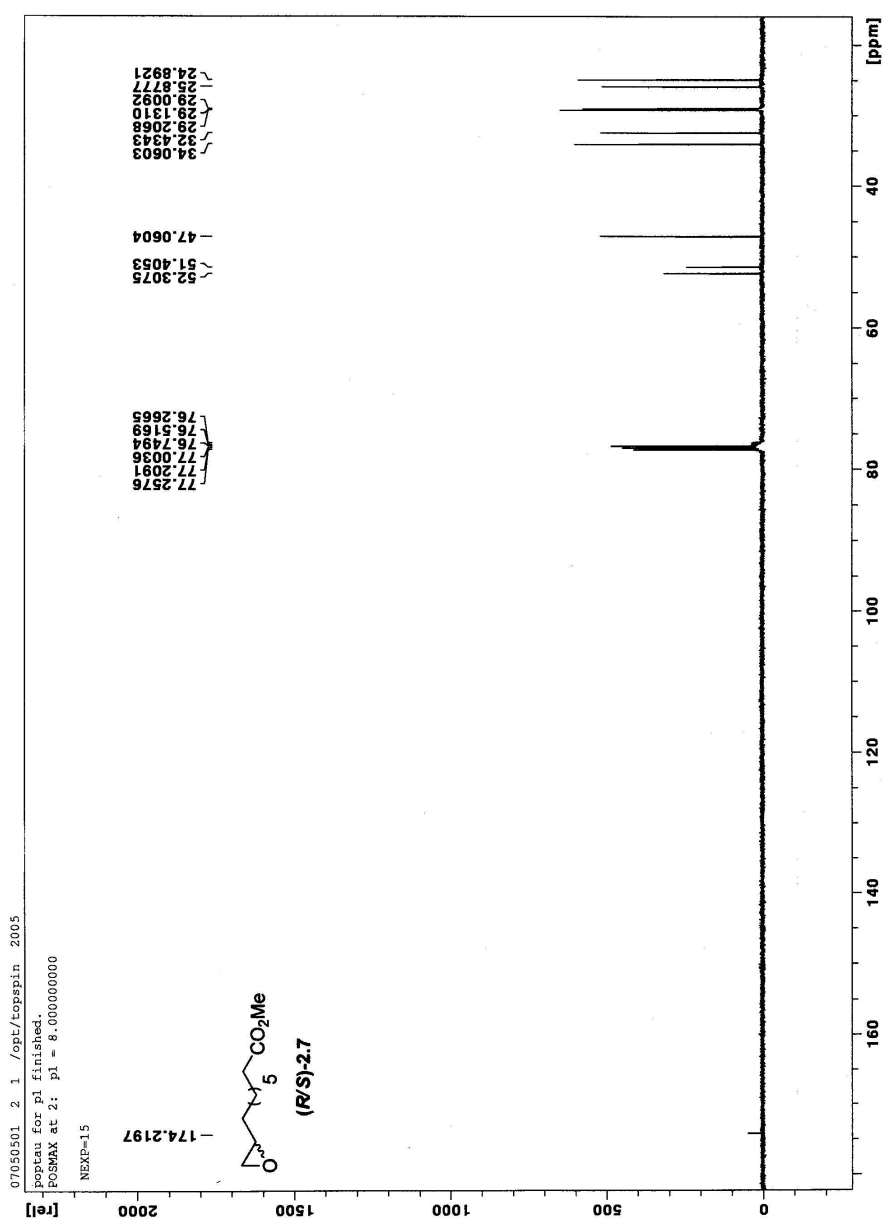
1.48 (m, 3H), 0.94- 1.08 (m, 2H); ^{13}C NMR (C_6D_6) δ 139.6, 135.6, 124.6, 132.8, 132.4, 129.2, 129.0, 128.9, 127.1, 116.7, 86.9, 82.4 (2), 81.3, 81.2, 81.1, 80.2, 79.6, 79.4, 77.5, 77.2, 77.0, 72.4, 53.1, 35.3, 35.2, 30.0, 29.0, 28.1, 27.3, 27.0, 26.4 (2), 23.4.

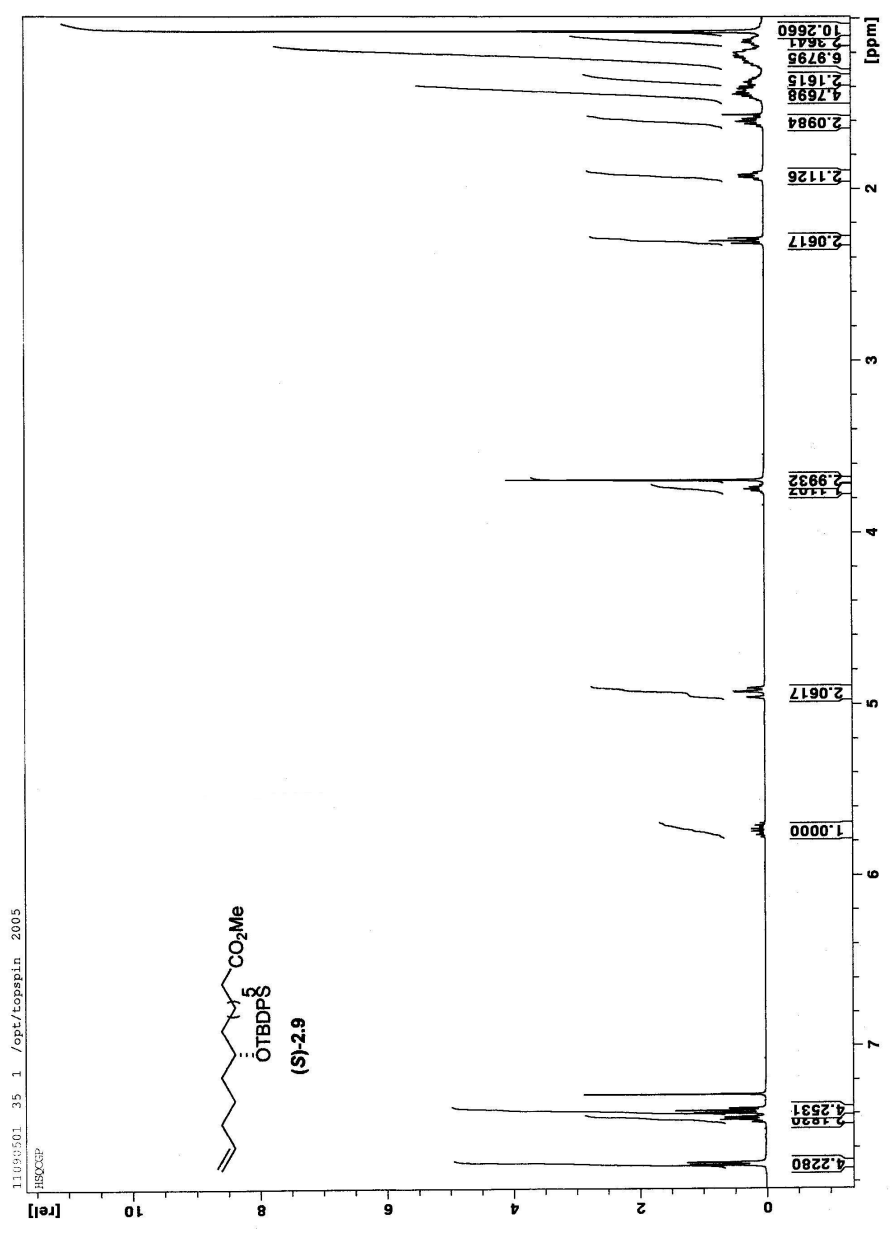
APPENDIX

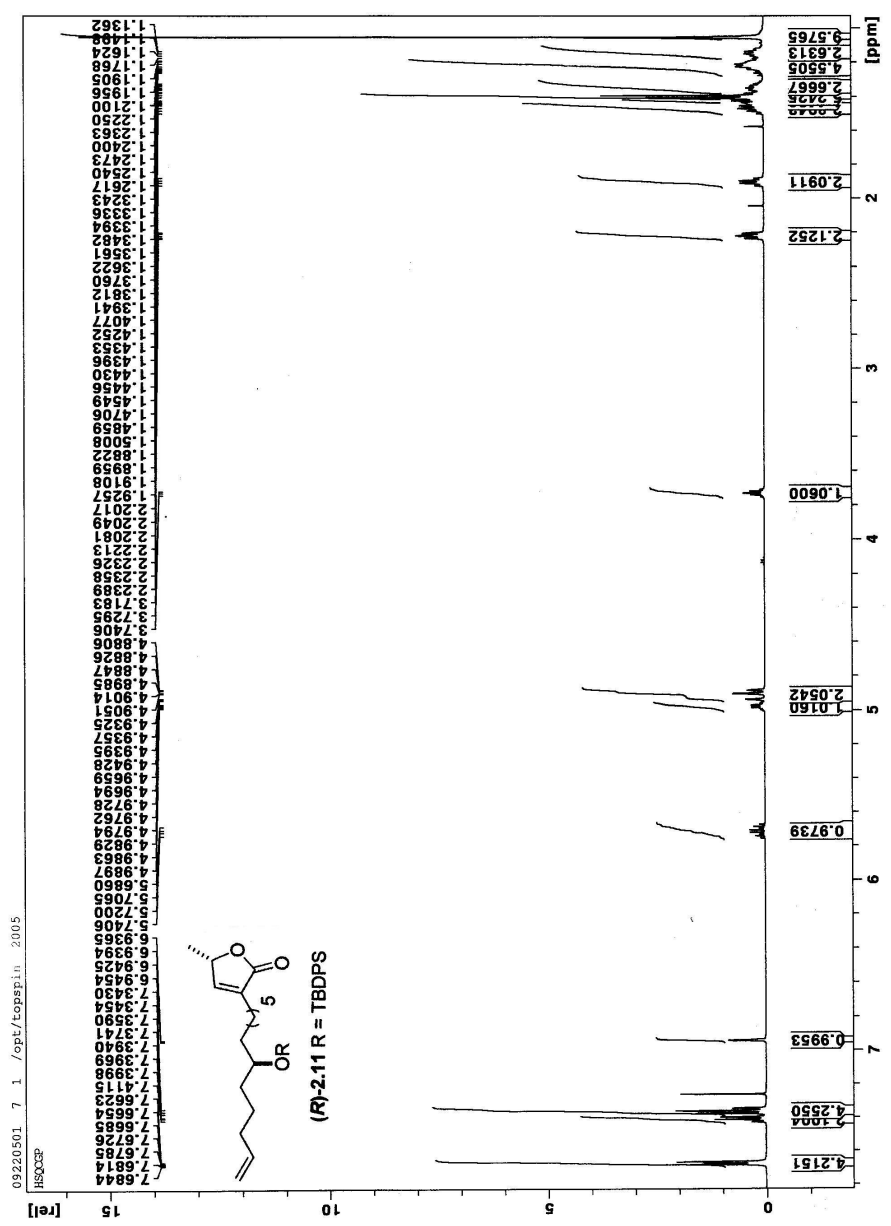


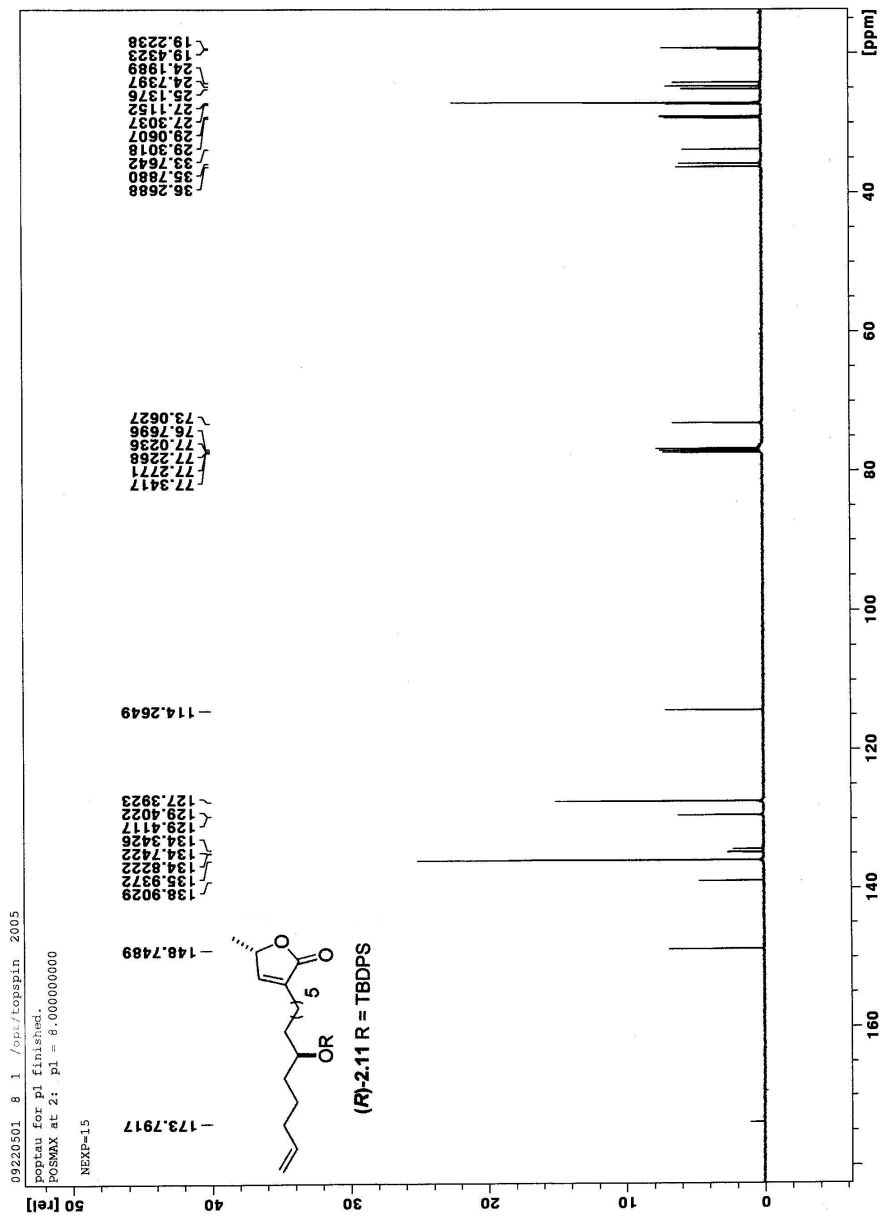


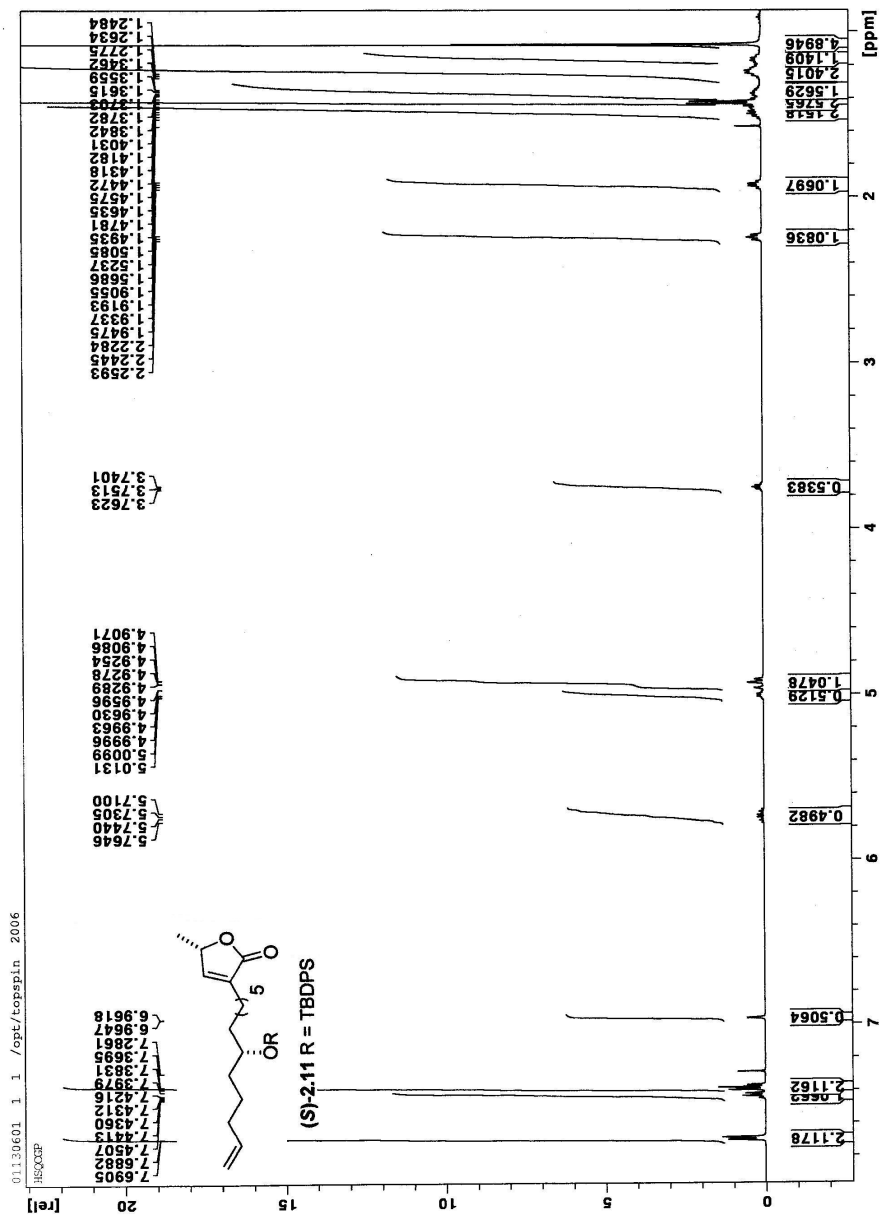


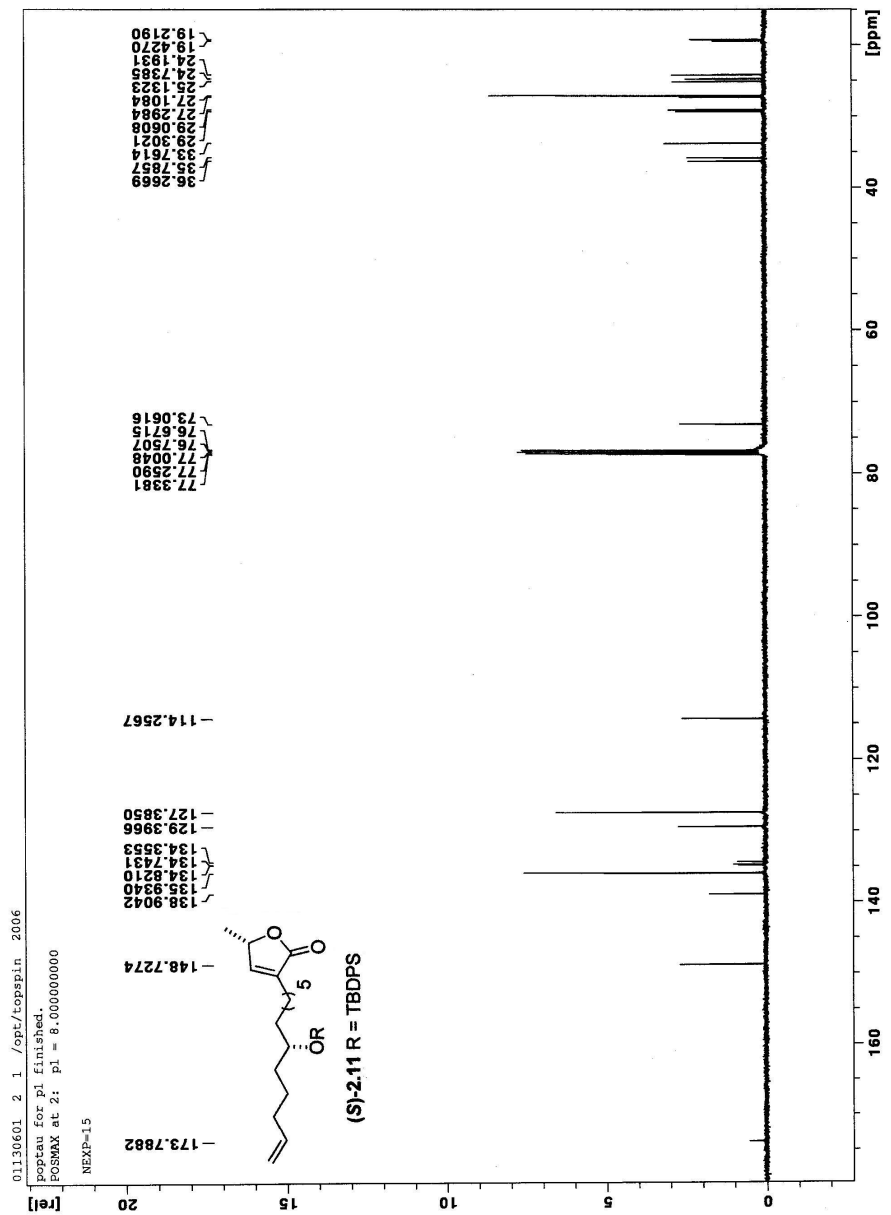


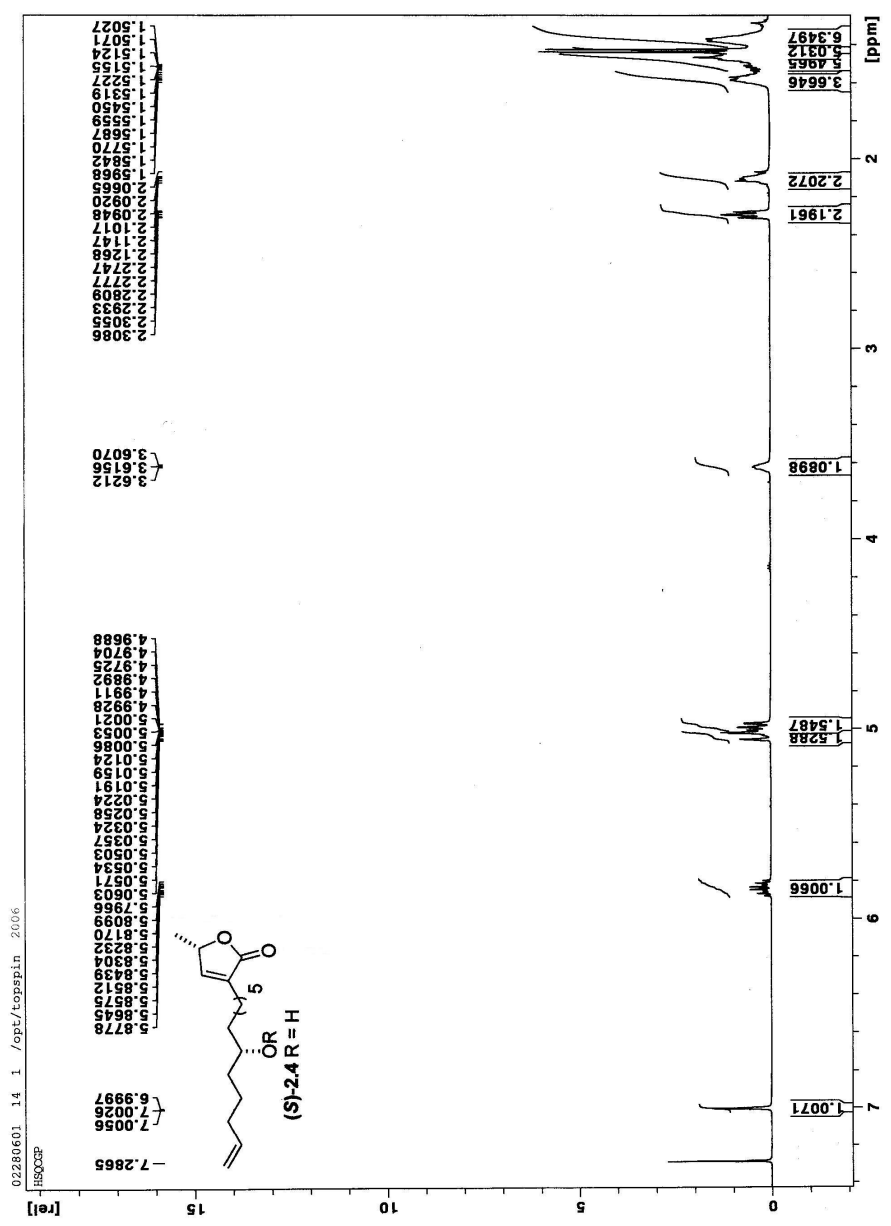


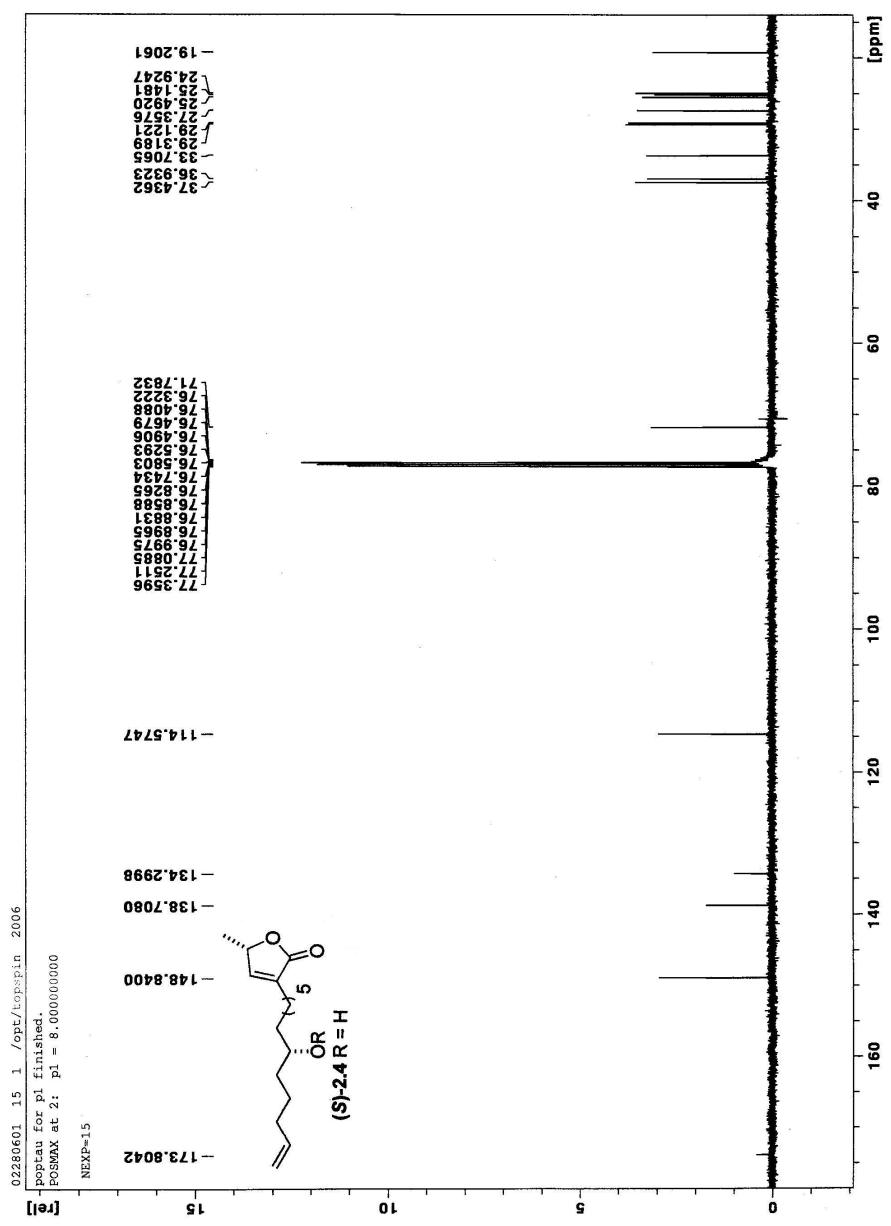


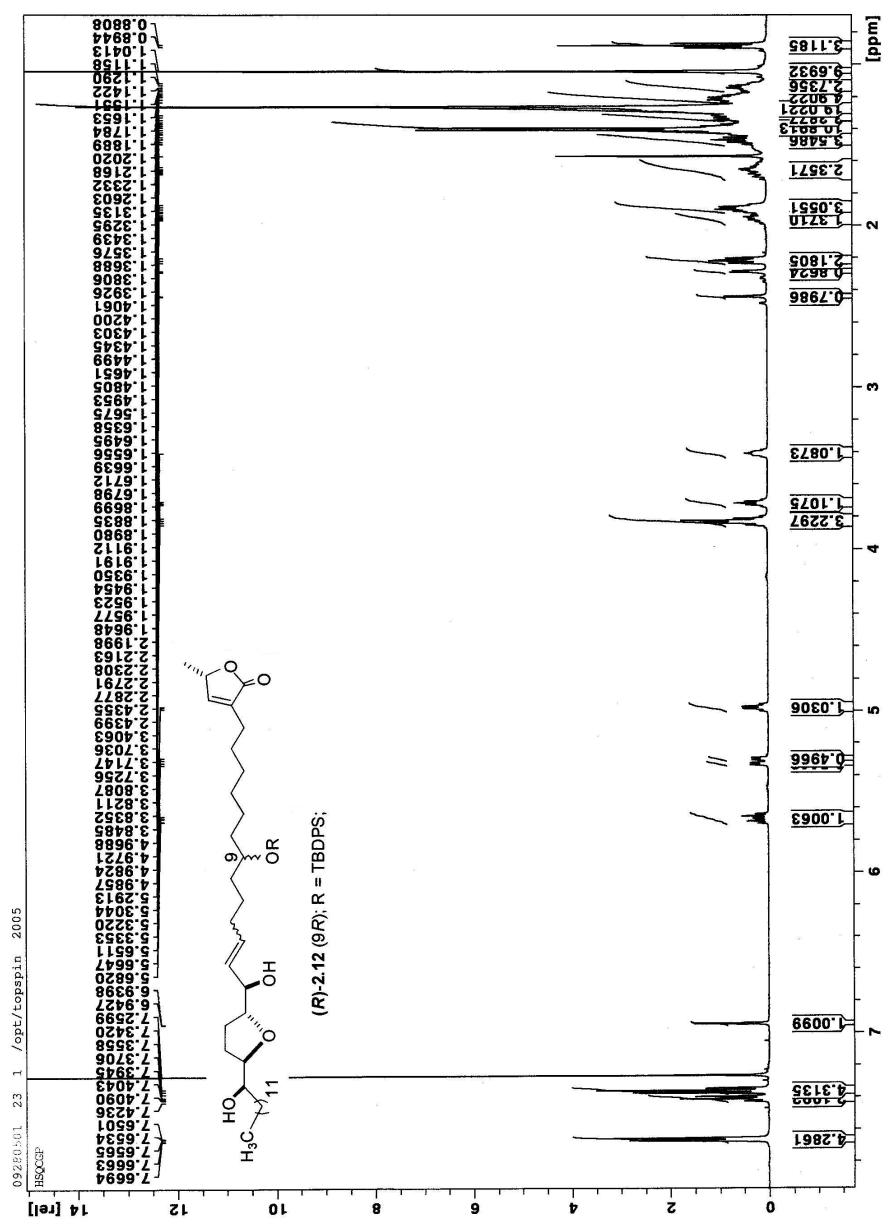


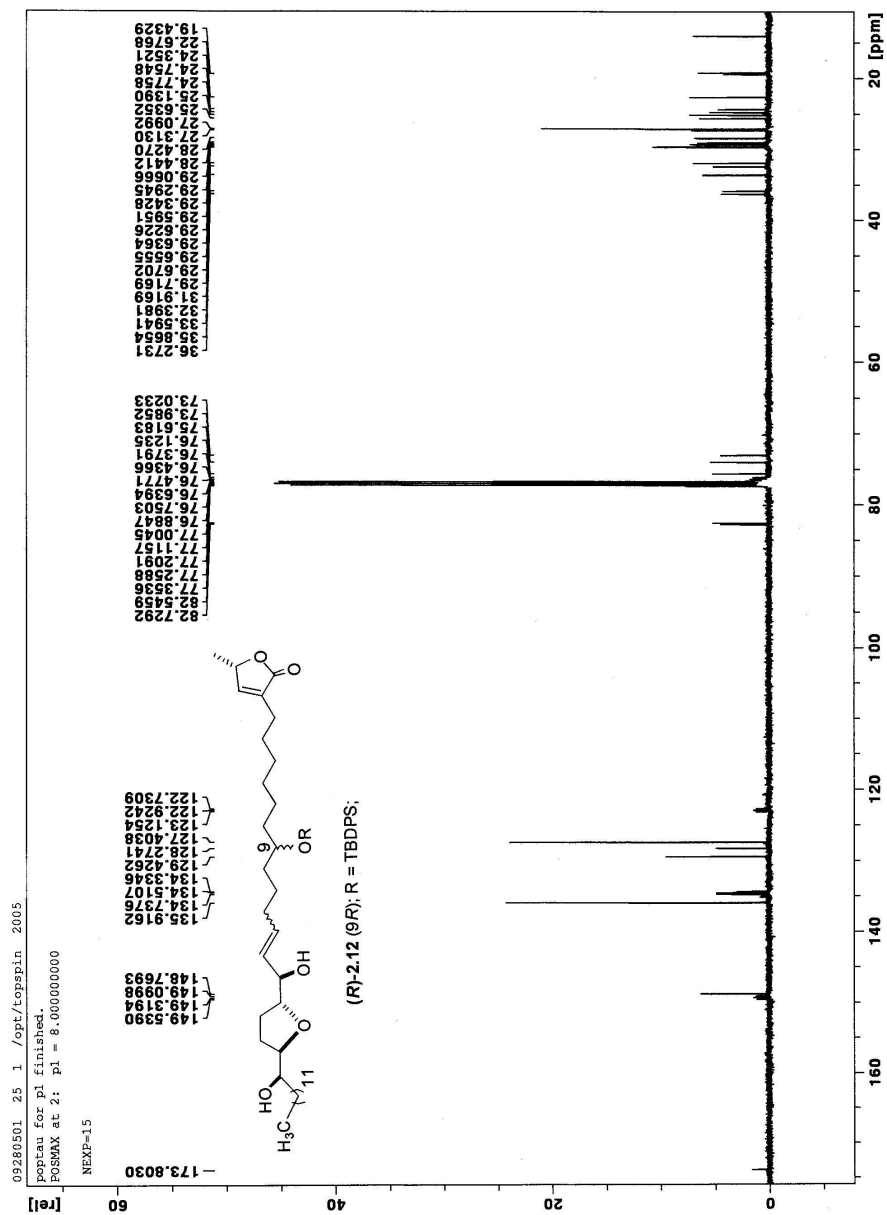


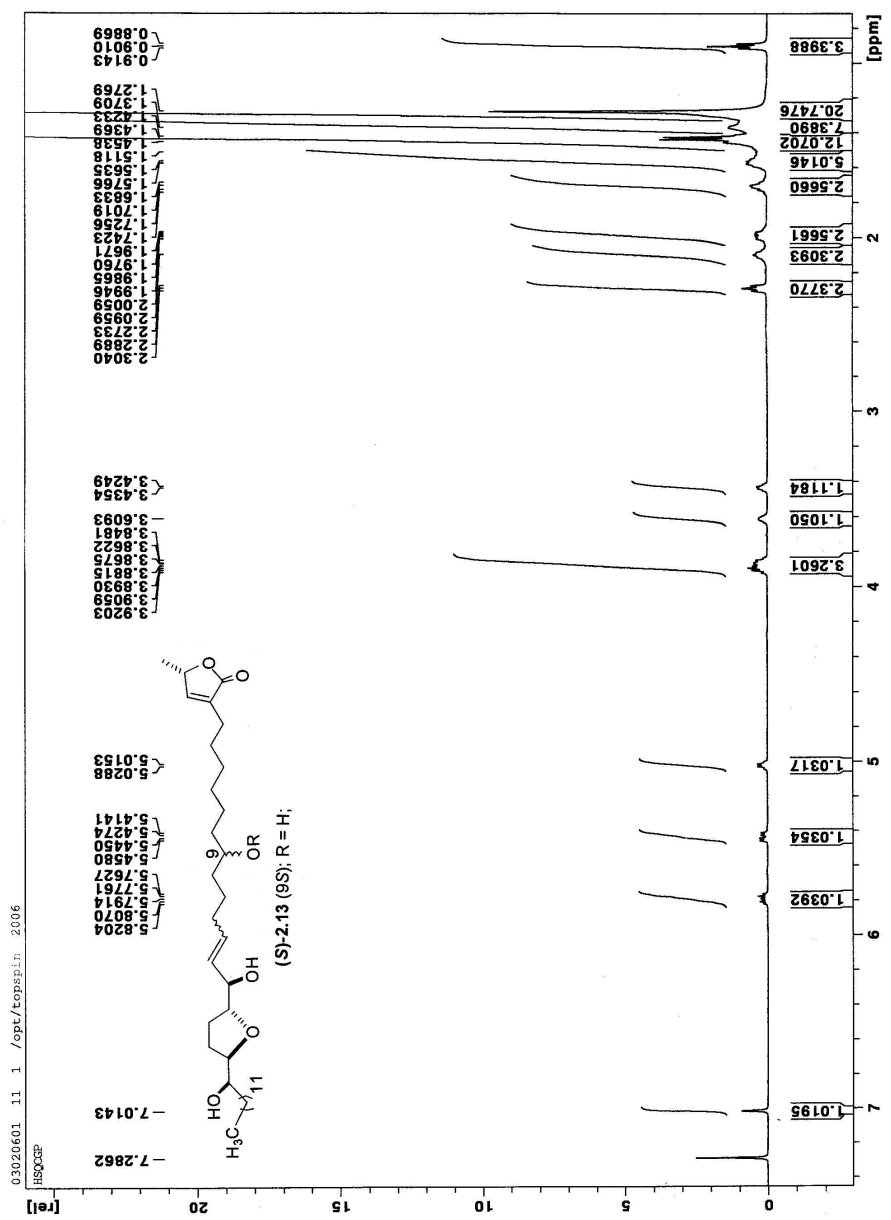


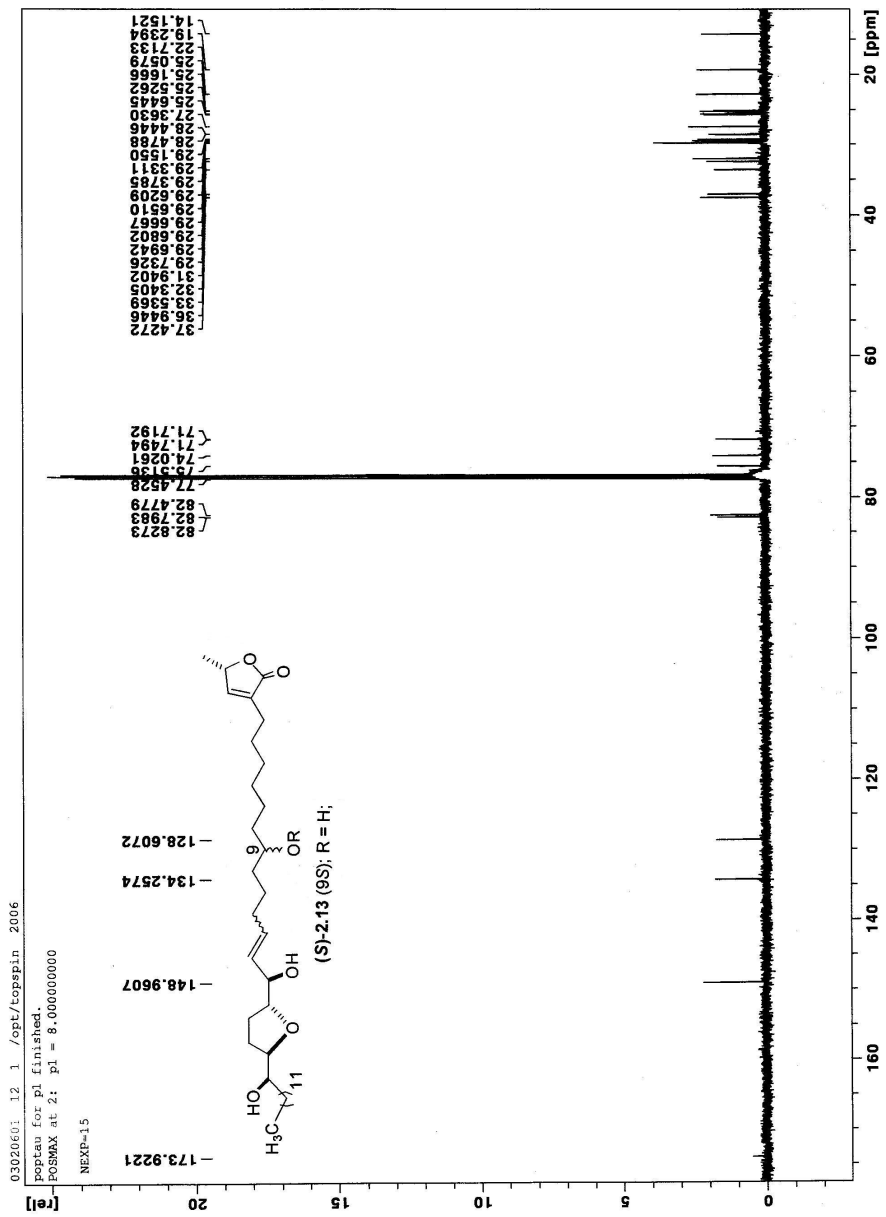


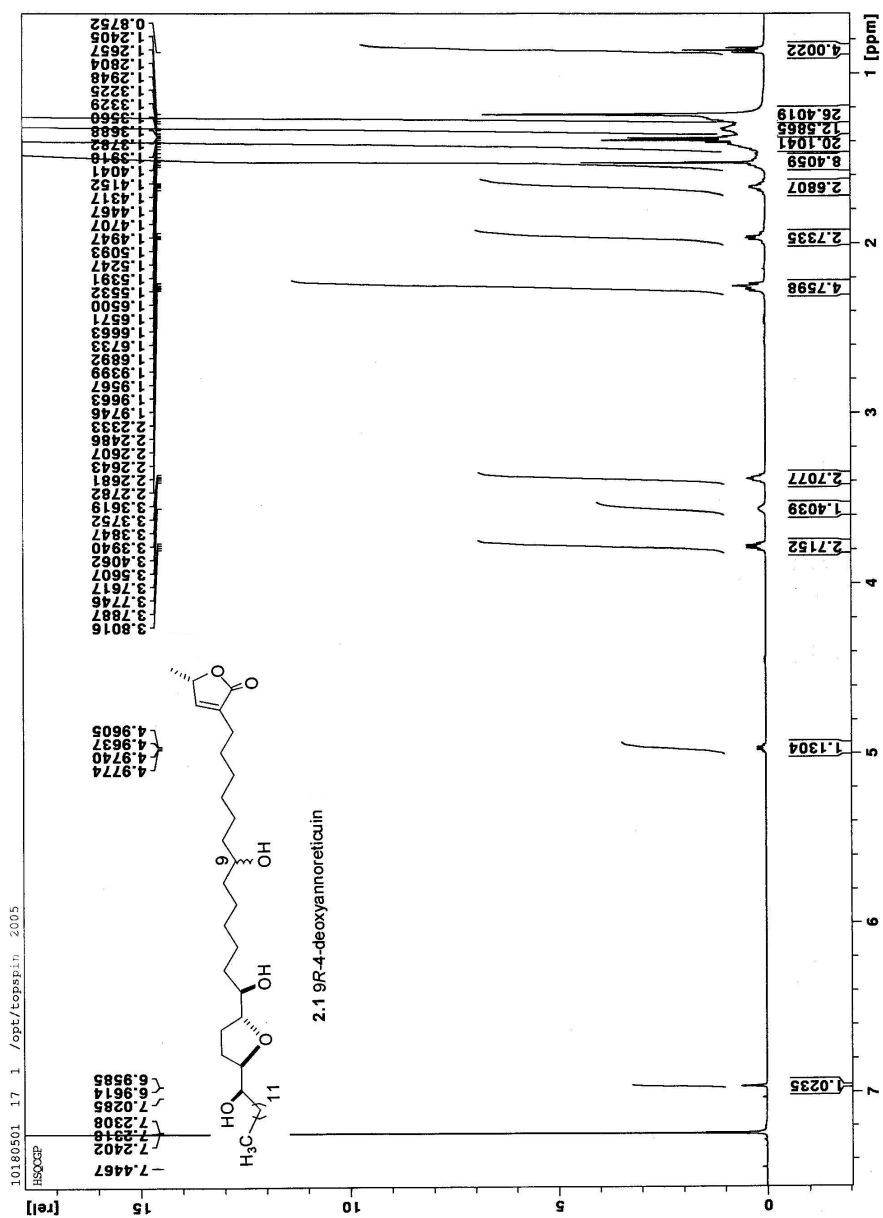


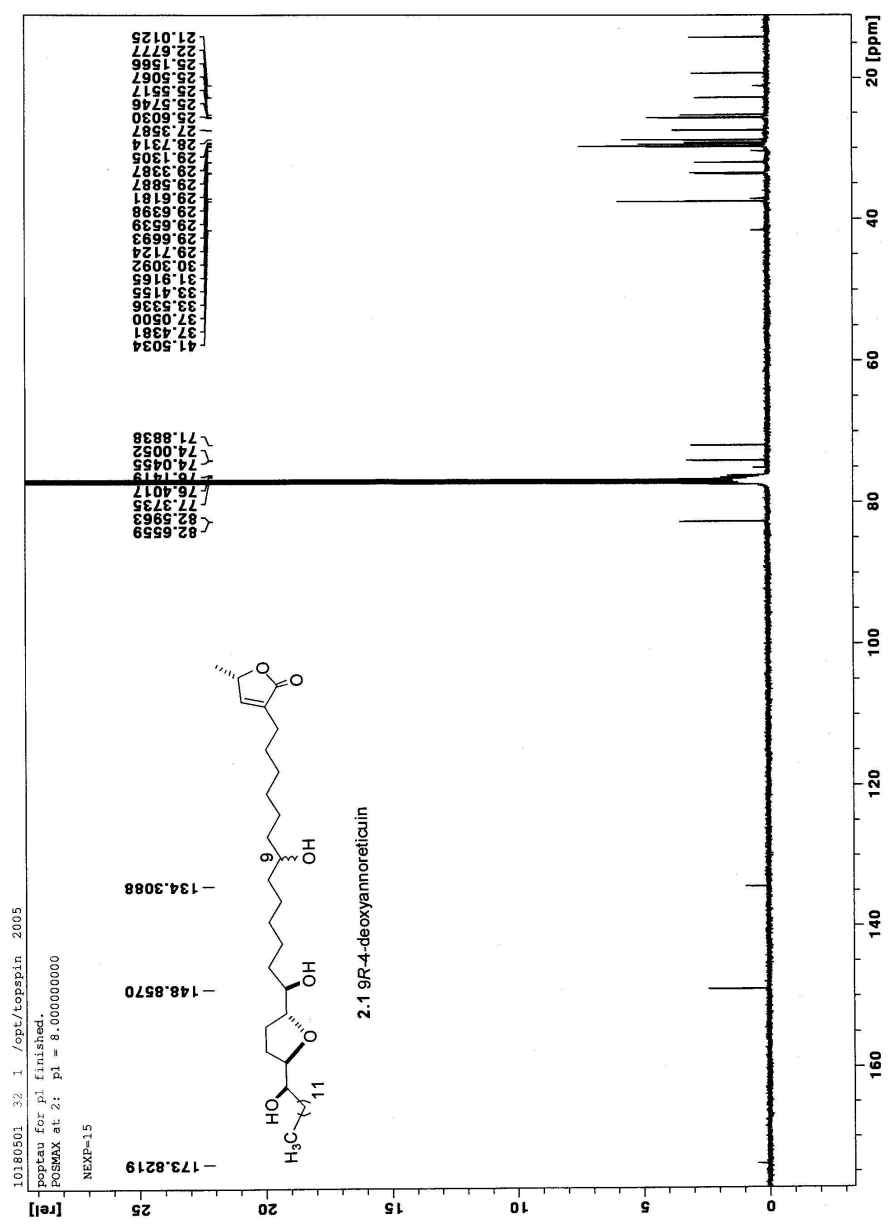


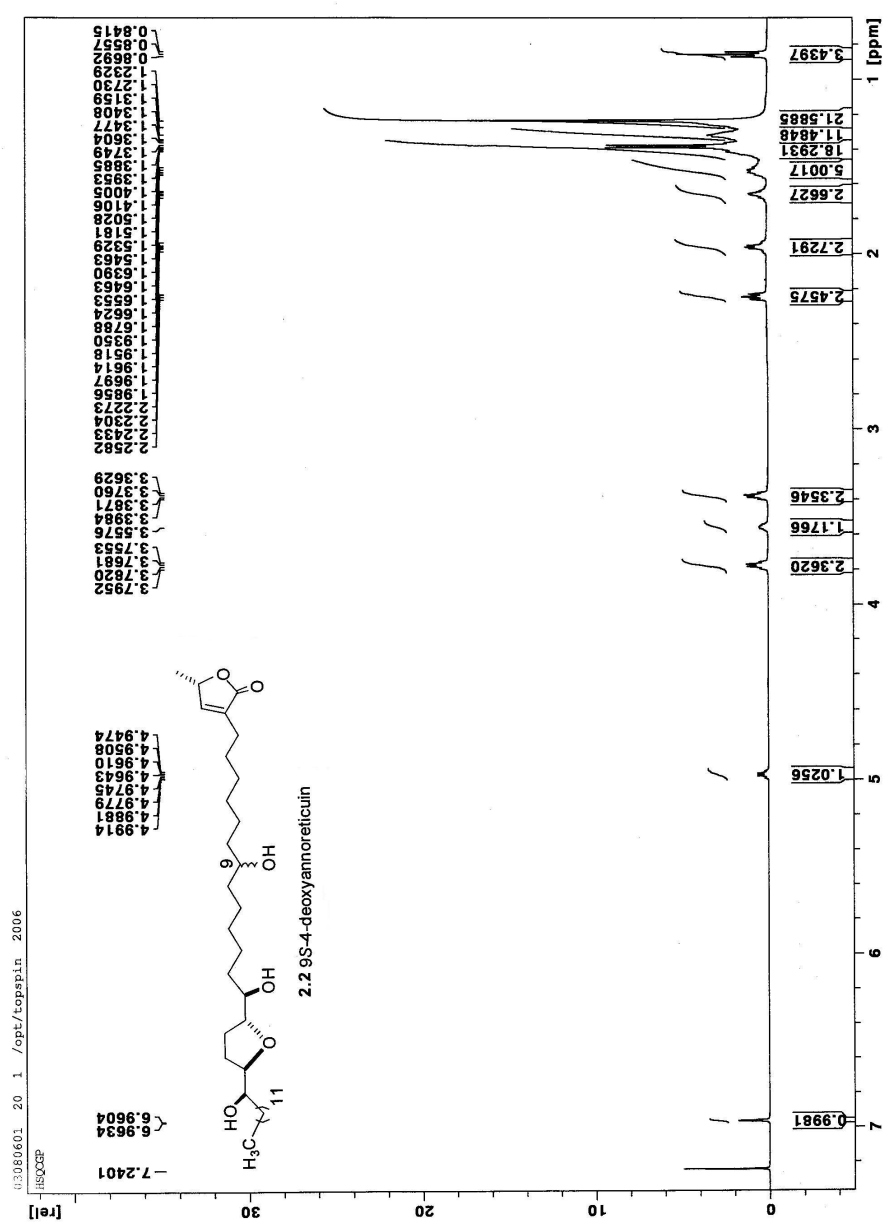


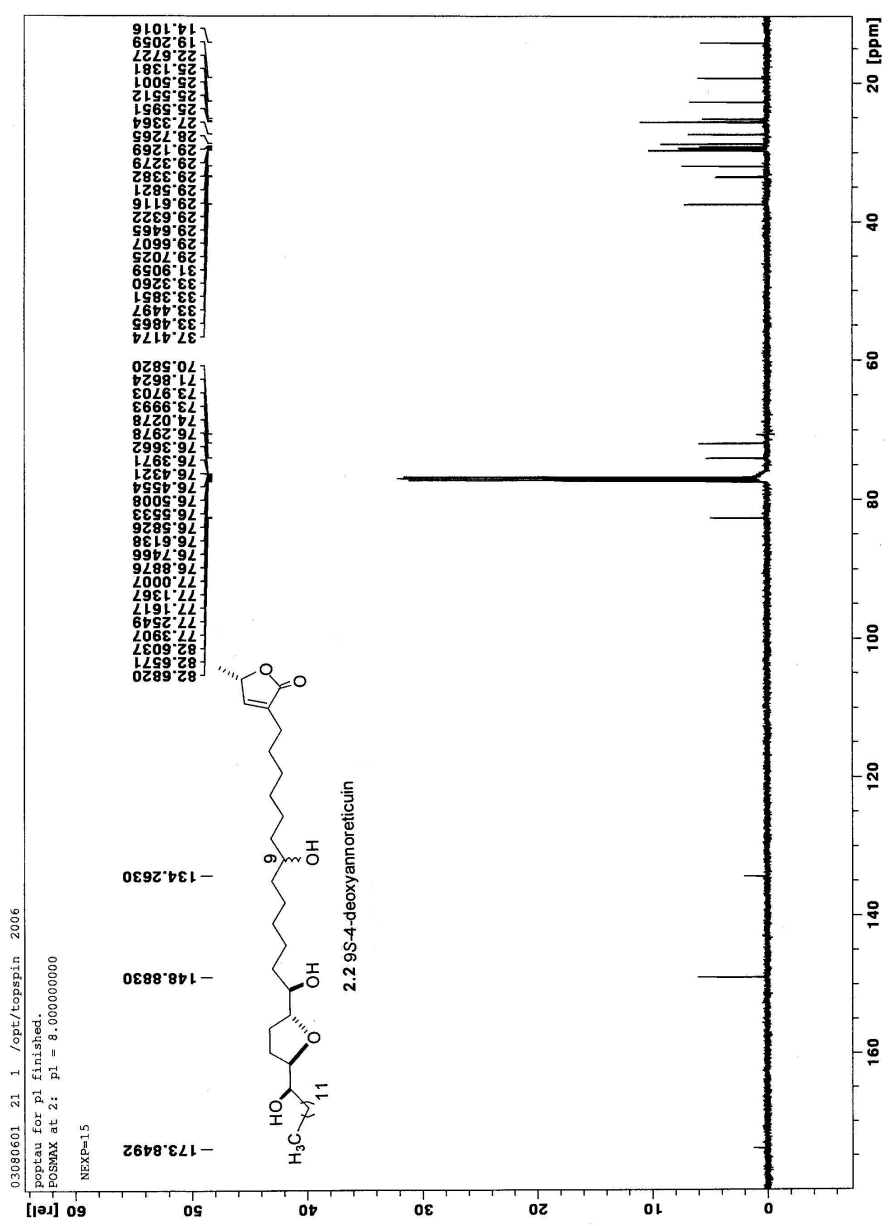


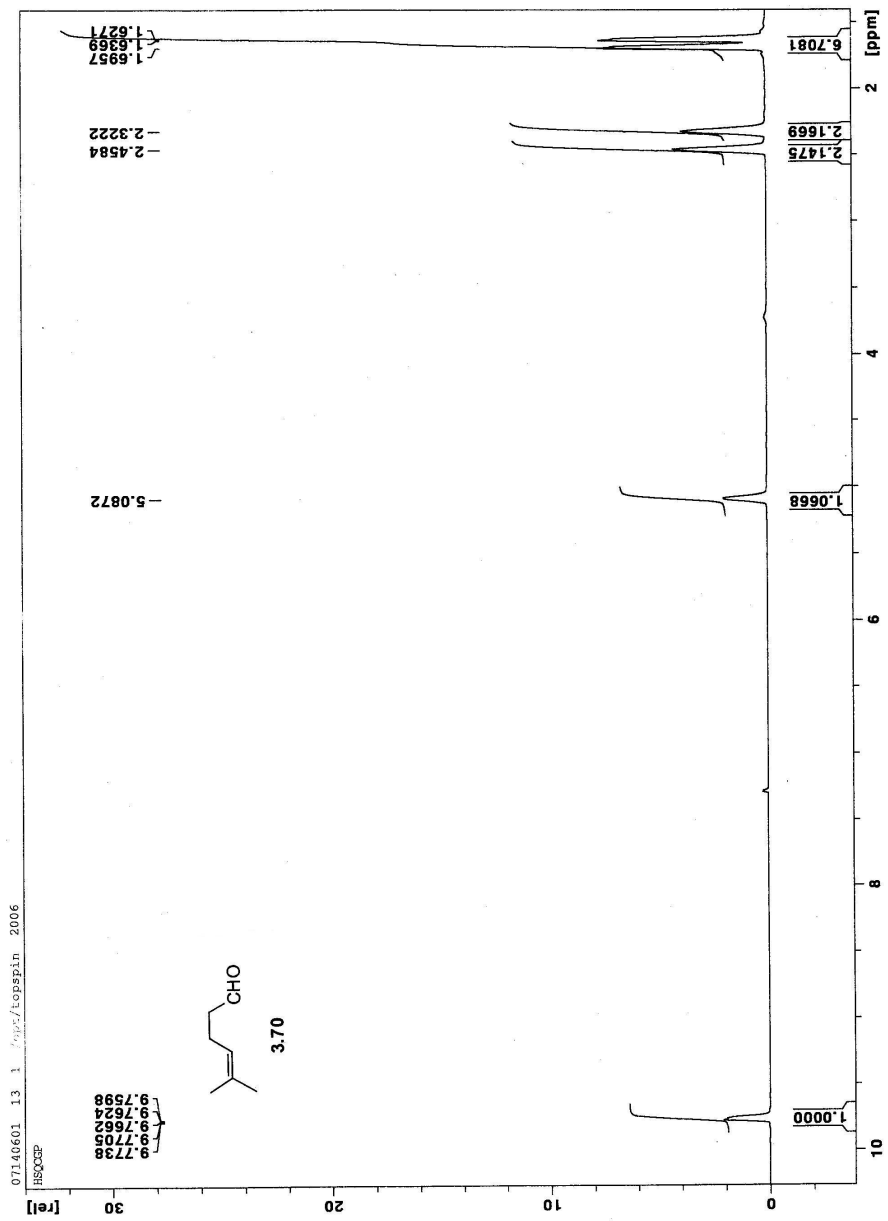


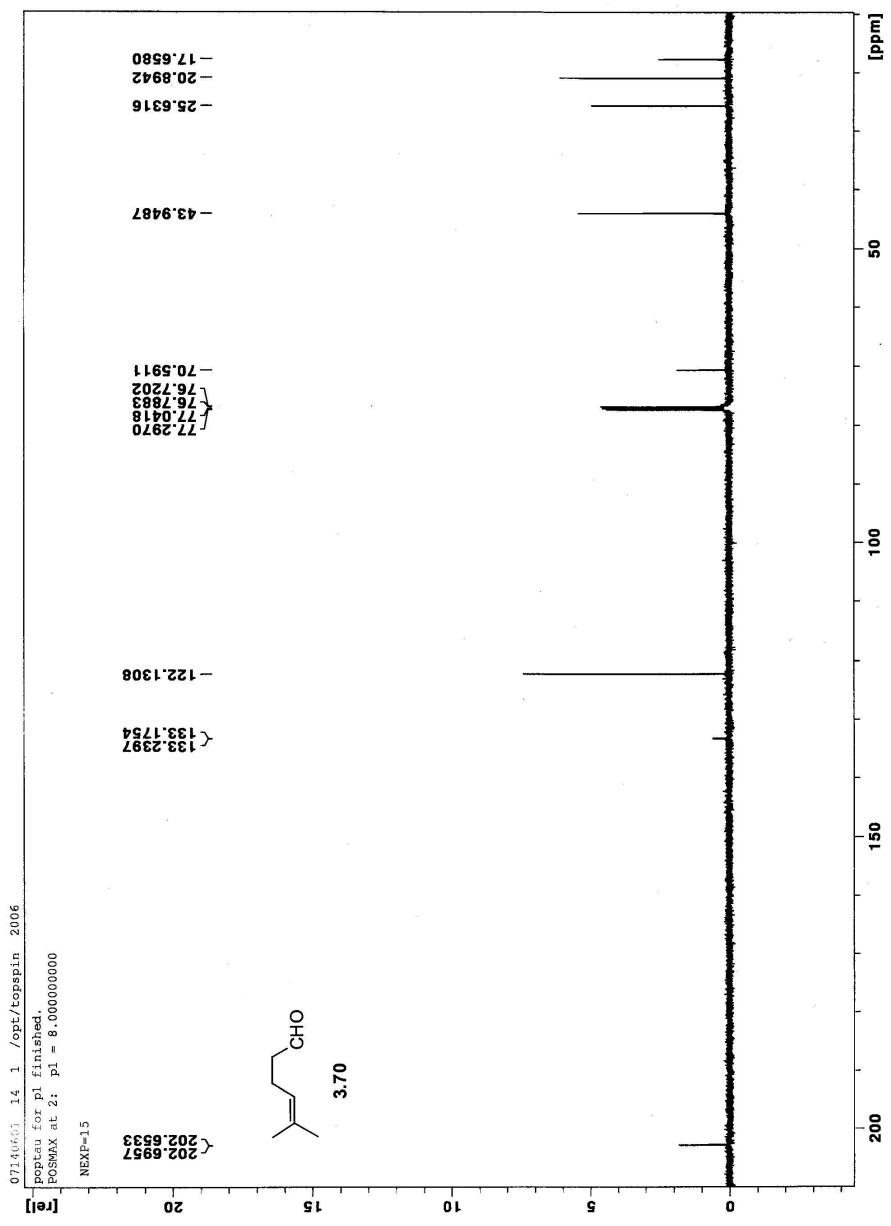


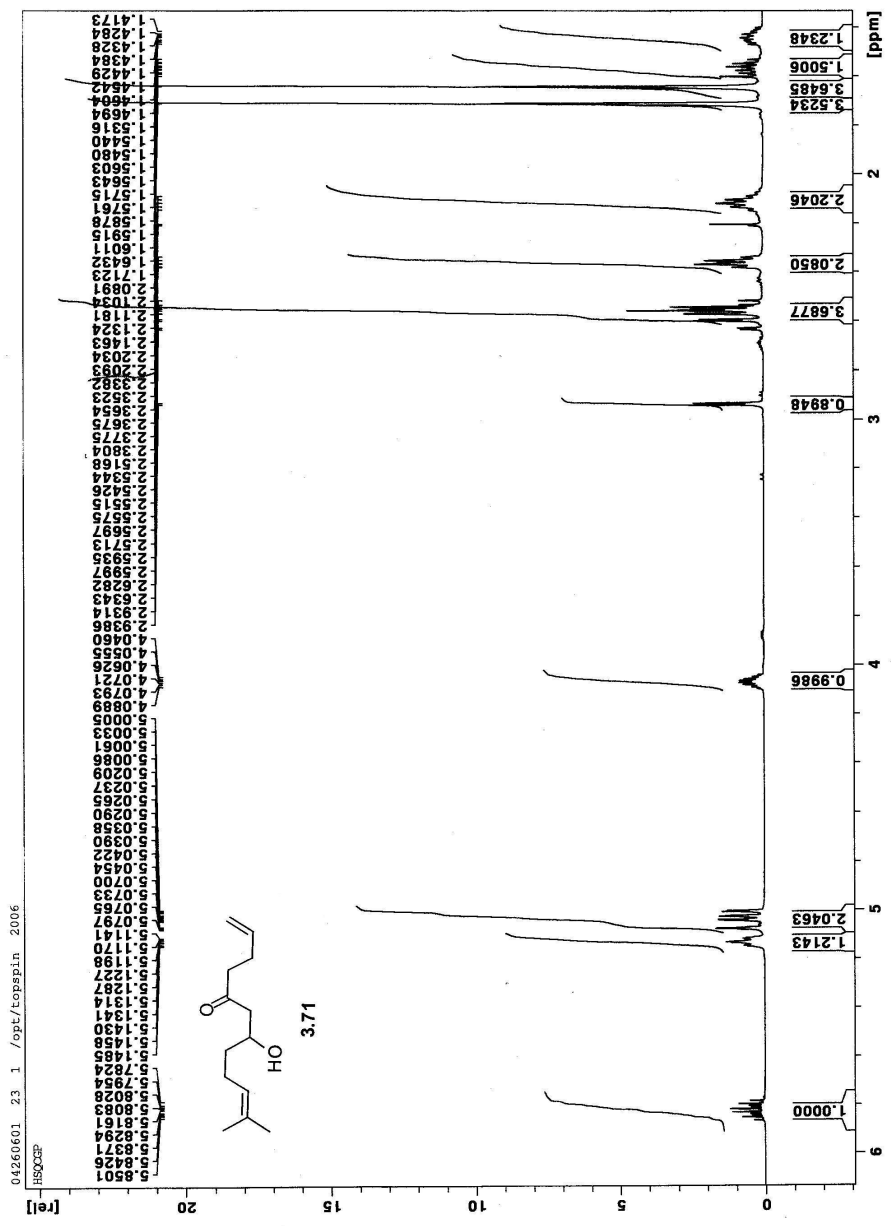


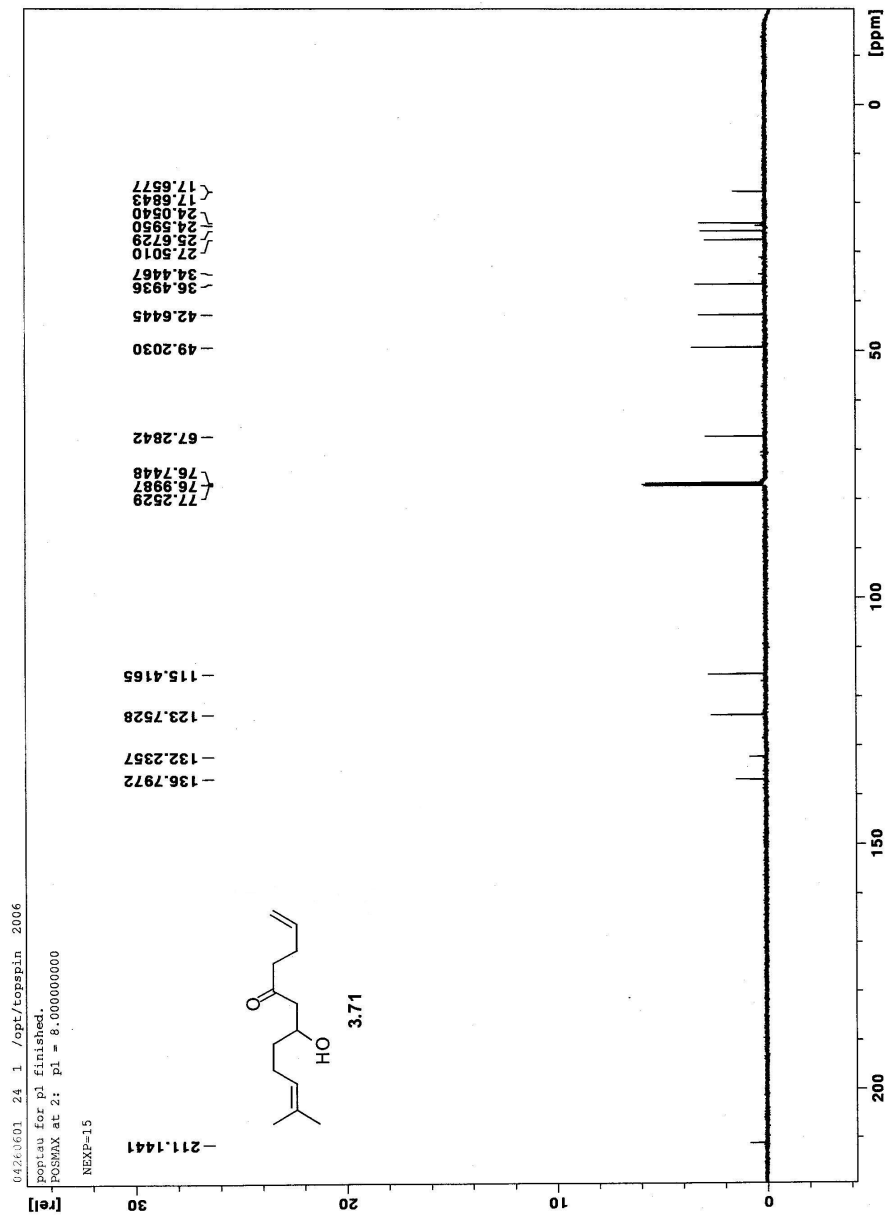


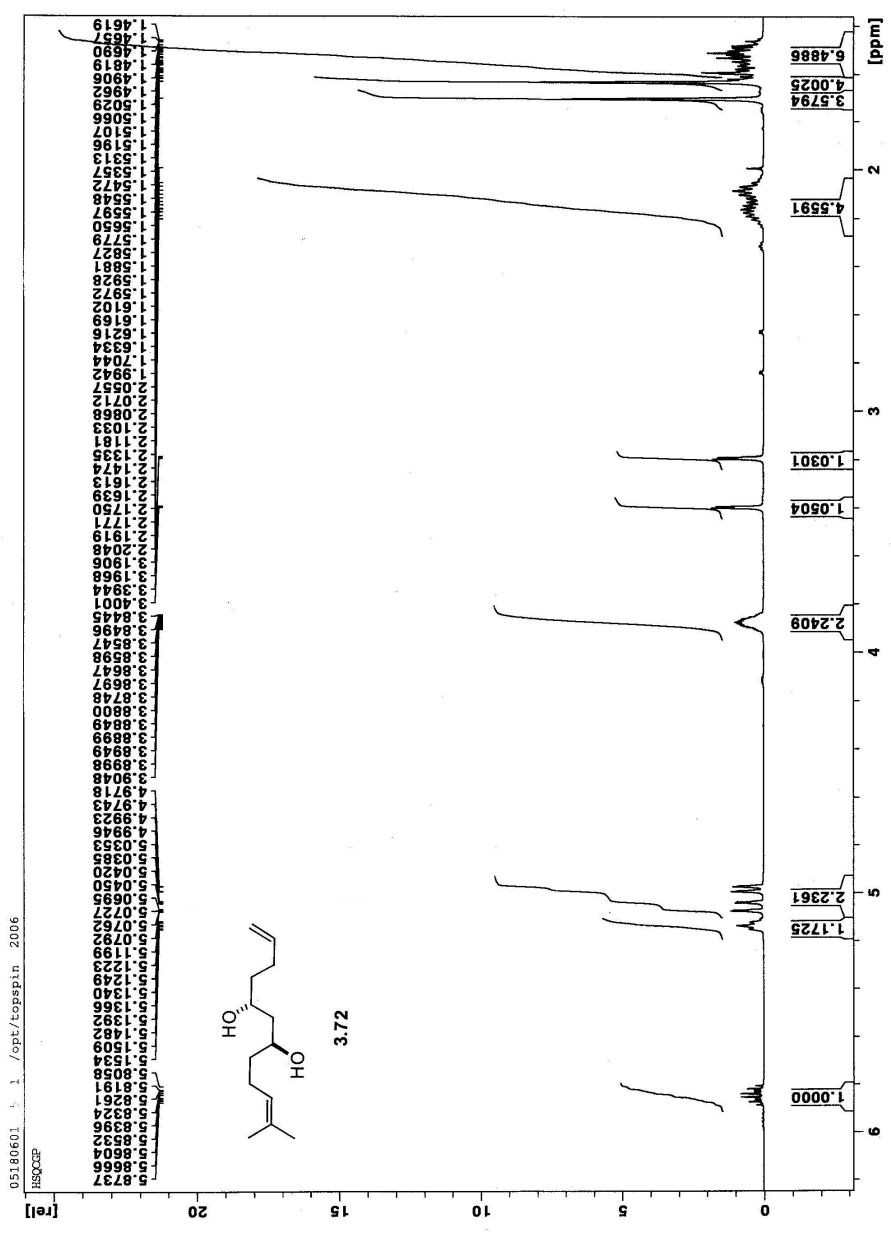


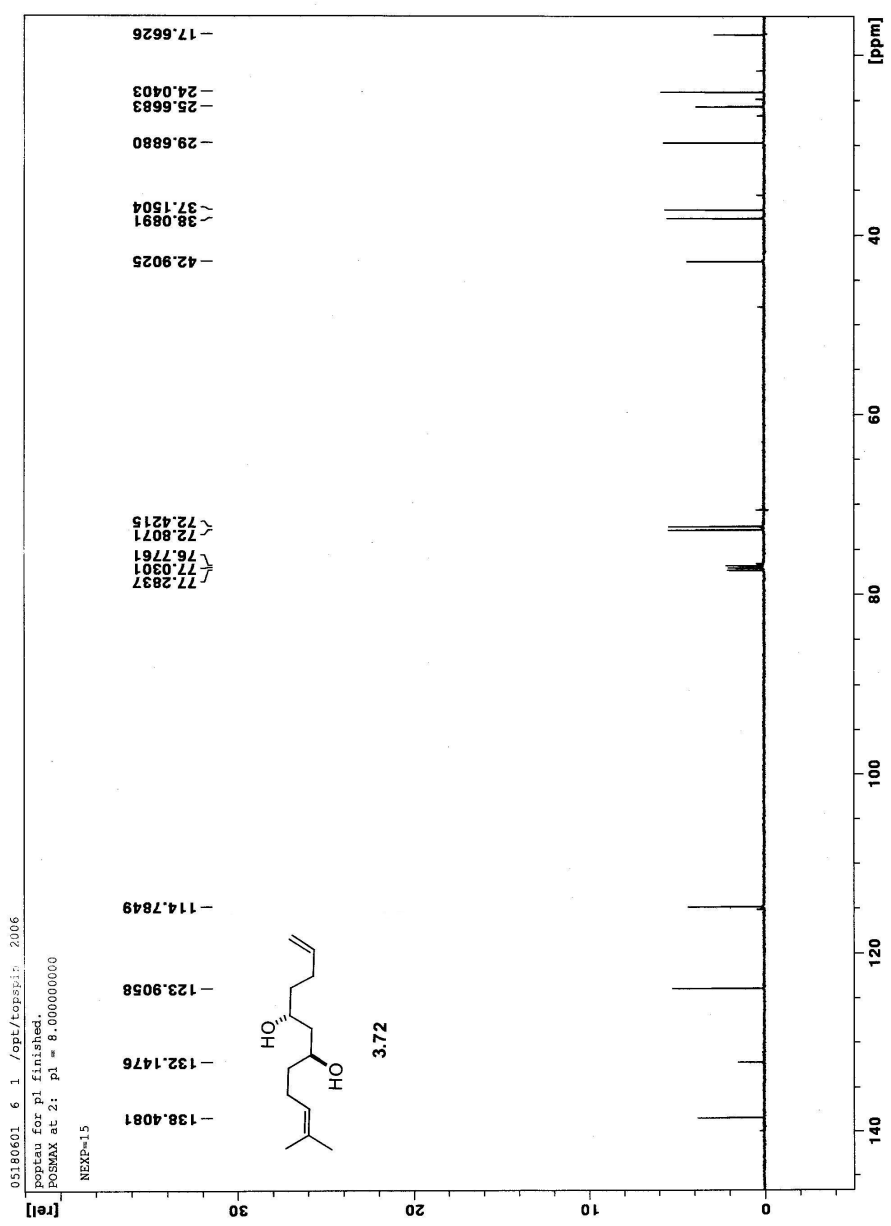


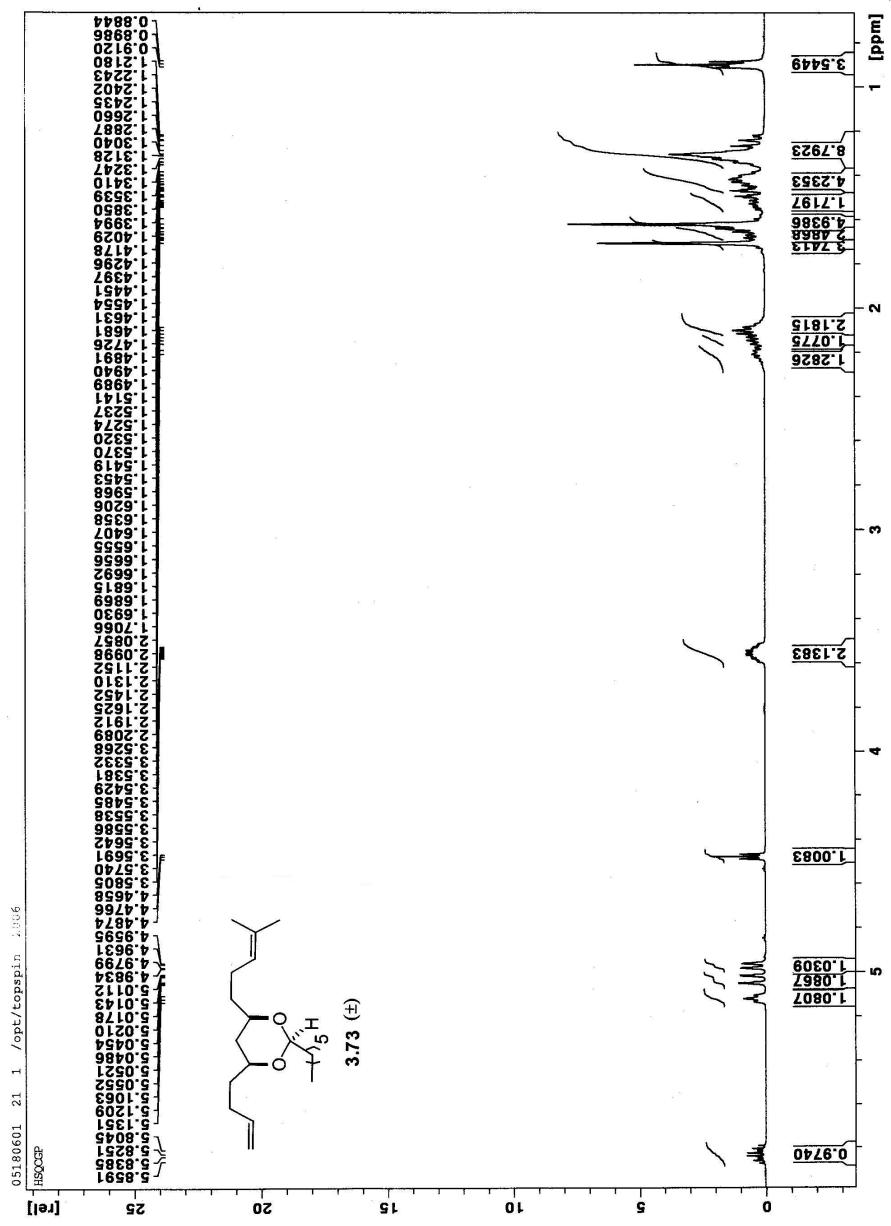


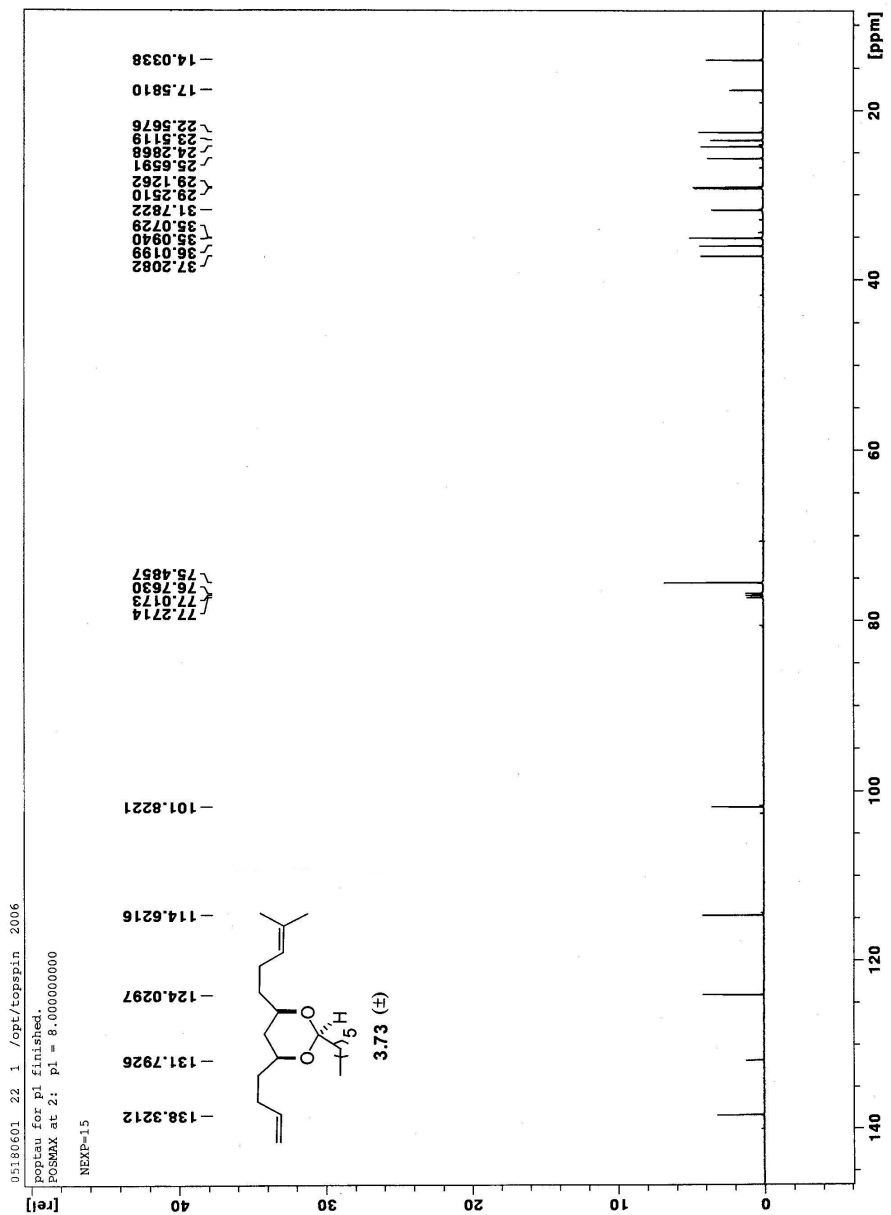


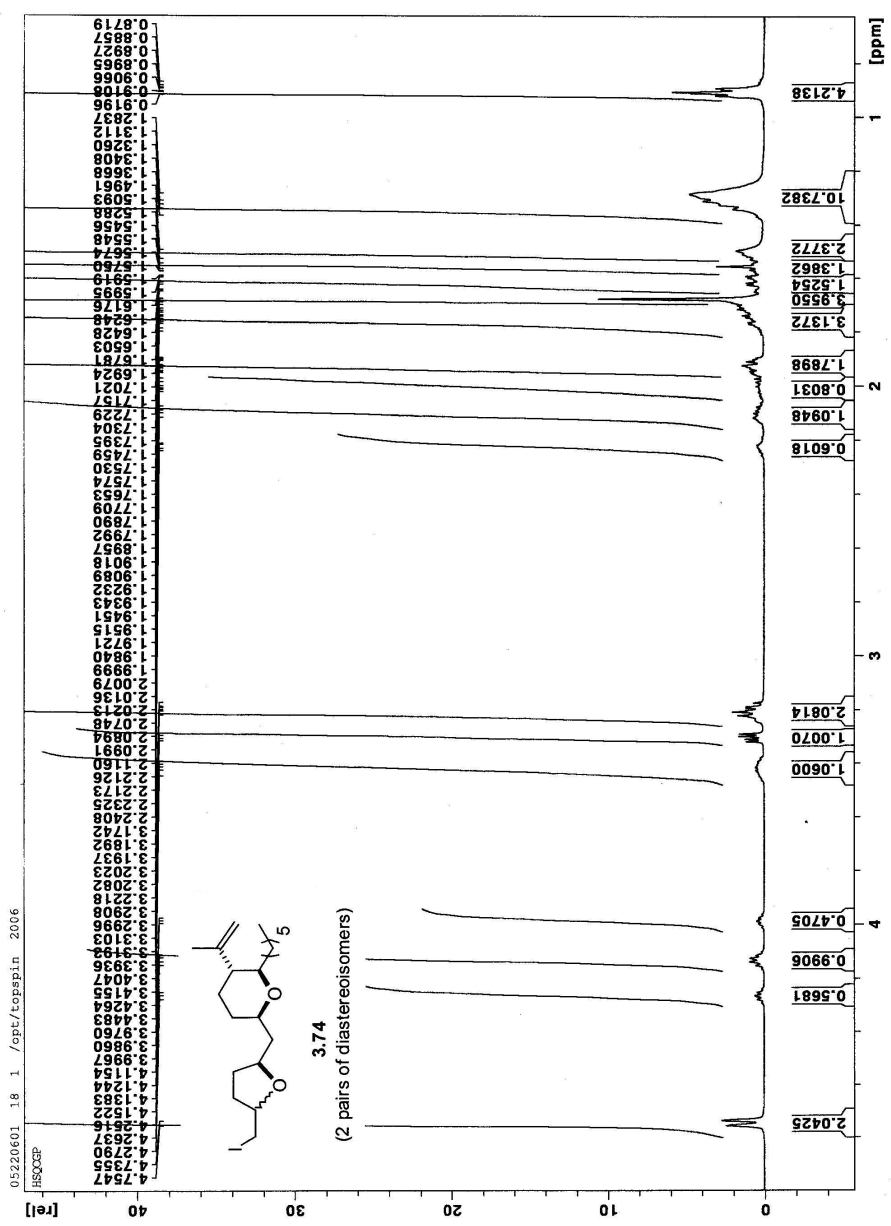


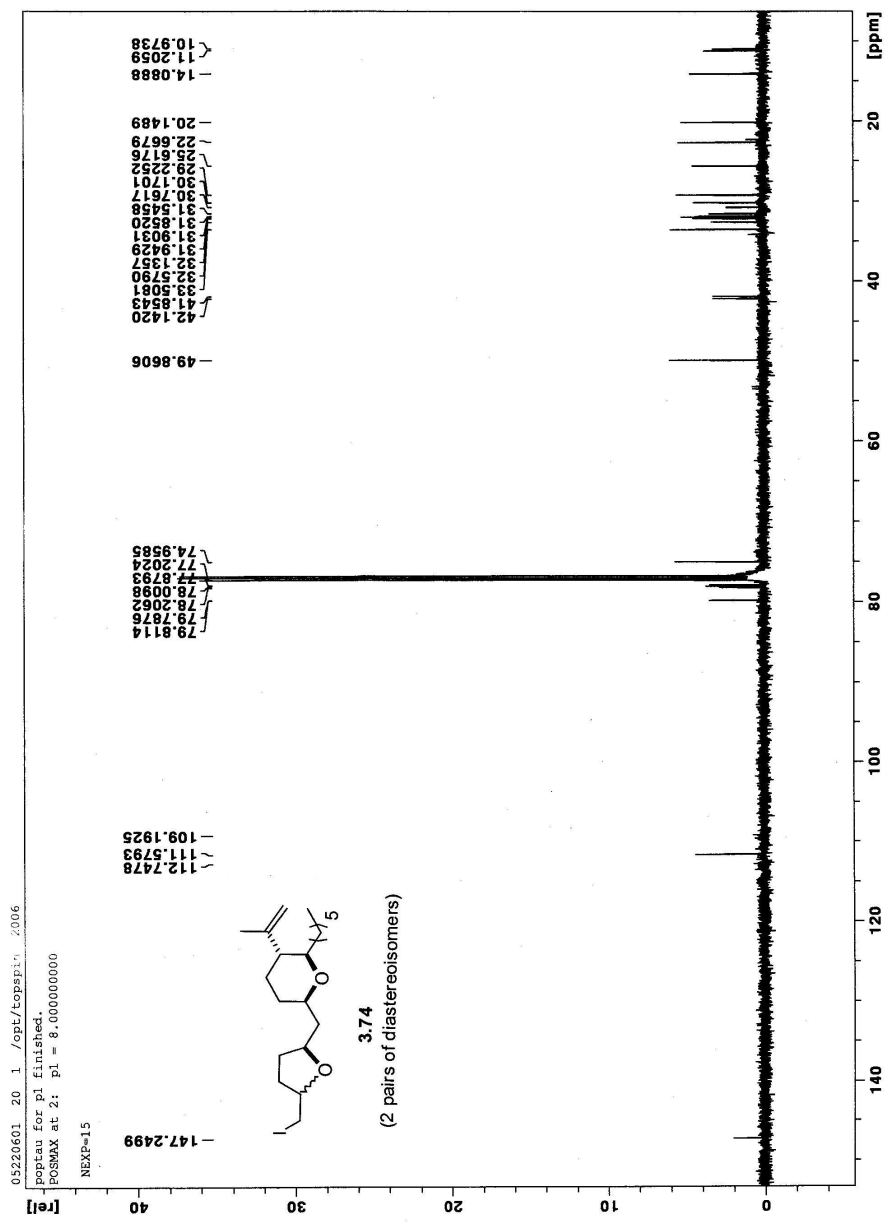


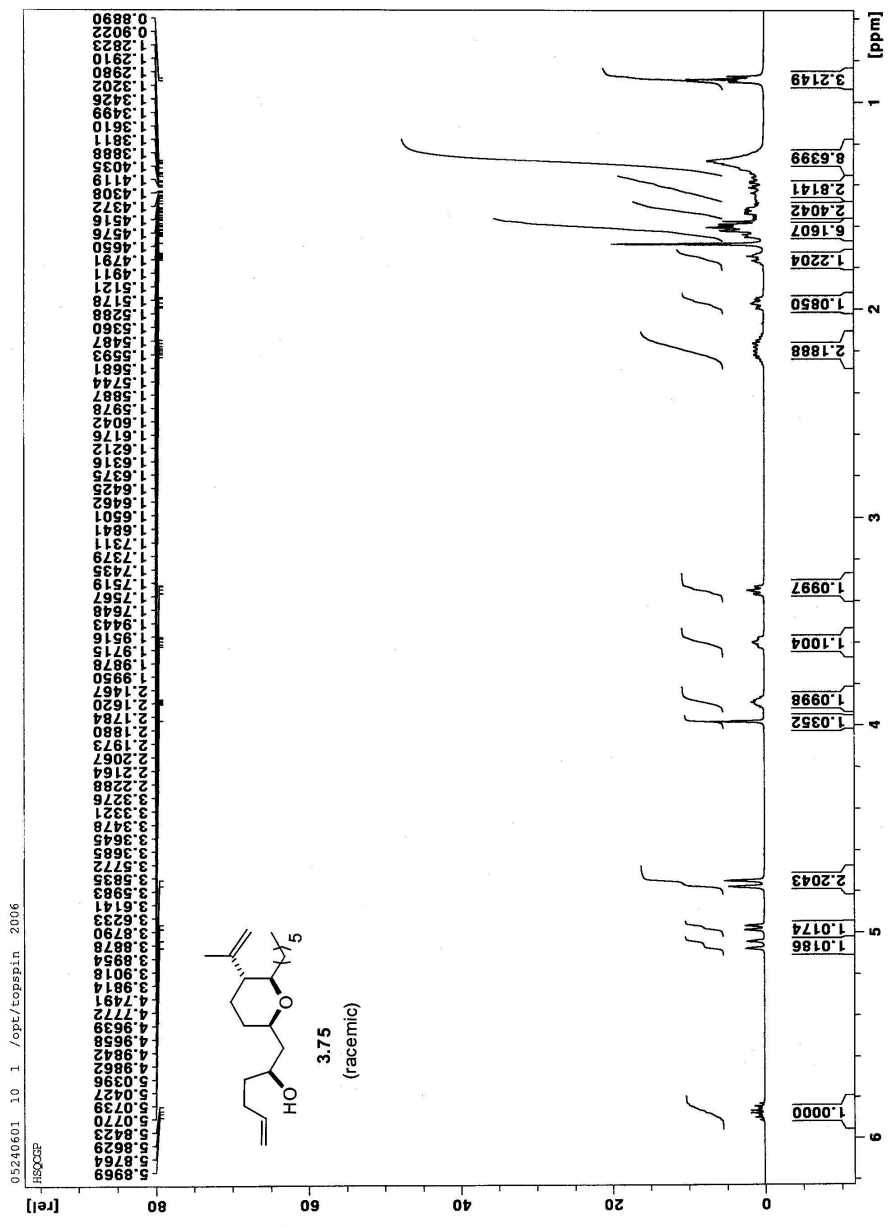


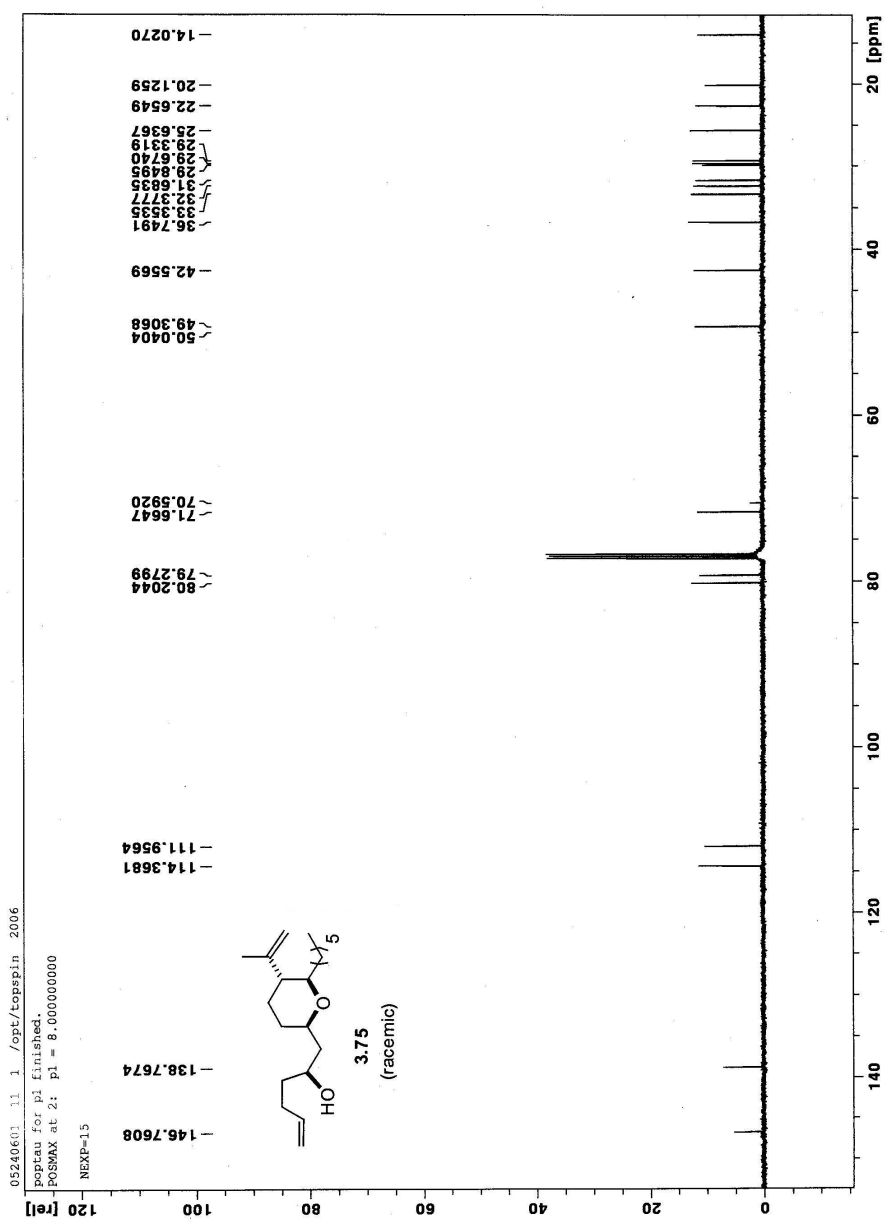


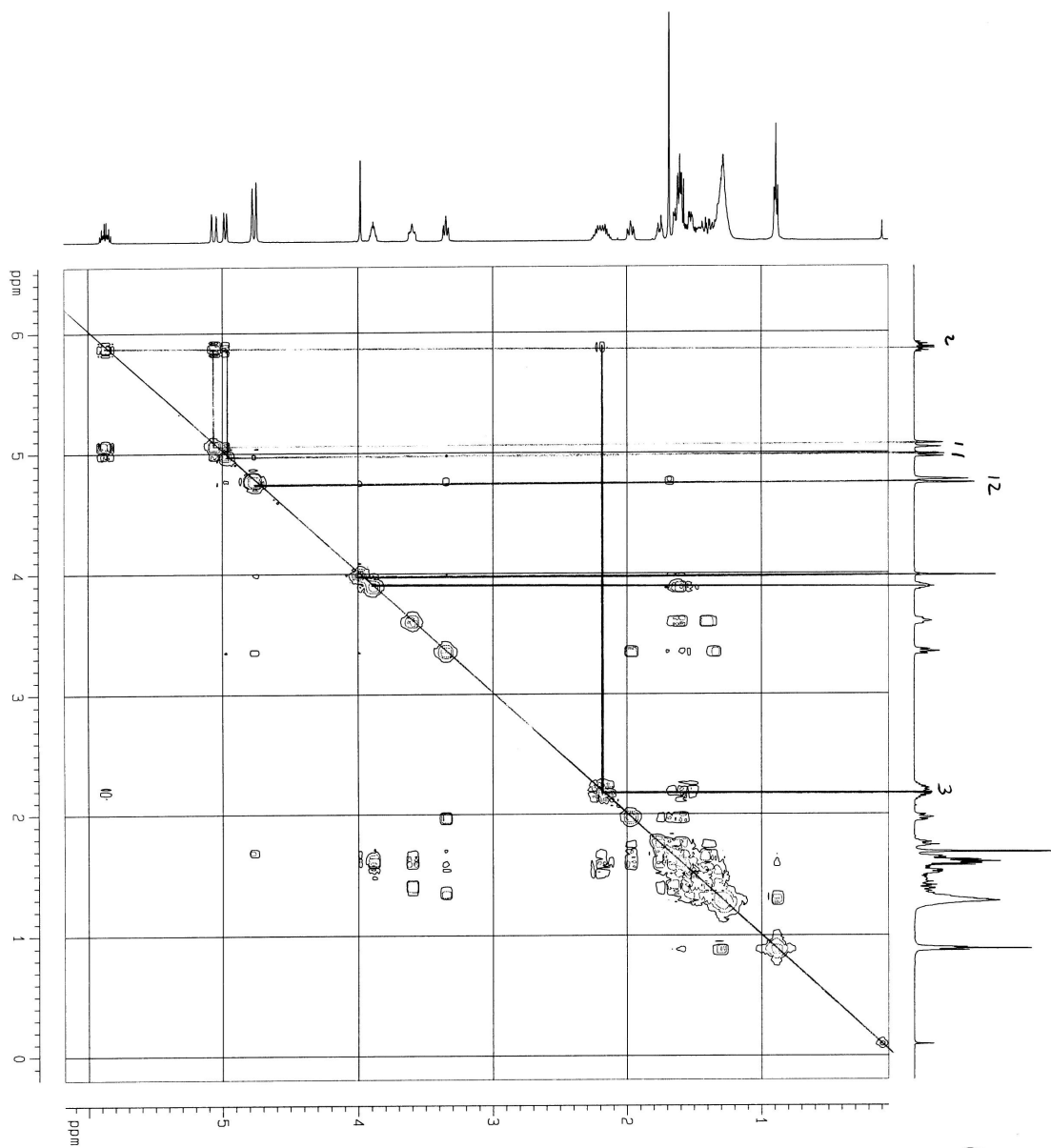




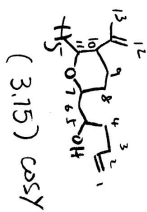


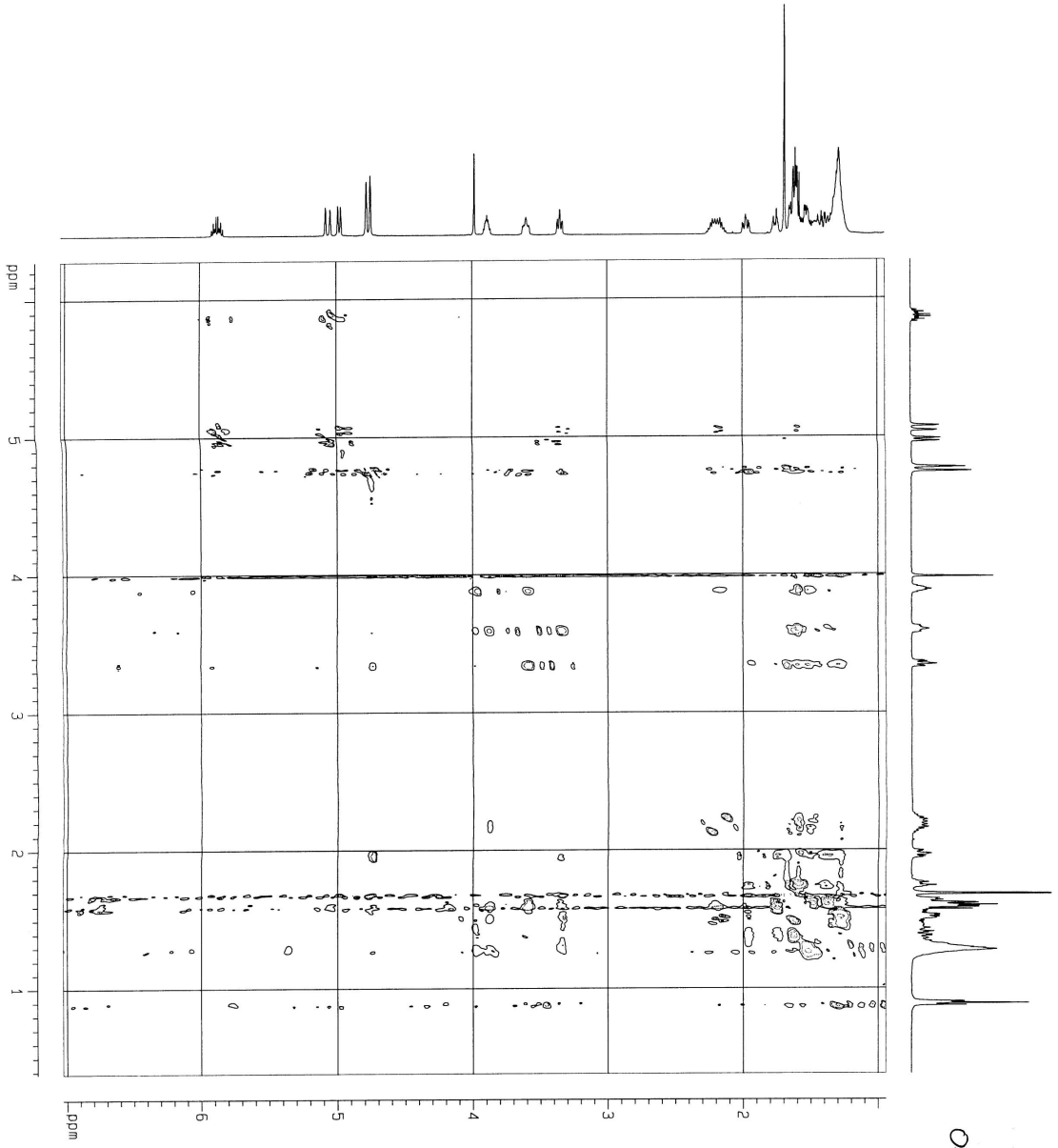




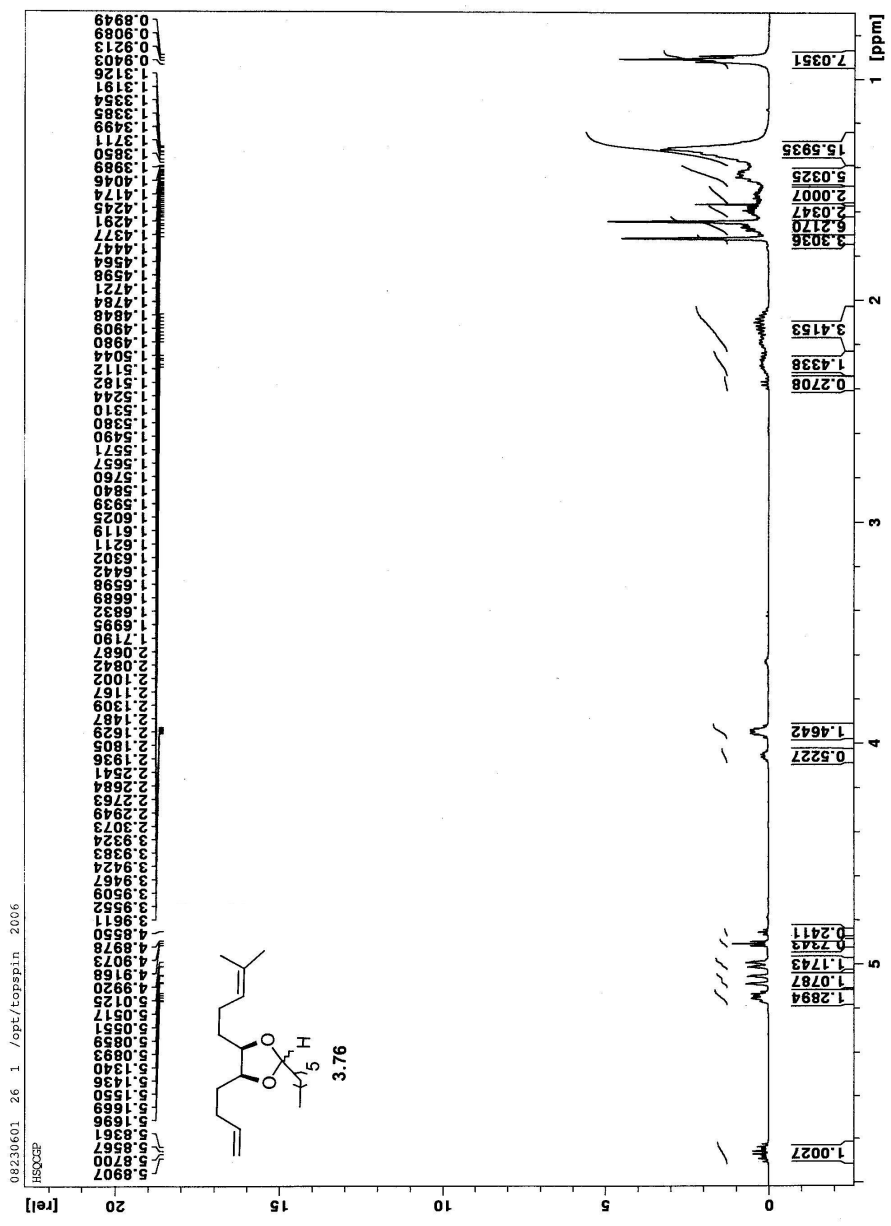


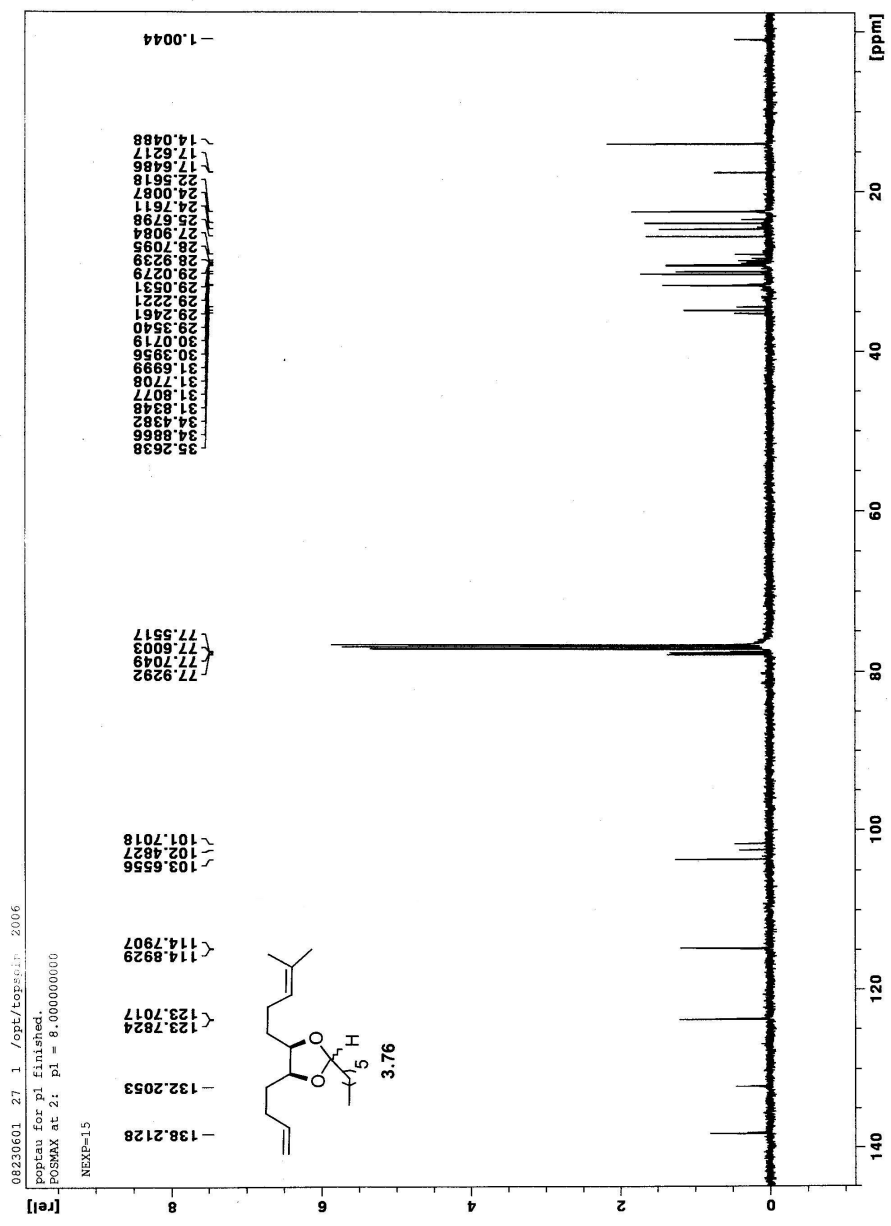
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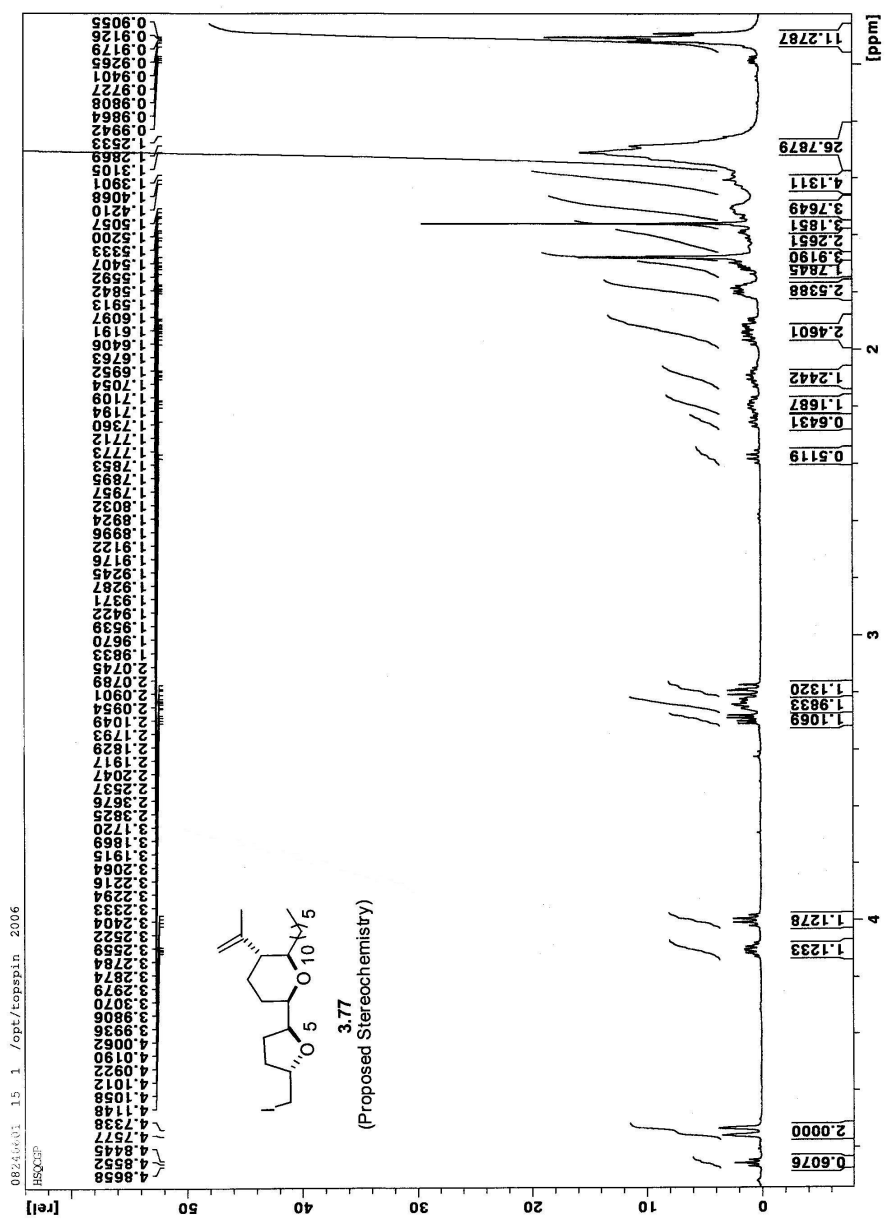


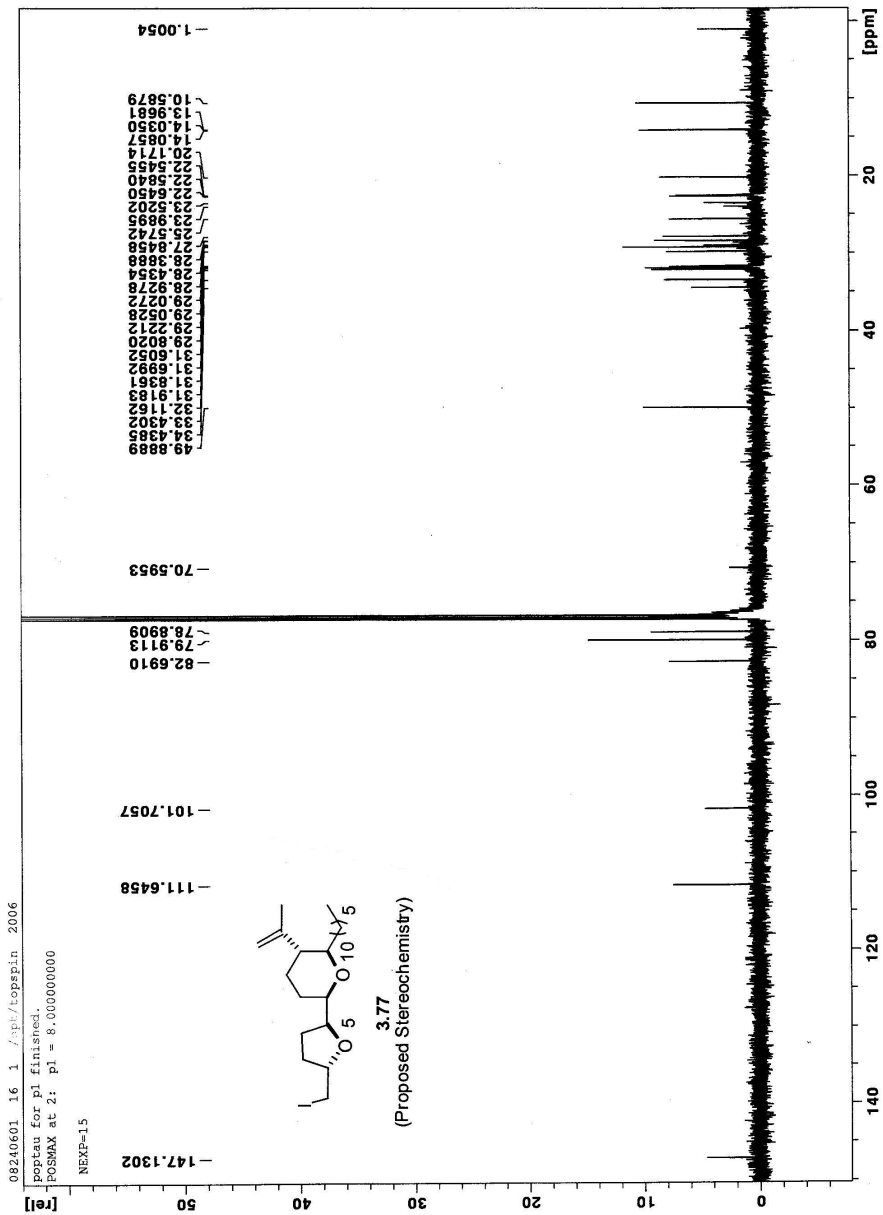


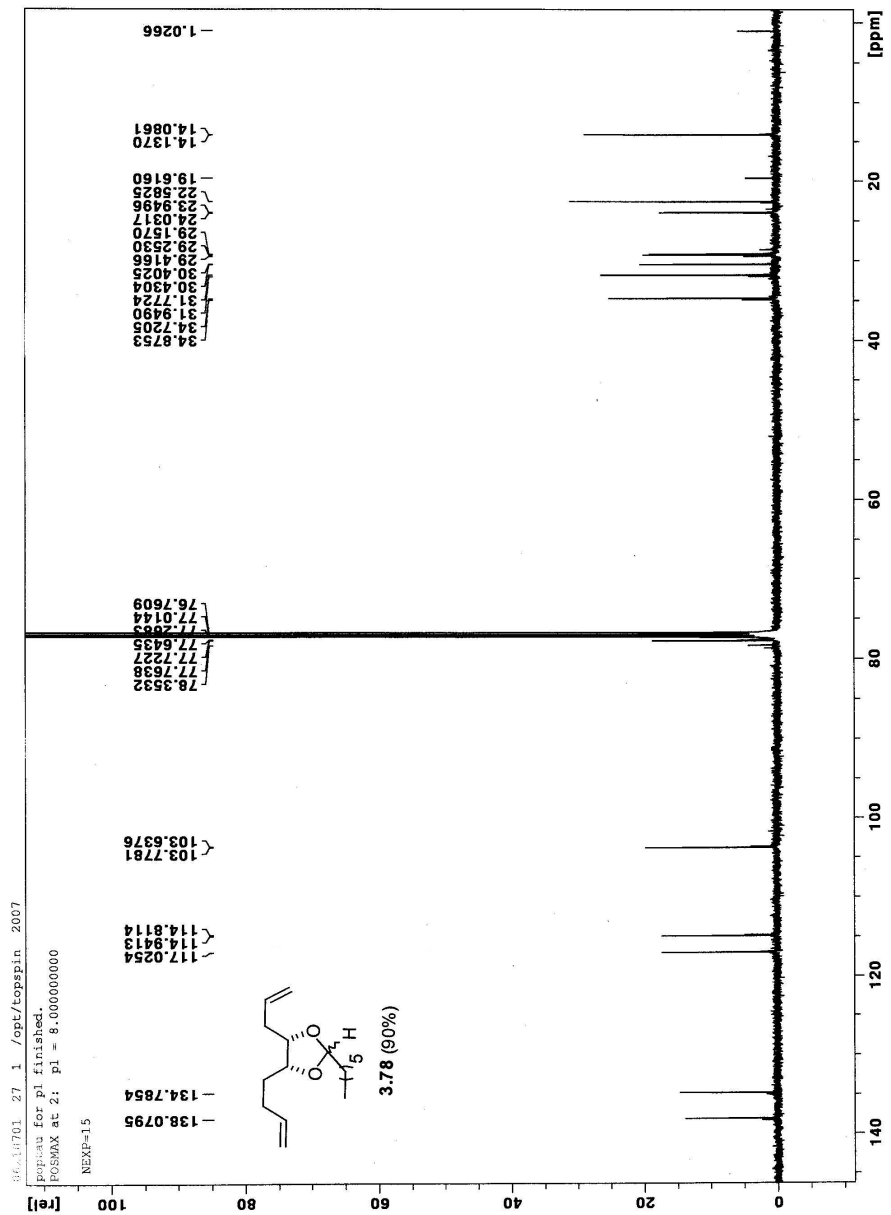
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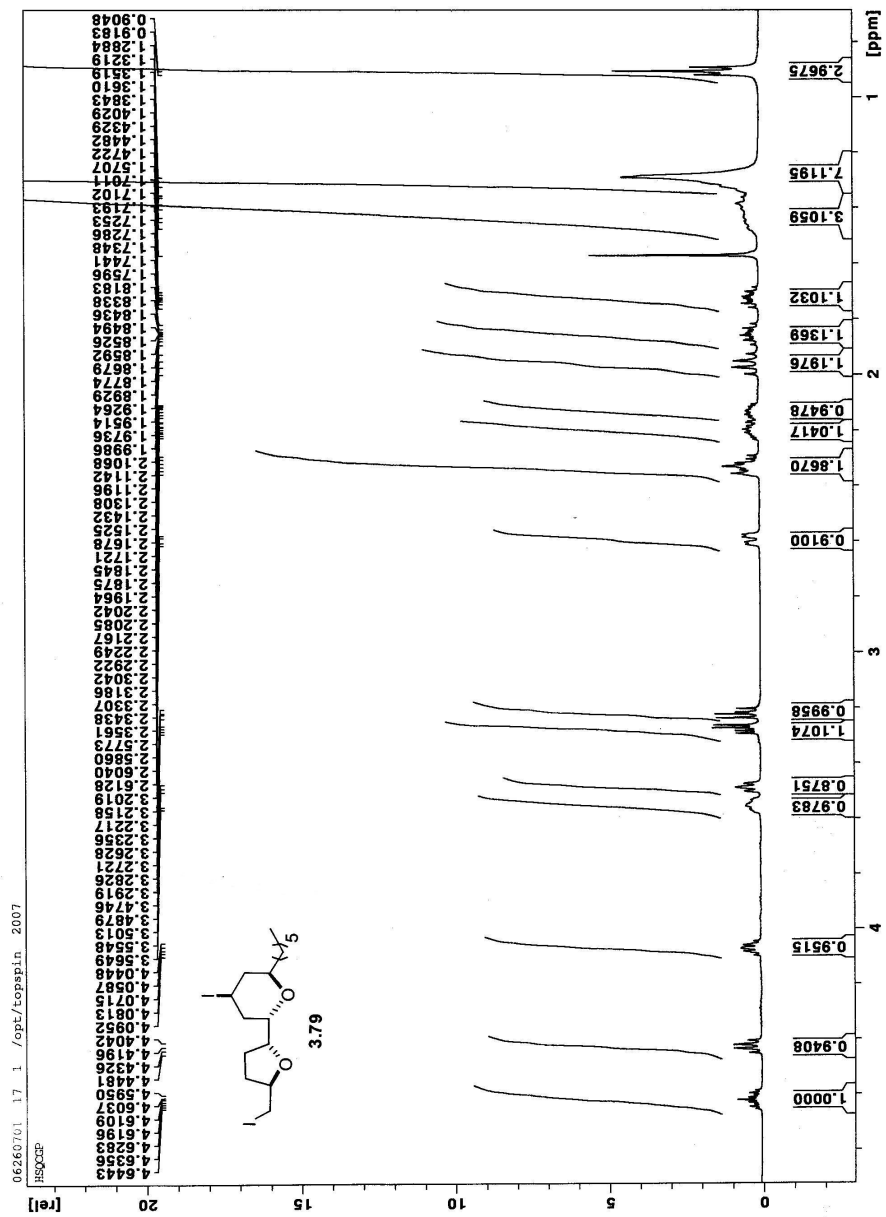


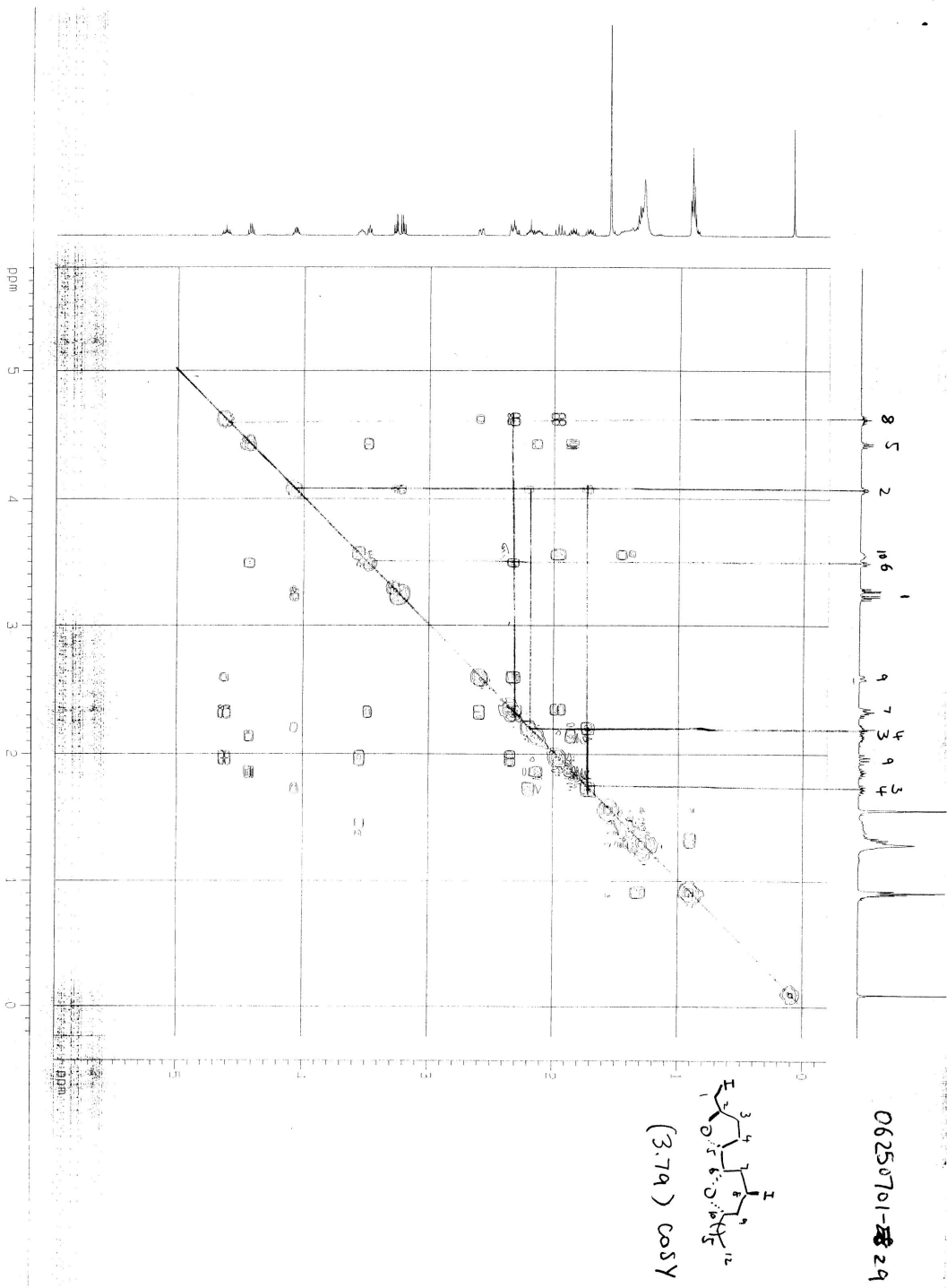


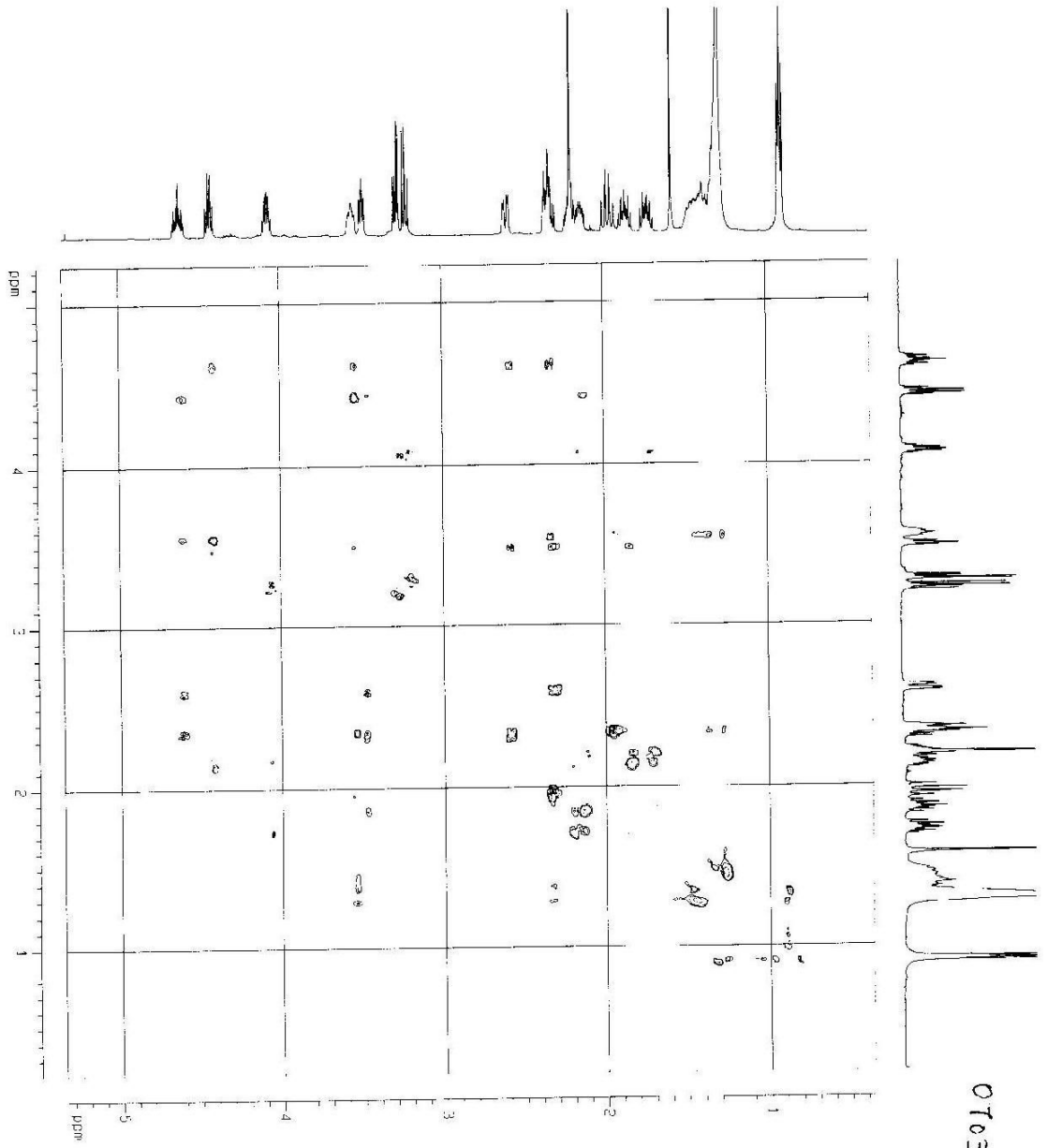




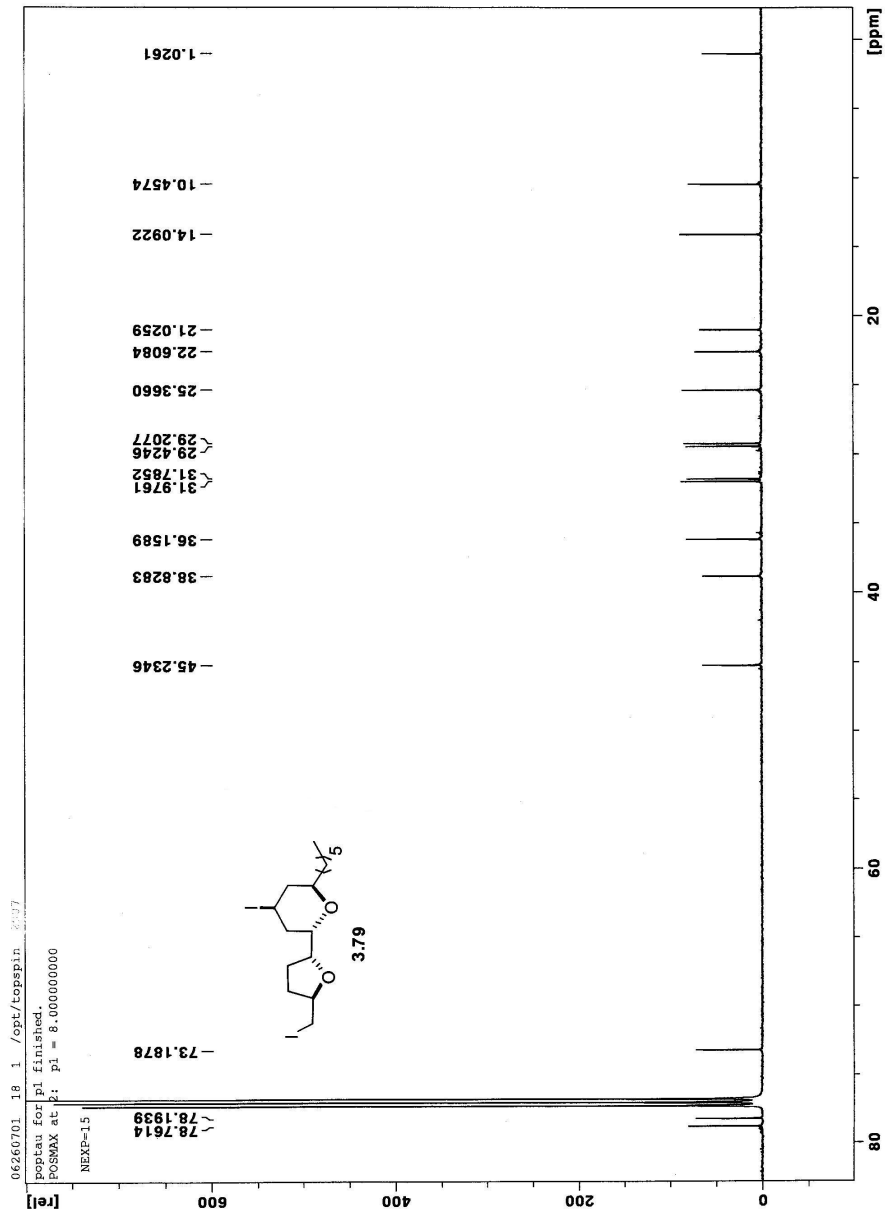


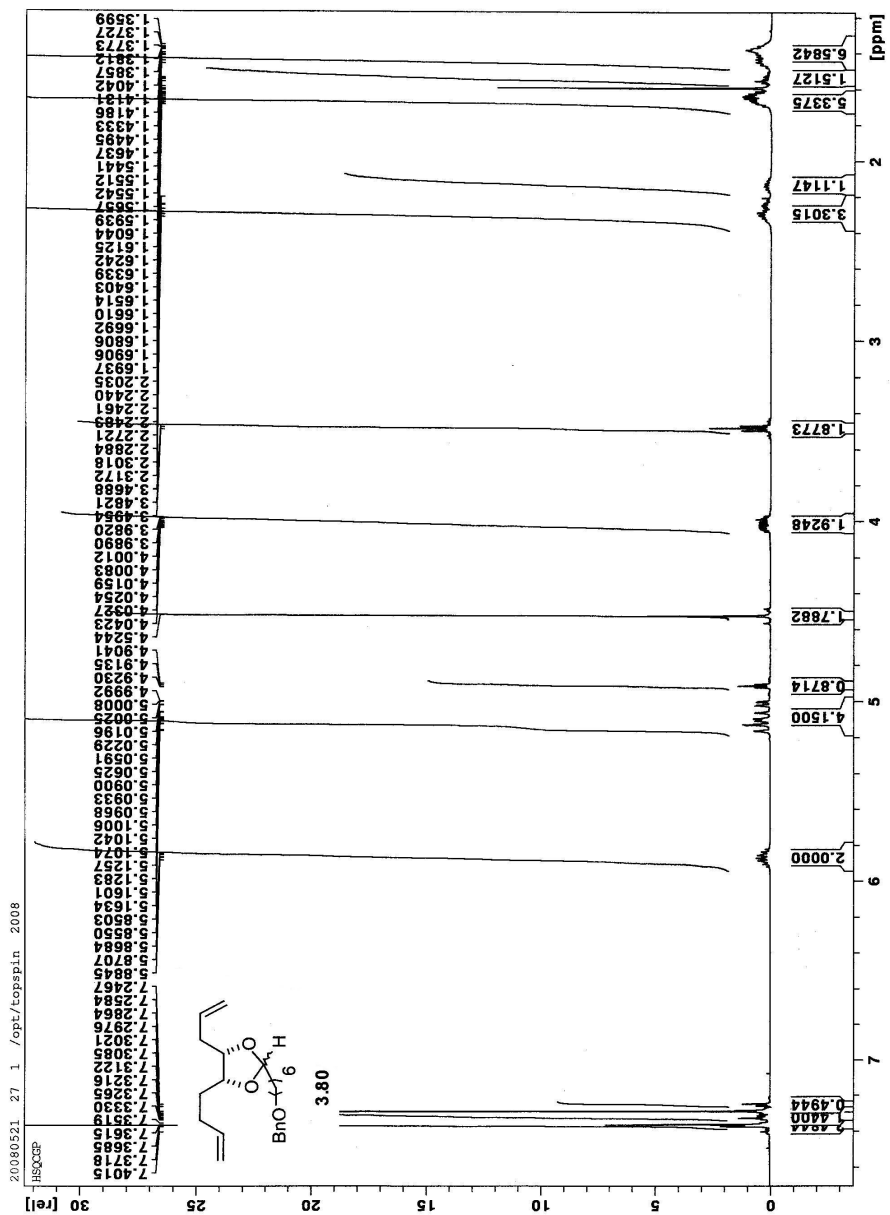


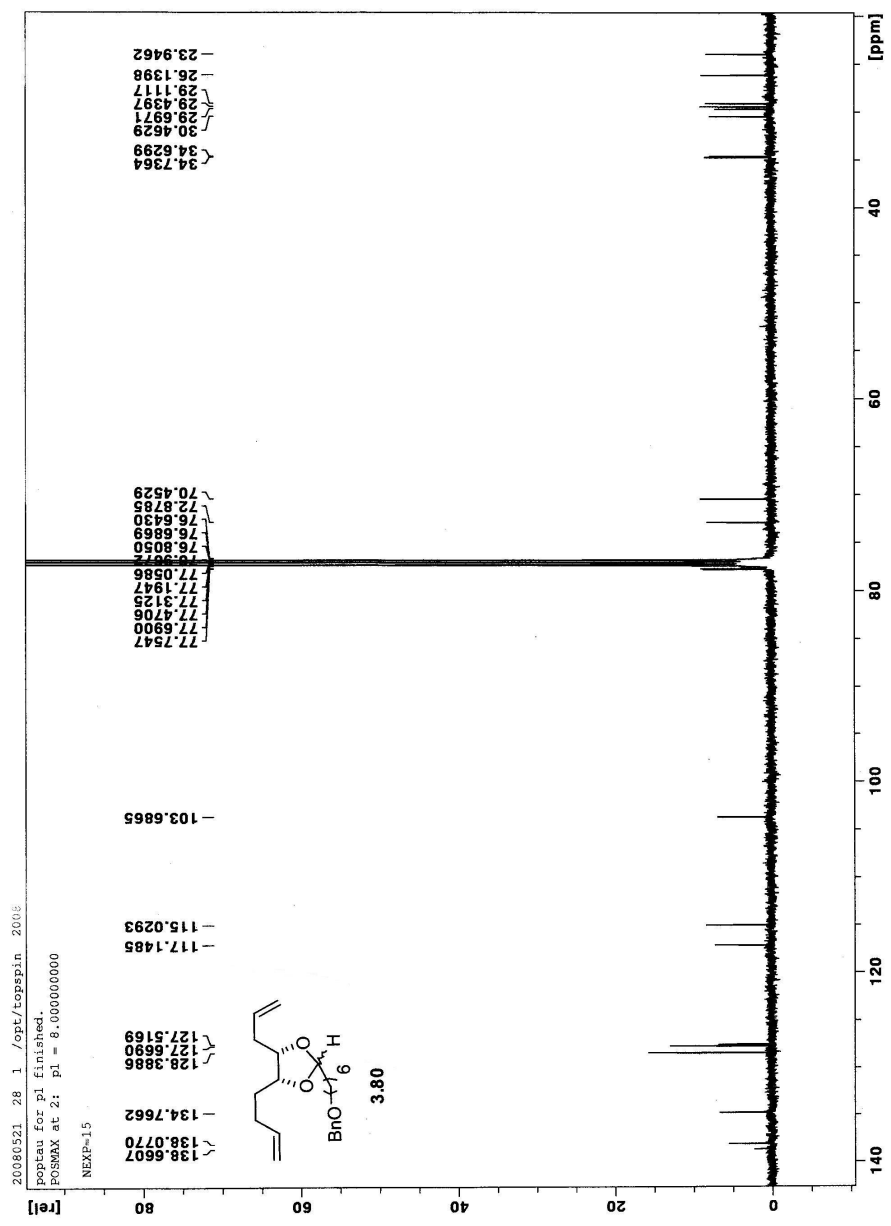


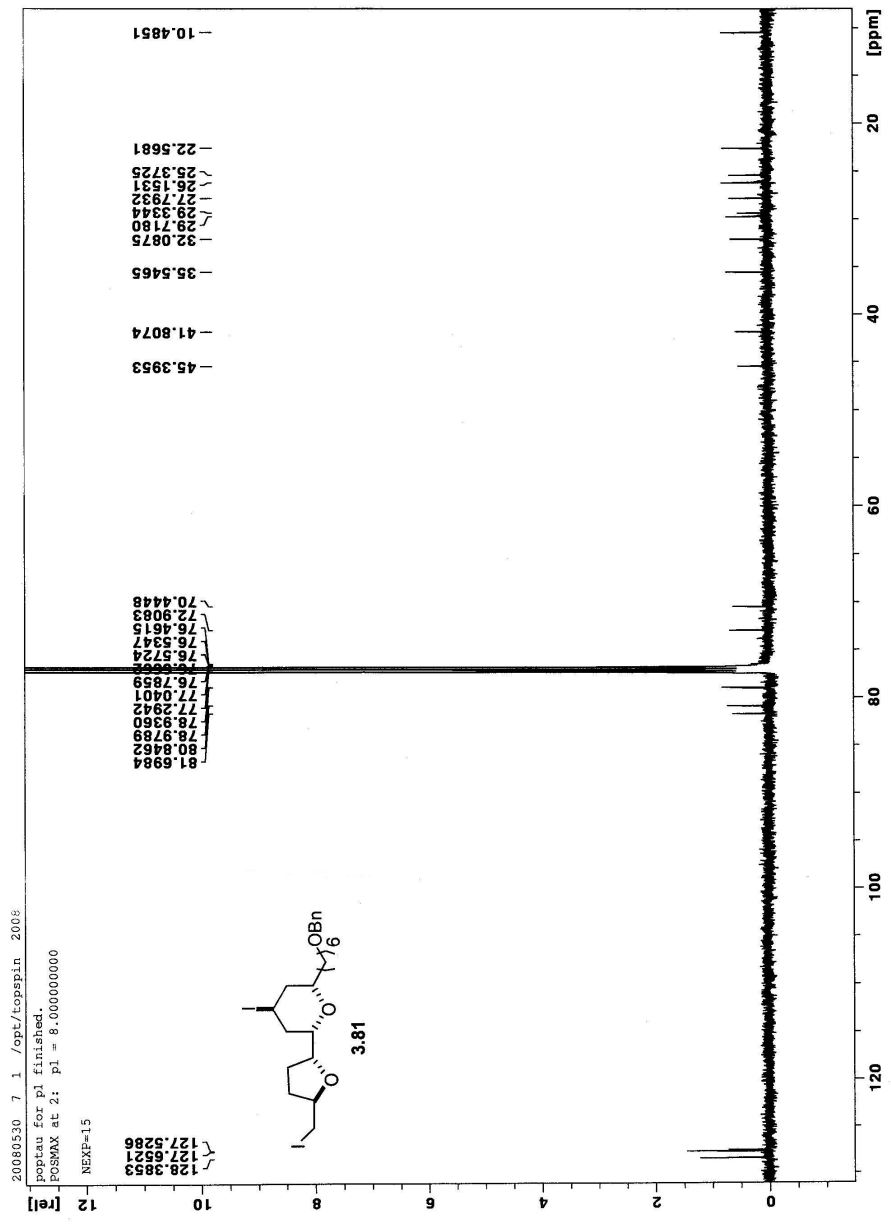


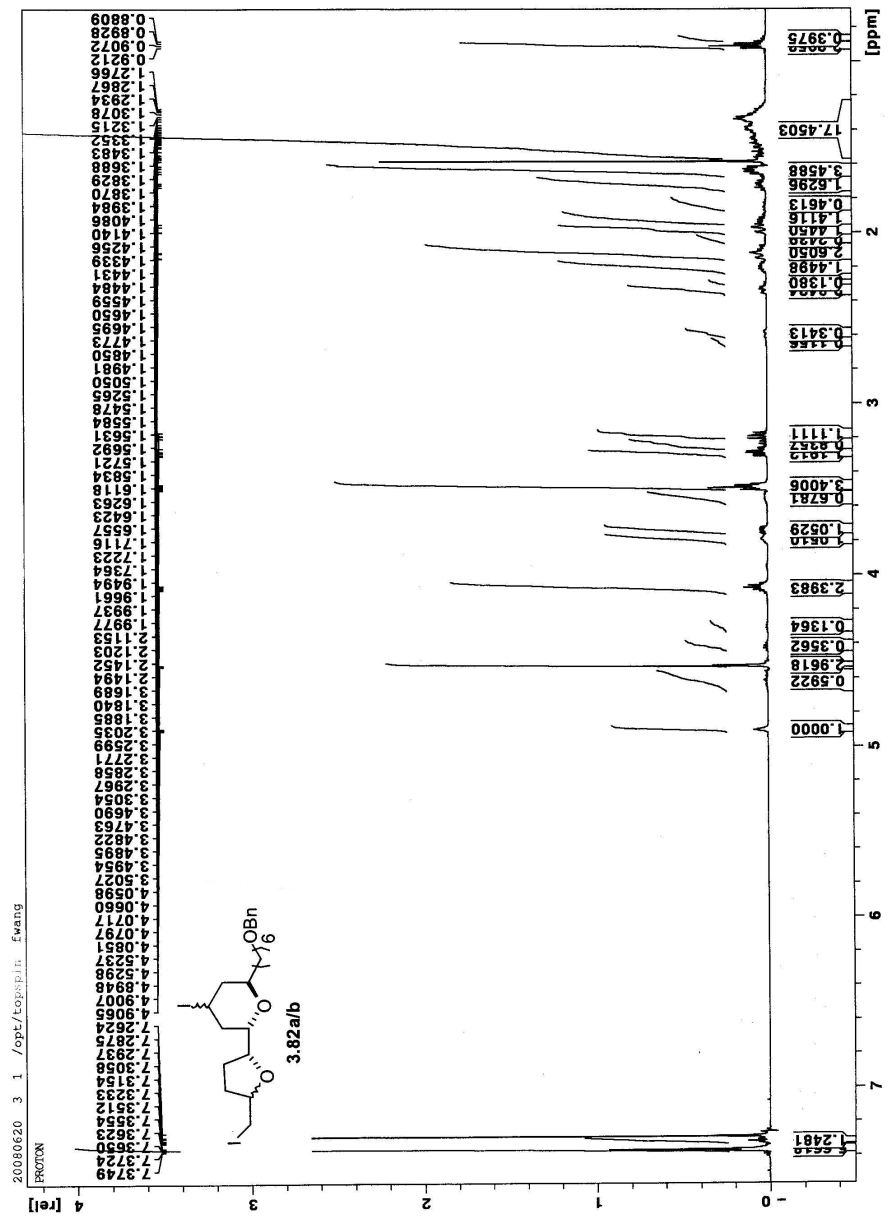
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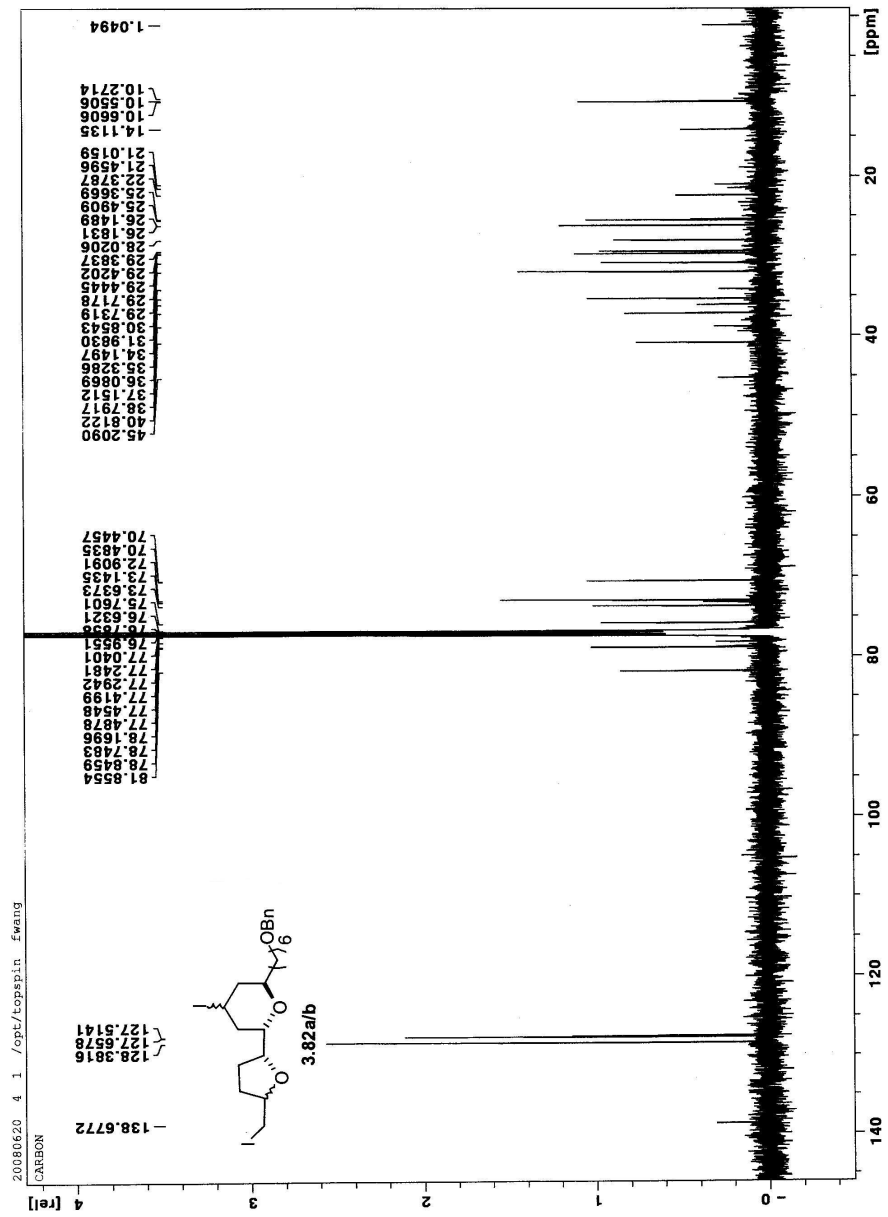


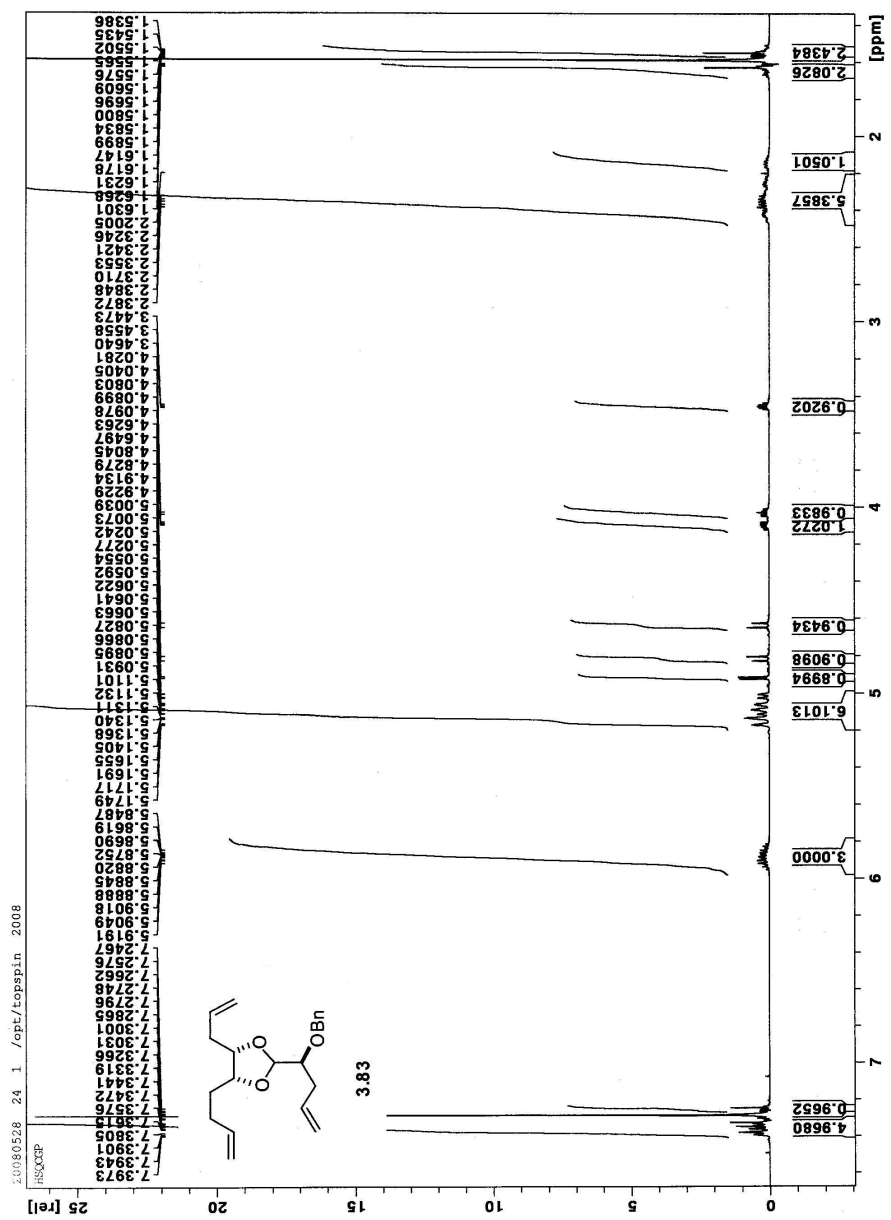


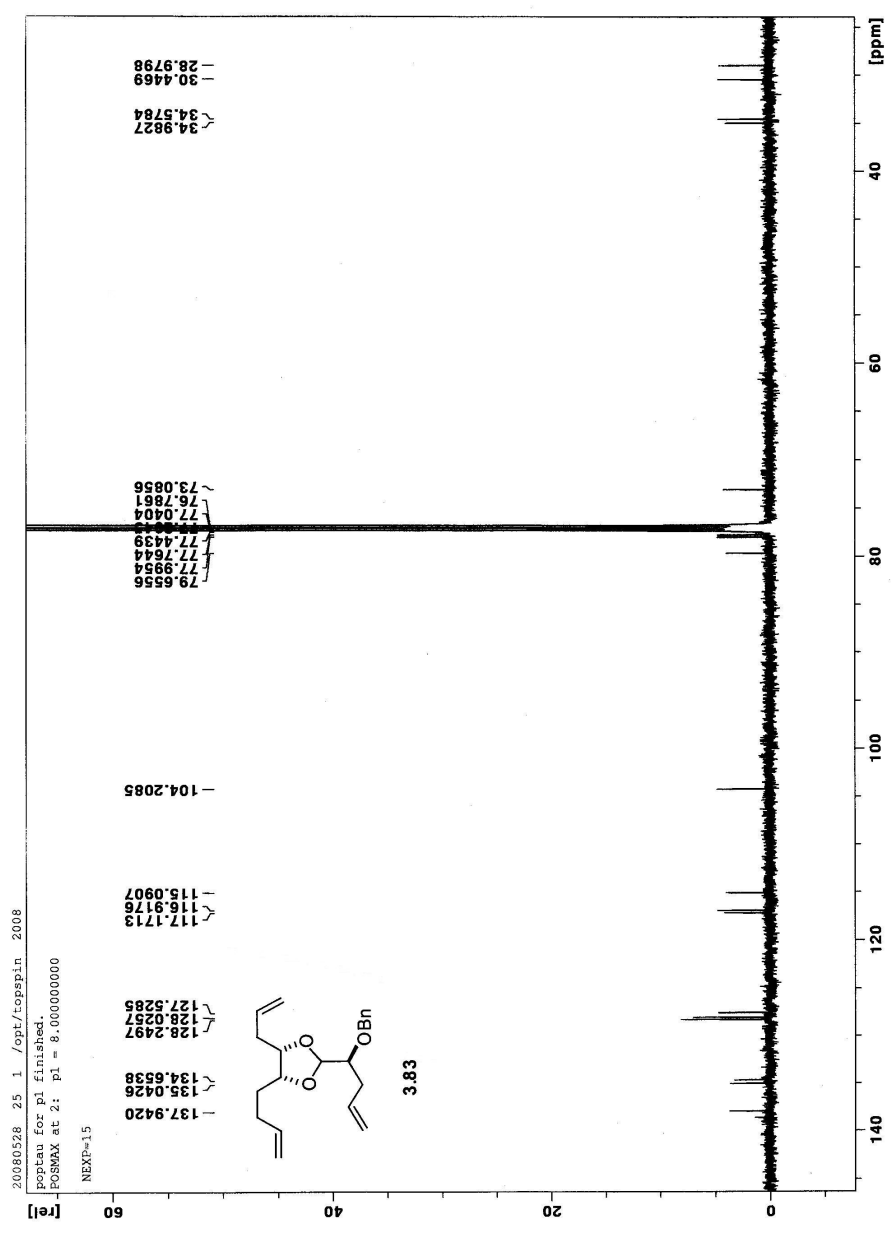


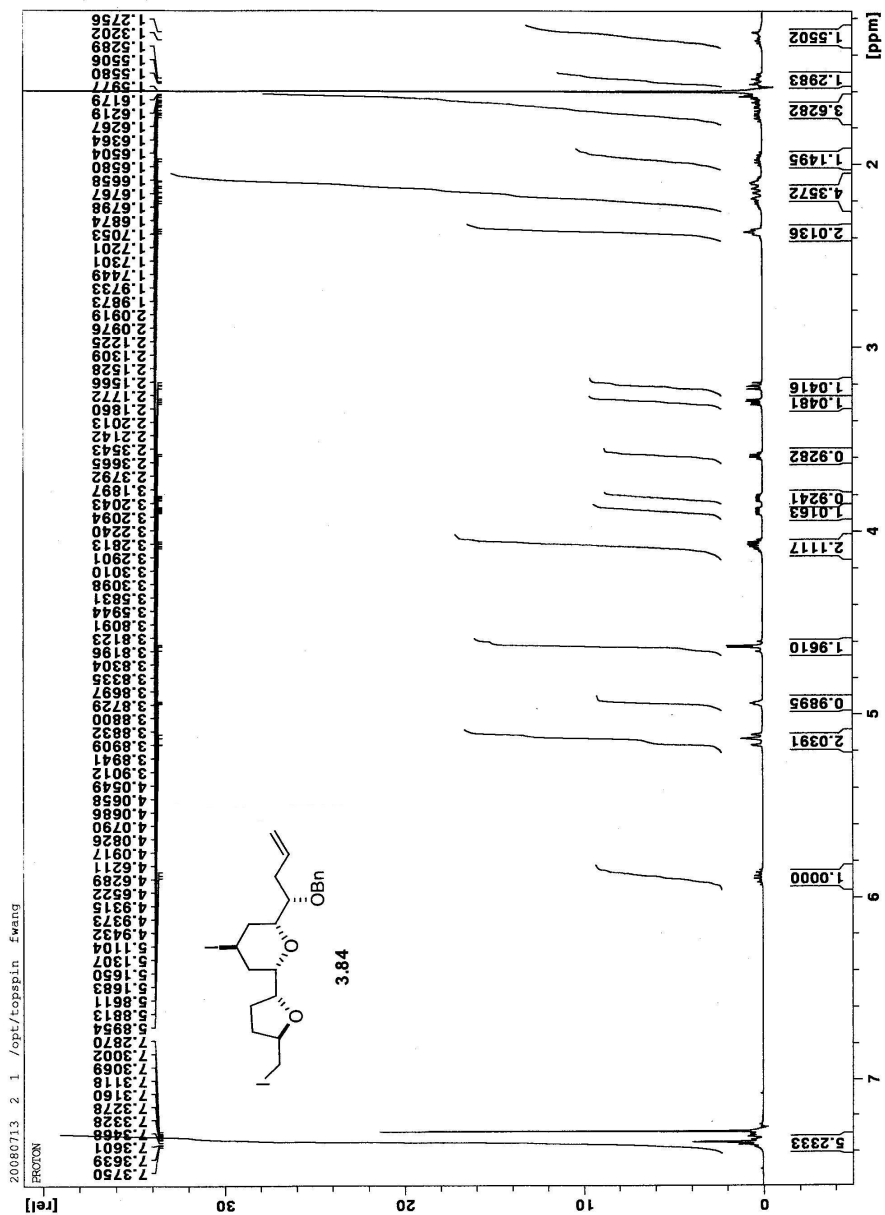


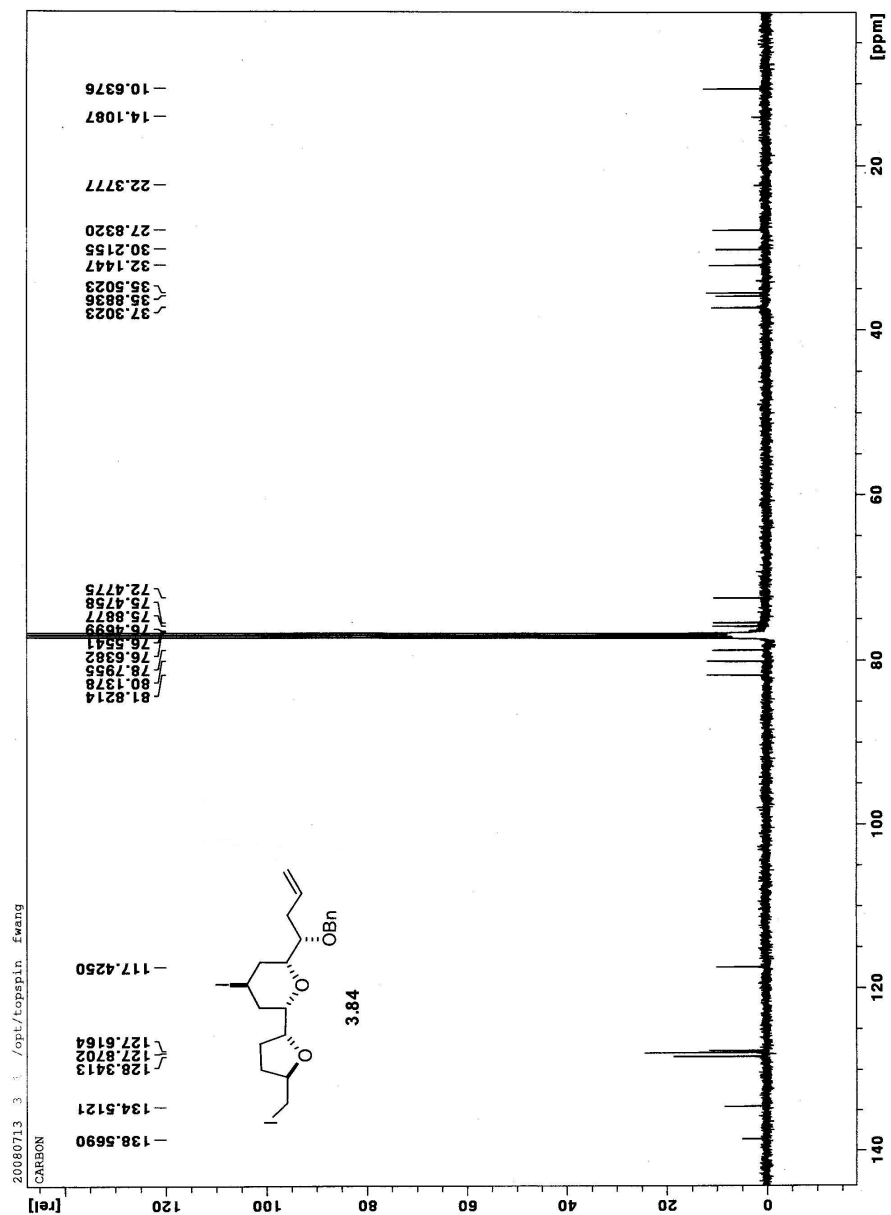


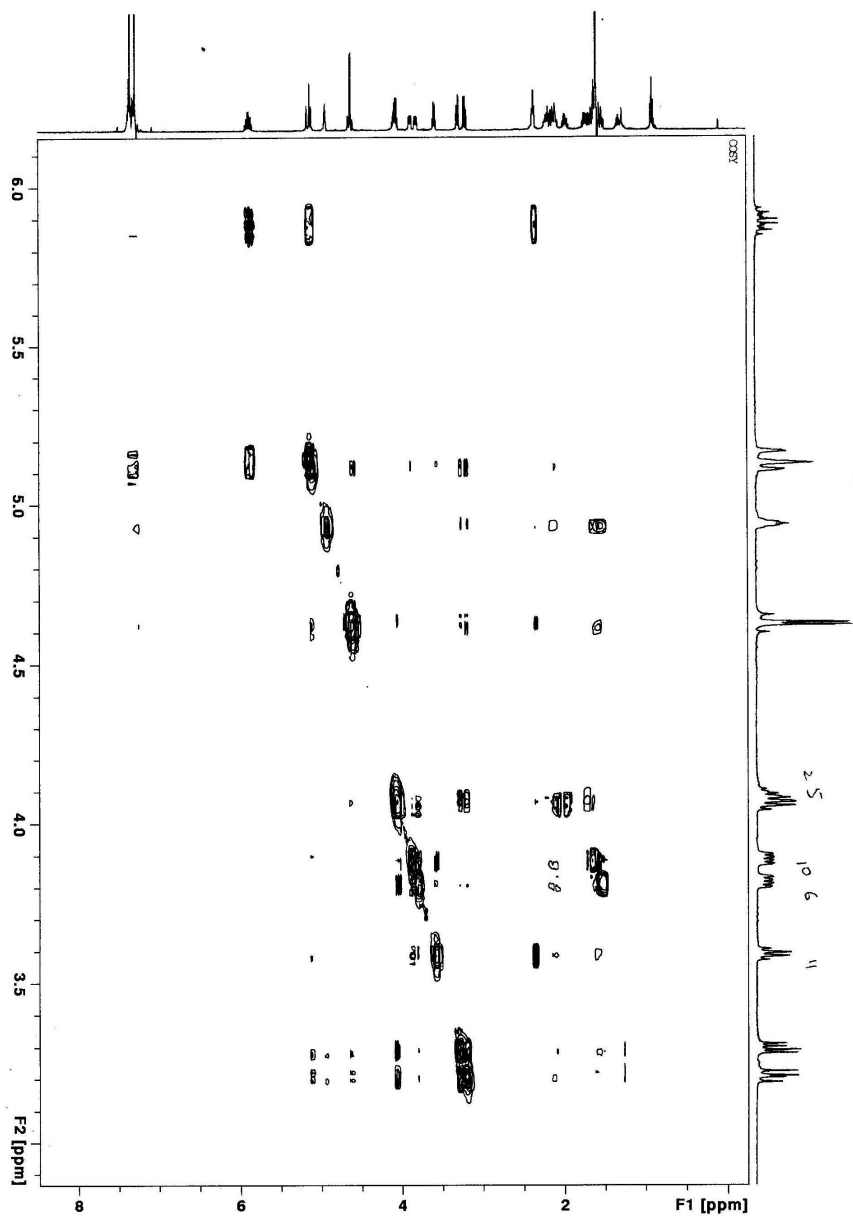


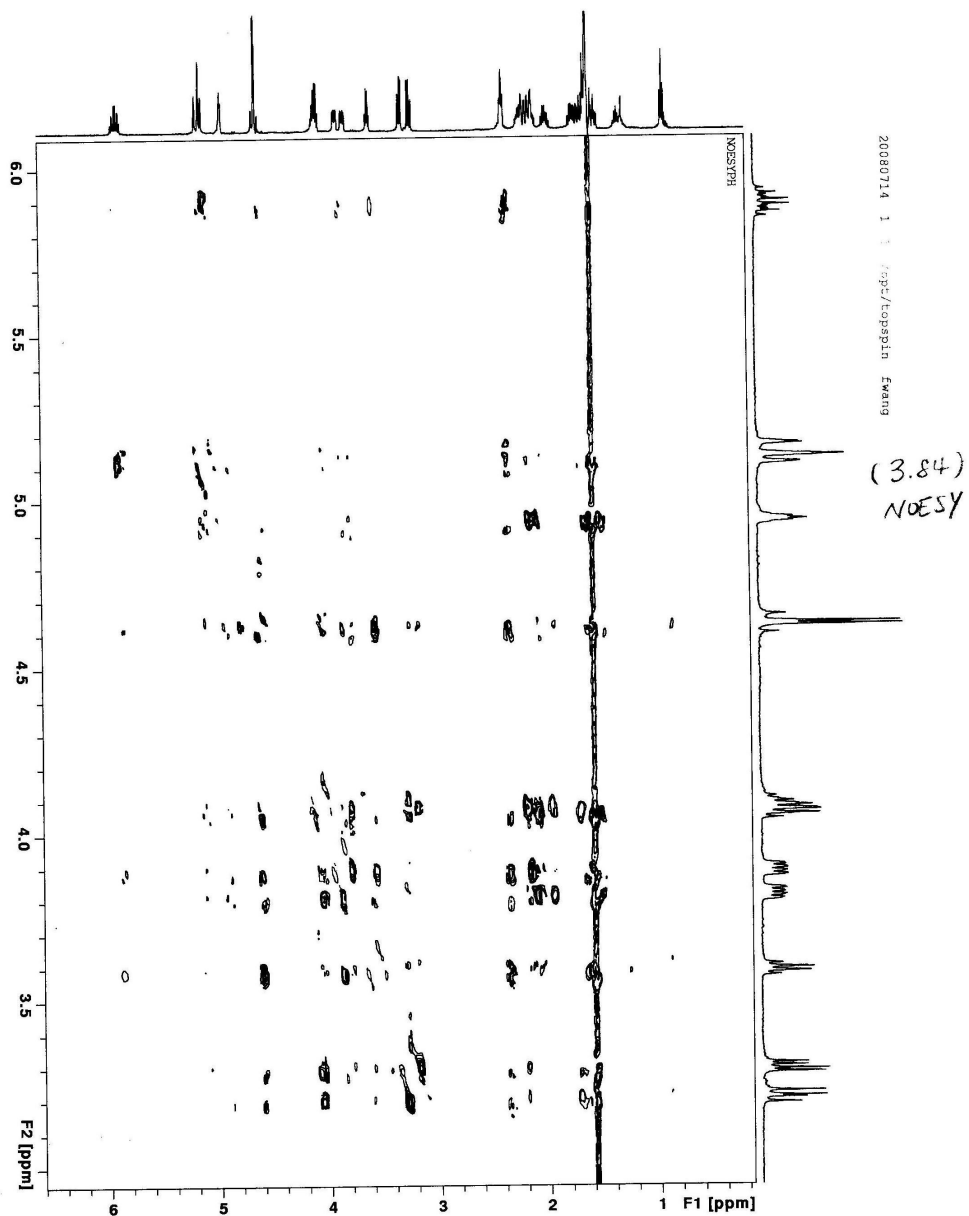


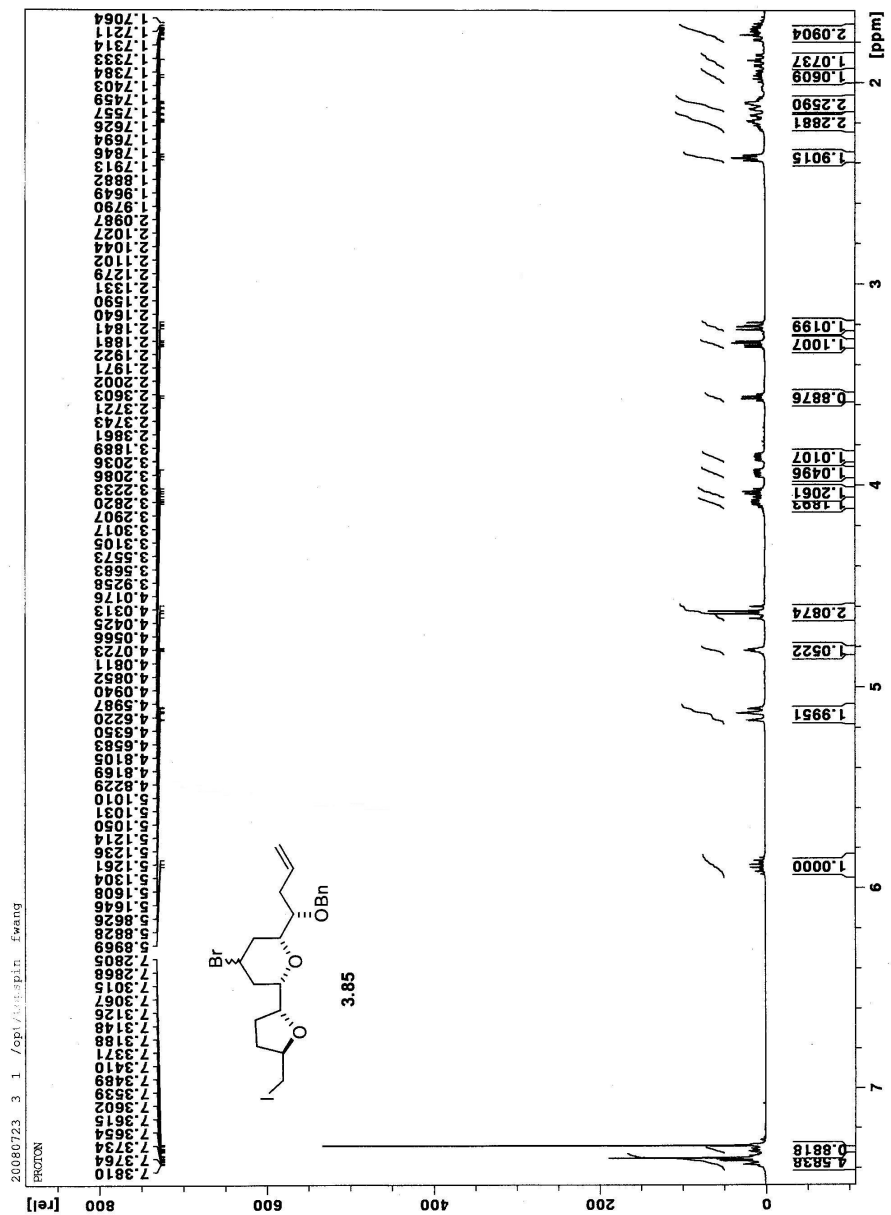


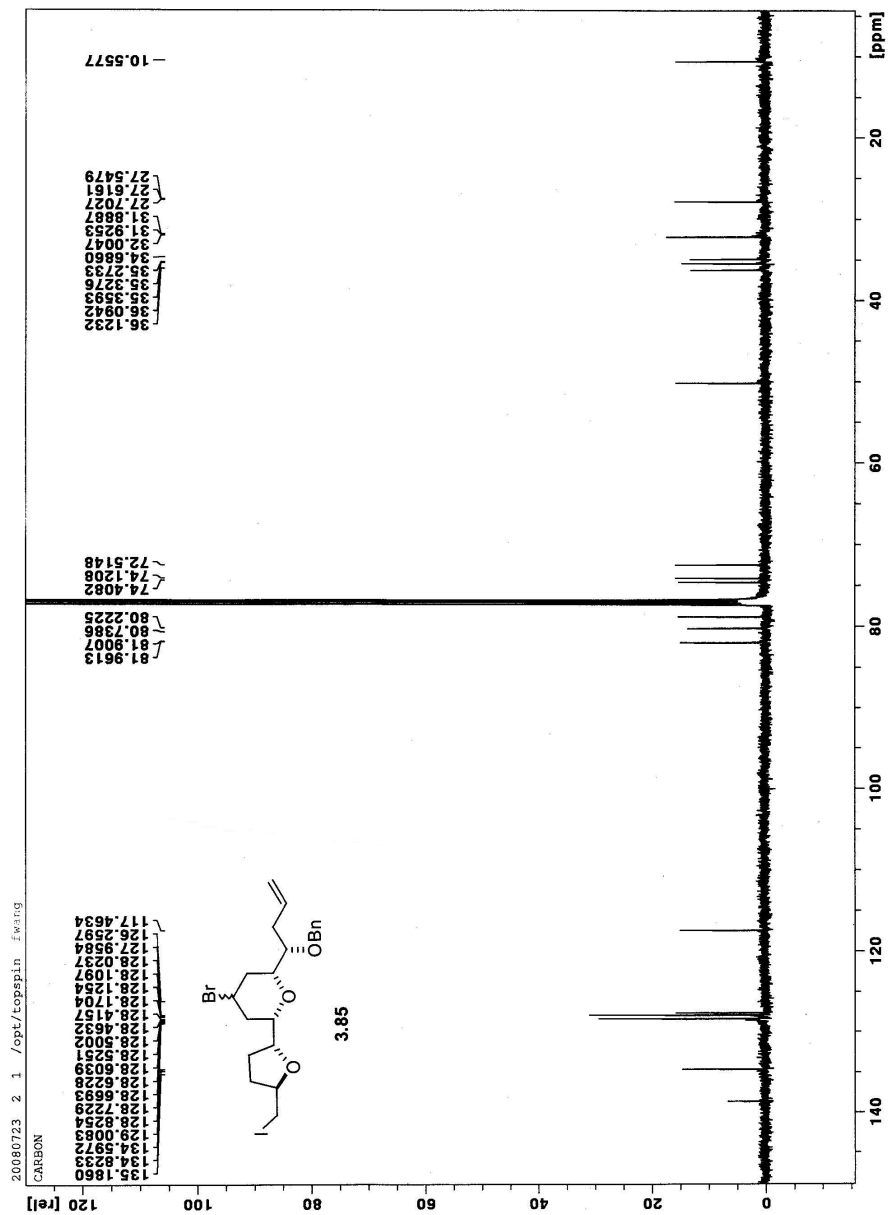


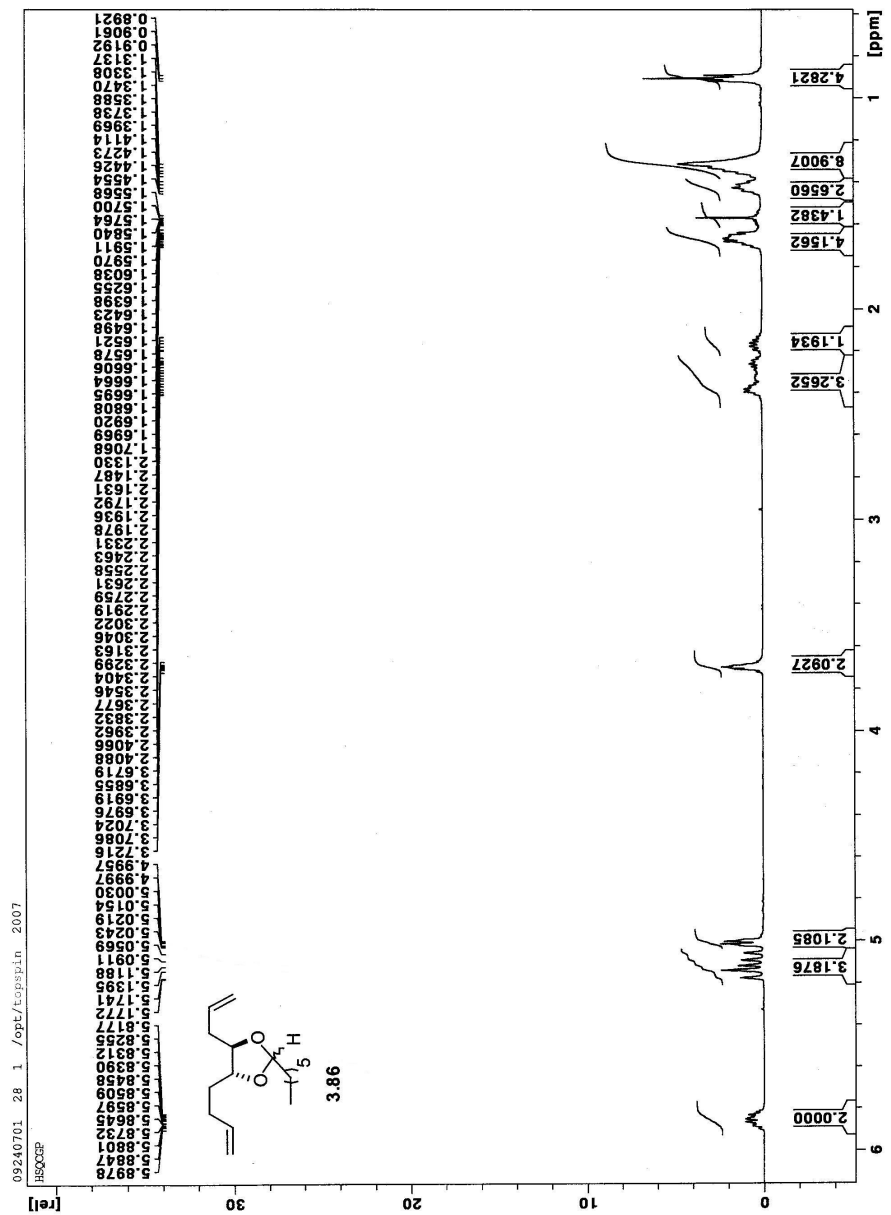


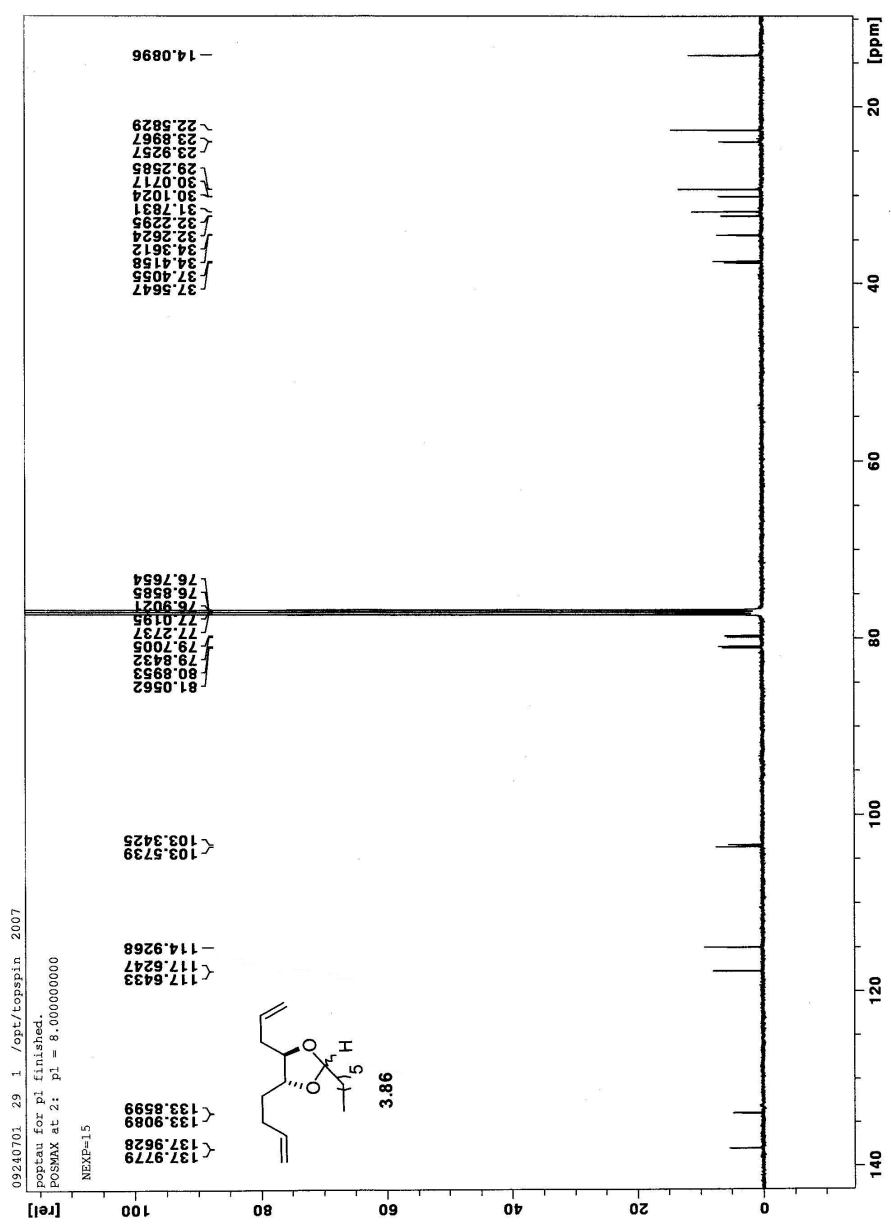


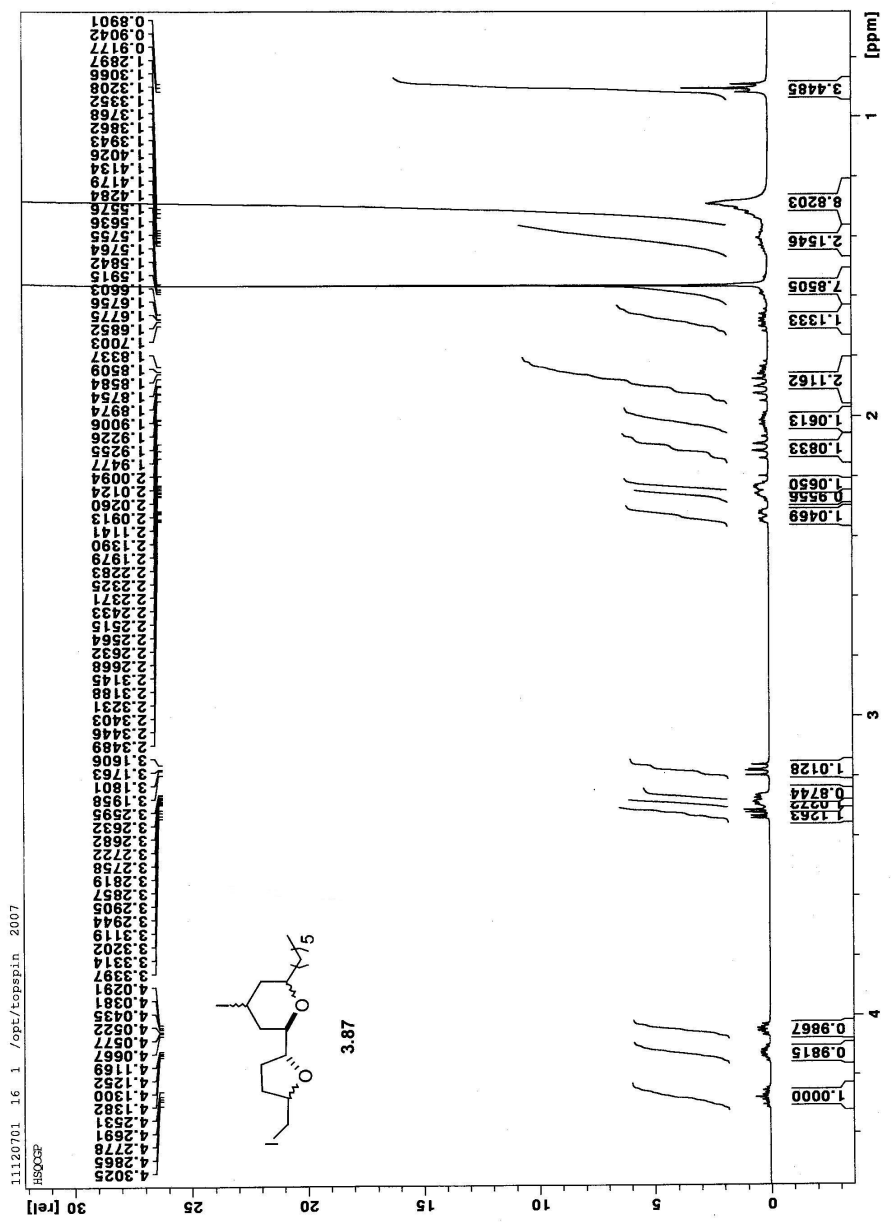


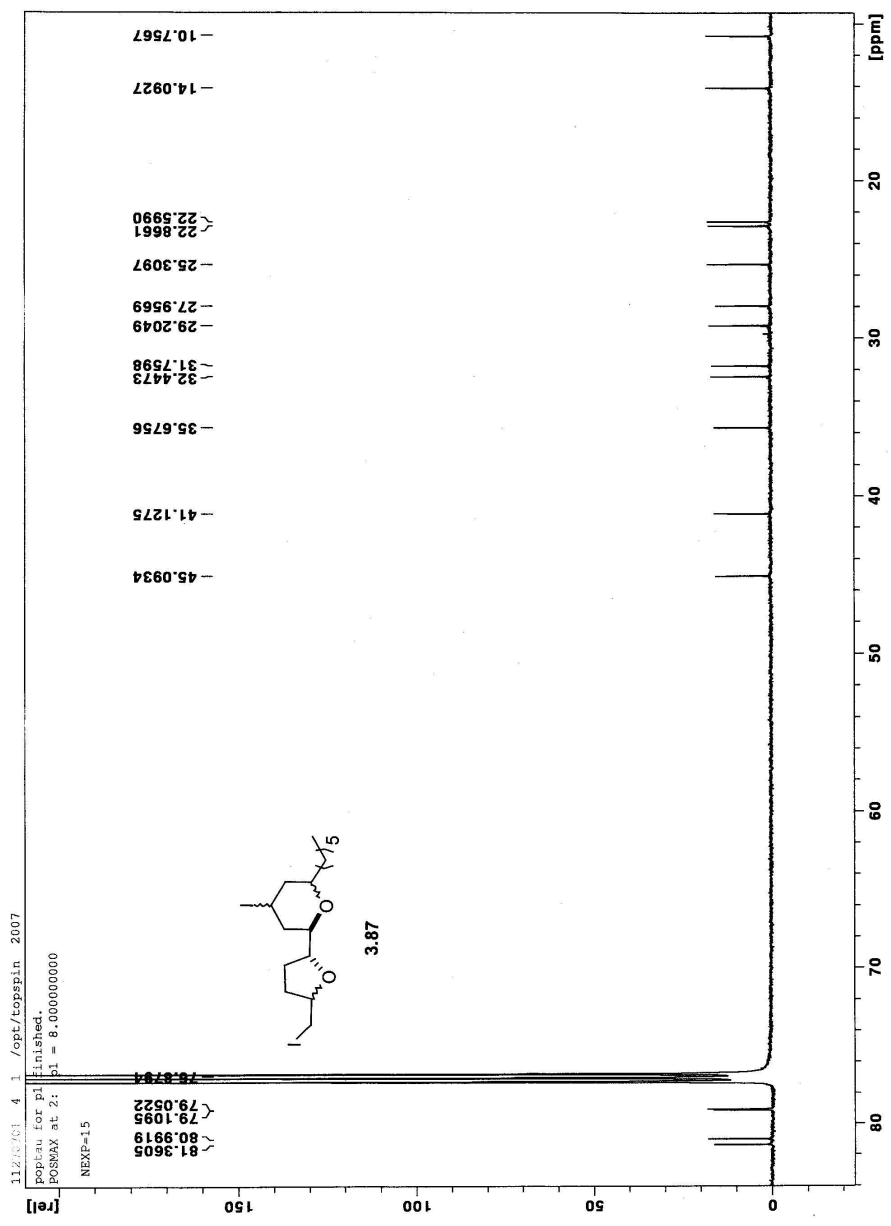


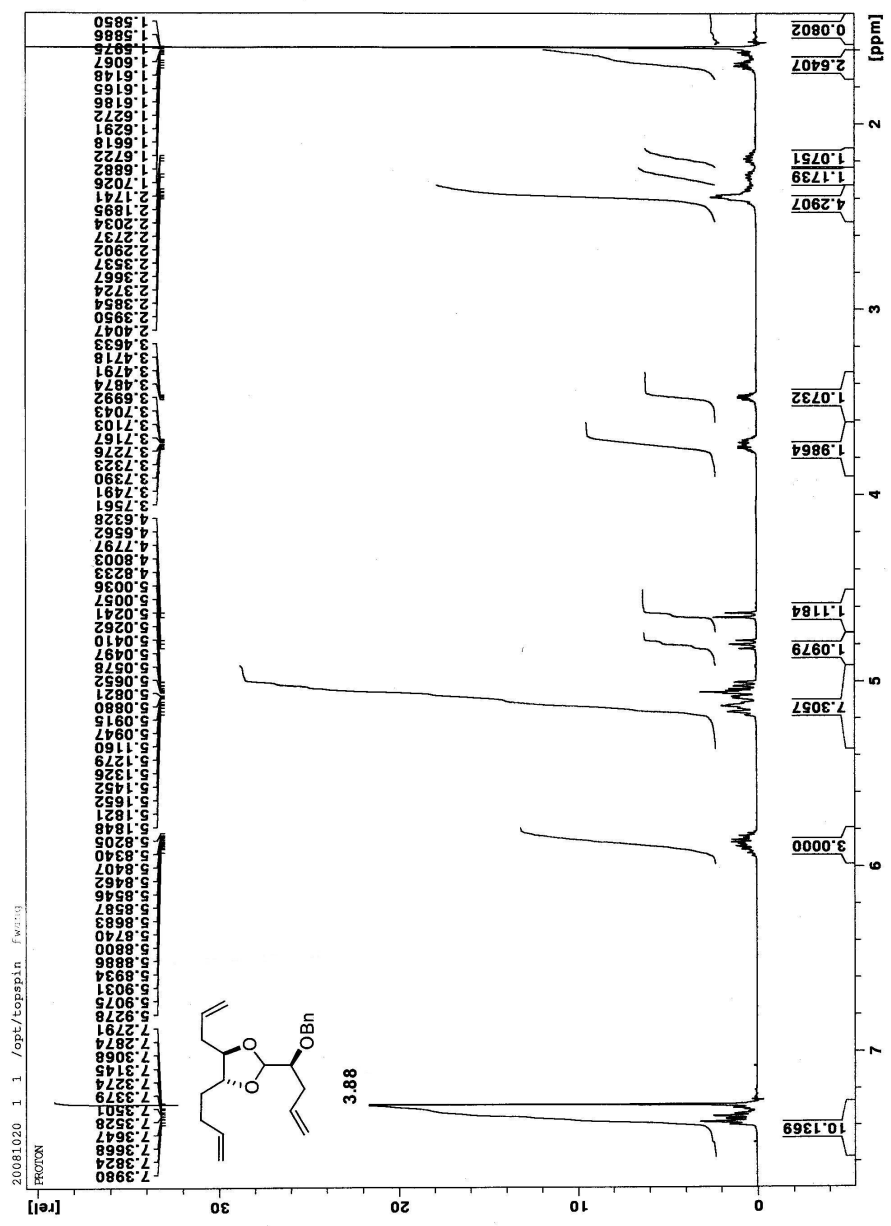


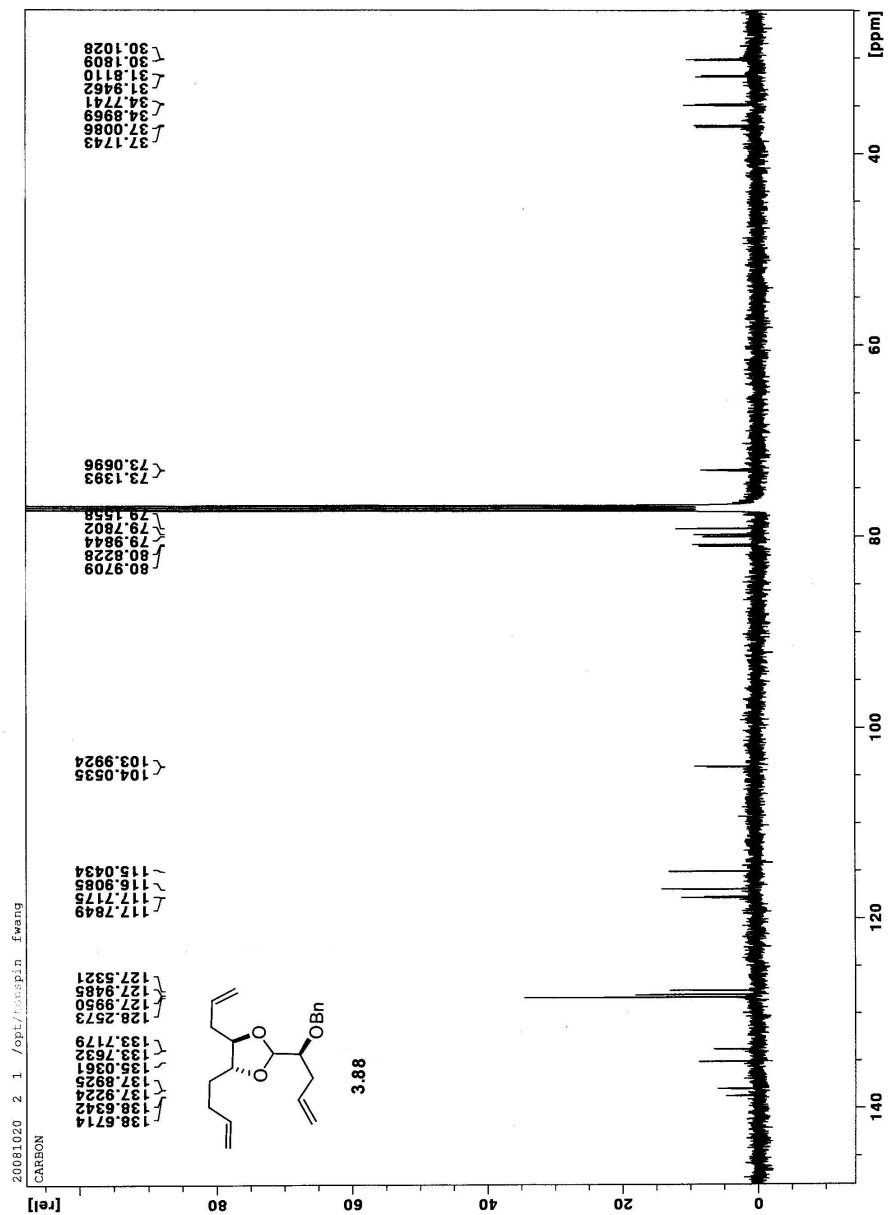


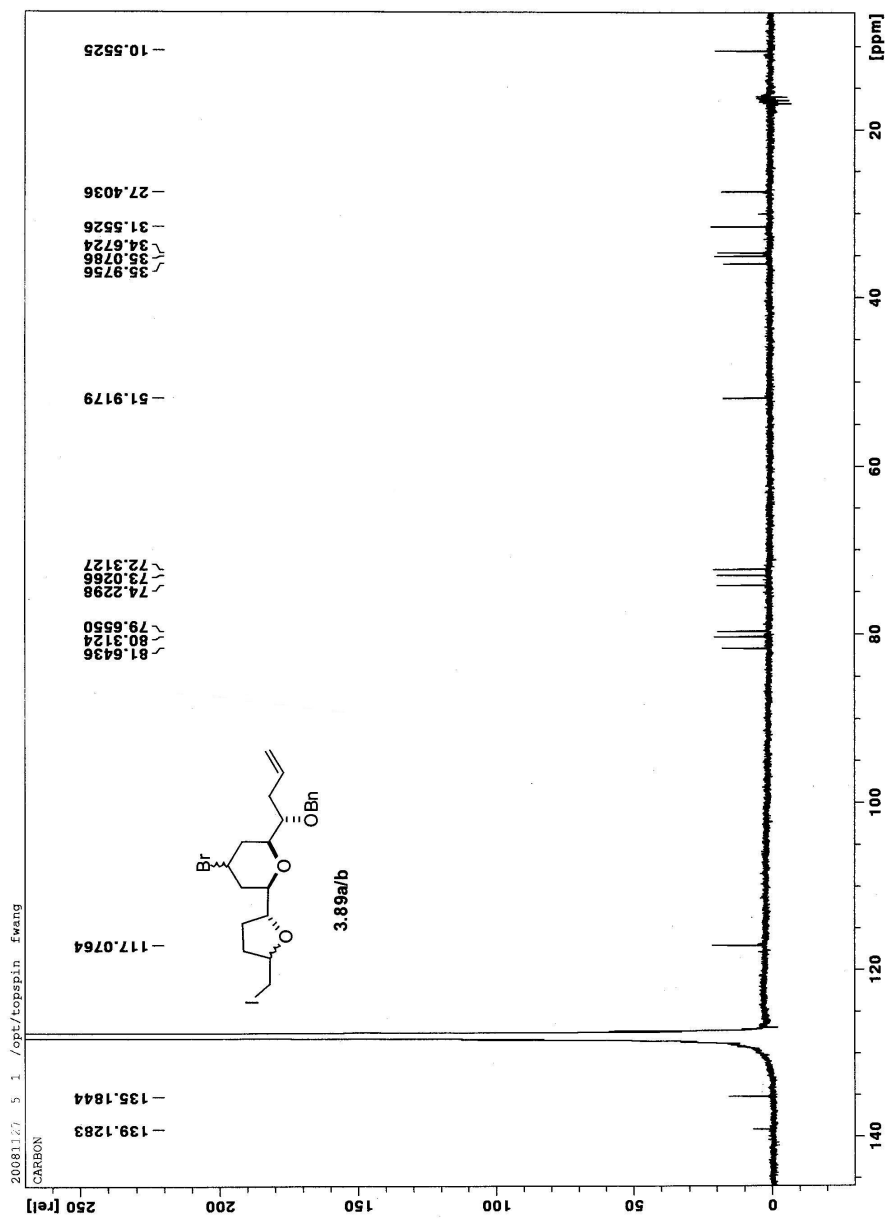


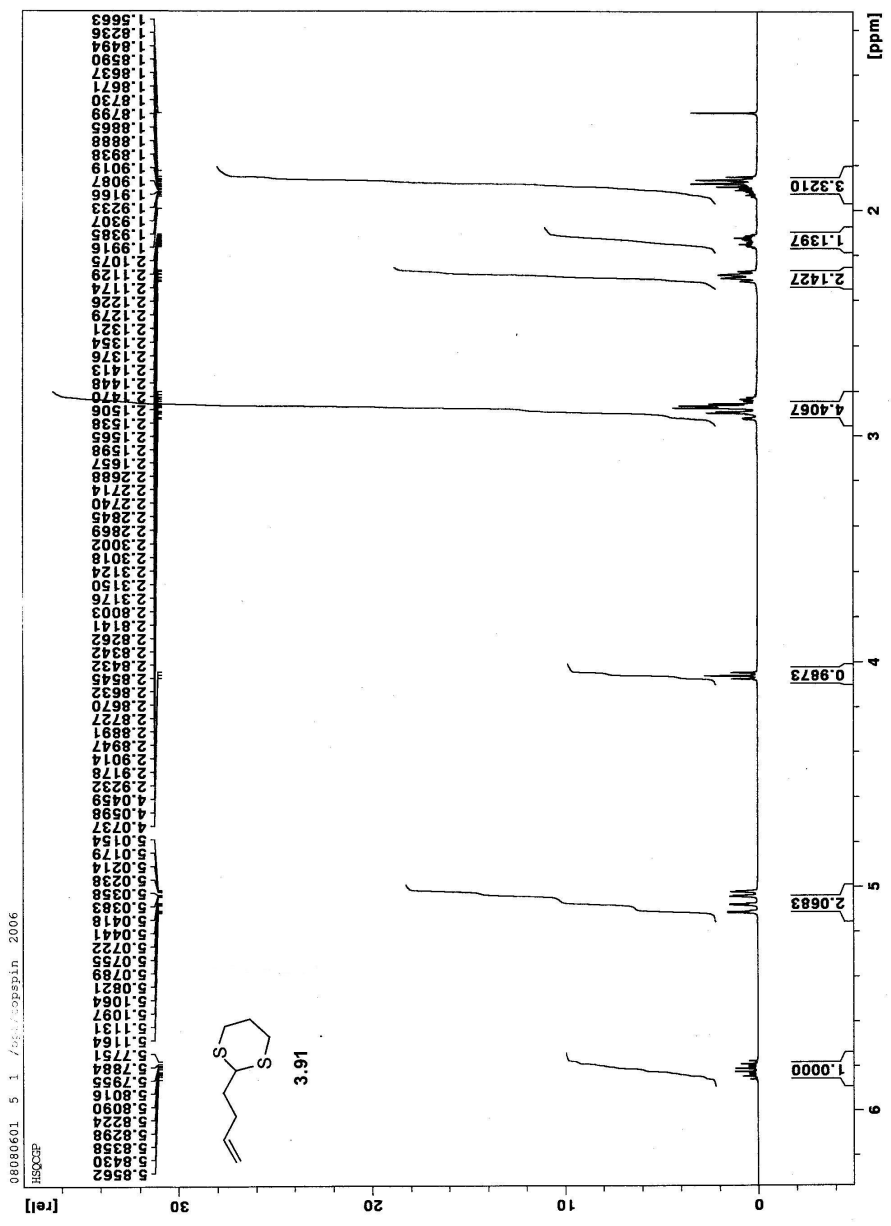


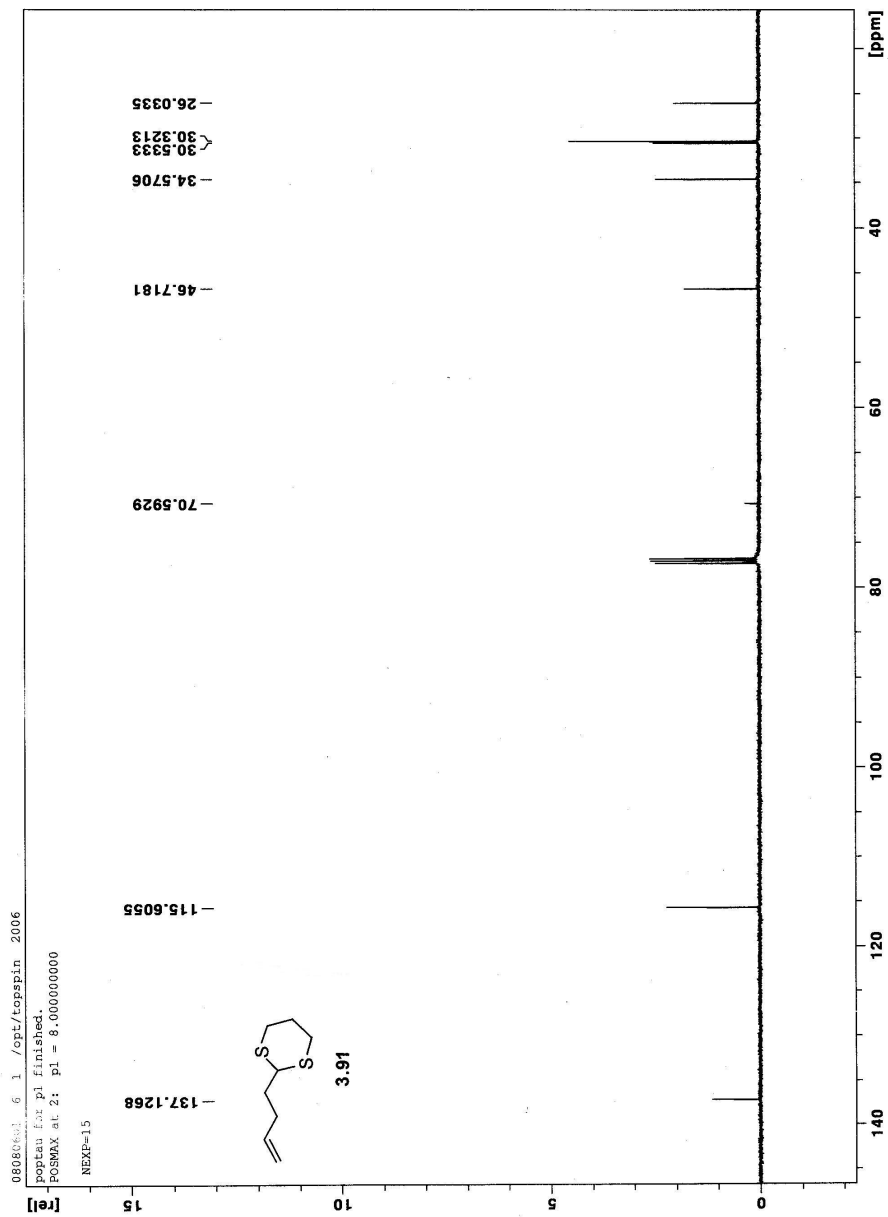


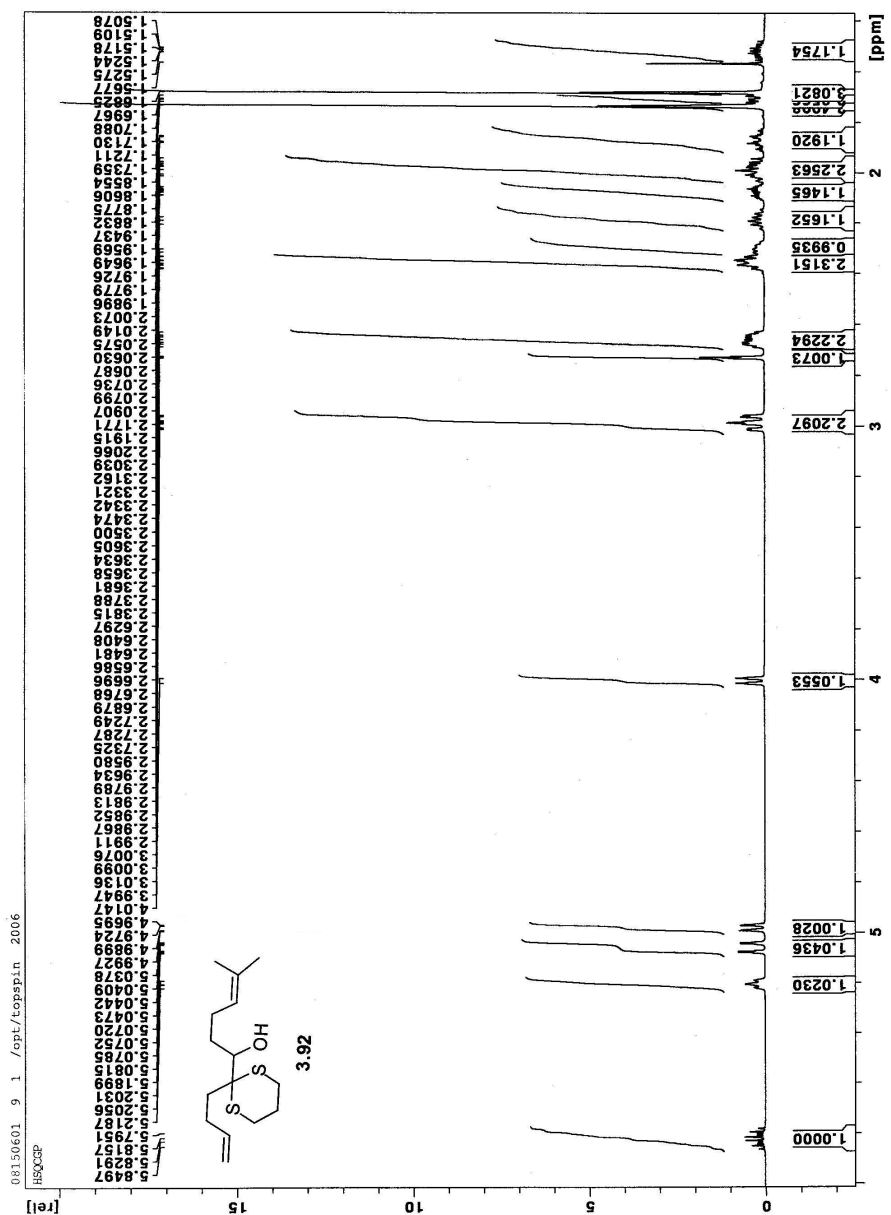


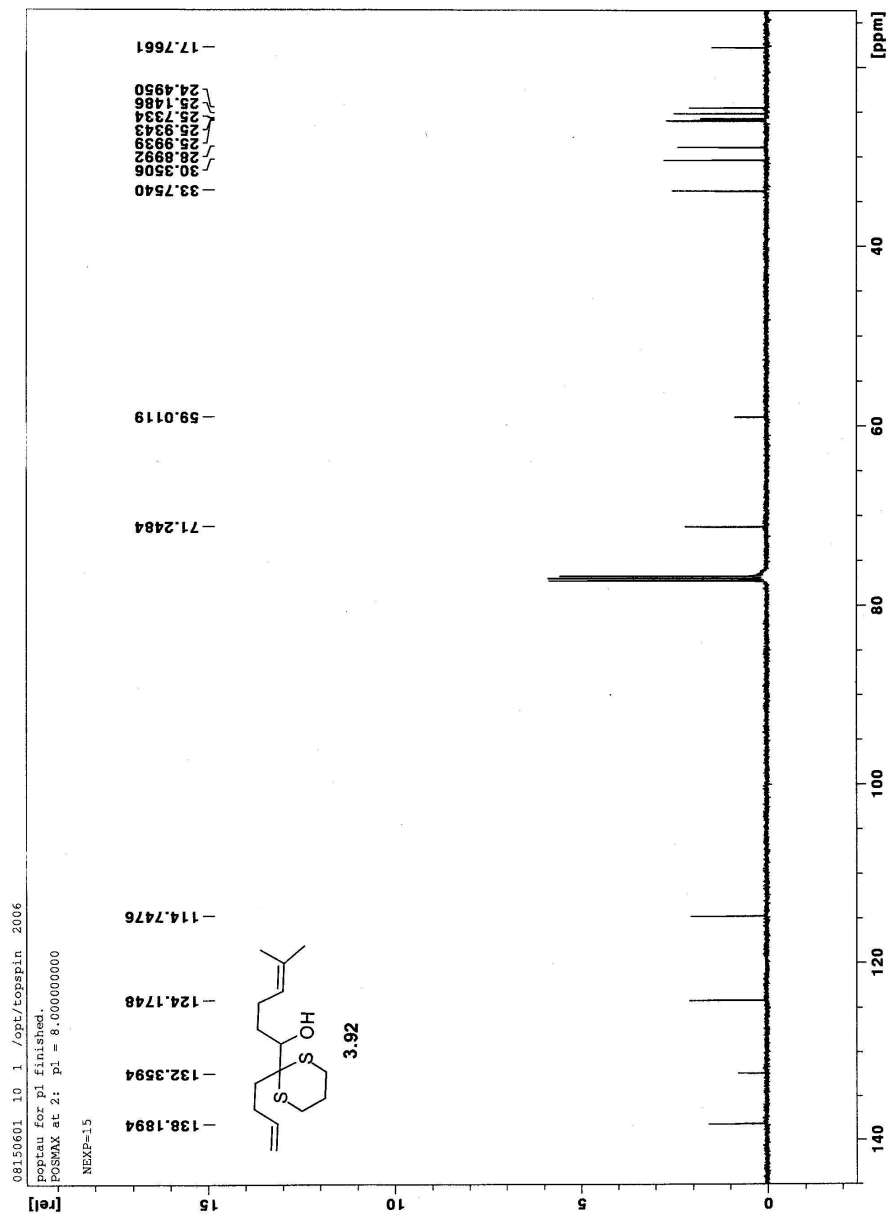


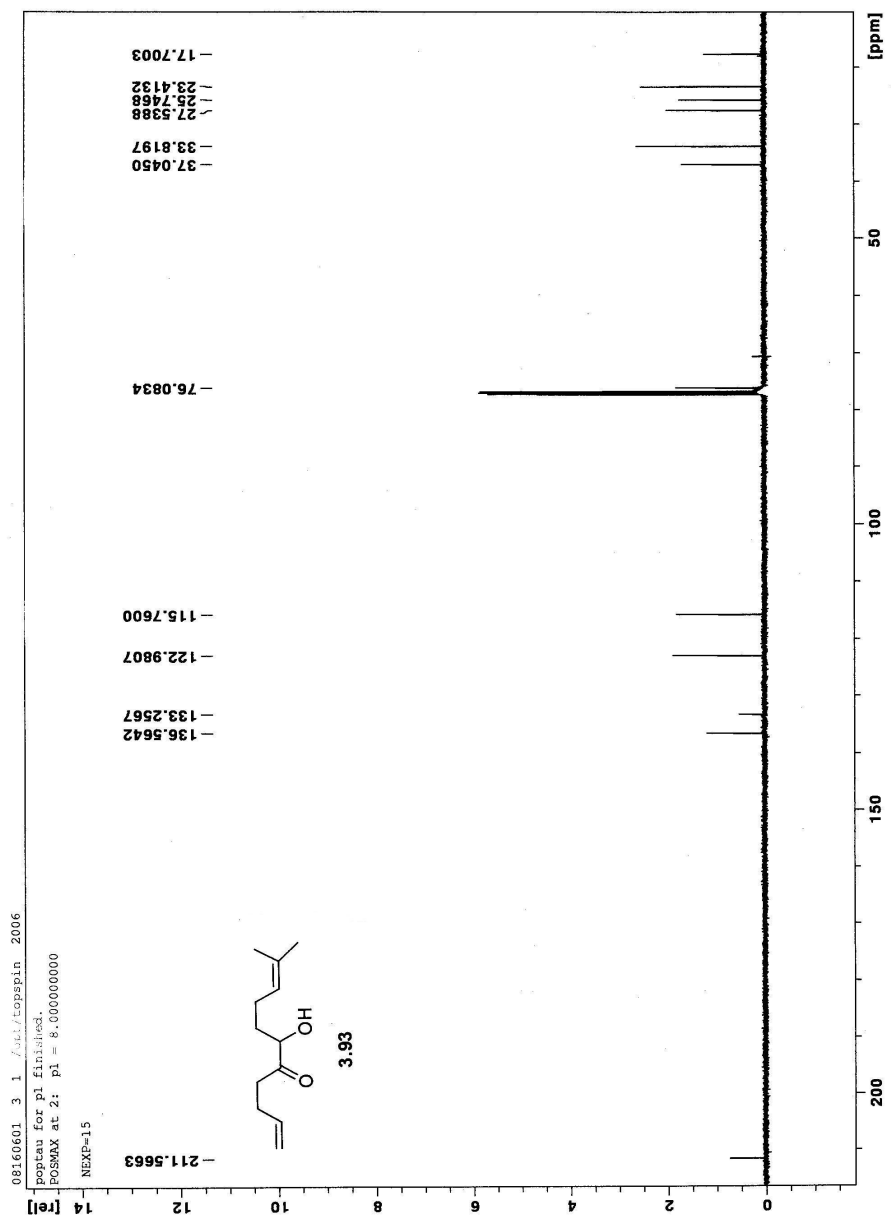


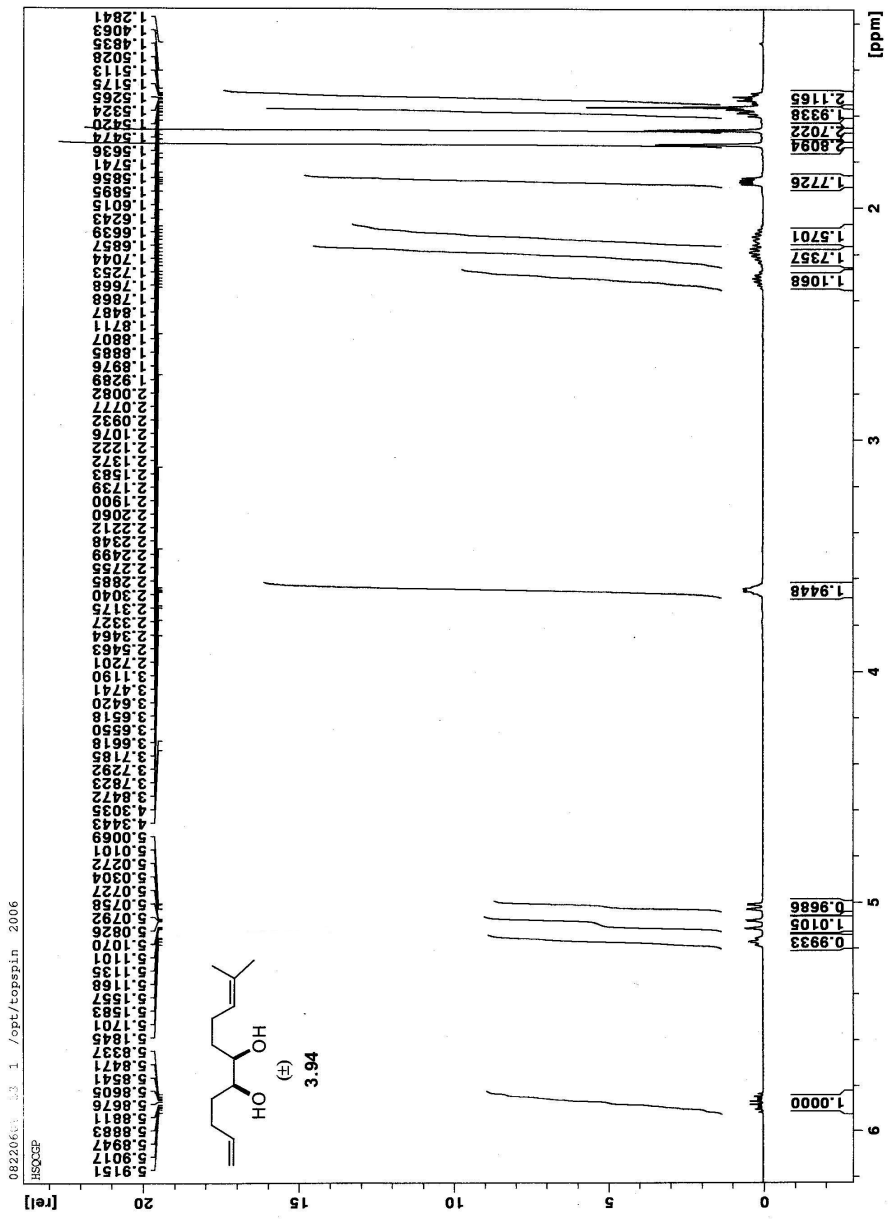


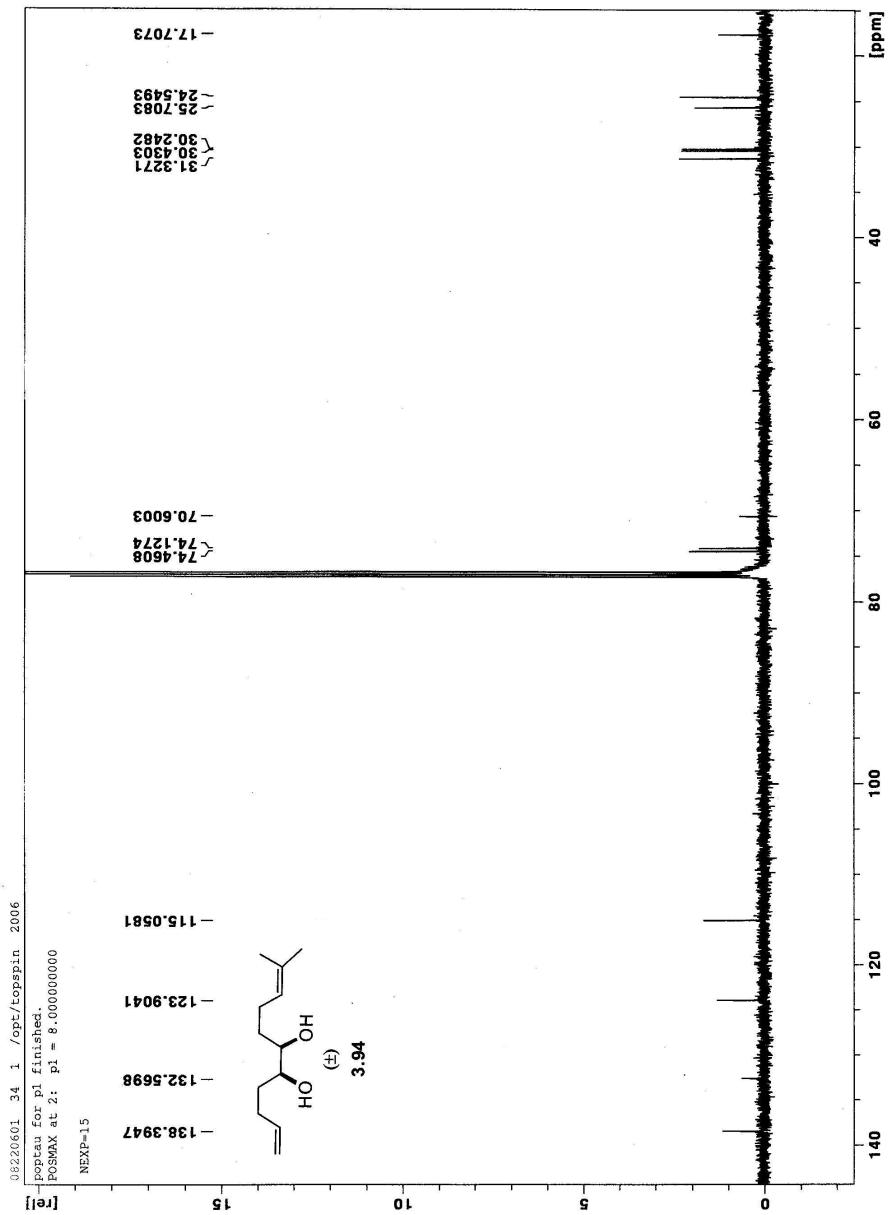


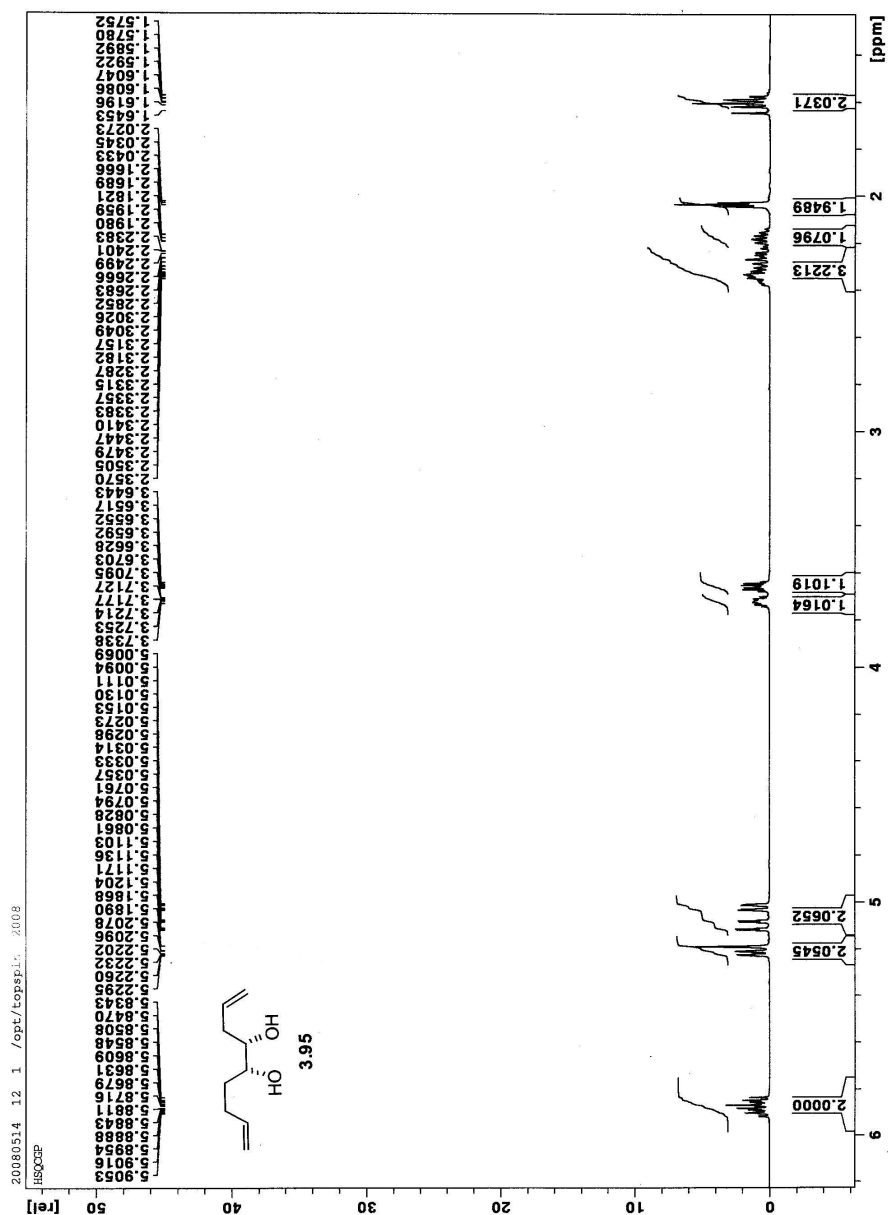


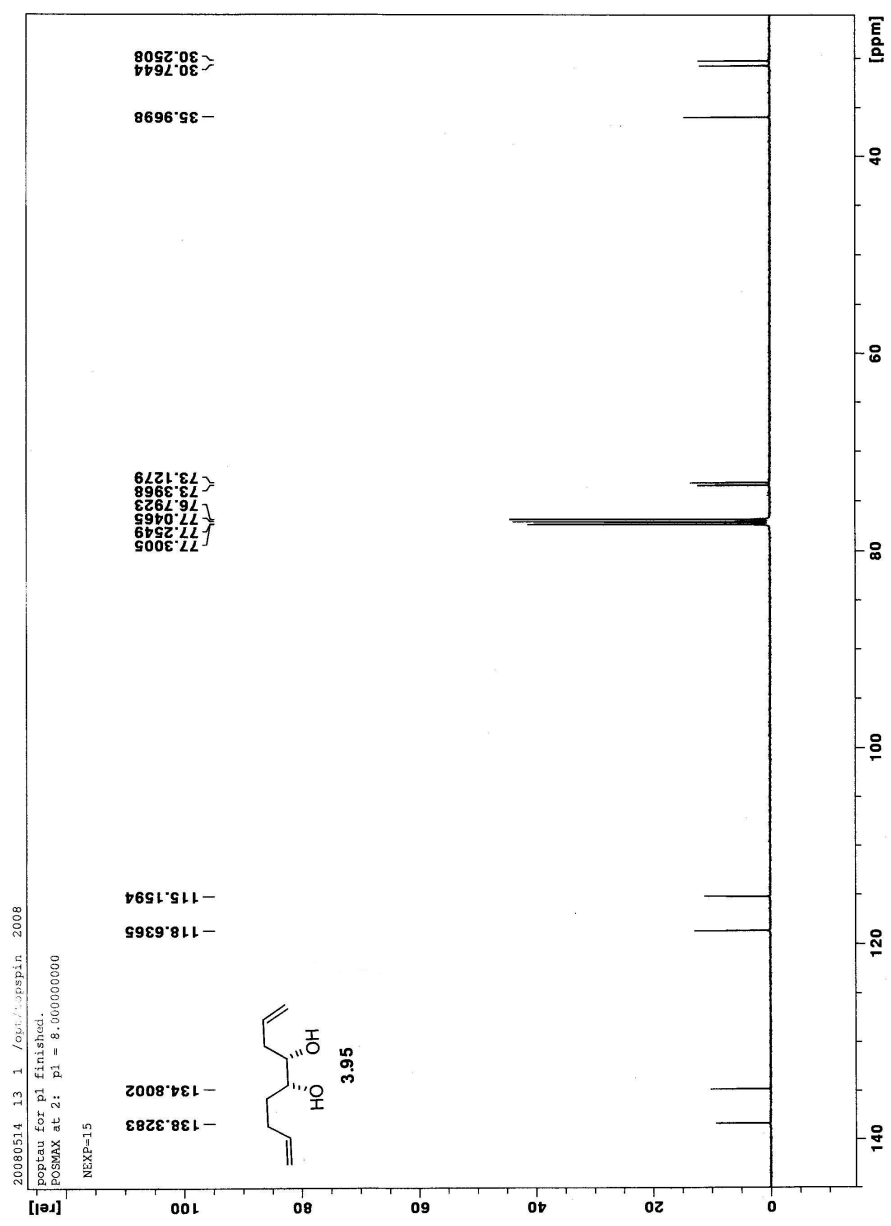


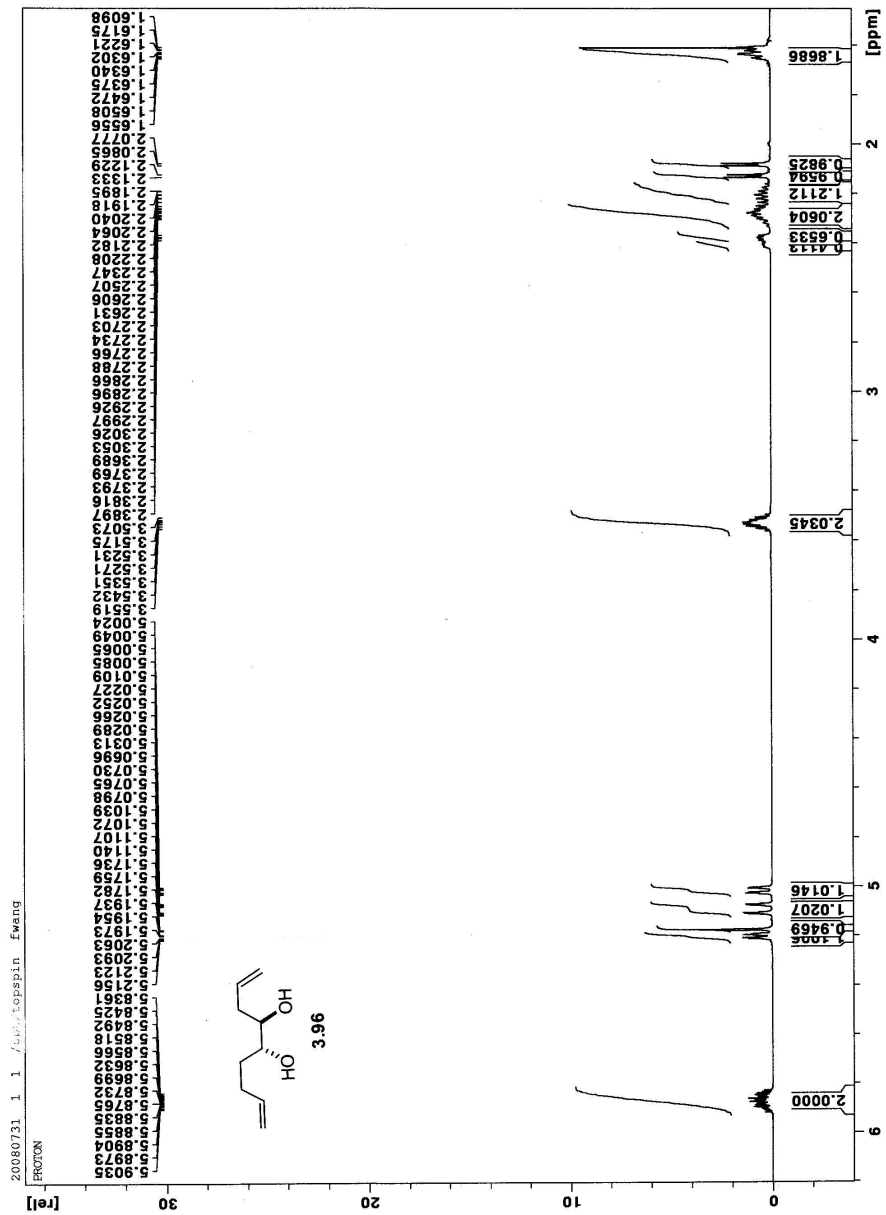


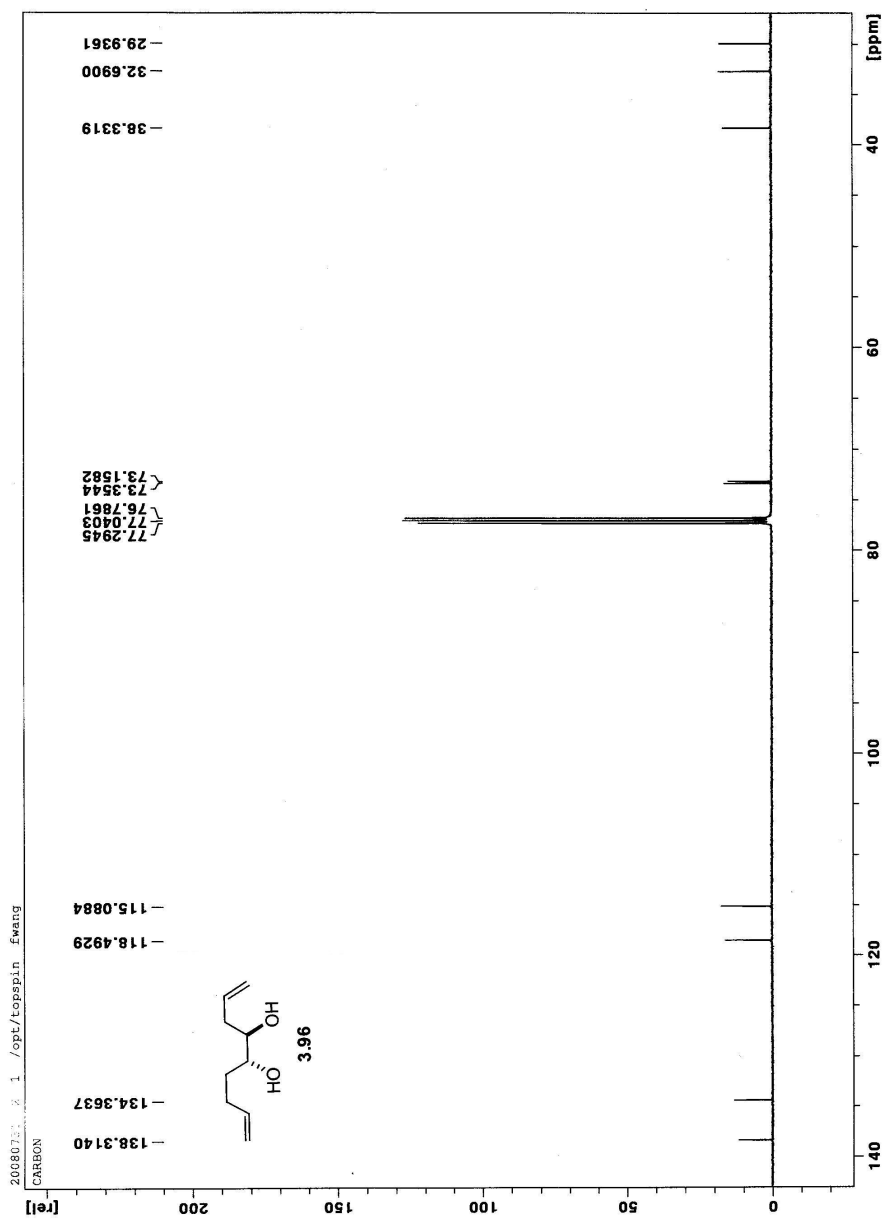


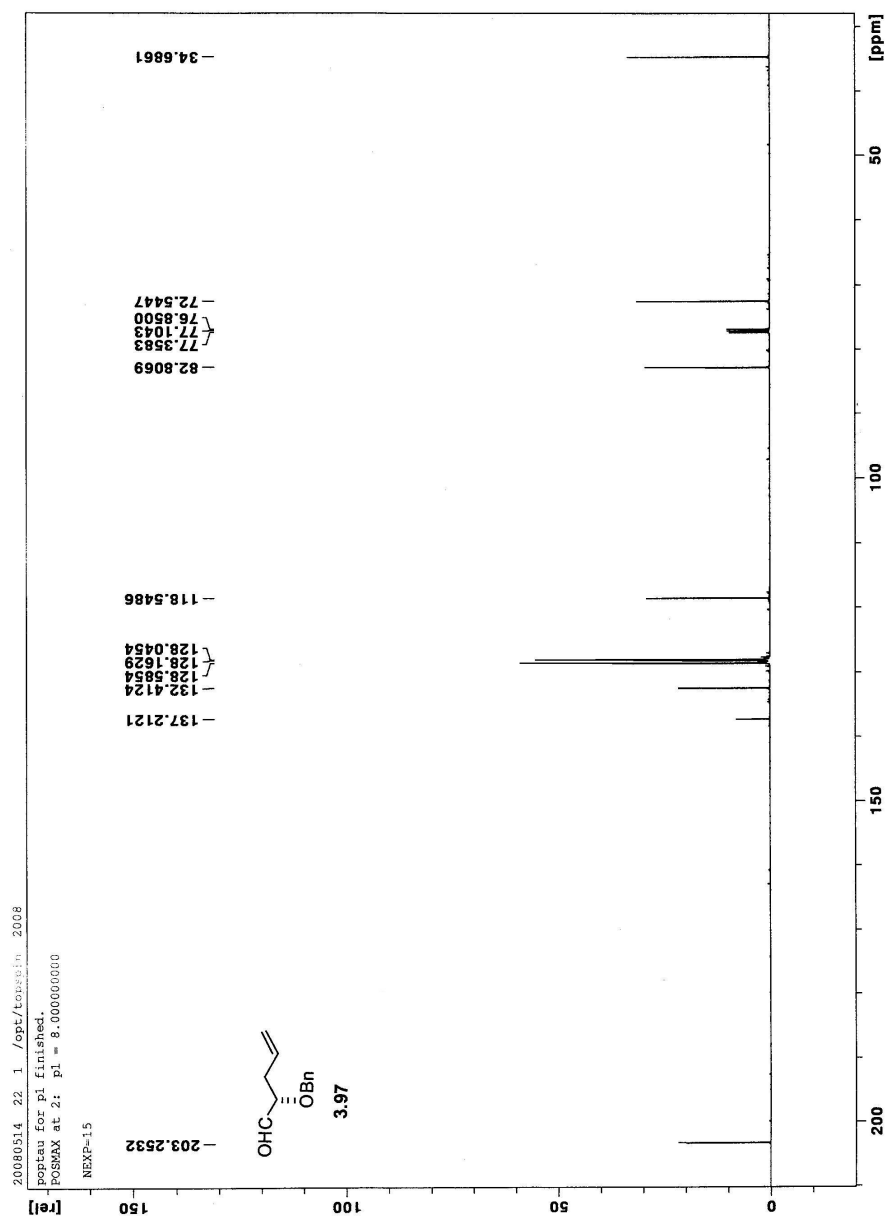


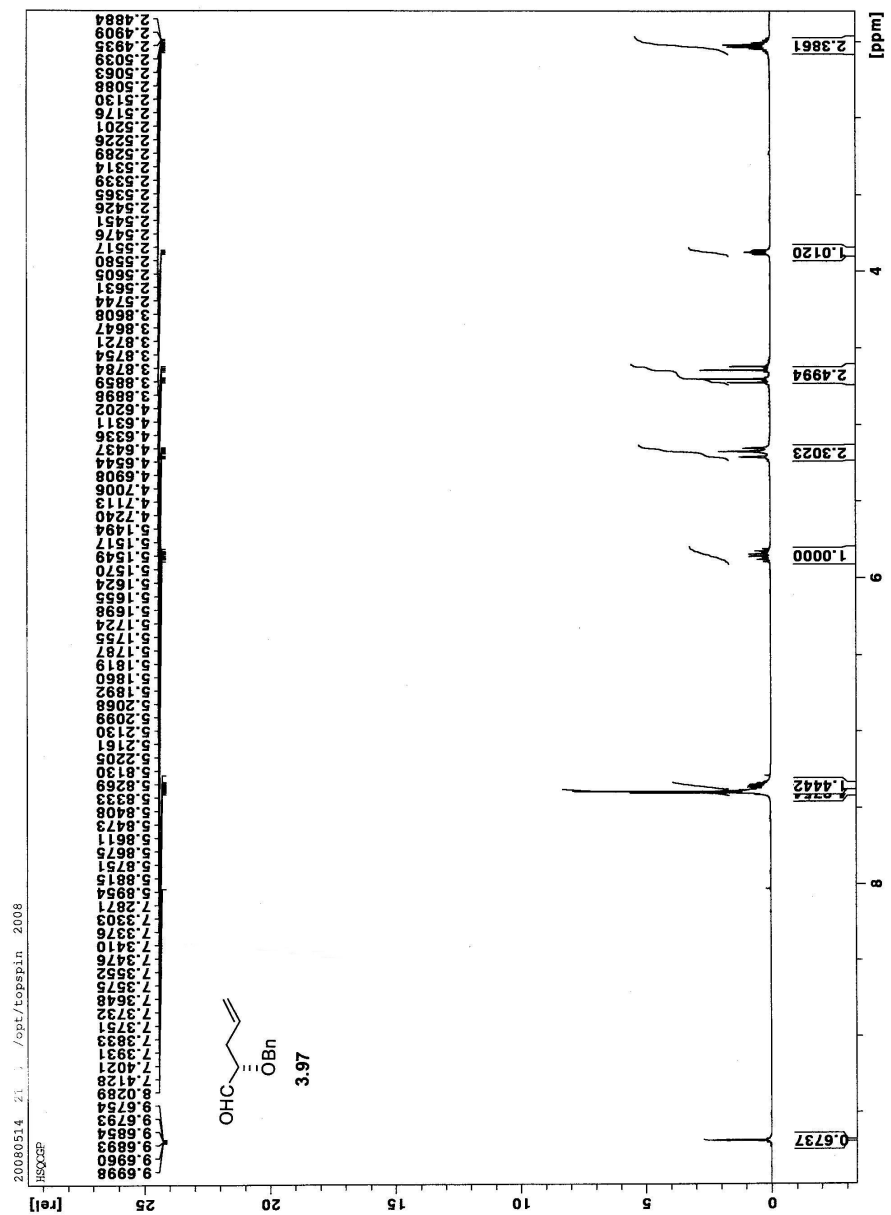


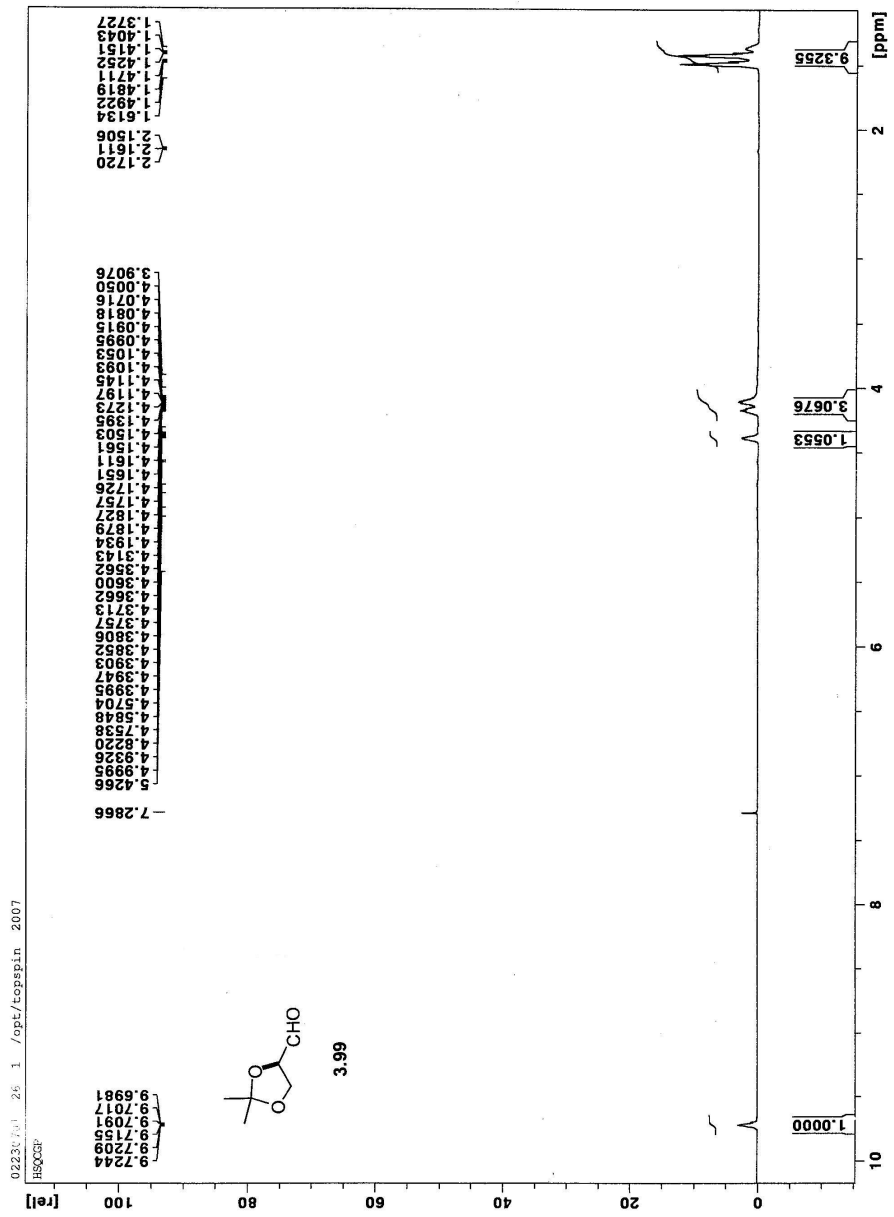


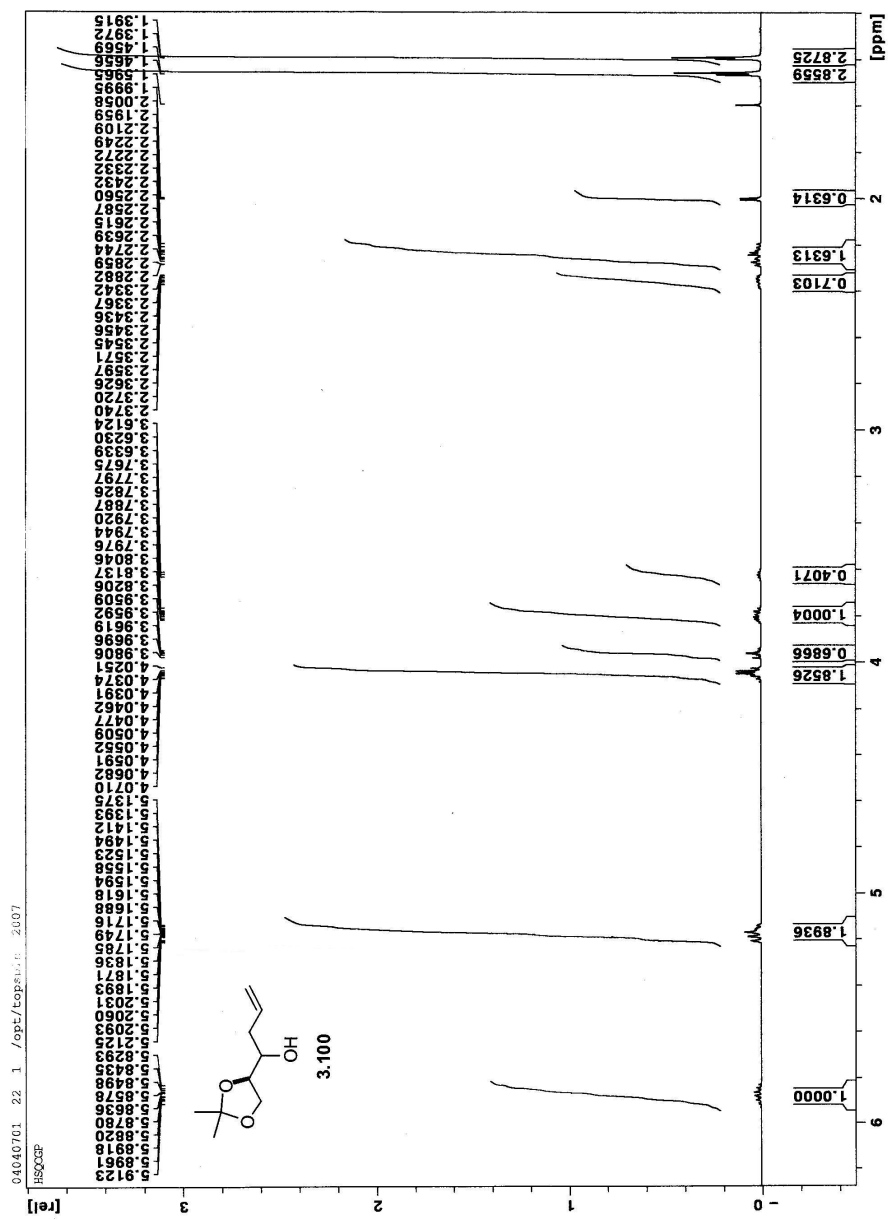


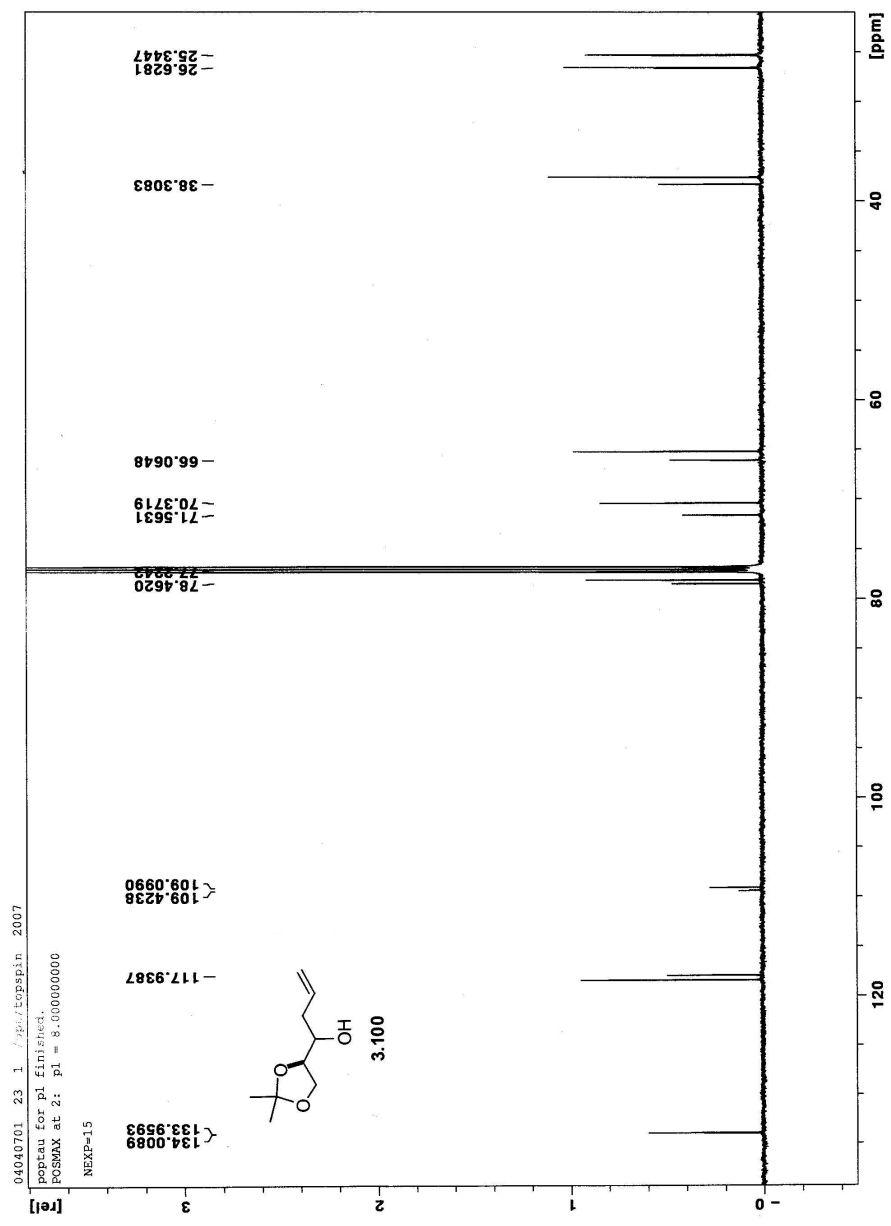


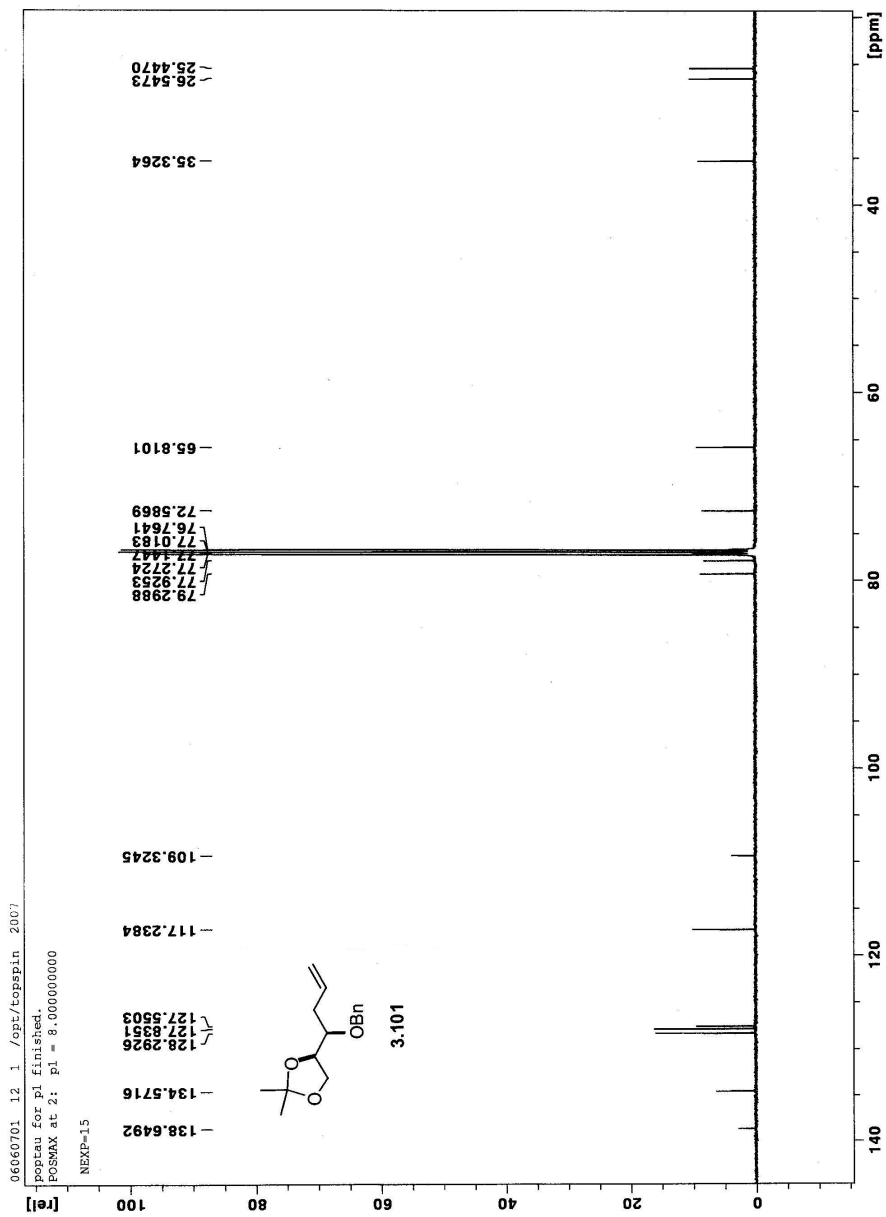


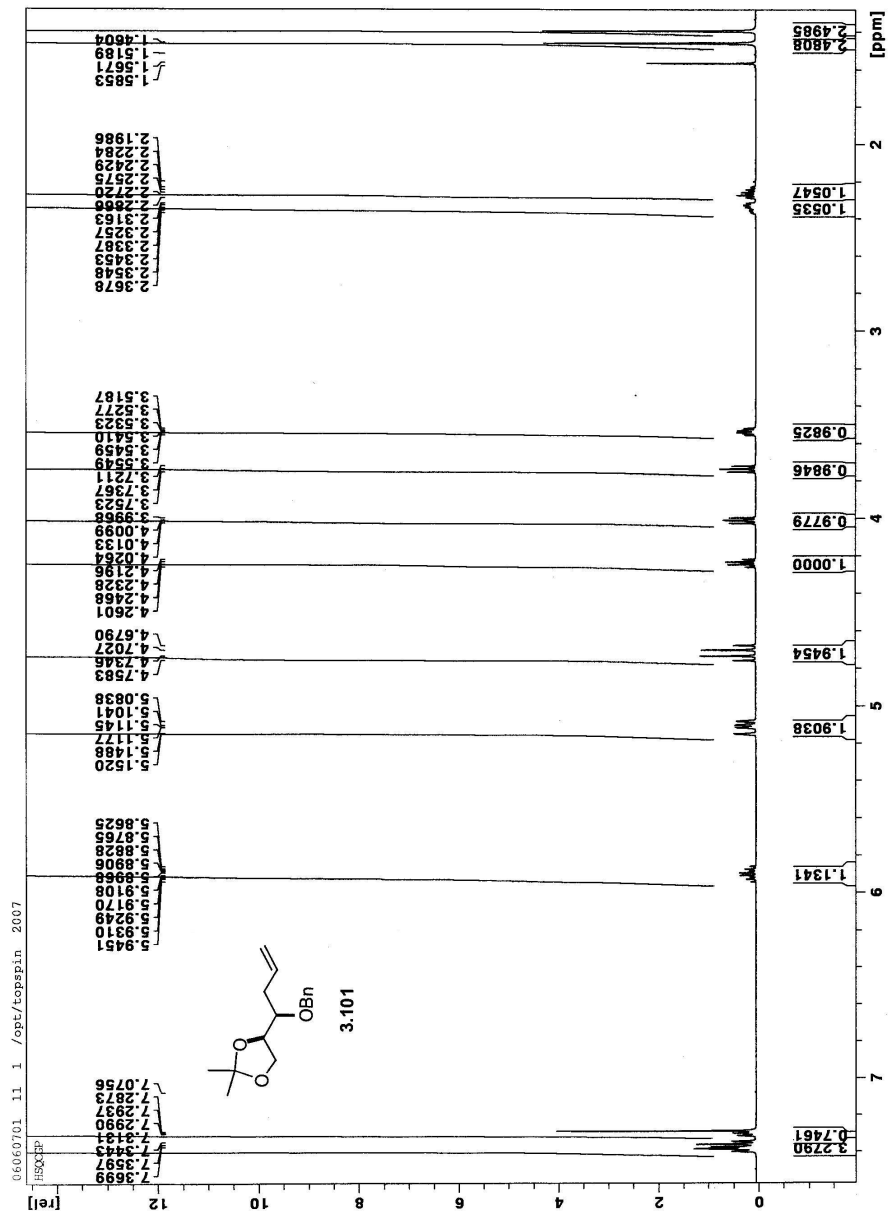


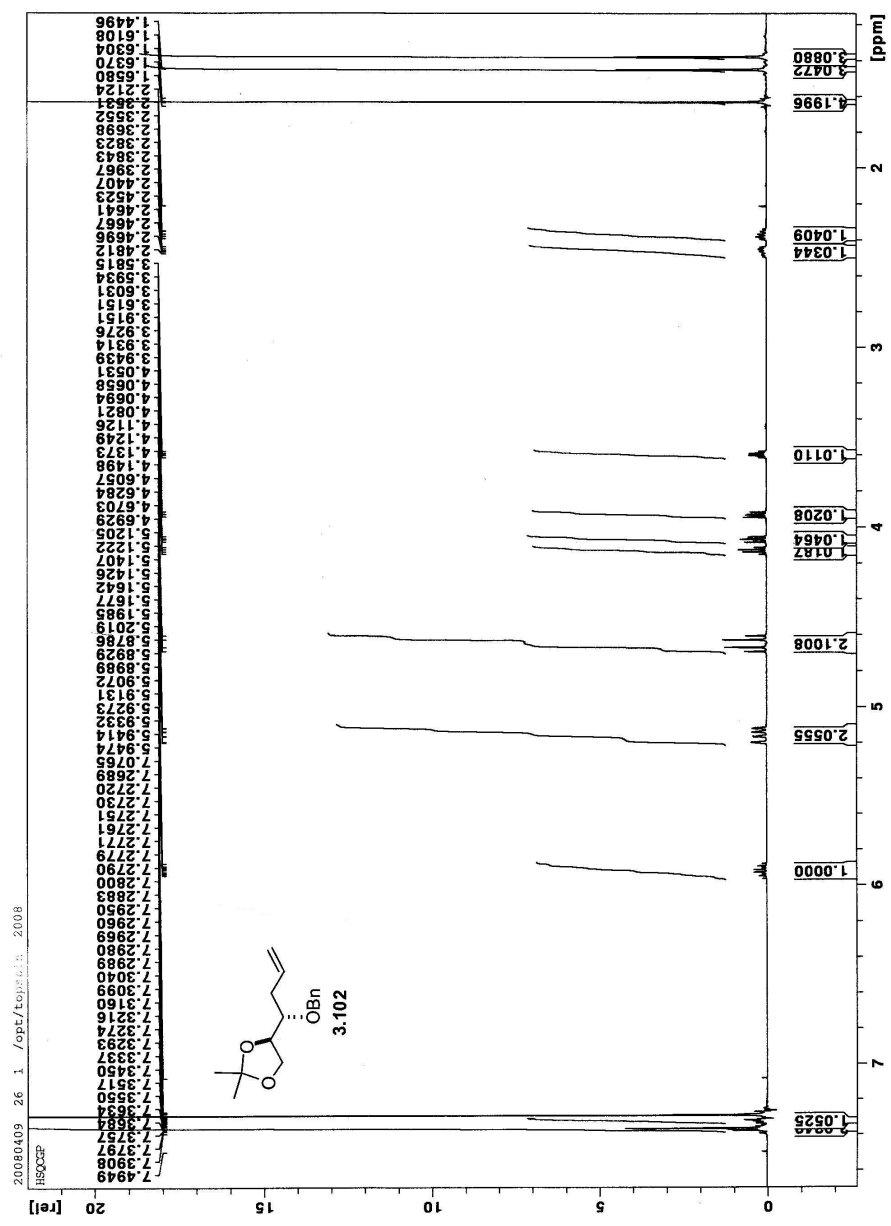


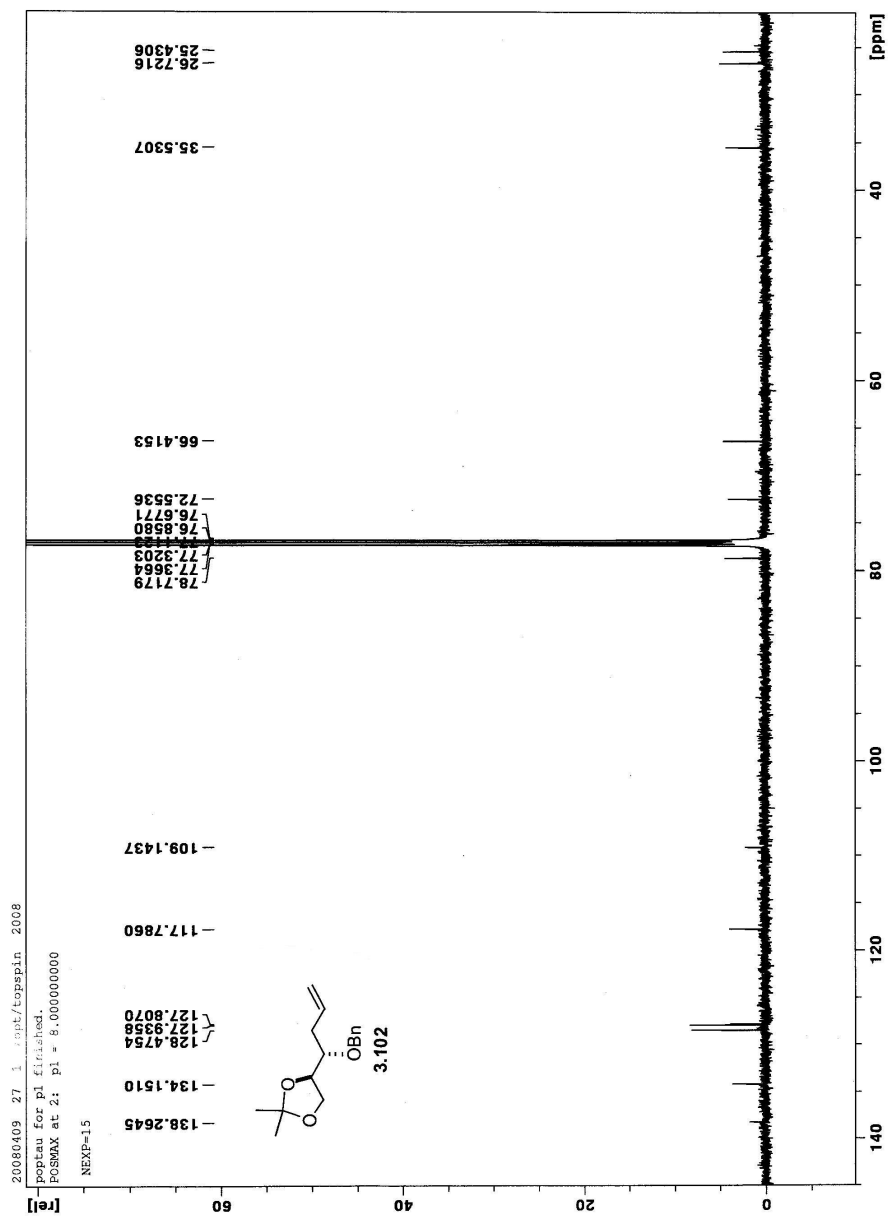


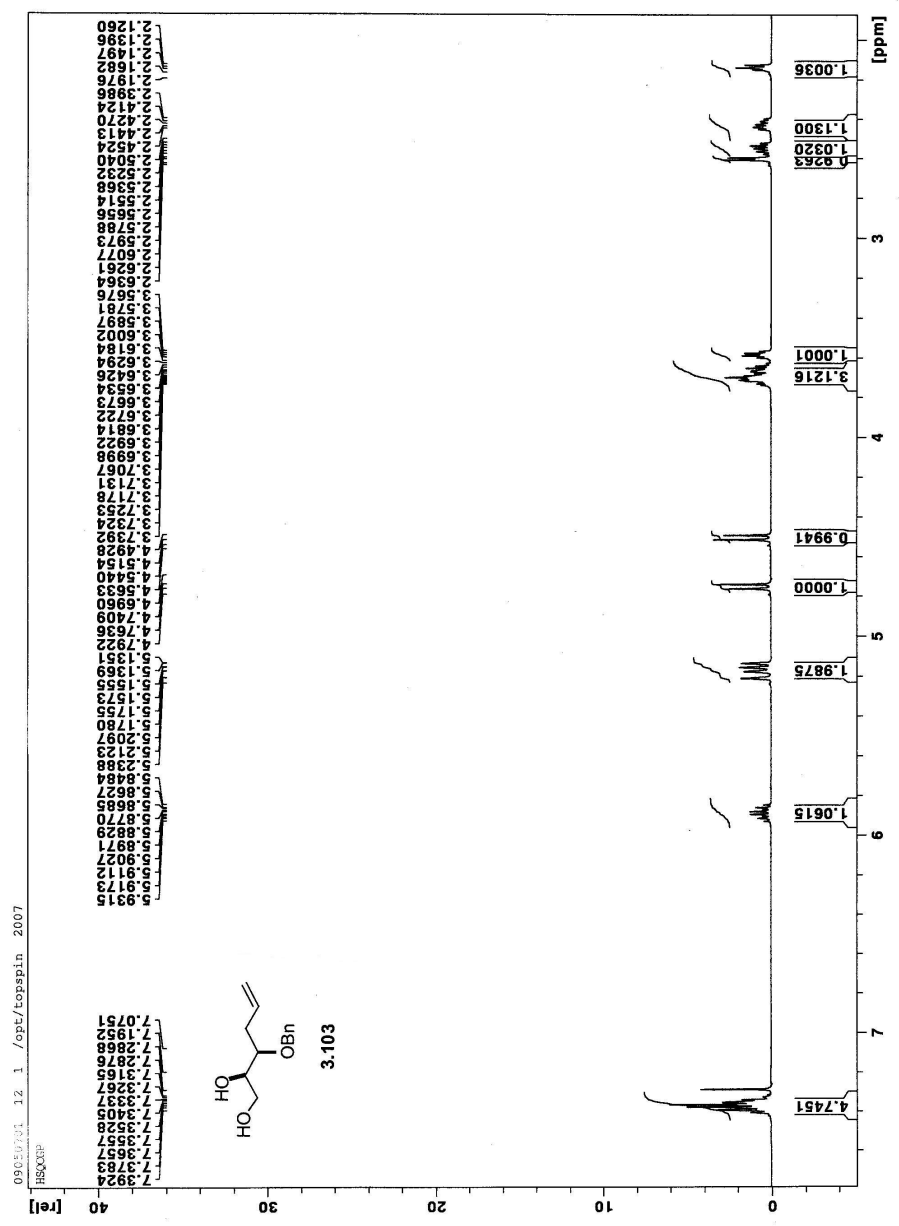


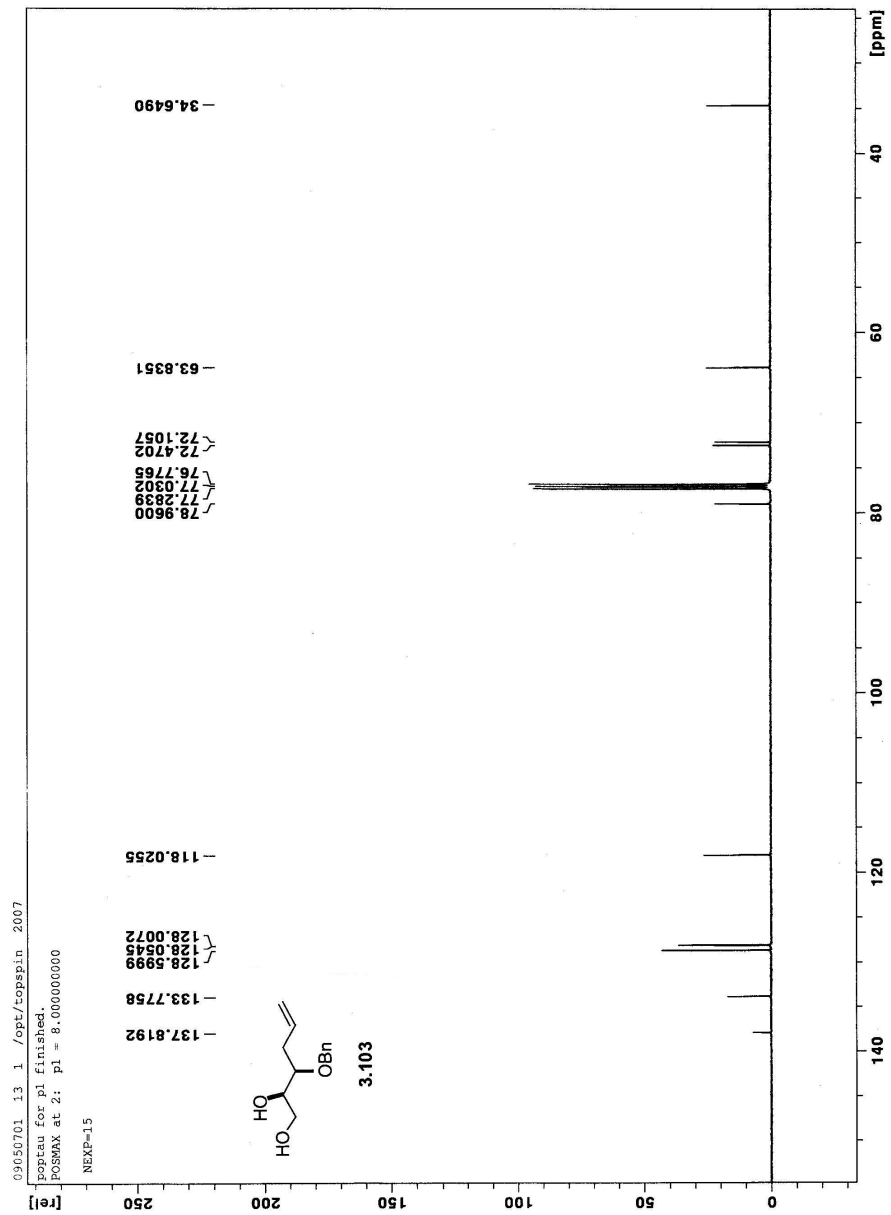


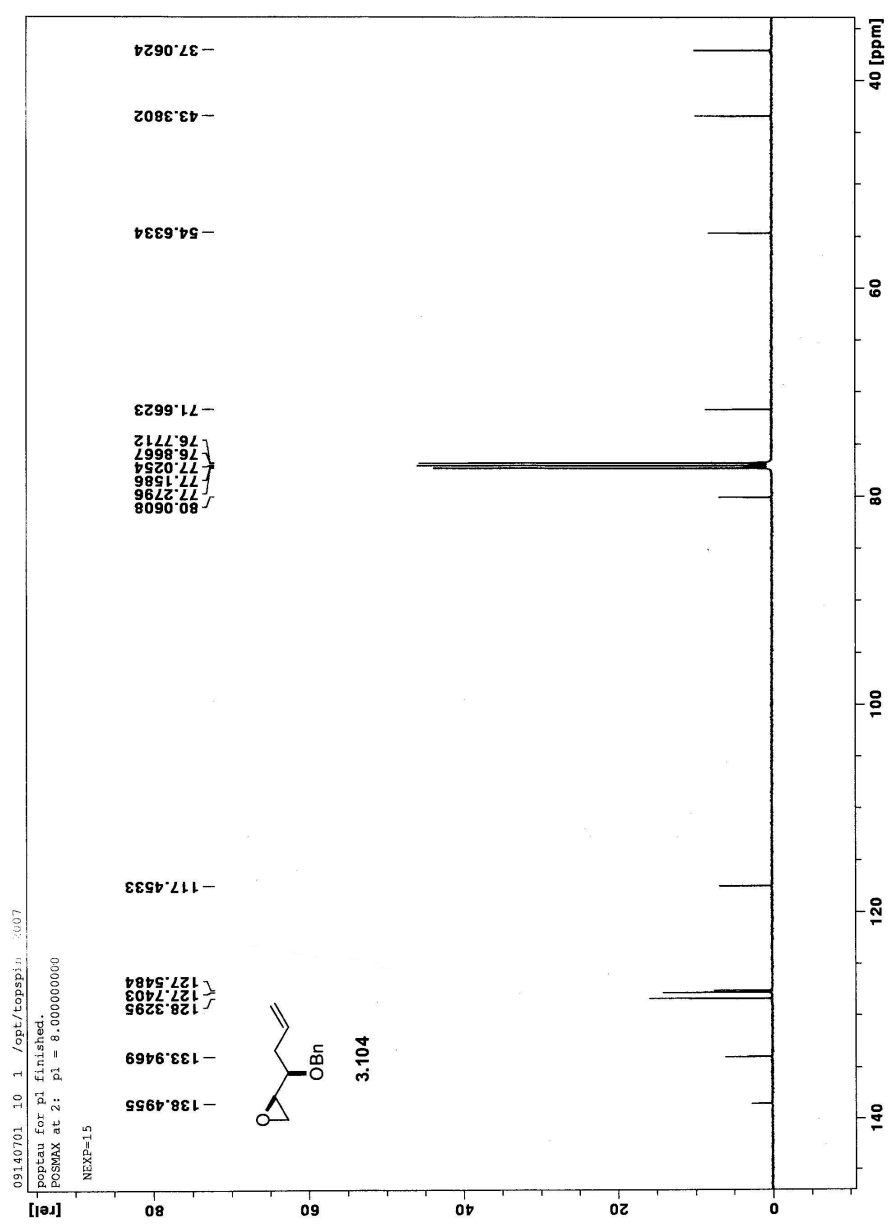


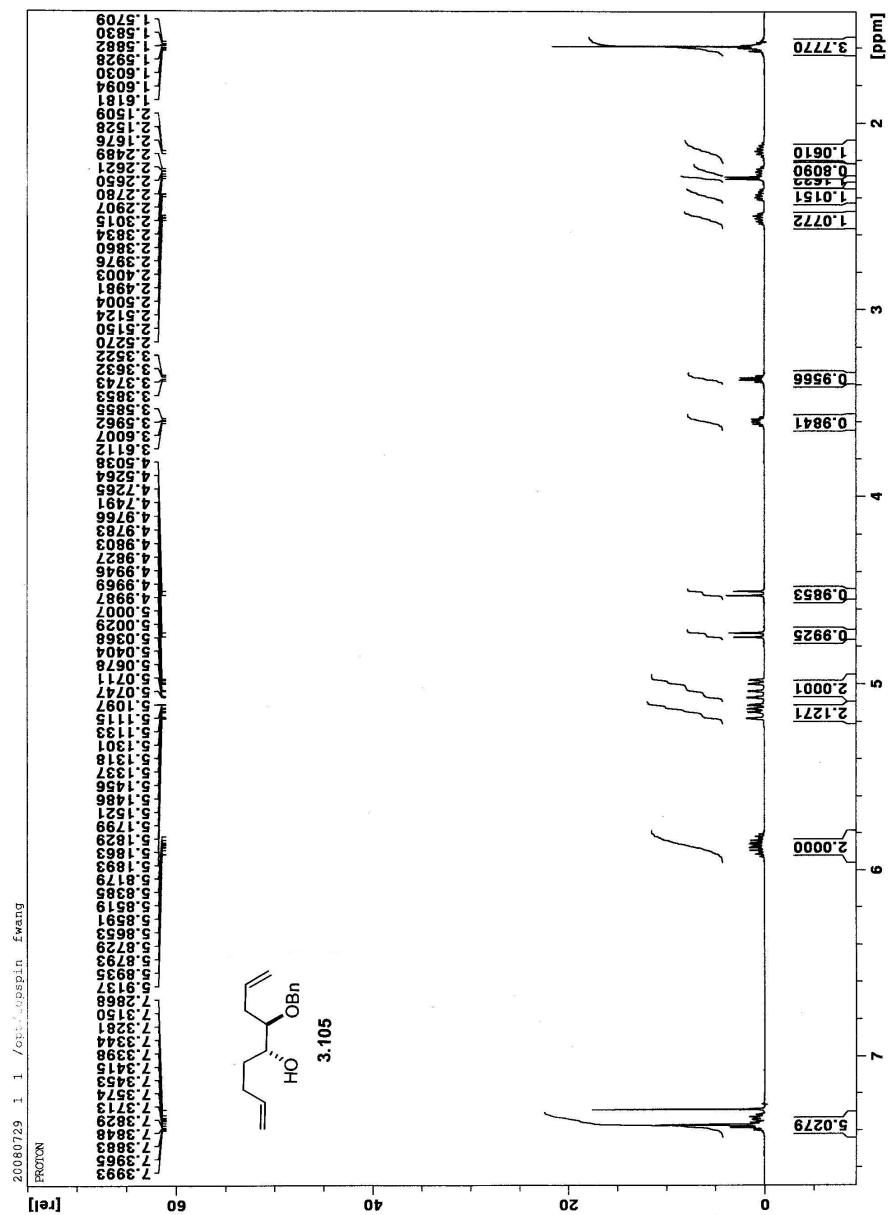


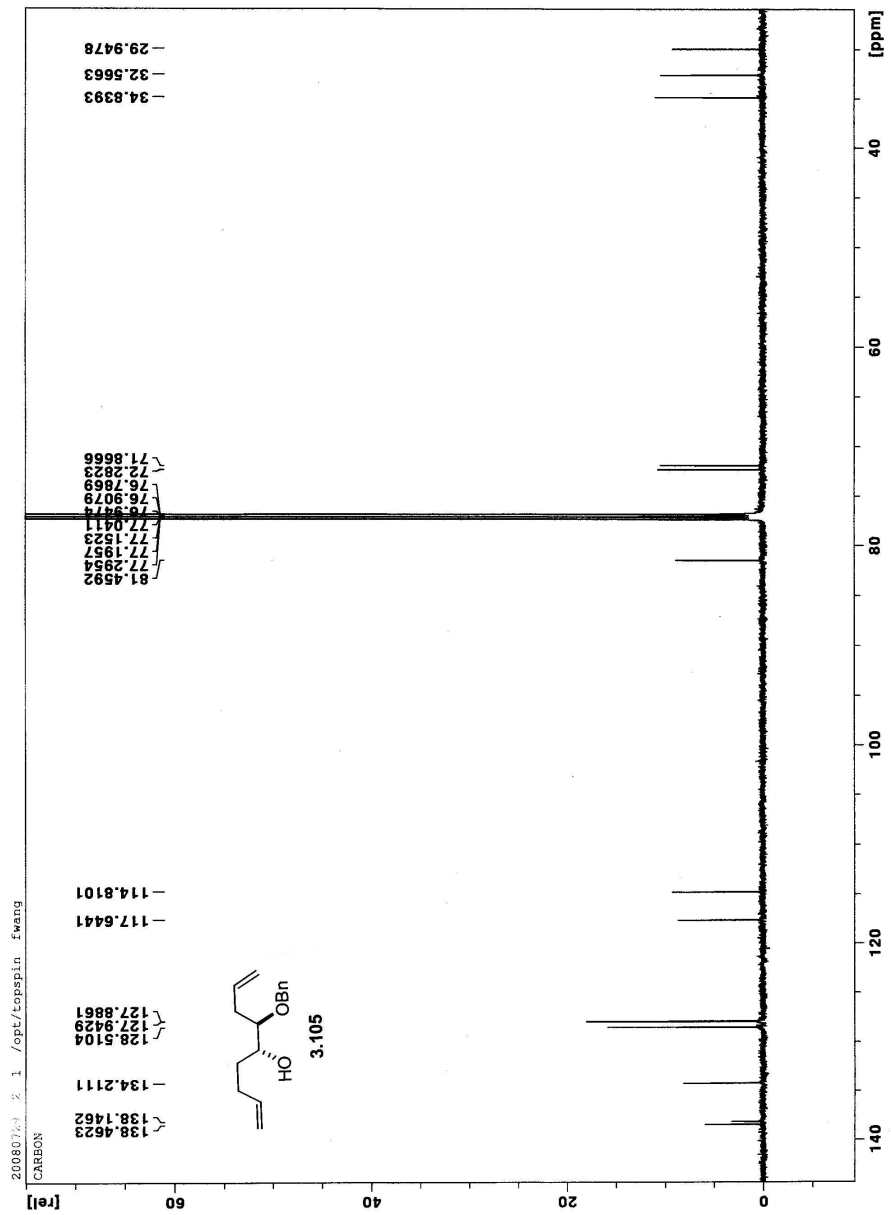


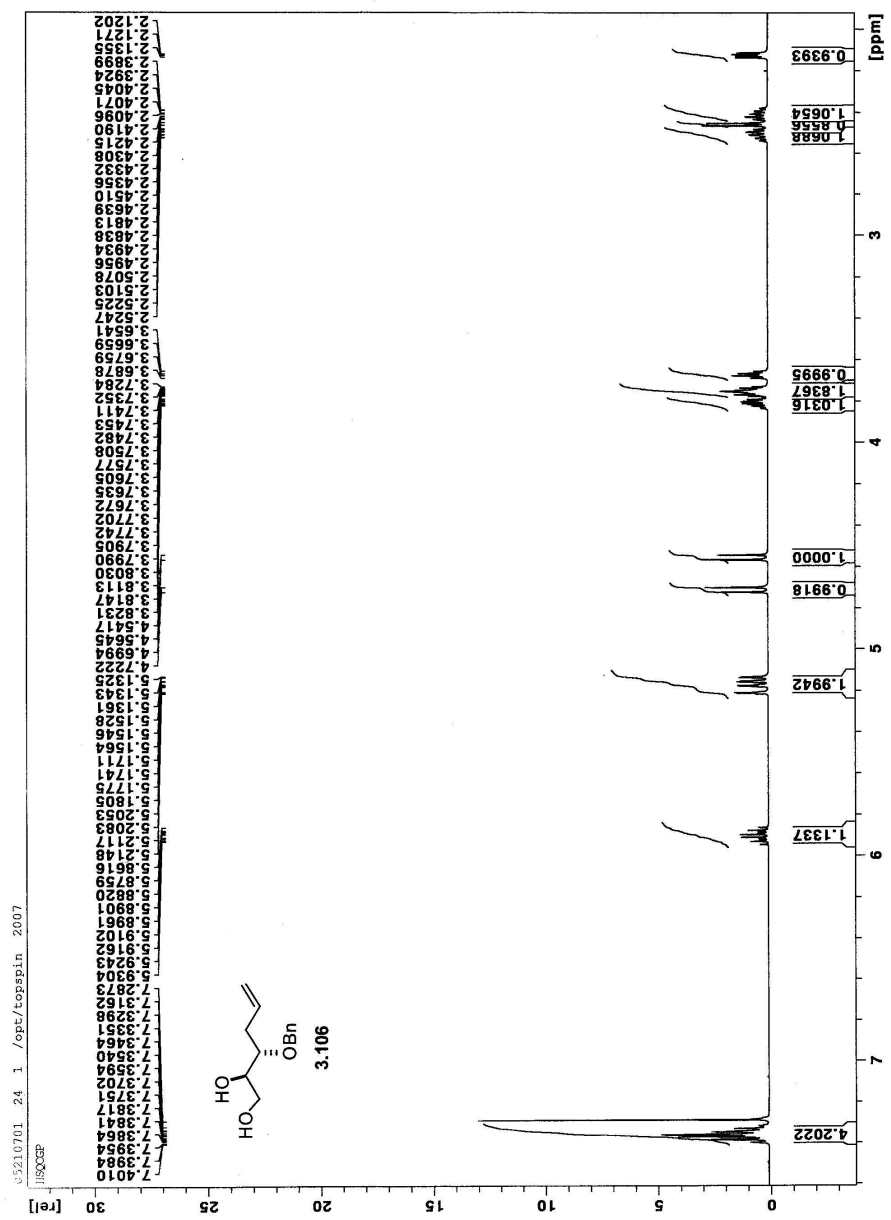


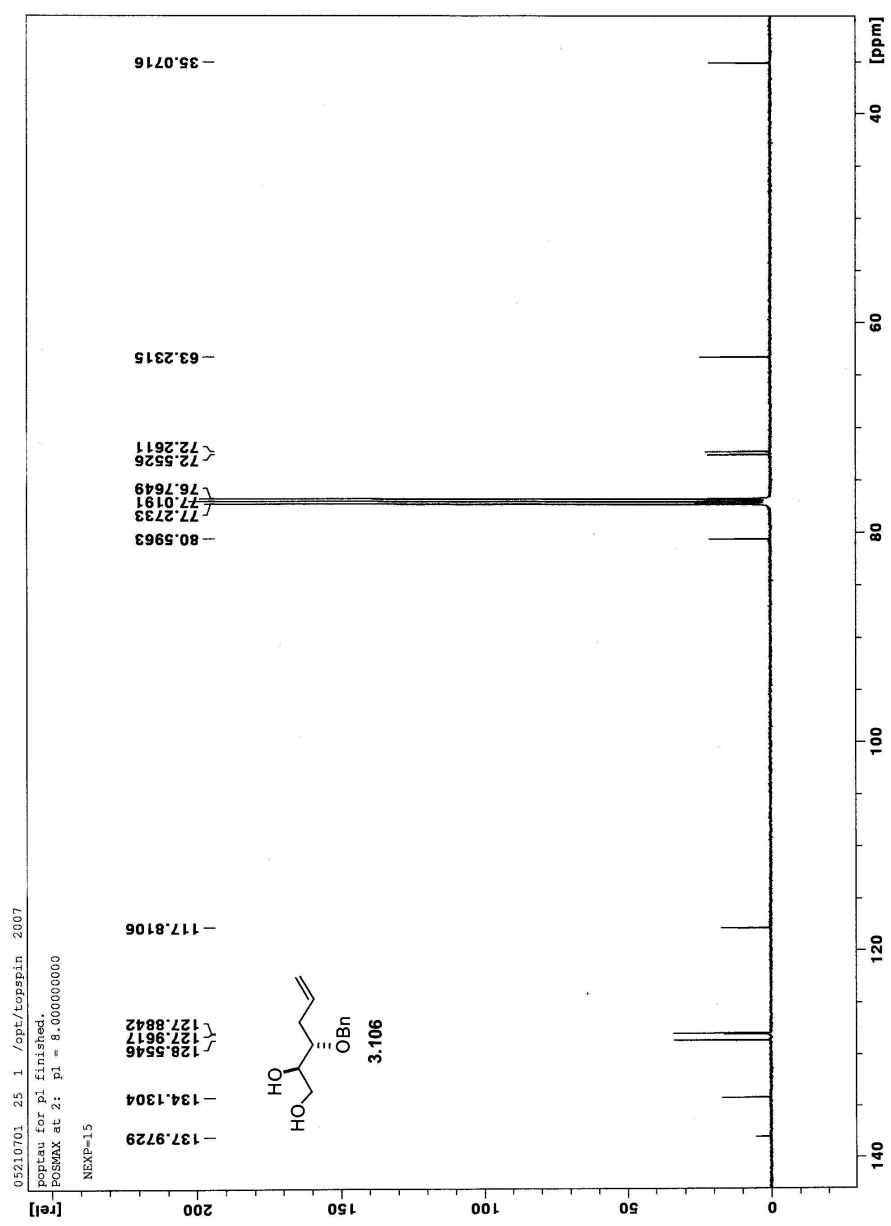


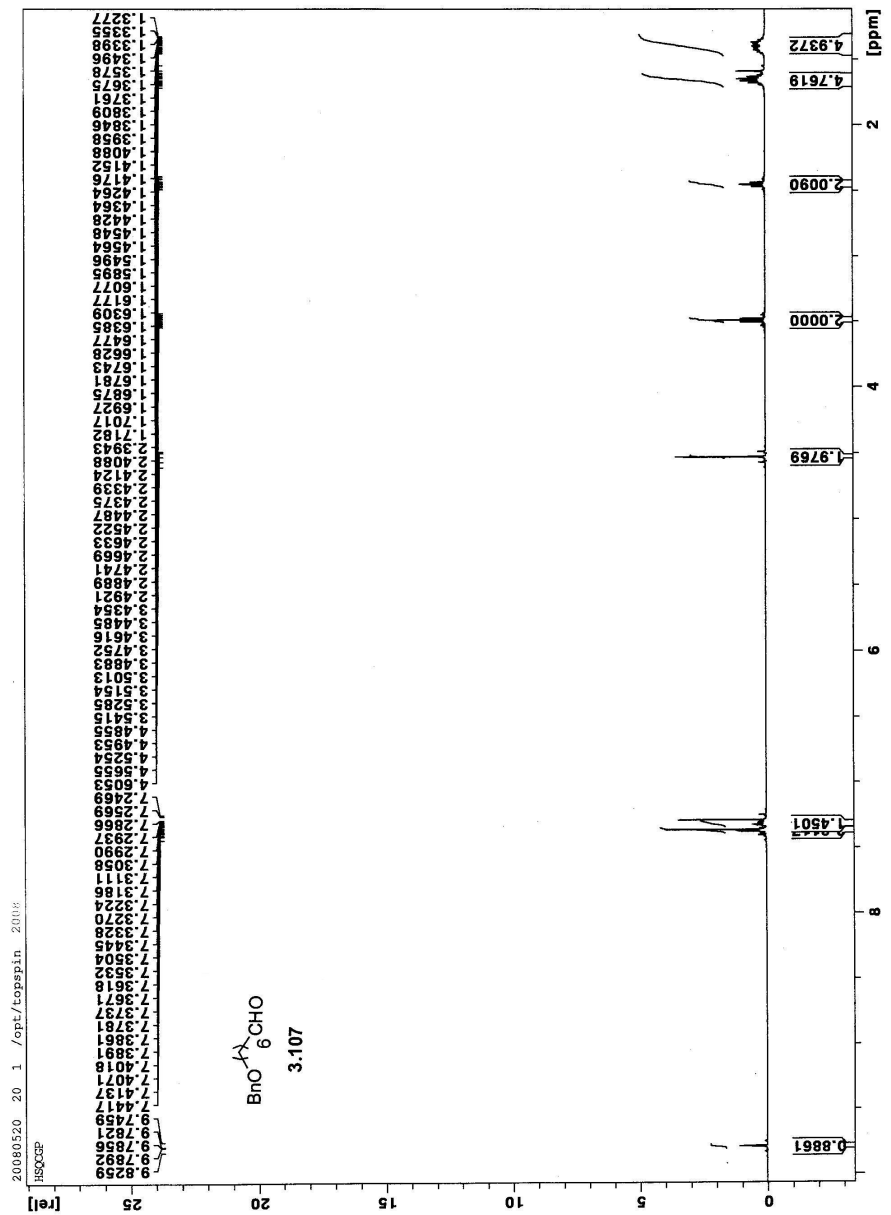


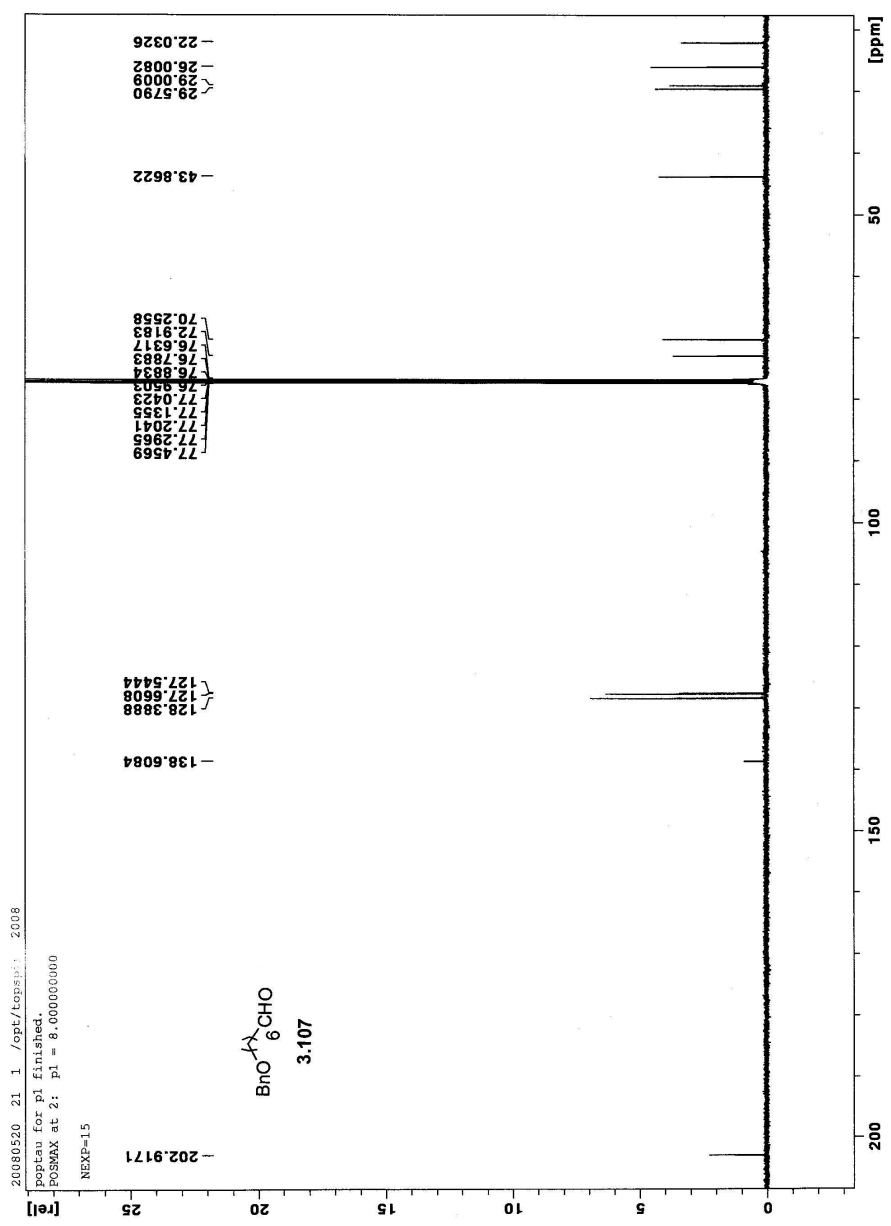


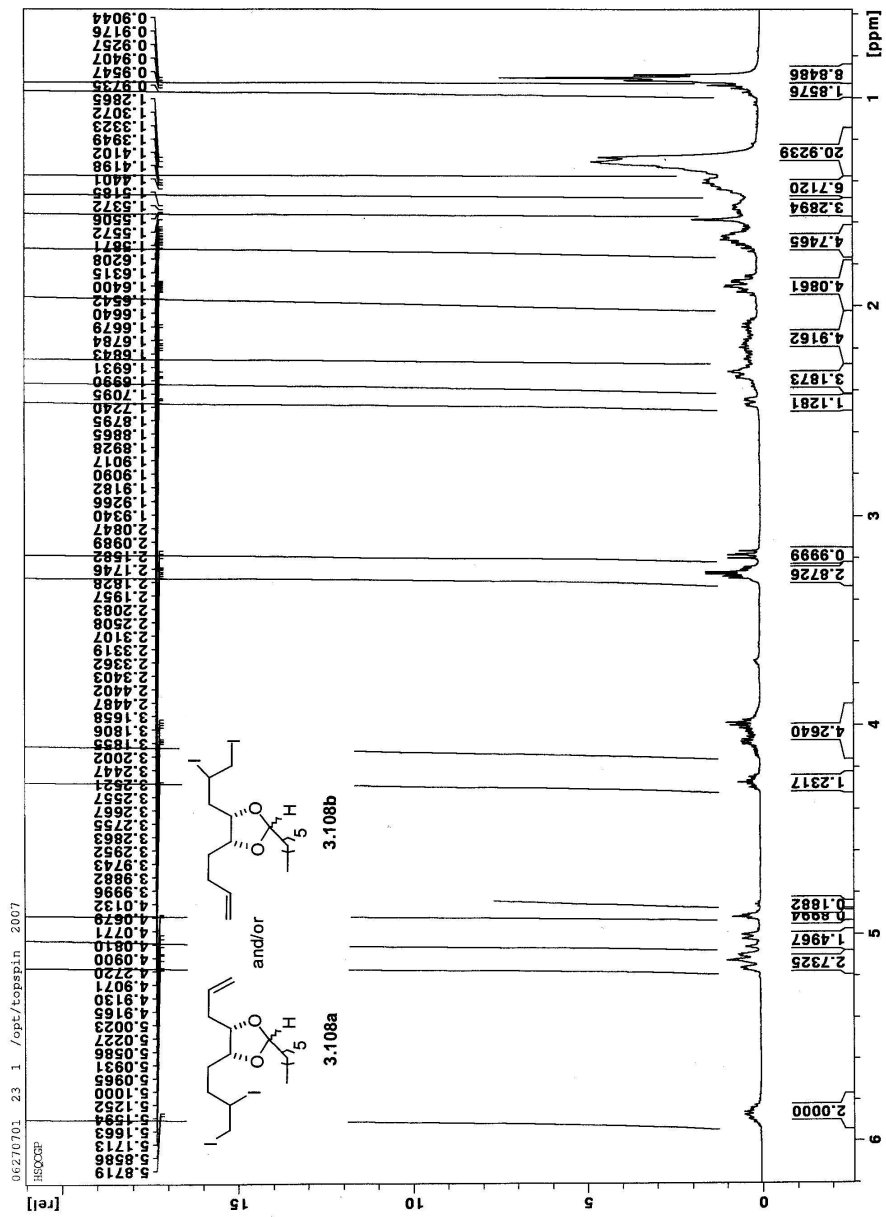


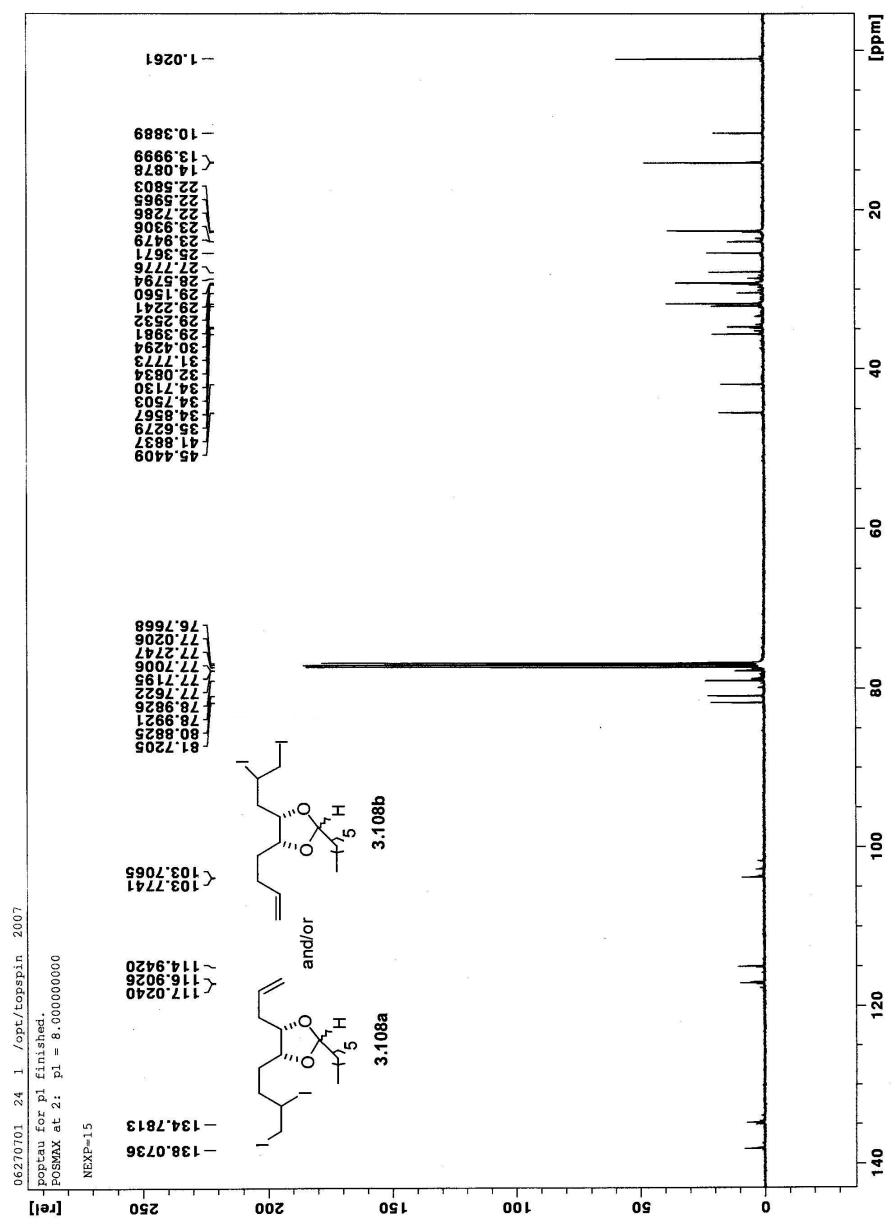


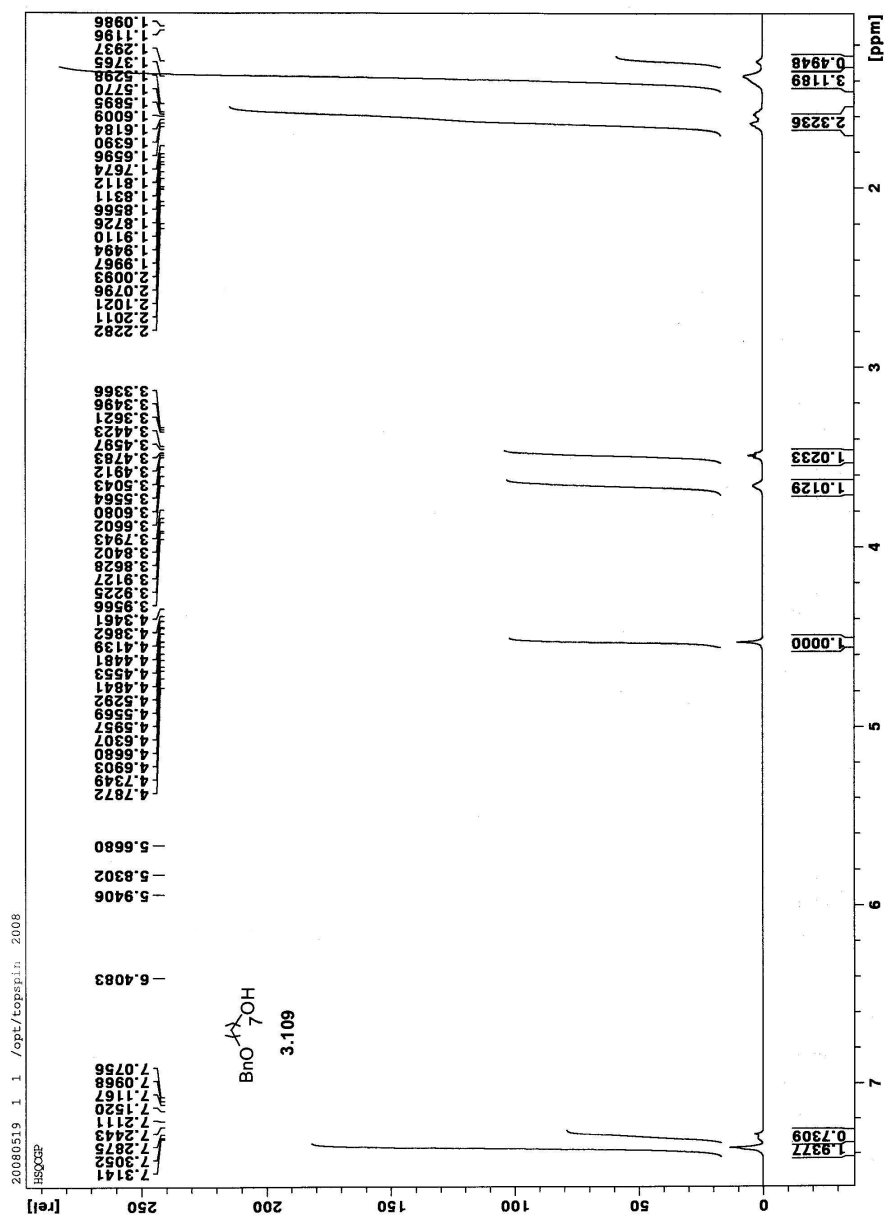


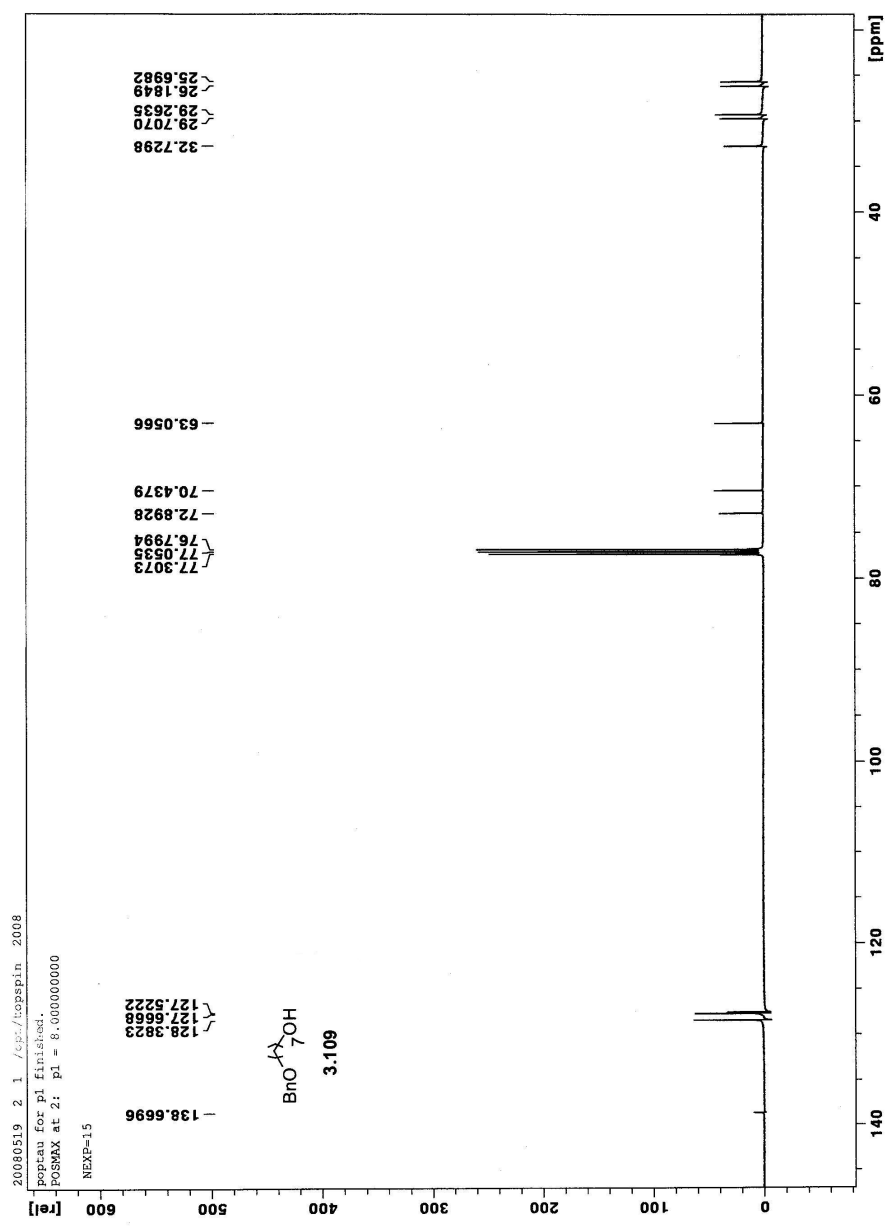


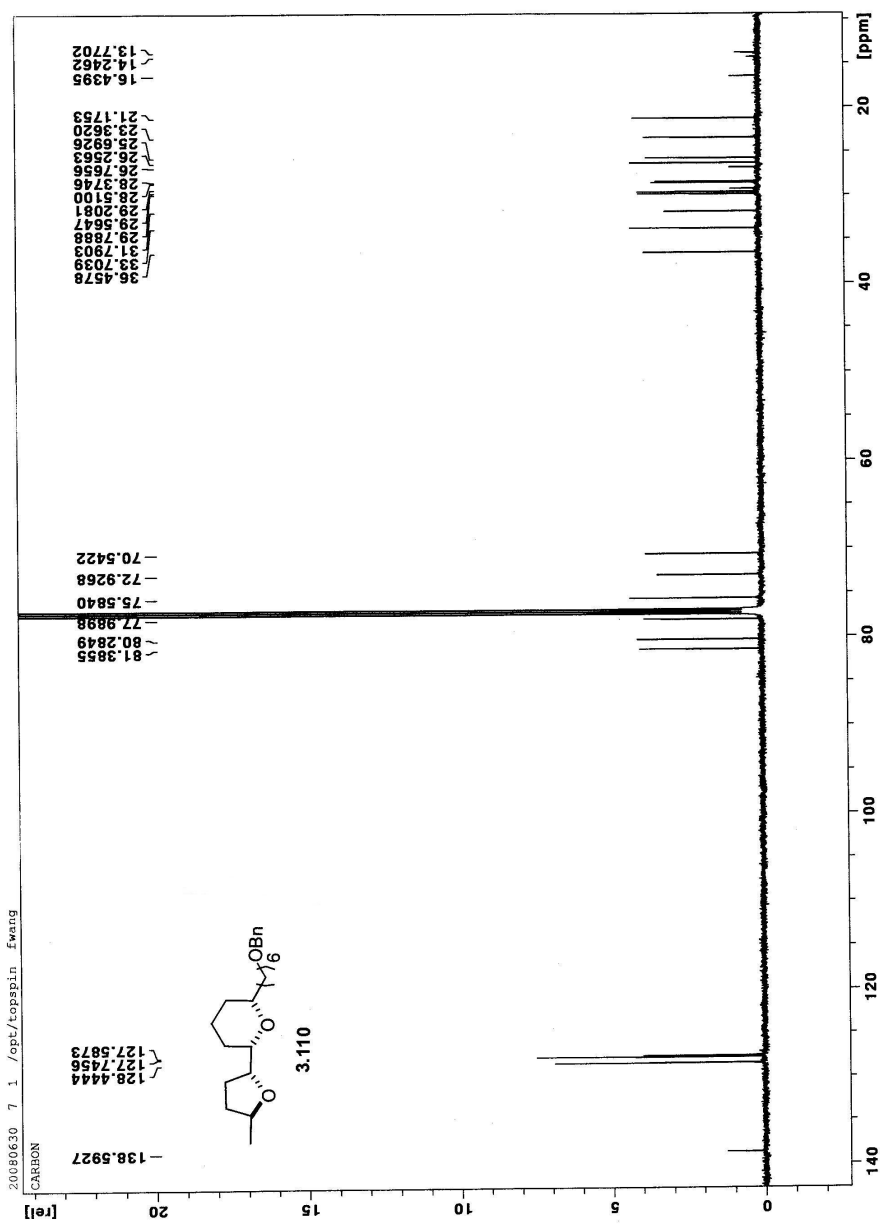


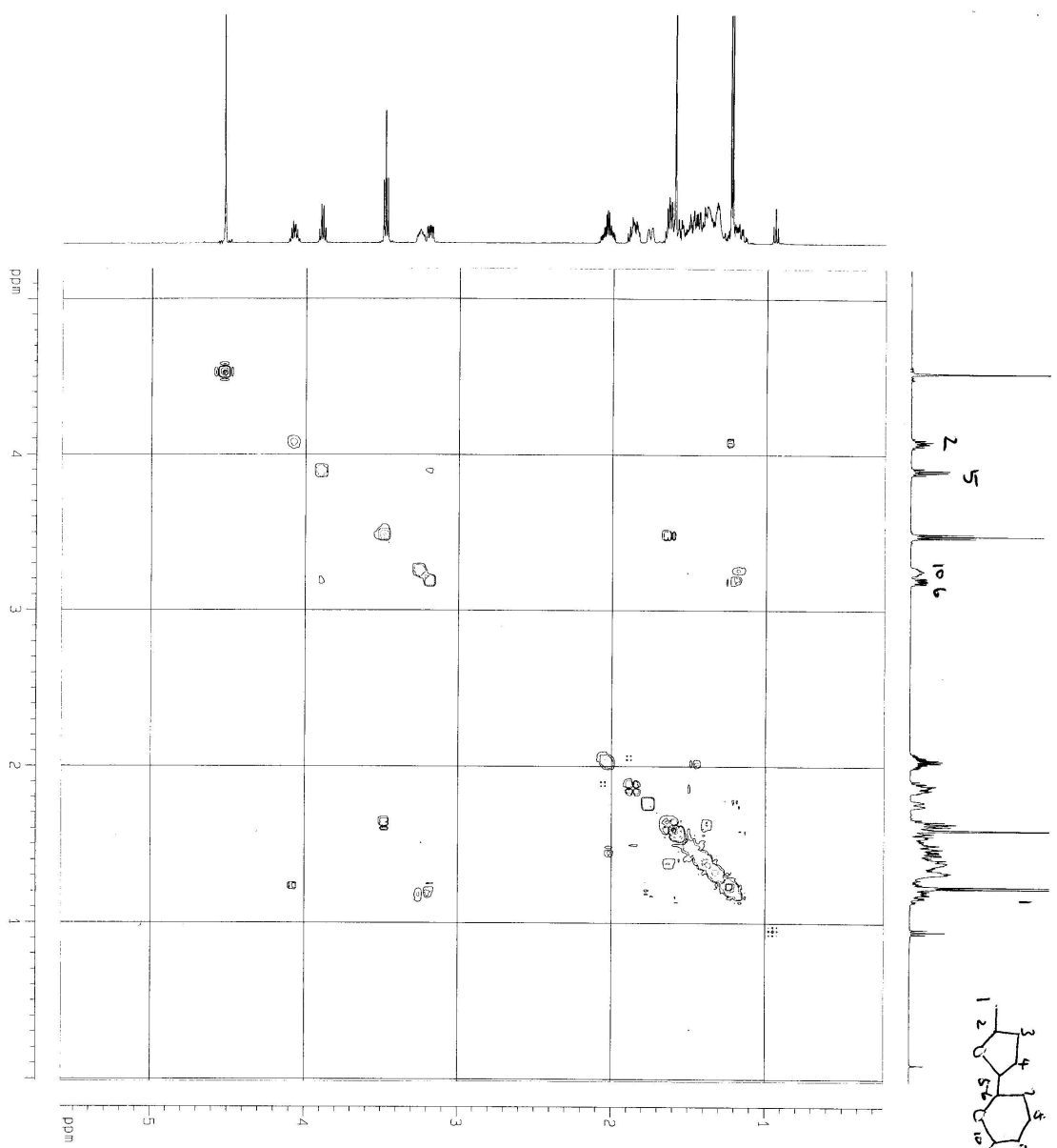




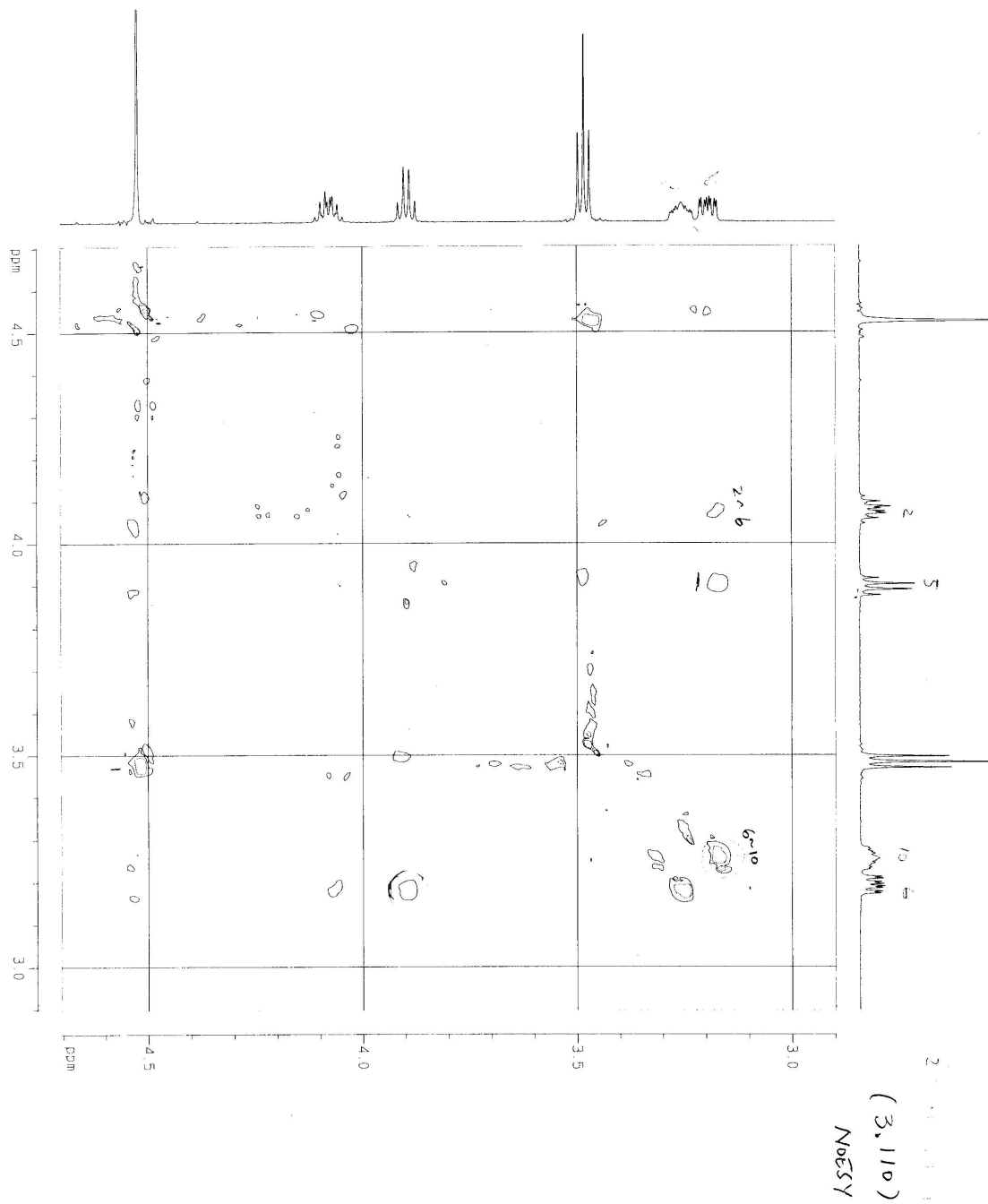


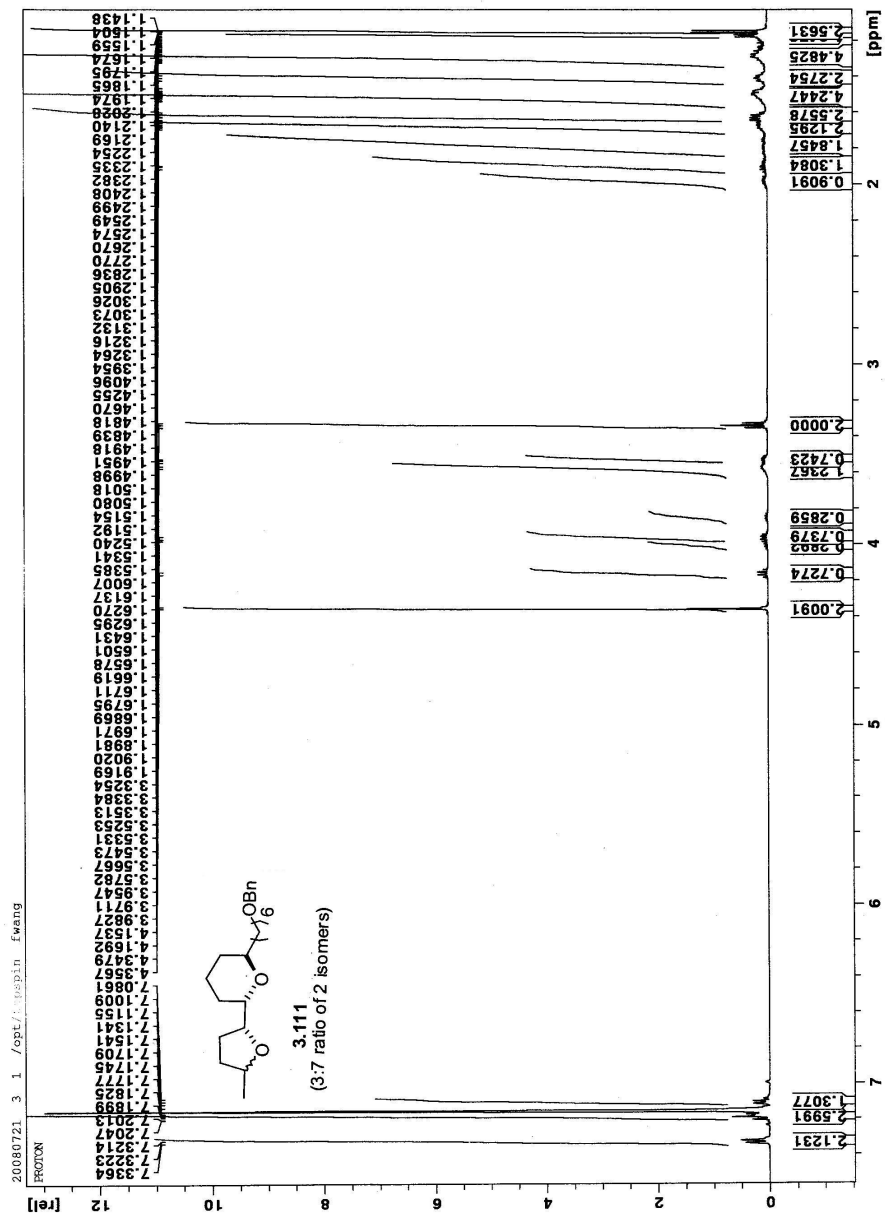


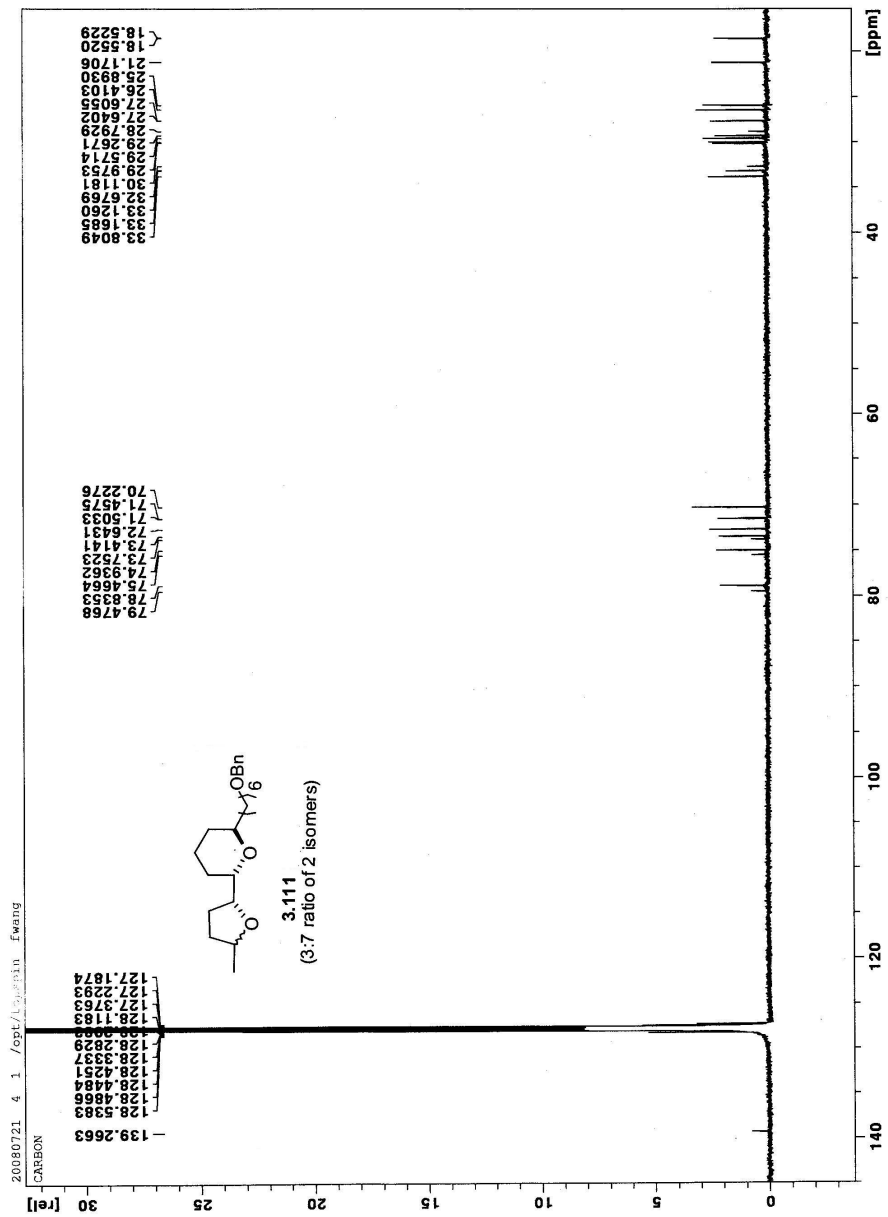


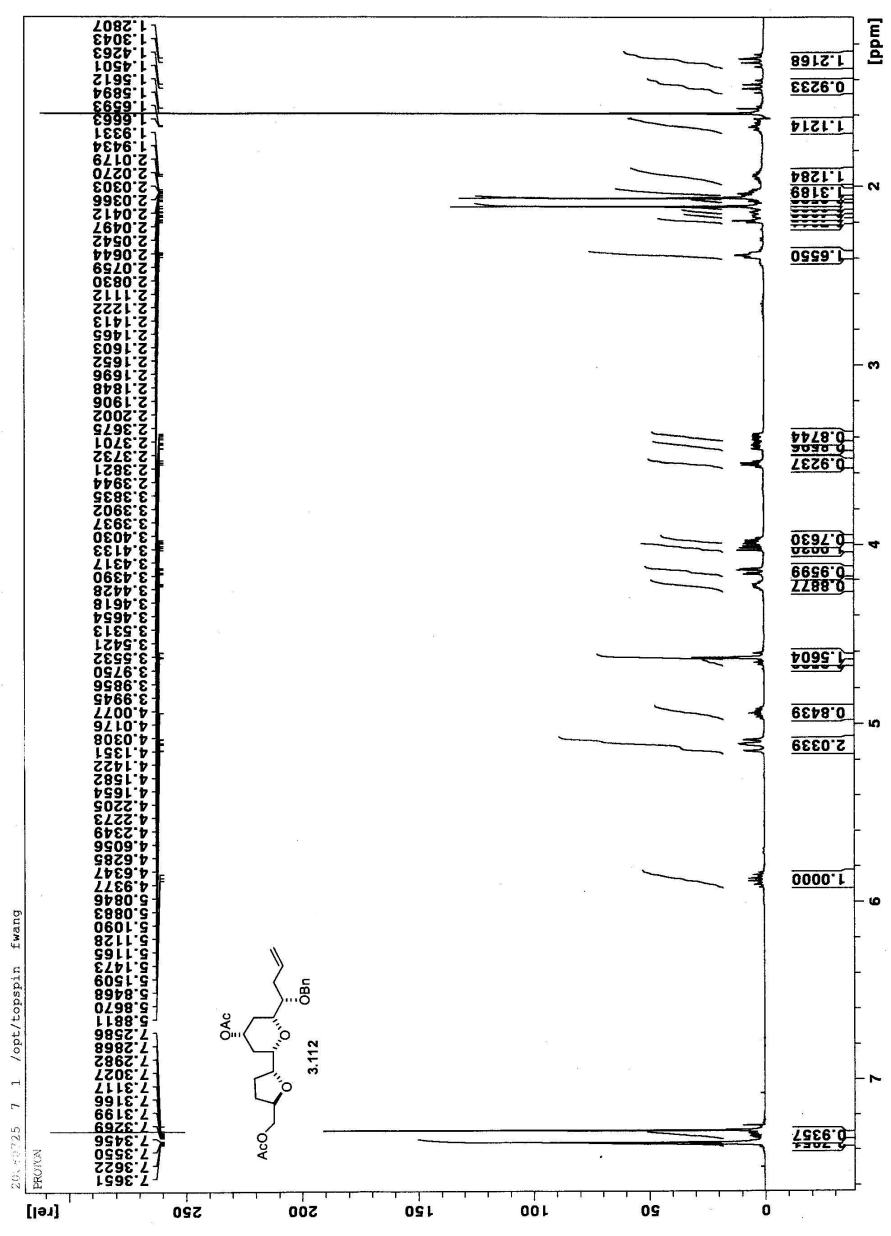


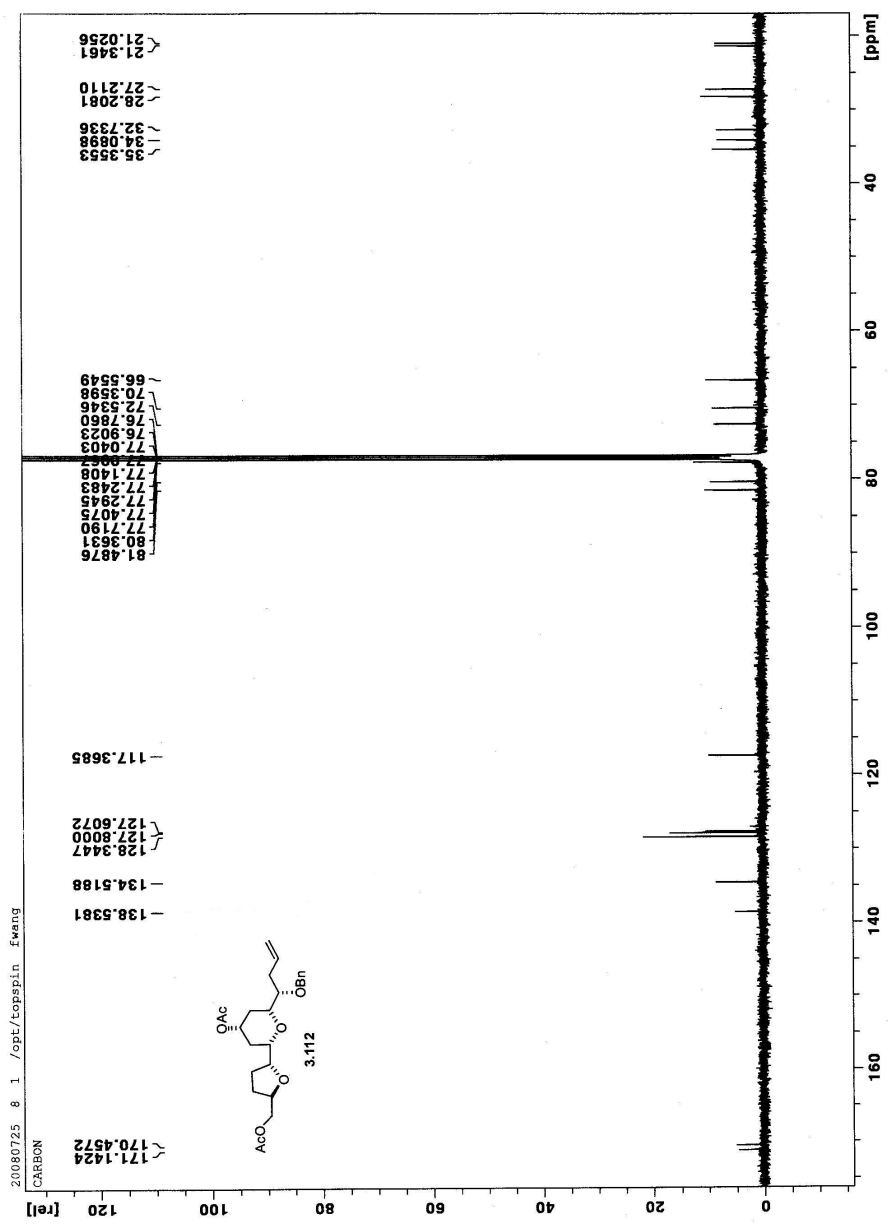
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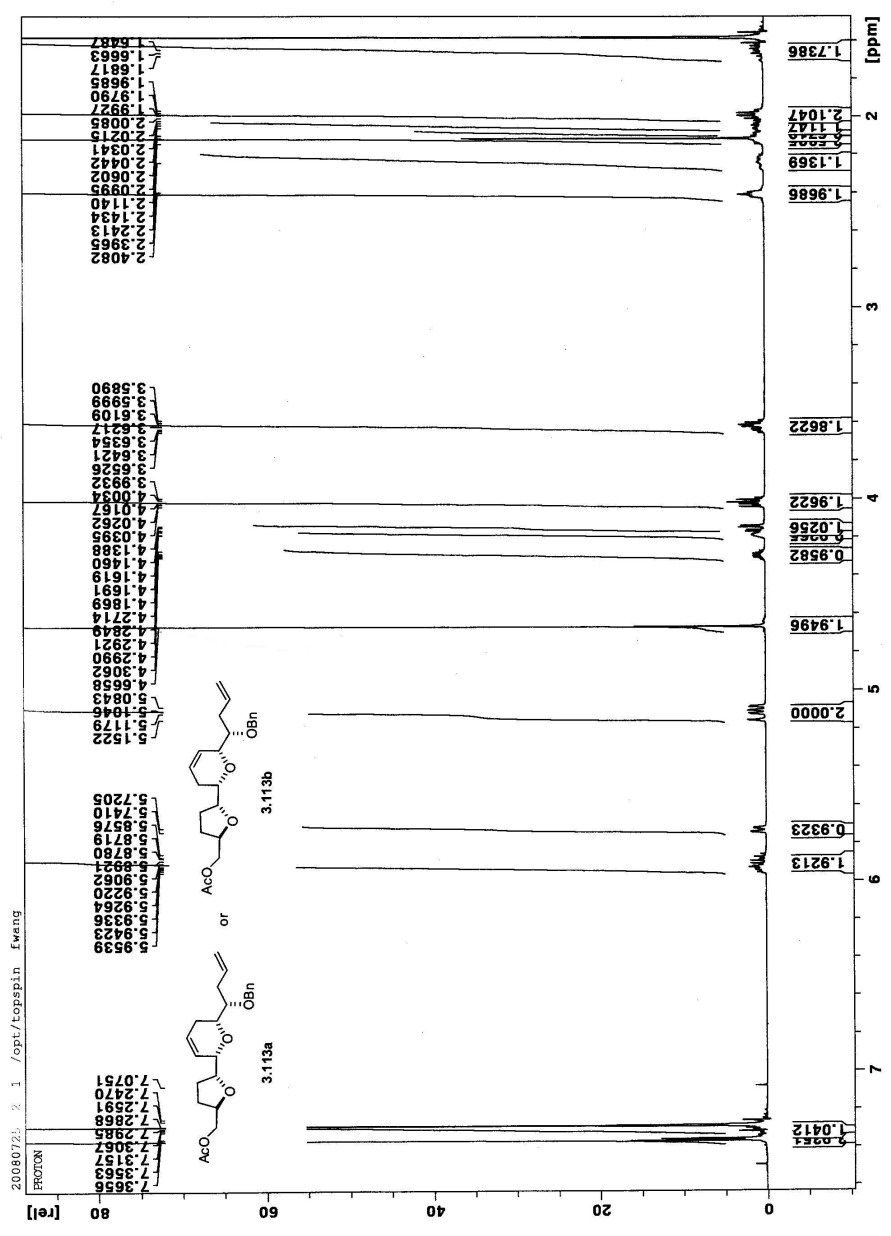


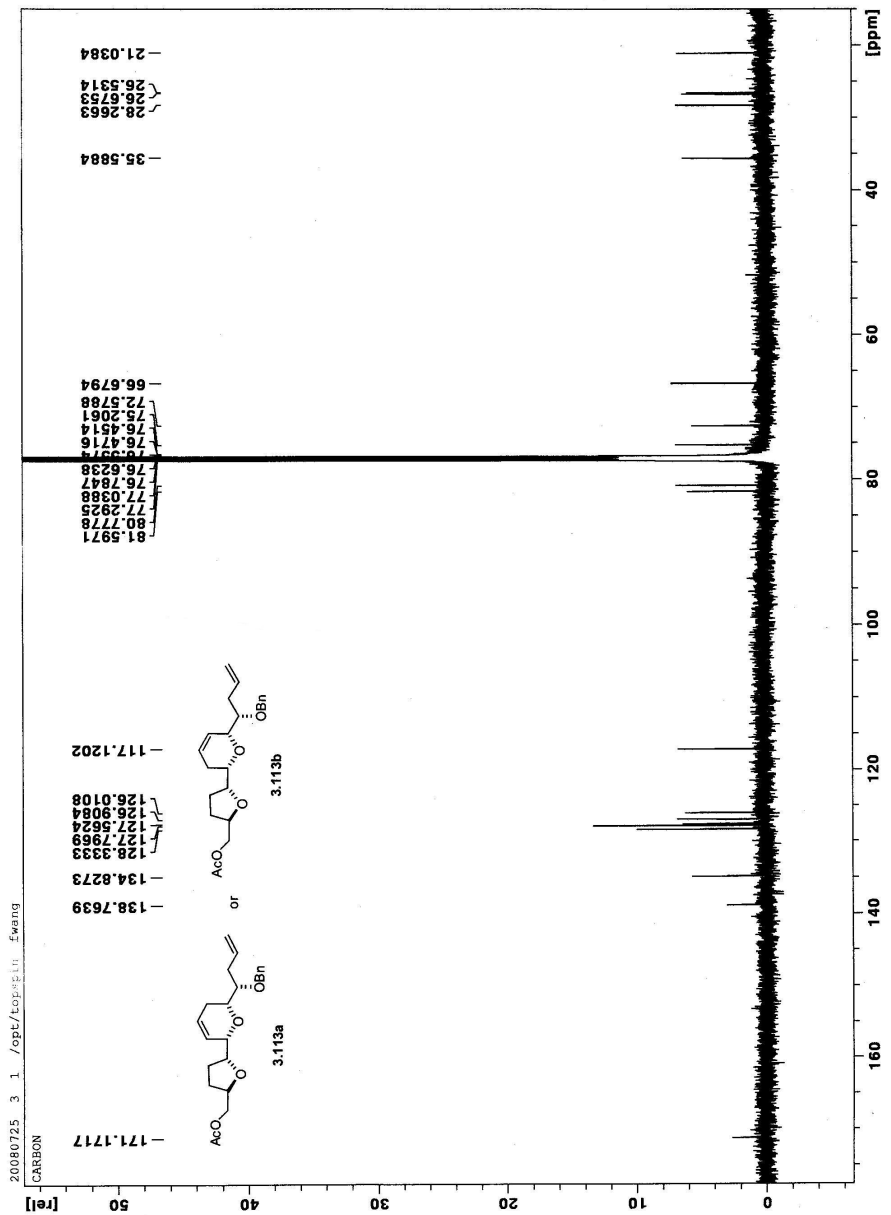


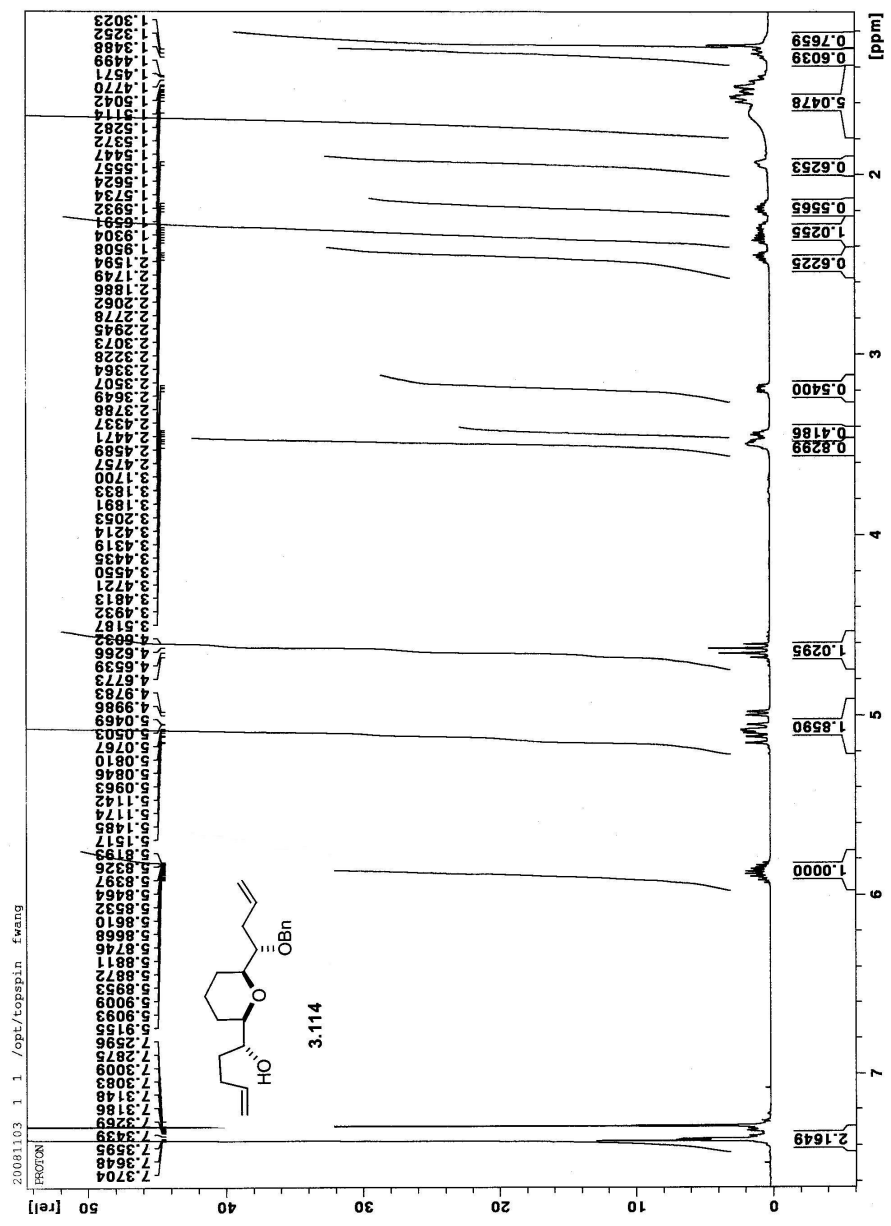


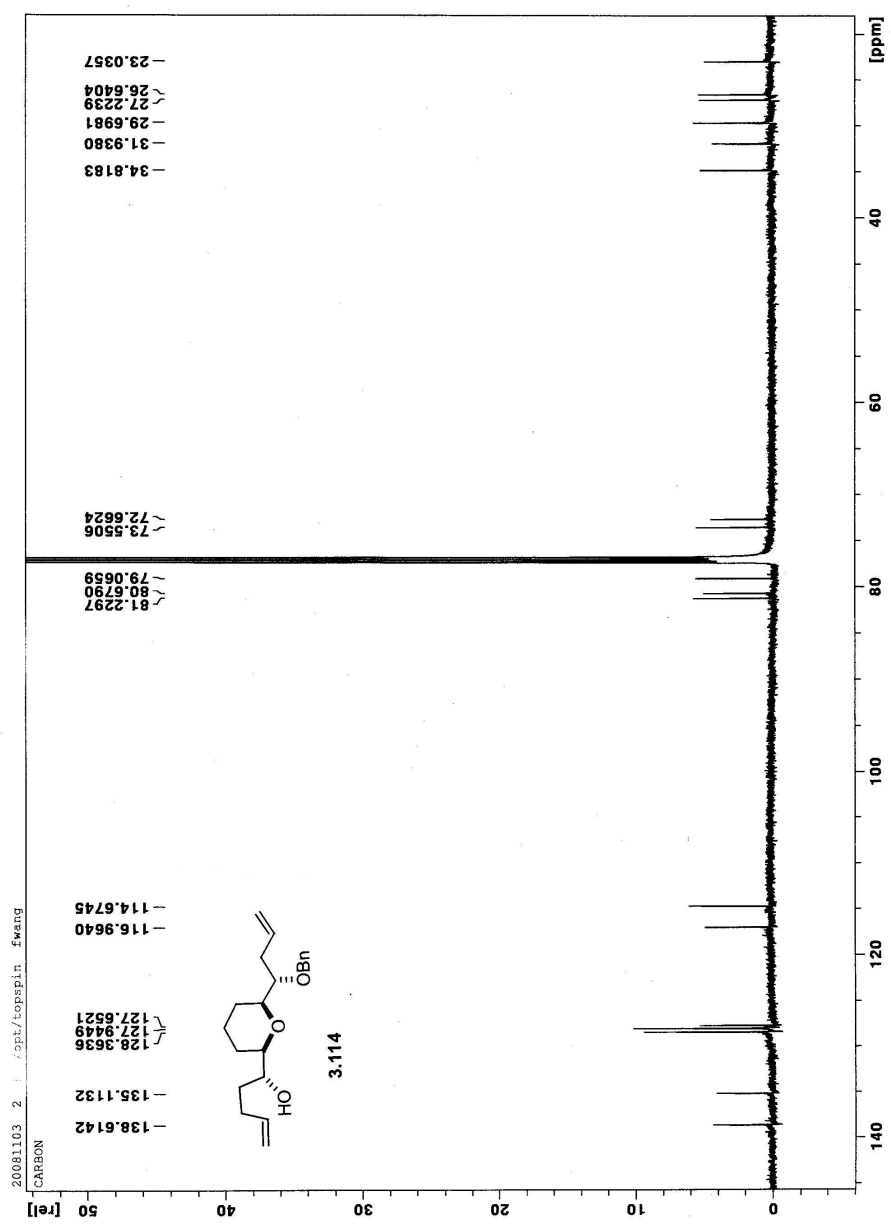


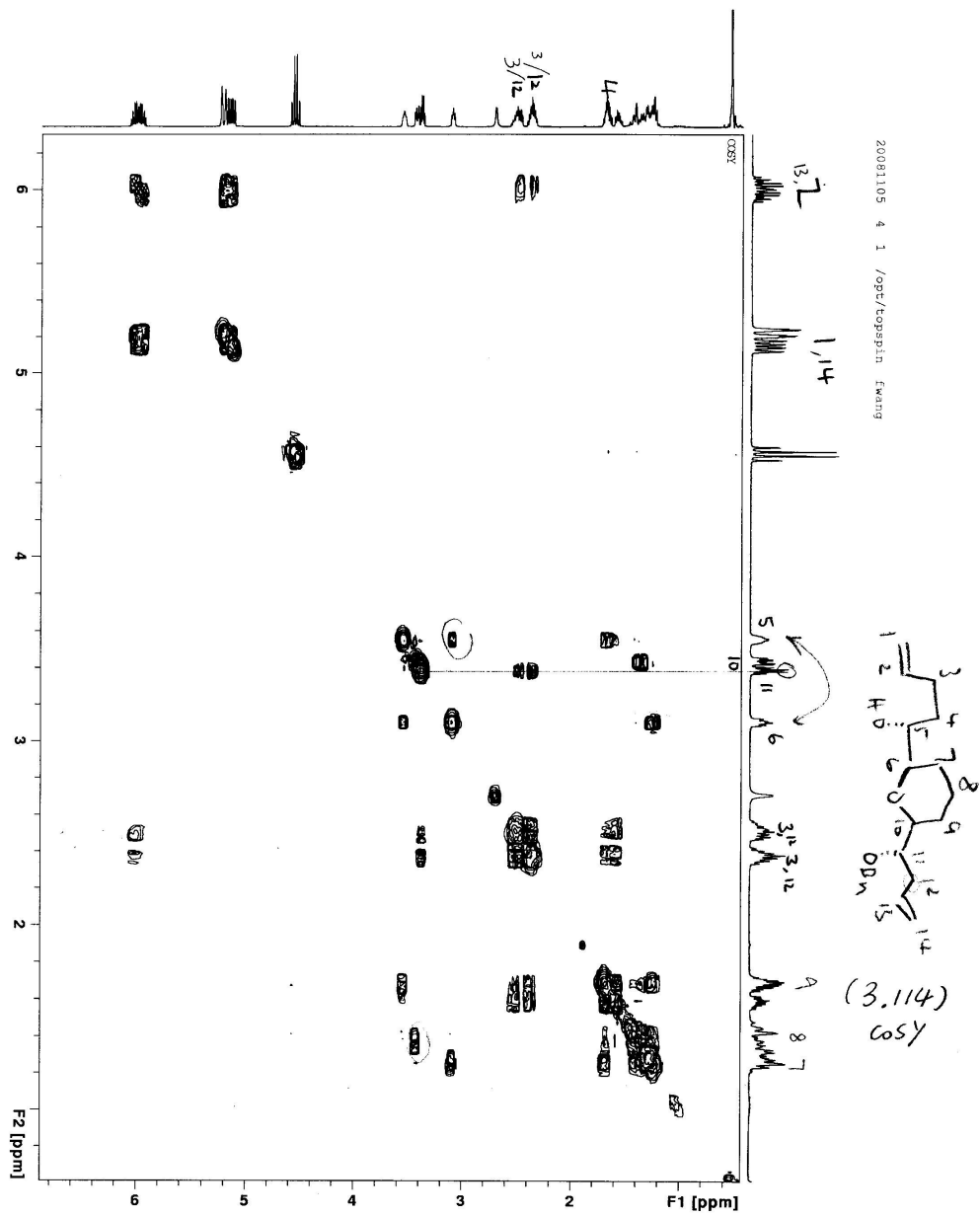


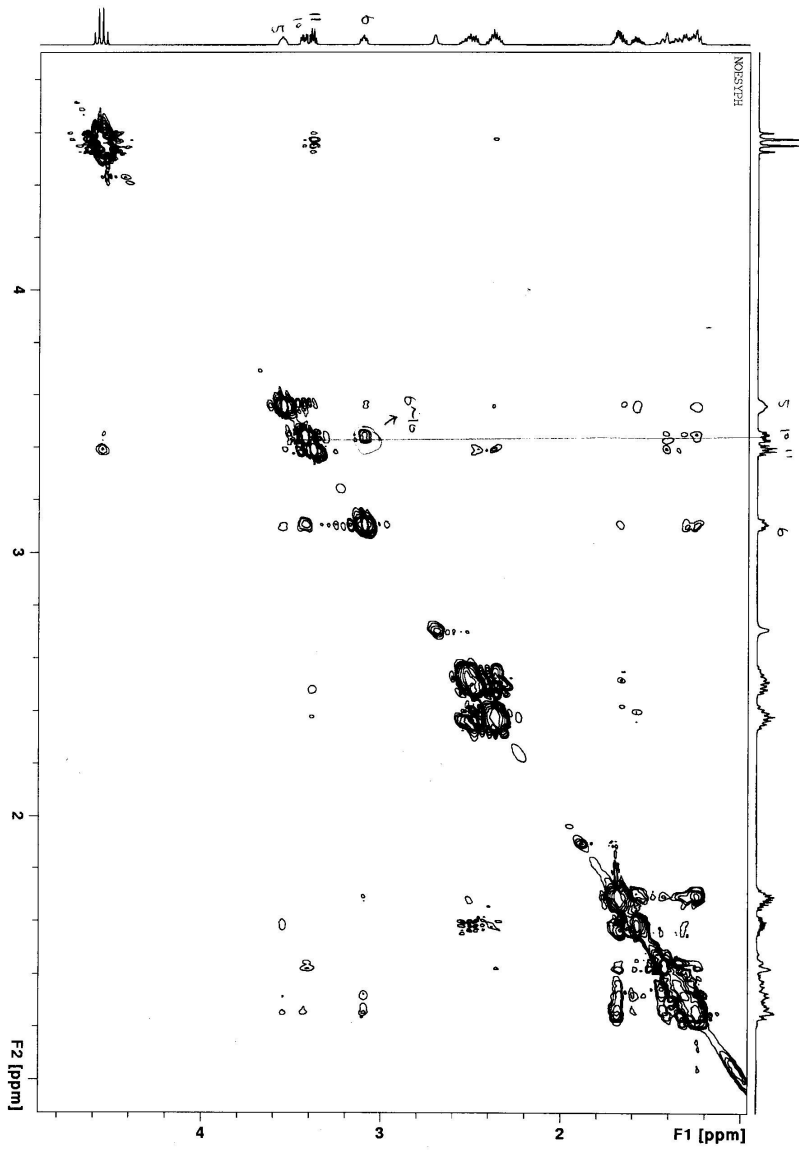






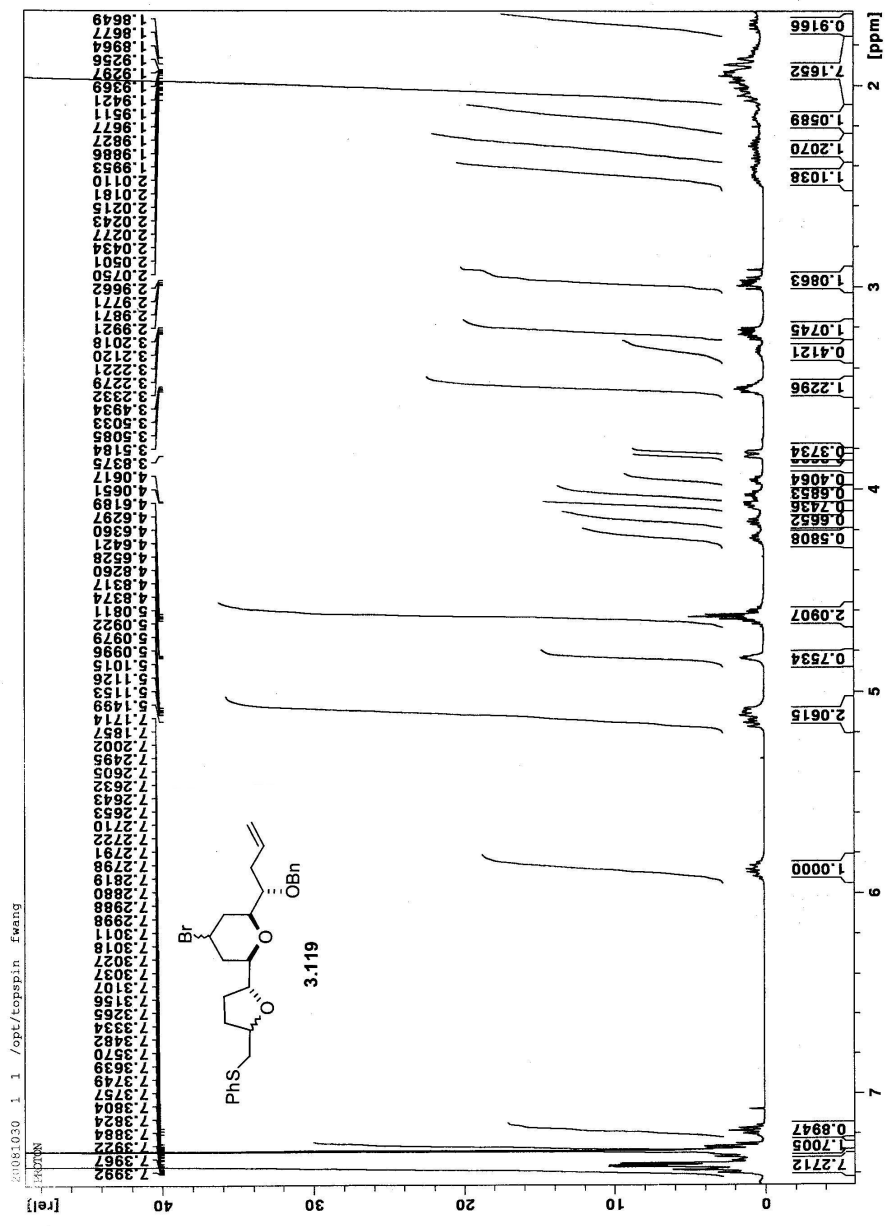


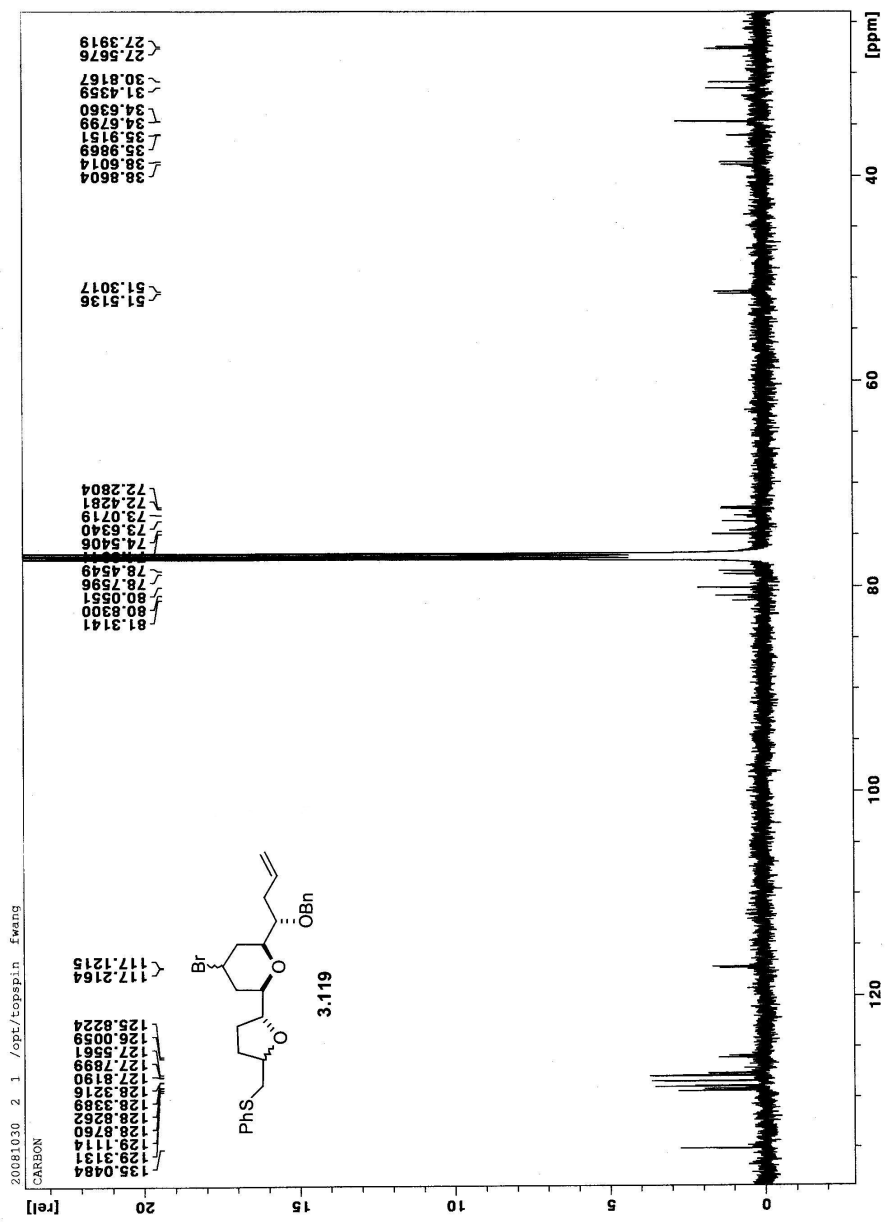


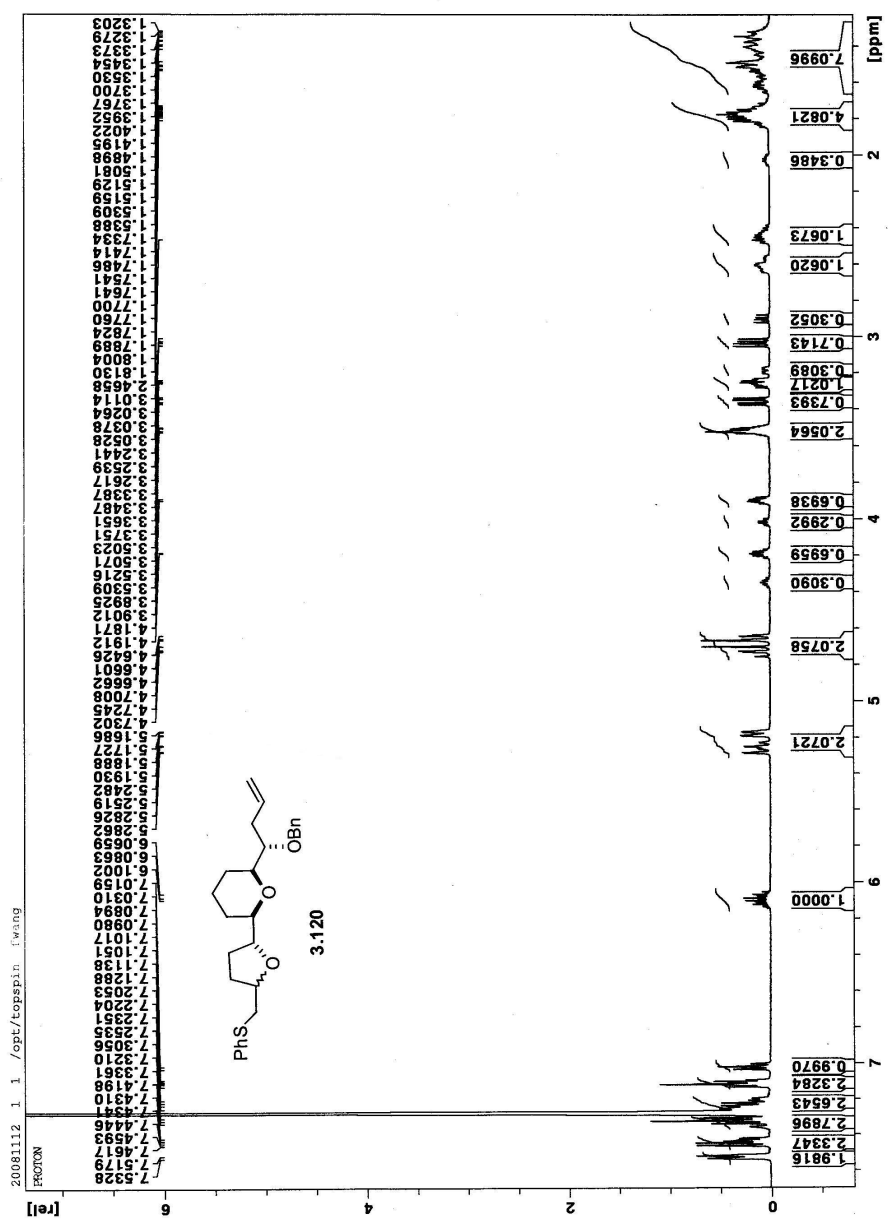


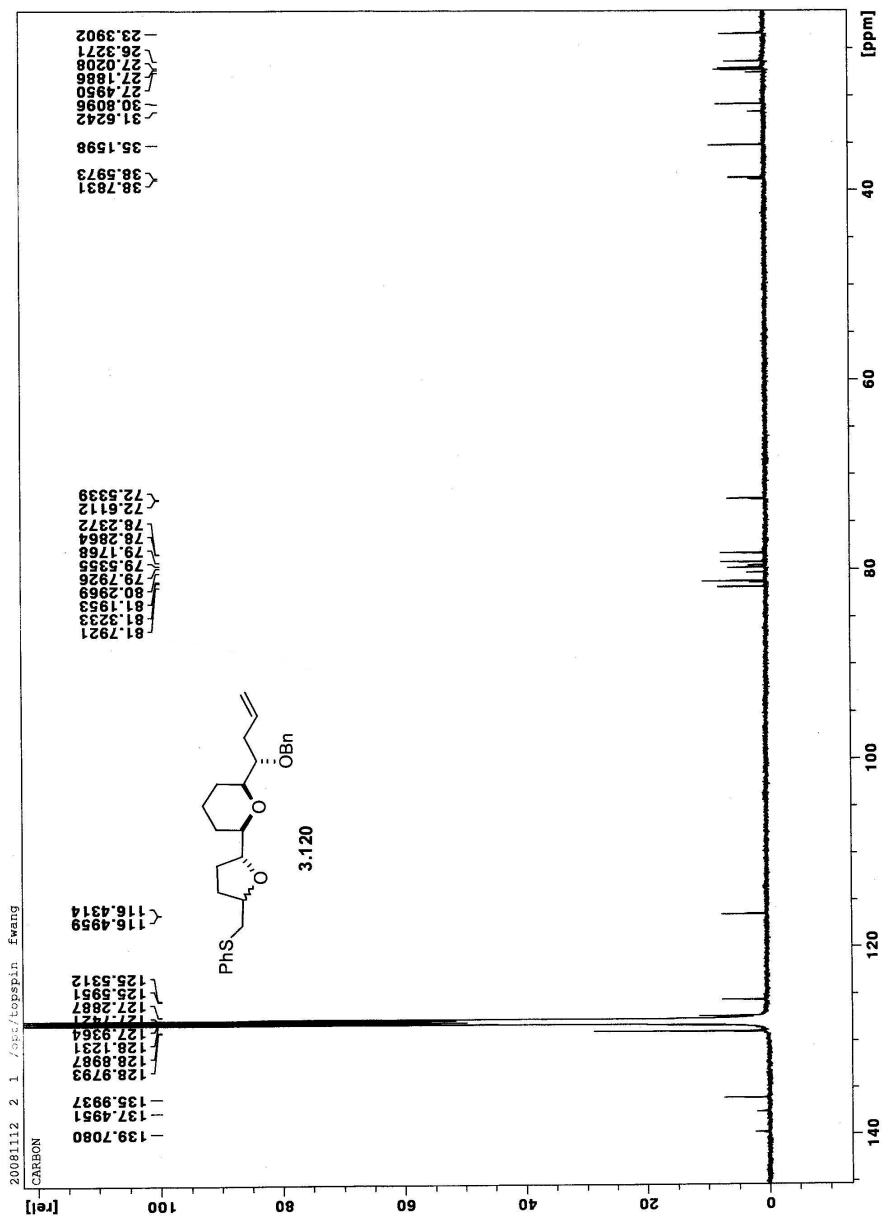
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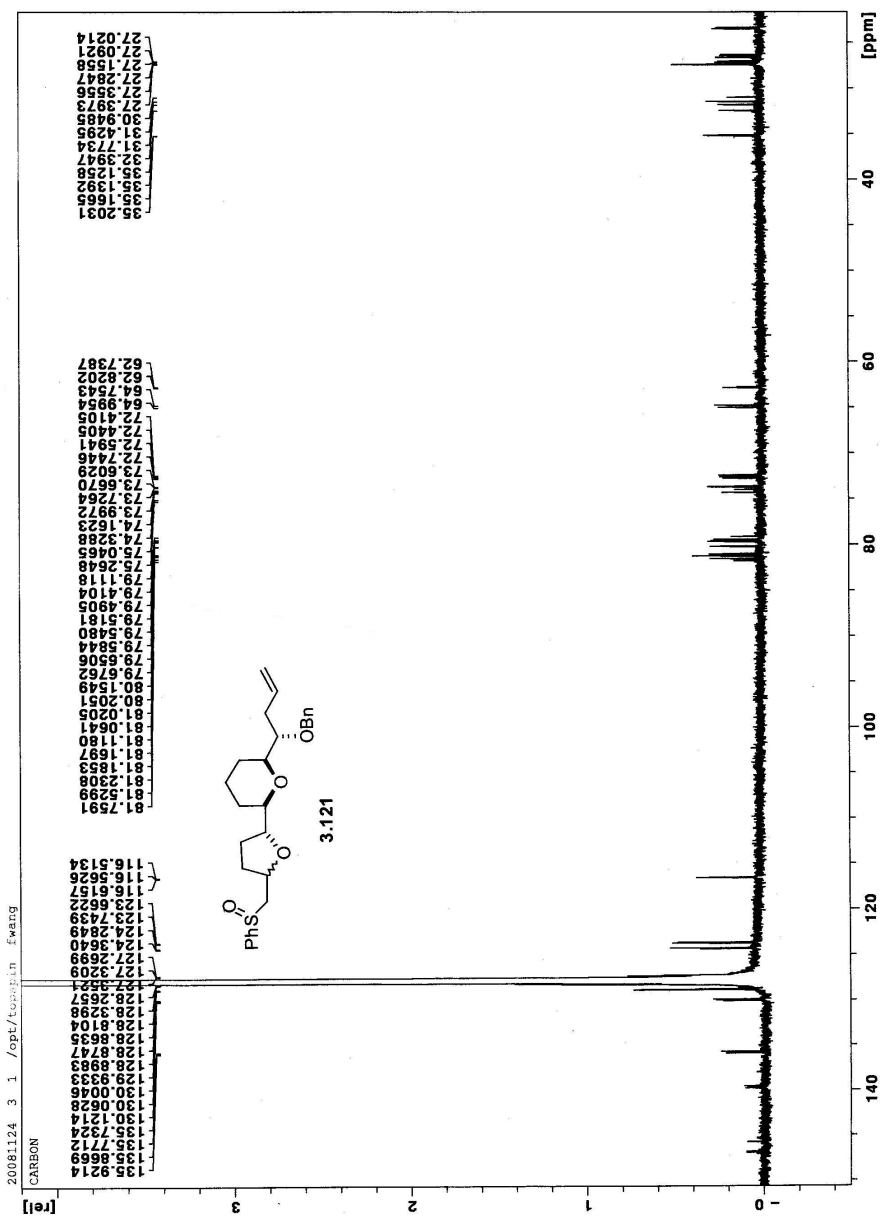
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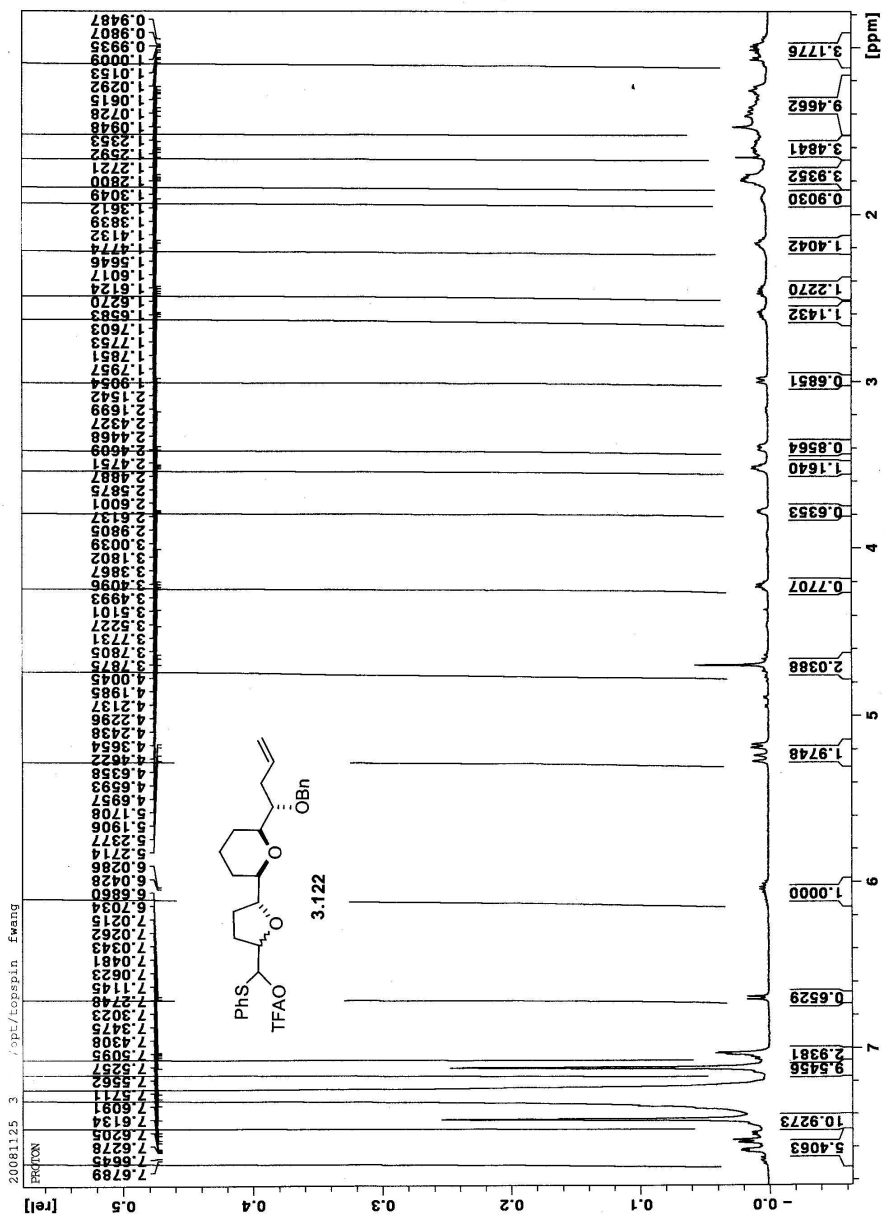


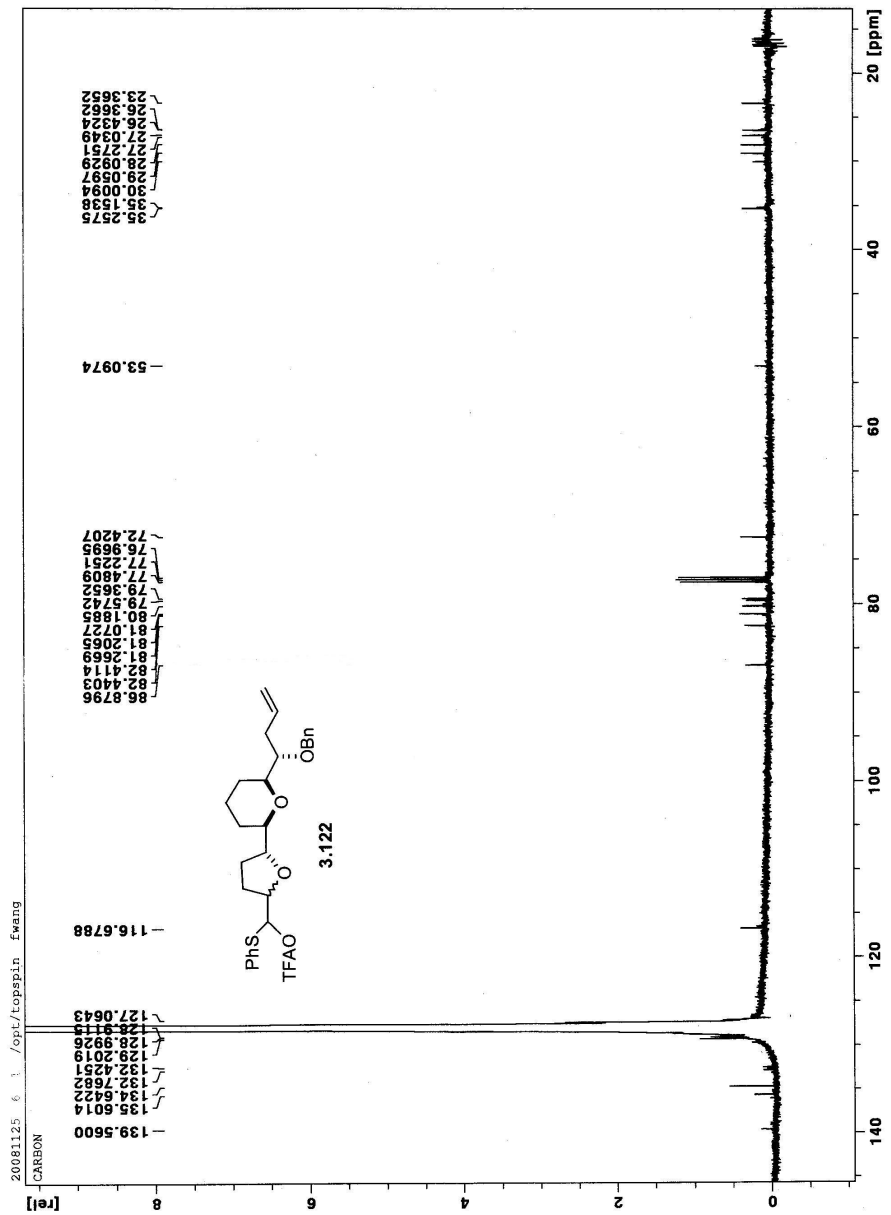












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